Supplementary Information

Long Wavelength Single Photon Like Driven Photolysis Via Triplet-Triplet Annihilation

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Supplementary Methods

Chemicals: perylene, tris [2-phenylpyridinato- C_2 , N] iridium (III) (Ir(ppy)₃) were purchased from TCI chemicals. The solvents of dimethylformamide (DMF), phosphoryl chloride (POCl₃), sodium borohydride (NaBH₄), methanol, acetyl chloride, pyridine, dicyclohexylcarbodiimide (DCC), 4-(dimethylamino) pyridine (DMAP); triethylamine (TEA) N-butylamine, dry CH₂Cl₂, 3-(4,5dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide (MTT), propidium iodide (PI), 9anthracenemethanol, 4-nitrophenylchloroformate, chloro (3-methylbutoxy) methanone, Npivaloyl glycine, 4-[(2,2-dimethylpropanoyl)amino] butanoic acid and poly(styrene-co-maleic anhydride) (PSMA, $M_w = 1600$) were purchased from Sigma-Aldrich (St. Louis, MO, USA). meso-tetraphenyltetrabenzoporphine palladium (PdTPBP) and complex platinum octaethylporphyrin (PtOEP) were purchased from Frontier Scientific, Inc. Dulbecco's Modified Eagle Medium (DMEM), fetal bovine serum (FBS), penicillin-streptomycin, trypsin-EDTA (EDTA = ethylenediaminetetraacetic acid), and phosphate-buffered saline solution (PBS) were obtained from Invitrogen (Carlsbad, CA, USA). Ultrapure water was prepared by using a Millipore Simplicity System (Millipore, Bedford, USA). All of the above-mentioned chemicals were used as received without further purification.

Characterization: ¹H NMR spectra were recorded with a Bruker 500 MHz spectrometer. UV–Vis spectra were recorded on an Agilent Cary-5 spectrophotometer. Steady-state fluorescence spectra were measured on a HITACHI F-7000 spectrometer. The morphology of the TTAP NPs was characterized by using a JEOL JEM-200CX transmission electron microscope (TEM) operated at 80 kV. The sample for TEM measurement was prepared by dropping the solution onto a carbon-coated copper grid following negative staining with 2.0% (w/v) phosphotungstic acid. The particle size and size distribution were measured by using dynamic light scattering (DLS) on a Malvern Nano-ZS particle size analyzer, excitation wavelength 532 nm. All samples in the Stern-Volmer quenching plots experiments were deaerated with argon for ca. 15 min before measurement and the gas flow was maintained during the measurements. For the TTA upconversion spectra measurement, the diode pumped solid-state laser (650 nm, continues wave, CW, Hi-Teach company, China) was used as the excitation light source and a modified spectrofluorometer was used to record the upconversion spectra. For the prodrug photolytic experiment, all the LEDs

(Mightex Company) were used as the excitation light source.

Preparation and characterization of PPGs-X.



Compound 1: 4-benzyl alcohol coumarin was synthesized according to the previous synthesis protocols.¹

Then 4-benzyl alcohol coumarin (247.1 mg, 1.0 mmol) was dissolved in dry dichloromethane (DCM) (10 mL) and pyridine (50 µL) were mixed. Acetyl chloride (2.0 mmol) was dissolved in dry DCM (10 mL) and added dropwise to the mixture over 10 min and then stirred for 2 hours. After the reaction, the solvents were removed under vacuum condition. The raw **1** was purified by silica gel flash column chromatography with DCM/n-hexane =1/1 (v/v) as elute. The yellow band was collected and yield determined to be 92% (266.9 mg). ¹H-NMR (500 MHz, CDCl₃): δ =7.29-7.28 (d, 1H, *J* = 5.0 Hz,), 6.59-6.57 (d, 1H, *J* =10.0Hz), 6.52 (s, 1H), 6.14-6.13 (d, 1H, *J* = 5.0 Hz), 5.22 (s, 2H), 3.44-3.39 (m, 4H), 2.19 (s, 3H), 1.22-1.20 ppm (m, 6H). ¹³C-NMR (125 MHz, CDCl₃): δ = 150.7, 149.4, 124.4, 108.7, 106.5, 106.0, 97.9, 61.4, 44.8, 20.8, 12.4 ppm. HRMS (ESI) exact mass calculated for [M + H]⁺ (C₁₆H₁₉NO₄+H⁺) m/z 290.1392, found m/z 290.1387.



Compound 2: 4-benzyl alcohol anthracene (208 mg, 1.0 mmol) dissolved in dry dichloromethane (DCM) (10 mL) and pyridine (50 µL) were mixed. Acetyl chloride (2.0 mmol) was dissolved in

dry DCM (10 mL) and added dropwise to the mixture over 10 min and then stirred for 2 hours. After the reaction, the solvents were removed under vacuum condition. The raw **2** was purified by silica gel flash column chromatography with DCM/n-hexane =1/2 (v/v) as elute. The yellow band was collected and yield determined to be 85% (212.5 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 8.52 (s, 1H), 8.34-8.32(d, 2H, *J* = 10.0Hz), 8.05-8.03 (d, 2H, *J* = 9.5 Hz), 7.58-7.56 (m, 2H), 7.50-7.48 (m, 2H), 6.16 (s, 2H), 2.09 ppm (s, 3H). ¹³C-NMR (125 MHz, CDCl₃): δ = 171.3, 131.4, 131.1, 129.2,129.1, 126.7, 126.2, 125.1, 123.9, 58.8, 21.0 ppm. HRMS (ESI) exact mass calculated for [M + Na⁺] (C₁₇H₁₄O₂ + Na⁺) m/z 273.0891, found m/z 273.0886.



Compound 3: 4-benzyl alcohol anthracene (208 mg, 1.0 mmol) dissolved in dry dichloromethane (DCM) (10 mL) and pyridine (50 µL) were mixed. Isobutyl chloroformate (2.0 mmol) was dissolved in dry DCM (10 mL) and added dropwise to the mixture over 10 min and then stirred for 2 hours. After the reaction, the solvents were removed under vacuum condition. The raw **3** was purified by silica gel flash column chromatography with DCM/n-hexane =1/2 (v/v) as elute. The yellow band was collected and yield determined to be 65% (200.2 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 8.51 (s, 1H), 8.40-8.38 (d, 2H, *J* = 10.0Hz), 8.03-8.01 (d, 2H, *J* = 9.5Hz), 7.60-7.56 (m, 2H), 7.50-7.47(m, 2H), 6.22(s, 2H), 3.96-3.94 (d, 2H, *J* = 10.0Hz), 1.98-1.91 (m, 1H), 0.92 (s, 3H), 0.90 ppm (s, 3H). ¹³C-NMR (125 MHz, CDCl₃): δ = 155.68, 131.4, 131.2, 129.5, 129.1, 126.8, 125.1, 124.0, 123.9, 74.4, 74.0, 62.1, 27.8, 18.9 ppm. HRMS (ESI) exact mass calculated for [M + Na]⁺ (C₂₀H₂₀O₃ + Na⁺) m/z 331.1310, found m/z 331.1305.



Compound 4: 4-benzyl alcohol anthracene (208 mg, 1.0 mmol) dissolved in dry dichloromethane (DCM) (10 mL) and pyridine (50 µL) were mixed. 4-nitrophenyl chloroformate (1.2 mmol) was dissolved in dry DCM (10 mL) and added dropwise to the mixture over 10 min and then stirred for 2 hours. Next, the butylamine (2.0 mmol) was added and stirred for another 12 hours in room temperature. After the reaction, 50 mL CH₂Cl₂ was added. The solutions were washed with saturated aqueous solution of sodium bicarbonate (30 mL). The organic layer was collected and dried by Na₂SO₄; the solvents were removed under vacuum condition. The raw **4** was purified by silica gel flash column chromatography with DCM/n-hexane =1/1 (v/v) as elute. The shallow yellow band was collected and yield determined to be 60% (184.2 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 8.50 (s, 1H), 8.40-8.38 (m, 2H), 8.03-8.01 (d, 2H, *J*=9.0 Hz), 7.57-7.55 (m, 2H), 7.50-7.47 (m, 2H), 6.14 (s, 2H), 4.66 (s, 1H), 3.23-3.20 (m, 2H), 1.50-1.44 (m, 2H), 1.35-1.30 (m, 2H), 0.92-0.90 ppm (m, 3H). ¹³C-NMR (125 MHz, CDCl₃): δ = 156.6, 131.4, 131.0, 129.1, 129.0, 126.9, 126.6, 125.1, 124.1, 59.0, 40.9, 32.0, 19.9, 13.7 ppm. HRMS (ESI) exact mass calculated for [M + Na]⁺ (C₂₀H₂₁NO₂ + Na⁺) m/z 330.1470, found m/z 330.1465.



