

# Supplementary Materials: Co-amorphization of Kanamycin with Amino acids Improves Aerosolization

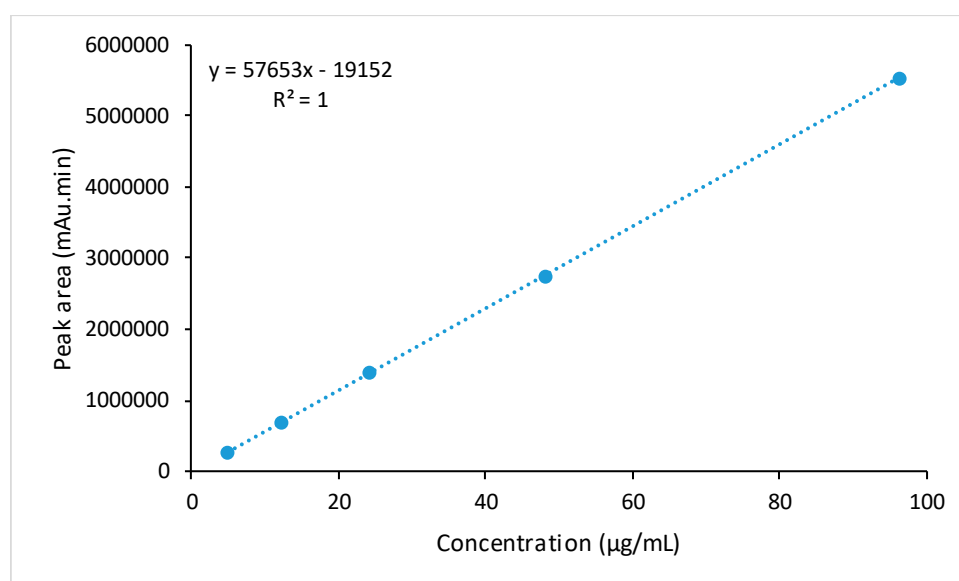
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**Table S1** IR band positions (selected) in kanamycin-methionine co-amorphous system (KM), kanamycin-valine co-amorphous system (KV), kanamycin-phenylalanine co-amorphous system (KP), kanamycin-tryptophan co-amorphous system (KT), physical mixture of amorphous kanamycin and amorphous tryptophan (PM), amorphous kanamycin, amorphous tryptophan.

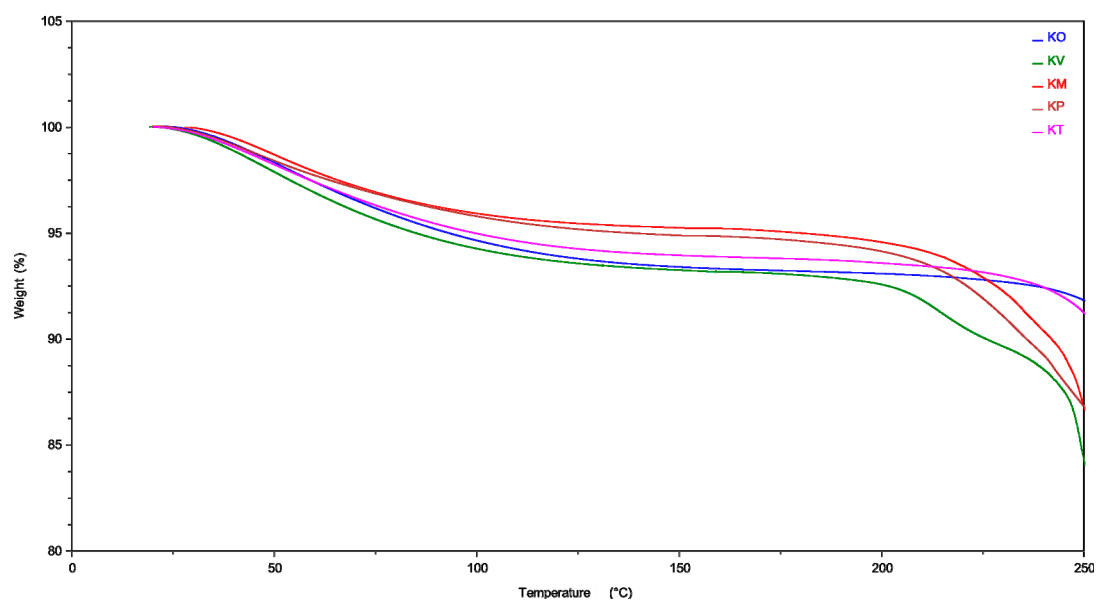
KM [IR bands( $\text{cm}^{-1}$ )]	KV [IR bands( $\text{cm}^{-1}$ )]	KP [IR bands( $\text{cm}^{-1}$ )]	KT [IR bands( $\text{cm}^{-1}$ )]	PM [IR bands( $\text{cm}^{-1}$ )]	Amorphous Kanamycin [IR bands( $\text{cm}^{-1}$ )]	Amorphous Tryptophan [IR bands( $\text{cm}^{-1}$ )]
			427	423		424 (Benzene ring deformation vibration)
			460	457		461 (Pyrrole ring deformation vibration)
609	609	609	608	608	608 (Sulfate ion bending)	
			749	742		740 (Benzene/ pyrrole ring deformation vibration)
1031	1030	1031	1033	1032	1029 (Sulfate ion stretching)	
1338	1332	1336	1342	1340	1337 (C-H bending)	1339 (C-H bending)
			1402	1397		1396 (-COOH sym. stretching)
1521	1520	1521	1522	1519	1528 (-NH <sub>2</sub> /NH <sub>3</sub> <sup>+</sup> sym. bending)	1490 (-NH <sub>2</sub> sym. bending)
1602	1600	1602	1602	1607	1605 (-NH <sub>2</sub> /NH <sub>3</sub> <sup>+</sup> asym. bending)	1601 (-COOH asym. stretch/ -NH <sub>2</sub> asym. bend/ Benzene ring stretching)

**Table S2** Summary of the particles size analysis of the formulations. Values are expressed as mean  $\pm$  standard deviation.

