

Supplementary information for

**Near-Room Temperature Ferromagnetic Insulating State in Highly
Distorted LaCoO_{2.5} with CoO₅ Square Pyramids**

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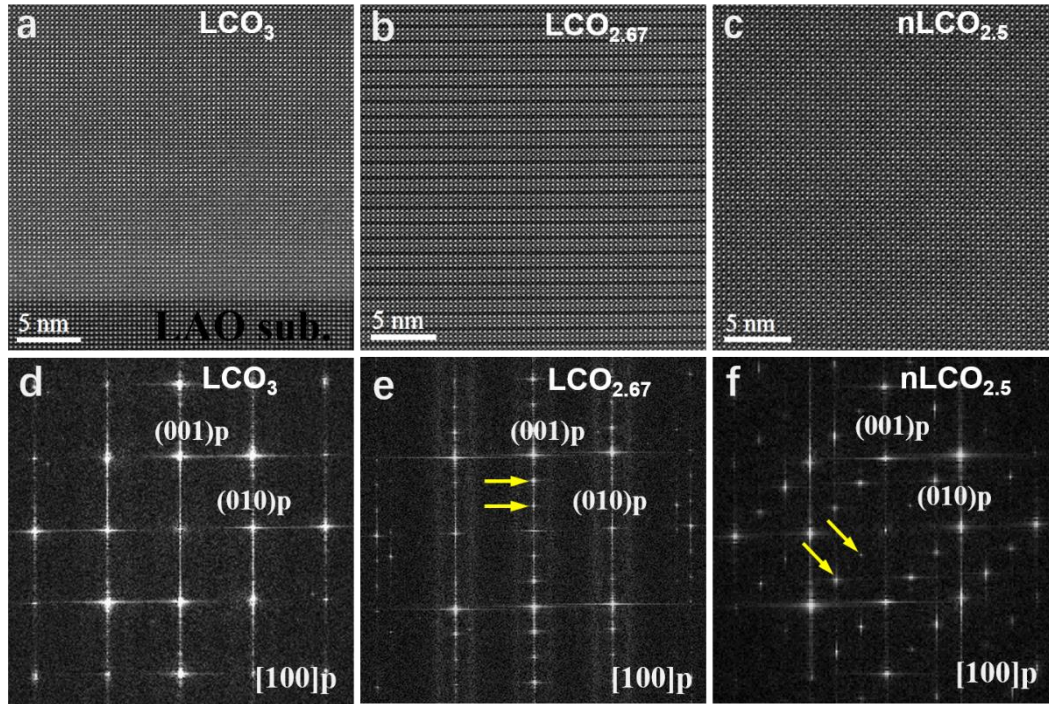