Compound 5: 4-benzyl alcohol anthracene (208 mg, 1.0 mmol) and Boc-Gly-OH (1.5 mmol) dissolved in dry dichloromethane (DCM) (10 mL) in 0 °C. Next, the 4-dimethylaminopyridine (DMAP, 0.2 mmol) and *N*, *N*'-dicyclohexylcarbodiimide (DCC, 1.2 mmol) were added to the mixture and then stirred for 12 hours. After the reaction, the solvents were removed under vacuum condition. The raw **5** was purified by silica gel flash column chromatography with DCM/n-hexane =1/1 (v/v) as elute. The shallow yellow band was collected and yield determined to be 71% (260 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 8.53 (s, 1H), 8.33-8.32 (d, 2H, *J* = 5.0 Hz), 8.04-8.03 (d, 2H, *J* = 5.0Hz), 7.60-7.57 (m, 2H), 7.52-7.49 (m, 2H), 6.24 (s, 2H), 4.98 (s, 1H), 3.93 (s, 2H), 1.42 ppm (s, 9H). ¹³C-NMR (125 MHz, CDCl₃): δ = 131.4, 131.1, 129.5, 129.2, 126.9, 125.2, 123.8, 59.7, 42.5, 28.3 ppm. HRMS (ESI) exact mass calculated for [M + Na]⁺ (C₂₂H₂₃NO₄ + Na⁺) m/z

388.1525, found m/z 388.1519.



Compound 6: 4-benzyl alcohol anthracene (208 mg, 1.0 mmol) and Boc-GABA-OH (1.5 mmol) dissolved in dry dichloromethane (DCM) (10 mL) in 0 °C. Next, 4-dimethylaminopyridine (DMAP, 0.2 mmol) and *N*, *N*'-dicyclohexylcarbodiimide (DCC, 1.2 mmol) were added to the mixture and then stirred for 12 hours. After the reaction, the solvents were removed under vacuum condition. The raw **6** was purified by silica gel flash column chromatography with DCM/n-hexane =1/1 (v/v) as elute. The shallow yellow band was collected and yield determined to be 80% (314 mg). 'H-NMR (500 MHz, CDCl₃): δ = 8.51 (s, 1H), 8.34-8.32 (d, 2H, *J* = 10.0Hz,), 8.04-8.03 (d, 2H, *J* = 5.0 Hz,), 7.59-7.56 (m, 2H), 7.51-7.48 (m, 2H), 6.17 (s, 2H), 4.54 (s, 1H), 3.13-3.12 (d, 2H, *J* = 5.0 Hz), 2.39-2.36 (m, 2H), 1.82-1.78 (m, 2H), 1.41 ppm (s, 9H). ¹³C-NMR (125 MHz, CDCl₃): δ = 173.4, 155.9, 131.4, 131.0, 129.2, 129.1, 126.7, 126.2, 125.1, 123.9, 58.9, 39.8, 31.1, 28.4, 25.3 ppm. HRMS (ESI) exact mass calculated for [M + Na]⁺ (C₂₄H₂₇NO₄ + Na⁺) m/z 416.1838, found m/z 416.1832.



Compound 7: 2,4-dimethylpyrrole (980 mg, 10 mmol) and acetoxyacetyl chloride (5.0 mmol) dissolved in dry dichloromethane (DCM) (300 mL) at 0 °C and then stirred for 12 hours. After the reaction, 5 mL BF₃.OEt₂ was added dropwise for 20 min at 0 °C and stirred another 4 hours, the

solvents were removed under vacuum condition. The raw **7** was purified by silica gel flash column chromatography with DCM/n-hexane =1/1 (v/v) as elute. The shallow yellow band was collected and yield determined to be 15% (480 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 6.09 (s, 2H), 2.53 (s, 6H), 2.36 (s, 6H), 2.13 (s, 3H), 1.54 ppm (s, 2H). ¹³C-NMR (125 MHz, CDCl₃): δ = 170.6, 156.6, 141.5, 133.3, 132.7, 122.3, 57.9, 20.6, 13.6, 14.7 ppm. HRMS (ESI) exact mass calculated for [M + H]⁺ (C₁₆H₁₉BF₂N₂O₂ + H⁺) m/z 321.1586, found m/z 321.1580.



Compound 8: 1-benzyl alcohol perylene was first synthesized according to the previous synthesis protocols.²4-benzyl alcohol perylene (140 mg, 0.5 mmol) dissolved in dry dichloromethane (DCM) (30 mL) and pyridine (50 µL) were mixed. Acetyl chloride (1.0 mmol) was dissolved in dry DCM (10 mL) and added dropwise to the mixture over 10 min and then stirred for 2 hours. After the reaction, the solvents were removed under vacuum condition. The raw **8** was purified by silica gel flash column chromatography with DCM/n-hexane =1/2 (v/v) as elute. The yellow band was collected and yield determined to be 95% (150 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 8.26-8.24 (d, 1H, *J* = 10.0Hz), 8.22-8.20 (m, 3H), 8.17-8.16 (d, 1H, *J* = 5.0Hz), 7.72-7.70 (m, 2H), 7.58-7.54 (m, 2H), 7.51-7.48 (m, 2H), 5.51 (s, 2H), 2.14 ppm (s, 3H). ¹³C-NMR (125 MHz, CDCl₃): δ = 171.1, 134.6, 133.0, 132.2, 131.9, 131.1, 130.9, 130.8, 129.0, 128.5, 128.3, 128.2, 128.1, 127.1, 126.6, 126.5, 123.3, 120.6, 120.5, 120.4, 119.6, 64.7, 21.1 ppm. HRMS (ESI) exact mass calculated for [M + Na]⁺ (C₂₃H₁₆O₂+Na⁺) m/z 347.1043, found m/z 347.1042.



Compound 9: 4-benzyl alcohol perylene (140 mg, 0.5 mmol) was dissolved in dry dichloromethane (DCM) (30 mL) and pyridine (50 μ L) were mixed. Isobutyl chloroformate (1.0 mmol) was dissolved in dry DCM (10 mL) and added dropwise to the mixture over 10 min and then stirred for 2 hours. After the reaction, the solvents were removed under vacuum condition. The raw **9** was purified by silica gel flash column chromatography with DCM/n-hexane =1/2 (v/v) as elute. The yellow band was collected and yield determined to be 85% (107 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 8.26-8.25 (d, 1H, *J* = 5.0 Hz), 8.23-8.21 (m, 2H), 8.18-8.16(d, 1H, *J* = 10.0Hz), 7.90-7.88 (d, 1H, *J* = 9.0 Hz), 7.72-7.70 (m, 2H), 7.60-7.56 (m, 2H), 7.51-7.48 (m, 2H), 5.57 (s, 2H), 3.97-3.96(d, *J* = 5.0 Hz, 2H), 2.00-1.94 (m, 1H), 0.95 (s, 3H), 0.94 ppm (s, 3H). ¹³C-NMR (125 MHz, CDCl₃): δ = 155.4, 134.6, 133.0, 132.4, 131.8, 130.8, 130.3, 129.0, 128.0, 127.2, 126.6, 126.5, 123.3, 120.7, 120.6, 120.4, 119.6, 74.3, 67.9, 27.8, 18.9 ppm. HRMS (ESI) exact mass calculated for [M + Na]⁺ (C₂₆H₂₃O₃+Na⁺) m/z 406.1539, found m/z 406.1495.



Compound 10: 4-benzyl alcohol perylene (140 mg, 0.5 mmol) was dissolved in dry dichloromethane (DCM) (30 mL) and pyridine (50 μ L) were mixed. 4-nitrophenyl chloroformate (0.6 mmol) was dissolved in dry DCM (10 mL) and added dropwise to the mixture over 10 min

and then stirred for 2 hours. Next, the butylamine (2.0 mmol) was added and stirred for another 12 hours in room temperature. After the reaction, 50 mL CH₂Cl₂ was added. The solutions were washed with saturated aqueous solution of sodium bicarbonate (30 mL). The organic layer was collected and dried by Na₂SO₄; the solvents were removed under vacuum condition. The raw **10** was purified by silica gel flash column chromatography with DCM/n-hexane =1/1 (v/v) as elute. The shallow yellow band was collected and yield determined to be 45% (91 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 8.25-8.23 (d, 1H, *J* =10.0Hz), 8.22-8.19 (m, 2H), 8.16-8.15 (d, 1H, *J* = 5.0 Hz), 7.89-7.87 (d, 1H, *J* = 10.0Hz), 7.70-7.69 (d, 2H, *J* = 5.0Hz), 7.57-7.54 (m, 2H), 7.50-7.47 (m, 2H), 5.50 (s, 2H), 4.74 (s, 1H), 3.24-3.21 (m, 2H), 1.49-1.47 (m, 2H), 1.39-1.32 (m, 2H), 0.94-0.91 ppm (m, 3H). ¹³C-NMR (125 MHz, CDCl₃): δ = 134.6, 133.0, 132.1, 131.8, 131.7, 131.1, 130.9, 129.0, 128.5, 128.1, 128.0, 127.0, 126.6, 123.5, 120.3, 119.7, 65.0, 40.9, 32.0, 19.9, 13.7 ppm. HRMS (ESI) exact mass calculated for [M + Na]⁺ (C₂₆H₂₃NO₂ + Na⁺) m/z 404.1626, found m/z 404.1621.



