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Supporting Information

Dinitrogen Fixation: Rationalizing Strategies Utilizing Molecular Complexes

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1. Summary of used complexes and reagents for N₂ to metal-nitride formation.

Table S1. Summary of used complexes and reagents for metal nitride formation.

Complex	Reagent	Selectivity^[a]	Ref.
Mo(N(^t BuAr) ₃) ₂ (1)	N ₂ <i>hν</i>	[Mo(N)(N(^t BuAr) ₃) (2)	[1] [2]
[Mo(OSi(O ^t Bu) ₃) ₃] (3)	N ₂	[Mo(N)(OSi(O ^t Bu) ₃) ₃] (4)	[3]
[Ti(Cp')(CH ₂ SiMe ₃) ₃] (90) Cp' = C ₅ Me ₄ SiMe ₃	N ₂ /H ₂	{[Ti(Cp')] ₄ (μ ₃ -NH) ₂ (μ ₂ -H) ₄ }	[4]
{[Ti(Cp')] ₃ (μ ₃ -H)(μ ₂ -H) ₆ } (5) Cp' = C ₅ Me ₄ SiMe ₃	N ₂	{[(Cp')Ti] ₃ (μ ₂ -NH)(μ ₃ -N)(μ ₂ -H) ₂ } (6)	[4]
{[Ti(Cp')] ₃ (μ ₂ -NH)(μ ₃ -N)(μ ₂ -H) ₂ }	¹⁵ N	{[(Cp')Ti] ₃ (μ ₃ - ¹⁵ N)(μ ₂ -NH) ₂ (μ ₂ - ¹⁵ NH)}	[4]
{[(<i>anti</i> -O ₃)Nb(μ-H) ₄][K(dme)] ₂ } (11) <i>anti</i> -O ₃ = CH{3,5-(^t Bu) ₂ Ph-2-O} ₃ ³⁻	N ₂	{[(<i>anti</i> -O ₃)Nb(μ-N) ₂][K(thf)] ₂ } (12)	[5]
[(<i>calix</i> -O ₄) ₂ Nb ₂] (13)	N ₂ N ₂ , TMEDA	[(<i>calix</i> -O ₄)Nb(N) ₂][Nb(<i>calix</i> -O ₄)] (14) [(<i>calix</i> -O ₄) ₂ Nb ₂ (μ-NNa(TMEDA)) ₂] (15)	[6, 7]
[Mo ₂ (PCP)] (18) PCP = 1,3-[OP(^t Bu) ₂] ₂ C ₆ H ₃ ⁻	Na(Hg)/N ₂	[Mo(N)(PCP)] (19)	[8]
[Mo(PNP)] ₃ (20) PNP = 1,3-[CH ₂ P(^t Bu) ₂] ₂ C ₅ H ₃ N	(CoCp* ₂)/N ₂ SmI ₂ /N ₂	[Mo(N)(PNP)] (21)	[9, 10]
[Mo(PNP)Cl ₃] (24) PNP = N(CH ₂ CH ₂ P ^t Bu ₂) ₂ ⁻	i.) Na(Hg)/N ₂ ii.) HOTf	[Mo(HPNP)(N)Cl] ⁺ (23)	[11]
[W(PNP)Cl ₃] PNP = N(CH ₂ CH ₂ P ^t Bu ₂) ₂ ⁻	Na(Hg)/N ₂	[W(HPNP)(N)Cl] ⁺	[12]
[Mo(PPP)Cl ₃] (29) PPP = PhP(CH ₂ CH ₂ PCy ₂) ₂	Na(Hg), NaI/N ₂	[Mo(N)(PPP)] (31)	[13]
[ReCl ₂ (PNP)] (32) PNP = N(CH ₂ CH ₂ P ^t Bu ₂) ₂ ⁻	CoCp* ₂ /N ₂ Na(Hg)/N ₂ <i>E</i> _{red} = -1.9 V vs. Fc/Fc ⁺	[Re ^v Cl(N)(PNP)] (33)	[14]
[ReCl ₂ (PNP)] (35) PNP = N(CH ₂ CH ₂ P ^t Bu ₂) ₂	CoCp* ₂ /N ₂ Na(Hg)/N ₂ <i>E</i> _{red} = -1.67 V vs. Fc/Fc ⁺	[Re ^v Cl(N)(PNP)] (36)	[15]
[(NNN ^{Si} V) ₂ (μ-Cl) ₂] (37) NNN ^{Si} = Me ₃ SiN(CH ₂ CH ₂ NSiMe ₃) ₂ ²⁻	KC ₈ /N ₂	[(NNN ^{Si} V) ₂ (μ-N) ₂] (38) [(NNN ^{Si} V) ₂ (μ-N) ₂ K] ⁻ (39)	[16]
[U(N ⁴)(DME)] ⁺ (40)	K-C ₁₀ H ₈ /N ₂	[(UN ⁴) ₂ (μ-NK) ₂] ²⁻ (41)	[17]

