



## Supporting Information

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# Defect Induced Polarization Loss in Multi-shelled Spinel Hollow Spheres for Electromagnetic Wave Absorption Application

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## Supplementary information:

### **Defect induced polarization loss in multi-shelled spinel hollow spheres for electromagnetic wave absorption application**

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**Table S1** The products prepared by different bivalent metal ions

Bivalent metal ions	Mg	Mn	Fe	Ni	Cu	Zn
Products	Co <sub>3</sub> O <sub>4</sub>	Co <sub>3</sub> O <sub>4</sub>	FeCo <sub>2</sub> O <sub>4</sub> Co <sub>3</sub> O <sub>4</sub>	NiO NiCo <sub>2</sub> O <sub>4</sub>	CuO CuCo <sub>2</sub> O <sub>4</sub>	ZnO ZnCo <sub>2</sub> O <sub>4</sub>

**Table S2** The calculated results of the Co<sup>2+</sup> ratios and oxygen vacancies form XPS

peak areas

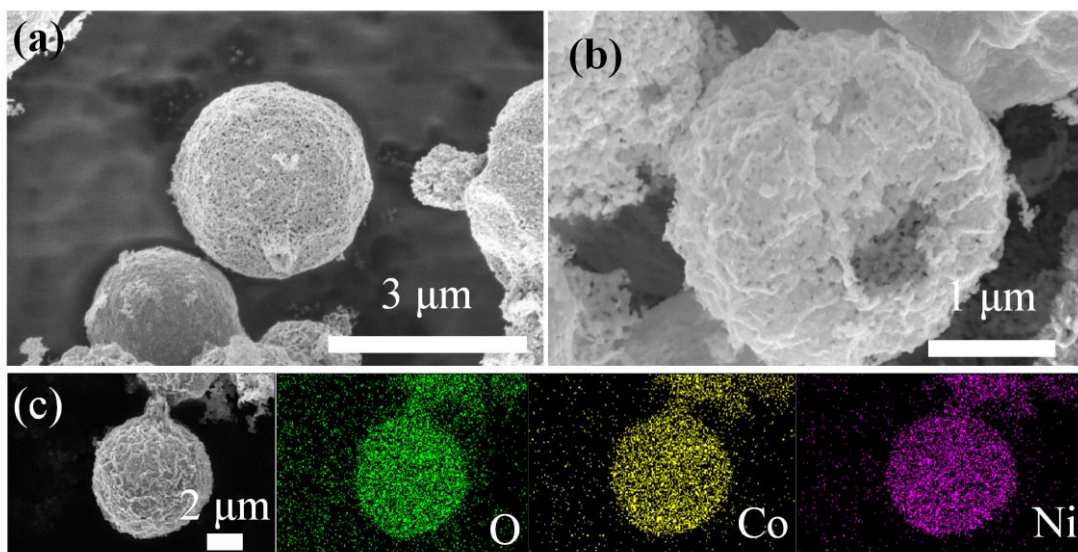
Samples	Co <sup>2+</sup> /%	Co <sup>3+</sup> /%	Co <sup>2+</sup> / Co <sup>3+</sup>	Ni <sup>2+</sup> /%	Ni <sup>3+</sup> /%	Oxygen vacancy/%
NCO-500	72.9	27.1	2.70	59.3	40.7	32.3
NCO-600	72.6	27.4	2.66	54.0	46.0	31.5
NCO-700	64.9	35.1	1.85	47.7	52.3	35.9