Compound 11: 4-benzyl alcohol perylene (140 mg, 0.5 mmol) was dissolved in dry dichloromethane (DCM) (30 mL) and triethylamine (100 μ L) were mixed. Benzoyl chloride (0.6 mmol) was dissolved in dry DCM (10 mL) and added dropwise to the mixture over 10 min and then stirred for 12 hours. After the reaction, 50 mL CH₂Cl₂ was added. The solutions were washed with saturated aqueous solution of sodium bicarbonate (30 mL). The organic layer was collected and dried by Na₂SO₄; the solvents were removed under vacuum condition. The raw **11** was purified by silica gel flash column chromatography with DCM/n-hexane =1/1 (v/v) as elute. The shallow yellow band was collected and yield determined to be 80% (154 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 8.27-8.25 (d, 1H, *J* = 10.0 Hz), 8.22-8.20 (m, 2H), 8.19-8.16 (m, 2H), 8.09-8.07 (m, 2H), 7.96-7.94 (d, 1H, *J* = 9.0 Hz), 7.72-7.70 (m, 2H), 7.55-7.53 (d, 2H, *J* = 10.0 Hz), 7.52-7.49 (m, 2H), 7.44-7.43 (m, 2H), 5.76 ppm (s, 2H). ¹³C-NMR (125 MHz, CDCl₃): δ = 166.6, 134.6, 134.5, 133.1,

133.0, 132.3, 131.9, 131.0, 130.9, 130.6, 130.1, 129.8, 129.1, 128.9, 128.4, 128.3, 128.1,128.0, 127.2, 126.7, 126.6, 123.4, 120.6, 120.5, 120.4, 119.6, 65.3 ppm. HRMS (ESI) exact mass calculated for $[M + Na]^+$ ($C_{28}H_{18}O_2 + Na^+$) m/z 409.1204, found m/z 409.1199.



Compound 12: 4-benzyl alcohol perylene (140 mg, 0.5 mmol) and Boc-Gly-OH (1.0 mmol) were dissolved in dry dichloromethane (DCM) (30 mL) in 0 °C. Next, the 4-dimethylaminopyridine (DMAP, 0.1 mmol) and *N*, *N*'-dicyclohexylcarbodiimide (DCC, 0.6 mmol) were added to the mixture and then stirred for 12 hours. After the reaction, the solvents were removed under vacuum condition. The raw **12** was purified by silica gel flash column chromatography with DCM/n-hexane =1/1 (v/v) as elute. The shallow yellow band was collected and yield determined to be 78% (171 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 8.26-8.20 (m, 3H), 8.17-8.15 (d, 1H, *J* = 10.0 Hz), 7.83-7.81 (d, 1H, *J* = 10.0Hz), 7.72-7.70 (m, 2H), 7.59-7.54 (m, 2H), 7.51-7.48(m, 2H), 5.59 (s, 2H), 5.01 (s, 1H), 3.99 (s, 2H), 1.44 ppm (s, 9H); ¹³CNMR (125 MHz, CDCl₃) δ = 170.4, 134.6, 133.0, 132.5, 131.9, 131.0, 130.8, 130.2, 129.0, 128.6, 128.5, 128.3, 128.1, 127.3, 126.7, 126.6, 123.2, 120.7, 120.6,120.4, 119.5, 65.5, 42.5, 28.3 ppm. HRMS (ESI) exact mass calculated for [M + Na]⁺ (C₂₄H₂₅NO₄ + Na⁺) m/z 462.1681, found m/z 462.1676.



Compound 13: 4-benzyl alcohol perylene (140 mg, 0.5 mmol) and Boc-GABA-OH (1.0 mmol) were dissolved in dry dichloromethane (DCM) (10 mL) in 0 °C. Next, the 4-dimethylaminopyridine (DMAP, 0.2 mmol) and *N*, *N*'-dicyclohexylcarbodiimide (DCC, 1.2 mmol) were added to the mixture and then stirred for 12 hours. After the reaction, the solvents were removed under vacuum condition. The raw **13** was purified by silica gel flash column chromatography with DCM/n-hexane =1/1 (v/v) as elute. The shallow yellow band was collected and yield determined to be 69% (156 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 8.26-8.24 (d, 1H, *J* = 10.0Hz), 8.23-8.20 (m, 2H), 8.17-8.16 (d, 1H, *J* = 5.0 Hz), 7.84-7.82 (d, 1H, *J* = 10.0Hz), 7.72-7.70 (m, 2H), 7.58-7.54 (m, 2H), 7.51-7.48 (m, 2H), 5.21 (s, 2H), 4.58 (s, 1H), 3.16 (s, 2H), 2.45-2.42 (m, 2H), 1.86-1.83 (m, 2H), 1.42 ppm (s, 9H). ¹³C NMR (125 MHz, CDCl₃) δ = 173.2, 155.9, 134.6, 133.0, 132.3, 131.8, 131.1, 130.9, 130.8, 129.0, 128.5, 128.4, 128.2, 128.1, 127.2, 126.7, 126.6, 123.3, 120.6, 120.4, 119.6, 64.8, 39.9, 31.6, 28.4, 25.3 ppm. HRMS (ESI) exact mass calculated for [M + Na]⁺ (C₃₀H₂₉NO₄ + Na⁺) m/z 490.1994, found m/z 490.1989.



Compound 14: 4-benzyl alcohol perylene (140 mg, 0.5 mmol) and Boc-Cys (StBu)-OH (1.0 mmol) were dissolved in dry dichloromethane (DCM) (10 mL) in 0 °C. Next, the 4-dimethylaminopyridine (DMAP, 0.2 mmol) and *N*, *N*'-dicyclohexylcarbodiimide (DCC, 1.2 mmol) were added to the mixture and then stirred for 12 hours. After the reaction, the solvents were removed under vacuum condition. The raw **14** was purified by silica gel flash column chromatography with DCM/n-hexane =1/1 (v/v) as elute. The yellow band was collected and yield determined to be 65% (186 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 8.24-8.19 (m, 3H), 8.16-8.13 (m, 1H), 7.83-7.82 (m, 1H), 7.71-7.70 (m, 2H), 7.57-7.48 (m, 4H), 5.58 (s, 2H), 5.43-5.41 (m, 1H),

4.66 (s, 1H), 3.21-3.11 (m, 2H), 1.43 (s, 9H), 1.26 ppm (s, 9H).¹³C NMR (125 MHz, CDCl₃) δ = 170.8, 155.1, 134.6, 133.0, 132.5, 131.8, 131.0, 130.8, 130.1, 129.0, 128.6, 128.4, 128.3, 128.1, 127.2, 126.7, 126.6, 123.3, 120.7, 120.6, 120.4, 119.5, 80.2, 65.9, 48.2, 42.9, 29.7, 28.3 ppm. HRMS (ESI) exact mass calculated for [M + Na]⁺ (C₃₃H₃₅NO₄S₂ + Na⁺) m/z 596.1905, found m/z 596.1900.



Compound 15: 4-benzyl alcohol perylene (140 mg, 0.5 mmol) and Boc-Phen-OH (1.0 mmol) were dissolved in dry dichloromethane (DCM) (10 mL) in 0 °C. Next, the 4dimethylaminopyridine (DMAP, 0.2 mmol) and *N*, *N*'-dicyclohexylcarbodiimide (DCC, 0.6 mmol) were added to the mixture and then stirred for 12 hours. After the reaction, the solvents were removed under vacuum condition. The raw **15** was purified by silica gel flash column chromatography with DCM/n-hexane =1/1 (v/v) as elute. The shallow yellow band was collected and yield determined to be 65% (163 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 8.25-8.20 (m, 3H), 8.15-8.13 (d, 1H, *J* = 10.0 Hz), 7.76-7.70 (m, 3H), 7.55-7.48 (m, 4 H), 7.14 (s, 3H), 6.99 (s, 2H), 5.58-5.49 (m, 2H), 5.02-4.98 (m, 1H), 4.68-4.64 (m, 1H), 3.09-3.04 (m, 2H), 1.40 ppm (s, 9H). ¹³C NMR (125 MHz, CDCl₃) δ = 171.9, 155.1, 135.7,134.6, 133.0, 132.4,131.8, 131.0, 130.8, 130.2, 129.3, 129.0, 128.7, 128.5, 128.3, 128.1, 127.2, 126.9, 126.7, 126.6, 123.2, 120.7, 120.6, 120.4, 119.5, 79.9, 65.4, 54.5, 38.4, 30.3, 28.3 ppm. HRMS (ESI) exact mass calculated for [M + Na]+ (C₃₅H₃₁NO₄ + Na⁺) m/z 552.2151, found m/z 552.2145.



Compound 16: 4-benzyl alcohol perylene (140 mg, 0.5 mmol) and naproxen (0.7 mmol) were dissolved in dry dichloromethane (DCM) (10 mL) in 0 °C. Next, the 4-dimethylaminopyridine (DMAP, 0.2 mmol) and *N*, *N*'-dicyclohexylcarbodiimide (DCC, 0.6 mmol) were added to the mixture and then stirred for 12 hours. After the reaction, the solvents were removed under vacuum condition. The raw **16** was purified by silica gel flash column chromatography with DCM/n-hexane =1/1 (v/v) as elute. The shallow yellow band was collected and yield determined to be 57% (141 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 8.10-8.07 (m, 2H), 8.05-8.04 (d, 1H, *J* = 5.0Hz), 7.99-7.97 (d, 1H, *J* = 10.0Hz), 7.65-7.58 (m, 5H), 7.53-7.52 (d, 1H, *J* = 5.0Hz), 7.46-7.41 (m, 2H), 7.36-7.34 (m, 2H), 7.22-7.19 (m, 1H), 7.09-7.05 (m, 2H), 5.44 (s, 2H), 3.91-3.90 (m, 1H), 3.87 (s, 3H), 1.66-1.58 ppm (d, *J* = 10.0Hz, 3H). ¹³C NMR (125 MHz, CDCl₃) δ = 174.6,157.6, 135.4, 134.5, 133.7, 132.8, 132.0, 131.5, 131.1, 130.8, 130.7, 129.3, 128.9, 128.4, 128.1, 127.9, 127.1, 126.7, 126.5, 126.4, 126.2, 126.0, 123.3, 120.5, 120.4, 120.2, 119.4, 118.9, 105.5, 65.2, 55.3, 45.5, 18.3 ppm. HRMS (ESI) exact mass calculated for [M + Na]⁺ (C₃₅H₂₆O₃ + Na⁺) m/z 517.1780, found m/z 517.1774.



