

Supporting Information

Impact of polypyrrole functionalization on the anodic performance of boron nitride nanosheets: insights from first-principles calculations

Chidera C. Nnadikewe^a, Ismail Abdulazeez^a, Muhammad Haroon^a, Qing Peng^{b,c}, Almaz Jalilov^a,
Abdulaziz Al-Saadi^{a,*}

^aChemistry Department, King Fahd University of Petroleum and Minerals, Dhahran 31261, Saudi Arabia

^bPhysics Department, King Fahd University of Petroleum and Minerals, Dhahran 31261, Saudi Arabia

^cK.A CARE Energy Research & Innovations Center at Dhahran, Dhahran 31261, Saudi Arabia

* Corresponding author: Dr. Abdulaziz Al-Saadi (asaadi@kfupm.edu.sa)

Contents

- 1. Density of states of the metal atoms/ions adsorbed onto BNNS.**
- 2. Density of states of metal atoms/ions adsorbed onto PP-BNNS.**
- 3. HOMO-LUMO maps of the metal atoms/ions on BNNS.**
- 4. HOMO-LUMO maps of the metal atoms/ions on PP-BNNS.**
- 5. Optimized structural cross-sections of the metal atoms/ions onto BNNS.**
- 6. Adsorption energies of the metal atoms/ions on BNNS and PP-BNNS.**

