Supporting Information: Human γ S-crystallin copper binding helps buffer against aggregation caused by oxidative damage

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Figure S1: Size distribution histograms for aggregates of γ S-crystallin point variants allowed to form without external provocation (left) and those formed through UV-A photodamage (right).

Figure S2: Irradiation of $\gamma\text{S-WT}$ from UVA (1.6 kJ/cm²) and UVB (104.4 J/cm²) produces aggregates with similar morphology independent of sample concentration during irradiation.

Figure S3: TEM images of aggregates formed from γS -WT after irradiation using UVA or UVB radiation for 10 (0.5 kJ/cm², 34.8 J/cm²), 20 (1.1 kJ/cm², 69.6 J/cm²), and 30 minutes $(1.6 \text{ kJ/cm}^2, 104.4 \text{ J/cm}^2)$. Under both treatments, globular aggregates form and these particles associate, yielding larger aggregates.

Figure S4: The first panel shows undamaged WT γ S-crysallin at the same concentration as for the variants shown in the other images. No aggregates are formed under these conditions. The other panels show TEM images of γ S-crystallin aggregates formed either by allowing the sample to precipitate over time as described in the Methods section of the main text, or by irradiation with UVA. Native aggregates from cataract-related point variants are very similar in size and morphology to those formed via UVA photodamage.

Figure S5: FTIR spectra of aggregated γ S-crystallins. (A) FTIR of soluble γ S-WT and tested point variants exhibit identical spectra. (B) Native aggregates of γ S point variants exhibit minor broadening of the amide I peak and slightly elevated amide II/I peak intensity ratios. (C) Soluble and aggregated γ S-WT resulting from CuCl₂ treatment show minor narrowing of the amide I peak. As is observed for the γ S-crystallin point variants, the amide II/I peak intensity ratio is increased. (D) UVB treatment of γ S-WT induced similar changes to $CuCl₂$ treatment for soluble and insoluble species.

Figure S6: (A) Structure of human γ S-crystallin (PDBID: 2M3T¹), with the location of the cysteine loop indicated by an open rectangle. (B) Predicted Cu⁺ binding sites for γ Scrystallin generated via the MIB webserver (http://bioinfo.cmu.edu.tw/MIB/).^{2,3} Binding predictions implicated some or all of C23, C25, C27, and C83 in the majority of the results.

Figure S7: Analytical SEC measurements for γ S-WT and γ S-C₀ collected separately following Cu^{2+} treatment. The soluble sample (top panel) was collected as the supernatant following treatment. The aggregates that were resuspended during the subsequent EDTA treatment were then collected (middle panel). Finally, the remaining aggregates were treated with DTT, with the resulting resuspension collection as the final samples. (bottom panel). Prior to loading onto the SEC, each sample was treated with EDTA and DTT in order to minimize potential damage to the column. The distribution of protein species for γS -WT is similar across the three chromatograms. Monomers compose the bulk of the elutants while a dimer peak is also clearly observable. In contrast to $\gamma S-WT$, in $\gamma S-C_0$, monomeric species are almost exclusively present in the remaining soluble and DTT-resolubilized chromatograms. The EDTA-resolubilized chromatogram of $\gamma S-C_0$ is primarily dimeric with a low level of monomer present.

Figure S8: SDS-PAGE separation of γ S-crystallin species after treatment with CuCl₂. The gradient (left to right) used for all gels was 0, 0.1, 0.2, 0.4, 0.6, 0.8, 1, 1.5, 2, 3, and 5 equivalents of CuCl2. The lanes of gels A and B were loaded with equal concentrations and volumes of protein, with the exception of the 5 equivalents lane, for which only trace amounts of protein remained soluble. For gel C the samples were prepared with equal volumes of the remaining soluble protein such that the observed loss of protein is representative of the increased insoluble fraction. (Gel A) Soluble γS -WT after incubation with increasing levels of CuCl₂. (Gel B) Soluble γ S-WT after incubation with increasing levels of CuCl₂ and reduction subsequent reduction with BME. (Gel C) Soluble γ S-WT after incubation with increasing levels of of CuCl₂. (Gel D) A: γ S-WT, B-D: γ S-WT treated 1 equiv CuCl₂ alone (B), with EDTA (C), with EDTA and BME (D), E-G: γS -C₀ treated 1 equiv CuCl₂ alone (E), with EDTA (F), or with EDTA and BME (G), H: soluble UV-B treated γS , I: soluble UV-B treated γS with BME.