Formulation	SEM ( $\mu\text{m}$ ) (n=300)
Kanamycin Only (KO)	1.1 $\pm$ 0.5
Kanamycin-valine (KV)	1.1 $\pm$ 0.5
Kanamycin-methionine (KM)	1.0 $\pm$ 0.5
Kanamycin-phenylalanine (KP)	1.2 $\pm$ 0.5
Kanamycin-tryptophan (KT)	1.1 $\pm$ 0.6

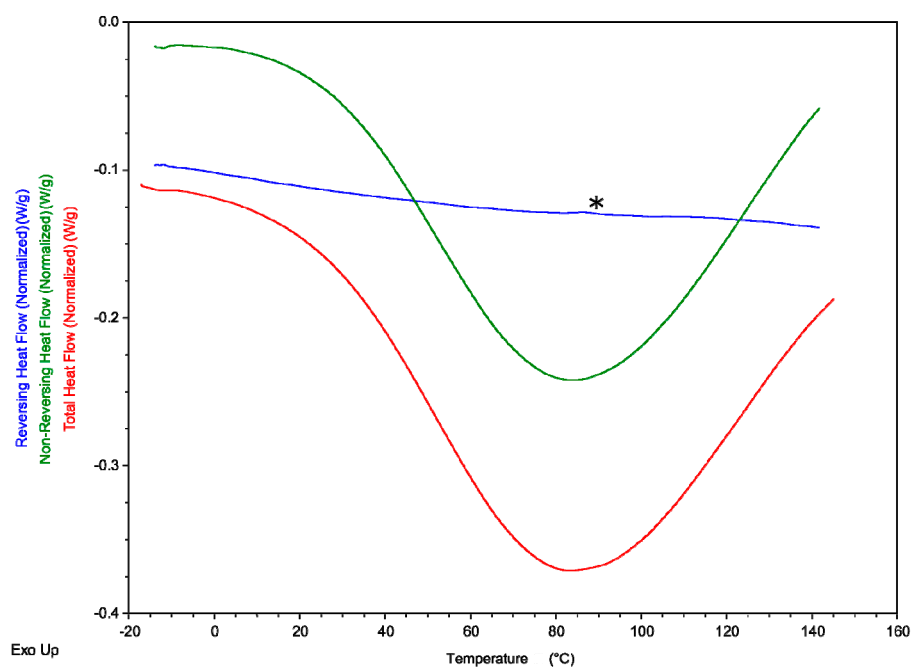


**Figure S1** Representative standard curve of concentration of kanamycin ( $R^2=0.99692$ ).

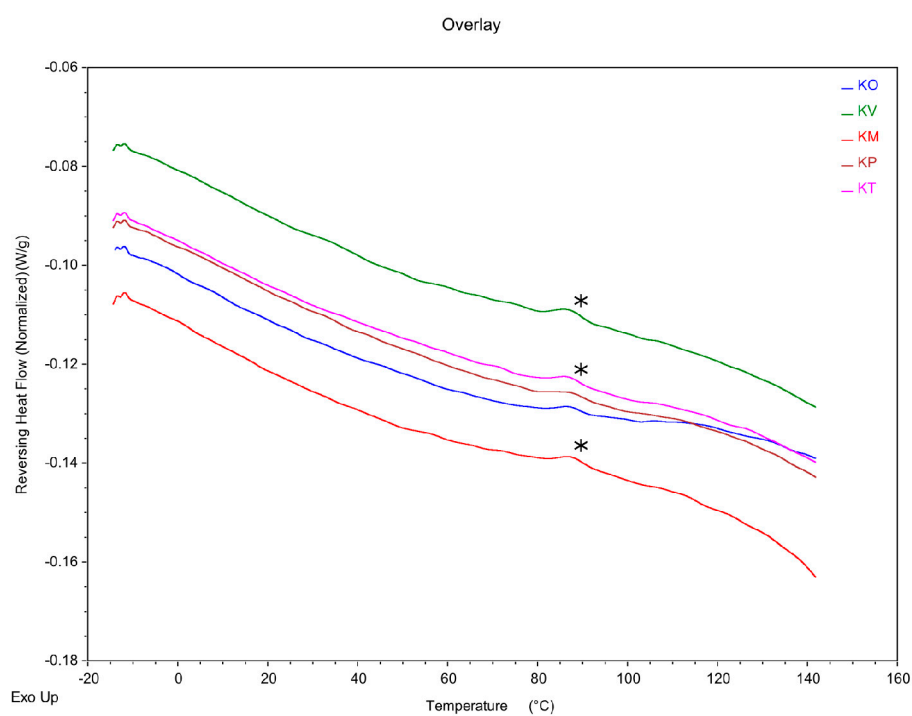


**Figure S2** Representative TGA thermograms of the formulations [kanamycin only (KO), kanamycin-valine (KV), kanamycin-methionine (KM), kanamycin-phenylalanine (KP), and kanamycin-tryptophan (KT)].