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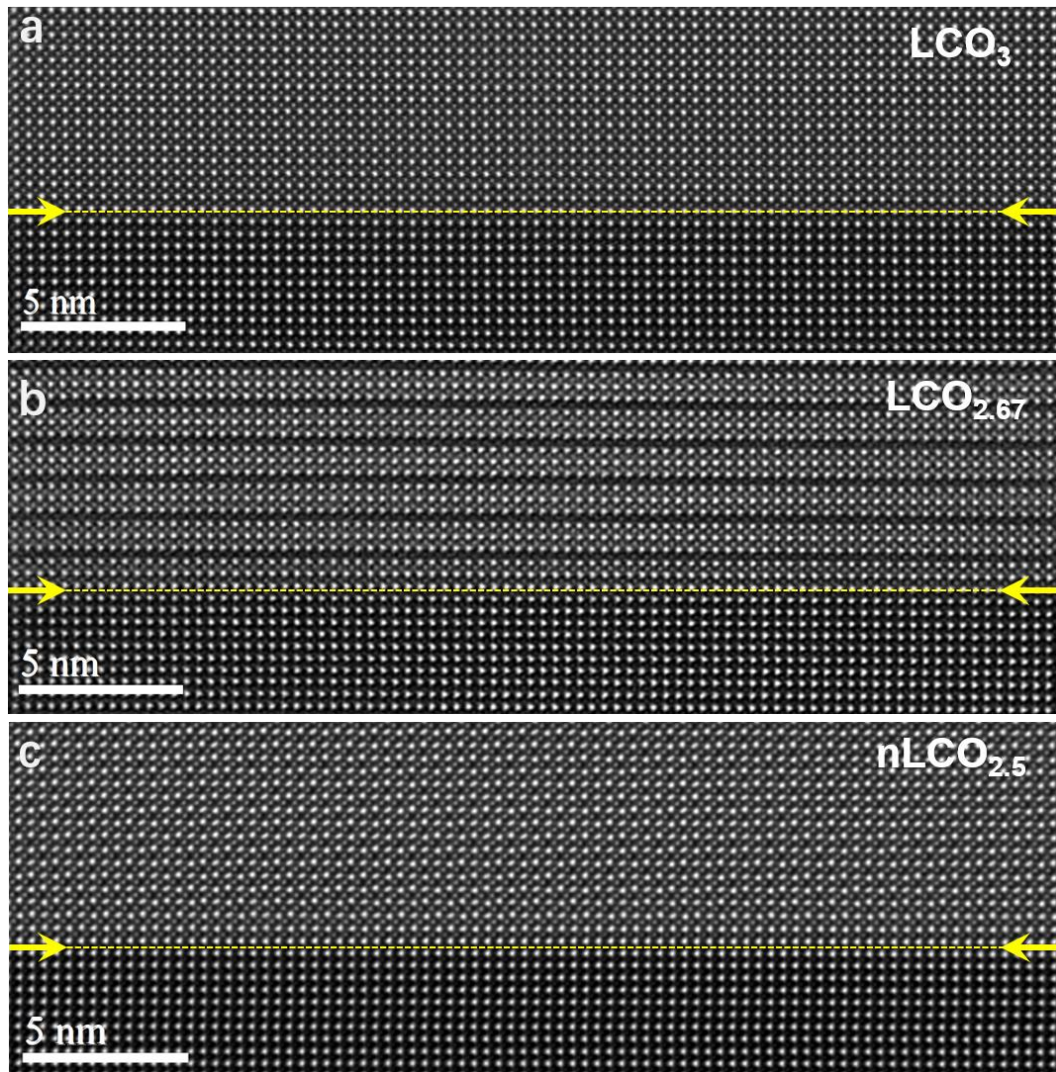
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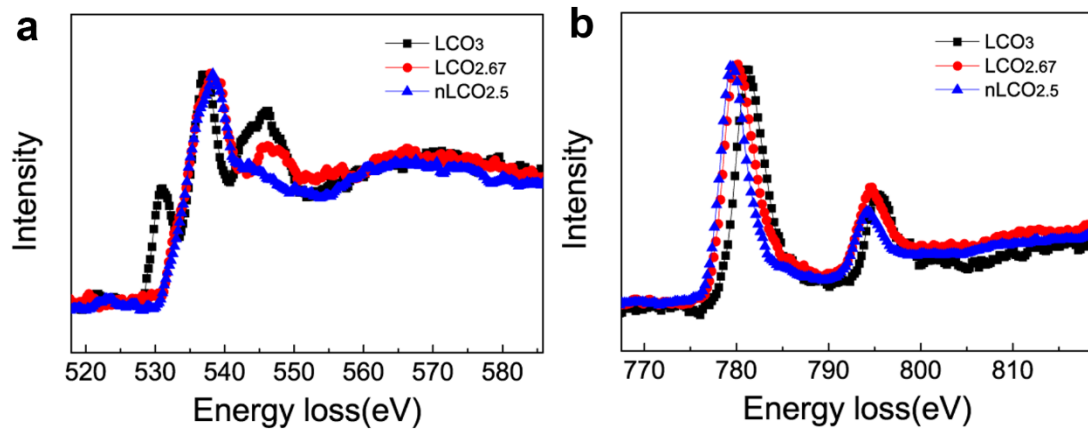
I. Microstructural characterizations and EELS analysis