Compound 17: 4-benzyl alcohol perylene (140 mg, 0.5 mmol) and indometacin (0.7 mmol) were dissolved in dry dichloromethane (DCM) (10 mL) in 0 °C. Next, the 4-dimethylaminopyridine (DMAP, 0.2 mmol) and *N*, *N*'-dicyclohexylcarbodiimide (DCC, 0.6 mmol) were added to the mixture and then stirred for 12 hours. After the reaction, the solvents were removed under vacuum condition. The raw **17** was purified by silica gel flash column chromatography with DCM/n-hexane =1/1 (v/v) as elute. The shallow yellow band was collected and yield determined to be 53% (164 mg). 'H-NMR (500 MHz, CDCl₃): δ = 8.21-8.19 (m, 3H), 8.13-8.11 (d, 1H, *J* = 10.0Hz), 7.72-7.70 (m, 2H), 7.67-7.65 (d, 1H, *J* = 10.0Hz), 7.55-7.54 (m, 2H), 7.52-7.49 (m, 3H), 7.39-7.35 (m, 3H), 6.90-6.88 (m, 2H), 6.65-6.63 (m, 1H), 5.52 (s, 2H), 3.72 (s, 2H), 3.66 (s, 3H), 2.29 ppm (s, 3H). ¹³C NMR (125 MHz, CDCl₃) δ = 170.8, 168.3, 156.1, 139.2, 135.8, 134.6, 133.9, 132.9, 132.4, 131.8, 131.1, 131.0, 130.8, 130.7, 130.5, 129.1, 129.0, 128.6, 128.4, 128.3, 128.1, 127.0, 126.7, 126.6, 123.3, 120.7, 120.6, 120.4, 119.5, 115.0, 112.5, 111.9, 101.1, 65.4, 55.5, 30.6, 13.4 ppm. HRMS (ESI) exact mass calculated for [M + Na]⁺ (C₄₀H₂₈NO₄Cl + Na⁺) m/z 644.1605, found m/z 644.1609.



Compound 18: 4-benzyl alcohol perylene (140 mg, 0.5 mmol) and ibuprofen (0.7 mmol) were dissolved in dry dichloromethane (DCM) (10 mL) in 0 °C. Next, the 4-dimethylaminopyridine (DMAP, 0.2 mmol) and *N*, *N*'-dicyclohexylcarbodiimide (DCC, 0.6 mmol) were added to the mixture and then stirred for 12 hours. After the reaction, the solvents were removed under vacuum condition. The raw **18** was purified by silica gel flash column chromatography with DCM/n-hexane =1/1 (v/v) as elute. The shallow yellow band was collected and yield determined to be 67% (152 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 8.21-8.16 (m, 3H), 8.10-8.08 (m, 1H), 7.70-7.68 (d, 2H, *J* = 10.0Hz), 7.65-7.63 (d, 1H, *J* = 10.0Hz), 7.50-7.47(m, 2H), 7.44-7.41 (m, 2H), 7.21-7.19 (d, 2H, *J* = 10.0Hz), 7.10-7.01 (m, 4H), 5.48 (s, 2H), 2.45-2.43 (m, 2H), 1.86-1.82 (m, 1H), 1.53-

1.51 (d, 3H), 0.90 ppm (s, 6H). ¹³C NMR (125 MHz, CDCl₃) δ = 174.7, 140.6, 137.5,134.6, 132.8, 132.0, 131.7, 130.9, 129.5, 129.4, 129.3, 128.9, 128.5, 128.1, 128.0, 127.9, 127.3, 126.9, 126.6, 126.5, 120.4, 120.3, 119.5, 65.0, 45.2, 45.0, 30.2, 22.4, 18.3 ppm. HRMS (ESI) exact mass calculated for [M + Na]⁺ (C₃₄H₃₀O₂ + Na⁺) m/z 493.2143, found m/z 493.2138.



Compound 19: 4-benzyl alcohol perylene (140 mg, 0.5 mmol) and chlorambucil (0.7 mmol) were dissolved in dry dichloromethane (DCM) (10 mL) in 0 °C. Next, the 4-dimethylaminopyridine (DMAP, 0.2 mmol) and *N*, *N*'-dicyclohexylcarbodiimide (DCC, 0.6 mmol) were added to the mixture and then stirred for 12 hours. After the reaction, the solvents were removed under vacuum condition. The raw **19** was purified by silica gel flash column chromatography with DCM/n-hexane =1/1 (v/v) as elute. The shallow yellow band was collected and yield determined to be 60% (168 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 8.26-8.20 (m, 3H), 8.17-8.16 (1H, d, *J* = 5.0 Hz), 7.86-7.84 (1H, d, *J* = 10.0Hz), 7.71-7.70 (2H, d, *J* = 10.0Hz), 7.58-7.55 (m, 2H), 7.52-7.48 (m, 2H), 6.98-6.96 (2H, d, *J* = 10.0Hz), 6.56-6.54 (2H, d, *J* = 9.0 Hz), 5.52 (s, 2H), 3.67-3.64 (m, 4H), 3.59-3.56 (m, 4H), 2.53-2.50 (m, 2H), 2.40-2.37 (m, 2H), 1.95-1.89 (m, 2H). ¹³C NMR (125 MHz, CDCl₃) δ = 173.5, 144.3, 134.6, 133.0, 131.9, 131.1, 130.9, 130.5, 129.7, 129.1, 128.5, 128.4, 128.2, 128.1, 127.1, 126.7, 126.6, 123.4, 120.6, 120.5, 120.4, 119.6, 112.1, 64.5, 53.6, 40.5, 33.9, 33.7, 26.8 ppm. HRMS (ESI) exact mass calculated for [M + H]⁺ (C₃₅H₃₁Cl₂NO₂ + H⁺) m/z 568.1810, found m/z 568.1805.



Compound 20: 4-benzyl alcohol perylene (140 mg, 0.5 mmol) and pyridine (50 μ L) were dissolved in dry dichloromethane (DCM) (10 mL) in 0 °C. Next, the cholesteryl chloroformate. (0.6 mmol) were added to the mixture and then stirred for 12 hours. After the reaction, the solvents were removed under vacuum condition. The raw **20** was purified by silica gel flash column chromatography with DCM/n-hexane =1/1 (v/v) as elute. The shallow yellow band was collected and yield determined to be 21% (72 mg). ¹H-NMR (500 MHz, CDCl₃): δ = 8.25-8.20 (m, 3H), 8.16-8.15 (d, 1H, *J* = 5.0 Hz), 7.89-7.87 (d, 1H, *J* =10.0 Hz), 7.71-7.69 (m, 2H), 7.59-7.55 (m, 2H), 7.51-7.48 (m, 2H), 5.56 (s, 2H), 5.39-5.38 (d, 1H, *J* =5.0 Hz), 4.55-4.49 (m, 1H), 2.44-2.38 (m, 2H), 2.01-1.95 (m, 3H), 1.88-1.82 (m, 2H), 1.68-1.63 (m, 1H), 1.50-1.43 (m, 5H), 1.36-1.31 (m, 3H), 1.26-1.24 (m, 3H), 1.17-1.04 (m, 8H), 0.99 (s, 4H), 0.92-0.90 (m, 3H), 0.87-0.85 (m, 6H), 0.67 ppm (s, 3H). ¹³C NMR (125 MHz, CDCl₃): δ = 154.6, 139.4, 134.6, 133.0, 132.4, 131.8, 131.1, 130.9, 130.4, 129.0, 128.4, 128.2, 128.1, 127.2, 126.6, 126.5, 123.3, 123.0, 120.7, 120.6, 120.4, 119.6, 78.2, 67.7, 56.7, 56.1, 50.0, 42.3, 39.7, 39.5, 38.0, 36.9, 36.5, 36.2, 35.8, 31.9, 31.8, 28.2, 27.7, 24.3, 23.8, 22.8, 22.6, 21.0, 19.3, 18.7, 11.9 ppm. HRMS (ESI) exact mass calculated for [M + Na]+ (C₄₉H₅₈O₃ + Na⁺) m/z 717.4284, found m/z 717.4278.

Supplementary Tables

	$\lambda_{\mathrm{abs}}{}^{b}$	$\lambda_{ m em}{}^c$	$\xi (10^4 \mathrm{M}^{-1} \mathrm{cm}^{-1})_d$	$\begin{array}{c} \mathbf{T}_1 \\ (\mathbf{eV})^e \end{array}$	S ₁ (eV) ^{<i>f</i>}	$ au_{\mathrm{T}}(\mu s)^{g}$
PtTNP	687	866	13.2	1.43	p	11.0 ^h
PdTPBP	630	790	11.0	1.55	<i>p</i>	223.7^{i}
PtOEP	533	647	7.0	1.92	<i>p</i>	90 ^j
Ir(ppy)3	377	513	1.7	2.40	<i>p</i>	1.6^{k}
BDP (7)	517	537	7.3	1.49 ^{<i>i</i>}	2.31	p
Py (8)	440	445	4.0	1.52 m	2.78	<i>p</i>
An (2)	366	411	1.1	1.77 ⁿ	3.02	<i>p</i>
Cou (1)	360	460	0.62	2.18°	2.52	<i>p</i>

Supplementary Table 1. The photophysical parameters of Sens and PPGs^a.

