Supplementary information

Automated synthesis of prexasertib and derivatives enabled by continuous-flow solid-phase synthesis

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SUPPLEMENTARY INFORMATION

Automated Synthesis of Prexasertib and Derivatives Enabled by Continuous-Flow Solid-Phase Synthesis

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1. General Information

Unless otherwise indicated, all commercially available reagents were used directly from the supplier without further purification. Tetrahydrofuran (THF), dichloromethane (DCM), dimethylformamide (DMF), 1.4-dioxane, diethyl ether ($Et₂O$) and dimethyl sulfoxide (DMSO) were high-performance liquid chromatography (HPLC) grade and anhydrous. Ethyl acetate (EtOAc), Methanol (MeOH) and ethanol (EtOH) were analytical grade, and directly used without further purification.

All microwave reactions were performed on a $CEMTM$ Discover SP Microwave System. Infrared spectra were recorded in potassium bromide (KBr) pellets using a Nicolet AVATER FTIR380 spectrometer. All high-resolution mass spectra (HRMS) were obtained on a Finnigan/MAT 95XL-T spectrometer. Optical rotation was obtained on an Anton Paar MCP-100 digital polarimeter using a 1 cm³ glass cell. The melting points were measured by Buchi B-540 melting point apparatus. Analytical-HPLC was performed using Agilent reversed phase column (4.6 x 150 mm, pore size: 95 Å, particle size: 5 µm, phase: eclipse Plus C18, part number: 959993-902).¹H NMR and ¹³C NMR spectra were recorded at ambient temperature in DMSO- d_6 (2.50 ppm for ¹H NMR, 39.52 ppm for ¹³C NMR) or methanol- d_4 (3.34 ppm for ¹H NMR, 49.0 ppm for ¹³C NMR) using Bruker 400 MHz and Bruker 500 MHz spectrometers. Chemical shift values are expressed as parts per million (ppm), and *J*-values are in hertz (Hz). Splitting patterns are indicated as s: singlet, d: doublet, t: triplet, q: quartet, p: pentet, or combination, brs: broad singlet, or m: multiplet. Analytic HPLC was performed using a Shimadzu Prominence System equipped with a Shim Pack GVP-ODS 2.0 mm C18 column (5 μM, 120Å, 250 mm x 4.60 mm i.d.) at room temperature. Preparative HPLC was performed using a Waters e2695 Separations Module equipped with a C18 column (Luna 5µM, C18(2) 100 Å, LC Column 250×10 mm) with a PDA detector at room temperature.

The human lymphocytic leukemia (non-T, non-B) cell line (REH) and human T-lymphoblastic leukemia (MOLT4) were cultured in RPMI 1640 media (HyClone, CA, USA) supplemented with 10% fetal bovine serum (FBS, HyClone, CA, USA) and 2 mM L-glutamine (Gibco, CA, USA). Cells were seeded in 96 well plates, followed by treatment of prexasertib derivatives for 48 hours. The Cell Counting Kit-8 colorimetric assay (CCK8, Dojindo, Japan) was used to determine the percentage of cell viability.

2. State of the art for multistep continuous-flow synthesis

Supplementary Figure 1. State of the art for multistep continuous-flow synthesis of APIs. a, Based on inline membrane-based phase separators. **b**, Based on polymer supported reagents and scavengers.

3. Description of the automated synthesizer and its components

Images of the automated SPS-flow platform and components are shown in Supplementary Figs. 2 to 7. Back pressure regulators (BPRs), high-purity perfluoroalkoxy polymer (HPFA) tubing, fittings (except stainless steel fittings), and bottom filters were purchased from IDEX Health & Science Technologies. Stainless steel tubing and fittings were purchased from Swagelok Ltd. The SF-10 peristaltic pump was purchased from Vaportec Ltd. The Asia syringe pump was purchased from Syrris Ltd. Selection valves were purchased from Valco Instruments Co. Inc. A remotecontrollable magnetic stirrer was purchased from Heidolph Instruments GmbH & CO. KG.

Supplementary Figure 2. Photograph and interpretation of the assembled automated SPSflow system.

Supplementary Figure 3. Photograph of the pumps. a, Pump #1, Vaportec SF-10 peristaltic pump, 10 bar max. back pressure, 10 mL/min max. flow rate. **b**, Pump #2, Syrris Asia syringe pump, 20 bar max. back pressure, 10 mL/min max. flow rate for the left channel, 2.5 mL/min max. flow rate for the right channel.

Supplementary Figure 4. Photograph of the multiway selector valves. a, 10-position selector valve, VICI EUTA-2SD10MWE. **b**, 16-position selector valve, VICI EUTA-2SD16MWE. **c**, Profile view of the selector valve.

Supplementary Figure 5. Photograph of the reagent storage bottles. a, 100 mL 3-port GL45 screw-capped bottle. **b**, Solvent inlet filter for the 2nd step.

Supplementary Figure 6. Photograph of the column reactor.

Supplementary Figure 7. Photograph of the oil bath and the heating plate.

4. Solution-batch synthesis of prexasertib TFA salt

Supplementary Figure 8. Scheme for the solution batch synthesis of prexasertib TFA salt.

Step 1-protection

To a dry flask containing **1** (2-chlorotrityl chloride) (9.2 mmol, 2.87 g) and amine hydrobromide $(11.0 \text{ mmol}, 2.40 \text{ g})$ was added CH₂Cl₂ (300 mL). Under stirring, Et₃N (22.0 mmol, 3.1 mL) was added dropwise to the resulting mixture over 5 min. The amine hydrobromide salt gradually disappeared within 2 min. The resulting solution was stirred at room temperature for 60 min. Then, saturated aqueous NH₄Cl (200 mL) was added. The aqueous phase was extracted with CH_2Cl_2 (100 mL \times 2). The combined organic phase was washed with water (100 mL) and brine (100 mL),

dried over Na2SO4, and concentrated in vacuum. **2** was obtained as a white solid (3.62 g, 95%) without purification and was ready for use in the following step.

To a solution of $2(200 \text{ mg})$ in 10 mL CH₂Cl₂ was added 0.1 mL CF₃COOH. The resulting mixture was stirred at room temperature for 10 min and then concentrated under vacuum. The residue was purified by flash column chromatography over silica gel (CH2Cl2/CH3OH 10:1) to afford **2'** (103 mg).

$$
\begin{matrix} \mathsf{NH}_{2} \cdot \mathsf{TFA} \\ \mathsf{Br} \\ \mathsf{2} \end{matrix}
$$

1 H NMR (500 MHz, methanol-*d*4) δ 3.53 (t, *J* = 6.4 Hz, 2H), 3.08 (t, *J* = 7.6 Hz, 2H), 2.23 – 2.16 (m, 2H); 13C NMR (125 MHz, methanol-*d*4) δ 39.4, 31.4, 29.8.

See Supplementary Fig. 24 for the NMR spectra of compound 2'.

Step $2-S_N2$

To a dry flask containing methyl 2-hydroxy-6-methoxybenzoate (18.1 mmol, 3.3 g) and $Cs₂CO₃$ (21.8 mmol, 7.08 g) was added anhydrous DMF (150 mL). The resulting mixture was stirred at rt for 30 min and then kept stationary for 10 min. Then, the upper clear solution was transferred to another dry flask containing **2** (7.26 mmol, 3.0 g) via a cannula. The reaction mixture was stirred at room temperature for 120 min and then quenched by the addition of aqueous 0.1 N HCl (50 mL) and extracted with EtOAc (100 mL \times 3). The combined organic layer was washed successively with H₂O (150 mL \times 3) and brine (200 mL) and then dried over Na₂SO₄ and concentrated under vacuum. Purification of the residue via flash chromatography on silica gel (*n*-hexane/EtOAc = 8:1) afforded **3** as a solid (3.45 g, 92% yield). Then, 200 mg **3** and CH2Cl2 (20 mL) was stirred at room temperature for 2 min, followed by the addition of $CF_3COOH (0.1 mL)$. The resulting mixture was stirred at room temperature for 5 min and then concentrated under vacuum. The residue was purified by flash column chromatography over silica gel $\left(CH_2Cl_2/CH_3OH = 10:1\right)$ to afford **3'** (110) mg).

1 H NMR (500 MHz, methanol-*d*4) δ 7.35 (t, *J* = 8.4 Hz, 1H), 6.71 (d, *J* = 6.8, 1H), 6.69 (d, *J* = 6.8, 1H), 4.16 (t, *J* = 5.6 Hz, 2H), 3.84 (s, 3H), 3.80 (s, 3H), 3.11 (t, *J* = 7.0 Hz, 2H), 2.12 – 2.07 (m, 2H); 13C NMR (125 MHz, methanol-*d*4) δ 169.1, 163.3 (q, *J* = 35 Hz, 1C), 159.1, 157.6, 132.9,

118.0 (q, *J* = 290 Hz, 1C), 114.0, 106.0, 105.7, 67.5, 56.5, 52.8, 38.8, 28.0; HRMS calculated for $C_{12}H_{18}NO_4^+ (M+H^+)$ 240.1230, found 240.1231.

See Supplementary Fig. 25 for the NMR spectra of compound 3'.

Step 3-Claisen condensation

A solution of *i*-Pr2NH (12.8 mmol, 1.79 mL) in THF (150 mL) was cooled to –78 °C, followed by the dropwise addition of *n*-BuLi (2 M in cyclohexane, 6.4 mL). The reaction solution was stirred at –78 °C for 5 min, followed by the dropwise addition of CH₃CN (11.6 mmol, 0.61 mL), and then stirred at –78 °C for 10 min. A solution of **3** (5.82 mmol, 3.00 g) in 10 mL THF was added to the above reaction solution. The resulting mixture continued to be stirred at -78 °C for 30 min, after which it was allowed to warm to room temperature and quenched by the addition of 0.5 N HCl (100 mL). The mixture was extracted with EtOAc (100 mL \times 2). The combined organic layer was washed successively with $H_2O(150 \text{ mL})$ and brine (150 mL), dried over Na₂SO₄, and concentrated under vacuum. Purification of the residue via flash chromatography over silica gel (*n*hexane/EtOAc = $8:1$ to $4:1$) afforded 4 (2.65 g, 87%) as a solid.

Then, 100 mg **4** and CH2Cl2 (20 mL) were stirred at room temperature for 2 min, followed by the addition of $CF₃COOH$ (0.1 mL). The resulting mixture was stirred at room temperature for 5 min and then concentrated under vacuum. Purification of the residue via preparative HPLC afforded **4'** (65 mg). Preparative HPLC separation was carried out at a flow rate of 5 mL/min (solvent $A =$ 0.1% v/v TFA in water; solvent $B = 0.1\%$ v/v TFA in acetonitrile).

1 H NMR (400 MHz, methanol-*d*4) δ 7.49 (t, *J* = 8.6 Hz, 0.8H), 7.44 – 7.37 (m, 0.32H), 6.82 (d, *J* = 8.6 Hz, 0.8H), 6.78 (d, *J* = 8.6 Hz, 0.8H), 6.76 – 6.71 (m, 0.4H), 4.22 (t, *J* = 5.6 Hz, 1.6H), 4.20 -4.15 (m, 0.4H), 3.93 (s, 2.2H), 3.88 (s, 0.33H), 3.86 (s, 0.21H), 3.23 – 3.16 (m, 2H), 2.19 – 2.13 (m, 2H); 13C NMR (125 MHz, methanol-*d*4) δ 193.4, 159.7, 159.7, 157.84, 157.83, 134.8, 133.0, 132.9, 117.5, 115.9, 106.3, 106.0, 105.9, 105.7, 105.6, 105.5, 67.9, 67.0, 66.8, 56.6, 56.5, 56.5, 39.0, 38.7, 38.6, 28.3, 28.3, 27.9; HRMS calculated for $C_{13}H_{17}N_2O_3^+$ (M+H⁺) 249.1234, found 249.1236.

See Supplementary Fig. 26 for the NMR spectra of compound 4'.

Step 4-hydrazine condensation

To a microwave tube (35 mL volume) containing **4** (3.81 mmol, 2.00 g) and a magnetic stirrer were added 8 mL THF and 8 mL EtOH followed by the addition of NH₂NH₂•H₂O (19.0 mmol, 0.93 mL) and HOAc (7.62 mmol, 0.45 mL). The tube was capped and subjected to microwave irradiation at 110 °C for 6 h. Then, the mixture was transferred to a flask and concentrated under vacuum. Purification of the residue via chromatography over silica gel $\rm CH_2Cl_2/MeOH/Et_3N =$ 120:3.5:1) afforded **5** as an amorphous foam (1.93 g, 94%).

Then, 200 mg **5** and CH₂Cl₂ (20 mL) was stirred at room temperature for 2 min, followed by the addition of CF₃COOH (0.1 mL). The resulting mixture was stirred at room temperature for 10 min and then concentrated under vacuum. Purification of the residue via flash chromatography over silica gel $\left(\text{CH}_2\text{Cl}_2/\text{CH}_3\text{OH} = 10:1\right)$ afforded **5'** (104 mg).