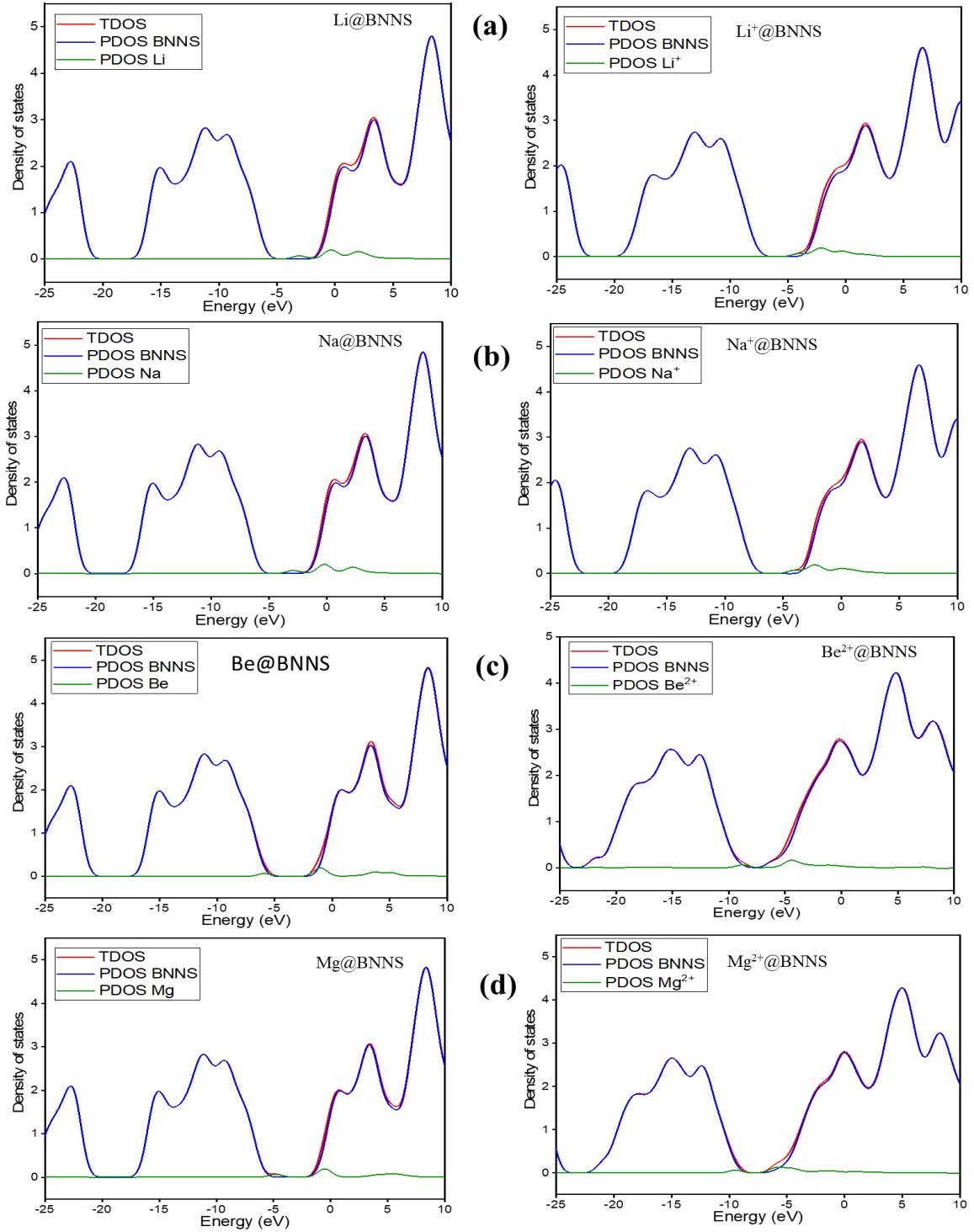


Figure S1: Density of States of the metal atoms and ions adsorbed onto the BNNS. It shows the total and partial state of the complexes and/or their components.

