Supplementary Information

Nucleophilic Dearomatisation of N-Heteroaromatics Enabled by Lewis Acids and Copper Catalysis

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1. General experimental information

All reactions using oxygen- and/or moisture-sensitive materials were carried out with anhydrous solvents under a nitrogen atmosphere using standard Schlenk techniques. Flash column chromatography was performed using Merck 60 Å 230-400 mesh silica gel. Thin layer chromatography was performed using 0.25 mm E. Merck silica plates (60F-254). The products were visualized by phosphomolybdic acid (PMA) and KMnO₄ staining. NMR data was collected on Varian VXR400 (¹H at 400 MHz; ¹³C at 100.58 MHz) equipped with a 5 mm *z*-gradient broadband probe. Chemical shifts are reported in parts per million (ppm) relative to residual solvent peak (CDCl₃, ¹H: 7.26 ppm; ¹³C: 77.16 ppm). Coupling constants are reported in Hertz. Multiplicity is reported with the usual abbreviations (s: singlet, bs: broad singlet, d: doublet, t: triplet, m: multiplet). Variabletemperature NMR spectra were acquired on a Bruker Advance III spectrometer paired with an Ascend 400 MHz magnet and BBFO dual-resonance probe. All temperatures were calibrated prior to acquisition with an external pure MeOH reference. ¹H-¹³C HSQCED spectra were recorded with the ${}^{1}J_{CH}$ constant set to 145 Hz while the ${}^{1}H$ - ${}^{13}C$ HMBC spectra were recorded with a ${}^{n}J_{XH}$ constant set to 8 Hz. Exact mass spectra were recorded on a LTQ Orbitrap XL apparatus with ESI ionization. Enantiomeric excess (ee) were determined by chiral HPLC analysis using a Shimadzu LC-10ADVP HPLC equipped with a Shimadzu SPD-M10AVP diode array detector.

2. Chemicals

Unless otherwise indicated, reagents and substrates were purchased from commercial sources and used as received. The substrates for synthesis of products **5j** and **5o** were synthesized according to the literature procedures respectively,¹ the rest are commercially available. Dry solvents were freshly collected from a dry solvent purification system prior to use. Inert atmosphere experiments were performed with standard Schlenk techniques with dried (P₂O₅) nitrogen gas. PhPrMgBr (2.6 M in Et₂O) and Hept-6-en-1-ylMgBr (1.4 M in Et₂O) was prepared from the corresponding alkyl bromides and Mg activated with I₂ in Et₂O. Me₃SiCH₂CH₂MgBr (0.4 M in 2-Me-THF) was prepared from the corresponding alkyl bromide and Mg in 2-Me-THF at 0 °C. Other Grignard reagents were purchased from Sigma-Aldrich: EtMgBr (3 M in Et₂O); PentMgBr, *i*BuMgBr, *i*PentMgBr (2 M in Et₂O); *i*PrMgBr (3 M in 2-Me-THF). The Grignard reagents were titrated by ¹H NMR before use. Chiral ligands (L1–L4) were purchased from Sigma-Aldrich and Strem Chemicals. All reported compounds were fully characterized by ¹H and ¹³C NMR and HRMS techniques.

3. Computational details

We have used the Density Functional Theory (DFT) in the Kohn-Sham formulation² to optimize all the stationary points presented in this manuscript. Geometries of all the stationary points were fully optimized at the M06³/def2svpp⁴ computational level. The effect of solvent was modeled using the polarizable continuum model (PCM)⁵ with the default parameters implemented in the Gaussian 09 package.⁶ All geometry optimizations have been optimized using tight convergence criteria in the SCF and requesting a pruned (99.590) grid to guarantee the accuracy of the reported results. Moreover, the calculations were performed considering 1 atm and 195 K to properly simulate the reaction conditions.

Harmonic analysis was used to establish the nature of all optimized structures as either minima or transition structures. For all stationary points, the stability of the wave function was also confirmed.⁷ Moreover, for all the copper species an analysis of the spin annihilation was also performed being the value 0.00.

IRC⁸ calculations were conducted for important transition states to ensure their connectivity with the expected reactants and products. The stability of the wave function was analyzed for all the presented stationary points.⁹ When the substrates showed conformational freedom, conformational analysis was performed manually, it must be indicated that only the most stable conformer of each stationary point was considered and reported unless otherwise indicated.

The visualization of the reported structures was performed using MOLDEN.¹⁰ The representation of the structures here presented were generated using CYLView.¹¹

The Grignard reagents used in this work are bought in ethereal solvents and then diluted. In line with this, we have confirmed the superior stability of the Grignard reagents with two solvent molecules (see Solvation studies).¹² Since solvent can be exchanged during the reaction process, we have evaluated the relative stability of each of the proposed stationary points here described with none, one and two solvent molecules directly coordinated to the magnesium center, for each step of the reaction, obtaining that the coordination two ethereal molecules (at -78 °C) lead to the more stable systems. Thus, along the mechanistic study two solvent molecules are considered in the first coordination sphere of the magnesium.

Moreover, the BF_3 was selected as the LA and L1 as a ligand for the molecular simulations.

4. Optimization of reaction conditions

Table S1. Optimization of the copper salt and LA for the enantioselective C-4-addition of EtMgBr to quinoline followed by trapping with acetyl chloride^a



^aReaction conditions: 0.1 M of **1a** in CH₂Cl₂, Cu(I) (5 mol%), **L1** (6 mol%), LA (2 equiv.), EtMgBr (2 equiv.) at -78 °C for 16 h, then ClCOMe (5 equiv.) at RT for 2 h. ^bThe ratio was determined by ¹H NMR of reaction crude. ^cTrimethoxybenzene was added after work-up and used as the internal standard. ^dEnantiomeric excess was determined by HPLC on a chiral stationary phase. ^eIn this case the reaction was quenched directly without trapping with ClCOMe. ^fIn this case THF (4 mL) was added after addition of ClCOMe. ^gIn this case the data reported corresponds to the isolated yield.

Table S2. Optimization of the of LA and EtMgBr quantities as well as the nature of solvent for the enantioselective C-4-addition of EtMgBr to quinoline followed by trapping with acetyl chloride^a

	1a	+ ΒF₃∙Et₂Ο + EtM	L1 (6 mol CuTc (5 mo -78 °C, 16 CICOMe, THF,	%) l %) h RT, 2 h	2a	
Entry	Solvent	BF ₃ ·Et ₂ O [equiv.]	EtMgBr [equiv.]	1a:2a:3a ^b	NMR yield (3a) [%] ^c	$ee~(3a)~[\%]^d$
1	CH ₂ Cl ₂	2	2	1:0:99	58 ^e	>99
2	CH_2Cl_2	1.2	2	1:0:99	54	>99
3	CH_2Cl_2	0.4	2	35:0:65	47	98
4	CH_2Cl_2	2	1.2	34:0:66	41	>99
5	CH_2Cl_2	1.2	1.2	35:0:65	39	>99
$6^{\rm f}$	CH_2Cl_2	2	2	1:0:99	57	>99
7 ^g	CH_2Cl_2	2	2	4:0:96	54	>99
$8^{\rm h}$	CH_2Cl_2	2	2	5:0:95	51	>99
9^i	THF	2	2	12:81:7	—	42
10^{i}	2-Me-THF	2	2	1:0:99	34	>99
11	MTBE	2	2	30:0:70	—	>99
12	MTBE	3	3	2:0:98	52	>99
13	Et ₂ O	3	3	2:0:98	41	>99
14	Toluene	3	3	1:0:99	44	>99

^aReaction conditions: 0.1 M of **1a**, CuTc (5 mol%), **L1** (6 mol%), $BF_3 \cdot Et_2O$, EtMgBr at -78 °C for 16 h, then ClCOMe (5 equiv.) and THF (4 mL) at RT for 2 h. ^bThe ratio was determined by ¹H NMR of reaction crude. ^cTrimethoxybenzene was added after work-up and used as the internal standard. ^dEnantiomeric excess was determined by HPLC on a chiral stationary phase. ^eIn this case the isolated yield is reported. ^fEtMgBr was diluted to 1 mL with MTBE and added with syringe pump in 1 h. ^gEtMgBr was first added followed by addition of $BF_3 \cdot Et_2O$. ^bThe mixture of chiral catalyst **L1**/Cu(I) and EtMgBr was added to the mixture of substrate and $BF_3 \cdot Et_2O$. ⁱIn this case THF (4 mL) was not needed after addition of ClCOMe.

Table S3. Optimization of the reductant for the enantioselective C-4-addition of EtMgBr to quinoline followed by reduction^a



^aReaction conditions: 0.2 mmol of **1a** in CH_2Cl_2 (2 mL), CuTc (5 mol%), **L1** (6 mol%), $BF_3 \cdot Et_2O$ (2 equiv.), EtMgBr (2 equiv.) at -78 °C for 2 h, then reductant (5 equiv.) at RT for 16 h. ^bConversion was determined by ¹H NMR of reaction crude. ^cEnantiomeric excess was determined by HPLC on a chiral stationary phase. ^dDuring work-up, the solvent was evaporated before saturated NaHCO₃ aqueous solution was added to the residue and the mixture was extracted with CH_2Cl_2 . ^eProduct is not pure.

5. Solvation studies for the computational mechanistic study

Table S4. Computed relative energies for the solvation of the EtMgBr species. Calculations were performed at the $PCM(CH_2Cl_2)/M06/def2svpp$ computational level using the Gaussian 09 program. The thermochemistry was obtained at 1 atm and 195 K.

ID	ΔG^{a}
EtMgBr	0.00
EtMgBr·Et ₂ O	-7.70
EtMgBr·2Et ₂ O	-8.84

^aEnergies are calculated with respect to EtMgBr and two molecules of Et₂O and are expressed in kcal/mol.

ID	ImFreqs	Stable	SCF ^a	SCF+ZPVE	H ^b	G ^c
EtMgBr	_	Yes	-2852.92586678	-2852.862580	-2852.858414	-2852.881361
Et ₂ O	-	Yes	-233.296065097	-233.161050	-233.156875	-233.179034
EtMgBr·Et ₂ O	_	Yes	-3086.24895436	-3086.048164	-3086.040115	-3086.072676
EtMgBr·2Et ₂ O	_	Yes	-3319.207607	-3319.19668	-3319.195446	-3319.252401

Table S5. Summary of the energies for solvation studies of the Grignard reagent.

^aEnergies are expressed in a.u. and the imaginary frequencies in cm⁻¹. ^bH represents enthalpies. ^cG Represents Gibbs free

Energies.

6. Lewis acid promoted copper-catalyzed enantioselective dearomatization of quinolines

General procedure for catalytic reactions

In a flame-dried Schlenk tube equipped with septum and magnetic stirring bar, the substrate (0.2 mmol, 1 equiv.), copper salt (0.01 mmol, 5 mol%), and ligand L (0.012 mmol, 6 mol%) were dissolved in the solvent (2 mL) and stirred under nitrogen atmosphere for 20 min. at RT. The mixture was cooled to -78 °C and LA (1.2–3 equiv.) was added. After 20 min., RMgBr (2–3 equiv.) was added dropwise in about 1 min.

General trapping procedure

After stirring for 16 h at -78 °C, acetyl chloride (71 µL, 1 mmol, 5 equiv.) was added and the reaction mixture was warmed up to RT. After stirring for 2 h, the resulting reaction mixture was quenched with saturated NaHCO₃ aqueous solution (2 mL) and stirred at RT for 1 h to hydrolyze the remaining acetyl chloride. The mixture was extracted with CH₂Cl₂ (10 mL × 3). The combined organic phase was dried over MgSO₄, filtered and evaporated on rotary evaporator. The crude was purified by flash chromatography on the silica gel. (Note: when BF₃·Et₂O was used as LA, 4 mL of THF has to be added after addition of acetyl chloride, because when using BF₃·Et₂O trapping was only successful in THF.)

General reduction procedure

After stirring for 2–16 h at –78 °C, BH₃ (1 M in THF, 1 mmol, 5 equiv.) was added. After stirring for 16 h at –78 °C, the resulting reaction mixture was quenched with saturated NaHCO₃ aqueous solution (2 mL) and warmed up to RT. The mixture was stirred at RT for 1 h to hydrolyze the remaining BH₃·THF, and extracted with CH₂Cl₂ (10 mL \times 3). The combined organic phase was dried over

MgSO₄, filtered and evaporated on rotary evaporator. The crude was purified by flash chromatography on the silica gel.

Procedure for the reaction with the reduced catalyst loading

In a flame-dried three-neck round-bottom flask equipped with septum and mechanistic stirring bar, CuTc (0.01 mmol) and ligand (*R*,*R*)-L1 (0.012 mmol) were dissolved in CH₂Cl₂ (10 mL) and the mixture was stirred under nitrogen atmosphere for 20 min. at RT, forming catalyst solution 1. 2 mL of catalyst solution 1 (2 µmol, 1 mol%) was transferred to substrate 1a (0.2 mmol, 1 equiv.) to perform the reaction with 1 mol% catalyst. 0.2 mL of catalyst solution 1 (0.2 µmol, 0.1 mol%) was transferred to substrate 1a (0.2 mmol, 1 equiv.) dissolved in CH₂Cl₂ (2 mL) to perform the reaction with 0.1 mol% catalyst. 1 mL of catalyst solution 1 (1 µmol) was transferred to another round-bottom flask and diluted with CH₂Cl₂ to 10 mL to form catalyst solution 2. 0.2 mL of catalyst solution 2 (0.02 µmol, 0.01 mol%) was transferred to substrate 1a (0.2 mmol, 1 equiv.) dissolved in CH₂Cl₂ (2 mL) to perform the reaction with 0.01 mol% catalyst.

Procedure for the preparative scale (10 mmol) reaction

In a flame-dried three-neck round-bottom flask equipped with septum and mechanistic stirring bar, the substrate **1a** (10 mmol, 1 equiv.), CuTc (0.01 mmol, 0.1 mol%) and ligand (*R*,*R*)-**L1** (0.012 mmol, 0.12 mol%) were dissolved in CH₂Cl₂ (50 mL) and the mixture was stirred under nitrogen atmosphere for 20 min. at RT. The mixture was cooled to -78 °C and BF₃·Et₂O (20 mmol, 2 equiv.) was added. After 20 min., EtMgBr (3 M in Et₂O, 20 mmol, 2 equiv.) was added with syringe pump in 15 min. After stirring for 2 h at -78 °C, BH₃ (1 M in THF, 50 mmol, 5 equiv.) was added. After stirring for 16 h at -78 °C, the resulting reaction mixture was quenched with saturated NaHCO₃ aqueous solution (20 mL) and warmed up to RT. The mixture was stirred at RT for 1 h to hydrolyze the BH₃·THF, and extracted with CH₂Cl (30 mL × 3). The combined organic phase was dried over MgSO₄, filtered and evaporated on rotary evaporator. The crude was purified by flash chromatography on the silica gel (SiO₂, pentane:Et₂O = 20:1) to yield the product **4a** as a colorless oil [71% yield, 96% *ee*].

Procedure for the synthesis of racemic 2a and 3a

In a flame-dried Schlenk tube equipped with septum and magnetic stirring bar, **1a** (0.2 mmol, 1 equiv.) was dissolved in CH₂Cl₂ (2 mL). The temperature was cooled down to -78 °C and EtMgBr (3 M in Et₂O, 0.4 mmol, 2 equiv.) was added. After stirring for 16 h, acetyl chloride (1 mmol, 5 equiv.) was added and the reaction mixture was warmed up to RT. After stirring for 2 h, the resulting reaction mixture was quenched with saturated NaHCO₃ aqueous solution (2 mL) and stirred at RT for 1 h to

hydrolyze the remaining acetyl chloride. The mixture was extracted with CH_2Cl (10 mL × 3). The combined organic phase was dried over MgSO₄, filtered and evaporated on rotary evaporator. Products **2a** and **3a** were obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 10:1) [**1a**:**2a**:**3a** = 37:53:10 in the crude product].

General procedure for the synthesis of racemic 4-substituted 1,2,3,4-tetrahydroquinolines

In a flame-dried Schlenk tube equipped with septum and magnetic stirring bar, the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%) and CH₂Cl₂ (2 mL) were stirred under nitrogen atmosphere for 20 min. at RT. The mixture was cooled to -78 °C and BF₃·Et₂O (0.6 mmol, 3 equiv.) was added. After 20 min., RMgBr (0.6 mmol, 3 equiv.) was added dropwise in about 1 min. After stirring for 16 h at -78 °C, NaBH₄ (1 mmol, 5 equiv.) was added and the reaction mixture was warmed up to RT. After stirring for 2 h, the resulting reaction mixture was quenched with saturated NaHCO₃ aqueous solution (2 mL) and stirred at RT for 1 h to hydrolyze the remaining NaBH₄. The mixture was extracted with CH₂Cl₂ (10 mL × 3). The combined organic phase was dried over MgSO₄, filtered and evaporated on rotary evaporator. The crude was purified by flash chromatography on the silica gel.

Specific experimental details and product characterization

1-acetyl-2-ethyl-1,2-dihydroquinoline (2a)



¹H NMR (CDCl₃, 400 MHz): δ 7.21-6.98 (m, 4H, CH_{Ar}), 6.43 (d, J = 9.5 Hz, 1H, CHCH=CH), 6.11 (dd, J = 9.6, 5.8 Hz, 1H, CHCH=), 5.38-5.17 (m, 1H, CH=CHCH), 2.18 (s, 3H, COCH₃), 1.51-1.37 (m, 1H, CH₃CHH), 1.37-1.21 (m, 1H, CH₃CHH), 0.83 (t, J = 7.4 Hz, 3H, CH₂CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 170.2, 135.3, 132.5, 128.6, 127.0, 126.3, 125.4, 124.8, 124.3, 51.7, 25.3, 22.8, 9.9.

HRMS (ESI, m/Z): calcd. for C₁₃H₁₆NO [M+H]⁺: 202.1226, found: 202.1225.

HPLC: Chiracel-ODH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 234 nm. Retention time (min): 13.6 and 16.2.



Figure S1. ¹H-¹³C-HSQCED spectrum of product 2a.



Figure S2. ¹H-¹³C-HMBC spectrum of product **2a**. Correlations of Hb with Ca, Cc and Cd are highlighted. No correlation of Hb with any aromatic carbon was observed, proving that product **2a** is the C-2-addition product.

(S)-1-acetyl-4-ethyl-1,4-dihydroquinoline (3a)



The reaction was performed with **1a** (24 μ L, 0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-L1 (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), EtMgBr (3 M in Et₂O, 0.4 mmol, 2 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 16 h, and following **General trapping procedure**. Product **3a** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 10:1) [99% conversion, 58% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 7.85 (d, J = 9.3 Hz, 1H, CH_{Ar}), 7.27-7.19 (m, 1H, CH_{Ar}), 7.19-7.08 (m, 2H, CH_{Ar}), 6.89 (d, J = 7.5 Hz, 1H, NCH=), 5.52 (dd, J = 7.5, 5.9 Hz, 1H, NCH=CH), 3.30-3.23 (m, 1H, CH=CHCH), 2.36 (s, 3H, COCH₃), 1.69-1.51 (m, 2H, CH₃CH₂), 0.88 (t, J = 7.4 Hz, 3H, CH₂CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 168.4, 136.8, 133.1, 128.3, 127.0, 126.0, 125.4, 123.0, 116.6, 39.9, 30.3, 24.2, 10.8.

HRMS (ESI, m/Z): calcd. for $C_{13}H_{16}NO [M+H]^+$: 202.1226, found: 202.1224.

HPLC: Chiracel-ASH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 251 nm. Retention time (min): 11.1 (major) and 12.4 (minor).



Figure S3. ¹H-¹³C-HSQCED spectrum of product 3a.



Figure S4. ¹H-¹³C-HMBC spectrum of product **3a**. Correlations of Hb with Ca, Cc, Cd and Ce are highlighted. Hb correlates not only with Ca, Cc and Cd but also with the aromatic Ce, proving that product **3a** is the C-4-addition product.

(S)-4-Ethyl-1,2,3,4-tetrahydroquinoline (4a)¹³



The reaction was performed with **1a** (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-**L1** (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), EtMgBr (3 M in Et₂O, 0.4 mmol, 2 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following **General reducing procedure**. Product **4a** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 20:1) [99% conversion, 83% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 7.07-7.03 (m, 1H, CH_{Ar}), 7.02-6.96 (m, 1H, CH_{Ar}), 6.68-6.61 (m, 1H, CH_{Ar}), 6.49 (dd, J = 8.0, 1.2 Hz, 1H, CH_{Ar}), 3.76 (bs, 1H, NH), 3.38-3.23 (m, 2H, NHCH₂), 2.72-2.63 (m, 1H, CH₃CH₂CH), 2.00-1.90 (m, 1H, CH₂), 1.89-1.71 (m, 2H, CH₂), 1.63-1.50 (m, 1H, CH₂), 1.01 (t, J = 7.4 Hz, 3H, CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 144.3, 129.3, 126.9, 125.7, 116.8, 114.3, 38.6, 37.3, 29.4, 25.8, 11.8.

HPLC: Chiracel-ADH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 247 nm. Retention time (min): 11.3 (major) and 12.8 (minor).

(S)-4-Pentyl-1,2,3,4-tetrahydroquinoline (4b)



The reaction was performed with **1a** (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-**L1** (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), PentMgBr (2 M in Et₂O, 0.6 mmol, 3 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following **General reducing procedure**. Product **4b** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 30:1) [99% conversion, 96% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 7.06-7.02 (m, 1H, CH_{Ar}), 7.01-6.95 (m, 1H, CH_{Ar}), 6.67-6.61 (m, 1H, CH_{Ar}), 6.49 (dd, J = 8.0, 1.2 Hz, 1H, CH_{Ar}), 3.82 (bs, 1H, NH), 3.34 (dt, J = 10.7, 3.6 Hz, 1H,

NHCH*H*), 3.26 (dt, *J* = 11.1, 4.7 Hz, 1H, NHC*H*H), 2.80-2.71 (m, 1H, NHCH₂CH₂C*H*), 2.00-1.89 (m, 1H, C*H*₂), 1.87-1.78 (m, 1H, C*H*₂), 1.75-1.64 (m, 1H, C*H*₂), 1.60-1.27 (m, 7H, C*H*₂), 0.93 (t, *J* = 6.6 Hz, 3H, C*H*₃).

¹³C NMR (CDCl₃, 100 MHz): δ 144.3, 129.2, 126.8, 125.9, 116.7, 114.2, 38.6, 36.8, 35.6, 32.2, 26.8, 26.2, 22.8, 14.3.

HRMS (ESI, m/Z): calcd. for C₁₄H₂₂N [M+H]⁺: 204.1747, found: 204.1746.

HPLC: Chiracel-ADH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 247 nm. Retention time (min): 10.0 (major) and 11.1 (minor).

(*R*)-4-(1-methylethyl)-1,2,3,4-tetrahydroquinoline (4c)¹⁴



The reaction was performed with **1a** (0.2 mmol, 1 equiv.), CuTc (0.02 mmol, 10 mol%), ligand (*R*,*R*)-L1 (0.024 mmol, 12 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), *i*PrMgBr (3 M in 2-Me-THF, 0.4 mmol, 2 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 16 h, and following **General reducing procedure**. Product **4c** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 30:1) [99% conversion, 82% yield, 80% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 7.06-6.96 (m, 2H, CH_{Ar}), 6.66-6.60 (m, 1H, CH_{Ar}), 6.48 (d, *J* = 7.9 Hz, 1H, CH_{Ar}), 3.82 (bs, 1H, NH), 3.40-3.26 (m, 2H, NHCH₂), 2.60-2.53 (m, 1H, CH₃CHCH), 2.11-2.00 (m, 1H, CH₃CH), 2.00-1.90 (m, 1H, NHCH₂CHH), 1.88-1.77 (m, 1H, NHCH₂CHH), 1.01 (d, *J* = 6.8 Hz, 3H, CH₃), 0.90 (d, *J* = 6.8 Hz, 3H, CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 144.7, 129.4, 126.8, 124.5, 116.3, 114.1, 42.3, 39.6, 30.5, 23.0, 21.6, 18.6.

HRMS (ESI, m/Z): calcd. for C₁₂H₁₈N [M+H]⁺: 176.1434, found: 176.1428.

HPLC: Chiracel-ADH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 247 nm. Retention time (min): 10.2 (major) and 14.4 (minor).

(S)-4-(2-methylpropyl)-1,2,3,4-tetrahydroquinoline (4d)



The reaction was performed with **1a** (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-L1 (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), *i*BuMgBr (2 M in Et₂O, 0.6 mmol, 3 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following **General reducing procedure**. Product **4d** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 30:1) [94% conversion, 72% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 7.05-6.94 (m, 2H, CH_{Ar}), 6.66-6.60 (m, 1H, CH_{Ar}), 6.49 (dd, J = 8.0, 1.2 Hz, 1H, CH_{Ar}), 3.85 (bs, 1H, NH), 3.34 (dt, J = 10.9, 3.4 Hz, 1H, NHCHH), 3.26 (dt, J = 11.2, 4.6 Hz, 1H, NHCHH), 2.89-2.80 (m, 1H, NHCH₂CH₂CH), 2.00-1.88 (m, 1H, CH₂), 1.83-1.71 (m, 2H, CH₂), 1.55-1.41 (m, 2H, CH₂, CH₃CH), 1.02-0.95 (m, 6H, CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 144.3, 129.1, 126.8, 126.2, 116.8, 114.2, 46.5, 38.3, 33.1, 26.1, 24.9, 23.8, 22.1.

HRMS (ESI, m/Z): calcd. for C₁₃H₂₀N [M+H]⁺: 190.1590, found: 190.1585.

HPLC: Chiracel-ADH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 247 nm. Retention time (min): 10.0 (major) and 11.7 (minor).

(S)-4-(3-methylbutyl)-1,2,3,4-tetrahydroquinoline (4e)



The reaction was performed with **1a** (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-**L1** (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), *i*PentMgBr (2 M in Et₂O, 0.6 mmol, 3 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following **General reducing procedure**. Product **4e** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 30:1) [99% conversion, 90% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 7.08-7.02 (m, 1H, CH_{Ar}), 7.02-6.96 (m, 1H, CH_{Ar}), 6.68-6.62 (m, 1H, CH_{Ar}), 6.49 (dd, J = 8.0, 1.2 Hz, 1H, CH_{Ar}), 3.85 (bs, 1H, NH), 3.34 (dt, J = 10.4, 5.5 Hz, 1H, NHCHH), 3.27 (td, J = 10.9, 4.7 Hz, 1H, NHCHH), 2.78-2.69 (m, 1H, NHCH₂CH₂CH), 2.00-1.89 (m, 1H, CH₂), 1.88-1.79 (m, 1H, CH₂), 1.77-1.66 (m, 1H, CH₂), 1.65-1.46 (m, 2H, CH₂, CH₃CH), 1.40-1.24 (m, 2H, CH₂), 0.96-0.90 (m, 6H, CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 144.3, 129.2, 126.8, 125.9, 116.8, 114.2, 38.5, 36.4, 35.9, 34.6, 28.4, 26.2, 23.1, 22.6.