1 H NMR (400 MHz, methanol-*d*4) δ 7.37 (t, *J* = 8.4 Hz, 1H), 6.97 (s, 1H), 6.82 (d, *J* = 8.4 Hz, 1H), 6.80 (d, *J* = 8.4 Hz, 1H), 4.20 (t, *J* = 5.8 Hz, 2H), 3.91 (s, 3H), 3.20 (t, *J* = 8.2, 7.8 Hz, 2H), 2.24 – 2.13 (m, 2H); 13C NMR (125 MHz, methanol-*d*4) δ 159.5, 158.2, 156.7, 137.6, 131.5, 108.4, 106.4, 105.7, 100.7, 67.0, 56.4, 38.6, 28.4; HRMS calculated for C₁₃H₁₉N₄O₂⁺ (M+H⁺) 263.1503, found 263.1504.

See Supplementary Fig. 27 for the NMR spectra of compound 5'.

Step $5-S_NAr$ $(2.5$ equiv $NH₂$ **MW** parameters N-ethyl morpholine $Time = 300 min$ $(2.5$ equiv) **HN HN** Temp. = $85 °C$ **DMSO** Safety pressure = 250 psi Ph MW, 85 °C, 5 h MeO MeO Power = $100 w$ step 5 Pre-mixing time $=$ 30 s **NH** ŃΗ \mathcal{C} C 89% stir speed: medium 5 6

To a microwave tube (35 mL volume) containing **5** (2.78 mmol, 1.50 g) and a magnetic stirrer were added 15 mL DMSO followed by the addition of 5-chloropyrazine-2-carbonitrile (7.0 mmol, 973 mg) and *N*-ethyl morpholine (7.0 mmol, 0.88 mL). The reaction mixture was subjected to microwave irradiation at 85 °C for 5 h. The mixture was transferred to a separating funnel followed by the addition of H₂O (50 mL). The aqueous phase was extracted with EtOAc (50 mL \times 3), dried

over Na2SO4 and concentrated under vacuum. Purification of the residue via flash chromatography over silica gel (*n*-hexane/EtOAc = 1:1 to $CH_2Cl_2/MeOH = 10:1$) afforded 6 as an amorphous foam $(1.58 \text{ g}, 89\%).$

¹H NMR (400 MHz, Methanol-*d*₄) δ 8.34 (s, 1H), 8.24 (s, 1H), 7.43 – 7.27 (m, 7H), 7.19 (td, *J* = 7.6, 1.7 Hz, 1H), 7.14 – 6.99 (m, 8H), 6.86 (d, *J* = 8.4 Hz, 1H), 6.83 (d, *J* = 8.7 Hz, 1H), 4.35 (t, *J* $= 5.6$ Hz, 2H), 3.97 (s, 3H), 2.38 (t, $J = 6.4$ Hz, 2H), 2.17 – 2.05 (m, 2H). ¹³C NMR (125 MHz, DMSO-*d*6) δ 157.5, 156.8, 151.9, 147.7, 146.3, 144.9, 140.8, 135.3, 134.9, 133.5, 131.5, 131.4, 129.7, 129.6, 128.6, 127.6, 126.3, 125.9, 117.7, 115.8, 107.0, 105.3, 104.1, 98.6, 70.5, 66.2, 55.8, 40.7, 29.7; HRMS calculated for $C_{37}H_{32}CIN_7O_2$ (M+H⁺) 642.2379, found 642.2377.

See Supplementary Fig. 28 for the NMR spectra of compound 6.

Step 6-cleavage

All of the purified **6** (1.58 g) and CH2Cl2 (100 mL) were added to a flask, followed by the addition of CF3COOH (0.5 mL). The resulting mixture was stirred at room temperature for 10 min and then concentrated under vacuum. The residue was obtained as a brown solid.

Protocol for crystallization: MeOH was gradually added to a 50 mL round-bottom flask containing the crude product from step 6 with gentle shaking until all crude product dissolved completely. In total, 5 mL MeOH was consumed. Then, EtOAc (50 mL) was added to the resulting solution with shaking. White solids gradually precipitated overnight. Filtration and washing with MeOH/EtOAc $(v/v = 1:20, 30 \text{ mL})$ and subsequent drying under vacuum at room temperature afforded 7 as a white powder $(1.06 \text{ g}, 90\%)$.

¹H NMR (500 MHz, DMSO-*d*₆) δ 12.34 (s, 1H), 10.76 (s, 1H), 8.69 (d, *J* = 1.3 Hz, 1H), 8.54 (s, 1H), 8.00 (s, 3H), 7.34 (t, *J* = 8.4 Hz, 1H), 6.87 (s, 1H), 6.78 (d, *J* = 8.4, 3.8 Hz, 1H), 6.77 (d, *J* = 8.4, 1H), 4.11 (t, *J* = 6.0 Hz, 2H), 3.81 (s, 3H), 2.99 – 2.90 (m, 2H), 2.08 – 2.00 (m, 2H); 13C NMR (125 MHz, DMSO-*d*6) δ 158.8, 157.7, 156.5, 152.1, 148.1, 146.5, 135.4, 134.7, 130.0, 120.8, 118.4, 117.9, 116.1, 113.6, 107.2, 105.4, 104.5, 98.4, 65.2, 55.9, 36.2, 26.9; HRMS calculated for $C_{18}H_{20}N_7O_2^+$ (M+H⁺) 366.1673, found 366.1675; FT-IR (cm⁻¹): 3428, 3168, 2983, 2227, 1668, 1612, 1571, 1540, 1501, 1464, 1163, 1128, 1096, 1022, 799, 698. Melting point: 225.1 – 226.4 °C.

See Supplementary Fig. 29 for the NMR spectra of compound 7.

5. SPS-batch synthesis of prexasertib TFA salt

Supplementary Figure 9. Scheme for the SPS-batch synthesis of prexasertib TFA salt.

Step 1-attachment of the starting compound to the resin

A mixture of 2-Cl trityl chloride resin (2.0 g, 1.02 mmol/g) and amine hydrobromide (2.45 mmol, 534 mg) in 60 mL CH₂Cl₂ was stirred at room temperature for 10 min, followed by the addition of Et3N (4.9 mmol, 0.68 mL) dropwise over 5 min. The amine hydrobromide salt gradually disappeared within 2 minutes. The resulting mixture was stirred at room temperature for 60 min, and then quenched with 3 mL MeOH. The resin $S6$ was successively washed with $CH_2Cl_2(50 \text{ mL})$, MeOH (50 mL), CH_2Cl_2 (50 mL), and MeOH (50 mL), and dried under vacuum at room temperature for 60 min, after which it was ready for the next step.

FT-IR (cm-1) of resin **S1**: 3567, 3441, 3059, 3024, 2920, 2850, 1600, 1492, 1452, 1062, 1040, 905, 828, 756, 697.

To identify the loading efficiency of the resin:

First, 500 mg 2-Cl trityl chloride resin was converted to resin **S1** by step 1, and then, a mixture of resin **S1** and CH_2Cl_2 (20 mL) was stirred at room temperature for 10 min, followed by the addition of CF3COOH (0.1 mL). The resulting mixture was stirred at room temperature for 10 min, filtered off, and washed with CH_2Cl_2 (20 mL) and MeOH (20 mL). The filtrate was concentrated using a rotary evaporator, and then dried under vacuum for 5 h. The resulting **2'** was obtained, and the loading efficiency was identified (97%, 255 mg), the crude NMR spectra of which coincided with that of **2'** from solution-batch synthesis.

Step $2-S_N2$

To a dry flask containing methyl 2-hydroxy-6-methoxybenzoate (2.55 mmol, 930 mg) and Cs_2CO_3 $(6.12 \text{ mmol}, 2.0 \text{ g})$ was added anhydrous DMF (30 mL) . The resulting mixture was stirred at room

temperature for 30 min, and then kept stationary for another 30 min. The upper clear solution was transferred to another dry flask containing resin **S1** (swelled in 30 mL DMF) from step 1. The resulting reaction mixture was stirred at room temperature for 120 min and then quenched by the addition of 0.1 N HCl (50 mL). The resin $S2$ was filtered off and washed successively with H_2O (100 mL), THF (50 mL), MeOH (50 mL), THF (50 mL), and MeOH (50 mL), and then dried under vacuum at room temperature for 60 min, after which it was ready for the next step.

FT-IR (cm-1) of resin **S2**: 3567, 3058, 3024, 2920, 2848, **1737**, 1597, 1493, 1464, 1453, 1254, 1108, 1072, 756, 697.

Then, resin **S2** (100 mg) was treated with TFA in CH₂Cl₂ (v/v = 0.5%, 50 mL) at room temperature for 10 min. The mixture was subsequently filtered off and washed with CH_2Cl_2 (10 mL) and MeOH (10 mL). The filtrate was concentrated under vacuum, the crude NMR data of which coincided with that of **3'** from solution-batch synthesis.

LDA solution: To a 100 mL dry 3-port GL45 screw-capped bottle containing a stir bar were added 70 mL anhydrous THF and *i*-Pr₂NH (5.5 equiv., 1.57 mL). The solution was cooled to -78 °C, followed by addition of *n*-BuLi (5.5 equiv., 2.0 M in cyclohexane, 5.6 mL). The resulting mixture was vigorously stirred at -78 °C for 5 min and then kept in a dry-ice bath, after which it was ready for use (delivered by Asia syringe pump channel #2).

CH3CN solution: To another 100 mL dry 3-port GL45 screw capped bottle were added 75 mL anhydrous THF and CH3CN (5.0 equiv., 0.53 mL). The resulting solution was mixed well and was ready for use (delivered by the peristaltic pump).

To a dry flask with argon protection containing resin **S2** from step 2 was added THF (20 mL). The flask and two pumps were connected by PFA tubing. Both pumps were set at the flow rate $= 2.5$ mL/min. After 30 min of pumping, to the flask containing the resin was added 0.5 N HCl (100 mL). The resin **S3** was filtered off and washed successively with 0.5 N HCl (100 mL), THF (50 mL), MeOH (50 mL), THF (50 mL), and MeOH (50 mL), and then dried under vacuum at room temperature for 60 min, after which it was ready for the next step.

FT-IR (cm-1) of resin **S3**: 3314, 3081, 3060, 3018, 2958, 2945, 2841, **2260**, **1694**, 1595, 1491, 1467, 1437, 1258, 1095, 1020, 755, 732, 705, 697.

Next, resin **S3** (100 mg) was treated with TFA in CH₂Cl₂ (v/v = 0.5%, 50 mL) at room temperature for 10 min. Then, the mixture was filtered off and washed with CH_2Cl_2 (10 mL) and MeOH (10 mL). The filtrate was concentrated under vacuum, the crude NMR data of which coincided with that of **4'** from solution-batch synthesis.

Step 4-hydrazine condensation

To a microwave tube (35 mL volume) containing resin **S3** and a magnetic stirrer were added 8 mL THF and 8 mL EtOH. The resulting mixture was stirred for 10 min, followed by the addition of NH₂NH₂•H₂O (10.2 mmol, 0.5 mL) and HOAc (4.08 mmol, 0.24 mL). The tube was capped and subjected to microwave irradiation for 6 h. The resulting resin **S4** was washed successively with THF (50 mL), MeOH (50 mL), THF (50 mL), and MeOH (50 mL), and then dried under vacuum at room temperature for 60 min, after which it was ready for the next step.

FT-IR (cm-1) of resin **S4**: **3464**, **3357**, 3058, 3024, 2920, 2850, 1597, 1492, 1452, 1438.

Next, resin **S4** (100 mg) was treated with TFA in CH₂Cl₂ ($v/v = 0.5\%$, 5 mL) at room temperature for 10 min. Then, the mixture was filtered off and washed with CH_2Cl_2 (10 mL) and MeOH (10 mL). The filtrate was concentrated under vacuum, the crude NMR data of which coincided with that of **5'** from solution-batch synthesis.

To a microwave tube (35 mL volume) containing resin **S4** and a magnetic stirrer were added 8 mL 1,4-dioxane and 8 mL DMSO. The resulting mixture was stirred for 10 min, followed by the addition of 5-chloropyrazine-2-carbonitrile (5.1 mmol, 709 mg) and *N*-ethyl morpholine (5.1 mmol, 0.64 mL). The tube was capped and subjected to microwave irradiation at 110 °C for 10 h. Resin **S5** was washed successively with THF (50 mL), MeOH (50 mL), THF (50 mL), and MeOH (50 mL), and then dried under vacuum at room temperature for 60 min, after which it was ready for the next step.

