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Supporting Information

Synthesis of Unprecedented 4d/4f-Polypnictogens

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General Methods:

All manipulations of water- and air-sensitive compounds were performed with exclusion of moisture and oxygen in flame-dried Schlenk-type glassware either on a dual manifold Schlenk line, interfaced to a high vacuum (10^{-3} mbar) line or in an argon-filled MBraun glove box.

Tetrahydrofuran was distilled under nitrogen from potassium benzophenoneketyl before storage in *vacuo* over LiAlH₄. Hydrocarbon solvents were dried by using an MBraun solvent purification system (SPS 800), degassed and stored *in vacuo* over LiAlH₄. Elemental analysis were carried out with an Elementar Vario Micro cube. IR spectra were obtained on a Bruker Tensor 37 spectrometer equipped with a room temperature DLaTGS detector and a diamond ATR (attenuated total reflection) unit.

[{Cp^t(CO)₂Mo}₂(μ , $\eta^{2:2}$ -As₂)] and [{Cp^t(CO)₂Mo}₂(μ , $\eta^{2:2}$ -Sb₂)] were synthesized at the University of Regensburg using following analytical equipment:

The IR spectra were recorded as solids using a ThermoFisher Nicolet iS5 FT-IR spectrometer with an iD7 ATR module and an ITX Germanium or Diamond crystal. The mass spectra and elemental analyses were performed by the analytical laboratories of the University of Regensburg. NMR spectra were recorded in C₆D₆ on a Bruker Avance 300 MHz NMR spectrometer (¹H: 300.132 MHz, ¹³C: 75.468 MHz) or a Bruker Avance 400 MHz NMR spectrometer (¹H: 400.130 MHz, ¹³C: 100.613 MHz) with external references of SiMe₄ (¹H, ¹³C). [(Cp^{*})₂Sm(thf)₂] and [(Cp^{Me_{4n}Pr})₂Sm] were prepared following literature procedures.^{1,2} [(Cp^{*})₂Sm(thf)₂] was sublimed to generate solvent free [(Cp^{*})₂Sm].

Synthesis of $\{[\text{Cp}^t(\text{CO})_2\text{Mo}]_2(\mu,\eta^{2:2}\text{-As}_2)\}$:

250 ml of a saturated solution of yellow arsenic in boiling xylene were transferred onto $[\text{Cp}^t(\text{CO})_2\text{Mo}]_2$ (0.81 g, 1.48 mmol) and refluxed for 4 h yielding an orange brown solution. The xylene was condensed off, the residue dissolved in 15 mL CH_2Cl_2 , combined with 10 g silica gel and dried till a flowing dark brown powder was formed. Chromatographic work-up (4 x 15 cm; silica gel) with pure *n*-hexane affords a yellow fraction of $[\text{Cp}^t(\text{CO})_2\text{Mo}(\eta^3\text{-As}_3)]$. After elution with a mixture of *n*-hexane/toluene of 5:1 an orange-red fraction of $\{[\text{Cp}^t(\text{CO})_2\text{Mo}]_2(\mu,\eta^{2:2}\text{-As}_2)\}$ and a small unidentified green fraction (maybe $[\text{Cp}^t(\text{CO})_2\text{Mo}]_3\text{As}$) were obtained.

$[\text{Cp}^t(\text{CO})_2\text{Mo}(\eta^3\text{-As}_3)]$: Yield: 195 mg (0.39 mmol, 13 % referred to one $[\text{Cp}^t(\text{CO})_2\text{Mo}]$ fragment)

Anal. calcd. (%) for $[\text{C}_{11}\text{H}_{13}\text{As}_3\text{MoO}_2]$: C: 26.53, H: 2.63; found (%): C: 26.68, H: 2.53.

NMR: ^1H NMR (400 MHz, C_6D_6 , 300 K) δ/ppm = 0.93 (s, *t*Bu, 9H), 4.36 (t, 2.4 Hz, Cp, 2H), 4.39 (t, 2.4 Hz, Cp, 2H); $^{13}\text{C}\{\text{H}\}$ NMR (101 MHz, C_6D_6 , 300 K) δ/ppm = 31.1 (s, *t*Bu), 31.6 (s, Me), 81.6 (s, Cp (C-H)), 81.8 (s, Cp (C-H)), 120.3 (s, Cp (quart.)), 223.4 (s, CO)

FD-MS (CH_2Cl_2): Cation *m/z* (%): 499.81 (100)

$\{[\text{Cp}^t(\text{CO})_2\text{Mo}]_2(\mu,\eta^{2:2}\text{-As}_2)\}$: Yield: 55 mg (0.08 mmol, 5.4 % referred $[\text{Cp}^t(\text{CO})_2\text{Mo}]_2$)

IR (ATR): $\tilde{\nu}/\text{cm}^{-1}$ = 2098 (w), 2955 (m), 2864 (w), 1965 (vs), 1934 (vs), 1893 (vs), 1872 (vs), 1478 (m), 1461 (m), 1444 (w), 1398 (w), 1360 (m), 1269 (m), 1197 (w), 1148 (m), 1060 (w), 1040 (w), 1019 (w), 921 (w), 906 (w), 880 (w), 842 (m), 830 (m), 815 (s), 801 (s), 673 (w)

Anal. calcd. (%) for $[\text{C}_{22}\text{H}_{26}\text{As}_2\text{Mo}_2\text{O}_4]$: C: 37.96, H: 3.76; found (%): C: 37.42, H: 3.50.

NMR: ^1H NMR (300 MHz, C_6D_6 , 300 K) δ/ppm = 1.06 (s, *t*Bu, 18H), 4.61 (t, 2.4 Hz, Cp, 4H), 4.75 (t, 2.4 Hz, Cp, 4H); $^{13}\text{C}\{\text{H}\}$ NMR (75 MHz, C_6D_6 , 300 K) δ/ppm = 31.3 (s, *t*Bu), 31.8 (s, Me), 83.0 (s, Cp (C-H)), 84.8 (s, Cp (C-H)), 119.0 (s, Cp (quart.)), 228.5 (s, CO)

FD-MS (CH_2Cl_2): Cation *m/z* (%): 695.86 (100) [M^+]; 789.72 (55) [not assignable]

Synthesis of $\{[\text{Cp}^t(\text{CO})_2\text{Mo}]_2(\mu,\eta^{2:2}\text{-Sb}_2)\}$:

Yellow $(t\text{BuSb})_4$ (1.25 g, 1.75 mmol) and dark brown $[\text{Cp}^t(\text{CO})_2\text{Mo}]_2$ (1.30 g, 2.38 mmol) were weighted into a 250 mL round flask and dissolved in 100 mL toluene. The dark orange brown solution was stirred under reflux for 6 h. Afterwards the solvent was removed, the residue dissolved in 15 mL CH_2Cl_2 , combined with 10 g silica gel, dried to a flowing dark brown powder under reduced pressure, and placed on a chromatography column (4 x 20 cm; silica gel). Elution with *n*-hexane yields a light yellow fraction of unreacted $(t\text{BuSb})_4$. Thereafter, a dark red fraction of $\{[\text{Cp}^t(\text{CO})_2\text{Mo}]_2(\mu,\eta^{2:2}\text{-Sb}_2)\}$ was isolated by elution with a mixture of *n*-hexane and toluene (3:1). The solvent was removed and recrystallisation in 5 mL CH_2Cl_2 at -30 °C for 5 days afforded dark red crystals.

Yield: 106 mg (0.14 mmol, 6 %)

IR (ATR): $\tilde{\nu}/\text{cm}^{-1} = 2983$ (w), 2965 (w), 2943 (w), 2902 (w), 1918 (vs), 1890 (vs), 1866 (vs), 1842 (w), 1476 (w), 1462 (w), 1443 (w), 1400 (w), 1362 (w), 1269 (w), 1147 (w), 1058 (w), 1041 (w), 1020 (w), 904 (w), 883 (w), 842 (w), 831 (w), 818 (w), 804 (w), 682 (w)

Anal. calcd. (%) for $\{(\text{C}_9\text{H}_{13})(\text{CO})_2\text{Mo}\}_2\text{Sb}_2$: C: 33.45, H: 3.32; found (%): C: 33.61, H: 3.34.

NMR: ^1H NMR (400 MHz, C_6D_6 , 300 K) $\delta/\text{ppm} = 1.08$ (s, *t*Bu, 9H), 4.43 (t, 2.4 Hz, Cp, 2H), 4.75 (t, 2.4 Hz, Cp, 2H); $^{13}\text{C}\{^1\text{H}\}$ NMR (101 MHz, C_6D_6 , 300 K) $\delta/\text{ppm} = 31.4$ (s, *t*Bu), 31.9 (s, Me), 81.1 (s, Cp (C-H)), 81.4 (s, Cp (C-H)), 116.8 (s, Cp (quart.)), 228.7 (s, CO)

LIFDI-MS (toluene): Cation *m/z* (%): 789.78 (100)

Synthesis of $\{(\text{Cp}^*\text{Sm})_2\text{As}_2(\text{Cp}^t\text{Mo}(\text{CO})_2)_2$ (1):

n-Hexane (10 mL) was condensed onto a mixture of $[\text{Cp}^*\text{Sm}]$ (52.0 mg, 0.12 mmol, 2.00 equiv.) and $\{[\text{Cp}^t(\text{CO})_2\text{Mo}]_2(\mu,\eta^{2:2}\text{-As}_2)\}$ (45.0 mg, 0.062 mmol, 1.00 equiv.) at -78 °C in a double ampule. The resulting reaction mixture was heated to 65 °C in an oven and after 3 days hot decanted. Crystals of **1** were obtained by slow evaporation of the solvent at ambient temperature to the other section of the ampule. Yield: 33.0 mg (34 %) of crystals.

IR (ATR): $\tilde{\nu}/\text{cm}^{-1} = 2958$ (s), 2929 (s), 2904 (s), 2853 (s), 1968 (m), 1944 (s), 1896 (vs), 1859 (s), 1620 (m), 1575 (m), 1482 (vw), 1463 (vw), 1445 (w), 1437 (w), 1378 (vw), 1362 (w), 1271 (vw), 1149 (vw), 1082 (vw), 1061 (vw), 1019 (vw), 841 (vw), 814 (vw), 792 (w), 677 (vw), 589 (vw), 558 (vw), 533 (vw), 485 (vw), 459 (vw).

NMR: ^1H NMR (400 MHz, C_6D_6 , 300 K) $\delta/\text{ppm} = -1.11$ (s, $t\text{Bu}$, 9H), 1.08 (s, Cp-Me , 60H), 3.81 (t, 2.2 Hz, Cp , 2H), 5.35 (t, 2.2 Hz, Cp , 2H); $^{13}\text{C}\{\text{H}\}$ NMR (101 MHz, C_6D_6 , 300 K) $\delta/\text{ppm} = 20.6$ (s, Cp-Me), 30.0 (s, CMe_3), 30.1 (s, CMe_3), 88.5 (s, Cp (C-H)), 96.0 (s, Cp (C-H)), 118.7 (s, Cp^*), 119.7 (s, Cp (quart.)), 274.8 (s, CO)

Anal. calcd. (%) for $[\text{C}_{62}\text{H}_{86}\text{Mo}_2\text{O}_4\text{As}_2\text{Sm}_2]$ (1537.81 g/mol⁻¹): C 48.42, H 5.64; found C 48.95, H 5.68.

Synthesis of $[(\text{Cp}^{\text{Me}4\text{nPr}})_2\text{Sm}]_2\text{As}_2(\text{Cp}^t\text{Mo}(\text{CO})_2)_2$ and $[(\text{Cp}^{\text{Me}4\text{nPr}})_2\text{Sm}]_2\text{As}_4(\text{Cp}^t\text{Mo}(\text{CO})_2)_2$ (2,3):

n-Heptane (10 mL) was added onto a mixture of $[\text{Cp}^{\text{Me}4\text{nPr}}_2\text{Sm}]$ (65.6 mg, 0.137 mmol, 2.00 equiv.) and $[\{\text{Cp}^t(\text{CO})_2\text{Mo}\}_2(\mu,\eta^{2:2}\text{-As}_2)]$ (50.0 mg, 0.069 mmol, 1.00 equiv.). The resulting reaction mixture was heated to 65 °C and stirred for 72 h. After cooling to rt the mixture was filtered. A crystal mixture of **2** and **3** was obtained by slow evaporation of the solvent at ambient temperature. Yield: 60.0 mg of mixed crystals of **2** and **3**.

Separation and further analytics of the two products (**2**, **3**) was not feasible due to similar solubility behavior.