(a)



(b)



**Figure S3** DSC thermograms. (a) Representative DSC thermogram of kanamycin only spray-dried particles. (b) Representative MDSC thermograms (reversing heat flow only) of the formulations [kanamycin only (KO), kanamycin-valine (KV), kanamycin-methionine (KM), kanamycin-phenylalanine (KP), and kanamycin-tryptophan (KT)]. [Asterisk (\*) shows the step change associated with glass transition temperature.]

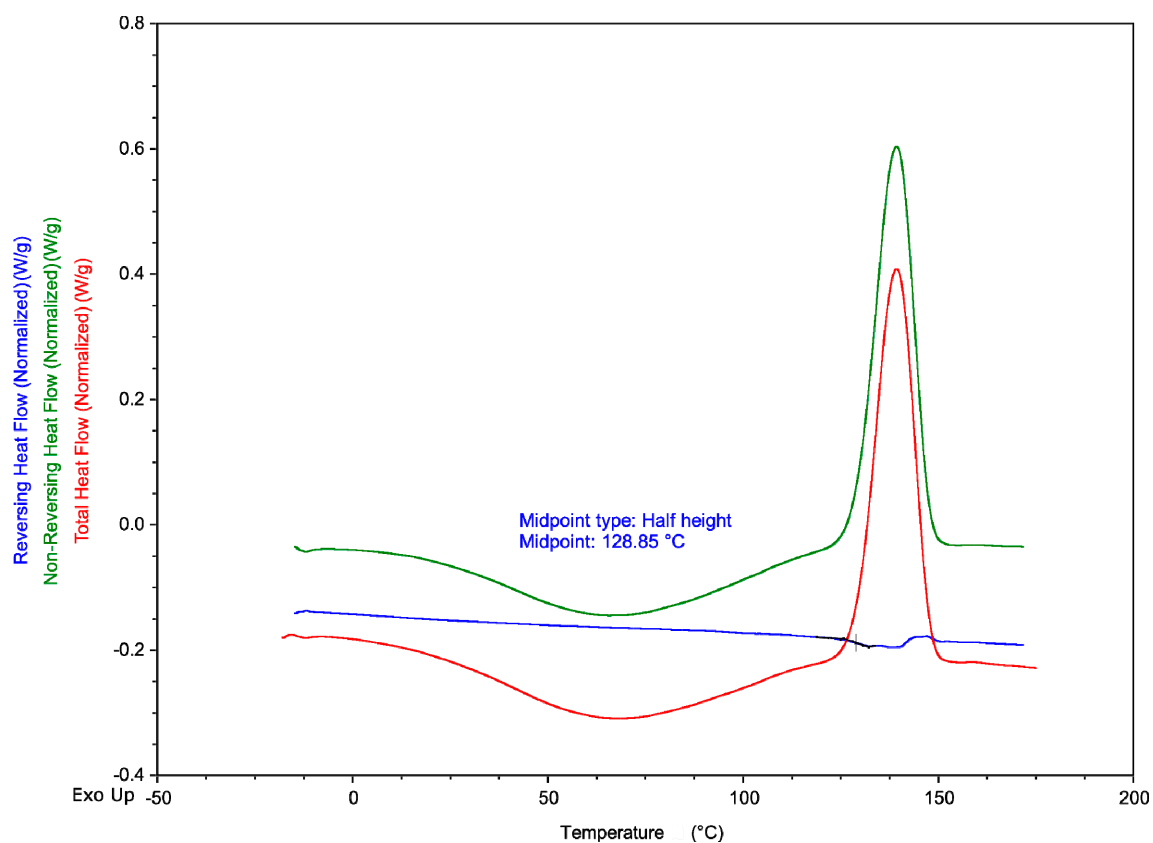


Figure S4 Representative MDSC thermogram of spray-dried tryptophan.