FT-IR (cm-1) of resin **S5**: ν 3430, 3234, 3038, 2969, **2231**, 1673, 1630, 1600, 1542, 1441, 1262, 1206, 1134, 1103, 814, 722.

Step 6-cleavage

All resin **S5** from step 5 was treated with TFA in CH_2Cl_2 (v/v = 0.5%, 50 mL) at room temperature for 10 min. Then, the mixture was filtered off and washed with CH_2Cl_2 (50 mL) and MeOH (50 mL). The filtrate was concentrated under vacuum for 10 min. The crude prexasertib TFA salt was obtained.

Protocol for crystallization: To a 50 mL round-bottom flask containing the crude product was added MeOH (2 mL) gradually with gentle shaking until all crude product dissolved completely. Then to the resulting solution was added EtOAc (20 mL) with shaking. White solids gradually precipitated overnight. Filtration and washing with MeOH/EtOAc ($v/v = 1:20$, 20 mL) and subsequent drying under vacuum at room temperature furnished a white powder (515 mg, 53% overall yield), the NMR data of which coincided with that of prexasertib TFA salt **7** from solutionbatch synthesis.

6. SPS-flow synthesis of prexasertib TFA salt in a column reactor

6.1 Determination of the size of the column reactor

• The matrix of the resin employed is polystyrene crosslinked with 1% divinylbenzene, the swelling factor of which varies with the kinds of solvents. We tested the swelling effect in the solvents involved in the synthesis:

Supplementary Table 1: Identifying the volumes of totally-swollen resins in different solvents.

Supplementary Figure 10. 2.0 g resin swollen in different solvents: (**a**) dry resin, (**b**) DMF, (**c**) CH₂Cl₂, (**d**) THF, (**e**) EtOH-THF ($v/v = 1:1$), (**f**) DMSO-dioxane ($v/v = 1:1$).

• The resin particles have a size of 100 to 200 mesh. Thus, frits with a porosity of 200 mesh (75 µm) can meet the requirement.

Based on the analysis, we customized the stainless steel column (Supplementary Fig. 6): outerdiameter (OD) = 0.5 inch, inner-diameter (ID) = 0.4 inch, length = 16 cm, volume = 13 mL, fitted with 200 mesh frits.

6.2 Establishing the solvent washing protocol

Supplementary Table 2: Establishing the washing protocol between each step.

7. Description of LabVIEW control

7.1 Description of the hardware

7.1.1 Instrument model number

SF-10 Reagent pump Manufacturer: Vapourtec Ltd

Asia Flow Chemistry Syringe Pump Manufacturer: Syrris Ltd

Valco Dead-end Selector with Universal Actuator serial interface: RS-232 Manufacturer: VICI Valco Instruments Co. Inc

Hei-Connect Magnetic Stirrer Manufacturer: Heidolph Instruments

7.1.2 Communication with the computer

All hardware was controlled via a computer through an RS232 cable interface (refer to instrument manuals for the communication setups).

7.1.3 Serial Commands List

For the SF-10 reagent pump, Valco dead-end selector valve, and Hei-connect stir plate, the lists of serial commands can be found in their respective operation manuals or acquired from the manufacturers. Syrris Ltd provides a free application programming interface in C language for controlling the Asia pump from a computer. We used AsiaPumpInterface.h (Version 1.2, Oct 2017). The relevant provided DLL file was placed in the work directory. The functions in the DLL were converted to LabVIEW VI by using the share library converter in LabVIEW.

Below is a list of all communication commands utilized, organized by hardware component.

SF-10 Reagent pump: **STOP SETFLOW X**: Sets flow rate to X ml/min (flow rate range 0.02ml/min to 10 ml/min) **START**

Valco Dead-end Selector: **CP:** Displays the current position **GOnn:** Sends the actuator to position, nn (from 1 to NP), via the shortest route

Asia Flow Chemistry Syringe Pump:

scanForAsiaPumps(): Returns the number of Asia Pumps attached to the PC. This command was sent before other functions.

enterRemoteMode(): Put pump into remote mode. This locks the front screen of the pump (shows EXTERNAL CONTROL)

This mode will automatically exit if the PC does not communicate with the pump at least once every 10 seconds.

pumpStatus(): Returns status of a given pump.

 $Error = -1$ Uninitialised $= 0$ Initialising $= 1$ $Idle = 2$ Filling $= 3$ $Full = 4$ Emptying $= 5$ Empty = 6 Pumping $= 7$

fill(): Tells target pump/channel to fill at specified fill rate.

pump(): Start pumping on the specified pump/channel at the specified rate.

pumpStop(): Halt asia control on the pump and stops the pump.

empty(): Tell pump to empty. Drives both syringes on the target pump/channel to the top.

clearUp(): Release all pumps and ensures everything is stopped. Always call this function as the last call before exiting.

exitRemoteMode(): Stops the pump, puts it back into manual mode and unlocks the menu system on the pump.

Hei-connect stir plate:

RESET: Reset all: activate old interface protocol, heating off, motor off, deactivate remote control **OUT_MODE_4 Y:** Set temperature control mode Y: $0 = \text{Precise-Mode}$; 1 = Fast-Mode **OUT** SP 1 Y: Set temperature to Y °C

OUT SP 3 Y: Set rotating speed to Y rpm

START 1: Start heating: Remote active; "PC" blinking in display MR

START_2: Start rotation: Remote active; "PC" blinking in display MR

STOP 1: Stop heating

STOP 2: Stop rotation

IN PV 1: Display the current sensor temperature

7.2 Software and programming

Code availability: The LabVIEWTM code for operating the SPS-flow automated synthesis in this study is available at https://github.com/nus-automated-flow-system/auto-SPS-Flow-Supplementary-Software.

7.2.1 Software version

For our control software, we utilized NI LabVIEW 2018 (32-bit) NI LabVIEW Runtime 2018

NI-Serial Runtime 17.5 NI-VISA Runtime 18.0

(The application installer includes required runtime engines.)

Below, we describe the different aspects of the control program that we developed.

7.2.2 Program overview

The SPF Flow System Automation program was written in LabVIEW 2018. The front panel contains three parts: user control panel, user monitor panel and condition table panel. The following paragraphs and subsections describe the functionalities available in the program interface shown in Supplementary Fig. 11.

The main process of the automation step is to move the valves to set port positions, fill the Asia pump if necessary, start both the SF-10 and Asia pumps simultaneously and start a built-in timer. Once the target duration has elapsed, the two pumps are stopped simultaneously, and the program will determine if the Asia pump needs to empty the syringes for the next step. Once the emptying process is finished, the program will read the next row conditions and repeat the whole process, starting from the setting of the valve port positions.

One highlight of the program's functionality is the temperature adjustment. The Hei-Connect stir plate uses a built-in proportional-integral-derivative (PID) controller to adjust the temperature. This controller has two modes, namely, precise-mode, which avoids an overshoot in the sample temperature, and fast-mode, in which the heater will heat rapidly but in our testing the overshoot could be as large as 15 °C. We chose to use the stir plate's precise-mode in our program, and in addition to this built-in control, we added the functionality that any pumps that are pumping will be paused whenever the measured temperature of the oil bath is out of a user-defined acceptable temperature range. After the pumps pause, the program will wait for the temperature to return to a different defined range of the set temperature plus or minus one. By setting two temperature ranges, one for pausing pumps (\pm 5 °C) and the other for resuming pumps (\pm 1 °C), the pumps do not rapidly switch on and off when the temperature is at the edge of the user set temperature range. This approach results in a reliable temperature and more precise pumping times.

In addition to the temperature adjustment functionality, another highlight of the program is the ability for users to pause the pumps whenever they are pumping. The pause functionality was vital during the developmental stage of the chemical process; it allowed a chemist user much flexibility in modifying the process.

Supplementary Figure 11. SPS flow system automation LabVIEWTM user interface. This user interface contains three parts: (a) the setup and control panels (labeled Serial Configuration and Parameters Setup) are used to set communication configurations and process parameters and to control the process; (b) the current status is a real-time monitor showing the current settings of the instruments; and (c) the condition table displays the user-defined table of conditions for keeping track of steps.

Condition Table

The condition table is a user-defined table containing the information of duration, flow rate of pump1 and pump2, port position of valve 1-4, hot plate temperature and rotation speed. We modified the condition table in Microsoft Excel and saved it as a tab delimited text file. To correctly interpret the condition table in LabVIEWTM, the titles and units in the text file match the table titles shown in the LabVIEWTM program user interface window. For the heating and cooling steps, we set the duration to "x" or "0" in the condition table, since the duration of this step depends on the stir plate heating speed and cooling speed.

Start from row number

Set the program to start from a user-defined step. The skipped steps are grayed out, and the current step is highlighted with a yellow background.

Serial Configuration

Set the serial communication between the instrument and computer, including COM port number, baud rate, parity and data bits. The automation program has default serial parameters corresponding to each instrument, although the COM port number is dependent on the user's

computer. The Syrris Asia pump hardware did not require serial configuration. After installing the driver, the pump was automatically detected by the computer.

Syringe Type

Set the syringe types loaded into the left and right channels of the Syrris Asia pump to calculate the maximum filling and emptying speed. Note that it is required to set the syringe types directly on the pump interface and to ensure that the same syringe types are specified in the automation program.

Temp Range

Set the acceptable temperature range for the sample. Since the PID heating control system inevitably has fluctuations in temperature, this program includes the functionality of running the process only within the acceptable temperature range to ensure a good outcome of the product. Once the temperature measured by the sensor is out of this range, the automation program will pause the timer and pumps and resume once the temperature is accurate again.

7.2.3 State machine

The program follows a state machine structure for executing the flow system process. Here, we describe the state machine logic that the control program is built around.

Supplementary Figure 12. SFS flow system automation program state machine chart.

Start operation (default):

This state is the default beginning state when the program enters the state machine structure. It checks if the condition table has entries and sends the initialize command to the Syrris Asia pump.

```
1. if the row number is greater than 0:
       send initialize command to Syrris Asia pump
       go to next state "Initialize"
  else:
       stay in the current state
       prompt: no entries
2. Reset pumps indicators
```
Initialize:

This state is used to wait for the Syrris Asia pump to finish the initializing process for both channels. This process will take several minutes. During this process, the operation buttons (START, STOP, PAUSE) are disabled in order to prevent intervention in initializing Syrris Asia pump. Intervening in this process may cause the pump to inaccurately determine its home position.

```
1. if both the left channel state and right channel state are at 
  initialized:
       go to next state "Wait to start"
  else:
       stay in the current state
```
Wait to start:

After completion of the "Initialize" state, the START and STOP buttons are enabled in this state. The "Wait to start" state is used to wait for the user to start the process by pressing the START button, or to exit the program by pressing the STOP button.

```
1. Enable Start and STOP button
2. if user press the START button:
      go to next state "Set valve position"
  else:
       stay in the current state
```
Set valve positions:

This state sends a command to the valves to go the set port position via the **shortest route** and highlights the current step in the condition table.

- 1. set the current row in the condition table with lime background
- 2. send commands to the actuator to set the valves port

```
3. check the current valve position
  if all four valves have reached the set positions:
      go to next state "Turn on heating"
  else:
       stay in the current state
```
Turn on heating:

This state sends a command to turn on the heater and motor of the stir plate if the temperature and rotation speed are set in the condition table. The heating function of the Hei-Connect stir plate has two modes: precise-mode and fast-mode. We set the **precise-mode** as default to enable heating without overshooting for temperature-sensitive samples.

Wait to reach temp:

This state is used to wait for the oil bath to reach the set temperature. During this state, the timer will be paused.

```
1. pause the timer
```
2. if the current temperature is within 1 degree around the set temperature: go to state "Resume pumps" else:

stay in the current state

Fill syringe pump:

In this state, the Syrris Asia pump will fill the syringes if needed. There are two cases where the syringes need to be filled before pumping:

1. The set flow rate is nonzero, and the valve 2 position has changed from the previous row

2. The set flow rate is nonzero, and the valve 2 position is the same as the previous row, but the set flowrate from the previous row is zero; i.e., the pump did not run in the previous row and was not filled with solution.

```
1. if set flow rate of the left channel is not zero and either 
  valve2 set position is different from previous row or set 
  flow rate of the left channel is different from previous row:
      if the left channel status is not full:
           fill the left channel
      else:
           return left channel filling finished as TRUE
  else:
       return left channel filling finished as TRUE
2. if set flow rate of the right channel is not zero and either 
  valve 2 set position is different from previous row or set 
  flow rate of the right channel is different from previous row:
      if the right channel status is not full:
           fill the right channel
      else:
           return right channel filling finished as TRUE
  else:
       return right channel filling finished as TRUE
3. if both left and right channel filling finished Booleans are 
  TRUE:
      go to state "Start pumps"
  else:
       stay in the current state
```
Start pumps:

This state sends commands to the corresponding pumps to start pumping.


```
and start pumping
  else:
       send command to Syrris Asia right channel to stop pumping
4. start the timer
5. go to state "Check temp & time"
```
Check temp & time:

This state continually updates the timer and measured temperature in cycles. The two cases that will cause the process to pause are as follows: 1. The measured temperature is not within the userdefined temperature range; 2. The user presses the PAUSE button. If either of the two cases occur, the process will go to the state "Pause pumps"; otherwise, it will stay in the current state until the set duration has elapsed and then go to state "Stop pumps".

```
1. if set temperature is nonzero but the measured temperature is 
  not in the range around set temperature, or user presses PAUSE 
  button:
      go to state "Pause pumps"
  else if the elapsed time has reached the set duration:
       go to state "Stop pumps"
  else:
       stay in the current state
```
Pause pumps:

This state pauses the running pumps and timer. If the user pressed the PAUSE button, this state will wait for the user to press the PAUSE button again to resume the process. Otherwise, if the pause state was triggered by an out-of-range measured temperature, this state will lead to the state "Wait to reach temp".

```
1. if any pump is pumping:
     send command to SF-10 to stop pumping
   send command to Syrris Asia left and right channels to 
stop pumping
2. pause the timer
3. if user pressed PAUSE button:
   stay in the current state
else:
     go to state "Wait to reach temp"
```
Resume pumps:

This state resumes the pumps to pump with the current flow rate.