Supplementary Fig. 1. High-magnification high-angle angular dark-field (HAADF) images of (a) pristine LCO₃, (b) LCO_{2.67} and (c) nLCO_{2.5} films grown on LAO substrates. The representative HAADF-STEM images were chosen randomly. For a LCO₃, no vertical or horizontal dark stripes were observed. For a LCO_{2.67}, we observed the horizontal dark stripes with a period of $3a_0$. For a nLCO_{2.5}, we observed more dense dark stripes aligned 45 degrees to the surface and interface. Such structural modulation is uniform within the entire nLCO_{2.5} sample. The corresponding FFT patterns of (d) pristine LCO₃, (e) LCO_{2.67} and (f) nLCO_{2.5} films are in good agreement with the SAED patterns shown in Figures 1a-c.

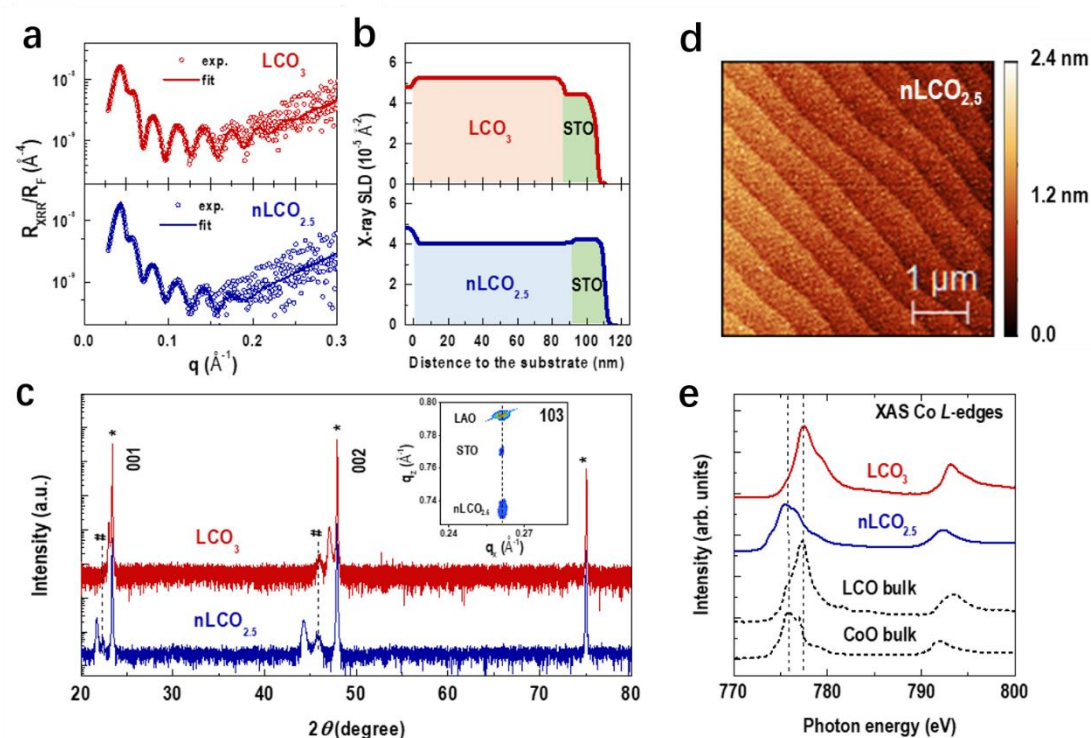


Supplementary Fig. 2. HAADF images of the interfaces between LaAlO₃ substrate and LCO₃ (a), LCO_{2.67} (b) and nLCO_{2.5} (c) films, respectively. Yellow arrows and dotted lines indicate the positions of the atomically sharp interfaces and highly coherent epitaxy of films.



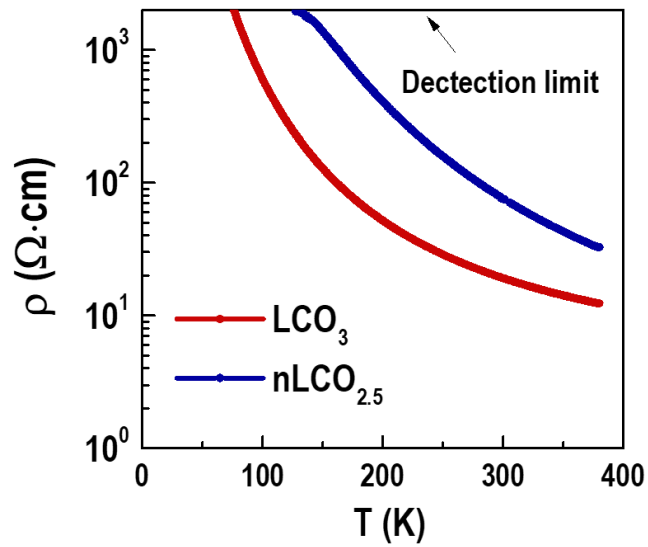
Supplementary Fig. 3. Comparison on the O *K*-edges (a) and Co *L*-edges (b) in three electronic states of LCO₃, LCO_{2.67}, and nLCO_{2.5} thin films.

