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Supporting Information

Catalytic Enantioselective Synthesis of Heterocyclic Vicinal Fluoroamines by Using Asymmetric Protonation: Method Development and Mechanistic Study**

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1. General

All reagents and solvents were obtained from commercial suppliers and were used without further purification unless otherwise stated. Purification was carried out according to standard laboratory methods.¹

1.1 Purification of Solvents

Dry THF for reactions were obtained from a PureSolv SPS-400-5 solvent purification system. Et₂O, EtOAc, and petroleum ether 40-60 °C for purification purposes were used as obtained from suppliers without further purification.

1.2 Experimental Details

Reactions were carried out using conventional glassware (preparation of intermediates), 2 mL HPLC vials, or in capped 5, 10, and 20 mL microwave vials. The glassware was oven-dried (150 °C) and purged with N₂ before use. Purging refers to a vacuum/nitrogen-refilling procedure. Prior to use, Mg turnings were oven-dried at 150 °C overnight. TRIP refers to 3,3'-bis(2,4,6-triisopropylphenyl)-1,1'-binaphthyl-2,2'-diyl hydrogenphosphate. Room temperature (rt) was generally *ca.* 18 °C. Reactions were carried out at elevated temperatures using a temperature-regulated hotplate/stirrer with a sand bath. Reactions were carried out at -10, -20, and -50 °C using an isopropanol bath cooled by a Thermo Haake EK90 cryocooler. Reactions were carried out at 0 °C using an ice/water bath.

1.3 Purification of Products

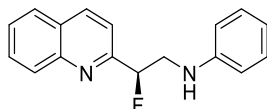
Thin layer chromatography was carried out using Merck silica plates coated with fluorescent indicator UV254. These were analyzed under 254 nm UV light, developed using potassium permanganate or vanillin solutions. Normal phase flash chromatography was carried out using ZEOprep 60 HYD 40-63 μm silica gel.

1.4 Analysis of Products

Fourier Transformed Infra-Red (FTIR) spectra were obtained on a Shimadzu IRAffinity-1 machine. ¹⁹F NMR spectra were obtained on either a Bruker AV 400 spectrometer at 376 or 377 MHz, or a Bruker AV 500 spectrometer at 470 MHz, respectively. ¹¹B NMR spectra were obtained on a Bruker AV 400 spectrometer at 128 MHz. ¹H and ¹³C NMR spectra were obtained on either a Bruker AV 400 at 400 MHz and 101 MHz, respectively, or Bruker DRX 500 at 500 MHz and 126 MHz, respectively. Chemical shifts are reported in ppm and coupling constants are reported in Hz with CDCl₃ referenced at 7.26 (¹H) and 77.0 ppm (¹³C) and DMSO-*d*₆ referenced at 2.50 (¹H) and 39.5 (¹³C). ¹¹B NMR spectra are referenced to BF₃·Et₂O. Unless otherwise stated, *J* refers to ³J_{HH} and *J*_{CF} in ¹H and ¹³C NMR, respectively. High-resolution mass spectra were obtained either through analysis at the EPSRC UK National Mass Spectrometry Facility at Swansea University, or through analysis at the University of St Andrews. High performance liquid chromatography (HPLC) was performed on an Agilent 1200 series HPLC using a chiral stationary phase column (column, Daicel Co. CHIRALCEL OJ-H, or CHIRALPAK IA; eluent: *n*-hexane/*i*-PrOH). All solvents used were HPLC-grade solvents purchased from Fisher. The column employed and the respective solvent mixture are indicated for each experiment. Optical rotations were obtained on a Perkin Elmer Model 341 polarimeter.

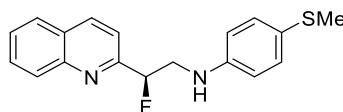
2. General Experimental Procedures

General Procedure A: Aza-Michael reaction catalyzed using a chiral acid (THF as solvent). For example, preparation of 7a.



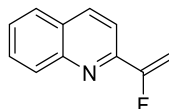
A 2 mL HPLC vial was charged with 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), and (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%). The vial was then capped and purged with N₂ before adding THF (200 μ L, 0.5 M). The reaction mixture was cooled to -10 $^{\circ}$ C for 15 min before the addition of aniline (27 μ L, 0.30 mmol, 3.0 equiv). The resulting mixture was then allowed to stir at -10 $^{\circ}$ C for 3 d before being quenched by sat. aq. K₂CO₃ solution (2 mL). The phases were separated and the aqueous phase extracted with EtOAc (5 mL \times 3). The organic phases were combined, dried over Na₂SO₄, filtered, and concentrated *in vacuo*. Purification of the crude *via* flash chromatography (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) gave the desired product as a pale yellow oil (21.8 mg, 82%, 96:4 e.r.).

General Procedure B: Aza-Michael reaction catalyzed using a chiral acid (CPME as solvent). For example, preparation of 7b.



A 2 mL HPLC vial was charged with 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), and (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%). The vial was then capped and purged with N₂ before adding CPME (200 μ L, 0.5 M). The reaction mixture was cooled to -20 $^{\circ}$ C for 15 min before the addition of 4-(methylthio)aniline (41.7 mg, 0.3 mmol, 3.0 equiv). The resulting mixture was then allowed to stir at -20 $^{\circ}$ C for 5 d before being quenched by sat. aq. K₂CO₃ solution (2 mL). The phases were separated and the aqueous phase extracted with EtOAc (5 mL \times 3). The organic phases were combined, dried over Na₂SO₄, filtered, and concentrated *in vacuo*. Purification of the crude *via* flash chromatography (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) gave the desired product as a pale yellow oil (29.0 mg, 93%, 97:3 e.r.).

General Procedure C: Hiyama coupling. For example, preparation of 5a.

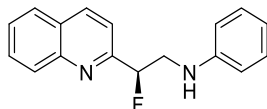


Prepared according to adapted literature procedure.²

An oven dried, 20 mL microwave vial was charged with 2-bromoquinoline (207 mg, 1.00 mmol, 1.0 equiv), (1-fluorovinyl)(methyl)diphenylsilane (356 mg, 1.5 mmol, 1.5 equiv), Pd(PPh₃)₄ (57.8 mg, 0.05 mmol, 5 mol%), CuI (9.7 mg, 0.05 mmol, 5 mol%) and CsF (357 mg, 2.35 mmol, 2.35 equiv). The microwave vial was then capped and purged with N₂ before addition of DMF (10 mL, 0.1 M). The reaction mixture was then allowed to stir at rt for 3.5 h. The reaction was quenched with water (20 mL)

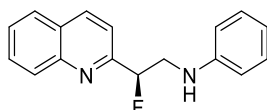
and diluted with Et₂O (20 mL). The organic phase was separated, washed with water (2 × 10 mL) and brine (10 mL), dried over Na₂SO₄, filtered, and concentrated *in vacuo*. Purification of the crude *via* flash chromatography (silica gel, Et₂O:petroleum ether, 1:99) gave the desired product as a pale yellow oil (161 mg, 93%).

General Procedure D: Catalyst e.r. vs product e.r.



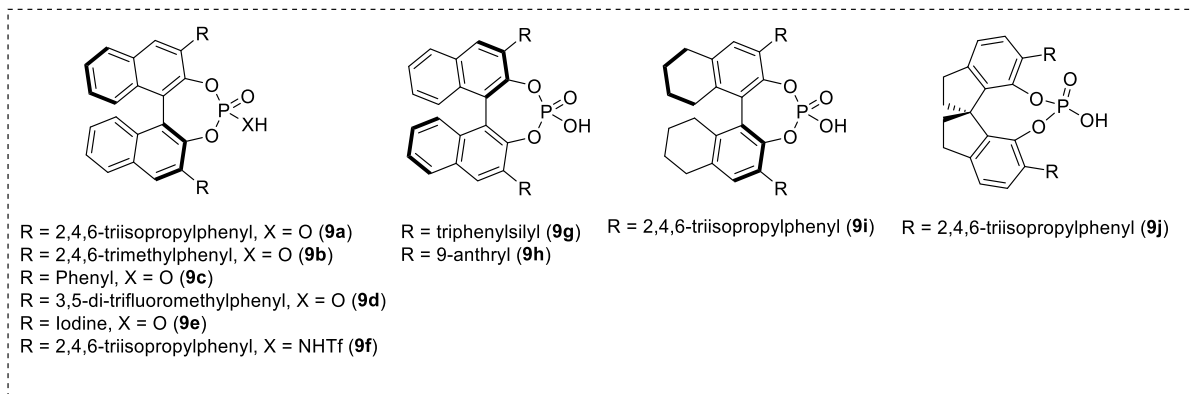
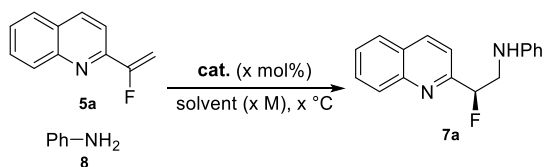
A 2 mL HPLC vial was charged with 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), and different ratios of (*S*)- and (*R*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%) (cat. e.r.: 60:40, 70:30, 80:20, 90:10 and 100:0). The vial was then capped and purged with N₂ before adding THF (200 μL, 0.5 M). The reaction mixture was cooled to -10 °C for 15 min before the addition of aniline (27 μL, 0.30 mmol, 3.0 equiv). The resulting mixture was then allowed to stir at -10 °C for 3 d before being quenched by sat. aq. K₂CO₃ solution (1 mL). The phases were separated and the aqueous phase was extracted with EtOAc (10 mL × 3). The organic phases were combined, dried over Na₂SO₄, filtered, and concentrated *in vacuo*. The e.r. of the product was determined by HPLC with an internal standard.

General Procedure E: Kinetic isotope effect



A 2mL HPLC vial was charged with 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and aniline (27 μL, 0.30 mmol, 3.0 equiv). The vial was then capped and purged with N₂ before adding *d*₈-THF (400 μL, 0.25 M). The resulting mixture was then transferred into a 0.5mm NMR tubes (7 inch) and ¹⁹F NMR was taken every 6 min for a period of 5 h to allow the determination of the conversion.

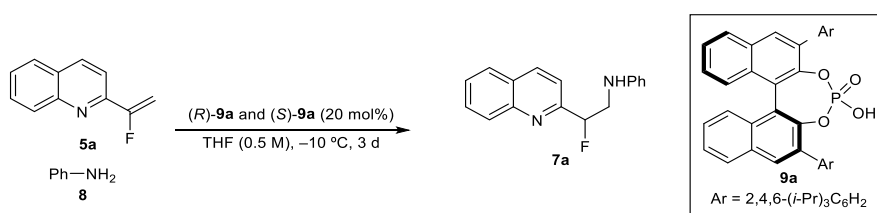
3. Optimization of Reaction Conditions



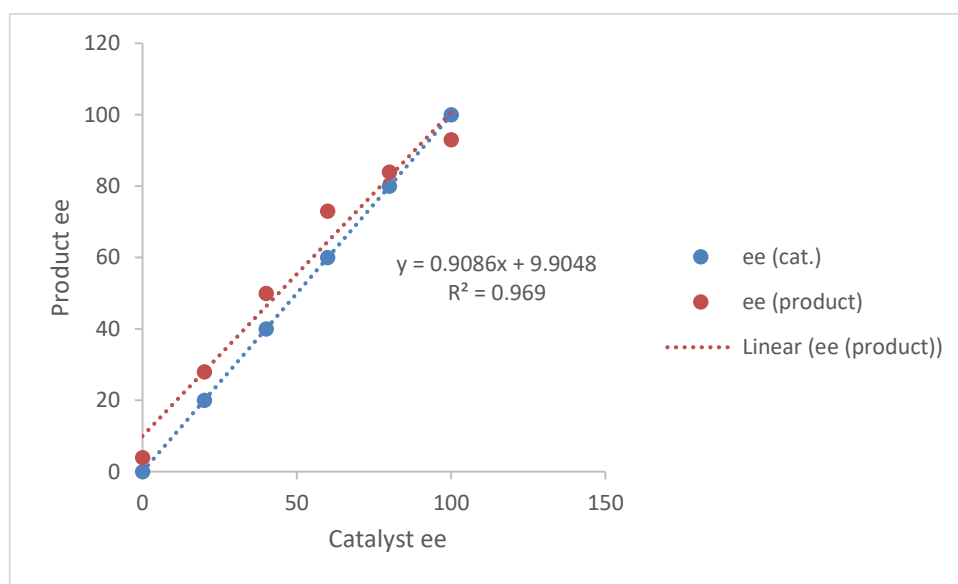
Entry	Solvent	Concentration (M)	Time (h)	Catalyst	Cat. loading (mol%)	Temp. (°C)	Aniline (equiv)	% conv. ^a (e.r.)
1	THF	0.5	48	9a	---	-20	3	0 (---)
2	THF	0.5	48	9a	20	-20	3	40 (96:4)
3	THF	0.5	96	9a	20	-20	3	68 (97:3)
4	THF	0.25	48	9a	20	-20	3	28 (97:3)
5	THF	0.1	48	9a	20	-20	3	15 (98:2)
6	THF	0.5	48	9a	20	-20	5	48 (95:5)
7	THF	0.5	48	9a	20	-20	10	58 (95:5)
8	THF	0.5	48	9a	10	-20	3	37 (95:5)
9	THF	0.5	48	9b	20	-20	3	25 (88:12)
10	THF	0.5	48	9c	20	-20	3	11 (58:42)
11	THF	0.5	48	9d	20	-20	3	n.d. (---)
12	THF	0.5	48	9e	20	-20	3	6 (55:45)
13	THF	0.5	48	9f	20	-20	3	n.d. (---)
14	THF	0.5	48	9g	20	-20	3	10 (62:38)
15	THF	0.5	48	9h	20	-20	3	25 (25:75)
16	THF	0.5	48	9i	20	-20	3	28 (89:11)
17	THF	0.5	48	9j	20	-20	3	n.d. (---)
18	THF	0.5	48	9a	20	-10	3	68 (96:4)
19	THF	0.5	72	9a	20	-10	3	82^b (96:4)
20	CPME	0.5	72	9a	20	-10	3	89^b (95:5)
21	hexane	0.5	72	9a	20	-10	3	98 (71:29)
22	toluene	0.5	72	9a	20	-10	3	95 (89:11)
23	Et ₂ O	0.5	72	9a	20	-10	3	94 (94:6)
24	CH ₂ Cl ₂	0.5	72	9a	20	-10	3	70 (75:25)
25	CPME	0.5	120	9a	20	-20	3	81 (95:5)
26	CPME	0.5	120	9a	10	-20	3	68 (93:7)

^a Determined by HPLC analysis with internal standard; ^b isolated yield.

4. Catalyst ee vs. Product ee



A series of mixtures of *(R)*-TRIP and *(S)*-TRIP were made in varying ratios. Reactions were carried out following General Procedure D, the enantioenrichment of each product was determined by HPLC analysis (Chiralpak IA, hexane/*i*-PrOH = 90/10, flow rate 0.5 mL/min, $\lambda = 250$ nm), t_r (major) = 23.4 min, t_r (minor) = 25.6 min. R^2 suggests no non-linear effects in these initial experiments.



5. Mechanistic Study

5.1 DFT Calculations

Conformation Generation

The starting geometry for the catalyst was adapted from the crystal structure of pyridinium complex in the crystal structure with Cambridge database code CAZKEU.³ It was assumed that the NH-O=P interaction present in this and all comparable crystal structures was retained and conformations of key structures were generated manually and where they yielded an optimized structure they are given below.

To study the conformational space of the bound complexes, we considered complexes with the catalyst bearing a proton on either of the two protruding phosphate oxygens (distal and proximal) and assumed a hydrogen bond with the quinoline nitrogen to be a critical interaction. Given this constraint, a full conformational exploration of both enantiomers of the product complex was undertaken. Two orientations of the quinoline ring were considered: one in which a secondary interaction between the the distal phosphate oxygen and the CH *peri* to the quinoline nitrogen was in place and one in which a 180° rotation about the POH...N_{quinoline} hydrogen bond takes place. All rotamers of the CHFCH₂NHPh sidechain that could be sterically accommodated by the catalyst were attempted and *i*-Pr groups at ortho positions in the catalyst were retained in their preferred conformation while the *i*-Pr at the para position was adjusted to minimise clashing with the bound species (the geom=nocrowd option in Gaussian 09⁴ was used in order to permit the relaxation of species featuring a small amount of steric clashing).

Calculations involving complexes containing chiral phosphoric acids

All frequency and geometry calculations were carried out using hybrid QM/MM calculations using the ONIOM method with B3LYP functional with the 6-31G** basis set for all atoms arising from aniline or quinoline and the PO₄H from the catalyst, and UFF is applied to the remainder of the catalyst. Additional single-point energy calculations were carried out on the previously optimised geometries using the M06-2X functional with the 6-31G** basis set employing a PCM solvation in diethyl ether and using an ultrafine grid. All calculations using the B3LYP, UFF or M06-2X functional have been performed using Gaussian 09.⁴

Calculations involving complexes not containing chiral phosphoric acids

All frequency and geometry calculations were carried out using the B3LYP functional with the 6-31G** basis set. Additional single-point energy calculations were carried out on the previously optimised geometries using the M06-2X functional with the 6-31G** basis set employing a PCM solvation in diethyl ether and using an ultrafine grid. All calculations using the B3LYP or M06-2X functional have been performed using Gaussian 09.⁴

Free energy calculations

Free energies are calculated from the generated electronic energies using the GoodVibes program (version 2.0.3)⁵ with a standard state of 1M concentration and temperature of 253K.

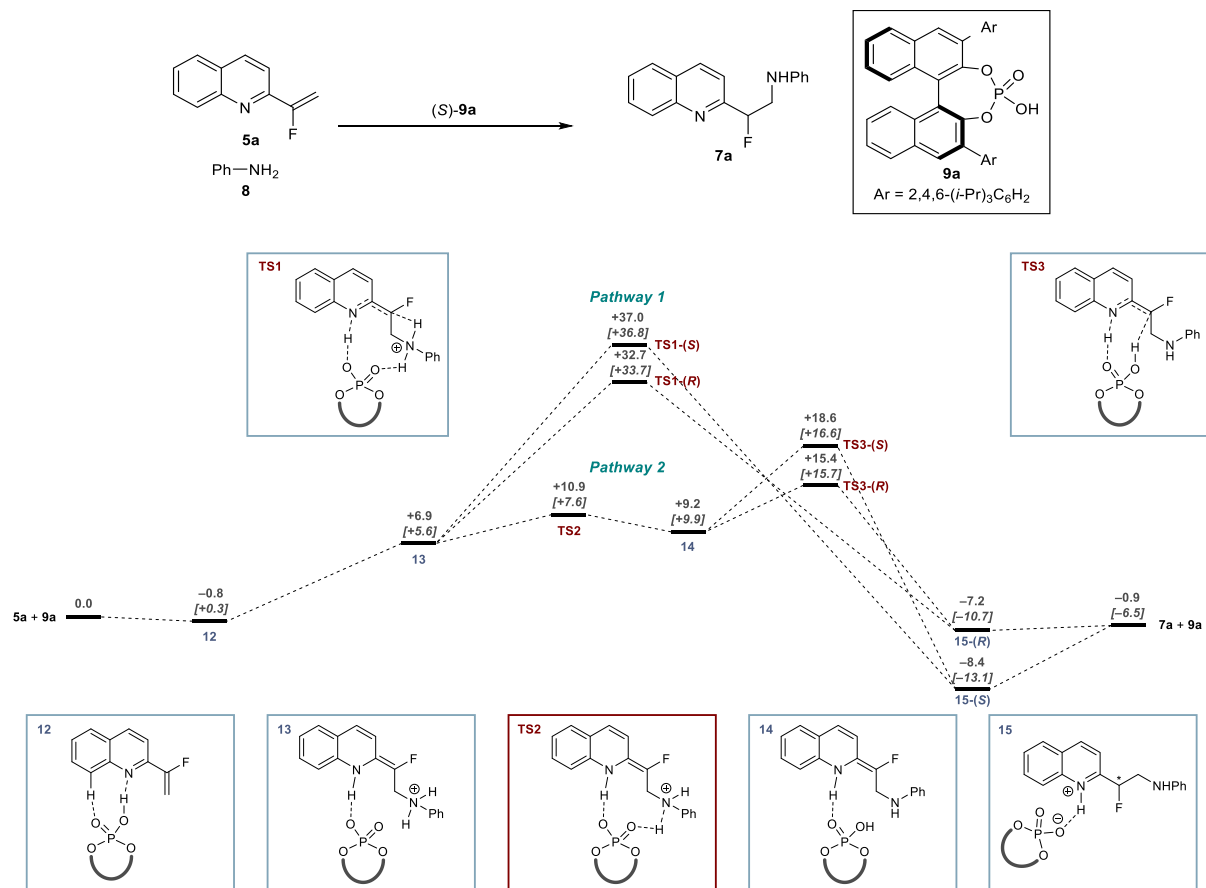
Transition State Generation

All of the optimized complexes that were within 3 kcal mol⁻¹ of the complex with the lowest energy were then subjected to a scan in which the proton at the stereocentre was moved towards either, the aniline nitrogen, the nearest (proximal) O=P or the furthest (distal) O=P. Each of these scans were then

used to initiate transition state optimizations. Subsequently, the lowest energy transition state structure was the starting point for a limited exploration of each of the steps that occurs prior to the stereoselectivity-determining step.

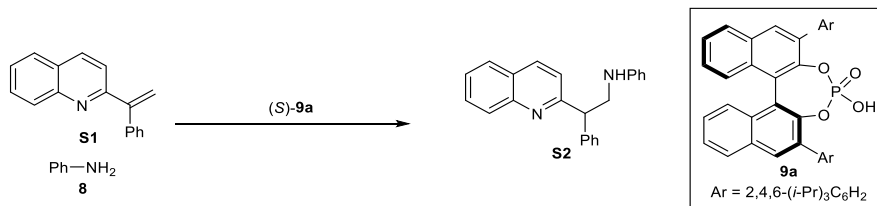
5.1.1 Reaction Profile – 5a

Free energy profile comparison (ONIOM (B3LYP/6-31G**:UFF), [single-point energy M06-2X/6-31G**(+PCM for Et₂O)]).

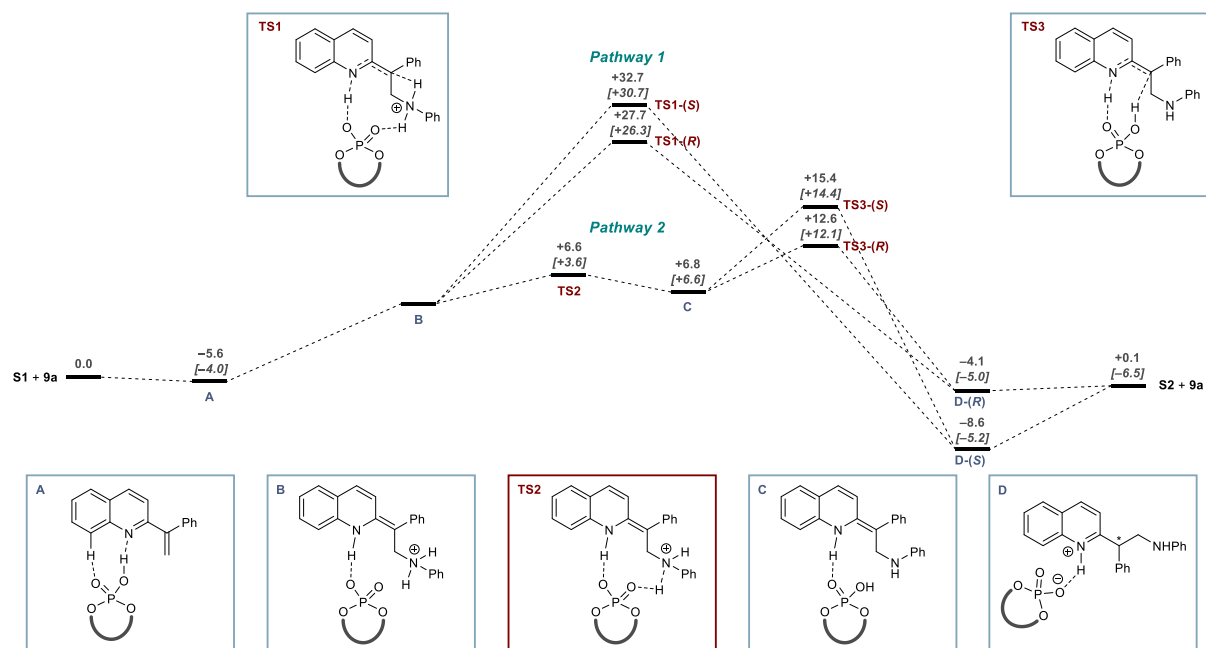


5.1.2 Reaction Profile – S1 (Phenyl System)

The reaction of compound **S1** studied previously was revisited with the insights provided by the calculations described in the main text.

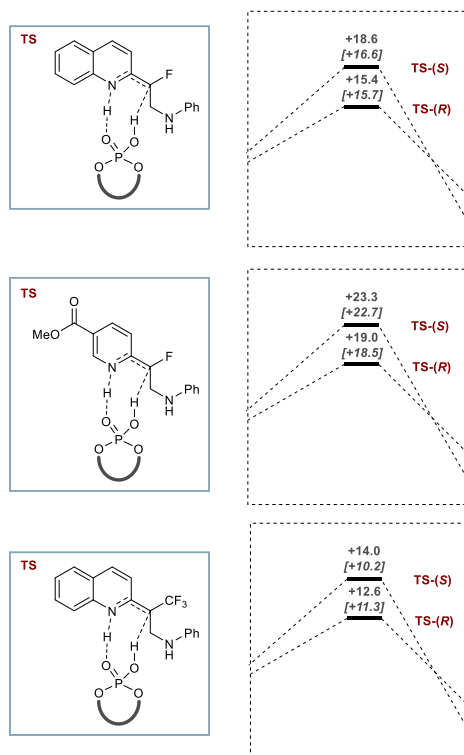


Free energy profile comparison (ONIOM (B3LYP/6-31G**:(UFF), [single-point energy M06-2X/6-31G**(+PCM for Et₂O)]).



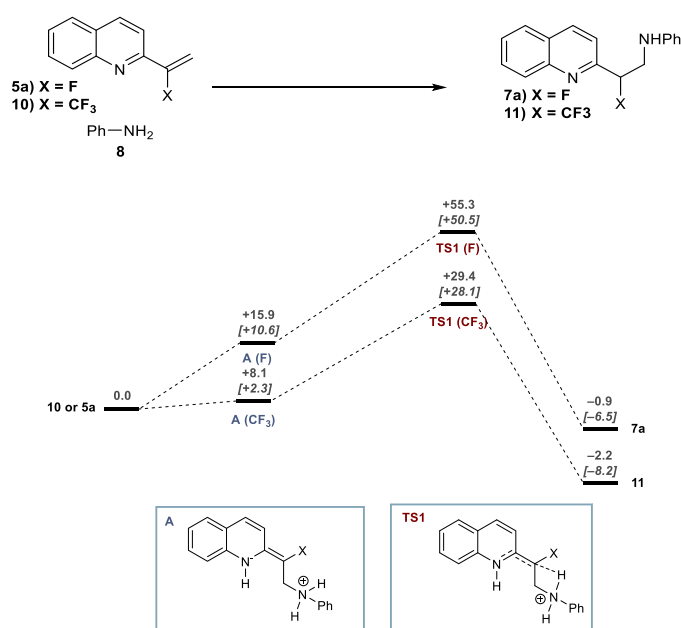
5.1.2 Reaction Profile – RDS Energy profile

A series of calculations, demonstrating the difference in energy for the RDS of different substrates. Free energy profile comparison (ONIOM (B3LYP/6-31G**:(UFF), [single-point energy M06-2X/6-31G**(+PCM for Et₂O)]).



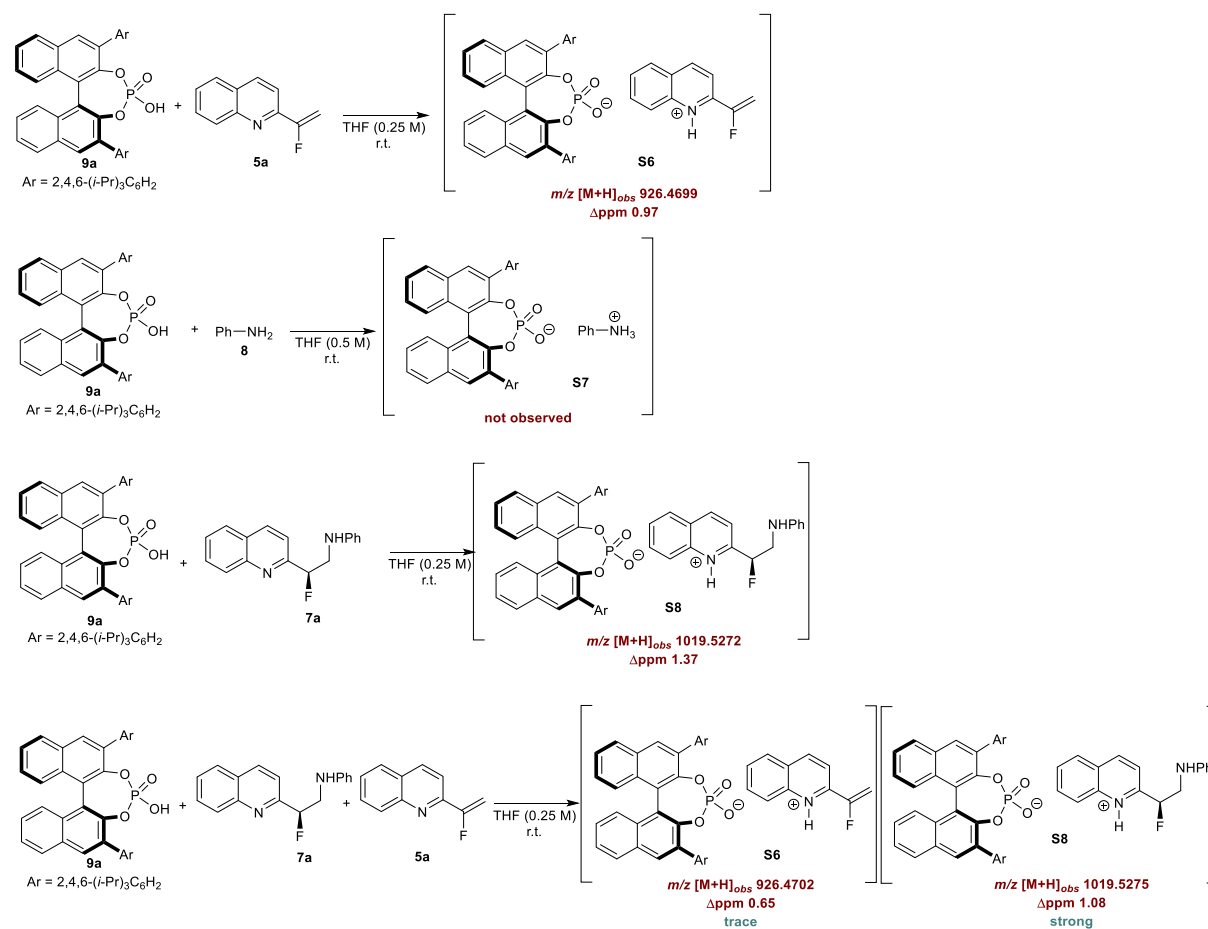
5.1.5 Background Reaction Profile – 5a & 10

A series of calculations, demonstrating the high energy barrier RDS for the uncatalyzed reaction, and decrease in the vinyl CF₃ system. Free energy profile comparison (ONIOM (B3LYP/6-31G**:(UFF), [single-point energy M06-2X/6-31G**(+PCM for Et₂O)]).



5.2 HRMS Study

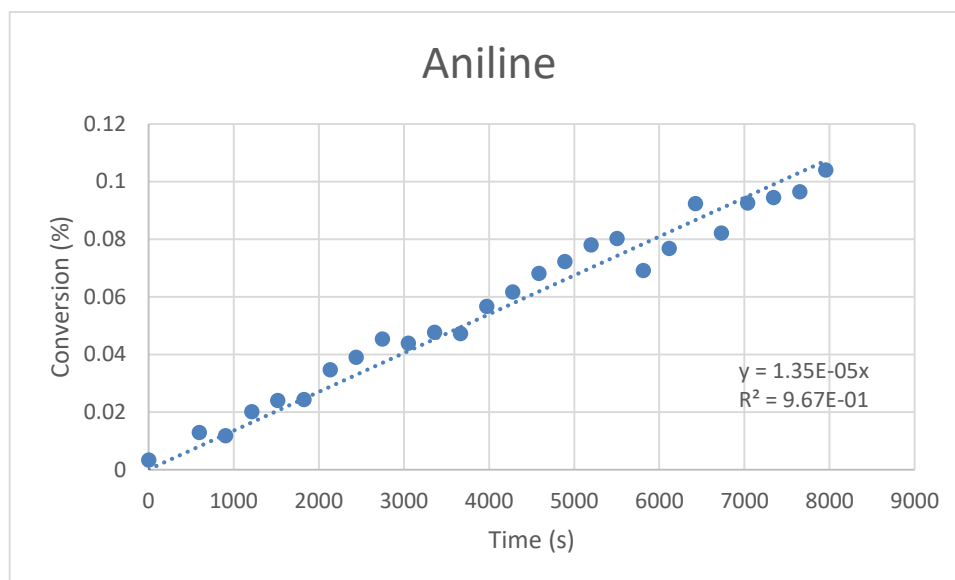
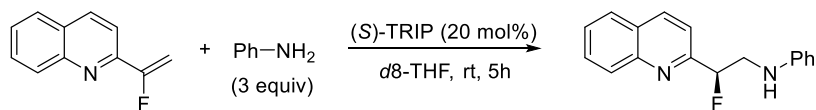
A series of reactions were analyzed by ESI analysis. Results suggest the product **7a** has the largest affinity towards CPA **9a** as compared to aniline **8** and quinoline **5a**.



5.3 Kinetic Isotope Effect Study

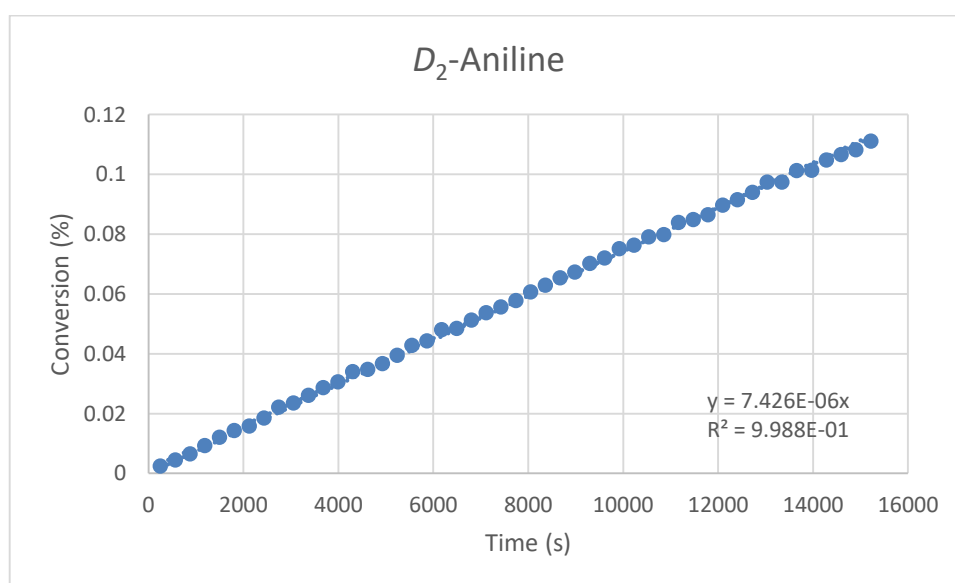
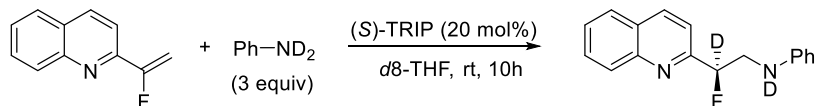
5.3.1 Kinetic Isotope Effect of Aniline

Data was obtained according to General Procedure E using aniline.



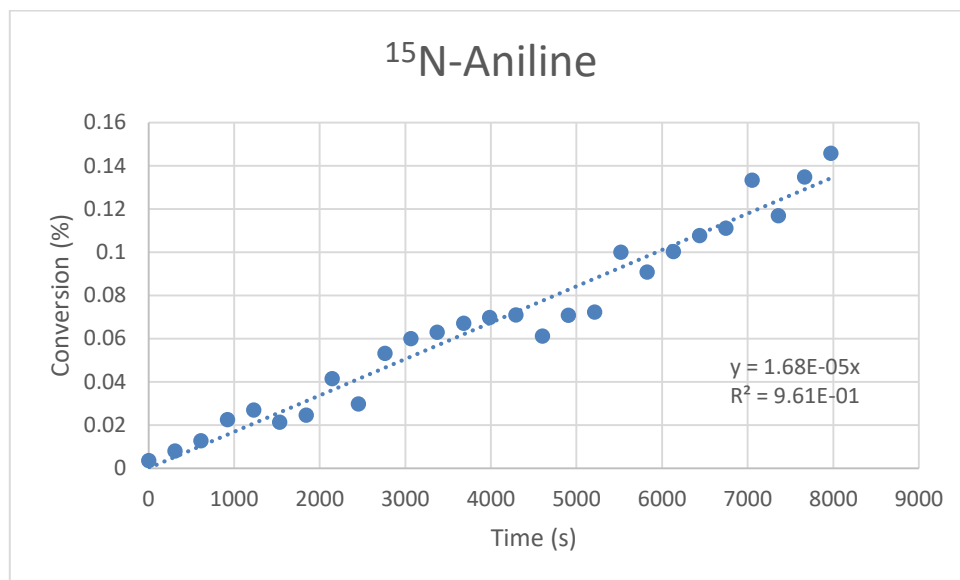
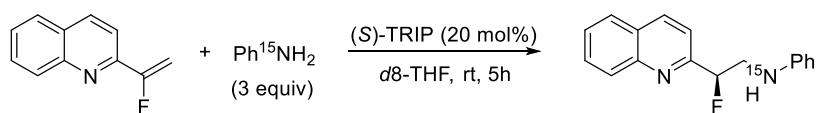
5.3.2 Kinetic Isotope Effect of D₂-Aniline

Data was obtained according to General Procedure E using D₂-aniline for 10 h.



5.3.3 Kinetic Isotope Effect of ¹⁵N-Aniline

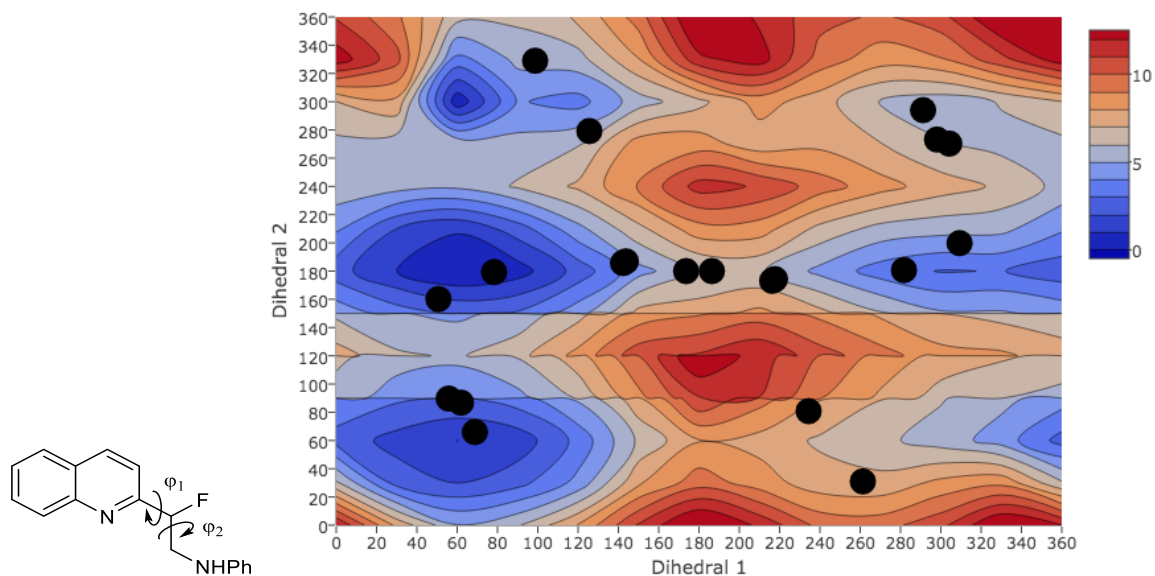
Data was obtained according to General Procedure E using ^{15}N -aniline.



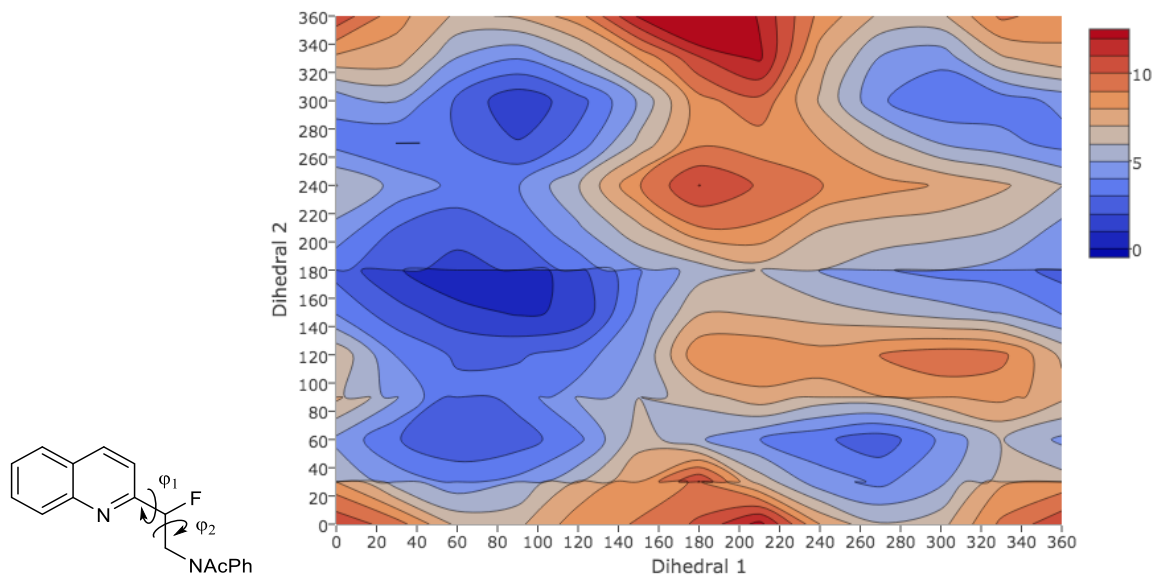
6. Ramachandran Plots

Ramachandran plots were computed at the M06-2X/6-31+G** level of theory. Calculations were performed in Gaussian09 using the modified redundant coordinate facility with dihedral angles being moved in 15° increments. Energy in kcal mol⁻¹.

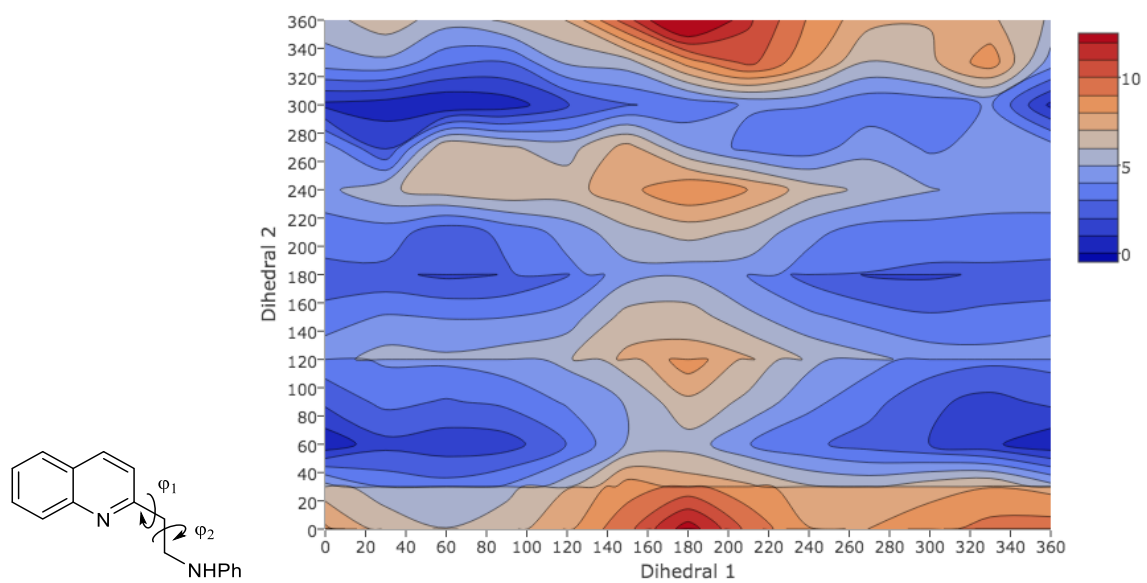
6.1 Ramachandran Plot for Compound 9a



6.2 Ramachandran Plot for Compound 16

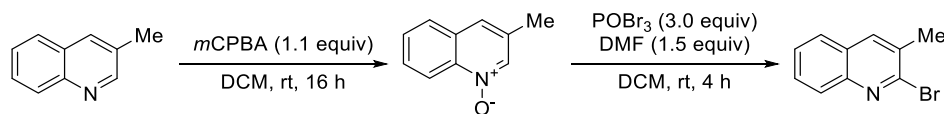


6.3 Ramachandran Plot for *N*-(2-(Quinolin-2-yl)ethyl)aniline



7. Characterization Data for Compounds

2-Bromo-3-methylquinoline S3



Prepared according to literature procedures.^{6,7}

An oven-dried, 250 mL round bottomed flask was charged with 3-methylquinoline (1.34 mL, 10.0 mmol, 1.0 equiv). The vessel was sealed and purged with N₂. CH₂Cl₂ (50 mL, 0.2 M) was added to the vessel and the mixture was stirred at 0 °C for 15 min before the addition of *m*CPBA (1.91 g, 11.0 mmol, 1.1 equiv). The reaction was left to stir at rt for 16 h. The mixture was then diluted with CH₂Cl₂ (50 mL), and washed with aq. KOH 6 M solution (3 × 50 mL). The organic layer was collected, dried over Na₂SO₄, filtered, and concentrated *in vacuo*.

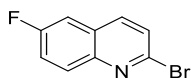
NB. Calculations based upon quantitative conversion to 2-bromo-3-methylquinoline 1-oxide in the previous step. An oven-dried, 250 mL round bottomed flask was charged with the crude 3-methylquinoline *N*-oxide (10.0 mmol, 1.0 equiv) and POBr₃ (8.60 g, 30.0 mmol, 3.0 equiv). The vessel was sealed and purged with N₂. Dry CH₂Cl₂ (50 mL, 0.2 M) and dry DMF (1.16 mL, 1.50 mmol, 1.5 equiv) were added. The reaction was left to stir at rt for 4 h. Sat. aq. Na₂SO₄ solution was added slowly until the pH was 7-8. The layers in the resulting mixture were separated and the aqueous was extracted with CH₂Cl₂ (3 × 50 mL). The organic phases were combined, dried over Na₂SO₄, filtered, and concentrated *in vacuo*. Purification of the crude product *via* flash chromatography (silica gel, EtOAc:petroleum ether, 1:99) gave the desired product as a white solid (506 mg, 23%).

¹H NMR (500 MHz, CDCl₃): δ 8.00 (d, *J* = 8.5 Hz, 1H), 7.90 (s, 1H), 7.71 (d, *J* = 8.1 Hz, 1H), 7.68 – 7.62 (m, 1H), 7.58 – 7.47 (m, 1H), 2.52 (d, *J* = 1.1 Hz, 3H).

¹³C NMR (126 MHz, CDCl₃): δ 147.1, 145.6, 137.1, 132.4, 129.6, 128.4, 127.8, 127.2, 127.0, 22.5.

Spectroscopic data in agreement with literature values.⁶

2-Bromo-6-fluoroquinoline S4



Prepared according to a literature procedure.⁶

An oven-dried, 10 mL microwave vial was charged with 6-fluoroquinolin-2(1*H*)-one (226 mg, 1.00 mmol, 1.0 equiv) and POBr₃ (56.2 mg, 1.10 mmol, 1.1 equiv). The vessel was capped and purged with N₂. The mixture was heated to 140 °C for 3 h, then cooled to rt and poured over ice. The resulting precipitate was collected, washed with petroleum ether and dried *in vacuo* to give the desired product as a pale brown solid (114 mg, 50%).

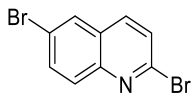
¹H NMR (400 MHz, CDCl₃): δ 8.04 (dd, *J* = 9.2, 5.2 Hz, 1H), 7.96 (d, *J* = 8.6 Hz, 1H), 7.54 (d, *J* = 8.6 Hz, 1H), 7.50 (td, *J* = 8.8, 2.8 Hz, 1H), 7.43 (dd, *J* = 8.6, 2.8 Hz, 1H).

¹³C NMR (126 MHz, CDCl₃): δ 160.7 (d, ¹*J* = 249.5 Hz), 145.7, 141.1, 137.8 (d, ⁴*J* = 5.2 Hz), 131.3 (d, ³*J* = 9.3 Hz), 127.8 (d, ³*J* = 10.3 Hz), 126.7, 120.9 (d, ²*J* = 25.5 Hz), 111.2 (d, ²*J* = 22.2 Hz).

¹⁹F NMR (376 MHz, CDCl₃): δ -111.94.

Spectroscopic data in agreement with literature values.⁷

2,6-Dibromoquinoline **S5**



Prepared according to a literature procedure.⁶

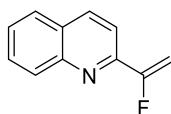
An oven-dried, 10 mL microwave vial was charged with 6-bromoquinolin-2(1H)-one (251 mg, 1.13 mmol, 1.13 equiv) and POBr₃ (287 mg, 1.00 mmol, 1.0 equiv). The vessel was capped and purged with N₂. The mixture was heated to 140 °C for 3 h, then cooled to rt and poured over ice. The resulting precipitate was collected, washed with petroleum ether and dried *in vacuo* to give the desired product as a brown solid (178 mg, 62%).

¹H NMR (500 MHz, CDCl₃): δ 7.98 (d, J = 2.2 Hz, 1H), 7.91 (dd, J = 8.7, 2.0 Hz, 2H), 7.80 (dd, J = 8.9, 2.2 Hz, 1H), 7.55 (d, J = 8.6 Hz, 1H).

¹³C NMR (126 MHz, CDCl₃): δ 147.3, 142.4, 137.4, 134.2, 130.5, 129.9, 128.2, 126.9, 121.2.

Spectroscopic data in agreement with literature values.⁶

2-(1-Fluorovinyl)quinoline **5a**



Prepared according to General Procedure C using 2-bromoquinoline (207 mg, 1.00 mmol, 1.0 equiv), (1-fluorovinyl)(methyl)diphenylsilane (356 mg, 1.50 mmol, 1.5 equiv), Pd(PPh₃)₄ (59.0 mg, 0.05 mmol, 5 mol%), CuI (9.7 mg, 0.05 mmol, 5 mol%), CsF (364 mg, 2.35 mmol, 2.35 equiv), and DMF (10 mL, 0.1 M). The reaction mixture was subjected to the purification outlined in the General Procedure C (silica gel, Et₂O:petroleum ether, 1:99) to afford the desired product as a pale yellow oil (161 mg, 93%).

ν_{\max} (film): 3056, 2926, 1654, 1284, 1108, 834, 758 cm⁻¹.

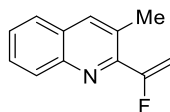
¹H NMR (400 MHz, CDCl₃): δ 8.13 (d, J = 8.6 Hz, 1H), 8.10 (d, J = 8.5 Hz, 1H), 7.81 (dd, J = 8.1, 0.9 Hz, 1H), 7.73 (ddd, J = 8.4, 6.9, 1.4 Hz, 1H), 7.69 (dd, J = 8.6, 1.2 Hz, 1H), 7.54 (ddd, J = 8.1, 6.9, 1.1 Hz, 1H), 5.87 (dd, $^3J_{\text{HF}}$ = 48.6 Hz, $^2J_{\text{HH}}$ = 2.9 Hz, 1H), 5.14 (dd, $^3J_{\text{HF}}$ = 16.4 Hz, $^2J_{\text{HH}}$ = 2.9 Hz, 1H).

¹³C NMR (126 MHz, CDCl₃): δ 162.3 (d, 1J = 251.2 Hz), 150.1 (d, 2J = 35.4 Hz), 147.9 (d, 4J = 3.8 Hz), 136.9, 130.0, 129.8, 128.0, 127.6, 127.1, 116.8 (d, 3J = 4.7 Hz), 93.8 (d, 2J = 18.4 Hz).

¹⁹F NMR (376 MHz, CDCl₃): δ -114.28.

HRMS (ESI): exact mass calculated for [M+H]⁺ (C₁₁H₉NF) requires m/z 174.0714, found m/z 174.0712.

2-(1-Fluorovinyl)-3-methylquinoline **5b**



Prepared according to General Procedure C using 2-bromo-3-methylquinoline (99.5 mg, 0.45 mmol, 1.0 equiv), (1-fluorovinyl)(methyl)diphenylsilane (164 mg, 0.675 mmol, 1.5 equiv), Pd(PPh₃)₄ (520 mg, 0.45 mmol, 1.0 equiv), CuI (85.7 mg, 0.45 mmol, 1.0 equiv), CsF (161 mg, 1.06 mmol, 2.35 equiv), and DMF (4.5 mL, 0.1 M). The reaction mixture was subjected to the purification outlined in General Procedure C (silica gel, Et₂O:petroleum ether, 1:99) to afford the desired product as a pale yellow oil (75.4 mg, 90%).

ν_{\max} (film): 2361, 2342, 1655, 1493, 1447, 1327, 1273, 1184, 1138, 1065, 930, 903, 860, 791, 756, 729 cm⁻¹.

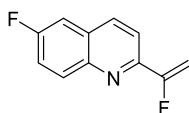
¹H NMR (500 MHz, CDCl₃): δ 8.08 (dd, J = 8.4, 1.3 Hz, 1H), 7.97 (s, 1H), 7.75 (dd, J = 8.2, 1.5 Hz, 1H), 7.67 (ddd, J = 8.4, 6.8, 1.5 Hz, 1H), 7.53 (ddd, J = 8.1, 6.8, 1.2 Hz, 1H), 5.37 (dd, $^3J_{\text{HF}}$ = 48.2 Hz, $^2J_{\text{HH}}$ = 3.0 Hz, 1H), 5.19 (dd, $^3J_{\text{HF}}$ = 16.8 Hz, $^2J_{\text{HH}}$ = 3.0 Hz, 1H), 2.60 (d, J = 4.9 Hz, 3H).

¹³C NMR (126 MHz, CDCl₃): δ 163.9 (d, 1J = 256.3 Hz), 151.2 (d, 2J = 32.3 Hz), 146.2, 146.2, 137.6, 129.6, 129.2, 128.4, 127.5, 126.8, 96.3 (d, 2J = 19.2 Hz), 20.0 (d, 4J = 7.6 Hz).

¹⁹F NMR (470 MHz, CDCl₃): δ -99.90.

HRMS (ESI): exact mass calculated for [M+H]⁺ (C₁₂H₁₁FN) requires m/z 188.0870, found m/z 188.0866.

6-Fluoro-2-(1-fluorovinyl)quinoline **5c**



Prepared according to General Procedure C using 6-fluoro-2-bromoquinoline (150 mg, 0.664 mmol, 1.0 equiv), (1-fluorovinyl)(methyl)diphenylsilane (241 mg, 0.995 mmol, 1.5 equiv), Pd(PPh₃)₄ (38.3 mg, 0.033 mmol, 5 mol%), CuI (6.3 mg, 0.033 mmol, 5 mol%), CsF (237 mg, 1.56 mmol, 2.35 equiv), and DMF (6.6 mL, 0.1 M). The reaction mixture was subjected to the purification outlined in General Procedure C (silica gel, Et₂O:petroleum ether, 1:99) to afford the desired product as a pale yellow oil (84.0 mg, 66%).

ν_{\max} (film): 1657, 1626, 1603, 1558, 1501, 1481, 1379, 1304, 1285, 1244, 1221, 1144, 1101, 916, 866, 829 cm⁻¹.

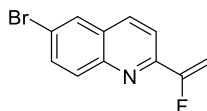
¹H NMR (400 MHz, CDCl₃): δ 8.12 (d, J = 8.7 Hz, 1H), 8.08 (dd, J = 9.3, 5.3 Hz, 1H), 7.68 (d, J = 8.7 Hz, 1H), 7.49 (td, J = 8.9, 2.8 Hz, 1H), 7.40 (dd, J = 8.7, 2.9 Hz, 1H), 5.84 (dd, $^3J_{\text{HF}}$ = 48.6 Hz, $^2J_{\text{HH}}$ = 3.0 Hz, 1H), 5.13 (dd, $^3J_{\text{HF}}$ = 16.4 Hz, $^2J_{\text{HH}}$ = 3.1 Hz, 1H).

¹³C NMR (126 MHz, CDCl₃): δ 162.0 (d, ¹J = 251.0 Hz), 160.8 (d, ¹J = 249.4 Hz), 149.6, (dd, ²J = 35.8, 3.1 Hz), 145.0 (d, ⁴J = 4.0 Hz), 136.4 (d, ³J = 6.6 Hz), 132.3 (d, ³J = 9.1 Hz), 128.6 (d, ³J = 10.1 Hz), 120.5 (d, ²J = 26.0 Hz), 117.6 (d, ⁴J = 4.7 Hz), 110.6 (d, ²J = 21.8 Hz), 93.8 (d, ²J = 18.5 Hz).

¹⁹F NMR (376 MHz, CDCl₃): δ -112.23, -114.42.

HRMS (ESI): exact mass calculated for [M+H]⁺ (C₁₁H₈F₂N) requires *m/z* 192.0619, found *m/z* 192.0618.

6-Bromo-2-(1-fluorovinyl)quinoline **5d**



Prepared according to General Procedure C using 2,6-dibromoquinoline (259 mg, 0.90 mmol, 1.0 equiv), (1-fluorovinyl)(methyl)diphenylsilane (328 mg, 1.38 mmol, 1.5 equiv), Pd(PPh₃)₄ (52.0 mg, 0.045 mmol, 5 mol%), CuI (8.6 mg, 0.045 mmol, 5 mol%), CsF (322 mg, 2.12 mmol, 2.35 equiv), and DMF (9 mL, 0.1 M). The reaction mixture was subjected to the purification outlined in General Procedure C (silica gel, Et₂O:petroleum ether, 1:99) to afford the desired product as a pale yellow solid (183 mg, 81%).

ν_{max} (film): 1659, 1595, 1489, 1283, 1190, 1107, 920, 874, 827 cm⁻¹.

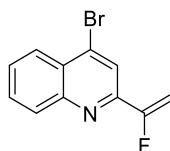
¹H NMR (400 MHz, CDCl₃): δ 8.11 (d, *J* = 8.6 Hz, 1H), 7.97 (d, *J* = 2.2 Hz, 1H), 7.95 (d, *J* = 9.0 Hz, 1H), 7.78 (dd, *J* = 9.0, 2.2 Hz, 1H), 7.70 (dd, *J* = 8.7, 1.3 Hz, 1H), 5.87 (dd, ³J_{HF} = 48.5 Hz, ²J_{HH} = 3.1 Hz, 1H), 5.16 (dd, ³J_{HF} = 16.3 Hz, ²J_{HH} = 3.1 Hz, 1H).

¹³C NMR (126 MHz, CDCl₃): δ 161.9 (d, ¹J = 251.1 Hz), 150.5 (d, ²J = 35.6 Hz), 146.5 (d, ⁴J = 4.0 Hz), 136.1, 133.7, 131.5, 129.7, 129.1, 121.2, 117.8 (d, ³J = 4.6 Hz), 94.4 (d, ²J = 18.3 Hz).

¹⁹F NMR (376 MHz, CDCl₃): δ -114.60.

HRMS (ESI): exact mass calculated for [M+H]⁺ (C₁₁H₈BrFN) requires *m/z* 251.9819, found *m/z* 251.9816.

4-Bromo-2-(1-fluorovinyl)quinoline **5e**



Prepared according to General Procedure C using 2,4-dibromoquinoline (259 mg, 0.90 mmol, 1.0 equiv), (1-fluorovinyl)(methyl)diphenylsilane (328 mg, 1.35 mmol, 1.5 equiv), Pd(PPh₃)₄ (52.0 mg, 0.045 mmol, 5 mol%), CuI (8.6 mg, 0.045 mmol, 5 mol%), CsF (322 mg, 2.12 mmol, 2.35 equiv), and DMF (9 mL, 0.1 M). The reaction mixture was subjected to the purification outlined in General Procedure C (silica gel, Et₂O:petroleum ether, 1:99) to afford the desired product as a pale yellow solid (105 mg, 42%).

ν_{max} (film): 2357, 1578, 1547, 1491, 1404, 1339, 1265, 1251, 1148, 1105, 928 cm⁻¹.

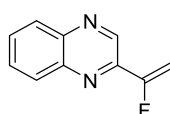
¹H NMR (500 MHz, CDCl₃): δ 8.12 (dd, *J* = 8.5, 1.4 Hz, 1H), 8.05 (dd, *J* = 8.5, 1.2 Hz, 1H), 7.94 (d, *J* = 1.2 Hz, 1H), 7.74 (ddd, *J* = 8.4, 6.8, 1.4 Hz, 1H), 7.60 (ddd, *J* = 8.2, 6.8, 1.2 Hz, 1H), 5.89 (dd, ³*J*_{HF} = 48.4 Hz, ²*J*_{HH} = 3.1 Hz, 1H), 5.16 (dd, ³*J*_{HF} = 16.2 Hz, ²*J*_{HH} = 3.1 Hz, 1H).

¹³C NMR (126 MHz, CDCl₃): δ 161.0 (d, ¹*J* = 251.5 Hz), 149.8 (d, ²*J* = 36.1 Hz), 148.3 (d, ⁴*J* = 4.0 Hz), 135.0 (d, ⁴*J* = 1.7 Hz), 130.9, 130.2, 128.3, 127.5, 126.7, 120.9 (d, ³*J* = 5.1 Hz), 94.7 (d, ²*J* = 17.9 Hz).

¹⁹F NMR (470 MHz, CDCl₃): δ -114.71.

HRMS (ESI): exact mass calculated for [M+H]⁺ (C₁₁H₈BrFN) requires *m/z* 251.9819, found *m/z* 251.9817.

2-(1-Fluorovinyl)quinoxaline **5f**



Prepared according to General Procedure C using 2-bromoquinoxaline (174 mg, 1.00 mmol, 1.0 equiv), (1-fluorovinyl)(methyl)diphenylsilane (363 mg, 1.50 mmol, 1.5 equiv), Pd(PPh₃)₄ (57.8 mg, 0.05 mmol, 5 mol%), CuI (6.6 mg, 0.05 mmol, 5 mol%), CsF (448 mg, 2.35 mmol, 2.35 equiv), and DMF (10 mL, 0.1 M). The reaction mixture was subjected to the purification outlined in General Procedure C (silica gel, Et₂O:petroleum ether, 1:99) to afford the desired product as a pale yellow solid (158 mg, 91%).

ν_{max} (film): 1655, 1506, 1489, 1364, 1341, 1105, 968, 924, 874, 764, 743 cm⁻¹.

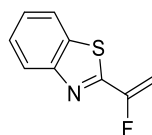
¹H NMR (400 MHz, CDCl₃): δ 9.13 (s, 1H), 8.16 – 8.07 (m, 2H), 7.87 – 7.74 (m, 2H), 5.92 (dd, ³*J*_{HF} = 48.2 Hz, ²*J*_{HH} = 3.4 Hz, 1H), 5.27 (dd, ³*J*_{HF} = 16.5 Hz, ²*J*_{HH} = 3.4 Hz, 1H).

¹³C NMR (126 MHz, CDCl₃): δ 160.7 (d, ¹*J* = 250.8 Hz), 145.0 (d, ²*J* = 34.0 Hz), 142.6, 141.9, 141.3 (d, ³*J* = 5.3 Hz), 130.9, 130.7, 129.8, 129.4, 95.9 (d, ²*J* = 17.3 Hz).

¹⁹F NMR (376 MHz, CDCl₃): δ -116.24.

HRMS (ESI): exact mass calculated for [M+H]⁺ (C₁₀H₈FN₂) requires *m/z* 175.0666, found *m/z* 175.0665.

2-(1-Fluorovinyl)benzo[d]thiazole **5g**



Prepared according to General Procedure C using 2-Bromobenzo[d]thiazole (214 mg, 1.00 mmol, 1.0 equiv), (1-fluorovinyl)(methyl)diphenylsilane (356 mg, 1.50 mmol, 1.5 equiv), Pd(PPh₃)₄ (1.16 g, 1.00 mmol, 1.0 equiv), CuI (132 mg, 1.00 mmol, 1.0 equiv), CsF (364 mg, 2.40 mmol, 2.4 equiv), and DMF (10 mL, 0.1 M). The reaction mixture was subjected to the purification outlined in General Procedure C (silica gel, Et₂O:petroleum ether, 1:99) to afford the desired product as a pale yellow solid (73.8 mg, 42%).

ν_{\max} (film): 2955, 2926, 2853, 1647, 1558, 1435, 1314, 1290, 1261, 1092, 760, 729, 700 cm^{-1} .

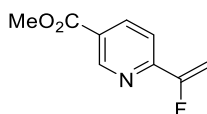
$^1\text{H NMR}$ (500 MHz, CDCl_3): δ 8.08 (d, $J = 8.2$ Hz, 1H), 7.92 (d, $J = 8.0$ Hz, 1H), 7.64 – 7.36 (m, 3H), 5.74 (dd, $^3J_{\text{HF}} = 47.1$ Hz, $^2J_{\text{HH}} = 3.7$ Hz, 1H), 5.21 (dd, $^3J_{\text{HF}} = 15.5$ Hz, $^2J_{\text{HH}} = 3.8$ Hz, 1H).

$^{13}\text{C NMR}$ (126 MHz, CDCl_3): δ 159.9 (d, $^2J = 39.3$ Hz), 157.3 (d, $^1J = 247.7$ Hz), 153.5 (d, $^4J = 2.0$ Hz), 135.0, 126.8, 126.1, 123.8, 121.9, 95.5 (d, $^2J = 17.2$ Hz).

$^{19}\text{F NMR}$ (470 MHz, CDCl_3): δ -106.57.

HRMS (ESI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_9\text{H}_7\text{FNS}$) requires m/z 180.0278, found m/z 180.0277.

Methyl 6-(1-fluorovinyl)nicotinate **5h**



Prepared according to General Procedure C using 6-bromonicotinate (216 mg, 1.00 mmol, 1.0 equiv), (1-fluorovinyl)(methyl)diphenylsilane (356 mg, 1.50 mmol, 1.5 equiv), $\text{Pd}(\text{PPh}_3)_4$ (59.0 mg, 0.05 mmol, 5 mol%), CuI (9.7 mg, 0.05 mmol, 5 mol%), CsF (364 mg, 2.40 mmol, 2.4 equiv), and DMF (10 mL, 0.1 M). The reaction mixture was subjected to the purification outlined in General Procedure C (silica gel, Et_2O :petroleum ether, 1:99) to afford the desired product as a pale yellow solid (164 mg, 91%).

ν_{\max} (film): 1722, 1655, 1593, 1560, 1437, 1366, 1287, 1273, 1196, 1130, 1101, 1022, 920, 897, 858, 789, 733 cm^{-1} .

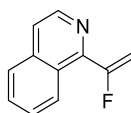
$^1\text{H NMR}$ (400 MHz, CDCl_3): δ 9.14 – 9.13 (m, 1H), 8.31 (dd, $J = 8.2, 2.1$ Hz, 1H), 7.59 (d, $J = 7.9$ Hz, 1H), 5.84 (dd, $^3J_{\text{HF}} = 48.5$ Hz, $^2J_{\text{HH}} = 2.9$ Hz, 1H), 5.11 (dd, $^3J_{\text{HF}} = 16.3$ Hz, $^2J_{\text{HH}} = 2.9$ Hz, 1H), 3.94 (s, 3H).

$^{13}\text{C NMR}$ (126 MHz, CDCl_3): δ 165.4, 161.3 (d, $^1J = 249.3$ Hz), 153.3 (d, $^2J = 36.8$ Hz), 150.8 (d, $^4J = 4.5$ Hz), 138.2 (d, $^4J = 1.8$ Hz), 125.8, 118.2 (d, $^3J = 4.5$ Hz), 95.2 (d, $^2J = 17.6$ Hz), 52.6.

$^{19}\text{F NMR}$ (376 MHz, CDCl_3): δ -115.94.

HRMS (ESI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_9\text{H}_9\text{FNO}_2$) requires m/z 182.0612, found m/z 182.0610.

1-(1-Fluorovinyl)isoquinoline **5i**



Prepared according to General Procedure C using 1-bromoisoquinoline (207 mg, 1.00 mmol, 1.0 equiv), (1-fluorovinyl)(methyl)diphenylsilane (356 mg, 1.50 mmol, 1.5 equiv), $\text{Pd}(\text{PPh}_3)_4$ (59.0 mg, 0.05 mmol, 5 mol%), CuI (9.7 mg, 0.05 mmol, 5 mol%), CsF (364 mg, 2.40 mmol, 2.4 equiv), and DMF (10 mL, 0.1 M). The reaction mixture was subjected to the purification outlined in General Procedure C (silica gel, Et_2O :petroleum ether, 1:99) to afford the desired product as a pale yellow oil (171 mg, 99%).

ν_{\max} (film): 1657, 1622, 1584, 1557, 1393, 1298, 1250, 1215, 1175, 1153, 1140, 926, 858, 827, 748, 685, 579 cm^{-1} .

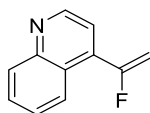
$^1\text{H NMR}$ (400 MHz, CDCl_3): δ 8.53 (d, $J = 5.6$ Hz, 1H), 8.37 (d, $J = 8.6$ Hz, 1H), 7.83 (d, $J = 8.2$ Hz, 1H), 7.76 – 7.65 (m, 2H), 7.61 (ddd, $J = 8.2, 6.9, 1.2$ Hz, 1H), 5.50 – 5.19 (m, 2H).

$^{13}\text{C NMR}$ (126 MHz, CDCl_3): δ 163.0 (d, $^1J = 256.4$ Hz), 151.0 (d, $^2J = 31.1$ Hz), 141.9, 136.9, 130.5, 128.0 (d, $^4J = 1.9$ Hz), 127.2, 126.3 (d, $^3J = 9.7$ Hz), 126.0 (d, $^4J = 2.3$ Hz), 122.2, 97.3 (d, $^2J = 19.2$ Hz).

$^{19}\text{F NMR}$ (376 MHz, CDCl_3): δ -97.19.

HRMS (ESI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{11}\text{H}_9\text{FN}$) requires m/z 174.0714, found m/z 174.0710.

4-(1-Fluorovinyl)quinoline **5j**



Prepared according to General Procedure C using 4-bromoquinoline (208 mg, 1.00 mmol, 1.0 equiv), (1-fluorovinyl)(methyl)diphenylsilane (356 mg, 1.50 mmol, 1.5 equiv), $\text{Pd}(\text{PPh}_3)_4$ (57.8 mg, 0.05 mmol, 5 mol%), CuI (9.7 mg, 0.05 mmol, 5 mol%), CsF (364 mg, 2.40 mmol, 2.4 equiv), and DMF (10 mL, 0.1 M). The reaction mixture was subjected to the purification outlined in General Procedure C (silica gel, Et_2O :petroleum ether, 1:99) to afford the desired product as a pale yellow oil (82.1 mg, 47%).

NB. Due to the reactive nature of this compound, an analytically pure sample was not obtained, data reported as observed. The compound was used directly in the next step.

