Supporting Information

Exploring the impacts of conformer selection methods on ion mobility collision cross-section predictions

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1. Example Chemical Property Prediction Methods

Many groups have developed methods for predicting chemical properties measured in several identification platforms including quantitative structure-retention relationship and machine learning models to predict liquid chromatography retention times,^{1,2} combinatorial approaches to predict MS/MS fragmentation patterns,³⁻⁵ quantum chemical calculations and artificial neural networks for NMR chemical shift predictions,⁶⁻⁹ and classical scattering and machine/deep learning to predict CCS for IMS.10-15 Similarly, our group recently developed the *In Silico* Chemical Library Engine (ISiCLE), which is an automated workflow for molecular property calculation. It has shown preliminary success for calculating collision cross sections (CCS) and NMR chemical shifts.16,17

2. Conformer Definition

We define a conformer as returned by conformer generation tools: Each structure is a conformer, regardless of energy or energy minima. This is important in applications like IMS, where any valid structure can contribute to the CCS. This is in contrast to the IUPAC definition, where a conformer is only a structure that sits at the minimum of a potential energy well. ¹⁸ This latter definition makes no reference to transition state structures that, although fleeting, are present during experiments and impact measured properties such as CCS.

Table S1. Summary of the small molecule test set. We used a benchmark set of 18 small molecules reported in Colby et al. (2019) ¹⁷ with masses ranging from 113 to 687 Da. Experimental CCS values for benchmark set adducts ($[M+H]^+$, $[M-H]^+$, or [M+Na]⁺) were obtained using an Agilent 6560 Ion Mobility Q-TOF MS (Agilent Technologies, Santa Clara) with nitrogen buffer gas, as described in Zheng et al. ¹⁹. This adduct set was also processed through ISiCLE ("Standard" calculation methods) to create an initial predicted CCS baseline. **Fig. 2** in the main plots the m/z vs CCS for the benchmark set molecules.

3. Hardware Architecture and Software Parameters

RDKit, CREST, and software used in the ISiCLE pipeline (AMBER, NWChem, MOBCAL-SHM) were run on PNNL supercomputing platforms, Cascade and Constance. Cascade has 1,440 compute nodes with 16 Intel Xeon cores (E5-2670, 2.6 GHz), 128 GB memory per node, and 14Gb/s data rate per lane (FDR InfiniBand). Constance has 464 dual socket compute nodes with 12-core Intel Haswell processors (E5-2670v3, 2.3 GHz), and 64 GB of DDR3-1600 memory per node.

CREST (v2.7.1):

Under the iMTD-GC workflow, CREST uses a mixture of meta-dynamics (MTD), MD, z-matrix crossing, and other methods to iteratively search for low energy conformers and fill out their conformations by finding their rotamers (conformers in this case being understood under the IUPAC definition, i.e. a conformer is only the lowest energy structure of a potential energy well). We used the following parameters and other default options.

GFN2-xTB (very tight or "vtight" optimization level) z-matrix sorting 6 kcal/mol energy threshold 40 ps MD simulations with 5 fs time step and other default options

RDKit (v 2019.03.1):

RDKit randomly generates conformers using distance geometry, where constraints bound the minimum and maximum pairwise distances between any two atoms ²⁰. We used the default parameters with and without UFF optimization.

MOBCAL-shm:

SEED_I2 5013489 BUFFER_GAS NITROGEN BUFFER_GAS_MASS 28.0134 TEMPERATURE 300 IPR 1000 ITN 10 INP 48 IMP 1024 NUM_THREADS 24

NWChem:

XC: B3LYP Basis: * library 6-31G* task dft energy

AmberTools17, Sanders:

As described by Colby et al. (2019)¹⁷

Fig. S1 Single point convergence plot on conformer variability (RMSD) for PE 16:1/16:1, the most flexible molecule in our set. At sample size S=500, the convergence plot shows the thoroughness of a MC simulation at increasing MC iterations. We used 10,000 which we found to be sufficient.

