

Supporting Information

Defining the qualities of high-quality palladium on carbon catalysts for hydrogenolysis

Conor J. Crawford,^{[a,b]*} Yan Qiao,^[c,d] Yequn Liu,^[c,d] Dongmei Huang,^[c,d] Wenjun Yan,^[c,d], Peter H. Seeberger,^[b] Stefan Oscarson,^{[a]*} and Shuai Chen,^{[d]*}

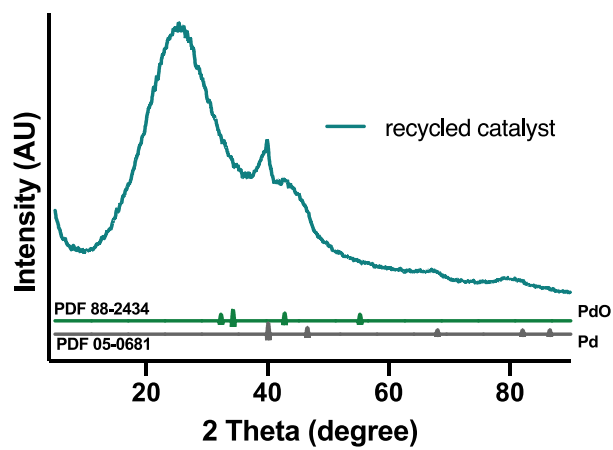
- [a] Centre for Synthesis and Chemical Biology, University College Dublin, Belfield, Dublin, Ireland. E-mail: conor.crawford@ucdconnect.ie stefan.oscarson@ucd.ie
- [b] Department of Biomolecular Systems, Max Planck Institute of Colloids and Interfaces, 14476 Potsdam, Germany.
- [c] Center of Materials Science and Optoelectronics Engineering, University of Chinese Academy of Sciences, Beijing 100049, People's Republic of China.
- [d] State Key Laboratory of Coal Conversion, Institute of Coal Chemistry, Chinese Academy of Sciences, Taiyuan 030001, People's Republic of China. Email: chenshuai@sxicc.ac.cn

Table of Contents

TEM and XRD of 5% Pd/C (pre-treated) recycled catalyst	3
Figure S1. TEM and XRD of 5% Pd/C (pre-treated) recycled catalyst.....	3
TEM of 5% Pd/C STREM Chemicals (1#)	4
TEM 10% Pd/C Sigma-Aldrich (2#).....	9
TEM of 20% Pd[OH] ₂ /C Sigma-Aldrich (3#).....	15
Palladium: PDF 05-0681.....	20
Palladium oxide: PDF 88-2434	21
SEM-EDS	22