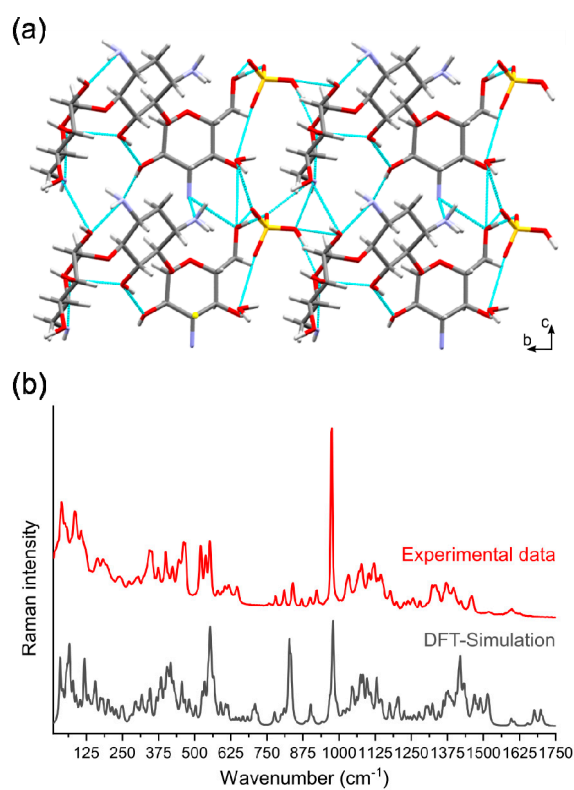
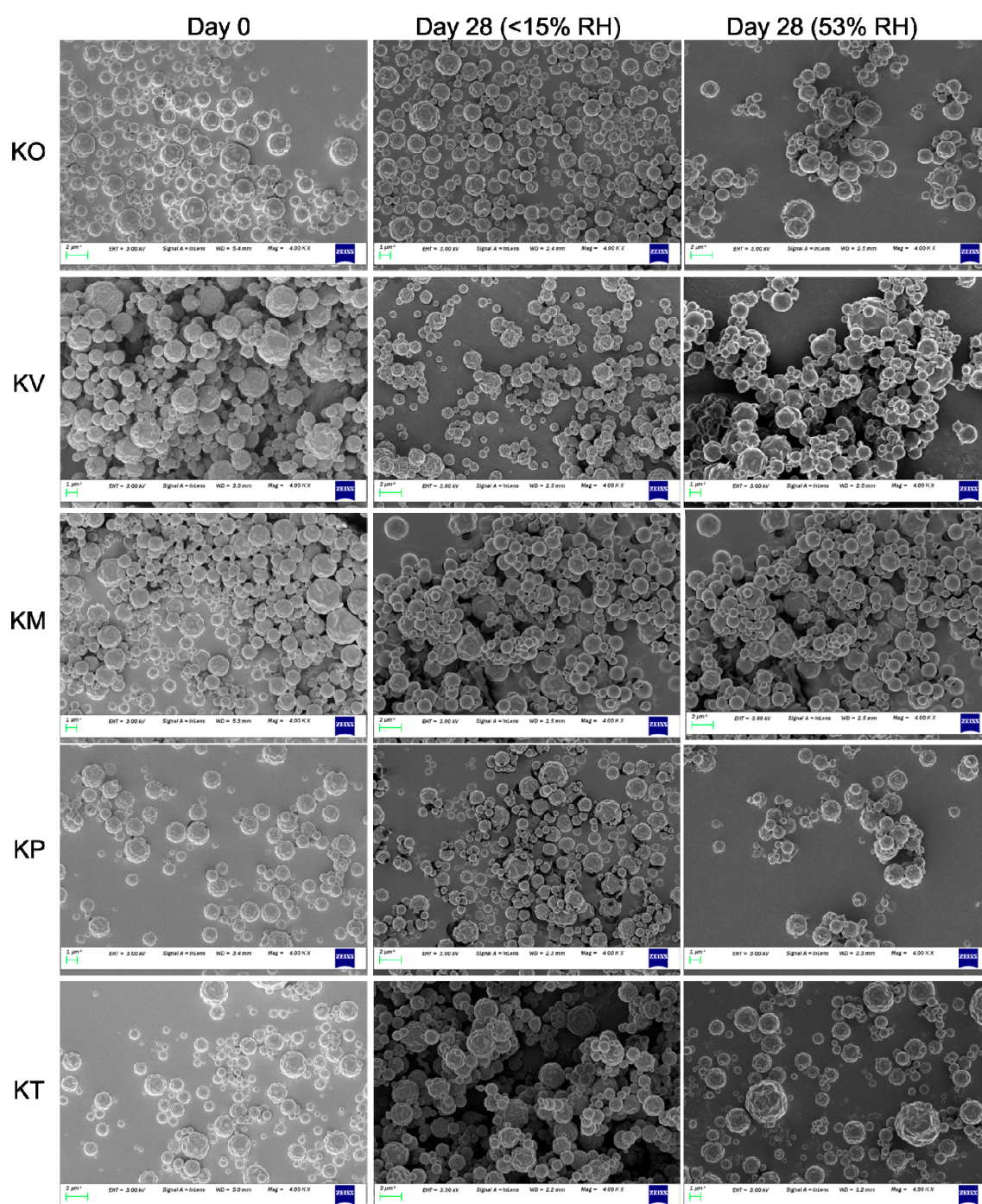
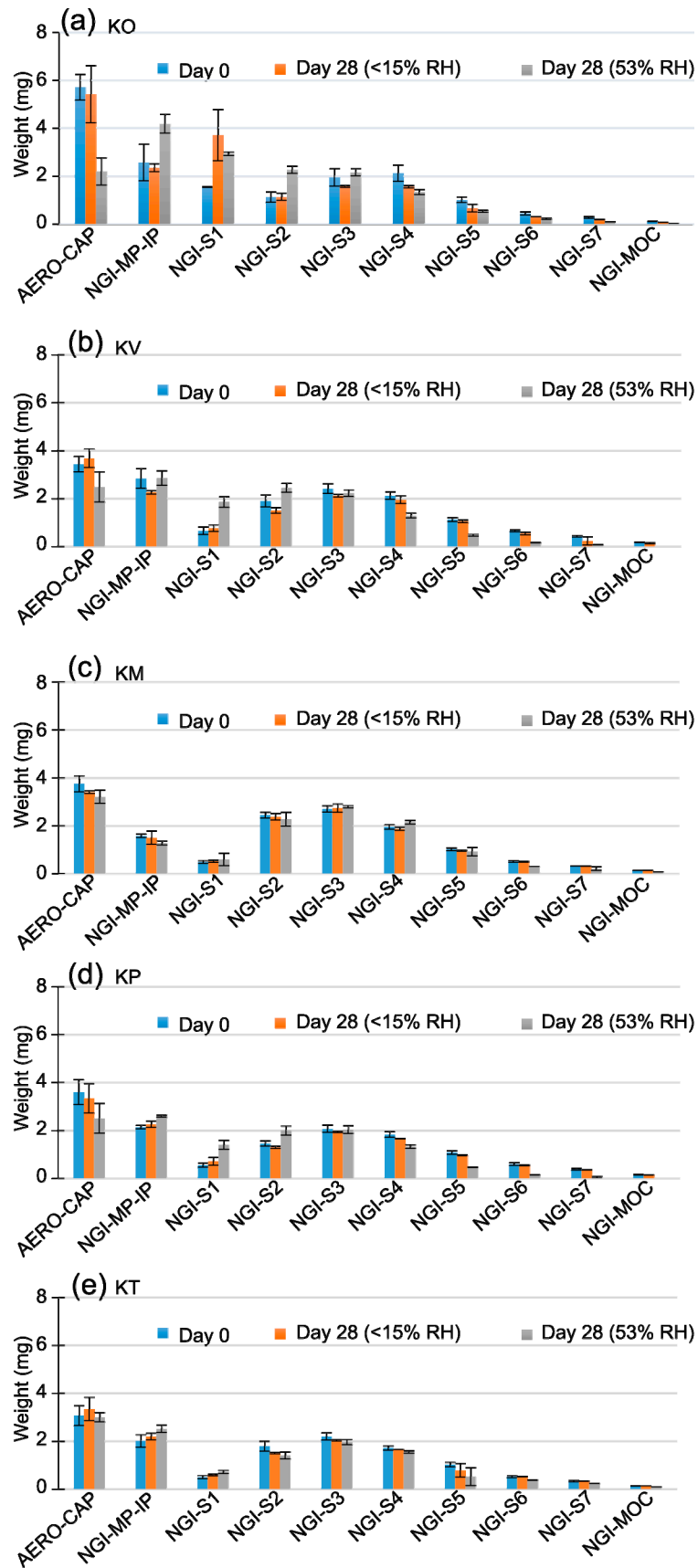