ν_{\max} (film): 2959, 2928, 2870, 1659, 1578, 1558, 1508, 1493, 1462, 1362, 1315, 1282, 1258, 1180, 1153, 926, 849, 764 cm^{-1} .

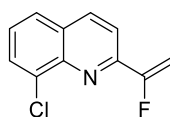
$^1\text{H NMR}$ (400 MHz, CDCl_3): δ 8.93 (d, $J = 4.5$ Hz, 1H), 8.23 – 8.15 (m, 2H), 7.79 – 7.73 (m, 1H), 7.67 – 7.58 (m, 1H), 7.49 (dd, $J = 4.4, 1.1$ Hz, 1H), 5.30 (dd, $^3J_{\text{HF}} = 16.7$ Hz, $^2J_{\text{HH}} = 3.4$ Hz, 1H), 5.09 (dd, $^3J_{\text{HF}} = 47.9$ Hz, $^2J_{\text{HH}} = 3.4$ Hz, 1H).

$^{13}\text{C NMR}$ (126 MHz, CDCl_3): δ 162.5, 160.5, 149.9, 149.9, 148.9, 148.7, 138.5, 138.3, 134.5, 134.1, 130.6, 130.1, 129.9, 128.1, 127.6, 127.0, 125.4, 125.4, 125.3, 125.1, 120.4, 97.7, 97.5, 85.5, 85.1, 82.9. *NB.* J_{CF} not indicated due to overlap of impurities.

$^{19}\text{F NMR}$ (376 MHz, CDCl_3): δ -92.69.

HRMS (ESI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{11}\text{H}_9\text{FN}$) requires m/z 174.0714, found m/z 174.0709.

8-Chloro-2-(1-fluorovinyl)quinoline **5k**



Prepared according to General Procedure C using 2-bromo-8-chloroquinoline (121 mg, 0.50 mmol, 1.0 equiv), (1-fluorovinyl)(methyl)diphenylsilane (182 mg, 0.75 mmol, 1.5 equiv), Pd(PPh₃)₄ (57.8 mg, 0.50 mmol, 1.0 equiv), CuI (6.6 mg, 0.50 mmol, 1.0 equiv), CsF (224 mg, 1.18 mmol, 2.35 equiv), and DMF (10 mL, 0.1 M). The reaction mixture was subjected to the purification outlined in General Procedure C (silica gel, Et₂O:petroleum ether, 1:99) to afford the desired product as a white solid (42.0 mg, 40%).

ν_{\max} (film): 1655, 1595, 1499, 1425, 1312, 1287, 1206, 1123, 989, 841, 758, 669 cm⁻¹.

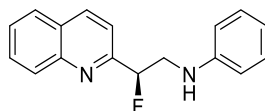
¹H NMR (400 MHz, CDCl₃): δ 8.20 (d, J = 8.6 Hz, 1H), 7.82 (dd, J = 7.5, 1.3 Hz, 1H), 7.72 (ddd, J = 8.1, 6.8, 1.4 Hz, 2H), 7.43 (dd, J = 8.2, 7.4 Hz, 1H), 6.06 (dd, $^3J_{\text{HF}}$ = 48.5 Hz, $^2J_{\text{HH}}$ = 2.9 Hz, 1H), 5.17 (dd, $^3J_{\text{HF}}$ = 16.0 Hz, $^2J_{\text{HH}}$ = 2.9 Hz, 1H).

¹³C NMR (126 MHz, CDCl₃): δ 161.9 (d, 1J = 250.7 Hz), 150.60 (d, 2J = 37.5 Hz), 144.1 (d, 4J = 4.5 Hz), 137.6, 133.9, 130.2, 129.2, 126.9, 126.7, 117.5 (d, 3J = 4.4 Hz), 94.7 (d, 2J = 17.6 Hz).

¹⁹F NMR (376 MHz, CDCl₃): δ -115.61.

HRMS (ESI): exact mass calculated for [M+H]⁺ (C₁₁H₈ClFN) requires m/z 208.0324, found m/z 208.0320.

(*R*)-*N*-(2-Fluoro-2-(quinolin-2-yl)ethyl)aniline **7a**



Prepared according to General Procedure A using 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), aniline (27 μ L, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and THF (200 μ L, 0.5 M). The reaction mixture was subjected to the purification outlined in General Procedure A (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale yellow solid (21.8 mg, 82%).

ν_{\max} (film): 3406 (br), 3051, 3019, 2926, 2846, 1600, 1504, 1061, 751 cm⁻¹.

¹H NMR (500 MHz, CDCl₃): δ 8.24 (d, J = 8.5 Hz, 1H), 8.11 (d, J = 8.5 Hz, 1H), 7.85 (d, J = 8.2 Hz, 1H), 7.79 – 7.71 (m, 1H), 7.64 (dd, J = 8.5, 1.2 Hz, 1H), 7.60 – 7.54 (m, 1H), 7.23 – 7.15 (m, 2H), 6.79 – 6.70 (m, 3H), 5.91 (ddd, $^2J_{\text{HF}}$ = 48.7 Hz, J = 7.3, 3.7 Hz, 1H), 3.95 (ddd, $^3J_{\text{HF}}$ = 26.3 Hz, $^2J_{\text{HH}}$ = 14.0 Hz, J = 3.8 Hz, 1H), 3.73 (ddd, $^3J_{\text{HF}}$ = 19.7 Hz, $^2J_{\text{HH}}$ = 14.0 Hz, J = 7.3 Hz, 1H).

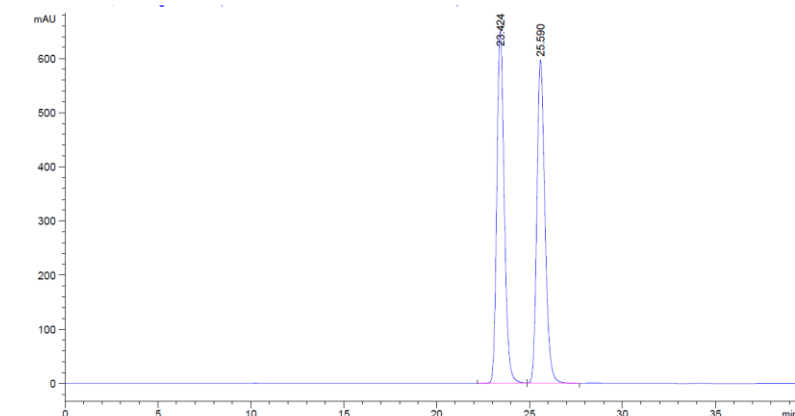
¹³C NMR (126 MHz, CDCl₃): δ 158.1 (d, 2J = 24.5 Hz), 147.6, 147.5, 137.3, 130.1, 129.5, 129.3, 127.9, 127.9, 127.0, 118.2, 117.9 (d, 3J = 6.7 Hz), 113.5, 93.4 (d, 1J = 174.8 Hz), 48.6 (d, 2J = 22.5 Hz).

¹⁹F NMR (376 MHz, CDCl₃): δ -191.33.

HRMS (NSI): exact mass calculated for [M+H]⁺ (C₁₇H₁₆N₂F) requires m/z 267.1292, found m/z 267.1293.

The enantiomeric purity of the product was determined by HPLC analysis: 96:4 e.r. (Chiralpak IA, hexane/*i*-PrOH = 90/10, flow rate 0.5 mL/min, λ = 250 nm), t_r (major) = 23.1 min, t_r (minor) = 25.2 min;

$[\alpha]_D^{23} = +21.7$ (c 1.01, CHCl₃).

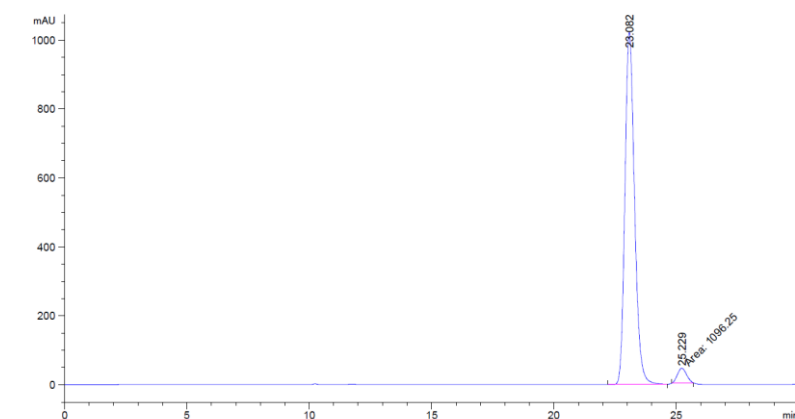


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Area Percent Report
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Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	23.424	BB	0.4287	1.80602e4	650.49353	50.0453
2	25.590	BB	0.4661	1.80275e4	596.60730	49.9547



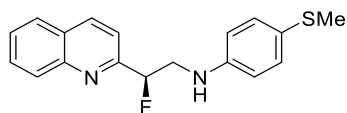
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Area Percent Report
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Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	23.082	BV	0.4204	2.79043e4	1022.16467	96.2199
2	25.229	MM	0.4293	1096.24890	42.55674	3.7801

(R)-N-(2-Fluoro-2-(quinolin-2-yl)ethyl)-4-(methylthio)aniline 7b



Prepared according to General Procedure B using 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), 4-(methylthio)aniline (41.7 mg, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and CPME (200 μ L, 0.5 M). The reaction mixture was subjected to the purification outlined in General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale brown oil (29.0 mg, 93%).

ν_{\max} (film): 3318 (br), 2970, 2916, 2839, 1597, 1504, 1319, 1034, 826 cm^{-1} .

$^1\text{H NMR}$ (500 MHz, CDCl_3): δ 8.22 (d, $J = 8.5$ Hz, 1H), 8.10 (d, $J = 8.5$ Hz, 1H), 7.84 (d, $J = 8.1$ Hz, 1H), 7.79 – 7.72 (m, 1H), 7.63 (dd, $J = 8.5, 1.2$ Hz, 1H), 7.57 (t, $J = 7.5$ Hz, 1H), 7.21 (d, $J = 8.6$ Hz, 2H), 6.66 (d, $J = 8.6$ Hz, 2H), 5.89 (ddd, $^2J_{\text{HF}} = 48.6$ Hz, $J = 7.3, 3.7$ Hz, 1H), 4.29 (s, 1H), 3.92 (ddd, $^3J_{\text{HF}} = 26.1$ Hz, $^2J_{\text{HH}} = 14.1$ Hz, $J = 3.5$ Hz, 1H), 3.71 (ddd, $^3J_{\text{HF}} = 21.0$ Hz, $^2J_{\text{HH}} = 14.1$ Hz, $J = 7.3$ Hz, 1H), 2.40 (s, 3H).

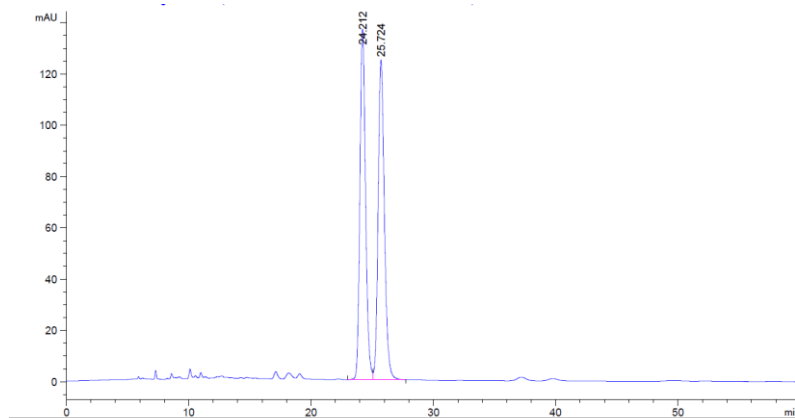
$^{13}\text{C NMR}$ (126 MHz, CDCl_3): δ 158.1, 157.9, 146.5, 137.3, 131.4, 130.1, 129.2, 127.9, 127.0, 125.8, 125.2, 117.9 (d, $^3J = 6.4$ Hz), 114.1, 93.3 (d, $^1J = 174.9$ Hz), 48.6 (d, $^2J = 22.4$ Hz), 19.1.

$^{19}\text{F NMR}$ (470 MHz, CDCl_3): δ -191.43.

HRMS (NSI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{18}\text{H}_{18}\text{N}_2\text{FS}$) requires m/z 313.1169, found m/z 313.1165.

The enantiomeric purity of the product was determined by HPLC analysis: 97:3 e.r. (Chiralpak IA, hexane/*i*-PrOH = 90/10, flow rate 0.5 mL/min, $\lambda = 250$ nm), t_r (major) = 25.2 min, t_r (minor) = 27.1 min;

$[\alpha]_{\text{D}}^{20} = +4.4$ (c 1.45, CHCl_3).

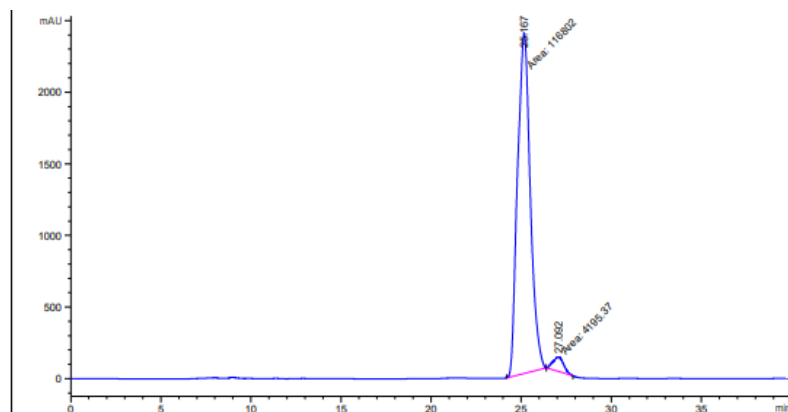


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 Area Percent Report
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Sorted By : Signal
 Multiplier : 1.0000
 Dilution : 1.0000
 Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	24.212	BV	0.4947	4403.19873	136.88387	49.7183
2	25.724	VB	0.5457	4453.10205	124.86521	50.2817



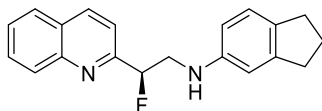
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 Area Percent Report
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Sorted By : Signal
 Multiplier : 1.0000
 Dilution : 1.0000
 Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	25.167	MM	0.8190	1.16802e5	2376.94727	96.5327
2	27.092	MM	0.6820	4195.36768	102.52654	3.4673

(*R*)-*N*-(2-Fluoro-2-(quinolin-2-yl)ethyl)-2,3-dihydro-1*H*-inden-5-amine **7c**



Prepared according to General Procedure B using 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), 2,3-dihydro-1*H*-inden-5-amine (39.9 mg, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg,

0.02 mmol, 20 mol%), and CPME (200 μ L, 0.5 M). The reaction mixture was subjected to the purification outlined in General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale yellow oil (27.2 mg, 89%).

ν_{\max} (film): 3395 (br), 3041, 3006, 2930, 2839, 1599, 1502, 1058, 833 cm^{-1} .

$^1\text{H NMR}$ (500 MHz, CDCl_3): δ 8.22 (d, $J = 8.5$ Hz, 1H), 8.10 (d, $J = 8.7$ Hz, 1H), 7.84 (d, $J = 8.1$ Hz, 1H), 7.75 (ddd, $J = 8.4, 6.9, 1.5$ Hz, 1H), 7.63 (dd, $J = 8.5, 1.5$ Hz, 1H), 7.57 (ddd, $J = 8.1, 6.9, 1.2$ Hz, 1H), 7.04 (d, $J = 8.0$ Hz, 1H), 6.67 – 6.61 (m, 1H), 6.54 (dd, $J = 8.0, 2.3$ Hz, 1H), 5.90 (ddd, $^2J_{\text{HF}} = 48.8$ Hz, $J = 7.4, 3.7$ Hz, 1H), 4.13 (s, 1H), 3.92 (ddd, $^3J_{\text{HF}} = 26.4$ Hz, $^2J_{\text{HH}} = 14.0$ Hz, $J = 3.7$ Hz, 1H), 3.70 (ddd, $^3J_{\text{HF}} = 19.6$ Hz, $^2J_{\text{HH}} = 14.0$ Hz, $J = 7.4$ Hz, 1H), 2.86 – 2.75 (m, 4H), 2.12 – 1.90 (m, 2H).

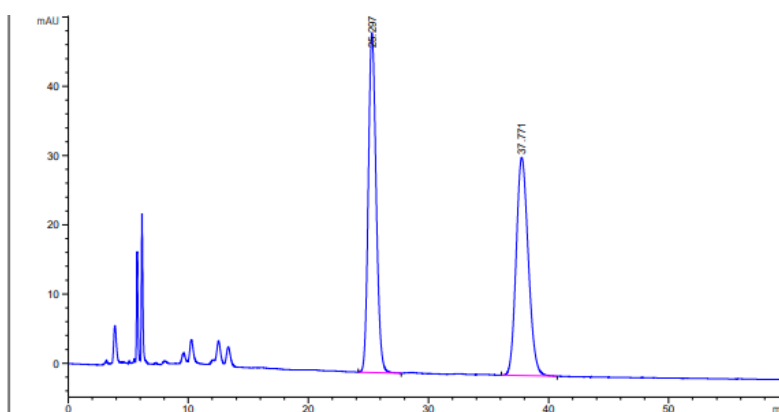
$^{13}\text{C NMR}$ (126 MHz, CDCl_3): δ 158.3 (d, $^2J = 24.5$ Hz), 147.5, 146.4, 145.6, 137.2, 134.0, 130.0, 129.4, 127.8, 126.9, 124.9, 118.0, 117.9, 112.0, 109.8, 93.4 (d, $^1J = 174.4$ Hz), 49.2 (d, $^2J = 22.4$ Hz), 33.2, 32.1, 25.8.

$^{19}\text{F NMR}$ (376 MHz, CDCl_3): δ -191.30.

HRMS (NSI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{20}\text{H}_{20}\text{N}_2\text{F}$) requires m/z 307.1605, found m/z 307.1607.

The enantiomeric purity of the product was determined by HPLC analysis: 91:9 e.r. (Chiralcel OJ-H, hexane/*i*-PrOH = 50/50, flow rate 1.0 mL/min, $\lambda = 210$ nm), t_r (major) = 25.7 min, t_r (minor) = 38.8 min;

$[\alpha]_{\text{D}}^{20} = +7.0$ (c 1.36, CHCl_3).

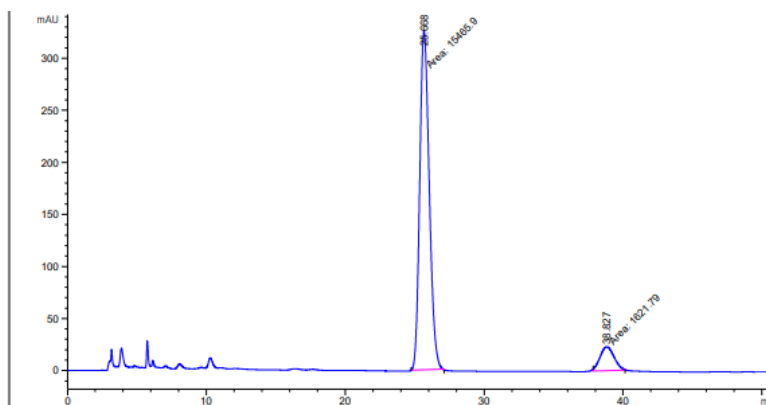


Area Percent Report

Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	25.297	BB	0.7026	2218.92896	49.04015	50.0879
2	37.771	BB	1.0750	2211.13745	31.50676	49.9121



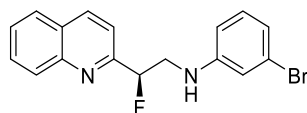
Area Percent Report

Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=210 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	25.668	MM	0.7928	1.54659e4	325.14124	90.5090
2	38.827	MM	1.1705	1621.78674	23.09332	9.4910

(*R*)-3-Bromo-*N*-(2-fluoro-2-(quinolin-2-yl)ethyl)aniline **7d**



Prepared according to General Procedure B using 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), 3-bromoaniline (53.1 mg, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and CPME (200 μ L, 0.5 M). The reaction mixture was subjected to the purification outlined in General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale yellow oil (30.3 mg, 88%).

ν_{\max} (film): 3418 (br), 3063, 2924, 2847, 1589, 1504, 1065, 833, 764 cm^{-1} .

$^1\text{H NMR}$ (400 MHz, CDCl_3): δ 8.23 (d, $J = 8.5$ Hz, 1H), 8.11 (d, $J = 8.5$ Hz, 1H), 7.85 (d, $J = 8.2$ Hz, 1H), 7.76 (ddd, $J = 8.4, 6.9, 1.4$ Hz, 1H), 7.63 (dd, $J = 8.5, 1.3$ Hz, 1H), 7.58 (ddd, $J = 8.0, 7.0, 1.0$ Hz, 1H), 7.01 (t, $J = 8.0$ Hz, 1H), 6.87 – 6.79 (m, 2H), 6.65 – 6.57 (m, 1H), 5.89 (ddd, $^2J_{\text{HF}} = 48.5$ Hz, $J = 7.0, 3.8$ Hz, 1H), 4.33 (s, 1H), 3.91 (ddd, $^3J_{\text{HF}} = 25.0$ Hz, $^2J_{\text{HH}} = 14.1$ Hz, $J = 3.8$ Hz, 1H), 3.73 (ddd, $^3J_{\text{HF}} = 20.8$ Hz, $^2J_{\text{HH}} = 14.1$ Hz, $J = 7.0$ Hz, 1H).

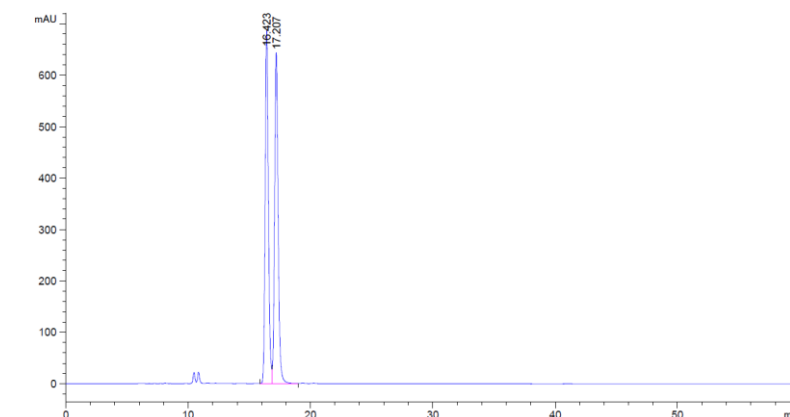
$^{13}\text{C NMR}$ (126 MHz, CDCl_3): δ 157.7 (d, $^2J = 24.6$ Hz), 149.0, 147.4 (d, $^4J = 2.3$ Hz), 137.4, 130.6, 130.2, 129.2, 127.9, 127.8, 127.0, 123.4, 120.8, 117.8 (d, $^3J = 6.5$ Hz), 115.9, 112.0, 93.2 (d, $^1J = 175.2$ Hz), 48.2 (d, $^2J = 22.4$ Hz).

$^{19}\text{F NMR}$ (377 MHz, CDCl_3): δ -191.34.

HRMS (ESI): exact mass calculated for $[M+H]^+$ ($C_{17}H_{15}N_2BrF$) requires m/z 345.0397, found m/z 345.0395.

The enantiomeric purity of the product was determined by HPLC analysis: 94:6 e.r. (Chiralpak IA, hexane/*i*-PrOH = 90/10, flow rate 0.5 mL/min, $\lambda = 250$ nm), t_r (major) = 16.7 min, t_r (minor) = 17.5 min;

$[\alpha]_D^{20} = +31.4$ (c 2.12, $CHCl_3$).

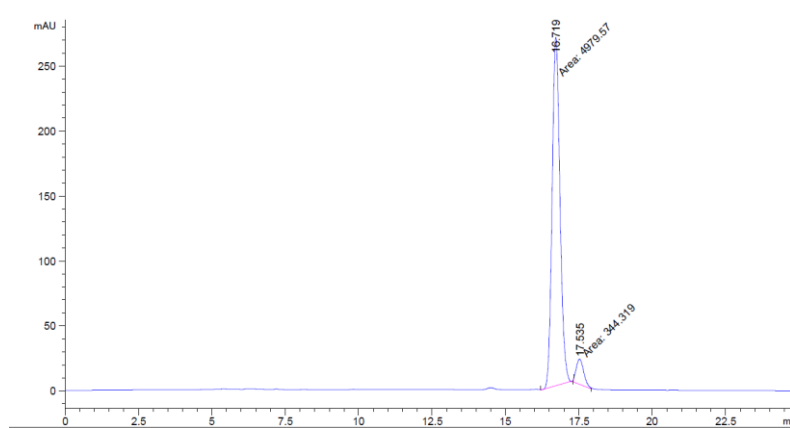


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Area Percent Report
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Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWDL A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	16.423	BV	0.2796	1.25106e4	686.43469	49.3029
2	17.207	VB	0.3046	1.28644e4	643.45532	50.6971



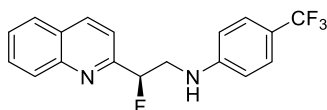
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Area Percent Report
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Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWDL A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	16.719	MM	0.3096	4979.56592	268.05878	93.5326
2	17.535	MM	0.2954	344.31888	19.42668	6.4674

(*R*)-*N*-(2-Fluoro-2-(quinolin-2-yl)ethyl)-4-(trifluoromethyl)aniline **7e**



Prepared according to General Procedure B using 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), 4-(trifluoromethyl)aniline (48.3 mg, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and CPME (200 μ L, 0.5 M). The reaction mixture was subjected to the purification outlined in General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale yellow oil (31.2 mg, 93%).

ν_{\max} (film): 3426 (br), 3055, 2963, 2839, 1612, 1535, 1319, 1103, 1065, 826 cm^{-1} .

$^1\text{H NMR}$ (500 MHz, CDCl_3): δ 8.24 (d, $J = 8.5$ Hz, 1H), 8.11 (d, $J = 8.5$ Hz, 1H), 7.85 (dd, $J = 8.1$, 1.4 Hz, 1H), 7.77 (ddd, $J = 8.4$, 6.8, 1.4 Hz, 1H), 7.64 (dd, $J = 8.5$, 1.6 Hz, 1H), 7.59 (ddd, $J = 8.2$, 6.9, 1.2 Hz, 1H), 7.40 (d, $J = 8.5$ Hz, 2H), 6.71 (d, $J = 8.5$ Hz, 2H), 5.90 (ddd, $^2J_{\text{HF}} = 48.5$ Hz, $J = 7.0$, 3.8 Hz, 1H), 4.61 (s, 1H), 3.99 (ddd, $^3J_{\text{HF}} = 25.1$ Hz, $^2J_{\text{HH}} = 14.2$ Hz, $J = 3.8$ Hz, 1H), 3.80 (ddd, $^3J_{\text{HF}} = 20.9$ Hz, $^2J_{\text{HH}} = 14.2$ Hz, $J = 7.0$ Hz, 1H).

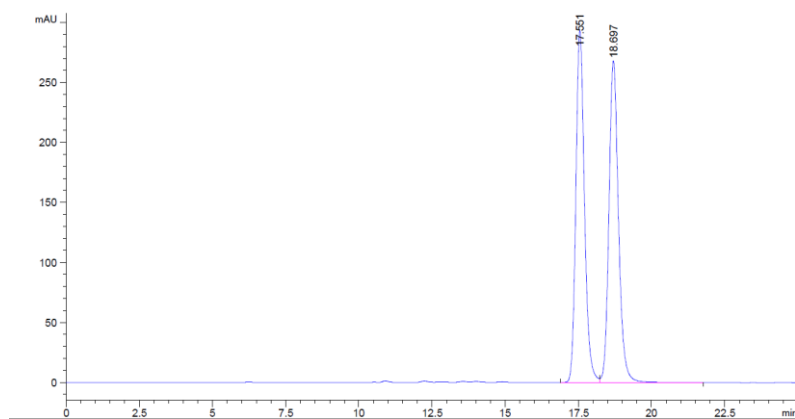
$^{13}\text{C NMR}$ (126 MHz, CDCl_3): δ 157.6 (d, $^2J = 24.8$ Hz), 150.2, 147.4, 137.5, 130.3, 129.2, 127.9, 127.9, 127.1, 126.8 (q, $^3J = 3.8$ Hz), 125.0 (q, $^1J = 270.4$ Hz), 119.6 (q, $^2J = 32.6$ Hz), 117.8 (d, $^3J = 6.7$ Hz), 112.5, 93.1 (d, $^1J = 175.6$ Hz), 47.9 (d, $^2J = 22.4$ Hz).

$^{19}\text{F NMR}$ (470 MHz, CDCl_3): δ -61.07, -191.49.

HRMS (NSI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{18}\text{H}_{15}\text{N}_2\text{F}_4$) requires m/z 335.1166, found m/z 336.1158.

The enantiomeric purity of the product was determined by HPLC analysis: 97:3 e.r. (Chiralpak IA, hexane/*i*-PrOH = 90/10, flow rate 0.5 mL/min, $\lambda = 250$ nm), t_r (major) = 17.2 min, t_r (minor) = 18.2 min;

$[\alpha]_{\text{D}}^{20} = +4.7$ (c 1.07, CHCl_3).

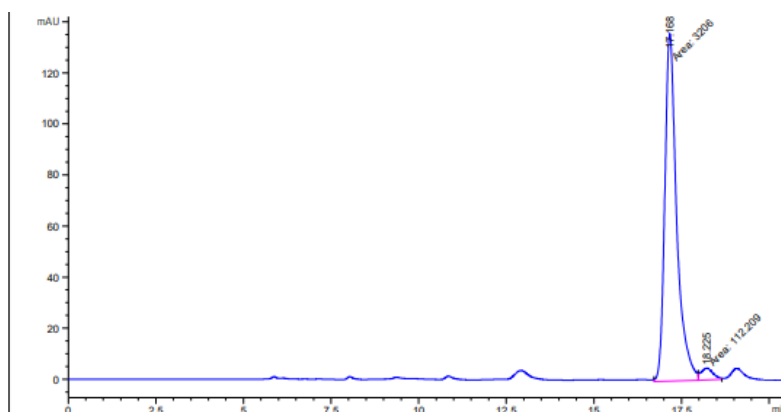


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 Area Percent Report
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Sorted By : Signal
 Multiplier : 1.0000
 Dilution : 1.0000
 Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	17.551	BV	0.3124	5980.82959	292.94598	49.6500
2	18.697	VB	0.3454	6065.15576	268.12650	50.3500



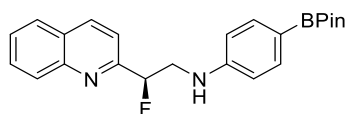
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 Area Percent Report
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Sorted By : Signal
 Multiplier : 1.0000
 Dilution : 1.0000
 Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	17.168	MF	0.3925	3205.99536	136.12822	96.6184
2	18.225	FM	0.4026	112.20947	4.64504	3.3816

(R)-*N*-(2-Fluoro-2-(quinolin-2-yl)ethyl)-4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)aniline **7f**



Prepared according to General Procedure A using 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), 4-(4,4,5,5-tetramethyl-1,3,2-dioxaborolan-2-yl)aniline (65.7 mg, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and THF (200 μ L, 0.5 M) at -20 $^{\circ}$ C for 5 d. The reaction mixture was subjected to the purification outlined in General Procedure A (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale yellow solid (31.0 mg, 79%).

ν_{\max} (film): 3401 (br), 3041, 2973, 2928, 1606, 1359, 1145 cm^{-1} .

^1H NMR (400 MHz, CD_2Cl_2): δ 8.27 (d, $J = 8.5$ Hz, 1H), 8.09 (d, $J = 8.5$ Hz, 1H), 7.88 (d, $J = 8.2$ Hz, 1H), 7.76 (ddd, $J = 8.4, 6.9, 1.4$ Hz, 1H), 7.63 (dd, $J = 8.5, 1.4$ Hz, 1H), 7.61 – 7.50 (m, 3H), 6.75 – 6.63 (m, 2H), 5.87 (ddd, $^2J_{\text{HF}} = 48.6$ Hz, $J = 7.1, 3.8$ Hz, 1H), 4.52 (s, 1H), 3.97 (m, 1H), 3.79 (ddd, $^3J_{\text{HF}} = 20.1$ Hz, $^2J_{\text{HH}} = 13.4$ Hz, $J = 6.3$ Hz, 1H), 1.30 (s, 12H).

^{13}C NMR (101 MHz, CD_2Cl_2): δ 158.2 (d, $^2J = 24.4$ Hz), 150.6, 137.5, 136.6, 136.6, 130.3, 129.5, 128.2, 128.1, 128.1, 127.2, 118.2 (d, $^3J = 6.1$ Hz), 112.5, 93.7 (d, $^1J = 174.8$ Hz), 83.6, 48.0 (d, $^2J = 22.4$ Hz), 25.1.

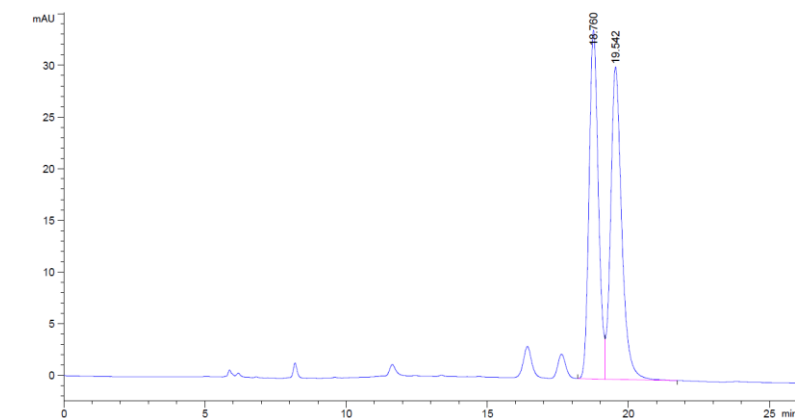
^{19}F NMR (376 MHz, CD_2Cl_2): δ -190.85 .

^{11}B NMR (128 MHz, CD_2Cl_2): δ 30.69.

HRMS (NSI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{23}\text{H}_{27}\text{N}_2\text{O}_2\text{FB}$) requires m/z 392.2154, found m/z 393.2202.

The enantiomeric purity of the product was determined by HPLC analysis: 93:7 e.r. (Chiralpak IA, hexane/*i*-PrOH = 90/10, flow rate 0.5 mL/min, $\lambda = 250$ nm), t_r (major) = 17.5 min, t_r (minor) = 18.9 min;

$[\alpha]_D^{20} = +43.4$ (c 0.15, CHCl_3).



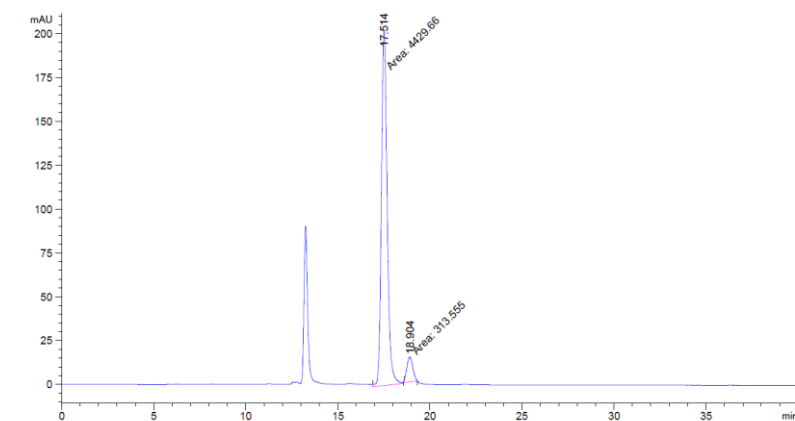
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                          Area Percent Report
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Sorted By      :      Signal
Multiplier     :      1.0000
Dilution       :      1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWd1 A, Wavelength=250 nm

Peak RetTime Type Width Area Height Area
# [min] [min] [min] [mAU*s] [mAU] %
-----|-----|-----|-----|-----|-----
1 18.760 BV 0.3529 772.33557 33.74578 48.4127
2 19.542 VB 0.4107 822.97955 30.23723 51.5873

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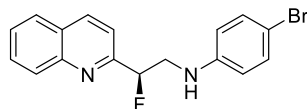
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 Area Percent Report
 =====

Sorted By : Signal
 Multiplier : 1.0000
 Dilution : 1.0000
 Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	17.514	MM	0.3651	4429.66260	202.23083	93.3894
2	18.904	MM	0.3693	313.55548	14.15067	6.6106

(R)-4-Bromo-N-(2-fluoro-2-(quinolin-2-yl)ethyl)aniline 7g



Prepared according to General Procedure A using 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), 4-bromoaniline (51.6 mg, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and THF (200 μ L, 0.5 M) at -20 $^{\circ}$ C for 5 d. The reaction mixture was subjected to the purification outlined in General Procedure A (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale yellow oil (24.4 mg, 71%).

ν_{max} (film): 3403 (br), 3026, 2928, 2851, 1599, 813 cm^{-1} .

^1H NMR (400 MHz, CDCl_3): δ 8.23 (d, $J = 8.5$ Hz, 1H), 8.09 (d, $J = 8.4$ Hz, 1H), 7.85 (d, $J = 8.1$ Hz, 1H), 7.76 (ddd, $J = 8.4, 6.9, 1.4$ Hz, 1H), 7.63 (dd, $J = 8.5, 1.5$ Hz, 1H), 7.58 (ddd, $J = 8.1, 6.9, 1.1$ Hz, 1H), 7.26 – 7.22 (m, 1H), 6.62 – 6.56 (m, 2H), 5.87 (ddd, $^2J_{\text{HF}} = 48.7$ Hz, $J = 7.1, 3.8$ Hz, 2H), 4.28 (t, $J = 5.8$ Hz, 1H), 3.91 (dddd, $^3J_{\text{HF}} = 25.4$ Hz, $^2J_{\text{HH}} = 14.0$ Hz, $J = 7.3, 3.8$ Hz, 1H), 3.78 – 3.65 (m, 1H).

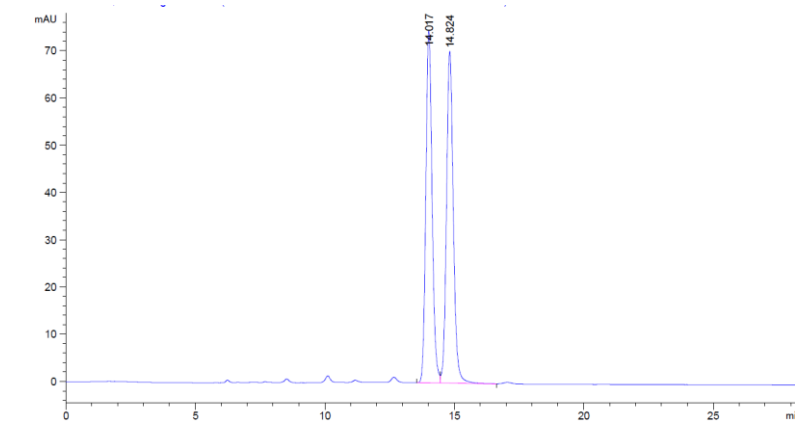
^{13}C NMR (101 MHz, CDCl_3): δ 157.9 (d, $^2J = 24.5$ Hz), 146.7, 137.3, 132.2, 132.1, 130.2, 129.4, 127.9, 127.0, 117.9 (d, $^3J = 6.9$ Hz), 115.0, 109.8, 94.1, 93.3 (d, $^1J = 175.0$ Hz), 48.5 (d, $^2J = 22.4$ Hz).

^{19}F NMR (376 MHz, CDCl_3): δ -191.44 .

HRMS (NSI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{17}\text{H}_{15}\text{N}_2\text{FBr}$) requires m/z 345.0403, found m/z 345.0400.

The enantiomeric purity of the product was determined by HPLC analysis: 90:10 e.r. (Chiralpak IA, hexane/*i*-PrOH = 50/50, flow rate 0.3 mL/min, $\lambda = 250$ nm), t_r (major) = 14.1 min, t_r (minor) = 15.0 min;

$[\alpha]_D^{20} = +58.6$ (c 1.22, CHCl_3).



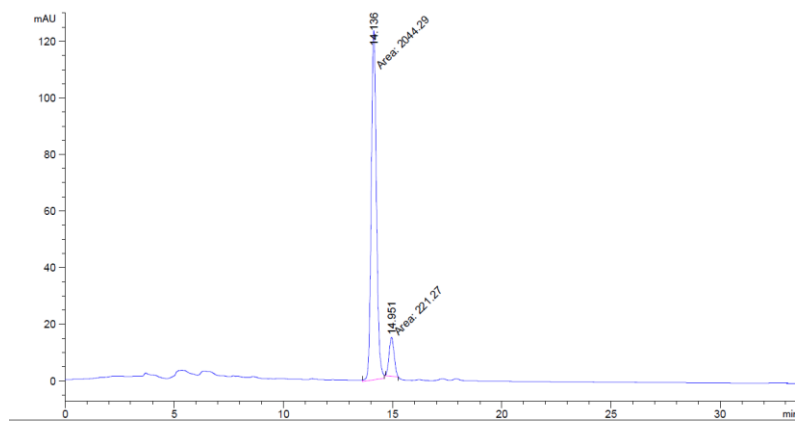
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Sorted By      :      Signal
Multiplier    :      1.0000
Dilution      :      1.0000
Use Multiplier & Dilution Factor with ISTDs
  
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Signal 1: VWd1 A, Wavelength=250 nm
  
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Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	14.017	VB	0.2586	1251.85400	74.50861	49.3576
2	14.824	VB	0.2801	1284.44275	70.30343	50.6424



```

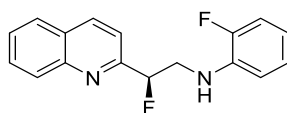
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                          Area Percent Report
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Sorted By      :      Signal
Multiplier    :      1.0000
Dilution      :      1.0000
Use Multiplier & Dilution Factor with ISTDs
  
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Signal 1: VWd1 A, Wavelength=250 nm
  
```

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	14.136	MM	0.2758	2044.29187	123.52139	90.2333
2	14.951	MM	0.2681	221.27008	13.75688	9.7667

(*R*)-2-Fluoro-*N*-(2-fluoro-2-(quinolin-2-yl)ethyl)aniline **7h**



Prepared according to General Procedure B using 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), 2-fluoroaniline (33.3 mg, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and CPME (200 μ L, 0.5 M). The reaction mixture was subjected to the purification outlined in General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale yellow oil (13.1 mg, 46%).

ν_{max} (film): 3422 (br), 3065, 2928, 2857, 1620, 1516, 1506, 1190, 829, 743 cm^{-1} .

$^1\text{H NMR}$ (400 MHz, CDCl_3): δ 8.24 (d, $J = 8.5$ Hz, 1H), 8.10 (d, $J = 8.8$ Hz, 1H), 7.85 (d, $J = 7.2$ Hz, 1H), 7.76 (ddd, $J = 8.4, 6.9, 1.4$ Hz, 1H), 7.65 (dd, $J = 8.5, 1.6$ Hz, 1H), 7.57 (ddd, $J = 8.1, 6.9, 1.1$ Hz, 1H), 7.04 – 6.91 (m, 2H), 6.84 (td, $J = 8.5, 1.4$ Hz, 1H), 6.69 – 6.55 (m, 1H), 5.91 (ddd, $^2J_{\text{HF}} = 48.6$ Hz, $J = 7.1, 3.8$ Hz, 1H), 4.56 (s, 1H), 3.97 (dddd, $^3J_{\text{HF}} = 25.3$ Hz, $^2J_{\text{HH}} = 14.1$ Hz, $J = 7.1, 3.7$ Hz, 1H), 3.84 – 3.71 (m, 1H).

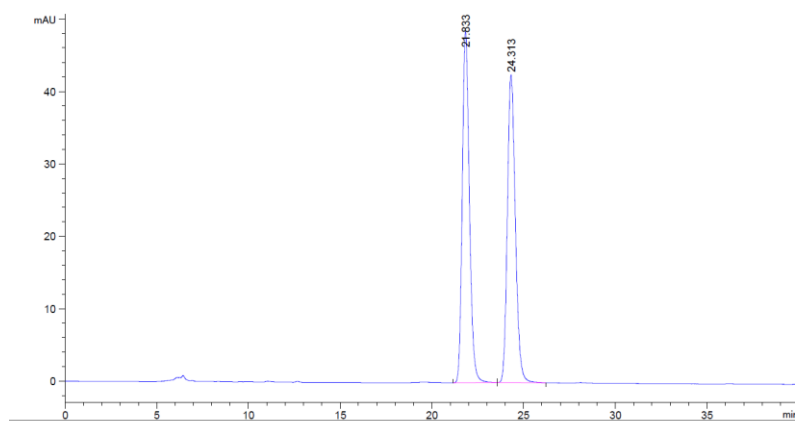
$^{13}\text{C NMR}$ (101 MHz, CDCl_3): δ 158.0 (d, $^2J = 24.6$ Hz), 151.9 (d, $^1J = 238.8$ Hz), 147.6 (d, $^4J = 1.8$ Hz), 137.3, 136.2 (d, $^3J = 11.0$ Hz), 130.1, 129.4, 127.9, 127.0, 124.7, 124.7, 117.9 (d, $^3J = 6.9$ Hz), 117.4 (d, $^3J = 7.1$ Hz), 114.7 (d, $^2J = 18.7$ Hz), 112.7 (d, $^4J = 2.7$ Hz), 93.3 (d, $^1J = 175.4$ Hz), 48.2 (d, $^2J = 22.4$ Hz).

$^{19}\text{F NMR}$ (376 MHz, CDCl_3): δ -136.12, -191.53.

HRMS (NSI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{17}\text{H}_{15}\text{N}_2\text{F}_2$) requires m/z 285.1203, found m/z 285.1202.

The enantiomeric purity of the product was determined by HPLC analysis: 89:11 e.r. (Chiralcel OJ-H, hexane/*i*-PrOH = 50/50, flow rate 0.5 mL/min, $\lambda = 250$ nm), t_r (major) = 19.7 min, t_r (minor) = 22.0 min;

$[\alpha]_{\text{D}}^{20} = +3.4$ (c 0.65, CHCl_3).

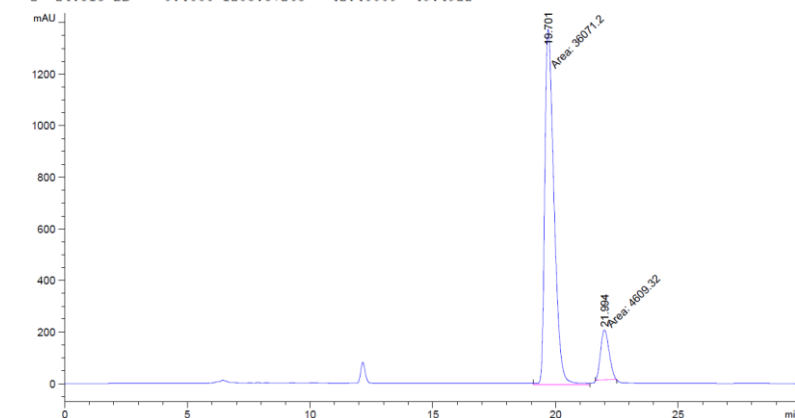


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 Area Percent Report
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Sorted By : Signal
 Multiplier : 1.0000
 Dilution : 1.0000
 Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	21.833	BB	0.4178	1311.95435	48.45583	50.5078
2	24.313	BB	0.4666	1285.57263	42.48009	49.4922



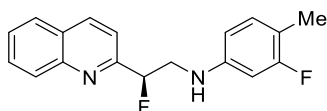
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 Area Percent Report
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Sorted By : Signal
 Multiplier : 1.0000
 Dilution : 1.0000
 Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	19.701	MM	0.4365	3.60712e4	1377.26685	88.6695
2	21.994	MM	0.3972	4609.32080	193.43065	11.3305

(R)-3-Fluoro-N-(2-fluoro-2-(quinolin-2-yl)ethyl)-4-methylaniline 7i



Prepared according to General Procedure B using 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), 3-fluoro-4-methylaniline (37.5 mg, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02

mmol, 20 mol%), and CPME (200 μ L, 0.5 M). The reaction mixture was subjected to the purification outlined in General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale yellow oil (29.2 mg, 98%).

ν_{max} (film): 3433 (br), 3048, 2924, 2862, 1636, 1597, 1520, 833 cm^{-1} .

$^1\text{H NMR}$ (500 MHz, CDCl_3): δ 8.23 (d, $J = 8.5$ Hz, 1H), 8.09 (d, $J = 8.5$ Hz, 1H), 7.84 (d, $J = 8.1$ Hz, 1H), 7.78 – 7.71 (m, 1H), 7.66 – 7.60 (m, 1H), 7.57 (t, $J = 7.4$ Hz, 1H), 6.82 (t, $J = 9.1$ Hz, 1H), 6.52 (dd, $J = 13.3, 2.7$ Hz, 1H), 6.41 (dd, $J = 8.7, 1.2$ Hz, 1H), 5.87 (ddd, $^2J_{\text{HF}} = 48.7$ Hz, $J = 7.2, 3.7$ Hz, 1H), 4.12 (s, 1H), 3.94 – 3.82 (m, 1H), 3.74 – 3.62 (m, 1H), 2.15 (s, 3H).

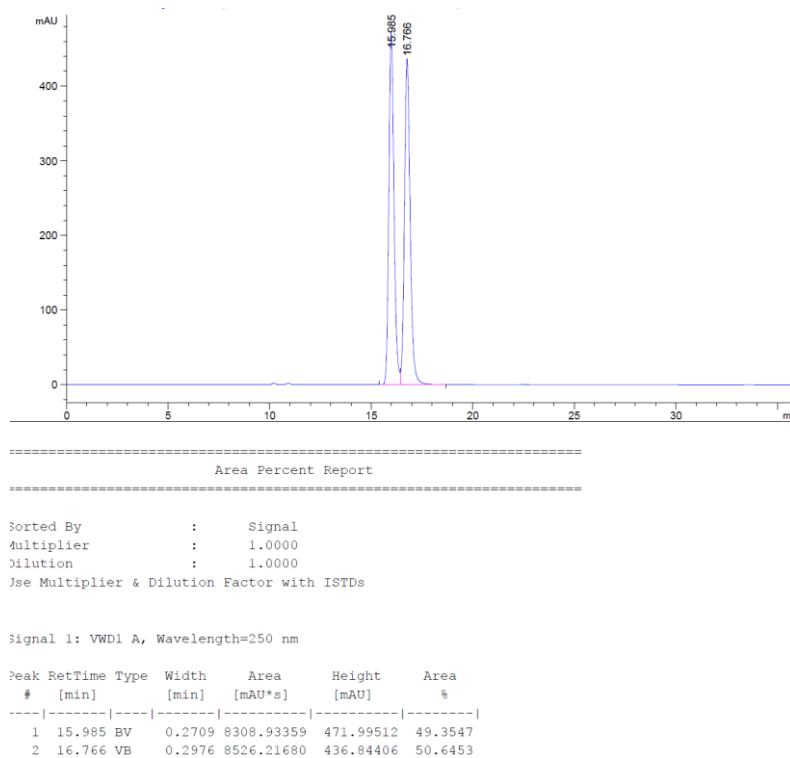
$^{13}\text{C NMR}$ (126 MHz, CDCl_3): δ 162.2 (d, $^1J = 242.4$ Hz), 157.9 (d, $^2J = 24.6$ Hz), 147.4, 147.3 (d, $^3J = 10.6$ Hz), 137.3, 131.8 (d, $^3J = 7.1$ Hz), 130.1, 129.3, 127.9, 127.8, 127.0, 117.8 (d, $^3J = 6.5$ Hz), 113.6 (d, $^2J = 17.7$ Hz), 109.15 (d, $^4J = 2.4$ Hz), 100.4 (d, $^2J = 26.1$ Hz), 93.2 (d, $^1J = 174.8$ Hz), 48.7 (d, $^2J = 22.4$ Hz), 13.7 (d, $^4J = 3.3$ Hz).

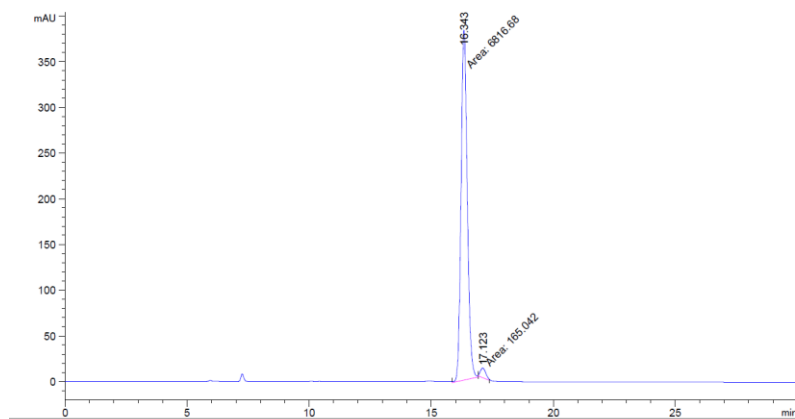
$^{19}\text{F NMR}$ (470 MHz, CDCl_3): δ -116.69, -191.35.

HRMS (ESI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{18}\text{H}_{17}\text{ON}_2\text{F}_2$) requires m/z 299.1354, found m/z 299.1351.

The enantiomeric purity of the product was determined by HPLC analysis: 98:2 e.r. (Chiralpak IA, hexane/*i*-PrOH = 90/10, flow rate 0.3 mL/min, $\lambda = 250$ nm), t_r (major) = 16.3 min, t_r (minor) = 17.1 min;

$[\alpha]_{\text{D}}^{20} = +14.6$ (c 1.6, CHCl_3).





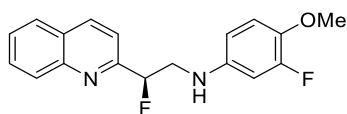
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Area Percent Report
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Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	16.343	MM	0.2963	6816.67725	383.37057	97.6361
2	17.123	MM	0.2606	165.04242	10.55658	2.3639

(R)-3-Fluoro-N-(2-fluoro-2-(quinolin-2-yl)ethyl)-4-methoxyaniline 7j



Prepared according to General Procedure A using 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), 3-fluoro-4-methoxyaniline (42.3 mg, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and THF (200 μ L, 0.5 M). The reaction mixture was subjected to the purification outlined in General Procedure A (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale yellow oil (24.6 mg, 72%).

ν_{max} (film): 3398 (br), 3056, 2948, 2932, 2833, 1519, 1227 cm^{-1} .

$^1\text{H NMR}$ (500 MHz, CDCl_3): δ 8.23 (d, $J = 8.5$ Hz, 1H), 8.09 (d, $J = 8.5$ Hz, 1H), 7.84 (d, $J = 8.1$ Hz, 1H), 7.78 – 7.71 (m, 1H), 7.66 – 7.60 (m, 1H), 7.57 (t, $J = 7.4$ Hz, 1H), 6.82 (t, $J = 9.1$ Hz, 1H), 6.52 (dd, $J = 13.3, 2.7$ Hz, 1H), 6.41 (dd, $J = 8.7, 1.2$ Hz, 1H), 5.87 (ddd, $^2J_{\text{HF}} = 48.7$ Hz, $J = 7.2, 3.7$ Hz, 1H), 4.12 (s, 1H), 3.94 – 3.82 (m, 1H), 3.80 (s, 3H), 3.74 – 3.62 (m, 1H).

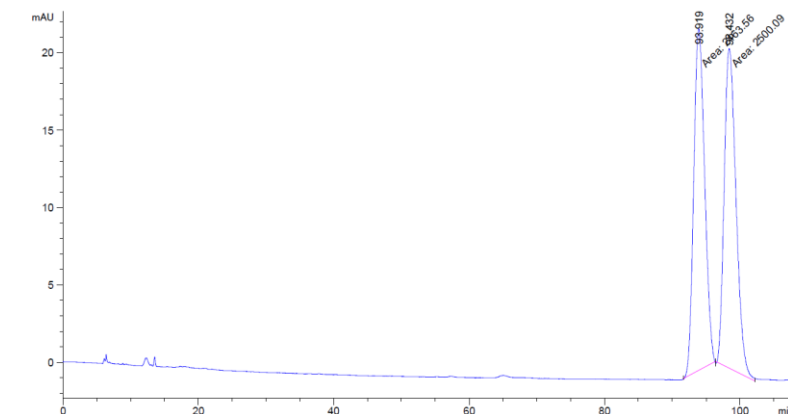
$^{13}\text{C NMR}$ (126 MHz, CDCl_3): δ 158.0 (d, $^2J = 24.6$ Hz), 153.7 (d, $^1J = 244.2$ Hz), 147.5, 142.8 (d, $^3J = 9.1$ Hz), 140.0 (d, $^3J = 11.3$ Hz), 137.3, 130.1, 129.4, 127.9, 127.9, 127.0, 117.8 (d, $^3J = 6.5$ Hz), 116.1 (d, $^4J = 3.1$ Hz), 108.7 (d, $^4J = 3.2$ Hz), 102.7 (d, $^2J = 22.0$ Hz), 93.3 (d, $^1J = 174.7$ Hz), 57.7, 49.3 (d, $^2J = 22.5$ Hz).

$^{19}\text{F NMR}$ (376 MHz, CDCl_3): δ -133.38, -191.48.

HRMS (NSI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{18}\text{H}_{17}\text{ON}_2\text{F}_2$) requires m/z 315.1303, found m/z 315.1306.

The enantiomeric purity of the product was determined by HPLC analysis: 93:7 e.r. (Chiralcel OJ-H, hexane/*i*-PrOH = 70/30, flow rate 0.5 mL/min, $\lambda = 250$ nm), t_r (major) = 92.3 min, t_r (minor) = 97.7 min;

$[\alpha]_D^{20} = +7.9$ (c 1.2, CHCl_3).



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                        Area Percent Report
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Sorted By      :      Signal
Multiplier     :      1.0000
Dilution      :      1.0000
Use Multiplier & Dilution Factor with ISTDs

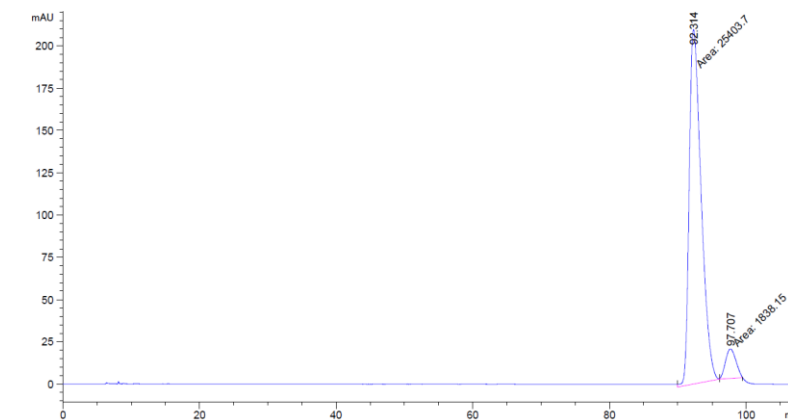
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Signal 1: VWD1 A, Wavelength=250 nm

Peak RetTime Type Width Area Height Area
# [min] [min] [min] [mAU*s] [mAU] [%]
-----|-----|-----|-----|-----|-----|
1 93.919 MM 1.8602 2463.55908 22.07241 49.6320
2 98.432 MM 2.0162 2500.09351 20.66677 50.3680

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                        Area Percent Report
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Sorted By      :      Signal
Multiplier     :      1.0000
Dilution      :      1.0000
Use Multiplier & Dilution Factor with ISTDs

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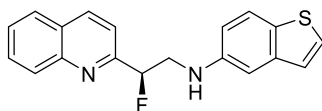
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Signal 1: VWD1 A, Wavelength=250 nm

Peak RetTime Type Width Area Height Area
# [min] [min] [min] [mAU*s] [mAU] [%]
-----|-----|-----|-----|-----|-----|
1 92.314 MM 2.0188 2.54037e4 209.72574 93.2525
2 97.707 MM 1.7613 1838.15466 17.39389 6.7475

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(*R*)-*N*-(2-Fluoro-2-(quinolin-2-yl)ethyl)benzo[*b*]thiophen-5-amine **7k**



Prepared according to General Procedure A using 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), benzo[*b*]thiophen-5-amine (44.7 mg, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and THF (200 μ L, 0.5 M) at -20 $^{\circ}$ C for 5 d. The reaction mixture was subjected to the purification outlined in General Procedure A (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale yellow solid (18.7 mg, 58%).

ν_{max} (film): 3389 (br), 3058, 3021, 2924, 2848, 1599, 1504, 1437, 1065, 831, 749, 691 cm^{-1} .

$^1\text{H NMR}$ (500 MHz, MeOD): δ 8.30 (d, $J = 8.5$ Hz, 1H), 8.07 (d, $J = 8.5$ Hz, 1H), 7.88 (d, $J = 8.1$ Hz, 1H), 7.82 – 7.73 (m, 1H), 7.65 (d, $J = 8.4$ Hz, 1H), 7.57 (t, $J = 7.2$ Hz, 1H), 7.54 (d, $J = 8.7$ Hz, 1H), 7.36 (d, $J = 5.4$ Hz, 1H), 7.08 (d, $J = 5.7$ Hz, 2H), 6.81 (dd, $J = 8.7, 2.2$ Hz, 1H), 5.86 (ddd, $^2J_{\text{HF}} = 48.7$ Hz, $J = 7.5, 3.4$ Hz, 1H), 3.86 (ddd, $^3J_{\text{HF}} = 27.5$ Hz, $^2J_{\text{HH}} = 14.6$ Hz, $J = 3.4$ Hz, 1H), 3.71 (ddd, $^3J_{\text{HF}} = 20.2$ Hz, $^2J_{\text{HH}} = 14.6$ Hz, $J = 7.5$ Hz, 1H). (N-H proton not observed)

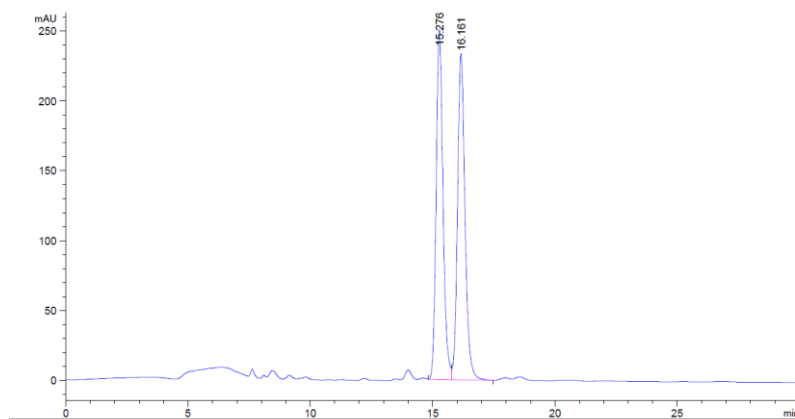
$^{13}\text{C NMR}$ (126 MHz, MeOD): δ 159.7 (d, $^2J = 23.9$ Hz), 148.4, 146.9, 142.4, 138.9, 131.3, 130.7, 129.3, 129.3, 129.1, 128.1, 127.5, 124.4, 123.5, 119.3 (d, $^3J = 6.0$ Hz), 115.3, 106.2, 94.4 (d, $^1J = 174.9$ Hz), 50.1 (d, $^2J = 23.2$ Hz).

$^{19}\text{F NMR}$ (376 MHz, MeOD): δ -190.65 .

HRMS (NSI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{19}\text{H}_{16}\text{N}_2\text{FS}$) requires m/z 323.1013, found m/z 323.1016.

The enantiomeric purity of the product was determined by HPLC analysis: 94:6 e.r. (Chiralpak IA, hexane/*i*-PrOH = 50/50, flow rate 0.5 mL/min, $\lambda = 250$ nm), t_r (major) = 14.9 min, t_r (minor) = 15.8 min;

$[\alpha]_{\text{D}}^{20} = +9.0$ (c 0.94, CHCl_3).

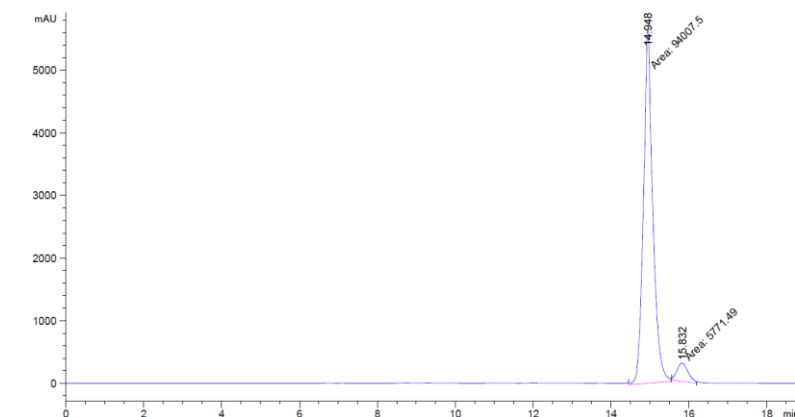


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 Area Percent Report
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Sorted By : Signal
 Multiplier : 1.0000
 Dilution : 1.0000
 Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	15.276	VV	0.2986	4870.37109	250.01653	49.5778
2	16.161	VB	0.3233	4953.32080	233.38797	50.4222



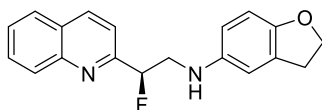
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 Area Percent Report
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Sorted By : Signal
 Multiplier : 1.0000
 Dilution : 1.0000
 Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	14.948	MM	0.2780	9.40075e4	5635.32910	94.2157
2	15.832	MM	0.3288	5771.49170	292.51791	5.7843

(R)-N-(2-Fluoro-2-(quinolin-2-yl)ethyl)-2,3-dihydrobenzofuran-5-amine 7I



Prepared according to General Procedure A using 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), 2,3-dihydrobenzofuran-5-amine (40.5 mg, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg,

0.02 mmol, 20 mol%), and THF (200 μ L, 0.5 M) at -20°C for 5 d. The reaction mixture was subjected to the purification outlined in General Procedure A (silica gel, EtOAc:petroleum ether, 5:95 to 10:100) to afford the desired product as a pale yellow oil (11.4 mg, 37%).

ν_{max} (film): 3376 (br), 3051, 2954, 2921, 2887, 2850, 1491, 1216 cm^{-1} .

$^1\text{H NMR}$ (400 MHz, CDCl_3): δ 8.22 (d, $J = 8.5$ Hz, 1H), 8.09 (dd, $J = 8.6, 1.1$ Hz, 1H), 7.84 (dd, $J = 8.2, 1.4$ Hz, 1H), 7.75 (ddd, $J = 8.4, 6.8, 1.5$ Hz, 1H), 7.63 (dd, $J = 8.5, 1.5$ Hz, 1H), 7.57 (ddd, $J = 8.1, 6.8, 1.2$ Hz, 1H), 6.70 – 6.62 (m, 2H), 6.51 (dd, $J = 8.5, 2.5$ Hz, 1H), 5.89 (ddd, $^2J_{\text{HF}} = 48.7$ Hz, $J = 7.4, 3.7$ Hz, 1H), 4.55 – 4.39 (m, 2H), 3.86 (ddd, $^3J_{\text{HF}} = 26.5$ Hz, $^2J_{\text{HH}} = 13.9$ Hz, $J = 3.7$ Hz, 1H), 3.67 (ddd, $^3J_{\text{HF}} = 19.6$ Hz, $^2J_{\text{HH}} = 13.9$ Hz, $J = 7.4$ Hz, 1H), 3.19 – 3.01 (m, 2H).

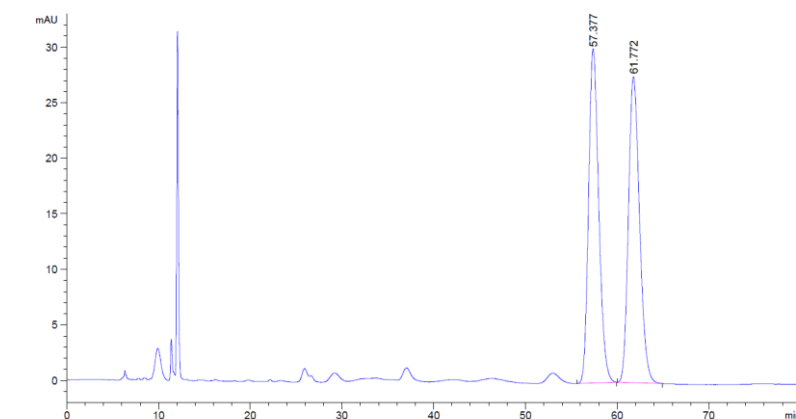
$^{13}\text{C NMR}$ (101 MHz, CDCl_3): δ 158.3 (d, $^2J = 24.8$ Hz), 153.2, 147.5, 141.9, 137.2, 130.0, 129.3, 128.0, 127.8, 126.9, 118.0, 117.9, 113.6, 111.5, 109.5, 93.4 (d, $^1J = 174.2$ Hz), 71.1, 50.2 (d, $^2J = 22.2$ Hz), 30.4.

$^{19}\text{F NMR}$ (376 MHz, CDCl_3): $\delta -191.64$.

HRMS (NSI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{19}\text{H}_{18}\text{ON}_2\text{F}$) requires m/z 309.1398, found m/z 309.1401.

The enantiomeric purity of the product was determined by HPLC analysis: 89:11 e.r. (Chiralcel OJ-H, hexane/*i*-PrOH = 50/50, flow rate 0.5 mL/min, $\lambda = 250$ nm), t_r (major) = 55.0 min, t_r (minor) = 60.2 min;

$[\alpha]_{\text{D}}^{20} = +2.5$ (c 0.57, CHCl_3).

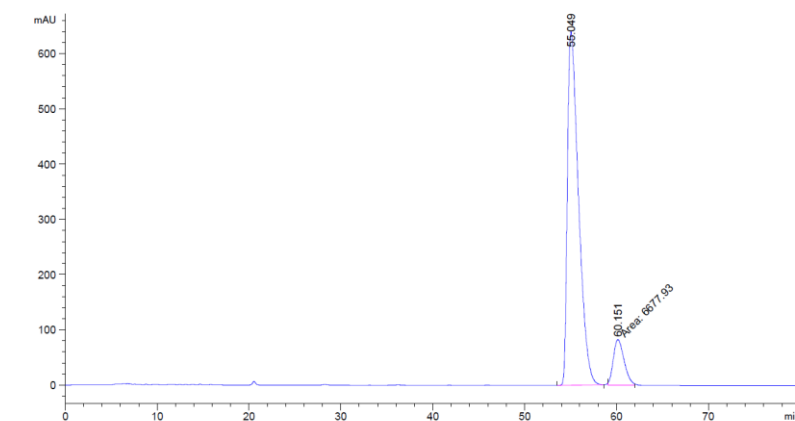


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Area Percent Report
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Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: WVD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	57.377	BB	1.1785	2316.01807	30.10777	50.0123
2	61.772	BB	1.2849	2314.87866	27.54219	49.9877



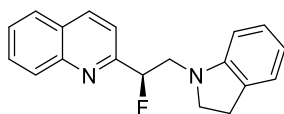
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 Area Percent Report
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Sorted By : Signal
 Multiplier : 1.0000
 Dilution : 1.0000
 Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	55.049	BB	1.2743	5.37853e4	640.15222	88.9554
2	60.151	MM	1.3544	6677.92578	82.17638	11.0446

(R)-2-(1-Fluoro-2-(indolin-1-yl)ethyl)quinoline 7m



Prepared according to General Procedure B using 2-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), indoline (35.7 mg, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and CPME (200 μ L, 0.5 M). The reaction mixture was subjected to the purification outlined in General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 10:100) to afford the desired product as a pale yellow solid (26.9 mg, 92%).

ν_{max} (film): 3048, 3024, 2949, 2922, 2845, 1607, 1599, 1489, 1244, 742 cm^{-1} .