TEM and XRD of 5% Pd/C (pre-treated) recycled catalyst

A



B

TEM

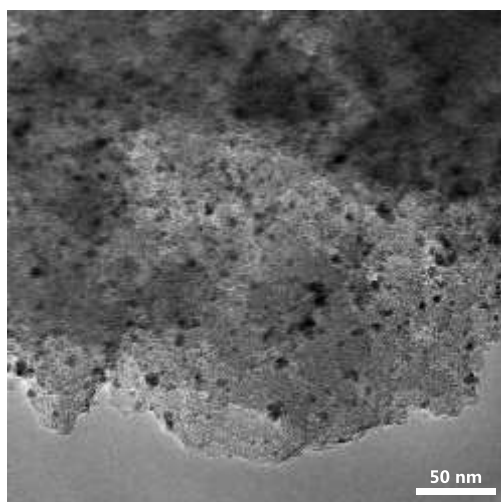
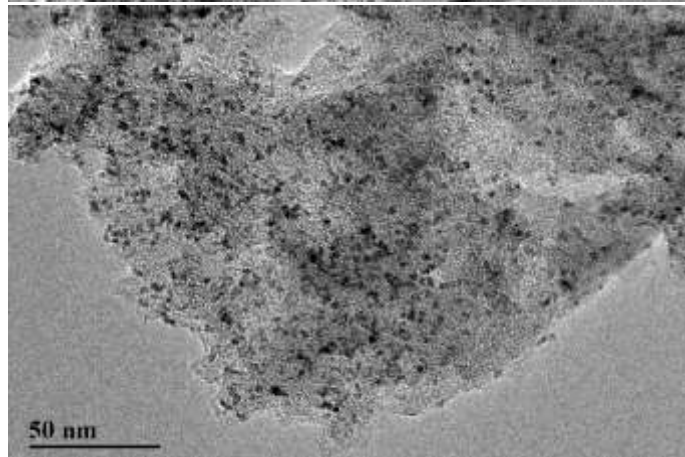
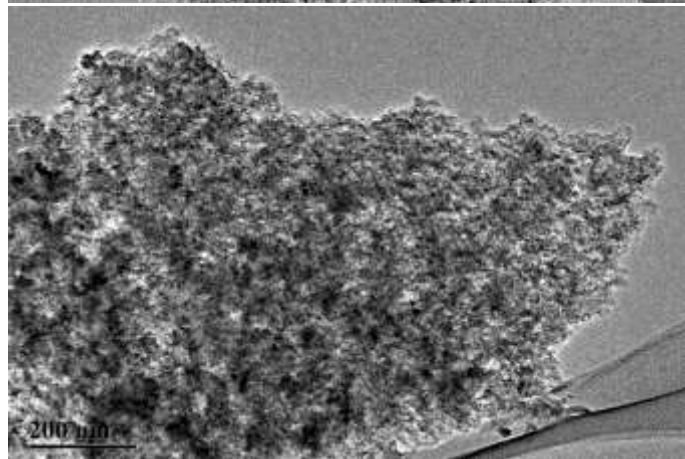
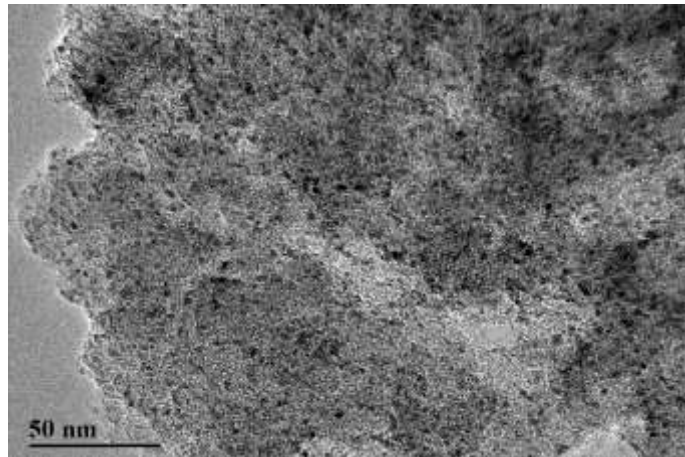
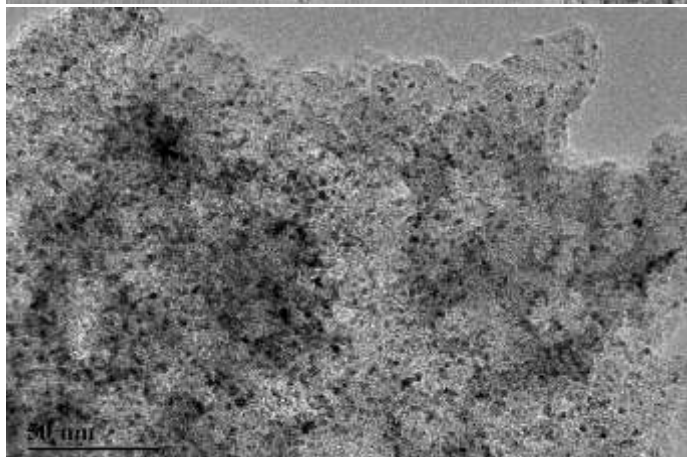
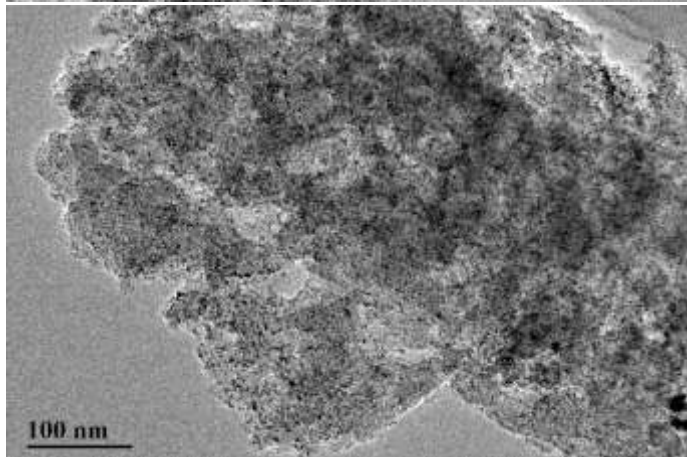
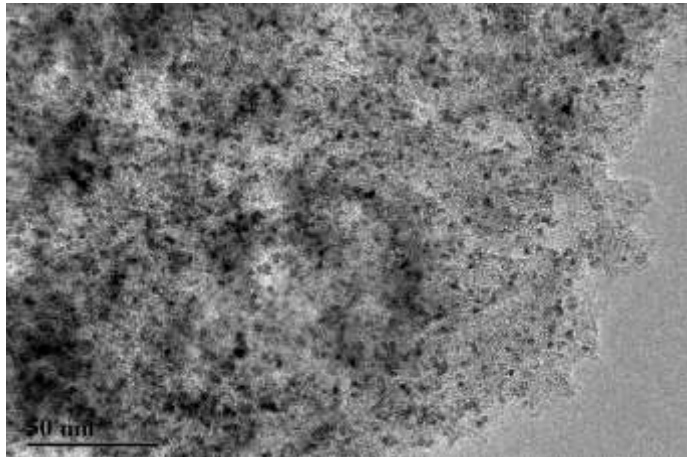
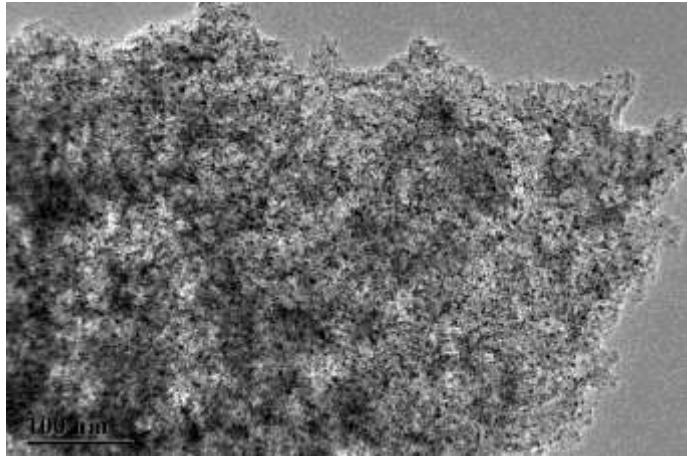
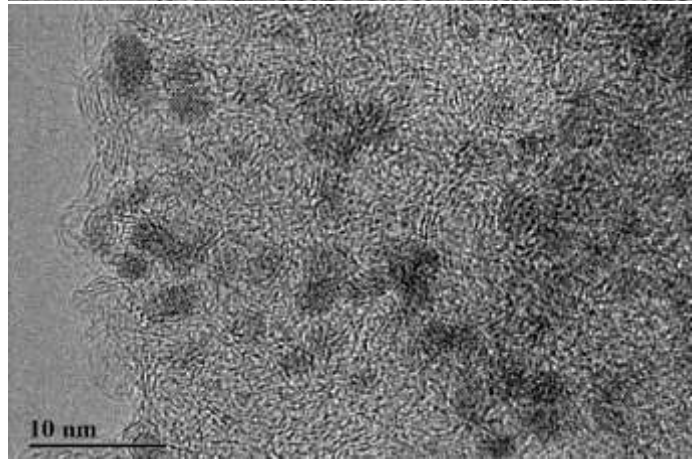
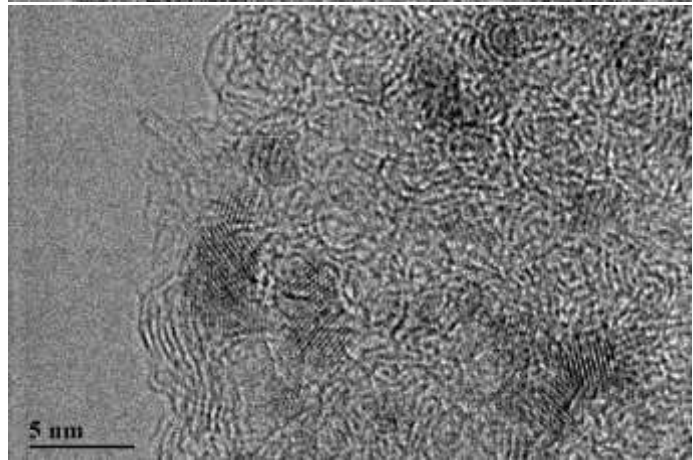
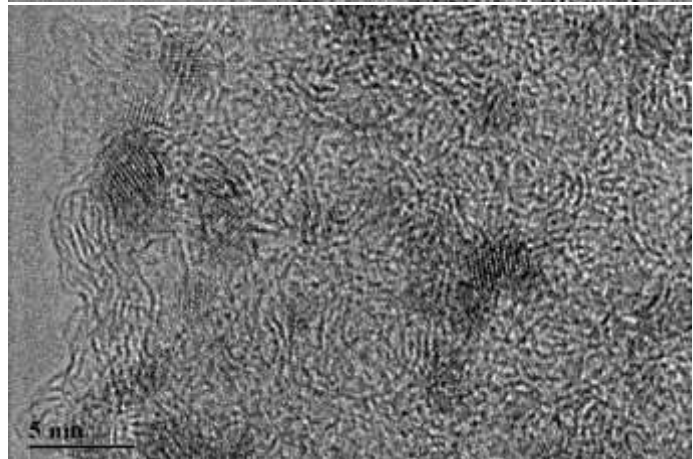
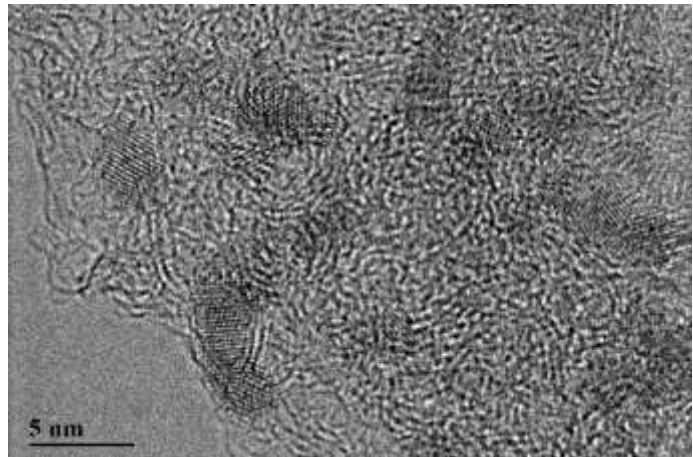