HRMS (ESI, m/Z): calcd. for C₁₄H₂₂N [M+H]⁺: 204.1747, found: 204.1740.

HPLC: Chiracel-ADH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 247 nm. Retention time (min): 9.8 (major) and 10.9 (minor).

(S)-4-(3-Phenylpropyl)-1,2,3,4-tetrahydroquinoline (4f)



The reaction was performed with **1a** (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-L1 (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), Ph(CH₂)₃MgBr (2.6 M in Et₂O, 0.6 mmol, 3 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following **General reducing procedure**. Product **4f** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 30:1) [99% conversion, 83% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 7.35-7.27 (m, 2H, CH_{Ar}), 7.25-7.18 (m, 3H, CH_{Ar}), 7.04-6.96 (m, 2H, CH_{Ar}), 6.67-6.60 (m, 1H, CH_{Ar}), 6.49 (d, *J* = 7.9 Hz, 1H, CH_{Ar}), 3.85 (bs, 1H, NH), 3.37-3.22 (m, 2H, NHCH₂), 2.84-2.75 (m, 1H, NHCH₂CH₂CH), 2.75-2.59 (m, 2H, PhCH₂), 2.01-1.90 (m, 1H, CH₂), 1.88-1.68 (m, 4H, CH₂), 1.67-1.53 (m, 1H, CH₂).

¹³C NMR (CDCl₃, 100 MHz): δ 144.3, 142.7, 129.2, 128.5, 128.4, 126.9, 125.8, 125.5, 116.8, 114.2, 38.5, 36.4, 36.2, 35.6, 29.1, 26.3.

HRMS (ESI, m/Z): calcd. for C₁₈H₂₂N [M+H]⁺: 252.1747, found: 252.1740.

HPLC: Chiracel-ADH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 247 nm. Retention time (min): 18.3 (minor) and 22.4 (major).

(S)-4-(Hept-6-en-1-yl)-1,2,3,4-tetrahydroquinoline (4g)



The reaction was performed with **1a** (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-**L1** (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), Hept-6-en-1-ylMgBr (1.4 M in Et₂O, 0.6 mmol, 3 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following **General reducing procedure**. Product **4g** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 30:1) [99% conversion, 84% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 7.06-6.94 (m, 2H, CH_{Ar}), 6.67-6.59 (m, 1H, CH_{Ar}), 6.48 (d, J = 7.9 Hz, 1H, CH_{Ar}), 5.83 (ddt, J = 16.9, 10.1, 6.7 Hz, 1H, CH₂=CH), 5.07-4.92 (m, 2H, CH₂=), 3.85 (bs, 1H, NH), 3.34 (dt, J = 10.7, 3.6 Hz, 1H, NHCHH), 3.26 (dt, J = 10.9, 4.7 Hz, 1H, NHCHH), 2.80-2.69 (m, 1H, NHCH₂CH₂CH), 2.13-2.01 (m, 2H, CH₂), 2.00-1.88 (m, 1H, CH₂), 1.86-1.76 (m, 1H, CH₂), 1.75-1.64 (m, 1H, CH₂), 1.59-1.30 (m, 7H, CH₂).

¹³C NMR (CDCl₃, 100 MHz): δ 144.3, 139.3, 129.2, 126.8, 125.8, 116.7, 114.3, 114.2, 38.6, 36.8, 35.6, 33.9, 29.4, 29.1, 27.0, 26.3.

HRMS (ESI, m/Z): calcd. for C₁₆H₂₄N [M+H]⁺: 230.1903, found: 230.1895.

HPLC: Chiracel-ADH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 247 nm. Retention time (min): 9.9 (major) and 11.1 (minor).

(R)-4-(2-Trimethylsilylethyl)-1,2,3,4-tetrahydroquinoline (4h)



The reaction was performed with **1a** (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-**L1** (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), Me₃Si(CH₂)₂MgBr (0.4 M in 2-Me-THF, 0.6 mmol, 3 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 16 h, and following **General reducing procedure**. Product **4h** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 40:1) [99% conversion, 67% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 7.07-7.02 (m, 1H, CH_{Ar}), 7.02-6.95 (m, 1H, CH_{Ar}), 6.68-6.61 (m, 1H, CH_{Ar}), 6.53-6.47 (m, 1H, CH_{Ar}), 3.74 (bs, 1H, NH), 3.35-3.22 (m, 2H, NHCH₂), 2.72-2.62 (m, 1H, NHCH₂CH₂CH₂CH), 1.98-1.85 (m, 2H, CH₂), 1.77-1.64 (m, 1H, CH₂), 1.59-1.46 (m, 1H, CH₂), 0.67 (dt, J = 13.6, 4.4 Hz, 1H, SiCHH), 0.67 (ddd, J = 14.1, 12.6, 4.8 Hz, 1H, SiCHH), 0.01 (s, 9H, CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 144.4, 129.3, 126.9, 125.8, 116.8, 114.2, 38.5, 38.4, 30.9, 25.3, 13.9, -1.56.

HRMS (ESI, m/Z): calcd. for C₁₄H₂₄NSi [M+H]⁺: 234.1673, found: 234.1665.

HPLC: Chiracel-ADH, *n*-heptane/*i*-PrOH 98:2, 0.5 mL/min., 40 °C, detection at 247 nm. Retention time (min): 9.7 (major) and 11.2 (minor).

(S)-4-Ethyl-6-fluoro-1,2,3,4-tetrahydroquinoline (5a)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (1.91 mg, 0.01 mmol, 5 mol%), ligand (*R*,*R*)-**L1** (6.08 mg, 0.012 mmol, 6 mol%), BF₃·Et₂O (49 μ L, 0.4 mmol, 2 equiv.), EtMgBr (3 M in Et₂O, 0.4 mmol, 2 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following **General**

reducing procedure. Product **5a** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 20:1) [99% conversion, 70% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 6.76 (dd, J = 9.8, 2.9 Hz, 1H, CH_{Ar}), 6.73-6.65 (m, 1H, CH_{Ar}), 6.40 (dd, J = 8.7, 4.9 Hz, 1H, CH_{Ar}), 3.74 (bs, 1H, NH), 3.33-3.19 (m, 2H, NHCH₂), 2.68-2.59 (m, 1H, CH₃CH₂CH), 1.96-1.86 (m, 1H, CH₂), 1.84-1.68 (m, 2H, CH₂), 1.61-1.47 (m, 1H, CH₂), 0.98 (t, J = 7.4 Hz, 3H, CH₃).

¹³C NMR (CDCl₃, 400 MHz): δ 155.5 (d, *J* = 234.1 Hz), 140.6 (d, *J* = 1.9 Hz), 127.0 (d, *J* = 6.2 Hz), 115.3 (d, *J* = 21.7 Hz), 114.9 (d, *J* = 7.4 Hz), 113.4 (d, *J* = 22.4 Hz), 39.0, 37.4, 29.2, 25.8, 11.6.

¹⁹F NMR (CDCl₃, 400 MHz): δ –128.2.

HRMS (ESI, m/Z): calcd. for C₁₁H₁₅NF [M+H]⁺: 180.1183, found: 180.1182.

HPLC: Chiracel-ADH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 242 nm. Retention time (min): 12.7 (major) and 14.6 (minor).

(S)-6-Bromo-4-ethyl-1,2,3,4-tetrahydroquinoline (5b)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-L1 (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), EtMgBr (3 M in Et₂O, 0.4 mmol, 2 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following General reducing procedure. Product **5b** was obtained as a white solid after column chromatography (SiO₂, pentane:Et₂O = 15:1) [99% conversion, 70% yield, >99% *ee*]. **5b** was dissolved in the mixed solvent of CH₂Cl₂ and pentane, and left for slow evaporation at RT in the air to afford the colorless single crystal.

¹H NMR (CDCl₃, 400 MHz): δ 7.14-7.10 (m, 1H, CH_{Ar}), 7.03 (dd, J = 8.5, 2.3 Hz, 1H, CH_{Ar}), 6.37 (d, J = 8.5 Hz, 1H, CH_{Ar}), 3.87 (bs, 1H, NH), 3.35-3.20 (m, 2H, NHCH₂), 2.67-2.58 (m, 1H, CH₃CH₂CH), 1.94-1.66 (m, 3H, CH₂), 1.59-1.45 (m, 1H, CH₂), 1.61-1.47 (m, 1H, CH₂), 0.98 (t, J = 7.4 Hz, 3H, CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 143.3, 131.6, 129.5, 127.5, 115.6, 108.1, 38.5, 37.3, 29.1, 25.3, 11.6.

HRMS (ESI, m/Z): calcd. for C₁₁H₁₅NBr [M+H]⁺: 240.0382, found: 240.0383.

HPLC: Chiracel-ADH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 253 nm. Retention time (min): 13.4 (major) and 15.2 (minor).

(S)-4-Ethyl-6-methyl-1,2,3,4-tetrahydroquinoline (5c)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-L1 (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), EtMgBr (3 M in Et₂O, 0.4 mmol, 2 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following General reducing procedure. Product **5c** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 30:1) [98% conversion, 68% yield, 99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 6.86 (d, J = 2.0 Hz, 1H, CH_{Ar}), 6.80 (dd, J = 8.1, 2.0 Hz, 1H, CH_{Ar}), 6.42 (d, J = 8.0 Hz, 1H, CH_{Ar}), 3.68 (bs, 1H, NH), 3.35-3.19 (m, 2H, NHCH₂), 2.68-2.58 (m, 1H, CH₃CH₂CH), 2.23 (s, 1H, PhCH₃), 1.98-1.88 (m, 1H, CH₂), 1.86-1.71 (m, 1H, CH₂), 1.61-1.48 (m, 1H, CH₂), 1.00 (t, J = 7.4 Hz, 3H, CH₂CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 142.0, 129.8, 127.4, 126.0, 125.9, 114.5, 38.8, 37.3, 29.5, 26.1, 20.7, 11.8.

HRMS (ESI, m/Z): calcd. for $C_{12}H_{18}N$ [M+H]⁺: 176.1434, found: 176.1435.

HPLC: Chiracel-ADH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 248 nm. Retention time (min): 10.4 (major) and 11.7 (minor).

(S)-4-Ethyl-6-methoxy-1,2,3,4-tetrahydroquinoline (5d)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-L1 (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), EtMgBr (3 M in Et₂O, 0.6 mmol, 3 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following General reducing procedure. Product **5d** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 20:1) [92% conversion, 76% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 6.66 (d, J = 2.9 Hz, 1H, CH_{Ar}), 6.61 (dd, J = 8.6, 2.9 Hz, 1H, CH_{Ar}), 6.46 (d, J = 8.6 Hz, 1H, CH_{Ar}), 3.74 (s, 1H, OCH₃), 3.47 (bs, 1H, NH), 3.34-3.16 (m, 2H, NHCH₂), 2.70-2.60 (m, 1H, CH₃CH₂CH), 1.98-1.88 (m, 2H, CH₂), 1.85-1.72 (m, 2H, CH₂), 1.63-1.49 (m, 1H, CH₂), 1.00 (t, J = 7.4 Hz, 3H, CH₂CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 152.2, 137.8, 127.7, 115.9, 114.9, 112.9, 56.0, 39.2, 37.4, 29.4, 26.0, 11.7.

HRMS (ESI, m/Z): calcd. for C₁₂H₁₈NO [M+H]⁺: 192.1383, found: 192.1383.

HPLC: Chiracel-ADH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 245 nm. Retention time (min): 16.9 (major) and 18.5 (minor).

(S)-Methyl 4-ethyl-1,2,3,4-tetrahydroquinoline-6-carboxylate (5e)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-L1 (0.012 mmol, 6 mol%), BF₃·Et₂O (0.24 mmol, 1.2 equiv.), EtMgBr (3 M in Et₂O, 0.4 mmol, 2 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following General reducing procedure. Product **5e** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 10:1) [98% conversion, 84% yield, 98% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 7.70 (d, J = 2.1 Hz, 1H, CH_{Ar}), 7.64 (dd, J = 8.4, 2.0 Hz, 1H, CH_{Ar}), 6.39 (d, J = 8.4 Hz, 1H, CH_{Ar}), 4.40 (bs, 1H, NH), 3.83 (s, 1H, OCH₃), 3.40-3.27 (m, 2H, NHCH₂), 2.70-2.62 (m, 1H, CH₃CH₂CH), 1.92-1.79 (m, 2H, CH₂), 1.77-1.68 (m, 1H, CH₂), 1.57-1.47 (m, 1H, CH₂), 0.98 (t, J = 7.4 Hz, 3H, CH₂CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 167.8, 148.4, 131.2, 129.2, 124.0, 117.2, 112.8, 51.5, 38.2, 37.2, 28.7, 25.0, 11.6.

HRMS (ESI, m/Z): calcd. for C₁₃H₁₈NO₂ [M+H]⁺: 220.1332, found: 220.1331.

HPLC: Chiracel-OJH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 225 nm. Retention time (min): 24.7 (major) and 26.4 (minor).

(S)-6-Cyano-4-ethyl-1,2,3,4-tetrahydroquinoline (5f)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-L1 (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), EtMgBr (3 M in Et₂O, 0.4 mmol, 2 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following General reducing procedure. Product **5f** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 10:1) [99% conversion, 70% yield, 99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 7.23 (d, J = 1.9 Hz, 1H, CH_{Ar}), 7.19 (dd, J = 8.3, 1.9 Hz, 1H, CH_{Ar}), 6.40 (d, J = 8.3 Hz, 1H, CH_{Ar}), 4.45 (bs, 1H, NH), 3.42-3.28 (m, 2H, NHCH₂), 2.67-2.57 (m, 1H, CH₃CH₂CH), 1.92-1.77 (m, 2H, CH₂), 1.72-1.59 (m, 1H, CH₂), 1.58-1.44 (m, 1H, CH₂), 0.97 (t, J = 7.4 Hz, 3H, CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 147.8, 133.1, 131.2, 124.9, 121.1, 113.5, 97.5, 38.2, 37.1, 28.4, 24.7, 11.4.

HRMS (ESI, m/Z): calcd. for C₁₂H₁₅N₂ [M+H]⁺: 187.1230, found: 187.1228.

HPLC: Chiracel-ADH, *n*-heptane/*i*-PrOH 90:10, 0.5 mL/min., 40 °C, detection at 281 nm. Retention time (min): 18.9 (major) and 19.7 (minor).

(S)-7-Bromo-4-ethyl-1,2,3,4-tetrahydroquinoline (5g)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-L1 (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), EtMgBr (3 M in Et₂O, 0.4 mmol, 2 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following General reducing procedure. Product **5g** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 20:1) [99% conversion, 77% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 6.85 (dd, J = 8.1, 0.8 Hz, 1H, CH_{Ar}), 6.70 (dd, J = 8.1, 2.0 Hz, 1H, CH_{Ar}), 6.60 (d, J = 1.9 Hz, 1H, CH_{Ar}), 3.90 (bs, 1H, NH), 3.35-3.21 (m, 2H, NHCH₂), 2.65-2.55 (m, 1H, CH₃CH₂CH), 1.93-1.76 (m, 2H, CH₂), 1.75-1.63 (m, 1H, CH₂), 1.58-1.45 (m, 1H, CH₂), 0.98 (t, J = 7.4 Hz, 3H, CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 145.6, 130.6, 124.2, 120.1, 119.2, 116.3, 38.3, 37.0, 29.0, 25.4, 11.6.

HRMS (ESI, m/Z): calcd. for C₁₁H₁₅NBr [M+H]⁺: 240.0382, found: 240.0382.

HPLC: Chiracel-OBH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 250 nm. Retention time (min): 12.8 (major) and 15.2 (minor).

(S)-4-Ethyl-7-methyl-1,2,3,4-tetrahydroquinoline (5h)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-L1 (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), EtMgBr (3 M in Et₂O, 0.4 mmol, 2 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following General reducing procedure. Product **5h** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 20:1) [96% conversion, 82% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 6.94 (d, J = 7.7 Hz, 1H, CH_{Ar}), 6.50-6.44 (m, 1H, CH_{Ar}), 6.33 (s, 1H, CH_{Ar}), 3.76 (bs, 1H, NH), 3.37-3.19 (m, 2H, NHCH₂), 2.68-2.58 (m, 1H, CH₃CH₂CH), 2.23 (s, 3H, PhCH₃), 1.98-1.88 (m, 1H, CH₂), 1.86-1.69 (m, 2H, CH₂), 1.60-1.47 (m, 1H, CH₂), 1.00 (t, J = 7.4 Hz, 3H, CH₂CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 144.3, 136.5, 129.2, 122.9, 117.8, 114.8, 38.7, 37.0, 29.4, 26.1, 21.2, 11.8.

HRMS (ESI, m/Z): calcd. for C₁₂H₁₈N [M+H]⁺: 176.1434, found: 176.1433.

HPLC: Chiracel-OBH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 248 nm. Retention time (min): 10.7 (major) and 13.0 (minor).

(S)-4-Ethyl-7-methoxy-1,2,3,4-tetrahydroquinoline (5i)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-L1 (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), EtMgBr (3 M in Et₂O, 0.6 mmol, 3 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following General reducing procedure. Product **5i** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 30:1) [98% conversion, 81% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 6.94 (d, J = 8.3 Hz, 1H, CH_{Ar}), 6.23 (dd, J = 8.4, 2.5 Hz, 1H, CH_{Ar}), 6.05 (d, J = 2.5 Hz, 1H, CH_{Ar}), 3.85 (bs, 1H, NH), 3.74 (s, 3H, OCH₃), 3.36-3.20 (m, 2H, NHCH₂), 2.66-2.57 (m, 1H, CH₃CH₂CH), 1.98-1.86 (m, 1H, CH₂), 1.85-1.66 (m, 2H, CH₂), 1.59-1.45 (m, 1H, CH₂), 0.99 (t, J = 7.4 Hz, 3H, CH₂CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 158.8, 145.2, 129.9, 118.4, 102.7, 99.3, 55.2, 38.6, 36.6, 29.4, 26.1, 11.7.

HRMS (ESI, m/Z): calcd. for C₁₂H₁₈NO [M+H]⁺: 192.1383, found: 192.1381.

HPLC: Chiracel-ADH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 245 nm. Retention time (min): 20.1 (minor) and 24.2 (major).

(S)-4-Ethyl-7-vinyl-1,2,3,4-tetrahydroquinoline (5j)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-L1 (0.012 mmol, 6 mol%), BF₃·Et₂O (0.24 mmol, 1.2 equiv.), EtMgBr (3 M in Et₂O, 0.4 mmol, 2 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following **General reducing procedure**. Product **5j** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 50:1) [98% conversion, 60% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 6.99 (d, J = 7.8 Hz, 1H, CH_{Ar}), 6.71 (dd, J = 7.1, 1.6 Hz, 1H, CH_{Ar}), 6.60 (dd, J = 17.6, 10.9 Hz, 1H, CH₂=CH), 6.53 (d, J = 1.6 Hz, 1H, CH_{Ar}), 5.64 (dd, J = 17.6, 1.1 Hz, 1H, CHH=CH), 5.14 (dd, J = 10.9, 1.1 Hz, 1H, CHH=CH), 3.85 (bs, 1H, NH), 3.40-3.19 (m, 2H, NHCH₂), 2.71-2.61 (m, 1H, CH₃CH₂CH), 1.99-1.87 (m, 1H, CH₂), 1.87-1.68 (m, 2H, CH₂), 1.61-1.47 (m, 1H, CH₂), 0.99 (t, J = 7.4 Hz, 3H, CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 144.4, 137.2, 136.4, 129.4, 125.6, 115.0, 112.7, 111.8, 38.7, 37.2, 29.3, 25.9, 11.7.

HRMS (ESI, m/Z): calcd. for C₁₃H₁₈N [M+H]⁺: 188.1434, found: 188.1432.

HPLC: Chiracel-OBH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 239 nm. Retention time (min): 15.9 (major) and 23.1 (minor).

(S)-5-Chloro-4-ethyl-1,2,3,4-tetrahydroquinoline (5k)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-**L1** (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), EtMgBr (3 M in Et₂O, 0.4 mmol, 2 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following **General reducing procedure**.

Product **5k** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 20:1) [99% conversion, 77% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 6.92-6.84 (m, 1H, CH_{Ar}), 6.65 (dd, J = 7.9, 1.1 Hz, 1H, CH_{Ar}), 6.35 (dd, J = 8.1, 1.1 Hz, 1H, CH_{Ar}), 4.00 (bs, 1H, NH), 3.38-3.21 (m, 2H, NHCH₂), 3.02-2.93 (m, 1H, CH₃CH₂CH), 2.03-1.94 (m, 1H, CH₂), 1.85-1.68 (m, 2H, CH₂), 1.45-1.30 (m, 1H, CH₂), 1.02 (t, J = 7.4 Hz, 3H, CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 145.2, 134.7, 127.3, 123.6, 117.4, 112.3, 36.8, 34.8, 27.0, 23.0, 11.8.

HRMS (ESI, m/Z): calcd. for C₁₁H₁₅NCl [M+H]⁺: 196.0888, found: 196.0887.

HPLC: Chiracel-OBH, *n*-heptane/*i*-PrOH 99:1, 0.5 mL/min., 40 °C, detection at 250 nm. Retention time (min): 11.9 (major) and 12.5 (minor).

(S)-4-Ethyl-5-methyl-1,2,3,4-tetrahydroquinoline (5l)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-**L1** (0.012 mmol, 6 mol%), BF₃·Et₂O (0.6 mmol, 3 equiv.), EtMgBr (3 M in Et₂O, 0.6 mmol, 3 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 24 h, and following **General reducing procedure**. Product **51** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 40:1) [95% conversion, 80% yield, 98% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 6.92-6.86 (m, 1H, CH_{Ar}), 6.49 (d, J = 7.4, Hz, 1H, CH_{Ar}), 6.34 (d, J = 8.0 Hz, 1H, CH_{Ar}), 3.81 (bs, 1H, NH), 3.41-3.20 (m, 2H, NHCH₂), 2.79-2.70 (m, 1H, CH₃CH₂CH), 2.26 (s, 1H, PhCH₃), 2.01-1.94 (m, 1H, CH₂), 1.83-1.71 (m, 1H, CH₂), 1.68-1.55 (m, 1H, CH₂), 1.52-1.38 (m, 1H, CH₂), 1.02 (t, J = 7.4 Hz, 3H, CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 143.7, 136.6, 126.5, 124.4, 118.9, 112.1, 36.8, 34.3, 27.2, 23.5, 18.9, 12.0.

HRMS (ESI, m/Z): calcd. for C₁₂H₁₈N [M+H]⁺: 176.1434, found: 176.1434.

HPLC: Chiracel-OJH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 248 nm. Retention time (min): 11.1 (major) and 12.8 (minor).

(S)-4-Ethyl-8-fluoro-1,2,3,4-tetrahydroquinoline (5m)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-**L1** (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), EtMgBr (3 M in Et₂O, 0.4 mmol, 2 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following **General reducing procedure**. Product **5m** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 20:1) [92% conversion, 76% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 6.85-6.77 (m, 2H, CH_{Ar}), 6.57-6.49 (m, 1H, CH_{Ar}), 6.34 (d, J = 8.0 Hz, 1H, CH_{Ar}), 4.05 (bs, 1H, NH), 3.40-3.29 (m, 2H, NHCH₂), 2.75-2.65 (m, 1H, CH₃CH₂CH), 2.00-1.81 (m, 2H, CH₂), 1.81-1.68 (m, 1H, CH₂), 1.63-1.48 (m, 1H, CH₂), 1.00 (t, J = 7.4 Hz, 3H, CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 151.0 (d, J = 237.2 Hz), 132.8 (d, J = 11.9 Hz), 127.8 (d, J = 3.3 Hz), 124.2 (d, J = 2.8 Hz), 115.2 (d, J = 7.4 Hz), 112.1 (d, J = 18.2 Hz), 37.9, 37.0 (d, J = 2.7 Hz), 29.2, 25.5, 11.7.

¹⁹F NMR (CDCl₃, 400 MHz): δ –138.9.

HRMS (ESI, m/Z): calcd. for $C_{11}H_{15}NF [M+H]^+$: 180.1183, found: 180.1182.

HPLC: Chiracel-ODH, *n*-heptane/*i*-PrOH 98:2, 0.5 mL/min., 40 °C, detection at 243 nm. Retention time (min): 10.9 (minor) and 11.7 (major).

(S)-4-Ethyl-8-methyl-1,2,3,4-tetrahydroquinoline (5n)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-L1 (0.012 mmol, 6 mol%), BF₃·Et₂O (0.6 mmol, 3 equiv.), EtMgBr (3 M in Et₂O, 0.6 mmol, 3 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following General reducing procedure. Product **5n** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 80:1) [99% conversion, C-2-addition:C-4-addition = 1:5, 73% yield, >99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 6.94 (d, *J* = 7.6 Hz, 1H, CH_{Ar}), 6.90 (d, *J* = 7.3 Hz, 1H, CH_{Ar}), 6.90 (d, *J* = 7.3 Hz, 1H, CH_{Ar}), 3.67 (bs, 1H, NH), 3.44-3.31 (m, 2H, NHCH₂), 2.74-2.64 (m, 1H, CH₃CH₂CH), 2.09 (s, 3H, PhCH₃), 1.99-1.70 (m, 3H, CH₂), 1.62-1.49 (m, 1H, CH₂), 1.00 (t, *J* = 7.4 Hz, 3H, CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 142.3, 128.0, 127.3, 125.1, 121.1, 116.1, 38.8, 37.5, 29.6, 25.6, 17.5, 11.9.

HRMS (ESI, m/Z): calcd. for C₁₂H₁₈N [M+H]⁺: 176.1434, found: 176.1433.

HPLC: Chiracel-ODH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 244 nm. Retention time (min): 9.6 (major) and 11.1 (minor).

(S)-4-Ethyl-1,2,3,4-tetrahydrobenzo[g]quinoline (50)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-L1 (0.012 mmol, 6 mol%), BF₃·Et₂O (0.6 mmol, 3 equiv.), EtMgBr (3 M in Et₂O, 0.6 mmol, 3 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following General reducing procedure. Product **50** was obtained as a white solid after column chromatography (SiO₂, pentane:Et₂O = 40:1) [96% conversion, 66% yield, 97% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 7.62 (d, *J* = 8.1 Hz, 1H, CH_{Ar}), 7.54-7.46 (m, 2H, CH_{Ar}), 7.32-7.24 (m, 1H, CH_{Ar}), 7.17-7.10 (m, 1H, CH_{Ar}), 6.75 (s, 1H, CH_{Ar}), 4.17 (bs, 1H, NH), 3.47 – 3.32 (m, 2H, NHCH₂), 2.91-2.82 (m, 1H, CH₃CH₂CH), 2.08-1.96 (m, 1H, CH₂), 1.95-1.78 (m, 2H, CH₂), 1.74-1.60 (m, 1H, CH₂), 1.04 (t, *J* = 7.4 Hz, 3H, CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 143.3, 134.0, 128.9, 127.4, 127.4, 127.2, 125.5, 124.9, 121.4, 38.8, 38.0, 28.6, 26.0, 11.7.