N ⁴ = Et ₃ -calix[4]tetrapyrrole			
[(O-μ-OO) ₂ Nb ₂] (42) OOO = {(3-Me,5- ^t Bu)Ph-(2-O)CH ₂ Ph(2-O)(4- ^t Bu)CH ₂ Ph-(2-O)(3-Me,5- ^t Bu)} ³⁻	LiBEt ₃ H/N ₂	{[(OO-μ-O)Nb] ₂ (μ-N) ₂ (Li-thf) ₂ } (43)	[18]
[(O-μ-NO) ₂ V ₂] (46) ONO = {(3-Me,5- ^t Bu)Ph-(2-O)CH ₂ Ph(2-NPhMe)(4- ^t Bu)CH ₂ Ph-(2-O)(3-Me,5- ^t Bu)} ³⁻	KH/N ₂	{[(ONO)V] ₂ (μ-N) ₂ (K-dme) ₂ } ²⁻ (47)	[19]
[Ti(N ₄ ^{Si})Cl] (48) N ₄ ^{Si} = {N(CH ₂ CH ₂ NSiMe ₃) ₃ } ³⁻	Mg/N ₂	{[Mg(N ₄ ^{Si})](μ-NTi) ₂ (Mg(N ₄ ^{Si}))} (50)	[20]
[[Fe(β-diketamin)] ₂ (μ-Cl) ₂] (52) β-diketamin = (2,6-Me ₂ Ph)NC(Me)CHC(Me)N(2,6-Me ₂ Ph)	KC ₈ /N ₂	{[Fe(β-diketamin)] ₂ (μ-N){(μ-N)K ₂ Cl ₂ Fe(β-diketamin)}]} (51)	[21]
	Na/N ₂	{[Fe(β-diketamin)] ₂ (μ ₂ -N){K(thf) ₂ (μ ₃ -N)Fe(β-diketamin)}]} (53)	[22]
[Fe ₃ (cyclophane-β-diketamin)Br ₂] (54) cyclophane-β-diketamin = (2,4,6-Et-Ph-1,3,5-CH ₂ -NC(Me)CHC(Me)N) ₃	KC ₈ /N ₂	[Fe ₃ (μ-NH) _{1,2} (cyclophane-diketamin)] (55)	[23]
[Cp*Ta{N(Pr)C(R)N(Pr)Cl ₃ }] (57) Cp* = η ⁵ -C ₅ Me ₅ , R = Me, NMe ₂ , Ph	KC ₈ /N ₂	{[Cp*Ta{N(Pr)C(R)N(Pr)}(μ-N)] ₂ } (56)	[24]
[Cp*Nb{N(Pr)C(R)N(Pr)Cl ₃ }] (58) R = Me, Ph	KC ₈ /N ₂	{[Cp*Nb{N(Pr)C(R)N(Pr)}(μ-N)] ₂ } (59)	[25]
{[Cp*MNEtC(R)NEt] ₂ (μ-N ₂)] M = Mo (60), W (61); R = Ph	Na(Hg)/N ₂	{[Cp*MNEtC(R)NEt] ₂ (μ-N) ₂ }	[26]
[MoCl ₄ -DME]	i.) MesMgBr/N ₂ ii.) <i>hν</i>	[Mo(Mes) ₃ -N=Mo(Mes) ₃] (63) Mes = 2,4,6-Me ₃ (C ₆ H ₂)	[27]
{Re(P'NP')Cl ₃ } P'NP' = (2,6-OP'Pr ₂)C ₅ H ₃ N	i.) LiBHEt ₃ /N ₂ ii.) <i>hν</i> (λ = 405)	{[Re(N)(P'NP')Cl]} (66)	[28]
{[MoCp*(depf)] ₂ (N ₂)} (67)	<i>hν</i> (λ > 400)	{[MoCp*(depf)(N)]} (68)	[29]
{[Cp*M{N(Pr)C(R)N(Pr)}] ₂ (μ-N ₂)] M = Mo (69), W (70); R = Me	<i>hν</i>	{[Cp*M{N(Pr)C(R)N(Pr)}] ₂ (μ-N) ₂ } (M = Mo, 71 ; M = W, 72) [[Cp*M(N){N(Pr)C(R)N(Pr)}] ₂] (73)	[30]
<i>trans</i> -[Mo(depe) ₂ (N ₂) ₂] (74) depe = Et ₂ PCH ₂ CH ₂ PET ₂	Oxidation, E _{red} = 0.5 V vs. Pt-wire	[Mo(depe) ₂ (N)] (75)	[31]

2. Summary of used complexes and reagents to functionalize nitride originating from N₂

Table S2. Summary of used complexes and reagents for nitride functionalization.

Nitride Complex	Reagent	Selectivity ^{bl}	Ref.
[Mo ^{VI} (N ^t BuAr) ₃ (N)] (2)	Mel	[Mo ^{VI} (N ^t BuAr) ₃ (NR)] ⁺	[32]
[Nb ^V (N ^t BuAr') ₃ N][Na] (77)	RCOCl	RCN	[33]
[Mo ^{VI} (N ^t BuAr) ₃ (¹⁵ N)] (2- ¹⁵ N)	(CF ₃ CO) ₂ O	CF ₃ CO ¹⁵ NH ₂	[34]
[Mo ^{VI} (N ^t BuAr) ₃ (N)] (2)	RC(O)Cl + R' ₃ SiOTf + Mg(anthracene) + MCl ₂ (M = Sn, Zn)	RCN (R = Me, ^t Bu, Ph)	[35]
[(ONO) ^V (μ-N)] ₂ [K(dme)] ₂ (47)	CO or isocyanide	potassium isocyanate 81 or carbodiimide complex 82 .	[19]
[[(anti-O ₃)Nb ^V] ₂ (μ-N)] ₂ [K(thf)] ₂ (12)	Mel, CO ₂	Imide, ureate complex 79 .	[5, 36]
[(Cp*Mo[N(Pr)C(Me)N(Pr)]) ₂ (μ-N ₂)] (69)	R ₃ ECl (R ₃ E = Me ₃ Si, Ph ₃ Si, Me ₃ Ge or Me ₃ C) + CO or CO ₂	RNCO	[37]
[Re ^V (PNP)Cl(N)] (33) (PNP = ((^t Bu) ₂ P(CH ₂ CH ₂)N)	MeOTf	Imide complex 83 .	[38]
[Re ^V (PNP)Cl(N)] (33) (PNP = ((^t Bu) ₂ P(CH ₂ CH ₂)N)	EtOTf, (KN(SiMe ₃) ₂), NCS	MeCN	[39]
[Re ^V (PNP)Cl(N)] (33) (PNP = ((^t Bu) ₂ P(CH ₂ CH ₂)N)	i.) BnBr, AgOTf, DTBMP, ii.) (KN(SiMe ₃) ₂) iii.) NCS	PhCN	[40]
[Re ^V (PNP)Cl(N)] (33) (PNP = ((Pr) ₂ P(CH ₂ CH ₂)N)	i.) hν (390 nm) ii.) PhC(O)Cl iii.) E = -1.67 V, N ₂	PhCN + PhC(O)NH ₂	[41]
[(Cp'Ti) ₄ (μ ³ -NH) ₂ (μ ³ -N)] (89) (Cp' = C ₅ Me ₄ SiMe ₃)	RC(O)Cl (R = aryl, Bn, Me, ^t Bu)	RCN	[42]
[Mo ^{IV} (PPP ^{Cy})(N)(I)] (31)	1,2-bisdimethylsilyethane	Bis(silyl)amine	[13]
[(Ta ^V (NN)Cp*)(μ-N)] ₂ (56)	PhSiH ₃	Silylimido complex 92 .	[24]
[(Mo ^{II} (NN)Cp*)(μ-N)] ₂ (60) ^a	Me ₃ SiCl + ^t PrOH	HN(SiMe ₃) ₂	[43]

[a] *in-situ* generation of the metal nitride

3. Summary of used complexes and reagents for stoichiometric N₂ protonation and N₂-hydrogenation

Table S3. Summary of used complexes and reagents for stoichiometric N-H bond formation.

