

Supplementary Materials

Dinuclear Lanthanide(III) Complexes from the Use of Methyl 2-pyridyl Ketoxime: Synthetic, Structural and Physical Studies

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Table S1. Crystallographic data and structural refinement parameters for complexes 1, 5 and 6.

Parameter	1	5	6
Formula	C ₂₂ H ₂₈ N ₆ Nd ₂ O ₁₆	C ₂₂ H ₂₈ N ₆ Dy ₂ O ₁₆	C ₂₆ H ₃₄ Dy ₂ N ₄ O ₁₄
Formula weight	920.98	957.50	951.57
Crystal color	colorless	colorless	colorless
Crystal size, mm	0.30 x 0.09 x 0.08	0.18 x 0.16 x 0.10	0.15 x 0.13 x 0.10
Crystal system	triclinic	triclinic	monoclinic
Space group	<i>P</i> $\bar{1}$	<i>P</i> $\bar{1}$	<i>P</i> 2 ₁ / <i>c</i>
Temperature, K	160	160	160
Radiation, Å	Cu K α , 1.54178	Cu K α , 1.54178	Cu K α , 1.54178
<i>a</i> , Å	8.1064(2)	8.0902(5)	8.7141(1)
<i>b</i> , Å	8.3855(2)	8.3503(3)	12.9093(2)
<i>c</i> , Å	12.0672(2)	11.8921(6)	15.3750(2)
α , °	98.087(1)	98.748(1)	90.0(1)
β , °	99.681(1)	100.335(1)	107.586(1)
γ , °	99.893(1)	99.857(1)	90.0(1)
Volume, Å ³	784.34(3)	764.70(7)	1648.75(4)
<i>Z</i>	1	1	2
Calculated density, g cm ⁻³	1.950	2.079	1.917
Absorption coefficient, mm ⁻¹	25.69	26.59	24.59
θ_{\min} - θ_{\max} , °	7.0-65.0	7.0-65.0	6.7-65.0
Reflections collected/unique (<i>R</i> _{int})	12135/2445 (0.078)	8564/2350 (0.113)	21531/2879 (0.076)
Reflections with <i>I</i> > 2 σ (<i>I</i>)	2374	1796	2601
Refined parameters/restraints	212/0	212/0	212/0
<i>R</i> ₁ ^a [<i>I</i> > 2 σ (<i>I</i>)], <i>R</i> ₁ ^a (all data)	0.0462, 0.0469	0.0875, 0.0996	0.0523, 0.0552
<i>wR</i> ₂ ^b [<i>I</i> > 2 σ (<i>I</i>)], <i>wR</i> ₂ (all data)	0.1158, 0.1170	0.1728, 0.1895	0.1324, 0.1351
Goodness-of-fit on <i>F</i> ²	1.03	1.15	1.06
Largest differences peak and hole (e Å ⁻³)	2.12/-2.07	2.46/-1.72	1.64/-2.41

^a $R_1 = \sum(|F_o| - |F_c|) / \sum(|F_o|)$. ^b $wR_2 = \{\sum[w(F_o^2 - F_c^2)^2] / \sum[w(F_o^2)^2]\}^{1/2}$, $w = 1/[\sigma^2(F_o^2) + (aP)^2 + bP]$, where $P = [\max(F_o^2, 0) + 2F_c^2]/3$ ($a = 0.0782$ and $b = 0.9824$ for **1**; $a = 0.0713$ and $b = 0.03264$ for **5**; $a = 0.724$ and $b = 10.8106$ for **6**).

Table S2. Continuous Shape Measures (CShM) values^a for the potential coordination polyhedra of the crystallographically independent Ln^{III} centers in the structures of complexes **1**, **5** and **6**.