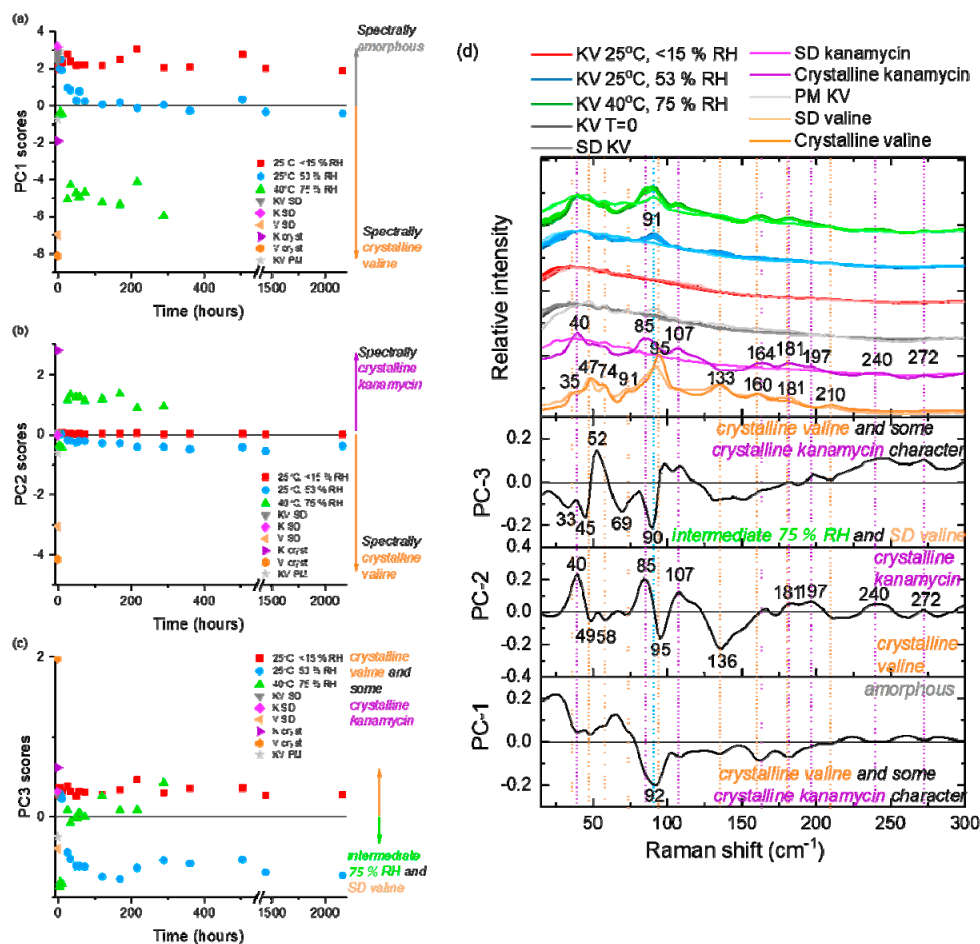
Figure S5 (a) View of the kanamycin sulfate monohydrate packing and different hydrogen-bonding patterns along the crystallographic *a*-axis. (b) Experimental and DFT-simulated Raman spectra of crystalline kanamycin sulfate (monohydrate).