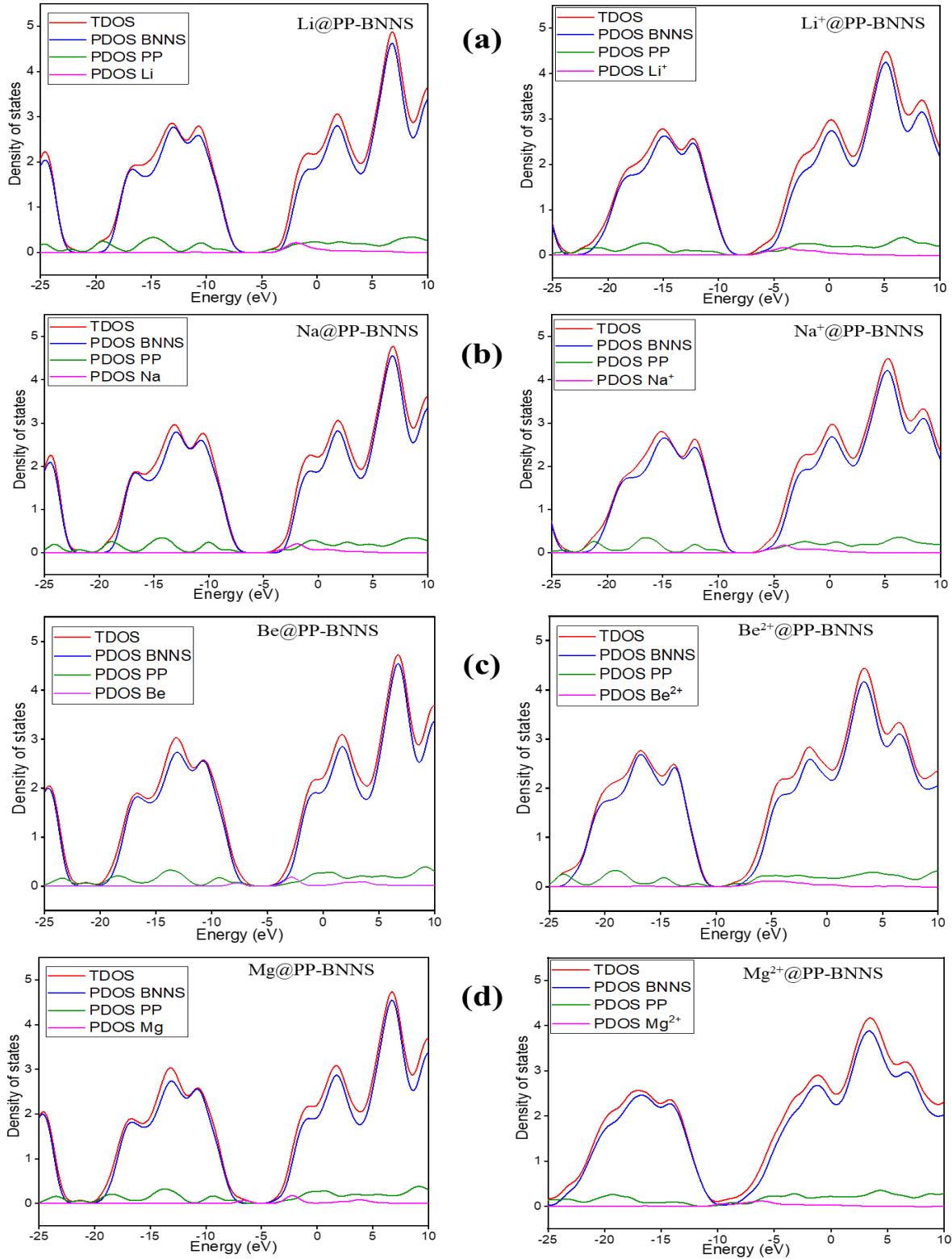


Figure S2: Density of States of the metal atoms and ions adsorbed onto the PP-BNNS. It shows the total and partial densities of the state of the complexes and/or their components.

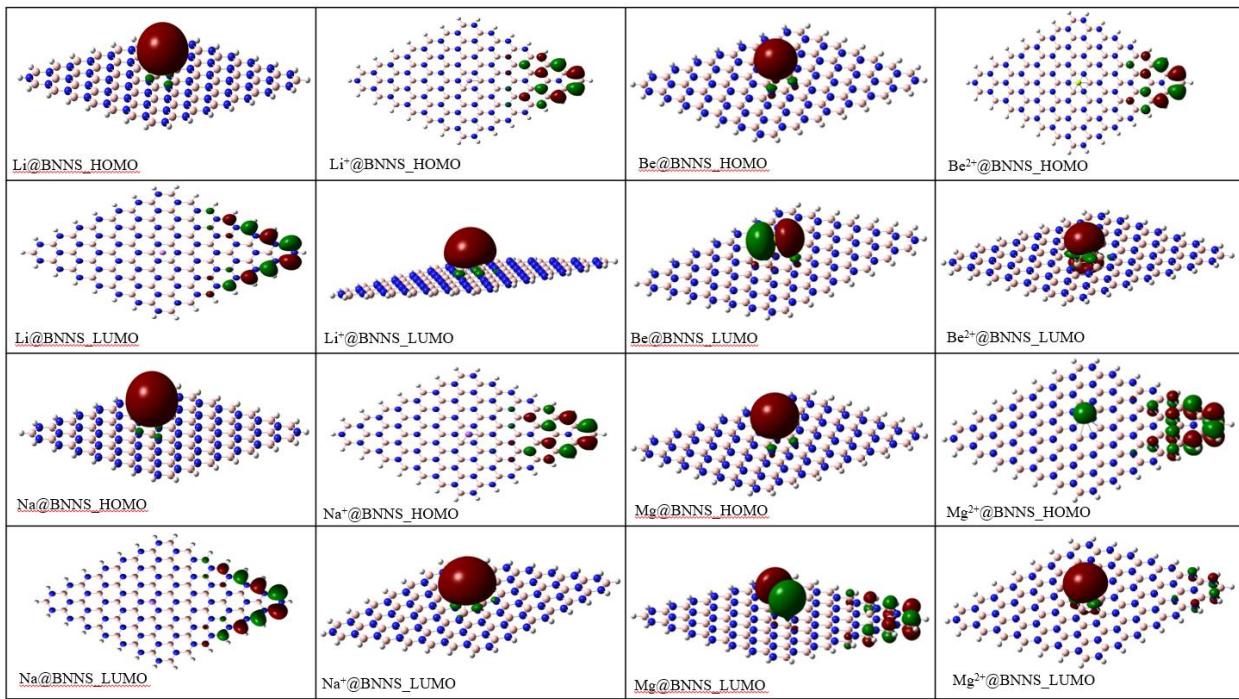


Figure S3: The HOMO and LUMO maps of the metal atoms and/or ions adsorbed onto the BNNS ($M/M^{n+}@BNNS$).