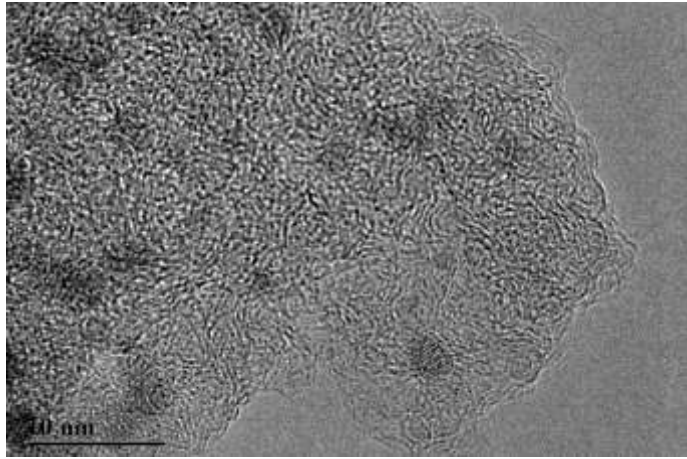
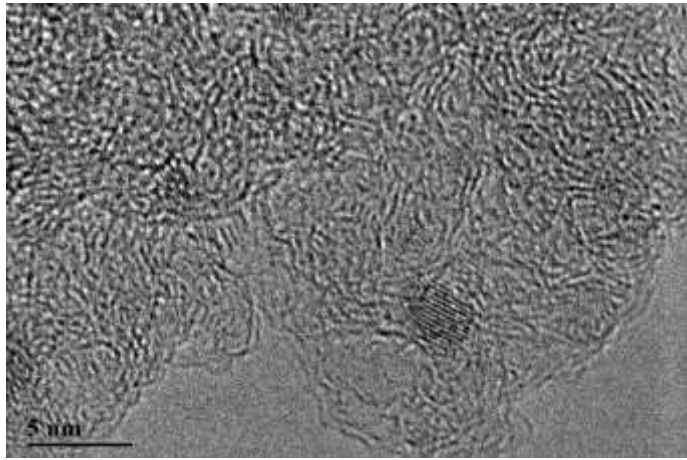
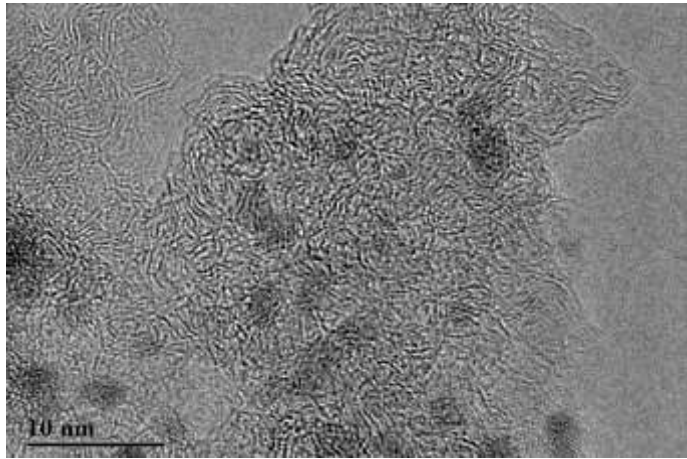
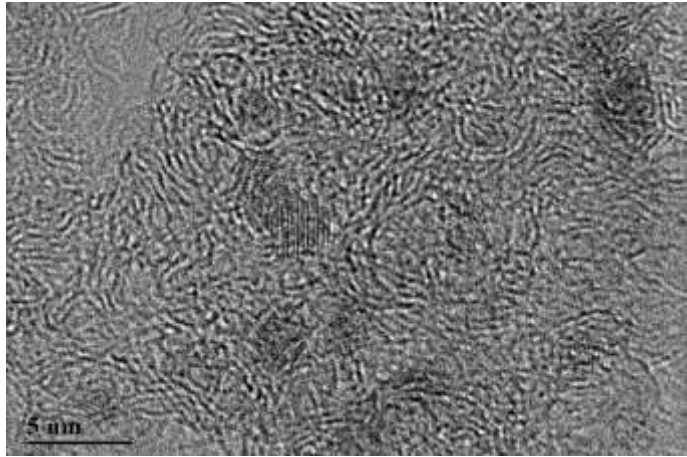
Figure S1. TEM and XRD of 5% Pd/C (pre-treated) recycled catalyst

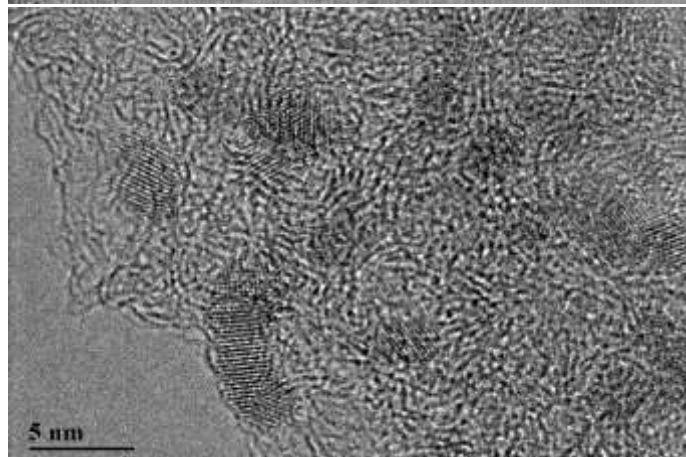
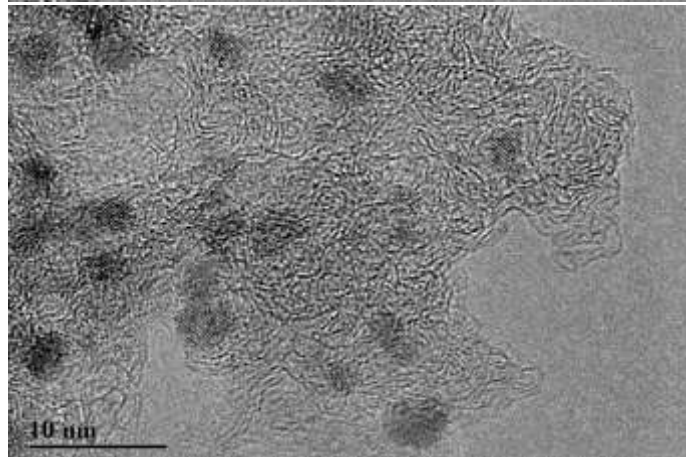
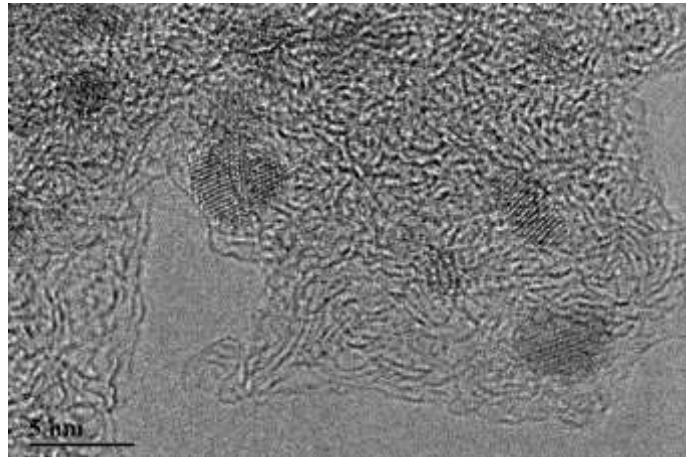
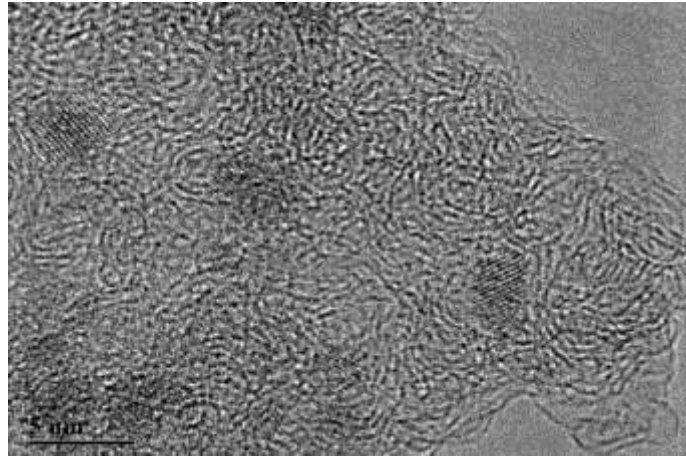
TEM of 5% Pd/C STREM Chemicals (1#)



