

Supporting Information

Nanophotonic Biosensors Harnessing van der Waals Materials

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2D vdW materials for nanopore sensing

Beyond electromagnetic field confinement in 2D materials, one of the most exciting applications to access molecular information is the ability to physically confine individual analyte molecules in a configuration with lower degrees of freedom. Depending on the 2D material and experiment, information about the physical and chemical properties of the analyte can be extracted, often at the single-molecule level. An emerging detection modality is based on the confinement of molecules into structures fabricated in the matrix of a 2D material. One very promising strategy is in the incorporation of nanoscale apertures or nanopores in such materials for applications in biomolecular sensing¹. Typically, the diameter of the nanopore is less than 10 nm, closely matching the dimension of individual molecules or within an order of magnitude of the screening length of individual ions. The selection of the correct membrane material is highly significant in ensuring efficient molecular transport through the nanopore. With this in mind, there has been a drive to design membranes from 2D materials, which offers unprecedented advantages over more conventional 3D materials.

The principles of conventional nanopore-based molecular sensing are straightforward; briefly, a constant voltage bias is applied across the nanopore that connects two reservoirs filled with electrolyte separated by a non-permeable and insulating membrane. Although 2D materials can be only a single atomic layer thick, they are often excellent ionic insulators, and the applied bias causes a steady-state ionic current to flow through the nanopore and as the resistance of the nanopore is greater than that of the electrolyte, majority of the potential drop occurs at or close to the nanopore. Ions and any other charged molecules are therefore transported (or translocated) through the nanopore as a result of the generated potential gradient. When a molecule enters the nanopore, the ionic current more often than not decreases, which results in a current blockade; however, current enhancement can also be obtained depending on the analyte properties, material properties of the membrane, and electrolytic conditions. The amplitude, duration, and in some cases the shape of the current change can reveal valuable information of the molecule being probed, including size, charge, lengths, and affinity amongst others.² Importantly these measurements are obtained for each translocated molecule, instead of measuring the average behaviour of an ensemble of molecules.

Although applications using nanopores are rapidly growing, they are not suitable for use in all circumstances. Both the nanopore diameter and membrane thickness are crucial for maximising the signal-to-noise (S/N) levels and require careful optimisation based on the target analyte. As a general rule, the smaller the nanopore diameter, the greater the S/N. Fortunately, nowadays there is an arsenal of established techniques for making nanoscale pores in solid-state materials, including transmission electron microscopy,^{3,4} ion milling and focused-ion-beam (FIB) microscopy^{5,6,7} electron-beam lithography^{8,9,10} dielectric breakdown,¹¹ laser etching¹². For a detailed discussion of these methods, the reader is referred to alternative reviews.¹³ Often single-digit-nanometer and even sub-nanometer pores can be achieved with high reproducibility. This flexibility in tuning nanopore dimensions has been arguably the single most crucial advantage of solid-state nanopores over their biological counterparts where the nanopore dimension is typically fixed depending on the protein being used.

Much like the nanopore diameter, the membrane thickness also plays a large role in maximising the signal-to-noise, where to a first approximation, the paradigm, less is more, is most certainly true. However, due to advances in materials science, it has recently only been possible to reliably fabricate nanopores within solid-state materials with sub-10 nm thickness, let alone 2D materials. The membrane thickness can be controlled using a bottom-up or top-down approach or a combination of both. In terms of bottom-up approaches, ultrathin membrane materials have been fabricated using ALD and chemical vapour deposition (CVD). Traditionally materials such as SiO₂, TiO₂, HfO₂, Al₂O₃ and graphene or even multiple layers of different materials can be deposited over a supporting dielectric structure usually made of SiN_x or SiO₂, fabricated on a Si wafer. An alternative fabrication strategy is the exfoliation or CVD growth of 2D materials such as graphene, MoS₂, hBN into nanosheets, followed by their transfer onto the supporting structure.^{14,15,16,17,18,19,20}

Alternatively, top-down approaches such as local thinning of existing solid-state membranes have also been documented. In most cases, silicon nitride (SiN_x), which is a traditionally used as the stoichiometry can be

modified to tune the Si to N ratio and to form high-aspect ratio membranes with low- tensile stress and high mechanical robustness. Local sections of the SiN membrane can be selectively etched, either using dry etching²¹, electron or ion-beam irradiation,^{22,23} or by laser-assisted photothermal etching²⁴. Nanopores can then be formed using conventional fabrication methods (usually using electron/ion microscope, dielectric breakdown, or laser etching), with the significant advantage of having a well-defined pore shape and thickness approaching atomic dimensions.