^{*a*} in toluene, 10 μ M; ^{*b*} maximum absorption wavelength; ^{*c*} maximum emission wavelength; ^{*d*} molar extinction coefficient; ^{*e*} the energy level of triplet excited state (calculated with the maximum emission wavelength of phosphorescence); ^{*f*} the energy level of singlet excited state (calculated with the maximum emission wavelength of fluorescence); ^{*g*} triplet excited lifetime of Sen; ^{*h*} reference 3; ^{*i*} reference 4; ^{*j*} reference 5; ^{*k*} reference 6; ^{*l*} reference 7; ^{*m*} reference 8; ^{*n*} reference 9; ^{*o*} reference 10; ^{*p*} not applicable.

Sens:	PtTNP	PdTPBP	PtOEP	Ir(ppy) ₃	PPGs:	BDP	Ру	An	Cou
$S_1 (eV)$	—	-	-	-		2.31	2.78	3.02	2.52
$T_1 (eV)$	1.43	1.55	1.92	2.4		1.49	1.52	1.77	2.18
$2 \times T_1 - S_1$						0.69	0.26	0.53	1.85
(eV)									

Supplementary Table 2. A summary of the singlet excited/triplet excited state energy levels of PPGs and Sens

	PtTNP	PdTPBP	PtOEP	Ir(ppy) ₃
BDP			×	×
Ру	\checkmark	\checkmark	\checkmark	×
An	×	×	\checkmark	×
Cou	×	×	×	\checkmark

Supplementary Table 3. The combination of Sen and PPGs in this study.

" $\sqrt{}$ ": a TTAP pair; " \times ": not a TTAP pair.

Supplementary Table 4. The photolytic reaction for Ir(ppy)₃ and Cou (compound 1).



Entry	Condition	Reaction time	Yield (%) ^{<i>a</i>}
1	light; air	20 min	4
2	dark, argon;	20 min	No reaction
3	light, Cou (compound 1) only, argon;	20 min	No reaction
4	light; argon;	20 min	73.6

^{*a*} Measured by HPLC

The photolysis yield of Cou (compound 1) at some condition without $Ir(ppy)_3$ under 365 nm illumination was measured. The photolytic yield is 79.8 %

	PtOEP	HO) + но
Entry	Condition	Reaction time	Yield (%) ^{<i>a</i>}
1	Light, air	1 h	Trace
2	Dark, argon	1 h	No reaction
3	An (compound 2) only, light,	1 h	No reaction
	argon		
4	Light, argon	1 h	83.2

Supplementary Table 5. The photolytic reaction for PtOEP and An (compound 2).

^a Measured by HPLC.

Supplementary Table 6. The Photolytic reaction for PtOEP and Py (compound 8).

	PtOEP LED 530 nm	HO	+ но
Entry	Condition	Reaction time	Yield (%) ^{<i>a</i>}
1	Light, air	1 h	2
2	Dark, argon	1 h	No reaction
3	Py (compound 8) only, light,	1 h	No reaction
	argon		
4	Light, argon	1 h	70.2
5	Light, argon, 520 nm long pass	1 h	61.3
	filter		

^{*a*} Measured by HPLC.

PdTPBP LED 650 nm	HO	+ но
Condition	Reaction time	Yield (%) <i>a</i>
Light, air	1 h	No reaction
Dark, argon	1 h	No reaction
Py (compound 8) only, light,		No reaction
argon		
Light, argon	1 h	87.1
	PdTPBP LED 650 nm Condition Light, air Dark, argon npound 8) only, light, argon Light, argon	PdTPBP LED 650 nm HO $\downarrow \downarrow \downarrow$ ConditionReaction timeLight, air1 hDark, argon1 hnpound 8) only, light,1 hargon1 hLight, argon1 h

Supplementary Table 7. The photolytic reaction for PdTPBP: Py (compound 8).

^{*a*} Measured by HPLC.

Supplementary Table 8. The photolytic reaction for PdTPBP: BDP (compound 7).

	PdTPBP B ^N F		
Entry	Condition	Reaction time	Yield (%) <i>a</i>
1	Light, air	1 h	No reaction
2	Dark, argon	1 h	No reaction
3	BDP (compound 7) only, light,	1 h	No reaction
	argon		
4	Light, argon	1 h	76.0

^{*a*} Measured by HPLC.

	PtTNP LED720 nm	HO	+ но
Entry	Condition	Reaction time	Yield (%) ^{<i>a</i>}
1	Light, air	1 h	No reaction
2	Dark, argon	1 h	No reaction
3	Py (compound 8) only, light,	1 h	No reaction
	argon		
4	Light, argon	1 h	14.7
5	Light, argon	2 h	24.3 %

Supplementary Table 9. The photolytic reaction for PtTNP: Py (compound 8).

^{*a*} Measured by HPLC.

Supplementary Table 10. The photolytic reaction for PtTNP: BDP (compound 7).

	PtTNP B ^N _F 720 nm LED		
Entry	Condition	Reaction time	Yield (%) ^{<i>a</i>}
1	Light, air	1 h	No reaction
2	Dark, argon	1 h	No reaction
3	BDP (compound 7) only, light,	1 h	No reaction
	argon		
4	Light, argon	1 h	17.7

^{*a*} Measured by HPLC.

Supplementary Table 11. The photolytic reactions for Py (compound **8**) at room light and blue light illumination.

	$THF / H_2O = 4/1, V$	HO HO	+ но [↓]
Entry	Condition	Reaction time	Yield (%) ^a
1	Py (compound 8) only, air, dark	60 min	No reaction
2	Py (compound 8) only, air, room light	60 min	No reaction
3	PdTPBP, Py (compound 8), air, room	60 min	No reaction
	light		
4	PdTPBP, Py (compound 8), argon, room	60 min	No reaction
	light		
5	Py (compound 8) only, air, blue light	60 min	59.0
	(445 nm)		
^a HPLC	yield.		

Supplementary	Table 12	2. The	photocleavage	quantum	yield (Φ_p) and	photolytic	efficiency	of
TTAP ($\varepsilon_2 \times \Phi_p \times \Phi_p$	$\Phi_{\text{TTET}} imes$	Φ_{TTA}).							

	$\mathbf{\Phi}_{\mathbf{p}}$ a	$\boldsymbol{\epsilon}_{2} \times \boldsymbol{\Phi}_{p} \times \boldsymbol{\Phi}_{\text{TTET}} \times \boldsymbol{\Phi}_{\text{TTA}}$
		TTAP (($\eta_{(TTAP)}$)
BDP (compound 7)	9.9×10 ^{-4 b}	$12.3^{c}/0.53^{d}$
Py (compound 8)	0.093 ^e	7.1^{d}
		679 ^c
		708 ^f
An (compound 2)	0.089 ^g	1255 ^f
Cou (compound 1)	0.13 ^{<i>h</i>}	_

^{*a*} photocleavage quantum yield of PPGs; ^{*b*} reference 11; ^{*c*} PdTPBP as sensitizer; ^{*d*} PtTNP as photosensitizer; ^{*e*} reference 12; ^{*f*} PtOEP as sensitizer; ^{*g*} reference 13; ^{*h*} reference 14.

Supplementary Table 13. Stern–Volmer quenching constant, bimolecular quenching constant of different TTAP pairs and triplet excited lifetime of photosensitizer (τ_T).

Pair:	PdTPBP:	PdTPBP:Py PtOEP: Py		PtOEP:An	Ir(ppy) ₃ : Cou	
	BDP					
k_{sv} (M -1) ^a	2.40×10^{5}	2.43×10^{5}	2.62×10^{5}	2.19× 10 ⁵	3.13×10 ³	
k _q (M -1 S -1) ^b	1.08× 10 ⁹	1.09×10^{9}	0^{9} 2.9 × 10 ⁹ 2.43 × 10		2.34 × 10 ⁹	
Photosensitizer	PdTPBP	PtOEP		Ir(ppy) ₃		
$\tau_{\rm T}(\mu s)$	222	91		1.5		

^{*a*} Stern–Volmer quenching constant; ^{*b*} bimolecular quenching constant, $k_{sv} = k_q \tau_T$. BDP (compound

7), Py (compound 8), An (compound 2) and Cou (compound 1)

Supplementary Table 14. The photolytic reaction for PdTPBP: compound **19** under far red LED irradiation.

CI	TTA-mediated photocleavage CI PdTPBP CI	HO OH + Chlorambucil	
Entry	Condition	Reaction time (h)	Yield ^a
1	Compound 19 only; red light;	2	None
2	Compound 19 only; argon; red	2	None
	light, 20 mW/cm ²		
3	PdTPBP, compound 19; argon; red	0.5	47
	light, 20 mW/cm ²		
4	PdTPBP, compound 19; argon; red	1	26
	light, 10 mW/cm ²		
5	PdTPBP, compound 19; argon; red	1	72.8
	light, 20 mW/cm ²		
6	PdTPBP, compound 19; argon; red	2	67.0
	light, 20 mW/cm ²		
7	PdTPBP, compound 19; air, red	2	7
	light, 20 mW/cm ²		
8	PdTPBP, compound 19; argon,	2	None
	dark;		

^a HPLC yield.

Supplementary Table 15. The photolytic reaction for PtTNP: pro-chlorambucil (compound **19**) under 720 nm LED irradiation.