^1H NMR (400 MHz, CDCl_3): δ 8.22 (d, J = 8.5 Hz, 1H), 8.12 (d, J = 8.4 Hz, 1H), 7.84 (dd, J = 8.2, 0.9 Hz, 1H), 7.75 (ddd, J = 8.4, 6.9, 1.4 Hz, 1H), 7.67 (dd, J = 8.5, 1.4 Hz, 1H), 7.57 (ddd, J = 8.1, 6.9, 1.1 Hz, 1H), 7.09 – 7.00 (m, 2H), 6.65 (td, J = 7.4, 0.8 Hz, 1H), 6.56 (d, J = 7.8 Hz, 1H), 5.98 (ddd, $^2J_{\text{HF}}$ = 48.7 Hz, J = 6.8, 2.9 Hz, 1H), 3.86 (ddd, $^3J_{\text{HF}}$ = 28.6 Hz, $^2J_{\text{HH}}$ = 15.1 Hz, J = 2.9 Hz, 1H), 3.72 (ddd, $^3J_{\text{HF}}$ = 24.0 Hz, $^2J_{\text{HH}}$ = 15.1 Hz, J = 6.8 Hz, 1H), 3.56 (q, J = 8.3 Hz, 1H), 3.48 (qd, J = 8.7, 0.8 Hz, 1H), 2.98 (t, J = 8.4 Hz, 2H).

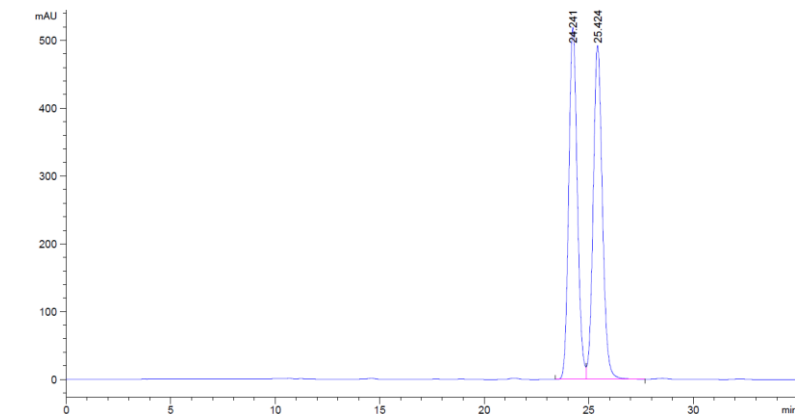
^{13}C NMR (101 MHz, CDCl_3): δ 158.5 (d, 2J = 24.8 Hz), 152.3, 147.6, 137.1, 130.0, 129.6, 129.4, 127.9, 127.8, 127.5, 126.8, 124.5, 118.1 (d, 3J = 6.2 Hz), 117.8, 106.9, 94.7 (d, 1J = 176.7 Hz), 54.9, 54.8 (d, 2J = 20.9 Hz), 28.9.

^{19}F NMR (376 MHz, CDCl_3): δ -187.59.

HRMS (NSI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{19}\text{H}_{18}\text{N}_2\text{F}$) requires m/z 293.1449, found m/z 293.1450.

The enantiomeric purity of the product was determined by HPLC analysis: 92:8 e.r. (Chiralcel OJ-H, hexane/*i*-PrOH = 50/50, flow rate 0.5 mL/min, $\lambda = 250$ nm), t_r (major) = 24.9 min, t_r (minor) = 24.0 min;

$$[\alpha]_D^{20} = +21.7 (c 1.35, \text{CHCl}_3).$$



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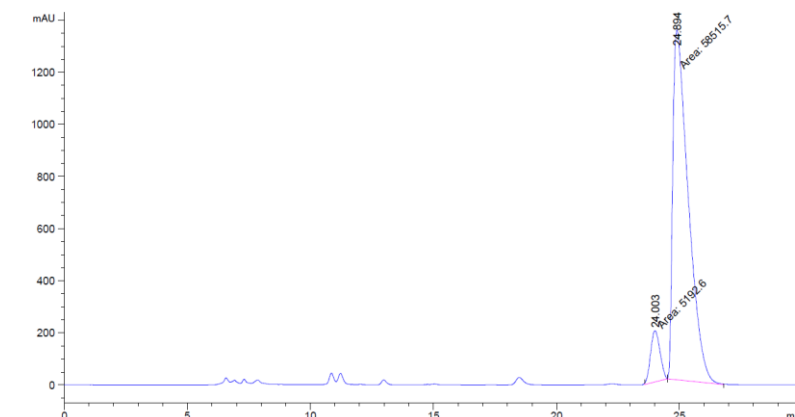
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                        Area Percent Report
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Sorted By      :      Signal
Multiplier     :      1.0000
Dilution      :      1.0000
Use Multiplier & Dilution Factor with ISTDs
  
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Signal 1: VWd1 A, Wavelength=250 nm

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Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	24.241	BV	0.4392	1.47304e4	518.54968	49.4921
2	25.424	VB	0.4713	1.50327e4	492.28079	50.5079



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                        Area Percent Report
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Sorted By      :      Signal
Multiplier     :      1.0000
Dilution      :      1.0000
Use Multiplier & Dilution Factor with ISTDs
  
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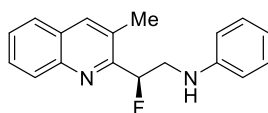
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Signal 1: VWd1 A, Wavelength=250 nm

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Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	24.003	MM	0.4411	5192.60156	196.21417	8.1506
2	24.894	MM	0.7268	5.85157e4	1341.82629	91.8494

(*R*)-*N*-(2-Fluoro-2-(3-methylquinolin-2-yl)ethyl)aniline **7n**



Prepared according to General Procedure B using 3-methyl-2-(1-fluorovinyl)quinoline (18.8 mg, 0.10 mmol, 1.0 equiv), aniline (27 μ L, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and CPME (200 μ L, 0.5 M) at -10 $^{\circ}$ C for 3 d. The reaction mixture was subjected to the purification outlined the General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 1:100) to afford the desired product as a pale yellow solid (13.8 mg, 50%).

ν_{\max} (film): 1603, 1506, 1497, 1447, 1325, 1261, 750, 692 cm^{-1} .

$^1\text{H NMR}$ (500 MHz, CDCl_3): δ 8.12 (d, $J = 8.4$ Hz, 1H), 7.96 (s, 1H), 7.76 (d, $J = 8.1$ Hz, 1H), 7.69 (t, $J = 7.6$ Hz, 1H), 7.55 (t, $J = 7.5$ Hz, 1H), 7.21 (t, $J = 7.8$ Hz, 2H), 6.74 (dd, $J = 10.9, 7.9$ Hz, 3H), 6.00 (ddd, $^2J_{\text{HF}} = 48.1$ Hz, $J = 7.2, 4.9$ Hz, 1H), 4.41 (s, 1H), 4.12 – 3.91 (m, 2H), 2.58 (s, 3H).

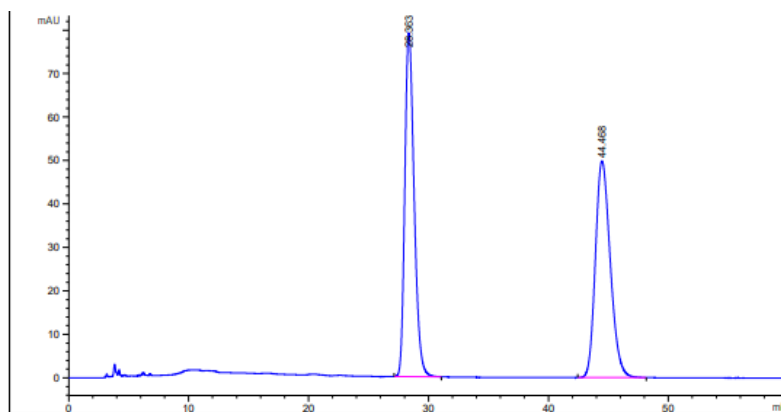
$^{13}\text{C NMR}$ (126 MHz, CDCl_3): δ 155.2 (d, $^3J = 18.1$ Hz), 147.7, 146.0, 137.3, 129.9, 129.5, 129.4, 129.0, 128.2, 127.2, 126.8, 117.9, 113.2, 90.2 (d, $^1J = 172.8$ Hz), 46.2 (d, $^2J = 24.6$ Hz), 18.6 (d, $^4J = 3.3$ Hz).

$^{19}\text{F NMR}$ (376 MHz, CDCl_3): δ -183.72 .

HRMS (ESI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{18}\text{H}_{18}\text{FN}_2$) requires m/z 281.1449, found m/z 281.1441.

The enantiomeric purity of the product was determined by HPLC analysis: 91:9 e.r. (Chiralcel OJ-H, hexane/*i*-PrOH = 50/50, flow rate 1.0 mL/min, $\lambda = 250$ nm), t_r (major) = 28.8 min, t_r (minor) = 45.1 min;

$[\alpha]_{\text{D}}^{20} = -71.9$ (c 0.21, in CHCl_3).

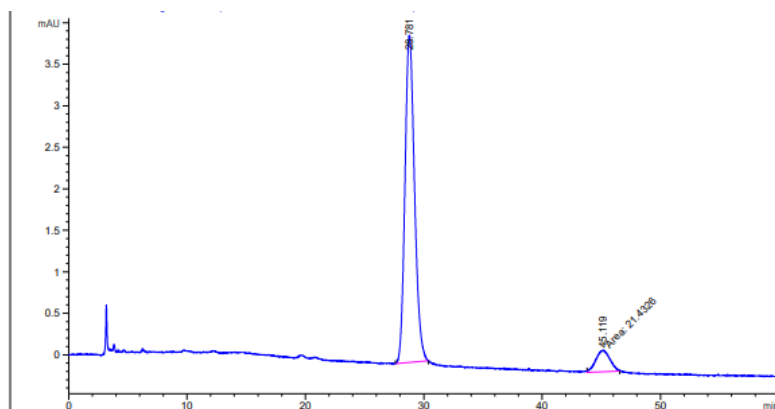


Area Percent Report

Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	28.363	BB	0.8188	4269.78467	79.12798	49.9060
2	44.468	BB	1.2874	4285.86523	49.82888	50.0940



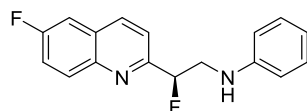
Area Percent Report

Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	28.781	BB	0.8186	221.00998	3.94218	91.1597
2	45.119	MM	1.3857	21.43265	2.57792e-1	8.8403

(R)-*N*-(2-Fluoro-2-(6-fluoroquinolin-2-yl)ethyl)aniline **7o**



Prepared according to General Procedure B using 6-fluoro-2-(1-fluorovinyl)quinoline (19.1 mg, 0.10 mmol, 1.0 equiv), aniline (27 μ L, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20

mol%), and CPME (200 μ L, 0.5 M) at -10 $^{\circ}$ C for 3 d. The reaction mixture was subjected to the purification outlined in General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 10:100) to afford the desired product as a pale brown oil (18.9 mg, 67%).

ν_{\max} (film): 1603, 1506, 1323, 1260, 1231, 1142, 1061, 872, 833, 750, 692 cm^{-1} .

^1H NMR (400 MHz, CDCl_3): δ 8.17 (d, J = 8.6 Hz, 1H), 8.10 (dd, J = 9.2, 5.3 Hz, 1H), 7.65 (d, J = 8.5 Hz, 1H), 7.52 (td, J = 8.7, 2.8 Hz, 1H), 7.46 (dd, J = 8.8, 2.8 Hz, 1H), 7.19 (t, J = 7.8 Hz, 2H), 6.78 – 6.70 (m, 3H), 5.88 (ddd, $^2J_{\text{HF}}$ = 48.6 Hz, J = 7.3, 3.6 Hz, 1H), 3.94 (ddd, $^3J_{\text{HF}}$ = 26.4 Hz, $^2J_{\text{HH}}$ = 14.1 Hz, J = 3.7 Hz, 1H), 3.72 (ddd, $^3J_{\text{HF}}$ = 20.0 Hz, $^2J_{\text{HH}}$ = 14.1 Hz, J = 7.3 Hz, 1H).

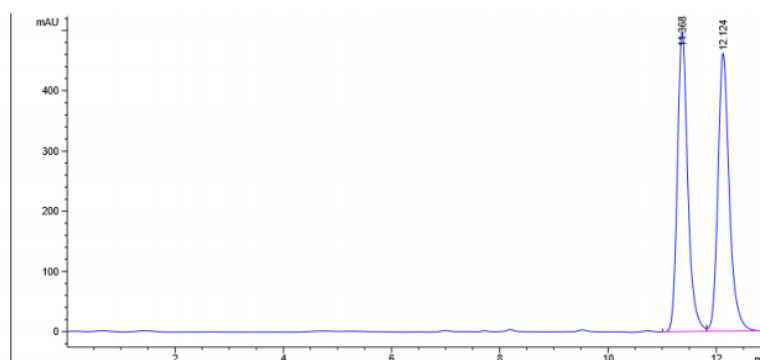
^{13}C NMR (101 MHz, CDCl_3): δ 160.7 (d, 1J = 248.7 Hz), 157.6 (d, 4J = 2.5 Hz), 157.4 (d, 4J = 2.9 Hz), 147.6, 144.6, 136.6 (d, 3J = 5.5 Hz), 131.9 (d, 3J = 9.2 Hz), 129.5, 120.4 (d, 2J = 25.8 Hz), 118.7 (d, 3J = 6.4 Hz), 118.2, 113.4, 110.8 (d, 2J = 21.7 Hz), 93.2 (d, 1J = 174.5 Hz), 48.5 (d, 2J = 22.4 Hz).

^{19}F NMR (376 MHz, CDCl_3): δ -112.80 , -191.35 .

HRMS (ESI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{17}\text{H}_{15}\text{F}_2\text{N}_2$) requires m/z 285.1198, found m/z 285.1190.

The enantiomeric purity of the product was determined by HPLC analysis: 94:6 e.r. (Chiralpak IA, hexane/*i*-PrOH = 50/50, flow rate 0.5 mL/min, λ = 250 nm), t_r (major) = 11.4 min, t_r (minor) = 12.1 min;

$[\alpha]_{\text{D}}^{20}$ = $+40.0$ (c 0.10, CHCl_3).

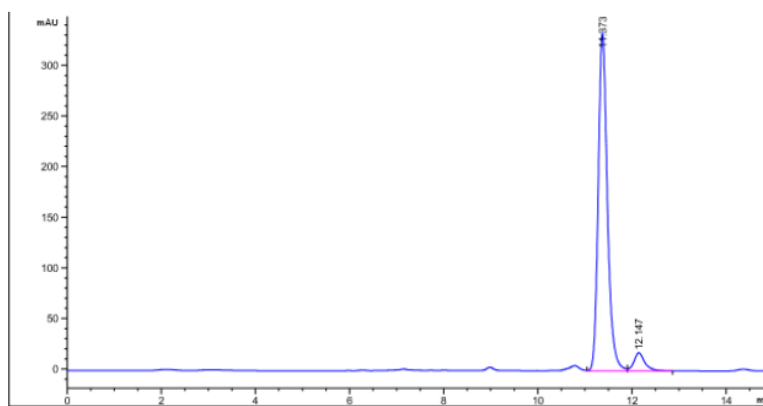


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Area Percent Report
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Sorted By : Signal
Multiplier: : 1.0000
Dilution: : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 A, Sig=254,4 Ref=360,100

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	11.368	VV	0.2018	6660.14502	498.86203	49.7094
2	12.124	VB	0.2223	6738.01025	461.47482	50.2906



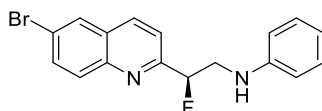
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Area Percent Report
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Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	11.373	VV	0.2056	4506.97461	333.68134	94.0452
2	12.147	VB	0.2389	285.37711	17.71371	5.9548

(R)-*N*-(2-(6-Bromoquinolin-2-yl)-2-fluoroethyl)aniline **7p**



Prepared according to General Procedure B using 6-bromo-2-(1-fluorovinyl)quinoline (25.2 mg, 0.10 mmol, 1.0 equiv), aniline (27 μ L, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and CPME (200 μ L, 0.5 M) at -10 $^{\circ}$ C for 3 d. The reaction mixture was subjected to the purification outlined in General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 10:100) to afford the desired product as a pale brown oil (33.1 mg, 96%).

ν_{\max} (film): 1603, 1508, 1491, 1059, 831, 750, 692 cm^{-1} .

^1H NMR (400 MHz, CDCl_3): δ 8.13 (d, $J = 8.5$ Hz, 1H), 8.00 (d, $J = 2.2$ Hz, 1H), 7.96 (d, $J = 9.0$ Hz, 1H), 7.81 (dd, $J = 9.0, 2.2$ Hz, 1H), 7.65 (d, $J = 8.5$ Hz, 1H), 7.19 (t, $J = 7.8$ Hz, 2H), 6.78 – 6.69 (m, 3H), 5.87 (ddd, $^2J_{\text{HF}} = 48.7$ Hz, $J = 7.3, 3.6$ Hz, 1H), 4.25 (s, 1H), 3.95 (ddd, $^3J_{\text{HF}} = 26.4$ Hz, $^2J_{\text{HH}} = 14.1$ Hz, $J = 3.6$ Hz, 1H), 3.71 (ddd, $^3J_{\text{HF}} = 20.9$ Hz, $^2J_{\text{HH}} = 14.1$ Hz, $J = 7.3$ Hz, 1H).

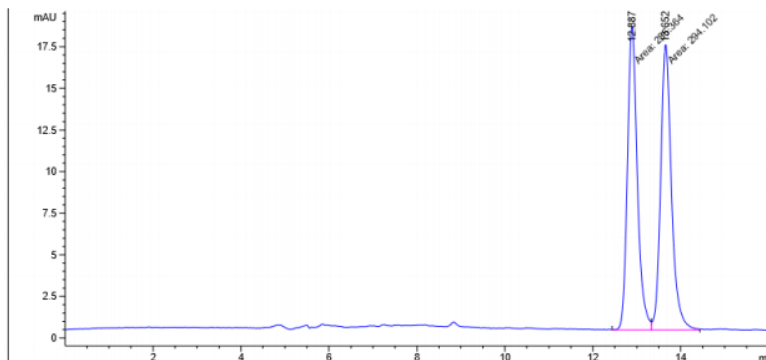
^{13}C NMR (101 MHz, CDCl_3): δ 158.6 (d, $^2J = 24.8$ Hz), 147.5, 146.1 (d, $^4J = 2.4$ Hz), 136.2, 133.5, 131.1, 129.9, 129.5, 128.9, 120.8, 118.8 (d, $^3J = 6.6$ Hz), 118.3, 113.4, 93.2 (d, $^1J = 174.8$ Hz), 48.5 (d, $^2J = 22.2$ Hz).

^{19}F NMR (376 MHz, CDCl_3): δ -191.78.

HRMS (ESI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{17}\text{H}_{15}\text{BrFN}_2$) requires m/z 345.0397, found m/z 345.0393.

The enantiomeric purity of the product was determined by HPLC analysis: 93:7 e.r. (Chiralpak IA, hexane/*i*-PrOH = 50/50, flow rate 0.5 mL/min, $\lambda = 250$ nm), t_r (major) = 12.7 min, t_r (minor) = 13.4 min;

$$[\alpha]_D^{20} = +11.8 (c 1.15, \text{CHCl}_3).$$

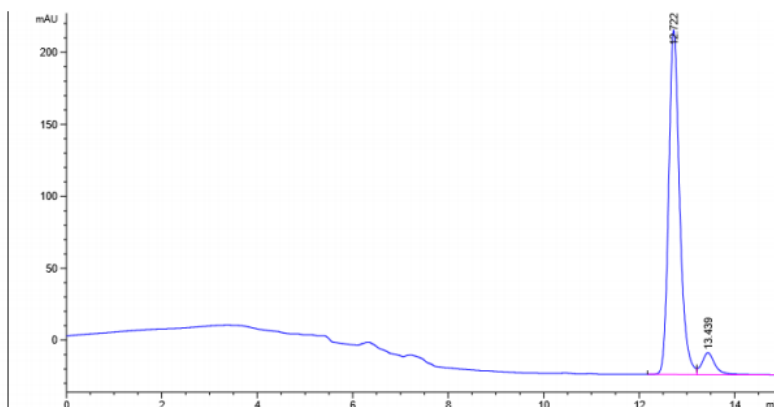


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 Area Percent Report
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Sorted By : Signal
 Multiplier: : 1.0000
 Dilution: : 1.0000
 Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 A, Sig=254,4 Ref=360,100

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	12.887	MF	0.2617	286.36404	18.24028	49.3335
2	13.652	FM	0.2859	294.10202	17.14728	50.6665



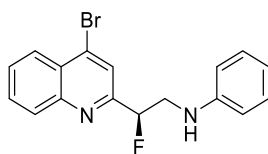
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 Area Percent Report
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Sorted By : Signal
 Multiplier : 1.0000
 Dilution : 1.0000
 Use Multiplier & Dilution Factor with ISTDs

Signal 1: WVD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	12.722	BV	0.2429	3816.47534	239.22588	92.9299
2	13.439	VBA	0.2828	290.35831	15.18612	7.0701

(*R*)-*N*-(2-(4-Bromoquinolin-2-yl)-2-fluoroethyl)aniline **7q**



Prepared according to General Procedure A using 4-bromo-2-(1-fluorovinyl)quinoline (25.2 mg, 0.10 mmol, 1.0 equiv), aniline (27 μ L, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and THF (200 μ L, 0.5 M). The reaction mixture was subjected to the purification outlined in General Procedure A (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale yellow solid (33.1 mg, 96%).

ν_{max} (film): 1603, 1508, 1491, 1319, 1261, 1186, 1059, 880, 831, 750, 692 cm^{-1} .

$^1\text{H NMR}$ (500 MHz, CDCl_3): δ 8.20 (dd, $J = 8.4, 1.4$ Hz, 1H), 8.09 (d, $J = 8.4$ Hz, 1H), 7.94 (d, $J = 1.5$ Hz, 1H), 7.80 (ddd, $J = 8.4, 6.9, 1.4$ Hz, 1H), 7.66 (ddd, $J = 8.3, 6.9, 1.2$ Hz, 1H), 7.19 (t, $J = 7.7$ Hz, 2H), 6.78 – 6.71 (m, 3H), 5.86 (ddd, $^2J_{\text{HF}} = 48.6$ Hz, $J = 7.3, 3.6$ Hz, 1H), 3.95 (ddd, $^3J_{\text{HF}} = 26.4$ Hz, $^2J_{\text{HH}} = 14.2$ Hz, $J = 3.6$ Hz, 1H), 3.78 – 3.66 (m, 1H).

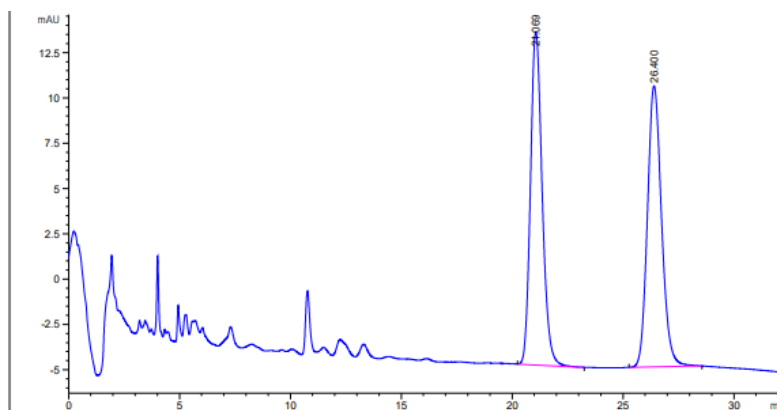
$^{13}\text{C NMR}$ (126 MHz, CDCl_3): δ 158.0 (d, $^2J = 25.0$ Hz), 148.1, 147.5, 135.4, 131.0, 129.8, 129.5, 128.2, 127.4, 127.0, 122.0 (d, $^3J = 7.5$ Hz), 118.3, 113.5, 92.7 (d, $^1J = 175.8$ Hz), 48.5 (d, $^2J = 21.9$ Hz).

$^{19}\text{F NMR}$ (470 MHz, CDCl_3): δ -192.02.

HRMS (ESI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{17}\text{H}_{15}\text{BrFN}_2$) requires m/z 345.0397, found m/z 345.0394.

The enantiomeric purity of the product was determined by HPLC analysis: 98:2 e.r. (Chiralcel OJ-H, hexane/*i*-PrOH = 50/50, flow rate 1.0 mL/min, $\lambda = 250$ nm), t_r (major) = 24.4 min, t_r (minor) = 20.5 min;

$[\alpha]_{\text{D}}^{20} = +30.7$ (c 0.27, CHCl_3).

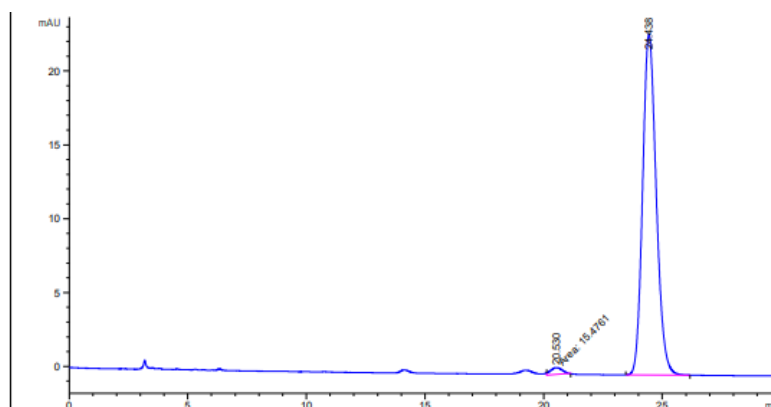


Area Percent Report

Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	21.069	BB	0.5653	673.27606	18.39787	49.5390
2	26.400	BB	0.6788	685.80560	15.51562	50.4610



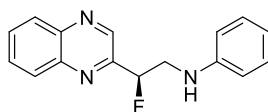
Area Percent Report

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Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	20.530	MM	0.5559	15.47610	4.63965e-1	1.6550
2	24.438	BB	0.6148	919.65417	23.08324	98.3450

(*R*)-*N*-(2-Fluoro-2-(quinoxalin-2-yl)ethyl)aniline **7r**



Prepared according to General Procedure B using 2-(1-fluorovinyl)quinoxaline (17.4 mg, 0.10 mmol, 1.0 equiv), aniline (27 μ L, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%),

and CPME (200 μ L, 0.5 M) at -10 $^{\circ}$ C for 3 d. The reaction mixture was subjected to the purification outlined in General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale brown oil (23.3 mg, 87%)

ν_{\max} (film): 1603, 1495, 1063, 750, 692 cm^{-1} .

^1H NMR (400 MHz, CDCl_3): δ 9.07 (s, 1H), 8.17 – 8.08 (m, 2H), 7.81 (qd, $J = 7.0, 3.4$ Hz, 2H), 7.19 (t, $J = 7.9$ Hz, 2H), 6.78 – 6.70 (m, 3H), 5.97 (ddd, $^2J_{\text{HF}} = 48.1$ Hz, $J = 7.1, 3.5$ Hz, 1H), 4.00 (ddd, $^3J_{\text{HF}} = 26.1$ Hz, $^2J_{\text{HH}} = 14.4$ Hz, $J = 3.6$ Hz, 1H), 3.79 (ddd, $^3J_{\text{HF}} = 21.3$ Hz, $^2J_{\text{HH}} = 14.4$ Hz, $J = 7.1$ Hz, 1H).

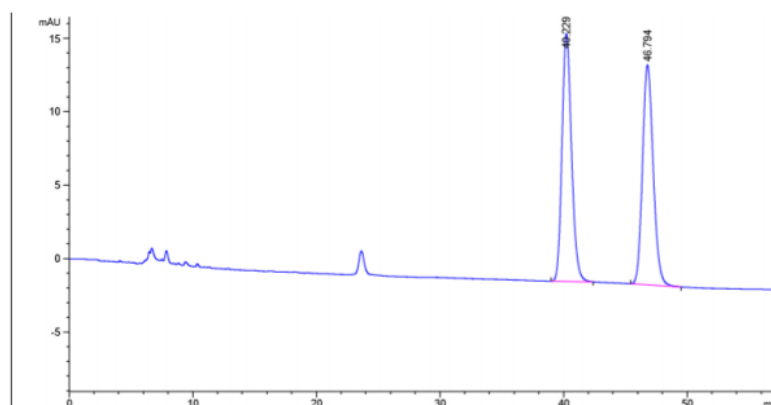
^{13}C NMR (101 MHz, CDCl_3): δ 152.4 (d, $^2J = 24.2$ Hz), 147.3, 142.8 (d, $^3J = 7.3$ Hz), 142.5, 141.5, 130.7, 130.4, 129.6, 129.5, 129.4, 118.5, 113.5, 92.3 (d, $^1J = 174.1$ Hz), 48.2 (d, $^2J = 22.0$ Hz).

^{19}F NMR (376 MHz, CDCl_3): δ -193.73 .

HRMS (ESI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{16}\text{H}_{15}\text{FN}_3$) requires m/z 268.1245, found m/z 268.1243.

The enantiomeric purity of the product was determined by HPLC analysis: 92:8 e.r. (Chiralcel OJ-H, hexane/*i*-PrOH = 50/50, flow rate 0.5 mL/min, $\lambda = 250$ nm), t_r (major) = 36.3 min, t_r (minor) = 42.2 min;

$[\alpha]_{\text{D}}^{20} = +5.4$ (c 0.13, CHCl_3).

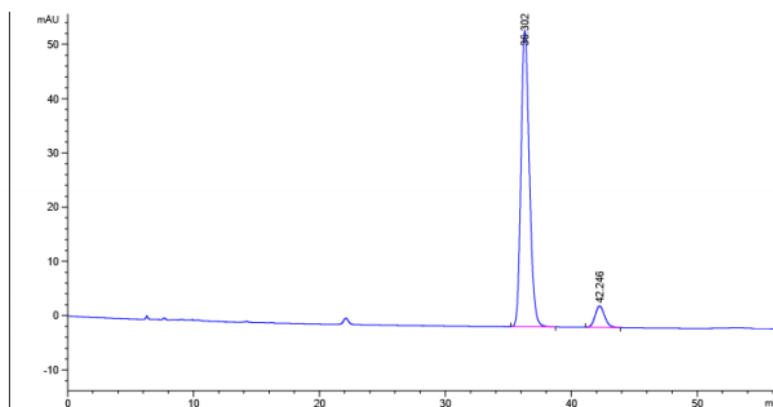


Area Percent Report

Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VwD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	40.229	BB	0.8238	902.49066	16.86399	49.9839
2	46.794	BB	0.9099	903.07245	14.98448	50.0161



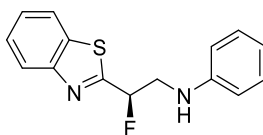
Area Percent Report

Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: Vwd1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	36.302	BB	0.7149	2526.63403	54.57219	92.4429
2	42.246	BB	0.7444	206.55081	3.94764	7.5571

(R)-N-(2-(Benzo[d]thiazol-2-yl)-2-fluoroethyl)aniline 7s



Prepared according to General Procedure B using 2-(1-fluorovinyl)benzo[d]thiazole (17.9 mg, 0.10 mmol, 1.0 equiv), aniline (27 μ L, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and CPME (200 μ L, 0.5 M) at -10 $^{\circ}$ C for 3 d. The reaction mixture was subjected to the purification outlined in General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale yellow solid (20.3 mg, 75%).

ν_{max} (film): 2357, 1603, 1508, 1317, 1260, 1061, 872, 754, 692 cm^{-1} .

$^1\text{H NMR}$ (400 MHz, CDCl_3): δ 8.07 (d, $J = 8.2$ Hz, 1H), 7.94 (d, $J = 7.9$ Hz, 1H), 7.53 (ddd, $J = 8.3, 7.2, 1.3$ Hz, 1H), 7.44 (ddd, $J = 8.3, 7.2, 1.2$ Hz, 1H), 7.22 (dd, $J = 8.6, 7.3$ Hz, 2H), 6.83 – 6.70 (m, 3H), 6.02 (ddd, $^2J_{\text{HF}} = 48.0$ Hz, $J = 7.2, 3.7$ Hz, 1H), 4.02 (ddd, $^3J_{\text{HF}} = 26.1$ Hz, $^2J_{\text{HH}} = 14.5$ Hz, $J = 3.7$ Hz, 1H), 3.80 (ddd, $^3J_{\text{HF}} = 19.7$ Hz, $^2J_{\text{HH}} = 14.5$ Hz, $J = 7.2$ Hz, 1H).

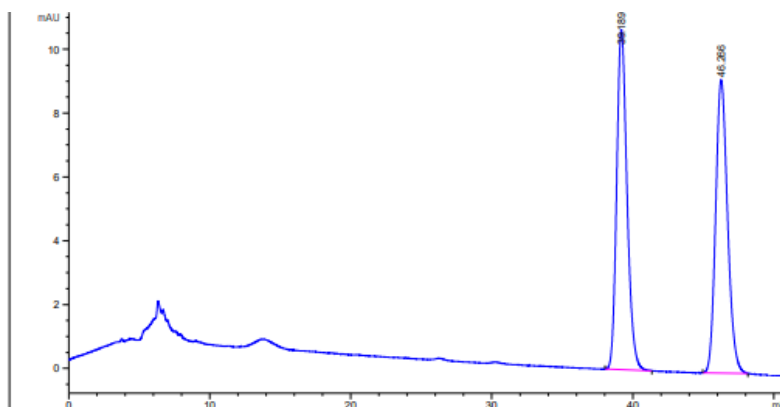
$^{13}\text{C NMR}$ (101 MHz, CDCl_3): δ 168.0 (d, $^2J = 28.0$ Hz), 153.1, 147.0, 134.9, 129.6, 126.6, 125.7, 123.6, 122.0, 118.7, 113.6, 90.7 (d, $^1J = 173.9$ Hz), 48.4 (d, $^2J = 22.0$ Hz).

$^{19}\text{F NMR}$ (377 MHz, CDCl_3): δ -181.92 .

HRMS (ESI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{15}\text{H}_{14}\text{FN}_2\text{S}$) requires m/z 273.0856, found m/z 273.0853.

The enantiomeric purity of the product was determined by HPLC analysis: 96:4 e.r. (Chiralcel OJ-H, hexane/*i*-PrOH = 50/50, flow rate 0.5 mL/min, $\lambda = 250$ nm), t_r (major) = 39.7 min, t_r (minor) 47.3 = min;

$$[\alpha]_D^{20} = -18.6 (c 0.43, \text{CHCl}_3).$$

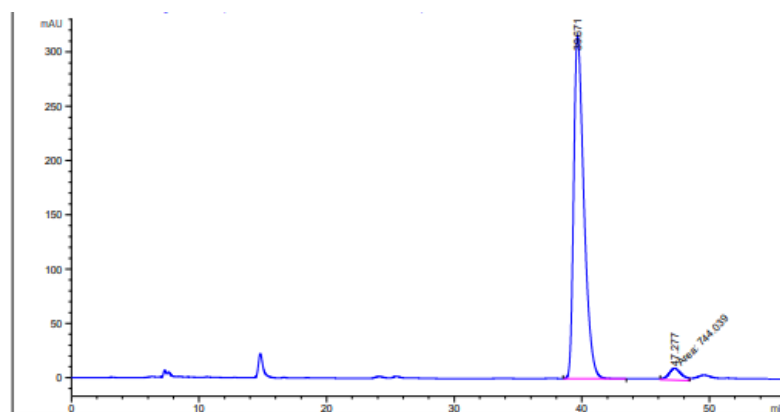


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Sorted By      :      Signal
Multiplier    :      1.0000
Dilution      :      1.0000
Use Multiplier & Dilution Factor with ISTDs
  
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Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	39.189	BB	0.7733	542.54297	10.65282	50.1830
2	46.266	BB	0.8892	538.58551	9.18738	49.8170



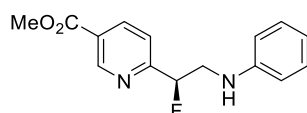
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Sorted By      :      Signal
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Dilution      :      1.0000
Use Multiplier & Dilution Factor with ISTDs
  
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Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	39.671	BB	0.8407	1.75361e4	316.16418	95.9298
2	47.277	MM	1.1413	744.03857	10.86501	4.0702

Methyl (*R*)-6-(1-fluoro-2-(phenylamino)ethyl)nicotinate **7t**



Prepared according to General Procedure B using methyl 6-(1-fluorovinyl)nicotinate (18.1 mg, 0.10 mmol, 1.0 equiv), aniline (27 μ L, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and CPME (200 μ L, 0.5 M) at rt for 3 d. The reaction mixture was subjected to the purification outlined in General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale brown oil (6.8 mg, 25%).

ν_{\max} (film): 1728, 1601, 1506, 1437, 1294, 1119, 750, 694 cm^{-1} .

$^1\text{H NMR}$ (400 MHz, CDCl_3): δ 9.20 – 9.19 (m, 1H), 8.35 (dd, $J = 8.2, 2.1$ Hz, 1H), 7.60 (d, $J = 8.2$ Hz, 1H), 7.25 – 7.13 (m, 2H), 6.80 – 6.65 (m, 3H), 5.78 (ddd, $^2J_{\text{HF}} = 48.4$ Hz, $J = 7.2, 3.3$ Hz, 1H), 4.03 – 3.82 (m, 4H), 3.61 (ddd, $^3J_{\text{HF}} = 21.3$ Hz, $^2J_{\text{HH}} = 14.3$ Hz, $J = 7.2$ Hz, 1H).

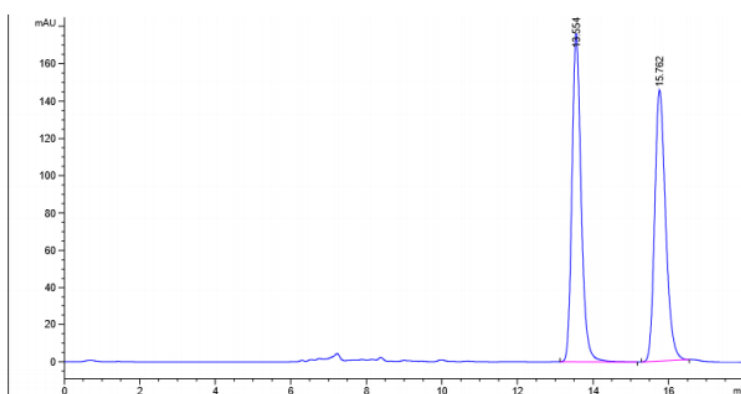
$^{13}\text{C NMR}$ (101 MHz, CDCl_3): δ 165.6, 162.0 (d, $^2J = 24.7$ Hz), 150.4 (d, $^4J = 2.7$ Hz), 147.4, 138.2, 129.5, 125.6, 119.8 (d, $^3J = 7.4$ Hz), 118.3, 113.4, 92.7 (d, $^1J = 175.3$ Hz), 52.6, 48.4 (d, $^2J = 21.6$ Hz).

$^{19}\text{F NMR}$ (376 MHz, CDCl_3): δ -193.31.

HRMS (ESI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{15}\text{H}_{16}\text{FN}_2\text{O}_2$) requires m/z 275.1190, found m/z 275.1190.

The enantiomeric purity of the product was determined by HPLC analysis: 88:12 e.r. (Chiralpak IA, hexane/*i*-PrOH = 50/50, flow rate 0.5 mL/min, $\lambda = 250$ nm), t_r (major) = 13.0 min, t_r (minor) = 14.8 min;

$[\alpha]_{\text{D}}^{20} = +2.95$ (c 0.88, CHCl_3).

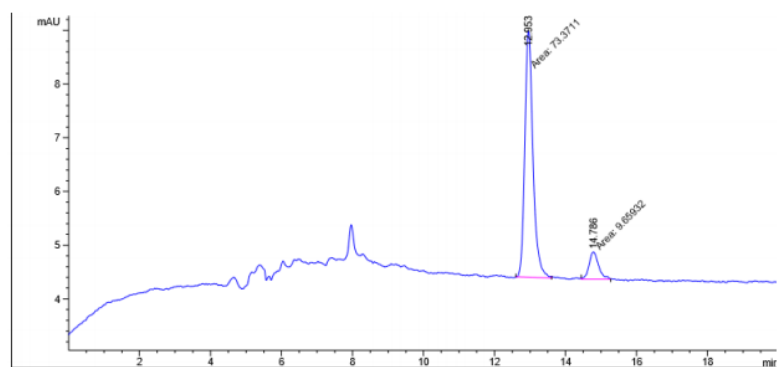


Area Percent Report

Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VMD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	13.554	BB	0.2588	2987.50024	176.31598	50.5592
2	15.762	BB	0.3083	2921.41113	145.62823	49.4408



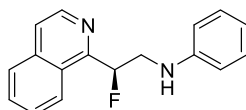
Area Percent Report

Sorted By : Signal
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Dilution: : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 A, Sig=254,4 Ref=360,100

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	12.953	MM	0.2664	73.37106	4.59014	88.3665
2	14.786	MM	0.3154	9.65932	5.10492e-1	11.6335

(R)-*N*-(2-Fluoro-2-(isoquinolin-1-yl)ethyl)aniline **7u**



Prepared according to General Procedure B using 1-(1-fluorovinyl)isoquinoline (17.3 mg, 0.10 mmol, 1.0 equiv), aniline (27 μ L, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and CPME (200 μ L, 0.5 M) at -10 $^{\circ}$ C for 3 d. The reaction mixture was subjected to the purification outlined in General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale brown oil (16.3 mg, 61%).

ν_{max} (film): 1603, 1504, 829, 748, 692 cm^{-1} .

^1H NMR (400 MHz, CDCl_3): δ 8.56 (d, J = 5.7 Hz, 1H), 8.31 (d, J = 8.0 Hz, 1H), 7.88 (d, J = 8.2 Hz, 1H), 7.77 – 7.60 (m, 3H), 7.25 – 7.16 (m, 2H), 6.79 – 6.69 (m, 3H), 6.37 (ddd, $^2J_{\text{HF}}$ = 48.4, J = 6.9, 5.2 Hz, 1H), 4.33 (s, 1H), 4.15 – 3.93 (m, 2H).

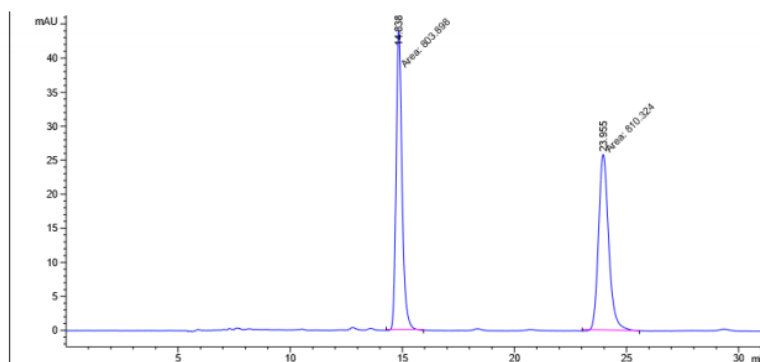
^{13}C NMR (101 MHz, CDCl_3): δ 155.2 (d, 2J = 18.6 Hz), 147.6, 141.7, 136.8, 130.4, 129.5, 128.0, 127.7, 126.9, 124.9 (d, 3J = 4.8 Hz), 122.0, 118.1, 113.4, 90.8 (d, 1J = 173.5 Hz), 46.9 (d, 2J = 24.1 Hz).

^{19}F NMR (376 MHz, CDCl_3): δ -180.92.

HRMS (ESI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{17}\text{H}_{16}\text{FN}_2$) requires m/z 267.1292, found m/z 267.1290.

The enantiomeric purity of the product was determined by HPLC analysis: 62:38 e.r. (Chiralpak IA, hexane/*i*-PrOH = 50/50, flow rate 0.5 mL/min, λ = 250 nm), t_r (major) = 14.8 min, t_r (minor) = 23.9 min;

$[\alpha]_{\text{D}}^{20}$ = +2.00 (c 0.10, CHCl_3).

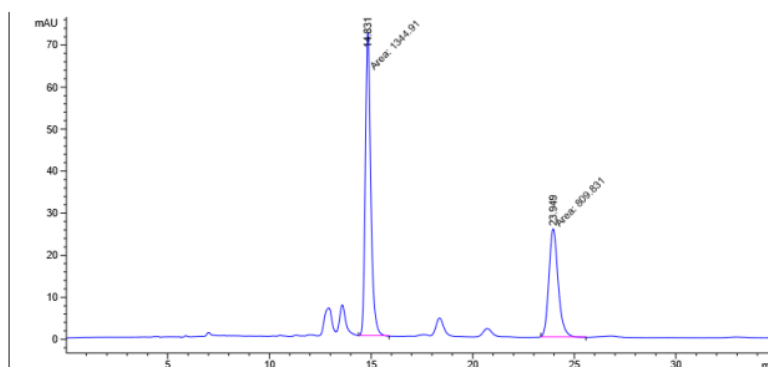


 Area Percent Report

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 Dilution: : 1.0000
 Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 A, Sig=254,4 Ref=360,100

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	14.838	MM	0.3049	803.89758	43.94274	49.8009
2	23.955	MM	0.5242	810.32446	25.76460	50.1991



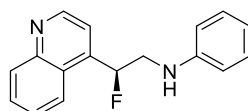
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Sorted By : Signal
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 Dilution: : 1.0000
 Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 A, Sig=254,4 Ref=360,100

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	14.831	MM	0.3106	1344.91199	72.15852	62.4163
2	23.949	MM	0.5258	809.83136	25.66826	37.5837

(S)-*N*-(2-Fluoro-2-(quinolin-4-yl)ethyl)aniline **7v**



Prepared according to General Procedure B using 4-(1-fluorovinyl)quinoline (17.3 mg, 0.10 mmol, 1.0 equiv), aniline (27 μ L, 0.30 mmol, 3.0 equiv), (*S*)-TRIP catalyst (15.1 mg, 0.02 mmol, 20 mol%), and CPME (200 μ L, 0.5 M) at rt for 3 d. The reaction mixture was subjected to the purification outlined in General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale yellow solid (17.9 mg, 68%).

ν_{\max} (film): 2361, 1601, 1509, 1323, 1250, 1069, 752 cm^{-1} .

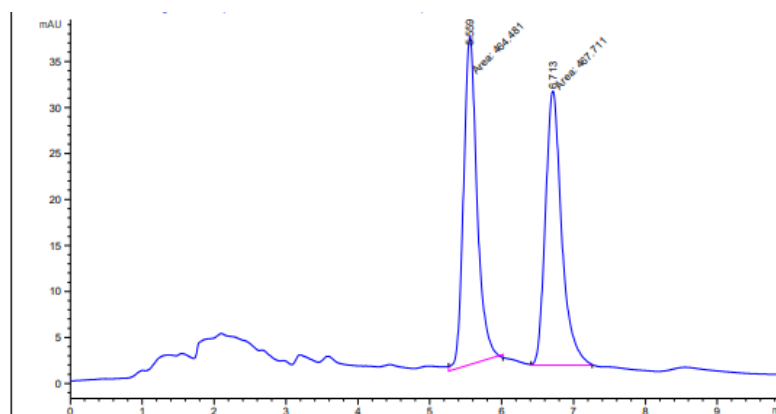
^1H NMR (400 MHz, CDCl_3): δ 8.96 (d, $J = 4.4$ Hz, 1H), 8.20 (dt, $J = 8.4, 1.1$ Hz, 1H), 7.92 (dt, $J = 8.5, 0.9$ Hz, 1H), 7.77 (ddd, $J = 8.4, 6.9, 1.4$ Hz, 1H), 7.62 (ddd, $J = 8.3, 6.9, 1.3$ Hz, 1H), 7.55 (dd, $J = 4.5, 0.8$ Hz, 1H), 7.31 – 7.12 (m, 2H), 6.79 (tt, $J = 7.3, 1.1$ Hz, 1H), 6.76 – 6.65 (m, 2H), 6.50 – 6.25 (m, 1H), 4.17 (s, 1H), 3.81 (ddd, $^3J_{\text{HF}} = 29.9$ Hz, $^2J_{\text{HH}} = 14.4$ Hz, $J = 3.0$ Hz, 1H), 3.61 (ddd, $^3J_{\text{HF}} = 18.3$ Hz, $^2J_{\text{HH}} = 14.4$ Hz, $J = 8.1$ Hz, 1H).

^{13}C NMR (101 MHz, CDCl_3): δ 150.2, 148.2, 147.1, 143.0 (d, $^2J = 19.2$ Hz), 130.7, 129.60, 129.53, 127.29, 122.41, 118.64, 117.68, 117.57, 113.45, 89.90 (d, $^1J = 175.4$ Hz), 49.54 (d, $^2J = 23.6$ Hz).

^{19}F NMR (377 MHz, CDCl_3): δ -190.24.

HRMS(ESI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{17}\text{H}_{16}\text{FN}_2$) requires m/z 267.1292, found m/z 267.1285.

The enantiomeric purity of the product was determined by HPLC analysis: 50:50 e.r. (Chiralpak IA, hexane/*i*-PrOH = 50/50, flow rate 0.5 mL/min, $\lambda = 250$ nm), t_r (major) = 5.7 min, t_r (minor) = 6.9 min;

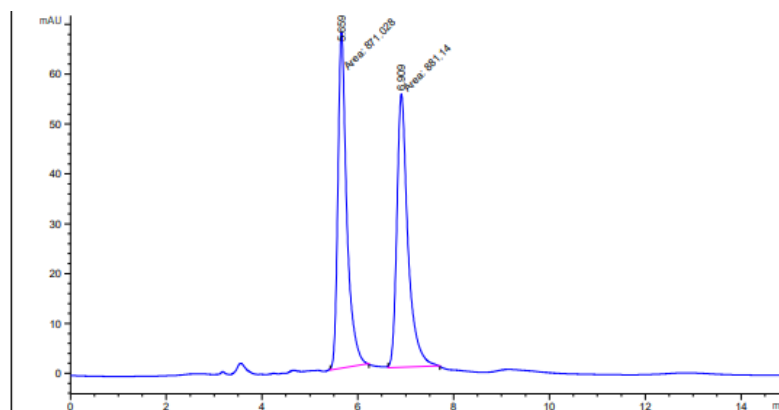


Area Percent Report

Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	5.559	MM	0.2170	464.48099	35.67741	49.8268
2	6.713	MM	0.2617	467.71078	29.78292	50.1732



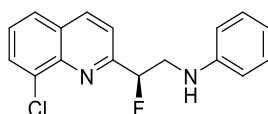
Area Percent Report

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Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=250 nm

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	5.659	MM	0.2156	871.02771	67.33373	49.7114
2	6.909	MM	0.2678	881.13983	54.83912	50.2886

(*R*)-*N*-(2-(8-Chloroquinolin-2-yl)-2-fluoroethyl)aniline **7w**



Prepared according to General Procedure B using 2-(1-fluorovinyl)-8-chloroquinoline (10.4 mg, 0.05 mmol, 1.0 equiv), aniline (14 μ L, 0.15 mmol, 3.0 equiv), (*S*)-TRIP catalyst (7.5 mg, 0.01 mmol, 20 mol%), and CPME (100 μ L, 0.5 M) at 40 $^{\circ}$ C. The reaction mixture was subjected to the purification outlined in General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale brown oil (9.5 mg, 50%).

ν_{\max} (film): 1601, 1501, 841, 752, 692, 669 cm^{-1} .

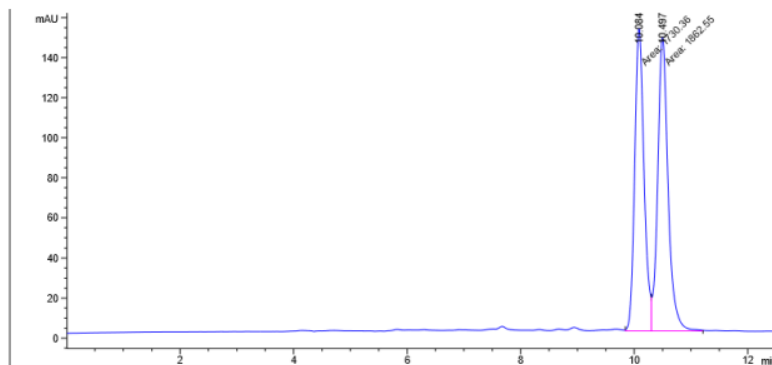
$^1\text{H NMR}$ (400 MHz, CDCl_3): δ 8.25 (d, $J = 8.5$ Hz, 1H), 7.87 (dd, $J = 7.5, 1.3$ Hz, 1H), 7.75 (ddd, $J = 16.5, 8.4, 1.6$ Hz, 2H), 7.49 (dd, $J = 8.2, 7.5$ Hz, 1H), 7.24 – 7.15 (m, 2H), 6.81 – 6.69 (m, 3H), 5.95 (ddd, $^2J_{\text{HF}} = 48.1$ Hz, $J = 6.2, 5.0$ Hz, 1H), 4.71 (s, 1H), 4.03 – 3.77 (m, 2H).

$^{13}\text{C NMR}$ (101 MHz, CDCl_3): δ 159.4 (d, $^2J = 25.7$ Hz), 147.8, 143.6, 137.7, 133.7, 130.1, 129.5, 129.0, 126.9, 126.9, 118.7 (d, $^3J = 6.9$ Hz), 118.1, 113.5, 92.7 (d, $^1J = 175.2$ Hz), 48.4 (d, $^2J = 23.1$ Hz).

$^{19}\text{F NMR}$ (376 MHz, CDCl_3): δ -191.99.

HRMS (ESI): exact mass calculated for $[\text{M}+\text{H}]^+$ ($\text{C}_{17}\text{H}_{15}\text{N}_2\text{ClF}$) requires m/z 301.0902, found m/z 301.0898.

The enantiomeric purity of the product was determined by HPLC analysis: 51:49 e.r. (Chiralpak IA, hexane/*i*-PrOH = 50/50, flow rate 0.5 mL/min, $\lambda = 250$ nm), t_r (major) = 10.5 min, t_r (minor) = 10.1 min;

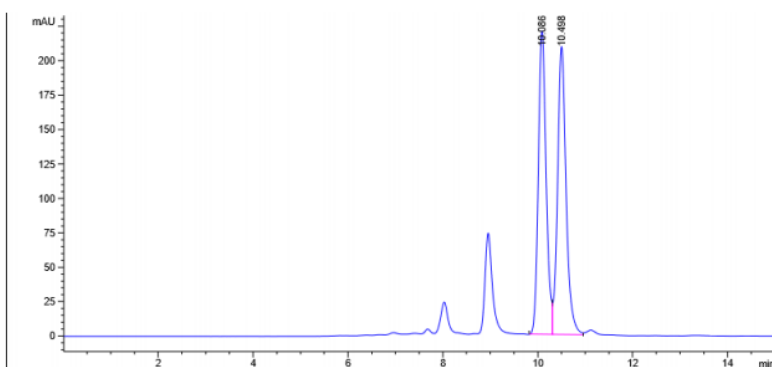


Area Percent Report

Sorted By : Signal
 Multiplier: : 1.0000
 Dilution: : 1.0000
 Use Multiplier & Dilution Factor with ISTDs

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Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
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2	10.497	FM	0.2118	1862.54712	146.57484	51.8396



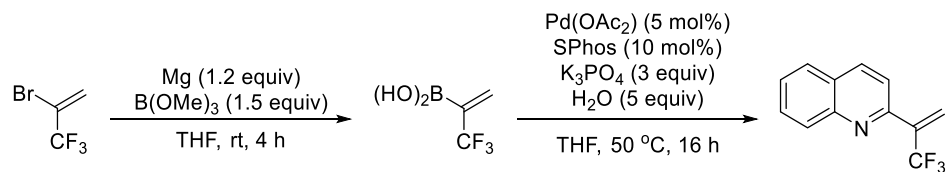
Area Percent Report

Sorted By : Signal
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 Dilution: : 1.0000
 Use Multiplier & Dilution Factor with ISTDs

Signal 1: DAD1 A, Sig=254,4 Ref=360,100

Peak #	RetTime [min]	Type	Width [min]	Area [mAU*s]	Height [mAU]	Area %
1	10.086	BV	0.1753	2519.43018	220.56381	49.0498
2	10.498	VV	0.1898	2617.04395	209.41792	50.9502

2-(3,3,3-Trifluoroprop-1-en-2-yl)quinoline **10**



This compound was prepared according to an adapted literature procedure.⁸

An oven-dried round 100 mL round bottom flask was charged with Mg turnings (262 mg, 10.8 mmol, 1.2 equiv), capped and purged with N₂. Then THF (18 mL, 0.5 M) and freshly distilled trimethylborate (1.50 mL, 13.5 mmol, 1.5 equiv) were added sequentially and the mixture was allowed to stir at rt for 5 min. 2-Bromo-1,1,1-trifluoropropene (0.96 mL, 9.00 mmol, 1.0 equiv) was then added dropwise over 5 min, then the reaction was stirred for 4 h at rt. The reaction was quenched using 6 M HCl (50 mL) and the solution was left to stir for 5 min. The organic layer was then separated and the aqueous layer was extracted with Et₂O (3 × 20 mL). The combined organic extracts were washed with brine (50 mL), dried over NaSO₄, filtered, and concentrated *in vacuo*. Formation of the boronic acid was confirmed *via* ¹¹B NMR, then the crude was used directly in the next step.

NB. Calculations based upon quantitative conversion to boronic acid in the previous step. An oven-dried 2-necked 100 mL round bottomed flask equipped with a reflux condenser was charged with Pd(OAc)₂ (50.5 mg, 0.225 mmol, 5 mol%), SPhos (185 mg, 0.45 mmol, 10 mol%), 2-bromoquinoline (936 mg, 4.50 mmol, 1.0 equiv), the crude boronic acid (9.00 mmol, 2.0 equiv), and K₃PO₄ (2.87 g, 13.5 mmol, 3.00 equiv). The vessel was capped and purged with N₂. THF (22.5 mL, 0.2 M) and water (405 μL, 22.5 mmol, 5.0 equiv) were then added to the mixture sequentially. The reaction was then heated to 50 °C for 16 h. The reaction mixture was then cooled to rt, filtered through Celite, eluting with EtOAc and concentrated *in vacuo*. Purification of the crude *via* flash chromatography (silica gel, EtOAc:petroleum ether, 10:90) gave the desired product as a pale yellow oil (102 mg, 9%).

NB. Due to the reactive nature of this compound, an analytically pure sample was not obtained, data reported as observed. The compound was used directly in the next step.

ν_{max} (film): 2361, 2342, 1504, 1323, 1196, 1126, 1096, 1069, 907, 833, 729, 648 cm⁻¹.

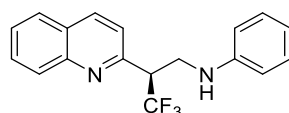
¹H NMR (400 MHz, CDCl₃): δ 8.20 (d, *J* = 8.7 Hz, 1H), 8.11 (d, *J* = 8.4 Hz, 1H), 7.83 (d, *J* = 8.1 Hz, 1H), 7.74 (ddd, *J* = 8.4, 6.9, 1.4 Hz, 1H), 7.65 – 7.60 (m, 1H), 7.56 (ddd, *J* = 8.1, 6.9, 1.1 Hz, 1H), 6.68 (q, ⁴*J*_{HF} = 1.8 Hz, 1H), 6.25 (q, ⁴*J*_{HF} = 1.4 Hz, 1H).

¹³C NMR (126 MHz, CDCl₃): δ 151.0, 148.0, 138.5 (q, ²*J* = 27.7 Hz), 137.0, 130.2, 130.0, 127.8, 127.6, 127.3, 124.5 (q, ¹*J* = 273.4 Hz), 123.3 (q, ³*J* = 6.3 Hz), 119.2.

¹⁹F NMR (470 MHz, CDCl₃): δ -63.61.

HRMS (ESI): exact mass calculated for [M+H]⁺ (C₁₂H₉F₃N) requires *m/z* 224.0682, found *m/z* 224.0677.

(*R*)-*N*-(3,3,3-Trifluoro-2-(quinolin-2-yl)propyl)aniline **11**



Prepared according to General Procedure B using 2-(3,3,3-trifluoroprop-1-en-2-yl)quinoline (11.2 mg, 0.05 mmol, 1.0 equiv), aniline (14 μL, 0.15 mmol, 3.0 equiv), (*S*)-TRIP catalyst (37.6 mg, 0.05 mmol, 1.0 equiv), and CPME (200 μL, 0.25 M) at -50 °C for 3 d. The reaction mixture was subjected to the purification outlined in General Procedure B (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) to afford the desired product as a pale yellow solid (11.5 mg, 73%).

ν_{max} (film): 2361, 2342, 1601, 1504, 1315, 1254, 1165, 1111, 826, 752, 694 cm⁻¹.

¹H NMR (500 MHz, CDCl₃): δ 8.18 – 8.14 (m, 2H), 7.85 (dd, *J* = 8.2, 1.4 Hz, 1H), 7.77 (ddd, *J* = 8.4, 6.9, 1.5 Hz, 1H), 7.59 (ddd, *J* = 8.1, 6.9, 1.2 Hz, 1H), 7.40 (d, *J* = 8.4 Hz, 1H), 7.20 – 7.14 (m, 2H), 6.73 (tt, *J* = 7.3, 1.1 Hz, 1H), 6.59 (dt, *J* = 7.7, 1.0 Hz, 2H), 4.20 – 3.82 (m, 4H).

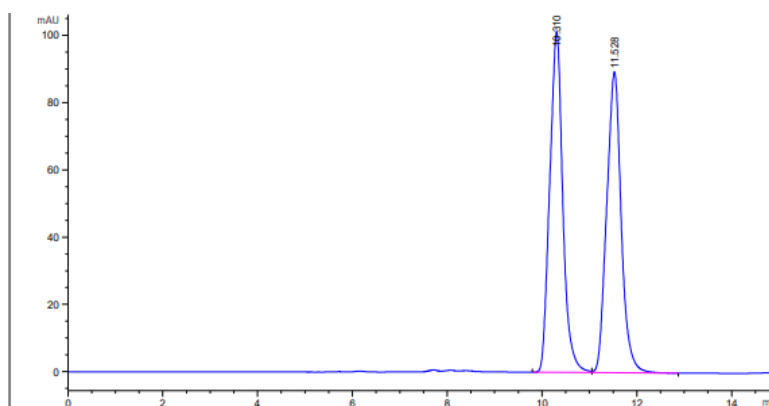
¹³C NMR (126 MHz, CDCl₃): δ 153.4, 148.0, 147.0, 137.1, 130.1, 129.7, 129.5, 127.7, 127.7, 127.2, 126.1 (q, ¹*J* = 280.5 Hz), 122.1, 118.2, 113.2, 51.1 (q, ²*J* = 25.0 Hz), 41.9.

¹⁹F NMR (470 MHz, CDCl₃): δ -67.02.

HRMS (ESI): exact mass calculated for [M+H]⁺ (C₁₈H₁₆F₃N₂) requires *m/z* 317.1260, found *m/z* 317.1251.

The enantiomeric purity of the product was determined by HPLC analysis: 90:10 e.r. (Chiralpak IA, hexane/*i*-PrOH = 50/50, flow rate 1.0 mL/min, λ = 250 nm), *t_r* (major) = 9.1 min, *t_r* (minor) = 11.6 min;

[α]_D²⁰ = -80.4 (*c* 1.00, CHCl₃).

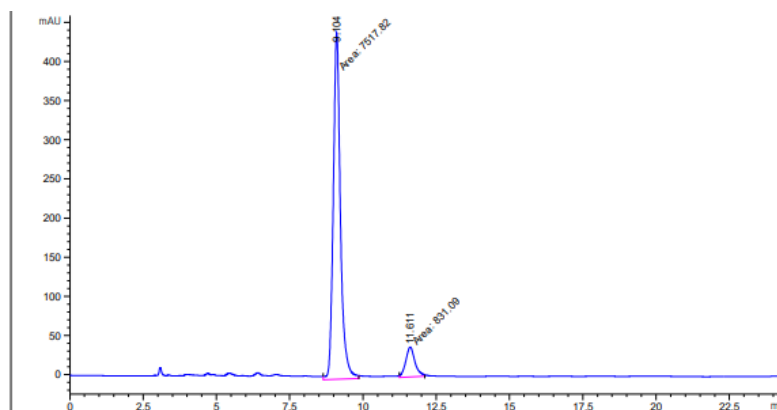


Area Percent Report

Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

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2	11.528	VB	0.3160	1898.84473	89.48219	49.4468



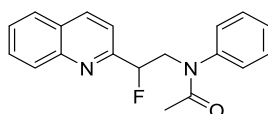
Area Percent Report

Sorted By : Signal
Multiplier : 1.0000
Dilution : 1.0000
Use Multiplier & Dilution Factor with ISTDs

Signal 1: VWD1 A, Wavelength=210 nm

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1	9.104	MM	0.2829	7517.82324	442.87985	90.0455
2	11.611	MM	0.3643	831.08960	38.02613	9.9545

N-(2-Fluoro-2-(quinolin-2-yl)ethyl)-*N*-phenylacetamide **16**



Prepared according to literature procedure.⁹

A 10 mL microwave vial was charged with *N*-(2-fluoro-2-(quinolin-2-yl)ethyl)aniline (151 mg, 0.57 mmol, 1.0 equiv). The vial was capped and purged with N₂. Dry CH₂Cl₂ (1 mL) and Ac₂O (54 μL, 0.57 mmol, 1.0 equiv) were added to the vessel and the reaction mixture was allowed to stir at rt for 16 h. The reaction was then quenched with 2 M HCl (1 mL) and the aqueous phase was extracted with CH₂Cl₂ (2 × 5 mL). The combined organic phases were dried over Na₂SO₄, filtered, and concentrated *in vacuo*. Purification of the crude *via* flash chromatography (silica gel, EtOAc:petroleum ether, 5:95 to 10:90) gave the desired product as a pale yellow oil (51.0 mg, 29%).

ν_{max} (film): 1659, 1597, 1497, 1427, 1389, 1300, 1281, 1072, 1026, 837, 756, 702 cm⁻¹.

¹H NMR (500 MHz, CDCl₃): δ 8.21 (d, *J* = 8.5 Hz, 1H), 8.03 (d, *J* = 8.5 Hz, 1H), 7.82 (d, *J* = 8.1 Hz, 1H), 7.71 (t, *J* = 7.7 Hz, 1H), 7.63 (d, *J* = 8.2 Hz, 1H), 7.55 (t, *J* = 7.5 Hz, 1H), 7.43 (t, *J* = 7.5 Hz, 2H), 7.35 (dd, *J* = 23.7, 7.4 Hz, 3H), 5.99 – 5.84 (m, 1H), 4.42 – 4.26 (m, 2H), 1.90 (s, 3H).

¹³C NMR (126 MHz, CDCl₃): δ 171.2, 157.4 (d, ²*J* = 22.2 Hz), 147.5, 143.2, 137.2, 130.0, 129.9, 129.4, 128.6, 128.3, 127.9, 127.8, 127.0, 118.3 (d, ³*J* = 5.6 Hz), 92.0 (d, ¹*J* = 176.9 Hz), 53.2 (d, ²*J* = 22.7 Hz), 23.0.

¹⁹F NMR (470 MHz, CDCl₃): δ -189.42.

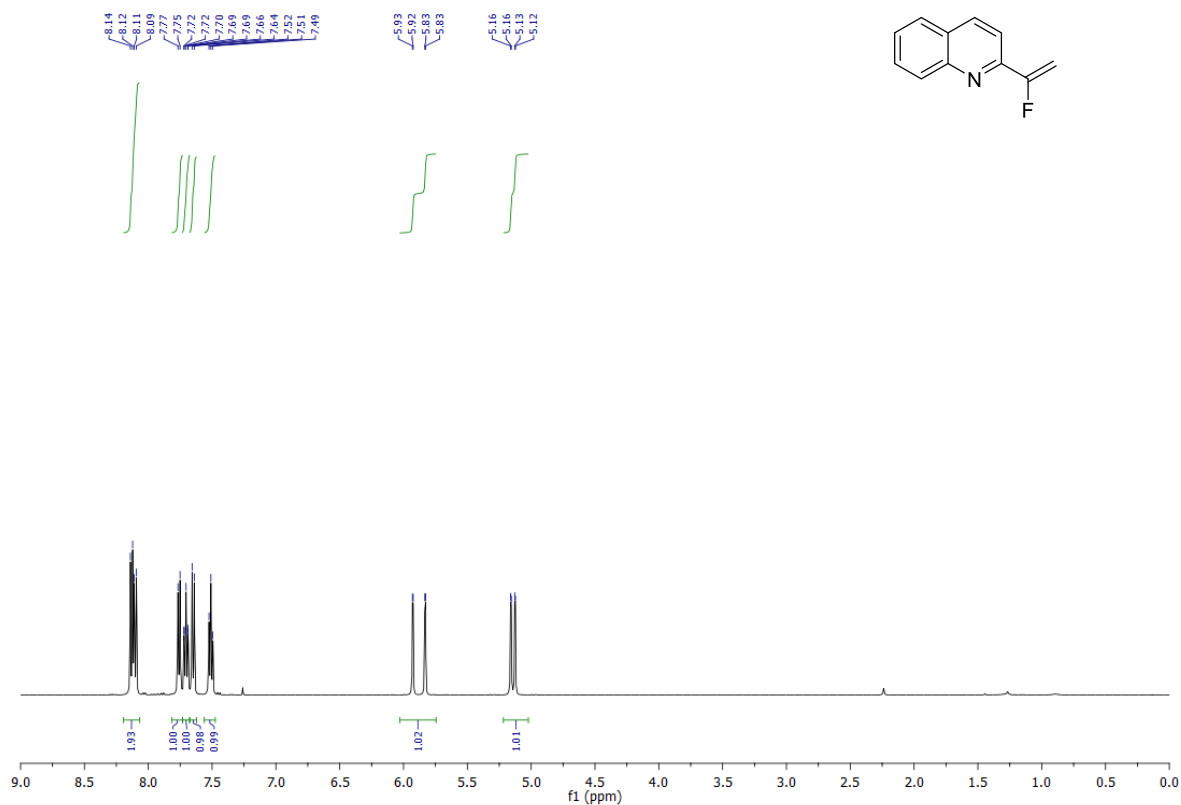
HRMS (ESI): exact mass calculated for [M+H]⁺ (C₁₉H₁₈FN₂O) requires *m/z* 309.1398, found *m/z* 309.1388.

