



One-Pot Synthesis of W_2C/WS_2 Hybrid Nanostructures for Improved Hydrogen Evolution Reactions and Supercapacitors

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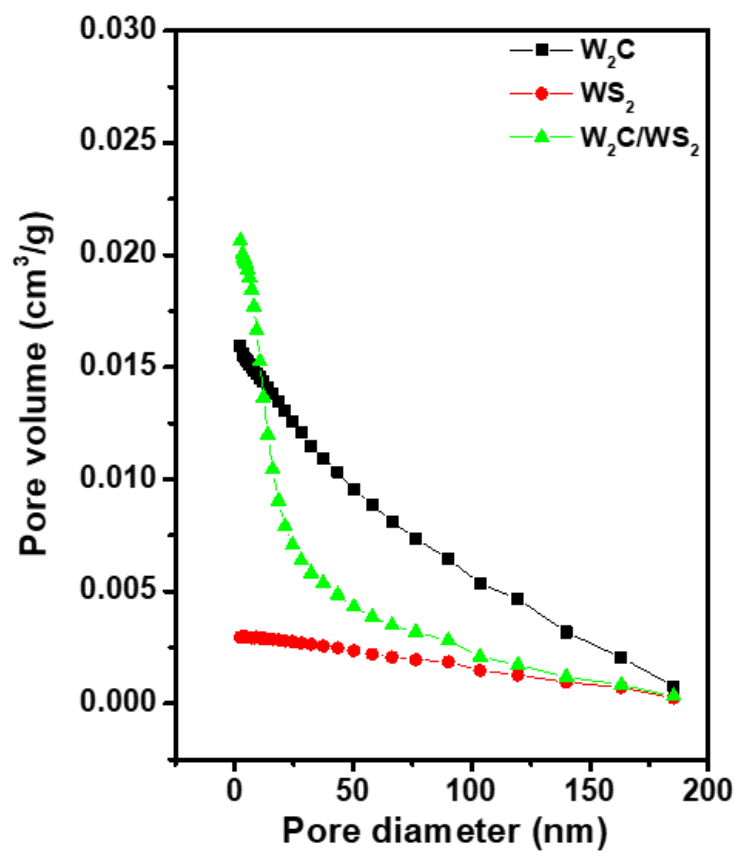


Figure S1. Pore diameter versus pore volume variations for the nanostructures.

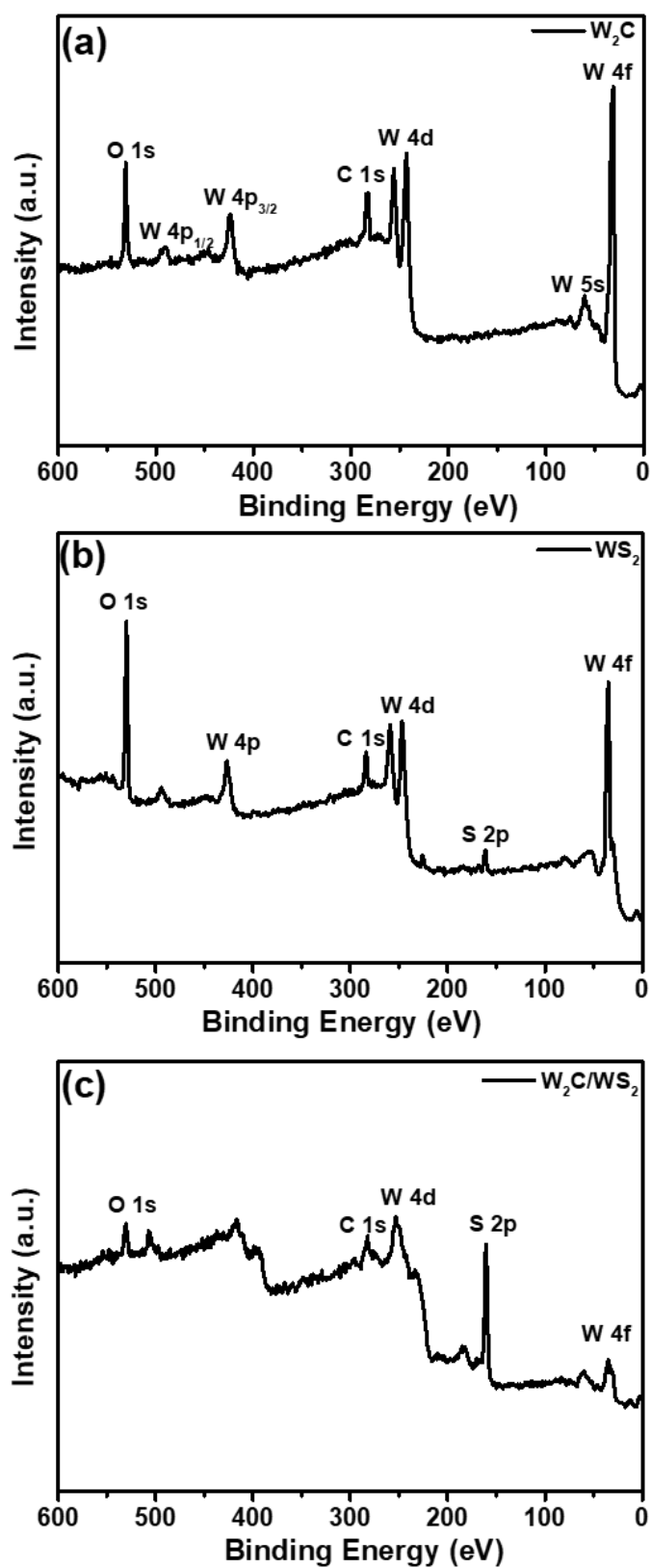


Figure S2. Survey XPS spectrum of (a) W_2C ; (b) WS_2 ; (c) W_2C/WS_2 .