Synthesis of $[(\text{Cp}^*_2\text{Sm})_2\text{Sb}_4(\text{Cp}^t\text{Mo}(\text{CO})_2)_2]$ (4):

Method A:

n-Hexane (10 mL) was condensed onto a mixture of $[\text{Cp}^*_2\text{Sm}]$ (32.0 mg, 0.076 mmol, 2.00 equiv.) and $[\{\text{Cp}^t\text{Mo}(\text{CO})_2\}_2(\mu,\eta^{2:2}\text{-Sb}_2)]$ (30.0 mg, 0.038 mmol, 1.00 equiv.) at -78 °C in a double ampule. The resulting reaction mixture was heated to 60 °C in an oven and hot decanted after 3 days. After cooling to room temperature crystals of **4** were obtained. Yield: 48.0 mg (67%) of crystals.

Method B:

Toluene (10 mL) was condensed onto a mixture of $[\text{Cp}^*_2\text{Sm}]$ (42.0 mg, 0.10 mmol, 2.00 equiv.) and $[\{\text{Cp}^t\text{Mo}(\text{CO})_2\}_2(\mu,\eta^{2:2}\text{-Sb}_2)]$ (40.0 mg, 0.05 mmol, 1.00 equiv.) at -78 °C in a double ampule. The resulting reaction mixture was heated to 60 °C in an oven and after 3 days hot decanted. Crystals of **4** were obtained by slow evaporation of the solvent at ambient temperature to the other section of the ampule. Yield: 52.0 mg (52%) of crystals.

IR (ATR): $\tilde{\nu}/\text{cm}^{-1}$ = 3243 (vw), 2955 (s), 2900 (s), 2855 (s), 2725 (w), 1898 (vs), 1889 (vs), 1858 (m), 1701 (vw), 1633 (s), 1480 (m), 1445 (m), 1435 (m), 1378 (m), 1358 (m), 1272 (m), 1148 (w), 1060 (vw), 1038 (vw), 1019 (w), 903 (w), 858 (w), 811 (w), 792 (m), 612 (w), 577 (w), 556 (m), 477 (w), 459 (w).

Anal. calcd. (%) for $[\text{C}_{62}\text{H}_{86}\text{Mo}_2\text{O}_4\text{Sb}_4\text{Sm}_2]$ (1874.88 g/mol⁻¹): C 39.72, H 4.62; found C 39.70, H 4.87.

Due to paramagnetism and decomposition upon redissolving single crystals of **4**, no suitable NMR spectra were obtainable.

Synthesis of $[(\text{Cp}^{\text{Me}4\text{nPr}})_2\text{Sm})_2\text{Sb}_4(\text{Cp}^t\text{Mo}(\text{CO})_2)_2]$ (**5**):

Toluene (10 mL) was added onto a mixture of $[\text{Cp}^{\text{Me}4\text{nPr}}_2\text{Sm}]$ (60.4 mg, 0.127 mmol, 2.00 equiv.) and $[\{\text{Cp}^t\text{Mo}(\text{CO})_2\}_2(\mu,\eta^{2:2}\text{-Sb}_2)]$ (50.0 mg, 0.063 mmol, 1.00 equiv.). The resulting reaction mixture was heated to 75 °C and stirred for 72 h. After cooling to rt the solvent was removed under reduced pressure and the solid subsequently recrystallized from hot *n*-heptane. Yield: 20 mg (31 %) of crystals.

IR (ATR): $\tilde{\nu}/\text{cm}^{-1}$ = 2955 (s), 2928 (s), 2899 (s), 2868 (s), 2851 (s), 2724 (w), 1896 (vs), 1857 (m), 1630 (s), 1481 (m), 1464 (m), 1434 (m), 1380 (m), 1360 (m), 1268 (w), 1147 (w), 1089 (w), 1038 (w), 1019 (w), 907 (w), 886 (w), 808 (w), 785 (m), 614 (w), 575 (w), 555 (m), 478 (w), 455 (m).

Anal. calcd. (%) for $[\text{C}_{70}\text{H}_{102}\text{Mo}_2\text{O}_4\text{Sb}_4\text{Sm}_2]$ (1987.20 g/mol⁻¹): C 42.31, H 5.17; found C 41.73, H 5.05.

Due to paramagnetism and decomposition upon redissolving single crystals of **5**, no suitable NMR spectra were obtainable.

IR Spectra:

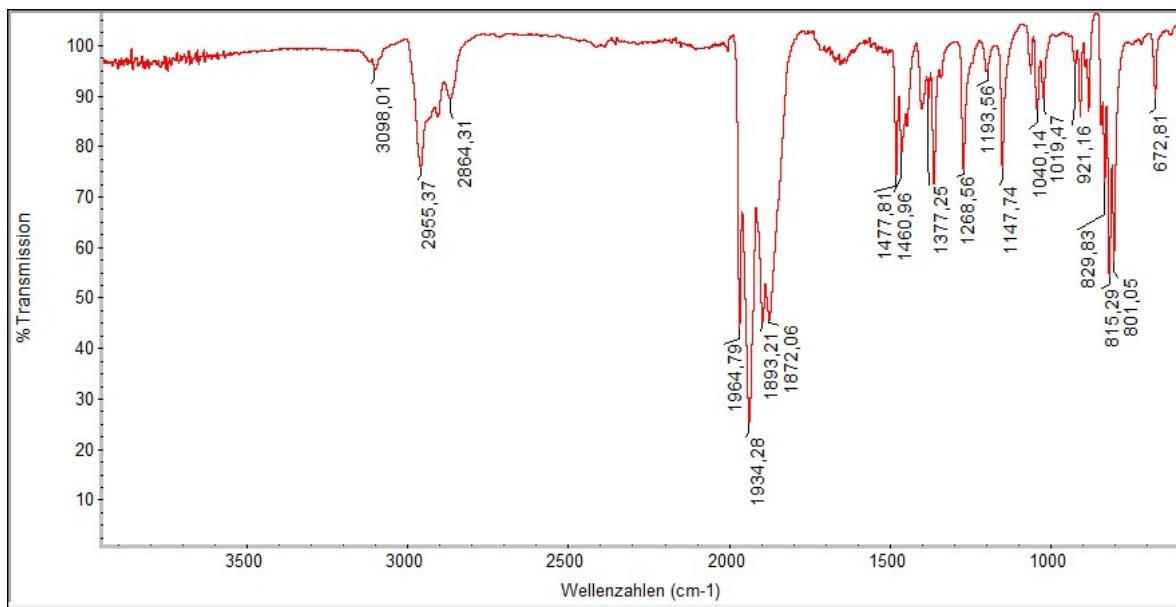


Fig. S1. IR Spectrum of $\left[\{\text{Cp}^t(\text{CO})_2\text{Mo}\}_2(\mu,\eta^{2:2}\text{-As}_2)\right]$.

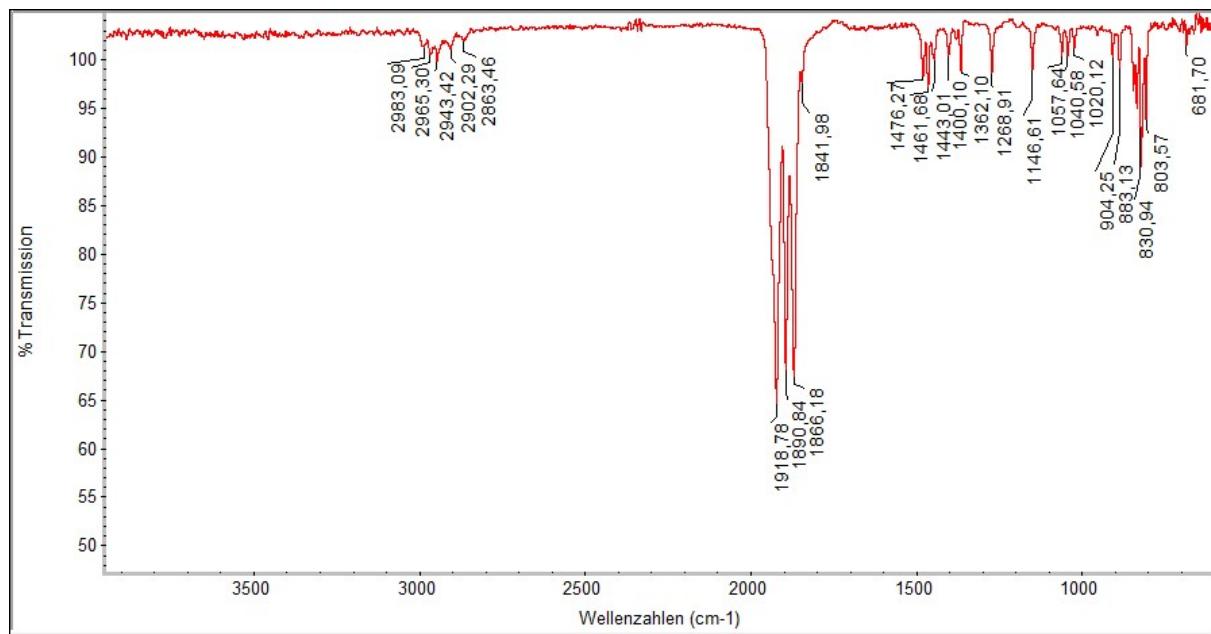


Fig. S2. IR Spectrum of $\left[\{\text{Cp}^t(\text{CO})_2\text{Mo}\}_2(\mu,\eta^{2:2}\text{-Sb}_2)\right]$.

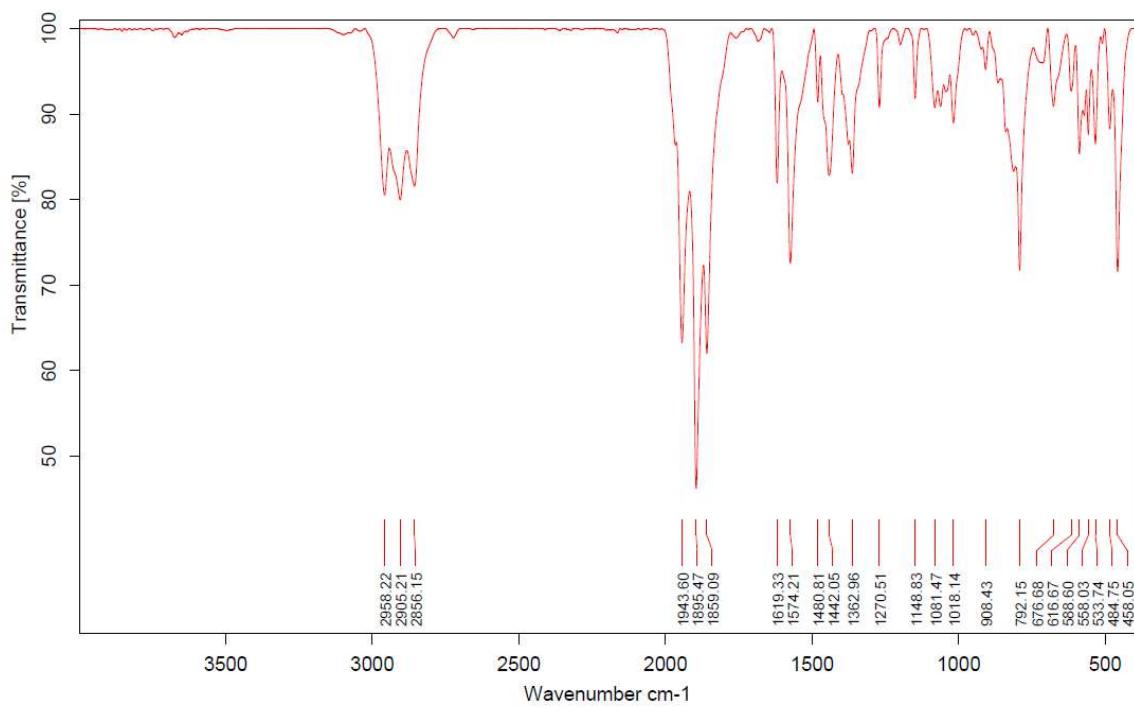


Fig. S3. IR Spectrum of **1** ($[(\text{Cp}^*)_2\text{Sm}]_2\text{As}_2(\text{Cp}^\ddagger\text{Mo}(\text{CO})_2)_2$).