4. Monte Carlo Sampling Across Versus Within Cycles

Because simulated annealing works in cycles, conformers were sampled from cycles using two different methods to distinguish possible cycle correlation. As shown in **[Fig. S3](#page-5-0)**, in one method the cycles were effectively pooled together by sampling across cycles. One conformer was randomly selected from each of the 1000 cycles for a total of 50 conformers. Their RMSD was then calculated for every conformer pair and then averaged. The second method was to keep the annealing cycles isolated, or to select all of the 50 conformers within a cycle, calculate their pairwise RMSD, and average the result. MC was then performed on these averages to simulate generating random cycles. Indeed, the lower average RMSD in **[Fig. S3](#page-5-0)** for the within cycle method, as well as the clustering shown in **[Fig. S2](#page-4-0)**, shows the conformers of a single cycle are, in some cases, more correlated with each other than they are with conformers of another cycle, as expected.

For BW and LE on CCS, there was no noticeable difference between sampling across cycles vs within cycles. SA would occasionally have a wider standard deviation (as shown in **[Fig. S4](#page-6-0)**) when sampling within cycles, suggesting conformational space is more thoroughly covered when sampling across cycles. We confirm it is best to sample across many cycles to achieve the higher variability between conformer geometries.

Fig. S2 A particularly distinct example of four sequential AMBER simulated annealing cycles clustering separately.

Fig. S3 Monte Carlo convergence plots of the RMSD between conformer geometries for 18 small molecules. The left shows random sampling across AMBER simulated annealing cycles, or treating the whole conformation as a single pool. The right shows sampling within cycle, or sampling a number of whole cycles together.

Fig. S4 Monte Carlo convergence plots of BW, LE, and SA demonstrating a lower standard deviation (higher precision) for SA when sampling across cycles (left) than when sampling within cycles (right).

5. Similarity Downselection Description

The goal of similarity downselection (SDS) is to sample conformational space with fewer conformers while still being representative of the larger population, thus saving on computational expense. Pairwise RMSD between conformations are used as a reciprocative similarity metric – the smaller the RMSD, the greater the similarity. SDS downselects based on this structural similarity to choose a subset of representative similar and most dissimilar conformers.

We developed a heuristic algorithm for performing SDS and created an open source Python package that can be found at [https://github.com/pnnl/sds.](https://github.com/pnnl/sds) The package includes relevant functions for performing SDS on conformers, but the SDS algorithm can also be generalized to any set of items where the items can be described as arrays whose elements are composed of the pairwise relations between the item in question and all other items of the set. Here, we employed the SDS algorithm to find the set of the *n* conformers most dissimilar from each other. To choose the most similar conformer, the pairwise RMSD between all conformers was summed, and the conformer with the smallest total RMSD was considered the most similar conformer.