**Table S3** Work function of materials and their differences collected from reference

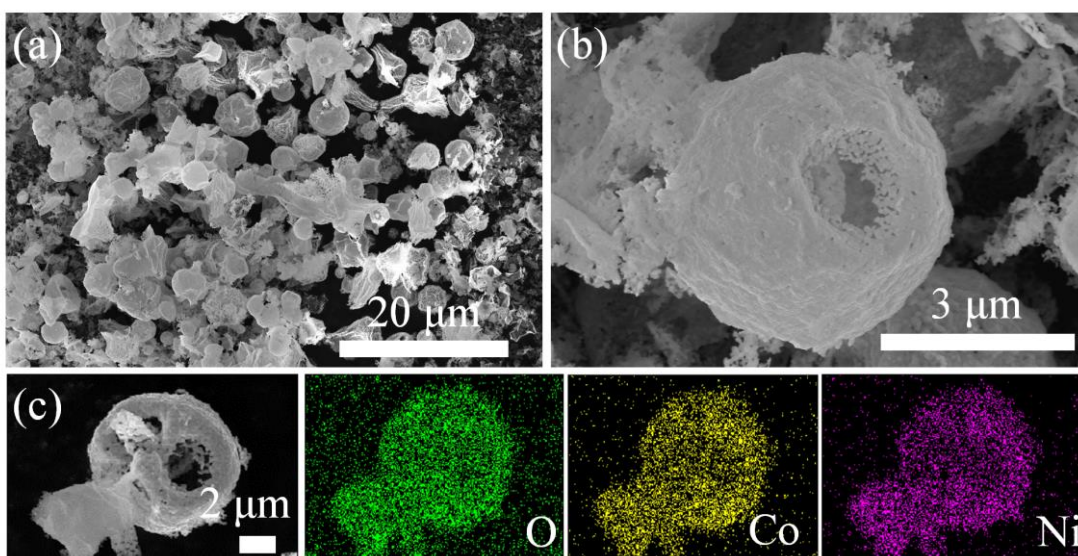
Materials	Work function (eV)	Work function difference (eV)	Refs.
NiCo <sub>2</sub> O <sub>4</sub>	6.1	-	[50]
NiO	5.4	-	
NiO/NiCo <sub>2</sub> O <sub>4</sub>	-	0.7	
CuCo <sub>2</sub> O <sub>4</sub>	4.87	-	[52]
CuO	5.31	-	[53]
CuO/CuCo <sub>2</sub> O <sub>4</sub>	-	0.44	
ZnO	4.45	-	[51]
ZnCo <sub>2</sub> O <sub>4</sub>	5.22	-	
ZnO/ZnCo <sub>2</sub> O <sub>4</sub>	-	0.77	

**Table S4** Comparison with our previously prepared NiCo<sub>2</sub>O<sub>4</sub>-based EM wave  
absorbing materials

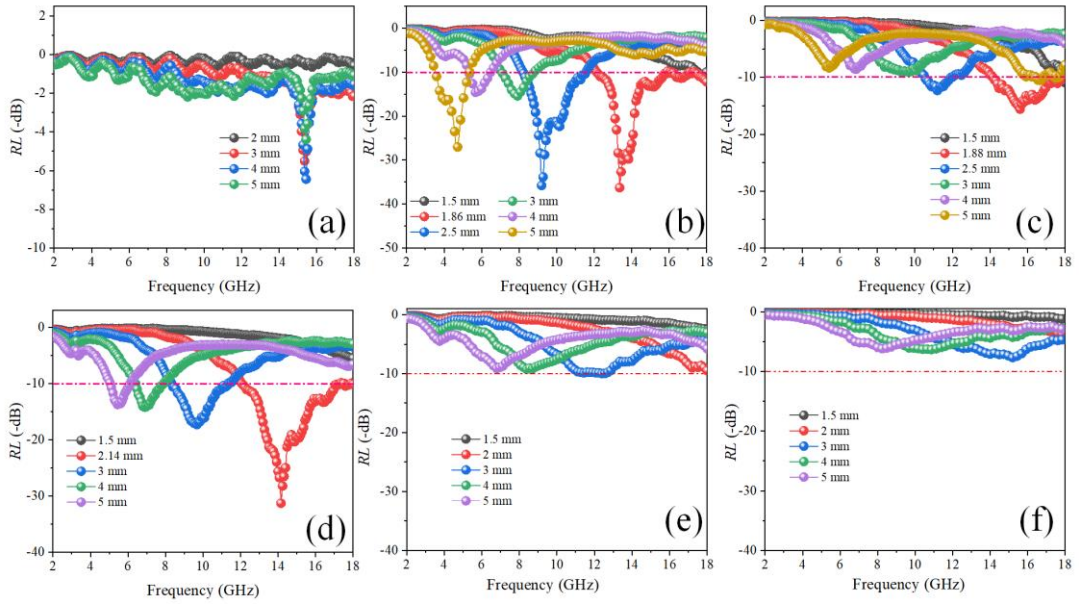
EM wave absorbers	Bandwidth (GHz)	Thickness (mm)	Density (mg/cm <sup>-3</sup> )	Refs.
NiCo <sub>2</sub> O <sub>4</sub> nanosheet	4.28	1.39	137	[12]
NCO-S	5.44	1.8	243	
NCO-P	4.64	1.6	106	[14]
NCO-C	4.96	1.9	120	
NiCo <sub>2</sub> O <sub>4</sub>	6.08	2.06	409	[18]
NiO/NiCo <sub>2</sub> O <sub>4</sub> microrod	6.08	1.88	294	[11]
NiCo <sub>2</sub> O <sub>4</sub>	7.44	2.1	478	[62]
<b>Multi-shelled</b>				
<b>NiO/NiCo<sub>2</sub>O<sub>4</sub> hollow spheres</b>	<b>5.84</b>	<b>1.86</b>	<b>41.1</b>	<b>This work</b>