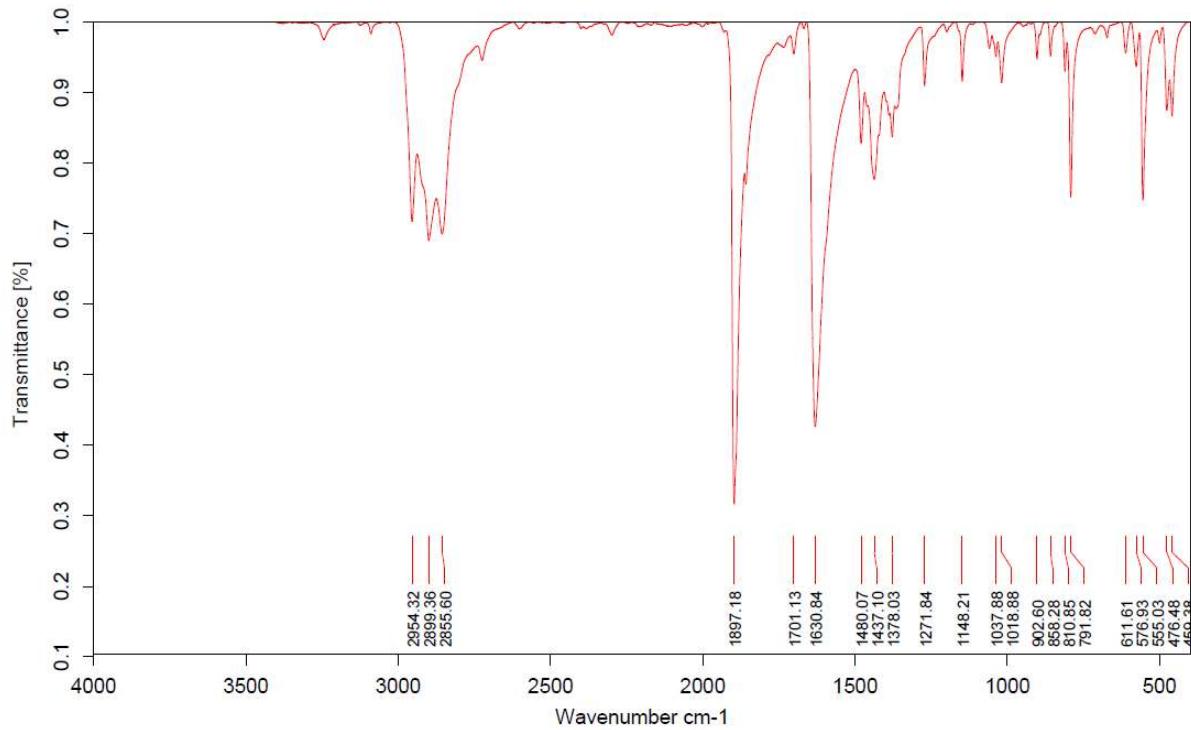


Fig. S4. IR Spectrum of **4** ($[(\text{Cp}^*{}^2\text{Sm})_2\text{Sb}_4(\text{Cp}^t\text{Mo}(\text{CO})_2)_2]$).

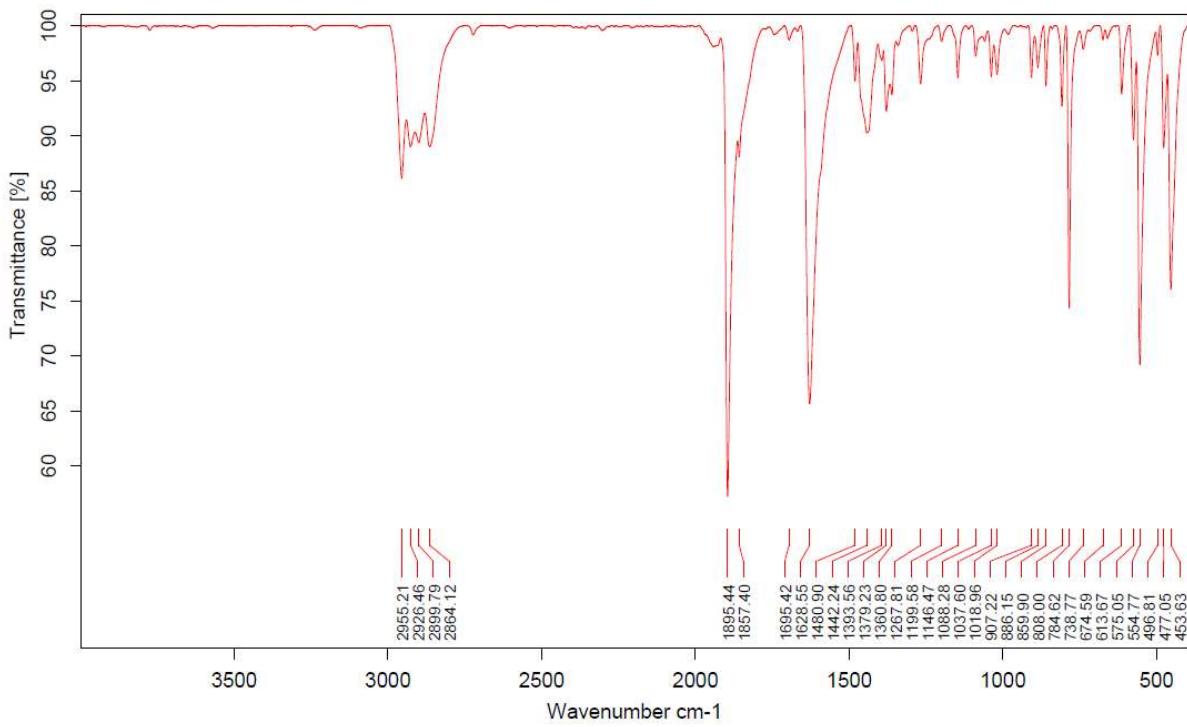


Fig. S5. IR Spectrum of **5** ($[(\text{Cp}^{\text{Me}4n\text{Pr}})_2\text{Sm})_2\text{Sb}_4(\text{Cp}^t\text{Mo}(\text{CO})_2)_2]$).

NMR Spectra:

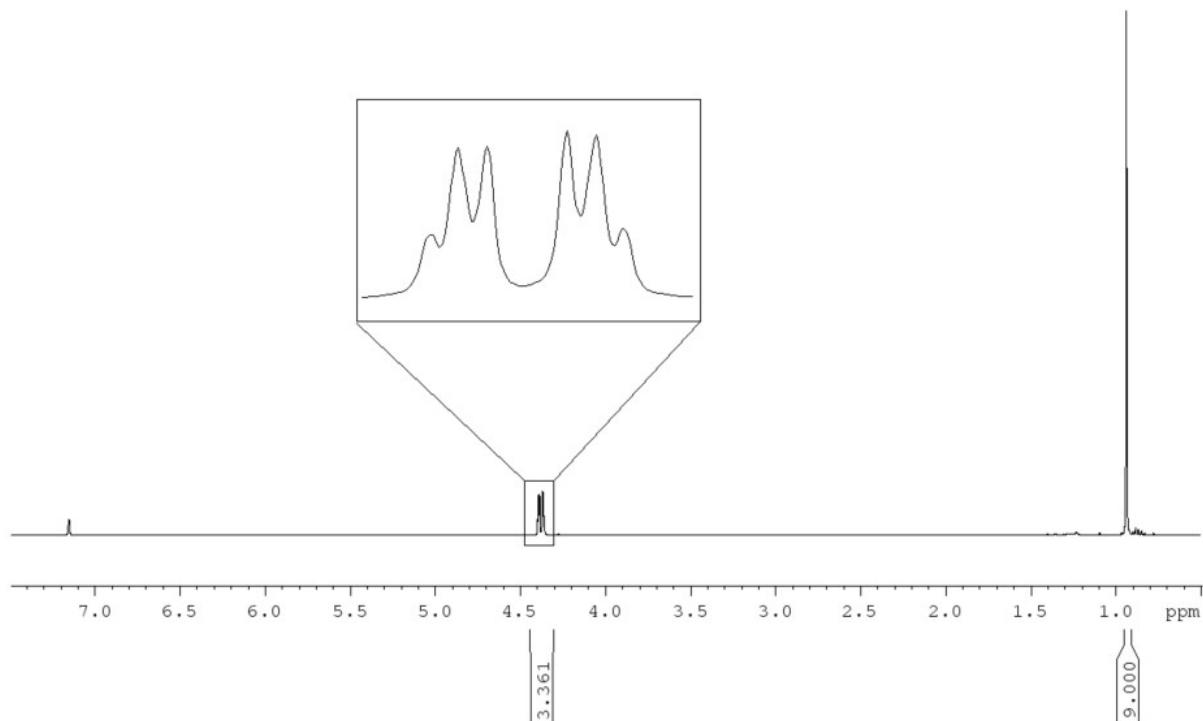


Fig. S6. ^1H -NMR Spectrum of $[\text{Cp}^t(\text{CO})_2\text{Mo}(\eta^3\text{-As}_3)]$.

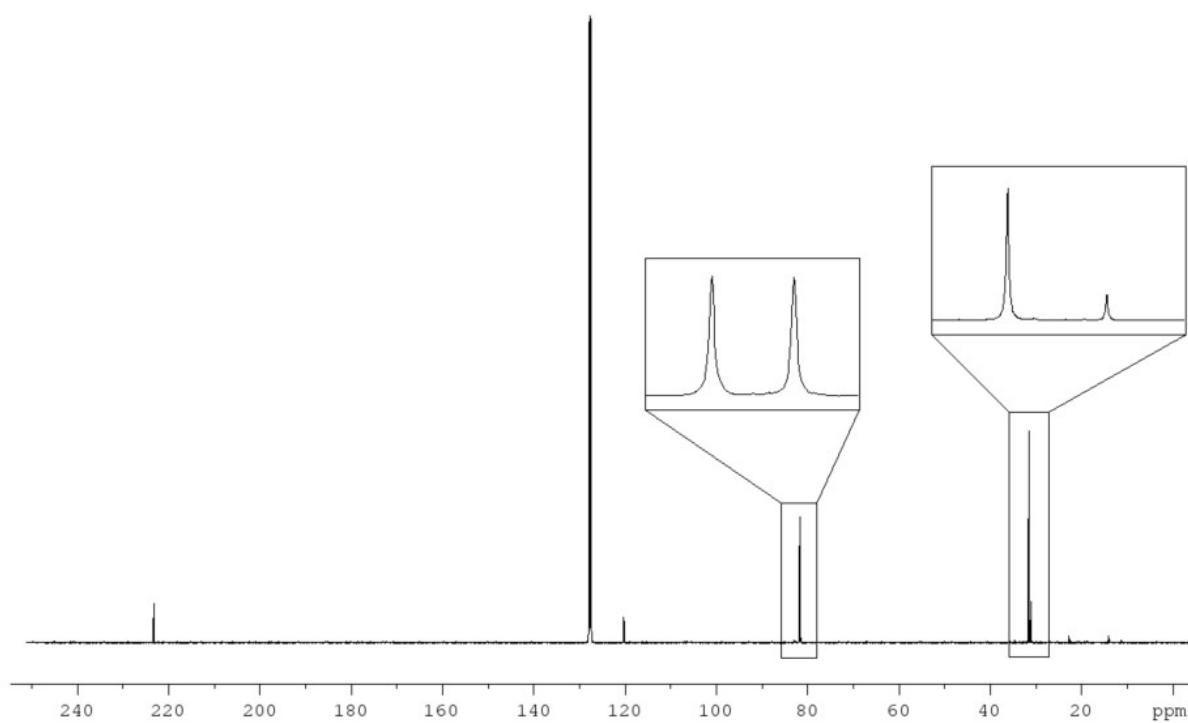


Fig. S7. $^{13}\text{C}\{\text{H}\}$ -NMR Spectrum of $[\text{Cp}^t(\text{CO})_2\text{Mo}(\eta^3\text{-As}_3)]$.