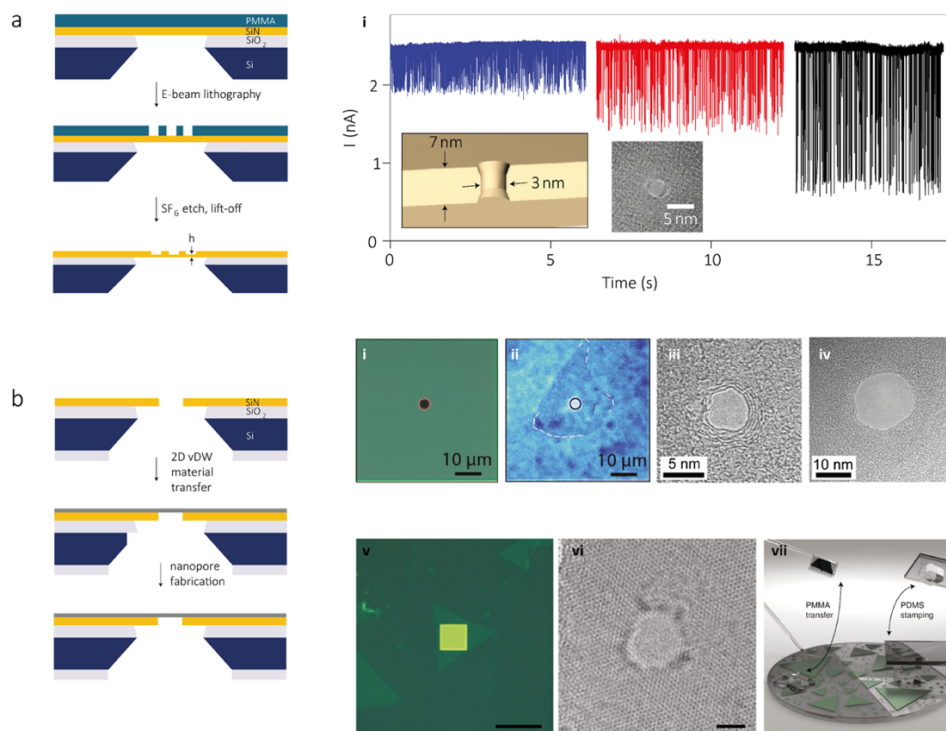


Figure S1 | Strategies for making 2D membranes. a) local thinning. In this example, a SiN membrane is locally thinned using SF₆ to locally etch the membrane around the nanopore. (i) The increase in signal to noise could clearly be seen for thinner membranes as is shown for the translocation of 3 kbp dsDNA, adapted from 21. b) transfer of 2D vdW material. (i-iv) Graphene membrane including optical micrographs depicting the transfer of graphene on Si/SiO₂. The nanopores were made using a highly focussed electron beam of a transmission electron microscope, adapted from 17. (v-vii) Overview of the fabrication process for making MoS₂ nanopores including an optical micrograph of the transferred material on a SiN membrane (scale bar 20 μm) and a TEM image of a drilled nanopore (scale bar 2 nm), adapted from 31.

A key advantage of utilising nanopores in 2D vdW materials is in the substantial increase in spatial resolution and S/N, due to the shorter length over which the ionic current signal is probed and the decrease of the nanopore resistance. These advantages can lead to enhanced sensitivity to small molecular differences; however, it is essential to highlight that to obtain transformative information at the submolecular level, the temporal resolution needs to be improved as well, or strategies for slowing down molecular transport need to be implemented. Hybrid 2D membrane materials can be used to tune interactions with specific analytes. For example, nucleic acid translocation can be slowed down due to increased electrostatic interaction with Al₂O₃ or ZnO membranes,^{25,26} or due to physicochemical interaction in HfO₂ membranes²⁷. Furthermore, single-stranded DNA has also been shown to be slowed down in graphene due to non-specific adsorption. Also, the unique electronic properties of that 2D materials can exhibit additional detection modalities. For example, the possibility of using in-plane electronic conductivity in graphene membranes or semiconducting MoS₂ membranes can enable additional channels for molecules detection or manipulation.

It should be noted that one of the challenges of using 2D membrane materials such as graphene is the increased hydrophobicity, leading to the unwanted adsorption of analytes such as DNA onto the membrane surface.²⁸ However, the strong hydrophobic interaction between DNA and graphene can be minimised using surface treatments of graphene or 2D vdW materials that pose enriched hydrophilic surface sites such as MoS₂.^{29,30,31}

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