



Supporting Information

for *Adv. Sci.*, DOI: 10.1002/adv.202000621

Combining Battery-Type and Pseudocapacitive Charge Storage in Ag/Ti₃C₂T_x MXene Electrode for Capturing Chloride Ions with High Capacitance and Fast Ion Transport

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Calculation method and equations

The Cl⁻-removal capacity (Cl⁻-RC), Cl⁻-removal rate (Cl⁻-RR) and energy consumption (EC) are calculated using following equations, respectively.

$$\text{Cl}^{-}\text{-RC (mg-Cl}^{-}\text{/g-electrode)} = \frac{(C_0 - C_e) \times V \times 35.5 \times 1000}{m} \quad (1)$$

$$\text{Cl}^{-}\text{-RR (mg-Cl}^{-}\text{/g-electrode/min)} = \frac{\text{Cl}^{-}\text{-RC}}{t} \times 60 \quad (2)$$

$$\text{EC (kWh/kg-Cl}^{-}\text{)} = \frac{i \times \int v dt}{3600 \times (C_0 - C_e) \times V \times 35.5} \quad (3)$$

Where the C_0 means the initial concentration of NaCl, mol/L; C_e represents the concentration of NaCl after charging, mol/L; V shows the volume of NaCl, L; m denotes the mass of Cl⁻ storage electrode, g; t signifies the desalination time, s; i and v expresses the current (A) and potential (V) during desalination process, respectively.

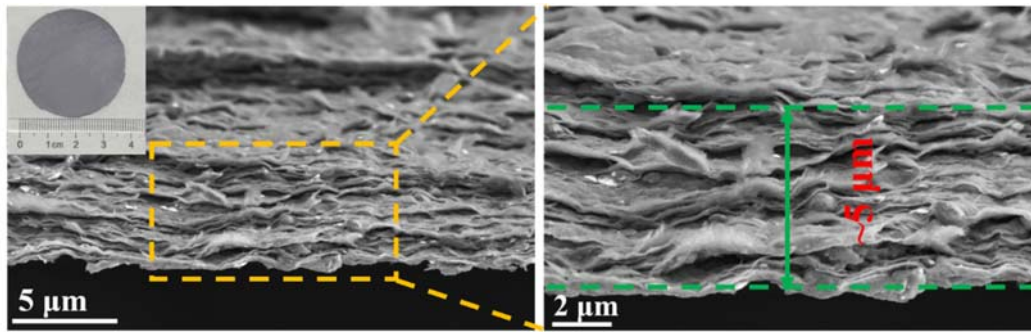


Figure S1. SEM images of Ti₃C₂T_x/Ag-3 hybrid (The inset: Digital photograph).

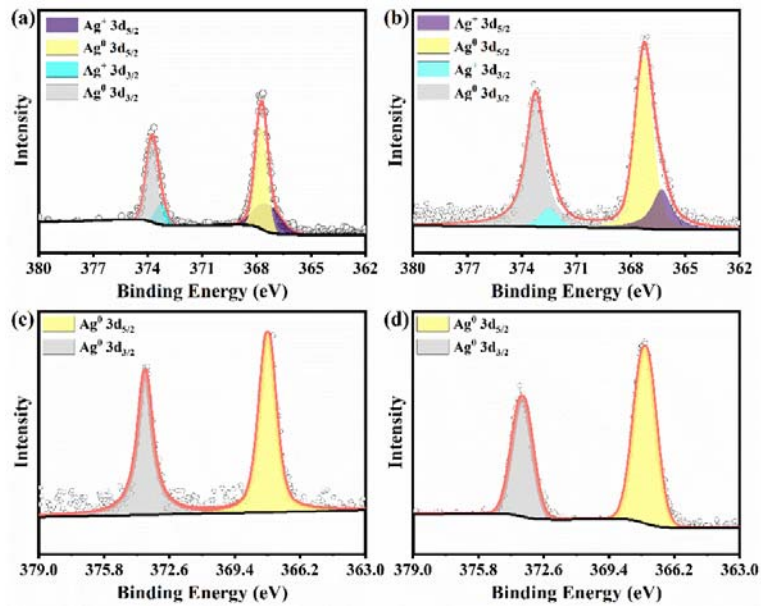


Figure S2. High-resolution Ag 3d XPS spectra of $\text{Ti}_3\text{C}_2\text{T}_x/\text{Ag}$ hybrid: a) $\text{Ti}_3\text{C}_2\text{T}_x/\text{Ag}$ -3, b) $\text{Ti}_3\text{C}_2\text{T}_x/\text{Ag}$ -6, c) $\text{Ti}_3\text{C}_2\text{T}_x/\text{Ag}$ -9 and d) $\text{Ti}_3\text{C}_2\text{T}_x/\text{Ag}$ -12.