II. Macroscopic structural and electronic state property characterizations



Supplementary Fig. 4. Structure, morphology, and electronic state characterizations in LCO₃ and nLCO_{2.5} thin films. (a) X-ray reflectivities of LCO₃ and nLCO_{2.5} films. The open circular symbols represent the experimental data, and the solid lines are the best fit. From XRR fittings, the chemical depth profiles [shown in (b)] of those two films can be obtained. XRR results indicate the bilayers have uniform chemical distributions, sharp interfaces, and smooth surfaces. The x-ray scattering length density (SLD), which is proportional to its chemical density, of LCO₃ is ~24% larger than that of nLCO_{2.5}. In addition, the thicknesses of LCO and STO layers in the STO/LCO₃ bilayer are approximately 84 nm and 20 nm, respectively. While, the thickness of nLCO_{2.5} is ~7 nm larger than that of LCO₃, whereas the thickness of STO is unchanged. (c) X-ray diffraction θ - 2θ scans of LCO₃ and nLCO_{2.5} films. * indicates the 00*l* peak positions of LAO substrate. # indicates the 00*l* peak positions of STO capping layer. Both curves reveal that the LCO₃ and nLCO_{2.5} films are single-phase and of high structural quality. The out-of-plane lattice constant of LCO₃ is 3.854 Å. After annealing in vacuum for 4 hours, the out-of-plane lattice constant of nLCO_{2.5} films increases to 4.088 Å. The elongation of lattice constant reaches ~6.1%, in consistent

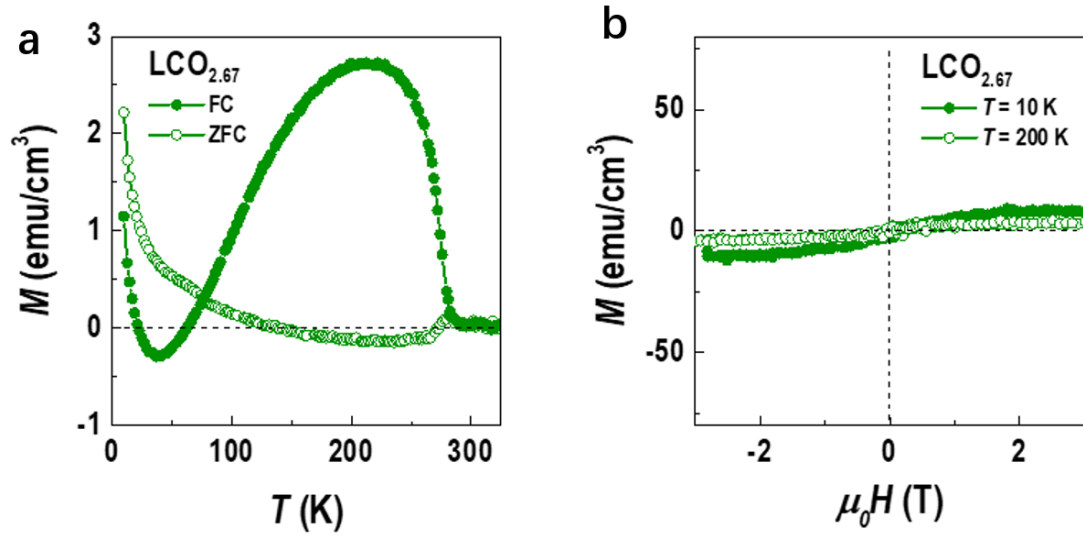
with the increment of nLCO_{2.5} layer's thickness. In contrast, the lattice constant of STO layer keeps almost constant, indicating the vacuum-annealing, at least under the same conditions, does not create significant Vo in STO. We also performed reciprocal space mapping (RSM) on a nLCO_{2.5} sample. Inset of (c) shows the RSM around 103 reflections around LAO substrate. Both nLCO_{2.5} layer and STO capping layer are coherently grown on LAO substrate, suggesting the identical in-plane lattice constant. The in-plane strain can be maintained up to ~ 200 u.c. due to the rather small lattice mismatch of ~ 1% between LAO and LCO. (d) Topography of a nLCO_{2.5} film capped with STO layer. Atomic force microscopy (AFM) image was recorded over an area of 5×5 μm² using Asylum Research AFM operated in the tapping mode. Surprisingly, we observe the step-and-terrace feature in our sample even without chemical etching the LAO substrate. The averaged r.m.s. is ~ 0.4 nm, revealing a smooth surface. The macroscopic structural and morphology measurements indicate that the as-grown LCO₃ films and newly formed nLCO_{2.5} films are chemically uniform, highly crystalline, and single-phase. (e) X-ray absorption spectra (XAS) for LCO₃ and nLCO_{2.5} films at Co *L*-edges. The reference spectra for Co³⁺ (from bulk LCO) [E. J. Guo *et al.* Science Advances 5, eaav5050 (2019)] and Co²⁺ (from bulk CoO) [N. Lu *et al.* Nature 546, 124 (2017)] are shown as black dashed lines for comparison. Apparently, XAS signal from LCO₃ at Co *L*-edges is nearly dominated by Co³⁺. The main peak left-shifts to ~ 777 eV in the XAS spectra from nLCO_{2.5}. The line shape of XAS from nLCO_{2.5} is almost identical to that of bulk CoO, suggesting that the Co³⁺ ions in LCO₃ transit into Co²⁺ ions in nLCO_{2.5}.