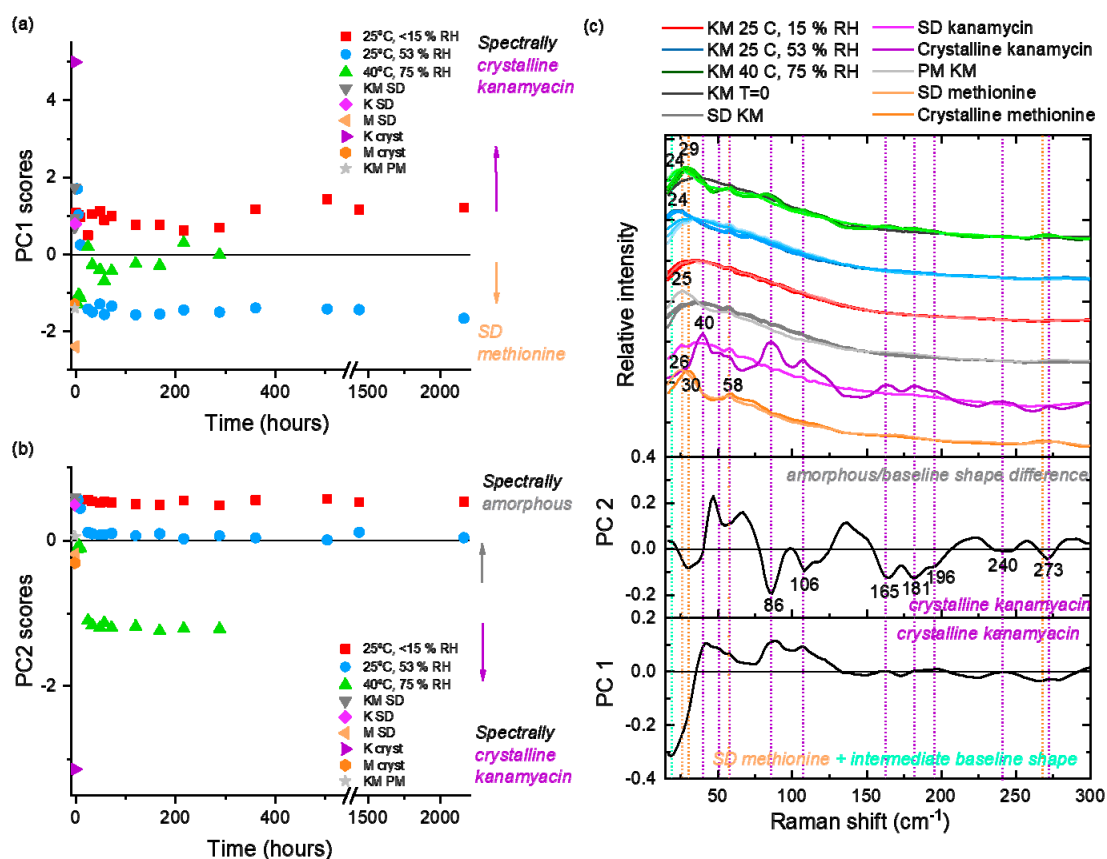
**Figure S6** Representative SEM images (low magnification) of the formulations [kanamycin only (KO), kanamycin-valine (KV), kanamycin-methionine (KM), kanamycin-phenylalanine (KP), and kanamycin-tryptophan (KT)] during the stability study on day 0 and day 28 when stored at 25 °C/<15 % RH and 25 °C/53 % RH.



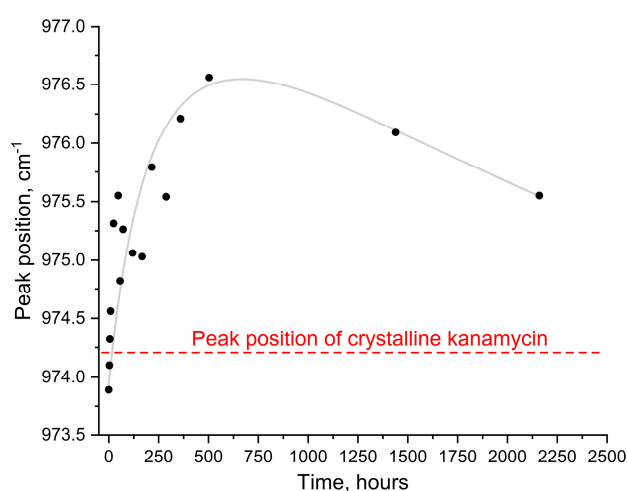
**Figure S7** Drug deposition behavior of the spray-dried particles over 28 days when stored at different stressed conditions (25 °C/<15% RH and 25 °C/ 53% RH) for (a) kanamycin only (KO), (b) kanamycin-valine (KV), (c) kanamycin-methionine (KM), (d) kanamycin-phenylalanine (KP), and (e) kanamycin-tryptophan (KT) formulations.



**Figure S8** Principal component analysis of the LFR spectra collected from KV samples stored under three different conditions (25 °C/<15% RH, 25 °C/53% RH and 40 °C/75% RH) over time. (a) PC 1 scores versus time, (b) PC 2 scores versus time, (c) PC3 scores versus time and (d) loadings with comparative spectra. PC 1 accounts for 86% of the explained spectral variance, PC 2 accounts for a further 9% and PC 3 a further 3% explained variance. PM represents physical mixture of amorphous kanamycin and crystalline valine in 1:1 molar ratio. Spectra from the different storage conditions in (c) have slight color graduations to highlight early (lighter coloring) versus latter (darker coloring) time points.