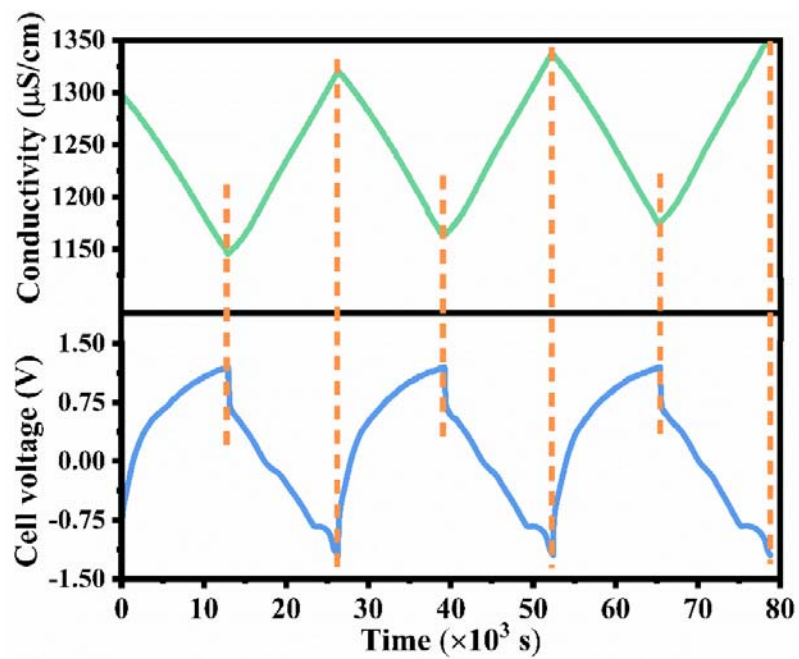


Figure S3. The conductivity (upper) and cell voltage change (lower) during the desalination process of $\text{Ti}_3\text{C}_2\text{T}_x/\text{Ag}$ -3 at 20 mA/g.

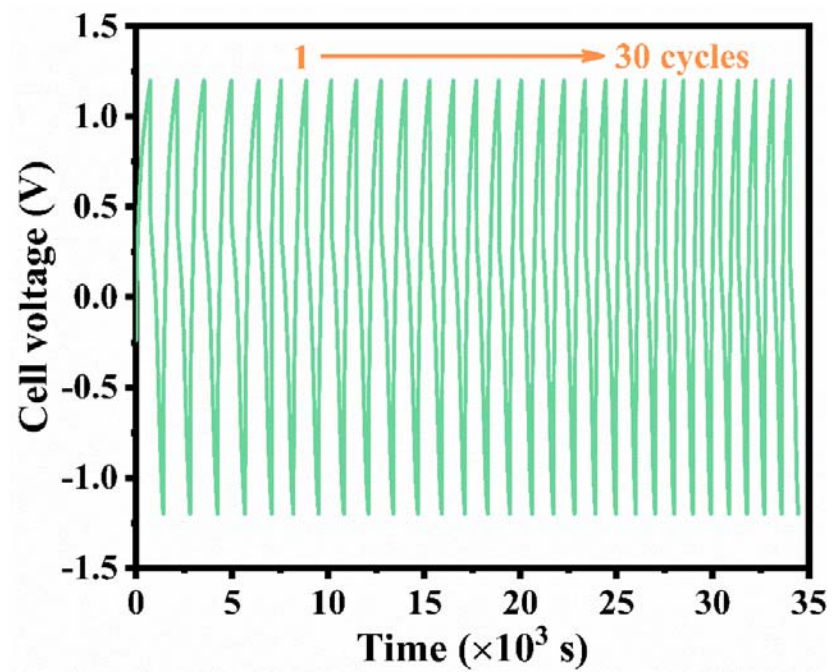


Figure S4. Potential curves of $\text{Ti}_3\text{C}_2\text{T}_x/\text{Ag}$ -3 electrode at 100 mA/g during 30 desalination cycles.

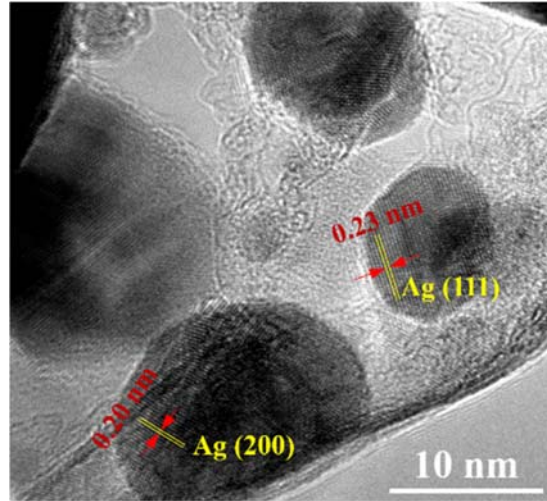


Figure S5. High-resolution TEM image of Ti₃C₂T_x/Ag-3 electrode after a long-term cycling.

Table S1. The content of Ag and AgCl species in Ti₃C₂T_x/Ag hybrid (EDX measurement).

Electrode materials	Weight percent of Ag⁰ (wt%)	Weight percent of AgCl (wt%)	Weight percent of total Ag (wt%)
Ti₃C₂T_x/Ag-3	7.90	17.68	21.22
Ti₃C₂T_x/Ag-6	12.52	8.64	19.03
Ti₃C₂T_x/Ag-9	19.79	-	19.79
Ti₃C₂T_x/Ag-12	21.01	-	21.01

Table S2. The content of Ag and AgCl species in Ti₃C₂T_x/Ag-3 hybrid after different operation (EDX measurement).

Operation	Weight percent of Ag⁰ (wt%)	Weight percent of AgCl (wt%)	Weight percent of total Ag (wt%)
Before inverse-voltage washing	7.90	17.68	21.22
After inverse-voltage washing	17.21	6.18	21.86
After long-term cycling	19.63	1.90	21.06