8. References

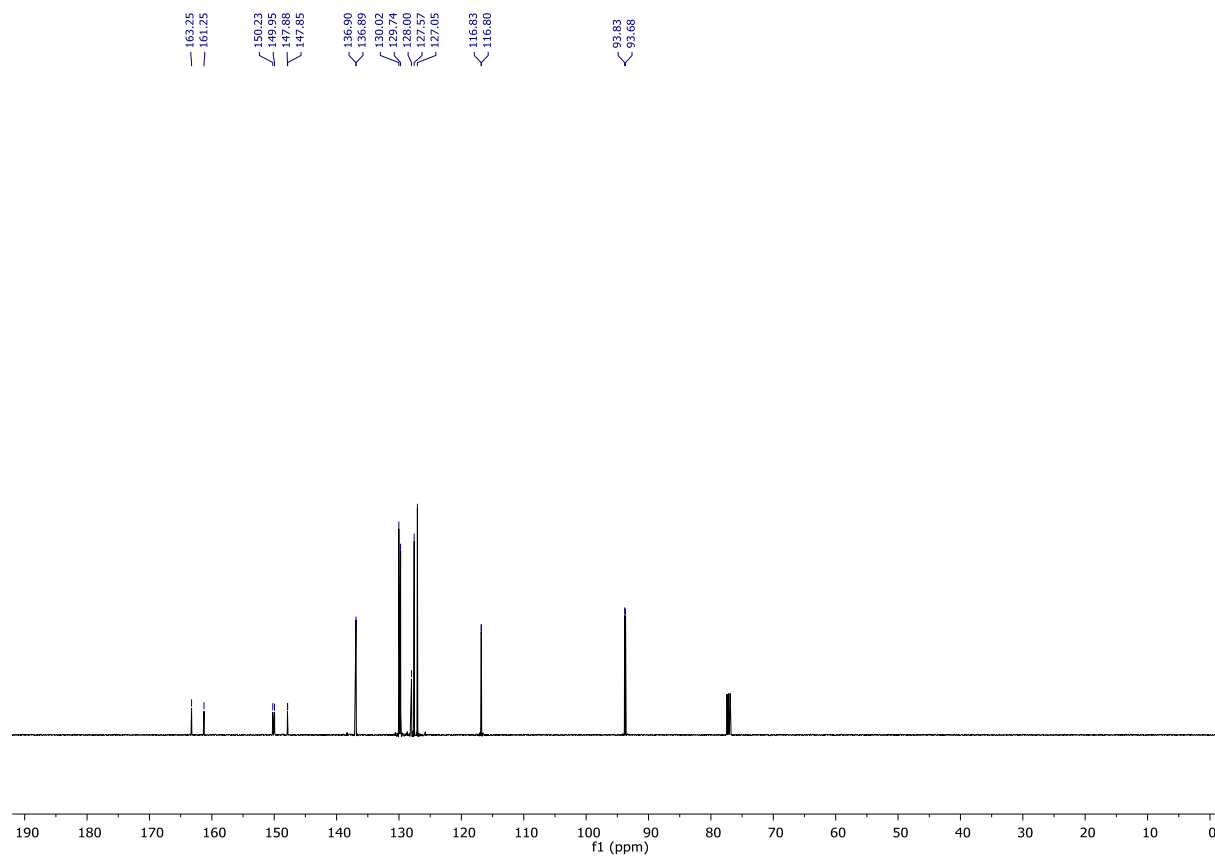
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9. NMR Spectra

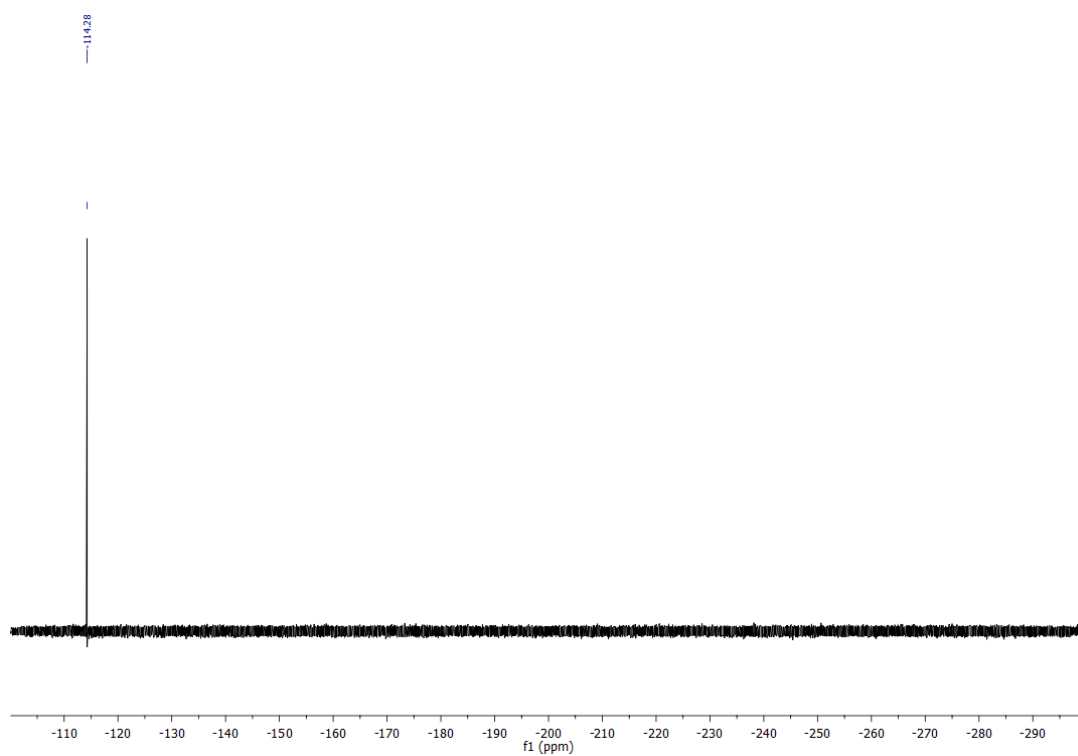
¹H NMR of 5a



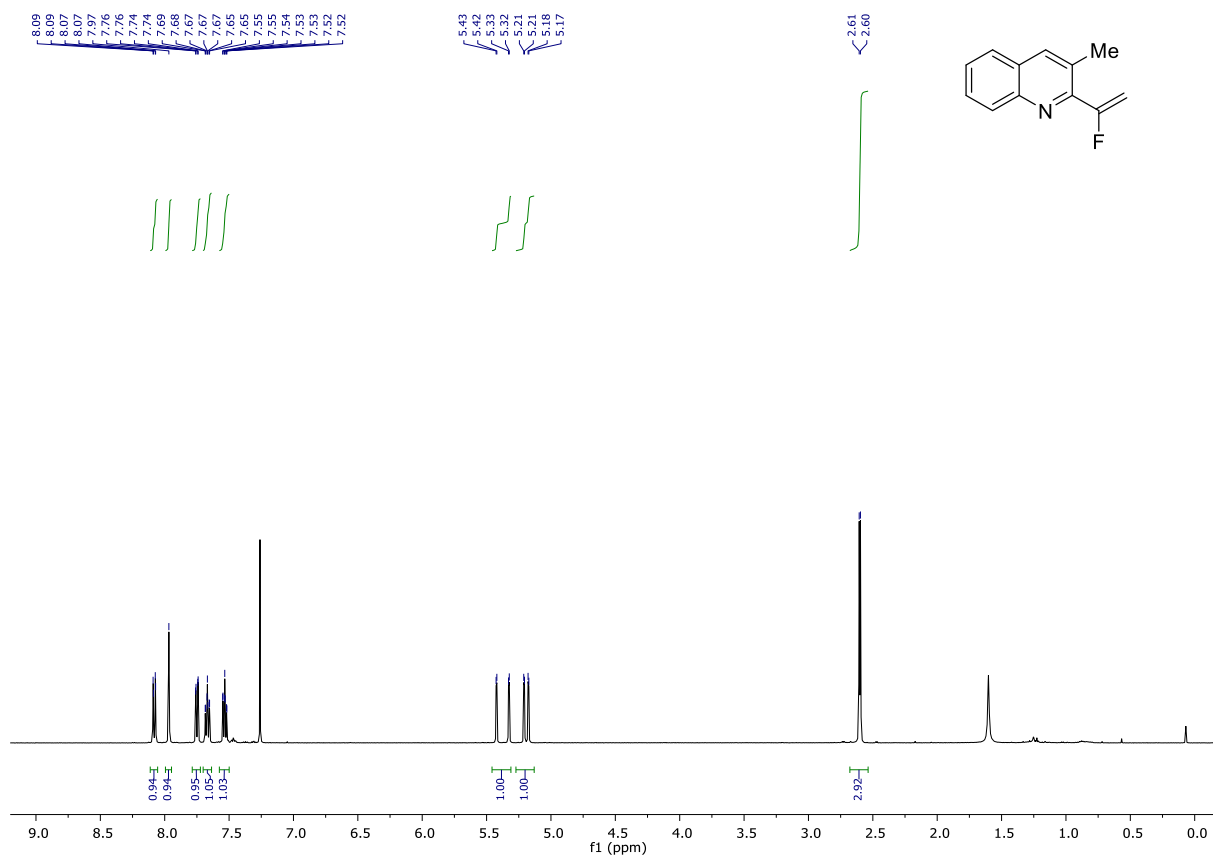
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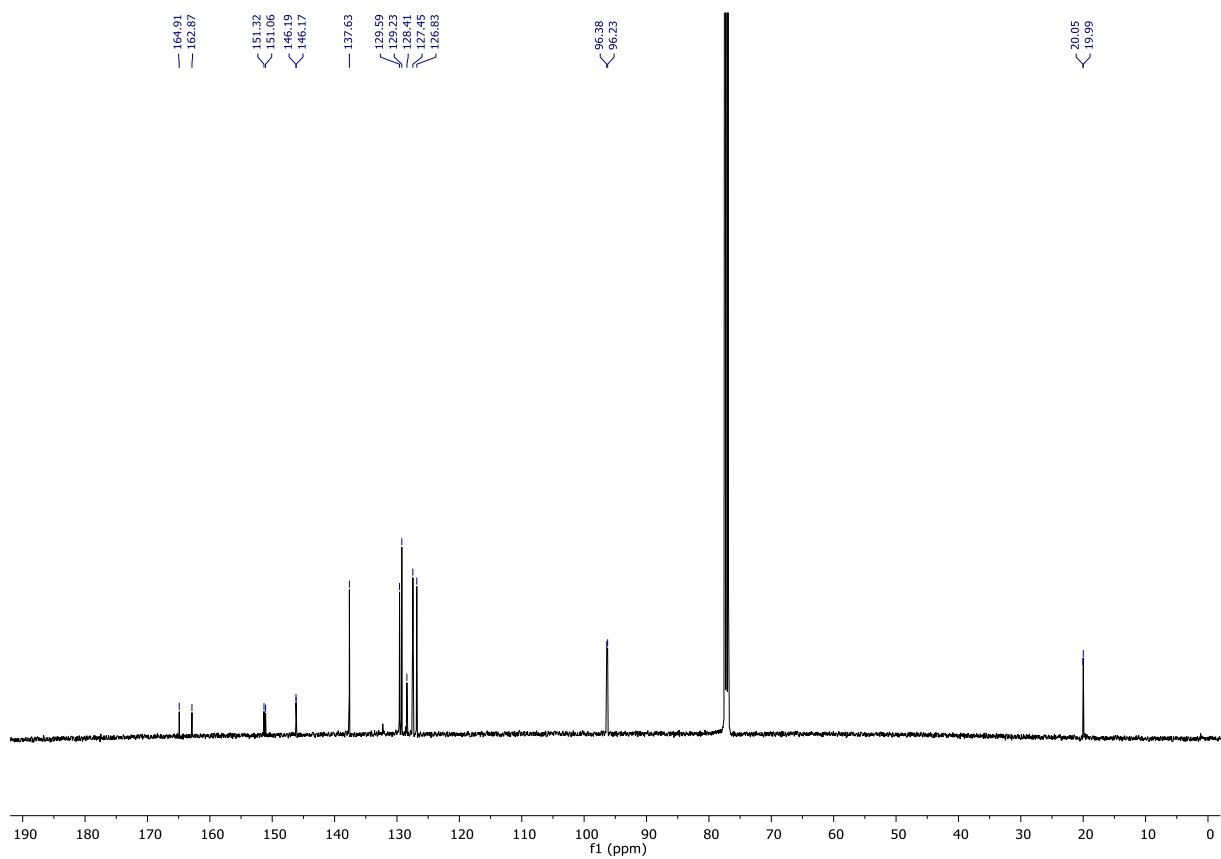
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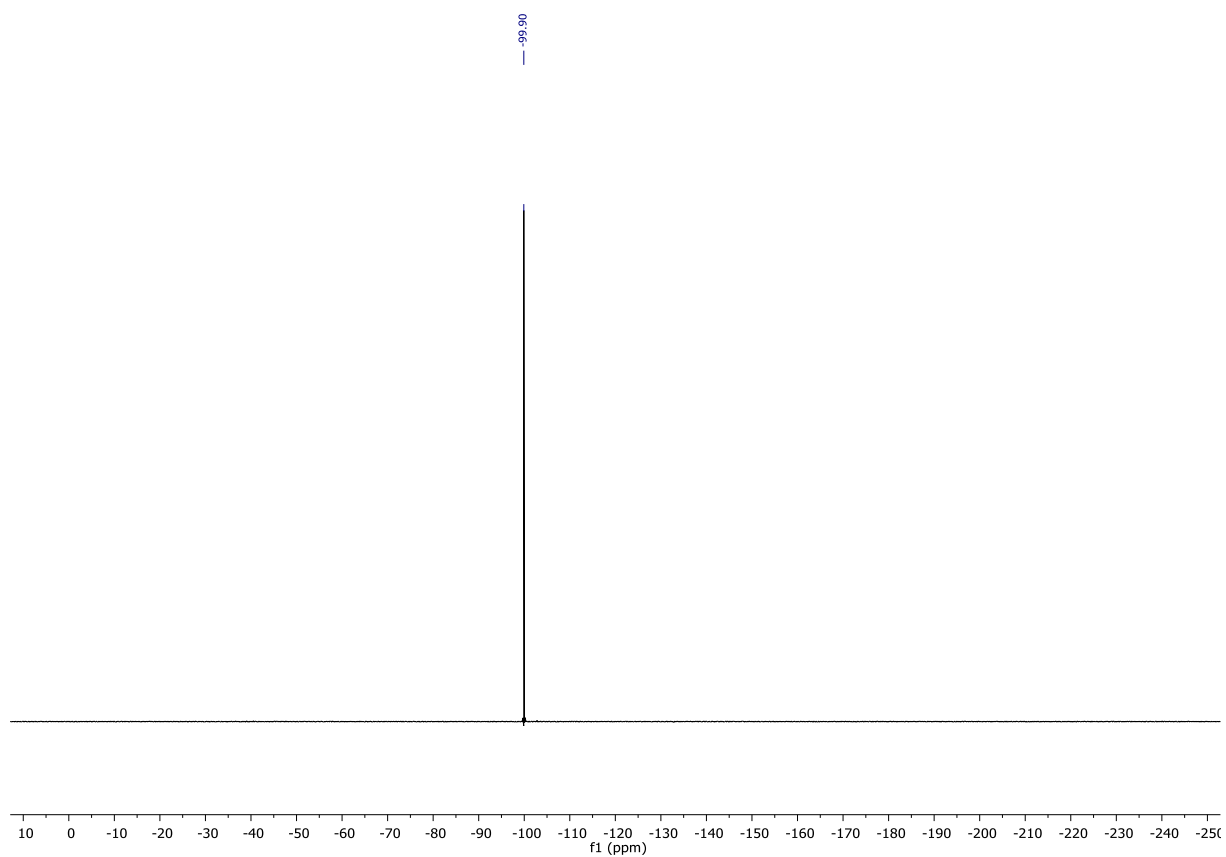
¹H NMR of 5b



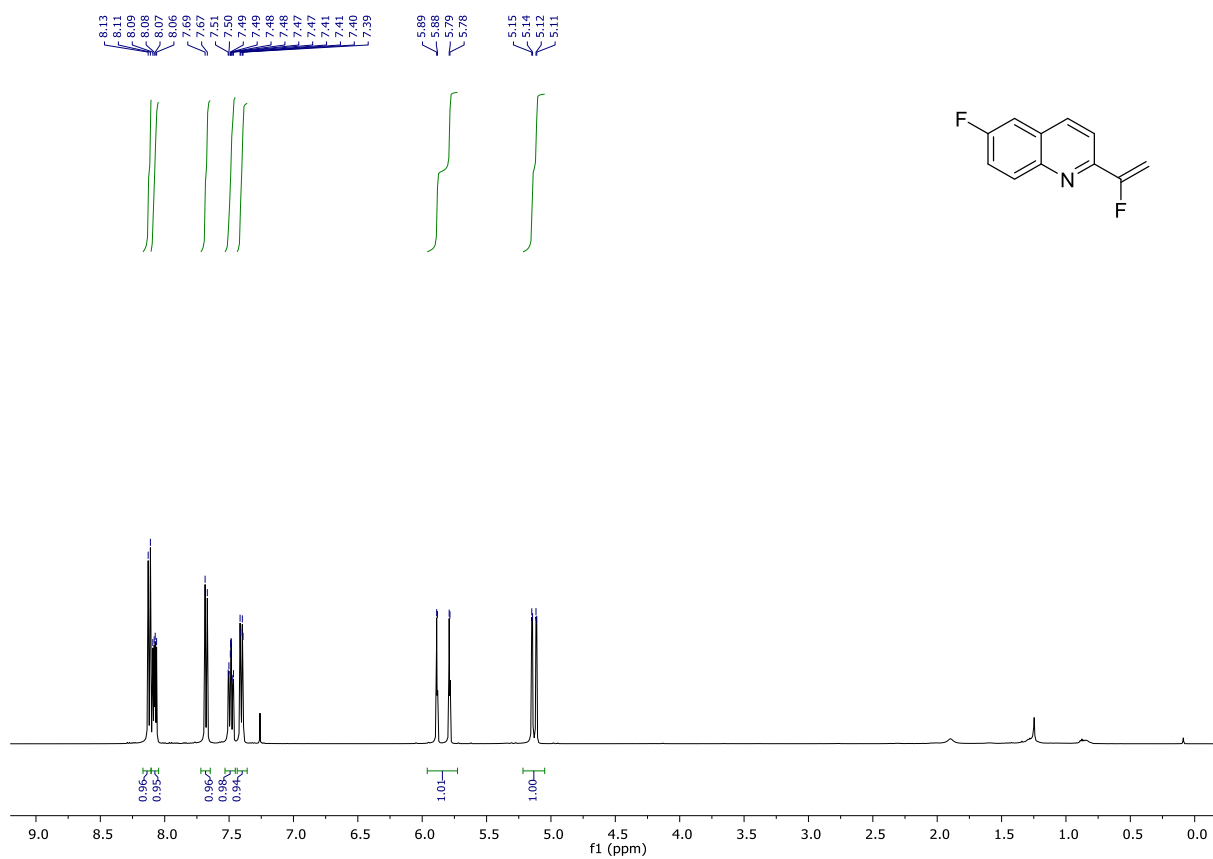
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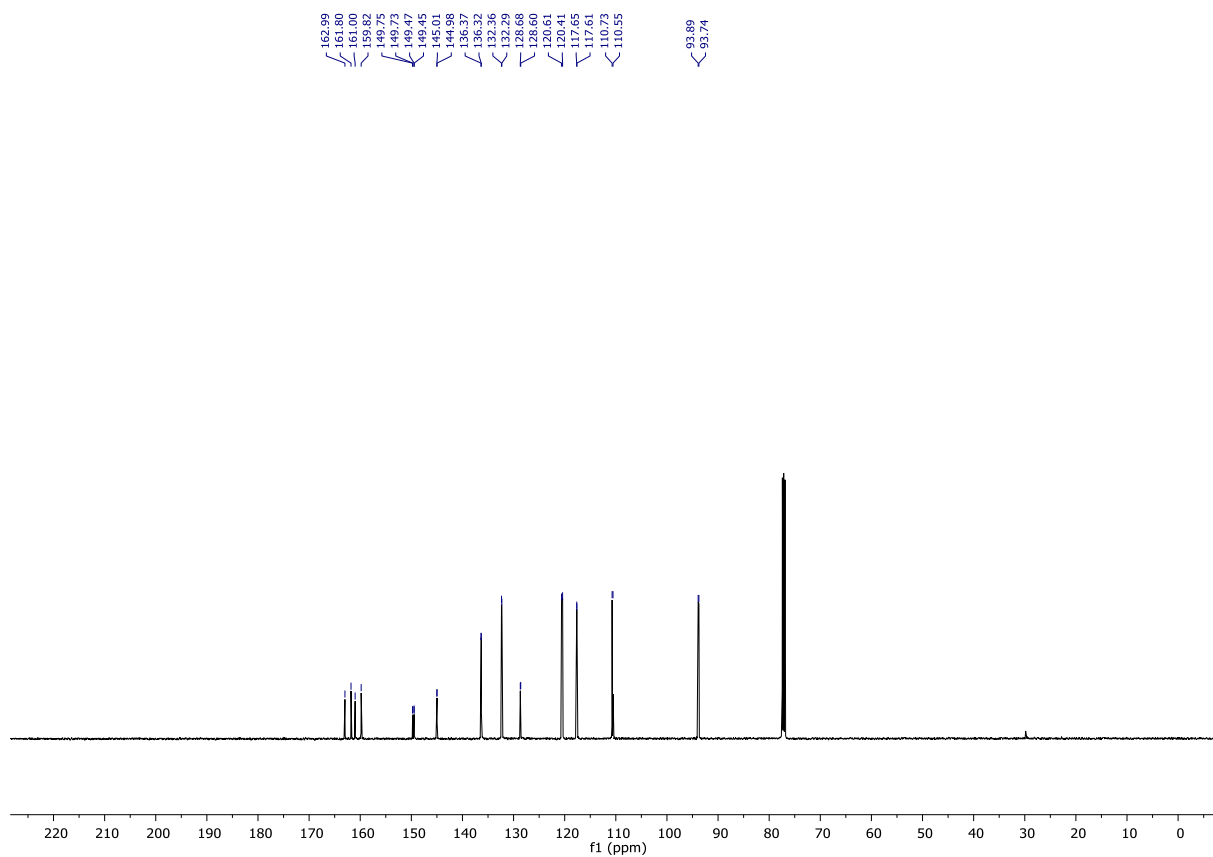
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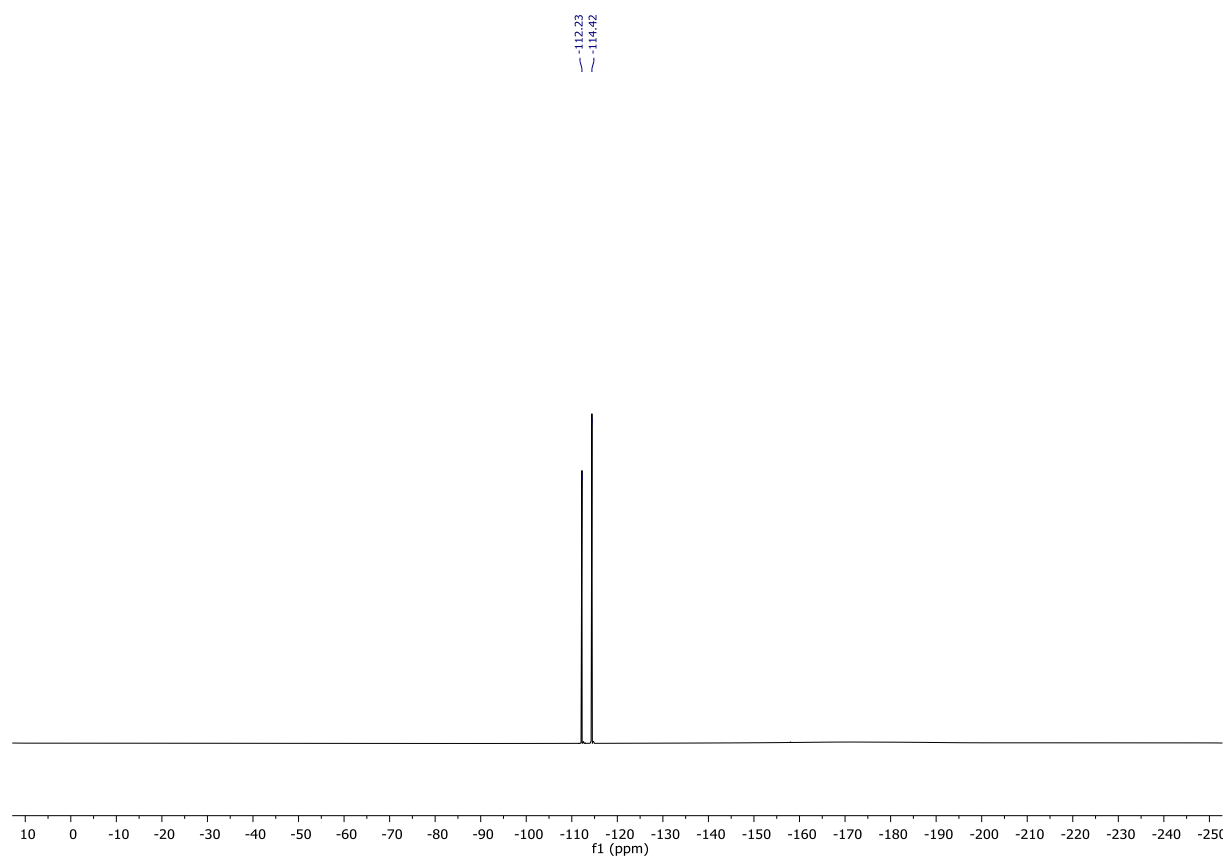
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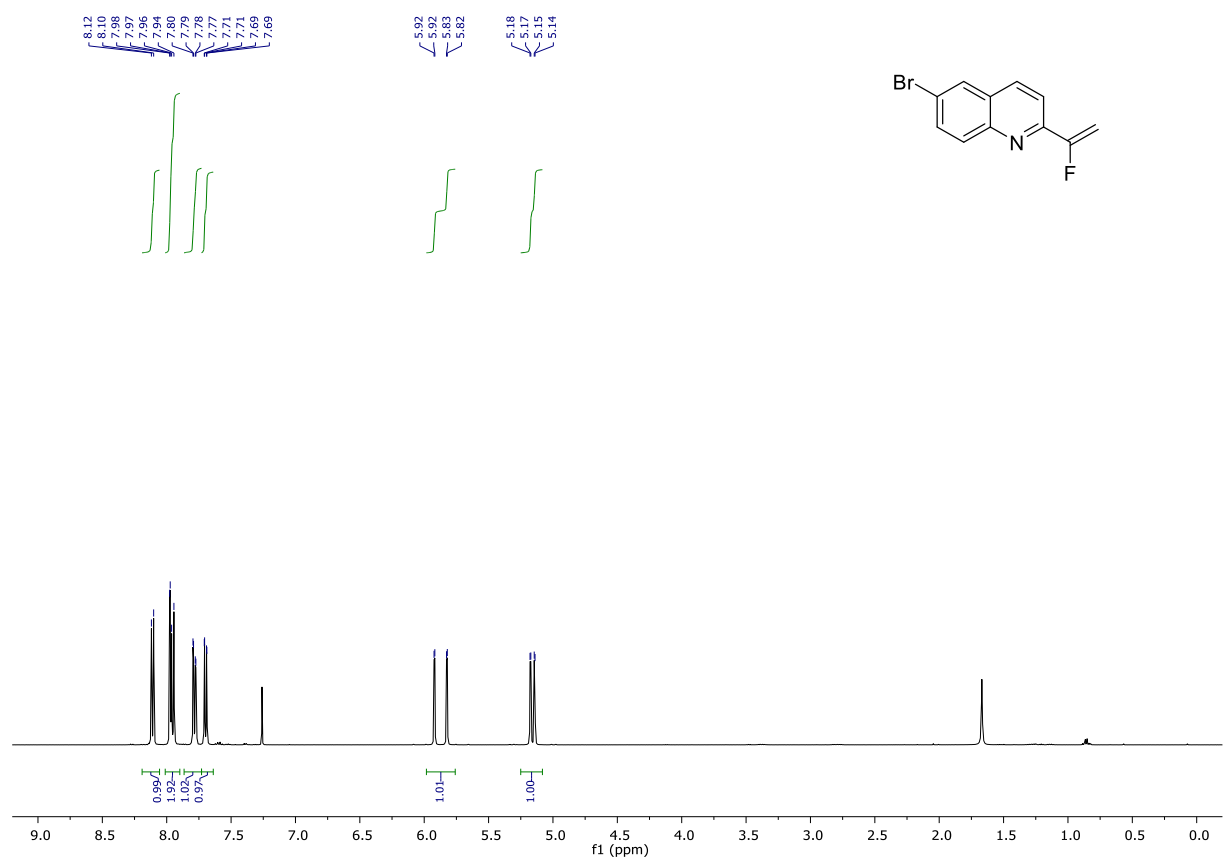
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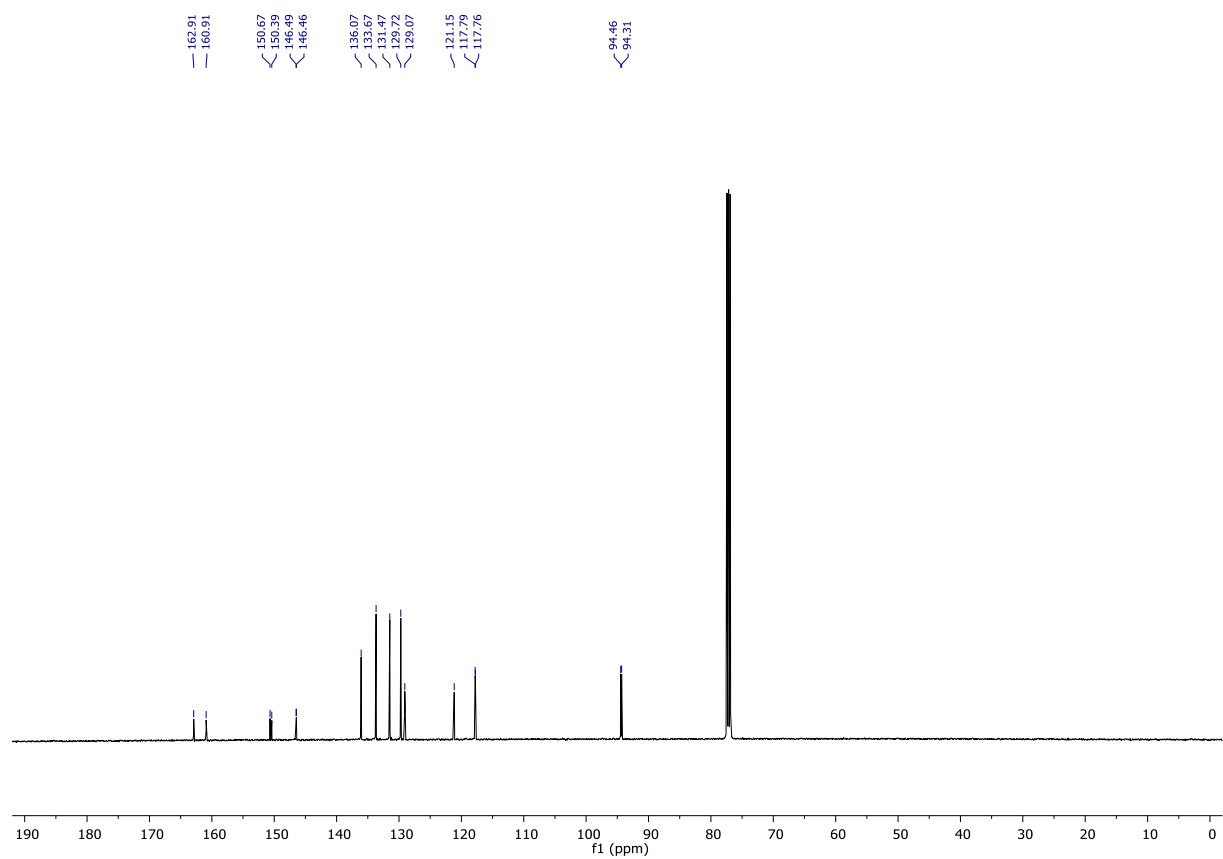
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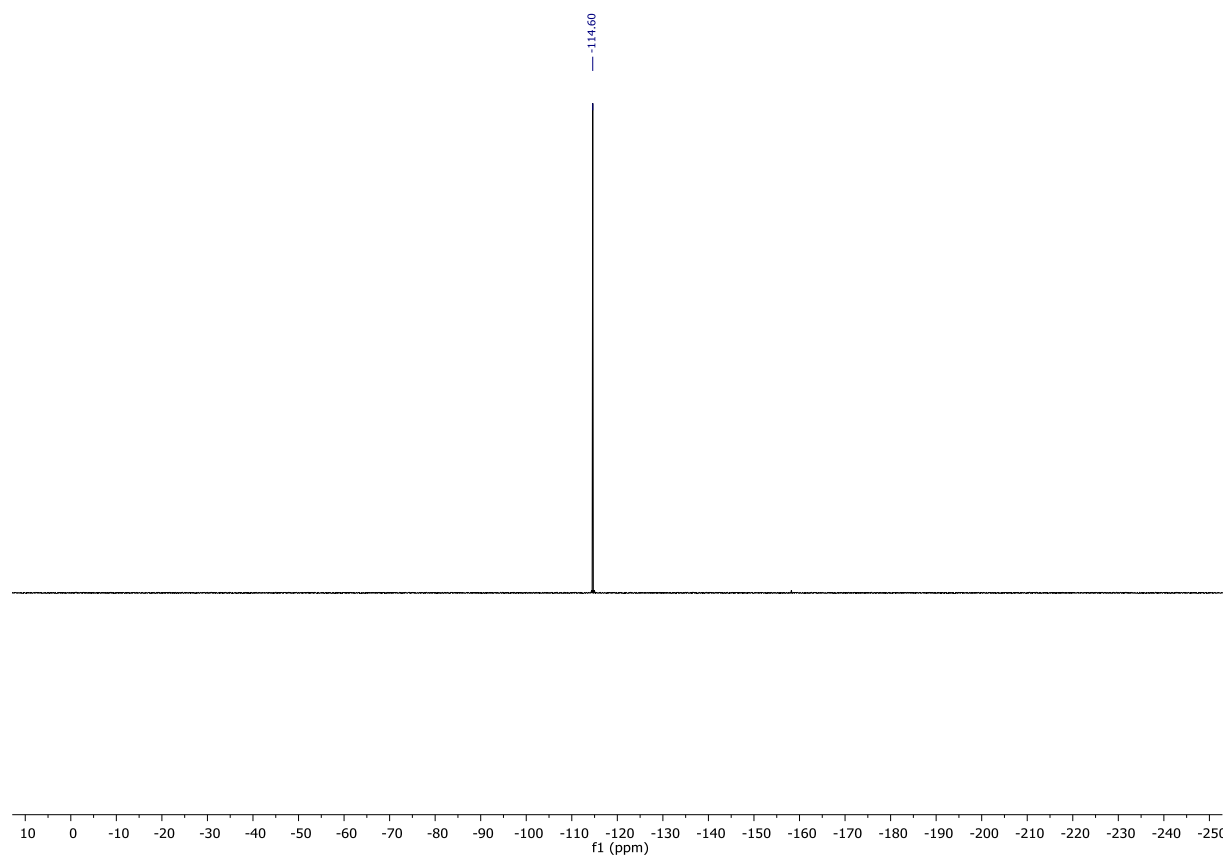
^1H NMR of 5d



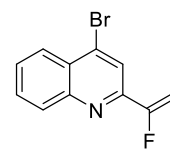
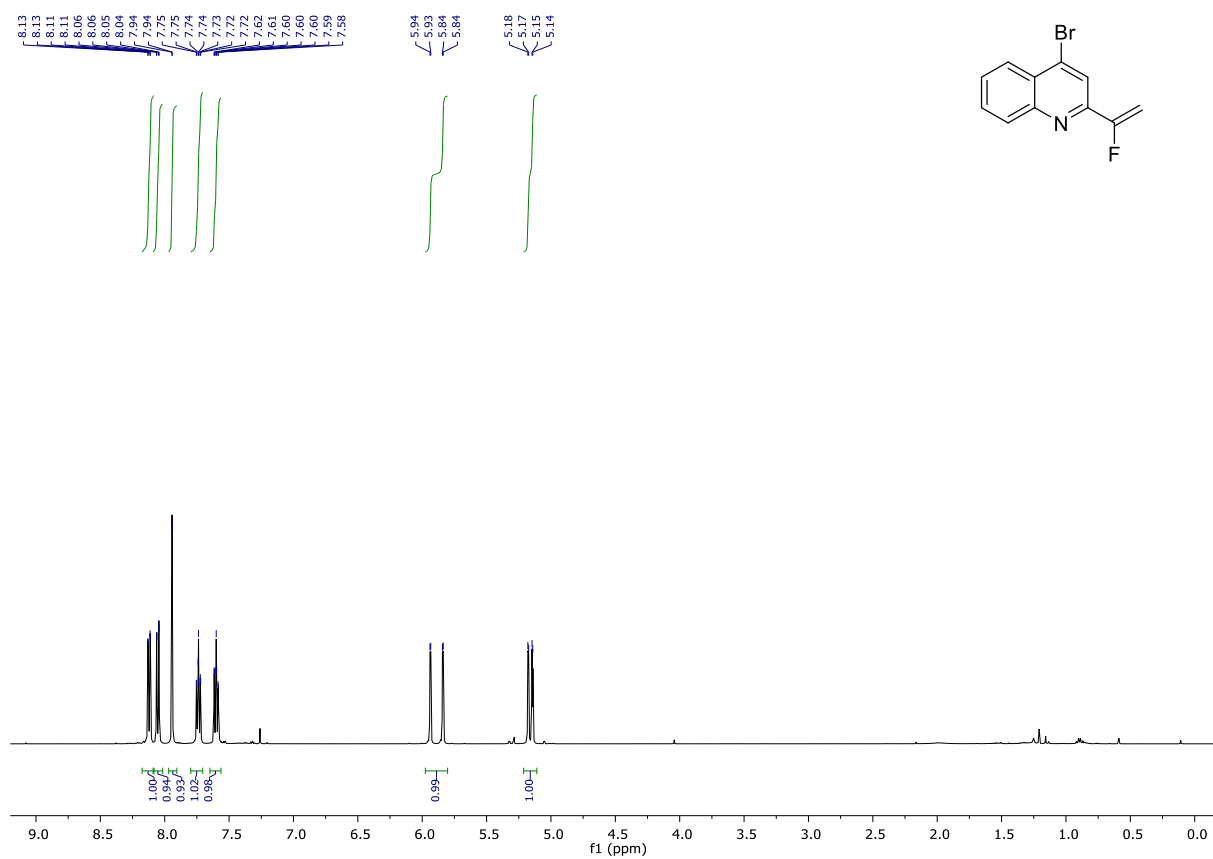
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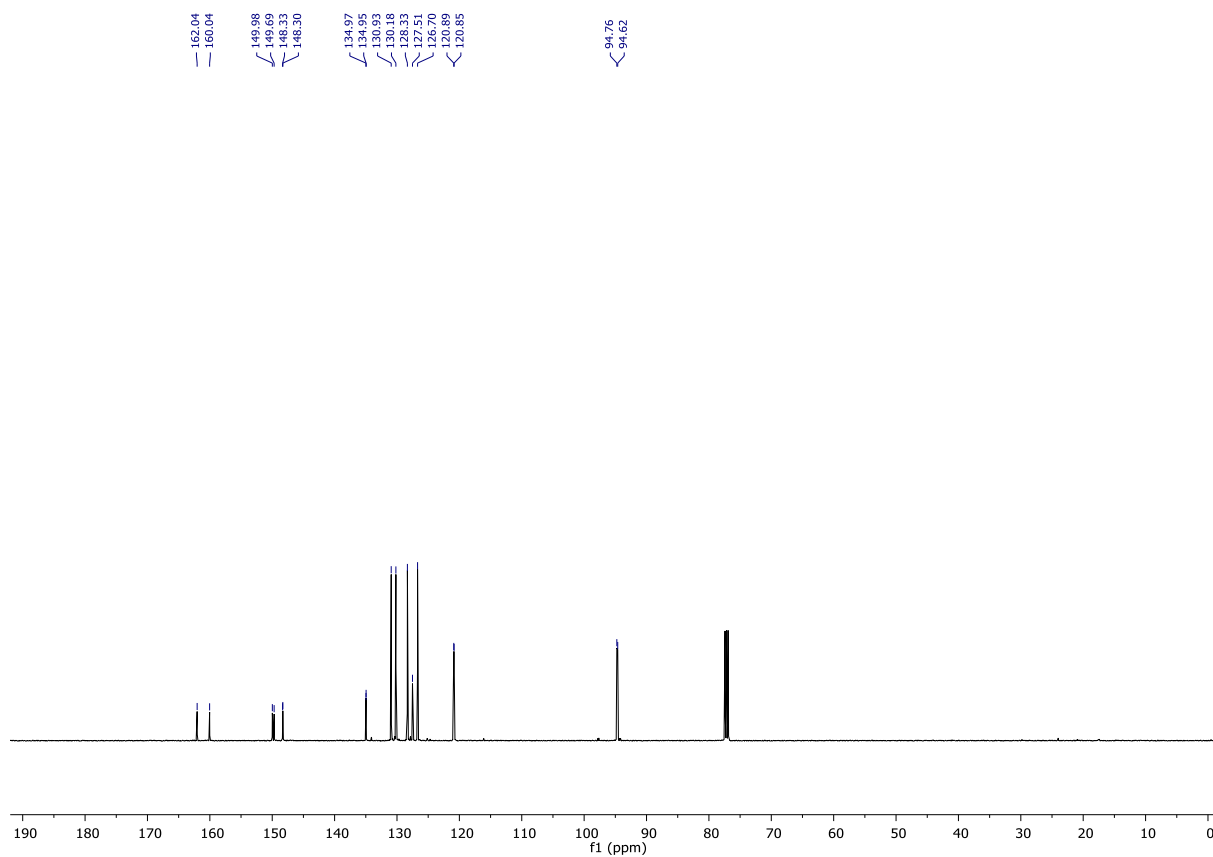
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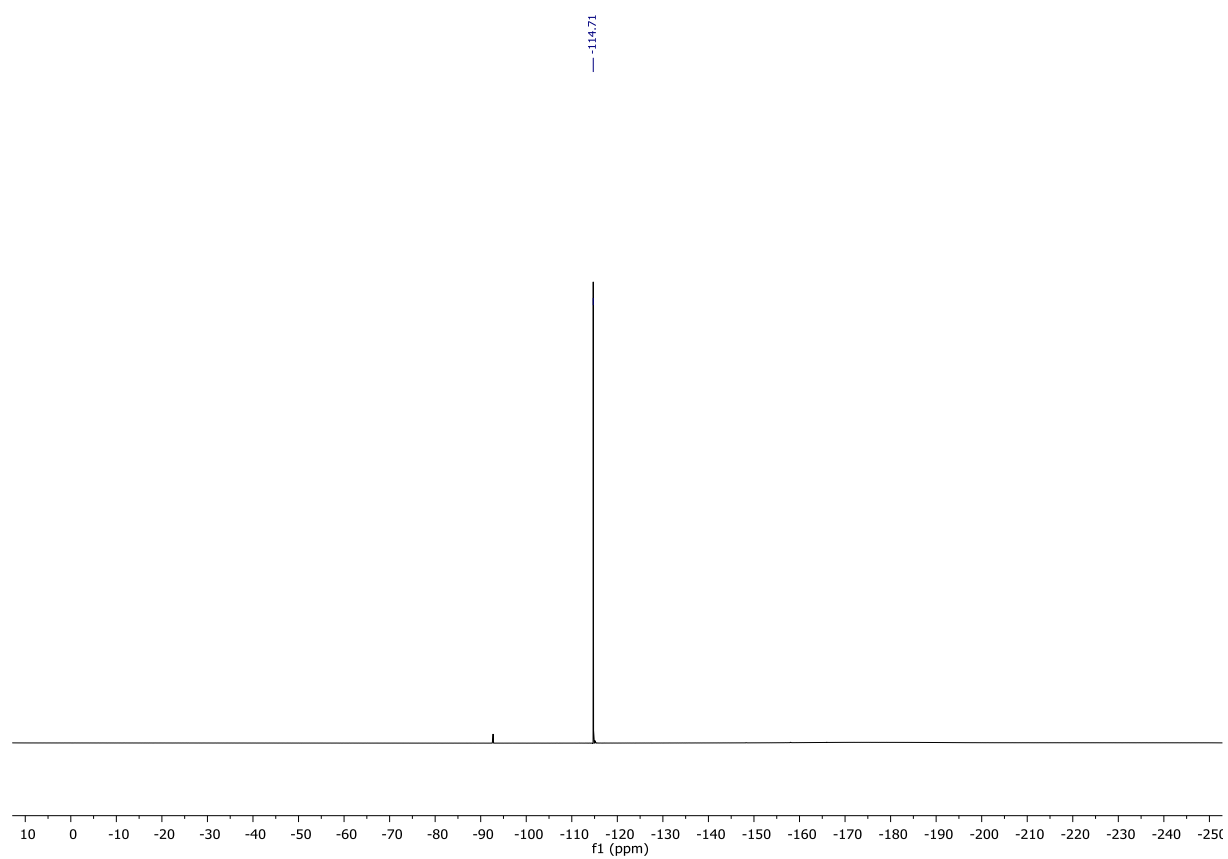
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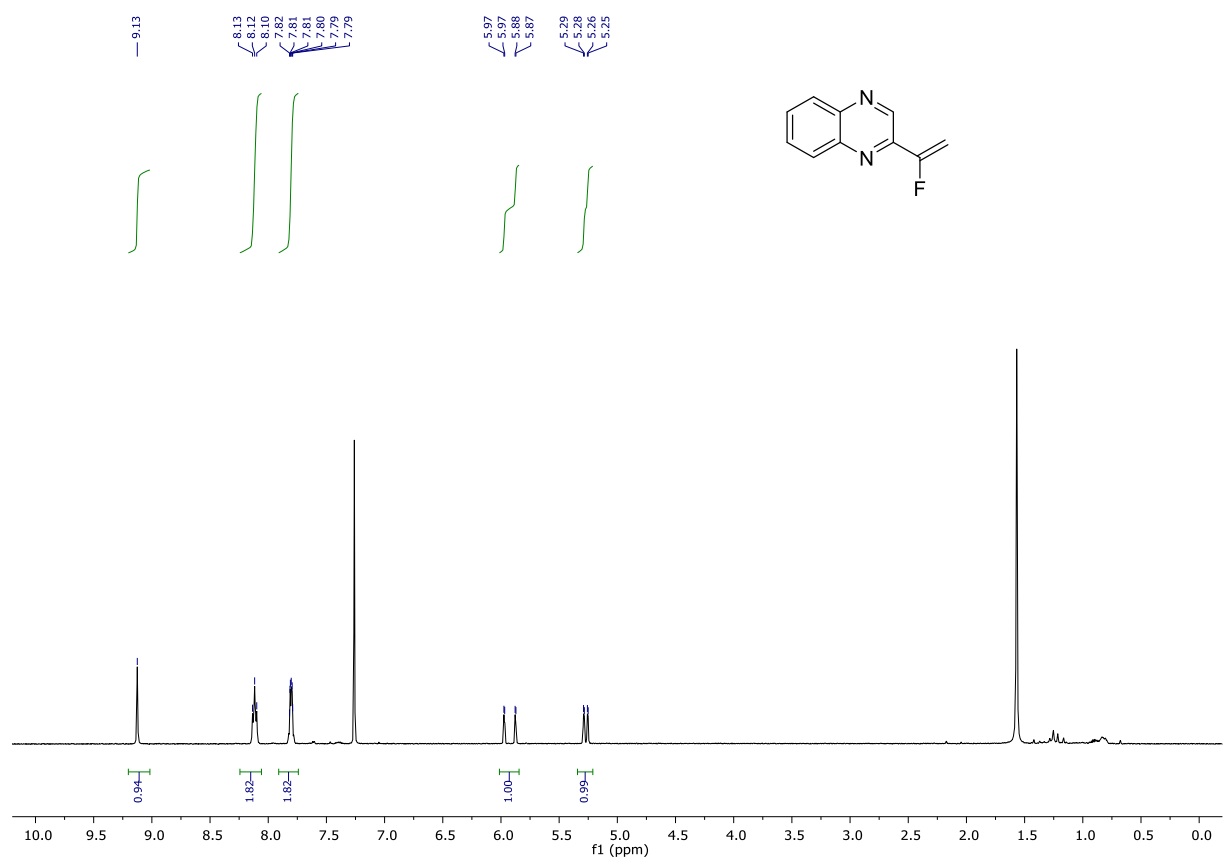
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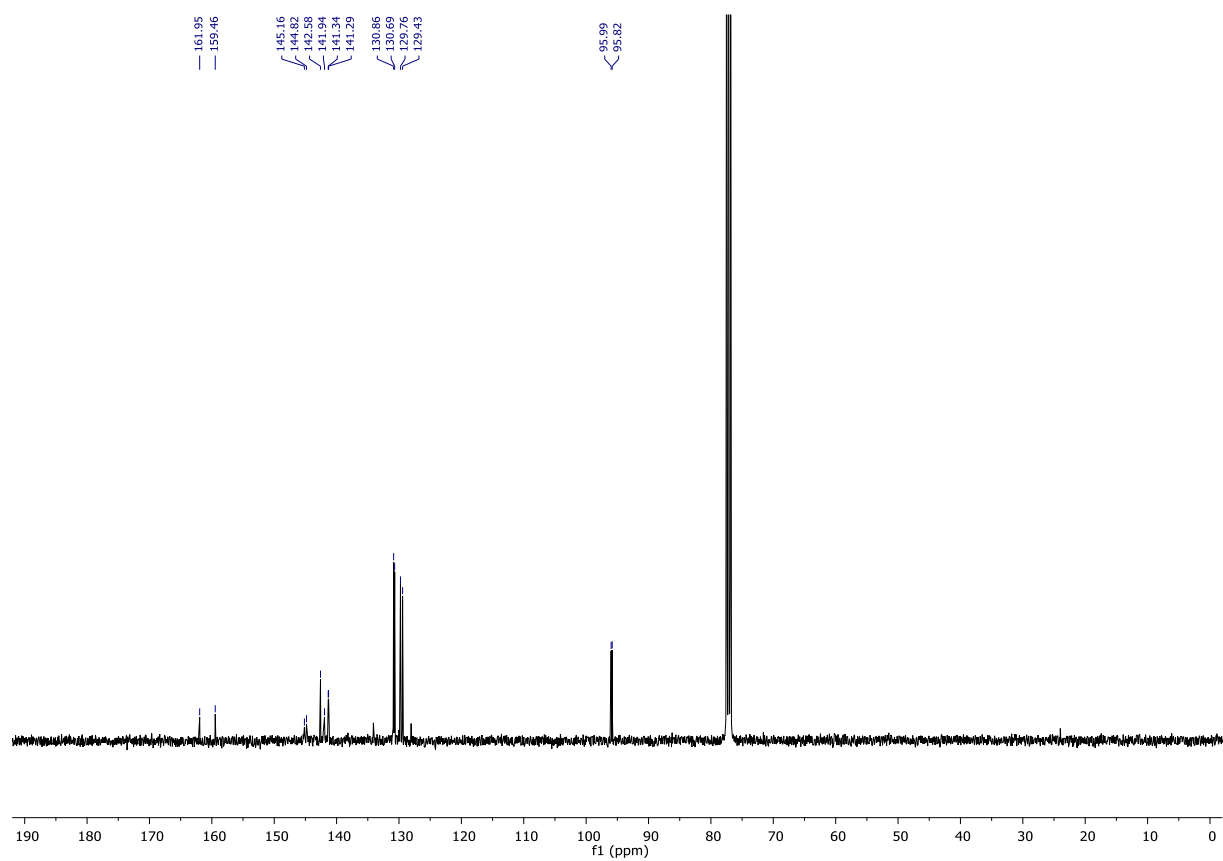
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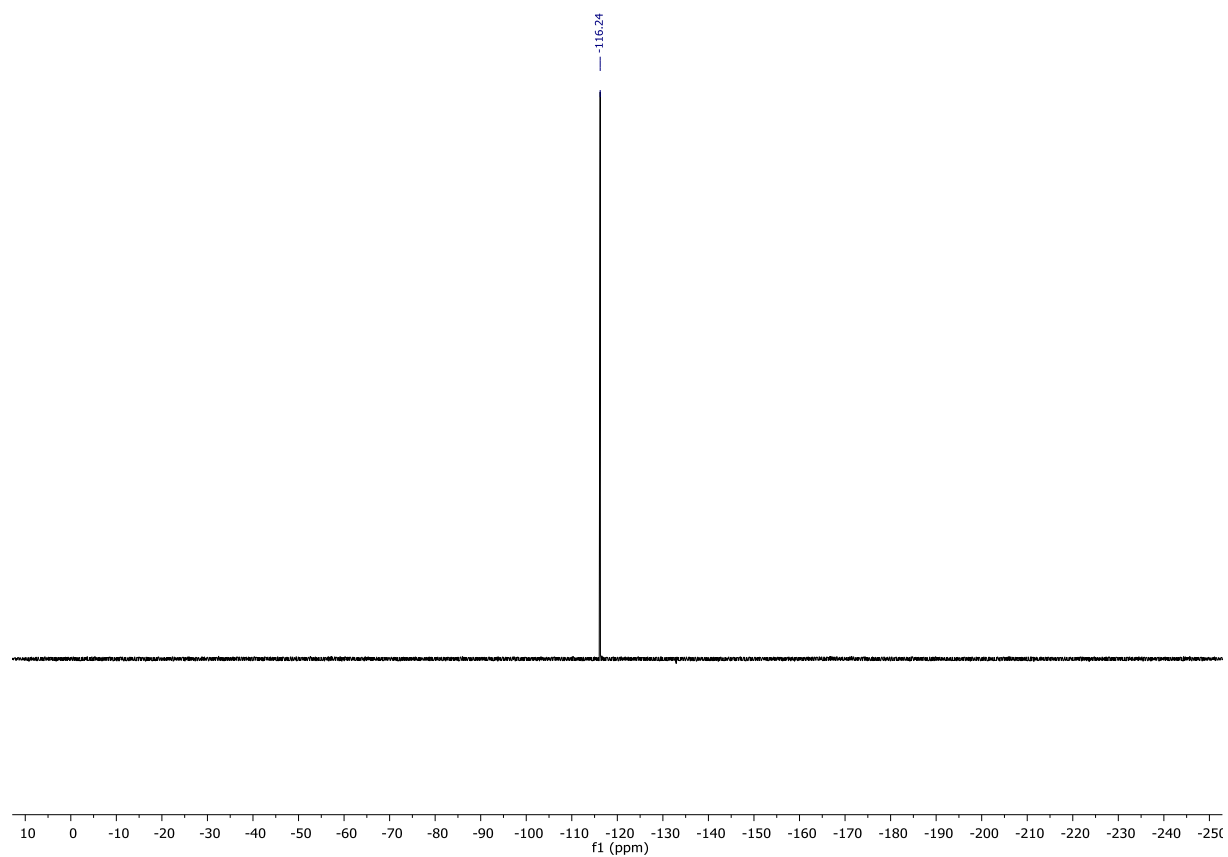
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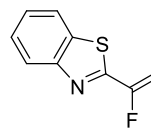
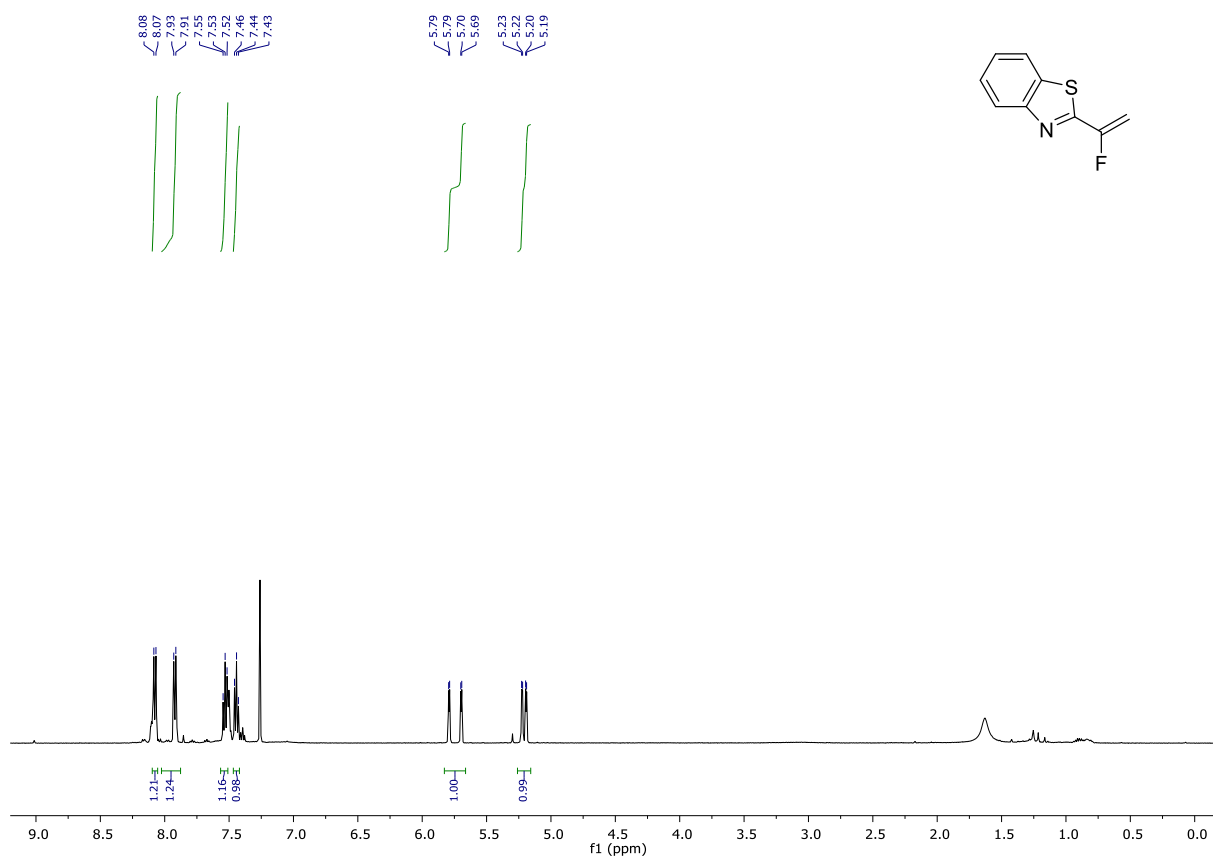
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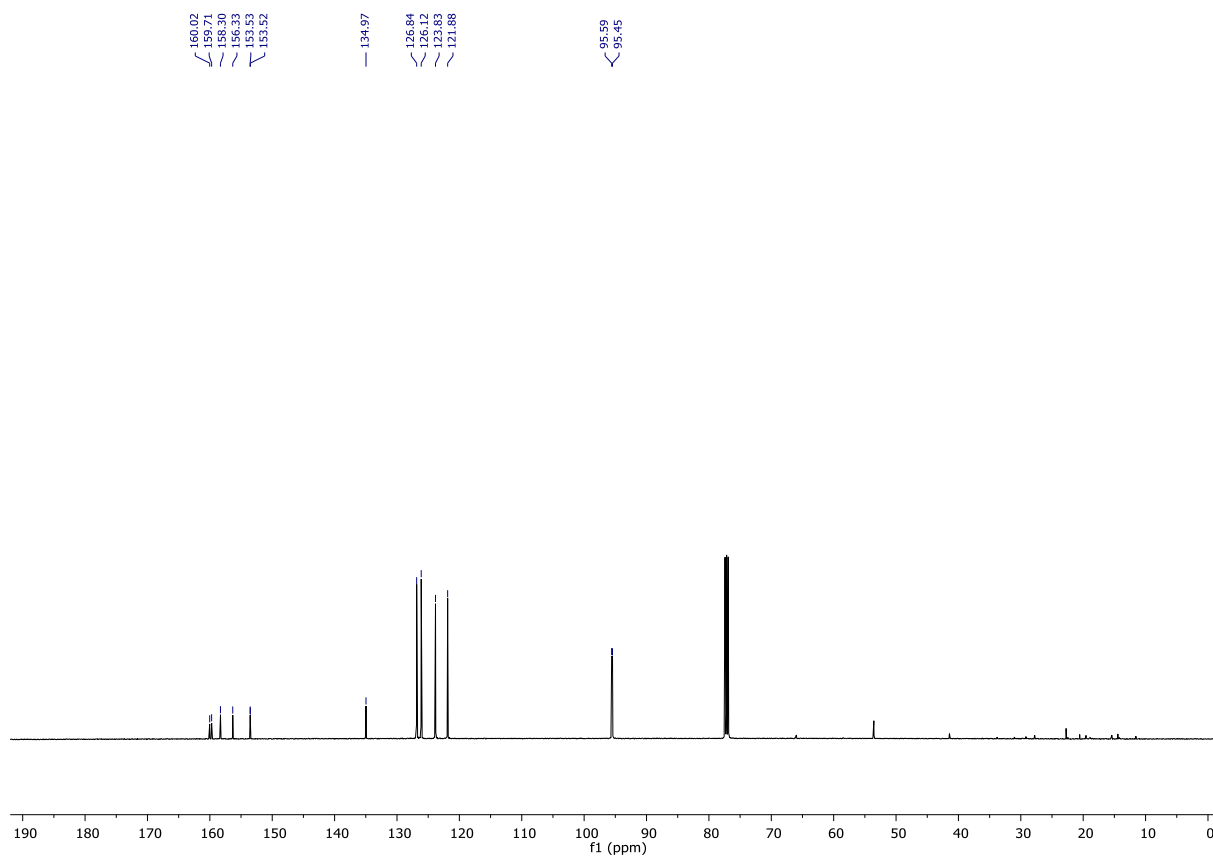
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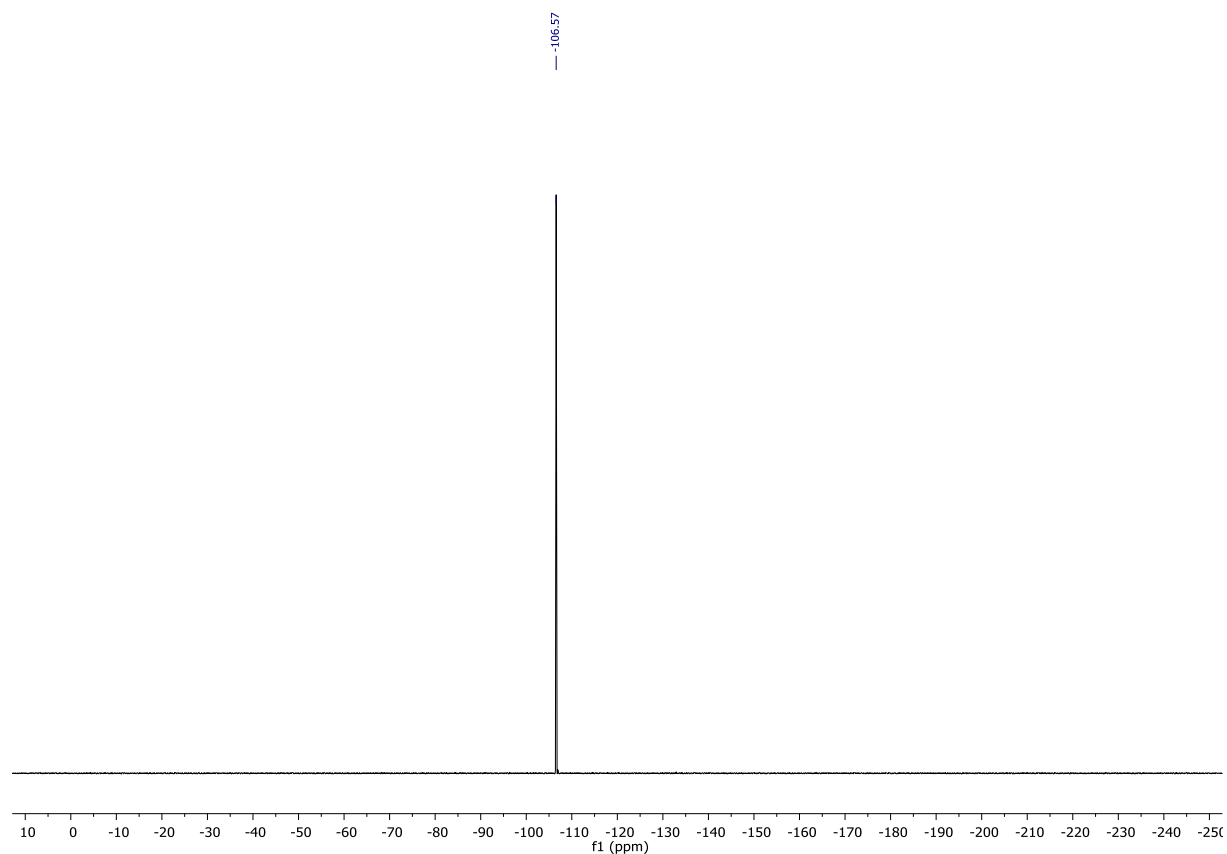
¹H NMR of 5g