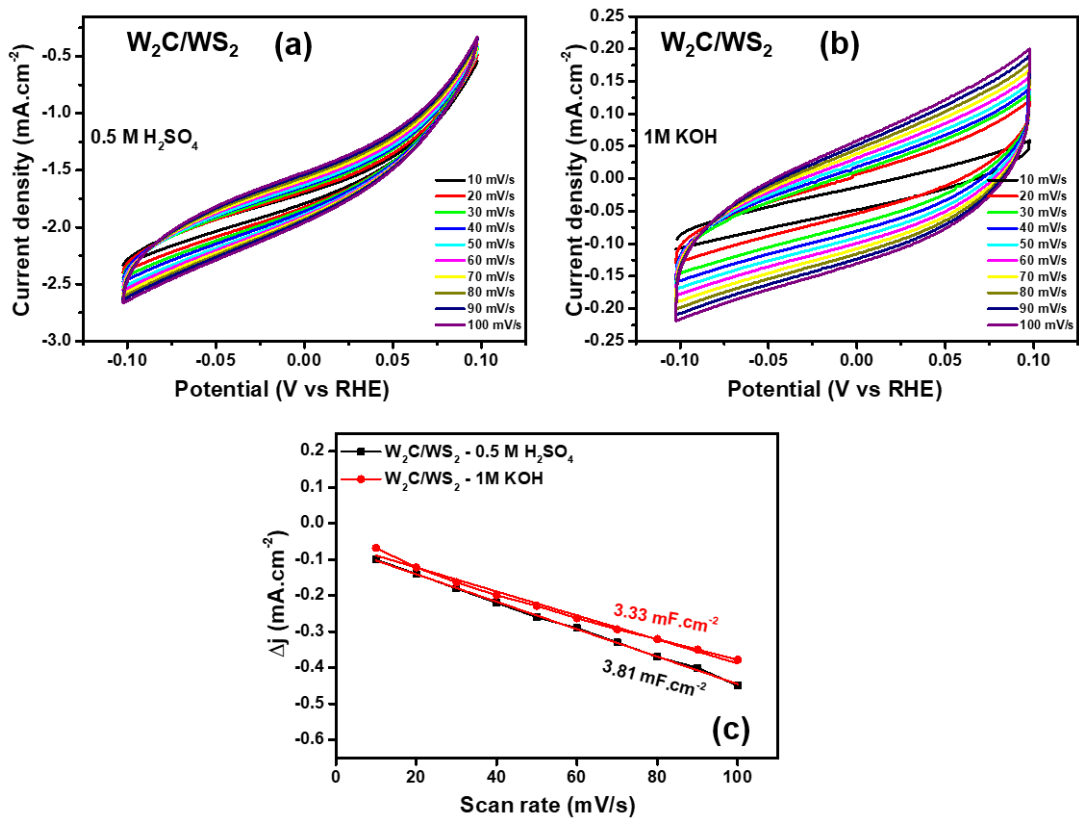


Figure S3. CV spectra in the non-Faradaic region with different scan rates for W_2C/WS_2 hybrid HER electrodes in (a) $0.5\text{ M H}_2\text{SO}_4$ and (b) 1 M KOH electrolyte and (c) their current differences.

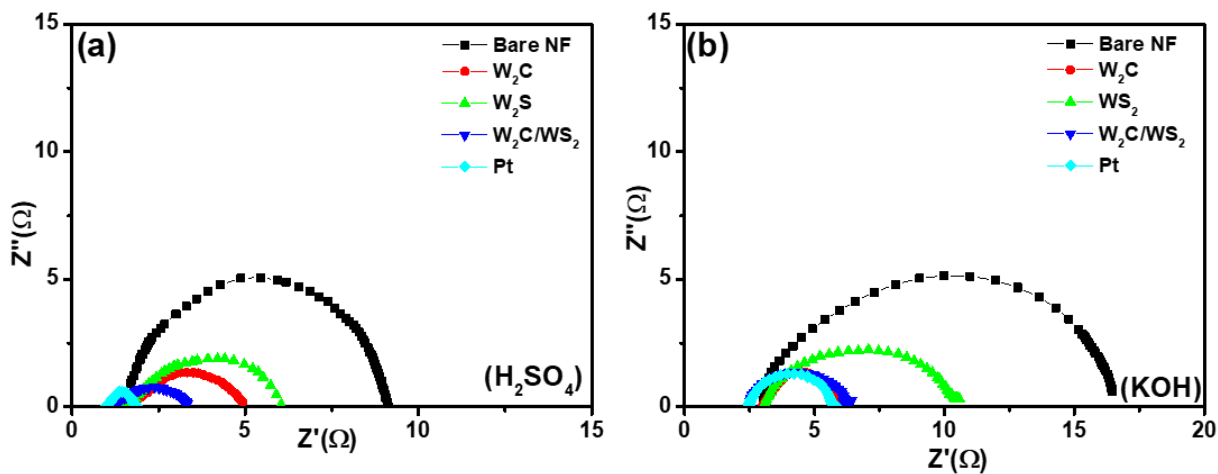


Figure S4. EIS spectra of Pt, W_2C , WS_2 , and W_2C/WS_2 hybrids electrodes in (a) $0.5\text{ M H}_2\text{SO}_4$ and (b) 1 M KOH electrolyte.