Supplementary Fig. 5. Transport measurements on LCO₃ and nLCO_{2.5} thin films.

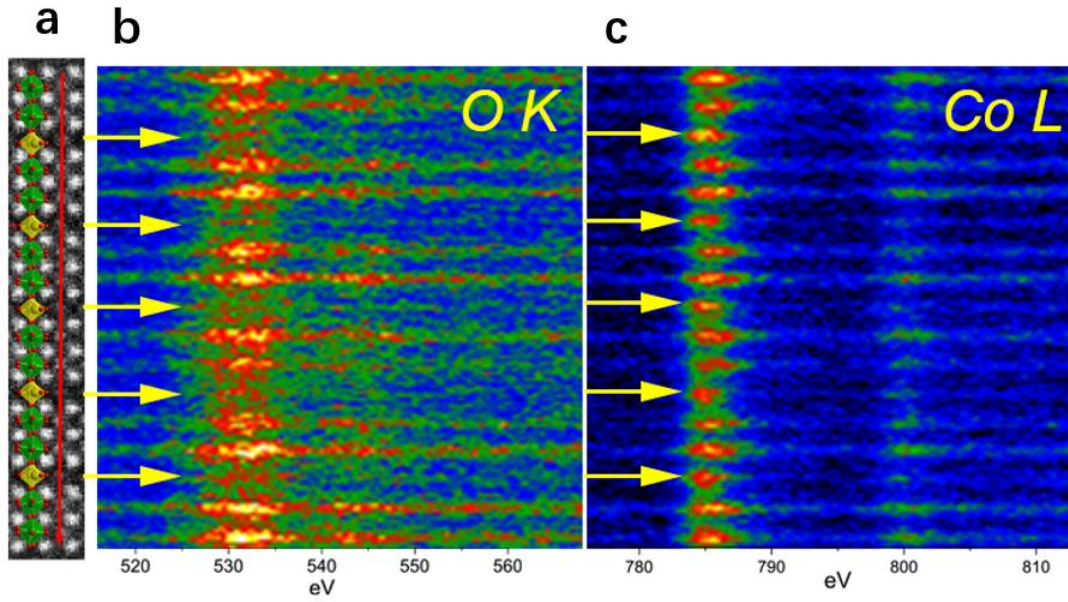
Resistivity (ρ) of LCO₃ and nLCO_{2.5} thin films were measured using standard van der Pauw method. The room-temperature resistivities of LCO₃ and nLCO_{2.5} thin films are above 1 Megaohms, corresponding to $\sim 1 \text{ } \Omega \cdot \text{cm}$. The ρ - T curves for LCO₃ and nLCO_{2.5} thin films were measured up to 380 K. The low temperature resistivities were recorded until the detection limit of our electrical device. The negative slopes of ρ - T curves indicate the insulating character of both LCO₃ and nLCO_{2.5} films.