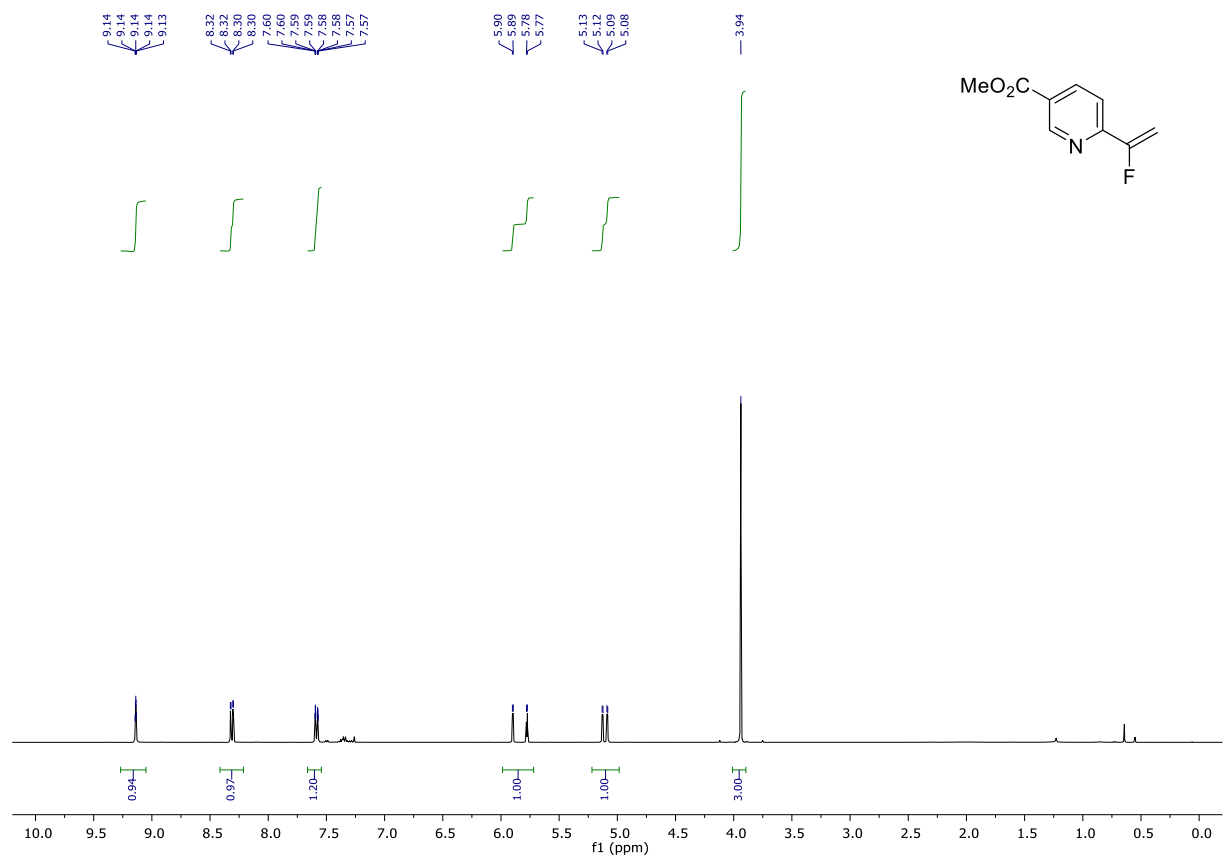
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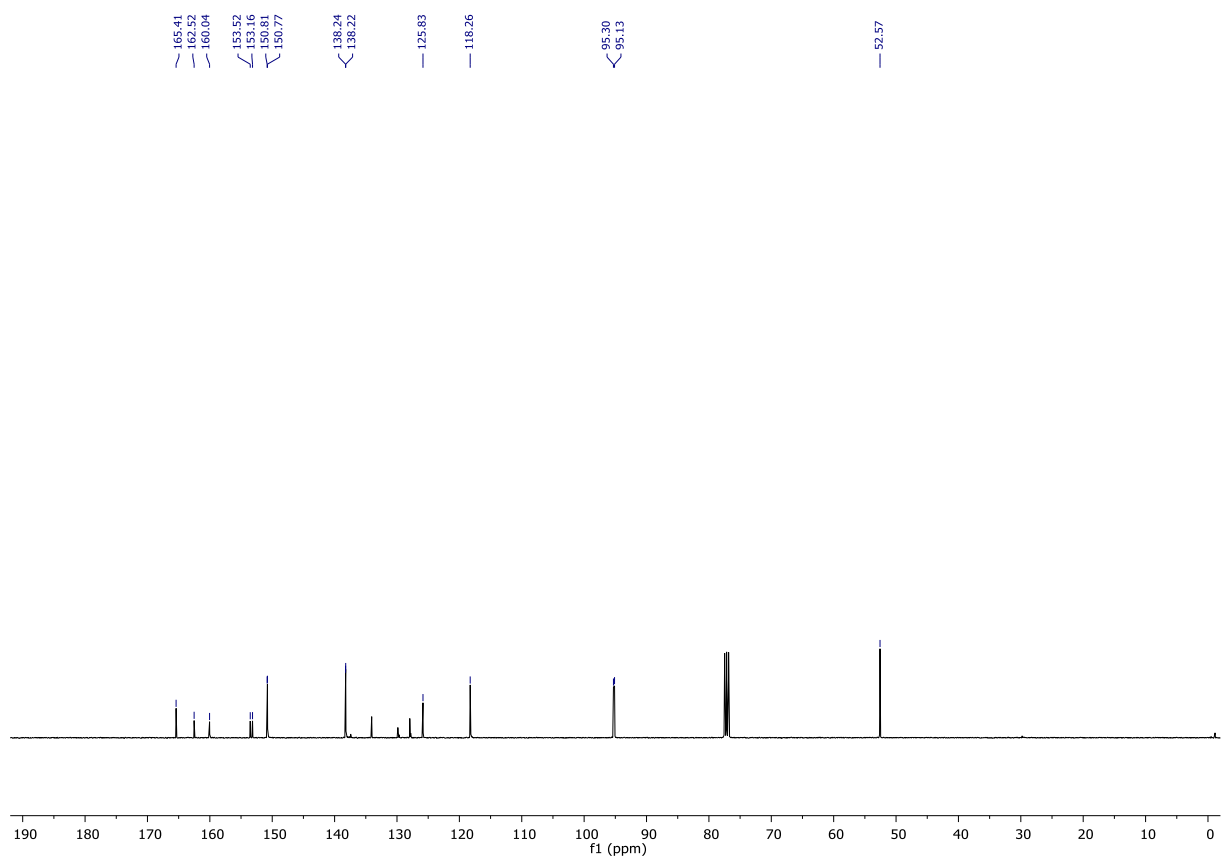
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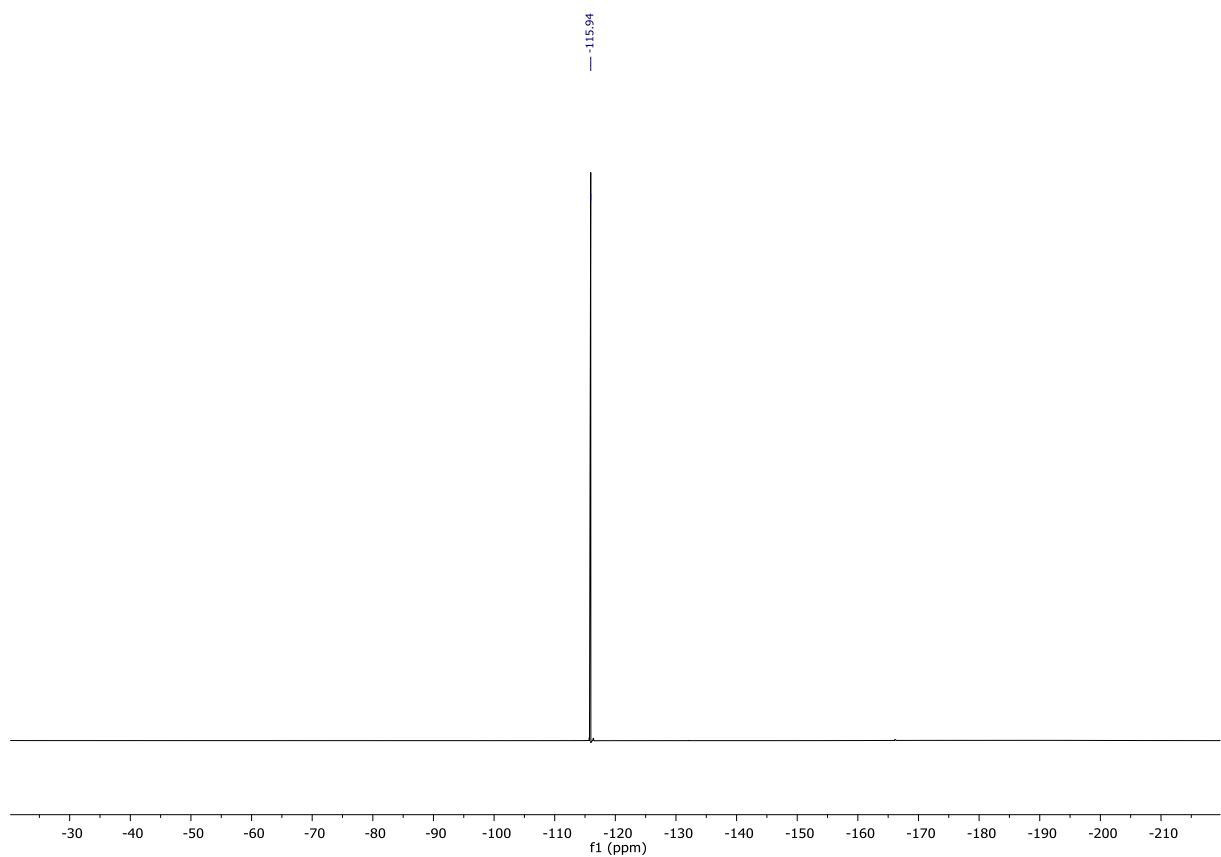
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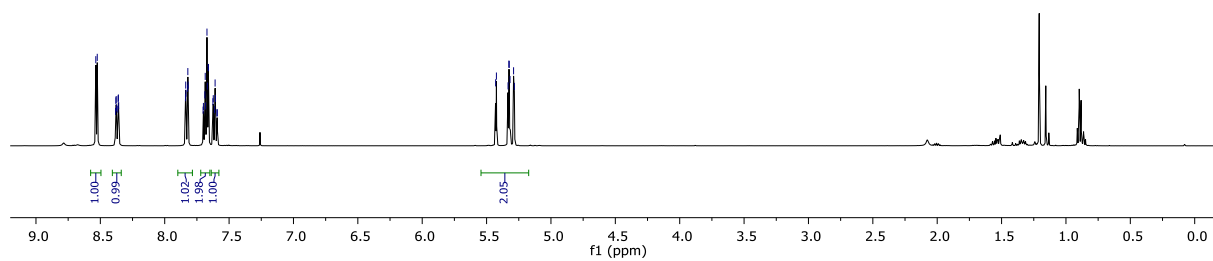
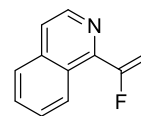
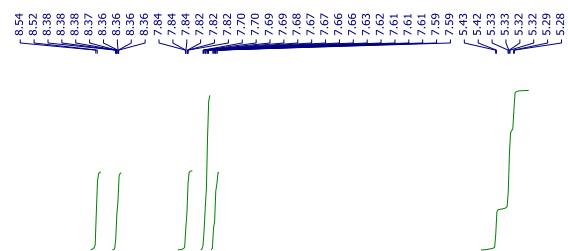
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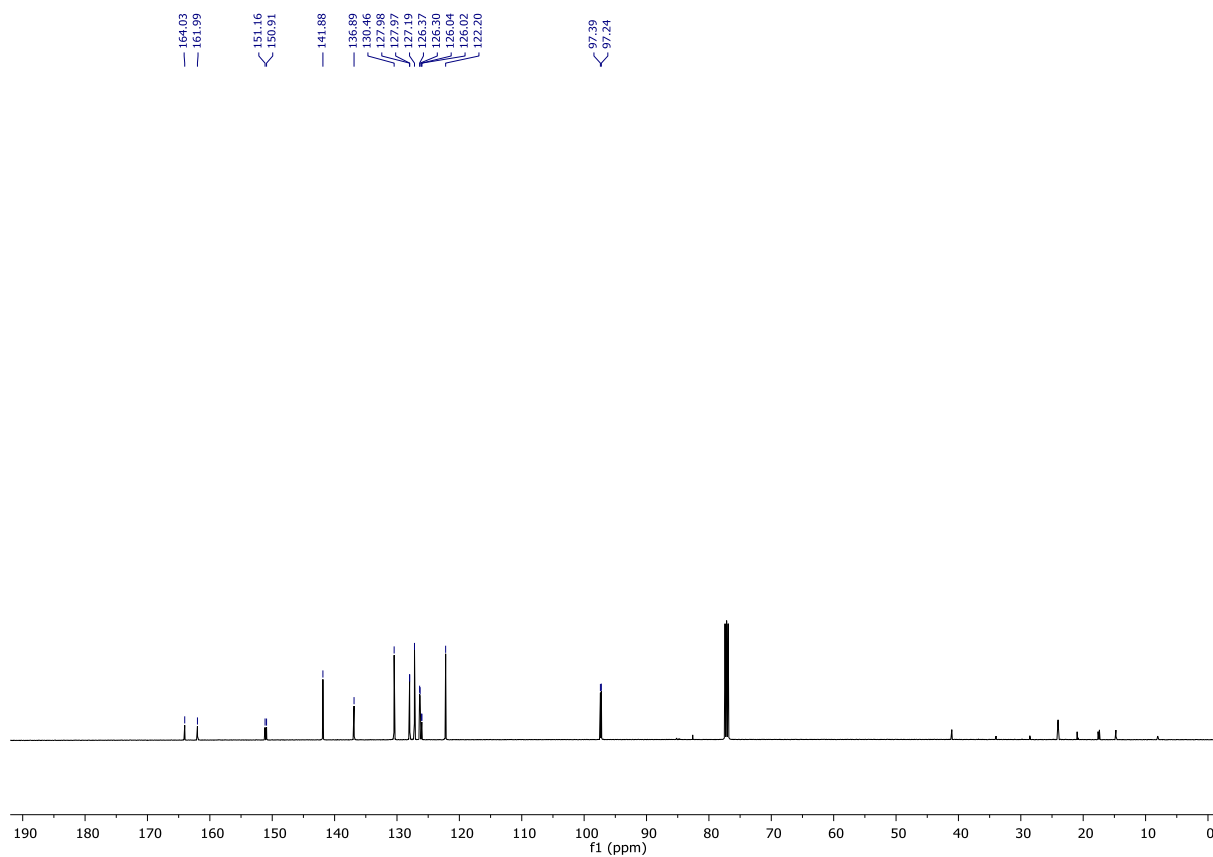
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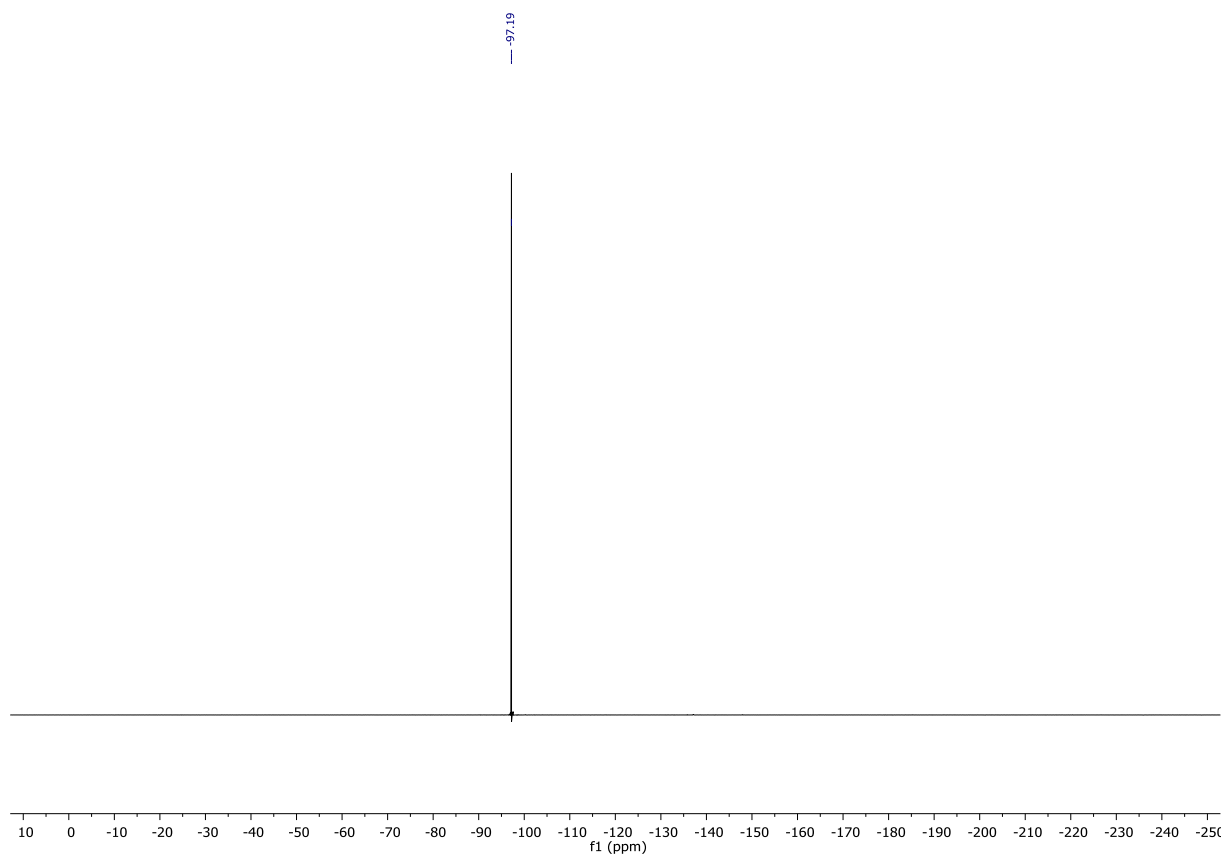
¹H NMR of 5i



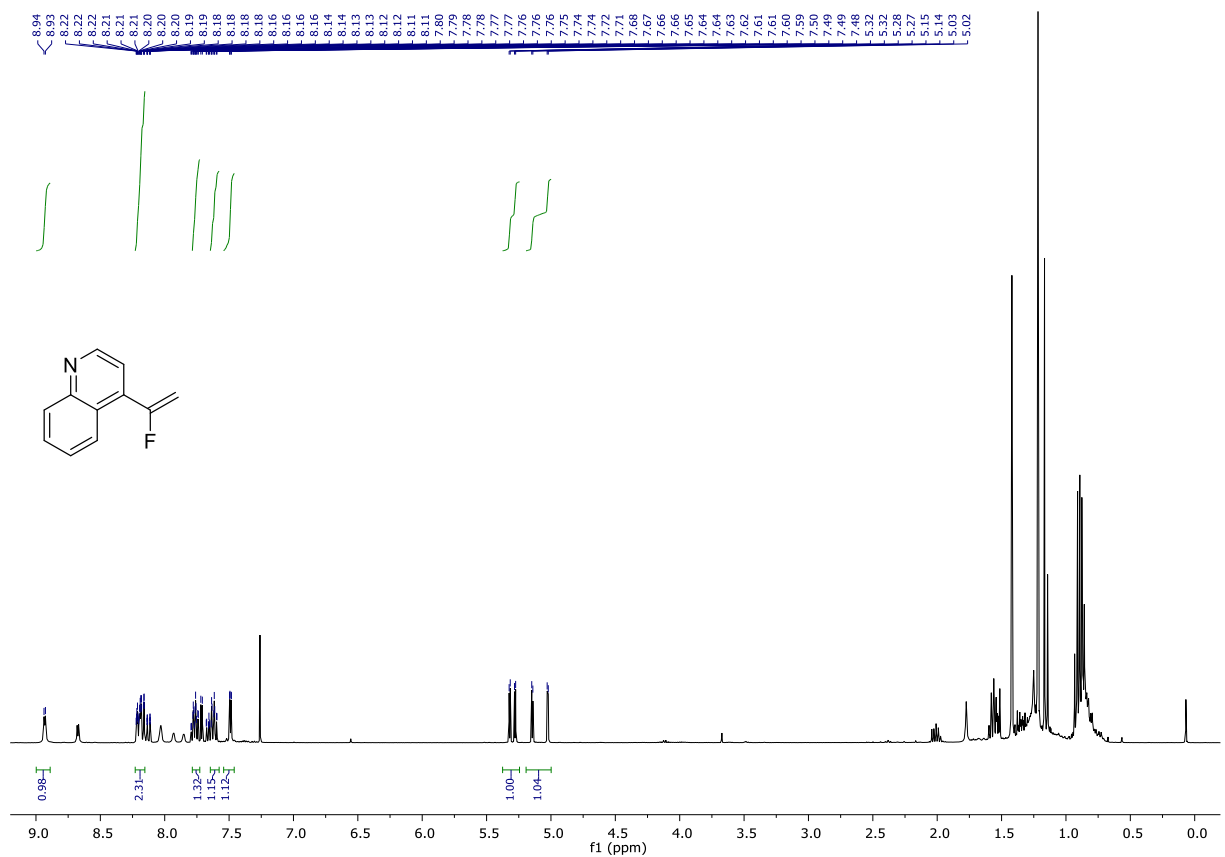
¹³C NMR of 5i



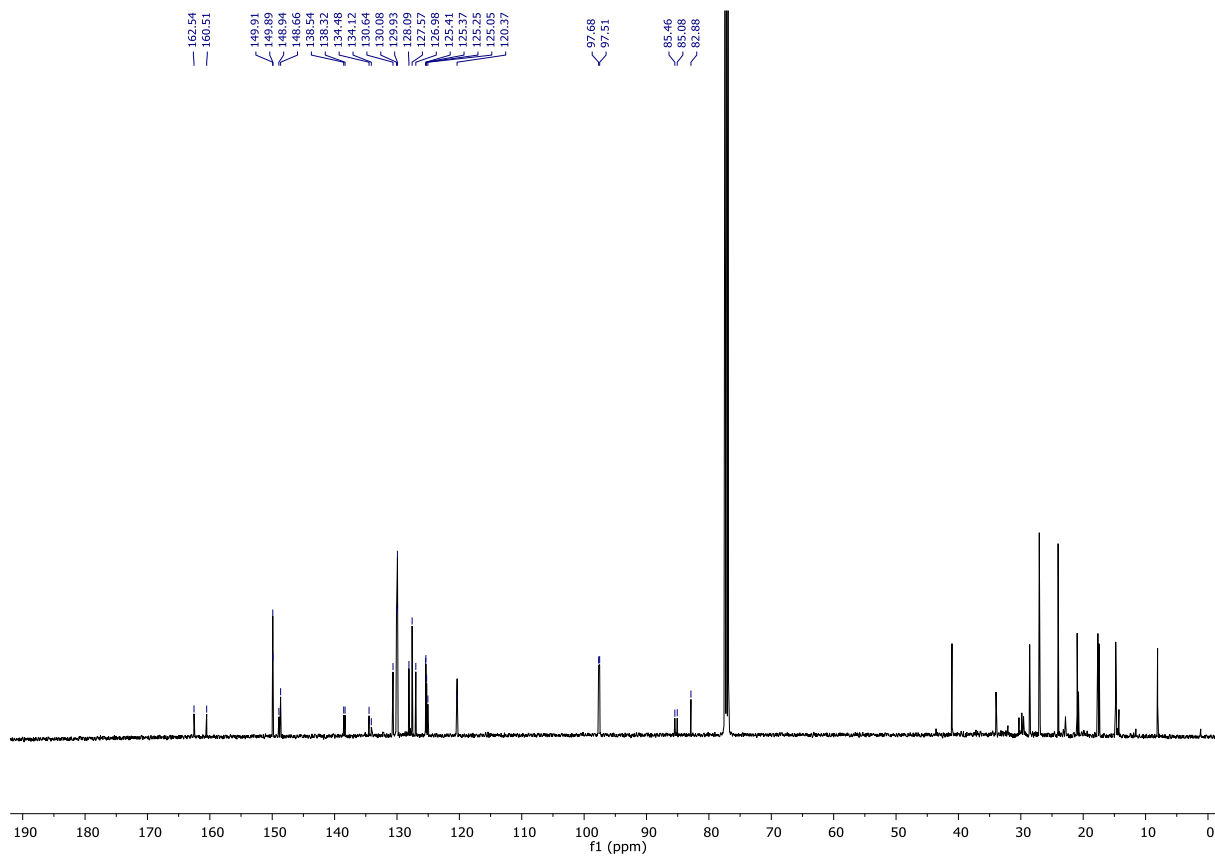
¹⁹F NMR of 5i



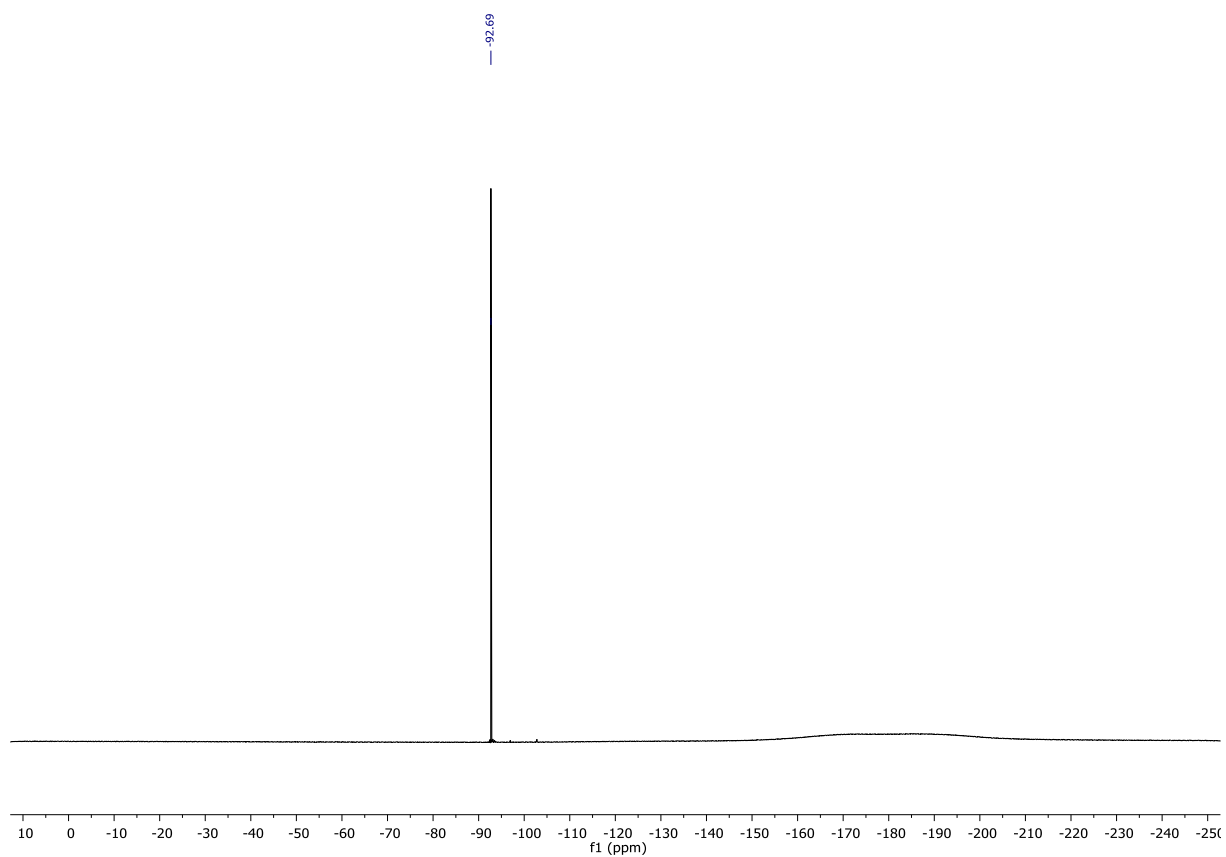
¹H NMR of 5j



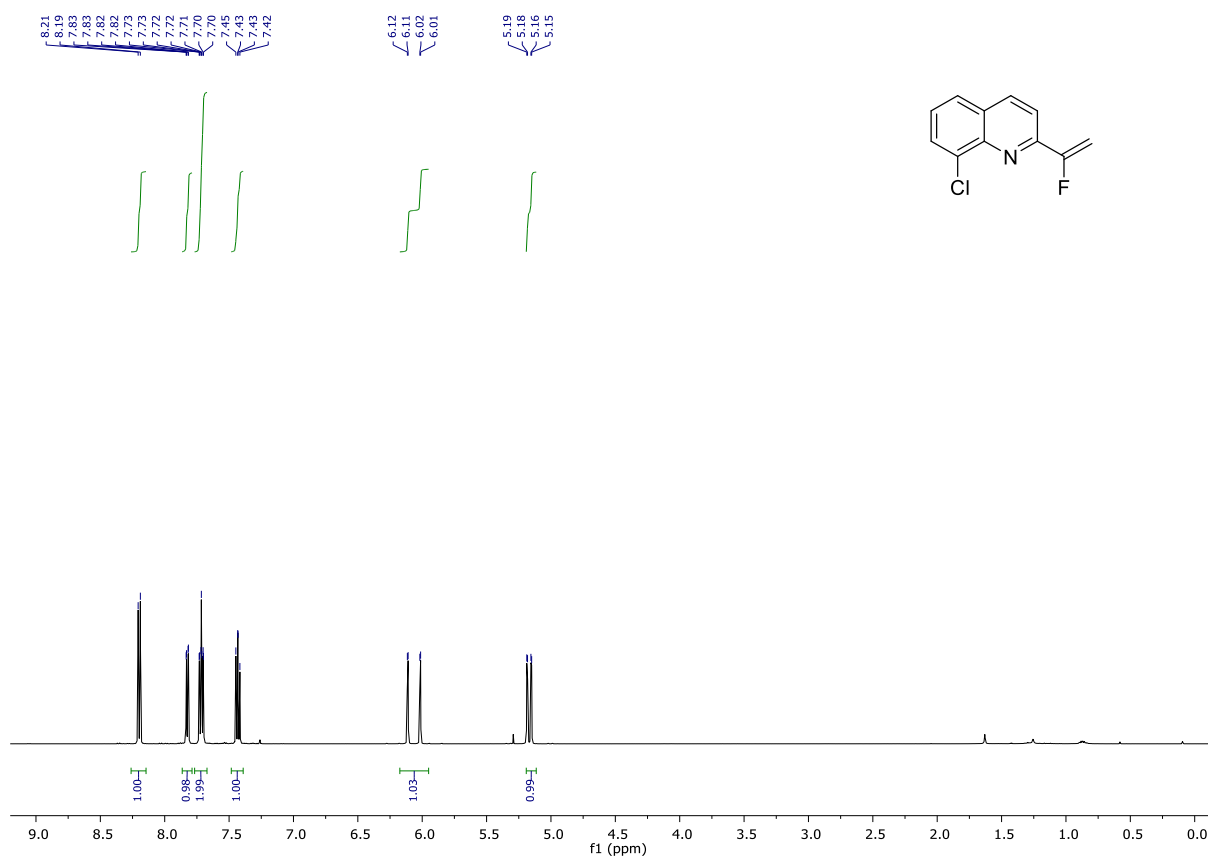
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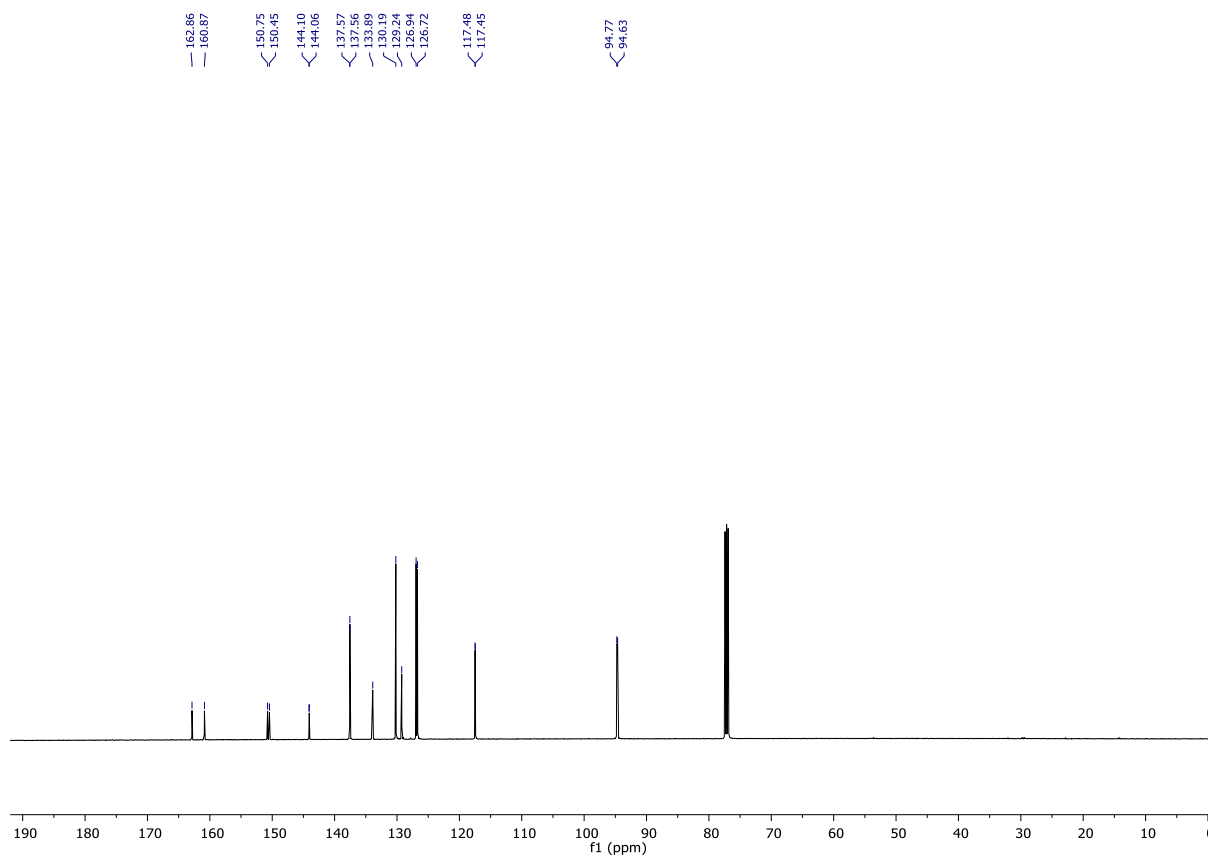
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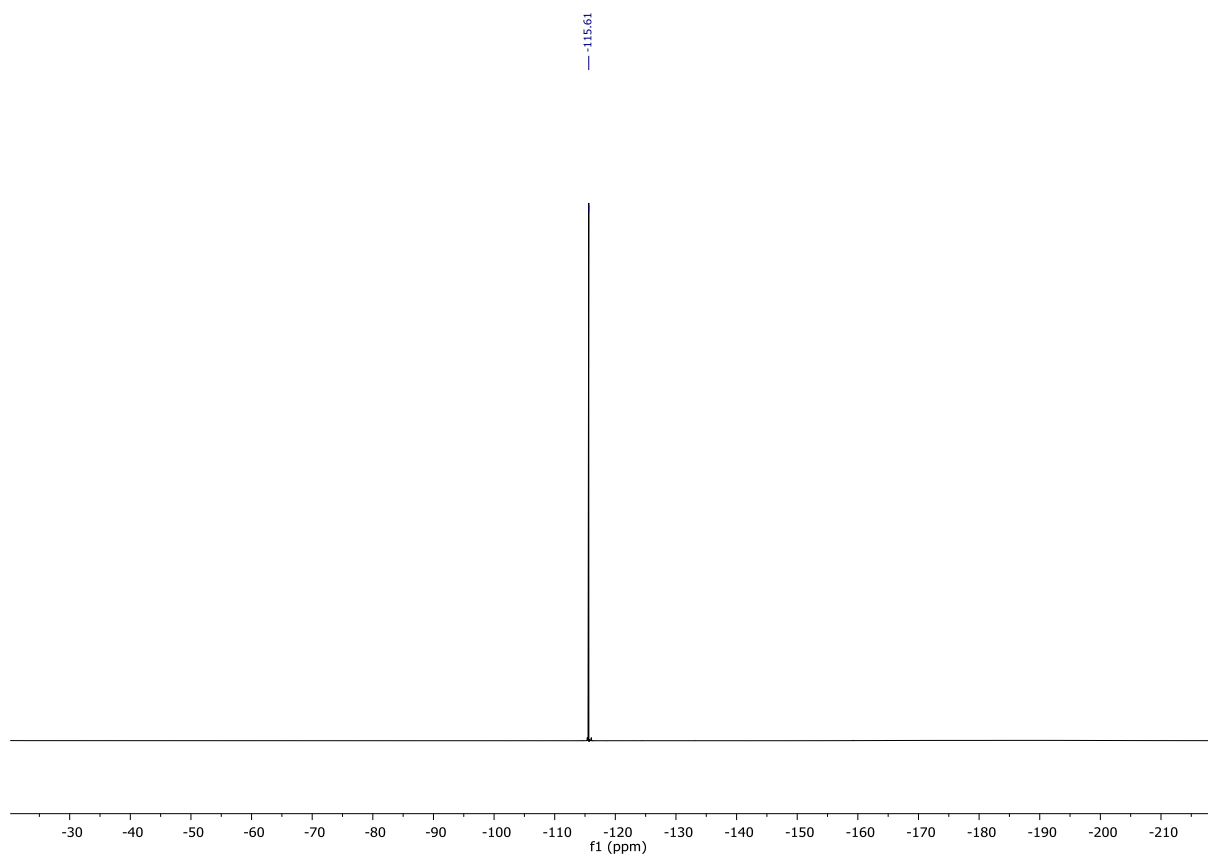
¹H NMR of 5k



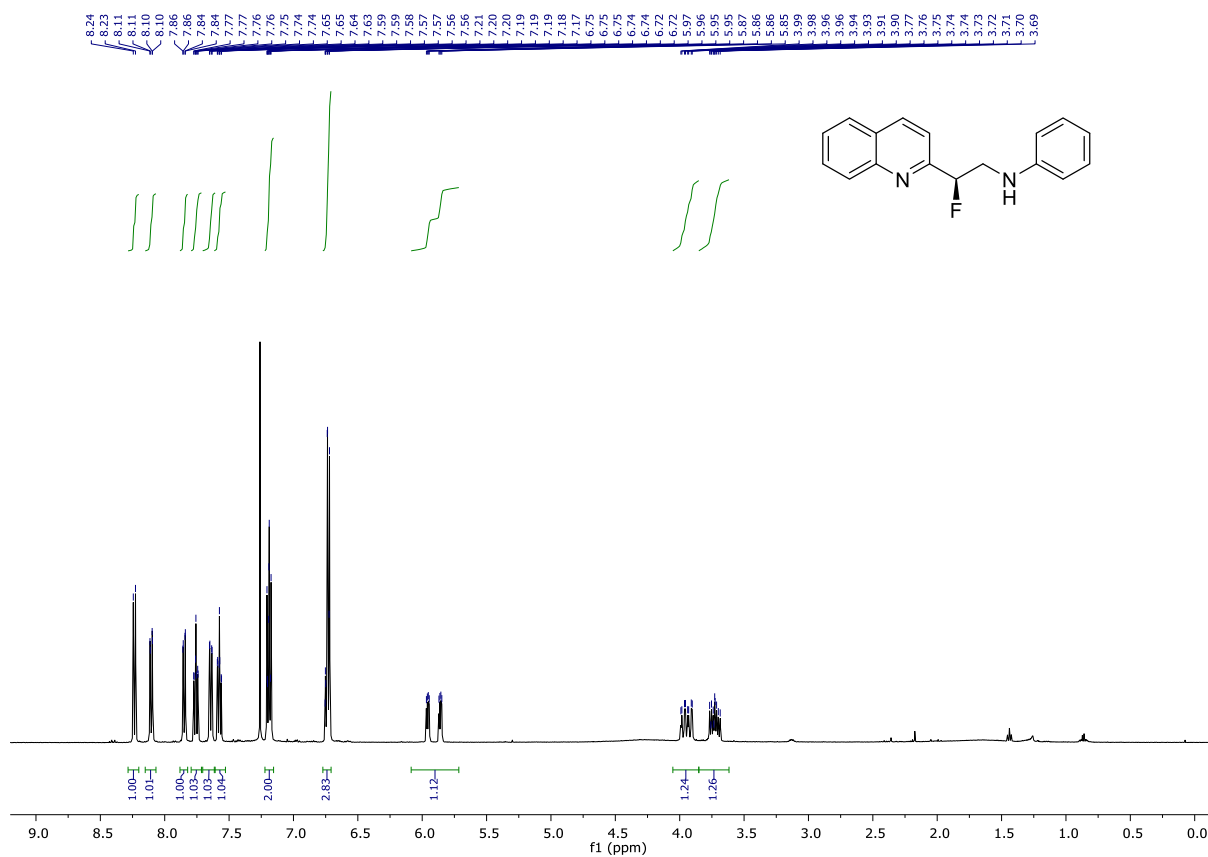
¹³C NMR of 5k



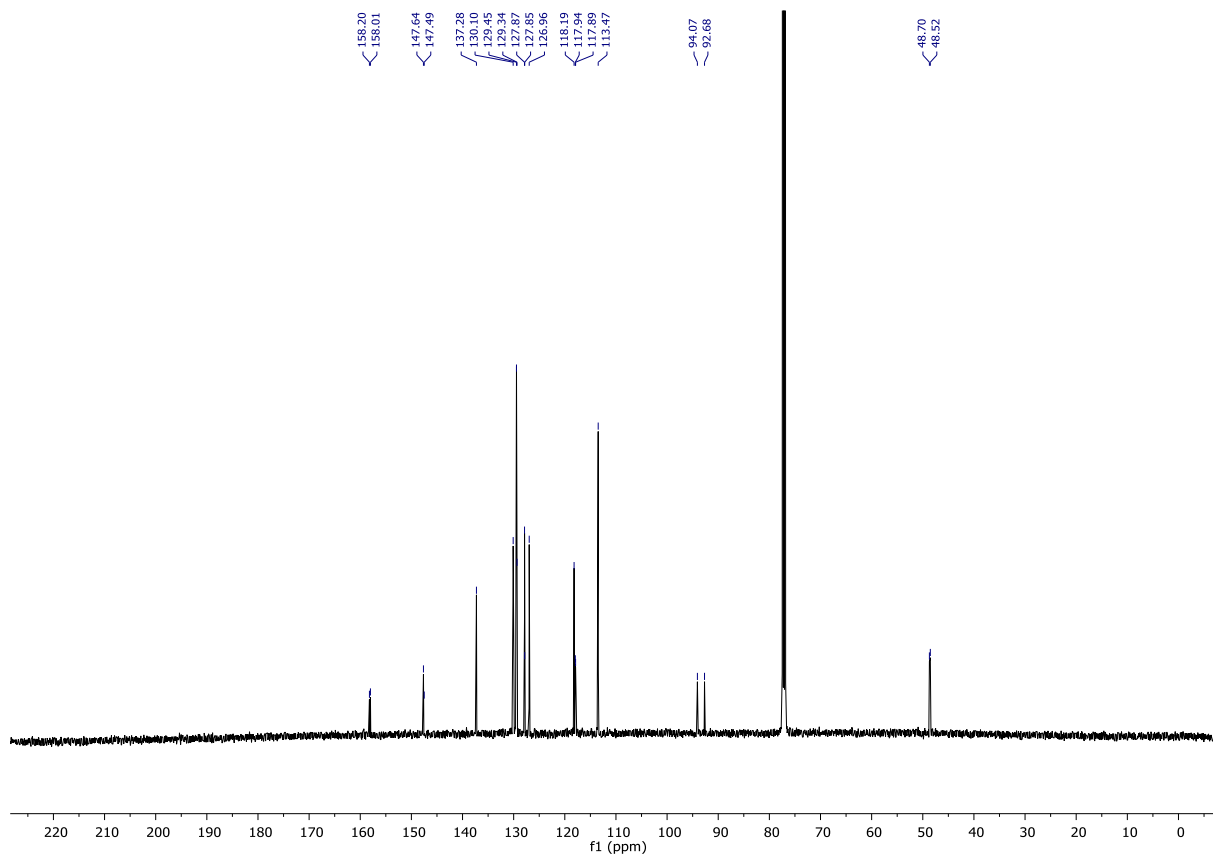
¹⁹F NMR of 5k



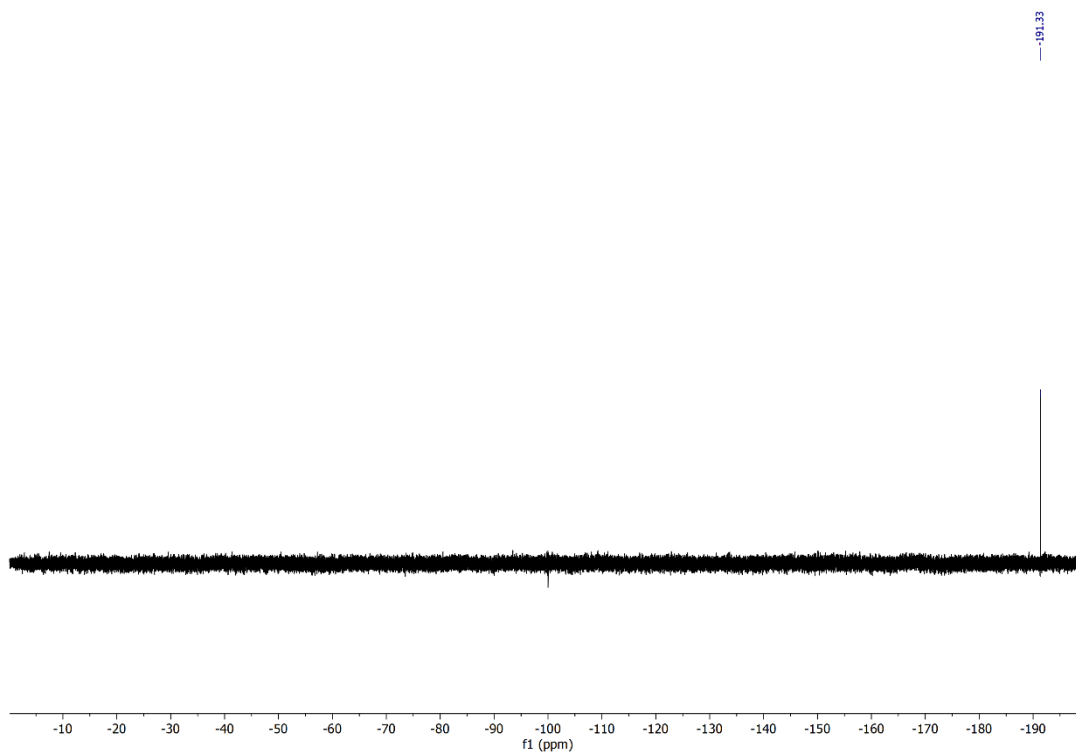
¹H NMR of 7a



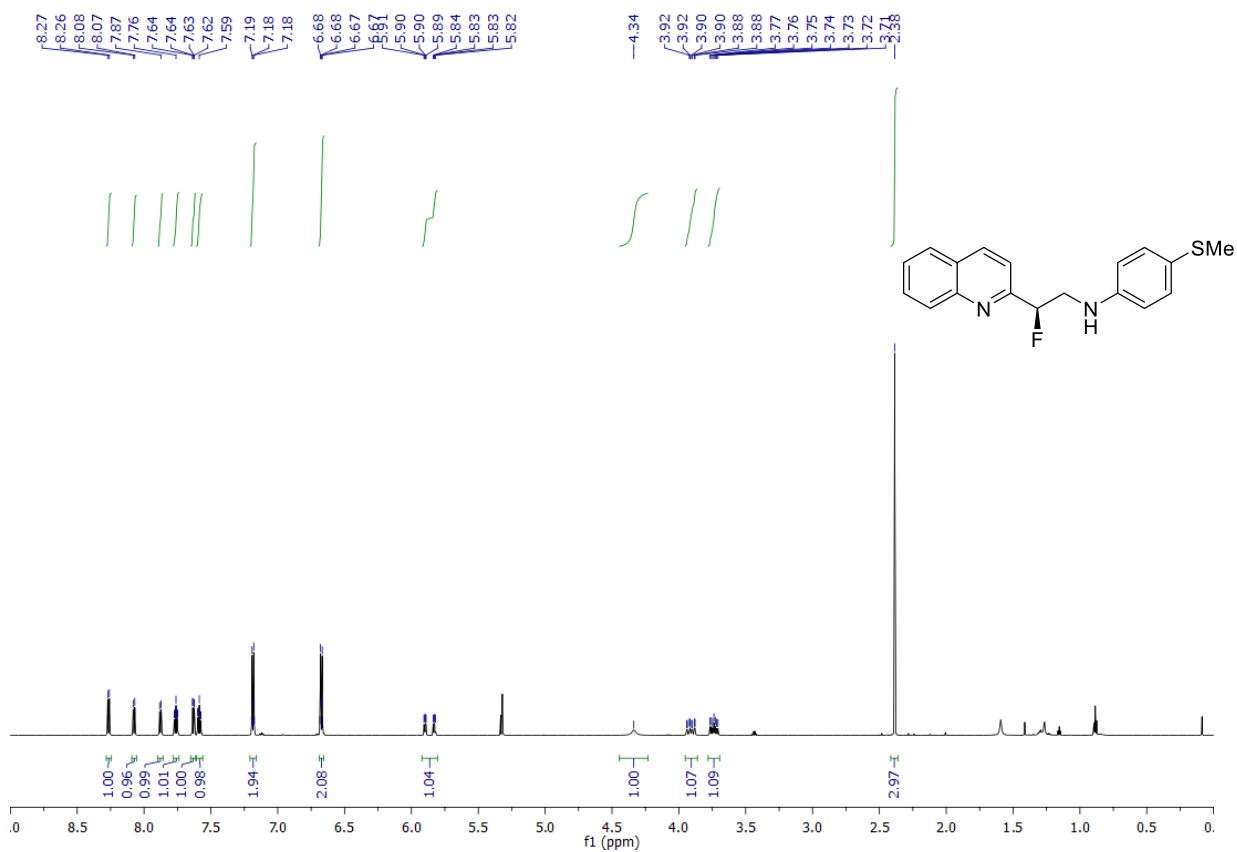
¹³C NMR of 7a



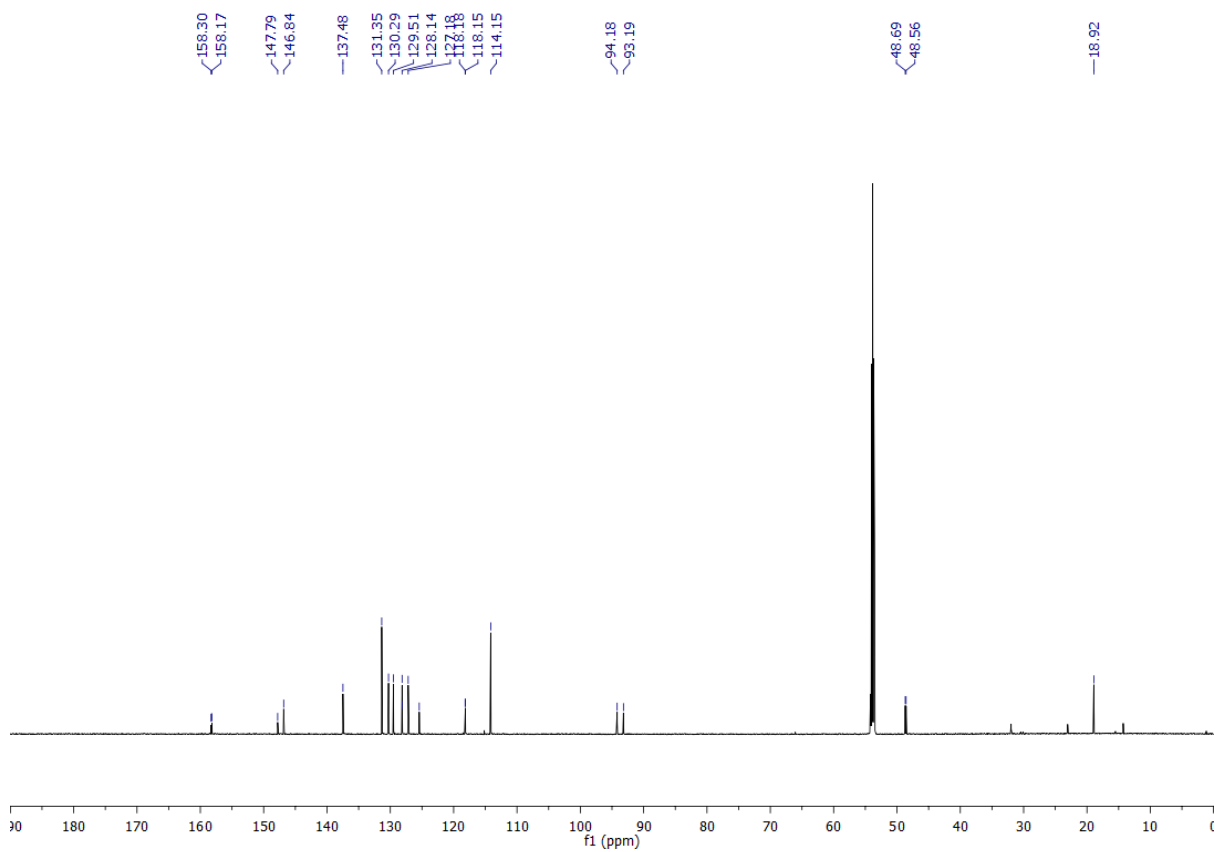
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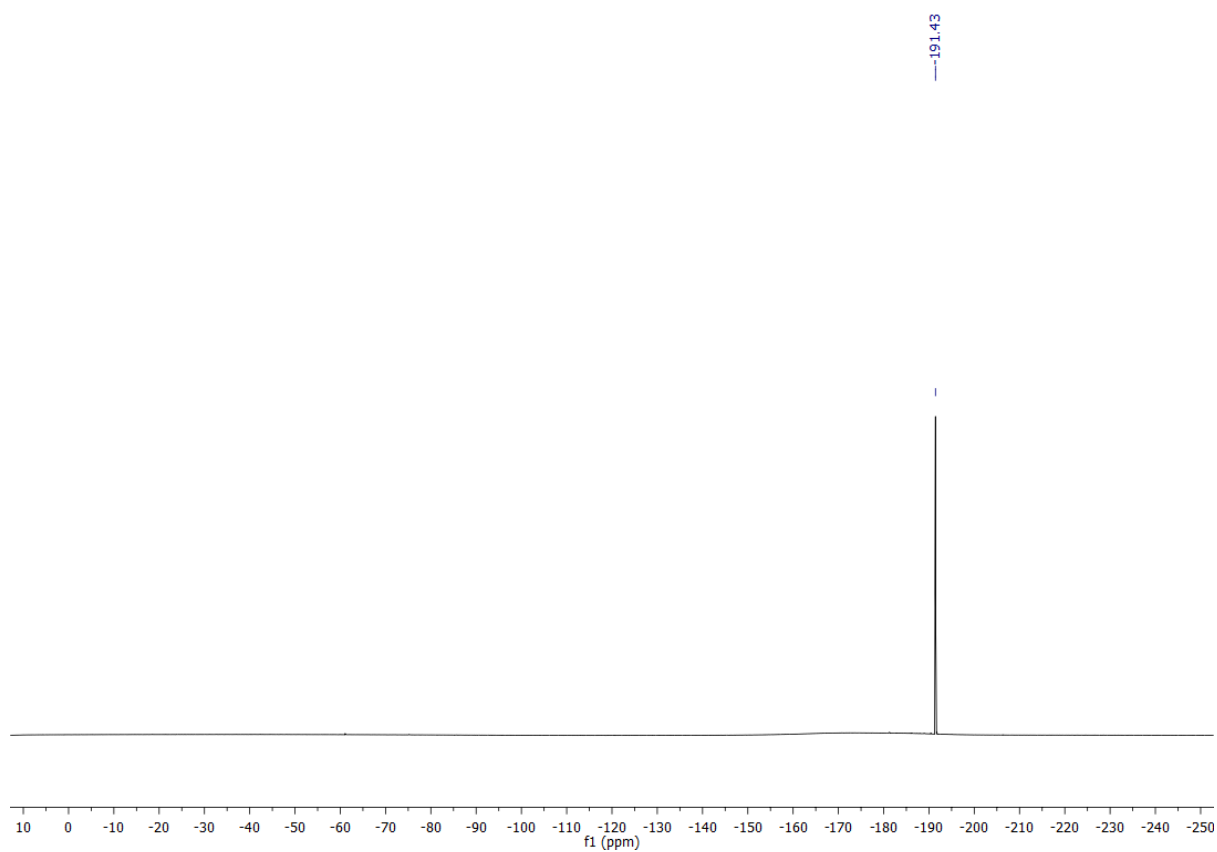
¹H NMR of 7b



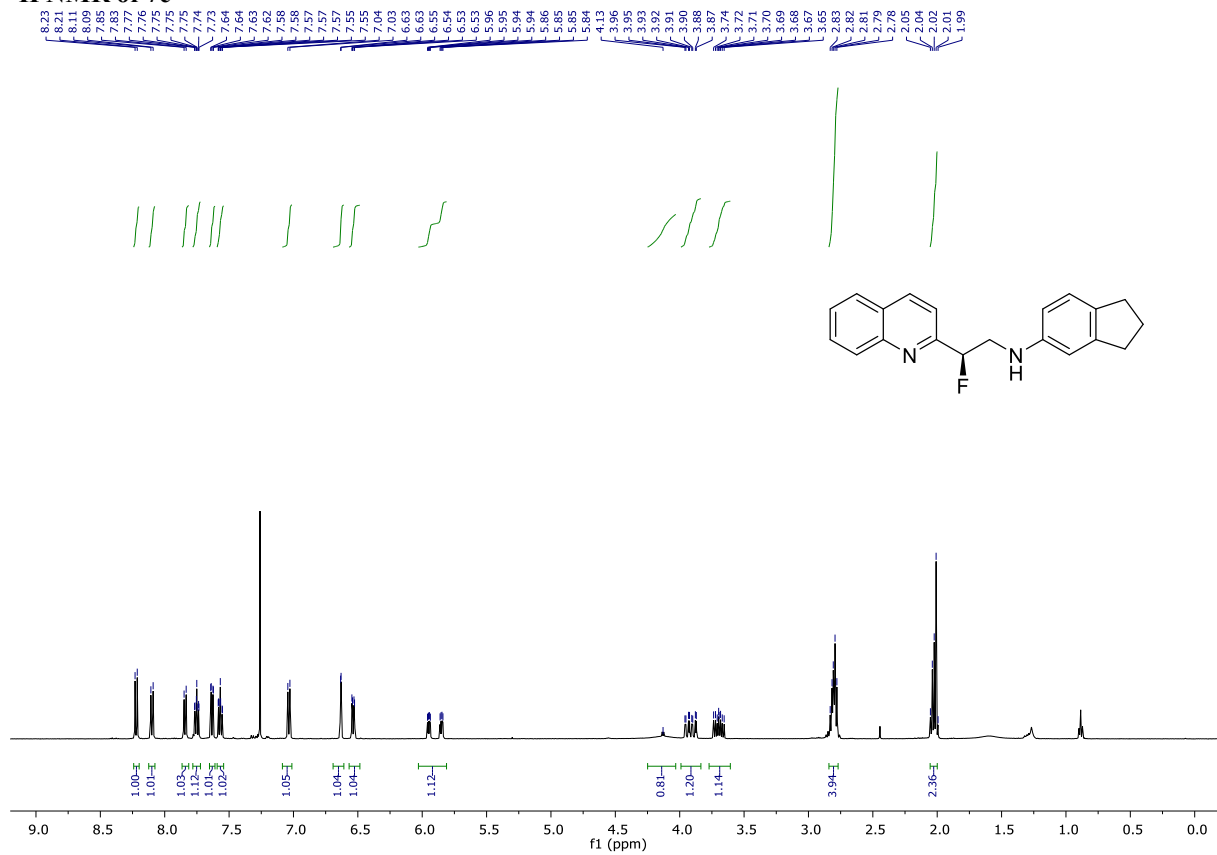
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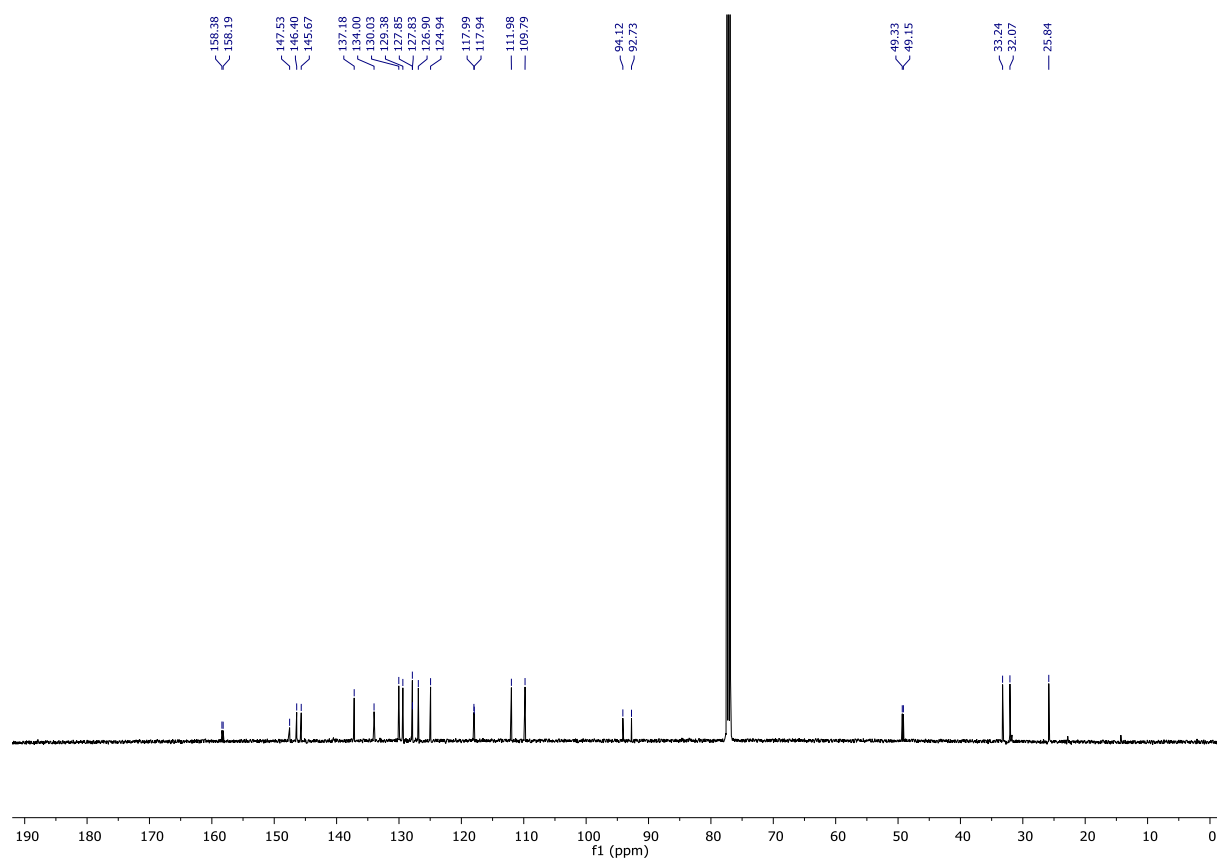
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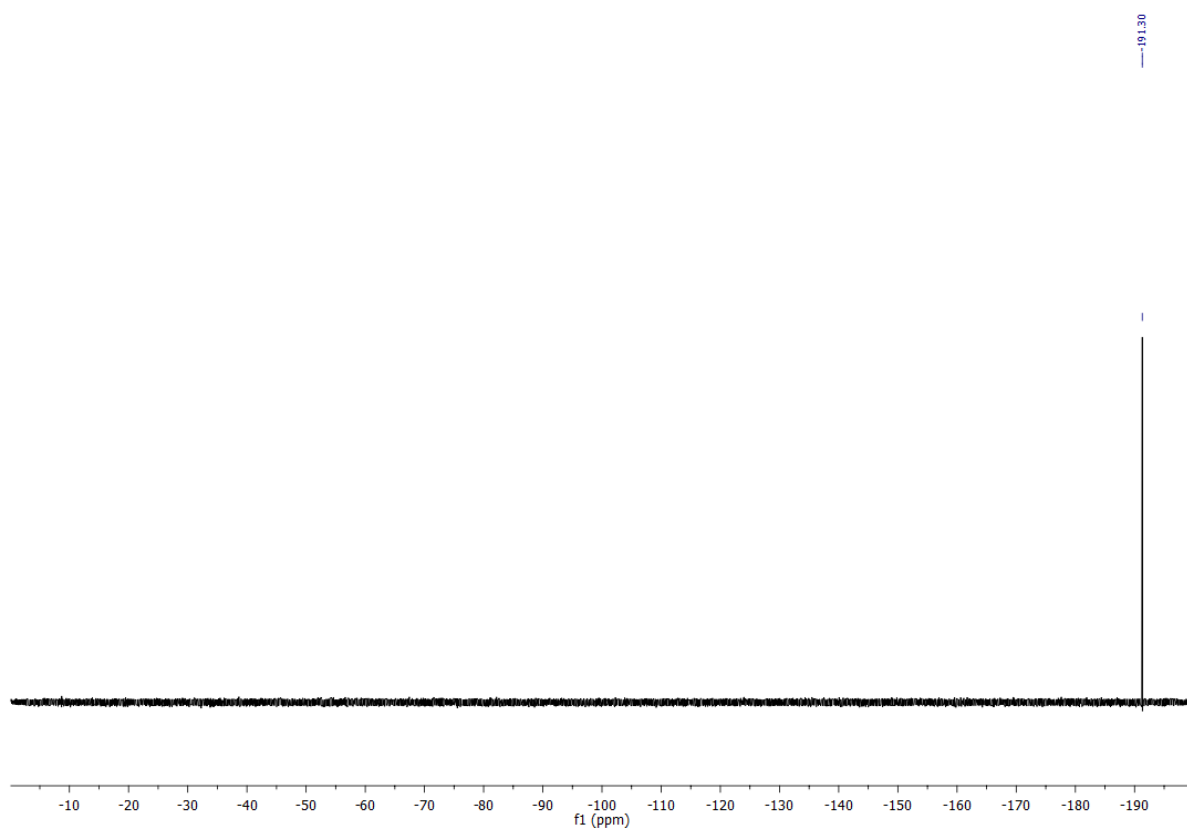
¹H NMR of 7c