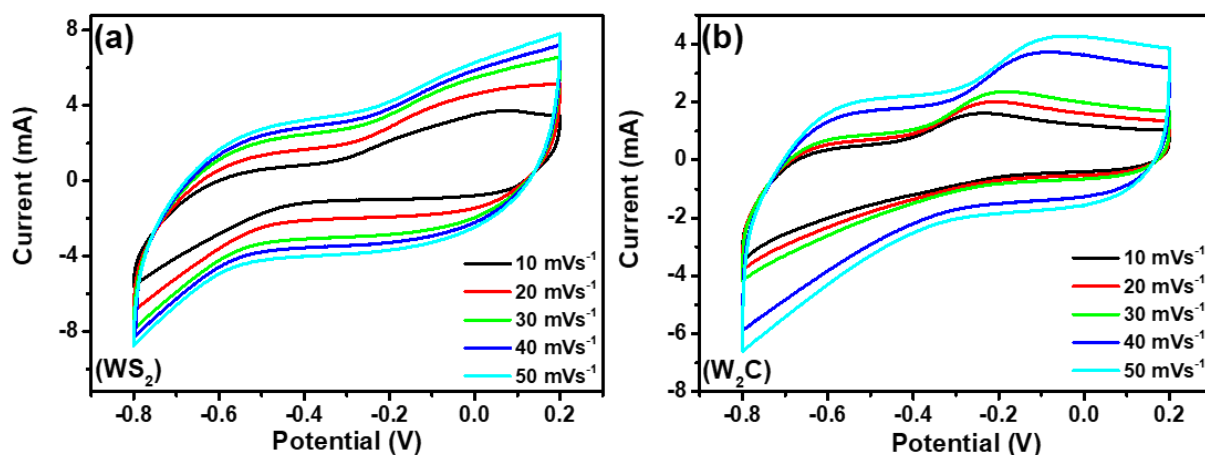


Figure S5. CV curves for the (a) WS₂ and (b) W₂C electrodes at various scan rates using half-cell (10-50 mV s⁻¹).

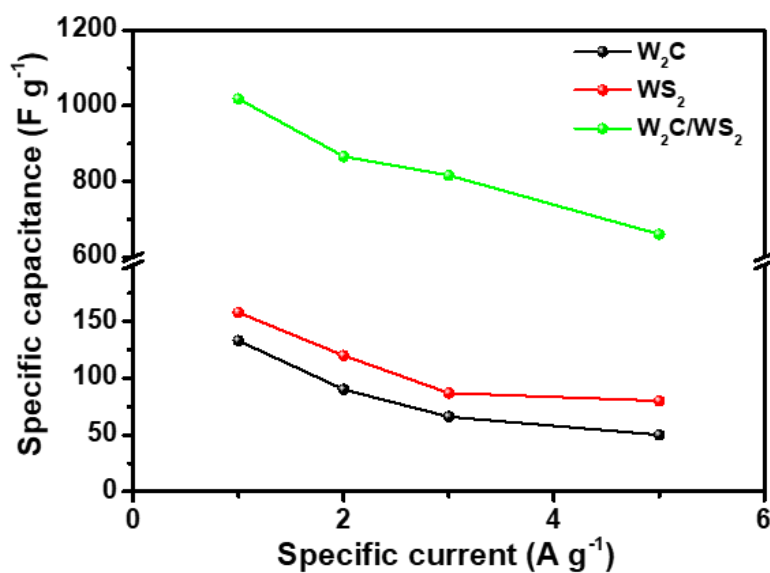


Figure S6. The specific capacitance variations at different current densities for W₂C, WS₂ and W₂C/WS₂ electrodes by half-cell measurements.

Table 1. Microstructural parameters of W₂C.

Lattice Plane	FWHM (°)	Crystallite Size (nm)	d-spacing (nm)
020	0.340	24.2	3.001
002	0.325	25.5	2.601
220	0.352	24.7	1.863
041	0.410	22.8	1.448
123	0.402	23.5	1.415
004	0.486	20.3	1.289
142	0.402	25.1	1.254
322	0.581	17.4	1.232

Table S2. Microstructural parameters of WS₂.

Lattice Plane	FWHM (°)	Crystallite Size (nm)	d-spacing (nm)
002	0.517	15.4	6.267
101	1.28	6.5	2.690
103	0.201	42.0	2.279
105	0.439	19.8	1.839
112	0.943	9.7	1.532
114	0.486	19.5	1.402
200	0.523	18.2	1.384

Table 3. Microstructural parameters of W₂C/WS₂.

Phase	Lattice Plane	FWHM (°)	Crystallite Size (nm)	d-spacing (nm)
WS ₂	002	0.765	10.4	6.159
	101	1.120	7.4	2.714
	103	0.956	8.8	2.281
	110	0.907	10.0	1.570
W ₂ C	020	0.283	29.1	2.981
	002	0.341	24.4	2.591
	220	0.362	24.0	1.877
	041	0.450	20.8	1.450
	123	0.444	21.2	1.427
	004	0.362	27.2	1.291
	142	0.620	16.1	1.256
	322	0.575	17.6	1.234

Table S4. Comparison of electrochemical parameters for different electrocatalysts.