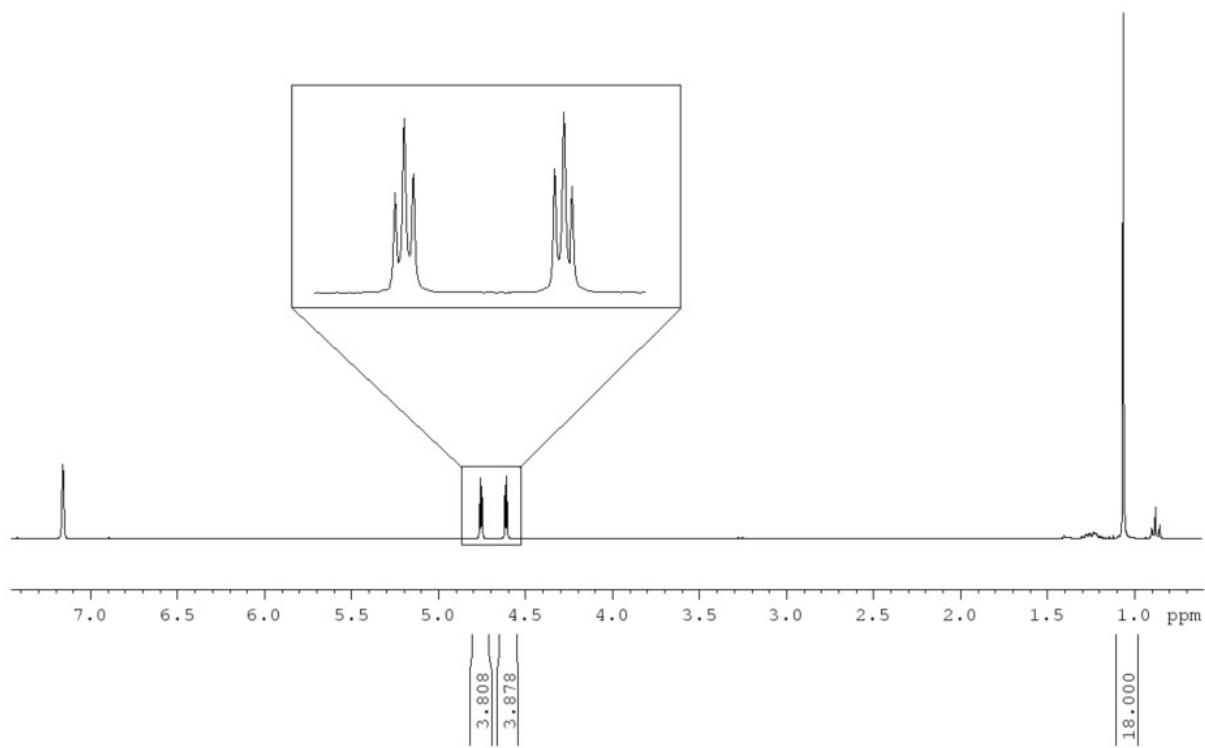


Fig. S8. ¹H-NMR Spectrum of [{Cp^t(CO)₂Mo}₂(μ,η^{2:2}-As₂)].

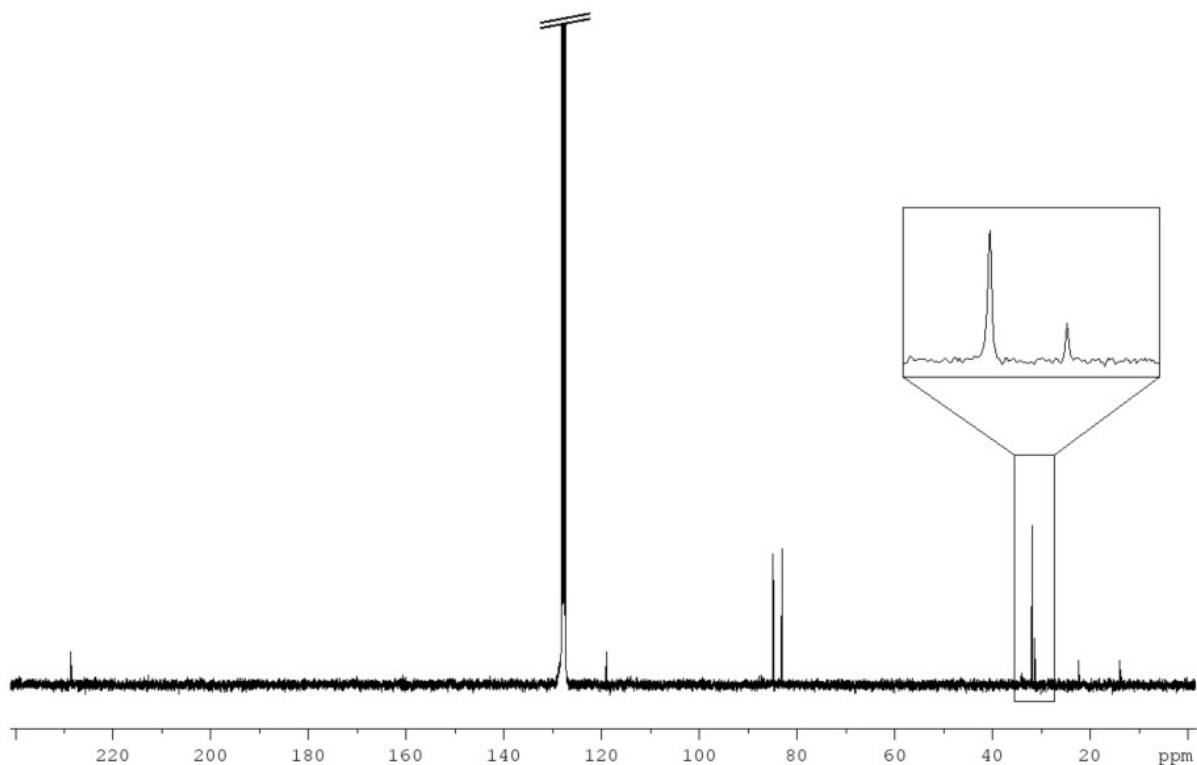


Fig. S9. ¹³C{¹H}-NMR Spectrum of [{Cp^t(CO)₂Mo}₂(μ,η^{2:2}-As₂)].

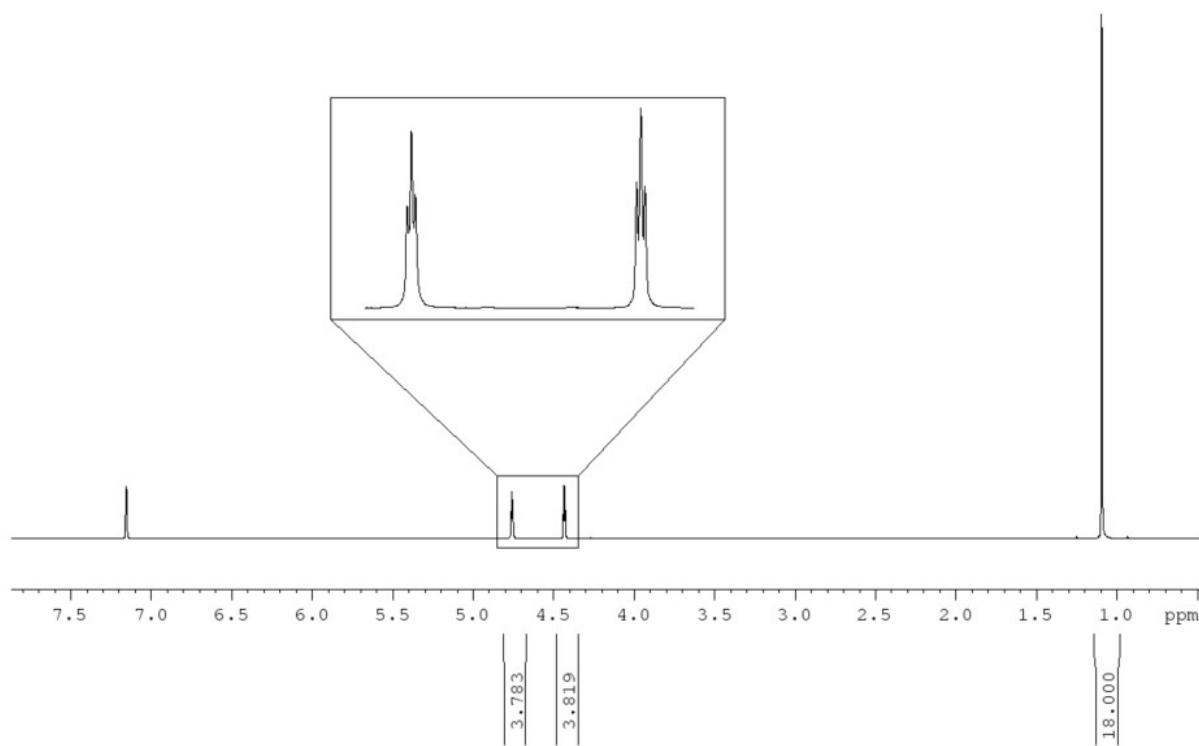


Fig. S10. ¹H-NMR Spectrum of [{Cpt(CO)₂Mo}₂(μ , $\eta^{2:2}$ -Sb₂)].

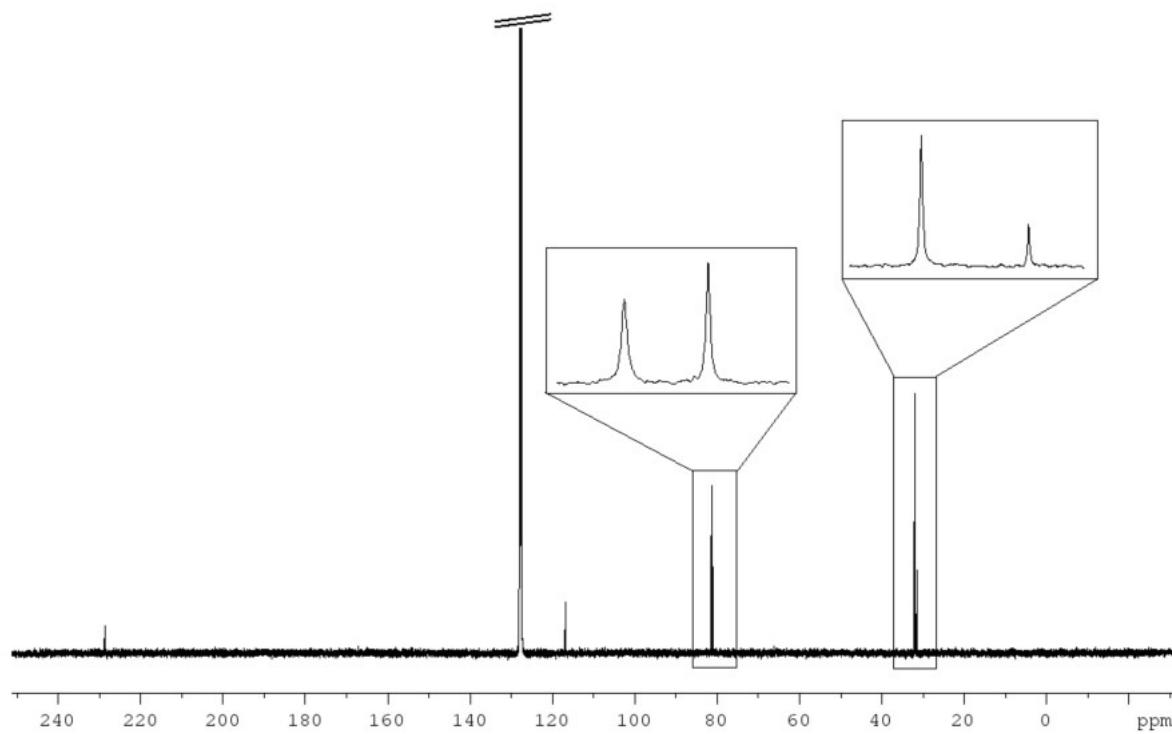


Fig. S11. ¹³C{¹H}-NMR Spectrum of [{Cpt(CO)₂Mo}₂(μ , $\eta^{2:2}$ -Sb₂)].

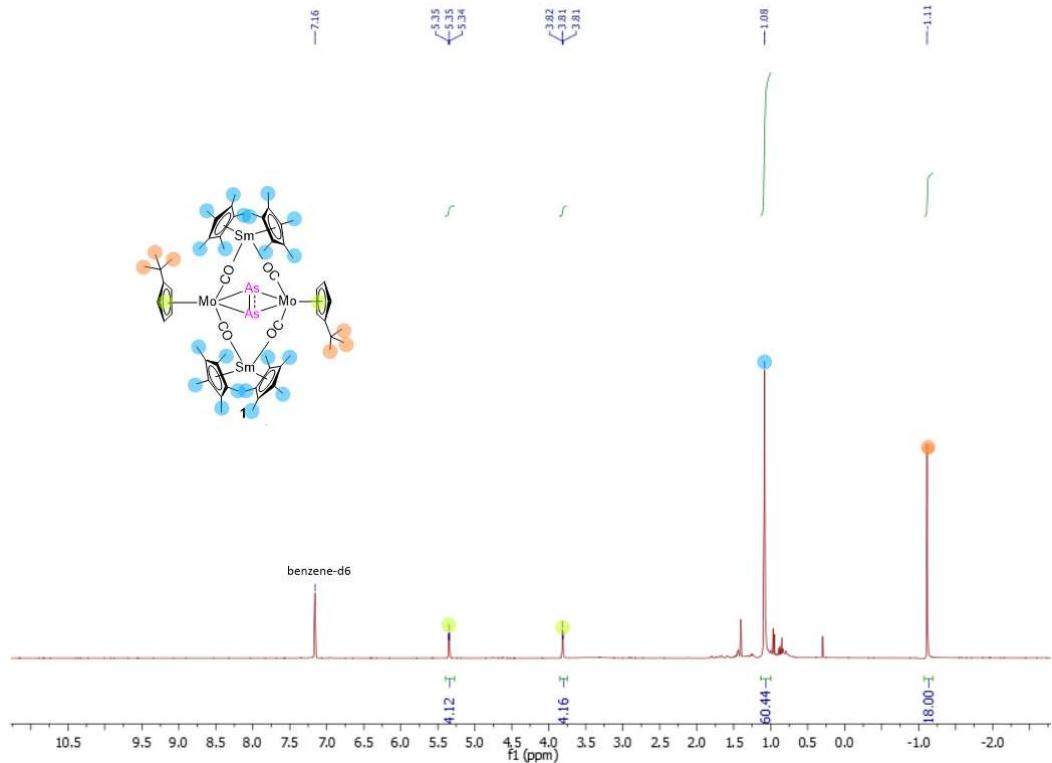


Fig. S12. ^1H -NMR Spectrum of **1**.