CI CI	19	HO OH + Chlorambucil	
Entry	Condition	Reaction time (h)	Yield ^a
1	Compound 19 only; 720 nm LED;	1	No reaction
2	Compound 19 only; argon; 720 nm	1	No reaction
	LED;		
3	PtTNP, compound 19; argon; 720 nm	1	30.4
	LED;		
4	PtTNP, compound 19; air, 720 nm	1	No reaction
	LED;		
6	PtTNP, compound 19 ; argon, dark;	1	No reaction
^a HPLC	yield.		

	Normal range	Control	3 h	6 h	12 h	24 h	48 h	96 h
WBC (K/µL)	1.8-10.7	6.42	9.78	10.08	8.90	6.22	6.04	6.4
NE (K/ μ L)	0.1-2.4	2.29	2.00	2.07	1.46	2.27	1.6	0.12
LY (K/ μ L)	0.9-9.3	3.59	6.55	6.70	4.82	3.38	4.08	4.0
MO (K/ μ L)	0.0-0.4	0.39	0.16	0.28	0.38	0.28	0.27	0.02
EO (K/ μ L)	0.0-0.2	0.11	0.05	0.02	0.18	0.08	0.07	0.01
BA (K/ μ L)	0.0-0.2	0.02	0.01	0.01	0.06	0.01	0.01	0.02
RBC (M/ μ L)	6.36-9.42	8.83	9.31	8.90	8.95	9.36	9.07	8.76
HGB (g/dL)	11.0-15.1	13.3	13.5	13.7	13.6	13.6	12.8	13.1
HCT (%)	35.1-45.4	44.1	40.6	45.1	43.7	41.3	38.7	36.3
MCV (fL)	45.4-60.3	55.0	54.4	55.7	54.0	54.8	53.7	53.8
MCH (pg)	14.1-19.3	14.5	14.5	16.8	17.7	14.5	14.1	14.6
MCHC (K/ μ L)	30.2-34.2	31.6	31.7	32.9	33.3	32.5	32.2	30.2
RDW (K/ μ L)	12.4-27.0	16.8	18.3	17.2	17.7	16.8	17.5	17.4
PLT (K/ μ L)	592-2972	630	669	649	743	605	671	778
MPV (fL)	5.0-20.0	5.6	6.1	6.2	5.3	5.6	5.7	6.2

Supplementary Table 16. Blood biochemistry and complete blood panel analysis of mice.^a

^a The Balb/c mice were *i.v.*-injected with TTAP NPs (100 μ g mL⁻¹, 150 μ L) and then sacrificed at the various time points (3, 6, 12, 24, 48, 96 hour) after injection. Another four Balb/c mice were used as the control. Before the mice were euthanatized, blood samples (~ 0.5 mL) were collected for blood panel analyses and blood chemistry tests. White blood cell (WBC), Neutrophils (NE), Lymphocytes (LY), Monocytes (MO), Eosinophils (EO), Basophils (BA), Red blood cell (RBC), Hemoglobin (HGB), Hematocrit (HCT), Mean corpuscular volume (MCV), mean corpuscular hemoglobin (MCH), Mean corpuscular hemoglobin concentration (MCHC), Red blood cell distribution width (RDW), Platelet Thrombocyte (PLT), Mean platelet volume (MPV).

Supplementary Figures



Supplementary Figure 1. (a) Normalized photoluminescence emission spectra of Ir(ppy)₃ (λ_{ex} = 410 nm), PtOEP (λ_{ex} = 530 nm), PdTPBP (λ_{ex} = 630 nm) and PtTNP (λ_{ex} = 700 nm); (b) Normalized photoluminescence emission spectra of Cou (compound **1**, λ_{ex} = 360 nm), An (compound **2**, λ_{ex} = 360 nm), Py (compound **8**, λ_{ex} = 410 nm) and BDP (compound **7**, λ_{ex} = 470 nm).



Supplementary Figure 2. A plot comparison of triple excited state and single excited state of Sens and PPGs.



Supplementary Figure 3. (a) Molecular structures of $Ir(ppy)_3$ and Cou (compound 1); (b) phosphorescence intensity quenching of $Ir(ppy)_3$ (10 μ M) in the presence of different concentration of Cou (compound 1), $\lambda_{ex} = 410$ nm; (c) Stern-Volmer plots generated from the phosphorescence intensity quenching of $Ir(ppy)_3$ as a function of Cou (compound 1) concentration.



Supplementary Figure 4. Normalized UV-vis absorption and emission spectra of Cou (compound 1). The fluorescence emission wavelength of compound 1 is very broad from 420 nm to 700 nm (peak at 498 nm), overlapping the excitation wavelength used for $Ir(ppy)_3$ photosensitizer (476 nm). Therefore, in this case, it is unlikely to separate excitation light from our upconverted luminescence to obtain upconversion spectra.



Supplementary Figure 5. (a) Molecular structures of PtOEP and An (compound 2); (b) phosphorescence intensity quenching of PtOEP (10 μ M) in the presence of different concentration of An (compound 2), $\lambda_{ex} = 530$ nm; (c) Stern-Volmer plots generated from the phosphorescence

intensity quenching of PtOEP as a function of An (compound 2) concentration. In toluene.



Supplementary Figure 6. Upconversion emission spectra of PtOEP (10 μ M) and An (compound 2) (500 μ M). long pass filter (>520 nm) was used to cut off the short-wavelength, $\lambda_{ex} = 530$ nm, 20 mW cm⁻², in toluene.



Supplementary Figure 7. (a) Molecular structures of PtOEP and Py (compound 8); (b) phosphorescence intensity quenching of PtOEP (10 μ M) in the presence of different concentration of Py (compound 8), $\lambda_{ex} = 530$ nm; (c) Stern-Volmer plots generated from the phosphorescence intensity quenching of PtOEP as a function of Py (compound 8) concentration, in toluene.



Supplementary Figure 8. Upconversion emission spectra of PtOEP (10 μ M) and Py (compound 8) (500 μ M) in air and in argon. Long pass filter (>520 nm) was used to cut off the short-wavelength, in toluene, $\lambda_{ex} = 530$ nm, 20 mW cm⁻².



Supplementary Figure 9. (a) Molecular structures of PdTPBP and Py (compound 8); (b) phosphorescence intensity quenching of PdTPBP (10 μ M) in the presence of different concentration of Py (compound 8), $\lambda_{ex} = 635$ nm; (c) Stern-Volmer plot generated from the phosphorescence intensity quenching of PdTPBP as a function of Py (compound 8) concentration; (d) the upconversion emission spectra of PdTPBP and Py (compound 8) paired in the air, in argon and in the absence of PdTPBP, $\lambda_{ex} = 650$ nm, 100 mW cm⁻², in toluene.



Supplementary Figure 10. (a) Molecular structures of PdTPBP and BDP (compound 7); (b) phosphorescence intensity quenching of PdTPBP (10 μ M) in the presence of different concentration of BDP (compound 7), $\lambda_{ex} = 635$ nm; (c) Stern-Volmer plot generated from the phosphorescence intensity quenching of PdTPBP as a function of BDP (compound 7) concentration; (d) the upconversion emission spectra of PdTPBP and BDP (compound 7) paired in the air, in argon and in the absence of PdTPBP, $\lambda_{ex} = 650$ nm, 100 mW cm⁻², in toluene.



Supplementary Figure 11. (a) Molecular structures of PtTNP, Py (compound 8) and BDP (compound 7); (b) phosphorescence intensity quenching of PtTNP (10 μ M) in the presence of different concentration of Py (compound 8), $\lambda_{ex} = 700$ nm; (c) The upconversion emission spectra of PtTNP and Py (compound 8) in the different concentration of Py (compound 8), $\lambda_{ex} = 720$ nm, 100 mW cm⁻², in toluene; (d) phosphorescence intensity quenching of PtTNP (10 μ M) in the presence of different concentration of BDP (compound 7), $\lambda_{ex} = 700$ nm; (e) the upconversion emission spectra of PtTNP and BDP (compound 7) in the different concentration of BDP (compound 7), $\lambda_{ex} = 720$ nm, 100 mW cm⁻², in toluene.



Supplementary Figure 12. (a) The upconversion emission spectra of the TTA-UC system of PtOEP (10 μ M) and An (compound **2**) (500 μ M) under argon with various excitation intensities ($\lambda_{ex} = 530$ nm). (b) The dependence of the TTA-UC intensity of the solution of PtOEP and An (compound **2**) on the various incident power density. The results fit the lines that have slopes of 2.23 (black, below) and 0.85 (red, above) in the low- and high-power regions, respectively. *I*_{th} is 25.4 mW cm⁻². (c) Upconversion emission spectra of the TTA-UC system of PtOEP (10 μ M) and Py (compound **8**) (500 μ M) under argon with various excitation intensities ($\lambda_{ex} = 530$ nm). (d) The dependence of the TTA-UC intensity of the solution of PtOEP and Py (compound **8**) on the various incident power regions, respectively. *I*_{th} is 25.4 mW cm⁻². The results fit the lines that have slopes of 1.89 (black, below) and 1.06 (red, above) in the low- and high-power regions, respectively. *I*_{th} is 21.8 mW cm⁻².