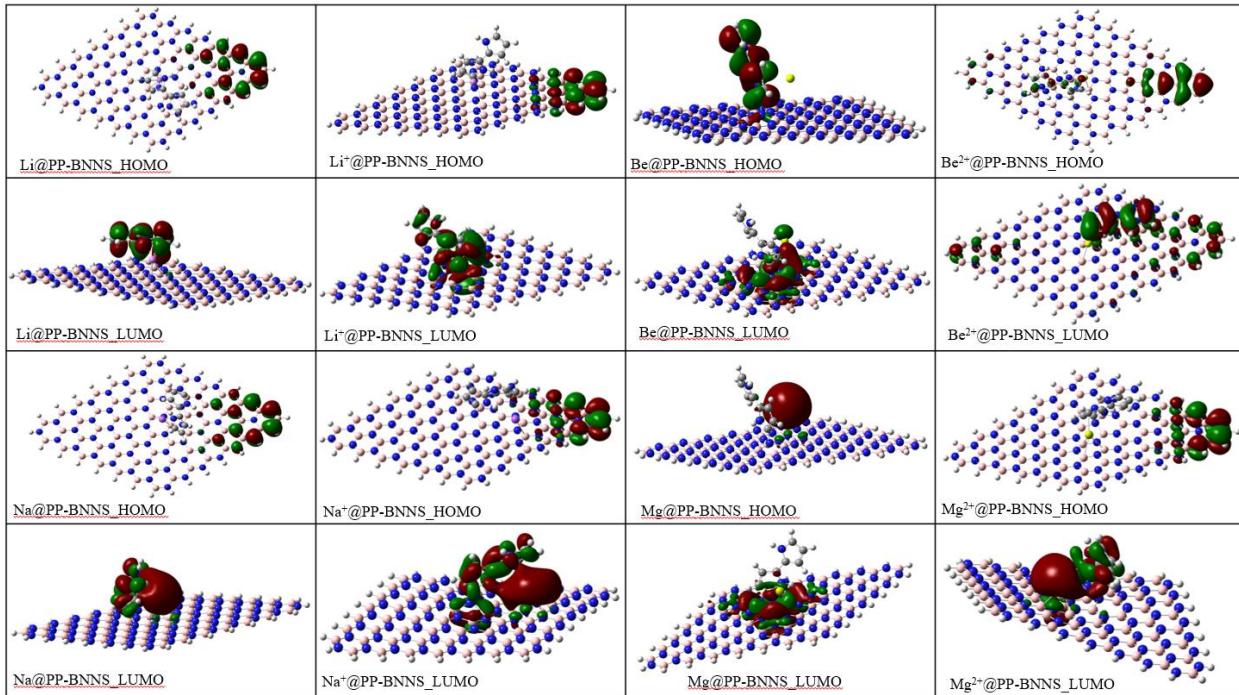


Figure S4: The HOMO and LUMO maps of the metal atoms and/or ions adsorbed onto the PP-BNNS ($M/M^{n+}@PP-BNNS$).