HRMS (ESI, m/Z): calcd. for C₁₅H₁₈N [M+H]⁺: 212.1434, found: 212.1433.

HPLC: Chiracel-ADH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 247 nm. Retention time (min): 17.4 (major) and 20.9 (minor).

(2R,4S)-4-Ethyl-2-methyl-1,2,3,4-tetrahydroquinoline (5p)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-L1 (0.012 mmol, 6 mol%), BF₃·Et₂O (0.4 mmol, 2 equiv.), EtMgBr (3 M in Et₂O, 0.4 mmol, 2 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following **General reducing procedure**. Product **5p** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 30:1) [98% conversion, 79% yield, 6:1 *dr*, 99% *ee*].

¹H NMR (CDCl₃, 400 MHz): δ 7.16 (dd, J = 7.7, 1.4 Hz, 1H, CH_{Ar}), 7.04-6.95 (m, 1H, CH_{Ar}), 6.72-6.60 (m, 1H, CH_{Ar}), 6.50 (d, J = 8.1 Hz, 1H, CH_{Ar}), 3.67 (bs, 1H, NH), 3.52-3.38 (m, 1H, NHCH), 2.93-2.82 (major) and 2.66-2.57 (minor) (m, 1H, CH₂), 2.11-1.97 (major) and 1.89-1.82 (minor) and 1.76-1.65 (minor) (m, 2H, CH₂), 1.65-1.50 (m, 1H, CH₂), 1.41-1.27 (m, 1H, CH₂), 1.23 (d, J = 6.2 Hz, 3H, CHCH₃), 1.05-0.94 (m, 3H, CH₂CH₃).

¹³C NMR (CDCl₃, 100 MHz): Major: δ 145.6, 127.1, 126.7, 125.2, 117.5, 114.4, 47.5, 36.8, 36.7, 27.1, 23.0, 10.7. Minor: δ 144.2, 129.7, 126.9, 125.4, 116.6, 113.9, 42.6, 37.9, 33.7, 31.0, 23.0, 12.1.

HRMS (ESI, m/Z): calcd. for C₁₂H₁₈N [M+H]⁺: 176.1434, found: 176.1433.

HPLC: Chiracel-ODH, *n*-heptane/*i*-PrOH 98:2, 0.5 mL/min., 40 °C, detection at 254 nm. Retention time of major disastereomer (min): 12.1 (major) and 12.7 (minor), retention time of minor disastereomer (min): 10.3 (minor) and 11.6 (major).

(2S,4S)-4-ethyl-2-phenyl-1,2,3,4-tetrahydroquinoline (5q)



The reaction was performed with the substrate (0.2 mmol, 1 equiv.), CuTc (0.01 mmol, 5 mol%), ligand (*R*,*R*)-L1 (0.012 mmol, 6 mol%), BF₃·Et₂O (0.6 mmol, 3 equiv.), EtMgBr (3 M in Et₂O, 0.6 mmol, 3 equiv.), CH₂Cl₂ (2.0 mL) at -78 °C for 2 h, and following General reducing procedure. Product **5q** was obtained as a colorless oil after column chromatography (SiO₂, pentane:Et₂O = 70:1) [95% conversion, 84% yield, 6:1 *dr*, 86% *ee*]. The relative configuration of **5q** was determined by NOE experiments after transforming to **5t** (see Figure S5).

¹H NMR (CDCl₃, 400 MHz): δ 7.50-7.30 (m, 5H, CH_{Ar}), 7.28-7.20 (m, 1H, CH_{Ar}), 7.11-7.00 (m, 1H, CH_{Ar}), 6.78-6.67 (m, 1H, CH_{Ar}), 6.56 (dd, *J* = 7.9, 1.2 Hz, 1H, CH_{Ar}), 4.50-4.42 (m, 1H, NHCH), 4.07 (minor) and 3.97 (major) (bs, 1H, NH), 3.10-2.99 (major) and 2.73-2.65 (minor) (m, 1H, CH₂), 2.26-1.92 (m, 2H, CH₂), 1.88-1.69 (m, 1H, CH₂), 1.69-1.53 (m, 1H, CH₂), 1.06 and 0.98 (t, *J* = 7.4 Hz, 3H, CH₃).

¹³C NMR (CDCl₃, 100 MHz): Major: δ 145.6, 144.8, 128.8, 127.7, 127.2, 126.9, 126.8, 125.1, 117.8, 114.6, 57.2, 38.0, 37.4, 27.0, 10.8. Minor: δ 145.0, 144.1, 129.6, 128.7, 127.6, 127.1, 126.8, 125.2, 52.4, 37.9, 34.9, 30.7, 12.0.

HRMS (ESI, m/Z): calcd. for C₁₇H₂₀N [M+H]⁺: 238.1590, found: 238.1590.

HPLC: Chiracel-ADH, *n*-heptane/*i*-PrOH 95:5, 0.5 mL/min., 40 °C, detection at 207 nm. Retention time of major disastereomer (min): 10.5 (minor) and 11.1 (major), retention time of minor disastereomer (min): 8.9 (major) and 12.9 (minor).

(2S,4S)-1-(3-Bromophenyl)carbonyl-4-ethyl-2-phenyl-1,2,3,4-tetrahydroquinoline (5t)



The reactions were performed according to the literature procedure.¹⁴ To a solution of **5q** (75 mg, 0.316 mmol, 1 equiv.) in dry CH₂Cl₂ (2 ml) was added benzoyl chloride (54 μ L, 0.411 mmol, 1.3 equiv.) dropwise, followed by DMAP (1.9 mg, 0.016 mmol, 5 mol%) and triethylamine (57 μ L, 0.411 mmol, 1.3 equiv.). The reaction mixture was stirred at RT overnight. Then it is quenched with water and extracted with CH₂Cl₂ for three times. The combined organic phases were washed with brine, dried over MgSO₄, and concentrated in vacuo to give the crude product. The major diastereomer of **5t** was obtained as a white solid after column chromatography (SiO₂, pentane:Et₂O = 20:1) followed by recrystallization in the mixed solvent of CH₂Cl₂ and pentane. Relative configuration was determined by NOE experiments (see Figure S5).

¹H NMR (CDCl₃, 400 MHz): δ 7.35-7.31 (m, 2H, CH_{Ar}), 7.26-7.22 (m, 1H, CH_{Ar}), 7.22-7.17 (m, 2H, CH_{Ar}), 7.15-7.08 (m, 4H, CH_{Ar}), 6.95-6.87 (m, 3H, CH_{Ar}), 6.55 (d, J = 7.1 Hz, 1H, CH_{Ar}), 5.56 (t, J = 9.5 Hz, 1H, NHCH), 2.27 (ddd, J = 12.6, 8.8, 3.3 Hz, 1H, NHCHCHH), 2.65-1.58 (m, 1H, CH₃CH₂CH), 2.19-2.10 (m, 1H, NHCHCHH), 1.59-1.48 (m, 1H, CH₃CHH), 1.38-1.29 (m, 1H, CH₃CHH), 1.10 (t, J = 7.4 Hz, 3H, CH₃).

¹³C NMR (CDCl₃, 100 MHz): δ 168.1, 144.1, 139.2, 139.0, 138.1, 133.0, 132.1, 129.3, 128.7, 127.2, 127.1, 127.1, 126.5, 126.4, 125.9, 123.7, 122.1, 58.6, 40.9, 37.3, 23.3, 11.8.

HRMS (ESI, m/Z): calcd. for C₂₄H₂₂NOBr [M+H]⁺: 420.0958, found: 420.0962.



Figure S5. ¹H NMR and 1D NOE experiment of **5t**. Selective irradiation on H-1 showed NOE with H-3 (highlighted) which are positioned on the same side of the ring.

7. Determination of absolute configuration

The absolute configuration of **5b** was determined by X-ray single crystallography (see Figure S6), and the relative configuration of **5q** and **5t** was determined by NOE experiment (see Figure S5). The absolute configurations of other compounds were assigned by analogy.

8. X-ray crystallographic analysis

A single crystal of compound **5b** was mounted on top of a cryoloop and transferred into the cold nitrogen stream (100 K) of a Bruker-AXS D8 Venture diffractometer. Data collection and reduction was done using the Bruker software suite APEX3.¹⁵ The final unit cell was obtained from the xyz centroids of 7072 reflections after integration. A multiscan absorption correction was applied, based on the intensities of symmetry-related reflections measured at different angular settings (SADABS). The structures were solved by direct methods using SHELXT¹⁶ and refinement of the structure was performed using SHLELXL.¹⁷ The hydrogen atoms were generated by geometrical considerations, constrained to idealized geometries and allowed to ride on their carrier atoms with an isotropic displacement parameter related to the equivalent displacement parameter of their carrier atoms. The

absolute configuration of the model was chosen based on anomalous dispersion. Refinement of the Flack x parameter converged 0.040(12).



Figure S6. Molecular structure of compound **5b**, showing 50% probability ellipsoids. Hydrogen atoms are omitted for clarity.

Table S6.	Crystallographic	data for	compound	5b
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chem. formula	$C_{11}H_{14}BrN$
M_{r}	240.14
cryst syst.	Monoclinic
color, habit	Orange, platelet
size (mm)	$0.42 \times 0.10 \ 0.03$
space group	P 21
a (Å)	8.8938(3)
b (Å)	5.9958(2)
c (Å)	9.5190(3)
a, deg	90
β, deg	94.8830(10)
γ, deg	90
V (Å ³)	505.76(3)
Ζ	2
$\rho_{calc}, g \cdot cm^{-3}$	1.577
μ(Mo Kα), cm ⁻¹	5.129
F(000)	244
temp (K)	100(2)
θ range (deg)	4.662 - 65.118
data collected (h,k,l)	-10:10, -6:7, -11:10
no. of rflns collected	6932
no. of indpndt reflns	1702
observed reflns	1689 ($F_o \ge 2(F_o)$)

R(F) (%)	1.74
$wR(F^2)$ (%)	4.21
GooF	1.076
Weighting a,b	0.0238, 0.0615
params refined	123
restraints	1
min, max resid dens	-0.552, 0.193
Flack x	0.040(12)

9. Mechanistic studies

DFT study of the copper catalyzed addition of EtMgBr·2Et₂O to 1a and 1a·BF₃

To unravel the mechanism behind the presented copper catalyzed nucleophilic addition of Grignard reagents to quinolines, we first explored the blank reaction, *i.e.* the direct addition of the Grignard reagent to the quinoline. We have found that – in line with the experimental evidence- this direct addition reaction (independently of the considered regioselectivity) is not accessible at the working temperature ($\Delta G^{\neq}_{C-2-addition} = 23.00 \text{ kcal/mol}, \Delta G^{\neq}_{C-4-addition} = 41.44 \text{ kcal/mol})$, see Figure S7. If the LA is added to the reaction media , it will bind to the nitrogen atom decreasing the electron density of the quinoline (the charge at C-2 changes from 0.33 a.u. to 0.24 a.u. and the charge at C-4 changes from 0.44 a.u. to 0.43 a.u.) and facilitate a subsequent addition of the Grignard reagent, both at C-2 and C-4; the latter is energetically less penalized, in agreement with the experimental data presented in the manuscript.



Figure S7. Computed mechanism for the (a) addition of $EtMgBr \cdot 2Et_2O$ to 1a. (b) addition of $EtMgBr \cdot 2Et_2O$ to the LA activated quinoline ($1a \cdot BF_3$). Calculations were performed at the PCM (CH_2Cl_2)/M06/def2svpp computational level using the Gaussian 09 program. The thermochemistry was obtained at 1 atm and 195 K. Short names have been used, but all the presented systems contain two Et_2O molecules except for BF_3 and quinoline, this information is omitted for clarity.

ID	ImFreqs	Stable	SCF^{a}	SCF+ZPVE	H^{b}	G ^c
1 a	_	Yes	-401.3167684	-401.181547	-401.17791	-401.200117
\mathbf{V}	-	Yes	-3720.911052	-3720.435202	-3720.419991	-3720.468347
TS-(V-VI)	-404.649	Yes	-3720.876478	-3720.400663	-3720.386453	-3720.431694
VI	-	Yes	-3487.268379	-3487.258404	-3487.257786	-3487.296207
TS-(V-VII)	-412.972	Yes	-3720.847843	-3720.371393	-3720.357214	-3720.402308
VII	-	Yes	-3720.94771	-3720.469152	-3720.454362	-3720.502078
BF ₃	-	Yes	-324.2164826	-324.203888	-324.201265	-324.219811
1a·BF ₃	-	Yes	-725.5809442	-725.430379	-725.424745	-725.452271
TS-(1a·BF ₃ -VIII)	-381.532	Yes	-4045.14233	-4044.650201	-4044.634257	-4044.682723
VIII	-	Yes	-4045.22728	-4044.733319	-4044.71804	-4044.765589
TS-(1a·BF ₃ -IX)	-510.418	Yes	-4045.135857	-4044.643974	-4044.62797	-4044.676943
IX	-	Yes	-3811.900283	-3811.542546	-3811.530313	-3811.57177

Table S7. Summary of the energies for the study of the blank reaction and the blank reaction with BF_3 in the addition of the Grignard reagent to quinoline

^aEnergies are expressed in a.u. and the imaginary frequencies in cm⁻¹. ^bH represents enthalpies. ^cG Represents Gibbs free Energies.

At this point, we wondered whether copper could also behave as a LA and compete with the BF_3 for binding tothe nitrogen atom of **1a**, or if the BF_3 could be engaged in the coordination of any other species present in the reaction. Thus, we performed a systematic exploration of all the potential coordination of BF_3 and **L1-CuBr** with **1a** (Figure S8). This study reveals that even **L1-CuBr** can behave as a LA and coordinate the quinoline, as expected BF_3 leads to a more thermodynamically stable speciesthus displacing the copper in case complex **X** or **XI** were formed in solution. We also explored potential coordination of BF_3 with the **L1** We have found that even BF_3 can coordinate **L1**, this coordination leads to a complex that is thermodynamically less stable than that resulting from its coordination to the **1a** (**1a**•**BF**₃) or that resulting from the coordination of **L1** with the copper source, as we detailed below (Figure S8). This is also in agreement with the NMR studies (Figures S15-16).


Figure S8. Computed relative stability of the coordination of BF_3 to the different species participating in the reaction. Calculations were performed at the PCM (CH₂Cl₂)/M06/def2svpp computational level using the Gaussian 09 program. The thermochemistry was obtained at 1 atm and 195 K.

Table S8. Summary of the energies for the study of the interaction of BF_3 with the different components of the catalytic system presented. All calculations were performed with the (*R*,*R*)-L1 ligand.

ID	ImFreqs	Stable	SCF ^a	SCF+ZPVE	H^{b}	G ^c
1 a	_	Yes	-401.3167684	-401.181547	-401.17791	-401.200117
1a·BF ₃	-	Yes	-725.5809442	-725.430379	-725.424745	-725.452271
IIa-proR ₃₋₄	-	Yes	-6937.441863	-6936.68338	-6936.660362	-6936.72743
IIa-proS ₃₋₄	-	Yes	-6937.438924	-6936.680365	-6936.657475	-6936.724125
II-proR ₃₋₄	-	Yes	-4442.685582	-4441.864414	-4441.841144	-4441.907423
II-proS ₃₋₄	_	Yes	-4442.684883	-4441.863422	-4441.840268	-4441.906948
II-proR ₁₋₂	-	Yes	-4442.682289	-4441.86133	-4441.810951	-4441.945899
II-pro S_{1-2}	-	Yes	-4442.683702	-4441.863418	-4441.812692	-4441.948965
X	-	Yes	-6613.203409	-6612.459973	-6612.439187	-6612.500643
XI	_	Yes	-6937.445992	-6936.688835	-6936.666203	-6936.731244
L1	_	Yes	-1997.721791	-1997.11694	-1997.101947	-1997.151917
XII	_	Yes	-2321.967603	-2321.348094	-2321.330934	-2321.385628
XIII	_	Yes	-2646.21189	-2645.57695	-2645.558094	-2645.61468
L1-CuBr	_	Yes	-6211.843657	-6211.236732	-6211.219631	-6211.274031
L1-CuBr·BF ₃	_	Yes	-6536.091876	-6535.470998	-6535.451685	-6535.50974

^aEnergies are expressed in a.u. and the imaginary frequencies in cm⁻¹. ^bH represents enthalpies. ^cG Represents Gibbs free Energies.

Once we have identified the role of the LA as well as its preferred disposition in the system, we moved to the study of the speciation of the copper source in the reaction. With this goal and taking into consideration that copper(I) is known for forming dimeric species and other aggregates in solution, we performed a systematic analysis on the relative stability of some of the copper species that might be present in the reaction media (Figure S9).

We have found that once the copper source interacts with **L1** it evolves towards the formation of the complex **L1-CuBr** (ΔG =-20.91 kcal/mol). This complex can further progress towards the formation of dimeric species and even transmetalate in the presence of the Grignard reagent (Figure S9). The formation of dimeric species is energetically favourable, nonetheless we have disregarded them as potential catalyst for two reasons: a) the dilution degree in which the copper species is present in the reaction media and b) when the dimeric species are allowed to interact with the organic substrate they break towards their monomeric constituents.

The L1-CuBr complex can alternatively evolve through a direct transmetalation leading to L1-CuEt, or I, this species is not only thermodynamically preferred (-32.70 kcal/mol) but also key for the evolution of the catalysed reaction, as we present in the main text.

Next we explored the possibility of species **I** further interacting with the Grignard reagent via a second transmetallation resulting in the formation of a cuprate species. We have found that the formation of cuprates is highly endergonic and consequently unfeasible in the working conditions used in this protocol.



Figure S9. Computed relative energies for the potential copper species present in the reaction media. The calculations were performed at the $PCM(CH_2Cl_2)/M06/def2svpp$ computational level using the Gaussian 09 program. The thermochemistry was obtained at 1 atm and 195 K. L1 has been represented as a curved line to clarify that it behaves as a chelating agent.

ID	Im. Freqs.	Stable	SCF ^a	SCF+ZPVE	H^{b}	G^{c}
$Cu_2Br_2 \cdot (SMe_2)_2$	_	Yes	-9383.671745	-9383.517717	-9383.506351	-9383.548312
L1-CuBr	_	Yes	-6211.843657	-6211.236732	-6211.219631	-6211.274031
$L1_2Cu_2Br_2$	_	Yes	-12423.732710	-12422.517483	-12422.483079	-12422.574618
L1 ₂ Cu ₂ BrEt	_	Yes	-9928.977519	-9927.699335	-9927.664871	-9927.754807
$L1_2Cu_2Et_2$	_	Yes	-7434.206748	-7432.867065	-7432.831901	-7432.923431
L1-CuEt	_	Yes	-3717.080114	-3716.411863	-3716.393885	-3716.450532
L1-CuEt ₂	_	Yes	-3796.216569	-3795.485075	-3795.465794	-3795.523754
CuEt ₂	_	Yes	-1798.483889	-1798.359603	-1798.354567	-1798.379112

Table S9. Summary of the energies for the study of the copper speciation in solution.

^aEnergies are expressed in a.u. and the imaginary frequencies in cm⁻¹. ^bH represents enthalpies. ^cG Represents Gibbs free Energies.

We then proceed with the exploration of the copper(I) catalysed reaction of the quinoline with the Grignard in the presence of the BF_3 as a LA, as described in the main text.

During the mechanistic exploration we also explored the potential evolution of $II_{3.4}$ through a C-2addition. Unfortunately, all our attempts to locate the transition state connecting $II_{3.4}$ and $III_{1.2}$ were unfruitful. In order to provide an estimation for the energy of TS-($II_{3.4}$ -III)- $R_{1.2}$ and TS-($II_{3.4}$ -III)- $S_{1.2}$, we explored the potential energy surface via a scan of the electronic energy variation with the elongation of the bond C-2-Et at III_{1-2} (see Figure S10).



Figure S10. Representation of the energy scan (top) and reaction profile (bottom) obtained through the evaluation of the energy penalty associated with the addition of the Grignard reagent to II_{3-4} at C-2. The energies reported correspond to electronic energies. The scan point corresponds to each of the steps of the scan, the step size selected was 0.1 Å, the initial C-C bond distance is 1.54 Å and the ending one 2.34 Å.

Moreover, we explored the possibility of a direct decoordination of the copper complex in III_{3-4} yielding directly the product while creating a highly electrophilic copper species (IIIa), naturally we found that the formation of such an ionic pair is energetically unfeasible (see Figure S11). Alternatively, we also explored the possibility of III_{3-4} to evolve via a structural reorganization that would involve the coordination of the copper at the nitrogen with the concominant release of BF₃. We

have found that the formation of III_b is too endothermic and thus it is unlikely that this species will be present in solution.



Figure S11. Computed potential reaction paths for the copper catalyzed addition of $EtMgBr \cdot 2Et_2O$ to quinoline in the presence of a LA. Calculations were performed at the PCM (CH₂Cl₂)/M06/def2svpp computational level using the Gaussian 09 program. The thermochemistry was obtained at 1 atm and 195 K.

ID	ImFreqs	Stable	SCF	SCF+ZPVE	Н	G
L1-CuBr	_	Yes	-6211.843657	-6211.236732	-6211.219631	-6211.274031
EtMgBr·2Et ₂ O	-	Yes	-3320.18153	-3319.840154	-3319.81692	-3319.893004
L1-CuEt	-	Yes	-3717.080114	-3716.411863	-3716.393885	-3716.450532
1a·BF ₃	_	Yes	-725.5809442	-725.430379	-725.424745	-725.452271
IIa-proR ₃₋₄	-	Yes	-6937.441863	-6936.68338	-6936.660362	-6936.72743
IIa-proS ₃₋₄	_	Yes	-6937.438924	-6936.680365	-6936.657475	-6936.724125
II-proR ₁₋₂	_	Yes	-4442.683702	-4441.863418	-4441.812692	-4441.948965
II -pro S_{1-2}	_	Yes	-4442.682289	-4441.86133	-4441.810951	-4441.945899
II-proR ₃₋₄	_	Yes	-4442.685582	-4441.864414	-4441.841144	-4441.907423
II-proS ₃₋₄	_	Yes	-4442.684883	-4441.863422	-4441.840268	-4441.906948
TS-(II-III)- R_{3-4}	-300.38	Yes	-4442.673217	-4441.851763	-4441.802325	-4441.933955
TS-(II-III)-S ₃₋₄	-318.93	Yes	-4442.683985	-4441.863683	-4441.813757	-4441.947534
III- <i>R</i> ₃₋₄	_	Yes	-4442.744373	-4441.920167	-4441.897521	-4441.9618
III-S ₃₋₄	-	Yes	-4442.755128	-4441.930912	-4441.908182	-4441.972896
III- <i>R</i> ₁₋₂	_	Yes	-4442.71626	-4441.891792	-4441.842248	-4441.974708
III-S ₁₋₂	_	Yes	-4442.736311	-4441.911465	-4441.88894	-4441.952942
III _a	_	Yes	-804.7923851	-804.577264	-804.57015	-804.600866
III _b	_	Yes	-4118.464794	-4117.656681	-4117.611087	-4117.735868
IV-S ₃₋₄	-	Yes	-804.7923853	-804.577273	-804.570153	-804.600902

Table S10. Summary of the energies for the study of the Computed mechanism for the copper catalyzed addition of $EtMgBr \cdot 2Et_2O$ to quinoline in the presence of a LA.^a

^aEnergies are expressed in a.u. and the imaginary frequencies in cm⁻¹. ^bH represents enthalpies. ^cG Represents Gibbs free Energies.

Scans of the variation of the energy with the reaction coordinate from II-*proR*₃₋₄ to TS-(II-III)- R_{3-4}

Due to the difficulties found when trying to locate **TS**-(**II-III**)- R_{1-2} and **TS**-(**II-III**)- S_{1-2} , we decided to explore the potential energy surface in the region of the reaction coordinate. To do so, we first confirmed in **TS**-(**II-III**)- R_{3-4} that the bending of C-4-Cu-Et is the main contributor to the energy penalization of the transition state (barrier from scan = 13.23, computed barrier = 9 kcal/mol), we chose this diastereomer and not the **II**-*proS*₃₋₄ one because it shows a higher barrier and thus it provides a clearer scan (see Figure S12).



Figure S12. Computed energy profile for the formation of **III**- R_{3-4} from **II**-*proR*₃₋₄. The represented energies correspond to pure electronic energies. Calculations were performed at the PCM (CH₂Cl₂)/M06/def2svpp computational level using the Gaussian 09 program. The scan step chosen is 10°.

Encouraged with these results we performed an analogous study for localizing $TS-(II-III)-R_{1-2}$ obtaining, as it was described in the main text, that the variation of this internal coordinate leads to the disaggregation of the complex and the formation of an alternative one in which the L1-CuEt complex is establishing a weak interaction with C-3-C-4. This reveals that $II-proR_{1-2}$ will not evolve via a C-2-addition but through a disaggregation of the complex and the formation of a new species $IIb-proR_{1-2}$ in which the copper is already sitting at C-3-C-4 thus predisposing the system to further progress via the formation of $II-_{3-4}$.



Figure S13. Scan of the internal coordinate C-2-Cu-Et at II-₁₋₂. The represented energies correspond to pure electronic energies. Calculations were performed at the PCM (CH₂Cl₂)/M06/def2svpp computational level using the Gaussian 09 program The scan step chosen is 10° . We have used black lines to represent the energy evolution of II-*proR*₁₋₂ and grey lines to represent the energy evolution of II-*proR*₁₋₂ for visualization purposes.

IRC obtained for TS-(II-III)-R₃₋₄

To rationalize the sensitivity of the reaction to the substituents at position C-3 but not to substituents at C-4 we have followed the intrinsic reaction coordinate (IRC) from **TS-(II-III)-R_{3-4}**. The IRC shows that once the transition state structure is met and the delivery of the *Et* group initiated, the **L1**/Cu(I) moiety starts relocating towards C-3 and it is not until the very end of the IRC that the relocation of the copper center towards C-4 is happening. As a result, the latter chemical event barely contributes to

the energy of the reaction whereas the relocation to C-3 has a sound effect on the energy of this TS, as a result if the access to that position is impeded the energy penalization for this reaction is going to be increased (Figure S14).