Supplementary Figure 13. (a) Upconversion emission spectra of the TTA-UC system of PdTPBP (10 μ M) and Py (compound **8**) (500 μ M) under argon with various excitation intensities ($\lambda_{ex} = 650$ nm). (b) The dependence of the TTA-UC intensity of the solution of PdTPBP and Py (compound **8**) on the various incident power density. The results fit the lines that have slopes of 2.05 (black, below) and 1.04 (red, above) in the low- and high-power regions, respectively. *I*_{th} is 48.1 mW cm⁻². (c) Upconversion emission spectra of the TTA-UC system of PdTPBP (10 μ M) and BDP (compound **7**) (500 μ M) under argon with various excitation intensities ($\lambda_{ex} = 650$ nm). (d) The dependence of the TTA-UC intensity of the solution of PdTPBP and BDP (compound **7**) on the various incident power density. The results fit the lines that have slopes of 1.89 (black, below) and 1.17 (red, above) in the low- and high-power regions, respectively. *I*_{th} is 46.7 mW cm⁻².



Supplementary Figure 14. (a) Upconversion emission spectra of the TTA-UC system of PtTNP (10 μ M) and Py (compound **8**) (10 mM) under argon with various excitation intensities ($\lambda_{ex} = 650$ nm). (b) The dependence of the TTA-UC intensity of the solution of PtTNP and Py (compound **8**) on the various incident power density. The results fit the lines that have slopes of 1.93 (black, below) and 1.13 (red, above) in the low- and high-power regions, respectively, I_{th} is 94.5 mW cm⁻². (c) Upconversion emission spectra of the TTA-UC system of PtTNP (10 μ M) and BDP (compound **7**) (10 mM) under argon with various excitation intensities ($\lambda_{ex} = 650$ nm). (d) The dependence of the TTA-UC intensity of the solution of PtTNP and BDP (compound **7**) on the various incident power density. The results fit the lines that have slopes of 1.82 (black, below) and 1.21 (red, above) in the low- and high-power regions, respectively, I_{th} is 119.8 mW cm⁻².



Supplementary Figure 15. Upconversion emission spectra of (a) PdTPBP (10 μ M) and Py (compound 8) (500 μ M), (b) PdTPBP (10 μ M) and BDP (compound 7) (500 μ M), (c) PtTNP (10 μ M) and Py (compound 8) (10 mM), (d) PtTNP (10 μ M) and BDP (compound 7) (10 mM), $\lambda_{ex} = 650$ nm, 20 mW cm⁻², in toluene.



Supplementary Figure 16. (a) Upper: A schematic diagram of the mechanism of PSMA-PEG-OAm scavenge oxygen and molecular structure of PSMA-PEG-OAm. Bottom: ¹H NMR monitor PSMA-PEG-OAm reacting with singlet oxygen. Black star is double bond, red star is products after double bond oxidation. (b) Normalized phosphorescence spectra of PdTPBP in toluene, in air and in argon, $\lambda_{ex} = 630$ nm. (c) Normalized phosphorescence spectra of PdTPBP NPs in PBS

buffer, in air and in argon, $\lambda_{ex} = 630$ nm.For the polymer of PSMA-PEG-OAm, it is noteworthy that the hydrophilic properties of PEG and carboxylic acids of the polymer enhance the biocompatibility of the nanoparticle. The unsaturated alkene sidechains of such amphiphilic polymers act as active decoys to react with and exhaust the oxygen from the external environment. We determined the molecular structure change of PSMA-PEG-OAm under light irradiation by nuclear magnetic resonance (NMR). Along with the illumination, the photosensitizer converts dioxygen into singlet oxygen, which reacts with unsaturated double bonds, thus generating new peaks at 5.75 ppm (a).. Moreover, since phosphorescence is extremely sensitive to di-oxygen, the phosphorescence of PdTPBP is further used to validate the oxygen scavenger role of PAA-OAm (b). The phosphorescence of PSMA-PEG-OAm encased PdTPBP nanoparticles (PdTPBP NPs) was measured in air and argon (c). The phosphorescence intensity in air was found to be quite similar to that under argon. These results demonstrated that PSMA-PEG-OAm can provide a hypoxia microenvironment to reduce oxygen quenching of the triplet state of PdTPBP.



Supplementary Figure 17. (a) TEM imaging of TTAP NPs, scar bar is 100 nm; (b) size distribution of TTAP NPs; (c) hydrodynamic diameter of TTAP NPs in PBS via DLS; (d) colloidal storage stability of TTAP NPs in PBS.



Supplementary Figure 18. UV-vis absorption spectra of TTAP NPs in PBS buffer. The absorption wavelength at 418, 441 nm comes from prodrug (compound **19**), and 627 nm come from PdTPBP. The result suggests TTAP NPs contain both PdTPBP and prodrug.



Supplementary Figure 19. UV-vis absorption spectra of TTAP NPs under 650 nm LED (20 mW cm⁻²) illumination. monitor time (0, 5, 10, 15, 20, 25, 30 min). We did not observe obvious photobleaching, suggesting the TTAP NPs' robust photostability.



Supplementary Figure 20. Monitoring the photoactivation chlorambucil releasing process of TTAP NPs with the fluorescence spectra of prodrug (compound **19**). (a) Blue light mediated photolytic reaction (455 nm LED, 20 mW cm⁻²) for Py NPs; (b) Far red light mediated photolytic reaction (650 nm LED, 20 mW cm⁻²) for Py NPs; (c) Far red light mediated photolytic reaction (650 nm laser, 100 mW cm⁻²) for TTAP NPs; (d) Far red light mediated photolytic reaction (650 nm LED, 20 mW cm⁻²) for TTAP NPs; (d) Far red light mediated photolytic reaction (650 nm LED, 20 mW cm⁻²) for TTAP NPs; (d) Far red light mediated photolytic reaction (650 nm LED, 20 mW cm⁻²) for TTAP NPs.As shown in b, A similar nanoparticle consisting of the prodrug (compound **19**) without the PdTPBP (Py NPs) was prepared as a stringent control. Upon 650 nm illumination of Py NPs for 60 min, no observable uncaged chlorambucil was detected, excluding the possibility of the direct uncaging of chlorambucil by 650 nm light. However, after uncaging via long wavelength light for TTAP NPs, after uncaging, the prodrug (compound **19**) is able to convert into hydrophilic chlorambucil, which can be released from the nanoparticles and kill tumor cells. When we illuminated TTAP NPs with a 650 nm LED(20 mW cm⁻²), we found

that the prodrug was uncaged, resulting in greater than 48% activation of the prodrug within 30 min, and a maximum photo-release of approximately 64% of the prodrug after 60 min (c-d, and supplementary Figure 21a).



Supplementary Figure 21. (a) The release profiles of chlorambucil from TTAP NPs in PBS buffer with 650 nm LED irradiation, 20 mW cm⁻². (b) The photolytic process for prodrug (compound **19**) from TTAP NPs with 650 nm LED irradiation. "ON" and "OFF" indicate the initiation and termination of LED irradiation, respectively; the LED (650 nm, 20 mW cm⁻²).



Supplementary Figure 22. (a) MTT assay of Hela cells viability of different concentrations of TTAP NPs and TTAP NPs + hv. (b) 4T1 cell viability in different concentrations of TTAP NPs with and without light irradiation. The value for IC₅₀ with respect to the Hela and 4T1 cells in the *in vitro* experiment: 17.3 ± 3.1 mg mL⁻¹ (Hela cells) and 20.6 ± 2.6 mg mL⁻¹ (4T1).



Supplementary Figure 23. Hela cell viability in different concentrations of PdTPBP NPs with and without light irradiation.



Supplementary Figure 24. Synergistic TTAP NPs and α -PD-L1 for inhibition of 4T1 tumor growth *in vivo*. (a-f) individual tumor growth kinetics in control and treated groups. In group 1 (PBS + *hv*), we did not observe tumor growth inhibition; after 15 days, the tumor volume increased 7.5-fold. This result suggested that low power intensity far red light did not have obvious photothermal effects in regard to cancer therapy. In group 2 (PdTPBP NPs + *hv*), the tumor volume also increased 6.5-fold and we did not observe significant tumor growth inhibition by PdTPBP NPs. This is likely due to the presence of an unsaturated bond on these nanoparticles that can exhaust singlet oxygen to reduce the possible photodynamic effect. In group 4 (TTAP NPs only), the tumor volume also increased 7.0-fold, suggesting a low inherent toxicity of TTAP NPs in the dark. However, in group 5 (TTAP NPs + *hv*), the tumor volume only increased 3.9-fold. These results demonstrated that the anticancer drug was efficiently photolyzed and released for tumor cell killing. In group 3, the injection of α -PD-L1 also led to tumor growth inhibition; a result that is consistent with previously reported results. In contrast, in group 6, the tumor showed a slow

growth rate from days 1 to 6, but post α -PD-L1 injection at 7th, 8th and 9th day, the tumor volume was remarkably suppressed. More significantly, we also measured the left tumor growth rates. In groups 1, 2, and 4, we did not observe tumor inhibition. In group 3, the tumor presented a slower tumor growth rate than seen in group 1, as α -PD-L1 can mitigate the metastasis. However, compared to group 5, group 6 presented significant tumor suppression (Figure 3b). We also measured tumor weight after 15 days' treatments. In group 6, both the right and left tumor showed obvious reduction (Figure 3d). These experimental results clearly demonstrated that the TTAP mediated prodrug photolytic system improves the checkpoint blockade immunotherapy efficacy and promotes abscopal effects.