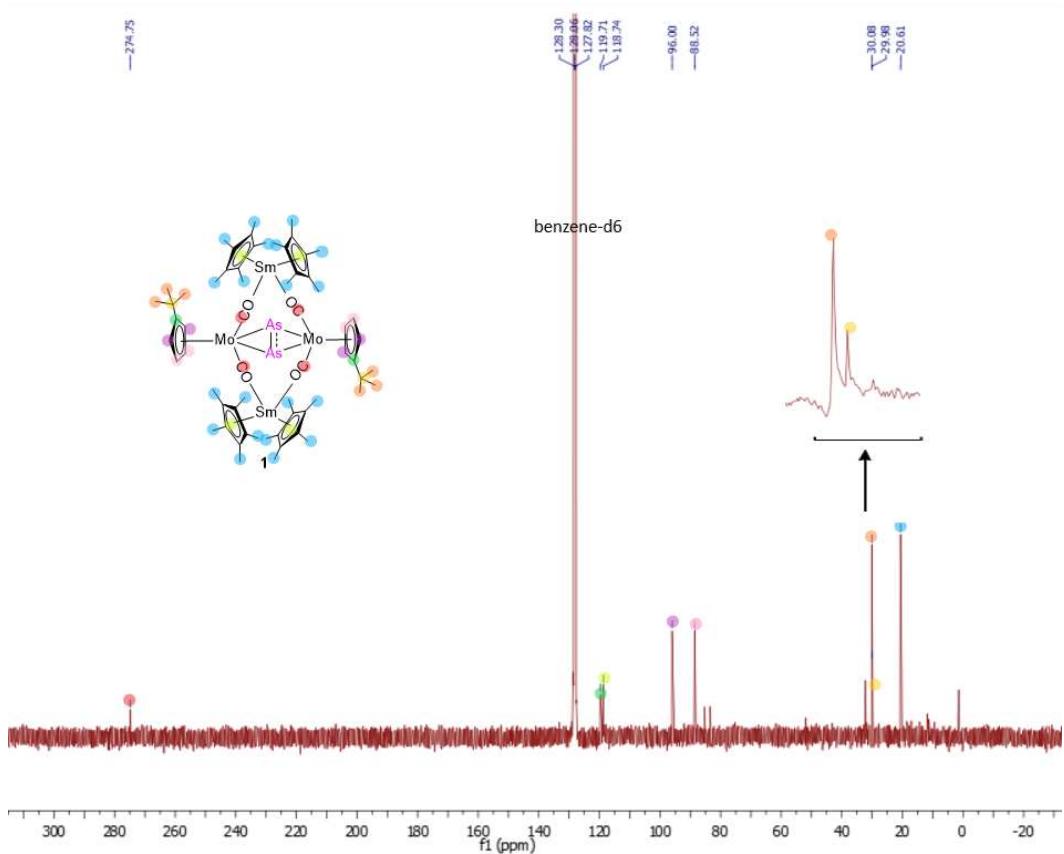


Fig. S13. $^{13}\text{C}\{^1\text{H}\}$ -NMR Spectrum of **1**. All unsigned peaks are traces of decomposition of **1** in solution. Single crystals were dissolved for recording the spectrum.

X-ray Crystallographic Studies of (1), (2), (3), (4) and (5):

A suitable crystal was covered in mineral oil (Aldrich) and mounted on a glass fiber. The crystal was transferred directly to the cold stream of a STOE IPDS 2 or a STOE StadiVari diffractometer. All structures were solved by using the program SHELXS/T³ and Olex2.⁴ The remaining non-hydrogen atoms were located from successive difference Fourier map calculations. The refinements were carried out by using full-matrix least-squares techniques on F^2 by using the program SHELXL.³ In each case, the locations of the largest peaks in the final difference Fourier map calculations, as well as the magnitude of the residual electron densities, were of no chemical significance.

Crystallographic data (excluding structure factors) for the structures reported in this paper have been deposited with the Cambridge Crystallographic Data Centre as a supplementary publication no. 2015996-2016000. Copies of the data can be obtained free of charge on application to CCDC, 12 Union Road, Cambridge CB21EZ, UK (fax: +(44)1223-336-033; email: deposit@ccdc.cam.ac.uk).

Refinement Details

The crystal structure of **1** contains 1.75 molecules of *n*-hexane per molecule. The molecules could not be modeled satisfactorily and were therefore removed from the electron density map using the OLEX2 solvent mask.

The crystal structure of **5** is reproducible slightly twinned and disordered. No crystals of better quality could be grown. The twinning could not be solved by using twin integration nor applying a twin-matrix. This results in A and B alerts in the cif check. One of the two half molecules in the asymmetric units is disordered by the turning of the Sb₄ unit in a 45 to 55 ratio. This is leading to a disorder with the CO moieties which could be modeled. Additionally,

the crystal structure contains one disordered molecule of toluene in the asymmetric unit which causes alerts dealing with short inter H...H contacts.

Table S1: Crystal data and structure refinement of **1**, **2** and **3**.

| Compound | 1 | 2 | 3 |
|---|---|---|---|
| Formula | C ₆₂ H ₈₆ Mo ₂ O ₄ As ₂ Sm ₂ · 1.75 <i>n</i> -hexane | C ₇₀ H ₁₀₂ Mo ₂ O ₄ As ₂ Sm ₂ | C ₇₀ H ₁₀₂ Mo ₂ O ₄ As ₄ Sm ₂ |
| D_{calc.}/ g cm⁻³ | 1.272 | 1.570 | 1.750 |
| μ/mm⁻¹ | 2.598 | 2.994 | 4.022 |
| Formula Weight | 1537.72 | 1649.93 | 1799.77 |
| Colour | clear red | red | clear reddish orange |
| Shape | irregular | irregular | plank |
| Size/mm³ | 0.43×0.38×0.34 | 0.38×0.26×0.12 | 0.36×0.16×0.05 |
| T/K | 150 | 100 | 100 |
| Crystal System | orthorhombic | monoclinic | monoclinic |
| Space Group | <i>Pccn</i> | <i>C2/c</i> | <i>P2₁/n</i> |
| a/Å | 25.760(5) | 30.778(6) | 17.800(4) |
| b/Å | 32.030(6) | 10.050(2) | 9.500(2) |
| c/Å | 19.460(4) | 25.375(5) | 20.200(4) |
| α° | 90 | 90 | 90 |
| β° | 90 | 117.24(3) | 91.10(3) |
| γ° | 90 | 90 | 90 |
| V/Å³ | 16056(6) | 6978(3) | 3415.2(12) |
| Z | 8 | 4 | 2 |
| Z' | 1 | 0.5 | 0.5 |
| Wavelength/Å | 0.71073 | 0.71073 | 0.71073 |
| Radiation type | MoK _α | MoK _α | MoK _α |
| Θ_{min}/° | 1.271 | 2.587 | 2.623 |
| Θ_{max}/° | 25.189 | 29.586 | 29.593 |
| Measured Refl. | 136018 | 43003 | 20036 |
| Independent Refl. | 14336 | 8944 | 8308 |
| Reflections with I > 2(I) | 12308 | 7556 | 6633 |
| R_{int} | 0.0322 | 0.0339 | 0.0219 |
| Parameters | 675 | 374 | 383 |
| Restraints | 0 | 0 | 0 |
| Largest Peak | 1.001 | 0.786 | 1.119 |
| Deepest Hole | -0.753 | -0.399 | -0.417 |
| GooF | 1.164 | 1.000 | 0.996 |
| wR₂ (all data) | 0.1179 | 0.0508 | 0.0649 |
| wR₂ | 0.1155 | 0.0497 | 0.0605 |
| R₁ (all data) | 0.0521 | 0.0282 | 0.0428 |
| R₁ | 0.0464 | 0.0218 | 0.0279 |

Table S2: Crystal data and structure refinement of **4** and **5**.

| Compound | 4 | 5 |
|---|--|--|
| Formula | C ₆₂ H ₈₆ Mo ₂ O ₄ Sb ₄ Sm ₂ | C ₇₀ H ₁₀₂ Mo ₂ O ₄ Sb ₄ Sm ₂ · 0.25 toluene |
| D_{calc.}/ g cm⁻³ | 1.953 | 1.709 |
| μ/mm⁻¹ | 3.899 | 3.190 |
| Formula Weight | 1874.88 | 2010.12 |
| Colour | clear orange | clear red |
| Shape | plate | plank |
| Size/mm³ | 0.17×0.16×0.15 | 0.34×0.18×0.07 |
| T/K | 100 | 210 |
| Crystal System | triclinic | triclinic |
| Space Group | P-1 | P-1 |
| a/Å | 10.650(3) | 14.6782(5) |
| b/Å | 10.613(2) | 16.5208(5) |
| c/Å | 14.627(3) | 18.0566(7) |
| α° | 88.52(2) | 83.156(3) |
| β° | 78.18(2) | 78.367(3) |
| γ° | 80.17(2) | 65.698(3) |
| V/Å³ | 1594.5(6) | 3905.5(3) |
| Z | 1 | 2 |
| Z' | 0.5 | 1 |
| Wavelength/Å | 0.71073 | 0.71073 |
| Radiation type | MoK _α | MoK _α |
| Θ_{min}/° | 1.947 | 1.354 |
| Θ_{max}/° | 26.133 | 27.175 |
| Measured Refl. | 11627 | 33430 |
| Independent Refl. | 6165 | 17178 |
| Reflections with I > 2(I) | 4840 | 10444 |
| R_{int} | 0.0686 | 0.0542 |
| Parameters | 347 | 841 |
| Restraints | 0 | 198 |
| Largest Peak | 2.367 | 5.390 |
| Deepest Hole | -2.063 | -3.119 |
| GooF | 1.019 | 1.040 |
| wR₂ (all data) | 0.1750 | 0.1807 |
| wR₂ | 0.1654 | 0.1586 |
| R_I (all data) | 0.0771 | 0.1099 |
| R_I | 0.0645 | 0.0601 |

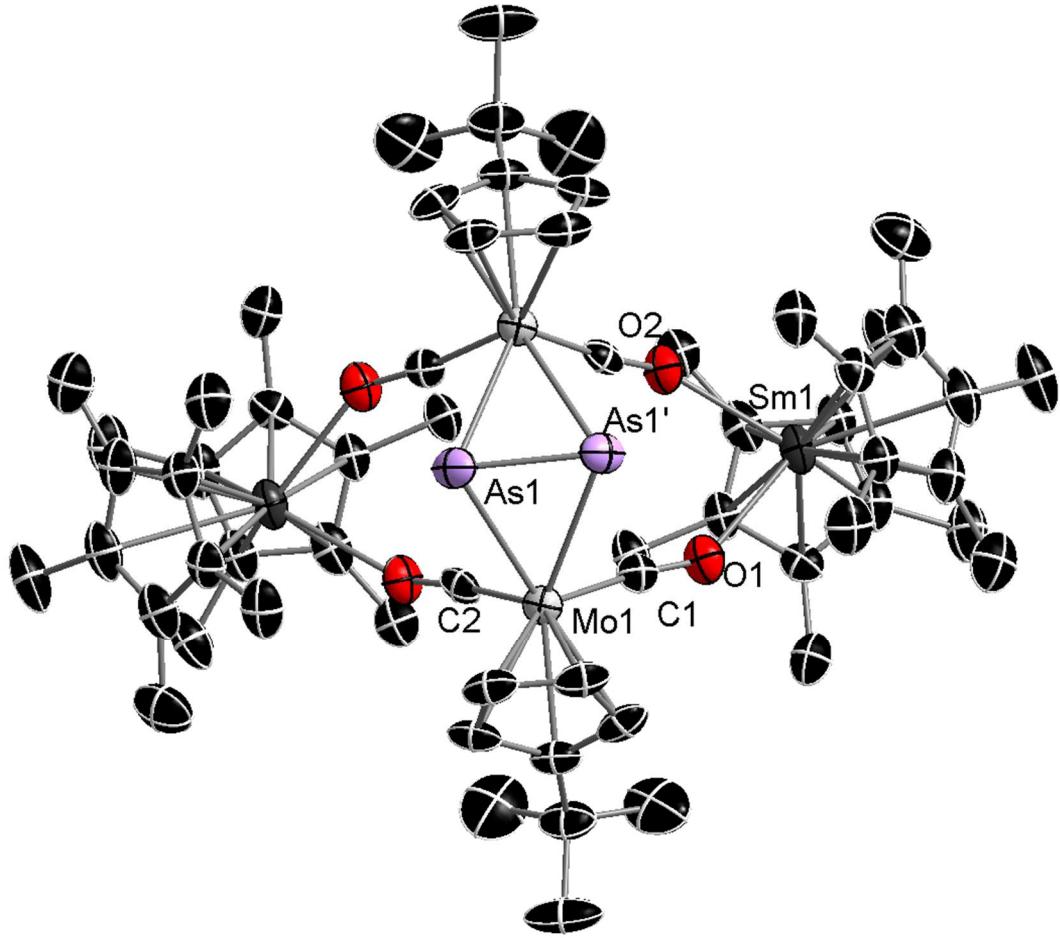


Fig. S14. Molecular structure of **1** in the solid state displayed with an ellipsoid probability of 50 %. Hydrogen atoms are omitted for clarity. Selected bond lengths [Å] and angles [°]: Sm1–O1 2.378(4), Sm1–O2' 2.391(4), Mo1–As1 2.6847(9), Mo1–As1' 2.6997(9), Mo1–C1 1.892(6), As1–As1' 2.238(2), C1–O1 1.212(3), O1–Sm–O2' 77.6(2), Mo1–As1–Mo1' 103.92(3).