Ideal Coordination Polyhedron	1	5	6
Enneagon (EP-9)	34.398	34.902	34.011
Octagonal pyramid (OPY-9)	21.909	22.085	22.489
Heptagonal bipyramid (HBPY-9)	17.795	17.915	17.333
Johnson triangular cupola (JTC-9)	15.576	15.552	14.592
Capped cube (JCCU-9)	11.176	10.707	9.358
Spherical-relaxed capped cube (CCU-9)	9.344	9.544	8.647
Capped square antiprism (JCSAPR-9)	2.865	2.219	2.620
Spherical-capped square antiprism (CSAPR-9)	2.110	1.598	1.992
Tricapped trigonal prism (JTCTPR-9)	3.584	3.190	3.117
Spherical tricapped trigonal prism (TCTPR-9)	2.754	2.405	2.517
Tridiminished icosahedron (JTDIC-9)	13.286	13.267	12.424
Hula-hoop (HH-9)	9.232	9.620	9.902
Muffin (MFF-9)	1.664	1.316	1.578

^aThe polyhedron with the smallest CShM value (in bold) is the real coordination polyhedron of the Ln/Ln1' center for the complex.

Table S3. Intra- and intermolecular H-bonding interactions (distances in Å and angles in °) in the crystal structures of the isomorphous complexes **1** and **5**. ^{a,b}

D-H...A	D-H	H...A	D...A	D-H...A
Complex 1				
O1-H(O1)...O5	0.84	2.01	2.734(7)	145
C7-H(C7)...O6'	0.95	2.57	3.261(8)	129
C4-H(C4)...O7 ⁱ	0.95	2.49	3.376(9)	155
C9-H _B (C9)...O4 ⁱⁱ	0.98	2.43	3.383(11)	164
C11-H _B (C11)...O2 ⁱⁱ	0.98	2.60	3.469(11)	149
C1-H _A (C1)...O1''	0.98	2.52	3.392(9)	148
C1-H _B (C1)...O4'''	0.98	2.47	3.319(9)	145
Complex 5				
O1-H(O1)...O5	0.84	2.02	2.690(12)	136
C7-H(C7)...O6'	0.95	2.44	3.107(14)	127
C4-H(C4)...O7 ⁱ	0.95	2.51	3.391(17)	155
C9-H _B (C9)...O4 ⁱⁱ	0.98	2.42	3.348(15)	157

C11-H _B (C11)···O2 ⁱⁱ	0.98	2.56	3.434(18)	148
C1-H _A (C1)···O1 ^{''}	0.98	2.55	3.402(15)	146
C1-H _B (C1)···O4 ^{'''}	0.98	2.59	3.35(2)	134

^a D = donor, A = acceptor. ^b Symmetry codes: (') = -x+2, -y+1, -z+2; (') = -x+1, -y+1, -z+1; (''') = -x+1, -y, -z+1; (i) = -x+2, -y+1, -z+1; (ii) = x, y+1, z.

Table S4. Intra- and intermolecular H-bonding interactions (distances in Å and angles in °) in the crystal structure of **6**. ^{a,b}

D-H···A	D-H	H···A	D···A	D-H···A
O1-H(O1)···O6'	0.84	2.03	2.603(8)	125
C7-H(C7)···O5	0.95	2.52	3.185(12)	127
C16-H _C (C16)···O2 ⁱⁱ	0.98	2.46	3.412(9)	164
C4-H(C4)···O3 ^{''}	0.95	2.35	3.114(11)	137
C6-H(C6)···O2 ^{'''}	0.95	2.44	3.200(10)	137
C14-H _C (C14)···O7 ⁱ	0.98	2.58	3.521(9)	162

^a D = donor, A = acceptor. ^b Symmetry codes: (') = -x, -y, -z; (') = -x+1, y+1/2, -z+1/2; (''') = x+1, y, z; (i) = -x+1, -y, -z; (ii) = x, -y+1/2, z-1/2.