Complex	Reagent	Selectivity ^[b]	Ref.
<i>trans</i> -[W(N ₂) ₂ (dppe) ₂] (94) dppe = Ph ₂ PCH ₂ CH ₂ PPh ₂	HX (X = Cl, Br)	<i>trans</i> -[W(NH-NH)(dppe) ₂ X] ⁺ (97)	[44]
<i>trans</i> -[Mo(N ₂) ₂ (depe) ₂] (95) depe = Et ₂ PCH ₂ CH ₂ PEt ₂	HX (X = Cl, Br)	<i>trans</i> -[Mo(NH-NH)(dppe) ₂ X] ⁺ (98)	[45]
<i>trans</i> -[W(N ₂) ₂ (depf) ₂] (103) depf = Et ₂ PFcPEt ₂	HOTf	<i>trans</i> -[W(N-NH ₂)(depf) ₂ OTf] ⁺ (101)	[46]
<i>trans</i> -[W(N ₂) ₂ (depr) ₂] (104) depr = Et ₂ PRuCP ₂ PEt ₂	HOTf	<i>trans</i> -[W(N-NH ₂)(depr) ₂ OTf] ⁺	[47]
<i>trans</i> -[W(N ₂) ₂ (depc) ₂] (105)	HOTf	<i>trans</i> -[W(N-NH ₂)(depc) ₂ OTf] ⁺⁺	[48]
<i>trans</i> -[Mo(N ₂) ₂ (depf) ₂] (106)	HOTf	<i>trans</i> -[Mo(N-NH ₂)(depf) ₂ OTf] ⁺ (102)	[46]
<i>trans</i> -[Mo(N ₂) ₂ (depc) ₂] (107)	HOTf	<i>trans</i> -[Mo(N-NH ₂)(depc) ₂ OTf] ⁺	[48]
<i>trans</i> -[W(N ₂) ₂ (depf)(PPh ₂ Me) ₂] (108)	H ₂ SO ₄	NH ₄ ⁺	[46]
<i>trans</i> -[Mo(N ₂) ₂ (depf)(PPh ₂ Me) ₂] (110)	H ₂ SO ₄	NH ₄ ⁺	[46]
<i>trans</i> -[W(N ₂) ₂ (depc)(PPh ₂ Me) ₂] (109)	H ₂ SO ₄	NH ₄ ⁺	[48]
<i>trans</i> -[W(N ₂) ₂ (PPh ₂ Me) ₄] (121)	HCl	<i>trans</i> -[W(NH=NH)(Me ₂ PPh) ₄ Cl] ⁺ (122)	[45]
<i>trans</i> -[W(N ₂) ₂ (MePPh ₂) ₄] (121)	H ₂ SO ₄	NH ₃	[49]
<i>cis</i> -[W(N ₂) ₂ (Me ₂ PPh) ₄] (123)	H ₂ SO ₄	NH ₃	[49]
<i>cis</i> -[Mo(N ₂) ₂ (Me ₂ PPh) ₄] (124)	H ₂ SO ₄	NH ₃ , NH ₂ -NH ₂	[50]
<i>cis</i> -[Mo(N ₂) ₂ (PPP)PPh ₃] (125) PPP = <i>bis</i> -(2-diphenylphosphinoethylene)phenylphosphine	HBr	NH ₃	[51]
<i>cis</i> -[Mo(N ₂) ₂ (PPP)(depf)] (127)	HBr	NH ₃	[52]
<i>trans</i> -[Cr(N ₂) ₂ (dmpe) ₂] dmpe = Me ₂ PCH ₂ CH ₂ PMe ₂	HOTf	[Cr(N-NH ₂)(dmpe) ₂ (OTf)] ⁺	[53]
[Cr(N ₂) ₂ (P ^{Ph} ₄ N ^{Bn} ₄)] (130) P ^{Ph} ₄ N ^{Bn} ₄ = 1,5,9,13-benzylaza-3,7,11,15-phenylphosphocyclohexadecane	HOTf HOTf, CoCp ₂	[Cr(N ₂ H)(P ^{Ph} ₄ N ^{Bn} ₄)] ⁺ (131) (DFT) N ₂ H ₅ ⁺ , NH ₄ ⁺	[54]
[Cr(N ₂) ₂ (P ^{Ph} ₄ N ^{Bn} ₄)] (130)	TEMPOH	NH ₃	[55]
[Fe(N ₂)(DMeOPrPE) ₂] (113)	HOTf	[Fe(N-NH ₂)(DMeOPrPE) ₂] ⁺	[56]