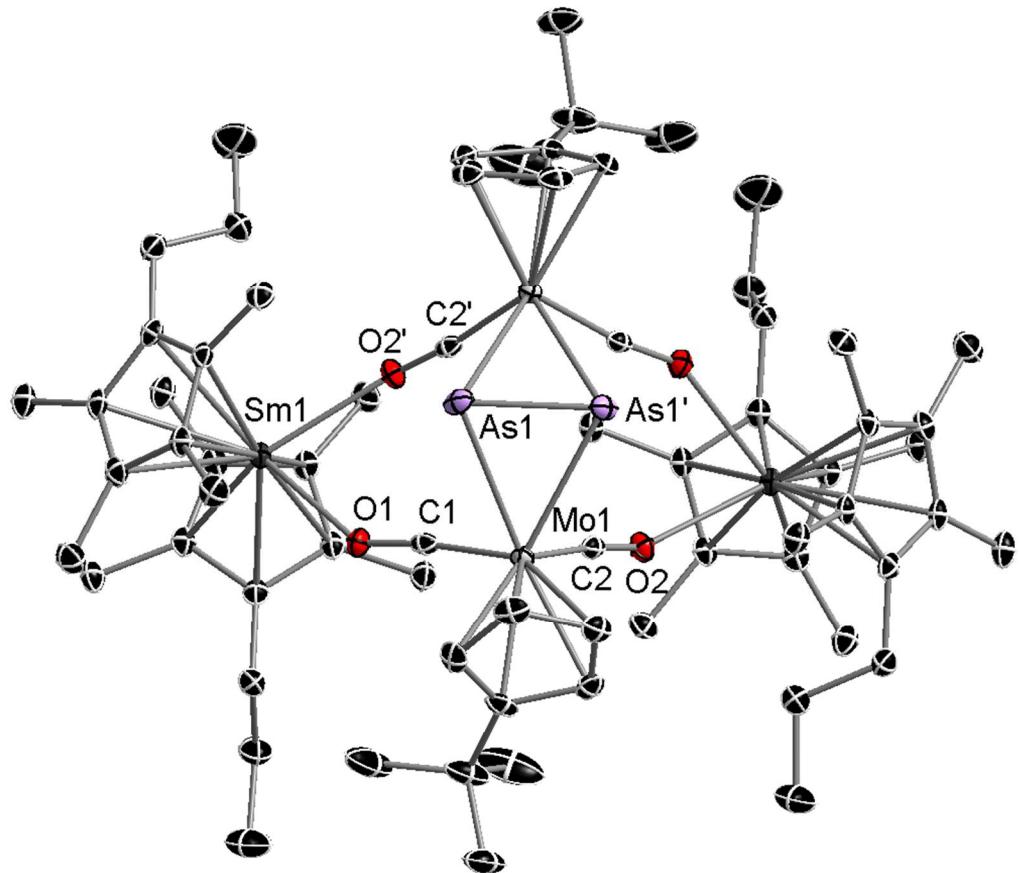


Fig. S15. Molecular structure of **2** in the solid state displayed with an ellipsoid probability of 50 %. Hydrogen atoms are omitted for clarity. Selected bond lengths [\AA] and angles [°]: Sm1–O1 2.383(2), Sm1–O2' 2.413(2), Mo1–As1 2.7047(5), Mo1–As1' 2.7092(10), Mo1–C1 1.887(2), As1–As1' 2.2411(7), C1–O1 1.212(3), O1–Sm–O2' 74.47(5), Mo1–As1–Mo1' 103.02(2).

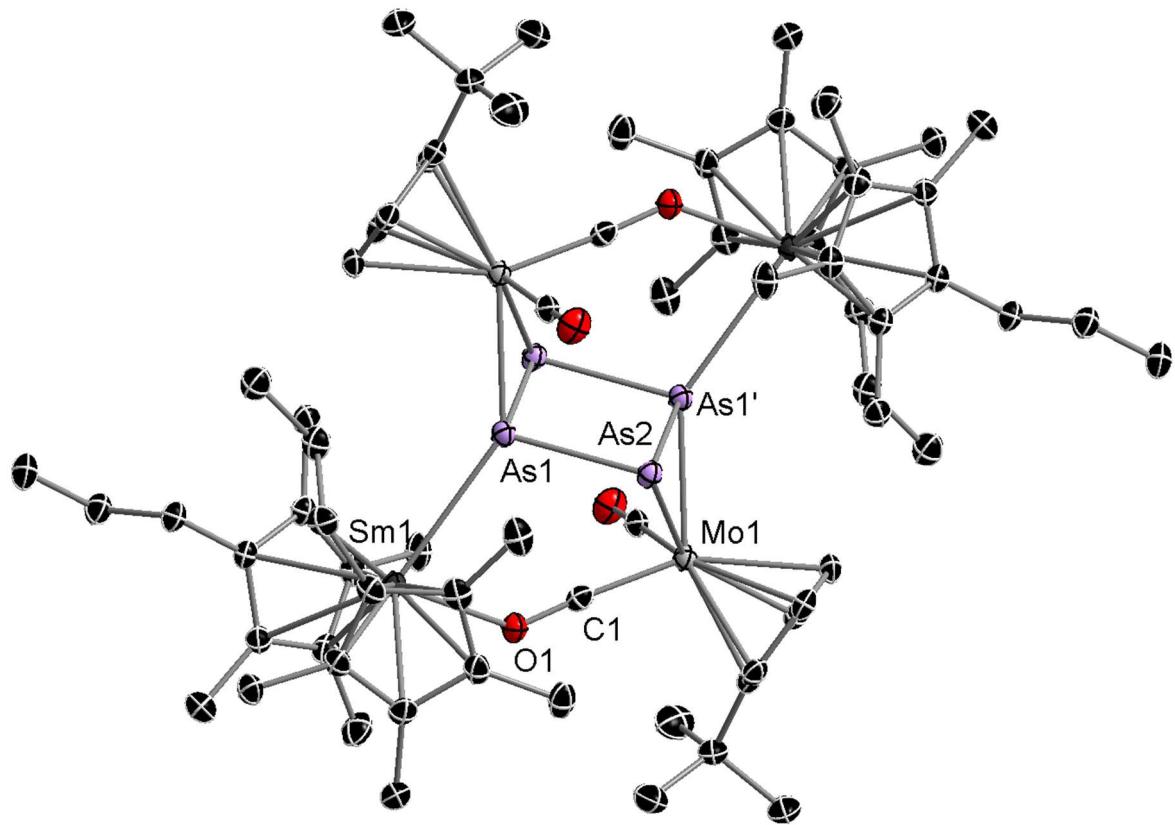


Fig. S16. Molecular structure of **3** in the solid state displayed with an ellipsoid probability of 50 %. Hydrogen atoms are omitted for clarity. Selected bond lengths [Å] and angles [°]: Sm1–As1 3.0300(8), Sm1–O1 2.380(2), Mo1–As2 2.6498(7), Mo1–As1' 2.6261(7), Mo1–C1 1.873(3), As1–As2 2.4862(7), As1'–As2 2.3503(6), C1–O1 1.206(4), O1–Sm–As1 76.99(6), As1'–Mo1–As2 52.91(2), Mo1–As1'–Sm1' 142.35(2), As1'–As2–Mo1 63.03(2), As1–As2–Mo1 105.01(3), As2–As1–As2' 91.16(2), As1–As2–As1' 88.84(2).

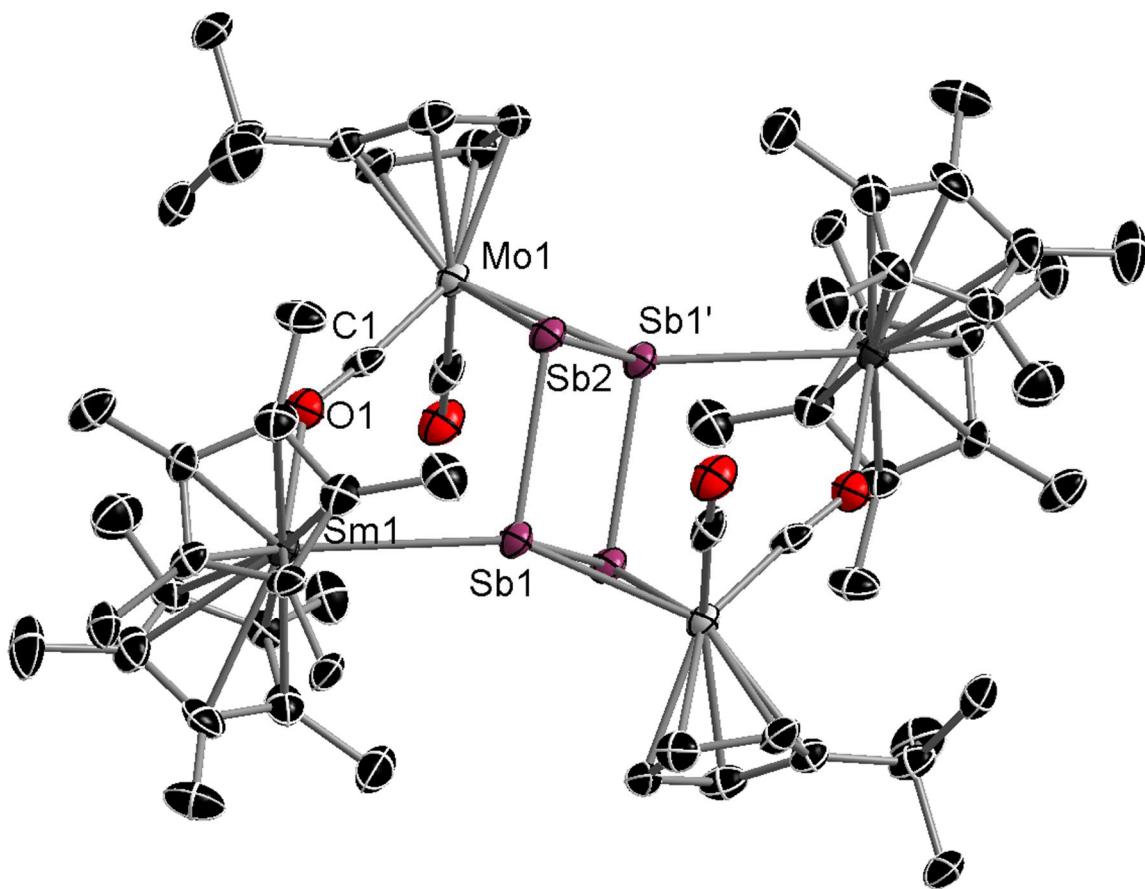


Fig. S17. Molecular structure of **4** in the solid state displayed with an ellipsoid probability of 50 %. Hydrogen atoms are omitted for clarity. Selected bond lengths [Å] and angles [°]: Sm1–Sb2' 3.2375(7), Sm1–O1 2.381(6), Mo1–Sb2 2.8146(9), Mo1–Sb1 2.8549(9), Mo1–C1 1.894(8), Sb1–Sb2 2.7313(8), Sb1–Sb2' 2.8608(7), C1–O1 1.209(10), O1–Sm–Sb2' 76.73(14), Sb1–Mo1–Sb2 57.59(2), Mo1–Sb2–Sm1' 153.18(2), Sb1–Sb2–Mo1 61.94(2), Sb2'–Sb1–Mo1 102.32(2), Sb2–Sb1–Sb2' 87.73(2), Sb1–Sb2–Sb1' 92.28(2).