Figure S14. IRC obtained for **TS-(II-III)**- R_{3-4} . The represented energies correspond to pure electronic energies. Calculations were performed at the PCM (CH₂Cl₂)/M06/def2svpp computational level using the Gaussian 09 program. Notice that for visualization purposes we have represented the mirror images of the compounds.

NMR experiments

NMR experiments were carried out in order to detect possible species formed in the reaction. NMR studies were performed with 2-methyl-quinoline in combination with $BF_3 \cdot Et_2O$, with CuBr \cdot SMe₂ and with $BF_3 \cdot Et_2O$ and L4-CuBr. Accordingly 3 different species were detected: 1) 2-methyl-quinoline/BF₃; 2) 2-methyl-quinoline/L4-CuBr; 3) 2-methyl-quinoline/BF₃/L4-CuBr. For the latter species we assume that Cu(I) complex binds to C-3-C-4 position of 2-methyl-quinoline, however we don't have a strong NMR evidence for this. All the attempts to detect species with 2-methyl-

quinoline/ $BF_3/L4$ -CuBr/RMgBr were not successful due the immediate addition product formation upon mixing all the reagents together.

Below one can find the overlapped spectra, separate spectra with assignments and procedures for the preparation of the species.

Overlapped spectra



Figure S15. NMR studies performed to evaluate the interaction of 2-Me-quinoline with $BF_3 \cdot Et_2O$, L4-CuBr and both of them. (a) ¹H-NMR spectra of 2-methyl-quinoline. (b) ¹H-NMR spectra of the 2-methyl-quinoline with 2 equiv. of $BF_3 \cdot Et_2O$. (c) ¹H-NMR spectra of the 2-methyl-quinoline with 1 equiv. of L4-CuBr. (d) ¹H-NMR spectra of the 2-Me-quinoline with 1 equiv of L4-CuBr and 2 equiv. of $BF_3 \cdot Et_2O$. (e) ¹H-NMR spectra of L4-CuBr.



Figure S16. Amplified aromatic part of ¹³C NMR spectra in the range of 118-162 ppm to evaluate the interaction of 2-Me-quinoline with BF₃·Et₂O, **L4**-CuBr and both of them (for full spectra, see below). (a) ¹³C-NMR spectra of 2-methyl-quinoline. (b) ¹³C-NMR spectra of the 2-Me-quinoline with 2 equiv of BF₃·Et₂O. (c) ¹³C -NMR spectra of the 2-methyl-quinoline with 1 equiv of **L4**-CuBr. (d) ¹³C-NMR spectra of the 2-methyl-quinoline with 1 equiv. of BF₃·Et₂O. (e) ¹³C-NMR spectra of the 2-methyl-quinoline with 1 equiv. of BF₃·Et₂O. (e) ¹³C-NMR spectra of the 2-methyl-quinoline with 1 equiv.

Full assignments of all the peaks

2-methyl-quinoline

2-methyl-quinoline was dissolved in CD_2Cl_2 in a dry NMR tube at -50 °C under N₂ atmosphere and measured by NMR spectroscopy at -50 °C. Full characterization was carried out by ¹H NMR (Figure S17), ¹³C NMR (Figure S18), ¹H-¹³C-HSQCED (Figure S19) and ¹H-¹³C-HMBC (Figure S20).

70.7 70.7

$$6 \underbrace{5}_{7} \underbrace{10}_{8} \underbrace{4}_{9} \underbrace{3}_{N} \underbrace{2}_{1}$$



-5.32

Figure S17. ¹H NMR spectrum of 2-methyl-quinoline (top) and amplified aromatic part of ¹H NMR spectra (bottom).

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-2.71



Figure S18. ¹³C NMR spectrum of 2-methyl-quinoline (top) and amplified aromatic part of ¹H NMR spectra (bottom).



Figure S19. ¹H-¹³C HSQCED spectrum of 2-methyl-quinoline (top) and amplified aromatic part of ¹H-¹³C HSQCED spectra (bottom).



Figure S20. ${}^{1}\text{H}{}^{-13}\text{C}$ HMBC spectrum of 2-methyl-quinoline (top) and amplified aromatic part of ${}^{1}\text{H}{}^{-13}\text{C}$ HMBC spectra (bottom).

2-methyl-quinoline-BF₃ complex

2 Equiv. of $BF_3 \cdot Et_2O$ was added to a solution of 2-methyl-quinoline in CD_2Cl_2 in a dry NMR tube at -50 °C under N₂ atmosphere, leading to instantaneous formation of a new species which was immediately measured by NMR spectroscopy at -50 °C. Full characterization was carried out by ¹H NMR (Figure S21), ¹³C NMR (Figure S22), ¹H-¹³C-HSQCED (Figure S23) and ¹H-¹³C-HMBC (Figure S24). The peaks of 2-methyl-quinoline for both ¹H NMR and ¹³C NMR have shifted after coordinating with BF₃, indicating the formation of 2-methyl-quinoline-BF₃ complex.



Figure S21. ¹H NMR spectrum of the mixture of 2-methyl-quinoline and $BF_3 \cdot Et_2O$ (top) and amplified aromatic part of ¹H NMR spectra (bottom).



Figure S22. ¹³C NMR spectrum of the mixture of 2-methyl-quinoline and $BF_3 \cdot Et_2O$ (top) and amplified aromatic part of ¹³C NMR spectra (bottom).



Figure S23. ¹H-¹³C HSQCED spectrum of the mixture of 2-methyl-quinoline and $BF_3 \cdot Et_2O$ (top) and amplified aromatic part of ¹H-¹³C HSQCED spectra (bottom).





Figure S24. ¹H-¹³C HMBC spectrum of the mixture of 2-methyl-quinoline and $BF_3 \cdot Et_2O$ (top) and amplified aromatic part of ¹H-¹³C HMBC spectra (bottom).

2-methyl-quinoline-L4-CuBr complex

1 Equiv. of L4-CuBr catalyst was added to a solution of 2-methyl-quinoline in CD_2Cl_2 in a dry NMR tube at -50 °C under N₂ atmosphere, leading to instantaneous formation of a new species which was immediately measured by NMR spectroscopy at -50 °C. Full characterization was carried out by ¹H NMR (Figure S25), ¹³C NMR (Figure S26), ¹H-¹³C-HSQCED (Figure S27) and ¹H-¹³C-HMBC (Figure S28). The peaks of 2-methyl-quinoline for both ¹H NMR and ¹³C NMR have shifted after coordinating with CuBr, indicating the formation of 2-methyl-quinoline- L4-CuBr complex.



Figure S25. ¹H NMR spectrum of the mixture of 2-methyl-quinoline and L4-CuBr catalyst (top) and amplified aromatic part of ¹H NMR spectra (bottom).





Figure S26. ¹³C NMR spectrum of the mixture of 2-methyl-quinoline and L4-CuBr catalyst (top) and amplified aromatic part of ¹³C NMR spectra (bottom).



Figure S27. ¹H-¹³C HSQCED spectrum of the mixture of 2-methyl-quinoline and L4-CuBr catalyst (top) and amplified aromatic part of ¹H-¹³C HSQCED spectra (bottom).





Figure S28. ¹H-¹³C HMBC spectrum of the mixture of 2-methyl-quinoline and L4-CuBr catalyst (top) and amplified aromatic part of ¹H-¹³C HMBC spectra (bottom).

Formation of 2-methyl-quinoline/BF₃/L4-CuBr complex

1 Equiv. of L4-CuBr catalyst and 2 equiv. of $BF_3 \cdot Et_2O$ was added to a solution of 2-methyl-quinoline in CD_2Cl_2 in a dry NMR tube at -50 °C under N₂ atmosphere, leading to instantaneous formation of a new species which was immediately measured by NMR spectroscopy at -50 °C. Full characterization was carried out by ¹H NMR (Figure S29), ¹³C NMR (Figure S30), ¹H-¹³C-HSQCED (Figure S31) and ¹H-¹³C-HMBC (Figure S32). The peaks of 2-methyl-quinoline for both ¹H NMR and ¹³C NMR have shifted after adding $BF_3 \cdot Et_2O$ and L4-CuBr, and the spectra are different from 2-methyl-quinoline- BF_3 complex and 2-methyl-quinoline- L4-CuBr complex.



Figure S29. ¹H NMR spectrum of the mixture of 2-methyl-quinoline, $BF_3 \cdot Et_2O$ and L4-CuBr catalyst (top) and amplified aromatic part of ¹H NMR spectra (bottom).



Figure S30. ¹³C NMR spectrum of the mixture of 2-methyl-quinoline, $BF_3 \cdot Et_2O$ and L4-CuBr catalyst (top) and amplified aromatic part of ¹³C NMR spectra (bottom).



Figure S31. ¹H-¹³C HSQCED spectrum of the mixture of 2-methyl-quinoline, $BF_3 \cdot Et_2O$ and L4-CuBr catalyst (top) and amplified aromatic part of ¹H-¹³C HSQCED spectra (bottom).



Figure S32. ¹H-¹³C HMBC spectrum of the mixture of 2-methyl-quinoline, $BF_3 \cdot Et_2O$ and L4-CuBr catalyst (top) and amplified aromatic part of ¹H-¹³C HMBC spectra (bottom).

Comparison of the ¹H NMR spectra of L4-CuBr with the mixture of L4-CuBr catalyst and BF₃·Et₂O

L4-CuBr catalyst was added to CD_2Cl_2 in a dry NMR tube at -50 °C under N_2 atmosphere, and immediately measured by NMR spectroscopy at -50 °C. 2 Equiv. of $BF_3 \cdot Et_2O$ was added to a solution of L4-CuBr in CD_2Cl_2 in a dry NMR tube at -50 °C under N_2 atmosphere, and immediately measured by NMR spectroscopy at -50 °C. Comparison were made for ¹H NMR (Figure S33), ¹³C NMR (Figure S34) and ³¹P NMR (Figure S35). The peaks of L4-CuBr catalyst for ¹H NMR, ¹³C NMR and ³¹P NMR have no shifted after addition of $BF_3 \cdot Et_2O$, indicating that BF_3 cannot displace CuBr to coordinate with L4.



Figure S33. Comparison of the ¹H NMR spectra of L4-CuBr with the mixture of L4-CuBr catalyst and $BF_3 \cdot Et_2O$.



Figure S34. Comparison of the ¹³C NMR spectra of L4-CuBr with the mixture of L4-CuBr catalyst and BF_3 ·Et₂O.



Figure S35. Comparison of the ³¹P NMR spectra of L4-CuBr with the mixture of L4-CuBr catalyst and $BF_3 \cdot Et_2O$.

10. NMR spectra



Figure S36. NMR spectra of 1-acetyl-2-ethyl-C2-dihydroquinoline (2a)



Figure S37. NMR spectra of (S)-1-acetyl-4-ethyl-C4-dihydroquinoline (3a)



Figure S38. NMR spectra of (S)-4-Ethyl-1,2,3,4-tetrahydroquinoline (4a)



Figure S39. NMR spectra of (S)-4-Pentyl-1,2,3,4-tetrahydroquinoline (4b)


Figure S40. NMR spectra of (*R*)-4-(1-methylethyl)-1,2,3,4-tetrahydroquinoline (**4c**)



Figure S41. NMR spectra of (S)-4-(2-methylpropyl)-1,2,3,4-tetrahydroquinoline (4d)



Figure S42. NMR spectra of (S)-4-(3-methylbutyl)-1,2,3,4-tetrahydroquinoline (4e)



Figure S43. NMR spectra of (S)-4-(3-Phenylpropyl)-1,2,3,4-tetrahydroquinoline (4f)



Figure S44. NMR spectra of (S)-4-(Hept-6-en-1-yl)-1,2,3,4-tetrahydroquinoline (4g)

7,7,05 7,7,03 7,

¹H NMR with CDCl₃, 400 MHz



Figure S45. NMR spectra of (*R*)-4-(2-Trimethylsilylethyl)-1,2,3,4-tetrahydroquinoline (4h)



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Figure S46. NMR spectra of (S)-4-Ethyl-6-fluoro-1,2,3,4-tetrahydroquinoline (5a)



Figure S47. NMR spectra of (S)-6-Bromo-4-ethyl-1,2,3,4-tetrahydroquinoline (5b)



Figure S48. NMR spectra of (*S*)-4-Ethyl-6-methyl-1,2,3,4-tetrahydroquinoline (**5c**)



Figure S49. NMR spectra of (S)-4-Ethyl-6-methoxy-1,2,3,4-tetrahydroquinoline (5d)



Figure S50. NMR spectra of (S)-Methyl 4-ethyl-1,2,3,4-tetrahydroquinoline-6-carboxylate (5e)



Figure S51. NMR spectra of (S)-6-Cyano-4-ethyl-1,2,3,4-tetrahydroquinoline (5f)



Figure S52. NMR spectra of (*S*)-7-Bromo-4-ethyl-1,2,3,4-tetrahydroquinoline (5g)



Figure S53. NMR spectra of (S)-4-Ethyl-7-methyl-1,2,3,4-tetrahydroquinoline (5h)



Figure S54. NMR spectra of (S)-4-Ethyl-7-methoxy-1,2,3,4-tetrahydroquinoline (5i)



Figure S55. NMR spectra of (S)-4-Ethyl-7-vinyl-1,2,3,4-tetrahydroquinoline (5j)



Figure S56. NMR spectra of (S)-5-Chloro-4-ethyl-1,2,3,4-tetrahydroquinoline (5k)



Figure S57. NMR spectra of (S)-4-Ethyl-5-methyl-1,2,3,4-tetrahydroquinoline (51)





Figure S58. NMR spectra of (S)-4-Ethyl-8-fluoro-1,2,3,4-tetrahydroquinoline (5m)



Figure S59. NMR spectra of (S)-4-Ethyl-8-methyl-1,2,3,4-tetrahydroquinoline (5n)



Figure S60. NMR spectra of (*S*)-4-Ethyl-1,2,3,4-tetrahydrobenzo[*g*]quinoline (**50**)





Figure S61. NMR spectra of (2*R*,4*S*)-4-Ethyl-2-methyl-1,2,3,4-tetrahydroquinoline (**5p**)

¹H NMR with CDCl₃, 400 MHz





Figure S62. NMR spectra of (2S,4S)-4-ethyl-2-phenyl-1,2,3,4-tetrahydroquinoline (5q)

¹H NMR with CDCl₃, 400 MHz





Figure S63. NMR spectra of (2*S*,4*S*)-1-(3-Bromophenyl)carbonyl-4-ethyl-2-phenyl-1,2,3,4-tetrahydroquinoline (**5t**)

11. HPLC and UV spectra



Figure S64. HPLC and UV spectra of 1-acetyl-2-ethyl-C2-dihydroquinoline (2a)



Figure S65. HPLC and UV spectra of (S)-1-acetyl-4-ethyl-C4-dihydroquinoline (3a)



Figure S66. HPLC and UV spectra of (S)-4-Ethyl-1,2,3,4-tetrahydroquinoline (4a)



Figure S67. HPLC and UV spectra of (S)-4-Pentyl-1,2,3,4-tetrahydroquinoline (4b)



Figure S68. HPLC and UV spectra of (*R*)-4-(1-methylethyl)-1,2,3,4-tetrahydroquinoline (4c)



Figure S69. HPLC and UV spectra of (S)-4-(2-methylpropyl)-1,2,3,4-tetrahydroquinoline (4d)



Figure S70. HPLC and UV spectra of (S)-4-(3-methylbutyl)-1,2,3,4-tetrahydroquinoline (4e)


Figure S71. HPLC and UV spectra of (S)-4-(3-Phenylpropyl)-1,2,3,4-tetrahydroquinoline (4f)



Figure S72. HPLC and UV spectra of (S)-4-(Hept-6-en-1-yl)-1,2,3,4-tetrahydroquinoline (4g)



Figure S73. HPLC and UV spectra of (*R*)-4-(2-Trimethylsilylethyl)-1,2,3,4-tetrahydroquinoline (4h)



Figure S74. HPLC and UV spectra of (*S*)-4-Ethyl-6-fluoro-1,2,3,4-tetrahydroquinoline (5a)



Figure S75. HPLC and UV spectra of (S)-6-Bromo-4-ethyl-1,2,3,4-tetrahydroquinoline (5b)



Figure S76. HPLC and UV spectra of (S)-4-Ethyl-6-methyl-1,2,3,4-tetrahydroquinoline (5c)



Figure S77. HPLC and UV spectra of (S)-4-Ethyl-6-methoxy-1,2,3,4-tetrahydroquinoline (5d)



Figure S78. HPLC and UV spectra of (S)-Methyl 4-ethyl-1,2,3,4-tetrahydroquinoline-6-carboxylate (5e)



Figure S79. HPLC and UV spectra of (S)-6-Cyano-4-ethyl-1,2,3,4-tetrahydroquinoline (5f)



Figure S80. HPLC and UV spectra of (S)-7-Bromo-4-ethyl-1,2,3,4-tetrahydroquinoline (5g)



Figure S81. HPLC and UV spectra of (S)-4-Ethyl-7-methyl-1,2,3,4-tetrahydroquinoline (5h)



Figure S82. HPLC and UV spectra of (S)-4-Ethyl-7-methoxy-1,2,3,4-tetrahydroquinoline (5i)



Figure S83. HPLC and UV spectra of (S)-4-Ethyl-7-vinyl-1,2,3,4-tetrahydroquinoline (5j)



Figure S84. HPLC and UV spectra of (S)-5-Chloro-4-ethyl-1,2,3,4-tetrahydroquinoline (5k)



Figure S85. HPLC and UV spectra of (S)-4-Ethyl-5-methyl-1,2,3,4-tetrahydroquinoline (51)



Figure S86. HPLC and UV spectra of (S)-4-Ethyl-8-fluoro-1,2,3,4-tetrahydroquinoline (5m)



Figure S87. HPLC and UV spectra of (S)-4-Ethyl-8-methyl-1,2,3,4-tetrahydroquinoline (5n)



Figure S88. HPLC and UV spectra of (S)-4-Ethyl-1,2,3,4-tetrahydrobenzo[g]quinoline (50)





11

Peak Table

Conc. 6,779 7,336 42,925 42,959

10

PDA Ch1 254nm Peak# Ret. Time 1 10,532 2 11,761 3 12,371 4 12,892 Total

Total

14 min

mAU

ģ

S127



Figure S89. HPLC and UV spectra of (2R,4S)-4-Ethyl-2-methyl-1,2,3,4-tetrahydroquinoline (5p)





Figure S90. HPLC and UV spectra of (2S,4S)-4-ethyl-2-phenyl-1,2,3,4-tetrahydroquinoline (5q)

12. Cartesian coordinates

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- Cu 0.021730 -0.184070 1.529246
- C 0.273710 -0.054244 -1.847178
- Н 0.868132 0.072295 -2.773002
- Н -0.468459 -0.850058 -2.060575
- C -0.433135 1.249928 -1.481254
- Н 0.301136 2.069686 -1.342136

- Н -1.108320 1.568677 -2.299296
- C 2.957813 0.164587 -0.865011
- C 1.921454 -2.381838 -0.882734
- C 3.537777 -0.778716 -1.933278
- Н 2.734053 1.152978 -1.314513
- C 3.343990 -2.208929 -1.443257
- Н 1.965413 -2.914387 0.086565
- Н 3.005884 -0.621959 -2.892639
- Н 4.605054 -0.552780 -2.125399
- Н 3.562745 -2.950355 -2.234404
- Н 4.067613 -2.404796 -0.628725
- C -2.927212 0.206392 -0.487532
- C -2.213469 2.729246 0.416056
- C -3.798945 1.365292 -0.969339
- Н -2.618438 -0.427926 -1.344131
- C -3.687324 2.505627 0.039030
- Н -2.129152 2.895645 1.505596
- Н -3.430432 1.697773 -1.959182
- Н -4.849249 1.056370 -1.130348
- Н -4.151816 3.436384 -0.337402
- Н -4.243365 2.234609 0.956282
- P 1.321690 -0.636956 -0.416911
- P -1.357014 1.055489 0.130482
- C -3.510439 -0.707482 0.565755
- C -2.723839 -1.769481 1.038352
- C -4.804335 -0.574517 1.077907
- C -3.210462 -2.672108 1.977244
- Н -1.699051 -1.887757 0.654733
- C -5.295698 -1.472785 2.029211
- Н -5.457845 0.235307 0.733616
- C -4.504702 -2.525059 2.480937
- Н -2.572651 -3.495223 2.322228
- Н -6.313358 -1.344225 2.416282
- Н -4.892290 -3.229789 3.225535
- C -1.473371 3.841918 -0.271542
- C -0.317246 4.357025 0.333148
- C -1.847126 4.346916 -1.522160
- C 0.449922 5.333812 -0.296000
- Н -0.013400 3.973620 1.317317
- C -1.083197 5.329296 -2.153005
- Н -2.747567 3.969921 -2.023106
- C 0.070064 5.823722 -1.546387
- Н 1.350066 5.719862 0.196508
- Н -1.395323 5.711135 -3.132233
- Н 0.669926 6.594117 -2.044348
- C 0.891952 -3.068224 -1.731401
- C -0.232299 -3.625601 -1.104529
- C 0.948825 -3.086698 -3.129507
- C -1.284094 -4.154244 -1.848802
- Н -0.279122 -3.637313 -0.005612
- C -0.096416 -3.628445 -3.877938
- Н 1.815474 -2.659900 -3.650516
- C -1.221190 -4.154161 -3.243068
- Н -2.157145 -4.573900 -1.334422
- Н -0.032553 -3.632871 -4.972443
- Н -2.044863 -4.571109 -3.833735
- C 3.898630 0.368761 0.301398
- C 5.022254 1.187665 0.116580

- C 3.723645 -0.238465 1.549467
- C 5.941128 1.391849 1.142285
- Н 5.172079 1.678467 -0.854987
- C 4.642370 -0.037177 2.580891
- Н 2.836464 -0.857947 1.739947
- C 5.754712 0.776757 2.381649
- H 6.808632 2.041218 0.975221
- Н 4.477504 -0.520959 3.551200
- Н 6.475034 0.937797 3.192064
- C 0.314200 -0.949651 3.332049
- Н -0.600054 -0.880090 3.961987
- Н 1.090935 -0.370656 3.882359
- C 0.751039 -2.411350 3.251595
- Н 0.906357 -2.907884 4.238928
- Н 0.010270 -3.038988 2.709516
- Н 1.707449 -2.532665 2.697754

II-pro R_{1-2}

- C -3.717465 -0.604320 -0.810611
- C -3.757500 2.085387 -1.279554
- C -5.084677 0.094063 -0.845520
- Н -3.447777 -0.740691 0.258334
- C -4.853618 1.521919 -0.381614
- Н -4.182743 2.210799 -2.296979
- Н -5.475989 0.087783 -1.884731
- Н -5.820465 -0.449471 -0.221053
- Н -5.771494 2.141231 -0.419884
- Н -4.522665 1.499698 0.676950

- C 2.544338 2.438103 0.570629
- C 1.133749 1.103624 2.526290
- C 3.186571 2.357653 1.959236
- H 1.983874 3.393363 0.501619
- C 2.039812 2.265060 2.954120
- Н 1.619287 0.152358 2.822084
- H 3.823526 1.453026 2.030861
- Н 3.836696 3.232994 2.144780
- Н 2.390769 2.121508 3.993320
- Н 1.479402 3.219908 2.932842
- P 1.253060 1.083797 0.633969
- P -2.492709 0.673645 -1.503607
- Cu 1.690175 -0.624261 -0.834835
- C 1.118905 -2.973746 1.214988
- C 1.788010 -2.652109 -1.067904
- C 1.425916 -3.018140 2.587442
- C -0.181888 -3.350451 0.787843
- C 0.491937 -3.054733 -1.517606
- Н 2.647008 -2.703720 -1.744998
- C 0.471853 -3.438898 3.501872
- Н 2.421948 -2.734709 2.932668
- C -1.137699 -3.745800 1.741970
- C -0.475761 -3.367621 -0.617058
- Н 0.309659 -3.089725 -2.597091
- C -0.817872 -3.800008 3.087525
- Н 0.735274 -3.479982 4.564743
- Н -2.135789 -4.032775 1.387414
- Н -1.477532 -3.674008 -0.942844
- Н -1.562924 -4.124838 3.821943

- N 2.081898 -2.609869 0.270063
- B 3.620153 -2.333331 0.708437
- F 4.121012 -3.479448 1.274513
- F 3.622291 -1.279261 1.612992
- F 4.299872 -1.986027 -0.442107
- C 1.565110 -0.212173 -2.767894
- Н 0.496082 -0.336056 -3.047008
- Н 1.794601 0.874945 -2.783887
- C 2.480039 -0.966056 -3.708210
- Н 3.536127 -0.930306 -3.368366
- Н 2.207285 -2.038209 -3.797090
- Н 2.466966 -0.564163 -4.745343
- C -3.718248 -1.962161 -1.459021
- C -4.099958 -3.076643 -0.698510
- C -3.401136 -2.160587 -2.807959
- C -4.148611 -4.351655 -1.259671
- Н -4.361081 -2.935121 0.360006
- C -3.445355 -3.435077 -3.374035
- Н -3.100153 -1.303079 -3.423932
- C -3.814679 -4.536184 -2.601896
- Н -4.444547 -5.208735 -0.642919
- Н -3.185542 -3.568332 -4.430845
- Н -3.844680 -5.537745 -3.046144
- C -3.093698 3.373501 -0.891446
- C -2.360494 4.080482 -1.857801
- C -3.125400 3.882708 0.412089
- C -1.664297 5.240003 -1.530529
- Н -2.332041 3.699117 -2.887943
- C -2.431820 5.048281 0.743394

- Н -3.695549 3.364615 1.192528
- C -1.692859 5.728513 -0.222354
- Н -1.096389 5.770906 -2.303761
- Н -2.473632 5.427972 1.771977
- Н -1.148501 6.643376 0.038870
- C 3.456732 2.313165 -0.614147
- C 4.341048 1.231128 -0.731688
- C 3.407701 3.253290 -1.648425
- C 5.161001 1.104312 -1.849911
- H 4.385391 0.464584 0.053098
- C 4.228861 3.127352 -2.769967
- Н 2.711599 4.099704 -1.572436
- C 5.110661 2.053151 -2.872740
- Н 5.842213 0.248098 -1.923494
- Н 4.178045 3.876762 -3.568658
- Н 5.756956 1.951806 -3.752412
- C -0.280536 1.098441 3.035633
- C -1.023104 2.270686 3.225623
- C -0.915199 -0.128633 3.270847
- C -2.358445 2.213799 3.626626
- Н -0.566840 3.253243 3.052324
- C -2.249064 -0.191002 3.663652
- Н -0.344877 -1.055736 3.130443
- C -2.980805 0.984519 3.841073
- Н -2.917607 3.145742 3.776028
- Н -2.718391 -1.167115 3.838327
- Н -4.029723 0.943182 4.156756
- C -0.288260 1.827052 -0.045621
- Н -0.024784 2.075641 -1.096645