[Fe(N ₂)(dppe) ₂] (112)	HCl	NH ₃ , NH ₂ -NH ₂	[57]
[Fe(dmpe) ₂ (N ₂)] (111)	H(OEt ₂) ₂ OTf/THF H(OEt ₂) ₂ OTf/Pentane	NH ₃ NH ₂ -NH ₂	[58]
[(Cp ⁺) ₂ Zr(N ₂)-(μ ² -N ₂)-Zr(N ₂)(Cp ⁺) ₂] (133)	HCl/Toluene	NH ₂ -NH ₂	[59]
[(NEt ₂ CS ₂) ₃ Nb] ₂ (μ-N ₂) (134)	HCl	NH ₂ -NH ₂ , NH ₃	[60, 61]
[(NEt ₂ CS ₂) ₃ Ta] ₂ (μ-N ₂) (135)	HCl	NH ₂ -NH ₂ , NH ₃	
{[(Me ₃ Si) ₂ N] ₂ (thf)Y} ₂ (μ-η ² :η ² -N ₂ [K(thf) ₆]) (138)	[Et ₃ NH][BPh ₄]	{[(Me ₃ Si) ₂ N] ₂ (thf)Y} ₂ (μ-N ₂ H ₂) (139)	[62]
{[bis-(2 ⁻ Bu-4-Me-phenolate)methylene]Th(dme)Cl} (140)	N ₂ /K-naphthalinide	{[bis-(2 ⁻ Bu-4-Me-phenolate)methylene]Th(dme)NH ₂ } (141)	[63]
M ₂ (mTP) ₂ (M = U, 142 and M = Th, 143) mTP = [{2-(OC ₆ H ₂ -Bu-2,4) ₂ CH}-C ₆ H ₄ -1,3] ⁺	KC ₈ /N ₂	K ₄ [M ₂ (μ-N ₂ H ₂)(mTP) ₂] (M = U, 144 ; Th, 145)	[64]
[K ₃ {[U(OR) ₃] ₂ (μ-N) (μ-η ² :η ² -N ₂)}] (R = Si(O ^t Bu)) (147)	HCl, H ₂	NH ₃	[65]
<i>cis</i> -[W(N ₂) ₂ (PPh ₂ Me) ₄] (123)	ZrHCPCl (148) CoH(CO) ₄ (151) HFeCo(CO) ₁₂ (149) (H) ₂ Fe(CO) ₄ (150)	NH ₃ , NH ₂ -NH ₂	[66]
<i>trans</i> -[W(N ₂) ₂ (dppe) ₂] (93)	Ru(H) ₂ Cp(dtfpe)Cl (152)	NH ₃	[67]
<i>trans</i> -[W(N ₂) ₂ (dppe) ₂] (93)	[Ru(H ₂)(dppe) ₂ Cl] ⁺ (153)	NH ₃	[68]
<i>cis</i> -[W(N ₂) ₂ (PPh ₂ Me) ₄] (123)	{[(Cp ⁺ M) ₂ (μ-SH) ₃] {M = Ir, Rh} (154 , 155) [(PPP-Fe) ₂ (μ-SH) ₃] (156) {PPP = MeC(CH ₂ PPh ₂) ₃ }	NH ₃	[69]
[Cr(diaminopyridine)Cl] (157)	NaH	[Cr(N-H)(diaminopyridine)Na(thf)] (158)	[70]
[(P ₂ N ₂)Zr-(μ-N ₂)-Zr(P ₂ N ₂)] (159) [P ₂ N ₂ = {(Ph-PCH ₂ SiMe ₂) ₂ N] ₂ }	H ₂	[(P ₂ N ₂)Zr-(μ-N-NH)-Zr(P ₂ N ₂)] (160)	[71]
[(CpMe ₄ H) ₂ Zr-(μ-N ₂)-Zr(CpMe ₄ H) ₂] (161)	H ₂	NH ₃	[72]
[(PNP)Ti-(μ-H) ₄ -Ti(PNP)] (163)	-	[(PNP)HTi-(μ-N)(μ-NH)-Ti(PNP)] (165)	[73]
[Ta(H) ₄ (SiO) _n] (167)	-	[(≡OSi) ₂ Ta(=NH)(NH ₂)] (169)	[74]

4. Summary of used complexes and reagents for stoichiometric N₂ silylation.

Table S4. Summary of used complexes and reagents for stoichiometric silylation of metal-dinitrogen complexes.

Complex	Reagent	Selectivity	Ref.
[Mo(N ₂) Cp*(depf)] (170)	Me ₃ SiCl	[Mo(N=N-SiMe ₃) Cp*(depf)] (171)	[75]
Mg(thf) ₄ [Co(PhBP ^{iPr} ₃)(N ₂) ₂] (172) [P ₃ ^B = Tris-[2-(diisopropylphosphino)-phenyl]borane]	Me ₃ SiCl	[(PhBP ^{iPr} ₃)CoN=N-TMS] (174)	[76]
[Co(EP ^{Ph} ₃)Cl] (E=N, 175 ; E=CMe, 176) EP ^{Ph} ₃ = E(CH ₂ CH ₂ PPh ₂) ₃		[Co(XP ^{Ph} ₃)(N=NTMS)] (E=N, 177 ; E=CMe, 178)	[77]
[Fe(P ₃ ^B)Cl] (179) (P ₃ ^B = [tris-(2-diisopropylphosphino)-borane])	N ₂ /Na(Hg) ClMe ₂ SiCH ₂ CH ₂ SiMe ₂ Cl	[(P ₃ ^B)Fe=N-N(Me ₂ SiCH ₂ CH ₂ SiMe ₂)] (181)	[78]
[Fe ₂ (P ₂ ^B) ₂ (N ₂)] (183) (P ₂ ^B = [bis-(2-diisopropylphosphino)-phenylborane])	Na/ClMe ₂ SiCH ₂ CH ₂ SiMe ₂ Cl	[(P ₂ ^B)Fe=N-N(Me ₂ SiCH ₂ CH ₂ SiMe ₂)] (184)	[79]
[Fe(N ₂)AltraPhos] ⁻ (188) AltraPhos = Al[N(o-C ₆ H ₄ NCH ₂ PIPr ₂) ₃]	KC ₈ /ClMe ₂ SiCH ₂ CH ₂ SiMe ₂ Cl	[Fe{N ₂ (SiMe ₂ CH ₂) ₂ }AltraPhos] (189)	[80]
[(μ-η ¹ :η ² -N ₂){Ta(NPN)H} ₂] (190) [NPN = (PhNSiMe ₂ CH ₂) ₂ PPh]	RSiH ₃ {R = ⁿ Bu, Ph}	[[Ta(NPN)] ₂ (μ-NSiH ₂ R) ₂] (194 , 195) (R = Ph, ⁿ Bu)	[81]
[[{η ⁵ -C ₅ H ₂ -1,2,4-Me ₃] ₂ Hf] ₂ (μ ₂ ,η ² ,η ² -N ₂)] (196)	CySiH ₃	[[{η ⁵ -C ₅ H ₂ -1,2,4-Me ₃] ₂ Hf](μ-H){μ,η ¹ ,η ² -N(SiH ₂ Cy)N}] (197)	[82]

5. Summary of used complexes and reagents for stoichiometric N-C bond formation.

Table S5. Summary of used complexes and reagents for stoichiometric N₂-functionalization for N-C bond formation.