Electrolyte	Electrocatalysts	Overpotential (mV vs RHE) @ 10 mA·cm ⁻²	Tafel Slope (mV·dec ⁻¹)	Exchange Current Density (j ₀ , mA·cm ⁻²)
0.5 M H ₂ SO ₄	Pt/C	51	36	1.03
	W ₂ C/WS ₂	133	70	1.02
	WS ₂	242	138	0.549
	W ₂ C	171	86	0.194
	Bare NF	450	168	0.022
1 M KOH	Pt/C	48	48	1.08
	W ₂ C/WS ₂	105	84	0.93
	WS ₂	189	127	0.382
	W ₂ C	123	141	0.717
	Bare NF	292	148	0.048

Table S5. HER catalytic performances TMDs and TMCs-based electrocatalysts.

Electrocatalyst	Electrolyte	η (mV)	Tafel Slope (mV·dec ⁻¹)	j ₀ (mA·cm ⁻²)	Ref
W ₂ C/WS ₂	0.5 M H ₂ SO ₄	133 @ 10 mA/cm ²	70	1.02	This work
W ₂ C/WS ₂	1.0 M KOH	105 @ 10 mA/cm ²	84	0.93	This work
WS ₂ /W ₂ C heterostructure	0.5 M H ₂ SO ₄ & 1M KOH	126 & 205 @ 10 mA/cm ²	68 & 72	0.501	[1]
W-W ₂ C/CNT	0.5 M H ₂ SO ₄ & 1M KOH	155 & 147 @ 10 mA/cm ²	63 & 64	-	[2]
p-WC _x NWs/CC	0.5 M H ₂ SO ₄ & 1M KOH	118 & 122 @ 10 mA/cm ²	56 & 56	0.165	[3]

W ₂ C@CNT-S hybrid	0.5 M H ₂ SO ₄ & 1M KOH	174 & 148@ 10 mA/cm ²	57.3 & 56.2	-	[4]
Mo ₂ C/CNT	0.1 M HClO ₄	250@ 10 mA/cm ²	251	1.43	[5]
Mo ₂ C/Ketjenblack carbon (KB)	1 M KOH	210@ 10 mA/cm ²	48	-	[6]
P-W ₂ C@NC	0.5 M H ₂ SO ₄	89@ 10 mA/cm ²	53	0.316	[7]
W/BrN	0.5 M H ₂ SO ₄ & 1M KOH	148 & 194@ 10 mA/cm ²	68 & 72	-	[8]
WC/W ₂ C heterojunction	0.5 M H ₂ SO ₄	69@ 10 mA/cm ²	52	0.361	[9]
Mo ₂ C QD/NGCL	1 M KOH	111@ 10 mA/cm ²	57.8	-	[10]
W _x C@WS ₂ Nanostructure	0.5 M H ₂ SO ₄	146@ 10 mA/cm ²	61	0.042	[11]
Mo ₂ C/N-CNT	1 M KOH	257@ 10 mA/cm ²	77	10	[12]
W ₂ C@GL	0.5 M H ₂ SO ₄	135@ 10 mA/cm ²	68	0.24	[13]
MoS ₂ /Ti ₃ C ₂ T _x hybrid	0.5 M H ₂ SO ₄	152@ 10 mA/cm ²	70	0.6165	[14]
CoS ₂ @WS ₂	0.5 M H ₂ SO ₄	97.2@ 10 mA/cm ²	66	-	[15]
Carbon coated cobalt– tungsten carbide Co ₆ W ₆ C	1 M KOH	73@ 10 mA/cm ²	25	-	[16]
MoS ₂ / Mo ₂ C-NCNTs	0.5 M H ₂ SO ₄	145@ 10 mA/cm ²	69	0.021	[17]
WC-CNTs	0.5 M H ₂ SO ₄ & 1M KOH	145 & 137@ 10 mA/cm ²	72 & 106	-	[18]
MoSSe@rGO composite	0.5 M H ₂ SO ₄	135@5 mA/cm ²	51		[19]
Nano MoC@graphite shell	0.5 M H ₂ SO ₄ & 1 M KOH	124 & 77@ 10 mA/cm ²	43 & 50	-	[20]
α-Mo ₂ C nanoparticles	0.5 M H ₂ SO ₄ & 1 M KOH	198 & 176@ 10 mA/cm ²	56 & 58	-	[21]
MoS ₂ -Co nanoboxes	0.5 M H ₂ SO ₄	155@ 10 mA/cm ²	55	0.013	[22]
MoC-Mo ₂ C HNWs	0.5 M H ₂ SO ₄ & 1 M KOH	126 & 120@ 10 mA/cm ²	43 & 42	-	[23]
MoS _{2-x} Se _{2(1-x)}	0.5 M H ₂ SO ₄	188@ 10 mA/cm ²	43	-	[24]
CNT-MoSeS	0.5 M H ₂ SO ₄	174@ 10 mA/cm ²	40	-	[25]
Mo ₂ C@NC	0.5 M H ₂ SO ₄ & 1 M KOH	124 & 60@ 10 mA/cm ²	60	-	[26]
S-MoSe ₂ @GGNR hybrids	0.5 M H ₂ SO ₄	153@ 10 mA/cm ²	46	-	[27]
1T/2HMoS ₂	0.5 M H ₂ SO ₄	220@ 10 mA/cm ²	61	0.443	[28]
N-doped Mo carbide and phosphide Hybrids (N@MoPC _x)	0.5 M H ₂ SO ₄	108@ 10 mA/cm ²	69.4	0.3424	[29]
WS _{2(1-x)} Se _{2x} NTs	0.5 M H ₂ SO ₄	298@ 10 mA/cm ²	105	0.029	[30]
FeS ₂ -RGO hybrid	0.5 M H ₂ SO ₄	226@ 10 mA/cm ²	61	0.25-0.65	[31]
Co/N, B co doped ultrathin carbon cages (Co@BCN)	1 M KOH	183@ 10 mA/cm ²	73.2	0.103	[32]