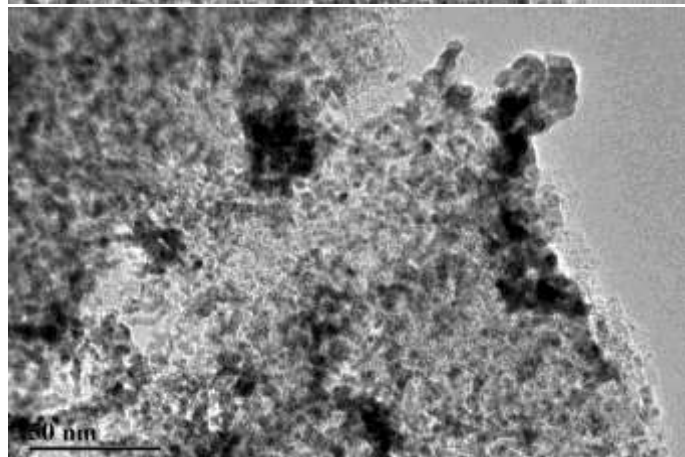
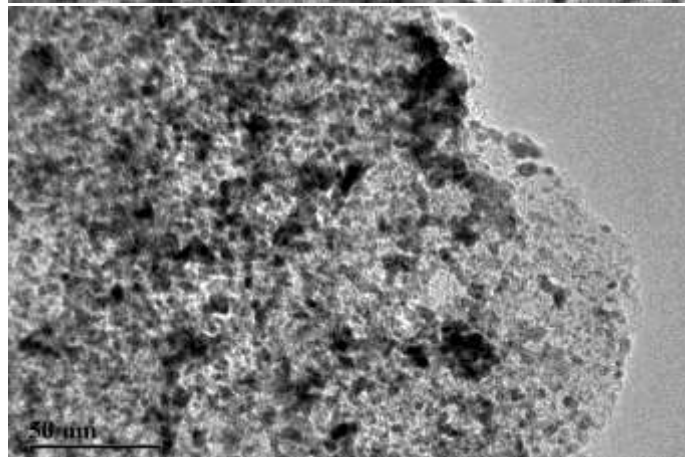
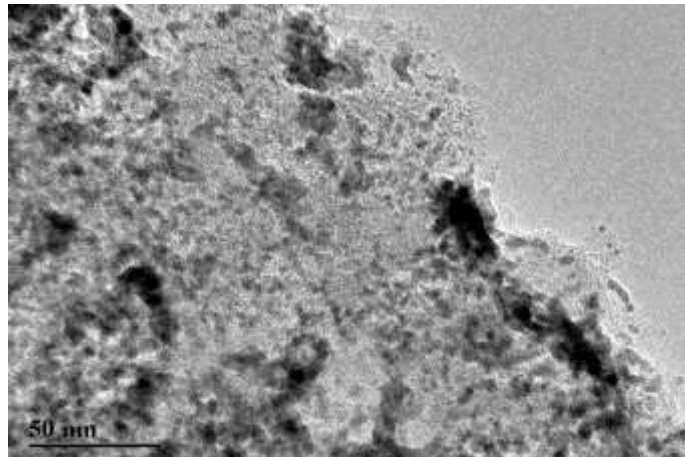


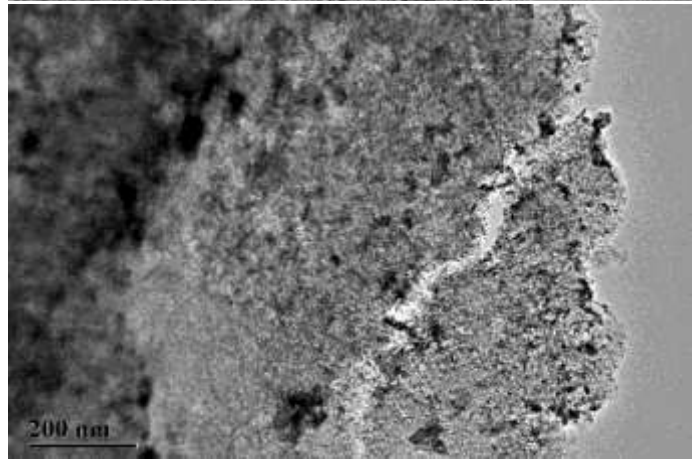
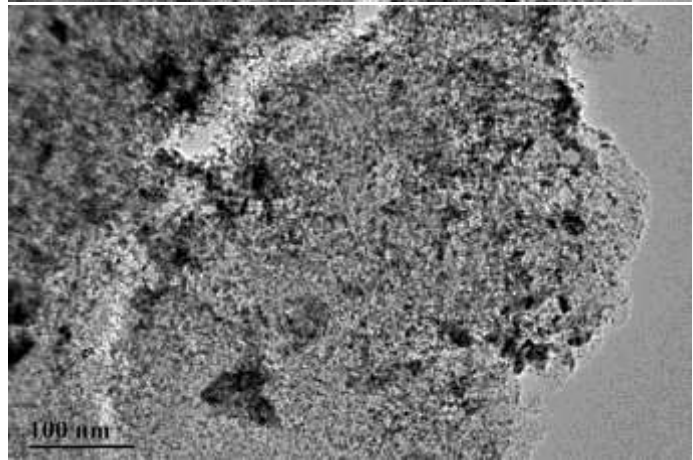
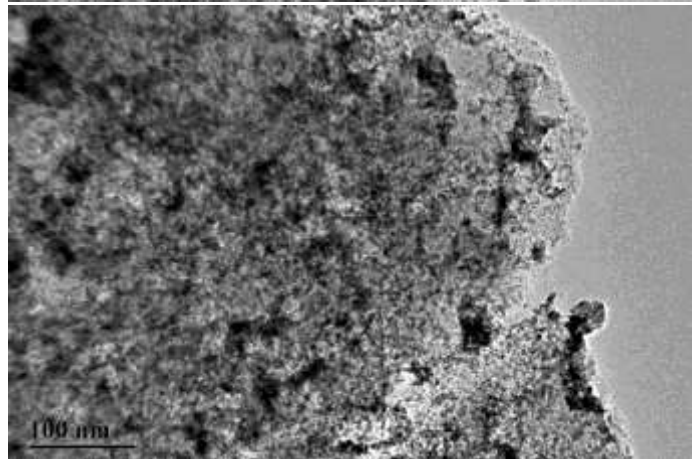
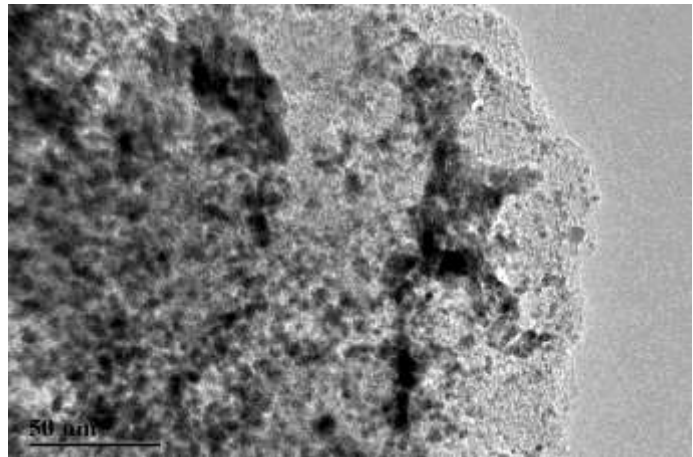