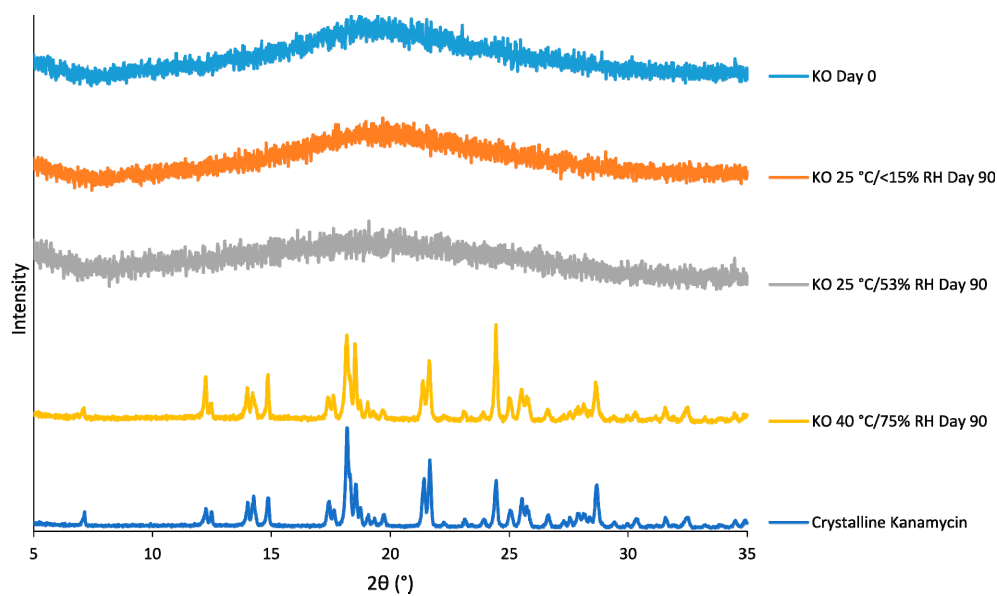


**Figure S9** Principal component analysis of the LFR spectra collected from KM samples stored under three different conditions (25 °C/<15% RH, 25 °C/53% RH and 40 °C/75% RH) over time. (a) PC 1 scores versus time, (b) PC 2 scores versus time, and (c) loadings with comparative spectra. PC 1 accounts for 67% of the explained spectral variance, PC 2 accounts for a further 21% explained variance. PM represents physical mixture of amorphous kanamycin and crystalline methionine in 1:1 molar ratio. Spectra from the different storage conditions in (c) have slight color graduations to highlight early (lighter coloring) versus latter (darker coloring) time points.

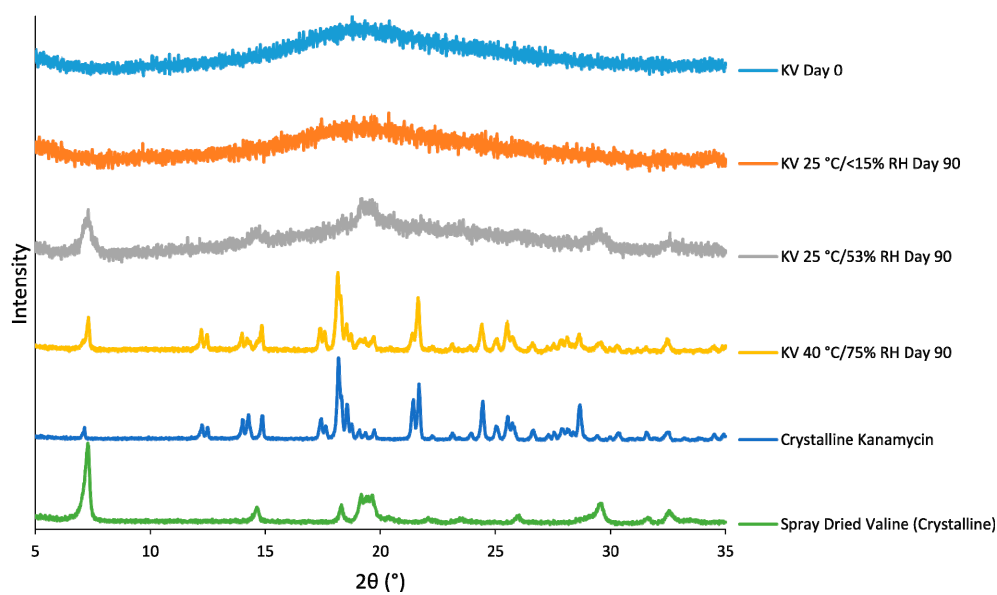


**Figure S10** Wavenumber position changes of the highest intensity peak for spray-dried kanamycin only (KO) formulation kept at 25 °C/53% RH. The line is drawn to assist in visualizing the trend.