6. Molecular Property Correlations

Exact Molecular Weight r											
					Converged Java Re						
	Convergence point ? Pvalue			Water Stock Private							
Chain bond count Chain atom count	0.71 0.63	0.01 0.01	0.26 0.21	0.03 0.05	0.93 0.91			0.08			1.0
Rotatable bond count	0.50	$^{< 0.01}$	0.16	0.10	0.90	0.01	0.11	0.18			
Van der Waals surface area (3D) ASA	0.90 0.66	0.01 $^{< 0.01}$	0.24 0.17	0.04 0.09	0.81 0.80	0.01 0.01	0.31 0.23	0.02 0.04			
Wiener index	0.79	$^{< 0.01}$	0.24	0.04	0.79	0.01	0.33	0.01			
Maximal projection radius $ASA+$	0.65	0.01	0.21	0.06	0.79	0.01	0.21	0.05			
Atom count	0.60 0.91	$^{< 0.01}$ 0.01	0.12 0.25	0.16 0.04	0.78 0.77	0.01 0.01	0.18 0.32	0.08 0.01			
Hyper wiener index	0.59	0.01	0.15	0.11	0.76	0.01	0.17	0.09			
Maximal projection area Randic index	0.90 0.93	$^{< 0.01}$ <0.01	0.26 0.26	0.03 0.03	0.76 0.76	0.01 0.01	0.37 0.35	0.01 0.01			
a(zz)	0.75	<0.01	0.33	0.01	0.74	0.01	0.32	0.01			
Average molecular polarizability Bond count	0.95 0.92	0.01 $^{< 0.01}$	0.27 0.25	0.03 0.03	0.73 0.73	0.01 0.01	0.39 0.34	0.01 0.01			0.8
Refractivity	0.95	0.01	0.31	0.02	0.73	0.01	0.47	<0.01			
Aliphatic atom count	0.87	$^{< 0.01}$	0.29	0.02	0.73	0.01	0.37	0.01			
Minimal projection area Molecular polarizability	0.93 0.95	0.01 0.01	0.28 0.26	0.02 0.03	0.72 0.72	0.01 0.01	0.39 0.41	<0.01 0.01			
Platt index	0.87	0.01	0.21	0.06	0.69	0.01	0.27	0.03			
Minimal projection radius measured ccs	0.78 0.94	$^{< 0.01}$ 0.01	0.17 0.34	0.09 0.01	0.68 0.68	0.01 0.01	0.22 0.51	0.05 0.01			
Exact molecular weight	1.00	0.01	0.33	0.01	0.67	0.01	0.47	0.01			
apKa2	0.66	0.01	0.38	0.01	0.66	0.01	0.50	0.01			
Szeged index Aliphatic bond count	0.87 0.89	0.01 0.01	0.26 0.30	0.03 0.02	0.65 0.65	0.01 0.01	0.42 0.42	0.01 0.01			
ASA_H	0.42	$^{< 0.01}$	0.07	0.29	0.60	0.01	0.11	0.17			
a(xx)	0.90	0.01	0.17	0.09	0.57	0.01	0.34	0.01			0.6
a(yy) ASA-	0.81 0.57	0.01 0.01	0.16 0.31	0.10 0.02	0.53 0.53	0.01 0.01	0.34 0.34	0.01 0.01			
Harary index		0.01	0.32	0.01	0.49	0.01	0.55	$^{< 0.01}$			
Balaban index Nitrogen %	0.04 0.15	0.41 0.19	0.03 $^{< 0.01}$	0.50 0.84	0.37 0.25	0.01 0.08	0.01 0.03	0.9 0.55			
Hydrogen %	0.06	0.34	0.01	0.84	0.23	0.04	0.01				