1. if set flow rate of pump1 (SF-10) is nonzero: send command to SF-10 to set the flow rate send command to SF-10 to start pumping

else: send command to SF-10 to stop pumping 2. if set flow rate of pump2 (Syrris Asia) left channel is nonzero: send command to Syrris Asia left channel to set the flow rate send command to Syrris Asia left channel to start pumping else: send command to Syrris Asia left channel to stop pumping 3. if set flow rate of pump2 (Syrris Asia) right channel is nonzero: send command to Syrris Asia right channel to set the flow rate send command to Syrris Asia right channel to start pumping else: send command to Syrris Asia right channel to stop pumping 4. go to state "Check temp & time"

Stop pumps:

This state determines whether the Syrris Asia pump syringes need to be emptied before going to the next step. It will wait until the emptying process is completed and then go to the next state.

There are two cases where the syringes need to be emptied:

1. The flow rate of the finished step is nonzero, but the flow rate of the next step is zero, i.e. the pump does not need to pump in the next step;

2. The flow rate of the finished step is nonzero, but the valve 2 port position needs to change for the next row.

If either one or both syringes need to be emptied, the process will stay in this state and wait for both syringes to get ready for the next state. If neither of the syringes need to be emptied, it will go to the next state "Check next step".

2. if the flow rate of Syrris Asia right channel was nonzero and either the next flow rate of right channel is zero or the valve 2 position in the next row is different from the current row: if valve 4 is at the waste port: if the right channel status is not empty yet: send empty command to Syrris Asia right channel else: return right channel emptying finished as TRUE else: send command to valve 4 to go to the waste port else: # No need to empty the right channel return right channel emptying finished as TRUE 3. if both left and right channel emptying finished Booleans are TRUE: go to state "Check next step" else: stay in the current state

Check the next step:

This state checks if the next row in the condition table has entries. If it does, the program will repeat the process from the state "Set valve position"; otherwise, the whole process has finished, and the program will clear indicators in the interface, close the communcation and exit

1. if the set duration, flow rates of two pumps, temperature and rotation speed of stir plate are all zero: go to state "No more entries" else: go to state "Set valves position"

7.3 Suggested operation procedure

In this chapter, we walk through an operation procedure for using the SPS Flow System Automation program to conduct a production process.

- Prepare the condition table: Fill in the condition table (in the form of an Excel file, **Supplementary Table 3**), and save as a **Tab delimited .txt file** Note: The heating and cooling steps require entering "x" or "0" under Duration (Minutes); otherwise, the system may not reach the expected temperature. In the next row, enter the duration and conditions that you would like the system to perform at the expected temperature. Note: For steps that do not need to be conducted at a specific temperature, enter "off" under Temperature (Celsius)
- Setup hardware: Setup the instruments, connect the system tubing and prepare the solutions
- Connect to computer: Connect instruments to the computer via USB cable and USB hub. Contact the instrument manufacturers for detailed connection information.
- Setup the communication: Open the SPS Flow System Automation program. Select the COM port corresponding to each instrument, except the Syrris Asia pump, which is automatically connected (refer to Syrris Asia pump Interface Library Reference Version 1.2). Refer to the communication configurations stated in each instrument's user manual.
- Input parameters: Select the syringe types used in the Syrris Asia pump. Set the acceptable temperature range in degree Celsius.
- Run the program: Click the run arrow ϕ to run the program
- Load the condition table: In the prompt window, choose the path to the condition table that you created in the first step. Check if the loaded condition table is correct by viewing the Condition Table on the user interface. The program will start initializing the Syrris Asia pump. Once the initialization is completed and the START button is enabled, it is ready to start the process
- Start the process by clicking the START button
- Pause/Stop the process if needed: During the process, if users need to pause or terminate the procedure, they can press the PAUSE or STOP button. If the procedure is paused, all pumps will stop pumping, the program timer will pause timing, and these will be resumed after the user resumes the process by pressing the PAUSE button again. If the procedure is terminated by pressing the STOP button, the program will stop and exit after closing communication with the instruments. Note: If the program is stopped when the Syrris Asia pump syringes are not empty, they will not be emptied automatically; they will be emptied in the initialization process when you run the program again.

Note: It is not recommended to abort the program using the Abort button \bullet , since the program will not stop pumps if it is aborted. Use the STOP button instead.

• Wait for the process to finish automatically.

Supplementary Table 3: The condition table in the form of an Excel file for automated synthesis of prexasertib.

8. Automated SPS-flow synthesis of prexasetib and analogs

8.1 Automated synthesis of prexasetib and analogs $(5th$ step as the nucleophilic aromatic substitution reaction)

8.1.1 Preparation and storage of reagents

a. $1st$ reagent solution: To a 100 mL dry 3-port GL45 screw capped bottle containing a stir bar were added 60 mL CH₂Cl₂ and 3-bromo-propan-*l*-amine hydrobromide (1.2 equiv., 533 mg). To the resulting mixture was added Et₃N $(2.4 \text{ equiv.}, 0.68 \text{ mL})$ dropwise. The amine hydrobromide salt gradually dissolved. The resulting solution was used as a stock solution for automated synthesis *(connecting to valve 1)*.

OMe COOMe

CN

- h_{\cdot} σ ^{OCs} for the 2nd step: To a 100 mL dry 3-port GL45 screw capped bottle containing a stir bar were added methyl 2-hydroxy-6-methoxybenzoate (2.5 equiv., 930 mg) and Cs_2CO_3 (3.0 equiv., 1990 mg), followed by the addition of anhydrous DMF (60 mL). The blue screw cap was then sealed. The resulting mixture was stirred at rt for 30 min, and kept stationary for 5 min. A GL45 screw cap equipped with a bottom filter (Supplementary Fig. 5b) was used to replace the blue cap. The resulting mixture was ready for use in automated synthesis *(connecting to valve 1).*
- c. LDA solution for the 3^{rd} step: To a 100 mL dry 3-port GL45 screw capped bottle containing a stir bar were added 70 mL anhydrous THF and *i*-Pr2NH (5.5 equiv., 1.57 mL). Then, the solution was cooled to -78 °C, followed by the addition of *n*-BuLi (5.5 equiv., 2.0 M in cyclohexane, 5.6 mL). The resulting mixture was stirred at -78 °C for 5 min, and then kept in a dry-ice bath for use in automated synthesis *(connecting to valve 2).*
- d. CH3CN solution for the 3rd step: To a 100 mL dry 3-port GL45 screw capped bottle were added 75 mL anhydrous THF and CH3CN (5.0 equiv., 0.53 mL). The resulting solution was mixed well and was ready for use as a stock solution for automated synthesis *(connecting to valve 1).*
- e. 4th reagent solution: To a 100 mL 3-port GL45 screw capped bottle were added 30 mL THF, 30 mL EtOH, NH2NH2•H2O (5.0 equiv., 0.5 mL) and HOAc (2.0 equiv., 0.24 mL). The resulting solution was mixed well and was ready for use as a stock solution for automated synthesis *(connecting to valve 2).*
- f. N N solution for the 5th step: To a 100 mL 3-port GL45 screw capped bottle were added 15 mL DMSO, 15 mL 1,4-dioxane, and 5-chloropyrazine-2-carbonitrile (2.5 equiv., 708 mg). The resulting solution was mixed well and was ready for use as a stock solution for automated synthesis *(connecting to valve 1).*
- g. *N*-Ethyl morpholine solution for the 5th step: To a 100 mL 3-port GL45 screw capped bottle were added 15 mL DMSO, 15 mL 1,4-dioxane, and 4-ethyl morpholine (2.5 equiv., 0.64 mL). The resulting solution was mixed well and was ready for use as a stock solution for automated synthesis *(connecting to valve 2).*

h. CF₃COOH solution ($v/v = 0.5\%$) for the 6th step: To a 500 mL 3-port GL45 screw capped bottle were added 360 mL CH₂Cl₂ and 1.6 mL CF₃COOH. The resulting solution was mixed well and was ready for use as a stock solution for automated synthesis (*connecting to valve 1).*

8.1.2 Establishing the chemical recipe file (CRF) #1 for automated synthesis ($5th$ step as the nucleophilic aromatic substitution reaction)

For all parameters of CRF #1, see Supplementary Table 4.

Supplementary Figure 13. CRF #1 (5th step as the nucleophilic aromatic substitution reaction).

Supplementary Figure 14. Schematic drawing of automated synthesizer for the synthesis of prexasetib.

Supplementary Table 4: Reagents and solvents corresponding to the storage slots and connection to the multiway valves for the synthesis of prexasertib.

Supplementary Figure 15. Comparison of 1H NMR spectra between pure prexasertib TFA salt and the crude product obtained from automated multistep synthesis.

Supplementary Figure 16. Chromatogram of prexasertib TFA salt produced automatically using the SPS-flow system after purification by crystallization (HPLC conditions:

temperature: 25 °C; solvent A = 0.1% (v/v) TFA in water; solvent B = 0.1% (v/v) TFA in acetonitrile; 5-80% solvent B over 15 min, flow rate: 1 mL/min). The HPLC analysis showed the purity of the product was >99.9%.

8.1.3 Comparison of waste generation for synthesis of 1 gram prexasertib TFA salt via solutionbatch synthesis *vs* automated SPS-flow synthesis

Supplementary Table 5: Waste generation in solution-batch synthesis vs automated SPS-flow synthesis.

The CRF for the automated synthesis of prexasertib was also applied to automated synthesis of compound **8** without any modification.

Compound **8**

Protocol for Purification: The collection was concentrated under vacuum. The residue was purified over silica gel chromatograph (CH₂Cl₂/MeOH = 5:1), and recrystallization with CH₂Cl₂/EtOAc. A pale-yellow solid was obtained (545 mg, 52% yield).

¹H NMR (500 MHz, DMSO-*d*₆) δ 12.32 (s, 1H), 9.39 (s, 1H), 7.91 (brs, 3H), 7.34 (t, *J* = 8.0 Hz, 1H), 6.78 (d, *J* = 8.0 Hz, 1H), 6.77 (d, *J* = 8.0 Hz, 1H), 6.32 (s, 1H), 4.11 (t, *J* = 6.0 Hz, 2H), 3.81 (s, 3H), 2.95 – 2.85 (m, 2H), 2.05 – 1.95 (m, 2H); ¹³C NMR (125 MHz, DMSO‐*d6*) δ 158.4 (q, *J* $= 31.3$ Hz), 157.6, 156.5, 146.3, 144.6 (d, $J = 231.2$, 15.7 Hz), 136.2-136.0 (m), 134.9, 132.1 (app. dd, *J* = 250.6 Hz, 34.6 Hz), 130.0, 107.2, 105.4, 104.5, 100.7, 65.2, 55.9, 36.1, 26.8; 19F NMR (376 MHz, DMSO-*d6*) δ -73.78 (s, 3F), -95.62 to -95.81 (m, 2F), -156.26 to -156.45 (m, 2F); HRMS calculated for $C_{18}H_{18}F_4N_5O_2^+(M+H^+)$ 412.1391, found 412.1395; FTIR (cm⁻¹) 3407, 3239, 3081, 2970, 1611, 1563, 1542, 1496, 1465; Melting point: 218.1 – 219.0 °C.

See Supplementary Fig. 30 and Fig. 30 for the NMR spectra of compound 8.

8.2 Automated synthesis with the 5th step as the amide coupling (CRF $#2$)

8.2.1 Preparation and storage of reagents for CRF #2

The 1st, $2nd$, $3rd$, $4th$, and $6th$ reagent solutions were the same as those in CRF #1.