III. Magnetic properties of $\text{LCO}_{2.67}$ films

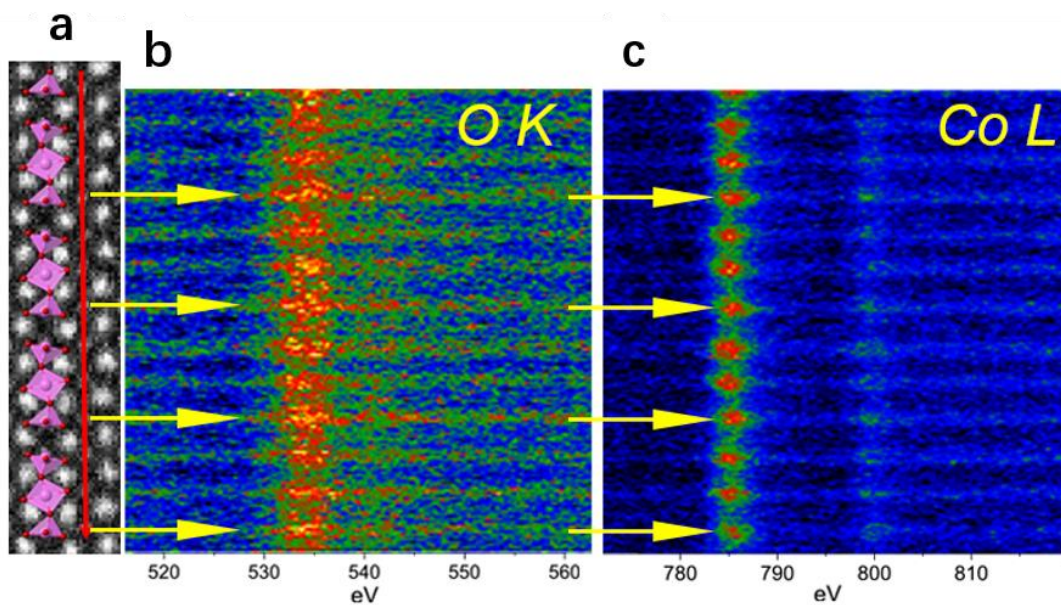


Supplementary Fig. 6. Magnetic properties of $\text{LCO}_{2.67}$ films. (a) M - T curves measured during the sample warming up at a magnetic field of 0.1 T. Solid and open circles represent the magnetizations were recorded after field-cooled (FC) and zero-field-cooled (ZFC), respectively. (b) M - H curves recorded at 10 K and 200 K. The saturation magnetization at 10 K and 200 K is $\sim 0.067 \mu_{\text{B}}/\text{Co}$ and $\sim 0.02 \mu_{\text{B}}/\text{Co}$, respectively.

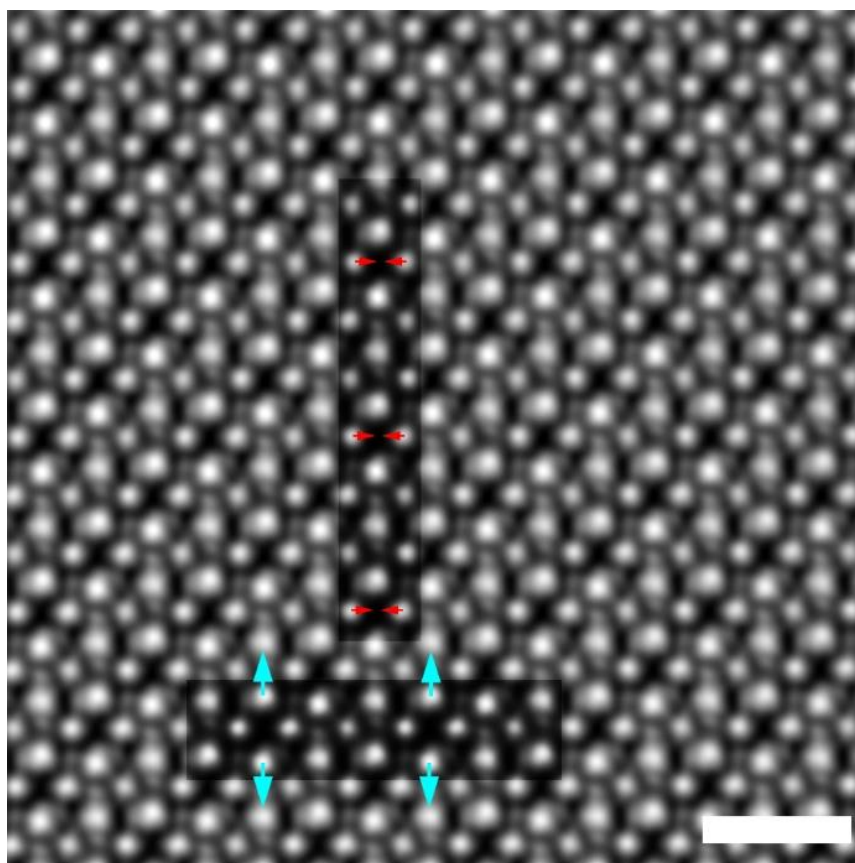
IV. Spectrum images of O *K*- and Co *L*-edges for the line scan EELS data



