

Biotransformation of food-grade and nanometric TiO₂ in the oral-gastro-intestinal tract: driving forces and effect on the toxicity toward intestinal epithelial cells

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Supplementary materials

Table S1. Composition of simulated OGI fluids

	Simulated saliva	Simulated gastric fluid	Simulated intestinal fluid
pH	6.5 ± 0.1	1.4 ± 0.1	8.1 ± 0.1
Inorganic components (g/L)	KCl 0.90 KSCN 0.20 NaH ₂ PO ₄ x H ₂ O 1.02 Na ₂ SO ₄ 0.57 NaCl 0.30	KCl 0.82 NH ₄ Cl 0.31 CaCl ₂ x 2H ₂ O 0.40 NaCl 2.75 NaH ₂ PO ₄ x H ₂ O 0.31	MgCl ₂ x 6H ₂ O 0.05 KCl 0.94 KH ₂ PO ₄ 0.08 NaHCO ₃ 9.17 NaCl 12.27 CaCl ₂ x 2H ₂ O 0.42
Organic components (g/L)	Urea 0.20	Urea 0.09 D-Glucose 0.65 Glucuronic acid 0.02 D-Glucosamine hydrochloride 0.33	Urea 0.35
Active components (g/L)	Mucin 0.05 Uric acid 0.016 α - amylase 0.145	Mucin 3 BSA 1 Pepsin 1	Pancreatin 3.00 Lipase 0.5 Bile 6.00 BSA 2.8

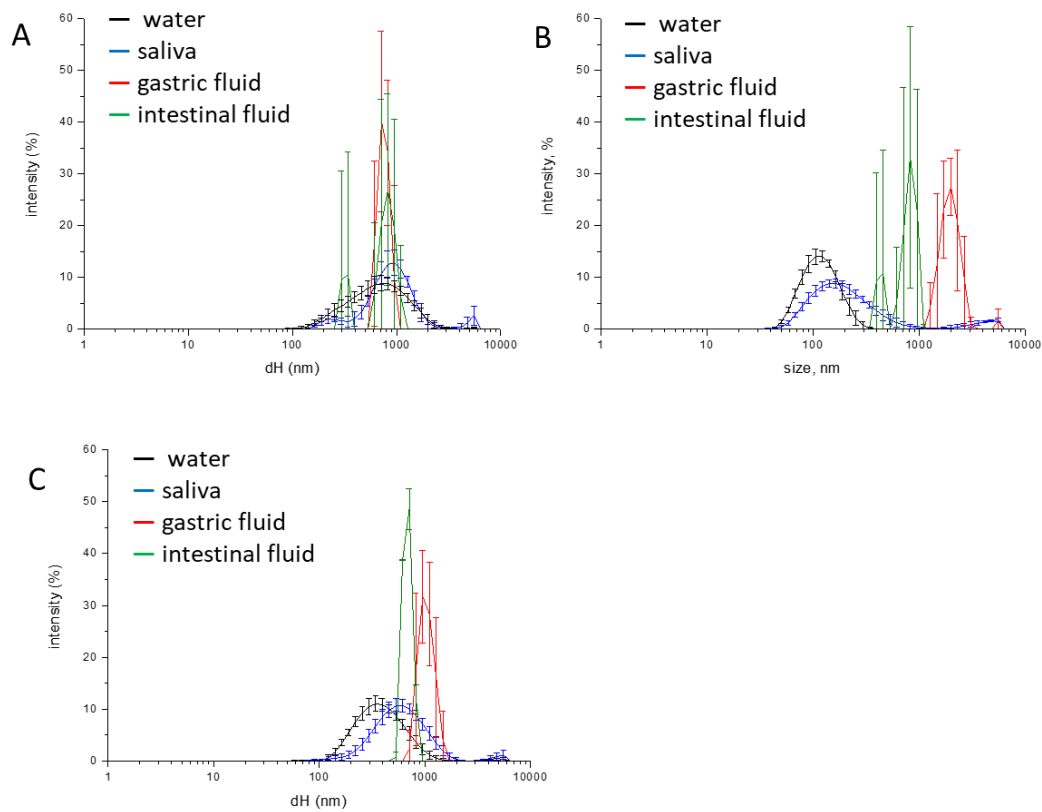


Figure S1. Size distribution monitored by DLS of A) TiO₂-NM1; B) TiO₂-NM2; C) TiO₂-FG during the digestion cascade in the absence of proteins. The hydrodynamic diameters (d_H) distribution (% intensity) is expressed as mean value of 5 measurements in three independent experiments \pm SD.