Polar surface area Wiener polarity	0.48 0	0.01 0.01	0.32 0.23	0.01 0.05	0.23 0.22	0.05 0.05	0.28 0.44	0.02			
logP	0.04	0.40	0.01		0.13	0.14	0.01	0.01			
donsitecount -	0.30	0.02	0.22	0.05	0.13	0.14	0.18	0.08			
Aromatic bond count Aromatic atom count -	0.01 0.01	0.75 0.79	0.01 0.01	0.79 0.84	0.11 0.11	0.17 0.17	0.01 0.01	0.77			
Aromatic ring count -	0.01	0.72	0.01		0.11	0.18	0.01				
accsitecount acceptorcount -	0.40 0.33	0.01 0.01	0.26 0.28	0.03 0.02	0.11 0.11	0.18 0.19	0.26 0.21	0.03 0.06			0.4
Carbo ring count -	0.01	0.78	0.11	0.19	0.10	0.21	0.02	0.61			
donorcount -	0.30	0.02	0.25	0.04	0.09	0.23	0.22	0.05			
Fused ring count - Tautomer count -	0.01 0.05	0.63 0.36	0.02 0.06	0.62 0.35	0.09 0.08	0.23 0.26	0.01 0.01	0.64 0.67			
bpKa1	0.01	0.86	0.03	0.46	0.08	0.27	0.01	0.89			
Fused aromatic ring count - Cyclomatic number -	0.03 0.09	0.53 0.23	0.02 0.02	0.61 0.63	0.06 0.06	0.31 0.32	0.07 0.12	0.30 0.15			
Ring count -	0.09	0.23	0.02	0.63	0.06	0.32	0.12	0.15			
Ring bond count -	0.10	0.21	0.02	0.59 0.21	0.06 0.06	0.33	0.14	0.13			
Largest ring size - Carboaromatic ring count -	0.01 0.01	0.69 0.88	0.12 0.01	0.63	0.06	0.39 0.34	0.19 0.01	0.10 0.67			
Ring atom count -	0.10	0.20	0.03	0.50	0.05	0.37	0.18	0.08			
apKa1- Heteroaromatic ring count -	0.02 0.01	0.59 0.7°	0.01 <0.01	0.82	0.04 0.04	0.45 0.43	0.01 0.01	0.99 0.78			
Heteroaliphatic ring count -	0.20	0.07	0.25	0.03	0.01	0.67	0.52	0.01			0.2
ASA_P- Resonant count -	0.06	0.33	0.09	0.23	0.01	0.70 0.71	0.08	0.26			
Oxygen % -	0.13 0.10	0.14 0.19	0.12 0.04	0.17 0.41	0.01 0.01	0.71	0.18 0.03	0.08 0.47			
Dreiding energy -	0.09	0.24	0.07	0.28	0.01	0.73	0.13	0.15			
Tetrahedral Stereoisomer count - $pl -$	0.12 0.11	0.16 0.29	<0.01 0.11	0.95 0.28	0.01 0.01	0.83 0.87	0.01 0.01	0.71 0.93			
Asymmetric atom count -	0.20	0.07	0.05	0.37	0.01	0.85	0.04	0.45			
Chiral center count - Hetero ring count -	0.20 0.08	0.07 0.26	0.05 0.14	0.37 0.12	0.01 0.01	0.85 0.85	0.04 0.24	0.45 0.04			
Carbon % -	0.01	0.9	0.09	0.22	0.01	0.87	0.01	0.88			
Aliphatic ring count -	0.20	0.07	0.05	0.40	0.01	0.89	0.18	0.08			
Smallest ring size - Stereoisomer count -	0.01 0.14	0.77 0.12	0.01 0.01	0.76 0.98	0.01 0.01	0.91 0.93	0.01 0.01	0.98 0.67			