W ₂ C/graphene nanoplatelets (GnP)	0.1m HClO ₄	186@ 10 mA/cm ²	64.7	0.024	[33]
N-doped Mo ₂ C	0.5 M H ₂ SO ₄ & 1 M KOH	177.5 & 160.4@ 10 mA/cm ²	59.6 & 48.9	-	[34]

Table S6. Performances of TMDs and TMCs-based electrodes for supercapacitors.

Electrode Materials	Electrolyte	Specific Capacitance	Capacitance Retention (%) / Cycle	Ref
W ₂ C/WS ₂ hybrid	1.0 M KOH	1018 F·g ⁻¹ @ 1.0 A·g ⁻¹	94/5000	This work
WS ₂ /RGO hybrids	1.0 M Na ₂ SO ₄	350 F/g @ 2mV/s	-	[35]
PANI-MoS ₂ nanocomposite	1 M H ₂ SO ₄	612 Fg ⁻¹ @ 0.2 A g ⁻¹	93/2000	[36]
WS ₂ -MWCNTs/PANI	1 M H ₂ SO ₄	760.1 Fg ⁻¹ @ 1 A g ⁻¹	82.5/2000	[37]
VSe ₂ /carbon-nanotube		1854 uF / cm ²	90/10000	[38]
TaSe ₂ Nanobelts	1 M KCl	835 mF cm ⁻² @ 2 mV s ⁻¹	98.7/10000	[39]
MoS ₂ -NH ₂ /PANI nanosheets	1 M H ₂ SO ₄	326.4 F g ⁻¹ at 0.5 A g ⁻¹	96.5/10000	[40]
MoS ₂ and WS ₂ nanoparticles	3 M KOH	1531.2 & 1439.5 F g ⁻¹ at 5 mA cm ⁻²	81.6 & 74.4/2000	[41]
MXene/polypyrrole composite	1 M H ₂ SO ₄	416 F g ⁻¹ at 0.5 A g ⁻¹	86.4/5000	[42]
WS ₂ QDs	H ₃ PO ₄ -PVA	28 mF cm ⁻² @ 0.1 mA cm ⁻²	80/10000	[43]
MoS ₂ /CNS	1 M Na ₂ SO ₄	231 F g ⁻¹ at 1 A g ⁻¹	-	[44]
MoS ₂ /RCF composite	1 M Na ₂ SO ₄	225 F g ⁻¹ at 0.5 A g ⁻¹	81/2000	[45]
1T' -MoTe ₂	2 M KOH	1393 F g ⁻¹ at A g ⁻¹	98/1000	[46]
MoS ₂ /graphene nanobelts	1 M Na ₂ SO ₄	445.71 F g ⁻¹ at 0.8 A g ⁻¹	96.7/1000	[47]
MoS ₂ /rGO	1 M H ₂ SO ₄	306 F g ⁻¹ at 0.5 A g ⁻¹	-	[48]
WS ₂ /CFC	1 M KCl	399 F g ⁻¹ at 1 A g ⁻¹	99/500	[49]
WS ₂ nanosheets	PVA/LiCl gel	0.93 F cm ⁻² at 4 mA cm ⁻²	82/10000	[50]
MoSe ₂	2 M KOH	243 F g ⁻¹ at 0.5 A g ⁻¹	90.3/100	[51]
VS ₂ /MWCNTs	PVA-LiClO ₄	182 F g ⁻¹ at 2 mV/s	93.2/5000	[52]
MnO ₂ /Ti ₃ C ₂ T _x hybrid	1.0 M KOH	212.1 F g ⁻¹ @ 1 A g ⁻¹	88/4000	[53]

References

- Li, Y.; Wu, X.; Zhang, H.; Zhang, J. Interface designing over WS₂/W₂C for enhanced hydrogen evolution catalysis. *ACS Applied Energy Materials* **2018**, *1*, 3377-3384.
- Hu, Y.; Yu, B.; Ramadoss, M.; Li, W.; Yang, D.; Wang, B.; Chen, Y. Scalable synthesis of heterogeneous W-W₂C nanoparticles embedded cnt networks for boosted hydrogen evolution reaction in both acidic and alkaline media. *ACS Sustainable Chemistry & Engineering* **2019**, *7*, 10016-10024.
- Ren, B.; Li, D.; Jin, Q.; Cui, H.; Wang, C. Novel porous tungsten carbide hybrid nanowires on carbon cloth for high-performance hydrogen evolution. *Journal of Materials Chemistry A* **2017**, *5*, 13196-13203.
- Hu, Y.; Yu, B.; Li, W.; Ramadoss, M.; Chen, Y. W₂C nanodot-decorated cnt networks as a highly efficient and stable electrocatalyst for hydrogen evolution in acidic and alkaline media. *Nanoscale* **2019**, *11*, 4876-4884.
- Šljukić, B.; Vujković, M.; Amaral, L.; Santos, D.; Rocha, R.; Sequeira, C.; Figueiredo, J.L. Carbon-supported Mo₂C electrocatalysts for hydrogen evolution reaction. *Journal of Materials Chemistry A* **2015**, *3*, 15505-15512.