Complex	Reagent	Selectivity	Ref.
<i>trans</i> -[W(N ₂) ₂ (dppe) ₂] (94)	RCOBr {R = Me, Et, Ph, <i>p</i> -OMePh}	<i>trans</i> -[W(N-NHC(O)R)(dppe) ₂ Br ₂] (201-204) (R = Et, Ph, <i>p</i> -OMePh)	[83]
<i>trans</i> -[Mo(N ₂) ₂ (dppe) ₂] (95)	RCOCl {R = Me, Et}	<i>trans</i> -[Mo(N-NHC(O)R)(dppe) ₂ Br ₂] (205-206)	[83]
<i>trans</i> -[W(N ₂) ₂ (dppe) ₂] (94)	R-X (R = Me, Et, ⁿ Pr, ⁱ Pr, ^t Bu., X = Cl, Br)	[(dppe) ₂ W(-N=NR)] {R = Me, Et, ⁿ Pr, ⁱ Pr, ^t Bu} (207-211) [(dppe) ₂ W(=N-NR ₂)] {R = Me} (214)	[84]
<i>trans</i> -[Mo(N ₂) ₂ (dppe) ₂] (95)	R-X {R = Me, Et}	[(dppe) ₂ Mo(-N=NR)] {R = Me, Et} (212-213)	[85]
<i>trans</i> -[W(N ₂) ₂ (dppe) ₂] (94)	ClCH ₂ CH ₂ CH ₂ CH ₂ Cl	[(dppe) ₂ W(=N-N(CH ₂) ₄)] (215)	[86]
[(PhMe ₂ P) ₃ Re(N ₂)Cl ₂]	RCOCl (R = Me, Ph)	[(PhMe ₂ P) ₃ Re(N ₂ COR)Cl ₂]	[84]
[M(PhB(CH ₂ P ^{Pr}) ₃)Cl]{M = Fe, Co} (216, 173)	N ₂ /Mg/MeOTs	M(PhB(CH ₂ P ^{Pr}) ₃)(N=NMe), M = Fe (218), Co (217)	[76]
[Fe(P ₃ ^X)L] (L = Cl, Br) (219-220) (X = B, Si)	N ₂ /Na(Hg)/MeOTf N ₂ /KC ₈ /MeOTf	Fe(P ₃ ^X)(N-NMe ₂) (221-222)	[87]
[Zr((η ⁵ -C ₅ Me ₄ R) ₂ SiMe ₂) ₂ (μ-η ² ,η ² -N ₂)] (223)	CO ₂	[Zr((η ⁵ -C ₅ Me ₄ R) ₂ SiMe ₂) ₂ (μ-η ² ,η ² -OC(O)N-NOCO)] (224)	[88]
[[{(η ⁵ -C ₅ Me ₄ H) ₂ Hf] ₂ (μ ₂ -η ² ,η ² -N ₂)] (225)	CO ₂	[[{(η ⁵ -C ₅ Me ₄ H) ₂ Hf] ₂ {NN(CO) ₂ }] (226)	[89]
[[{(η ⁵ -C ₅ Me ₄ H) ₂ Hf] ₂ (μ ₂ -η ² ,η ² -N ₂)] (225)	PhNCO	[[{(η ⁵ -C ₅ Me ₄ H) ₂ Hf] ₂ (N ₅ C ₃ O ₃ Ph ₃)]	[89]
[[{(η ⁵ -C ₅ Me ₄ H)Sc((NEt) ₂ CNEt ₂) ₂ (μ ₂ -η ² ,η ² -N ₂)] (227)	EtBr	[[{(η ⁵ -C ₅ Me ₄ H)Sc((NEt) ₂ CNEt ₂) ₂ (μ ₂ -η ² ,η ² -N ₂ Me)]	[24]
[[{(η ⁵ -C ₅ Me ₄ H)Sc((N ⁱ Pr) ₂ C ^t Bu) ₂ (μ ₂ -η ² ,η ² -N ₂)] (228)	Xs MeOTf	[[{(η ⁵ -C ₅ Me ₄ H)Sc((N ⁱ Pr) ₂ C ^t Bu) ₂ (μ ₂ -η ² ,η ² -(MeN-NMe))] (229)	[90]
[(μ-η ¹ :η ² -N ₂)Ta ₂ (NPN) ₂ (μ-H) ₂] (190)	PhCH ₂ Br	[(NPN)Ta-(η ¹ -NCH ₂ Ph:μ ² -N)(μ-H) ₂ TaBr(NPN)] (231)	[91]
[(μ-η ¹ :η ² -N ₂)Ta ₂ (NPN) ₂ (μ-H) ₂] (190)	Ph-N=C=N-Ph X=N=C=S	[(NPN)Ta-(μ-η ¹ :η ² -{κ ² -N(Ph)C(N-Ph)N-N})(μ-H) ₂ TaBr(NPN)] [(NPN)(X=N=C=N)Ta-(μ-N)(μ-S)-Ta(NPN)]	[92]
[(P ₂ N ₂)Zr-(μ-N ₂)-Zr(P ₂ N ₂)] (159)	H-C≡C-Ar Ar = Ph, <i>p</i> -MePh, <i>p</i> - ^t BuPh	[(P ₂ N ₂)Zr-(μ-η ² :η ² -N-N(CH=CHAr))(μ-C≡CPh)-Zr(P ₂ N ₂)] (233-235)	[93]
[(O ₃ C) ₂ Ti ₂ (μ-N ₂)K ₃ (thf) ₃][K(thf) ₆] (236) O ₃ C = (σ-OPh) ₃ C ⁺	CO ₂ ^t BuNCO	[(O ₃ C) ₂ Ti ₂ {μ-(κ ² :κ ² -(OCO) ₂ N-N(COO-))}] (237) [(O ₃ C) ₂ Ti ₂ {μ-(κ ² :κ ² -(N ^t BuCON) ₂ }] (238)	[94]