- Н -0.535490 2.783922 0.454542
- C -1.453195 0.839355 0.033747
- Н -1.085973 -0.201304 0.187350
- Н -2.086107 1.062521 0.914766

II-pro S_{1-2}

С	-3.905993	1.824165	-0.545570
С	-4.718801	-0.449877	0.727714
С	-5.314239	1.294784	-0.852682
Η	-3.255385	1.576959	-1.408189
С	-5.279007	-0.213807	-0.669611
Η	-5.461350	-0.065029	1.458295
Η	-6.045327	1.736837	-0.144683
Η	-5.633624	1.595965	-1.868757
Η	-6.275804	-0.681370	-0.795098
Η	-4.617782	-0.658290	-1.442031
С	1.472030	-2.205714	1.626311
С	0.771182	-2.640866	-0.999902
С	2.199369	-3.328117	0.883298
Η	0.638854	-2.663434	2.197629
С	1.213640	-3.838381	-0.157355
Η	1.616630	-2.331262	-1.646811
Η	3.110540	-2.941723	0.381409
Η	2.524919	-4.120546	1.582623
Η	1.637967	-4.638904	-0.793192
Η	0.340154	-4.272483	0.371794
Р	0.654569	-1.223938	0.249206
Р	-3.264910	0.768891	0.910169

- Cu 1.476940 0.931701 0.184945
- C 4.200333 0.603271 -1.151606
- C 2.653797 2.369467 -0.624425
- C 4.533587 -0.584203 -1.827093
- C 5.136894 1.160069 -0.241679
- C 3.600015 2.955193 0.273666
- Н 1.861341 2.999477 -1.045930
- C 5.753706 -1.203223 -1.596320
- Н 3.828120 -1.022483 -2.534174
- C 6.369257 0.513249 -0.034435
- C 4.808398 2.370022 0.458970
- Н 3.315216 3.877953 0.789354
- C 6.680306 -0.660887 -0.696474
- Н 5.989915 -2.129260 -2.132367
- H 7.078612 0.961634 0.672550
- Н 5.552485 2.804818 1.137137
- Н 7.640391 -1.158148 -0.522579
- N 2.965247 1.241993 -1.359788
- B 2.064459 1.015173 -2.689830
- F 2.683668 1.669073 -3.724537
- F 0.818689 1.560325 -2.401373
- F 1.936467 -0.335746 -2.945169
- C 0.602758 2.085468 1.538098
- Н -0.072745 1.404753 2.098539
- H 1.383110 2.430954 2.248312
- C -0.177429 3.243485 0.947990
- Н -0.791338 3.792224 1.697973
- Н 0.479823 4.000850 0.472797
- Н -0.888599 2.911350 0.161941

- C -4.343417 -1.841279 1.140526
- C -4.108429 -2.870991 0.222378
- C -4.164751 -2.119251 2.505759
- C -3.696486 -4.133313 0.655198
- Н -4.234612 -2.695823 -0.853778
- C -3.746633 -3.373404 2.938827
- Н -4.350930 -1.322376 3.239080
- C -3.506907 -4.390003 2.011195
- Н -3.519940 -4.923623 -0.084647
- Н -3.610758 -3.562678 4.010284
- Н -3.183176 -5.381764 2.347803
- C -3.818782 3.302256 -0.296091
- C -4.459364 3.889742 0.803854
- C -3.078491 4.128310 -1.149077
- C -4.360904 5.257698 1.044398
- Н -5.034795 3.258252 1.495445
- C -2.973611 5.498993 -0.911098
- Н -2.563517 3.683151 -2.011294
- C -3.614472 6.068882 0.187412
- Н -4.866849 5.695444 1.913121
- Н -2.384168 6.125629 -1.590899
- Н -3.532578 7.145250 0.378384
- C -0.447819 -2.815434 -1.868648
- C -0.846014 -1.774973 -2.723722
- C -1.203659 -3.993363 -1.870187
- C -1.970454 -1.900578 -3.533749
- Н -0.259739 -0.848618 -2.746195
- C -2.325984 -4.125475 -2.690801
- Н -0.917185 -4.836790 -1.231623

- C -2.719905 -3.079019 -3.520126
- Н -2.261631 -1.070411 -4.188323
- Н -2.895839 -5.062796 -2.679438
- Н -3.601694 -3.182917 -4.163023
- C 2.261703 -1.321943 2.546241
- C 1.752410 -0.992097 3.807703
- C 3.480414 -0.751904 2.150791
- C 2.434779 -0.117420 4.652686
- Н 0.798859 -1.429690 4.132222
- C 4.168171 0.115140 2.997268
- Н 3.906924 -0.985899 1.163952
- C 3.647225 0.439652 4.250222
- H 2.015066 0.127768 5.635191
- Н 5.124648 0.542433 2.671441
- H 4.188722 1.124312 4.912876
- C -1.912077 -0.120665 -0.019990
- Н -1.254252 0.702003 -0.379184
- Н -2.312186 -0.609925 -0.928984
- C -1.100136 -1.109096 0.808173
- Н -1.029000 -0.777739 1.866650
- Н -1.559335 -2.117710 0.819533

II-proR₃₋₄

- C 3.256710 2.111440 0.009200
- C 3.972726 0.398742 -1.980197
- C 4.752642 2.022392 -0.295454
- Н 3.057508 1.340111 0.776996
- C 4.956666 0.608616 -0.826805

- H 4.403762 0.880365 -2.882738
- Н 5.049167 2.771150 -1.059644
- Н 5.359241 2.222719 0.608483
- Н 6.000645 0.416140 -1.142330
- Н 4.734753 -0.097828 -0.001262
- C -0.378127 -2.589261 -0.982992
- C 0.661333 -3.003227 1.536467
- C -0.213667 -4.039066 -0.503957
- Н 0.476934 -2.360563 -1.652878
- C 0.956236 -4.057721 0.470941
- Н -0.176887 -3.361752 2.168705
- Н -1.135275 -4.385511 0.006428
- Н -0.051032 -4.711959 -1.367325
- Н 1.114972 -5.054759 0.924439
- Н 1.880033 -3.801351 -0.086545
- P -0.078529 -1.566364 0.566674
- P 2.468392 1.543030 -1.611956
- Cu -1.924330 -0.764023 1.642709
- C -3.442020 1.610337 -0.169548
- C -1.414347 2.045959 0.935521
- C -4.256066 1.706671 -1.314377
- C -3.960477 0.978909 0.994108
- C -1.824151 1.386811 2.101531
- Н -0.433420 2.535270 0.893840
- C -5.547825 1.211880 -1.295272
- Н -3.866861 2.180853 -2.217103
- C -5.283312 0.497078 0.986646
- C -3.116841 0.813269 2.155952
- Н -1.174615 1.428156 2.984166

- C -6.073968 0.612837 -0.140480
- Н -6.165953 1.296826 -2.196234
- Н -5.669541 0.023874 1.897850
- Н -3.585921 0.553610 3.112647
- Н -7.102740 0.236762 -0.134441
- N -2.162647 2.165569 -0.152305
- B -1.557900 2.965012 -1.433832
- F -1.452015 2.064206 -2.473371
- F -0.320843 3.441268 -1.052870
- F -2.426708 3.990289 -1.732293
- C -3.107161 -2.165313 2.436440
- Н -4.131643 -2.010612 2.034890
- Н -2.753707 -3.143966 2.044901
- C -3.106389 -2.187599 3.953771
- Н -3.449957 -1.228916 4.394662
- Н -3.764355 -2.975635 4.383423
- Н -2.092925 -2.374873 4.364328
- C 3.619489 -1.013263 -2.358180
- C 4.122734 -2.133295 -1.683572
- C 2.739716 -1.238533 -3.430427
- C 3.736607 -3.426251 -2.044497
- Н 4.835144 -2.005763 -0.859695
- C 2.345221 -2.525142 -3.787574
- Н 2.350487 -0.377574 -3.992731
- C 2.836713 -3.629824 -3.088316
- Н 4.149367 -4.284218 -1.499123
- Н 1.651041 -2.668178 -4.624590
- Н 2.529611 -4.644449 -3.367394
- C 2.690501 3.405587 0.506940

- C 1.932478 3.424164 1.684646
- C 2.859518 4.607781 -0.192495
- C 1.347862 4.603395 2.147375
- Н 1.799152 2.490177 2.251701
- C 2.285253 5.788631 0.270274
- Н 3.437758 4.614495 -1.126022
- C 1.521893 5.790961 1.439706
- Н 0.753815 4.592731 3.069440
- Н 2.426941 6.718773 -0.292656
- H 1.064343 6.720317 1.798123
- C -1.658343 -2.237866 -1.688067
- C -1.631968 -1.582271 -2.923800
- C -2.905679 -2.524274 -1.113920
- C -2.813959 -1.223739 -3.572480
- Н -0.665131 -1.348632 -3.390312
- C -4.087359 -2.168217 -1.758938
- Н -2.958662 -3.032770 -0.143142
- C -4.046045 -1.515357 -2.991795
- Н -2.767443 -0.709228 -4.539272
- Н -5.051536 -2.399481 -1.289821
- Н -4.976743 -1.230966 -3.497006
- C 1.766531 -2.507523 2.423128
- C 3.120325 -2.668749 2.110113
- C 1.431292 -1.746746 3.554148
- C 4.109410 -2.063968 2.887802
- Н 3.418077 -3.265608 1.239342
- C 2.415121 -1.139718 4.329451
- Н 0.370709 -1.619717 3.817216
- C 3.762085 -1.290051 3.993175

- Н 5.164563 -2.200943 2.622505
- Н 2.129528 -0.545450 5.205196
- Н 4.539458 -0.811298 4.599356
- C 1.391478 -0.536264 0.139111
- Н 2.242937 -1.208672 -0.086732
- H 1.646568 0.012730 1.067481
- C 1.113006 0.409950 -1.011344
- Н 0.776756 -0.141332 -1.911569
- Н 0.277791 1.094674 -0.774521

II-pro S_{3-4}

С	-4.289677	0.720279	-0.444702
С	-2.956512	2.407178	-2.147402
С	-5.012228	1.994574	-0.900033
Η	-3.892574	0.904610	0.574723
С	-3.941070	3.039733	-1.168092
Η	-3.458818	2.325695	-3.133145
Η	-5.581376	1.797444	-1.831709
Н	-5.748069	2.322457	-0.140871
Η	-4.360011	3.984555	-1.566440
Н	-3.441320	3.290629	-0.210563
С	-0.428218	1.148780	2.799539
С	0.181394	-1.529118	2.923213
С	-0.133630	0.555657	4.183566
Η	-1.530052	1.134773	2.666731
С	-0.553072	-0.906528	4.111275
Н	1.246680	-1.661381	3.204815
Н	0.949556	0.636849	4.411964

- Н -0.673389 1.114287 4.970779
- Н -0.345099 -1.458709 5.047281
- Н -1.649124 -0.948460 3.946766
- P 0.235614 -0.163438 1.630088
- P -2.767923 0.590000 -1.589160
- Cu 2.356351 0.286727 0.848553
- C 1.647375 -0.174544 -1.758551
- C 3.643090 -1.429998 -1.733444
- C 2.253767 0.947730 -1.167033
- C 4.334358 -0.351400 -1.124769
- C 3.589456 0.816582 -0.707635
- Н 4.152356 1.700413 -0.379242
- N 2.268260 -1.322850 -1.977347
- B 1.407603 -2.577505 -2.548483
- F 0.079015 -2.214710 -2.470850
- F 1.682558 -3.660102 -1.737577
- F 1.800773 -2.805898 -3.847014
- C 3.714412 -0.131995 2.238116
- C 4.847062 0.851803 2.458594
- Н 3.161331 -0.294162 3.189217
- C 5.723710 -0.446991 -0.935161
- C 6.416575 -1.582433 -1.317856
- Н 7.499506 -1.643814 -1.165586
- C 4.357778 -2.577588 -2.117674
- C 5.725402 -2.649683 -1.906349
- Н 6.268048 -3.550089 -2.214700
- Н 6.251506 0.401978 -0.484506
- Н 3.835169 -3.409119 -2.592943
- Н 1.711689 1.902662 -1.140504

- Н 0.595426 -0.141471 -2.066949
- C -0.325260 -2.827183 2.356974
- C 0.378183 -3.425966 1.300152
- C -1.492156 -3.447743 2.814197
- C -0.075573 -4.601663 0.710817
- Н 1.295385 -2.950852 0.917322
- C -1.945144 -4.632541 2.229772
- Н -2.060718 -3.013884 3.644963
- C -1.242121 -5.211227 1.176661
- Н 0.485318 -5.032145 -0.125112
- Н -2.860933 -5.104201 2.605389
- Н -1.603207 -6.137004 0.714176
- C 0.070767 2.541163 2.533509
- C -0.830407 3.592752 2.332811
- C 1.443015 2.824301 2.480897
- C -0.377663 4.888795 2.082404
- Н -1.910384 3.392560 2.374209
- C 1.899899 4.114130 2.219878
- Н 2.169921 2.017413 2.652231
- C 0.989331 5.153356 2.020437
- Н -1.102339 5.697743 1.930808
- Н 2.978532 4.309317 2.180637
- Н 1.346726 6.169929 1.819129
- C -5.122345 -0.528269 -0.407516
- C -5.246311 -1.267893 0.774687
- C -5.780153 -0.997382 -1.553276
- C -6.007017 -2.436561 0.816951
- Н -4.734516 -0.920727 1.683345
- C -6.544450 -2.160635 -1.514791

- Н -5.681530 -0.447501 -2.499296
- C -6.662099 -2.885743 -0.327717
- Н -6.086087 -3.000077 1.754281
- Н -7.049113 -2.509469 -2.423463
- Н -7.260599 -3.803585 -0.298462
- C -1.615932 3.049281 -2.356682
- C -1.040151 3.933717 -1.436043
- C -0.860081 2.689550 -3.484311
- C 0.248007 4.435274 -1.631073
- Н -1.593549 4.242592 -0.540579
- C 0.425114 3.187422 -3.682415
- Н -1.294968 1.996414 -4.217671
- C 0.987242 4.065092 -2.753287
- Н 0.672779 5.126789 -0.892458
- Н 0.992796 2.889197 -4.571773
- Н 1.998275 4.460148 -2.906507
- C -1.182058 -0.451385 0.486515
- Н -0.921017 -1.288088 -0.193610
- Н -2.070565 -0.771310 1.070311
- C -1.444635 0.823432 -0.297995
- Н -0.523744 1.146731 -0.821401
- Н -1.714415 1.662525 0.374061
- Н 4.107963 -1.123836 1.921750
- Н 5.539150 0.543758 3.273070
- Н 5.474471 0.978750 1.554625
- H 4.480607 1.863344 2.732825

TS-(II-III)- *R*₃₋₄

- C -2.160424 2.713496 -0.138907
- C -0.076407 3.628997 -1.659930
- C -2.022336 4.225671 -0.317810
- Н -1.726817 2.455981 0.848893
- C -0.533332 4.499981 -0.493386
- Н -0.529194 4.035712 -2.587425
- Н -2.562894 4.569075 -1.223258
- Н -2.456464 4.770683 0.542615
- Н -0.317239 5.570930 -0.675371
- Н -0.005297 4.219740 0.440993
- C -0.233514 -1.456399 -2.679541
- C 2.225828 -2.321724 -1.807441
- C 0.661364 -2.212869 -3.677039
- Н -0.271608 -0.388824 -2.972259
- C 2.113348 -1.970257 -3.290700
- Н 2.012775 -3.405149 -1.697127
- Н 0.447408 -3.298881 -3.629975
- Н 0.441477 -1.891368 -4.711584
- Н 2.816404 -2.572176 -3.897364
- Н 2.368112 -0.904391 -3.462911
- P 0.728438 -1.480571 -1.055425
- P -0.972698 1.961783 -1.424394
- Cu -0.231753 -2.099149 0.865978
- C -0.048791 0.376900 3.152152
- C 1.534674 -1.299140 2.669314
- C -0.357694 1.714259 3.468372
- C -1.071465 -0.599558 3.269961
- C 0.612597 -2.317578 2.789226
- Н 2.594346 -1.536225 2.519332

- C -1.621253 2.068507 3.918625
- Н 0.414262 2.479422 3.364730
- C -2.328498 -0.216732 3.761714
- C -0.816339 -2.010038 2.896313
- Н 0.965907 -3.355855 2.820663
- C -2.615634 1.100800 4.086299
- Н -1.827807 3.117760 4.159892
- Н -3.099038 -0.988393 3.889838
- Н -1.393512 -2.744053 3.473871
- Н -3.606504 1.375812 4.463633
- N 1.249416 0.017620 2.775308
- B 2.468953 1.044938 2.752642
- F 3.529393 0.407438 2.129171
- F 2.785881 1.385054 4.056380
- F 2.118481 2.186773 2.035405
- C -2.199372 -2.565840 1.380678
- Н -2.961754 -2.129761 2.043223
- Н -2.389973 -2.133345 0.371703
- C -2.221995 -4.075507 1.358415
- Н -1.978930 -4.509828 2.349041
- Н -3.218084 -4.468661 1.069202
- Н -1.495952 -4.494459 0.632317
- C 1.385478 3.406492 -1.906771
- C 2.348309 3.495556 -0.895117
- C 1.799952 2.975371 -3.177963
- C 3.674430 3.133171 -1.135157
- Н 2.066268 3.824599 0.110395
- C 3.124205 2.621980 -3.423670
- Н 1.057064 2.904288 -3.984935

- C 4.068889 2.692762 -2.397076
- Н 4.403079 3.189574 -0.317443
- Н 3.422568 2.288593 -4.425124
- Н 5.111743 2.409114 -2.583977
- C -3.521059 2.086767 -0.222179
- C -3.794652 0.946633 0.548208
- C -4.516741 2.549236 -1.091808
- C -5.025444 0.299204 0.467806
- Н -3.022902 0.564097 1.233095
- C -5.751945 1.905811 -1.171807
- Н -4.330912 3.429274 -1.720422
- C -6.014568 0.780903 -0.389770
- Н -5.214464 -0.588484 1.084868
- Н -6.519940 2.291872 -1.852821
- Н -6.987668 0.279559 -0.450931
- C -1.637872 -1.970258 -2.525205
- C -2.733210 -1.110152 -2.667621
- C -1.882813 -3.310869 -2.196583
- C -4.036458 -1.574237 -2.487074
- Н -2.558346 -0.054764 -2.914167
- C -3.182977 -3.778504 -2.021299
- Н -1.040483 -4.005477 -2.065450
- C -4.266550 -2.909858 -2.162479
- Н -4.877969 -0.879632 -2.598198
- Н -3.351333 -4.831313 -1.764459
- Н -5.289823 -3.276883 -2.021126
- C 3.494757 -2.008217 -1.069747
- C 4.352087 -0.970561 -1.454136
- C 3.811594 -2.741421 0.082609

- C 5.490186 -0.673359 -0.705000
- Н 4.132740 -0.371959 -2.347133
- C 4.946810 -2.445015 0.832480
- Н 3.148016 -3.561171 0.393376
- C 5.791547 -1.406887 0.440363
- Н 6.146240 0.146038 -1.022436
- Н 5.175352 -3.031264 1.730421
- Н 6.683717 -1.168466 1.030336
- C 1.272904 0.252792 -0.759589
- Н 1.717260 0.680652 -1.680271
- Н 2.093478 0.189196 -0.014118
- C 0.153560 1.121763 -0.194291
- Н -0.526843 0.509980 0.439606
- Н 0.586764 1.872105 0.492583

TS-(II-III)-S₃₋₄

- C -2.235278 3.227324 -0.349997
- C -4.178203 1.312261 -0.492977
- C -3.684960 3.689934 -0.558888
- Н -1.704963 3.319738 -1.319034
- C -4.408232 2.582980 -1.304831
- Н -4.704776 1.437427 0.475803
- H -4.169063 3.854855 0.425961
- Н -3.711666 4.658388 -1.093773
- Н -5.489147 2.789933 -1.431820
- Н -3.977474 2.488814 -2.321876
- C -1.712464 -2.208274 0.936605
- C 0.368402 -3.082624 -0.544175

- C -2.066540 -3.434644 0.090392
- Н -2.551301 -1.483949 0.921960
- C -0.815498 -4.070865 -0.542856
- Н 1.039390 -3.342796 0.297085
- Н -2.645178 -4.166455 0.685725
- Н -2.756858 -3.106049 -0.708625
- Н -0.511871 -4.981540 0.006909
- Н -1.055135 -4.414604 -1.564053
- P -0.295341 -1.410526 0.003815
- P -2.341771 1.360255 0.020485
- Cu 1.366021 -0.492778 1.341190
- C 1.462336 2.075425 0.043427
- C 3.530440 1.146099 -0.577619
- C 1.595414 1.686876 1.379431
- C 3.766343 0.745002 0.764580
- C 2.740535 0.949977 1.767270
- Н 3.012288 0.848964 2.824170
- N 2.352549 1.822544 -0.907570
- B 1.976701 2.208136 -2.433557
- F 3.021456 2.914102 -2.985746
- F 0.832821 2.990076 -2.380089
- F 1.740925 1.024889 -3.102723
- C 2.308804 -1.952633 2.333327
- C 2.813709 -1.674948 3.735053
- Н 2.028069 -1.222217 4.374224
- Н 3.149367 -2.595598 4.262428
- Н 3.681165 -0.984114 3.750684
- C -4.606455 0.000279 -1.077724
- C -4.603653 -0.252553 -2.455908

- C -5.002366 -1.037229 -0.219050
- C -4.966544 -1.504170 -2.955177
- Н -4.322799 0.538350 -3.161537
- C -5.368846 -2.285986 -0.714091
- Н -5.028044 -0.851130 0.863856
- C -5.347785 -2.527111 -2.088596
- Н -4.958119 -1.676508 -4.037857
- Н -5.674738 -3.078249 -0.020142
- Н -5.637288 -3.507783 -2.483330
- C -1.483949 4.016169 0.686835
- C -0.474402 4.910267 0.310044
- C -1.785358 3.893269 2.050997
- C 0.222966 5.650977 1.265235
- Н -0.219934 5.013238 -0.753331
- C -1.095810 4.634800 3.007503
- Н -2.572770 3.195618 2.369710
- C -0.084822 5.515331 2.617529
- Н 1.014468 6.339491 0.946973
- Н -1.347099 4.521929 4.068764
- Н 0.462318 6.095105 3.369878
- C -1.319161 -2.444273 2.372264
- C -1.213669 -1.338910 3.232735
- C -1.041699 -3.712979 2.887550
- C -0.846566 -1.499359 4.564500
- Н -1.438223 -0.334893 2.842782
- C -0.669160 -3.876393 4.224385
- Н -1.114615 -4.598624 2.245059
- C -0.571129 -2.773362 5.067163
- Н -0.777929 -0.622784 5.219828

- Н -0.456905 -4.881571 4.607077
- Н -0.281344 -2.902848 6.116330
- C 1.224403 -2.954005 -1.775752
- C 0.742967 -3.192857 -3.068734
- C 2.543577 -2.495880 -1.637072
- C 1.554488 -2.989919 -4.184877
- Н -0.288768 -3.533974 -3.219425
- C 3.356053 -2.293482 -2.750099
- Н 2.939194 -2.294568 -0.630361
- C 2.864749 -2.541475 -4.031443
- Н 1.153506 -3.184577 -5.186484
- Н 4.384368 -1.939331 -2.610997
- Н 3.501983 -2.383806 -4.909042
- C -1.106005 -0.747523 -1.531677
- Н -1.993341 -1.374296 -1.753272
- Н -0.383288 -0.918155 -2.354153
- C -1.466658 0.731801 -1.507676
- Н -0.539349 1.328708 -1.577243
- Н -2.039120 0.985950 -2.422229
- Н 0.848478 2.032899 2.105010
- Н 0.586199 2.654115 -0.263812
- C 4.980134 0.113311 1.090481
- C 5.933894 -0.139889 0.121883
- Н 6.874361 -0.633563 0.389016
- C 5.687556 0.241457 -1.204895
- Н 6.436616 0.041610 -1.979597
- C 4.507714 0.876966 -1.553424
- Н 4.336003 1.180130 -2.587699
- Н 5.149977 -0.183368 2.132628

- Н 3.136807 -2.291141 1.671920
- Н 1.554817 -2.767189 2.364775

III-*R*₁₋₂

- C -4.150875 -1.853425 0.721740
- C -3.570263 0.343472 2.240354
- C -4.253194 -1.995712 2.243781
- Н -3.179208 -2.288338 0.412221
- C -3.247060 -1.022799 2.835745
- Н -4.535681 0.680933 2.670764
- Н -5.274469 -1.733681 2.588530
- Н -4.069039 -3.041816 2.554126
- Н -3.273838 -0.998656 3.942749
- Н -2.227094 -1.349551 2.547688
- C 0.088543 0.184569 -2.725178
- C 0.585788 2.791432 -2.013476
- C 0.831674 1.077368 -3.729907
- Н -0.966462 0.111460 -3.057543
- C 0.367864 2.508613 -3.500359
- Н 1.680078 2.761141 -1.830243
- Н 1.926011 1.010048 -3.563342
- Н 0.641370 0.728682 -4.761574
- Н 0.919889 3.237060 -4.123272
- Н -0.704447 2.599464 -3.767263
- P -0.010169 1.232553 -1.161717
- P -4.010806 0.021047 0.406203
- Cu 1.101793 0.678554 0.703910
- C 4.278266 -0.691608 0.290129

- C 2.724894 -1.437558 2.034725
- C 5.176663 -0.976304 -0.771152
- C 3.815738 0.656123 0.381486
- C 2.469136 0.024358 2.324655
- Н 2.915491 -1.935863 3.003176
- C 5.524530 -0.008584 -1.698674
- Н 5.582120 -1.986912 -0.864757
- C 4.192884 1.622806 -0.565268
- C 3.007172 1.004566 1.533116
- H 1.985983 0.280186 3.278353
- C 5.028674 1.303539 -1.622621
- Н 6.205071 -0.282830 -2.514290
- Н 3.820584 2.649622 -0.436575
- H 2.919391 2.067084 1.812513
- Н 5.317266 2.057946 -2.362121
- N 3.914757 -1.632541 1.208765
- B 4.787657 -2.914742 1.410690
- F 4.673384 -3.777796 0.306237
- F 4.354694 -3.572015 2.557871
- F 6.124863 -2.547912 1.552203
- C 1.457411 -2.082743 1.449733
- Н 0.599658 -1.765724 2.084739
- Н 1.285857 -1.657211 0.431563
- C 1.485982 -3.596080 1.355963
- Н 1.673583 -4.045625 2.348934
- Н 0.524067 -3.984198 0.972321
- Н 2.287470 -3.935692 0.677088
- C -2.579655 1.456815 2.407767
- C -1.234950 1.245236 2.740833