6. Wang, D.; Wang, J.; Luo, X.; Wu, Z.; Ye, L. In situ preparation of Mo₂C nanoparticles embedded in ketjenblack carbon as highly efficient electrocatalysts for hydrogen evolution. *ACS Sustainable Chemistry & Engineering* **2017**, *6*, 983-990.
7. Yan, G.; Wu, C.; Tan, H.; Feng, X.; Yan, L.; Zang, H.; Li, Y. N-carbon coated PW₂C composite as efficient electrocatalyst for hydrogen evolution reactions over the whole pH range. *Journal of Materials Chemistry A* **2017**, *5*, 765-772.
8. Li, Q.; Han, C.; Ma, X.; Wang, D.; Xing, Z.; Yang, X. Bromine and nitrogen co-doped tungsten nanoarrays to enable hydrogen evolution at all pH values. *Journal of Materials Chemistry A* **2017**, *5*, 17856-17861.
9. Zhang, L.-N.; Ma, Y.-Y.; Lang, Z.-L.; Wang, Y.-H.; Khan, S.U.; Yan, G.; Tan, H.-Q.; Zang, H.-Y.; Li, Y.-g. Ultrafine cable-like WC/W₂C heterojunction nanowires covered by graphitic carbon towards highly efficient electrocatalytic hydrogen evolution. *Journal of Materials Chemistry A* **2018**, *6*, 15395-15403.
10. Pu, Z.; Wang, M.; Kou, Z.; Amiin, I.S.; Mu, S. Mo₂C quantum dot embedded chitosan-derived nitrogen-doped carbon for efficient hydrogen evolution in a broad pH range. *Chemical Communications* **2016**, *52*, 12753-12756.
11. Wang, F.; He, P.; Li, Y.; Shifa, T.A.; Deng, Y.; Liu, K.; Wang, Q.; Wang, F.; Wen, Y.; Wang, Z. Interface engineered W_xC@WS₂ nanostructure for enhanced hydrogen evolution catalysis. *Advanced Functional Materials* **2017**, *27*, 1605802.
12. Zhang, K.; Zhao, Y.; Fu, D.; Chen, Y. Molybdenum carbide nanocrystal embedded n-doped carbon nanotubes as electrocatalysts for hydrogen generation. *Journal of Materials Chemistry A* **2015**, *3*, 5783-5788.
13. Zhou, Y.; Ma, R.; Li, P.; Chen, Y.; Liu, Q.; Cao, G.; Wang, J. Ditungsten carbide nanoparticles encapsulated by ultrathin graphitic layers with excellent hydrogen-evolution electrocatalytic properties. *Journal of Materials Chemistry A* **2016**, *4*, 8204-8210.
14. Liu, J.; Liu, Y.; Xu, D.; Zhu, Y.; Peng, W.; Li, Y.; Zhang, F.; Fan, X. Hierarchical "nanoroll" like MoS₂/Ti₃C₂T_x hybrid with high electrocatalytic hydrogen evolution activity. *Applied Catalysis B: Environmental* **2019**, *241*, 89-94.
15. Zhou, X.; Yang, X.; Li, H.; Hedhili, M.N.; Huang, K.-W.; Li, L.-J.; Zhang, W. Symmetric synergy of hybrid CoS₂-WS₂ electrocatalysts for the hydrogen evolution reaction. *Journal of Materials Chemistry A* **2017**, *5*, 15552-15558.
16. Liu, Y.; Li, G.-D.; Yuan, L.; Ge, L.; Ding, H.; Wang, D.; Zou, X. Carbon-protected bimetallic carbide nanoparticles for a highly efficient alkaline hydrogen evolution reaction. *Nanoscale* **2015**, *7*, 3130-3136.
17. Zhang, K.; Zhao, Y.; Zhang, S.; Yu, H.; Chen, Y.; Gao, P.; Zhu, C. MoS₂ nanosheet/Mo₂C-embedded n-doped carbon nanotubes: Synthesis and electrocatalytic hydrogen evolution performance. *Journal of Materials Chemistry A* **2014**, *2*, 18715-18719.
18. Fan, X.; Zhou, H.; Guo, X. WC nanocrystals grown on vertically aligned carbon nanotubes: An efficient and stable electrocatalyst for hydrogen evolution reaction. *ACS Nano* **2015**, *9*, 5125-5134.
19. Konkana, B.; Masa, J.; Xia, W.; Muhler, M.; Schuhmann, W. MoSSe@reduced graphene oxide nanocomposite heterostructures as efficient and stable electrocatalysts for the hydrogen evolution reaction. *Nano Energy* **2016**, *29*, 46-53.
20. Shi, Z.; Wang, Y.; Lin, H.; Zhang, H.; Shen, M.; Xie, S.; Zhang, Y.; Gao, Q.; Tang, Y. Porous nanoMoC@graphite shell derived from a MOFs-directed strategy: An efficient electrocatalyst for the hydrogen evolution reaction. *Journal of Materials Chemistry A* **2016**, *4*, 6006-6013.
21. Ma, L.; Ting, L.R.L.; Molinari, V.; Giordano, C.; Yeo, B.S. Efficient hydrogen evolution reaction catalyzed by molybdenum carbide and molybdenum nitride nanocatalysts synthesized via the urea glass route. *Journal of Materials Chemistry A* **2015**, *3*, 8361-8368.
22. Chen, G.; Dong, W.F.; Li, B.L.; Deng, Y.H.; Wang, X.H.; Zhang, X.F.; Luo, H.Q.; Li, N.B. Cobalt incorporated MoS₂ hollow structure with rich out-of-plane edges for efficient hydrogen production. *Electrochimica Acta* **2018**, *276*, 81-91.
23. Lin, H.; Shi, Z.; He, S.; Yu, X.; Wang, S.; Gao, Q.; Tang, Y. Heteronanowires of MoC-Mo₂C as efficient electrocatalysts for hydrogen evolution reaction. *Chemical science* **2016**, *7*, 3399-3405.
24. Zhen, C.; Zhang, B.; Zhou, Y.; Du, Y.; Xu, P. Hydrothermal synthesis of ternary MoS_{2-x}Se_{2(1-x)} nanosheets for electrocatalytic hydrogen evolution. *Inorganic Chemistry Frontiers* **2018**, *5*, 1386-1390.
25. Yang, J.; Liu, Y.; Shi, C.; Zhu, J.; Yang, X.; Liu, S.; Li, L.; Xu, Z.; Zhang, C.; Liu, T. Carbon nanotube with vertical 2D molybdenum sulphoselenide nanosheet arrays for boosting electrocatalytic hydrogen evolution. *ACS Applied Energy Materials* **2018**, *1*, 7035-7045.