	PhC=C=CH ₂	[(O ₃ C) ₂ Ti ₂ {μ-η ² :η ² -NN(C(CH ₃)CHPh)}] (239)or, [(O ₃ C) ₂ Ti ₂ {μ-η ² :η ² -NN(C(CH ₂)CH ₂ Ph)}] (240)	
[K ₃ {[U(OR) ₃] ₂ (μ-N) (μ-η ² :η ² -N ₂)}] (R = Si(O ^t Bu)) (147)	CO	KCN + [K ₂ {[U(OR) ₃] ₂ (μ-O)(μ-NCO) ₂ }] (241)	[65]
[K ₂ {[U(OR) ₃] ₂ (μ-O) (μ-η ² :η ² -N ₂)}] (243) (R = Si(O ^t Bu))	CO	[K ₂ {[U(OR) ₃] ₂ (μ-O) ₂ (μ-NCN)}] (244) +'NCNCO' (<i>not identified</i>)	[95]

6. Summary of complexes and reagents for catalytic N₂ protonation.

Table S6. Molecular catalysts for N₂ protonation, conditions for highest TON reported

Catalyst	Reductant (equiv.) ^a	Proton source (equiv.)	Solvent	Reaction conditions	TON ^a	Ref.
[[{(Tren ^{TMS})Ti] ₂ (μ-η ¹ :η ¹ :η ² :η ² N ₂ K ₂)] (49)	KC ₈ (600)	[Cy ₃ PH]I (600)	Et ₂ O	1 atm of N ₂ , - 78 °C	9	[96]
[V(OXyl)(PNP')(N ₂)] (262)	KC ₈ (200)	[HOEt ₂][BAR ^F ₄] (184)	Et ₂ O	1 atm of N ₂ , - 78 °C	6	[97]
Mg[Mg ₂ Mo ₈ O ₂₂ (OMe) ₈ (MeOH) ₄] (245)	Na/Hg	solvent	Methanol	1 atm of N ₂ , 20 °C	1000 ^b	[98]
[Mo(N ₂)(HIPTN ₃ N)] (246)	CrCp ⁺ ₂ (36)	[LutH][BAR ^F ₄]	Hexane	1 atm of N ₂ , 25 °C	3.78	[99]
[Mo(N ₂) ₂ (PNP)] ₂ (μ-N ₂) (22)	CoCp ₂ (72)	[LutH][OTf] (96)	Toluene	1 atm of N ₂ , 25 °C	11.5	[100]
[Mo(N)(PNP)Cl][OTf] (253)	CoCp ₂ (36)	[LutH][OTf] (48)	Toluene	1 atm of N ₂ , 25 °C	3	[101]
[Mo(N ₂) ₂ (p-OMe-PNP)] ₂ (μ-N ₂) (254)	CoCp ₂ (216)	[LutH][OTf] (288)	Toluene	1 atm of N ₂ , 25 °C	17	[102]
[Mo(Cl ₃)(p-Ofc-PNP)](255)	CoCp ⁺ ₂ (360)	[LutH][OTf] (480)	Toluene	1 atm of N ₂ , 25 °C	41.5	[103]
[Mo(N ₂) ₂ (p-Fc-PNP)] ₂ (μ-N ₂) (256)	CoCp ₂ (216)	[LutH][OTf] (288)	Toluene	1 atm of N ₂ , 25 °C	18.5	[104]
[Mo(Cl ₃)(p-Ofc(PNP) ₂)](257)	CoCp ⁺ ₂ (360)	[LutH][OTf] (480)	Toluene	1 atm of N ₂ , 25 °C	22	[103]
[Mo(N)(PPP)Cl] (26)	CoCp ⁺ ₂ (36)	[CoIH][OTf] (48)	Toluene	1 atm of N ₂ , 25 °C	5.5	[105]
[Mo(N ₂) ₂ (PCP)] ₂ (μ-N ₂) (263)	CrCp ⁺ ₂ (72)	[LutH][OTf] (96)	Toluene	1 atm of N ₂ , 25 °C	58	[106]
[Mo ₃ (PNP)] (20)	CoCp ⁺ ₂ (180)	[CoIH][OTf] (180)	Toluene	1 atm of N ₂ , 25 °C	26	[9]
[Mo(N)(PNP)] (21)	CoCp ⁺ ₂ (180)	[CoIH][OTf] (180)	Toluene	1 atm of N ₂ , 25 °C	6.1	[9]
[Mo ₃ (PNP)] (20)	Sml ₂ (360)	HOCH ₂ CH ₂ OH (360)	THF	1 atm of N ₂ , 25 °C	21.4	[10]
[Mo ₂ (PNP')] (258)	Sml ₂ (thf) ₂ (180)	HOCH ₂ CH ₂ OH (180)	THF	1 atm of N ₂ , 25 °C	6.1	[107]
[Mo(N ₂) ₂ (PCP)] ₂ (μ-N ₂) (263)	Sml ₂ (360)	HOCH ₂ CH ₂ OH (360)	THF	1 atm of N ₂ , 25 °C	18	[10]
[Mo(N ₂) ₂ (PCP)] ₂ (μ-N ₂) (263)	Sml ₂ (14400)	H ₂ O (14400)	THF	1 atm of N ₂ , 25 °C	2175	[10]
[Fe(PNP')(N ₂)] (260)	KC ₈ (40)	[HOEt ₂][BAR ^F ₄] (38)	Et ₂ O	1 atm of N ₂ , - 78 °C	3.5	[108]
[Na(12-crown-4)] ₂ [Fe(P ₃ ^B)(N ₂)] (264)	KC ₈ (50)	[HOEt ₂][BAR ^F ₄] (46)	Et ₂ O	1 atm of N ₂ , - 78 °C	3.5	[109]
[Na(12-crown-4)] ₂ [Fe(P ₃ ^B)(N ₂)] (264)	CoCp ⁺ ₂ (486)	[H ₂ NPh ₂][OTf] (966)	Et ₂ O	1 atm of N ₂ , - 78 °C	6.4	[110]
[K(Et ₂ O) _{0.5}][Fe(P ₃ ^C)(N ₂)] (265)	KC ₈ (40)	[HOEt ₂][BAR ^F ₄] (38)	Et ₂ O	1 atm of N ₂ , - 78 °C	2.3	[111]