Supplementary Figure 25. Tumor-specific immune responses. (a-b) Splenocytes were stimulated for 4 h with PMA/ionomycin, the frequency of IFN- γ – producing CD4⁺ and CD8⁺ T cells was measured by flow cytometry. (c-d) The percentage of tumor-infiltrating (c) CD4⁺ T cells, (d) CD8⁺ T cells in the primary tumor. (e-f) The percentage of tumor-infiltrating (e) CD4⁺ T cells, (f) CD8⁺ T cells in the distant tumor. Data are expressed as means ± s.d. (n = 5). *p < 0.05, **p < 0.01, and ***p < 0.001.



Supplementary Figure 26. The percentage of tumor-infiltrating (a) CD4⁺ T cells or (b) CD8⁺ T cells in the spleen. Data are expressed as means \pm s.d. (n = 5). *P < 0.05, **P < 0.01, and ***P < 0.001.



Supplementary Figure 27. The percentage of tumor-infiltrating (a) B cells, (b) NK T cells in the primary tumor, and (c) B cells, (d) NK T cells in the distant tumor. Data are expressed as means \pm s.d. (n = 5).



Supplementary Figure 28. The mice body weight change of anti-tumor immunotherapy model.



Supplementary Figure 29. H&E staining images of major organs (heart, liver, spleen, lung and kidney) of 4T1 tumor bearing mice after injection of TTAP NPs for 14 days, scale bar represents 100 μm.



Supplementary Figure 30. IVS imaging of the mice feces for i.v. injection of PBS (control) and i.v. injection of IR806-TTAP NPs.



Supplementary Figure 31. In vivo distributions and biosafety of IR806-RGD-TTAP-NPs. (a)

Time dependent ex vivo NIR fluorescence in major mice organs (heart, liver, spleen, lung, kidney, tumor) including 10, 24, 48, 96 and 168 hours. (b) Normalized fluorescence intensity in mice organs, Values are the means \pm s.e.m. (n = 3 mice per group). We can see that ten hours after treatment with the IR806-RGD-TTAP NPs by intravenous injection, the tumor and organs, including the brain, heart, liver, spleen, lung, and kidney were excised from tumor-bearing mice and imaged ex vivo. An intense NIR fluorescence was seen in the excised tumor, which indicates efficient IR806-RGD-TTAP NPs accumulation in the tumor. After 48 h, the fluorescence from the liver and the tumor had reduced. By 96 h, the fluorescence of the liver and the tumor had significantly weakened. We quantified the fluorescence intensity of the tumor and each organ from the IR806-RGD-TTAP NPs treated group at different time-points. The fluorescence intensity of the tumor was higher than that in organs at 10 h, except in the liver. However, after 48 h, the fluorescence from the liver was reduced by 50%, but that from the tumor only reduced by 10%. At 96 h, the fluorescence from the liver could hardly be observed. A possible reason for this result might be that IR806-RGD-TTAP NPs is mainly eliminated by macrophage cells in the liver. Meanwhile, the fluorescence was observed at kidney at 10, 24, 48 hours and disappeared at 96 and 168 hours, suggesting the RGD-TTAP NPs may be excluded from the body through kidney.



Supplementary Figure 32. (a) Schematic illustration of TTAP NPs mediated targeted tumor therapy. (b) Specific-targeted IR806 fluorescence tumor imaging *in vivo*. (c) Normalized fluorescence intensity of IR806-RGD-TTAP NPs and IR806-TTAP NPs in 4T1 tumors at 10, 24, 48 and 96 h. Data are means \pm s.e.m. (n = 3 mice). (d) Tumor growth inhibition by TTAP NPs

mediated photoactivatable prodrug release in 4T1 tumors; light irradiation was performed 24 h after injection of TTAP NPs. Values are means \pm s.e.m. (n = 5 mice per group). (e) The mice body weight change of targeted tumor therapy. (d) H&E staining of tumor tissue sections from different treatment groups 10 days after treatment; scale bar represents 50 µm. No tumor growth inhibition or tumor tissue necrosis was observed in the group of mice with LED irradiation only (*hv* only), indicating that irradiation with such low power intensity red light had little photothermal effect. No tumor inhibition was observed for the group with injection of only RGD-TTAP NPs, which showed that RGD-TTAP NPs has negligible dark toxicity. Group TTAP NPs plus light illumination also did not display an obvious therapeutic effect, suggesting that cRGD was important for targeting tumor tissue. PdTPBP NPs plus light illumination also did not show obvious tumor inhibition effects, which suggests that PdTPBP has less photo-toxicity. In contrast, the tumor growth in the treatment RGD-TTAP NPs plus light illumination was remarkably suppressed. H&E staining analysis of the tumor tissue also showed clear necrosis, which indicates that RGD-TTAP NPs NPs can be effectively activated by red LED irradiation for intense anticancer effect.



Supplementary Figure 33. ¹H NMR of 1 (500 MHz, CDCl₃).



Supplementary Figure 34. ¹³C NMR of 1 (125 MHz, CDCl₃).



Supplementary Figure 35. ¹H NMR of 2 (500 MHz, CDCl₃).



Supplementary Figure 36. ¹³C NMR of 2 (125 MHz, CDCl₃).



Supplementary Figure 37. ¹H NMR of 3 (500 MHz, CDCl₃).



Supplementary Figure 38. ¹³C NMR of 3 (125 MHz, CDCl₃).



Supplementary Figure 39. ¹H NMR of 4 (500 MHz, CDCl₃).



Supplementary Figure 40. ¹³C NMR of 4 (125 MHz, CDCl₃).



Supplementary Figure 41. ¹H NMR of 5 (500 MHz, CDCl₃).



Supplementary Figure 42. ¹³C NMR of 5 (125 MHz, CDCl₃).



Supplementary Figure 43. ¹H NMR of 6 (500 MHz, CDCl₃).



Supplementary Figure 44. ¹³C NMR of 6 (125 MHz, CDCl₃).



Supplementary Figure 45. ¹H NMR of 7 (500 MHz, CDCl₃).



Supplementary Figure 46. ¹³C NMR of 7 (125 MHz, CDCl₃).



Supplementary Figure 47. ¹H NMR of 8 (500 MHz, CDCl₃).



Supplementary Figure 48. ¹³C NMR of 8 (125 MHz, CDCl₃).



Supplementary Figure 49. ¹H NMR of 9 (500 MHz, CDCl₃).



Supplementary Figure 50. ¹³C NMR of 9 (125 MHz, CDCl₃).



Supplementary Figure 51. ¹H NMR of 10 (500 MHz, CDCl₃).



Supplementary Figure 52. ¹³C NMR of 10 (125 MHz, CDCl₃).



Supplementary Figure 53. ¹H NMR of 11 (500 MHz, CDCl₃).



Supplementary Figure 54. ¹³C NMR of 11 (125 MHz, CDCl₃).



Supplementary Figure 55. ¹H NMR of 12 (500 MHz, CDCl₃).



Supplementary Figure 56. ¹³C NMR of 12 (125 MHz, CDCl₃).



Supplementary Figure 57. ¹H NMR of 13 (500 MHz, CDCl₃).



Supplementary Figure 58. ¹³C NMR of 13 (125 MHz, CDCl₃).



Supplementary Figure 59. ¹H NMR of 14 (500 MHz, CDCl₃).



Supplementary Figure 60. ¹³C NMR of 14 (125 MHz, CDCl₃).



Supplementary Figure 61. ¹H NMR of 15 (500 MHz, CDCl₃).



Supplementary Figure 62. ¹³C NMR of 15 (125 MHz, CDCl₃).



Supplementary Figure 63. ¹H NMR of 16 (500 MHz, CDCl₃).



Supplementary Figure 64. ¹³C NMR of 16 (125 MHz, CDCl₃).



Supplementary Figure 65. ¹H NMR of 17 (500 MHz, CDCl₃).



Supplementary Figure 66. ¹³C NMR of 17 (125 MHz, CDCl₃).


Supplementary Figure 67. ¹H NMR of 18 (500 MHz, CDCl₃).



Supplementary Figure 68. ¹³C NMR of 18 (125 MHz, CDCl₃).



Supplementary Figure 69. ¹H NMR of 19 (500 MHz, CDCl₃).



Supplementary Figure 70. ¹³C NMR of 19 (125 MHz, CDCl₃).



Supplementary Figure 71. ¹H NMR of 20 (500 MHz, CDCl₃).



Supplementary Figure 72. ¹³C NMR of 20 (125 MHz, CDCl₃).



Supplementary Figure 73. HRMS (ESI) of 1.



Supplementary Figure 74. HRMS (ESI) of 2.



Supplementary Figure 75. HRMS (ESI) of 3.



Supplementary Figure 76. HRMS (ESI) of 4.



Supplementary Figure 77. HRMS (ESI) of 5.



Supplementary Figure 78. HRMS (ESI) of 6.



Supplementary Figure 79. HRMS (ESI) of 7.



Supplementary Figure 80. HRMS (ESI) of 8.



Supplementary Figure 81. HRMS (ESI) of 9.



Supplementary Figure 82. HRMS (ESI) of 10.



Supplementary Figure 83. HRMS (ESI) of 11.



Supplementary Figure 84. HRMS (ESI) of 12.



Supplementary Figure 85. HRMS (ESI) of 13.



Supplementary Figure 86. HRMS (ESI) of 14.



Supplementary Figure 87. HRMS (ESI) of 15.



Supplementary Figure 88. HRMS (ESI) of 16.



Supplementary Figure 89. HRMS (ESI) of 17.



Supplementary Figure 90. HRMS (ESI) of 18.



Supplementary Figure 91. HRMS (ESI) of 19.



Supplementary Figure 92. HRMS (ESI) of 20.

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