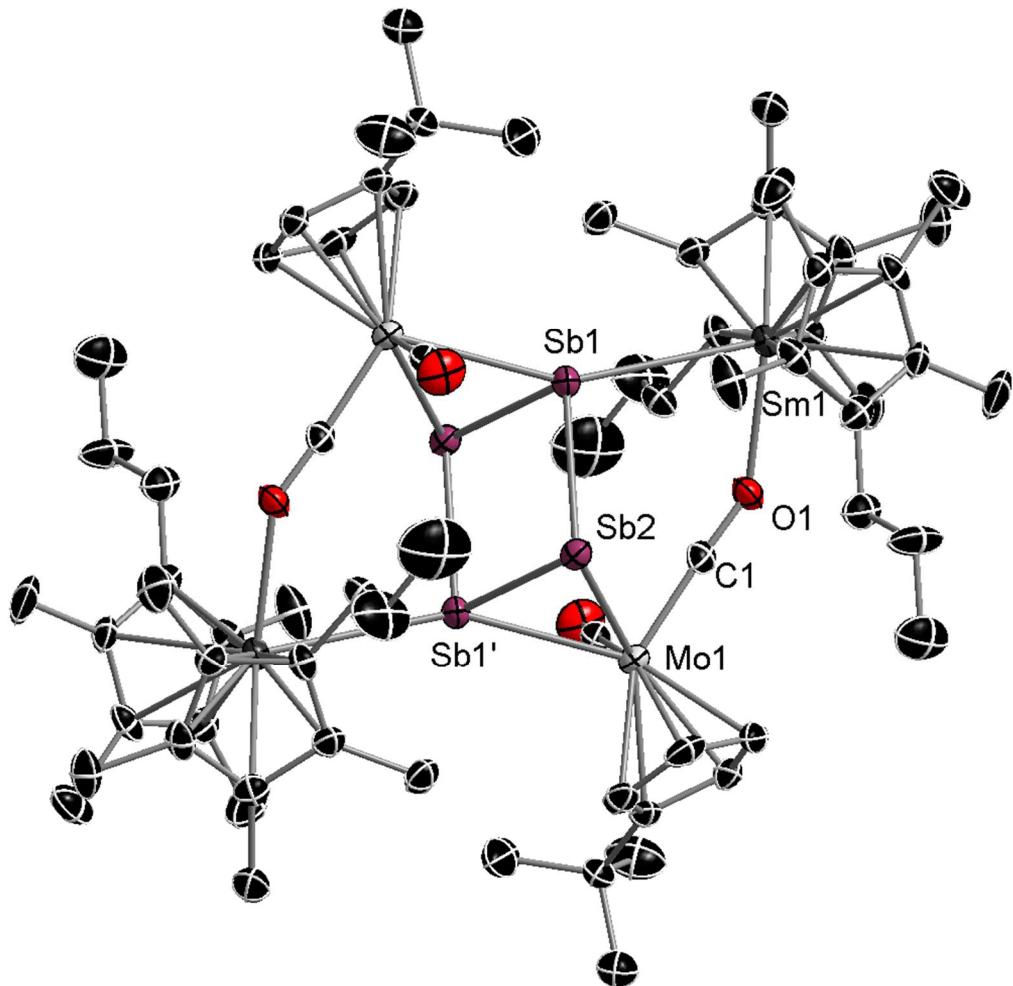


Fig. S18. Molecular structure of **5** in the solid state displayed with an ellipsoid probability of 30 %. Hydrogen atoms are omitted for clarity. Selected bond lengths [Å] and angles [°]: Sm1–Sb1 3.2794(8), Sm1–O1 2.386(7), Mo1–Sb2 2.8464(11), Mo1–Sb1' 2.8315(11), Mo1–C1 1.888(10), Sb1–Sb2 2.8618(10), Sb1'–Sb2 2.7254(10), C1–O1 1.221(12), O1–Sm–Sb1 76.83(17), Sb1'–Mo1–Sb2 57.37(3), Mo1–Sb1'–Sm1' 152.00(3), Sb1'–Sb2–Mo1 61.59(3), Sb1–Sb2–Mo1 103.46(3), Sb2–Sb1–Sb2' 90.16(3), Sb1–Sb2–Sb1' 89.84(3).

Theoretical calculations

Computational Details:

The quantum chemical RI-DFT calculations were performed by means of the program system TURBOMOLE⁵ using the RI-BP86 functional.⁶ The basis sets for each atom were of def-SV(P) quality.⁷ For Mo⁸ or Sm⁹ effective core potentials (ecp) containing 28 or 51 electrons including the f-electrons were taken. Population analyses based on occupation numbers were performed to calculate shared electron numbers (SEN) as reliable measures for covalent bonding.¹⁰

Table S3: Total energies (given in a.u.) of the molecules under discussion. The energy considerations of Fig. 5 are based on these values.

| | symmetry | P | As | Sb |
|--|-----------------|-------------|-------------|------------|
| [{CpMo(CO) ₂ } ₂ (μ, η ^{2:2} -E ₂)] | C ₂ | -1659.22071 | -5448.4499 | -987.61943 |
| [(Cp [*]) ₂ Sm) ₂ E ₂ (CpMo(CO) ₂) ₂] | C _i | -3288.27198 | -7077.4962 | -2616.6594 |
| [(Cp [*]) ₂ Sm) ₂ E ₄ (CpMo(CO) ₂) ₂] | C _i | -3970.90026 | -11549.3500 | -2627.6896 |
| [(Cp [*]) ₂ Sm) ₂ E ₂ (CpMo(CO) ₂) ₄] | C ₂ | -4264.9307 | -8054.1418 | -3593.2969 |
| E ₂ | D _{∞h} | -682.563468 | -4471.7904 | -10.95832 |
| E ₂ H ₂ | C _{2h} | | | |
| E ₂ H ₄ | C _{2h} | | | |

Table S4: E-E Bond distances (pm) and SEN values obtained from theoretical calculations.

| | P-P | | As-As | | Sb-Sb | |
|--|---------------|-----------|---------------|----------|---------------|-----------|
| | d | SEN | d | SEN | d | SEN |
| [{CpMo(CO) ₂ } ₂ (μ, η ^{2:2} -E ₂)] | 218.45 | 1.23 | 240.94 | 1.06 | 280.18 | 1.05 |
| [(Cp [*]) ₂ Sm) ₂ E ₂ (CpMo(CO) ₂) ₂] | 206.51 | 1.57 | 227.9 | 1.38 | 266.32 | 1.38 |
| [(Cp [*]) ₂ Sm) ₂ E ₄ (CpMo(CO) ₂) ₂] | 217.79/233.15 | 1.24/0.97 | 240.02/255.73 | 1.1/0.85 | 279.07/293.61 | 1.01/0.85 |
| [(Cp [*]) ₂ Sm) ₂ E ₂ (CpMo(CO) ₂) ₄] | 207.75 | 1.38 | 237.07 | 1.03 | 281.1 | 0.87 |
| E ₂ | 191.6 | 2.46 | 212.25 | 2.28 | 253.72 | 2.21 |
| E ₂ H ₂ | 205.48 | 1.73 | 227.08 | 1.59 | 268.8 | 1.56 |
| E ₂ H ₄ | 227.66 | 1.08 | 249.41 | 0.99 | 290.05 | 0.97 |

**Cartesian Coordinates of the theoretically calculated Molecules
(given in a. u.)**

[{CpMo(CO)₂}₂(μ, η^{2:2}-P₂)]

\$coord

| | | | |
|-------------------|-------------------|-------------------|----|
| -0.30834973622480 | 4.08196070174724 | 0.21989572725268 | mo |
| 0.30834973622480 | -4.08196070174724 | 0.21989572725268 | mo |
| 3.22246887805232 | 5.11469331993020 | 0.99423286581254 | c |
| -3.22246887805232 | -5.11469331993020 | 0.99423286581254 | c |
| 0.86406397276028 | 3.07043633515838 | -3.17279808796864 | c |
| -0.86406397276028 | -3.07043633515838 | -3.17279808796864 | c |
| 5.24943953927126 | 5.88328400908424 | 1.41251924929999 | o |
| -5.24943953927126 | -5.88328400908424 | 1.41251924929999 | o |
| 1.52203364781404 | 2.62877701380224 | -5.23450599479011 | o |
| -1.52203364781404 | -2.62877701380224 | -5.23450599479011 | o |
| -3.72588111475749 | 6.09967048683332 | 2.45529805208185 | c |
| -4.66670155047934 | 5.38504589977637 | 0.01159876146113 | c |
| -3.23141508213772 | 6.72757500131574 | -1.85491072893377 | c |
| -1.42355696996306 | 8.29467105979273 | -0.57772957993773 | c |
| -1.72471979585176 | 7.88532458875162 | 2.10145061390659 | c |
| -4.42884968837578 | 5.40103053285627 | 4.27889195948667 | h |
| -6.23096947074277 | 4.07609768699070 | -0.36280548438742 | h |
| -3.51299109290457 | 6.62203399810001 | -3.90881766237223 | h |
| -0.09906883857625 | 9.61421559841998 | -1.47657788763415 | h |
| -0.65601795160886 | 8.83411431593896 | 3.60630246063442 | h |
| 3.72588111475749 | -6.09967048683332 | 2.45529805208185 | c |
| 1.72471979585176 | -7.88532458875162 | 2.10145061390659 | c |
| 4.66670155047934 | -5.38504589977637 | 0.01159876146113 | c |
| 1.42355696996306 | -8.29467105979273 | -0.57772957993773 | c |
| 3.23141508213772 | -6.72757500131574 | -1.85491072893377 | c |
| 4.42884968837578 | -5.40103053285627 | 4.27889195948667 | h |
| 0.65601795160886 | -8.83411431593896 | 3.60630246063442 | h |
| 6.23096947074277 | -4.07609768699070 | -0.36280548438742 | h |
| 0.09906883857625 | -9.61421559841998 | -1.47657788763415 | h |
| 3.51299109290457 | -6.62203399810001 | -3.90881766237223 | h |
| 2.06408773875717 | -0.00613498001178 | 1.50795573608825 | p |
| -2.06408773875717 | 0.00613498001178 | 1.50795573608825 | p |

\$end

[{CpMo(CO)₂}₂(μ, η^{2:2}-As₂)]

\$coord

| | | | |
|-------------------|-------------------|-------------------|----|
| -0.37006651411595 | 4.19446113049219 | 0.17685310610832 | mo |
| 0.37006651411595 | -4.19446113049219 | 0.17685310610832 | mo |
| 3.17742963606254 | 5.23663666917989 | 0.81252768663992 | c |
| -3.17742963606254 | -5.23663666917989 | 0.81252768663992 | c |
| 0.65543901925306 | 3.17903158706586 | -3.26146053558911 | c |
| -0.65543901925306 | -3.17903158706586 | -3.26146053558911 | c |
| 5.20647849516273 | 6.04899382194473 | 1.14731076438739 | o |
| -5.20647849516273 | -6.04899382194473 | 1.14731076438739 | o |
| 1.21872500177326 | 2.76897866807403 | -5.35854434585220 | o |
| -1.21872500177326 | -2.76897866807403 | -5.35854434585220 | o |
| -3.71797189227186 | 6.20292641853547 | 2.51985205846138 | c |
| -4.71042870481563 | 5.55832892086894 | 0.07504744089007 | c |
| -3.28690407393905 | 6.92401467710171 | -1.78480202220594 | c |
| -1.43607104108615 | 8.43366494707606 | -0.50216565805544 | c |
| -1.69982560258598 | 7.96971402684038 | 2.17330672669332 | c |
| -4.40940890290394 | 5.48494117673408 | 4.34045485124214 | h |

| | | | |
|-------------------|-------------------|-------------------|----|
| -6.31135729257492 | 4.29532187263999 | -0.30298184948094 | h |
| -3.60596066437848 | 6.87311103621068 | -3.83516009625259 | h |
| -0.10820147339750 | 9.75546842058912 | -1.39278954777306 | h |
| -0.59297002975358 | 8.86977221657889 | 3.68047089214327 | h |
| 3.71797189227186 | -6.20292641853547 | 2.51985205846138 | c |
| 1.69982560258598 | -7.96971402684038 | 2.17330672669332 | c |
| 4.71042870481563 | -5.55832892086894 | 0.07504744089007 | c |
| 1.43607104108615 | -8.43366494707606 | -0.50216565805544 | c |
| 3.28690407393905 | -6.92401467710171 | -1.78480202220594 | c |
| 4.40940890290394 | -5.48494117673408 | 4.34045485124214 | h |
| 0.59297002975358 | -8.86977221657889 | 3.68047089214327 | h |
| 6.31135729257492 | -4.29532187263999 | -0.30298184948094 | h |
| 0.10820147339750 | -9.75546842058912 | -1.39278954777306 | h |
| 3.60596066437848 | -6.87311103621068 | -3.83516009625259 | h |
| 2.27608940311289 | 0.04477197673941 | 1.51208052864353 | as |
| -2.27608940311289 | -0.04477197673941 | 1.51208052864353 | as |