Solutoion-1 for the $5th$ step: To a 100 mL dry 3-port GL45 screw capped bottle containing HATU (2.0 equiv., 1550 mg) and *iso*-nicotinic acid (2.0 equiv., 500 mg) was added 30 mL anhydrous DMF. The resulting solution was ready for the use as a stock solution for automation synthesis (*connecting to valve 1*).

Solutoion-2 for the $5th$ step: To a 100 mL dry 3-port GL45 screw capped bottle was added 30 mL DMF and $(i-Pr)_2$ NEt (4.0 equiv., 1.42 mL). The resulting solution was ready for use as a stock solution for automation synthesis (*connecting to valve 2*).

8.2.2 Establishing the CRF #2 ($5th$ step as the amide coupling reaction)

Supplementary Figure 17. CRF #2 ($5th$ **step as the amide coupling reaction).** HATU = 1-[Bis(dimethylamino)methylene]-1*H*-1,2,3-triazolo[4,5-b]pyridinium 3-Oxide Hexafluorophosphate.

Supplementary Table 6: The condition table in the form of an Excel file for CRF #2.

Compound **9**

Protocol for purification: The collection was concentrated under vacuum. The resulted crude product was dissolved in CH_2Cl_2 (5 mL), followed by addition of EtOAc (10 mL) slowly. Overnight crystallization following by washing with $EtOAc-CH₂Cl₂ ($v/v = 4:1, 20 \text{ mL}$) and drying$ under vacuum at room temperature afforded the pure compound as a yellow powder (540 mg, 55% yield).

¹H NMR (500 MHz, DMSO-*d*₆) δ 10.62 (s, 1H), 9.34 (d, *J* = 1.0 Hz, 1H), 8.92 (d, *J* = 2.5 Hz, 1H), 8.80 (t, *J* = 1.5 Hz, 1H), 7.91 (s, 3H), 7.34 (t, *J* = 8.5 Hz, 1H), 6.97 (s, 1H), 6.78 (d, *J* = 8.5 Hz, 1H), 6.77 (d, *J* = 8.5 Hz, 1H), 4.13 (t, *J* = 6.0 Hz, 2H), 3.81 (s, 3H), 3.06 – 2.96 (m, 2H), 2.08 – 2.00 (m, 2H); 13C NMR (125 MHz, DMSO-*d6*) δ 160.8, 158.7 (q, *J =* 35.1 Hz), 157.8, 156.6, 147.9, 145.1, 144.6, 144.0, 143.5, 135.1, 130.1, 116.1 (q, *J =* 291.1 Hz), 107.5, 105.4, 104.6, 98.9, 65.4,

55.9, 36.4, 27.0; HRMS calculated for $C_{18}H_{21}N_6O_3^+$ (M+H⁺) 369.1670, found 369.1673; FTIR (cm^{-1}) : 3407, 3239, 3081, 2970, 1611, 1563, 1542, 1496, 1465; Melting point: 218.0 – 219.1 °C.

See Supplementary Fig. 32 for the NMR spectra of compound 9.

Compound **10**

Protocol for purification: The collection was concentrated under vacuum. The resulted crude product was dissolved in CH_2Cl_2 (5 mL), followed by addition of EtOAc (10 mL) slowly. Overnight crystallization with EtOAc-CH₂Cl₂ ($v/v = 4:1$) washing (20 mL) and drying under vacuum at room temperature afforded the pure compound as a pale-yellow powder (605 mg, 56% yield).

1 H NMR (500 MHz, DMSO-*d6*) δ 10.95 (s, 1H), 9.05 (d, *J =* 8.5 Hz, 1H), 8.62 (d, *J =* 5.5 Hz, 1H), 8.12 – 8.05 (m, 2H), 7.93 (brs, 3H), 7.90 – 7.84 (m, 1H), 7.80 – 7.75 (m, 1H), 7.35 (t, *J* = 8.5 Hz, 1H), 7.03 (s, 1H), 6.80 (d, *J* = 8.5 Hz, 1H), 6.80 (d, *J* = 8.5 Hz, 1H), 4.14 (t, *J* = 6.2 Hz, 2H), 3.83 (s, 3H), 3.07 – 2.96 (m, 1H), 2.11 – 1.99 (m, 1H); 13C NMR (125 MHz, DMSO-*d6*) δ 163.6, 158.7 (q, *J =* 35.0 Hz), 157.8, 156.7, 150.2, 145.6, 140.9, 136.8, 135.0, 130.9, 130.1, 128.8, 127.3, 126.6, 125.8, 123.9, 116.1 (q, *J =* 291.1 Hz), 107.6, 105.4, 104.6, 98.6, 65.4, 55.9, 36.4, 27.0; HRMS calculated for $C_{23}H_{24}N_5O_3$ (M+H⁺) 418.1874, found 418.1875; FTIR (cm⁻¹): 3407, 3239, 3081, 2970, 1611, 1563, 1542, 1496, 1465; Melting point: 190.4 – 191.8 °C.

See Supplementary Fig. 33 for the NMR spectra of compound 10.

Compound **11**

Protocol for purification: The collection was concentrated under vacuum. To a 50 mL flask containing the crude solid was added MeOH (10 mL). Then resulted suspension was stirred vigorously at 40 °C for 10 min, followed by addition of CH_2Cl_2 (5 mL). The mixture was kept stationary at room temperature for 10 min, filtrated, washed with cooled MeOH (10 mL) and CH_2Cl_2 (5 mL). Drying under vacuum at room temperature furnished a white solid (768 mg, 70%) yield).

¹H NMR (500 MHz, DMSO-*d*₆) δ 9.70 (s, 1H), 7.84 (brs, 3H), 7.31 (t, *J* = 8.4 Hz, 1H), 6.76 (d, *J* = 8.4 Hz, 1H), 6.75 (d, *J* = 8.4 Hz, 2H), 6.68 (brs, 1H), 4.08 (t, *J* = 6.0 Hz, 2H), 3.77 (s, 3H), 2.95 $(q, J = 6.5 \text{ Hz}, 2\text{H})$, 2.01 – 1.96 (m, 5H), 1.92 (d, $J = 2.9 \text{ Hz}, 6\text{H}$), 1.72 – 1.67 (m, 6H); ¹³C NMR (125 MHz, DMSO-*d6*) δ 175.3, 158.4 (q, *J =* 33.9 Hz), 157.7, 156.5, 146.1, 134.6, 129.8, 116.4 (q, *J =* 295.7 Hz), 108.0, 105.3, 104.5, 98.8, 65.3, 55.8, 40.6, 38.3, 36.4, 36.0, 27.7, 26.8; HRMS calculated for $C_{18}H_{21}N_6O_3$ (M+H⁺) 425.2547, found 425.2548; FTIR (cm⁻¹): 3407, 3239, 3081, 2970, 1611, 1563, 1542, 1496, 1465; Melting point: 212.5 – 214.1 °C.

See Supplementary Fig. 34 for the NMR spectra of compound 11.

Compound **12**

Protocol for purification: The solution was concentrated under vacuum. To a 50 mL flask containing the crude solid was added MeOH (10 mL). Then resulted suspension was stirred vigorously at 40 °C for 10 min, followed by addition of CH_2Cl_2 (5 mL). The mixture was kept stationary at room temperature for 10 min, filtrated, washed with cooled MeOH (10 mL) and $CH₂Cl₂$ (5 mL). Drying under vacuum at room temperature furnished a white solid (790 mg, 68%) yield).

1 H NMR (500 MHz, DMSO-*d6*) δ 9.76 (s, 1H), 7.83 (brs, 3H), 7.31 (t, *J* = 8.5 Hz, 1H), 6.75 (d, *J* = 8.5 Hz, 1H), 6.74 (d, *J* = 8.5 Hz, 1H), 6.66 (brs, 1H), 4.08 (t, *J* = 6.0 Hz, 2H), 3.77 (s, 3H), 3.59 $(s, 3H)$, 2.94 (q, $J = 6.5$ Hz, 2H), 2.02 – 1.95 (m, 2H), 1.85 – 1.72 (m, 12H); ¹³C NMR (125 MHz, DMSO-*d6*) δ 177.2, 174.8, 158.5 (q, *J* = 34.9 Hz), 157.7, 156.5, 146.1, 134.7, 129.9, 116.4 (q, *J* = 291.6 Hz), 107.8, 105.3, 104.5, 98.8, 65.3, 55.8, 51.6, 38.8, 38.2, 36.4, 27.7, 27.2, 26.8; HRMS calculated for $C_{18}H_{21}N_6O_3$ (M+H⁺) 457.2446, found 457.2453; FTIR (cm⁻¹) 3407, 3239, 3081, 2970, 1611, 1563, 1542, 1496, 1465. Melting point: 219.8 – 222.1 °C.

See Supplementary Fig. 35 for the NMR spectra of compound 12.

Protocol for purification: The collection was concentrated under vacuum. The resulted crude product was dissolved in CH_2Cl_2 (6 mL), followed by addition of EtOAc (10 mL) slowly. Overnight crystallization with EtOAc-CH₂Cl₂ ($v/v = 4:1$) washing (20 mL) and drying under vacuum at room temperature afforded the pure compound as a pale-yellow powder (720 mg, 63% yield).

1 H NMR (500 MHz, DMSO-*d6*) δ 10.64 (s, 1H), 8.24 (d, *J =* 8.0 Hz, 1H), 8.19 (d, *J =* 8.5 Hz, 1H), 7.91 – 7.87 (m, 1H), 7.83 (brs, 3H), 7.74 (s, 1H), 7.73 – 7.69 (m, 1H), 7.36 (t, *J* = 8.4 Hz, 1H), 7.00 (s, 1H), 6.80 (d, *J* = 8.4 Hz, 1H), 6.79 (d, *J* = 8.4 Hz, 1H), 4.18 (s, 3H), 4.13 (t, *J* = 6.0 Hz, 2H), 3.82 (s, 3H), 3.05 – 2.94 (m, 2H), 2.08 – 1.97 (m, 2H); 13C NMR (125 MHz, DMSO-*d6*) δ 163.5, 161.3, 158.3 (q, *J =* 34.8 Hz), 157.7, 156.5, 150.9, 146.9, 145.3, 135.0, 131.0, 130.1, 129.1, 127.6, 121.7, 121.4, 116.1 (q, *J =* 292.1 Hz), 107.5, 105.4, 104.5, 98.2, 98.0, 65.3, 56.5, 55.9, 36.3, 26.9; HRMS calculated for $C_{24}H_{26}N_5O_4$ (M+H⁺) 448.1979, found 448.1987; FTIR (cm⁻¹) 3407, 3239, 3081, 2970, 1611, 1563, 1542, 1496, 1465. Melting point: 220.9 – 221.4 °C.

See Supplementary Fig. 36 for the NMR spectra of compound 13.

Compound **14**

Protocol for purification: The collection was concentrated under vacuum. The resulted crude product was dissolved in CH₂Cl₂ (4 mL), followed by addition of *n*-Hexane/EtOAc ($v/v = 1:2$) (5 mL) slowly. Overnight crystallization with *n*-hexane/EtOAc (*v/v* = 1:2) washing (5 mL) and drying under vacuum at room temperature afforded the pure compound as a yellow powder (610 mg, 54% yield).

¹H NMR (500 MHz, DMSO-*d*₆) δ 9.06 (s, 1H), 7.83 (brs, 3H), 7.48 – 7.41 (m, 4H), 7.31 (t, *J* = 8.4 Hz, 1H), 6.75 (d, *J* = 8.4, 1H), 6.74 (d, *J* = 8.4, 1H), 6.66 (s, 1H), 4.07 (t, *J* = 6.0 Hz, 2H), 3.76 (s, 3H), 2.97-2.87 (m, 2H), 2.01 – 1.92 (m, 2H), 1.48 (dd, *J* = 6.8 Hz, 4.2 Hz, 2H), 1.10 (dd, *J* = 6.8

Hz, 4.2 Hz, 2H); 13C NMR (125 MHz, DMSO-*d6*) δ 169.9, 158.5 (q, *J =* 34.8 Hz), 157.7, 156.5, 145.8, 139.0, 134.7, 132.0, 131.8, 130.0, 128.6, 116.1 (q, *J =* 291.8 Hz), 107.7, 105.3, 104.5, 98.5, 65.3, 55.8, 36.3, 30.6, 26.8, 15.0; HRMS calculated for $C_{23}H_{26}CIN_4O_3$ (M+H⁺) 441.1688, found 441.1689; FTIR (cm⁻¹) 3407, 3239, 3081, 2970, 1611, 1563, 1542, 1496, 1465. Melting point: $179.6 - 181.5$ °C.

See Supplementary Fig. 37 for the NMR spectra of compound 14.