- C -3.002372 2.778055 2.185607
- C -0.345560 2.320995 2.839687
- Н -0.875352 0.232740 2.966499
- C -2.114759 3.846655 2.262069
- Н -4.055809 2.961772 1.933512
- C -0.776444 3.621671 2.589575
- Н 0.690321 2.146067 3.155717
- Н -2.468337 4.867097 2.071872
- Н -0.077632 4.461804 2.669405
- C -5.237642 -2.504777 -0.082317
- C -4.922335 -3.425285 -1.088566
- C -6.587925 -2.195958 0.134886
- C -5.923177 -4.027835 -1.851462
- Н -3.868882 -3.676614 -1.273957
- C -7.590169 -2.797664 -0.621276
- Н -6.859518 -1.461509 0.905503
- C -7.261535 -3.718160 -1.618334
- Н -5.652707 -4.747770 -2.632882
- Н -8.639961 -2.542797 -0.433931
- Н -8.050430 -4.191287 -2.214427
- C 0.626980 -1.203768 -2.531782
- C -0.217492 -2.312014 -2.670014
- C 1.967156 -1.424420 -2.189303
- C 0.261694 -3.607514 -2.475060
- Н -1.269753 -2.158475 -2.946132
- C 2.447438 -2.716315 -1.986397
- Н 2.653585 -0.575230 -2.054596
- C 1.596729 -3.812191 -2.132747
- Н -0.415986 -4.461342 -2.592997

- Н 3.487939 -2.866681 -1.675204
- Н 1.976602 -4.826897 -1.965034
- C 0.049635 4.059767 -1.417798
- C -1.177617 4.612771 -1.806132
- C 0.785587 4.709078 -0.418089
- C -1.650402 5.782321 -1.212497
- Н -1.775983 4.133819 -2.590843
- C 0.317615 5.880037 0.173512
- Н 1.752534 4.288945 -0.106028
- C -0.904912 6.421168 -0.221872
- Н -2.610265 6.202295 -1.534564
- Н 0.916269 6.376682 0.946272
- Н -1.274883 7.344974 0.237521
- C -1.816813 1.362039 -0.853558
- Н -2.354436 1.678617 -1.768909
- Н -1.962210 2.153755 -0.092907
- C -2.308510 0.003066 -0.363416
- Н -2.375528 -0.708381 -1.212145
- Н -1.585000 -0.445005 0.353535

III- S_{1-2}

- C -3.959835 0.163008 -1.309831
- C -2.145107 1.942317 -2.325633
- C -4.328057 0.905069 -2.604449
- Н -4.594709 0.544430 -0.485547
- C -3.610582 2.249237 -2.613026
- Н -1.757396 1.315439 -3.157059
- Н -3.996379 0.326220 -3.487257

- H -5.424817 1.012731 -2.691518
- Н -3.732139 2.779737 -3.577031
- Н -4.037418 2.907881 -1.829949
- C 1.599911 2.222717 1.916408
- C -0.092822 0.453602 3.159522
- C 1.953504 1.755653 3.337236
- Н 1.155556 3.236101 1.994248
- C 0.648308 1.447338 4.050541
- Н 0.484760 -0.490455 3.171330
- Н 2.579952 0.841077 3.276307
- H 2.556916 2.525514 3.853728
- Н 0.802950 1.029282 5.063679
- Н 0.071467 2.385915 4.173084
- P 0.161275 1.116299 1.407747
- P -2.215075 0.747545 -0.877726
- Cu -0.178493 -0.306892 -0.398062
- C 1.720640 -3.362696 0.671893
- C 2.013467 -1.993377 -1.332374
- C 0.321976 -3.635247 0.197482
- C 0.590377 -1.960288 -1.544351
- C -0.225572 -2.921282 -0.798606
- Н -1.261638 -3.087811 -1.127823
- N 2.563013 -2.786984 -0.380701
- B 4.091492 -3.129651 -0.415596
- F 4.858137 -1.982196 -0.145683
- F 4.433594 -3.610451 -1.678125
- F 4.356196 -4.099860 0.543021
- C 0.057411 -1.265259 -2.663390
- C 0.864587 -0.534273 -3.522163

- Н 0.436898 -0.000610 -4.378050
- C 2.807753 -1.218407 -2.221567
- C 2.245654 -0.514206 -3.271098
- Н 2.906976 0.063921 -3.928736
- Н -1.018444 -1.373889 -2.862029
- Н 3.890452 -1.204586 -2.083552
- Н -0.259697 -4.401143 0.727127
- Н 2.188099 -4.315297 0.979769
- C -1.522322 0.119280 3.469229
- C -2.046769 -1.105394 3.027520
- C -2.365888 0.990040 4.169053
- C -3.372952 -1.447382 3.278033
- Н -1.395599 -1.806245 2.484194
- C -3.699118 0.654462 4.407907
- H -1.983013 1.945908 4.545616
- C -4.208761 -0.563959 3.963329
- Н -3.755284 -2.419335 2.944147
- Н -4.342072 1.351108 4.958210
- Н -5.252662 -0.832257 4.161852
- C 2.782274 2.293300 0.992889
- C 3.390859 3.532564 0.758537
- C 3.353078 1.149465 0.422055
- C 4.535091 3.632810 -0.030492
- Н 2.953926 4.436744 1.204063
- C 4.501320 1.244649 -0.363614
- H 2.902956 0.158227 0.583855
- C 5.093021 2.486347 -0.595964
- Н 4.994450 4.613340 -0.203249
- Н 4.935149 0.328435 -0.778212

- Н 5.995560 2.558906 -1.213891
- C -4.031220 -1.340692 -1.311359
- C -4.335571 -2.006999 -0.116849
- C -3.717975 -2.113614 -2.436505
- C -4.322506 -3.398163 -0.042908
- Н -4.584876 -1.415112 0.775377
- C -3.710614 -3.507241 -2.368252
- Н -3.472285 -1.632587 -3.391632
- C -4.009734 -4.156348 -1.171562
- Н -4.568424 -3.896188 0.902774
- Н -3.464493 -4.090438 -3.263167
- Н -4.002973 -5.250889 -1.119165
- C -1.171427 3.057888 -2.083407
- C -1.571334 4.362324 -1.778181
- C 0.201944 2.765320 -2.101986
- C -0.622690 5.341953 -1.472985
- Н -2.635572 4.626831 -1.773217
- C 1.147653 3.738329 -1.798064
- Н 0.533443 1.748417 -2.360598
- C 0.736151 5.033173 -1.472319
- Н -0.954593 6.359175 -1.234262
- H 2.214439 3.482653 -1.812146
- Н 1.479010 5.801883 -1.227665
- C -1.201115 2.370872 1.210570
- Н -1.403353 2.848634 2.189803
- Н -0.777846 3.159286 0.557357
- C -2.493195 1.823860 0.611072
- Н -3.188816 2.653542 0.374385
- Н -3.013758 1.171895 1.343998

- C 1.652328 -2.456756 1.907643
- Н 1.218677 -1.496162 1.555780
- Н 0.908544 -2.890798 2.610573
- C 2.963729 -2.208036 2.626685
- Н 3.692288 -1.689328 1.977537
- Н 2.805509 -1.586280 3.529681
- Н 3.428468 -3.157670 2.950703

EtMgBr

С	3.665331	-0.579973	0.000017
Н	3.519564	-1.231160	0.885705
Н	3.519604	-1.231090	-0.885728
Н	4.741290	-0.296031	0.000053
С	2.712398	0.614969	0.000042
Н	2.932050	1.259539	-0.879254
Н	2.932029	1.259483	0.879386
Mg	0.639519	0.186040	0.000021
Br	-1.816718	-0.062949	-0.000022

Et₂O

0	0.074236	-0.590758	-0.295823
С	-1.238248	-0.626216	0.200003
С	0.913893	0.349567	0.318735
С	2.329842	0.105674	-0.138276
Η	0.605332	1.388573	0.065481
Η	0.837191	0.257885	1.427558
Н	3.023002	0.835334	0.316312

- Н 2.659764 -0.911734 0.140545
- Н 2.401651 0.197739 -1.237525
- Н -1.699881 -1.533260 -0.232872
- Н -1.220991 -0.767730 1.305642
- C -2.063347 0.593358 -0.157041
- Н -3.116891 0.448261 0.143324
- Н -1.699099 1.507907 0.346398
- Н -2.036806 0.768790 -1.248799

V·2Et₂O

С	-3.622484	-0.704994	1.755425
С	-2.355823	-0.531554	1.246125
С	-2.047203	-0.987087	-0.060723
С	-3.061656	-1.630678	-0.828429
С	-4.355882	-1.794366	-0.273713
С	-4.632313	-1.337819	0.993067
Η	-3.852564	-0.350431	2.766342
Η	-1.568446	-0.046380	1.837490
С	-2.725284	-2.082193	-2.126761
Η	-5.126208	-2.291060	-0.875687
Η	-5.633763	-1.464620	1.418405
С	-1.451286	-1.893219	-2.605544
С	-0.514676	-1.235841	-1.781569
Η	-3.492040	-2.578611	-2.734120
Η	-1.155806	-2.229002	-3.603647
Η	0.504226	-1.042292	-2.148990
N	-0.792001	-0.796024	-0.568938
Mg	0.774770	0.380319	0.455606

- Br 2.250179 1.377171 -1.403123
- C 0.884139 0.434971 2.606945
- Н 0.633230 -0.582500 2.991761
- C 2.299481 0.789565 3.066651
- Н 0.155213 1.110165 3.113442
- Н 2.479295 0.731948 4.166054
- Н 3.056945 0.123339 2.597777
- H 2.584051 1.821435 2.764444
- O -0.587005 2.120262 0.186447
- C -0.306855 3.224866 1.038256
- Н 0.138298 4.049565 0.446079
- Н 0.477101 2.860690 1.725528
- C -1.518649 3.683532 1.814019
- Н -1.225921 4.442795 2.561159
- Н -2.288897 4.139267 1.164179
- Н -1.977584 2.830275 2.349667
- C -1.487922 2.361924 -0.888253
- Н -1.498846 1.424279 -1.471217
- Н -2.512769 2.499583 -0.483173
- C -1.092801 3.513879 -1.783078
- Н -0.043938 3.397699 -2.114447
- Н -1.742163 3.521186 -2.676919
- Н -1.203101 4.497372 -1.290639
- O 1.926076 -1.539224 0.420009
- C 3.284788 -1.700369 0.024537
- Н 3.892661 -2.002385 0.900816
- Н 3.612340 -0.690094 -0.274166
- C 3.452359 -2.665673 -1.124744
- Н 4.503183 -2.664256 -1.465773

- Н 3.196378 -3.706096 -0.850144
- Н 2.817006 -2.363065 -1.978938
- C 1.283896 -2.687853 0.959405
- Н 0.276916 -2.344147 1.256016
- Н 1.139438 -3.443364 0.158008
- C 1.991931 -3.284193 2.153451
- Н 2.945877 -3.774209 1.885908
- Н 2.197379 -2.503045 2.909780
- Н 1.347880 -4.052386 2.617561

TS-(V-VI)·2Et₂O

С	-3.388601	2.041363	0.762698
С	-2.264879	1.578500	0.103255
С	-2.280927	0.334838	-0.576765
С	-3.496852	-0.411453	-0.575643
С	-4.626283	0.084866	0.102793
С	-4.581323	1.294243	0.773989
Н	-3.348803	3.006663	1.282145
Н	-1.338568	2.165572	0.087471

- C -3.509635 -1.650632 -1.294276
- H -5.548366 -0.511056 0.087510
- Н -5.464788 1.671680 1.300532
- C -2.382229 -2.105903 -1.901742
- C -1.169814 -1.343127 -1.781571
- Н -4.450243 -2.213766 -1.356726
- Н -2.370249 -3.046541 -2.463097
- Н -0.386683 -1.540882 -2.527002
- N -1.166145 -0.104726 -1.230689

- C 0.011048 -2.758048 -0.372797
- Н -0.050317 -3.463839 -1.225190
- Н 1.076205 -2.843112 -0.044276
- C -0.914861 -3.179002 0.749095
- Н -0.729296 -2.603295 1.677332
- Н -1.982813 -3.015109 0.487641
- Н -0.833852 -4.253769 1.027399
- Mg 0.740250 -0.483250 -0.345984
- Br 3.130113 -0.930116 -1.116915
- O 0.774585 -0.403500 1.742173
- O 1.170506 1.649680 -0.504995
- C -0.049620 0.455683 2.544304
- H 0.543657 0.788139 3.418407
- Н -0.251937 1.351547 1.929758
- C -1.339861 -0.203161 2.962405
- Н -1.166597 -1.089078 3.600641
- Н -1.950051 0.514538 3.540648
- Н -1.928356 -0.513826 2.077787
- C 1.632605 -1.279055 2.494261
- Н 1.082674 -1.610696 3.395669
- Н 1.792017 -2.172074 1.864546
- C 2.943560 -0.622506 2.844889
- Н 2.797924 0.292412 3.449144
- Н 3.561626 -1.319077 3.439376
- Н 3.503453 -0.362257 1.927669
- C 2.105002 2.264581 0.381177
- Н 3.124040 2.150378 -0.042322
- Н 2.074597 1.663136 1.305884
- C 1.779354 3.708327 0.675714

- H 2.473871 4.098996 1.440943
- Н 1.879771 4.349332 -0.219054
- Н 0.747539 3.810829 1.062787
- C 1.224569 2.125221 -1.858540
- Н 2.199771 2.628985 -2.008210
- Н 1.231776 1.237234 -2.518639
- C 0.069073 3.027157 -2.214192
- Н 0.005966 3.901958 -1.542056
- Н 0.194543 3.398302 -3.247846
- Н -0.885132 2.474031 -2.161940

VI·2Et₂O

С	-2.316472	2.473229	1.331594
С	-1.485462	1.415228	0.981898
С	-1.936616	0.333519	0.179739
С	-3.300832	0.391175	-0.248000
С	-4.117313	1.463990	0.117423
С	-3.647819	2.514146	0.906704
Н	-1.915099	3.281968	1.956013
Н	-0.451336	1.413154	1.357828
С	-3.749895	-0.645470	-1.160544
Н	-5.154137	1.471507	-0.248598
Н	-4.303657	3.346663	1.185242
С	-2.949279	-1.672109	-1.486942
С	-1.603974	-1.863263	-0.844342
Н	-4.748689	-0.543026	-1.608882
Н	-3.280598	-2.439118	-2.201121
Н	-0.890301	-2.164587	-1.647174
- N -1.080997 -0.648112 -0.216869
- C -1.621209 -3.056633 0.129388
- Н -1.954215 -3.956901 -0.429955
- Н -0.571481 -3.259420 0.432720
- C -2.490307 -2.855874 1.355794
- Н -2.464176 -3.740061 2.019796
- Н -2.153776 -1.982431 1.946014
- Н -3.546771 -2.678102 1.076450
- Mg 0.889652 -0.296464 -0.270709
- Br 2.406854 -1.770056 -1.549810
- O 1.624587 -0.268618 1.665542
- O 1.088294 1.582636 -1.109136
- C 1.113372 -1.252069 2.592701
- Н 1.969312 -1.638682 3.178355
- H 0.730100 -2.100925 1.993133
- C 0.029623 -0.690344 3.476406
- Н -0.814037 -0.312440 2.870970
- Н -0.355424 -1.486499 4.139005
- Н 0.400633 0.135597 4.109002
- C 2.918447 0.255871 1.990784
- Н 3.226980 0.854567 1.114646
- Н 3.629325 -0.590376 2.072561
- C 2.913877 1.108672 3.235031
- Н 2.710306 0.515779 4.144456
- H 3.903305 1.582128 3.363520
- H 2.153230 1.909012 3.163662
- C 1.629182 2.754617 -0.492403
- Н 1.536677 2.597141 0.598255
- Н 0.978857 3.616135 -0.739798

- C 3.060970 3.002640 -0.895386
- Н 3.154965 3.206073 -1.977103
- Н 3.459740 3.878696 -0.353831
- Н 3.698086 2.130097 -0.660576
- C 0.617722 1.767945 -2.457387
- Н 1.301995 2.474101 -2.964248
- Н 0.742444 0.790100 -2.957661
- C -0.812817 2.241857 -2.509444
- Н -0.946178 3.193883 -1.962655
- Н -1.107191 2.411706 -3.560265
- Н -1.497849 1.495688 -2.069151

TS-(V-VII)·2Et₂O

С	5.120567	-0.228570	0.032548
С	4.223783	-0.840427	-0.821478
С	2.976888	-0.246790	-1.130627
С	2.647374	0.992831	-0.512342
С	3.592105	1.611041	0.329458
С	4.811799	1.017688	0.606265
Н	6.082119	-0.708358	0.251834

- Н 4.452266 -1.799545 -1.301346
- C 1.334101 1.579501 -0.763666
- Н 3.345678 2.581514 0.778400
- Н 5.530669 1.515672 1.266754
- C 0.591538 0.948457 -1.812504
- C 1.065708 -0.249761 -2.380136
- Н 1.250936 2.668098 -0.666283
- Н -0.273208 1.456823 -2.262409

- Н 0.454777 -0.717390 -3.170155
- N 2.173205 -0.877701 -2.046903
- C 0.466089 1.642333 1.289677
- Н -0.474133 1.300636 1.785971
- Н 1.275621 1.099136 1.816941
- C 0.558145 3.135764 1.526679
- Н -0.159316 3.706172 0.900576
- Н 0.355295 3.420487 2.579993
- Н 1.560586 3.540970 1.289447
- Mg -0.614965 -0.018418 0.000347
- Br -1.396521 -2.219851 -1.041066
- O -2.454657 1.014732 -0.189126
- O -0.670746 -1.069843 1.864426
- C -3.512922 0.526489 -1.045134
- H -4.441676 1.051503 -0.753768
- Н -3.634689 -0.544602 -0.807034
- C -3.212869 0.708829 -2.511412
- Н -4.077382 0.360996 -3.105874
- Н -3.024725 1.766396 -2.769753
- Н -2.338629 0.104117 -2.814367
- C -2.743958 2.266556 0.449136
- Н -2.036569 2.355156 1.289567
- Н -3.758046 2.201899 0.889762
- C -2.617304 3.442399 -0.486317
- Н -3.398168 3.438582 -1.268397
- Н -2.717200 4.385460 0.080812
- Н -1.627001 3.442281 -0.981698
- C -1.955061 -1.112805 2.491756
- Н -2.640839 -0.536256 1.841575

- Н -2.315393 -2.161158 2.482642
- C -1.926882 -0.532225 3.885062
- Н -1.586142 0.519527 3.863987
- Н -2.939381 -0.558136 4.325760
- Н -1.256761 -1.098317 4.556679
- C 0.171725 -2.172349 2.205511
- Н 0.232934 -2.233874 3.309625
- Н -0.308438 -3.101424 1.834701
- C 1.541201 -1.998132 1.606100
- Н 1.521533 -2.057579 0.501097
- Н 1.997916 -1.034988 1.903764
- H 2.203808 -2.808166 1.958811

VII·2Et₂O

С	-2.135000	2.989758	-0.002348
С	-1.307094	1.873793	0.003760
С	-1.795813	0.570808	-0.231935
С	-3.192097	0.424263	-0.444456
С	-4.003185	1.560126	-0.470479
С	-3.499991	2.843710	-0.257053
Η	-1.705008	3.982793	0.181670
Η	-0.230952	2.008101	0.184366
С	-3.805665	-0.955243	-0.535608
Η	-5.079446	1.421624	-0.651066
Η	-4.164194	3.715230	-0.278870
С	-2.753581	-1.935949	-0.970087
С	-1.445933	-1.682840	-0.750396
Н	-4.639072	-0.928376	-1.274740

- Н -3.050248 -2.903476 -1.393566
- Н -0.697706 -2.458899 -0.983013
- N -0.915871 -0.494146 -0.279828
- C -4.437512 -1.347986 0.814641
- Н -3.638826 -1.326275 1.587248
- Н -5.167869 -0.562479 1.102240
- C -5.116817 -2.704586 0.800570
- Н -5.618377 -2.920098 1.762091
- Н -5.885556 -2.755669 0.003788
- Н -4.392595 -3.519976 0.614812
- Mg 1.073489 -0.420600 0.052925
- Br 2.354083 -2.530940 -0.074878
- O 1.515417 0.482072 1.865976
- O 2.009787 0.898217 -1.234292
- C 0.418564 0.661464 2.784570
- Н 0.547988 -0.058028 3.616350
- Н -0.493231 0.368588 2.232175
- C 0.322522 2.084124 3.273266
- Н -0.546537 2.187591 3.947033
- Н 1.220127 2.395294 3.837778
- Н 0.180992 2.781749 2.427282
- C 2.780326 0.277085 2.521261
- Н 2.810016 -0.768940 2.887198
- H 2.821081 0.945916 3.400663
- C 3.922121 0.567881 1.586255
- H 4.876033 0.392356 2.113857
- Н 3.908492 -0.095908 0.700717
- Н 3.907028 1.620808 1.252924
- C 2.225395 2.296798 -1.008981

- Н 1.915567 2.475989 0.037742
- Н 3.315943 2.489969 -1.062172
- C 1.463093 3.182094 -1.962710
- Н 0.378719 2.966663 -1.926124
- Н 1.615786 4.239834 -1.682543
- Н 1.810602 3.065246 -3.005186
- C 2.516502 0.375234 -2.477985
- Н 3.224549 1.112783 -2.898855
- Н 3.095529 -0.532219 -2.223122
- C 1.394643 0.054972 -3.433268
- Н 0.703900 -0.692287 -2.999226
- Н 0.805836 0.954737 -3.686508
- Н 1.803257 -0.368860 -4.368073

TS-(1a·BF₃-VIII)·2Et₂O

- C 3.981840 -1.825269 1.274446
- C 2.920960 -0.960702 1.494490
- C 2.780581 0.203843 0.718465
- C 3.746518 0.472607 -0.281122
- C 4.817961 -0.415230 -0.478071
- C 4.940100 -1.561739 0.286703
- Н 4.063205 -2.731915 1.885044
- Н 2.186765 -1.200571 2.265462
- C 3.594913 1.660530 -1.071332
- Н 5.551365 -0.180652 -1.259587
- Н 5.772937 -2.254070 0.122972
- C 2.499108 2.442602 -0.936591
- C 1.461851 2.076886 -0.006762

- Н 4.385731 1.919918 -1.786004
- Н 2.366525 3.347204 -1.537592
- N 1.725399 1.099273 0.908764
- B 0.693142 0.904343 2.053908
- F -0.143274 -0.226009 1.586208
- F 1.249784 0.564917 3.246224
- F -0.134601 1.992139 2.149088
- Н 0.791634 2.860108 0.356059
- C -0.066150 1.779421 -1.670169
- Н 0.678873 1.317872 -2.341055
- Н -1.029596 1.272549 -1.959489
- C -0.275899 3.260380 -1.976367
- Н 0.039518 3.940736 -1.153417
- Н 0.286604 3.591098 -2.871279
- Н -1.334861 3.510727 -2.175787
- Mg -0.771904 -0.054262 -0.327621
- Br 0.777189 -1.595607 -1.619783
- O -2.412499 1.037222 0.502199
- O -2.152481 -1.700115 -0.226553
- C -2.677242 2.427288 0.271698
- Н -3.082797 2.861423 1.205158
- Н -1.703055 2.914503 0.091894
- C -3.616547 2.647598 -0.888644
- Н -4.600216 2.174029 -0.718685
- Н -3.778823 3.730624 -1.039307
- Н -3.199064 2.238555 -1.827900
- C -3.076004 0.487675 1.647580
- Н -2.878740 1.153670 2.510755
- Н -2.577463 -0.469302 1.863473

- C -4.552148 0.273535 1.428838
- Н -4.987080 -0.237129 2.307278
- Н -5.100308 1.223869 1.294999
- Н -4.727300 -0.363308 0.541037
- C -1.788794 -2.792304 0.628278
- Н -1.114524 -2.362630 1.389131
- Н -1.193120 -3.513796 0.033591
- C -2.978936 -3.449816 1.284709
- Н -2.625584 -4.177721 2.036243
- Н -3.608763 -2.702876 1.803690
- Н -3.611291 -4.001583 0.566777
- C -2.876468 -2.104034 -1.388853
- Н -3.866503 -2.490416 -1.078151
- Н -2.321083 -2.931955 -1.876488
- C -3.040086 -0.939839 -2.327304
- Н -3.646880 -1.242050 -3.198114
- Н -3.556547 -0.095598 -1.833927
- Н -2.062534 -0.597408 -2.722592

VIII·2Et₂O

С	1.810446	3.298571	0.447981
С	1.755039	1.997109	0.941657
С	2.310674	0.920876	0.223871
С	2.948145	1.213238	-1.013385
С	2.990798	2.529067	-1.488570

- C 2.425009 3.579598 -0.773079
- Н 1.371567 4.109401 1.042424
- Н 1.292799 1.816030 1.914681

- C 3.476962 0.108008 -1.797851
- Н 3.480433 2.716248 -2.453904
- Н 2.470485 4.605266 -1.155945
- C 3.589792 -1.116369 -1.263980
- C 3.276072 -1.334489 0.189322
- Н 3.758384 0.304151 -2.841725
- Н 3.966225 -1.962048 -1.853302
- N 2.244950 -0.399829 0.662519
- B 1.138645 -1.032112 1.425540
- F 0.252888 -0.101422 2.059767
- F 0.125122 -1.652483 0.487728
- F 1.516750 -2.013645 2.286439
- Н 2.868649 -2.358391 0.306882
- Mg -1.363513 -0.248958 0.691505
- Br -2.760070 1.410669 1.910288
- O -0.833459 0.679117 -1.077048
- C -1.150086 2.063796 -1.318566
- Н -0.957279 2.597848 -0.369981
- Н -0.423857 2.448552 -2.060195
- C -2.578100 2.253911 -1.765422
- Н -2.776572 3.328720 -1.929655
- Н -2.794328 1.718272 -2.707096
- Н -3.277420 1.900835 -0.984621
- C -0.087140 0.027552 -2.116939
- Н 0.797162 0.656202 -2.339761
- Н 0.295811 -0.903448 -1.666276
- C -0.901269 -0.250950 -3.354483
- Н -0.303965 -0.861280 -4.055916
- Н -1.819529 -0.817191 -3.108355