\$end

[{CpMo (CO) 2} 2 (μ, η^{2:2}-Sb₂)]

| | | | |
|-------------------|-------------------|-------------------|----|
| \$coord | | | |
| -0.45345131803413 | 4.42800394865069 | 0.16176160803301 | mo |
| 0.45345131803413 | -4.42800394865069 | 0.16176160803301 | mo |
| 3.09700858432799 | 5.50481188707297 | 0.65393444292648 | c |
| -3.09700858432799 | -5.50481188707297 | 0.65393444292648 | c |
| 0.40175703356391 | 3.34495297743235 | -3.30124399999736 | c |
| -0.40175703356391 | -3.34495297743235 | -3.30124399999736 | c |
| 5.10552475391943 | 6.41000555388012 | 0.89701406239714 | o |
| -5.10552475391943 | -6.41000555388012 | 0.89701406239714 | o |
| 0.84308393076595 | 2.93300288457553 | -5.42917195990210 | o |
| -0.84308393076595 | -2.93300288457553 | -5.42917195990210 | o |
| -3.73097152278013 | 6.47104541741548 | 2.57573542923309 | c |
| -4.77790492383457 | 5.85038197010962 | 0.14413404605557 | c |
| -3.36443285867382 | 7.19903340870142 | -1.73561343207828 | c |
| -1.46570113681106 | 8.67139856270526 | -0.48057491280442 | c |
| -1.69448538569756 | 8.20847393958017 | 2.20003451490318 | c |
| -4.41452656269867 | 5.77635537136996 | 4.40831583076423 | h |
| -6.42397062249803 | 4.63953911640639 | -0.20925053602742 | h |
| -3.72326534033797 | 7.16503390082723 | -3.77959686706325 | h |
| -0.13526646074734 | 9.97802870009392 | -1.38960836485444 | h |
| -0.54772448690933 | 9.08727790885893 | 3.68976155943350 | h |
| 3.73097152278013 | -6.47104541741548 | 2.57573542923309 | c |
| 1.69448538569756 | -8.20847393958017 | 2.20003451490318 | c |
| 4.77790492383457 | -5.85038197010962 | 0.14413404605557 | c |
| 1.46570113681106 | -8.67139856270526 | -0.48057491280442 | c |
| 3.36443285867382 | -7.19903340870142 | -1.73561343207828 | c |
| 4.41452656269867 | -5.77635537136996 | 4.40831583076423 | h |
| 0.54772448690933 | -9.08727790885893 | 3.68976155943350 | h |
| 6.42397062249803 | -4.63953911640639 | -0.20925053602742 | h |
| 0.13526646074734 | -9.97802870009392 | -1.38960836485444 | h |
| 3.72326534033797 | -7.16503390082723 | -3.77959686706325 | h |
| 2.64415732746676 | 0.12857455279356 | 1.59436857898121 | sb |
| -2.64415732746676 | -0.12857455279356 | 1.59436857898121 | sb |

\$end

[(Cp*₂Sm) 2P₂ (CpMo (CO) 2) 2]

| | | | |
|-------------------|-------------------|-------------------|----|
| \$coord | | | |
| -7.67113085999244 | -0.80946155383979 | -0.71715955024543 | sm |

| | | | |
|--------------------|-------------------|-------------------|----|
| -4.19944720917236 | 0.68178687293195 | -3.37853701525469 | o |
| -4.38475387979769 | 0.34690380511289 | 2.35862543939967 | o |
| -2.34403027854076 | 2.03806822948726 | -3.48961921350678 | c |
| -9.84623933852756 | 3.71518028850610 | 0.83804168929031 | c |
| -11.63040899384906 | 1.91484923501928 | -2.75273827783282 | c |
| -11.56859835070598 | 1.76972179149900 | 1.63718393230009 | c |
| -9.87383618545662 | 3.79882245986307 | -1.87228015492134 | c |
| -2.73023394249022 | 1.85594635644285 | 2.90009579079038 | c |
| -5.27718987710786 | -5.28789841080847 | -1.80145652857965 | c |
| -12.72208924093034 | 0.70335784620322 | -0.57745863046797 | c |
| -8.07555356589703 | 4.70108854598549 | 4.42647566003863 | h |
| -9.65816089591586 | -5.53684014682791 | -2.05247834486509 | c |
| -2.53160554431455 | -5.23076452041737 | -2.54606720279632 | c |
| -1.42999209523568 | -3.81299291679851 | -1.45803291246466 | h |
| -6.24153312054156 | -5.59039848236718 | 0.71629576525730 | c |
| -4.68272231103578 | -5.95403880117953 | 3.07365376711951 | c |
| -2.85794568402656 | -4.92555633954667 | 2.97953103637963 | h |
| -12.30475947020734 | 1.20310550256195 | 4.33421577947391 | c |
| -7.38805126956673 | -5.21746689447658 | -3.50731884496549 | c |
| -8.95349801304629 | -5.72855034440282 | 0.56852313552390 | c |
| -10.05642561246422 | -5.46417636775576 | 4.54655316549091 | h |
| -10.69405014447210 | 1.43363165162064 | 5.66533740717241 | h |
| -10.83422471064482 | 1.81217099325665 | -6.81039534787725 | h |
| 0.36900238993291 | 4.34698552222225 | -4.00079221938703 | mo |
| -8.54091871060639 | 5.55901399433160 | 2.56721392283781 | c |
| -12.42889945743948 | 1.54507415469061 | -5.46663340308924 | c |
| -8.55081815598245 | 5.71929526949994 | -3.50851517064921 | c |
| -0.36909915463495 | 4.34993882192857 | 3.99122854198240 | mo |
| -15.04725277544996 | -0.93864935534958 | -0.56453743256708 | c |
| -12.21306670626869 | -6.13151253681001 | -3.16213200508734 | c |
| -1.61503130699490 | -7.10030487638874 | -2.22142675972202 | h |
| -2.29114084809121 | -4.76324222177416 | -4.57585283745836 | h |
| -5.67965135583872 | -5.28809697761922 | 4.79940406481573 | h |
| -4.23400277309253 | -7.99305706408630 | 3.38055805459781 | h |
| -13.82851664164373 | 2.49856682192501 | 5.00868136993995 | h |
| -13.03253498973445 | -0.75449115625582 | 4.56364546359586 | h |
| -7.25038702564870 | -5.11881501383757 | -6.35527329249052 | c |
| -10.68543535511078 | -6.35164325798928 | 2.74703000828753 | c |
| -1.94523251558011 | 6.58103833036069 | -0.19385886091674 | p |
| 1.93902462998208 | 6.58320098462072 | 0.18210005141541 | p |
| 2.72380885612789 | 1.84717339840892 | -2.90709204499610 | c |
| 2.23397428433679 | 7.84192858285311 | -6.37535457055061 | c |
| 3.57523562830843 | 9.15954441945118 | -5.49755943267488 | h |
| -0.45429292946788 | 8.14950788854219 | -6.47922433884074 | c |
| -1.52546553888553 | 9.74140262561789 | -5.68786440249075 | h |
| 0.55363626265862 | 4.36907925475596 | -8.46843736062661 | c |
| 2.85574539647667 | 5.51182226063320 | -7.60034821033318 | c |
| 4.76652007938917 | 4.75122332081415 | -7.87651760997873 | h |
| -1.49338763224645 | 6.01792540366473 | -7.77706205209704 | c |
| -3.50117281275674 | 5.71338340707948 | -8.20506144540514 | h |
| -6.74856465574450 | 6.29039943661477 | 1.75547467901046 | h |
| -9.78095686505626 | 7.22241397236491 | 2.95420455393082 | h |
| -13.22000995277523 | -0.37049898618965 | -5.80916410580315 | h |
| -13.92049441991042 | 2.92773362904707 | -6.02896617513157 | h |
| -8.11261020065162 | 4.96238383267736 | -5.41907999735915 | h |
| -9.75860343581808 | 7.42630413649490 | -3.79233839720634 | h |
| -6.74114316070773 | 6.37393280876070 | -2.66679576144219 | h |
| 2.33226759269852 | 2.02761921897602 | 3.47964925762125 | c |
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\$end

[(Cp⁺₂Sm) ₂As₂ (CpMo (CO) ₂) ₂]

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| 3.47189799959648 | 5.63268387141185 | 8.32758853258836 | h |
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| 5.19910914930123 | -5.18235080462211 | 1.86453557965827 | c |
| 12.78122893071782 | 0.58410634607523 | 0.51646870218496 | c |
| 8.24877924429551 | 4.61474894886477 | -4.55309530633702 | h |
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| 2.44857205046702 | -5.06462091817834 | 2.57784521893478 | c |
| 1.37424154579505 | -3.67991206920059 | 1.42319705461175 | h |
| 6.18549530079661 | -5.55432735425855 | -0.63547954025934 | c |
| 4.64825922991027 | -5.93108069486738 | -3.00467085085014 | c |
| 2.88185528461053 | -4.80178411848487 | -2.98932718588224 | h |
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| 8.73207599223265 | 5.49569061823058 | -2.70923088341714 | c |
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| 12.09892211540430 | -6.13567713077602 | 3.32600030331293 | c |
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| 2.19207778949835 | -4.51609469915438 | 4.58516973005322 | h |
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\$end

[(Cp*₂Sm) ₂Sb₂ (CpMo (CO) ₂) ₂]

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| -2.42262503026516 | 2.37519618485328 | -3.53573105618317 | c |
| -10.24321573735476 | 3.56173934691955 | 1.09612440075850 | c |
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| -10.37054450255754 | 3.76804748595766 | -1.60497655590355 | c |
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| -12.88106416145273 | 0.36521026849119 | -0.38151080020017 | c |
| -8.30873594075096 | 4.49594341688918 | 4.61794558518966 | h |
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| -1.30697418260378 | -3.28367910735436 | -1.48073472302696 | h |
| -6.07660412671894 | -5.44643454703276 | 0.46814761595488 | c |
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| 2.80192909185895 | 12.34462280719612 | -7.35456262319836 | h |
| 1.06117803236533 | 10.56719374272989 | -9.64124272919426 | h |

\$end

[(Cp⁺₂Sm) ₂As₄ (CpMo (CO)₂)₂]

\$coord

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| -4.09272569790213 | 0.14236044721159 | 13.12920231360851 | h |
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| 4.09272569790213 | -0.14236044721159 | -13.12920231360851 | h |
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| 3.15378242787162 | 8.60738293012259 | -7.33283999496193 | c |
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| 2.79013119242207 | 8.75354831040690 | -2.38676494072799 | c |
| 0.89858752907096 | 9.63651835755186 | -2.60258961379115 | h |
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| 8.15333237402849 | 2.92170055307073 | -3.48198134637663 | h |
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\$end

[(Cp*₂Sm) ₂Sb₄ (CpMo (CO) ₂) ₂]

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| 0.57653888088139 | -0.05920707260514 | 10.73103827583295 | c |
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| 2.71008248673173 | -3.89853525145930 | 10.65529124981851 | c |
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| -2.82080579232591 | -1.03807619864156 | 14.19550013072626 | c |

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| 0.40725002966870 | -6.2055222034233 | 14.35246876694171 | c |
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| 4.11099070493279 | 1.40416222940307 | 6.92247031553395 | h |
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| 8.73451939114348 | -6.15694565431192 | -4.81690632967977 | h |
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| 2.24877743339870 | 12.79198877561821 | -8.36095471115043 | h |
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\$end

[(Cp^{*}₂Sm) ₂P₂ (CpMo (CO)₂)₄]

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\$end

[(Cp⁺₂Sm) ₂As₂ (CpMo (CO)₂)₄]

\$coord

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| 3.10596286501576 | -13.38241352792335 | 11.15571784254987 | h |
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P₂

\$coord

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\$end

As₂

\$coord

| | | | |
|-------------------|------------------|------------------|----|
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Sb₂

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