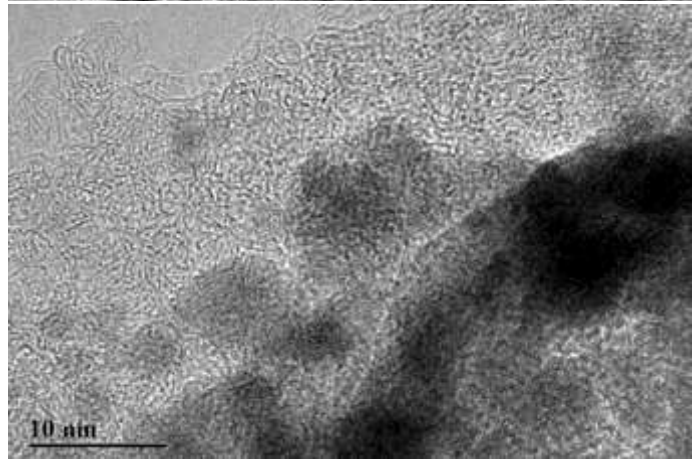
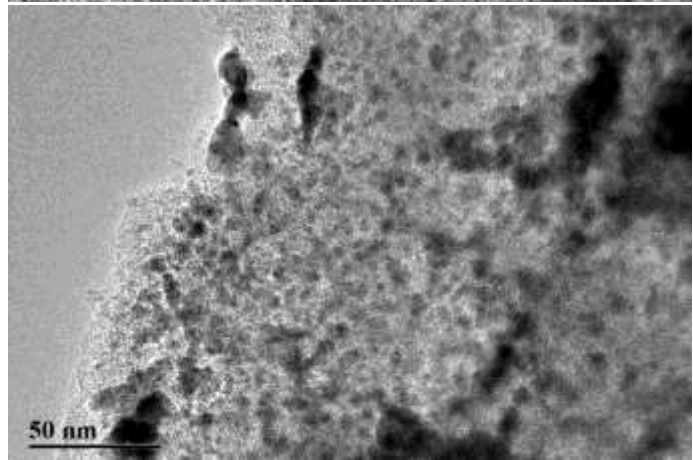
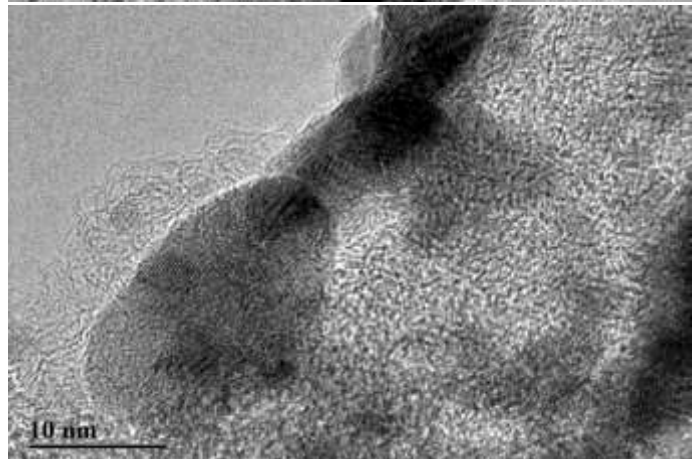
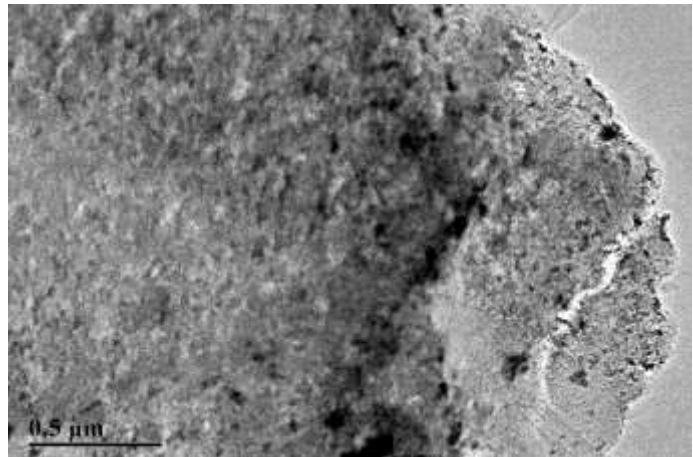