26. Liu, Y.; Yu, G.; Li, G.D.; Sun, Y.; Asefa, T.; Chen, W.; Zou, X. Coupling Mo₂C with nitrogen-rich nanocarbon leads to efficient hydrogen-evolution electrocatalytic sites. *Angewandte Chemie International Edition* **2015**, *54*, 10752-10757.
27. Fan, W.; Wang, D.; Sun, Z.; Ling, X.Y.; Liu, T. Graphene/graphene nanoribbon aerogels decorated with s-doped MoSe₂ nanosheets as an efficient electrocatalyst for hydrogen evolution. *Inorganic Chemistry Frontiers* **2019**, *6*, 1209-1216.
28. Liu, Z.; Gao, Z.; Liu, Y.; Xia, M.; Wang, R.; Li, N. Heterogeneous nanostructure based on 1T-phase MoS₂ for enhanced electrocatalytic hydrogen evolution. *ACS applied materials & interfaces* **2017**, *9*, 25291-25297.
29. Huang, Y.; Ge, J.; Hu, J.; Zhang, J.; Hao, J.; Wei, Y. Nitrogen-doped porous molybdenum carbide and phosphide hybrids on a carbon matrix as highly effective electrocatalysts for the hydrogen evolution reaction. *Advanced Energy Materials* **2018**, *8*, 1701601.
30. Xu, K.; Wang, F.; Wang, Z.; Zhan, X.; Wang, Q.; Cheng, Z.; Safdar, M.; He, J. Component-controllable WS₂(1-x)Se_{2x} nanotubes for efficient hydrogen evolution reaction. *Acs Nano* **2014**, *8*, 8468-8476.
31. Jiang, J.; Zhu, L.; Chen, H.; Sun, Y.; Qian, W.; Lin, H.; Han, S. Highly active and stable electrocatalysts of FeS₂-reduced graphene oxide for hydrogen evolution. *Journal of materials science* **2019**, *54*, 1422-1433.
32. Zhang, H.; Ma, Z.; Duan, J.; Liu, H.; Liu, G.; Wang, T.; Chang, K.; Li, M.; Shi, L.; Meng, X. Active sites implanted carbon cages in core-shell architecture: Highly active and durable electrocatalyst for hydrogen evolution reaction. *ACS Nano* **2015**, *10*, 684-694.
33. Chen, W.F.; Schneider, J.M.; Sasaki, K.; Wang, C.H.; Schneider, J.; Iyer, S.; Iyer, S.; Zhu, Y.; Muckerman, J.T.; Fujita, E. Tungsten carbide-nitride on graphene nanoplatelets as a durable hydrogen evolution electrocatalyst. *ChemSusChem* **2014**, *7*, 2414-2418.
34. Jiang, R.; Fan, J.; Hu, L.; Dou, Y.; Mao, X.; Wang, D. Electrochemically synthesized n-doped molybdenum carbide nanoparticles for efficient catalysis of hydrogen evolution reaction. *Electrochimica Acta* **2018**, *261*, 578-587.
35. Ratha, S.; Rout, C.S. Supercapacitor electrodes based on layered tungsten disulfide-reduced graphene oxide hybrids synthesized by a facile hydrothermal method. *ACS applied materials & interfaces* **2013**, *5*, 11427-11433.
36. Raghu, M.; Kumar, K.Y.; Rao, S.; Aravinda, T.; Prasanna, B.; Prashanth, M. Fabrication of polyaniline-few-layer MoS₂ nanocomposite for high energy density supercapacitors. *Polymer Bulletin* **2018**, *75*, 4359-4375.
37. Gao, J.; Ma, Y.; Li, J.; Fan, J.; Shi, P.; Xu, Q.; Min, Y. Free-standing WS₂-MWCNTs hybrid paper integrated with polyaniline for high-performance flexible supercapacitor. *Journal of Nanoparticle Research* **2018**, *20*, 298.
38. Wang, C.; Wu, X.; Xu, H.; Zhu, Y.; Liang, F.; Luo, C.; Xia, Y.; Xie, X.; Zhang, J.; Duan, C. VSe₂/carbon-nanotube compound for all solid-state flexible in-plane supercapacitor. *Applied Physics Letters* **2019**, *114*, 023902.
39. Wang, M.; Zhang, L.; Zhong, Y.; Huang, M.; Zhen, Z.; Zhu, H. In situ electrodeposition of polypyrrole onto TaSe₂ nanobelts quasi-arrays for high-capacitance supercapacitor. *Nanoscale* **2018**, *10*, 17341-17346.
40. Zeng, R.; Li, Z.; Li, L.; Li, Y.; Huang, J.; Xiao, Y.; Yuan, K.; Chen, Y. Covalent connection of polyaniline with MoS₂ nanosheets towards ultrahigh rate capability supercapacitors. *ACS Sustainable Chemistry & Engineering* **2019**, *7*, 11540-11549.
41. Nagaraju, C.; Gopi, C.V.M.; Ahn, J.-W.; Kim, H.-J. Hydrothermal synthesis of MoS₂ and WS₂ nanoparticles for high-performance supercapacitor applications. *New Journal of Chemistry* **2018**, *42*, 12357-12360.
42. Jian, X.; He, M.; Chen, L.; Zhang, M.-m.; Li, R.; Gao, L.-j.; Fu, F.; Liang, Z.-h. Three-dimensional carambola-like MXene/polypyrrole composite produced by one-step co-electrodeposition method for electrochemical energy storage. *Electrochimica Acta* **2019**, *318*, 820-827.
43. Ghorai, A.; Midya, A.; Ray, S.K. Superior charge storage performance of WS₂ quantum dots in a flexible solid state supercapacitor. *New Journal of Chemistry* **2018**, *42*, 3609-3613.
44. Khawula, T.N.; Raju, K.; Franklyn, P.J.; Sigalas, I.; Ozoemena, K.I. Symmetric pseudocapacitors based on molybdenum disulfide (MoS₂)-modified carbon nanospheres: Correlating physicochemistry and synergistic interaction on energy storage. *Journal of Materials Chemistry A* **2016**, *4*, 6411-6425.
45. Zhao, C.; Zhou, Y.; Ge, Z.; Zhao, C.; Qian, X. Facile construction of MoS₂/RCF electrode for high-performance supercapacitor. *Carbon* **2018**, *127*, 699-706.
46. Liu, M.; Wang, Z.; Liu, J.; Wei, G.; Du, J.; Li, Y.; An, C.; Zhang, J. Synthesis of few-layer 1T'-MoTe₂ ultrathin nanosheets for high-performance pseudocapacitors. *Journal of Materials Chemistry A* **2017**, *5*, 1035-1042.
47. Jia, Y.; Wan, H.; Chen, L.; Zhou, H.; Chen, J. Hierarchical nanosheet-based MoS₂/graphene nanobelts with high electrochemical energy storage performance. *Journal of Power Sources* **2017**, *354*, 1-9.