- Н -1.189643 0.674524 -3.884996
- C 4.537199 -1.245015 1.058142
- Н 4.927091 -0.207456 0.995822
- Н 4.227247 -1.403823 2.109970
- C 5.617746 -2.238490 0.672100
- Н 6.465100 -2.206393 1.380509
- Н 5.225847 -3.274965 0.669640
- Н 6.025152 -2.034185 -0.336101
- O -2.731718 -1.636385 -0.016496
- C -4.110477 -1.270365 -0.190982
- Н -4.179848 -0.220965 0.146725
- Н -4.716502 -1.877973 0.509946
- C -4.590796 -1.421936 -1.611187
- Н -5.626187 -1.044769 -1.689311
- Н -3.961926 -0.836744 -2.308318
- Н -4.598754 -2.476327 -1.942542
- C -2.443114 -3.042111 -0.131785
- Н -1.779415 -3.301584 0.710978
- Н -3.390382 -3.594432 0.014953
- C -1.793255 -3.387491 -1.447387
- Н -0.841934 -2.838791 -1.569328
- Н -1.562848 -4.467797 -1.476196
- Н -2.448125 -3.155016 -2.306285

TS-(1a·BF₃-IX)·2Et₂O

- C 4.422906 1.139177 -0.714511
- C 3.591722 0.106321 -1.119890
- C 2.946060 -0.693221 -0.159776

- C 3.155024 -0.429603 1.214869
- C 3.985765 0.637489 1.601163
- C 4.621867 1.413211 0.647637
- Н 4.933422 1.744753 -1.472078
- Н 3.442046 -0.103295 -2.181419
- C 2.431082 -1.239871 2.156530
- H 4.123615 0.836367 2.671114
- Н 5.278928 2.234754 0.953364
- C 2.069072 -2.542312 1.729195
- C 1.832177 -2.707407 0.386502
- Н 2.528561 -1.014092 3.224830
- Н 1.821326 -3.338144 2.437327
- N 2.087594 -1.731647 -0.524819
- B 1.073208 -1.531015 -1.715917
- F 0.267753 -0.383414 -1.329765
- F 1.693047 -1.238129 -2.900873
- F 0.265817 -2.633768 -1.810500
- Н 1.343094 -3.604808 -0.007861
- C 0.335668 -0.088999 1.969937
- Н 0.885971 0.849064 2.132942
- Н 0.657260 -0.688336 1.114443
- C -0.388188 -0.732308 3.121324
- Н -1.491106 -0.585061 3.123090
- Н -0.224395 -1.831767 3.144915
- Н -0.029494 -0.339647 4.093614
- Mg -1.058521 0.479339 0.242890
- Br -3.118432 1.651102 1.166593
- O -2.138006 -1.248571 -0.225242
- O -0.311019 2.220855 -0.628541

- C 1.071619 2.563656 -0.518374
- Н 1.592027 1.655769 -0.165029
- Н 1.470138 2.780783 -1.530255
- C -1.060639 2.974702 -1.588417
- Н -2.068832 3.106906 -1.155562
- Н -0.603641 3.978123 -1.679609
- C -1.108913 2.259201 -2.915724
- Н -1.622577 1.282879 -2.819654
- Н -1.658690 2.861937 -3.660623
- Н -0.092029 2.066953 -3.305863
- C 1.279915 3.709961 0.438945
- H 2.356438 3.942817 0.528889
- H 0.761218 4.626435 0.102206
- Н 0.894178 3.445936 1.441792
- C -1.946001 -2.501175 0.431260
- Н -1.713160 -3.275479 -0.324015
- Н -1.030316 -2.368462 1.036235
- C -3.114676 -2.885653 1.305435
- Н -2.845335 -3.757632 1.927788
- Н -3.389306 -2.050319 1.977160
- Н -4.006982 -3.165324 0.715718
- C -3.224401 -1.163190 -1.150495
- Н -4.180907 -1.267253 -0.600827
- Н -3.196901 -0.125083 -1.527685
- C -3.109684 -2.146088 -2.288369
- Н -3.871365 -1.914975 -3.053837
- Н -2.108839 -2.082095 -2.752372
- Н -3.275770 -3.189558 -1.963808

IX·Et₂O

С	3.127318	2.088594	1.183433
С	1.816014	2.036869	0.724927
С	1.447203	1.168593	-0.316440
С	2.425363	0.334172	-0.888386
С	3.741892	0.423858	-0.426367
С	4.106099	1.286340	0.601065
Η	3.382861	2.775830	1.997915
Η	1.071855	2.692284	1.185504
С	2.087804	-0.699080	-1.936836
Η	4.499573	-0.220346	-0.892747
Η	5.144731	1.330518	0.946837
С	0.773941	-0.369906	-2.573477
С	-0.120168	0.437782	-1.995513
Η	2.883611	-0.681312	-2.714078
Η	0.514503	-0.823536	-3.537305
N	0.081016	1.102874	-0.757607
В	-0.871519	2.313933	-0.473692
F	-1.306136	2.044133	0.940706
F	-0.284494	3.535685	-0.512381
F	-2.002728	2.239229	-1.235264
Η	-1.094606	0.633759	-2.454119
С	2.103491	-2.115239	-1.329624
Η	3.093587	-2.277239	-0.857834
Η	1.367513	-2.152781	-0.498344
С	1.833624	-3.217152	-2.337338
Η	1.943914	-4.215947	-1.877115
Н	0.810587	-3.159821	-2.755359

- Н 2.541554 -3.161031 -3.187659
- Mg -0.895800 0.078182 1.011830
- Br 0.407864 -1.441120 2.412734
- O -2.651440 -0.604200 0.165864
- C -2.899748 -1.842950 -0.522287
- Н -3.733170 -2.352694 0.001833
- Н -3.237029 -1.604596 -1.548506
- C -1.672136 -2.708796 -0.562173
- Н -1.910709 -3.648456 -1.090753
- Н -0.851299 -2.217053 -1.119952
- Н -1.312425 -2.967173 0.451273
- C -3.857818 0.130683 0.464596
- Н -4.435068 -0.461778 1.202495
- Н -3.524030 1.058598 0.958549
- C -4.683469 0.455837 -0.754053
- Н -5.479518 1.165239 -0.466913
- Н -4.060078 0.936399 -1.528046
- Н -5.176216 -0.433106 -1.185253

IIa-proR₃₋₄

- C -3.749576 1.480906 -0.386426
- C -4.241072 -0.192039 1.709189
- C -5.197256 1.019228 -0.204584
- Н -3.278605 0.756394 -1.078028
- C -5.101208 -0.349665 0.454455
- Н -4.876484 0.271147 2.493340
- Н -5.755788 1.722587 0.447941
- Н -5.729881 0.984607 -1.174201

- Н -6.094057 -0.779048 0.692575
- Н -4.614474 -1.042736 -0.262097
- C 0.697834 -2.199081 1.421722
- C 0.025830 -3.154808 -1.071093
- C 0.907973 -3.694595 1.140774
- Н -0.246160 -2.096531 1.996519
- C -0.135637 -4.106710 0.112343
- Н 0.992790 -3.374454 -1.568634
- Н 1.924323 -3.867596 0.731932
- Н 0.831099 -4.274936 2.079820
- Н -0.024051 -5.160725 -0.206060
- Н -1.141736 -4.006437 0.568380
- P 0.321555 -1.476069 -0.272308
- P -2.997324 1.224874 1.331217
- Cu 1.983879 -0.394065 -1.373658
- C 2.819187 2.388853 0.184252
- C 0.872965 2.310353 -1.122378
- C 3.472707 2.727000 1.387835
- C 3.568901 1.772715 -0.860171
- C 1.533225 1.644227 -2.178701
- Н -0.187172 2.580467 -1.213392
- C 4.819909 2.469208 1.532682
- Н 2.912255 3.187755 2.202340
- C 4.946516 1.519402 -0.677968
- C 2.904108 1.411318 -2.072653
- Н 0.985585 1.451218 -3.107295
- C 5.566072 1.867085 0.500593
- Н 5.314728 2.735456 2.473250
- Н 5.498398 1.032363 -1.489926

- Н 3.494637 1.067891 -2.929093
- Н 6.634364 1.671773 0.640944
- N 1.477516 2.679821 -0.010376
- B 0.603586 3.485849 1.142612
- F 0.603838 2.687809 2.260637
- F -0.654774 3.644347 0.620485
- F 1.240808 4.681580 1.347066
- Br 3.595442 -1.920635 -2.199005
- C -3.633913 -1.431958 2.305407
- C -3.835382 -2.708988 1.765386
- C -2.827125 -1.326942 3.450597
- C -3.227619 -3.833649 2.327746
- Н -4.481179 -2.839469 0.888652
- C -2.213561 -2.444888 4.010009
- Н -2.673699 -0.339448 3.909066
- C -2.404498 -3.707548 3.444658
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- Н -1.582839 -2.330291 4.899975
- Н -1.923229 -4.589725 3.882451
- C -3.514169 2.857080 -0.931073
- C -2.736722 3.030792 -2.082528
- C -4.030350 3.997513 -0.301442
- C -2.470767 4.303898 -2.588326
- Н -2.335014 2.144732 -2.596266
- C -3.773657 5.269346 -0.806299
- H -4.630494 3.887283 0.611611
- C -2.988763 5.428894 -1.950133
- Н -1.857579 4.415817 -3.490646
- Н -4.184425 6.149119 -0.296972

- Н -2.782800 6.431372 -2.342729
- C 1.798608 -1.492354 2.160400
- C 1.518778 -0.765736 3.322947
- C 3.124357 -1.537148 1.703404
- C 2.532814 -0.110902 4.022470
- Н 0.485830 -0.719944 3.694388
- C 4.139012 -0.888893 2.403519
- Н 3.371136 -2.081485 0.780990
- C 3.847773 -0.175638 3.567528
- Н 2.290094 0.450393 4.932201
- Н 5.169753 -0.940362 2.032191
- H 4.648275 0.333910 4.117174
- C -1.052764 -3.061426 -2.109395
- C -2.376440 -3.443462 -1.864201
- C -0.747517 -2.470388 -3.345176
- C -3.372107 -3.215895 -2.815629
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- C -1.739602 -2.240170 -4.294424
- Н 0.292846 -2.183703 -3.556337
- C -3.060638 -2.605928 -4.029273
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- Н -1.479570 -1.774845 -5.252398
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С	3.256710	2.111440	0.009200
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Η	6.000645	0.416140	-1.142330
Η	4.734753	-0.097828	-0.001262
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Н	1.114972	-5.054759	0.924439
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- C -3.960477 0.978909 0.994108
- C -1.824151 1.386811 2.101531
- Н -0.433420 2.535270 0.893840
- C -5.547825 1.211880 -1.295272
- Н -3.866861 2.180853 -2.217103
- C -5.283312 0.497078 0.986646
- C -3.116841 0.813269 2.155952
- Н -1.174615 1.428156 2.984166
- C -6.073968 0.612837 -0.140480
- Н -6.165953 1.296826 -2.196234
- Н -5.669541 0.023874 1.897850
- Н -3.585921 0.553610 3.112647
- Н -7.102740 0.236762 -0.134441
- N -2.162647 2.165569 -0.152305
- B -1.557900 2.965012 -1.433832
- F -1.452015 2.064206 -2.473371
- F -0.320843 3.441268 -1.052870
- F -2.426708 3.990289 -1.732293
- C -3.107161 -2.165313 2.436440
- Н -4.131643 -2.010612 2.034890
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- C -3.106389 -2.187599 3.953771
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- Н -2.092925 -2.374873 4.364328
- C 3.619489 -1.013263 -2.358180
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- C 2.739716 -1.238533 -3.430427
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- Н 4.835144 -2.005763 -0.859695
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- Н 2.350487 -0.377574 -3.992731
- C 2.836713 -3.629824 -3.088316
- Н 4.149367 -4.284218 -1.499123
- Н 1.651041 -2.668178 -4.624590
- Н 2.529611 -4.644449 -3.367394
- C 2.690501 3.405587 0.506940
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- C 2.859518 4.607781 -0.192495
- C 1.347862 4.603395 2.147375
- Н 1.799152 2.490177 2.251701
- C 2.285253 5.788631 0.270274
- Н 3.437758 4.614495 -1.126022
- C 1.521893 5.790961 1.439706
- H 0.753815 4.592731 3.069440
- Н 2.426941 6.718773 -0.292656
- Н 1.064343 6.720317 1.798123
- C -1.658343 -2.237866 -1.688067
- C -1.631968 -1.582271 -2.923800
- C -2.905679 -2.524274 -1.113920
- C -2.813959 -1.223739 -3.572480
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- C -4.087359 -2.168217 -1.758938
- Н -2.958662 -3.032770 -0.143142
- C -4.046045 -1.515357 -2.991795
- Н -2.767443 -0.709228 -4.539272
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- Н -4.976743 -1.230966 -3.497006

- C 1.766531 -2.507523 2.423128
- C 3.120325 -2.668749 2.110113
- C 1.431292 -1.746746 3.554148
- C 4.109410 -2.063968 2.887802
- Н 3.418077 -3.265608 1.239342
- C 2.415121 -1.139718 4.329451
- Н 0.370709 -1.619717 3.817216
- C 3.762085 -1.290051 3.993175
- Н 5.164563 -2.200943 2.622505
- Н 2.129528 -0.545450 5.205196
- Н 4.539458 -0.811298 4.599356
- C 1.391478 -0.536264 0.139111
- Н 2.242937 -1.208672 -0.086732
- H 1.646568 0.012730 1.067481
- C 1.113006 0.409950 -1.011344
- Н 0.776756 -0.141332 -1.911569
- Н 0.277791 1.094674 -0.774521

Х

С	1.700844	1.723807	2.314339
С	-0.715338	0.699387	3.136508
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Η	1.537470	2.820439	2.229154
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Η	-0.558182	-0.380349	3.338040
Η	1.867858	0.284165	3.928662
Η	2.211821	1.959044	4.435107
Н	-0.128441	1.187305	5.185855

- Н -0.163070 2.584230 4.083784
- C -3.171796 0.166474 -1.957749
- C -1.012507 1.525908 -2.966324
- C -3.172370 0.505861 -3.453728
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- Н -0.482682 0.763360 -3.574380
- Н -2.698726 -0.315347 -4.029089
- Н -4.208074 0.602502 -3.830390
- Н -2.244554 2.070013 -4.688204
- Н -2.897660 2.612929 -3.124634
- P -1.430528 0.604549 -1.353171
- P 0.190043 0.937242 1.491023
- Cu -0.002401 -0.756756 -0.090817
- Br -0.888203 -2.761882 1.019975
- C 2.783203 -1.996392 -0.467674
- C 1.725734 -1.400850 -2.433111
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- C 2.804327 -1.843581 -3.227002
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- C 3.900199 -2.383454 -2.598100
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- C 3.813564 0.430327 -1.234561
- C 3.804745 -0.219158 0.005865
- C 4.877338 1.292875 -1.519288
- C 4.816555 -0.014597 0.937159
- Н 2.967975 -0.884450 0.257819
- C 5.897207 1.500274 -0.588559
- Н 4.924372 1.814228 -2.482207
- C 5.871879 0.849347 0.642619

- Н 4.770626 -0.522253 1.908953
- Н 6.719836 2.182914 -0.832299
- Н 6.669578 1.019363 1.374924
- C -3.945905 -0.356475 0.902263
- C -5.202127 -0.305648 0.287799
- C -3.541639 0.730405 1.686844
- C -6.030767 0.805043 0.440637
- Н -5.531213 -1.151739 -0.330770
- C -4.361033 1.848286 1.832346
- Н -2.557559 0.715371 2.175252
- C -5.609052 1.889914 1.209065
- Н -7.010359 0.826458 -0.051475
- H -4.018182 2.695113 2.438862
- Н -6.253755 2.768999 1.323705
- C 0.470951 -3.468939 1.165819
- C 0.435925 -4.619110 0.367544
- C 1.722893 -2.940227 1.513996
- C 1.615749 -5.214692 -0.079509
- Н -0.523863 -5.069848 0.088531
- C 2.901874 -3.535308 1.070892
- Н 1.767502 -2.038022 2.141494
- C 2.852865 -4.674254 0.266400
- Н 1.564152 -6.115505 -0.702065
- Н 3.869573 -3.105131 1.355889
- Н 3.779206 -5.144469 -0.082521
- C 0.388816 -1.549129 -1.797760
- Н 0.415230 -1.760109 -2.885770
- Н 1.160059 -2.200548 -1.334080
- C -0.988796 -1.881626 -1.228807

- Н -1.188193 -2.969072 -1.306994
- Н -1.780063 -1.371798 -1.815026
- B 1.943801 2.080347 2.318719
- F 2.121718 2.614156 3.537245
- F 1.824604 0.720142 2.305766
- F 2.682493 2.600532 1.329044

$Cu_2Br_2 \cdot (SMe_2)_2$

Cu	-1.186680	-0.074589	-0.272952
Br	0.396508	-1.889383	-0.118612
Br	-0.441304	2.229344	-0.032088
Cu	1.199109	0.477401	0.033810
S	-3.428659	-0.500968	-0.423254
S	3.492204	0.497582	0.193635
С	-4.073822	0.768413	0.700223
Н	-5.145608	0.593623	0.898316
Н	-3.502313	0.773412	1.644832
Н	-3.941985	1.739061	0.193751
С	-3.542418	-1.963018	0.644505
Н	-3.012562	-1.786514	1.596920
Н	-4.599988	-2.217280	0.832843
Н	-3.050151	-2.789846	0.105903
С	3.905610	-0.548550	-1.230363
Н	3.751167	0.056302	-2.139242
Н	4.964539	-0.855035	-1.169768
Н	3.243817	-1.432627	-1.255880
С	3.734396	-0.717601	1.518986
Н	4.795262	-1.021725	1.552747

- Н 3.464243 -0.227428 2.468966
- Н 3.081697 -1.593412 1.354043

$L1_2Cu_2Br_2$

- Cu 1.616502 -0.331108 -0.047830
- C 3.826743 1.885101 0.938480
- H 4.808780 2.022463 1.432269
- Н 3.675057 2.771856 0.291154
- C 2.708330 1.793608 1.972366
- Н 2.832593 0.899739 2.615354
- Н 2.707713 2.682934 2.634576
- C 5.225072 -0.612414 0.603415
- C 4.844847 0.847027 -1.697427
- C 6.486056 0.041215 0.012293
- Н 5.180415 -0.453471 1.699151
- C 6.240288 0.237622 -1.479538
- Н 4.328441 0.327638 -2.528446
- Н 6.663701 1.017713 0.506278
- Н 7.382257 -0.581051 0.204229
- Н 7.030487 0.848867 -1.954037
- Н 6.277255 -0.750314 -1.976708
- C 0.772096 3.426097 0.603430
- C -0.251378 1.628373 2.489320
- C -0.220222 3.958002 1.654510
- Н 1.746149 3.941740 0.731178
- C -1.128708 2.808272 2.071947
- Н -0.785050 0.671921 2.321616

- H 0.347330 4.333341 2.526959
- Н -0.810878 4.806833 1.262019
- Н -1.839248 3.094848 2.871482
- Н -1.746925 2.508203 1.199358
- P 3.828399 0.372571 -0.157293
- P 1.060542 1.624844 1.123148
- C 0.336717 3.682702 -0.826400
- C 0.136498 2.684162 -1.780757
- C 0.170382 5.016291 -1.231914
- C -0.225839 3.000268 -3.092505
- Н 0.283576 1.631999 -1.512840
- C -0.203046 5.338426 -2.533006
- Н 0.344497 5.824818 -0.508640
- C -0.406713 4.326929 -3.474813
- Н -0.350999 2.189261 -3.822429
- Н -0.326505 6.389986 -2.819317
- Н -0.692350 4.576030 -4.503693
- C 0.238092 1.628945 3.917851
- C 0.789054 0.450077 4.444280
- C 0.128943 2.736117 4.769957
- C 1.225048 0.382903 5.764077
- Н 0.889874 -0.430713 3.791986
- C 0.566030 2.670762 6.094186
- Н -0.318325 3.671325 4.416279
- C 1.118318 1.496240 6.598137
- Н 1.650328 -0.552977 6.145533
- Н 0.465446 3.552426 6.738308
- Н 1.459443 1.446085 7.638702
- C 4.751753 2.321869 -1.956331

- C 3.578260 2.825210 -2.534903
- C 5.738192 3.229613 -1.554697
- C 3.385338 4.195100 -2.692063
- Н 2.790107 2.124180 -2.847932
- C 5.550002 4.602261 -1.715779
- Н 6.665972 2.867923 -1.093635
- C 4.371500 5.091417 -2.277889
- Н 2.451760 4.562622 -3.136481
- Н 6.333920 5.297503 -1.392538
- Н 4.223603 6.170903 -2.398307
- C 5.169324 -2.094895 0.330500
- C 4.569471 -2.645761 -0.807521
- C 5.792214 -2.963411 1.235523
- C 4.600506 -4.020884 -1.035931
- Н 4.038309 -2.001854 -1.522517
- C 5.817045 -4.338492 1.017026
- Н 6.262735 -2.546282 2.136375
- C 5.222695 -4.873161 -0.125848
- Н 4.127658 -4.427486 -1.937152
- Н 6.303785 -4.997489 1.745741
- Н 5.241193 -5.954770 -0.305352
- Br 0.680908 -1.001235 -2.270413
- Br 1.618640 -1.991756 1.767302
- Cu -0.887546 -0.551406 -0.474930
- P -2.112034 -1.929539 0.896386
- P -2.960646 0.339206 -1.423651
- C -3.866702 -1.465958 0.485112
- C -2.176419 -3.797351 0.575135
- C -2.028815 -2.064039 2.784268

- $C \quad -3.958966 \ -1.184128 \ -1.010215$
- C -4.352174 1.624827 -1.196966
- C -2.969237 0.414471 -3.324716
- Н -4.567226 -2.267444 0.792329
- Н -4.138512 -0.565264 1.073708
- C -2.357085 -4.419112 1.967549
- Н -3.099800 -3.941122 -0.018511
- C -1.033976 -4.309876 -0.263311
- C -1.684790 -3.541141 3.021148
- Н -1.158681 -1.442456 3.074733
- C -3.256576 -1.520259 3.466560
- Н -3.557454 -2.033522 -1.598925
- Н -5.010244 -1.051427 -1.333739
- C -4.867360 1.892209 -2.624121
- Н -5.127733 1.081012 -0.626846
- C -4.031098 2.858092 -0.396671
- C -3.745322 1.698534 -3.641127
- Н -1.902366 0.539981 -3.589485
- C -3.449471 -0.851635 -3.985899
- Н -3.441003 -4.501102 2.176735
- Н -1.960900 -5.452205 1.992158
- C 0.162031 -4.790675 0.281236
- $C \qquad -1.168844 \quad -4.302091 \quad -1.658292 \\$
- Н -1.954719 -3.851263 4.048716
- Н -0.585621 -3.638650 2.936192
- C -3.374099 -0.136854 3.670774
- C -4.319555 -2.333009 3.878342
- Н -5.690064 1.183507 -2.835764
- Н -5.307612 2.905235 -2.695339

- C -4.530348 2.974207 0.909097
- C -3.318514 3.941042 -0.928235
- Н -4.133263 1.682810 -4.677665
- Н -3.041143 2.552943 -3.590208
- C -2.549365 -1.913365 -4.167594
- C -4.770732 -1.031232 -4.416014
- C 1.190853 -5.245974 -0.547483
- Н 0.307695 -4.807478 1.367568
- C -0.143791 -4.746771 -2.486143
- Н -2.103436 -3.935351 -2.108558
- C -4.516211 0.416971 4.243121
- Н -2.550143 0.522501 3.368000
- C -5.465122 -1.782686 4.454076
- Н -4.262651 -3.420192 3.752997
- C -4.351450 4.140895 1.650278
- Н -5.090407 2.135359 1.346607
- C -3.144667 5.110825 -0.191394
- Н -2.894285 3.887702 -1.939834
- C -2.960300 -3.113018 -4.745232
- Н -1.506084 -1.794856 -3.838330
- C -5.183445 -2.231867 -4.993785
- Н -5.502667 -0.222683 -4.310152
- C 1.043264 -5.223305 -1.932192
- Н 2.129226 -5.601605 -0.100700
- Н -0.269565 -4.722409 -3.575322
- C -5.572923 -0.405532 4.633993
- Н -4.575723 1.501709 4.391406
- Н -6.282645 -2.443153 4.766197
- C -3.665170 5.220348 1.097374

- Н -4.765646 4.213850 2.663084
- Н -2.593141 5.947909 -0.636457
- C -4.282192 -3.281116 -5.157150
- Н -2.237932 -3.927038 -4.883015
- Н -6.223511 -2.343237 -5.322313
- Н 1.854000 -5.576297 -2.580244
- Н -6.474018 0.025735 5.084827
- Н -3.533215 6.145386 1.670798
- Н -4.605539 -4.224375 -5.612094

L1₂Cu₂BrEt

Cu	-1.423738	0.448208	-0.002935
С	-3.992458	-1.523900	0.825392
Η	-4.953108	-1.518543	1.376918
Η	-4.081269	-2.321725	0.061260
С	-2.835543	-1.845231	1.767667
Η	-2.786842	-1.117497	2.599983
Η	-2.957569	-2.857426	2.204100
С	-4.892757	1.239871	0.826681
С	-4.969676	-0.067770	-1.583480
С	-6.296633	0.845969	0.341245
Η	-4.766763	1.023780	1.904598
С	-6.238609	0.704730	-1.175805
Η	-4.458029	0.467595	-2.407042
Η	-6.593562	-0.114768	0.807936
Η	-7.049798	1.595615	0.654794
Н	-7.151708	0.229327	-1.579559
Η	-6.199331	1.714610	-1.625827

- C -1.210517 -3.343756 -0.091863
- C 0.136695 -2.221904 2.083905
- C -0.249623 -4.239560 0.712763
- Н -2.237677 -3.757337 -0.016564
- C 0.828358 -3.346496 1.314990
- Н 0.782511 -1.324386 2.117002
- Н -0.819619 -4.744329 1.515783
- Н 0.192143 -5.033163 0.080872
- Н 1.540992 -3.905674 1.952555
- Н 1.429807 -2.904506 0.491163
- P -3.752721 0.084949 -0.107225
- P -1.219383 -1.708123 0.865417
- C -0.875736 -3.296981 -1.571504
- C -0.526563 -2.145041 -2.281717
- C -0.950689 -4.505895 -2.282592
- C -0.255491 -2.203897 -3.652371
- Н -0.466948 -1.176338 -1.763177
- C -0.671264 -4.571475 -3.643626
- Н -1.246398 -5.421341 -1.751380
- C -0.318708 -3.412963 -4.339886
- Н 0.001373 -1.281857 -4.191491
- Н -0.739902 -5.531713 -4.169151
- Н -0.104603 -3.453326 -5.414356
- C -0.272728 -2.536575 3.502340
- C -0.627081 -1.480525 4.357007
- C -0.271950 -3.833334 4.034002
- C -0.981134 -1.711991 5.683438
- Н -0.644015 -0.455006 3.958155
- C -0.627396 -4.067719 5.363572