Compound **15**

61% yield, 681 mg, 26 hrs

Protocol for purification: The collection was concentrated under vacuum. The crude compound was dissolved in a mixture solvent of $CH_2Cl_2/MeOH$ (5 mL/1mL), followed by addition of EtOAc (2 mL) slowly. Overnight crystallization with EtOAc washing (5 mL) and drying under vacuum at room temperature afforded the pure compound as a chartreuse powder (681 mg, 61% yield).

¹H NMR (500 MHz, DMSO-*d*₆) δ 12.38 (s, 1H), 10.93 (s, 1H), 9.00 (s, 1H), 8.02 (dd, *J* = 7.9, 1.6 Hz, 1H), 7.83 – 7.79 (m, 1H), 7.78 (s, 3H), 7.58 (d, *J* = 8.4 Hz, 1H), 7.50 (td, *J* = 7.6, 1.1 Hz, 1H), 7.35 (t, *J* = 8.4 Hz, 1H), 6.93 (s, 1H), 6.79 (d, *J* = 8.3 Hz), 6.78 (d, *J* = 8.3 Hz), 4.11 (t, *J* = 6.0 Hz, 2H), 3.81 (s, 3H), 3.35 (s, 5H), 3.00 – 2.91 (m, 2H), 2.05– 1.95 (m, 2H); 13C NMR (125 MHz, DMSO-*d6*) δ 161.0, 158.4, 158.1 (q, *J =* 31 Hz), 157.7, 156.6, 154.0, 147.9, 145.5, 134.8, 134.5, 130.3, 130.1, 125.4, 118.9, 118.6, 117.8 (q, *J =* 300 Hz), 116.3, 107.4, 105.4, 104.5, 98.5, 65.2, 55.9, 36.3, 26.9; HRMS calculated for C₂₄H₂₆N₅O₄ (M+H⁺) 435.1663, found 435.1668; FTIR (cm⁻ ¹) 3407, 3239, 3081, 2970, 1611, 1563, 1542, 1496, 1465; Melting point: 248.1 – 249.6 °C.

See Supplementary Fig. 38 for the NMR spectra of compound 15.

Compound **16**

Protocol for purification: The collection was concentrated under vacuum. The crude solid was dissolved in CH₂Cl₂/MeOH (5 mL/1 mL) followed by addition of Et₂O (5 mL). Overnight crystallization with Et_2O washing (10 mL) and drying under vacuum at room temperature furnished a white powder (707 mg, 60% yield).

1 H NMR (500 MHz, DMSO-*d*6) δ 10.54 (s, 1H), 7.83 (s, 3H), 7.60 (d, *J* = 2.6 Hz, 1H), 7.38 (dd, *J* = 8.9, 2.6 Hz, 1H), 7.32 (t, *J* = 8.4 Hz, 1H), 7.12 (d, *J* = 8.9 Hz, 1H), 6.78 – 6.73 (m, 3H), 4.87 (s, 2H), 4.09 (t, *J* = 6.2 Hz, 2H), 3.78 (s, 3H), 3.01 – 2.85 (m, 2H), 2.04 – 1.94 (m, 2H); 13C NMR (125 MHz, DMSO-*d6*) δ 164.7, 158.5 (q, *J* = 35.4 Hz), 157.7, 156.5, 152.7, 145.6, 134.6, 129.9, 129.4, 128.1, 125.0, 122.4, 116.0 (q, *J* = 292.9 Hz), 115.2, 107.5, 105.4, 104.5, 98.2, 67.5, 65.2 , 55.8, 36.2, 26.8; HRMS calculated for $C_{21}H_{23}Cl_2N_4O_4^+$ (M+H⁺) 465.1091, found 465.1088; FT-IR (cm-1): 3447, 3181, 3099, 2982, 2946, 2893, 2854, 1675, 1634, 1595, 1481, 1430, 1287, 1247, 1207, 1174, 1131, 1105, 834, 725; Melting point: 190.4 – 192.7 °C.

See Supplementary Fig. 39 for the NMR spectra of compound 16.

Compound **17**

Protocol for purification: The collection was concentrated under vacuum. The crude solid was dissolved in CH₂Cl₂ (5 mL), followed by addition of Et₂O (3 mL). Overnight crystallization with $Et₂O$ washing (20 mL) and dying under vacuum at room temperature furnished a pale-yellow powder (637 mg, 65% yield).

1 H NMR (500 MHz, DMSO-*d6*) δ 11.28 (s, 1H), 8.87 (dd, *J =* 4.4 Hz, 1.9 Hz, 2H), 8.10 (dd, *J =* 4.4 Hz, 1.9 Hz, 2H), 7.87 (s, 3H), 7.34 (t, *J =* 8.5 Hz, 1H), 6.94 (s, 1H), 6.79 (d, *J* = 8.5 Hz, 1H), 6.78 (d, *J* = 8.5 Hz, 1H), 4.13 (t, *J* = 6.0 Hz, 2H), 3.82 (s, 3H), 3.05 – 2.94 (m, 2H), 2.07 – 1.98 (m, 2H); 13C NMR (125 MHz, DMSO-*d6*) δ 162.4, 158.6 (q, *J* = 35.3 Hz), 157.7, 156.6, 148.5, 146.0, 143.3, 134.6, 130.0, 122.6, 116.0 (q, *J* = 290.9 Hz), 107.5, 105.4, 104.5, 99.4, 65.3, 55.9, 36.4, 26.9; HRMS calculated for C₁₉H₂₂N₅O₃ (M+H⁺) 368.1717, found 368.1714; FTIR (cm⁻¹) 3407, 3239, 3081, 2970, 1611, 1563, 1542, 1496, 1465; Melting point: 184.2 – 186.3 °C.

See Supplementary Fig. 40 for the NMR spectra of compound 17.

8.3 Automated synthesis with the 5th step as the reductive amination (CRF #3)

8.3.1 Preparation and storage of reagents for CRF #3:

The 1^{st} , 2^{nd} , 3^{rd} , 4^{th} , and 6^{th} reagent solutions were the same as those in CRF #1.

Reagent solution for the 5th step (imine generation): To a 100 mL dry 3-port GL45 screw capped bottle were successively added 20 mL EtOH, 20 mL THF and furfural (3.0 equiv., 0.51 mL). The resulting solution was ready for use as a stock solution for automated synthesis (*connecting to valve 2*).

Reagent solution for the $5th$ step (imine reduction): To a 100 mL dry 3-port GL45 screw capped bottle were added 60 mL anhydrous THF and LiAlH4 (1.0 M in THF, 2 mL). The resulting solution was ready for use as a stock solution for automated synthesis (*connecting to valve 2*).

8.3.2 Establishing the CRF #3 ($5th$ step as the reductive amination reaction)

Supplementary Figure 18. CRF #3 (5th step as the reductive amination reaction).

Supplementary Table 7: The condition table in form of an Excel file for CRF #3.

Protocol for purification: The solution was concentrated under vacuum. Purification of the residue over silica gel chromatography (CH₂Cl₂: MeOH = 20 : 1) afforded a pale-yellow powder (400 mg, 43% yield).

¹H NMR (500 MHz, DMSO-*d*₆) δ 11.40 (brs, 1H), 8.15 (s, 3H), 7.53 (dd, *J* = 1.9, 0.9 Hz, 1H), 7.25 (t, *J* = 8.4 Hz, 1H), 6.72 (d, *J* = 8.4 Hz, 2H), 6.36 (dd, *J* = 3.2, 1.9 Hz, 1H), 6.26 (dd, *J* = 3.2, 0.9 Hz, 1H), 5.83 (s, 1H), 5.47 (brs, 1H), 4.24 (s, 2H), 4.08 (t, *J* = 6.1 Hz, 2H), 3.76 (s, 3H), 2.95 $(t, J = 7.3 \text{ Hz}, 2\text{H})$, $2.08 - 1.97 \text{ (m, 2H)}$; ¹³C NMR (125 MHz, DMSO- d_6) δ 158.4 (q, $J = 31.9 \text{ Hz}$), 157.8, 156.6, 155.4, 154.6, 141.7, 135.2, 129.4, 117.3 (q, *J* = 298 Hz), 110.5, 108.5, 106.5, 105.6, 104.7, 93.0, 65.5, 55.9, 41.2, 36.5, 26.9; HRMS calculated for $C_{18}H_{23}N_4O_3^+$ (M+H⁺) 343.1765, found 343.1773; FT-IR (cm⁻¹): 3420, 1683, 1507, 1470, 1439, 1207, 1141, 1103, 802, 725; Melting point: 119.8 – 120.9 °C.

See supplementary Fig. 41 for the NMR spectra of compound 18.

8.4 Automated synthesis (5th step as *N*-triflation)

8.4.1 Preparation and storage of reagents for CRF #4 (5th step as *N*-triflation):

The $1st$, $2nd$, $3rd$, $4th$, and $6th$ reagent solutions were the same as those in CRF #1.

Reagent solution-1 for the 5th step: To a 100 mL dry 3-port GL45 screw capped bottle were added CH_2Cl_2 (80 mL) and pyridine (2.0 equiv., 0.33 mL). The resulting solution was ready for use as a stock solution for automated synthesis (*connecting to valve 1*).

Reagent solution-2 for the 5th step: To a 100 mL dry 3-port GL45 screw capped bottle were added CH_2Cl_2 (80 mL) and Tf₂O (2.0 equiv., 0.69 mL). The resulting solution was ready for use as a stock solution for automation synthesis (*connecting to valve 2*).

8.4.2 Establishing the CRF #4 (5th step as the *N*-triflation reaction)

Supplementary Figure 19. CRF #4 (5th step as the *N***-triflation reaction). Tf₂O =** trifluoromethanesulfonic anhydride.

45% yield, 466 mg, 22 hrs

Protocol for purification: The collection was concentrated under vacuum. Purified over silica gel chromatography (CH₂Cl₂: MeOH = 8 : 1) afforded a pale-yellow amorphous foam (466 mg, 45%) yield).

1 H NMR (500 MHz, DMSO-*d*6) δ 8.05 (s, 3H), 7.34 (t, *J* = 8.4 Hz, 1H), 6.71 (d, *J* = 8.4 Hz, 1H), 6.70 (d, *J* = 8.4 Hz, 1H), 6.48 (s, 2H), 5.48 (s, 1H), 4.01 (t, *J* = 6.1 Hz, 1H), 3.69 (s, 3H), 2.85 (t, *J* = 7.3 Hz, 2H), 1.92 – 1.83 (m, 2H); ¹³ C NMR (125 MHz, DMSO-*d*6) 158.4, 158.2 (q, *J* = 32 Hz), 157.2, 155.2, 152.9, 131.1, 119.2 (q, *J* = 320 Hz), 117.4 (q, *J* = 300 Hz), 109.9, 105.4, 104.6, 91.7, 65.4, 56.0, 36.2, 26.9; 19F NMR (377 MHz, DMSO-*d*6) δ -73.70, -73.81; HRMS calculated for C₁₄H₁₈F₃N₄O₄S⁺ (M+H⁺) 395.0995, found 395.0988; FT-IR (cm⁻¹): 3446, 2948, 2846, 1676, 1603, 1488, 1475, 1436, 1411, 1235, 1207, 1142, 1098, 1030, 803, 723, 630.

See Supplementary Fig. 42 and Fig. 43 for the NMR spectra of compound 19.

8.5 Automated early-stage diversification $(2nd$ step as the Mitsunobu reaction)

8.5.1 Preparation and storage of reagents for CRF $#5$ (2nd step as the Mitsunobu reaction)

The preparation of the 3^{rd} , 4^{th} , 5^{th} , and 6^{th} reagent solutions was the same as those in CRF #1.

Reagent solution-1 for the 2nd step: To a 100 mL dry 3-port GL45 screw capped bottle were added 1,1'-(azodicarbonyl) dipiperidine (*ADDP*, 2.5 equiv., 1285 mg) and 75 mL THF. The resulting solution was ready for use as a stock solution for automated synthesis (*connecting to valve 1*).

Reagent solution-2 for 2nd step: To a 100 mL dry 3-port GL45 screw capped bottle were added methyl *2*-hydroxy-*6*-methoxybenzoate (2.5 equiv., 928 mg), *n*-Bu3P (2.5 equiv., 1.27 mL) and 75 mL THF. The resulting solution was ready for automated synthesis (*connecting to valve 2*).

8.5.2 Establishing the CRF #5 ($2nd$ step as the Mitsunobu reaction)

Supplementary Figure 20. CRF #5 ($2nd$ **step as the Mitsunobu reaction).** ADDP = 1,1-(azodicarbonyl)dipiperidine.

Supplementary Table 9: The condition table in form of an Excel file for CRF #5.

Protocol for purification: The collection was concentrated under vacuum. The residue was purified over silica gel chromatography $(CH_2Cl_2$: MeOH = 8:1 to 3:1) followed by recrystallization with EtOAc/Et₂O, furnishing a pale-yellow powder $(457 \text{ mg}, 50\% \text{ yield})$.