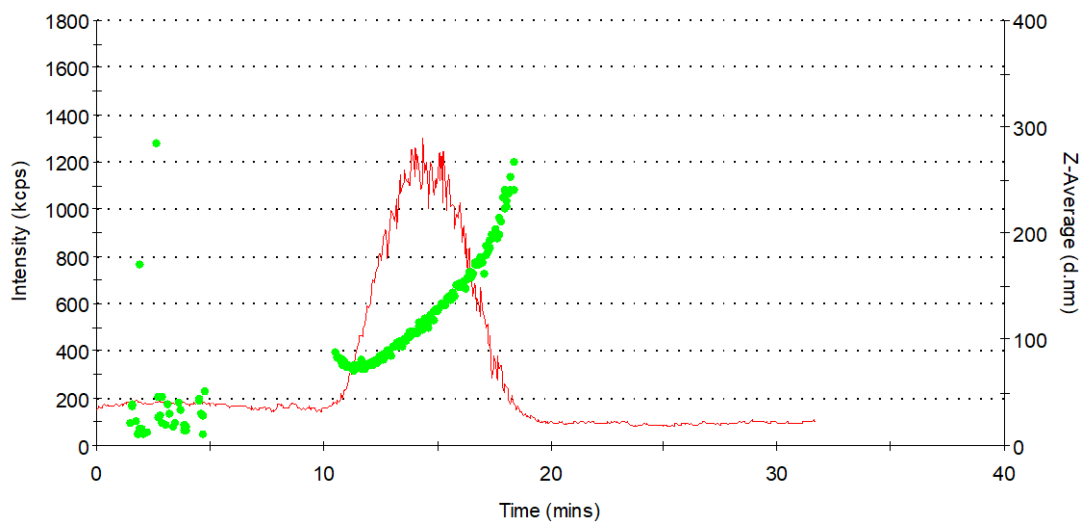


Figure S2. AF4/DLS fractogram of TiO₂-S. Concentration of the injected suspension: 50 mg L⁻¹. Mean diameter: 77.3 nm.

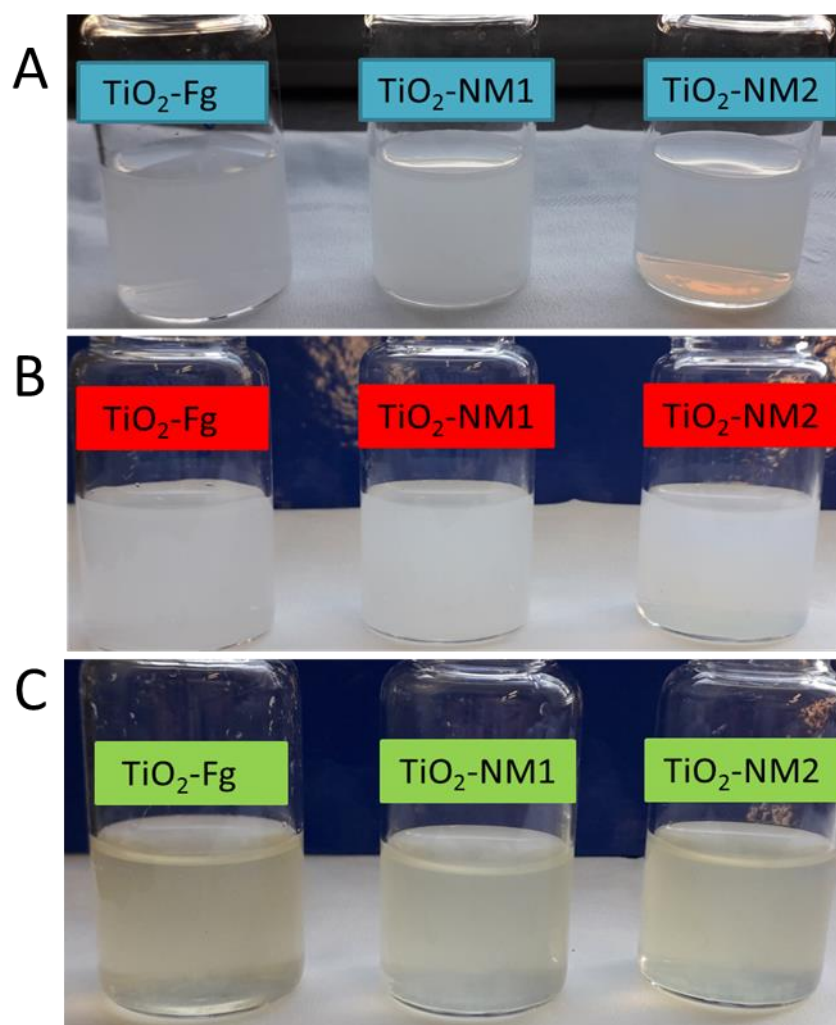


Figure S3. Appearance of the suspensions of the TiO₂ samples during the digestion cascade. A) saliva; B) gastric fluid; C) intestinal fluid.