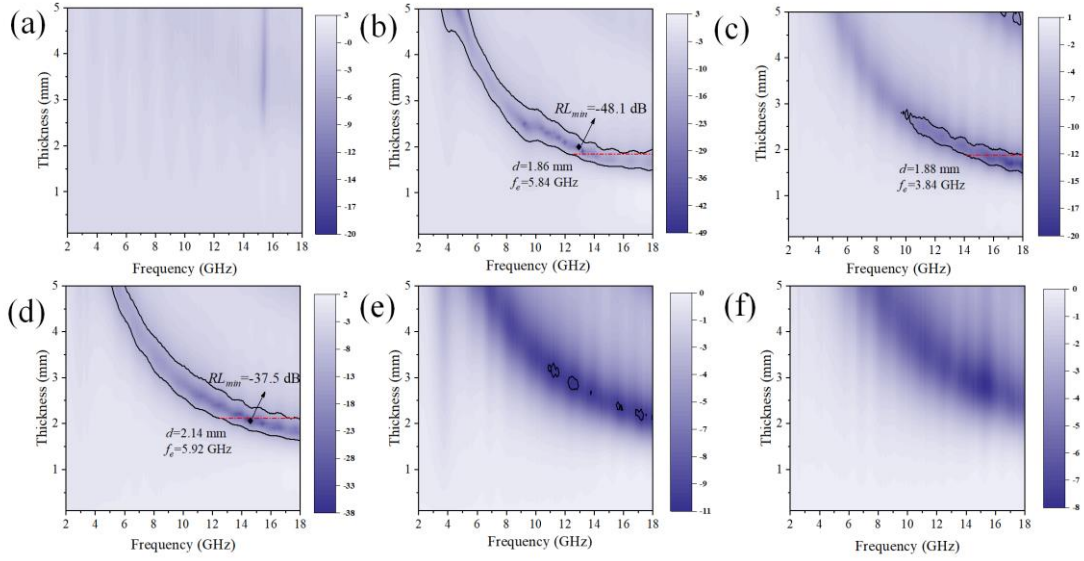
**Fig. S1** SEM images and EDS mapping of NCO-550.



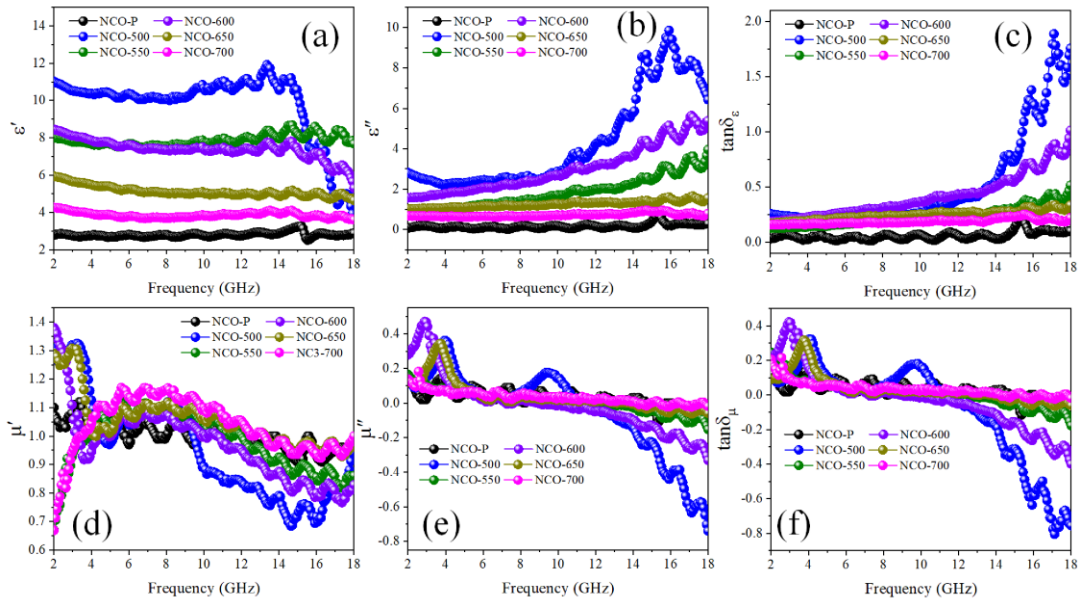
**Fig. S2** SEM images and EDS mapping of NCO-650.