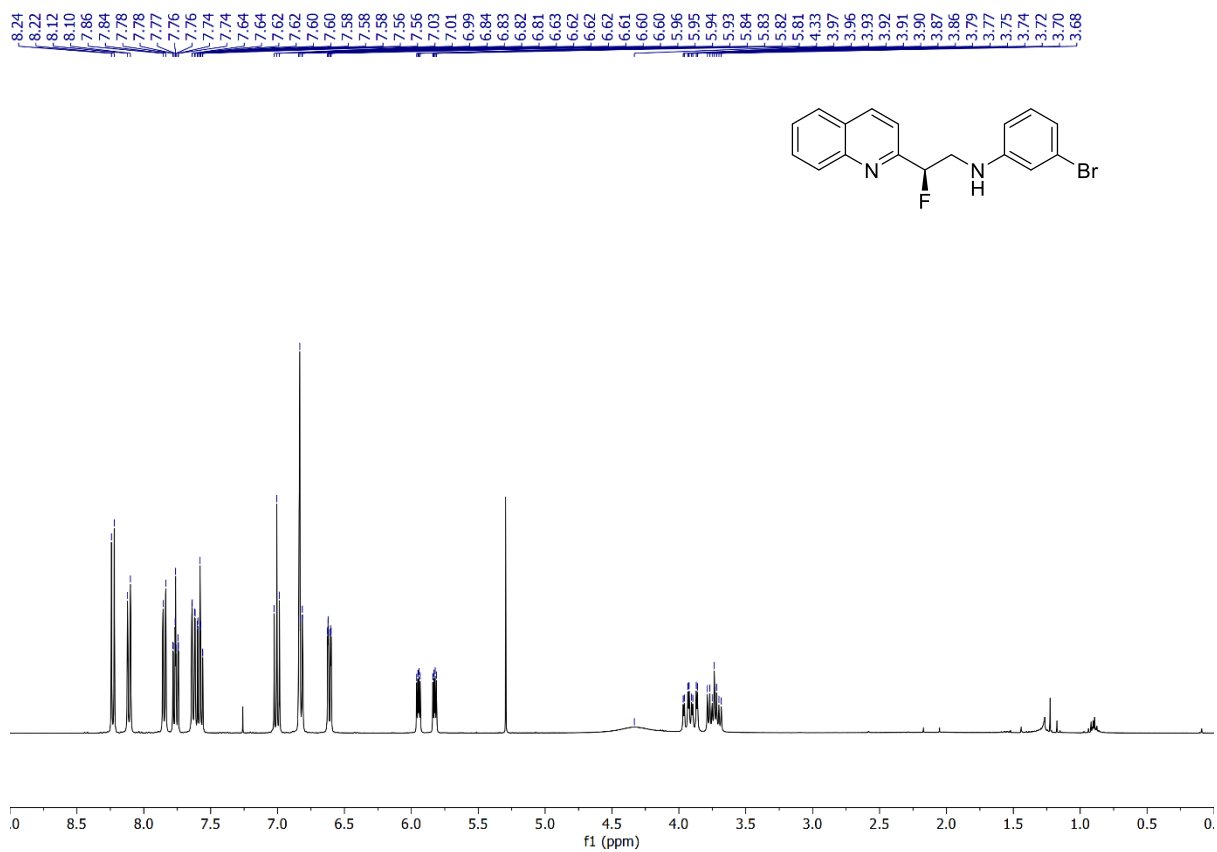
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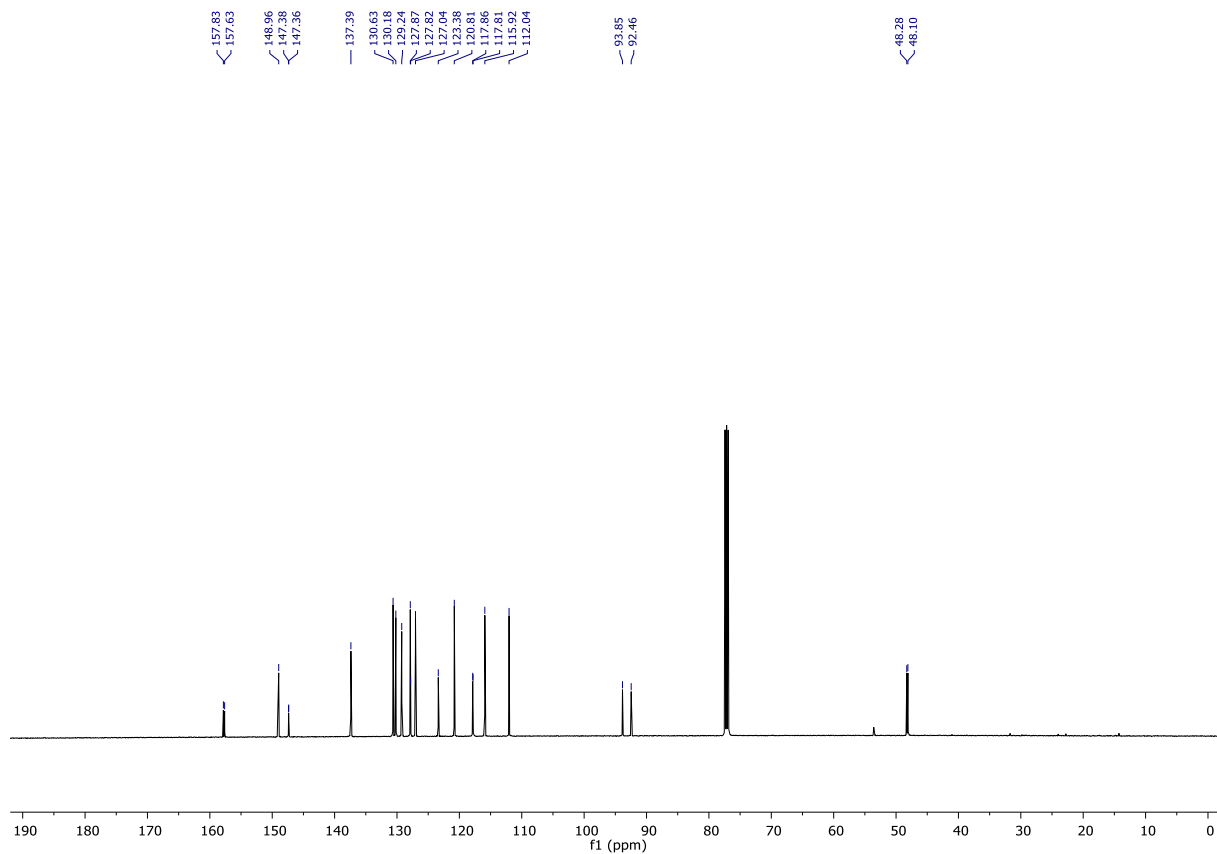
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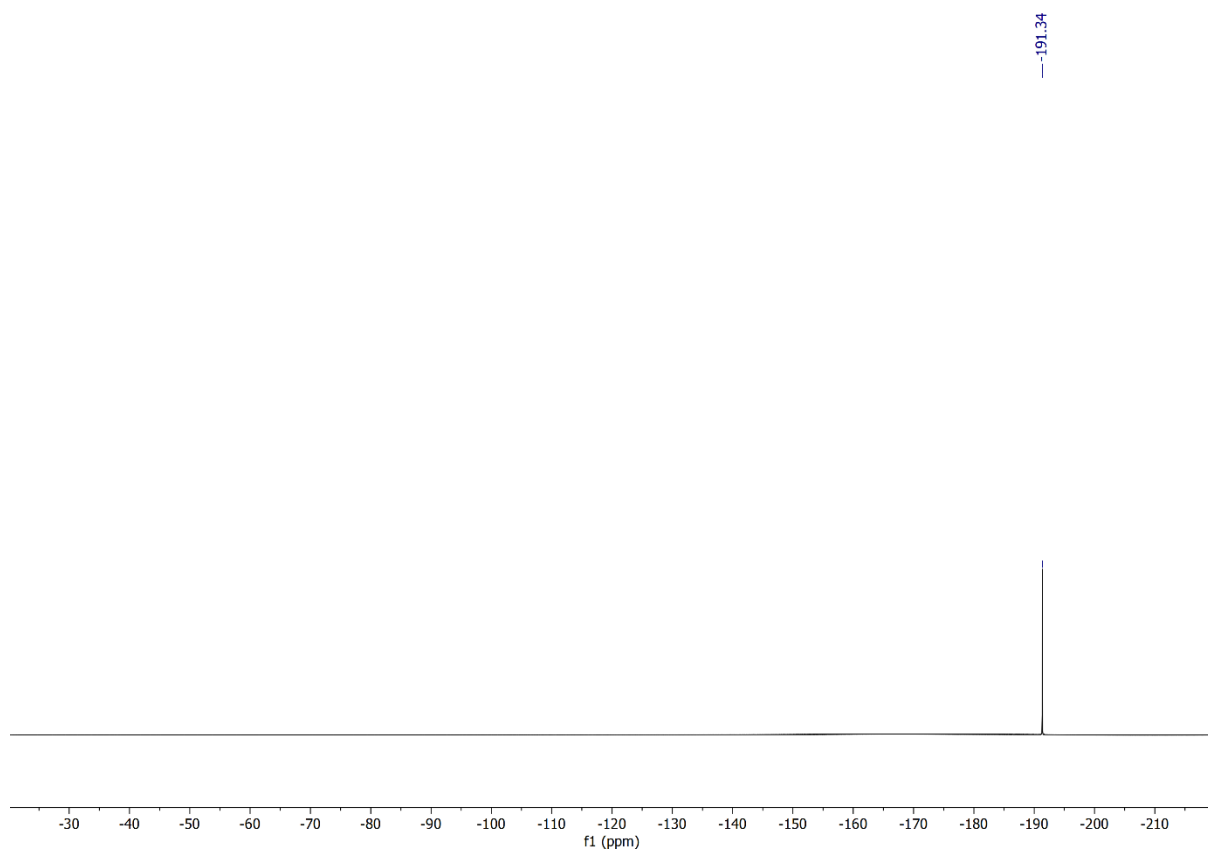
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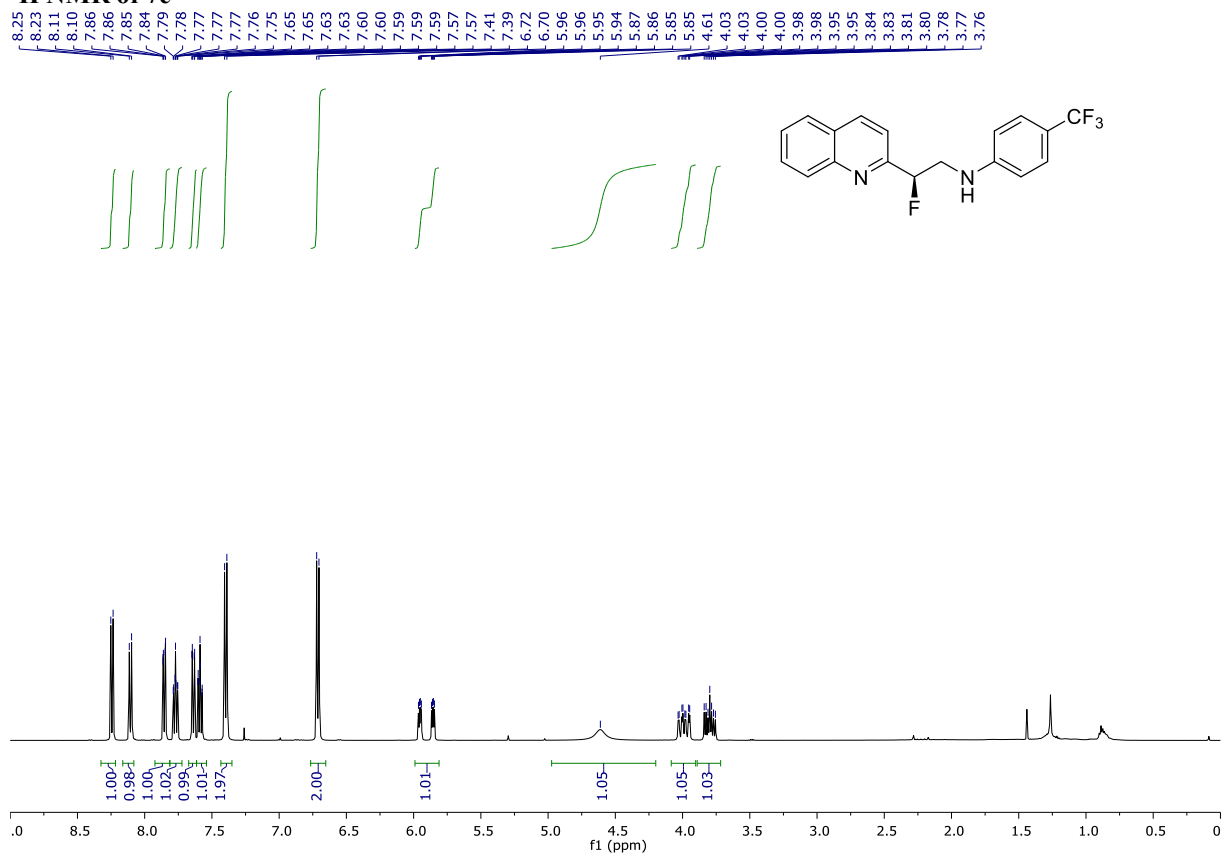
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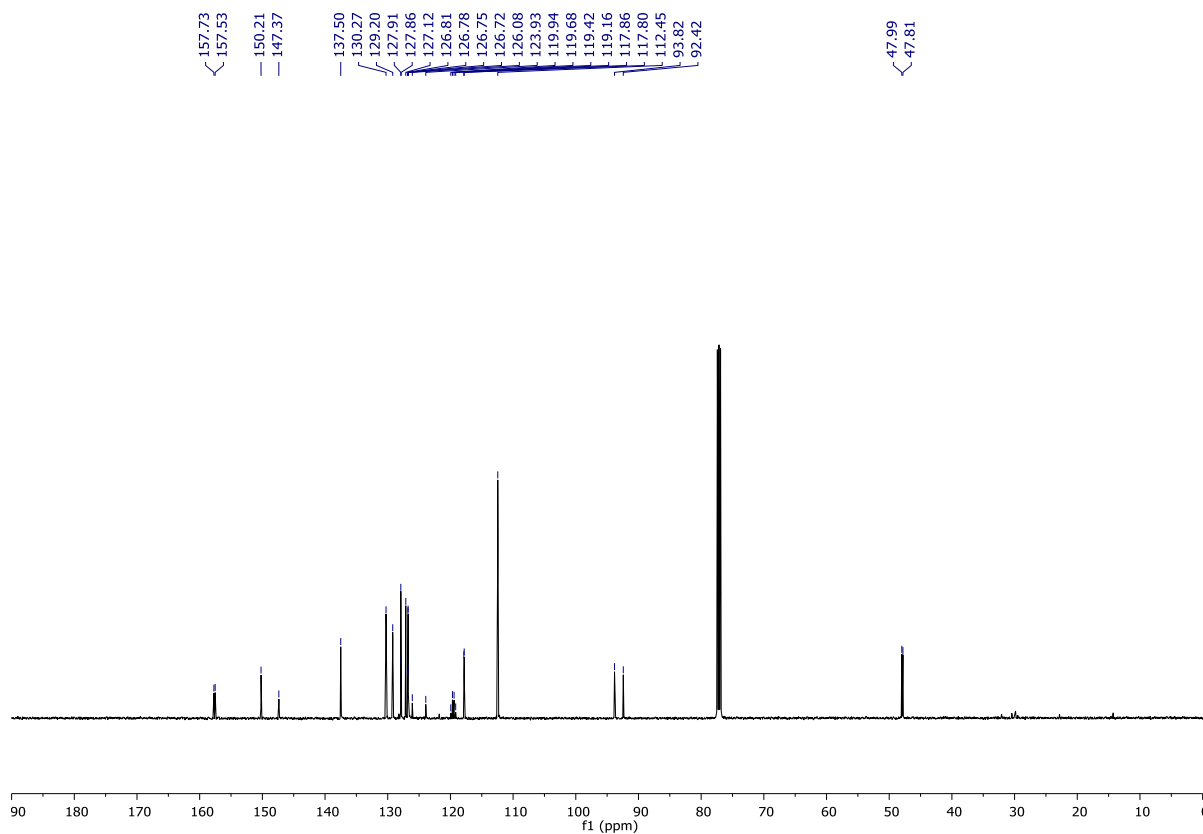
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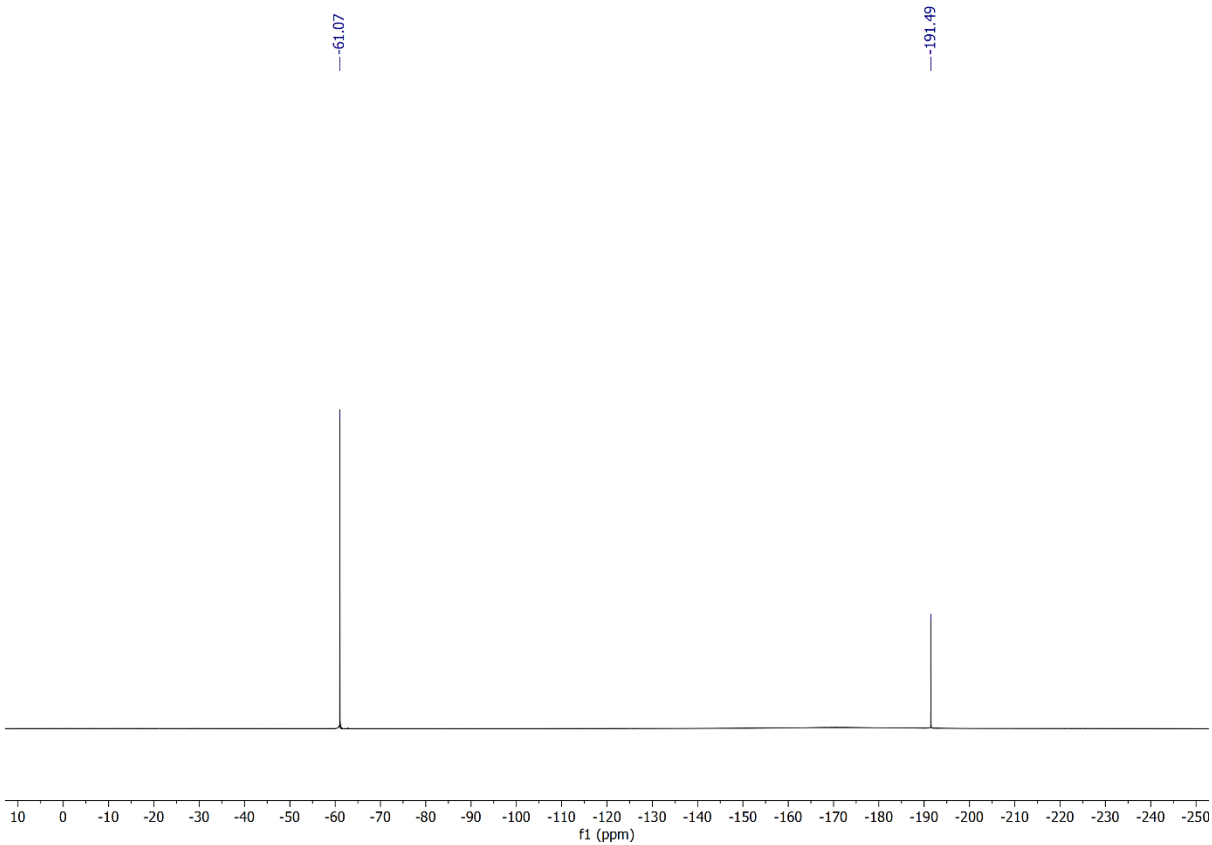
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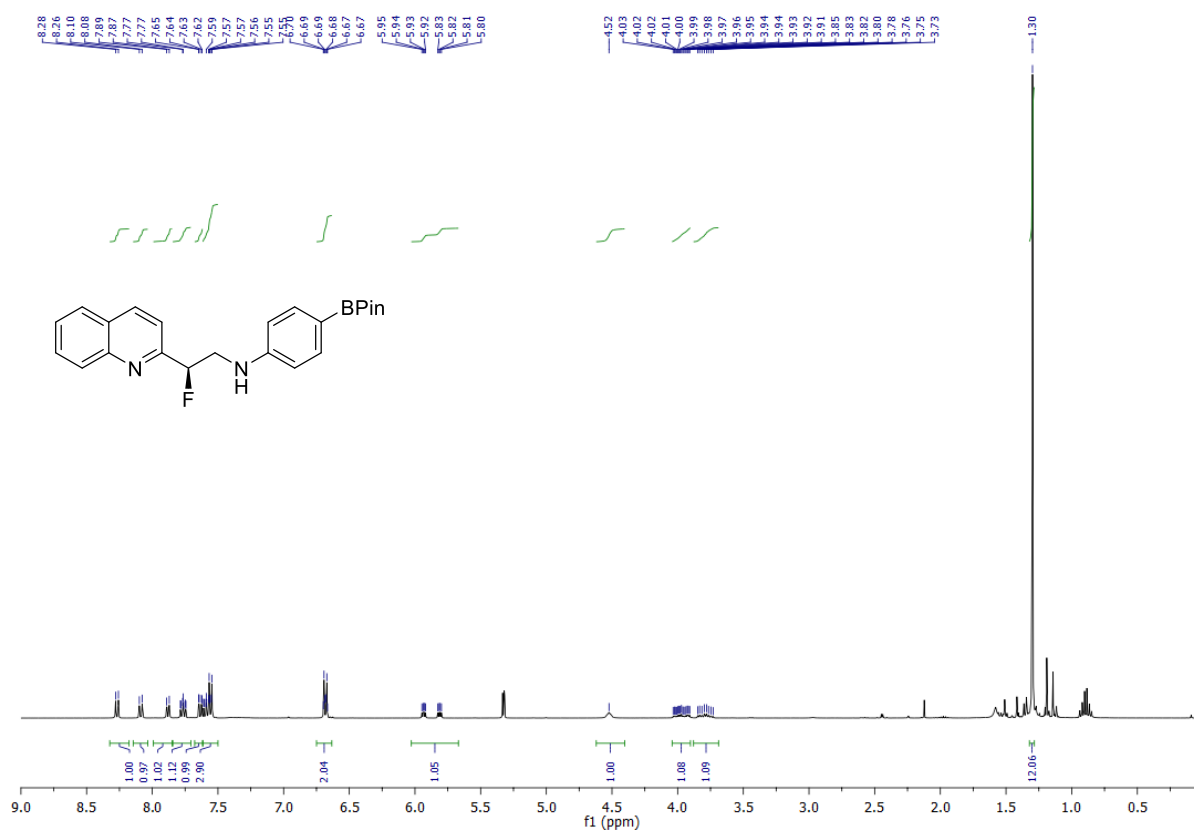
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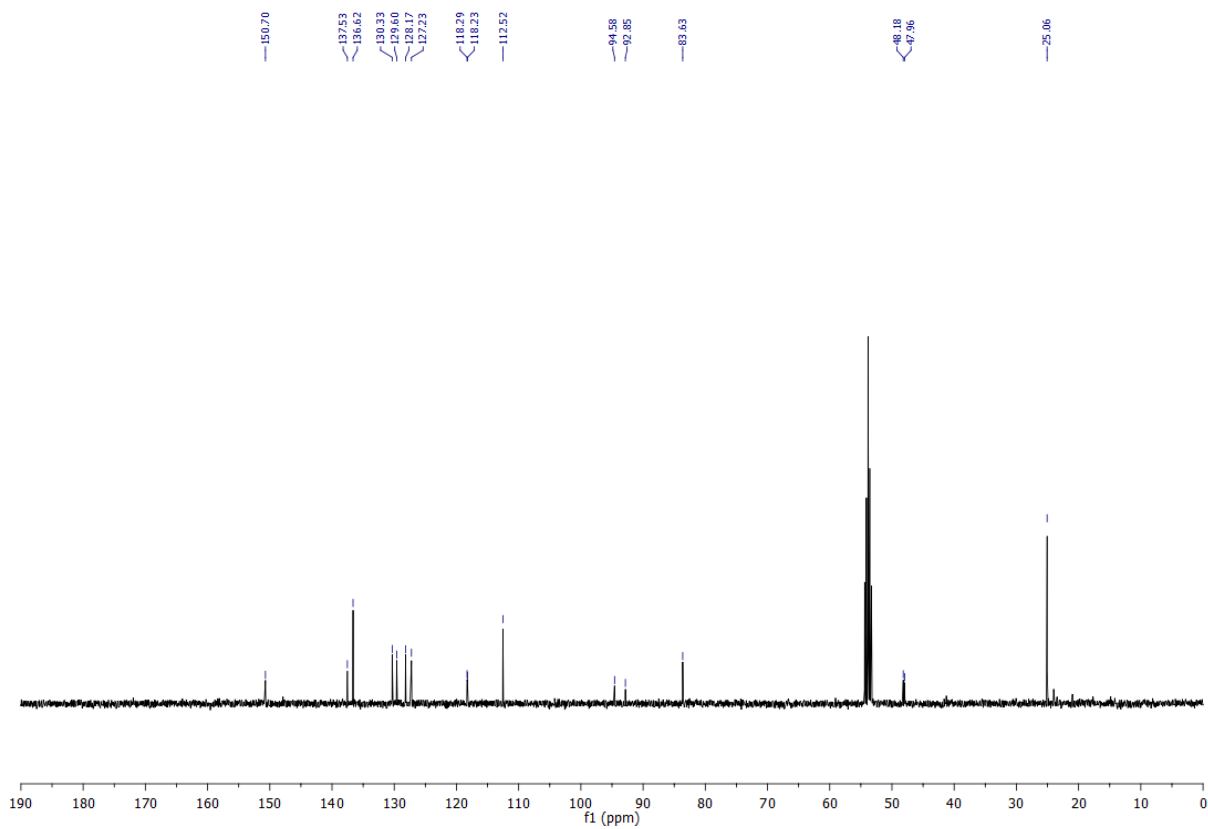
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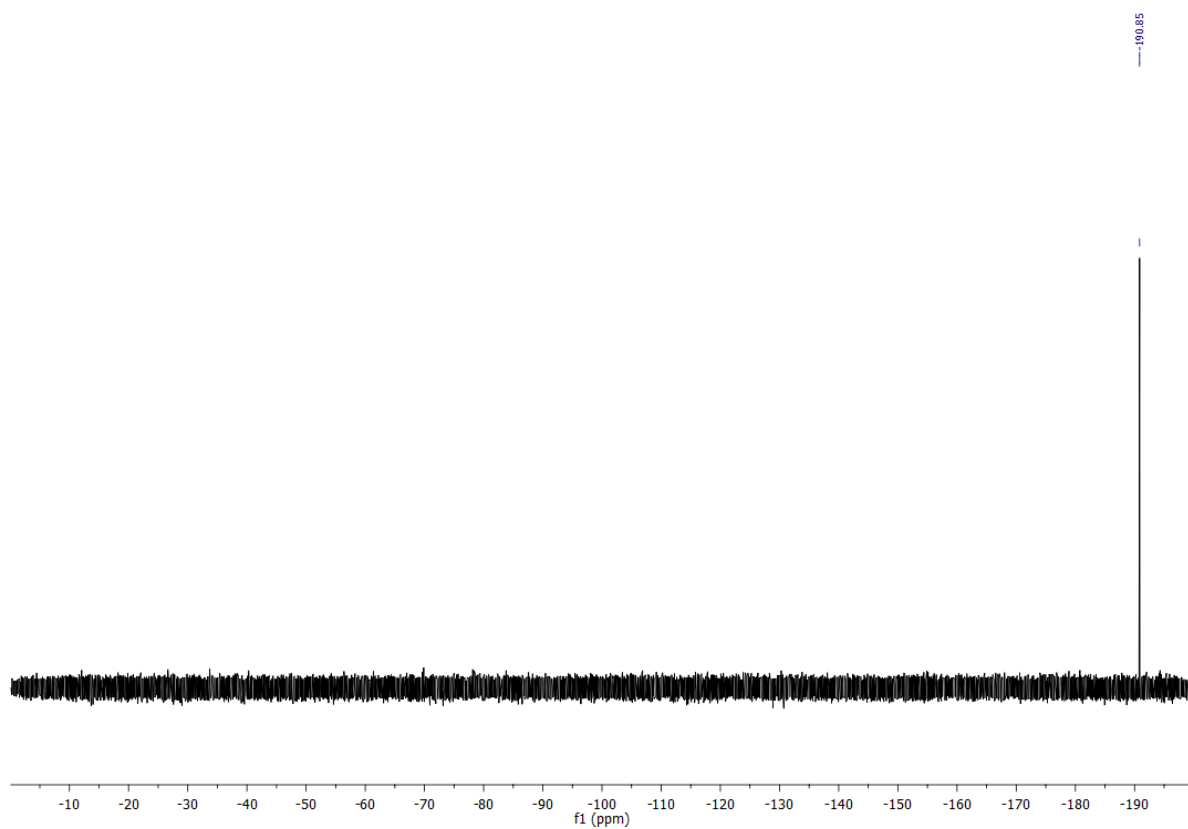
¹H NMR of 7f



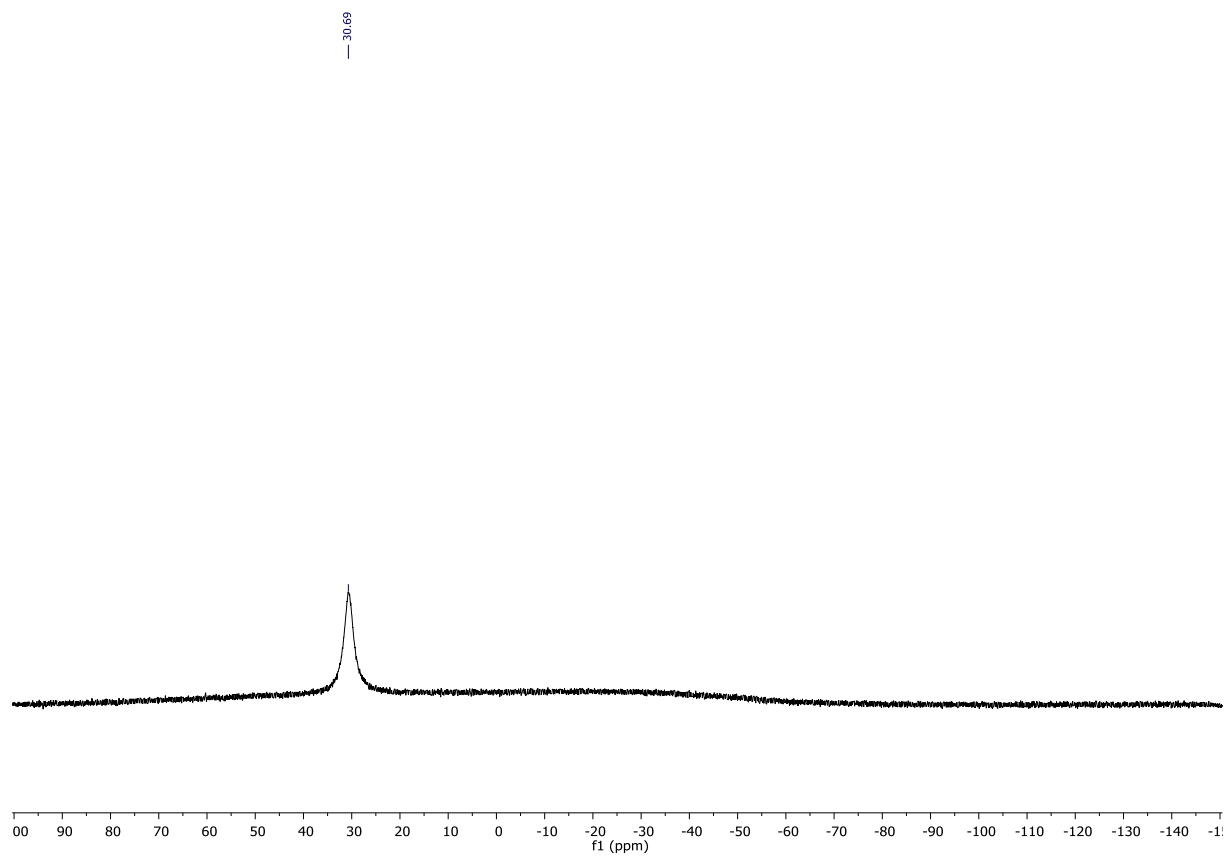
¹³C NMR of 7f



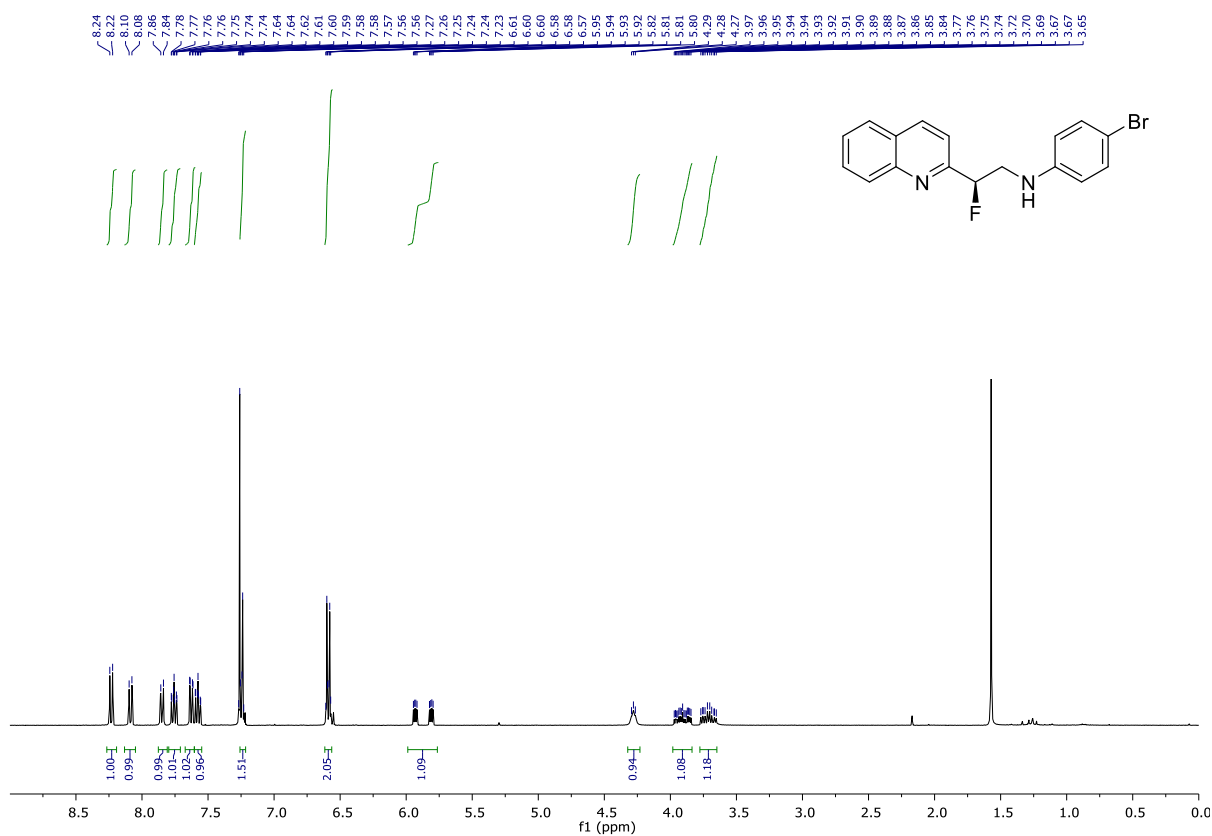
^{19}F NMR of 7f



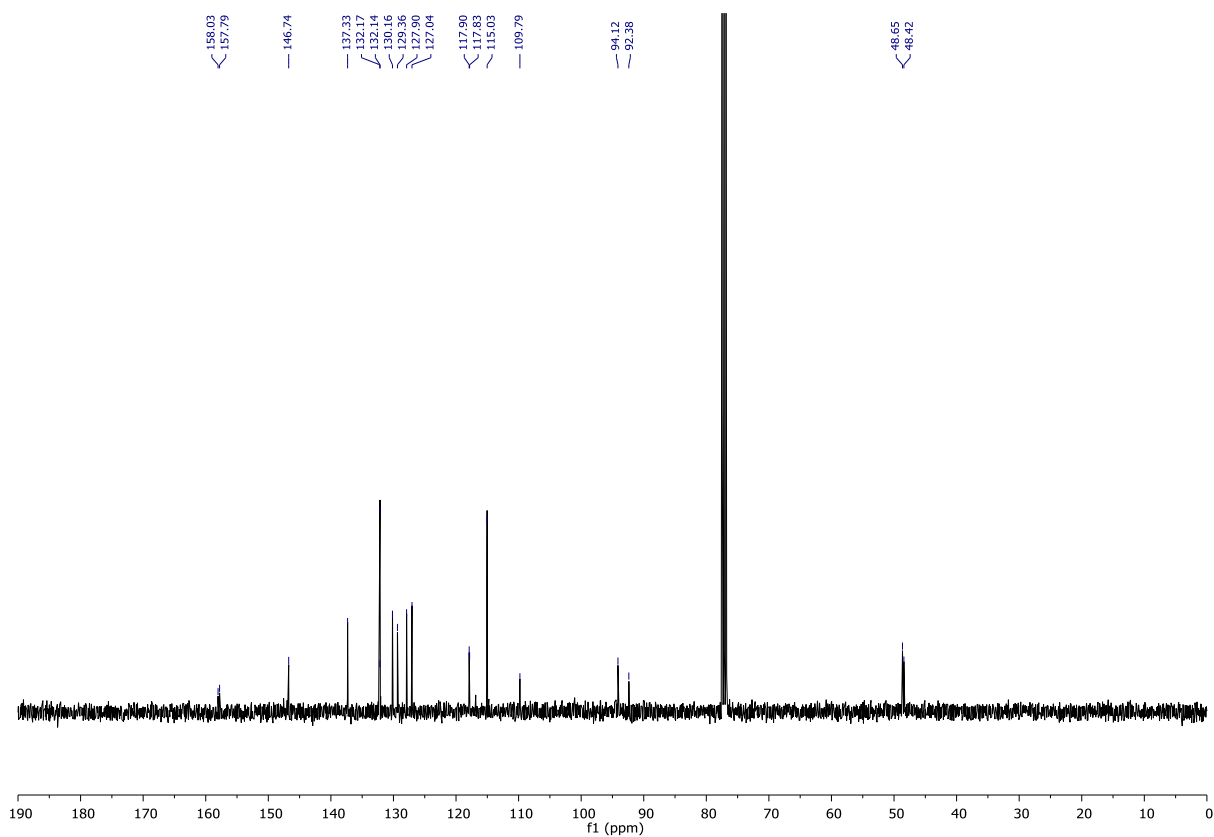
^{11}B NMR of 7f



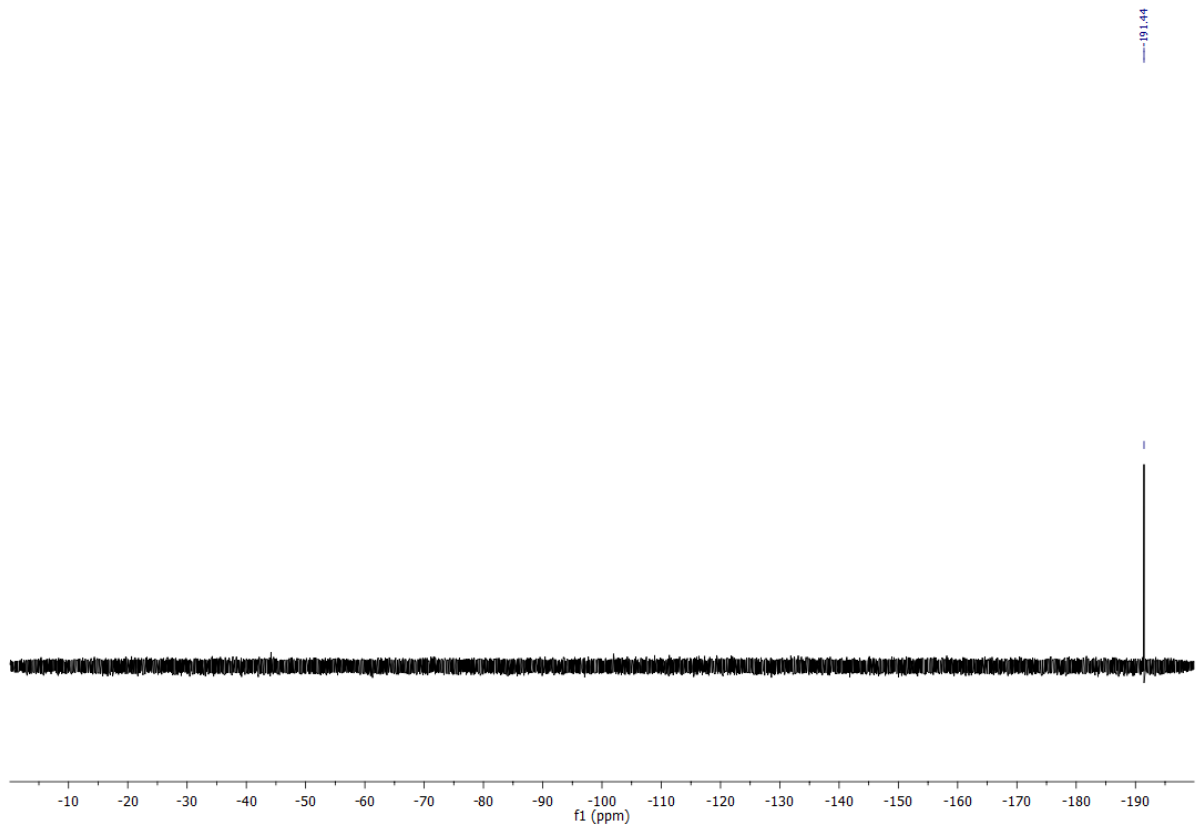
¹H NMR of 7g



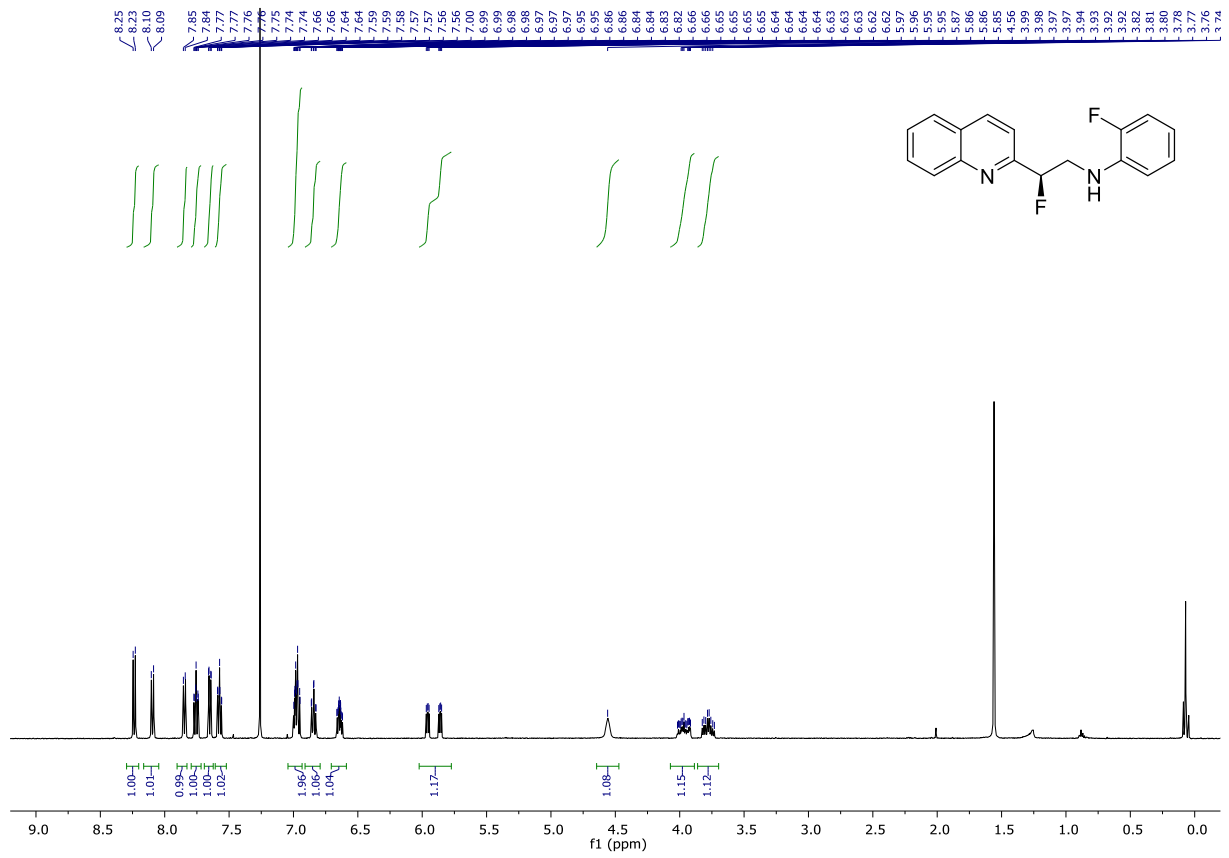
¹³C NMR of 7g



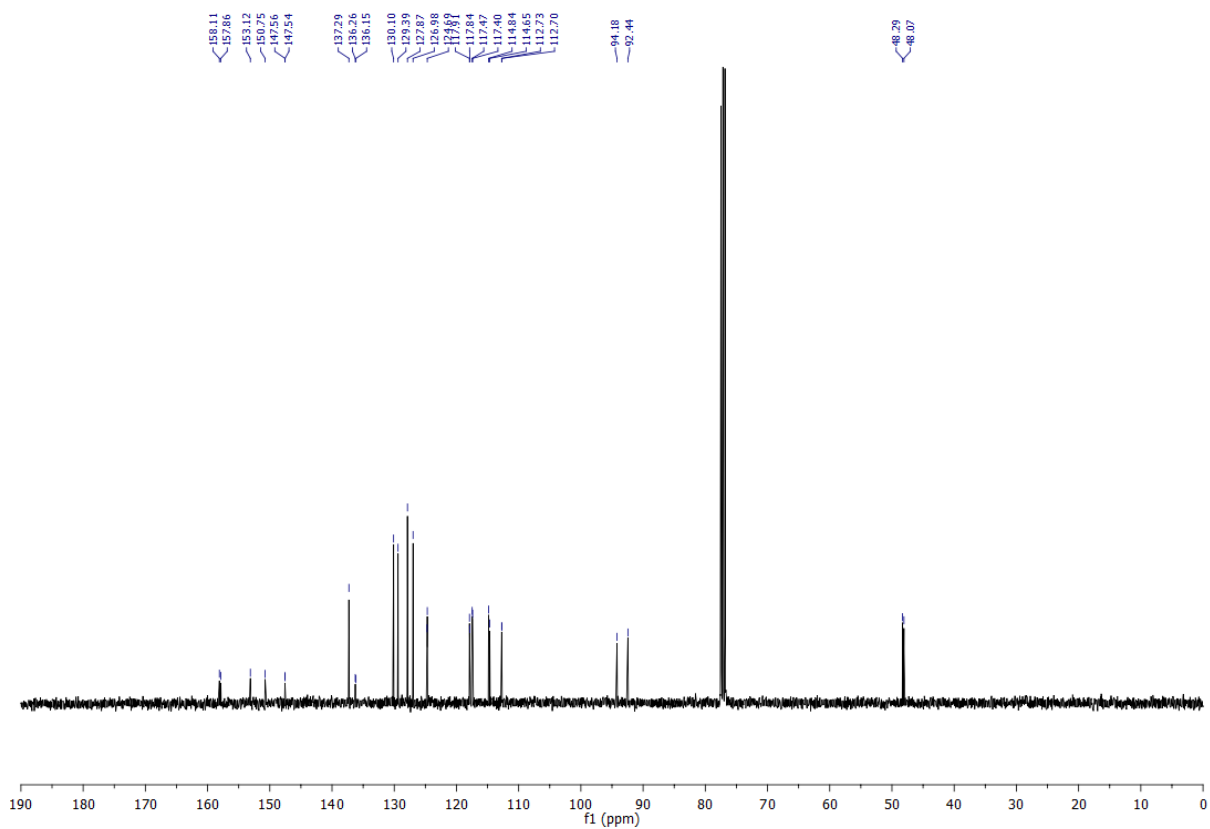
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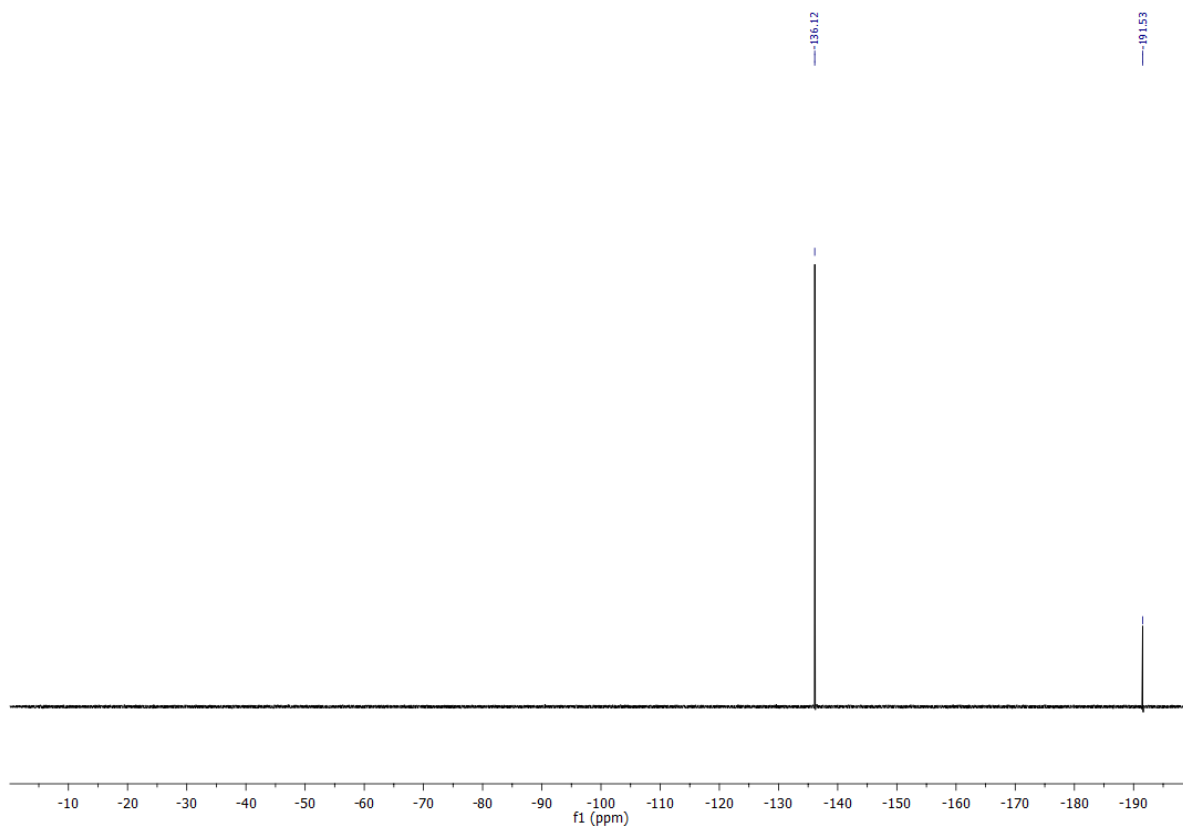
¹H NMR of 7h



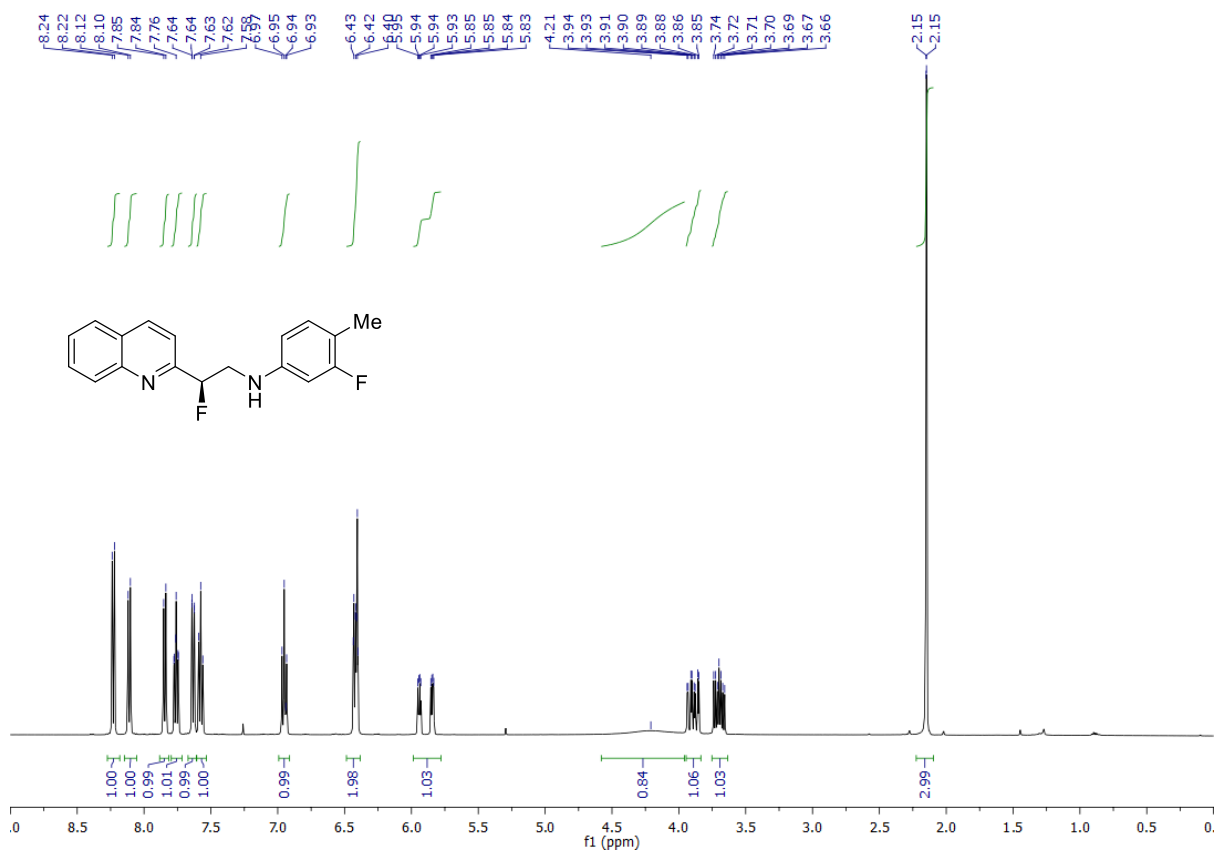
¹³C NMR of 7h



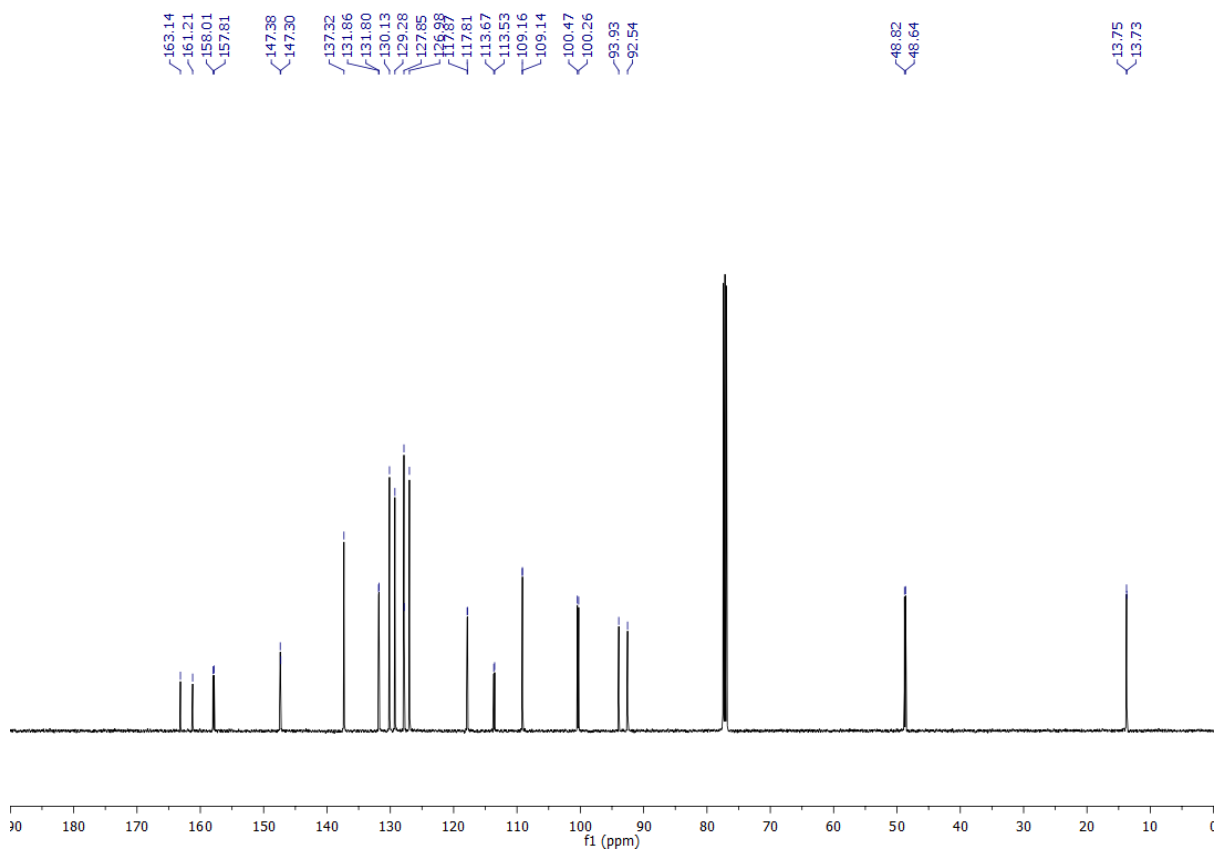
¹⁹F NMR of 7h



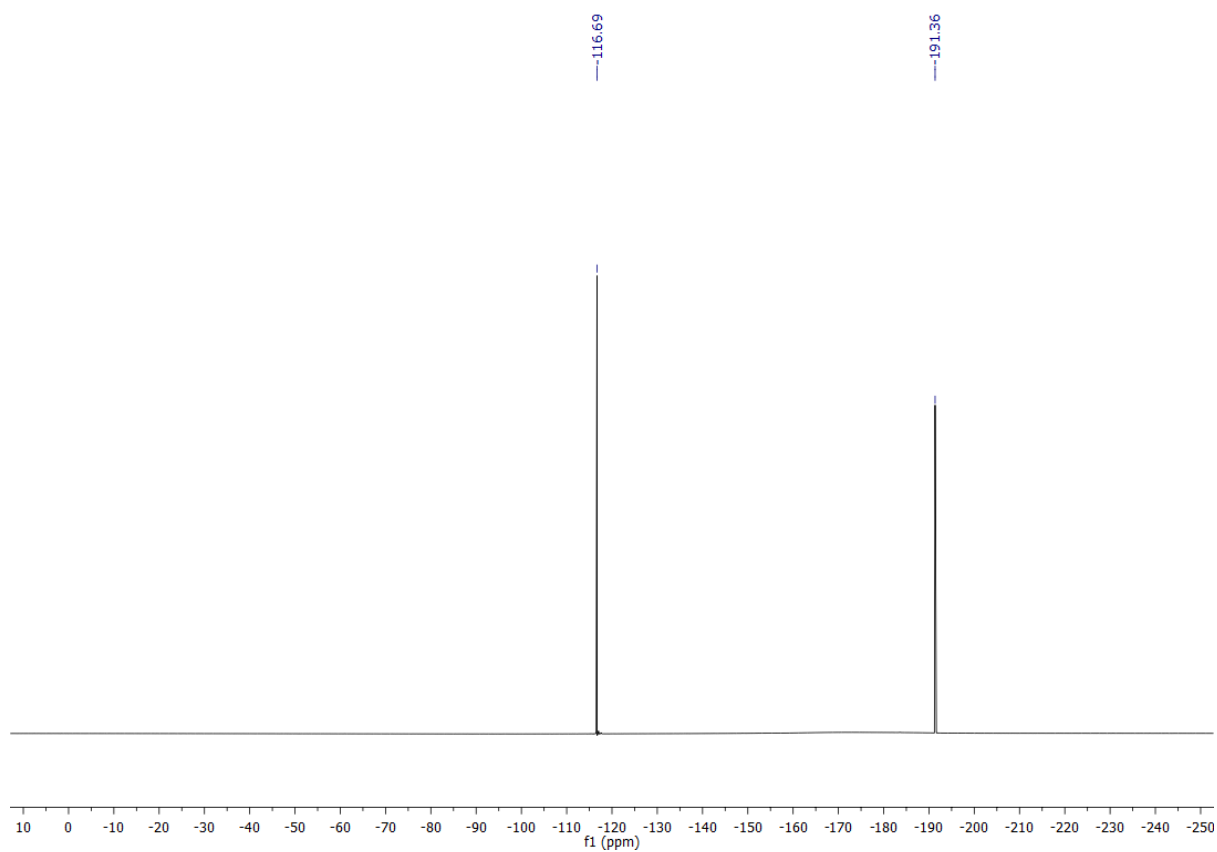
¹H NMR of 7i



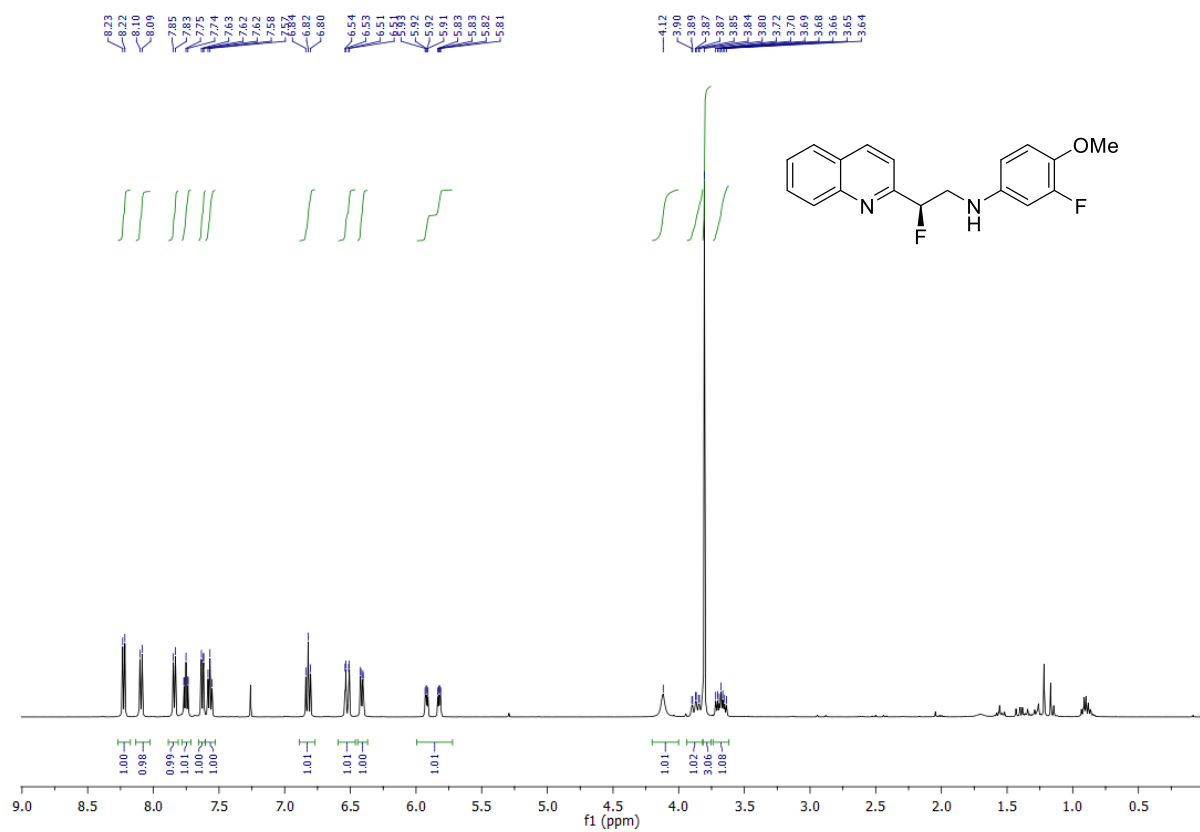
¹³C NMR of 7i



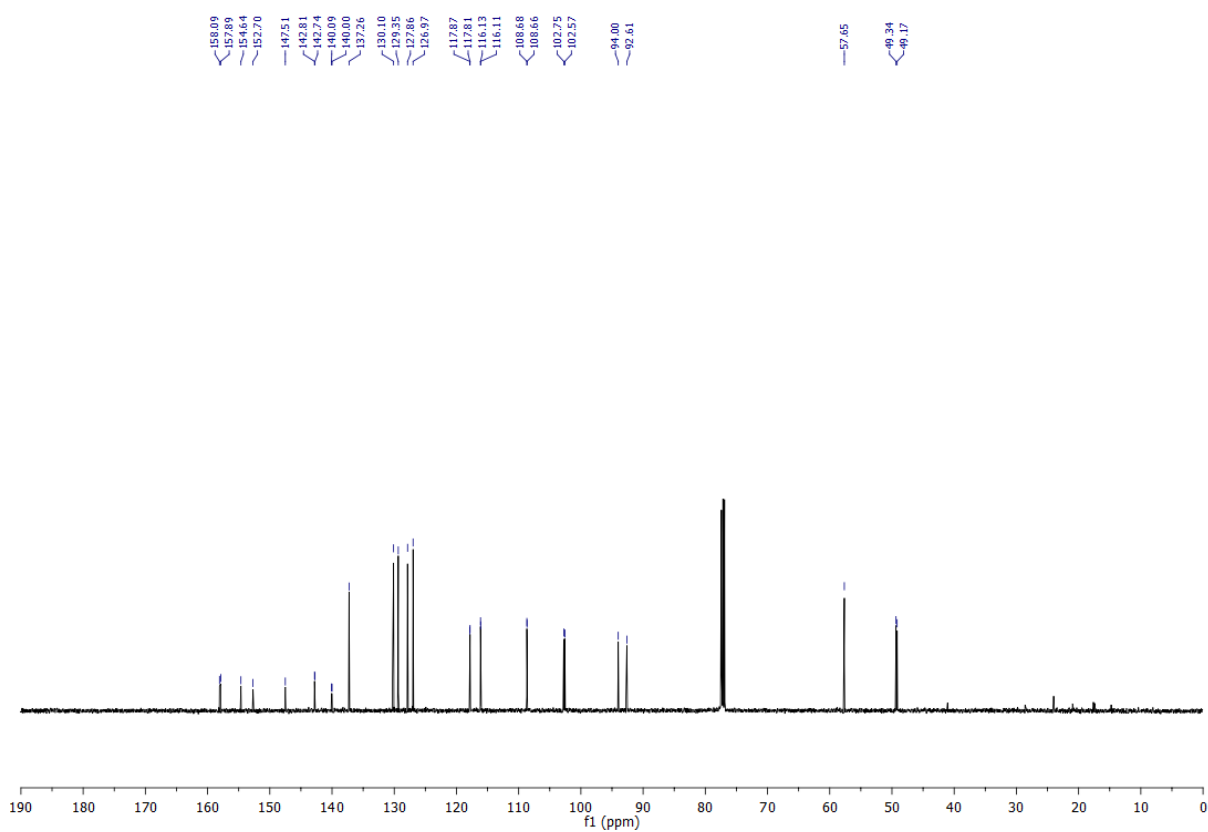
^{19}F NMR of 7i



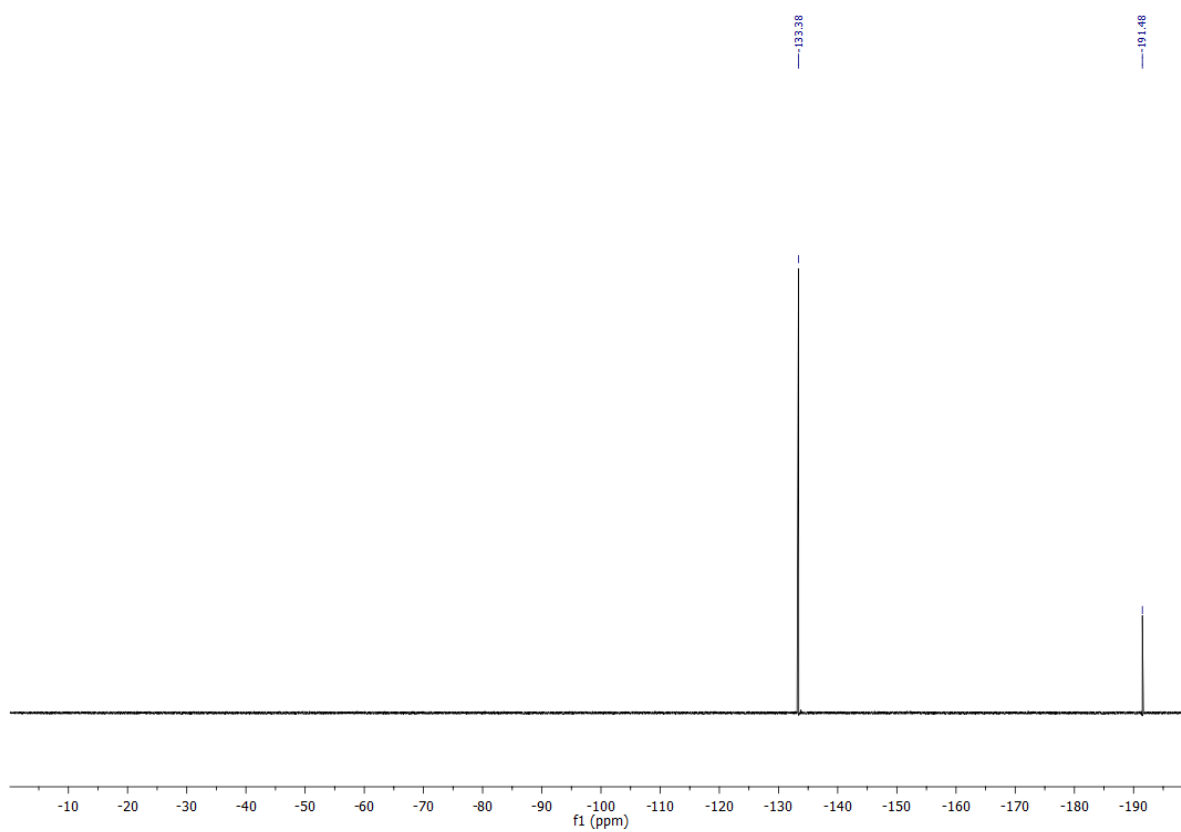
^1H NMR of 7j



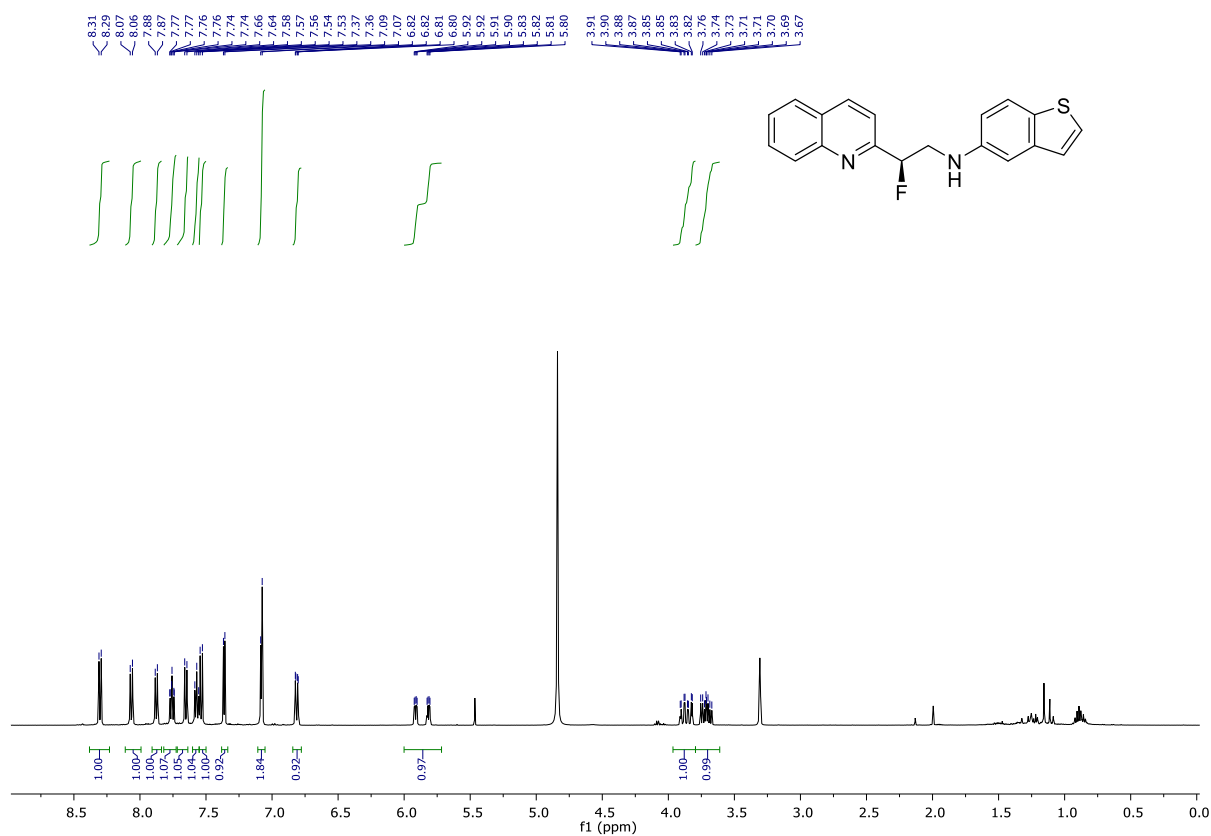
¹³C NMR of 7j



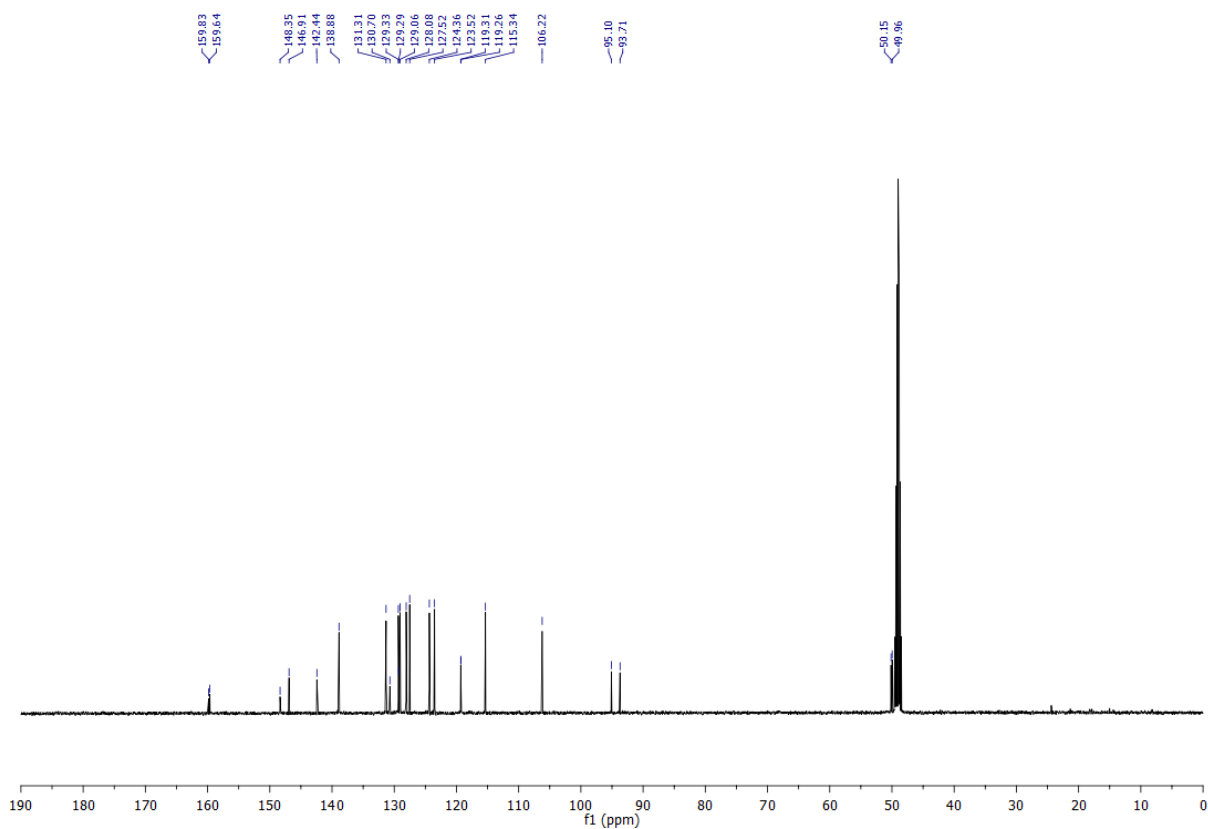
¹⁹F NMR of 7j



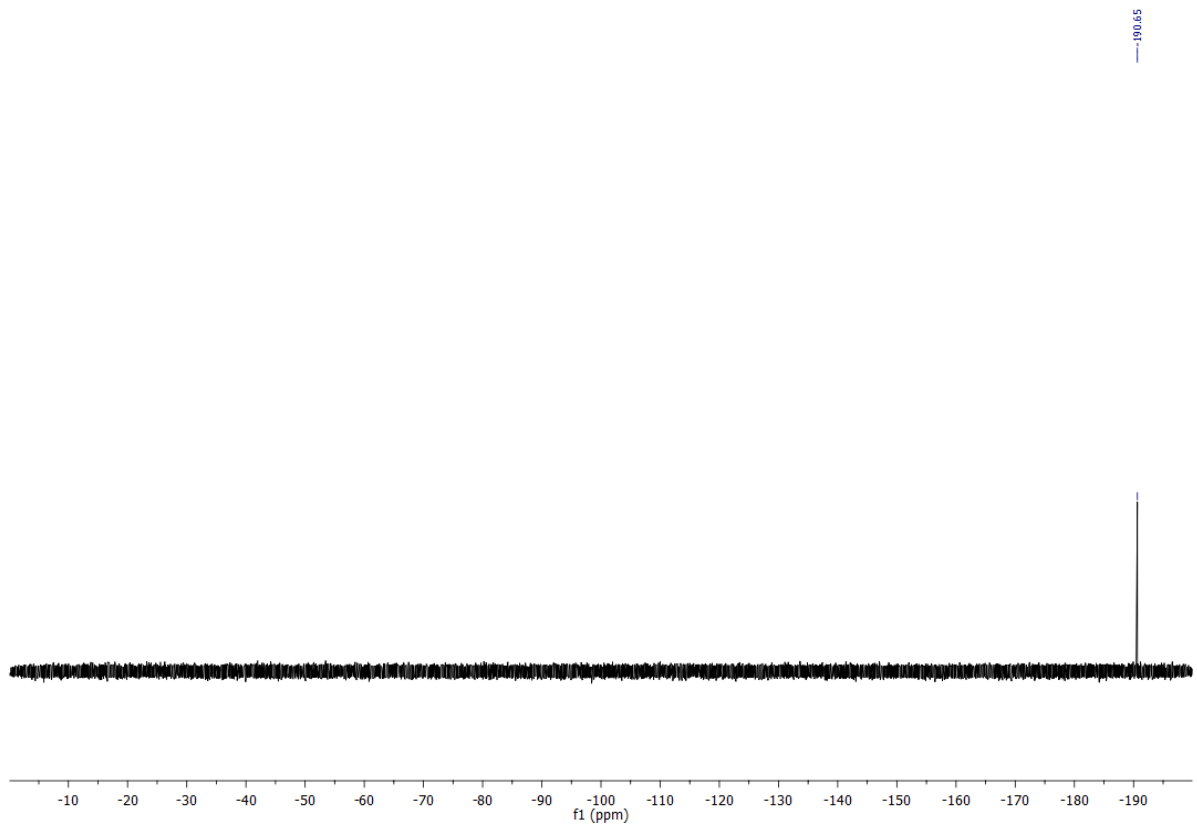
¹H NMR of 7k



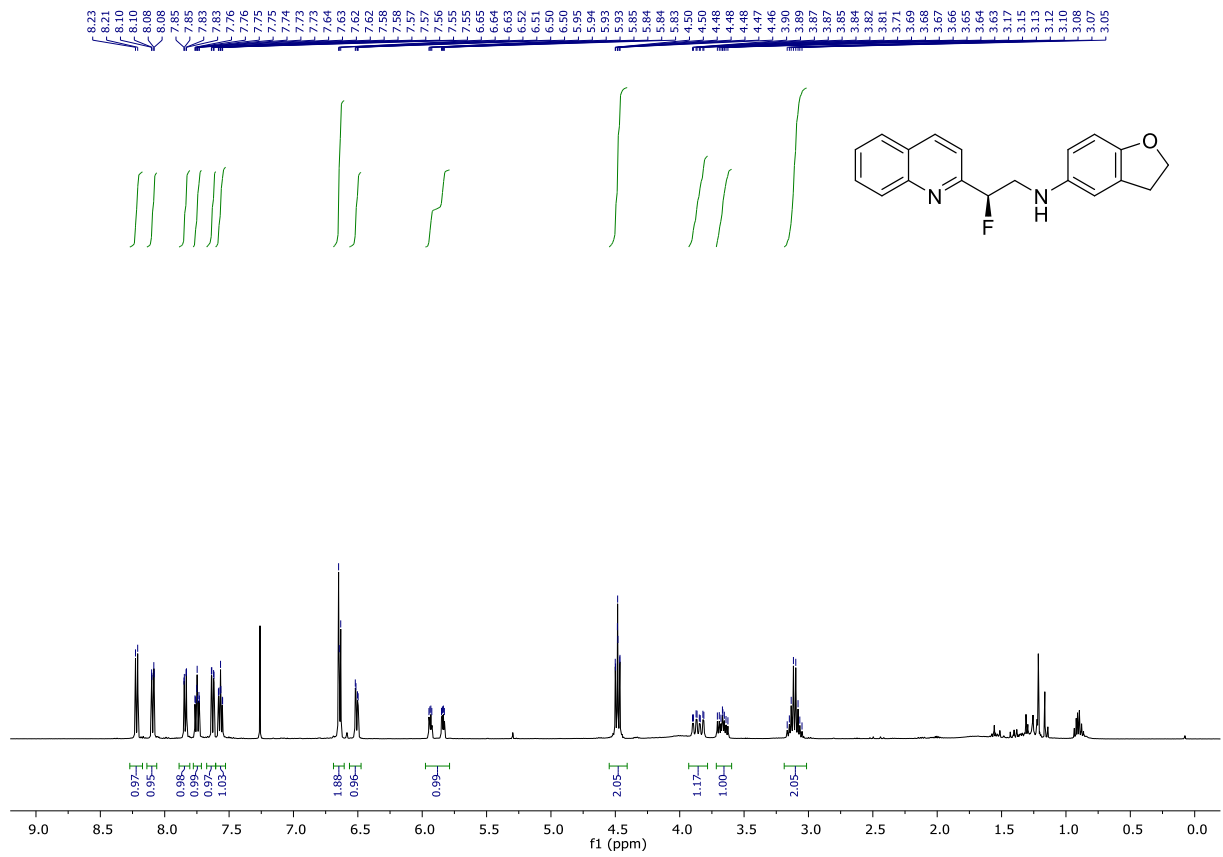
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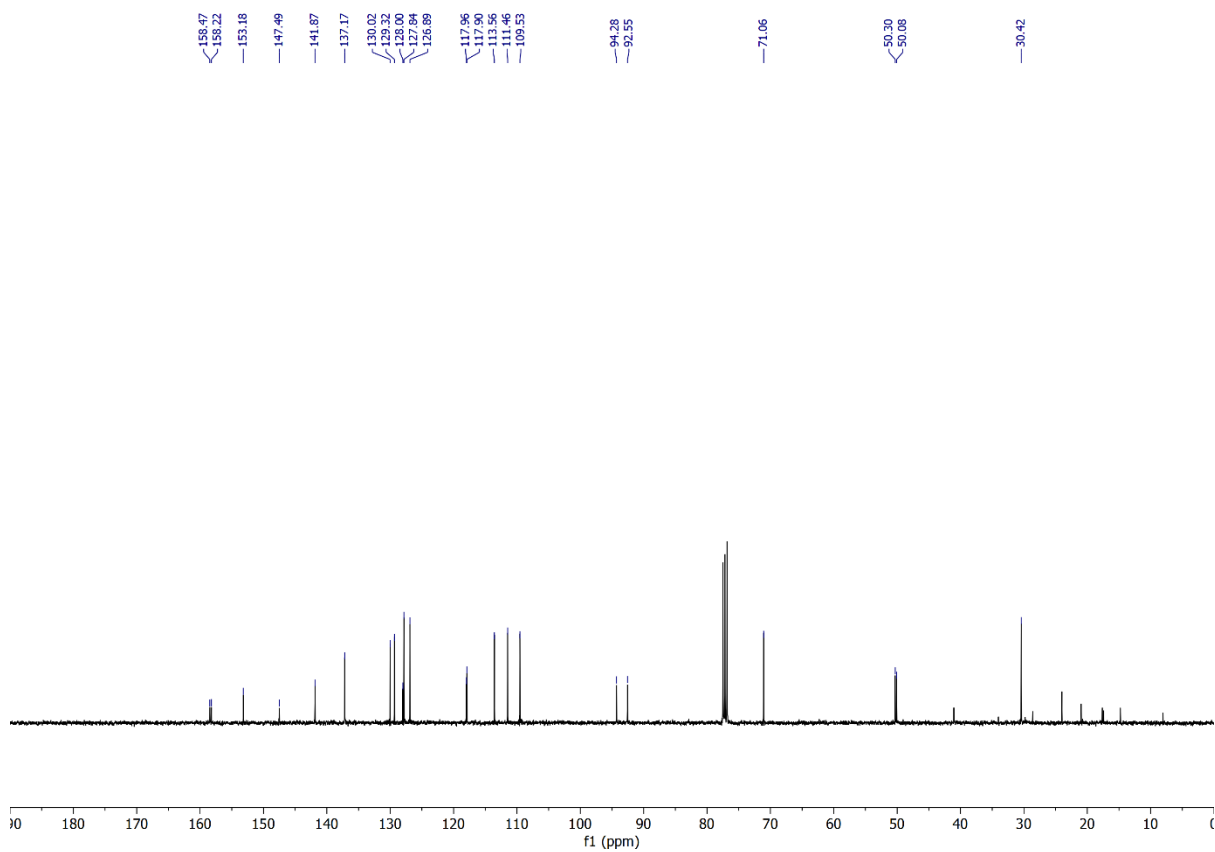
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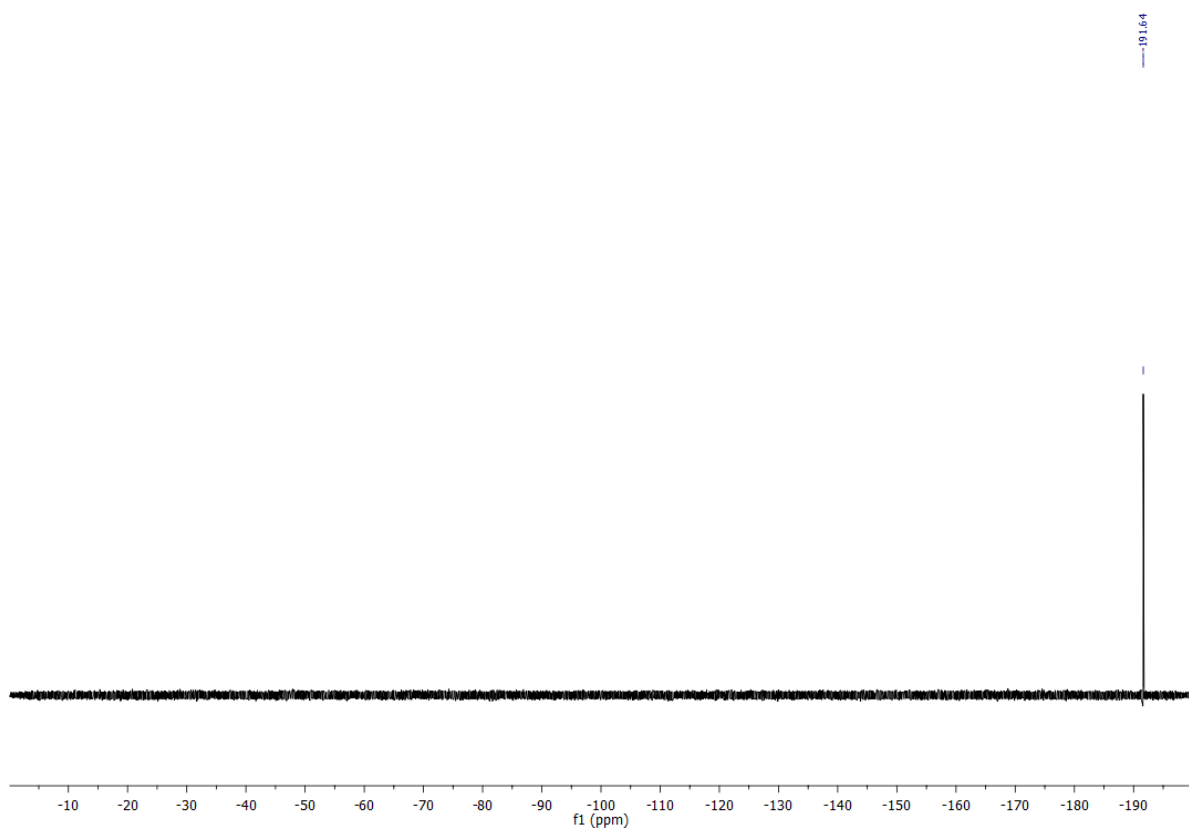
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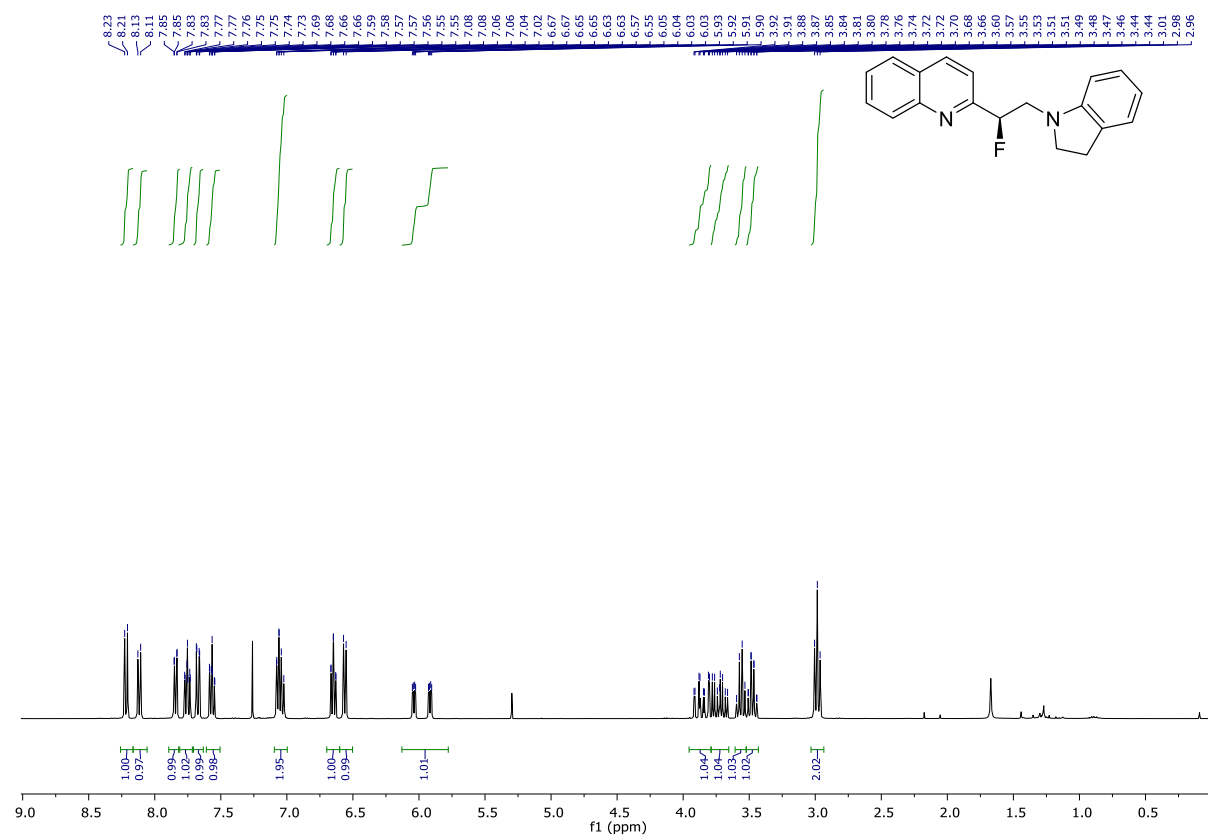
¹³C NMR of 7l