Supplementary Fig. 7. Atomic-resolved EELS spectra of the $\text{LCO}_{2.67}$ phase. (a) HAADF image with overlaid CoO_6 octahedra and CoO_4 tetrahedra. The red line indicates the position of line scanning. Two-dimensional spectra image of (b) O *K*- and (c) Co *L*-edges, the yellow arrows indicate the oxygen-deficient CoO_4 tetrahedra. The difference of O *K*- and (c) Co *L*-edges between CoO_6 octahedra and CoO_4 tetrahedra can be identified clearly, with periodicity of $3a_0$.



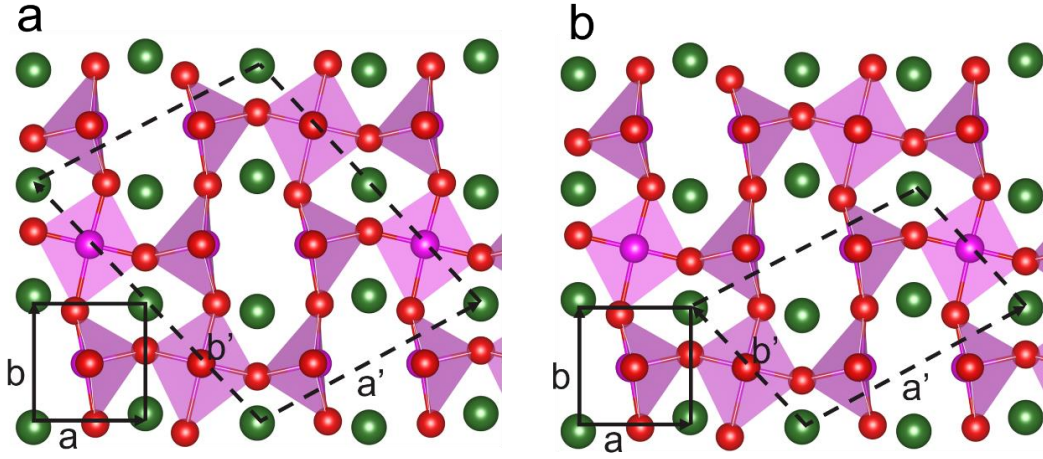
Supplementary Fig. 8. Atomic-resolved EELS spectra of the nLCO_{2.5} phase. (a) HAADF image with overlaid CoO₅ square pyramids. The red line indicates the position of line scanning. Two-dimensional spectra image of (b) O *K*- and (c) Co *L*-edges, the yellow arrows indicate the - Co columns in Co₂-O₅ pyramids. A roughly uniform intensity distribution of O *K*- and Co *L*-edges' signals can be visualized.



Supplementary Fig. 9. The iDPC image of nLCO_{2.5} phase. Image contrast in two rectangular regions were adjusted to highlight atomic displacements of La and Co, as indicated by blue and red arrows, respectively. The scale bar is 1 nm.

V. Structure determination and the summaries of structure parameters

With the accurate projected position of each atom, we construct two most possible structural models (nLCO-A and nLCO-B) as shown in Supplementary Fig. 9. The difference between them is relative V_o position of the Co_2-O_5 square pyramids in the projected direction, where V_o s reside alternatively in $(100)_p$ plane among neighbored Co_2-O_5 square pyramids in nLCO-A and in the same $(100)_p$ plane in nLCO-B. We optimized these two structures and find the energy of the nLCO-A is lower than that of nLCO-B, which is also favorable for the alternative arrangement of V_o in the neighbored Co_2-O_5 square pyramids.