[Na(12-crown-4) ₂][Fe(P ₃ Si)(N ₂)] (266)	KC ₈ (1800)	[HOEt ₂][BAr ^F ₄] (1500)	Et ₂ O	1 atm of N ₂ , - 78 °C	1.9	[112]
[K(thf) ₂][Ru(P ₃ Si)(N ₂)]	KC ₈ (50)	[HOEt ₂][BAr ^F ₄] (46)	Et ₂ O	1 atm of N ₂ , - 78 °C	2.2	[113]
[K(thf) ₂][Os(P ₃ Si)(N ₂)]	CoCp* ₂ (1800)	[H ₂ NPh ₂][OTf] (1500)	Et ₂ O	1 atm of N ₂ , - 78 °C	60	[113]
[FeH(PPP) ₂](N ₂) (268)	KC ₈ (3600)	[HOEt ₂][BAr ^F ₄] (3000)	Et ₂ O	1 atm of N ₂ , - 78 °C, Hg lamp	33.4	[114]
[Fe(N ₂)(P ^{Ph} P ₂ ^{Cy})(H) ₂] (268)	KC ₈ (200)	[HOEt ₂][BAr ^F ₄] (200)	Et ₂ O	1 atm of N ₂ , - 80 °C	1.4	[115]
[Fe(CAAC) ₂][BAr ^F ₄] (269)	KC ₈ (50)	[HOEt ₂][BAr ^F ₄] (50)	Et ₂ O	1 atm of N ₂ , - 95 °C	1.7	[116]
[Fe(depe) ₂ N ₂] (270)	CoCp* ₂ (270)	[H ₂ NPh ₂][OTf] (360)	Et ₂ O	1 atm of N ₂ , - 78 °C	12.3 ^c	[117]
[Co(PNP')(N ₂)] (261)	KC ₈ (40)	[HOEt ₂][BAr ^F ₄] (38)	Et ₂ O	1 atm of N ₂ , - 78 °C	2.2	[118]

[a] equivalents per catalyst. [b] based on the equivalents of reductant. [c] in N₂H₄

7. Summary of complexes and reagents for catalytic N₂ silylation

Table S7. Molecular catalysts for N₂ silylation, conditions for highest TON reported.

Catalyst	Reductant (equiv.) ^a	Silyl source (equiv.)	Solvent	Reaction condition:	TON ^a	Ref.
K ₂ {[(Xy-N ₃ N)Ti] ₂ (μ ₂ -N ₂)]} (306)	K (1500)	Me ₃ SiCl (1500)	THF	1 atm of N ₂ , 50 °C	6.3	[119]
[V{(Me ₃ SiNCH ₂ CH ₂) ₂ NSiMe ₃ }(μ-N)] ₂ (38)	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	11.5	[120]
K[V{(Me ₃ SiNCH ₂ CH ₂) ₂ NSiMe ₃ }(μ-N)] ₂ (39)	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	12	[120]
[V{(Me ₃ SiNCH ₂ CH ₂) ₂ NSiMe ₃ }(μ-Cl)] ₂	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	5.5	[120]
[VCl(N(SiMe ₃) ₂)(thf)]	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	5.5	[120]
[VCl ₂ (tmeda) ₂] (307)	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	6	[120]
[Na(thf)][V(N ₂) ₂ (dppe) ₂] (308)	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	10.5	[120]
[CrCl ₃]	Li (50)	Me ₃ SiCl (50)	THF	1 atm of N ₂ , 25 °C	2.7	[121]
[Cr(N ₂) ₂ (P ^{Ph} ₄ N ^{Bn} ₄)] (130)	Na (10 ⁵)	Me ₃ SiCl (10 ⁵)	THF	1 atm of N ₂ , 25 °C	17	[55]
[Cr(Cy ₂ PC ₆ H ₄ -(η ⁵ -C ₅ H ₄ Et ₄)(N ₂) ₂)] (275)	K (2000)	Me ₃ SiCl (2000)	Et ₂ O	1 atm of N ₂ , 25 °C	13	[122]
<i>trans</i> -[W(N ₂) ₂ (dppe) ₂] (94)	Na (100)	Me ₃ SiCl (100)	THF	1 atm of N ₂ , 30 °C	1.7	[123]
<i>trans</i> -[Mo(N ₂) ₂ (dppe) ₂] (95)	Na (100)	Me ₃ SiCl (100)	THF	1 atm of N ₂ , 30 °C	4.9	[123]
[Mo(N ₂) ₂ (PMe ₂ Ph) ₄] (124)	Na (200)	Me ₃ SiCl (200)	THF	1 atm of N ₂ , 30 °C	18.3	[123]
[Mo(N ₂) ₂ (depf) ₂] (106)	Na (8000)	Me ₃ SiCl (8000)	THF	1 atm of N ₂ , 25 °C	226	[124]
[Mo(PP ^{Cy} ₃)Cl][BPh ₄] (276)	K (200)	Me ₃ SiCl (200)	THF	1 atm of N ₂ , 50 °C	5.9	[125]
[Mo(PP ^{Cy} ₃)=(N-N(SiMe ₃) ₂)] (278)	K (200)	Me ₃ SiCl (200)	THF	1 atm of N ₂ , 50 °C	7.5	[125]
[Mo(PP ^{Cy} ₃)=(N(SiMe ₃))] (279)	K (200)	Me ₃ SiCl (200)	THF	1 atm of N ₂ , 50 °C	5.9	[125]
[Mo(PPP)Cl ₃] (28)	K (400)	Me ₃ SiCl (400)	THF	1 atm of N ₂ , 25 °C	19.2	[126]
[[Mo(PMe ₃)Cp*] ₂ (μ ₂ -H) ₈ {FeN(SiMe ₃) ₂ }] (295)	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	34.5	[127]
[[Mo(PMe ₃)Cp*] ₂ (μ ₂ -H) ₈ {MnN(SiMe ₃) ₂ }] (296)	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	6	[127]
[[Mo(PMe ₃)Cp*] ₂ (μ ₂ -H) ₈ {FeS(2,4,6- ^t Pr ₃ C ₆ H ₂)}] (297)	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	32.5	[127]
[[Mo(PMe ₃)Cp*] ₂ (μ ₂ -H) ₈ {FeS(2,6-(SiMe ₃) ₂ C ₆ H ₃)}] (298)	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	34.5	[127]