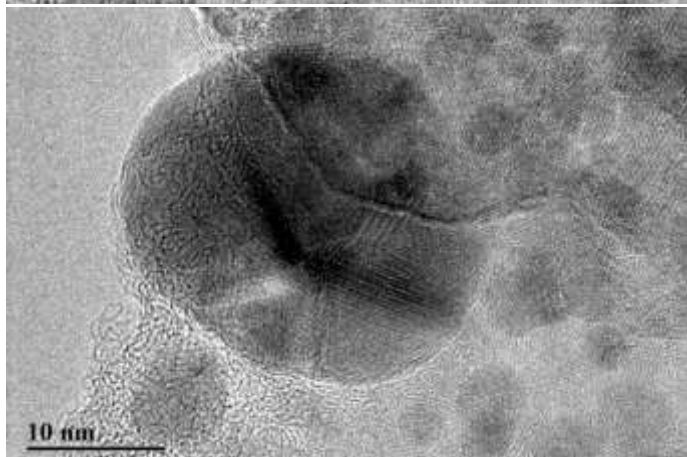
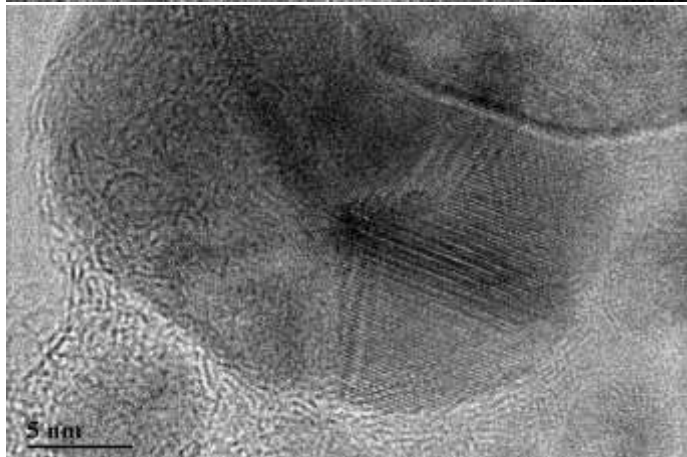
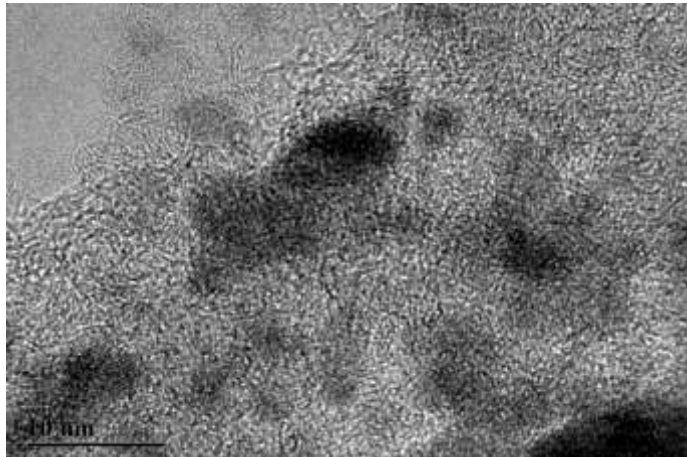
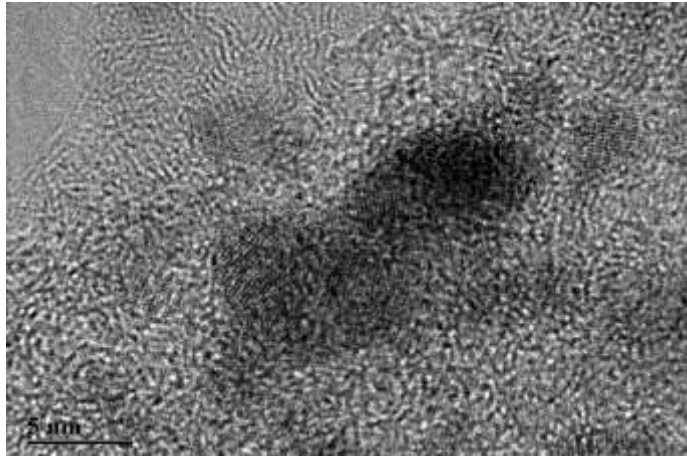


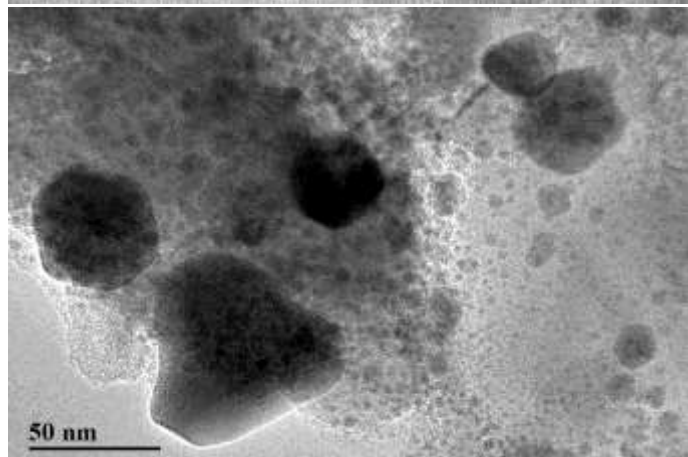
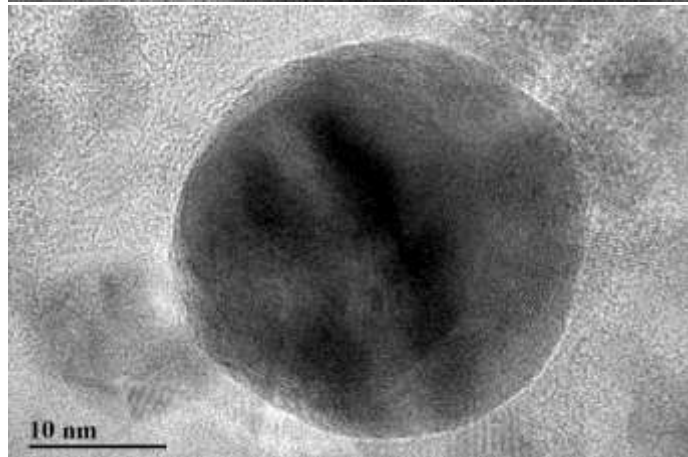
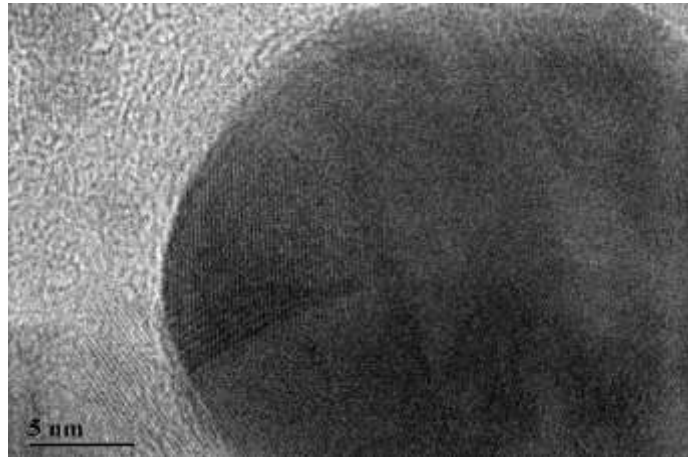
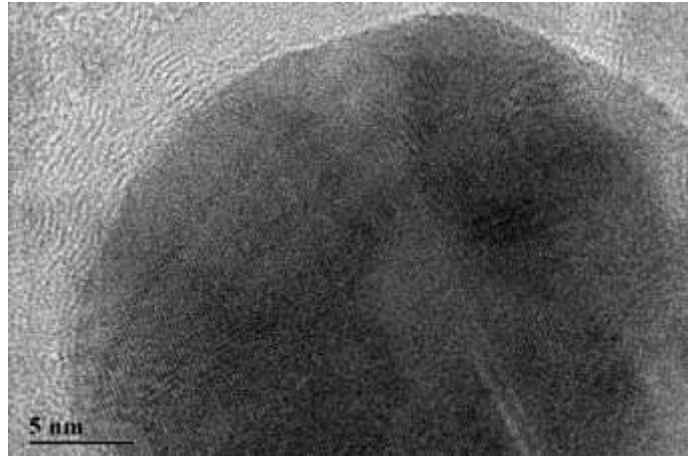
TEM 10% Pd/C Sigma-Aldrich (2#)