Supplementary Fig. 10. Two possible crystal structures of nLCO_{2.5}. a , b are the basic vectors of the unit cell ($1 \times 1 \times 1$) and a' , b' are the basic vectors of the extended cell ($\sqrt{5} \times 2 \sqrt{2} \times 2$) for (a) nLCO-A and ($\sqrt{5} \times \sqrt{2} \times 2$) for (b) nLCO-B, respectively, as indicated by dark dotted lines. The corresponding transformation matrix with the basic unit cell is $[2, -1, 0; 2, 2, 0; 0, 0, 2]$ and $[2, -1, 0; 1, 1, 0; 0, 0, 2]$, respectively.

Supplementary Table 1.Structure parameters summary of the nLCO_{2.5} phase from DFT calculation

supercell	<i>a</i> (Å)	<i>b</i> (Å)	<i>c</i> (Å)	α (°)	β (°)	γ (°)
$\sqrt{5}\times 2\sqrt{2}\times 2$	8.589	11.079	7.58	90.00	90.00	105.11
1×1×1	3.79	4.04	3.79	90.00	90.00	90.00

Atoms	label	number	<i>x</i>	<i>y</i>	<i>z</i>
La	La1	1	-0.007	0.015	0.000
La	La2	1	0.017	0.478	0.000
La	La3	1	0.017	-0.022	0.500
La	La4	1	0.993	0.515	0.500
La	La5	1	0.687	0.193	0.000
La	La6	1	0.687	0.702	0.000
La	La7	1	0.687	0.202	0.500
La	La8	1	0.687	0.693	0.500
La	La9	1	0.323	0.317	0.000
La	La10	1	0.322	0.813	0.000
La	La11	1	0.322	0.313	0.500
La	La12	1	0.323	0.817	0.500
Co	Co1	1	0.344	0.084	0.249
Co	Co1	1	0.344	0.584	0.251
Co	Co1	1	0.344	0.084	0.751
Co	Co1	1	0.344	0.584	0.749
Co	Co1'	1	0.658	0.427	0.247

Co	Co1'	1	0.658	0.927	0.253
Co	Co1'	1	0.658	0.427	0.753
Co	Co1'	1	0.658	0.927	0.747
Co	Co2	1	0.004	0.247	0.273
Co	Co2	1	0.004	0.747	0.227
Co	Co2	1	0.004	0.247	0.727
Co	Co2	1	0.004	0.747	0.773
O	O1	1	0.342	0.076	0.000
O	O2	1	0.342	0.576	0.000
O	O3	1	0.342	0.076	0.500
O	O4	1	0.342	0.576	0.500
O	O5	1	1.000	0.250	0.000
O	O6	1	1.000	0.750	0.500
O	O7	1	0.655	0.418	0.000
O	O8	1	0.655	0.918	0.000
O	O9	1	0.655	0.418	0.500
O	O10	1	0.655	0.918	0.500
O	O11	1	0.131	0.128	0.279
O	O12	1	0.130	0.628	0.221
O	O13	1	0.131	0.128	0.721
O	O14	1	0.130	0.628	0.779
O	O15	1	0.865	0.365	0.276

O	O16	1	0.865	0.865	0.224
O	O17	1	0.865	0.365	0.724
O	O18	1	0.865	0.865	0.776
O	O19	1	0.232	0.416	0.250
O	O20	1	0.232	0.916	0.250
O	O21	1	0.232	0.416	0.750
O	O22	1	0.232	0.916	0.750
O	O23	1	0.777	0.101	0.251
O	O24	1	0.777	0.601	0.249
O	O25	1	0.777	0.101	0.749
O	O26	1	0.777	0.601	0.751
O	O27	1	0.476	0.255	0.250
O	O28	1	0.476	0.755	0.250
O	O29	1	0.476	0.255	0.750
O	O30	1	0.476	0.755	0.750

Supplementary Table 2

The Co-O bond length of CoO₅ square pyramids in nLCO_{2.5} phase from DFT calculation

bond length (Å)	O1	O2	O3	O4	O5
Co1	2.027	1.908	1.937	1.886	1.852
Co2	2.072	1.982	2.189	1.923	2.340
Co1'	2.077	1.878	1.940	1.917	2.126