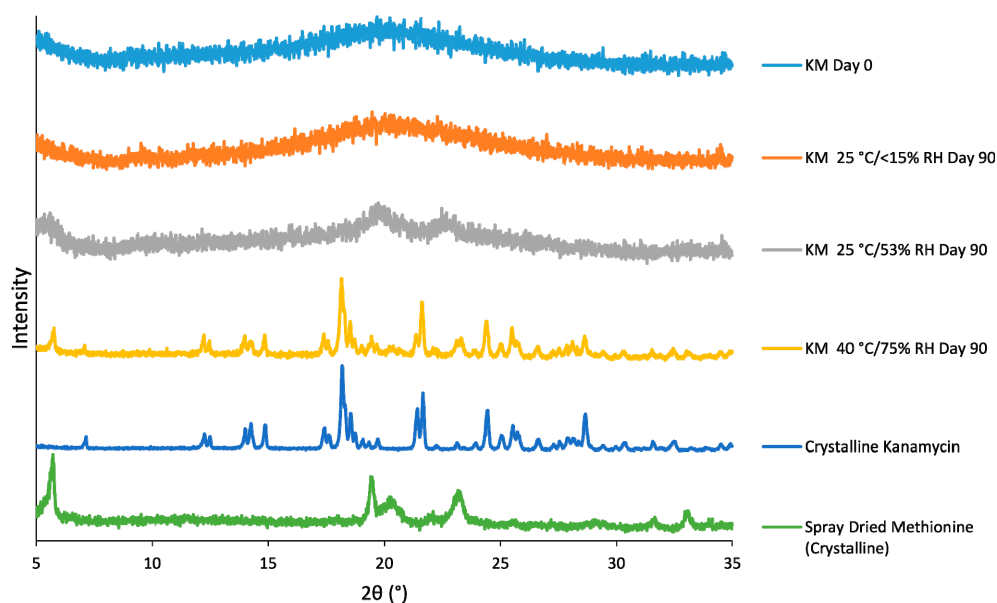




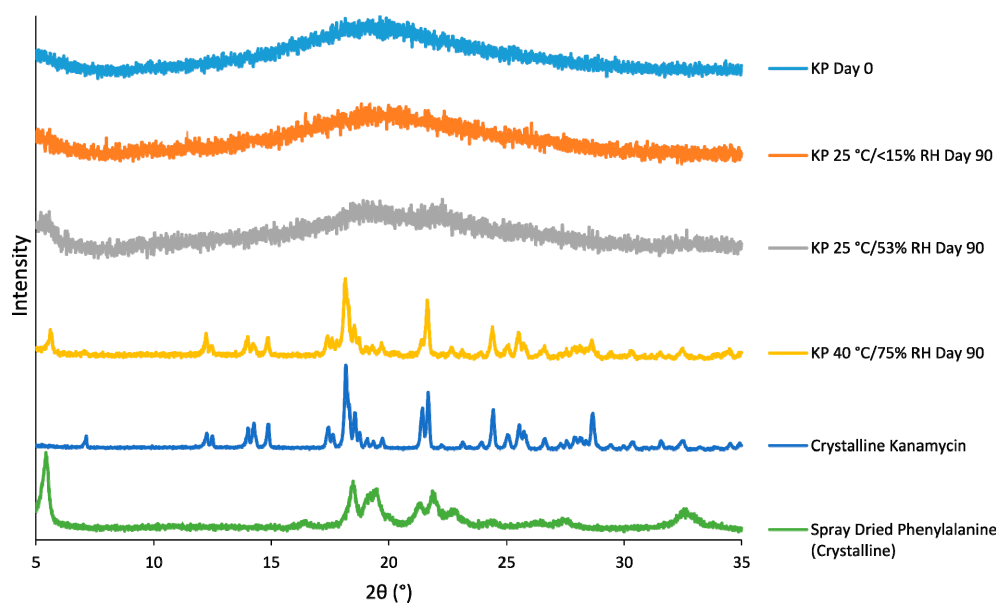
**Figure S11** Stability Study: PXRD of the kanamycin only formulation (KO) on day 0 and day 90 when kept at 25 °C/<15% RH, 25 °C/53% RH, and 40 °C/75% RH.



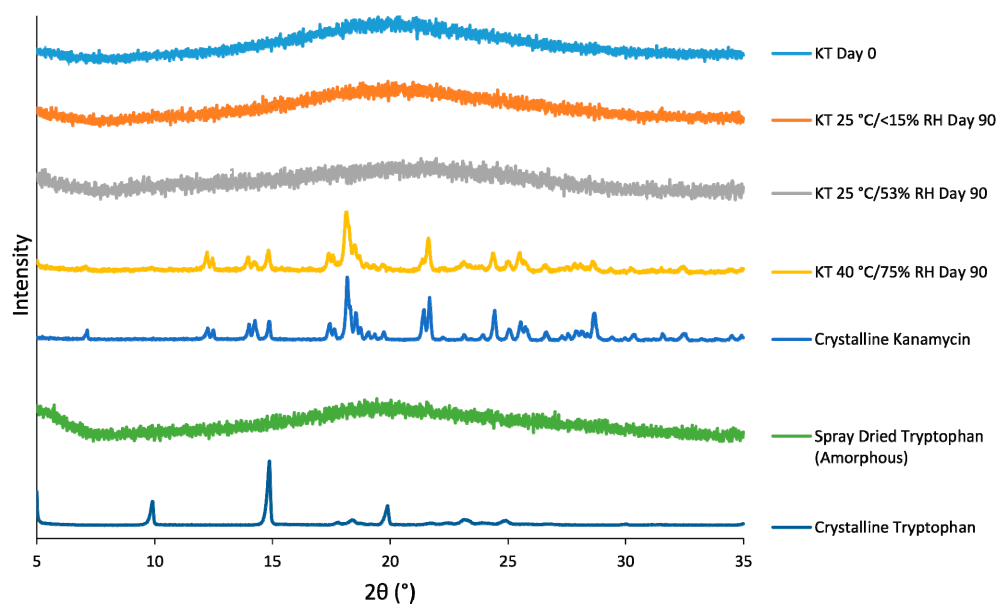
**Figure S12** Stability Study: PXRD of the kanamycin-valine formulation (KV) on day 0 and day 90 when kept at 25 °C/<15% RH, 25 °C/53% RH, and 40 °C/75% RH.



**Figure S13** Stability Study: PXRD of the kanamycin-methionine formulation (KM) on day 0 and day 90 when kept at 25 °C/<15% RH, 25 °C/53% RH, and 40 °C/75% RH.



**Figure S14** Stability Study: PXRD of the kanamycin-phenylalanine formulation (KP) on day 0 and day 90 when kept at 25 °C/<15% RH, 25 °C/53% RH, and 40 °C/75% RH.



**Figure S15** Stability Study: PXRD of the kanamycin-tryptophan formulation (KT) on day 0 and day 90 when kept at 25 °C/<15% RH, 25 °C/53% RH, and 40 °C/75% RH.