¹H NMR (500 MHz, DMSO-*d*₆) δ 10.78 (s, 1H), 8.65 (d, *J* = 1.3 Hz, 1H), 8.50 (s, 1H), 8.11 (s, 3H), 7.70 (d, *J* = 8.7 Hz, 2H), 7.08 (d, *J* = 8.8 Hz, 2H), 6.85 (s, 1H), 4.76 – 4.67 (m, 1H), 3.16 (dt, *J* = 14.4, 4.5 Hz, 1H), 3.11 – 2.98 (m, 1H), 1.29 (d, *J* = 6.2 Hz, 3H); 13C NMR (125 MHz, DMSO*d*6) δ 156.8, 152.0, 147.8, 147.5, 142.5, 135.4, 126.6, 122.8, 117.7, 116.6, 116.4, 93.1, 70.8, 43.3, 16.9; HRMS calculated for C₁₇H₁₈N₇O⁺ (M+H⁺) 336.1567, found 336.1572; FT-IR (cm⁻¹): 3474, 3135, 2982, 2236, 1680, 1607, 1535, 1491, 1377, 1247, 1181, 1134, 1022, 833, 798, 722; Melting point: 142.1 – 144.5 °C (decomposed); $[\alpha]_{25}^D = +8$ (c = 0.1, MeOH).

See Supplementary Fig. 44 for the NMR spectra of compound 20.

Compound **21**

Protocol for purification: The collection was concentrated under vacuum. The residue was dissolved in MeOH (5 mL), followed by addition of EtOAc (10 mL) slowly. Overnight crystallization with EtOAc washing (20 mL) and vacuum drying at room temperature furnished a yellow powder (566 mg, 58% yield).

¹H NMR (400 MHz, DMSO-*d*₆) δ 12.28 (s, 1H), 10.71 (s, 1H), 8.65 (d, *J* = 1.4 Hz, 1H), 8.55 (s, 1H), 8.13 (s, 3H), 7.35 (t, *J* = 8.4 Hz, 1H), 6.89 (s, 1H), 6.85 (d, *J* = 8.4 Hz, 1H), 6.83 (d, *J* = 8.4 Hz, 1H), 4.80 – 4.67 (m, 1H), 3.82 (s, 3H), 3.15 (dd, *J* = 13.3, 3.7 Hz, 1H), 3.06 (dd, *J* = 13.3, 7.9 Hz, 1H), 1.16 (d, *J* = 6.1 Hz, 3H); 13C NMR (125 MHz, DMSO-*d*6) δ 158.3 (q, *J* = 31.0 Hz), 157.7, 154.7, 152.1, 148.0, 146.3, 135.2, 134.4, 129.9, 117.7, 117.5 (q, *J* = 299 Hz), 116.1, 109.1, 107.9, 105.3, 98.8, 71.9, 55.9, 43.4, 16.8; HRMS calculated for $C_{18}H_{20}N_7O_2^+$ (M+H⁺) 366.1673, found 366.1672; FT-IR (cm-1): 3395, 3175, 2981, 2224, 1684, 1617, 1569, 1542, 1506, 1475, 1206, 1182, 1131, 1092, 1023, 799, 722; Melting point: $165.2 - 166.5$ °C; $[\alpha]_{25}^D = +10$ (c = 0.1, MeOH).

See Supplementary Fig. 45 for the NMR spectra of compound 21.

Compound **22**

Protocol for purification: The collection was concentrated under vacuum. The residue was purified over silica gel chromatography (CH_2Cl_2 : MeOH = 10:1 to 2:1) followed by recrystallization with EtOAc/Et₂O, furnishing a pale-yellow powder $(165 \text{ mg}, 18\% \text{ yield})$.

¹H NMR (500 MHz, DMSO-*d*₆) δ 10.96 (s, 1H), 8.65 (s, 1H), 8.57 (s, 1H), 8.42 (s, 1H), 8.33 (s, 3H), 7.88 (s, 1H), 7.03 (s, 1H), 4.94 – 4.81 (m, 1H), 3.32 – 3.19 (m, 1H), 3.18 – 2.98 (m, 1H), 1.36 (d, *J* = 6.3 Hz, 3H); 13C NMR (126 MHz, DMSO-*d*6) δ 159.4 (q, *J* = 32.5 Hz), 154.0, 152.0, 147.8, 147.0, 140.2, 138.1, 137.7, 135.7, 127.3, 119.7, 117.8, 116.9, 116.8 (q, *J* = 293 Hz), 94.6, 72.0, 43.5, 17.0; HRMS calculated for $C_{16}H_{17}N_8O^+$ (M+H⁺) 337.1520, found 337.1527; FT-IR (cm⁻¹): 3396, 3190, 2956, 2849, 2227, 1682, 1646, 1539, 1491, 1418, 1385, 1285, 1192, 1017, 800, 748, 722; Melting point: 230.2 – 233.1 °C (decomposed); $[\alpha]_{25}^D = +14$ (c = 0.1, MeOH)

See Supplementary Fig. 46 for the NMR spectra of compound 22.

Compound **23**

Protocol for purification: The collection was concentrated under vacuum. The residue was purified over silica gel chromatography (CH₂Cl₂: MeOH = 6 : 1) followed by recrystallization with EtOAc/Et₂O to afford a yellow powder (285 mg, 30% yield).

1 H NMR (500 MHz, DMSO-*d*6) δ 10.82 (s, 1H), 8.68 (d, *J* = 1.4 Hz, 1H), 8.51 (s, 1H), 7.88 (d, *J* = 2.4 Hz, 1H), 7.52 (dd, *J* = 8.9, 2.4 Hz, 1H), 7.23 (d, *J* = 8.9 Hz, 1H), 7.08 (s, 1H), 4.90 – 4.81 (m, 1H), 3.28 – 3.15 (m, 2H), 1.24 (d, *J* = 6.1 Hz, 3H); 13C NMR (125 MHz, DMSO-*d*6) δ 158.4 (q, *J* = 32.0 Hz), 153.9, 152.0, 147.8, 144.4 (d, *J* = 10.0 Hz), 141.4, 135.4, 124.0, 121.6 (d, *J* = 3.6 Hz), 118.3, 117.6, 117.5 (q, *J* = 300.0 Hz), 116.5, 113.3, 113.1, 93.7, 73.1, 43.4, 17.0; HRMS calculated for C₁₇H₁₇FN₇O⁺ (M+H⁺) 354.1473, found 354.1476; FT-IR (cm⁻¹): 3172, 3121, 3055, 2984, 2231, 1679, 1624, 1538, 1515, 1275, 1204, 1135, 1016, 800, 723; Melting point: 202.1 – 204.9 °C (decomposed); $[\alpha]_{25}^D = +8$ (c = 0.1, MeOH).

See supplementary Fig. 47 for the NMR spectra of compound 23.

Compound **24**

Protocol for purification: The collection was concentrated under vacuum. The residue was purified over silica gel chromatography CH_2Cl_2 : MeOH = 8:1 to 3:1), followed by recrystallization with EtOAc/Et₂O, furnishing a pale-yellow powder $(274 \text{ mg}, 30\% \text{ yield})$.

¹H NMR (500 MHz, DMSO-*d*₆) δ 10.80 (s, 1H), 8.64 (s, 1H), 8.55 (s, 1H), 7.68 (d, *J* = 7.7 Hz, 1H), 7.35 (t, *J* = 7.8 Hz, 1H), 7.24 (d, *J* = 8.4 Hz, 1H), 7.07 (t, *J* = 7.5 Hz, 2H), 6.99 (s, 1H), 4.90 (s, 1H), 3.32 – 3.18 (m, 2H), 1.23 (d, *J* = 6.1 Hz, 3H); 13C NMR (125 MHz, DMSO-*d*6) δ 159.0 (q, *J* = 32.5 Hz), 153.1, 152.0, 147.8, 146.9, 139.3, 135.4, 129.5, 128.4, 121.8, 119.7, 117.6, 117.2 (q, $J = 294$ Hz), 116.4, 115.3, 96.3, 71.7, 43.3, 16.7; HRMS calculated for C₁₇H₁₈N₇O⁺ (M+H⁺) 336.1567, found 336.1565; FT-IR (cm-1): 3421, 2961, 2928, 2230, 1683, 1617, 1539, 1499, 1386, 1207, 1140, 1019, 801, 724; Melting point: 214.6 – 217.0 °C; [α]^D₂₅= +22 (c = 0.1, MeOH).

See Supplementary Fig. 48 for the NMR spectra of compound 24.

Compound **25**

Protocol for purification: The collection was concentrated under vacuum. The residue was purified over silica gel chromatography (CH_2Cl_2 : MeOH = 10 : 1) followed by recrystallization with EtOAc/Et₂O to afford a yellow powder $(510 \text{ mg}, 55\% \text{ yield})$.

1 H NMR (500 MHz, Methanol-*d*4) δ 8.55 (d, *J* = 1.4 Hz, 1H), 8.42 (s, 1H), 7.42 (d, *J* = 5.6 Hz, 1H), 7.07 (d, *J* = 5.6 Hz, 1H), 6.66 (s, 1H), 4.72 – 4.63 (m, 1H), 3.27 – 3.22 (m, 2H), 1.35 (d, *J* =

6.2 Hz, 3H); 13C NMR (125 MHz, Methanol-*d*4) δ 163.1 (q, *J* = 34.8 Hz), 153.4, 152.9, 148.5, 146.6, 139.7, 136.5, 125.2, 120.3, 119.1, 118.1 (q, *J* = 293 Hz), 118.0, 114.8, 94.7, 76.9, 45.7, 17.8; HRMS calculated for $C_{15}H_{16}N_7OS^+(M+H^+)$ 342.1132, found 342.1141; FT-IR (cm⁻¹): 3368, 3344, 3112, 3085, 2220, 1682, 1605, 1535, 1489, 1240, 1197, 1139, 1022, 775, 724; Melting point: 185.9 $-188.3 \text{ °C}; [\alpha]_{25} = +59 \text{ (c = 0.1, MeOH)}.$

See Supplementary Fig. 49 for the NMR spectra of compound 25.

Compound **26**

63% yield, 615 mg, 36 hrs

Protocol for purification: The collection was concentrated under vacuum. The residue was purified over silica gel chromatography $(CH_2Cl_2$: MeOH = 4 : 1) followed by recrystallization with MeOH/EtOAc to afford a yellow powder (615 mg, 63% yield).

¹H NMR (500 MHz, DMSO-*d*₆) δ 12.96 (s, 1H), 10.80 (s, 1H), 8.66 (d, *J* = 1.4 Hz, 1H), 8.49 (s, 1H), 7.97 (s, 3H), 7.42 (d, *J* = 2.2 Hz, 1H), 7.29 (dd, *J* = 8.3, 2.2 Hz, 1H), 7.14 (d, *J* = 8.3 Hz, 1H), 6.91 (s, 1H), 4.60 – 4.52 (m, 1H), 3.87 (s, 3H), 3.13 (dd, *J* = 13.3, 3.6 Hz, 1H), 3.05 (dd, *J* = 13.3, 8.1 Hz, 1H), 1.23 (d, *J* = 6.2 Hz, 3H); 13C NMR (125 MHz, DMSO-*d*6) δ 152.0, 150.9, 147.8, 145.7, 142.2, 135.4, 124.1, 118.6, 118.4, 117.7, 117.6, 116.4, 109.5, 93.4, 72.7, 55.8, 43.4, 17.0; HRMS calculated for $C_{18}H_{20}N_7O_2^+$ (M+H⁺) 366.1673, found 366.1674; FT-IR (cm⁻¹): 3094, 3061, 2983, 2944, 2227, 1680, 1610, 1594, 1536, 1507, 1398, 1260, 1198, 1130, 1053, 1024, 800, 722; Melting point: 198.9 – 201.2 °C (decomposed); $[\alpha]_{25} = +14$ (c = 0.1, MeOH).

See Supplementary Fig. 50 for the NMR spectra of compound 26.

Protocol for purification: The collection was concentrated under vacuum. The residue was dissolved in MeOH (1.5 mL), after that, EtOAc (2 mL) was slowly added to the solution. The compound was crystallized as a yellow powder (140 mg, 13% yield) overnight.