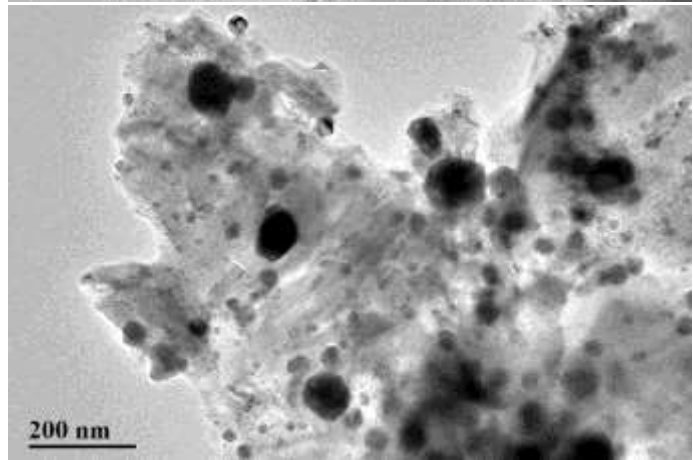
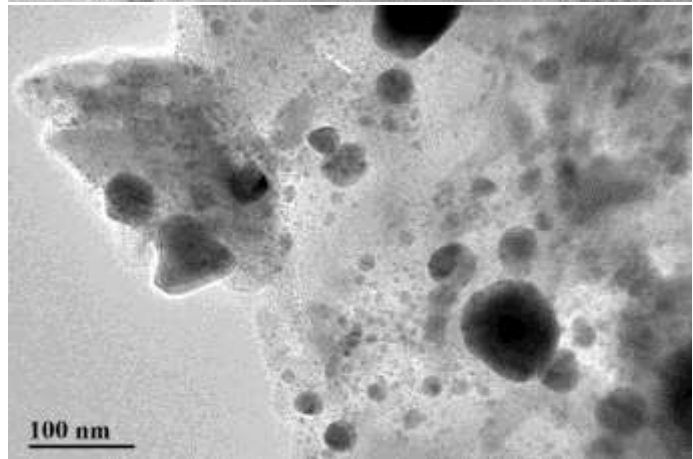
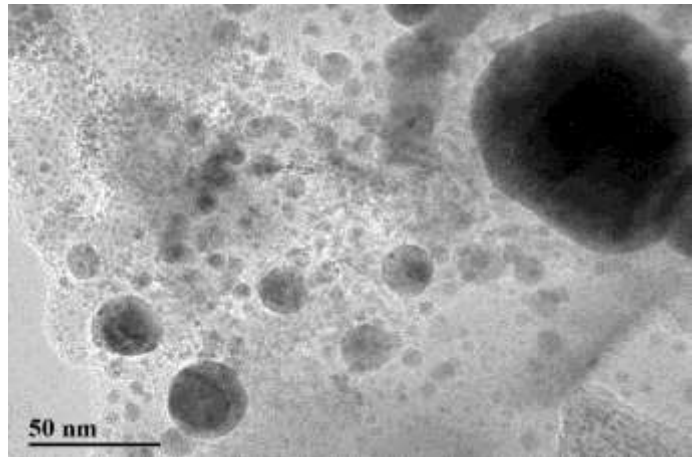