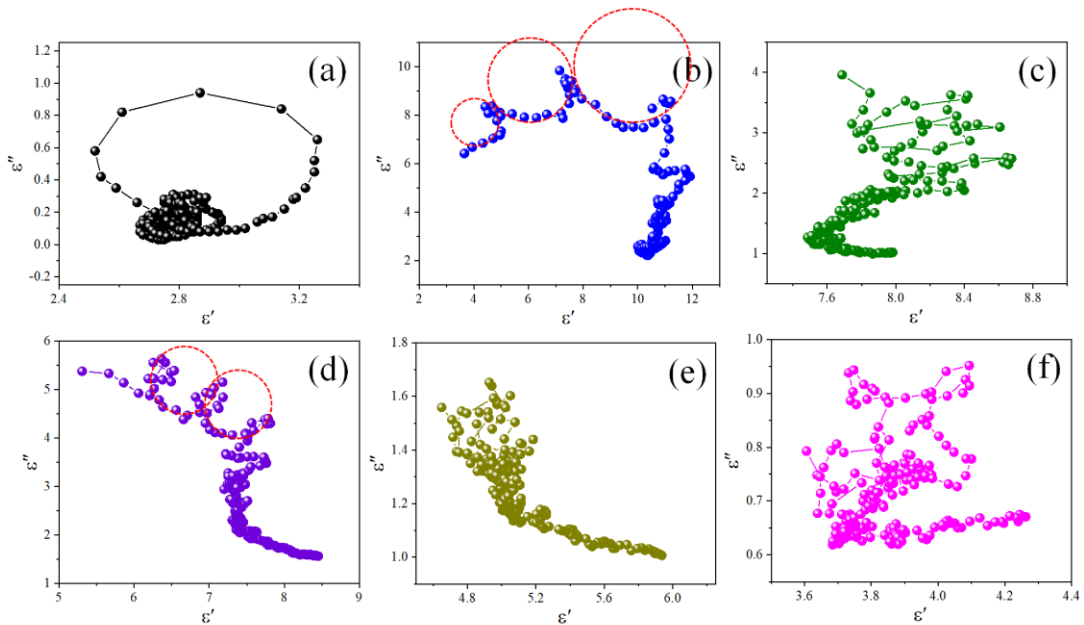
**Fig. S3**  $RL$  values of (a)  $\text{NiCo}_2\text{O}_4$  precursors, (b) NCO-500, (c) NCO-550, (d) NCO-600, (e) NCO-650 and (f) NCO-700 as a function of frequency.



**Fig. S4** 2D plot of (a)  $\text{NiCo}_2\text{O}_4$  precursors, (b) NCO-500, (c) NCO-550, (d) NCO-600, (e) NCO-650 and (f) NCO-700 as a function of frequency.

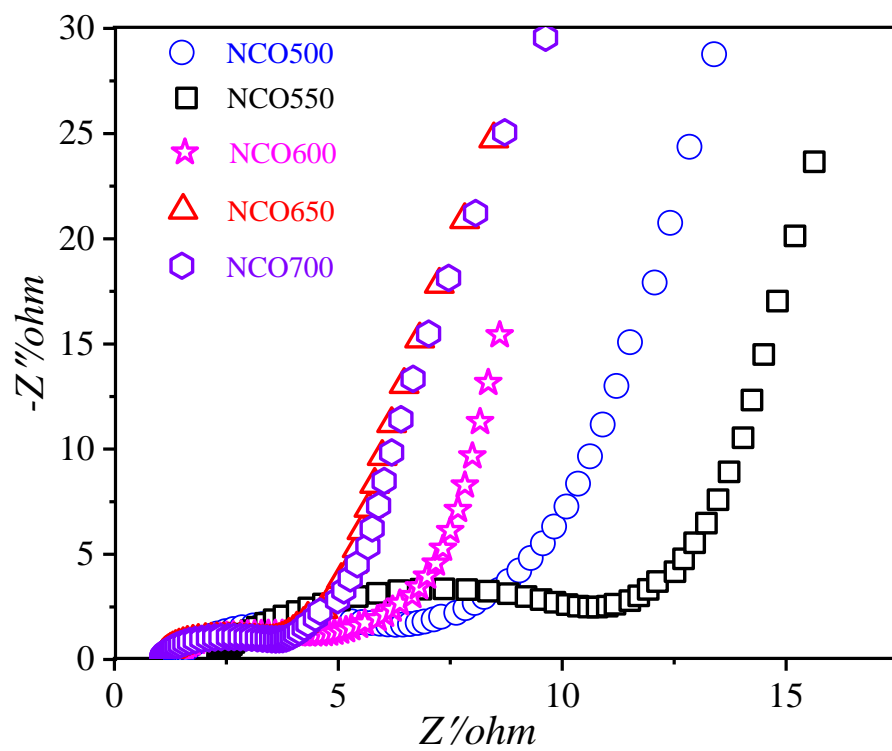


**Fig. S5** (a) The real part and (b) imaginary part of complex permittivity, (c) the dielectric loss tangent, the (d) real part, (e) imaginary part of complex permeability, (f) the magnetic loss tangent of the as-obtained NCO samples.