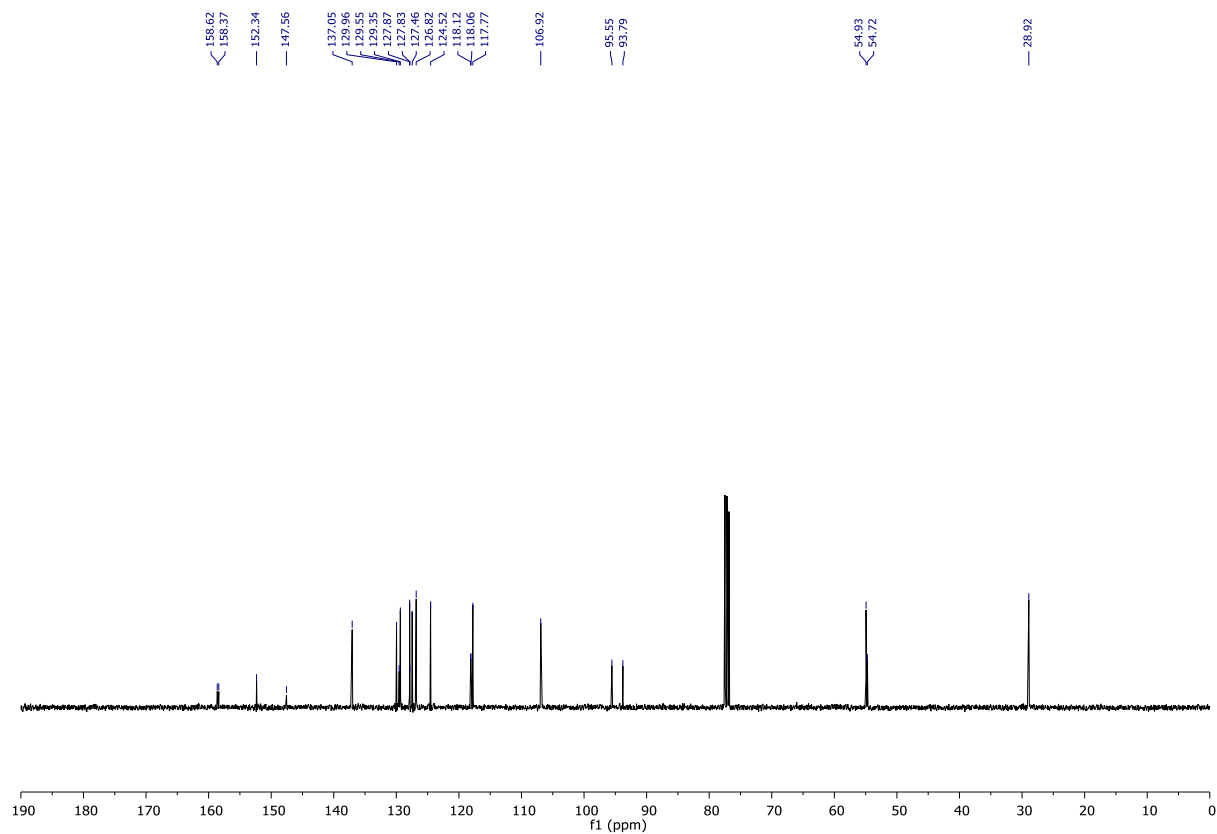
¹⁹F NMR of 7l



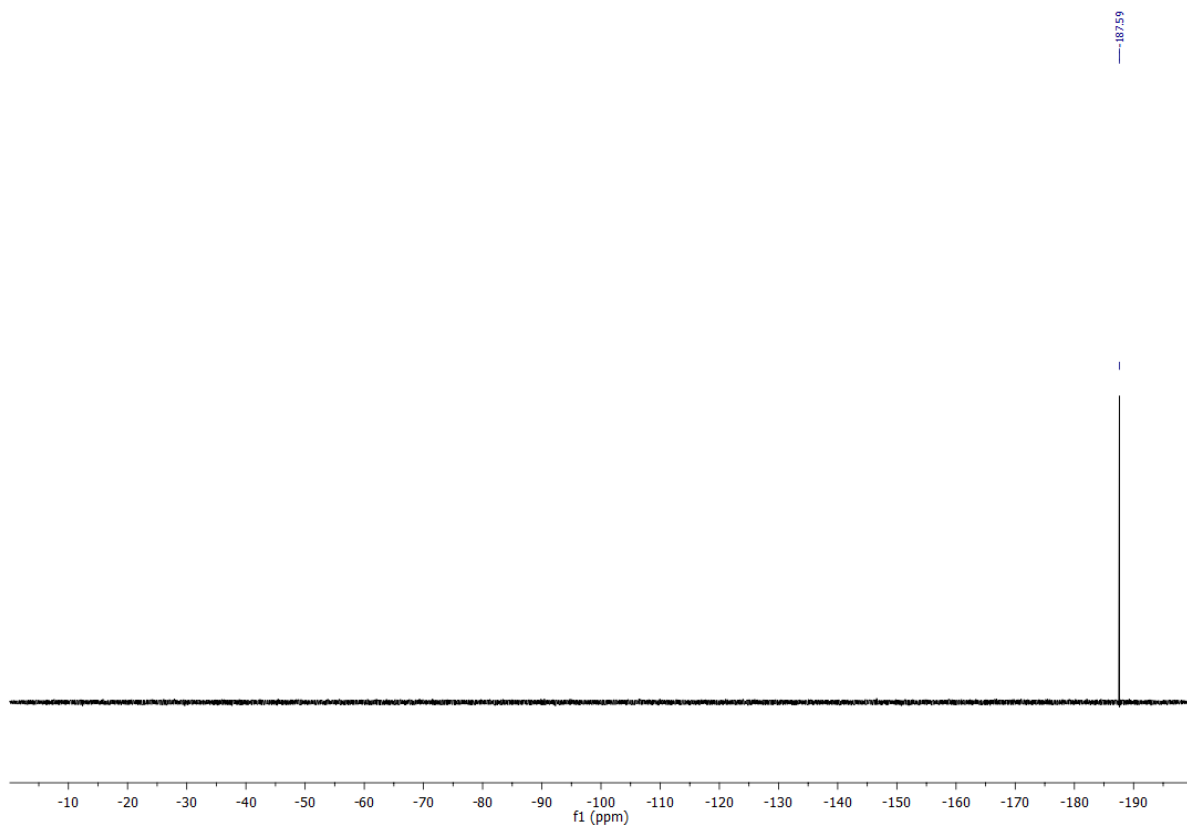
¹H NMR of 7m



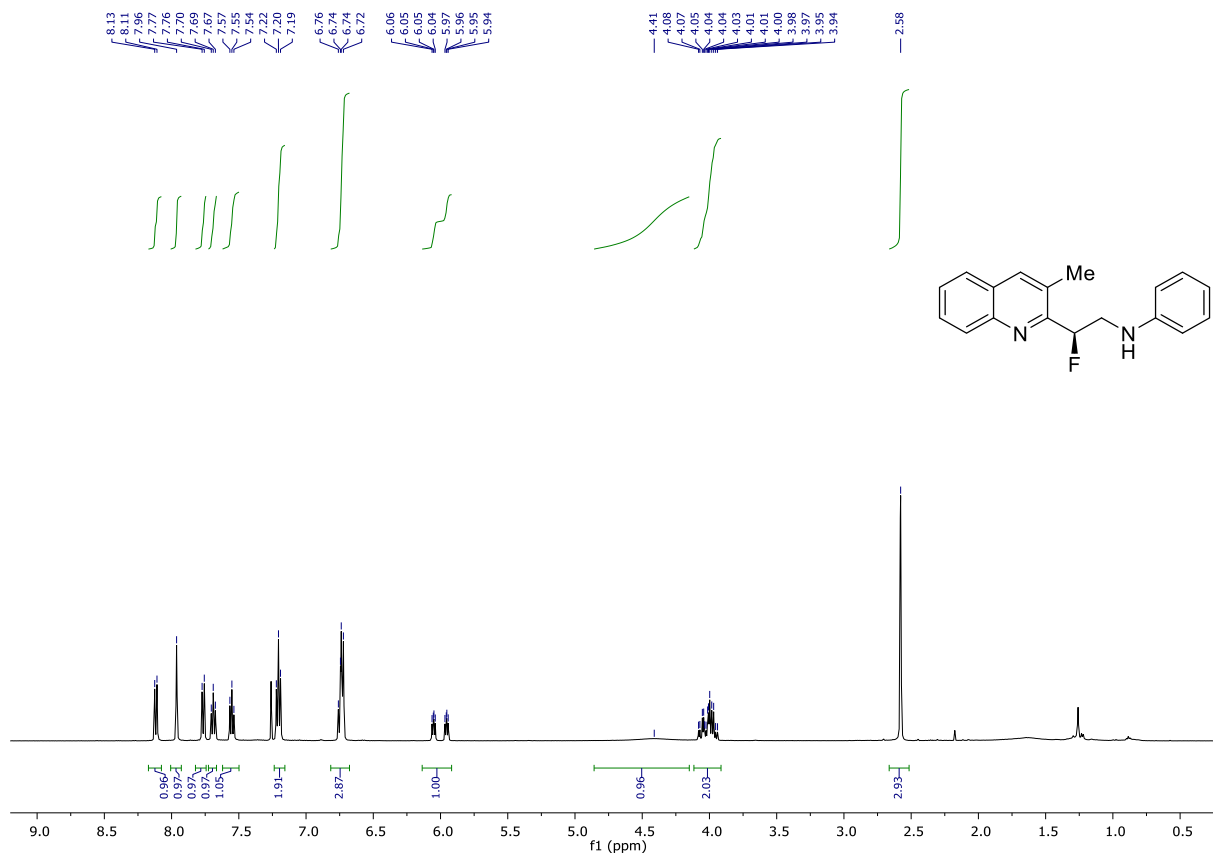
¹³C NMR of 7m



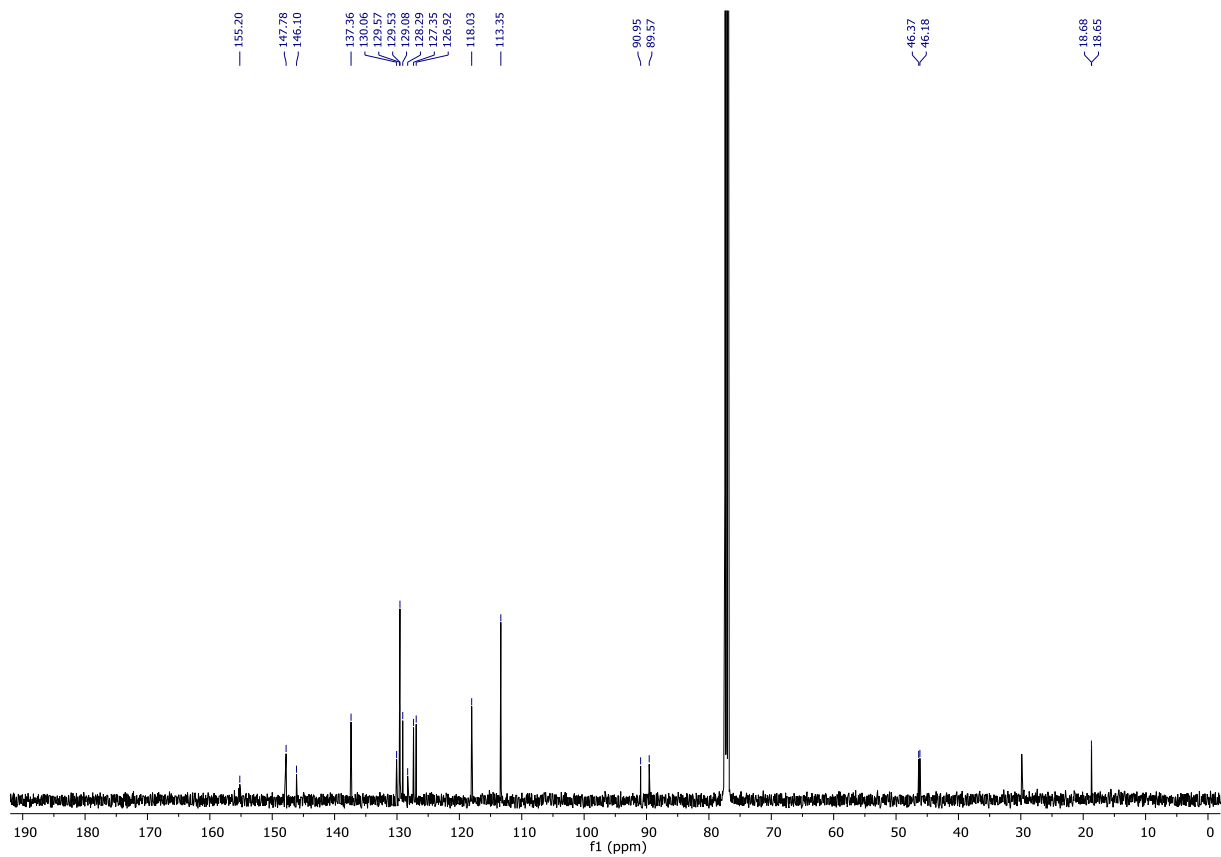
^{19}F NMR of 7m



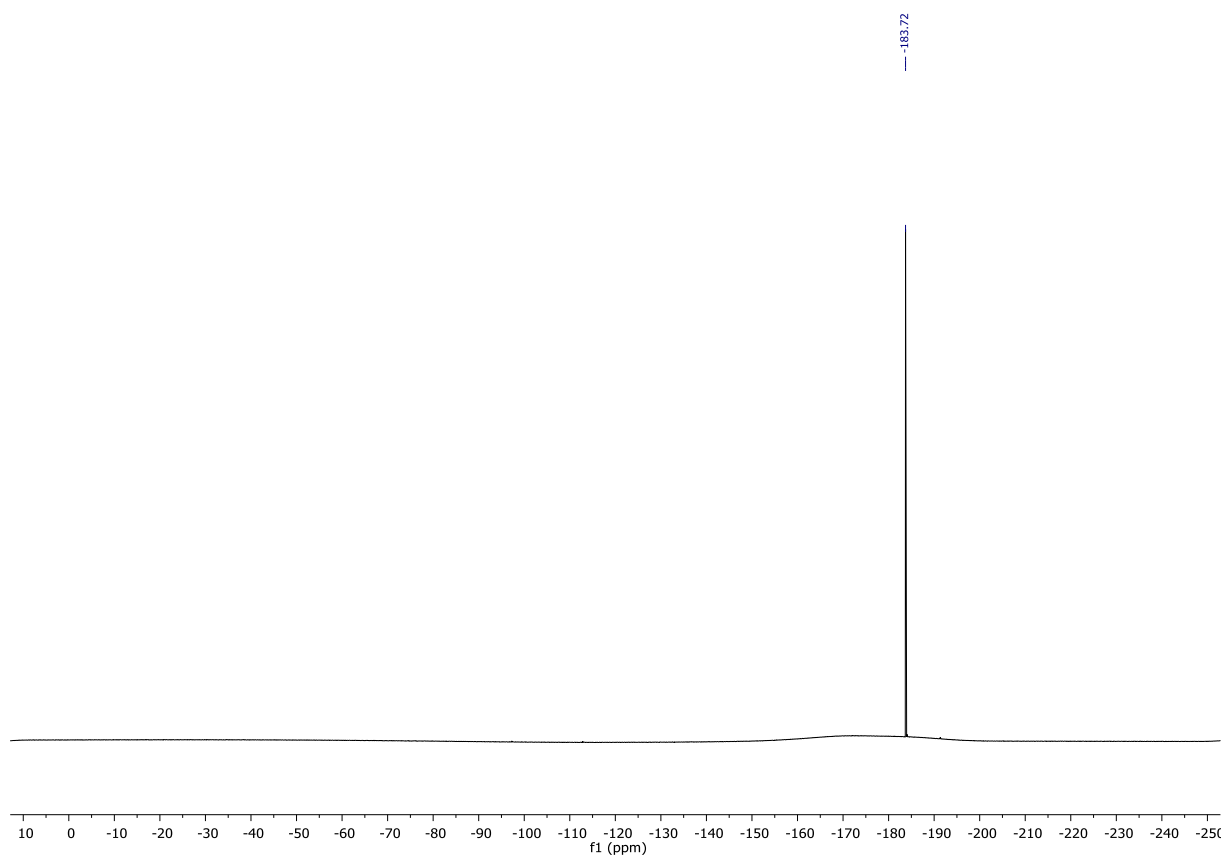
^1H NMR of 7n



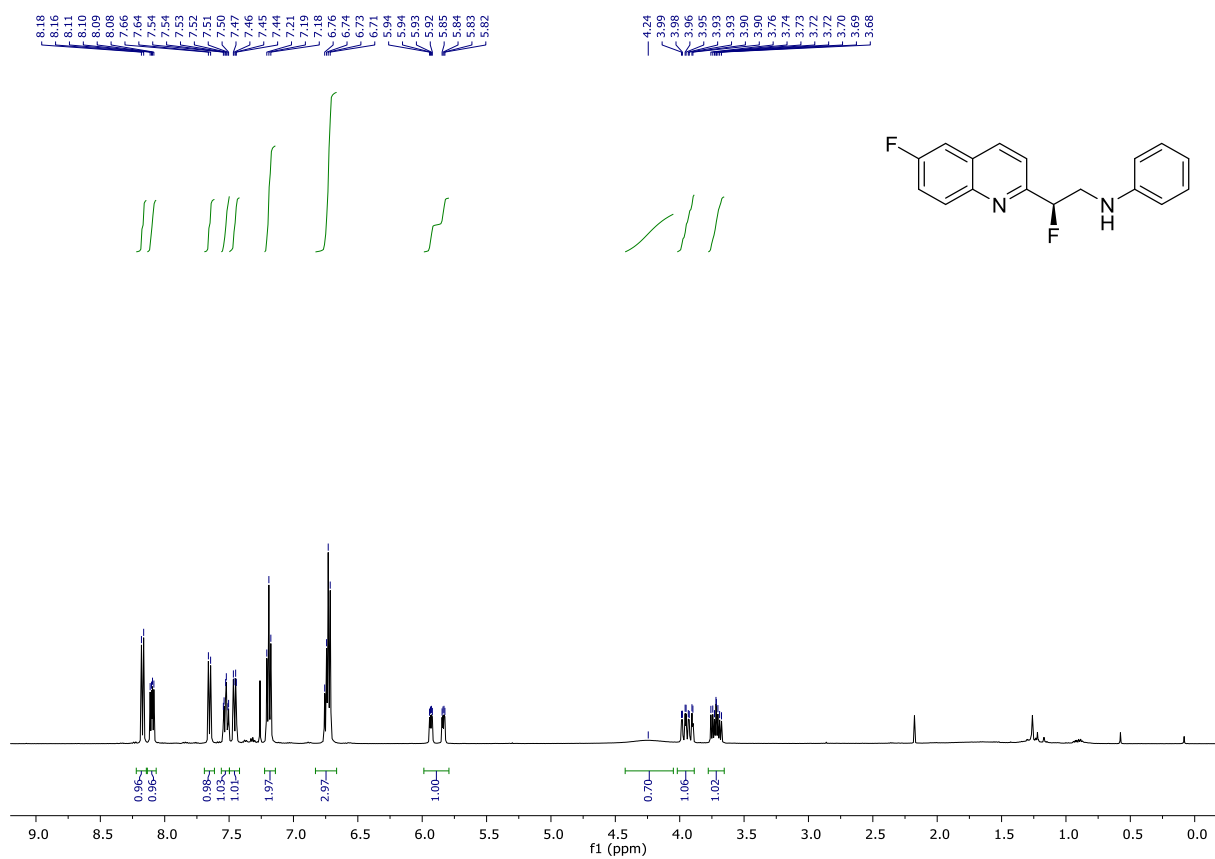
¹³C NMR of 7n



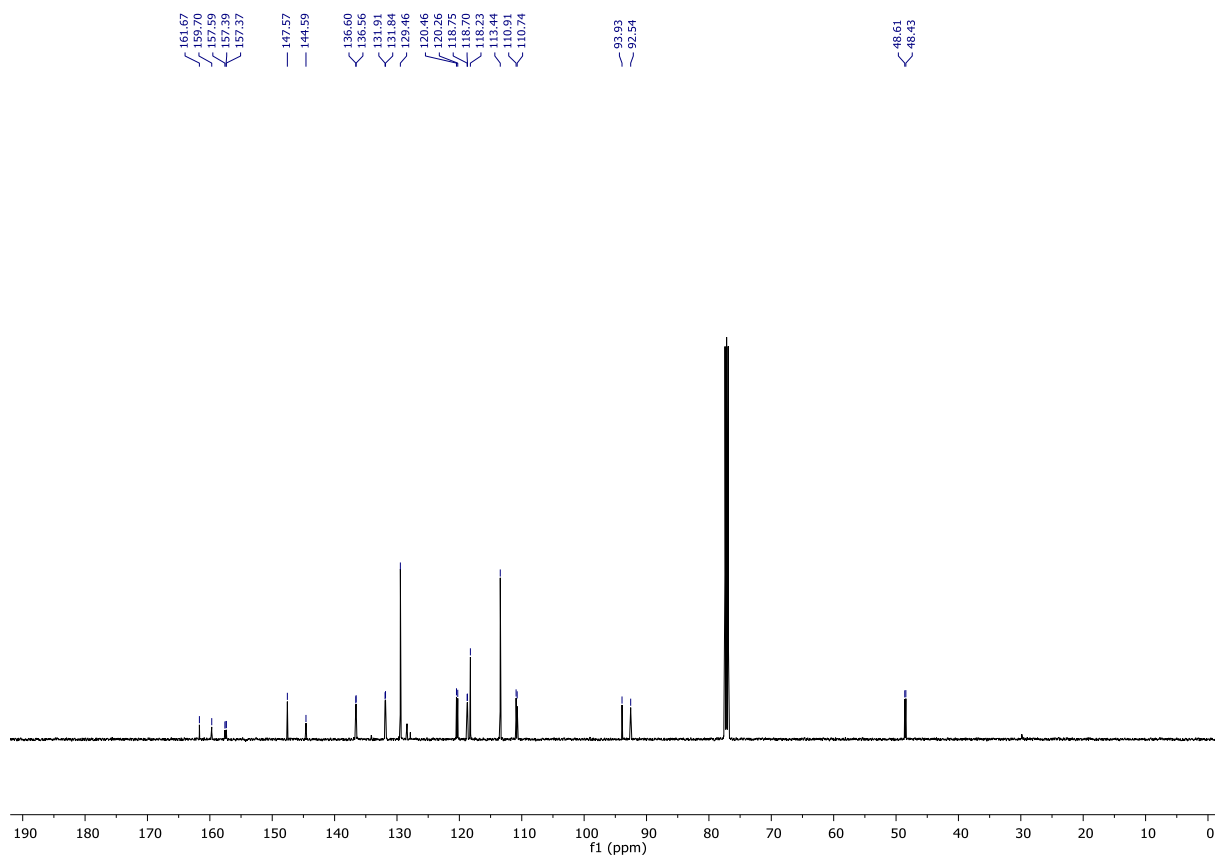
¹⁹F NMR of 7n



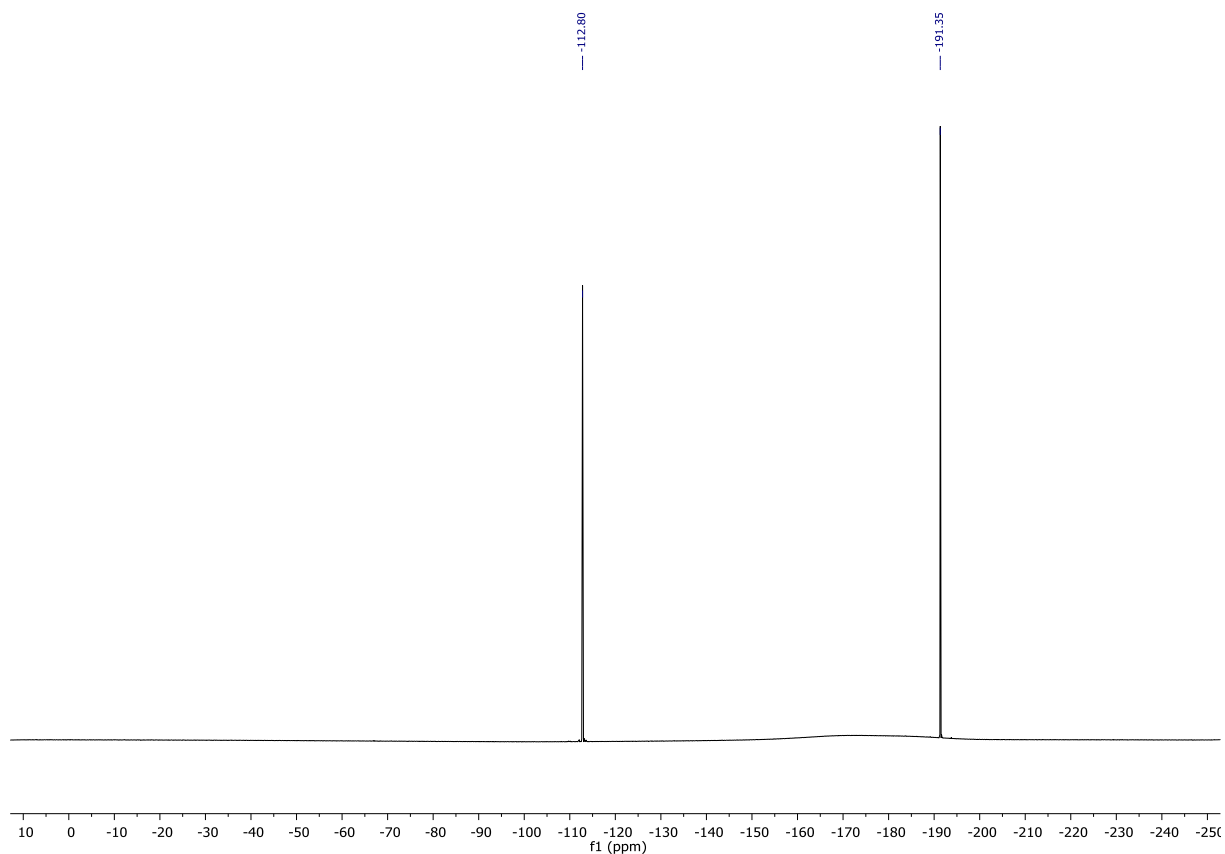
¹H NMR of 7o



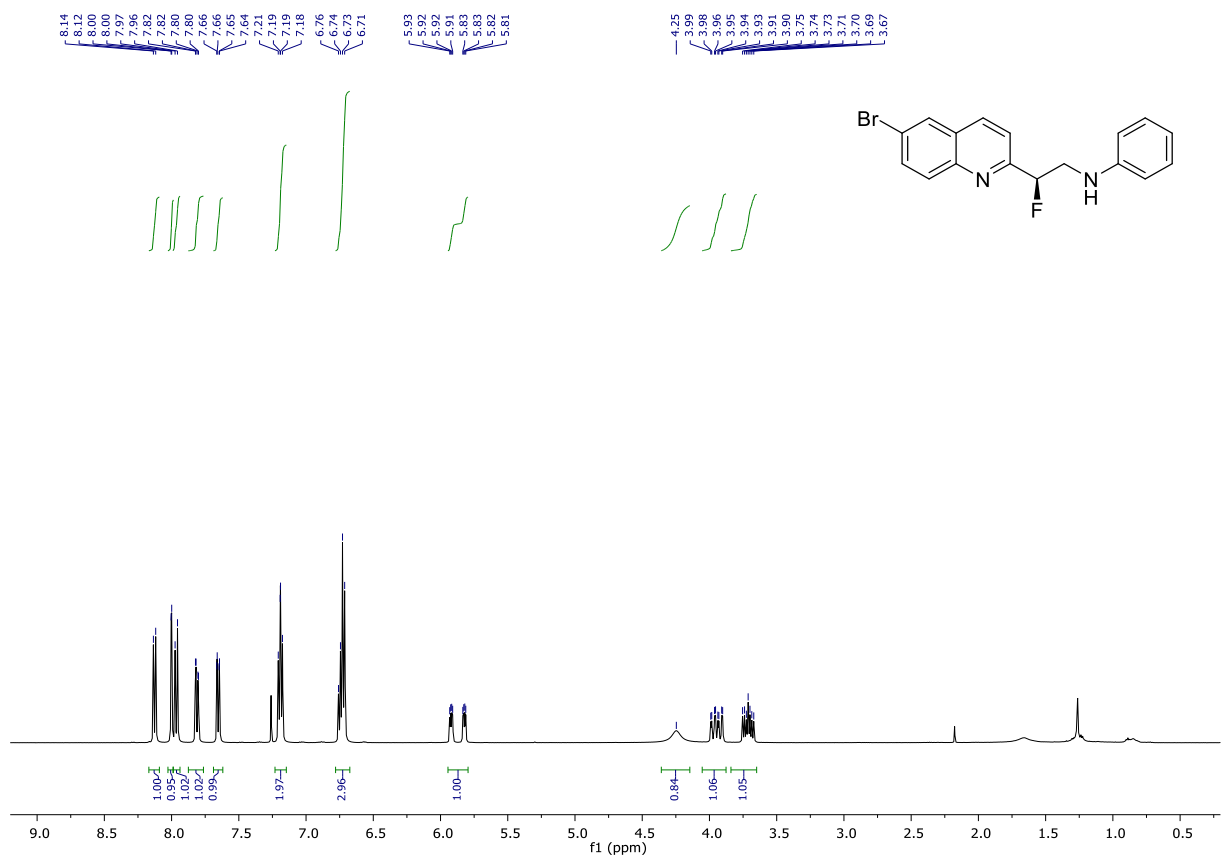
¹³C NMR of 7o



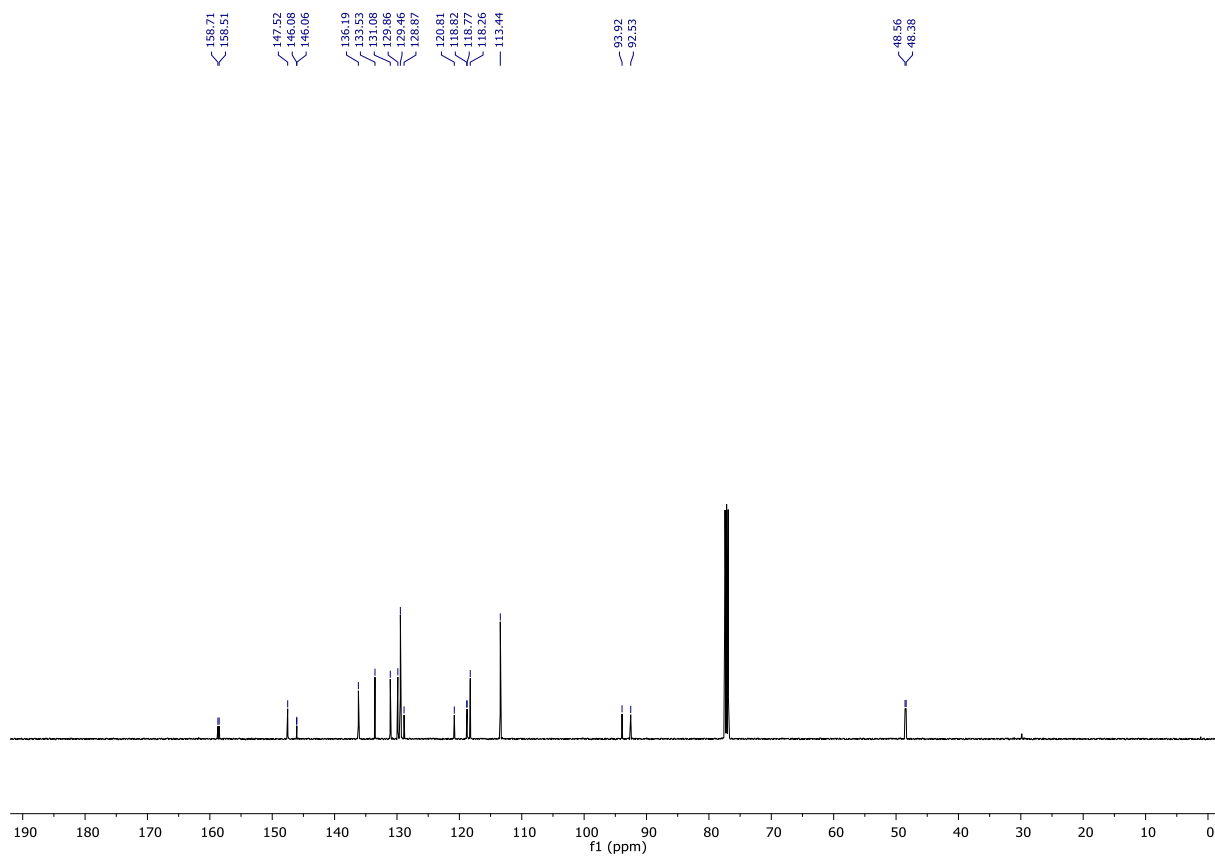
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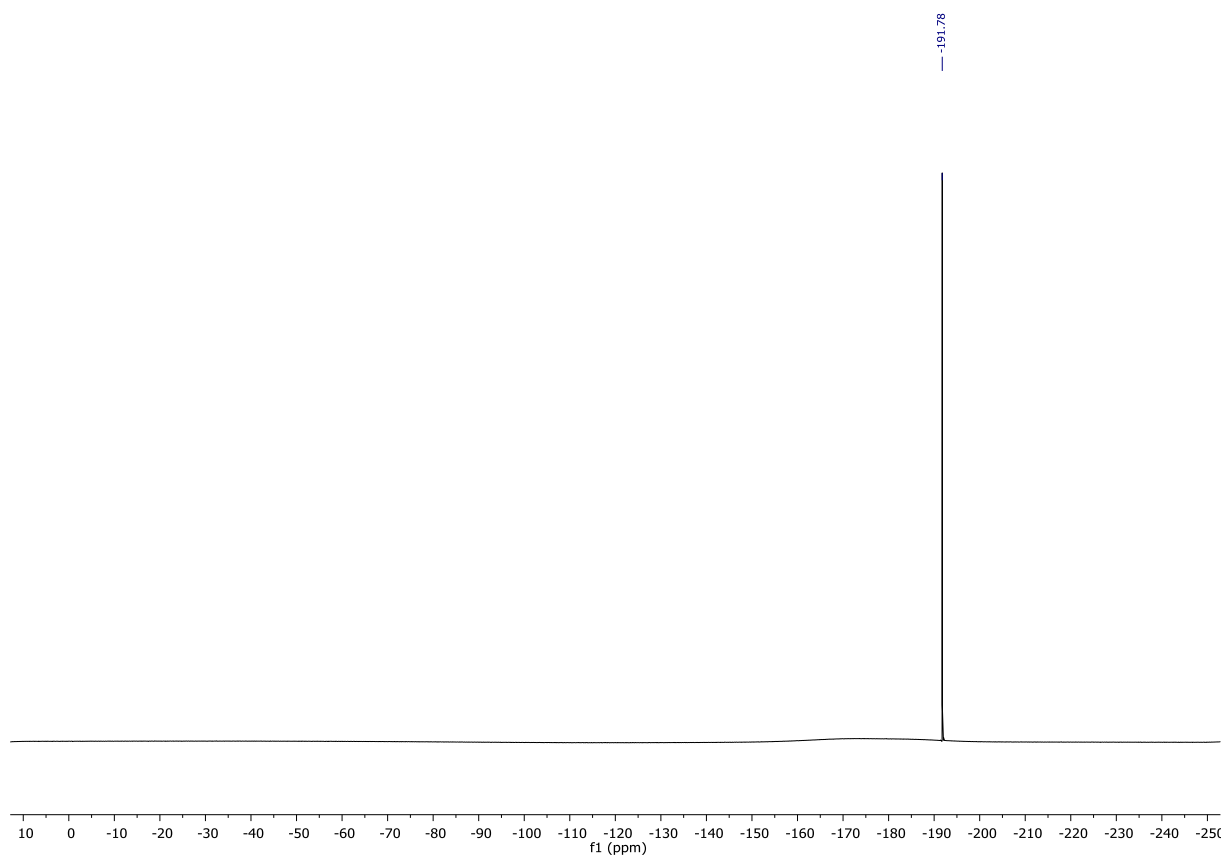
¹H NMR of 7p



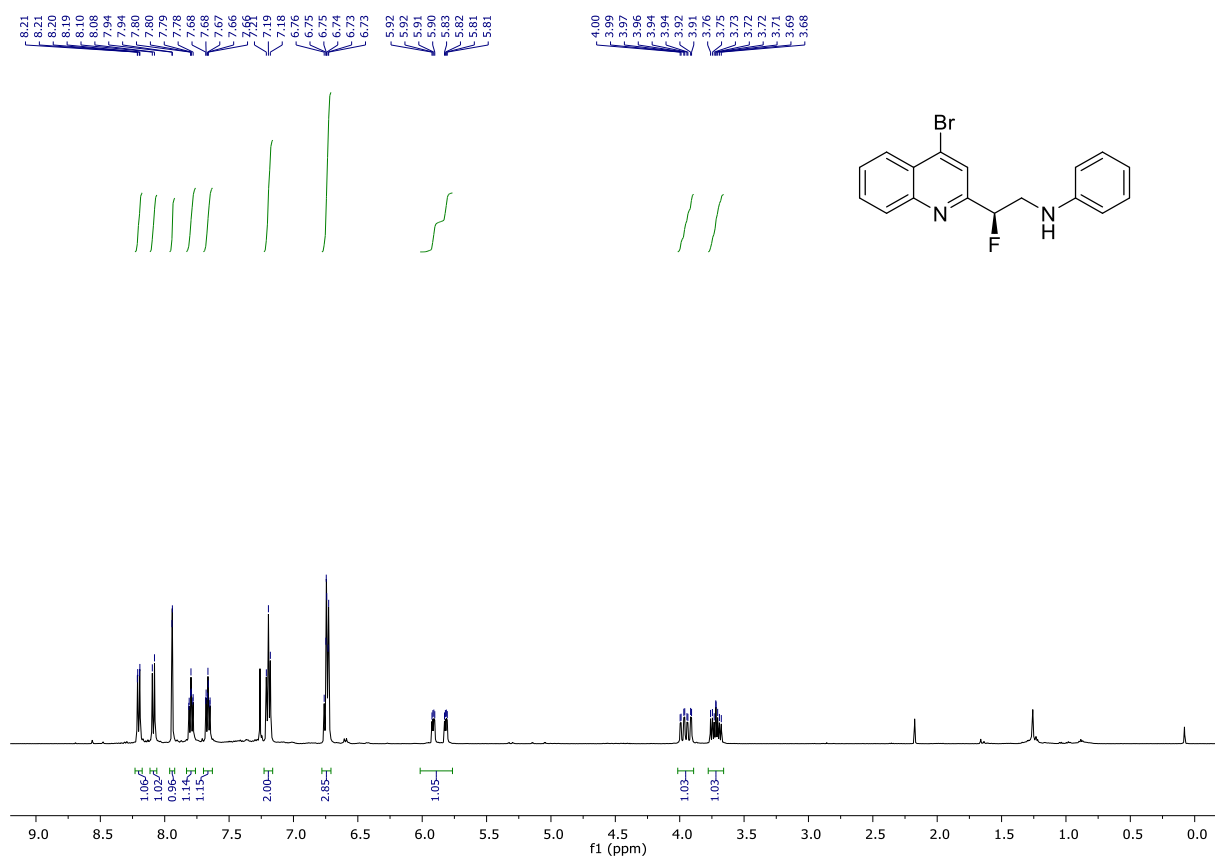
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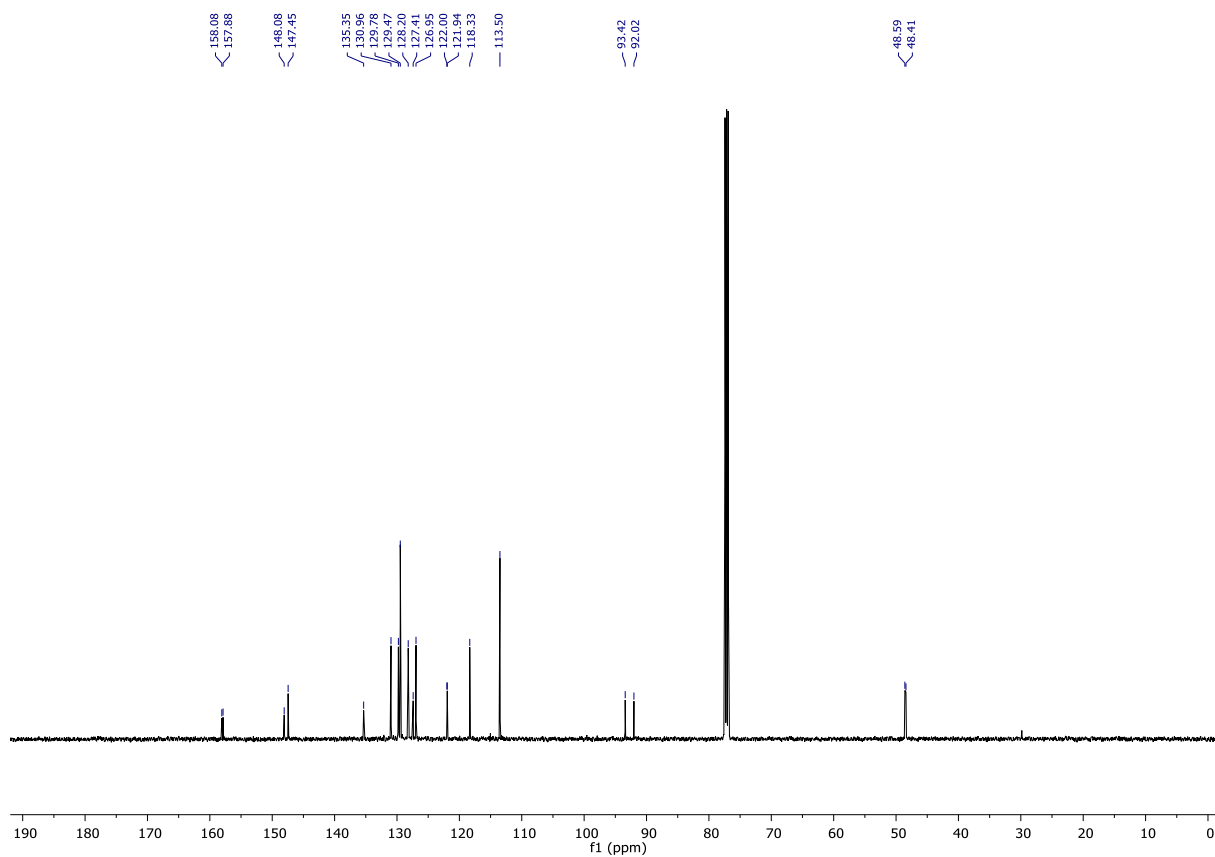
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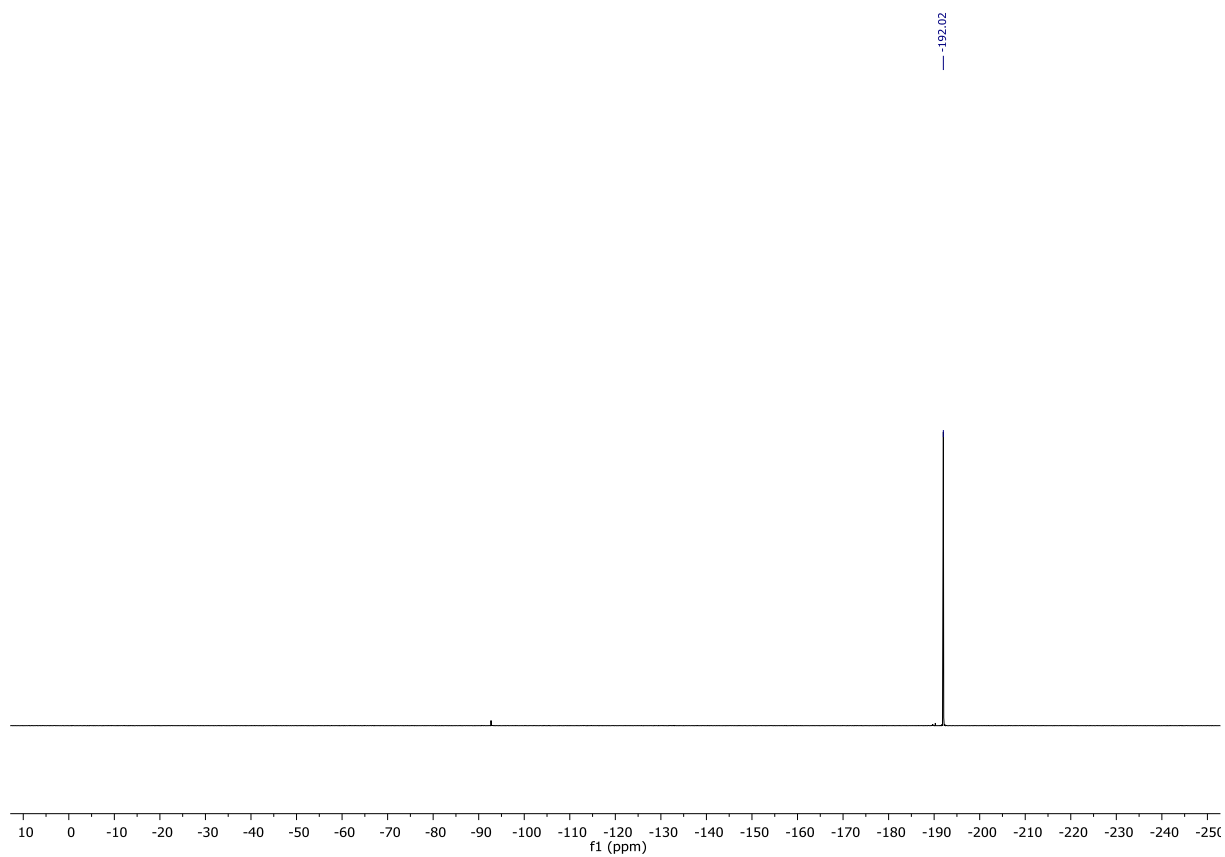
¹H NMR of 7q



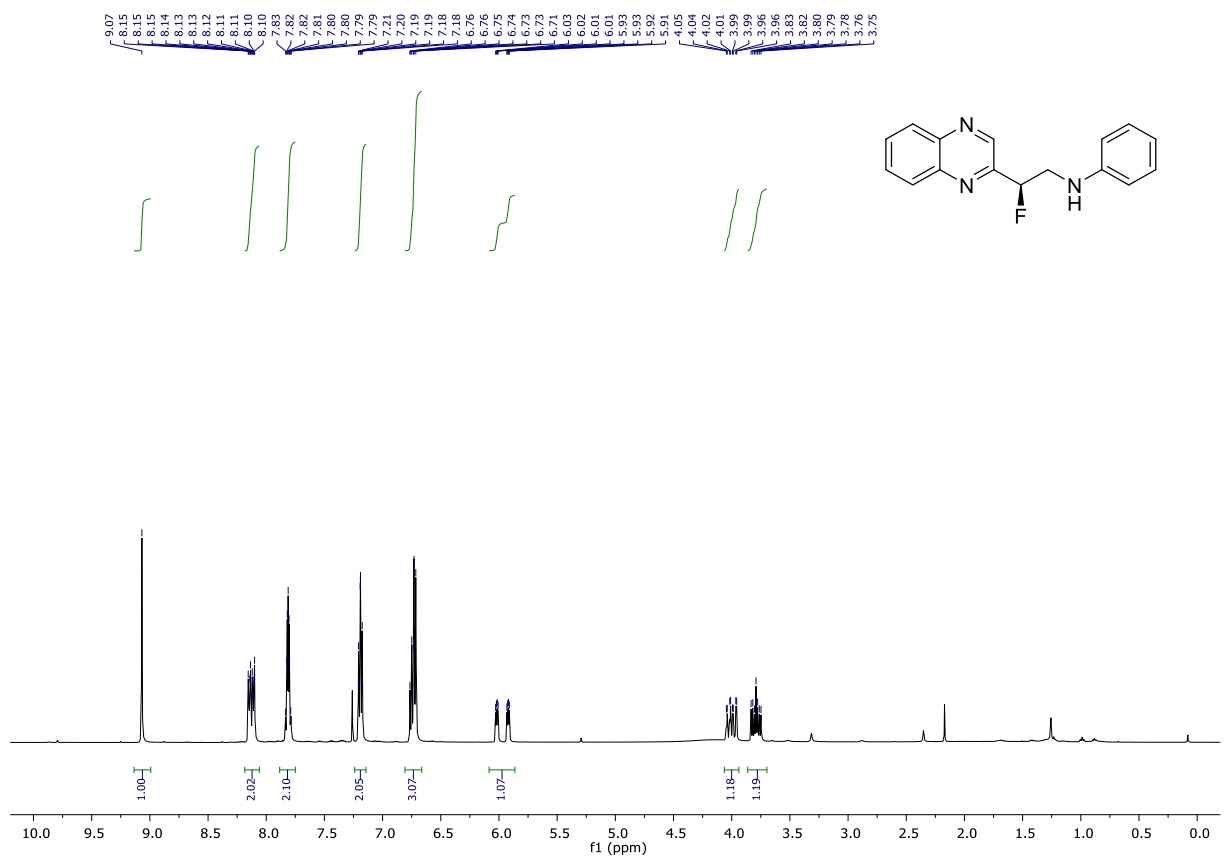
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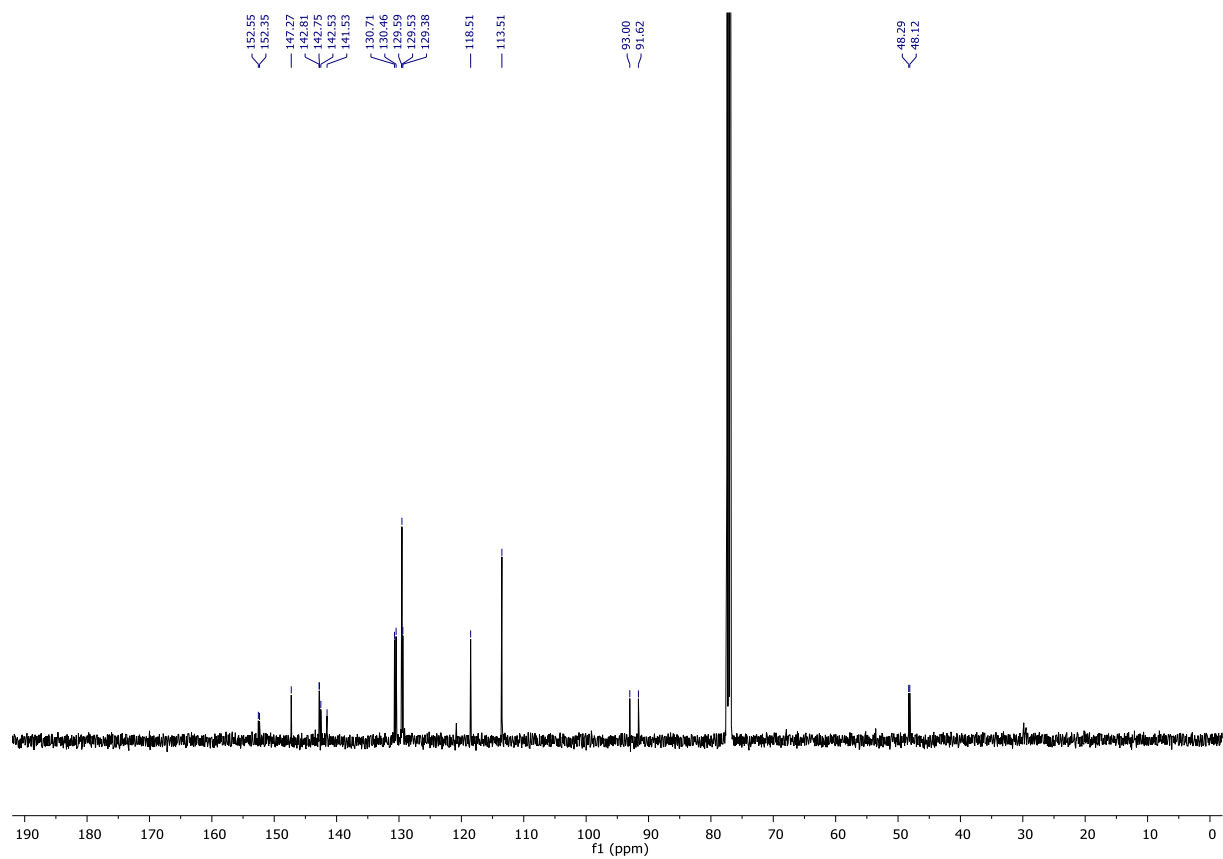
¹⁹F NMR of 7q



¹H NMR of 7r



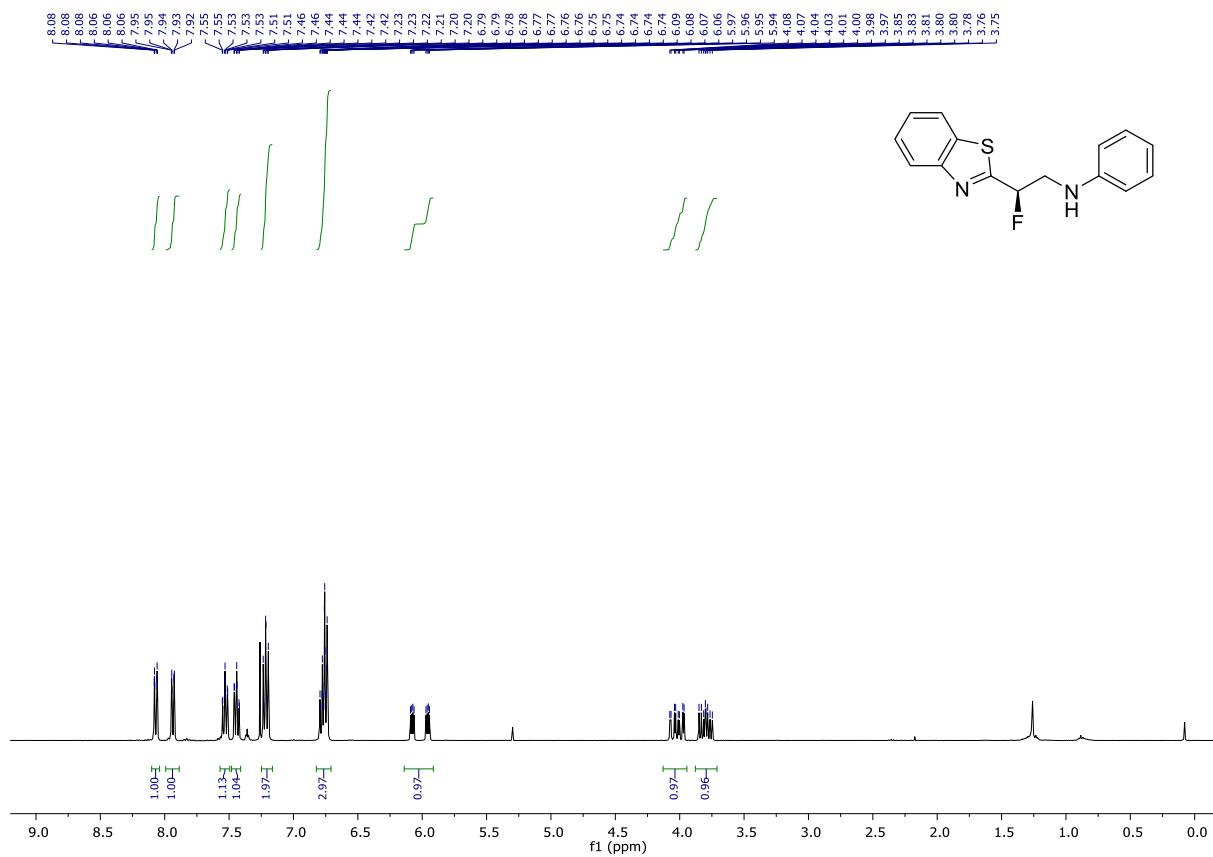
¹³C NMR of 7r



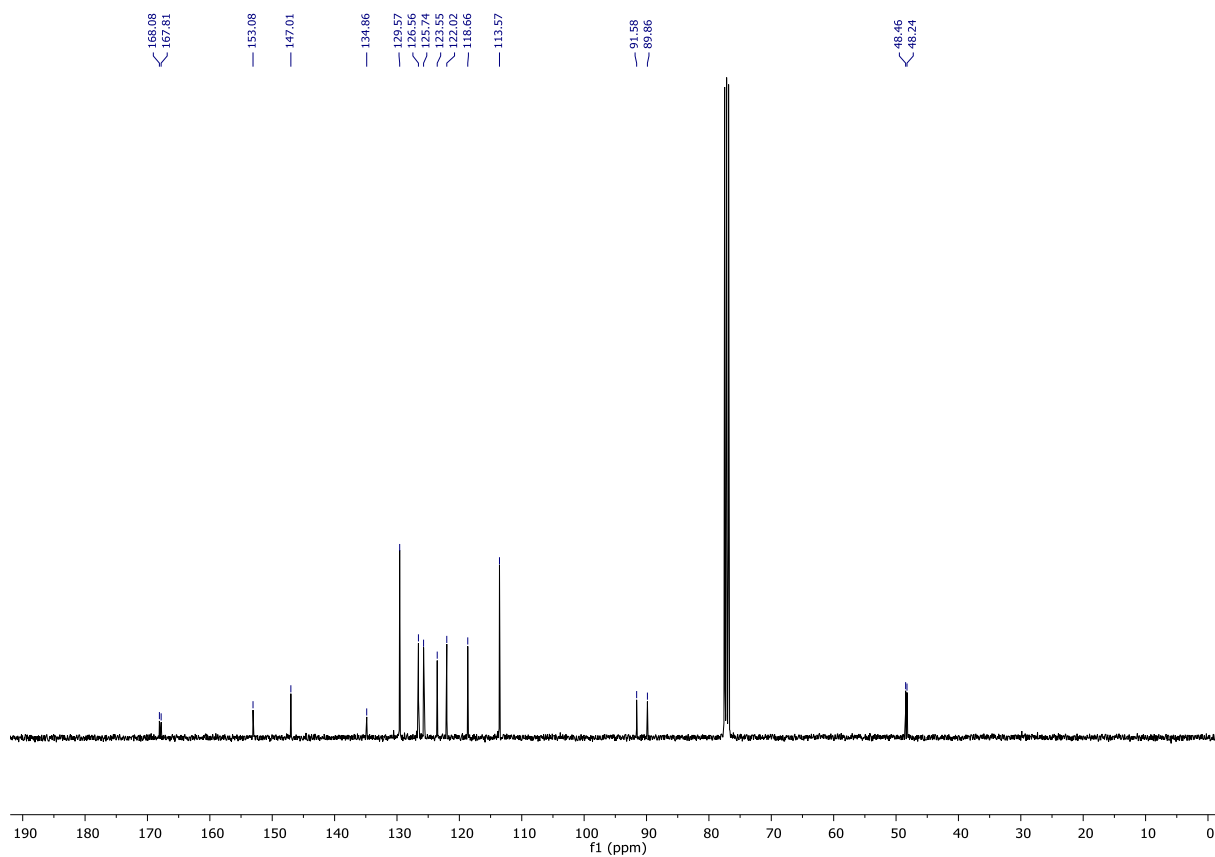
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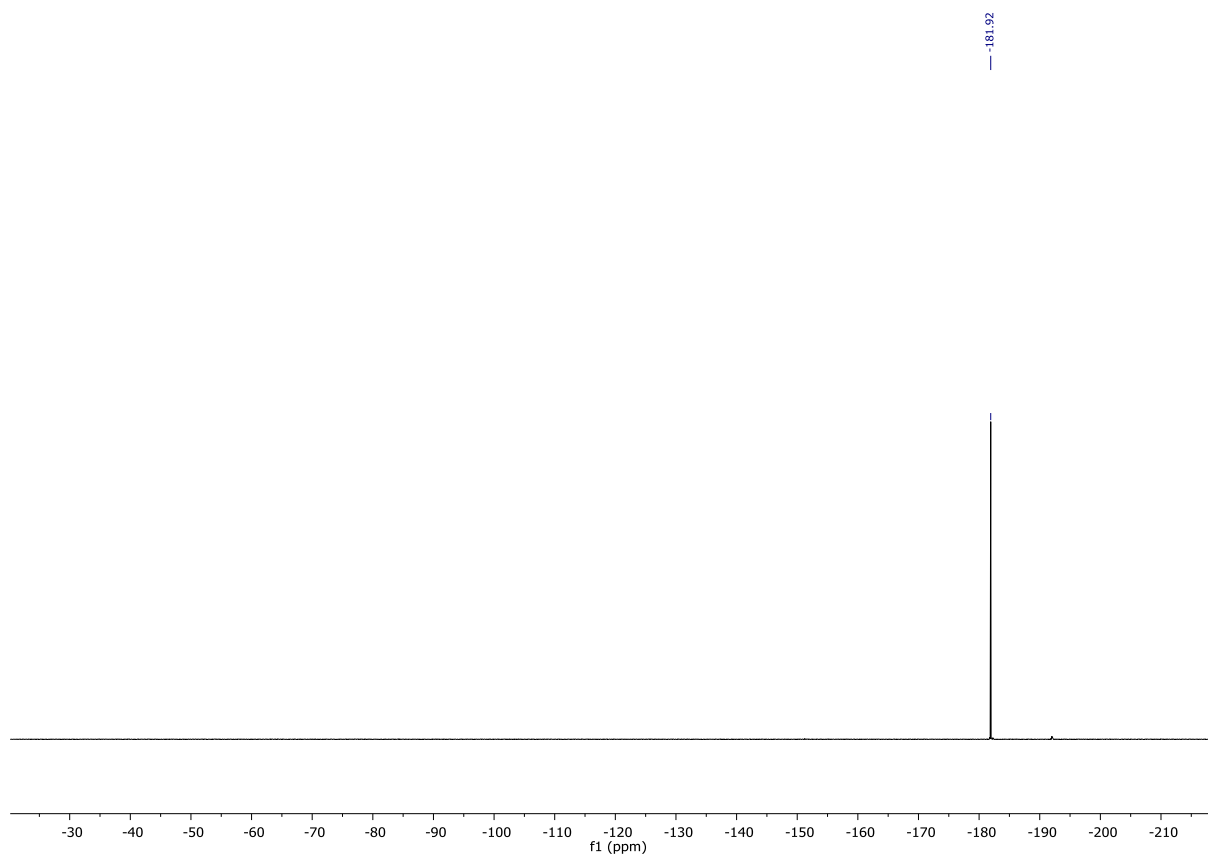
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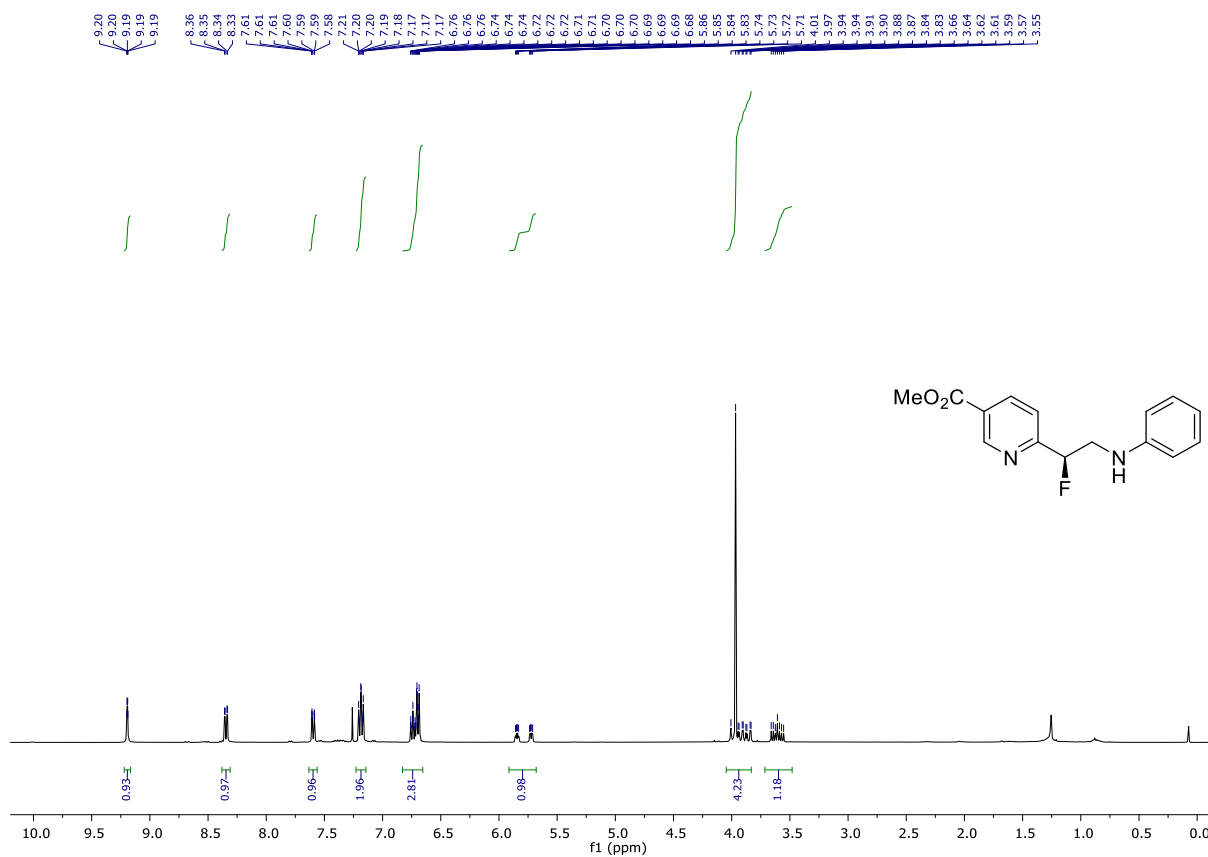
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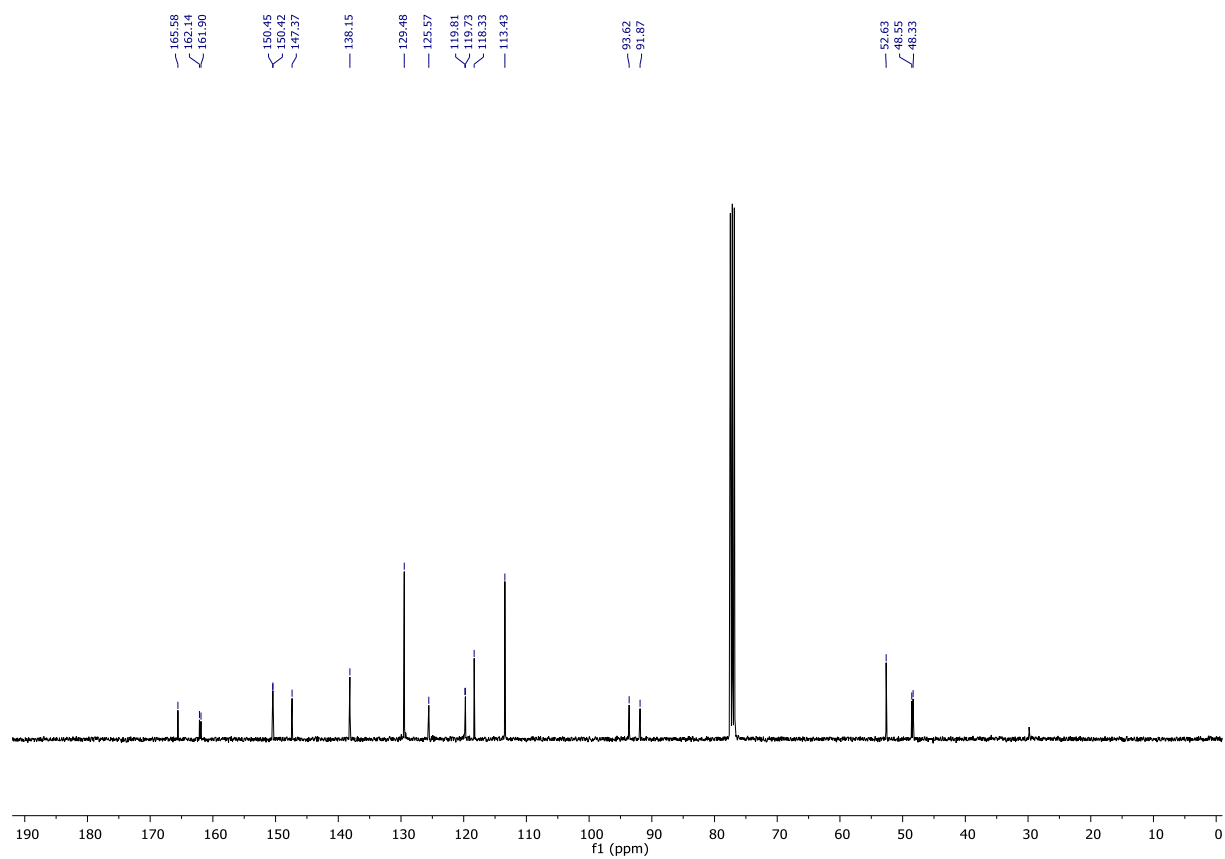
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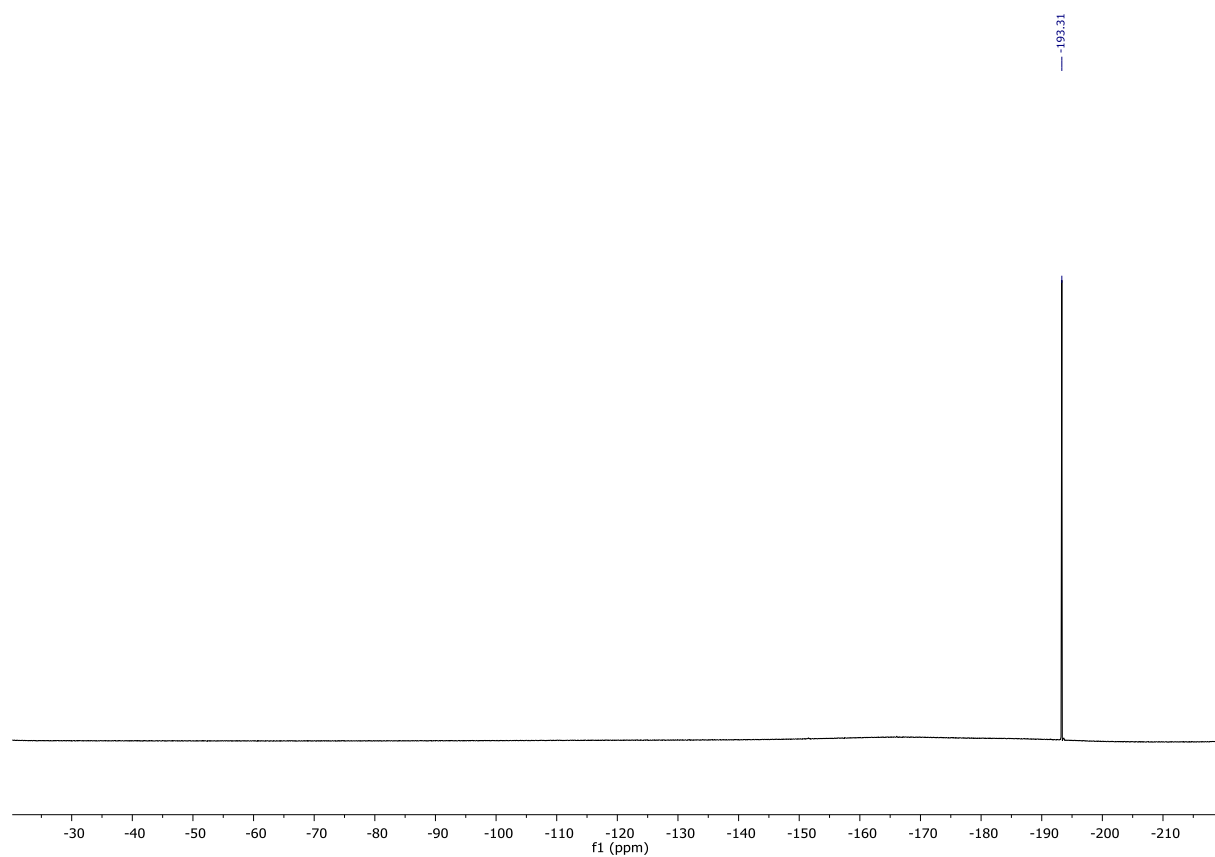
¹H NMR of 7t