48. Ji, H.; Hu, S.; Jiang, Z.; Shi, S.; Hou, W.; Yang, G. Directly scalable preparation of sandwiched MoS₂/graphene nanocomposites via ball-milling with excellent electrochemical energy storage performance. *Electrochimica Acta* **2019**, *299*, 143-151.
49. Shang, X.; Chi, J.-Q.; Lu, S.-S.; Gou, J.-X.; Dong, B.; Li, X.; Liu, Y.-R.; Yan, K.-L.; Chai, Y.-M.; Liu, C.-G. Carbon fiber cloth supported interwoven WS₂ nanosplates with highly enhanced performances for supercapacitors. *Applied Surface Science* **2017**, *392*, 708-714.
50. Liu, S.; Zeng, Y.; Zhang, M.; Xie, S.; Tong, Y.; Cheng, F.; Lu, X. Binder-free WS₂ nanosheets with enhanced crystallinity as a stable negative electrode for flexible asymmetric supercapacitors. *Journal of Materials Chemistry A* **2017**, *5*, 21460-21466.
51. Gao, Y.-P.; Huang, K.-J.; Shuai, H.-L.; Liu, L. Synthesis of sphere-feature molybdenum selenide with enhanced electrochemical performance for supercapacitor. *Materials Letters* **2017**, *209*, 319-322.
52. Pandit, B.; Karade, S.S.; Sankapal, B.R. Hexagonal VS₂ anchored MWCNTs: First approach to design flexible solid-state symmetric supercapacitor device. *ACS applied materials & interfaces* **2017**, *9*, 44880-44891.
53. Rakhi, R.B.; Ahmed, B.; Anjum, D.; Alshareef, H.N. Direct chemical synthesis of MnO₂ nanowhiskers on transition-metal carbide surfaces for supercapacitor applications. *ACS applied materials & interfaces* **2016**, *8*, 18806-18814.