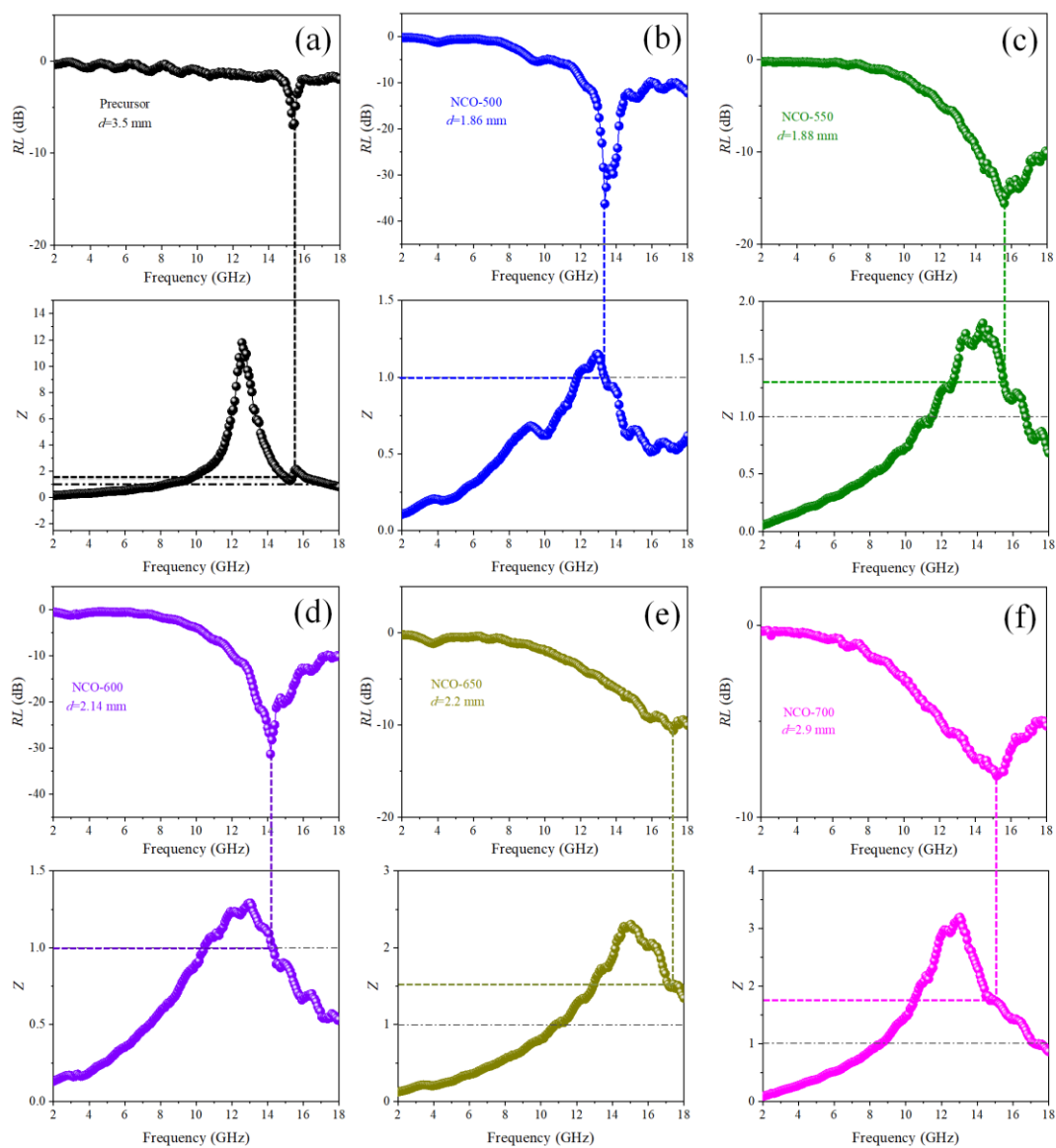


**Fig. S6** Cole-Cole semicircles of (a) NiCo<sub>2</sub>O<sub>4</sub> precursors, (b) NCO-500, (c) NCO-550, (d) NCO-600, (e) NCO-650 and (f) NCO-700.