$DTT (5$ equvi) – $+$

Figure S9: SDS-PAGE analysis of γ S-WT stored for 5 months. Two independent samples were stored at 10 mg/mL in the absence of reducing agent at $4 \degree C$ for 5 months. Prior to electrophoresis, lanes 2 and 4 were treated with 5 equivalents of DTT. The first two lanes show that the intramolecular disulfide bonds formed are completely reducible, whereas the intramolecular disulfide formed in lane 3 shows negligible reduction following DTT treatment (lane 4). In both cases, the low levels of dimeric species observed are eliminated following reduction.

Figure S10: SDS-PAGE analysis of chelating agent-treated γ S-crystallin aggregates. Lanes are labeled from left to right, following the ladded.. Lane 1: γ S-WT, Lane 2: γ S-WT treated 1 equiv CuCl₂ alone, Lane 3: sample 2 with EDTA, Lane 4: sample 2 with EDTA and BME, Lane 5: γ S-C₀ treated 1 equiv CuCl₂, Lane 6: sample 5 with EDTA, Lane 7: sample 5 with EDTA and BME, Lane 8: soluble UV-B-treated γ S-WT, Lane 9: sample 7 with BME.

Figure S11: The observed secondary fragmentation observed for the YDCDCD-CADFHTY/LSSCR peptides containing an intramolecular disulfide bond. Fragments containing the intramolecular disulfide bond are colored red, while fragments of only one of the peptides are colored blue. The first peptide, YDCDCDCADFHTY, and second peptide, LSSCR, are indicated by the numbers '1' and '2', respectively. A slash between fragments indicates an intramolecular disulfide bond. The observed masses for each fragment are listed in Table S1.

Table S1: MS-MS fragmentation of disulfide-bonded YDCDCDCADFHTY and LSSCR peptides Table S1: MS-MS fragmentation of disulfide-bonded YDCDCDCADFHTY and LSSCR peptides

Figure S12: Digested fragments containing an intramolecular disulfide bond were identified by their disappearance following the addition of DTT. The peptide trace without DTT is colored black and the peptide trace with DTT is colored yellow. The predicted isotope pattern for the fragment is shown as green bars. Secondary fragmentation was not sufficient to confirm the location of the intramolecular disulfides were between peptides. The Na⁺ bound versions of fragments 1 and 4 were also observed and are shown in the respective traces.

Protein	No. Residues	No. Cysteines	Percentage Cysteine	UniProt ID
α A	173	2	1.16\%	CRYAA HUMAN
α B	175	0	0.00%	CRYAB HUMAN
β A1*	198	$\overline{5}$	2.53%	CRBA1 HUMAN
β A2	197	6	3.05%	CRBA2 HUMAN
β A3	215	$\overline{5}$	2.33%	CRBA1 HUMAN
β A4	196	5	2.55%	CRBA4 HUMAN
β B1	252		0.40%	CRBB1 HUMAN
β B2	205	$\overline{2}$	0.98%	CRBB2 HUMAN
β B3	211	$\overline{2}$	0.95%	CRBB3 HUMAN
γ A	174	9	5.17%	CRGA HUMAN
γB	175	7	4.00%	CRGB HUMAN
γC	174	8	4.60%	CRGC HUMAN
γD	174	6	3.45%	CRGD HUMAN
γ S	178		3.93%	CRGS HUMAN

Table S2: Cysteine content of human α -, β -, and γ -crystallins

*Note: β A1 is an isoform of β A3 generated via an alternate translation initiation site

Figure S13: The mass spectra and reconstructed masses for soluble fractions of γ S-WT (A, B) and $\gamma S-C_0$ (C,D) following incubation with 2 equivalents of Cu²⁺. Dimeric species are evidenced as a doubly-charged species in the raw mass spectrum of $\gamma S-WT$ and the resulting mass reconstruction. In contrast, minimal levels of dimer are observed for $\gamma S-C_0$. The expected monomeric and dimeric molecular weights are shown for $\gamma S-WT$ and $\gamma S-C_0$ in panels C and D. The dimer masses assume a singular intermolecular disulfide bond. Inset images show the mass reconstruction for the monomeric peak. The observed mass for γ S-WT is consistent with an m-2 species, while the mass of the $\gamma S-C_0$ peak is consistent with the expected mass (m).

References

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