Fig. S5 Heat map of molecular properties (y-axis) correlated with RMSD MC convergence properties and exact molecular weight $(x-axis)$, showing the $r²$ correlation with its associated p-value.

Fig. S6 Volcano plot of MC convergence properties (converged value, convergence point, maximum standard deviation) correlated against calculated molecular properties. The plot shows possible statistical significance of these correlations, although the p-value of 0.05 was arbitrarily chosen, and having a lower p-value does not necessarily mean higher statistical significance once the target p-value threshold his crossed. Interestingly, we note the one molecular property most negatively correlated with all three MC convergence properties (metrics of the variability between a molecule's conformers) was the second acidic pKa site being more weakly acidic.

Table S2 CCS values for various conformer selection methods. See **[Table S3](#page-9-0)** for description of columns.

Molecule	BW	LE	SΑ	ET ₅	ET 2	ET ₁	ET 0.5	CREST	ISiCLE	Best Combo	Experimental
Harmine +H	149.2	148	149.7	149.4	149.1	149.3	149.3	149.2	148.7	149.7	146
1-Methylguanosine +H	164.5	164.4	166.7	165.4	165	164.5	164.5	164	165.7	165.8	168.8
Sphingosine +H	201.8	207.4	207.4	204.9	202.8	202.8	196.8	170.9	180.6	184.5	186
riboflavin +H	192.8	193.1	197.3	193.7	193.9	191.6	191.6	182.4	192.9	193.4	188.3
Mandelonitrile +H	129	128.5	130.4	130.3	129.7	128.5	128.5	131.3	127.1	129.2	128.9
Creatinine +Na	119.4	119.5	119.7	119.3	119.3	119.4	119.6	118.7	118.9	119.7	133.4
Methyleugenol +Na	143.3	143	143	143	143	143.4	144.2	139.4	141.8	143.1	160.4
N6-methyladenosine +Na	166.8	168	165.1	165.7	166.2	166.7	167.3	164.9	166	165.5	170.4
Cholic Acid +Na	186.5	186.6	186.9	186.2	186.4	186.7	187.4	183.6	185.8	187.1	197.3
Astilbin +Na	206.5	208.3	201.2	202.7	204	204	208.3	202.8	200.9	203.3	212.6
SDGRG +Na	227.5	227.5	218.4	227.5	227.5	227.5	227.5	209.6	226.6	224.3	203.5
Biliverdin +Na	256.8	256.4	251.8	250	263.4	256.4	256.4	258.3	257.6	265.2	246.7
Anthranilic acid -H	127.8	127.7	128.1	127.9	127.7	127.6	127.7	126.7	127.9	126.8	124
Aminohippuric acid -H	154.7	154.7	155.4	154.7	154.6	154.7	154.7	152.3	152.1	157.1	147.6
3'-O-methylguanosine -H	169	167.8	176.1	169.4	169.2	169.4	169.4	168.7	168.7	171.9	163.8
Sucrose -H	163.2	163.5	166.3	165.3	163.1	163.1	163.1	157.6	163.4	162.9	168.5
Naringin-H	238.8	238.1	235.9	238.8	240.5	238.1	238.1	219.3	240.9	234.7	217.3
PE 16:1/16:1 -H	292.4	292.4	309.7	305.2	291.9	291.9	291.9	NaN	311.3	314.5	256.3

Table S3 Mean absolute percent error of various conformer selection method results relative to ISiCLE CCS. The ISiCLE method selects the most similar and two most dissimilar conformers out of 10 AMBER simulated annealing cycles (for a total of 30 conformers), applies DFT geometry optimization, and averages the CCS with Boltzmann weighting. Boltzmann weighting (BW), lowest energy (LE), simple average (SA), and simple averaging under energy thresholds 5, 2, 1, and 0.5 *kcal/mol* (ET 5, 2, 1, 0.5) were applied to 50k AMBER conformers and using DFT energies. CREST is the single lowest energy CREST conformer. The best combo is the statistical best combination we found on the AMBER conformers for the 18 molecules—10 AMBER cycles, selecting the most similar and 10 most dissimilar set under an AMBER energy threshold of 10 *kcal/mol,* and choosing the lowest energy according to the conformer's DFT energy.

Table S4 Same **[Table S3](#page-9-0)** except with mean absolute percent error calculated relative to experimental CCS.

Table S5. Shows at which Sanders (AmberTools17) simulated annealing cycle the nth lowest energy conformer was generated for n=1-10. The maximum number of cycles generated for this project was 1000. Based on this data it is reasonable to assume new low energy conformers would be generated past 1000 cycles.

7. **Monte Carlo Simulations and CCS vs Energy Space**

Fig. S7 MC simulation convergence plots on CCS (left) for SA, BW, and LE, and how they relate to CCS vs energy space (right). Black and gray represent standard deviation from the average (pink). SA converges to the CCS where the conformers are most dense in the CCS versus energy space, BW convergences to the average of low energy conformers or clusters of conformers, and LE converges to the single lowest energy CCS after 50k conformers are selected.