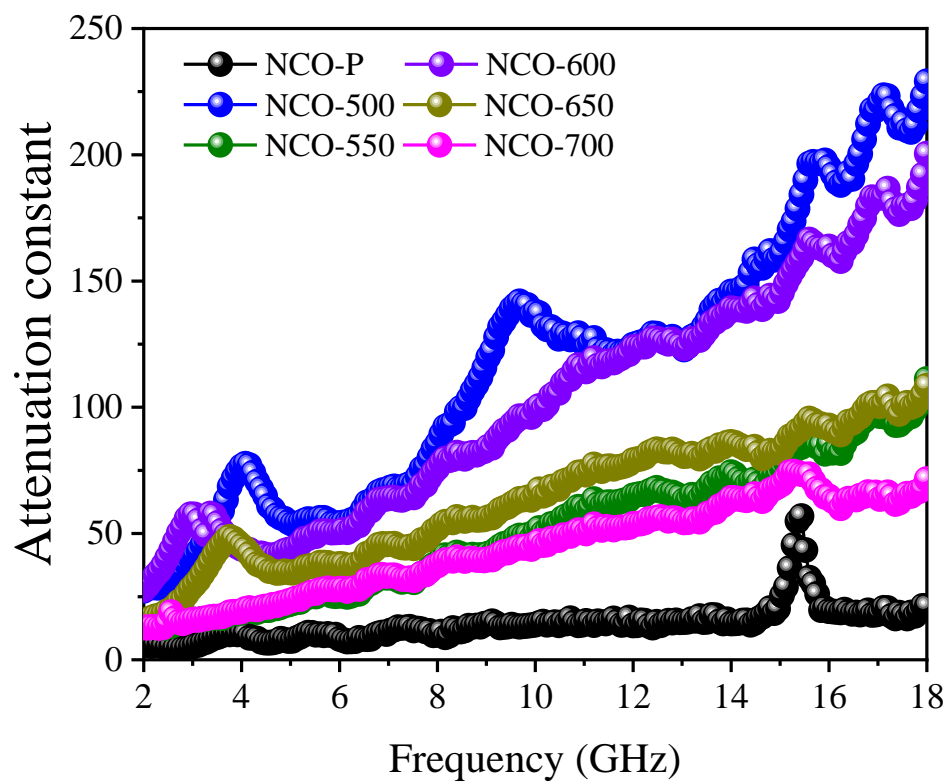




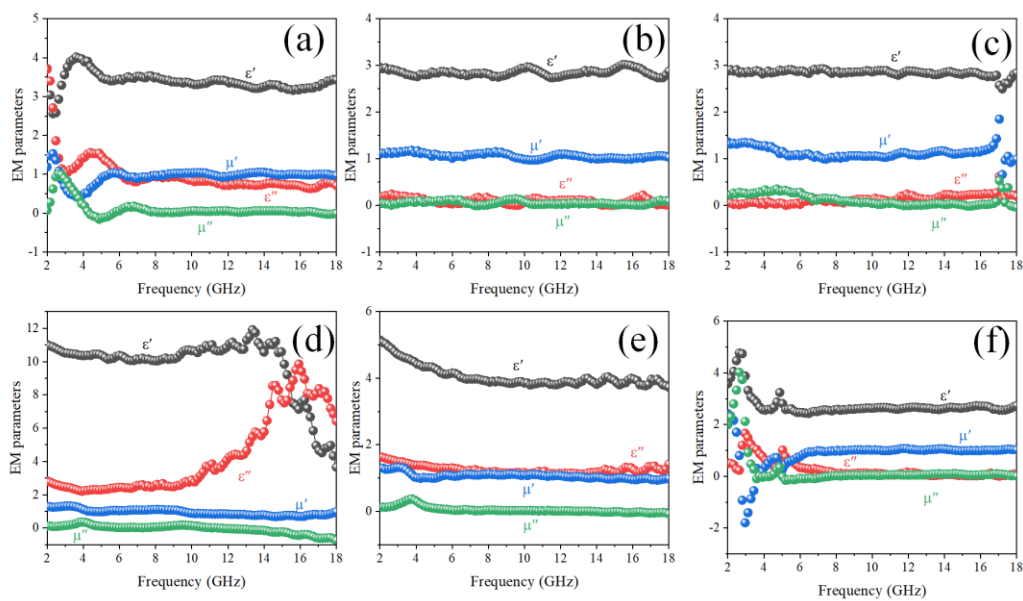
**Fig. S7** EIS curves of serial NCO samples.