[Fe(CO) ₅]	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	12.5	[128]
[Fe(SiMe ₃) ₂ (CO) ₄]	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	14.5	[128]
[FeCp(CO) ₂] ₂	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	8.5	[128]
[FeCp ₂]	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	6.5	[128]
[Fe(η ⁵ -C ₅ H ₄ SiMe ₃) ₂]	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	11.5	[128]
[Fe(η ⁵ -C ₅ H ₂ (SiMe ₃) ₃) ₂]	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	17	[128]
[Fe(PSiP)H(PMe ₃) ₂] (285)	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	13	[129]
[Fe[SiCHSi](H)(N ₂)(PMe ₃)] (287)	KC ₈ (1800)	Me ₃ SiCl (1800)	Dioxane	1 atm of N ₂ , 25 °C	37.2	[130]
[Fe(CAAC) ₂][BAF ₄] (269)	KC ₈ (600)	Me ₃ SiCl (600)	Et ₂ O	1 atm of N ₂ , 25 °C	12.2	[116]
[Fe(depe) ₂ N ₂] (270)	KC ₈ (1500)	Me ₃ SiCl (1500)	Et ₂ O	1 atm of N ₂ , 25 °C	60.5	[131]
[Fe(P ₄ N ₂) ₂] (289)	KC ₈ (500)	Me ₃ SiCl (500)	Toluene	1 atm of N ₂ , 25 °C	5.5	[132]
[Fe(N ₂)(P ^{Ph} P ₂ ^{Cy})(H) ₂]	K (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	14.5	[115]
[Fe(N ₂) ₂ (P ^{Ph} P ₂ ^{Cy}) (290)	K (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	16	[115]
[Fe(N ₂) ₂ (P ^{Bu} P ₂ ^{Cy}) (291)	K (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	7.3	[115]
[Fe ₃ (cyclophane-β-diketamin)Br ₂] (54)	KC ₈ (500)	Me ₃ SiCl (500)	Et ₂ O	1 atm of N ₂ , - 34 °C	41.5	[133]
[Fe ₄ (μ-H) ₄ (μ ₃ -H) ₂ {N(SiMe ₃) ₂ } ₂ (PMe ₃) ₄] (292)	Na (600)	Me ₃ SiCl (600)	DME	1 atm of N ₂ , 25 °C	80	[134]
[Fe ₆ (μ-H) ₁₀ (μ ₃ -H) ₂ (PMe ₃) ₁₀] (293)	Na (600)	Me ₃ SiCl (600)	DME	1 atm of N ₂ , 25 °C	91.5	[134]
[Fe ₄ (η ⁶ -C ₇ H ₈)(μ-H) ₂ {μ-N(SiMe ₃) ₂ } ₂ {N(SiMe ₃) ₂ }] (294)	Na (600)	Me ₃ SiCl (600)	DME	1 atm of N ₂ , 25 °C	74	[134]
[Fe ₄ (μ-H) ₄ (μ ₃ -H) ₂ (SDmp) ₂ (PMe ₃) ₄]	Na (600)	Me ₃ SiCl (600)	DME	1 atm of N ₂ , 25 °C	52	[134]
[Co ₂ (CO) ₈]	Na (600)	Me ₃ SiCl (600)	DME	1 atm of N ₂ , 25 °C	18	[135]
[Co ₂ (CO) ₈]	Na (600)	Me ₃ SiCl (600)	DME	1 atm of N ₂ , 25 °C + 2 equiv. bipy.	20	[135]
[Co(SiMe ₃)(CO) ₄]	Na (600)	Me ₃ SiCl (600)	DME	1 atm of N ₂ , 25 °C	18	[135]
[CoCp ₂]	Na (600)	Me ₃ SiCl (600)	DME	1 atm of N ₂ , 25 °C	4	[135]
[Co ₂ (P ₃ N ₃)] (299)	KC ₈ (2000)	Me ₃ SiCl (2000)	THF	1 atm of N ₂ , 25 °C	195	[135]
[Co(ICy) ₂ (N ₂)] (300)	KC ₈ (2000)	Me ₃ SiCl (2000)	Et ₂ O	1 atm of N ₂ , 25 °C	125	[136]
[Co(NpNP)] (302)	KC ₈ (1500)	Me ₃ SiCl (2000)	THF	1 atm of N ₂ , - 40 °C	100	[137]
[Co(QuiNacNacP)Cl] (303)	KC ₈ (1000)	Me ₃ SiCl (1500)	THF	1 atm of N ₂ , 25 °C	38	[138]

[Co(PSiP)(PMePh ₂)N ₂] (304)	Na (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , 25 °C	20.5	[129]
[Rh(PNP')N ₂] (305)	KC ₈ (600)	Me ₃ SiCl (600)	THF	1 atm of N ₂ , - 40 °C	11.5	[139]
[U ₂ (<i>m</i> TP) ₂] (309)	K (85)	Me ₃ SiCl (60)	C ₆ H ₆	1 atm of N ₂ , 25 °C	3.2 ^c	[64]
<i>m</i> TP = [{2-(OC ₆ H ₂ - ^t Bu-2,Me-4) ₂ CH}-C ₆ H ₄ -1,3] ⁴⁻		[HNEt ₃][BPh ₄]				

[a] equivalents per catalyst. [b] based on the equivalents of reductant. [c] in HMDS

8. References

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