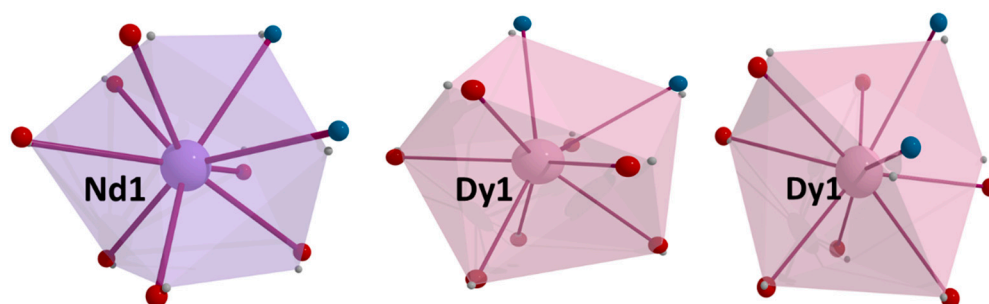
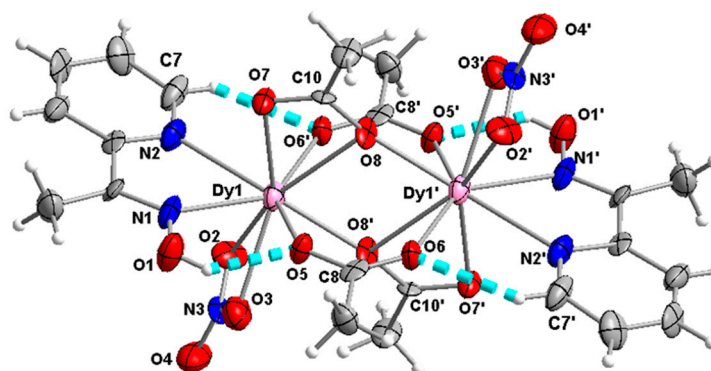
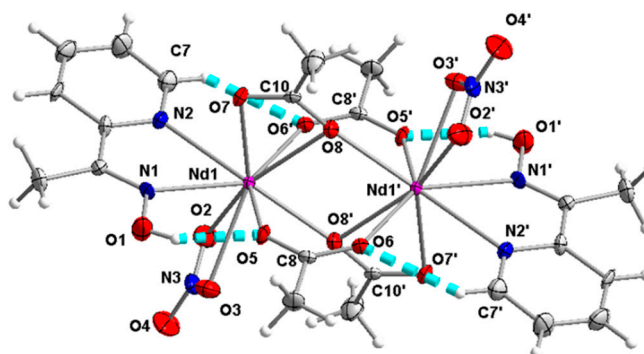


Figure S1. The muffin-type coordination polyhedra of Ln1/Ln1' in complexes **1** (left), **5** (middle) and **6** (right). The plotted polyhedra with the small spheres as vertices are the best-fit polyhedra created by using the program SHAPE.



(a)



(b)

Figure S2. Intramolecular H bonds (dashed cyan lines) in the dinuclear complexes $[\text{Dy}_2(\text{O}_2\text{CMe})_4(\text{NO}_3)_2(\text{mepaoH})_2]$ (**5**, a) and $[\text{Nd}_2(\text{O}_2\text{CMe})_4(\text{NO}_3)_2(\text{mepaoH})_2]$ (**1**, b). The ellipsoids are drawn at the 50% thermal probability level. Symmetry code: (') = $-x+2, -y+1, -z+2$.

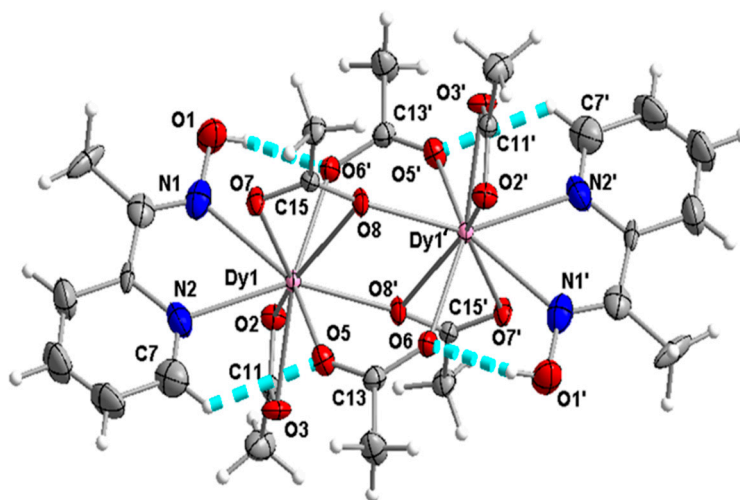


Figure S3. Intramolecular H bonds (dashed cyan lines) in the dinuclear complex $[\text{Dy}_2(\text{O}_2\text{CMe})_6(\text{mepaoH})_2]$ (**6**). The ellipsoids are drawn at the 50% thermal probability level. Symmetry code: (') = $-x, -y, -z$.