**Fig. S8** Normalized impedance matching characteristic of NCO samples.

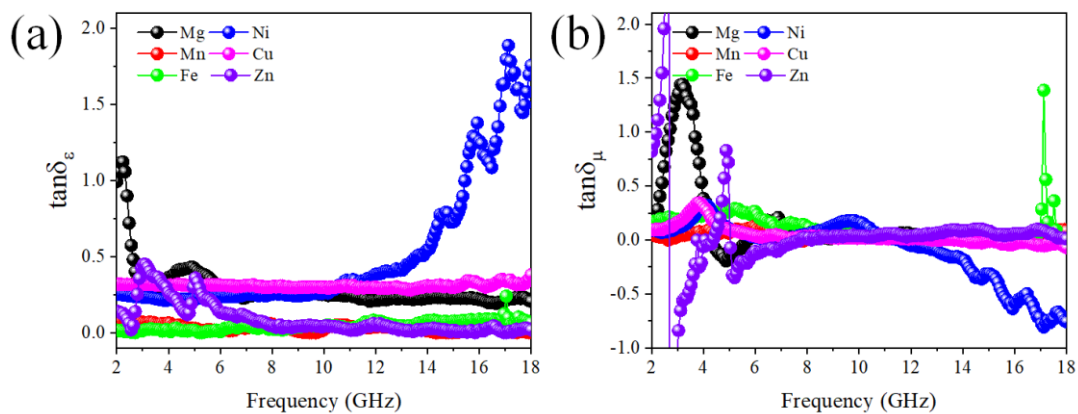


**Fig. S9** Attenuation constant of as-prepared NCO samples.



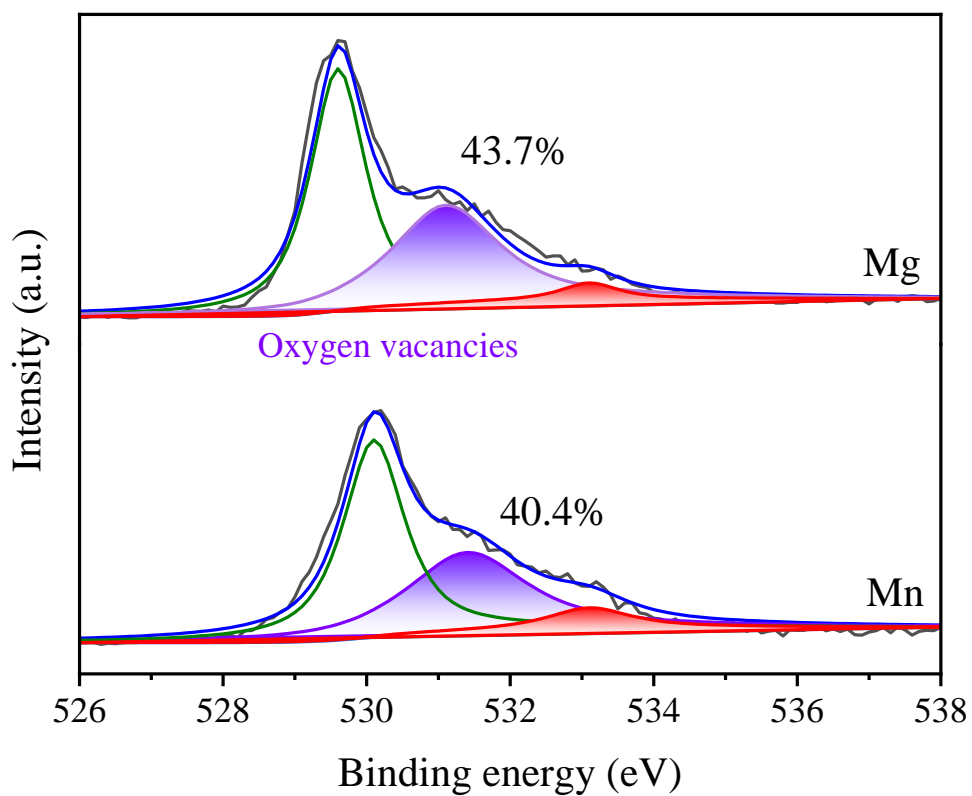
**Fig. S10** EM parameters of samples obtained from different divalent metal ions (a)

$\text{Mg}^{2+}$ , (b)  $\text{Mn}^{2+}$ , (c)  $\text{Fe}^{2+}$ , (d)  $\text{Ni}^{2+}$ , (e)  $\text{Cu}^{2+}$  and (f)  $\text{Zn}^{2+}$ .