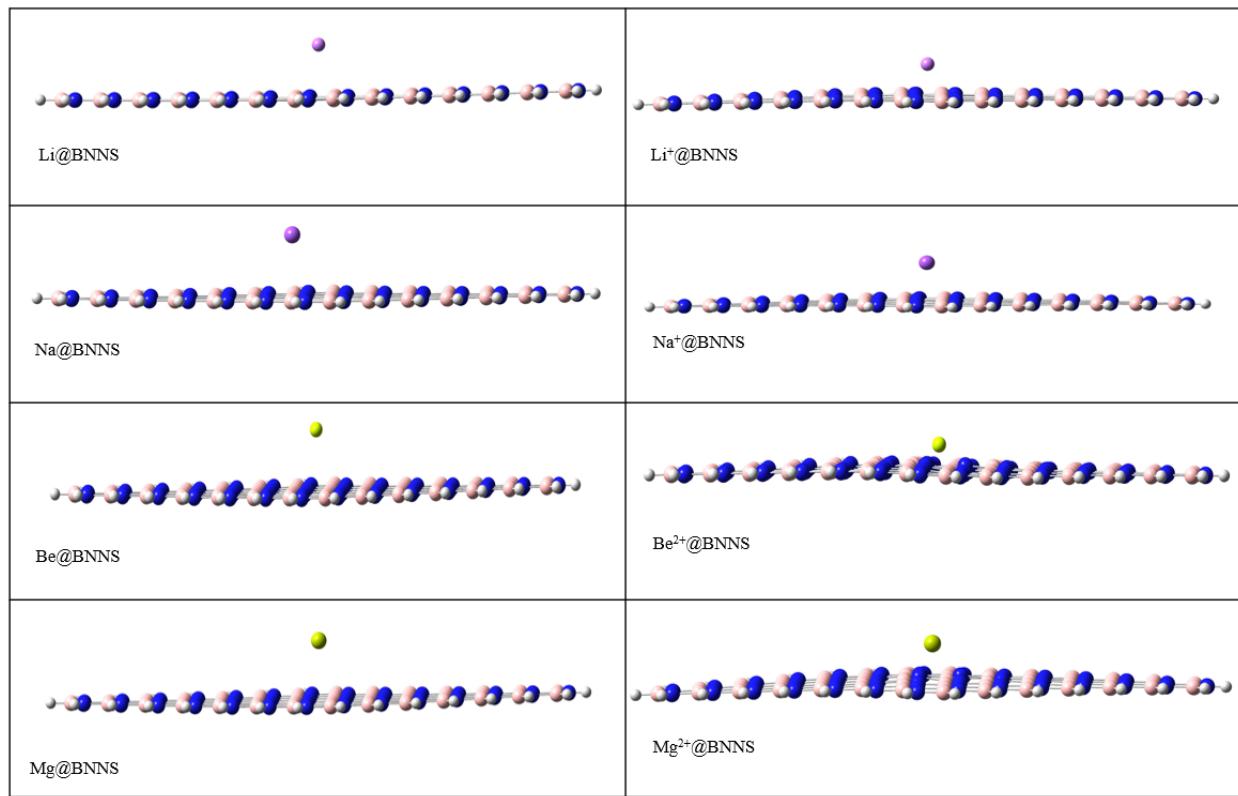


Figure S5: Optimized structural cross-section of (a) Li@BNNS, (b) Na@BNNS, (c) Be@BNNS, (d) Mg@BNNS, (e) Li⁺@BNNS, (f) Na⁺@BNNS, (g) Be²⁺@BNNS, and (h) Mg²⁺@BNNS complexes showing the distance between the M/Mⁿ⁺ and the BNNS. It can be seen that the distance is shorter for Mⁿ⁺@BNNS than M@BNNS due to strong adsorption.

Table S1: Adsorption energies of metal atoms and ions on BNNS and PP-BNNS complexes

	E _{HOMO} (eV)	E _{LUMO} (eV)	E _g (eV)	%ΔE _g	E _{ad} (kcal/mol)
BNNS	-6.449	-0.639	5.809	-	-
PP-BNNS	-7.380	-3.479	3.901	-32.846	-542.71
Li ⁺ @BNNS	-7.979	-4.065	3.914	-32.622	-41.12
Li@BNNS	-3.070	-0.662	2.409	-58.530	-5.45
Na ⁺ @BNNS	-7.953	-3.994	3.959	-31.847	-25.08
Na@BNNS	-2.906	-0.655	2.251	-61.250	-1.82
Be ²⁺ @BNNS	-9.465	-8.815	0.650	-88.810	-308.73
Be@BNNS	-5.877	-1.109	4.768	-17.920	-0.79
Mg ²⁺ @BNNS	-9.695	-9.456	0.239	-95.886	-173.83
Mg@BNNS	-4.801	-0.654	4.147	-28.611	-1.57
Li ⁺ @PP-BNNS	-9.500	-6.318	3.182	-18.431	-1.55
Li@PP-BNNS	-8.014	-3.975	4.038	3.512	-49.17
Na ⁺ @PP-BNNS	-9.787	-5.548	4.240	8.690	-8.57
Na@PP-BNNS	-8.037	-3.444	4.594	17.765	-40.24
Be ²⁺ @PP-BNNS	-11.761	-11.573	0.188	-95.181	-282.29
Be@PP-BNNS	-7.446	-3.539	3.907	0.154	-0.210
Mg ²⁺ @PP-BNNS	-10.886	-9.685	1.201	-69.213	-106.99
Mg@PP-BNNS	-6.592	-3.518	3.074	-21.120	-2.69