¹H NMR (500 MHz, DMSO-*d*₆) δ 12.99 (s, 1H), 10.82 (s, 1H), 8.65 (d, *J* = 1.4 Hz, 1H), 8.48 (s, 1 H), 8.03 (brs, 3H), 8.00 (d, *J* = 2.2 Hz, 1H), 7.74 (dd, *J* = 8.6, 2.2 Hz, 1H), 7.19 (d, *J* = 8.6 Hz, 1H), 6.90 (s, 1H), 4.20 (t, *J* = 6.0 Hz, 2H), 3.03 (t, *J* = 7.4 Hz, 2H), 2.14 – 1.99 (m, 2H); 13C NMR (125 MHz, DMSO-*d*6) δ 158.4 (q, *J* = 31 Hz), 154.3, 152.0, 147.8, 140.9, 140.8, 135.4, 129.3, 125.8, 123.6, 117.6, 117.2 (q, *J* = 300 Hz), 116.4, 114.1, 111.6, 93.5, 65.8, 36.2, 26.8; HRMS calculated for C₁₇H₁₇BrN₇O⁺ (M+H⁺) 414.0673, found 414.0668; FT-IR (cm⁻¹): 3392, 3184, 3115, 2963, 2228, 1678, 1616, 1538, 1494, 1394, 1288, 1253, 1203, 1136, 1015, 800, 723; Melting point: $225.1 - 226.4$ °C (decomposed).

See Supplementary Fig. 51 for the NMR spectra of compound 27.

Compound **28**

Protocol for purification: The collection was concentrated under vacuum. The residue was purified over silica gel chromatography $(CH_2Cl_2$: MeOH = 10:1) followed by recrystallization with EtOAc/Et₂O to afford a yellow powder (462 mg, 43% yield).

1 H NMR (500 MHz, DMSO-*d*6) δ 8.68 (d, *J* = 1.4 Hz, 1H), 8.51 (s, 1H), 7.88 (d, *J* = 2.5 Hz, 1H), 7.52 (dd, *J* = 9.0, 2.5 Hz, 1H), 7.23 (d, *J* = 9.0 Hz, 1H), 7.08 (s, 1H), 4.91 – 4.81 (m, 1H), 3.28 – 3.18 (m, 2H), 1.24 (d, *J* = 6.2 Hz, 3H); 13C NMR (125 MHz, DMSO-*d*6) δ 158.9 (appear s, 1C), 152.9, 152.5, 148.3, 147.2, 138.7, 136.0, 132.2, 130.6, 122.6, 118.1, 118.0, 117.2 (q, *J* = 300 Hz, 1C), 116.9, 113.7, 97.6, 72.8, 43.8, 17.1; HRMS calculated for $C_{17}H_{17}BrN_7O^+ (M+H^+)$ 414.0673, found 414.0668; FT-IR (cm⁻¹): 3391, 3172, 3118, 2986, 2229, 1680, 1611, 1542, 1499, 1475, 1385, 1237, 1204, 1139, 1015, 802, 723; Melting point: $225.1 - 226.4 \degree C$ (decomposed); $[\alpha]_{25}^D = +13$ (c $= 0.1$, MeOH).

See Supplementary Fig. 52 for the NMR spectra of compound 28.

Compound **29**

Purification: The collection was concentrated under vacuum. The residue was dissolved in MeOH (3 mL), followed by addition of EtOAc (3 mL) slowly. Overnight crystallization with EtOAc washing (5 mL) and vacuum drying at room temperature for 60 min afforded a pale-yellow powder (232 mg, 25% yield).

¹H NMR (500 MHz, DMSO-*d*₆) δ 12.57 (s, 1H), 10.81 (s, 1H), 8.68 (d, *J* = 1.3 Hz, 1H), 8.46 (brs, 1H), 7.95 (brs, 3H), 7.52 (s, 1H), 7.11 (d, *J* = 5.5 Hz, 1H), 6.83 (s, 1H), 4.23 (t, *J* = 5.9 Hz, 2H), 3.05 (t, *J* = 7.4 Hz, 2H), 2.11 – 2.00 (m, 2H); 13C NMR (125 MHz, DMSO-*d*6) δ 158.4 (q, *J* = 31.3 Hz), 153.3, 151.9, 147.9, 147.6, 135.5, 135.2, 124.3, 117.8, 117.7, 117.2 (q, *J* = 300.8 Hz), 116.3, 108.0, 94.1, 68.0, 36.3, 27.2; HRMS calculated for $C_{15}H_{16}N_7OS^+$ (M+H⁺) 342.1132, found 342.1140. Melting point: 223.0 – 225.6 °C (decomposed).

See Supplementary Fig. 53 for the NMR spectra of compound 29.

8.6 Automated early-stage diversifications $(2nd$ step as the click reaction)

8.6.1 Preparation and storage of reagents for CRF $#6$ ($2nd$ step as the click reaction)

The 1st, 3rd, 4th, 5th, and 6th reagent solutions were the same as those in CRF #1.

Reagent solution-1 for the 2nd step: To a 100 mL dry 3-port GL45 screw capped bottle were added $[Cu(CH₃CN)₄]BF₄$ (0.2 equiv., 128 mg) and 30 mL CH₂Cl₂. The resulting solution was ready for use as a stock solution for automated synthesis (*connecting to valve 1*).

Reagent solution-2 for the 2nd step: To a 100 mL dry 3-port GL45 screw capped bottle was added methyl *4*-(azidomethyl) benzoate (2.0 equiv., 780 mg). The resulting solution was ready for use as a stock solution for automated synthesis (*connecting to valve 2*).

8.6.2 Establishing the CRF #6 ($2nd$ step as the click reaction)

Supplementary Figure 21. CRF #6 (2nd step as the Click reaction).

Supplementary Table 10: The condition table in form of an Excel file for CRF #6.

Protocol for purification: The collection was concentrated under vacuum. The residue was dissolved in CH_2Cl_2 (5 mL), followed by addition of EtOAc (5 mL) slowly. Overnight crystallization with EtOAc washing (5 mL) and vacuum drying at room temperature afforded a white solid (485 mg, 49% yield).

¹H NMR (500 MHz, DMSO-*d*₆) δ 13.10 (s, 1H), 10.85 (s, 1H), 8.66 (d, *J* = 1.4 Hz, 1H), 8.48 (s, 1H), 8.32 (s, 3H), 8.22 (s, 1H), 7.76 (d, *J* = 8.4 Hz, 2H), 7.43 (d, *J* = 8.4 Hz, 2H), 6.94 (s, 1H), 5.69 (s, 2H), 4.14 (s, 2H); 13C NMR (125 MHz, DMSO-*d*6) δ 158.2(q, *J* = 30 Hz), 151.9, 147.8, 141.9, 140.5, 135.7, 135.4, 129.3, 129.0, 128.8, 125.4, 124.5, 117.6, 116.5, 93.9, 52.6, 33.9; HRMS calculated for $C_{18}H_{17}N_{10}$ ⁺ (M+H⁺) 373.1632, found 373.1628; FT-IR (cm⁻¹): 3412, 3130, 3060, 2954, 2228, 1728, 1684, 1540, 1501, 1203, 1134, 801, 723; Melting point: 191.2 – 193.7 °C (decomposed).

See Supplementary Fig. 54 for the NMR spectra of compound 30.

Supplementary Figure 22. Schematic drawing for the cleaning protocol between each molecule synthesis implementation.

9. Analysis of the feasibility of synthesizing top-selling drugs using SPS-flow technology

Upon close analysis of the top 200 small-molecule pharmaceuticals (by retail sales in 2018), we envision that 73 out of 155 single small molecules (among the top 200 pharmaceuticals, 39 are mixtures of molecules and 6 are feedstock compounds) possess functional group handles suitable for attachment to solid supports, 28 molecules can be achieved by cyclization-cleavage or substitution-cleavage strategies, and 12 molecules can potentially be approached through SPS assisted by a traceless linker. Taken together, 73% of these single pharmaceutical molecules can potentially be synthesized by SPS:

Supplementary Table 11: SPS Feasibility Analysis for Top 200 Small Molecule Pharmaceuticals (by Retail Sales in 2018)

10. Bioactivity evaluation

The half maximal inhibitory concentration (IC_{50}) of prexasertib was first determined in a human lymphocytic leukemia REH cell line $(IC_{50} = 4.57 \text{ nM})$ and a T-lymphoblastic leukemia Molt-4 cell line (IC₅₀=0.89 nM). The anti-cancer activity of a series of prexasertib analogs was evaluated by CCK8 assay using concentrations of 7.5 nM and 1.5 nM in REH and Molt4, respectively. Preliminary evaluation suggests that prexasertib **7** remains the most potent compound among all the tested derivatives.

Supplementary Fig. 23. Cell viability of compounds at a) 7.5 nM on REH cell line and b) 1.5nM on Molt4 cell line after treatment for 48 hr. The experiment was conducted in n=3 biologically independent samples. Two independent experimental replicates were performed and showed similar results. Error bars, mean +/- s.d. Differences between NT vs compound treatment with *p*

value < 0.01 are indicated; ****, $p < 0.0001$. Two-tailed unpaired t-test statistical analysis was used.

11. NMR spectra of new compounds

Supplementary Figure 24. ¹ H NMR and 13C NMR spectra of compound **2'**.

Supplementary Figure 25. ¹ H NMR and 13C NMR spectra of compound **3'**.

Supplementary Figure 27. ¹ H NMR and 13C NMR spectra of compound **5'**.

Supplementary Figure 28. ¹ H NMR and 13C NMR spectra of compound **6**.

Supplementary Figure 29. ¹ H NMR and 13C NMR spectra of compound **7**.

Supplementary Figure 30. ¹ H NMR and 13C NMR spectra of compound **8**.

Supplementary Figure 32. ¹ H NMR and 13C NMR spectra of compound **9**.

Supplementary Figure 33. ¹ H NMR and 13C NMR spectra of compound **10**.

Supplementary Figure 34. ¹ H NMR and 13C NMR spectra of compound **11**.

Supplementary Figure 35. ¹ H NMR and 13C NMR spectra of compound **12**.

Supplementary Figure 36. ¹ H NMR and 13C NMR spectra of compound **13**.

Supplementary Figure 37. ¹ H NMR and 13C NMR spectra of compound **14**.

Supplementary Figure 38. ¹ H NMR and 13C NMR spectra of compound **15**.

Supplementary Figure 39. ¹ H NMR and 13C NMR spectra of compound **16**.

Supplementary Figure 40. ¹ H NMR and 13C NMR spectra of compound **17**.

Supplementary Figure 41. ¹ H NMR and 13C NMR spectra of compound **18**.

Supplementary Figure 42. ¹ H NMR and 13C NMR spectra of compound **19**.

Supplementary Figure 43. 19F NMR spectra of compound **19**.

Supplementary Figure 44. 1 H NMR and 13C NMR spectra of compound **20**.

Supplementary Figure 45. ¹ H NMR and 13C NMR spectra of compound **21**.

Supplementary Figure 46. ¹ H NMR and 13C NMR spectra of compound **22**.

Supplementary Figure 47. ¹ H NMR and 13C NMR spectra of compound **23**.

Supplementary Figure 48. ¹ H NMR and 13C NMR spectra of compound **24**.

 $\bar{\mathbf{r}}$ $rac{1}{30}$ $\frac{1}{20}$ $\frac{1}{10}$ **Supplementary Figure 49.** ¹ H NMR and 13C NMR spectra of compound **25**.

Supplementary Figure 50. ¹ H NMR and 13C NMR spectra of compound **26**.

Supplementary Figure 51. ¹ H NMR and 13C NMR spectra of compound **27**.

Supplementary Figure 52. ¹ H NMR and 13C NMR spectra of compound **28**.

Supplementary Figure 53. ¹ H NMR and 13C NMR spectra of compound **29**.

Supplementary Figure 54. ¹ H NMR and 13C NMR spectra of compound **30**.

12. Representative crude 1H NMR spectra of the cleaved derivatives

Notes: The rude ¹H NMR spectra of derivatives were acquired without any purification after cleaving the products from the solid resin in automated SPS-flow synthesis.

Supplementary Figure 55. Crude ¹H NMR and purified ¹H NMR spectra of compound 10 (5th step amide coupling)

Supplementary Figure 56. Crude ¹H NMR and purified ¹H NMR spectra of compound 12 (5th step amide coupling).

Supplementary Figure 57. Crude ¹H NMR and purified ¹H NMR spectra of compound 15 (5th step amide coupling).

Supplementary Figure 58. Crude ¹H NMR and purified ¹H NMR spectra of compound 16 (5th step amide coupling).

Supplementary Figure 59. Crude ¹H NMR and purified ¹H NMR spectra of compound 18 (5th step reductive amination).

Supplementary Figure 60. Crude ¹H NMR and purified ¹H NMR spectra of compound 19 (5th step *N*-triflation).

Supplementary Figure 61. Crude ¹H NMR and purified ¹H NMR spectra of compound 24 (2nd step Mitsunobu reaction).

Supplementary Figure 62. Crude ¹ H NMR and purified ¹ H NMR spectra of compound **25** (2nd step Mitsunobu reaction).

Supplementary Figure 63. Crude ¹ H NMR and purified ¹ H NMR spectra of compound **28** (2nd step Mitsunobu reaction).

Supplementary Figure 64. Crude ¹ H NMR and purified ¹ H NMR spectra of compound **30** (2nd step the Click reaction).