Fig. S8 RDKit with UFF optimization on ~6k mandelonitrile [M+H]⁺ conformers. Left gray cluster shows ~50k DFT geometry optimized AMBER structures, right gray cluster shows ~50k RDKit without UFF, and middle clusters with density coloring indicate the ~6k RDKit structures with UFF optimization. In this example, UFF optimization clustered the RDKit conformers into tight energy intervals, which would likely greatly increase precision for BW, LE, and other low energy dependent conformer selection methods. However, the UFF conformers have energies much higher than the DFT geometry optimized conformers, and different CCS as well, making it unclear how this affects accuracy.

Fig. S9 MC convergence plots on CCS using three sampling techniques (SA, BW, LE) for conformers generated in AMBER and the same AMBER conformers after a DFT geometry optimization for mandelonitrile $[M+H]^+$, creatinine $[M+Na]^+$, and sucrose [M-H]⁻. Note that for sucrose, only about 25k of the 50k AMBER conformers were DFT geometry optimized here.

Fig. S10 MC convergence plots on CCS applying SA under 5, 2, 1, and 0.5 *kcal/mol* energy thresholds. Black represents standard deviation from the average (pink).

Fig. S11 Diagram of conformer selection and down selection methods for all molecules in the test set. Simple average (SA), lowest energy (LE), Boltzmann weighting (BW), energy threshold (ET), and similarity down selection (SDS). SA shows 50 randomly selected conformers, BW is shaded based off real weighted values, ET is a 5 kcal/mol threshold, and SDS shows the one most similar and 49 most dissimilar conformers.

8. Using MD vs DFT energy on MD structures

Fig. S12 Comparison of run times for NWChem energy optimization on MD structures (node minutes), NWChem geometry optimization (node minutes), MOBCAL-shm (node minutes), and AMBER simulated annealing (averaged per conformer, wall minutes).

Table S6 Average and range of runtimes for NWChem (node), MOBCAL-shm (node), and AMBER (wall).

Fig. S13 Comparing CCS vs energy space for MD and DFT calculations. Top row: AMBER conformers with MD energies relative to the minimum energy. Middle row: DFT geometry optimized and non-optimized AMBER conformers with DFT energies, relative to the minimum energy of both sets. There are 30 DFT geometry optimized conformers for all molecules except for mandelonitrile and creatinine which have about 50k and sucrose which has about 25k. Bottom row: DFT vs MD energy correlations.

9. Limitations of the study

- The RMSD for SDS and geometry variability were calculated by excluding the non-backbone hydrogens. However, it appears even rotating a methyl group can lead to significantly different CCS calculations, and this appears to contribute to the CCS range of DFT geometry optimized conformers.
- Duplicate or nearly identical conformers generated by chance may give those conformers more weight than they should when Boltzmann weighting. This is a general problem for conformer

generation tools like AMBER that don't screen for duplicate conformers. Tools like CREST check for this duplicity.

The stereochemistry of generated conformers was spot checked. Differences in stereochemistry can lead to differences in CCS and energy. We recommend using automated checking software, such as Bond Locator Utilizing Electronic Structure (BLUES,

https://github.com/quantum2classical/blues) in conjunction with smiles canonicalizers.

10. Conformer selection analyses with AMBER energies

The following are example figures with the original AMBER energies instead of DFT energies, which were calculated on the AMBER structures. DFT energies were found to have better correlation with the conformers after DFT geometry optimization.

Fig. S14 CCS vs energy landscapes for 50k AMBER generated conformers for creatinine [M+Na]⁺, sucrose [M-H]⁻, PE 16:1/16:1 [M-H]- respectively. Highlighted are the most similar (dashed) and two most dissimilar (solid) conformers chosen heuristically with a structural RMSD metric.

Fig. S15 MC convergence plots on CCS for harmine [M+H]⁺.

Fig. S16 MC convergence plots on CCS using three sampling techniques (SA, BW, LE) for conformers generated in AMBER, RDKit, and the AMBER conformers after a DFT geometry optimization for mandelonitrile $[M+H]^+$.

Fig. S17 MC convergence plots on CCS for mandelonitrile $[M+H]^+$ for 5, 2, 1, and 0.5 kcal/mol energy thresholds.

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