- Н 0.024365 -4.689494 3.418418
- C -0.986833 -3.010152 6.195075
- Н -1.253981 -0.865670 6.325124
- Н -0.615641 -5.093511 5.751091
- Н -1.264616 -3.195142 7.239332
- C -5.134233 -1.493307 -2.020397
- C -4.140854 -2.073462 -2.821489
- C -6.197442 -2.299789 -1.595581
- C -4.197888 -3.418655 -3.178457
- Н -3.297472 -1.456605 -3.164354
- C -6.258841 -3.646525 -1.952460
- Н -6.987390 -1.879018 -0.960319
- C -5.257520 -4.213816 -2.740522
- Н -3.403519 -3.845372 -3.802910
- Н -7.099164 -4.260251 -1.606692
- Н -5.306880 -5.272962 -3.018725
- C -4.610530 2.702880 0.589098
- C -4.117641 3.197343 -0.625051
- C -4.928820 3.627796 1.592145
- C -3.972064 4.567859 -0.839370
- Н -3.826718 2.506988 -1.427755
- C -4.779477 4.997595 1.386715
- Н -5.302337 3.258695 2.556842
- C -4.307259 5.475163 0.163329
- Н -3.586814 4.925006 -1.802254
- Н -5.034923 5.698963 2.189940
- Н -4.190040 6.552914 -0.003150
- C -0.585509 1.768014 -1.441046
- Br -1.580999 1.524922 2.407116

- Cu 0.915352 0.661781 -0.466508
- P 2.365176 1.492565 1.154511
- P 2.823790 -0.111464 -1.708585
- C 4.044581 1.038092 0.495291
- C 2.604681 3.363202 1.361641
- C 2.365325 1.097926 3.006190
- C 3.999744 1.173692 -1.025694
- C 4.094421 -1.526355 -1.829895
- C 2.770481 0.217935 -3.583290
- Н 4.833704 1.676295 0.940005
- Н 4.276326 -0.005093 0.794256
- C 2.827678 3.564120 2.871414
- Н 3.540002 3.594286 0.817504
- C 1.509940 4.172496 0.720474
- C 2.127031 2.462233 3.664189
- Н 1.464639 0.471636 3.157286
- C 3.569894 0.305016 3.433353
- Н 3.638746 2.179211 -1.325260
- Н 5.005685 1.053219 -1.473639
- C 4.596689 -1.474258 -3.285221
- Н 4.913531 -1.206789 -1.159959
- C 3.659217 -2.881127 -1.344588
- C 3.488885 -0.982415 -4.211205
- Н 1.695041 0.180372 -3.839157
- C 3.266587 1.591973 -3.939013
- H 3.916987 3.532987 3.068592
- Н 2.482915 4.567377 3.187821
- C 0.262277 4.352073 1.327797
- C 1.719837 4.744808 -0.541491

- Н 2.445152 2.453927 4.724395
- Н 1.033543 2.639005 3.665568
- C 3.603954 -1.074181 3.172021
- C 4.686818 0.881813 4.049202
- Н 5.459620 -0.782569 -3.332839
- Н 4.976814 -2.464739 -3.601140
- C 4.199318 -3.377843 -0.150038
- C 2.784956 -3.704431 -2.066712
- Н 3.877804 -0.732193 -5.216843
- Н 2.748135 -1.791957 -4.363059
- C 2.382524 2.677264 -3.845181
- C 4.590197 1.850822 -4.316543
- C -0.750209 5.063386 0.688183
- Н 0.046864 3.889757 2.297411
- C 0.716687 5.476319 -1.176183
- Н 2.694093 4.618975 -1.037218
- C 4.714972 -1.846432 3.498553
- Н 2.740217 -1.551507 2.688346
- C 5.802171 0.110526 4.380153
- H 4.695655 1.952550 4.284380
- C 3.899987 -4.662661 0.299615
- Н 4.885095 -2.746261 0.433232
- C 2.489506 -4.992549 -1.623246
- Н 2.320013 -3.342747 -2.993358
- C 2.810726 3.978548 -4.096887
- Н 1.335015 2.494465 -3.567045
- C 5.021072 3.152792 -4.570535
- Н 5.308236 1.028053 -4.415947
- C -0.525868 5.635480 -0.563494

- Н -1.734982 5.145207 1.167695
- Н 0.909109 5.926703 -2.158037
- C 5.824869 -1.254845 4.103635
- Н 4.709240 -2.922159 3.283365
- Н 6.663282 0.587130 4.863241
- C 3.049742 -5.480206 -0.442631
- Н 4.344621 -5.030816 1.232298
- Н 1.807521 -5.618878 -2.211363
- C 4.136532 4.223922 -4.454833
- Н 2.097800 4.808333 -4.020103
- Н 6.062630 3.329512 -4.863550
- Н -1.321361 6.202591 -1.062578
- Н 6.701633 -1.858496 4.364472
- Н 2.818349 -6.495465 -0.099270
- Н 4.476675 5.246580 -4.653879
- Н 0.372546 2.347370 -1.549285
- C -1.103298 1.444694 -2.838180
- Н -1.231141 2.535728 -0.957912
- Н -1.271776 2.347077 -3.468641
- Н -0.398978 0.796114 -3.398738
- Н -2.063399 0.887846 -2.813165

$L1_2Cu_2Et_2\\$

- Cu -1.173250 0.176221 0.276355
- C -4.431310 1.075991 0.086349
- Н -5.207352 1.764263 0.475976
- Н -4.707964 0.858856 -0.963700
- C -4.395100 -0.218067 0.889781

- Н -4.173039 -0.012778 1.958062
- Н -5.379619 -0.725136 0.862164
- C -3.022824 3.192494 1.410899
- C -2.883264 3.180036 -1.346761
- C -3.747822 4.341098 0.690364
- Н -3.685491 2.737907 2.174449
- C -3.066044 4.540557 -0.658171
- Н -1.891693 3.137558 -1.836254
- Н -4.813607 4.071873 0.547525
- Н -3.732459 5.265231 1.301224
- Н -3.616776 5.252906 -1.301012
- Н -2.067302 4.989893 -0.490807
- C -3.847940 -2.122740 -1.226969
- C -3.160970 -2.881822 1.325509
- C -4.692335 -3.259812 -0.632334
- Н -4.502667 -1.372401 -1.712904
- C -3.877084 -3.937921 0.464609
- Н -2.106995 -3.184136 1.475281
- Н -5.623153 -2.831401 -0.209209
- Н -5.003490 -3.979724 -1.415226
- Н -4.503661 -4.607874 1.083711
- Н -3.111940 -4.589572 -0.000233
- P -2.751425 1.899939 0.071226
- P -3.026373 -1.312403 0.255872
- C -2.823551 -2.583199 -2.232602
- C -1.627783 -3.211816 -1.854204
- C -3.061462 -2.392926 -3.599707
- C -0.719044 -3.653465 -2.812563
- Н -1.369955 -3.346086 -0.793509

- C -2.142203 -2.812783 -4.560714
- Н -3.990742 -1.898802 -3.914416
- C -0.966712 -3.452954 -4.170173
- Н 0.198424 -4.157499 -2.483043
- Н -2.349994 -2.642681 -5.623833
- Н -0.242974 -3.790986 -4.921122
- C -3.734501 -2.581121 2.679596
- C -2.897146 -2.034600 3.663501
- C -5.089050 -2.756102 2.987802
- C -3.397392 -1.659354 4.908044
- Н -1.832889 -1.884528 3.431917
- C -5.591937 -2.387592 4.235663
- Н -5.773408 -3.178151 2.240570
- C -4.750838 -1.833642 5.200073
- Н -2.721618 -1.227334 5.656289
- Н -6.656566 -2.532909 4.454524
- Н -5.148734 -1.540910 6.178624
- C -3.912690 2.768490 -2.361003
- C -3.609844 1.724037 -3.248908
- C -5.195999 3.324807 -2.411447
- C -4.559125 1.246373 -4.148645
- Н -2.610587 1.268046 -3.218051
- C -6.149293 2.847227 -3.312084
- Н -5.471137 4.141459 -1.732847
- C -5.838372 1.804226 -4.182232
- Н -4.293837 0.433662 -4.835922
- Н -7.148623 3.298145 -3.330485
- Н -6.588185 1.430221 -4.888826
- C -1.755719 3.646348 2.086767

- C -0.523648 3.690605 1.426668
- C -1.808895 4.078278 3.419051
- C 0.617413 4.156459 2.077465
- Н -0.426222 3.321896 0.394862
- C -0.668637 4.535412 4.076313
- Н -2.768975 4.045914 3.952247
- C 0.553480 4.578002 3.403936
- Н 1.565124 4.185240 1.526920
- Н -0.734671 4.858941 5.121952
- Н 1.455917 4.936575 3.914299
- Cu 1.110323 0.322553 -0.448956
- P 3.109136 1.258093 0.226655
- P 2.543906 -2.051188 -0.417272
- C 4.472462 0.082383 -0.255498
- C 3.895168 2.846597 -0.446813
- C 3.496420 1.720527 2.047248
- C 4.220980 -1.378928 0.076879
- C 2.821720 -3.807579 0.269585
- C 2.815420 -2.625745 -2.227360
- Н 4.567367 0.215731 -1.350339
- H 5.436120 0.398432 0.186861
- C 4.945032 3.191429 0.608315
- Н 4.378923 2.548543 -1.400451
- C 2.893053 3.937537 -0.761883
- C 4.294333 3.037908 1.980714
- Н 2.500265 1.901668 2.497679
- C 4.149414 0.595241 2.804114
- Н 5.039999 -1.994793 -0.343929
- Н 4.251413 -1.521726 1.175289

- C 3.655654 -4.523596 -0.793097
- Н 3.420758 -3.663713 1.190581
- C 1.528251 -4.461215 0.683597
- C 3.164523 -4.123942 -2.182396
- Н 1.819715 -2.491332 -2.692075
- C 3.766432 -1.726584 -2.962211
- Н 5.800872 2.497479 0.508896
- Н 5.372580 4.201137 0.466130
- C 3.043331 5.260950 -0.328422
- C 1.786862 3.637610 -1.570085
- Н 5.039891 3.093804 2.795807
- Н 3.600480 3.884479 2.148884
- C 3.357864 -0.351745 3.471074
- C 5.541116 0.431812 2.846777
- Н 4.713603 -4.220348 -0.671701
- Н 3.640069 -5.623033 -0.655314
- C 0.909382 -4.026988 1.867606
- C 0.919893 -5.501482 -0.028062
- Н 3.906130 -4.386207 -2.960625
- Н 2.257484 -4.705105 -2.437546
- C 3.269752 -0.556047 -3.553842
- C 5.142892 -1.972920 -3.043601
- C 2.115432 6.243913 -0.676386
- Н 3.894455 5.548392 0.298767
- C 0.863900 4.616886 -1.926954
- Н 1.643346 2.604646 -1.911387
- C 3.930872 -1.441483 4.123661
- Н 2.267754 -0.228163 3.475455
- C 6.118556 -0.656770 3.500029

- H 6.197202 1.165743 2.361902
- C -0.256374 -4.624697 2.334284
- Н 1.364465 -3.202070 2.435041
- C -0.264476 -6.088956 0.424984
- Н 1.376839 -5.884957 -0.947666
- C 4.117411 0.356272 -4.176621
- Н 2.190574 -0.365116 -3.510751
- C 5.995930 -1.064350 -3.671793
- Н 5.565739 -2.887563 -2.607650
- C 1.019708 5.928334 -1.475971
- Н 2.256524 7.269761 -0.316147
- Н 0.015883 4.351880 -2.572561
- C 5.316461 -1.604978 4.133631
- Н 3.286967 -2.165521 4.637382
- Н 7.209821 -0.762068 3.514159
- C -0.854109 -5.660051 1.610697
- Н -0.708604 -4.277468 3.272200
- Н -0.720082 -6.901013 -0.154419
- C 5.490375 0.108150 -4.232983
- Н 3.699821 1.265598 -4.627283
- Н 7.070415 -1.277254 -3.722190
- Н 0.291660 6.700472 -1.751026
- Н 5.770713 -2.460956 4.645861
- Н -1.778472 -6.126356 1.973164
- Н 6.163265 0.820710 -4.723976
- C -0.110396 0.098707 -2.072875
- Н 0.240593 -0.884887 -2.460233
- Н -1.200074 -0.093428 -1.911150
- C -0.012574 1.136665 -3.186800

- Н -0.424553 2.119147 -2.876459
- Н 1.032418 1.343868 -3.498824
- Н -0.556567 0.852927 -4.118849
- C 0.031010 -0.328475 1.848386
- Н -0.289335 -1.383890 2.008590
- Н 1.135510 -0.412954 1.756666
- C -0.289162 0.509850 3.082540
- Н 0.123666 1.536331 2.992240
- Н -1.383159 0.643410 3.230459
- Н 0.101356 0.090458 4.043508

L1-CuEt₂

Cu	-1.643595	0.812514	-2.210233
С	0.192403	-0.343847	1.721937
Η	-0.202872	-0.263555	2.754727
Η	0.974197	-1.127537	1.765997
С	0.812466	0.987736	1.310996
Η	0.041601	1.784367	1.275835
Η	1.550860	1.301170	2.076587
С	-2.616422	-0.205982	1.420700
С	-1.516794	-2.708810	1.250322

- C -2.865194 -1.119103 2.630909
- Н -2.319481 0.807626 1.758984
- C -2.757600 -2.557800 2.143325
- Н -1.778158 -3.302557 0.352651
- Н -2.107873 -0.918316 3.414459
- Н -3.853227 -0.914185 3.090680
- Н -2.756248 -3.282652 2.980499

- Н -3.656126 -2.790357 1.538745
- C 3.103120 -0.163331 0.029125
- C 2.667585 2.544277 -0.382257
- C 4.181122 0.790188 0.535661
- Н 2.779619 -0.840427 0.845434
- C 4.120045 2.059543 -0.306120
- Н 2.480245 2.994417 -1.375503
- Н 3.964988 1.031260 1.594754
- Н 5.190633 0.334278 0.530397
- H 4.795461 2.850961 0.074032
- Н 4.465903 1.826135 -1.332394
- P -1.124746 -0.952935 0.531317
- P 1.599175 0.938804 -0.390379
- C 3.449078 -1.038673 -1.152449
- C 2.462218 -1.905891 -1.651105
- C 4.702965 -1.054950 -1.771737
- C 2.719896 -2.764790 -2.713890
- Н 1.462353 -1.895438 -1.191376
- C 4.964975 -1.910636 -2.845395
- Н 5.503634 -0.394875 -1.417402
- C 3.978765 -2.770748 -3.319924
- Н 1.926818 -3.429421 -3.078847
- Н 5.956798 -1.900313 -3.313476
- Н 4.185472 -3.440731 -4.162716
- C 2.221372 3.544589 0.646114
- C 1.043355 4.270680 0.409603
- C 2.895565 3.768138 1.852872
- C 0.547504 5.172360 1.345897
- Н 0.495596 4.107000 -0.528760

- C 2.404587 4.676612 2.792876
- Н 3.823909 3.227764 2.075327
- C 1.227303 5.380181 2.547785
- Н -0.378895 5.720185 1.133196
- Н 2.952999 4.833337 3.729714
- H 0.841581 6.090724 3.288226
- C -0.288228 -3.320603 1.848091
- C 0.690789 -3.847653 0.990634
- C -0.017408 -3.307185 3.222287
- C 1.908970 -4.311199 1.480428
- Н 0.489998 -3.878804 -0.090291
- C 1.196853 -3.781885 3.718510
- Н -0.758996 -2.903344 3.923301
- C 2.170449 -4.276731 2.851444
- H 2.659938 -4.704723 0.784210
- Н 1.386341 -3.757470 4.798501
- Н 3.127452 -4.641115 3.242666
- C -3.848050 -0.068666 0.559300
- C -4.863023 0.805576 0.970660
- C -4.039628 -0.790068 -0.622904
- C -6.027197 0.965888 0.223077
- Н -4.724621 1.383912 1.895154
- C -5.204546 -0.636291 -1.375578
- Н -3.257620 -1.470753 -0.984807
- C -6.200957 0.244124 -0.959328
- Н -6.802200 1.664449 0.561302
- Н -5.325997 -1.205564 -2.305529
- Н -7.112284 0.370820 -1.555877
- C -1.120958 -0.655239 -3.429256

- Н -0.003390 -0.676375 -3.420643
- Н -1.381480 -0.422846 -4.489771
- C -1.623184 -2.060857 -3.113957
- Н -1.229001 -2.868249 -3.779775
- Н -1.361419 -2.363678 -2.075158
- Н -2.731145 -2.132119 -3.180442
- C -2.082893 2.441466 -1.184568
- $C \quad -1.766837 \quad 3.730669 \quad -1.940354$
- Н -3.159842 2.443459 -0.891832
- Н -1.529852 2.452566 -0.212677
- Н -1.930671 4.679181 -1.367740
- Н -0.709132 3.759768 -2.283773
- Н -2.372838 3.824571 -2.866304

CuEt₂

Cu	0.000002	0.000008	-0.075759
С	-1.842341	-0.677331	-0.070996
Η	-1.957512	-1.479895	0.695599
Η	-2.074399	-1.185931	-1.036607
С	-2.904618	0.392549	0.182891
Η	-3.958189	0.020250	0.188647
Η	-2.755369	0.900347	1.159110
Η	-2.870591	1.199000	-0.579629
С	1.842350	0.677336	-0.070986
С	2.904600	-0.392573	0.182889
Η	2.074412	1.185902	-1.036614
Η	1.957560	1.479914	0.695588
Η	3.958183	-0.020307	0.188603

- Н 2.755366 -0.900344 1.159126
- Н 2.870523 -1.199041 -0.579610

V-R₃₋₄

- C 1.934007 -0.372031 2.794257
 C -0.410780 -1.762060 2.583423
 C 1.815779 -1.702387 3.550786
 H 1.756794 0.437052 3.535871
- C 0.349448 -1.906062 3.893558
- Н -0.115138 -2.611323 1.931211
- Н 2.178570 -2.527745 2.902802
- Н 2.444983 -1.701333 4.459373
- Н 0.156406 -2.888099 4.366265
- Н 0.035587 -1.130248 4.620847
- C -3.696289 1.746622 -0.351808
- C -1.375047 3.154846 -0.691635
- C -3.760234 3.071439 -1.133377
- Н -4.148217 1.907247 0.648215
- C -2.657456 3.976942 -0.614216
- Н -1.170102 2.948440 -1.763324
- Н -3.595189 2.865051 -2.210760
- Н -4.767168 3.520622 -1.044445
- Н -2.567490 4.913369 -1.197539
- Н -2.880520 4.268259 0.431854
- P -1.856875 1.461990 -0.022990
- P 0.381950 -0.290118 1.713241
- Cu -0.388556 -0.295731 -0.481338
- C 2.493824 -1.469210 -1.987716

- C 0.477240 -0.468294 -2.687165
- C 3.859733 -1.341370 -1.699703
- C 1.786105 -2.593150 -1.492860
- C -0.289082 -1.504242 -2.175981
- Н -0.000880 0.419226 -3.124046
- C 4.507112 -2.260168 -0.881636
- Н 4.417136 -0.493497 -2.103109
- C 2.447377 -3.471156 -0.630598
- C 0.377639 -2.843778 -1.997668
- Н -1.360460 -1.501831 -2.432178
- C 3.795882 -3.311878 -0.311882
- Н 5.573677 -2.127124 -0.664664
- Н 1.904720 -4.324938 -0.209831
- Н 0.480240 -3.298959 -3.016120
- H 4.289862 -4.023221 0.360148
- N 1.823628 -0.455583 -2.704158
- B 2.574025 0.809219 -3.285010
- F 1.626818 1.617448 -3.900650
- F 3.532001 0.392063 -4.192276
- F 3.178448 1.521261 -2.243793
- C -0.468464 -3.822402 -1.180560
- Н -0.280721 -3.663343 -0.097675
- Н -1.536078 -3.563080 -1.336375
- C -0.278528 -5.284333 -1.553336
- Н 0.774356 -5.611023 -1.468665
- Н -0.887527 -5.948198 -0.913000
- Н -0.585679 -5.459953 -2.601969
- C -1.908175 -1.687334 2.599267
- C -2.639892 -1.336139 3.738258

- C -2.607882 -1.938157 1.409003
- C -4.030918 -1.219280 3.681739
- Н -2.126443 -1.149146 4.688835
- C -3.992849 -1.828927 1.350698
- Н -2.048414 -2.234459 0.508878
- C -4.711406 -1.459045 2.489701
- Н -4.586585 -0.941866 4.585133
- Н -4.516874 -2.035842 0.410131
- Н -5.803468 -1.368808 2.445941
- C 3.270425 -0.099346 2.149814
- C 4.445307 -0.494676 2.810277
- C 3.398150 0.614129 0.955015
- C 5.700591 -0.167542 2.304761
- Н 4.383723 -1.056649 3.749574
- C 4.654977 0.947641 0.448295
- Н 2.505956 0.915378 0.387724
- C 5.810688 0.564895 1.122320
- Н 6.601132 -0.484806 2.844120
- Н 4.714310 1.488701 -0.502640
- Н 6.797572 0.826636 0.722748
- C -4.423608 0.623619 -1.034594
- C -3.866441 -0.045396 -2.131195
- C -5.708067 0.262102 -0.615990
- C -4.558034 -1.073044 -2.769130
- Н -2.866532 0.242729 -2.487680
- C -6.411192 -0.756106 -1.258896
- Н -6.159743 0.783688 0.238605
- C -5.833621 -1.434707 -2.331381
- Н -4.098891 -1.592570 -3.618805

- Н -7.415923 -1.026799 -0.913555
- Н -6.379500 -2.242227 -2.832580
- C -0.120768 3.692396 -0.073031
- C -0.131849 4.553552 1.031141
- C 1.115651 3.285642 -0.592736
- C 1.063399 4.981008 1.610750
- Н -1.082585 4.905395 1.449762
- C 2.308259 3.716501 -0.018941
- Н 1.152482 2.625430 -1.471420
- C 2.287434 4.560649 1.091907
- H 1.034803 5.655618 2.474601
- Н 3.257102 3.383046 -0.453764
- Н 3.225580 4.899545 1.546430
- C -1.776387 1.614137 1.828895
- Н -2.570343 0.935376 2.201493
- Н -2.061332 2.637035 2.146050
- C -0.431378 1.226792 2.437222
- Н 0.314067 2.034474 2.271057
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- H 5.007061 1.468676 3.047851
- Н 3.473703 3.566975 3.120634
- Н 4.132544 3.228244 1.502313
- C -1.422479 1.047733 -2.684702
- C 0.381713 -1.013935 -2.866846
- C -1.621117 0.011197 -3.799897
- Н -0.970115 1.953180 -3.136856
- $C \qquad -0.241605 \quad -0.521466 \quad -4.169851$
- Н -0.205862 -1.884151 -2.514651
- Н -2.247249 -0.824648 -3.433858
- Н -2.144421 0.468206 -4.660626
- Н -0.293342 -1.335825 -4.917796
- Н 0.360063 0.293791 -4.621352
- P -0.049440 0.322532 -1.616159
- P 2.096307 1.005074 0.813501
- Cu 0.116201 -0.218210 0.643501
- C -0.926283 -2.136179 1.251587
- C -3.277681 -2.066990 1.047070
- C -0.889586 -1.220967 2.281433
- C -3.362922 -1.189435 2.152588
- C -2.140725 -0.479216 2.694056
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- N -2.035872 -2.477607 0.543978
- B -1.885818 -3.342544 -0.752224
- F -0.527604 -3.587093 -0.952408
- F -2.390869 -2.622121 -1.843808
- F -2.566706 -4.545143 -0.635406
- C -2.049516 0.997067 2.261596
- Н -1.059779 1.363167 2.614233

- Н -2.029435 1.034965 1.147155
- C -3.133367 1.924252 2.778378
- Н -3.206417 1.879562 3.883067
- Н -2.919260 2.973670 2.498473
- Н -4.123953 1.671530 2.357699
- C -4.613585 -0.955156 2.733045
- C -5.785032 -1.489210 2.204637
- Н -6.752534 -1.280589 2.675101
- C -4.467379 -2.583015 0.499433
- C -5.702260 -2.289168 1.064399
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- Н -4.661998 -0.322005 3.628730
- Н -4.414107 -3.249272 -0.364897
- Н -0.042699 -1.255259 2.979996
- Н -0.027665 -2.726567 1.018004
- C 1.839497 -1.357429 -2.799290
- C 2.283966 -2.159121 -1.735282
- C 2.788690 -0.851394 -3.693798
- C 3.638446 -2.415417 -1.548022
- Н 1.539854 -2.589140 -1.051358
- C 4.148693 -1.112657 -3.510518
- Н 2.472219 -0.239438 -4.547359
- C 4.580036 -1.883213 -2.432088
- Н 3.963358 -3.036128 -0.703757
- Н 4.877829 -0.705878 -4.221290
- Н 5.648682 -2.081007 -2.284673
- C -2.633522 1.461125 -1.901237
- C -2.922146 2.818419 -1.718677
- C -3.472587 0.510989 -1.303099

- C -4.027521 3.221889 -0.968409
- Н -2.269879 3.572796 -2.180259
- C -4.579560 0.911613 -0.558230
- Н -3.248998 -0.559990 -1.406592
- C -4.863201 2.268056 -0.390220
- Н -4.239734 4.290887 -0.842573
- Н -5.219127 0.149484 -0.093754
- Н -5.734367 2.580669 0.197900
- C 4.025767 -0.917096 1.746878
- C 5.247243 -1.497983 1.388364
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- C 5.503670 -2.844168 1.646107
- Н 6.008811 -0.882675 0.890453
- C 3.312495 -3.068096 2.626190
- Н 2.104104 -1.277532 2.667025
- C 4.533949 -3.635712 2.260278
- Н 6.466855 -3.280228 1.356148
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- Н 4.728834 -4.696548 2.454817
- C 1.420086 3.755451 0.943627
- C 2.175690 4.676181 0.207814
- C 0.021634 3.848223 0.881636
- C 1.551412 5.650450 -0.572691
- Н 3.271269 4.645418 0.239430
- C -0.604131 4.822472 0.109166
- Н -0.586873 3.135246 1.453731
- C 0.161027 5.727677 -0.627746
- Н 2.163650 6.361570 -1.139362
- Н -1.699449 4.873576 0.085444

- Н -0.326149 6.497831 -1.236571
- C 1.260708 1.634683 -1.815723
- Н 1.543899 1.726376 -2.883343
- Н 0.751498 2.582809 -1.539370
- C 2.496162 1.425160 -0.949564
- Н 3.160667 2.311423 -0.989282
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- H 1.477761 2.245269 4.685233
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- C 4.976479 1.339705 0.562247
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- C -2.414426 1.683623 0.103316
- Н -2.964462 2.504938 -0.397584
- Н -3.157343 1.125378 0.708109

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