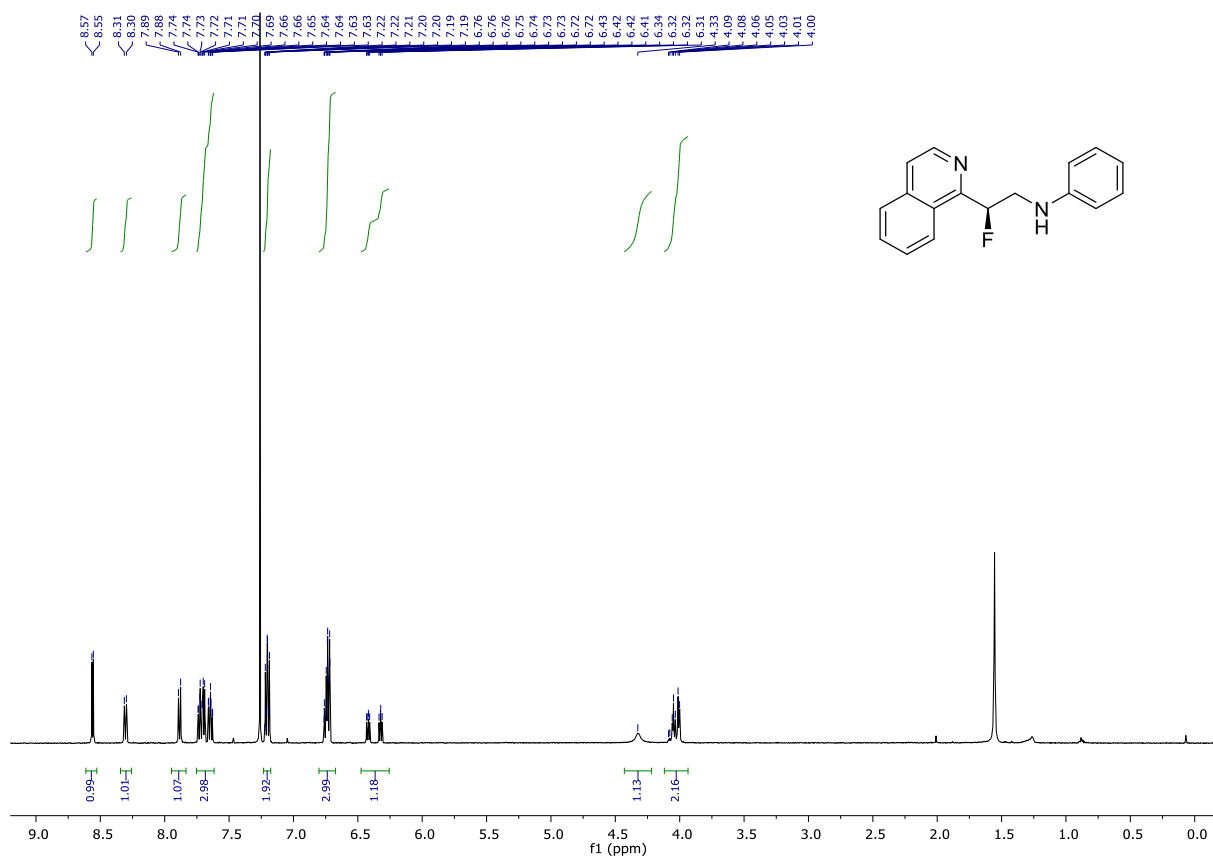
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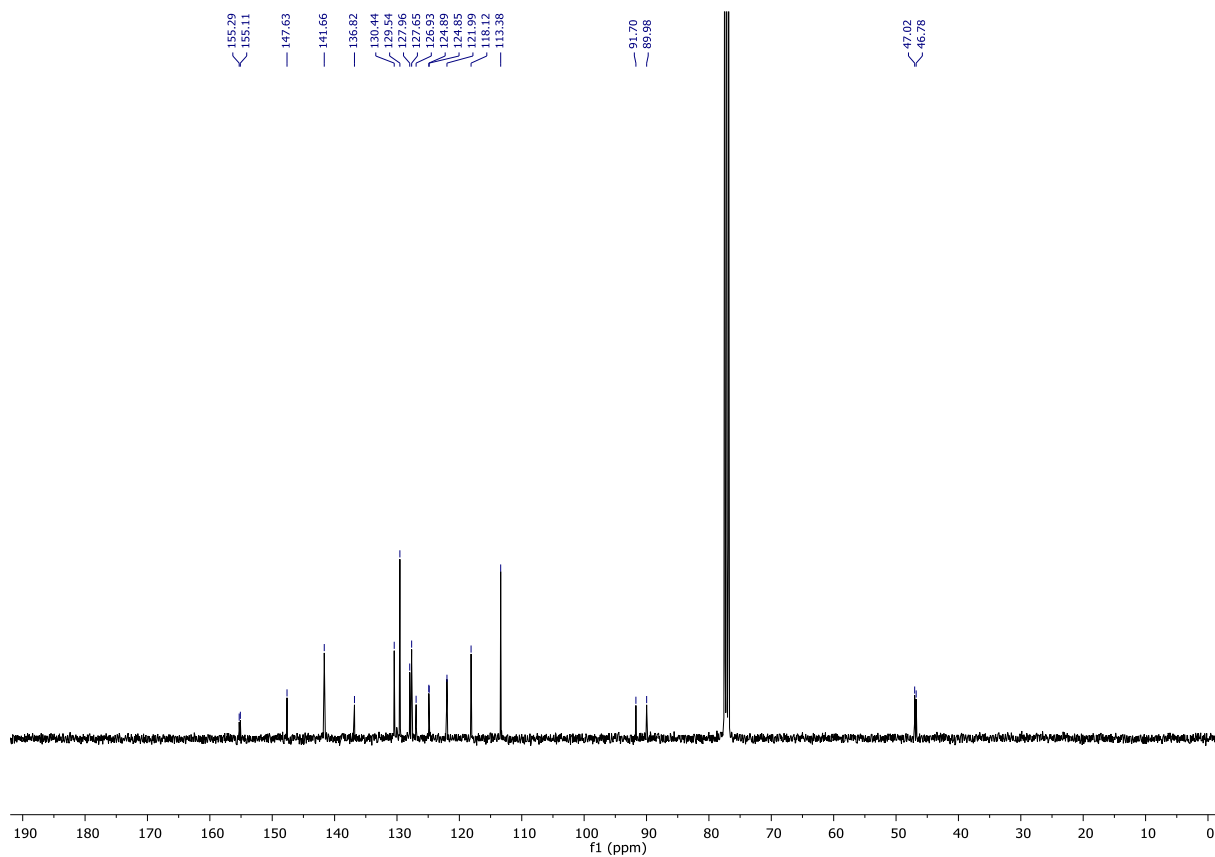
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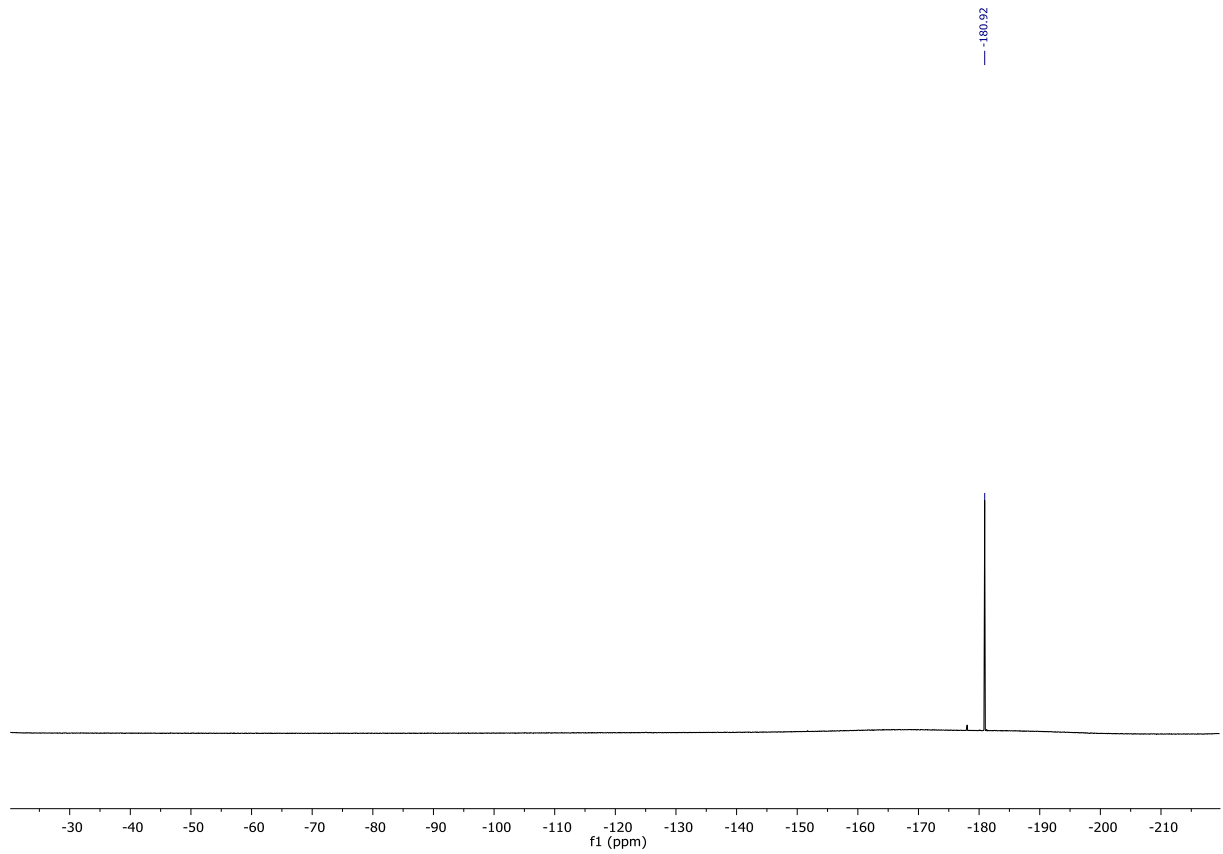
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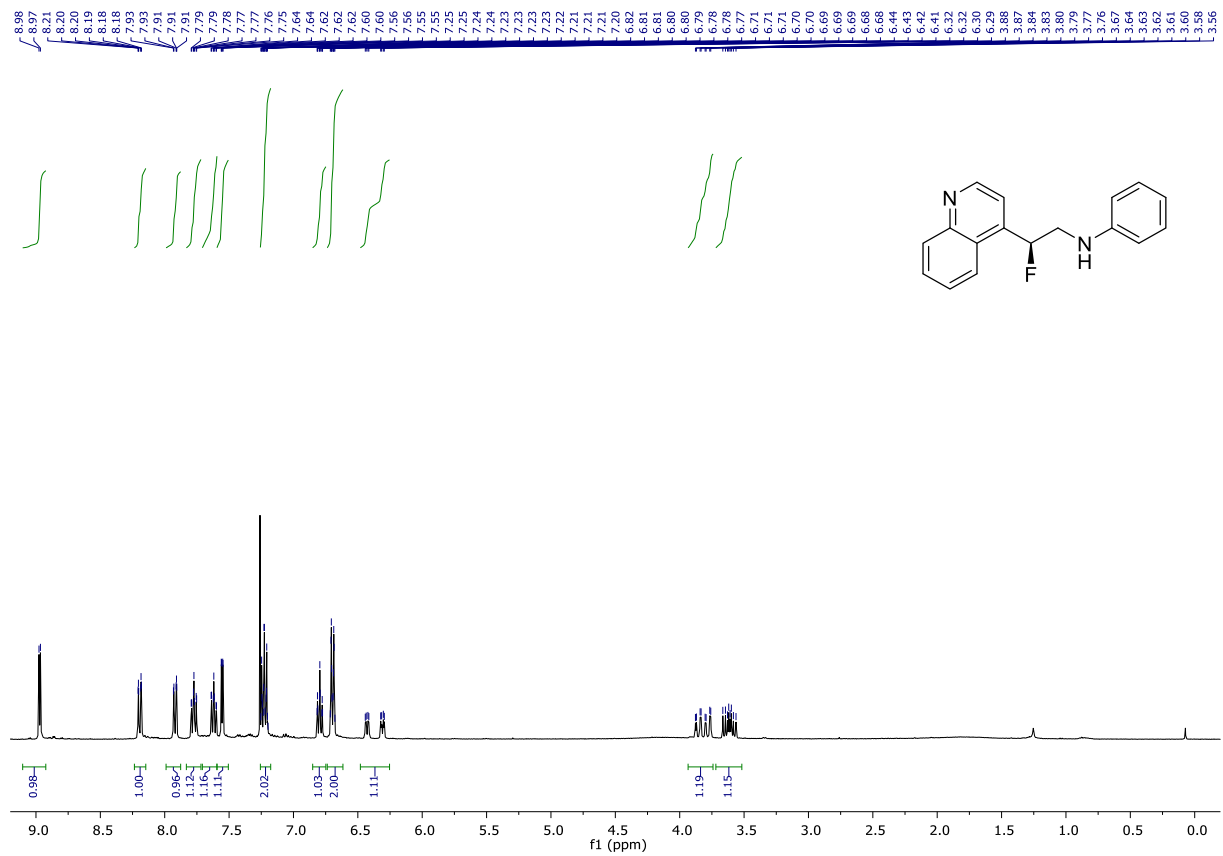
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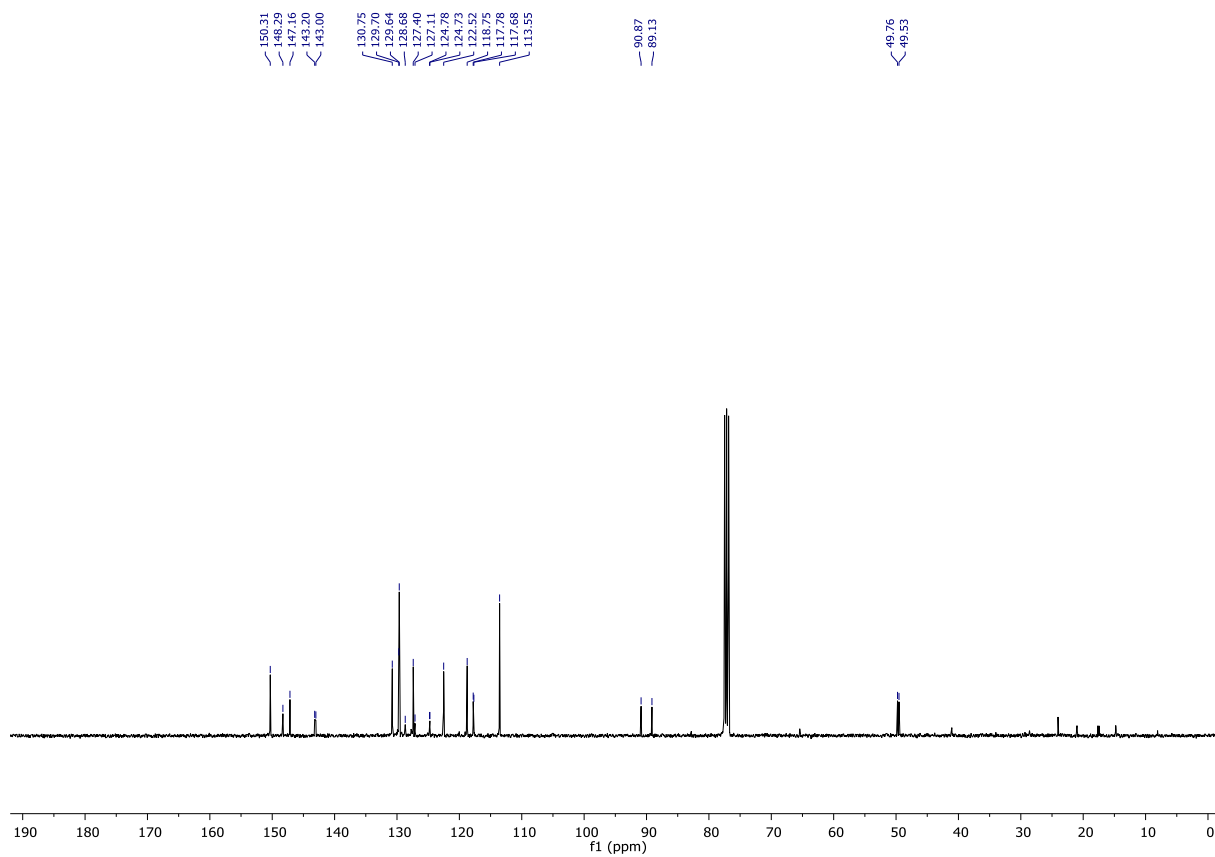
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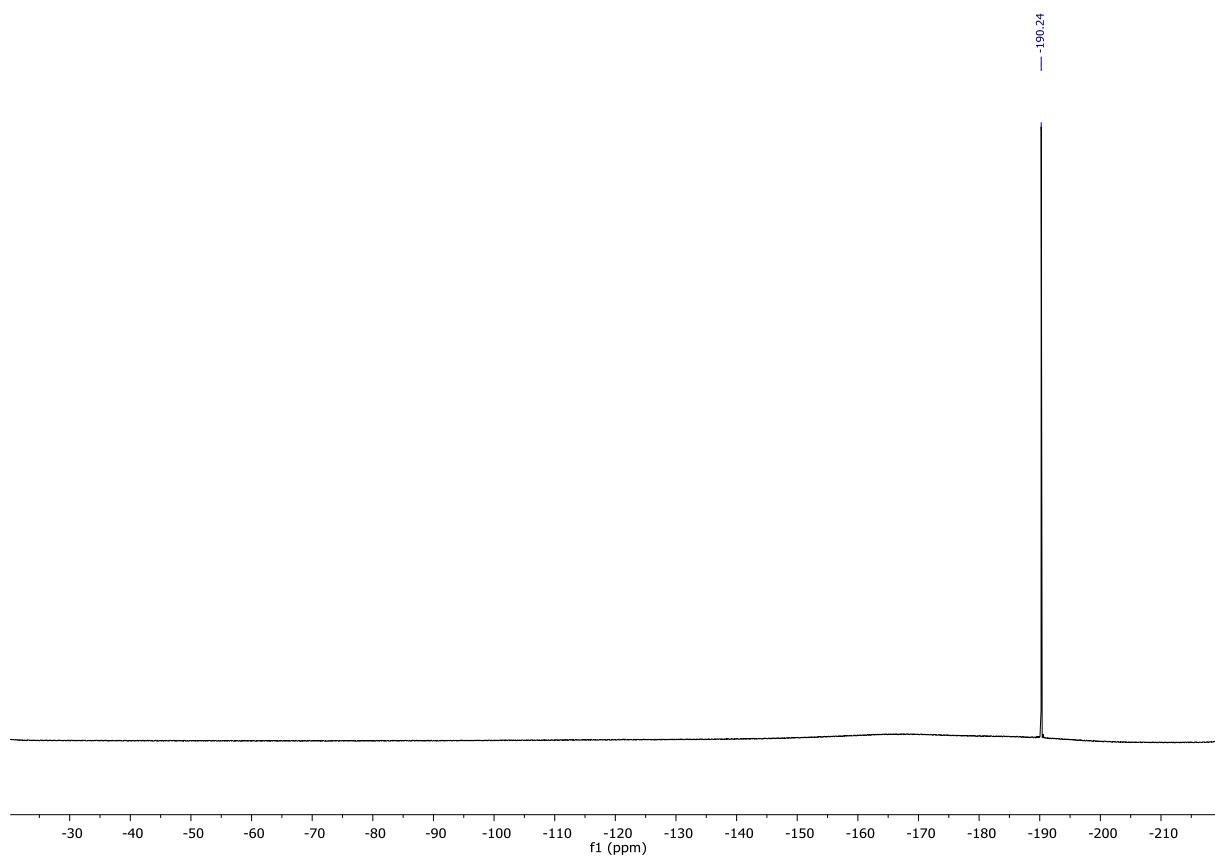
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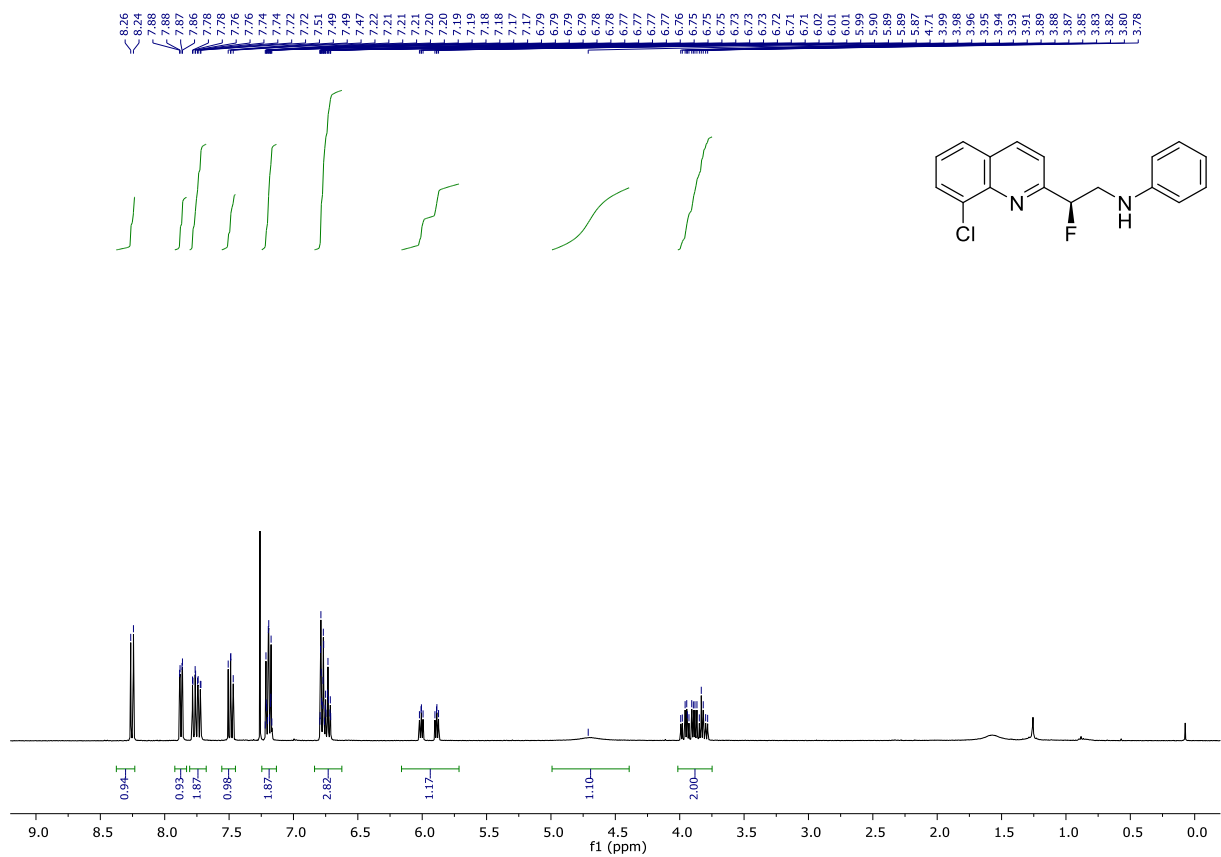
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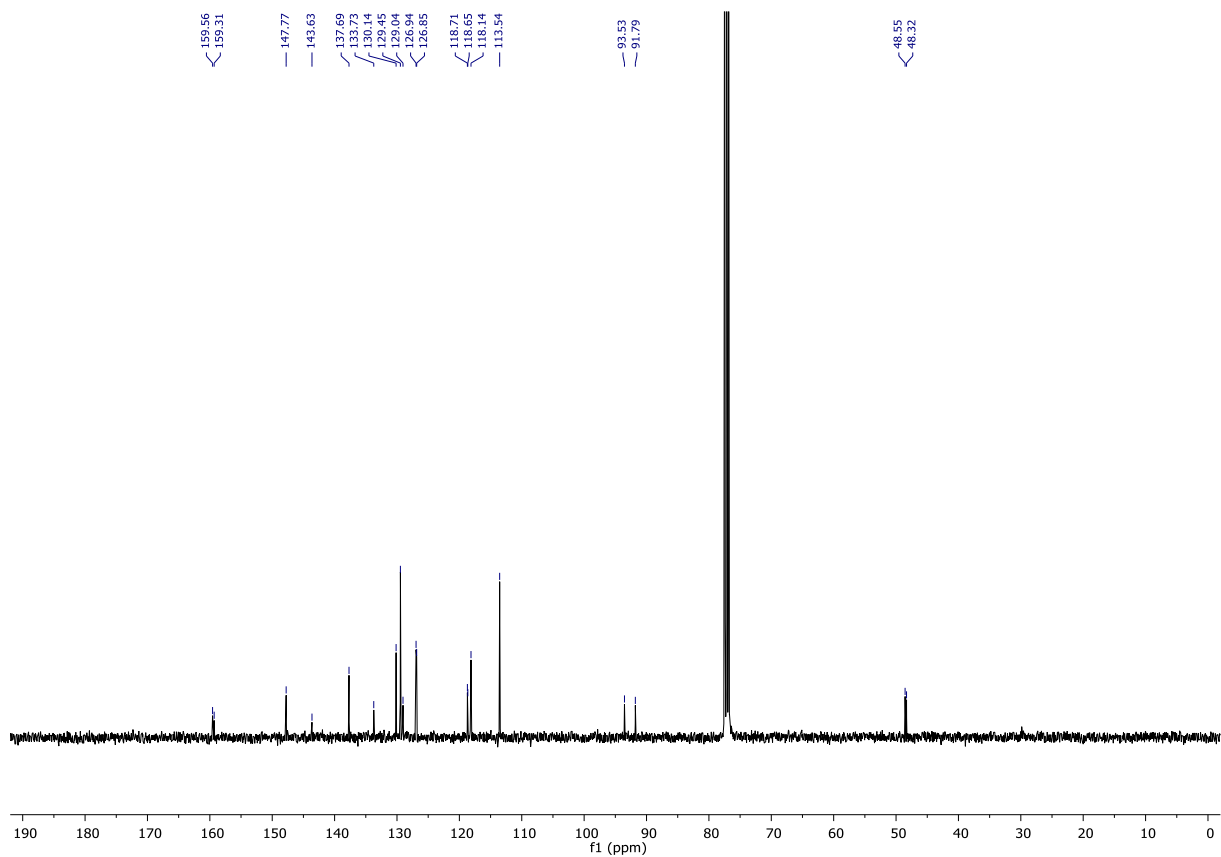
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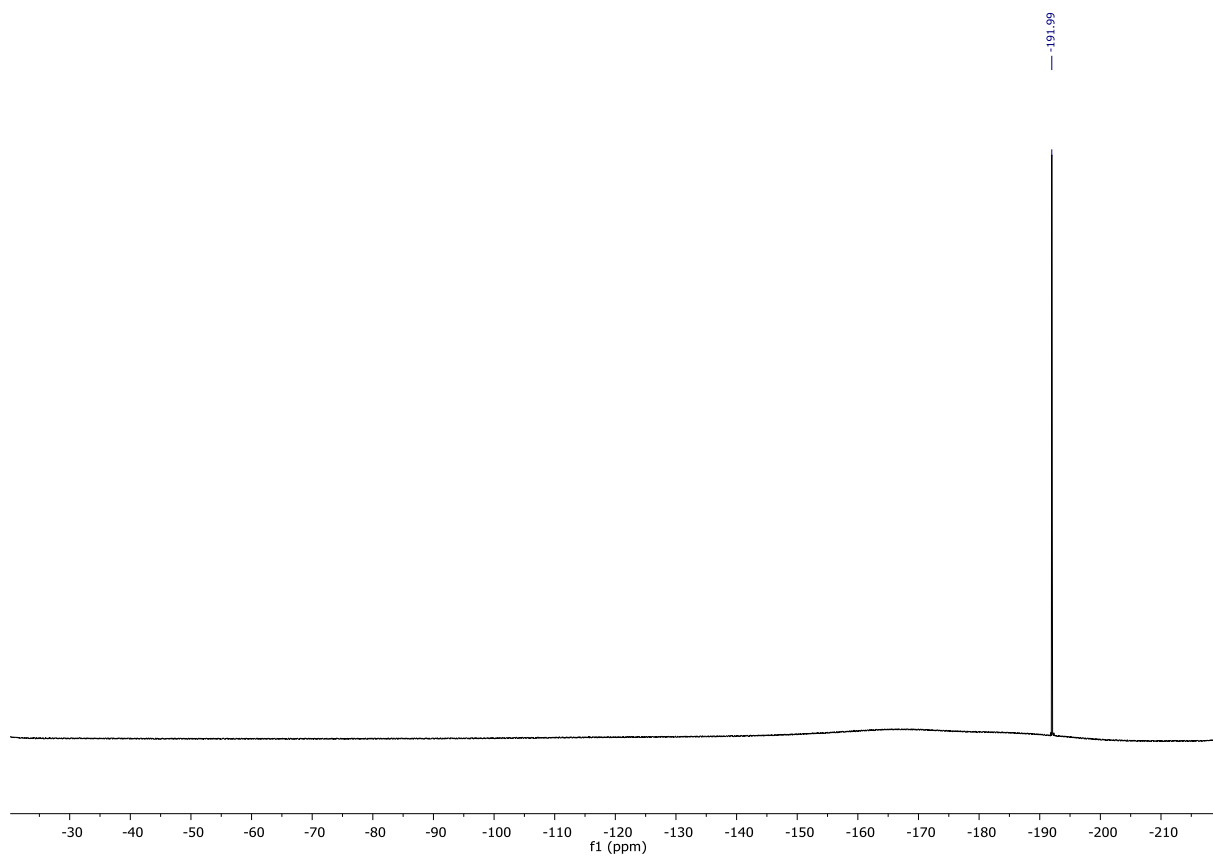
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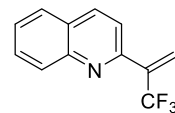
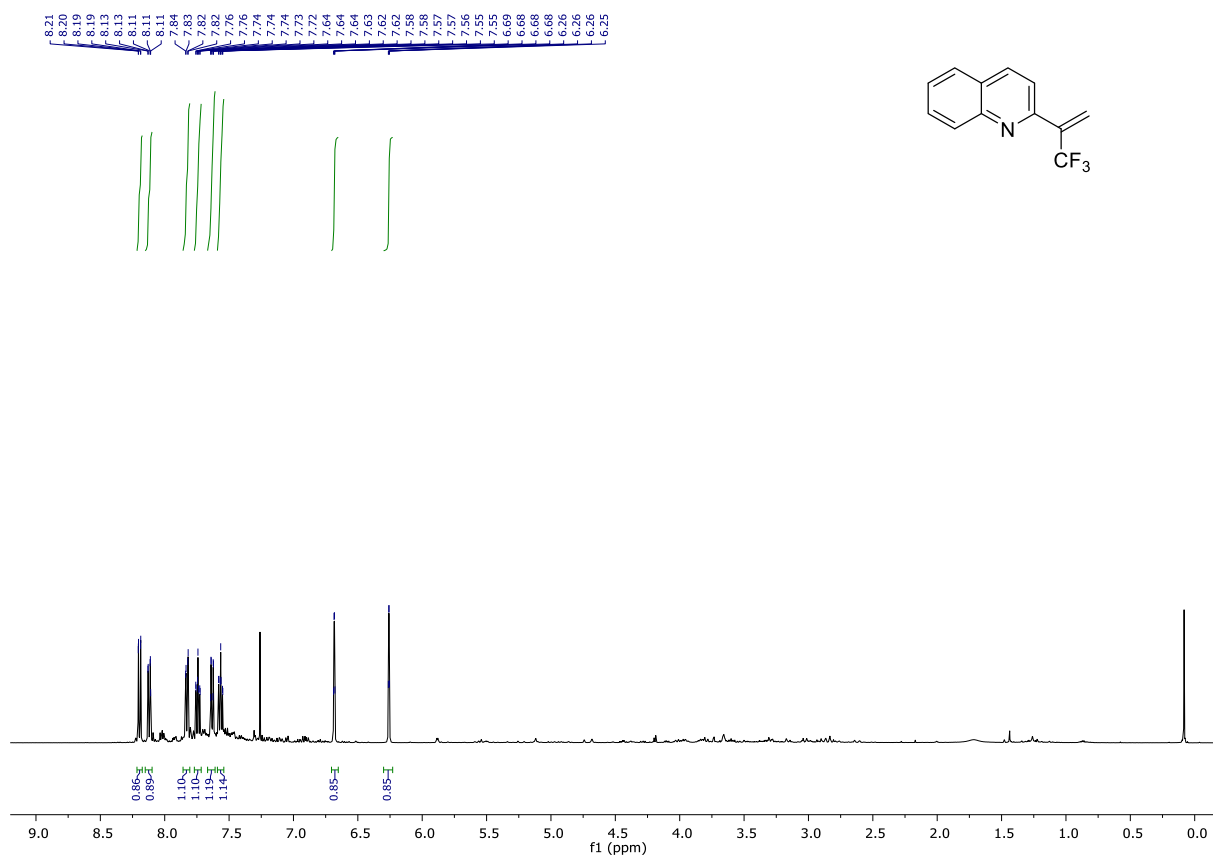
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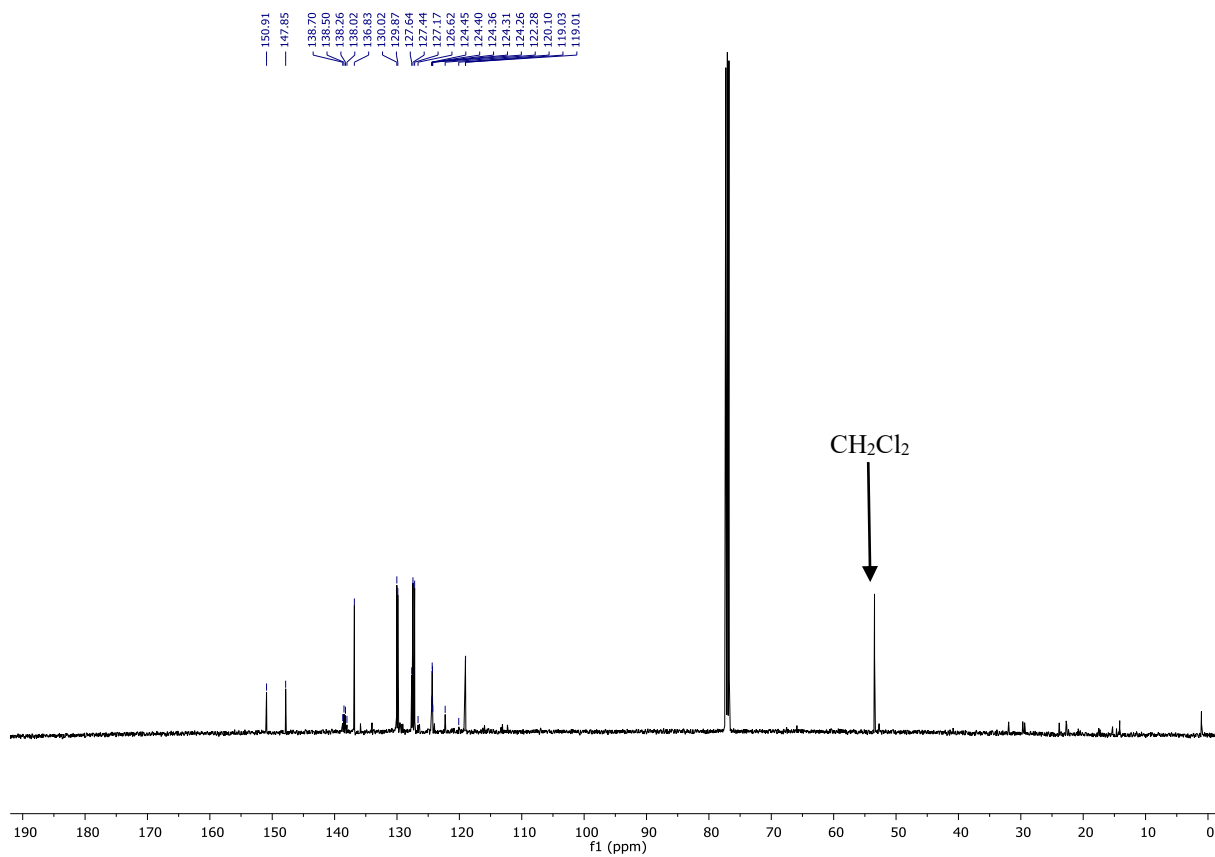
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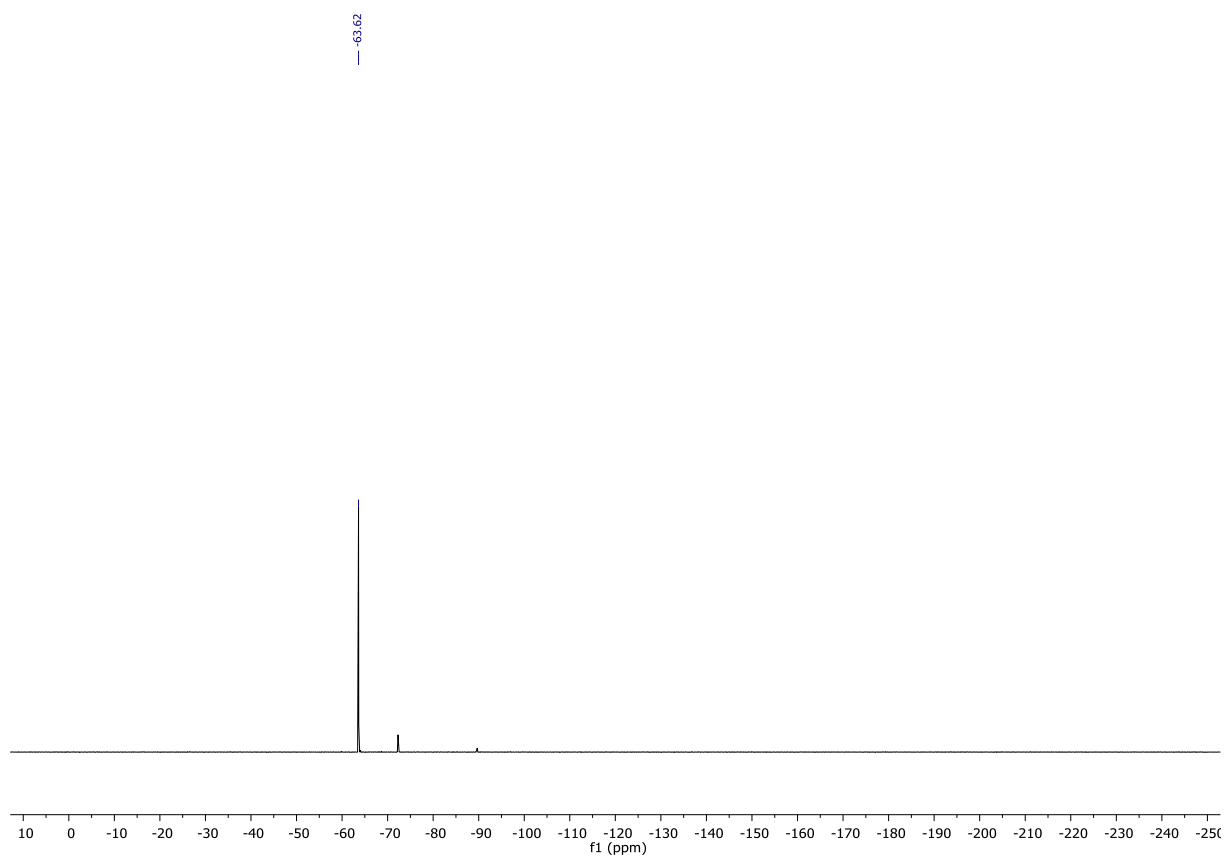
¹H NMR of 10



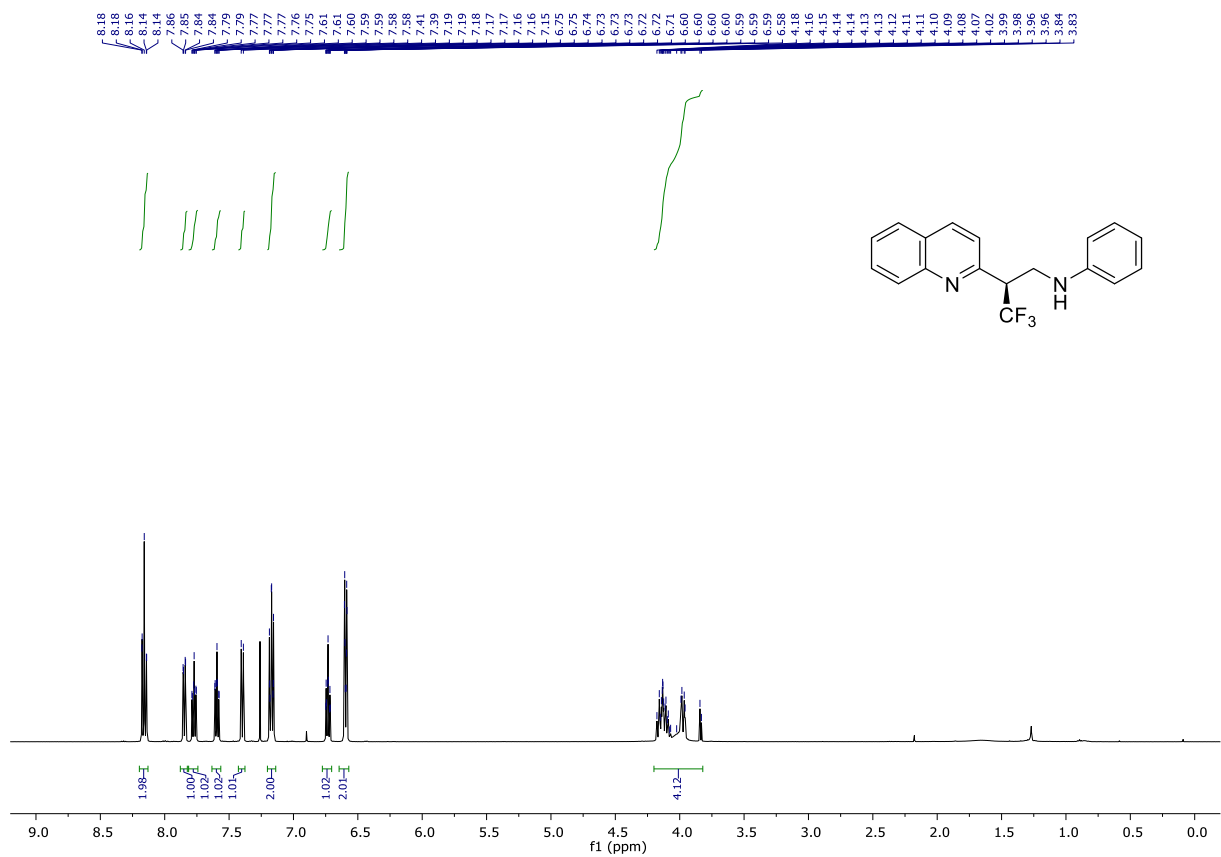
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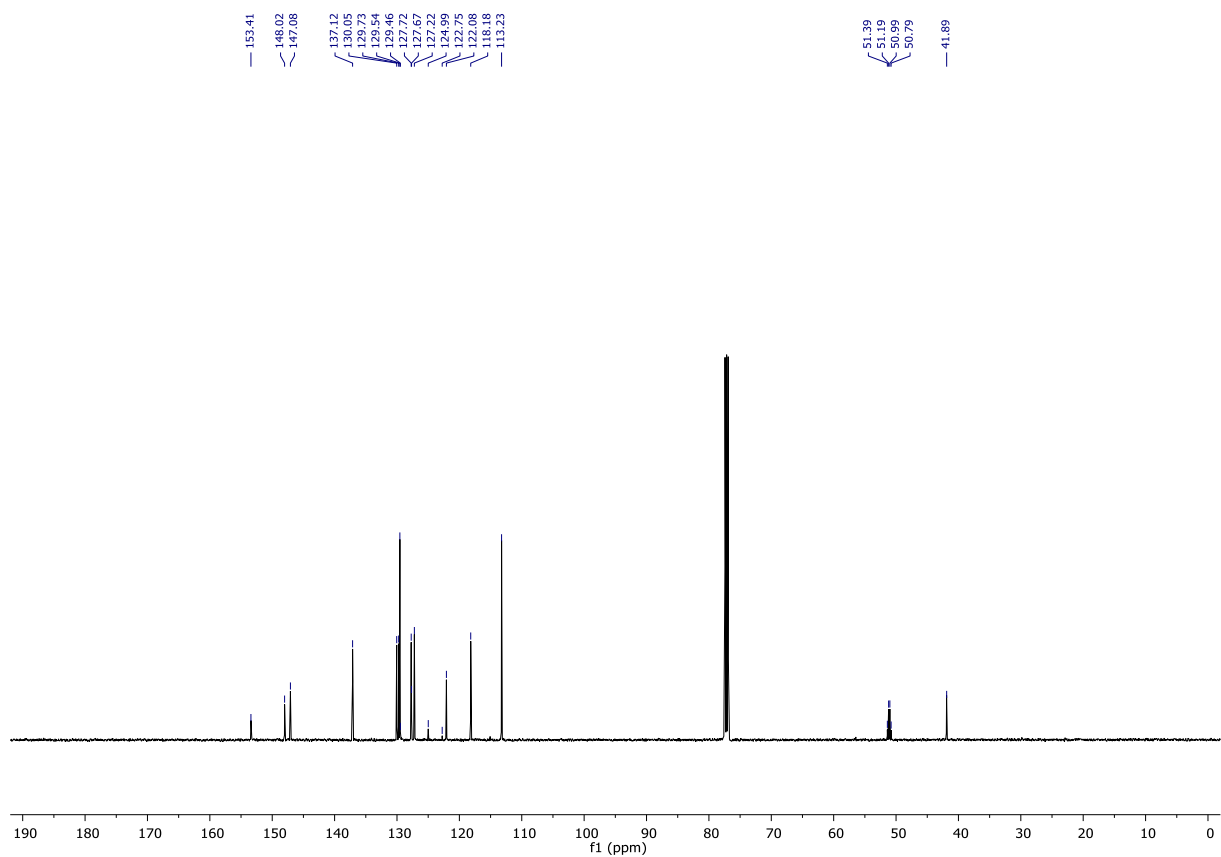
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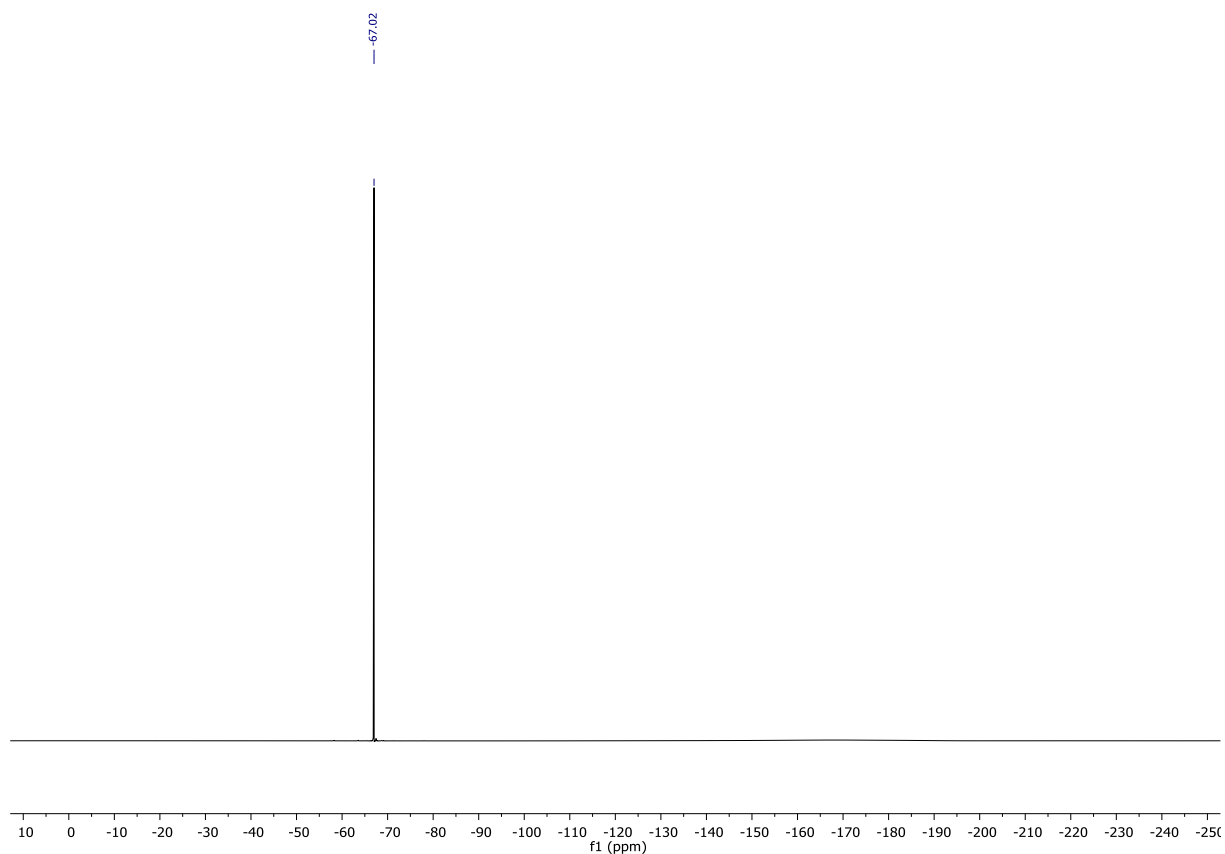
¹H NMR of 11



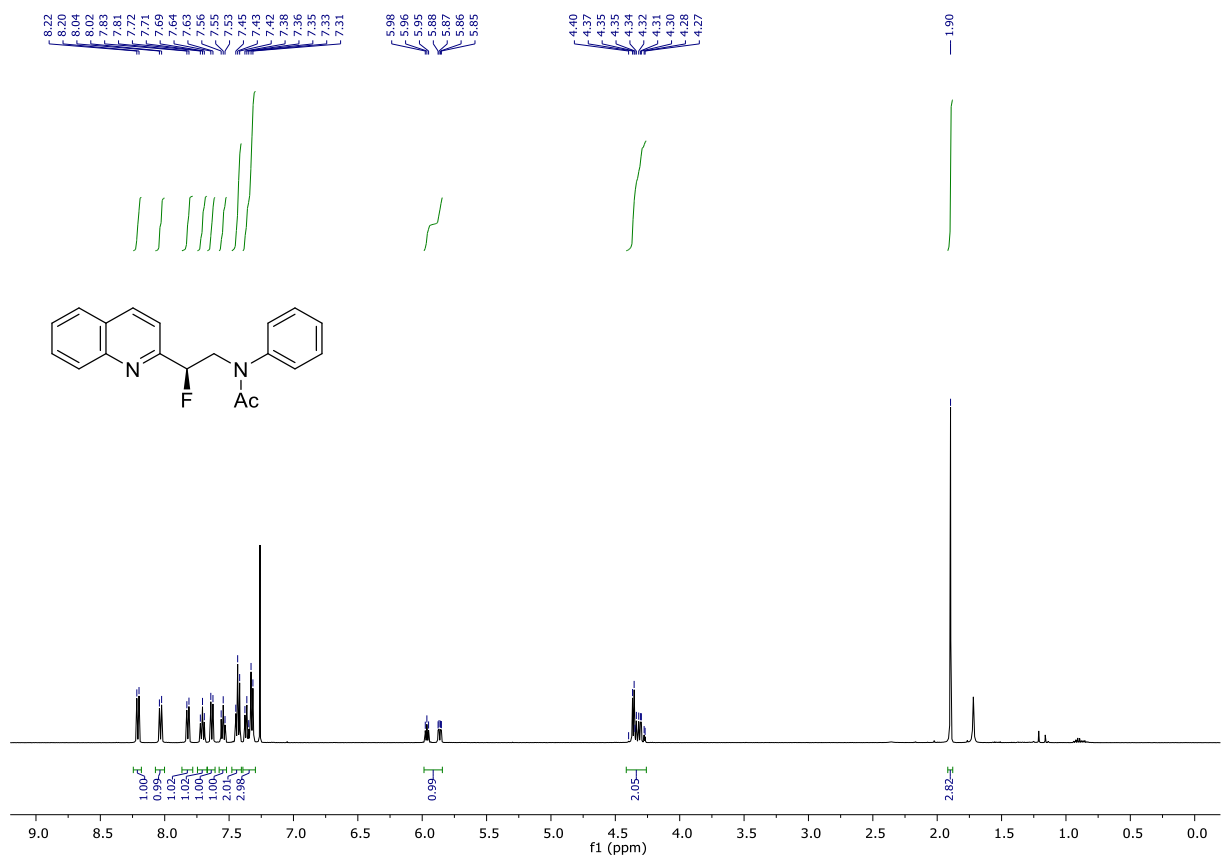
¹³C NMR of 11



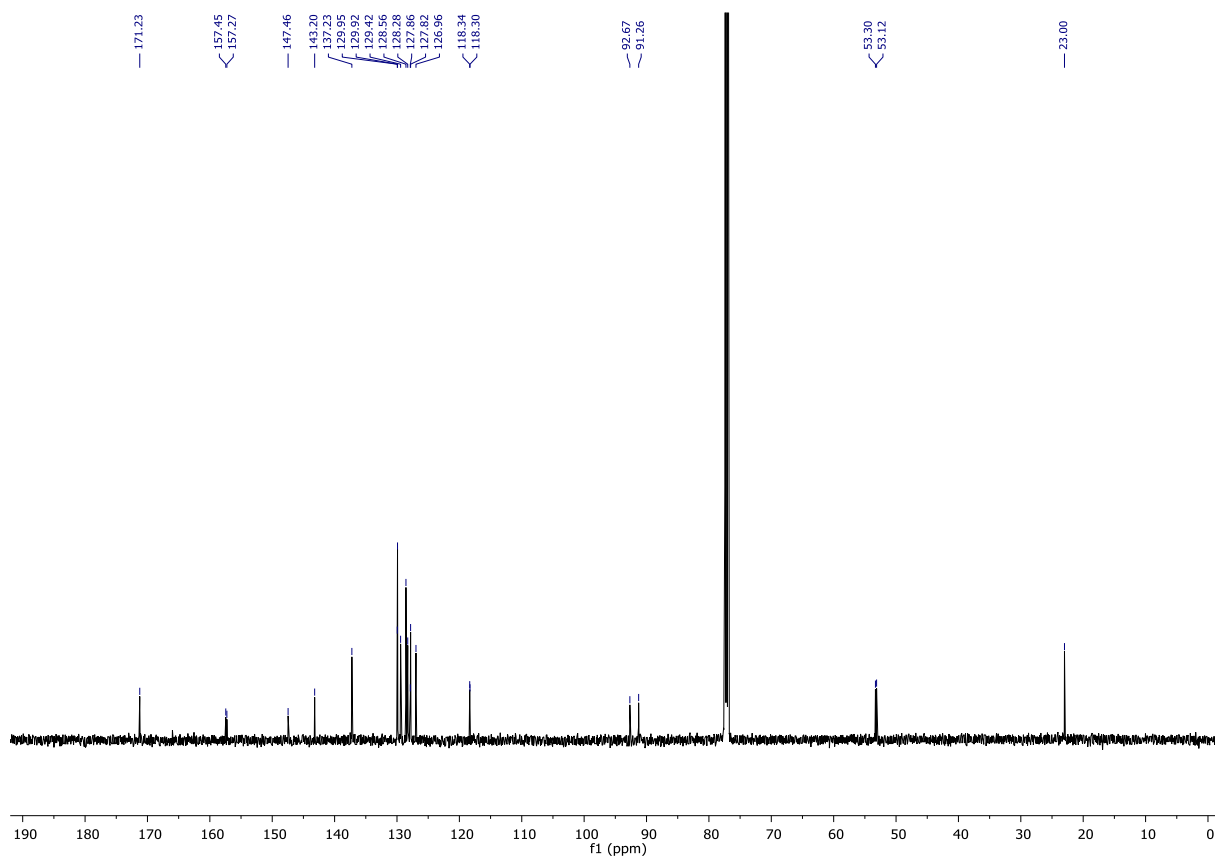
¹⁹F NMR of 11



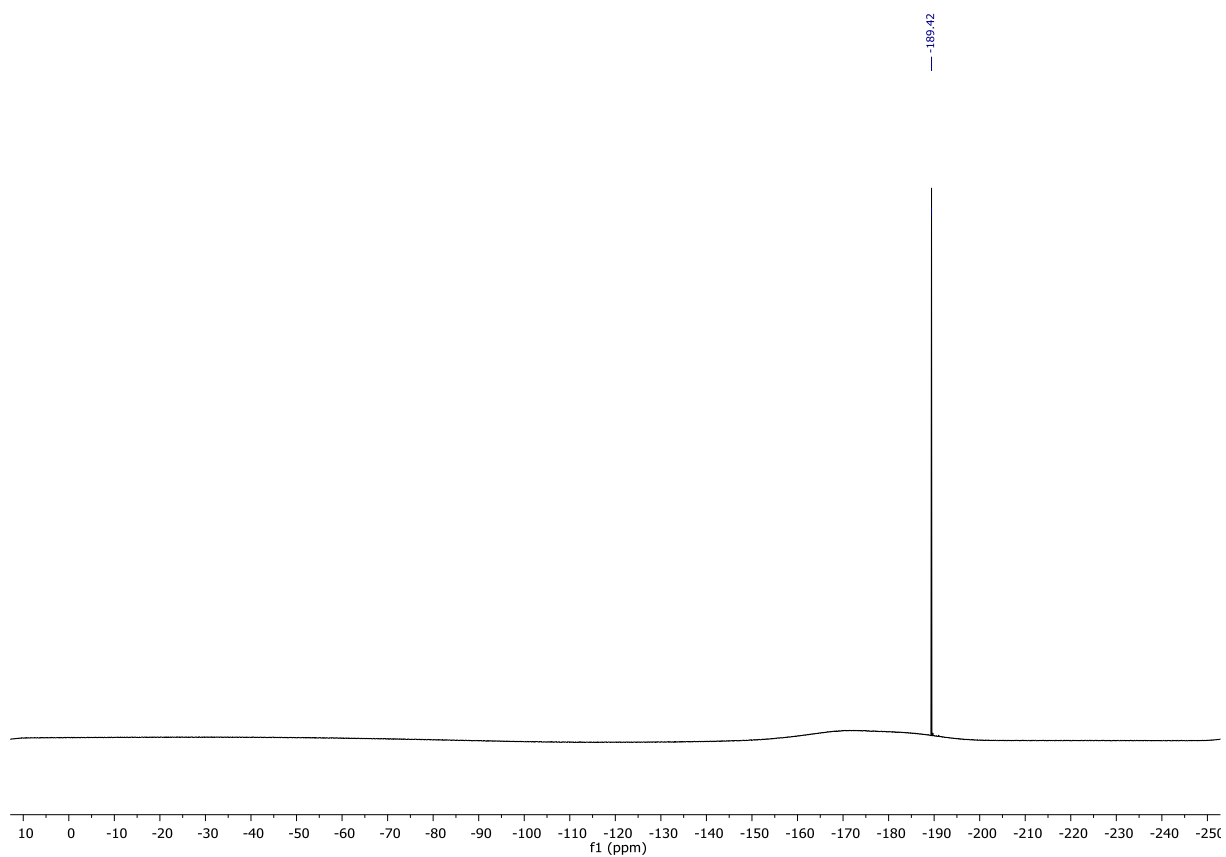
¹H NMR of 16



¹³C NMR of 16



¹⁹F NMR of 16



10. X-Ray Data

The datasets for all three compounds were collected using a Rigaku FR-X Ultrahigh Brilliance Microfocus RA generator/confocal optics with XtaLAB P200 diffractometer. The data for **7a** was collected at 125K with Cu radiation ($\lambda = 1.54187 \text{ \AA}$), whereas the data for **11** and **16** were collected at 93 K with Mo radiation ($\lambda = 0.71075 \text{ \AA}$). Intensity data were collected using ω steps accumulating area detector images spanning at least a hemisphere of reciprocal space. All data were corrected for Lorentz polarization effects. A multiscan absorption correction was applied by using CrysAlisPro. The structure of **7a** was solved using direct methods (SIR2011), whereas **11** and **16** were solved using dual space methods (SHELXT). The structures were refined by full-matrix least-squares against F^2 (SHELXL-2013). Non-hydrogen atoms were refined anisotropically, and hydrogen atoms were refined using a riding model, with the exception of H11 in **7a**, which was located via the electron density map. All calculations for **7a** were performed using the CrystalStructure interface and Olex2 for structures **11** and **16**. Selected crystallographic data are presented in Table 1. CCDC 1900849-1900851 contains the supplementary crystallographic data for this paper. The data can be obtained free of charge from the Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/structures.

	7a	16	11
Crystal data			
Chemical formula	C ₁₇ H ₁₅ FN ₂	C ₁₉ H ₁₇ FN ₂ O	C ₁₈ H ₁₅ F ₃ N ₂
M_r	266.32	308.35	316.32
Crystal system, space group	Orthorhombic, $P2_12_12_1$	Monoclinic, $C2/c$	Monoclinic, $P2_1$
Temperature (K)	125	93	93
a, b, c (Å)	5.44863 (10), 10.49560 (18), 23.5843 (4)	18.6107 (4), 7.75338 (14), 23.2547 (5)	10.1249 (3), 5.4916 (1), 13.3402 (5)
α, β, γ (°)	90, 90, 90	90, 111.212 (2), 90	90, 93.978 (3), 90
V (Å ³)	1348.71 (4)	3128.21 (12)	739.95 (4)
Z	4	8	2
Radiation type	Cu $K\alpha$	Mo $K\alpha$	Mo $K\alpha$
μ (mm ⁻¹)	0.71	0.09	0.11
Crystal size (mm)	0.12 × 0.10 × 0.10	0.27 × 0.18 × 0.06	0.36 × 0.09 × 0.03
Data collection			
T_{\min}, T_{\max}	0.889, 0.932	0.881, 0.995	0.871, 1.000
No. of measured, independent and observed reflections	15722, 2741, 2729	22904, 3460, 3247	8314, 2968, 2587
R_{int}	0.019	0.018	0.022
$(\sin \theta/\lambda)_{\text{max}}$ (Å ⁻¹)	0.628	0.664	0.660

Refinement			
$R[F^2 > 2\sigma(F^2)],$ $wR(F^2), S$	0.027, 0.071, 1.09	0.040, 0.107, 1.02	0.039, 0.103, 1.06
No. of reflections	2741	3460	2968
No. of parameters	186	209	256
No. of restraints	1	0	8
$\Delta\rho_{\max}, \Delta\rho_{\min}$ (e Å ⁻³)	0.16, -0.18	0.50, -0.31	0.24, -0.23
Absolute structure parameter	0.06 (4)	–	-0.2 (3)

Computer programs: *CrysAlis PRO* 1.171.39.8d (Rigaku OD, 2015), *CrysAlis PRO* 1.171.40.29a (Rigaku OD, 2018), *SIR2011* (Burla, *et al.*, 2012), *SHELXT* Version 2014/4 (Sheldrick, 2014), *SHELXT* (Sheldrick, 2015), *SHELXL* Version 2014/7 (Sheldrick, 2008), *SHELXL* (Sheldrick, 2015), *Olex2* (Dolomanov *et al.*, 2009), *CrystalStructure* 4.2 (Rigaku, 2015), *CrystalStructure* 4.3 (Rigaku, 2018).