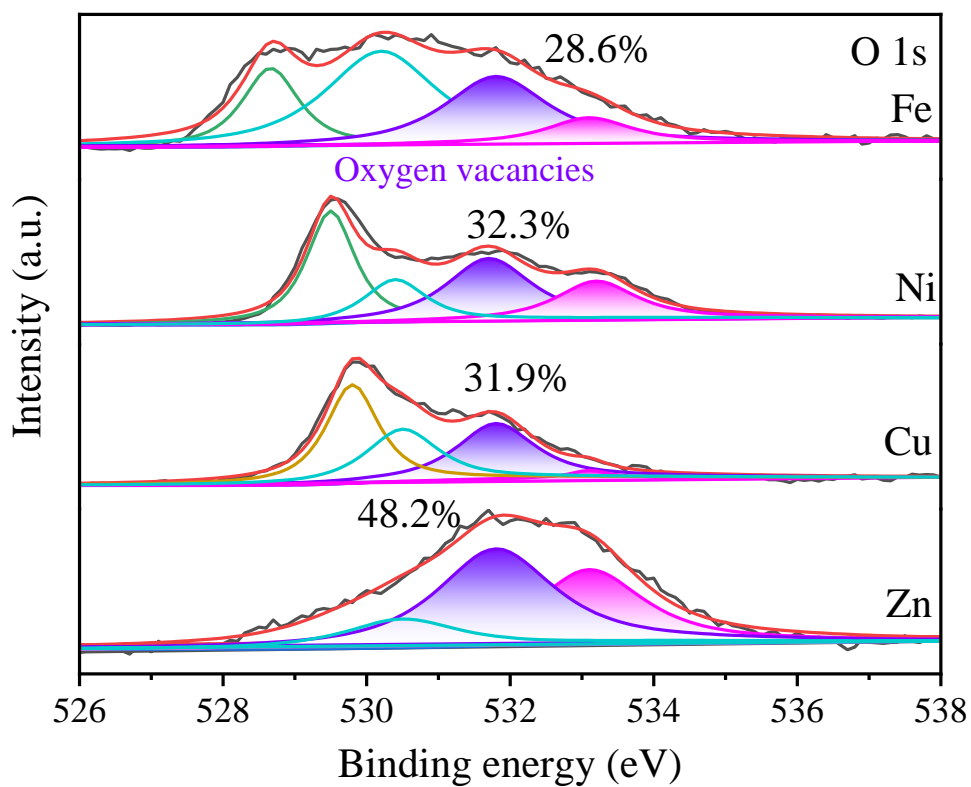


**Fig. S11** (a) the dielectric loss tangent and (b) the magnetic loss tangent of the

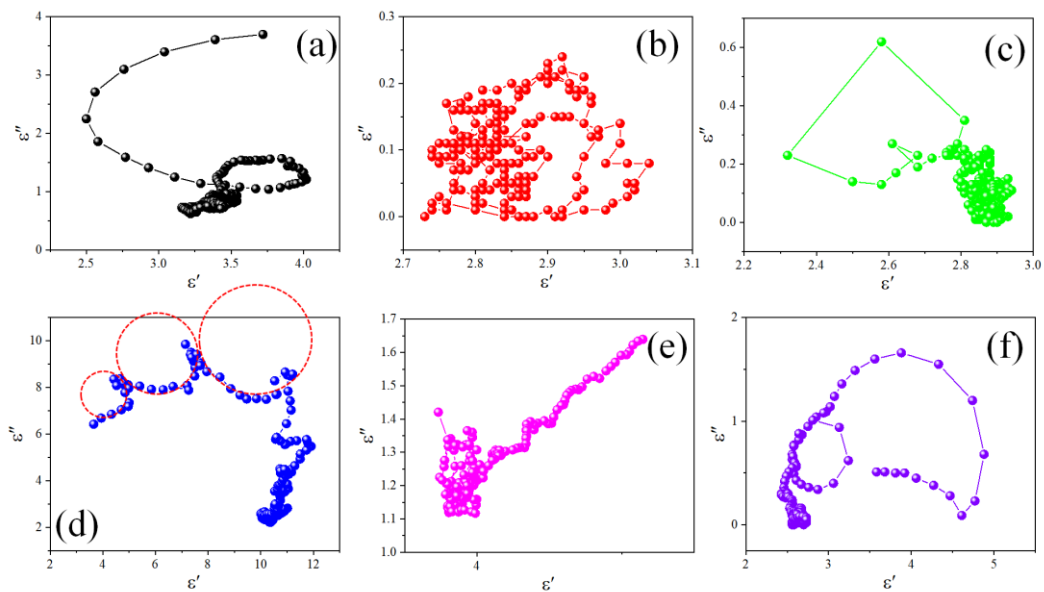
as-obtained samples.



**Fig. S12** High-resolution XPS spectrum survey of O 1s of as-prepared samples obtained from divalent metal ions (a)  $\text{Mg}^{2+}$  and (b)  $\text{Mn}^{2+}$ .



**Fig. S13** XPS spectrum survey of O 1s of as-prepared samples obtained from divalent metal ions  $\text{Fe}^{2+}$ ,  $\text{Ni}^{2+}$ , and  $\text{Cu}^{2+}$ .



**Fig. S14** Cole-Cole semicircles of samples obtained from different divalent metal ions

(a)  $\text{Mg}^{2+}$ , (b)  $\text{Mn}^{2+}$ , (c)  $\text{Fe}^{2+}$ , (d)  $\text{Ni}^{2+}$ , (e)  $\text{Cu}^{2+}$  and (f)  $\text{Zn}^{2+}$ .