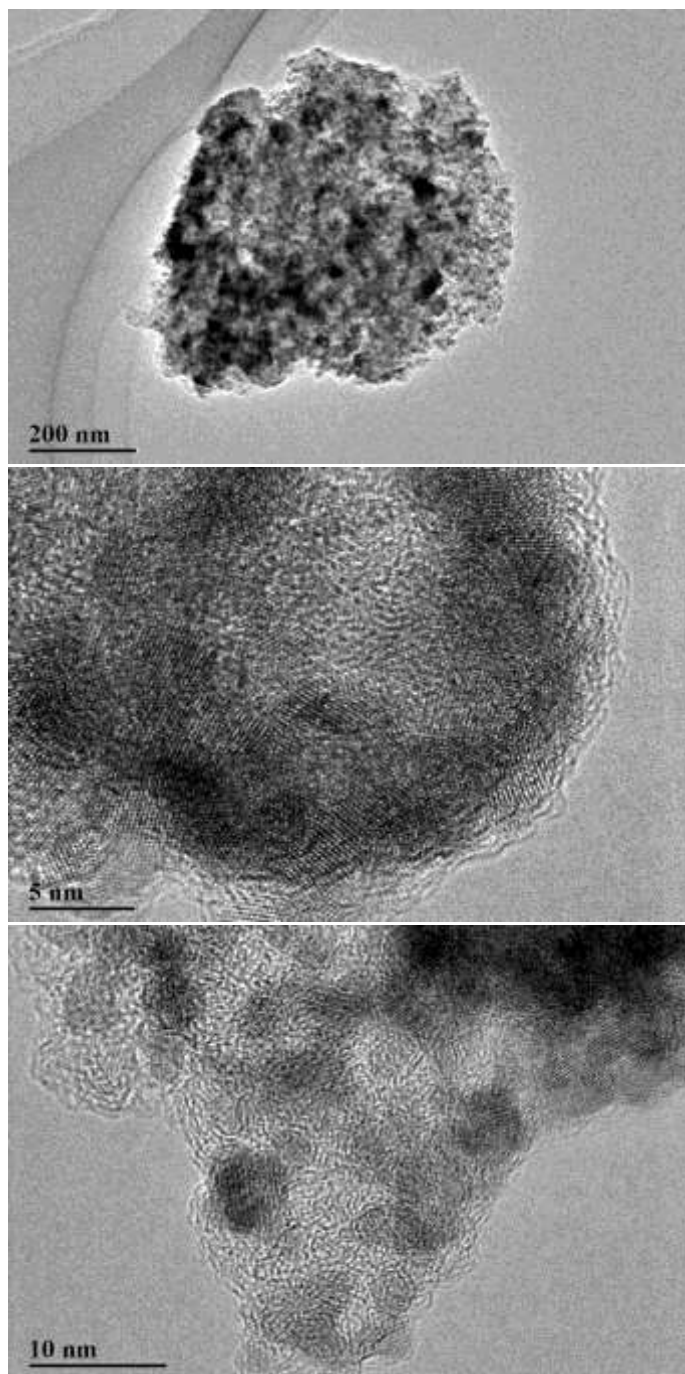


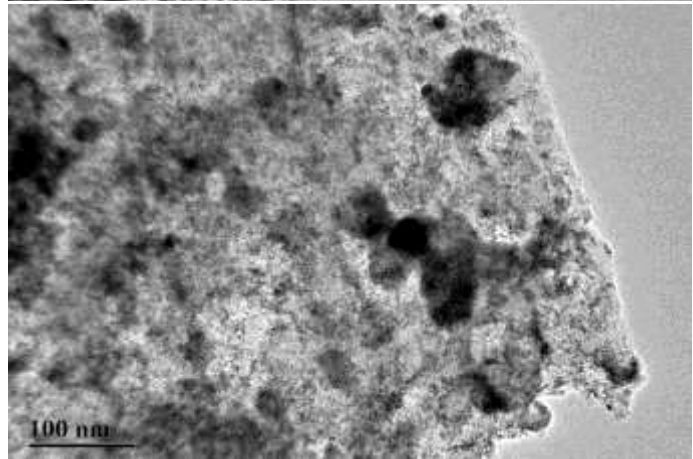
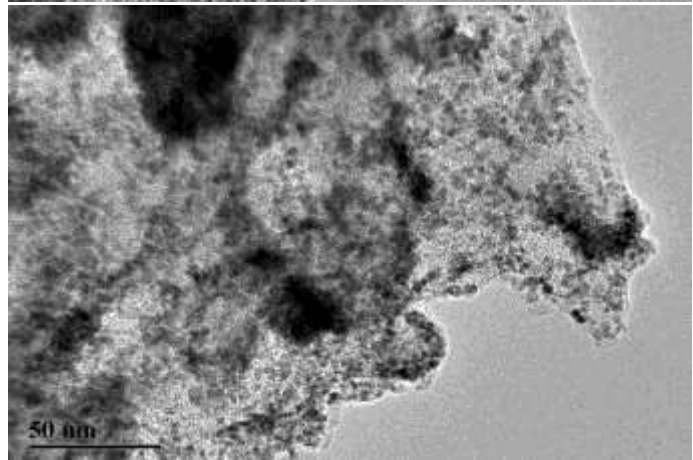
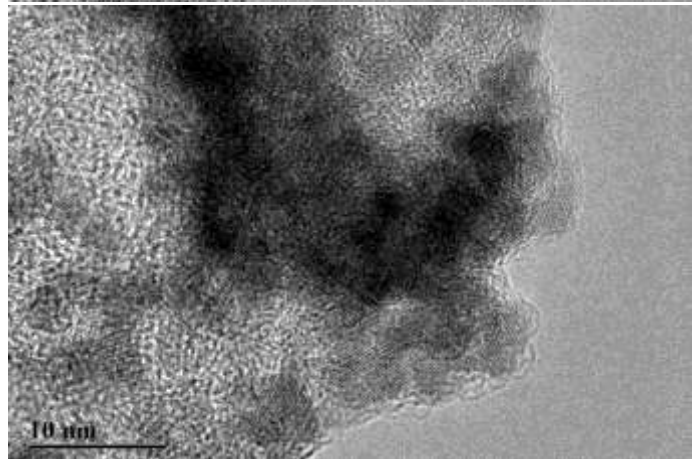
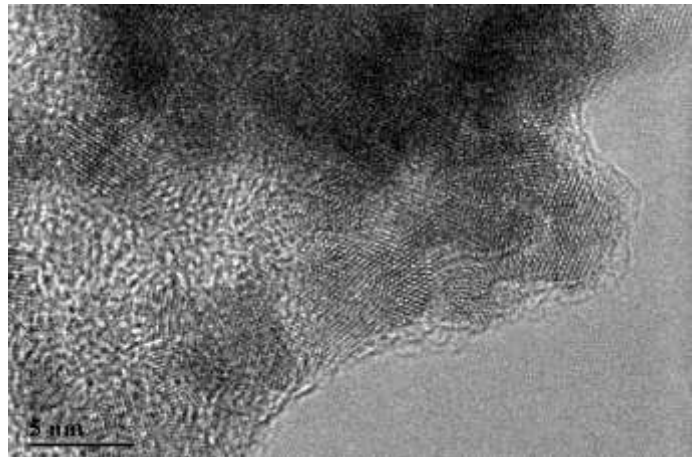


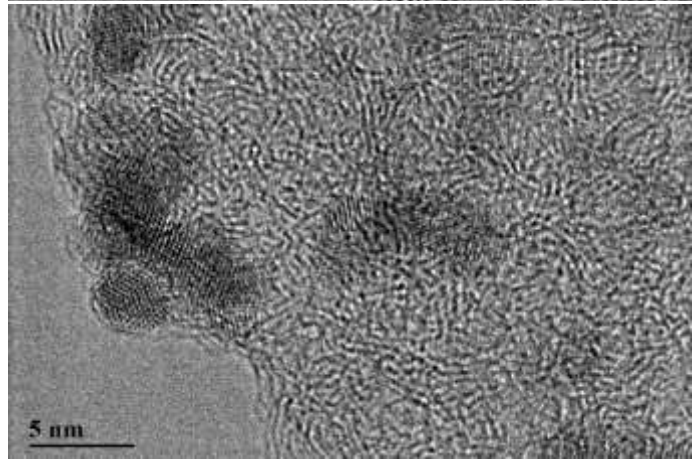
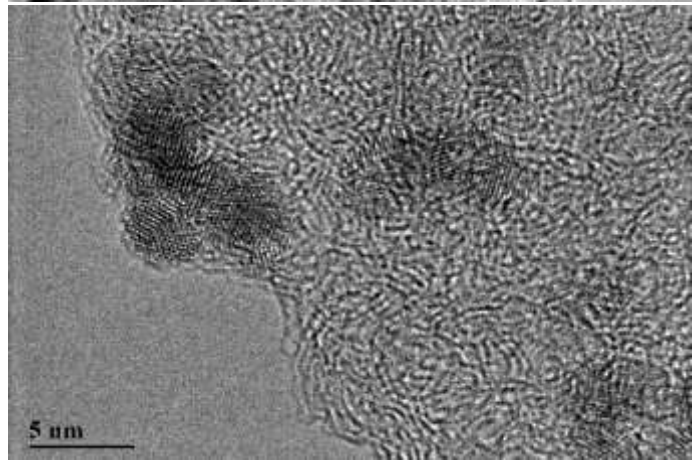
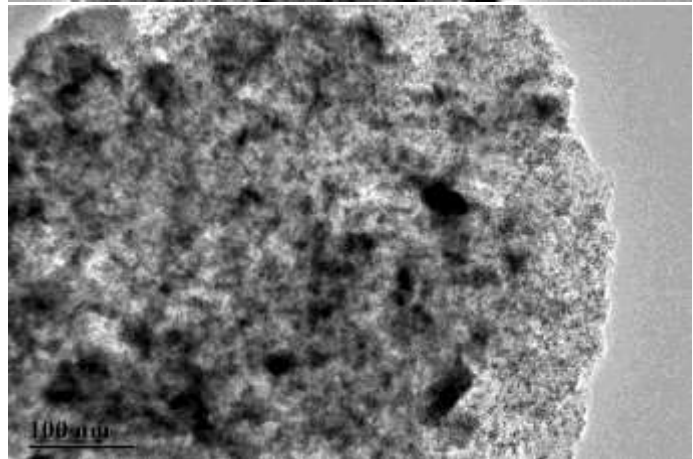
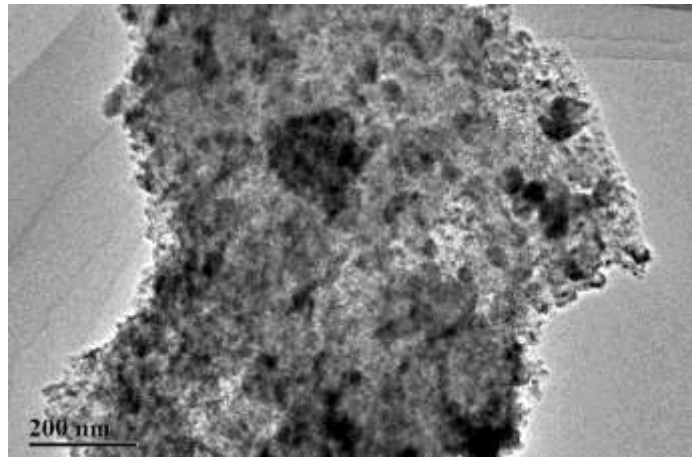


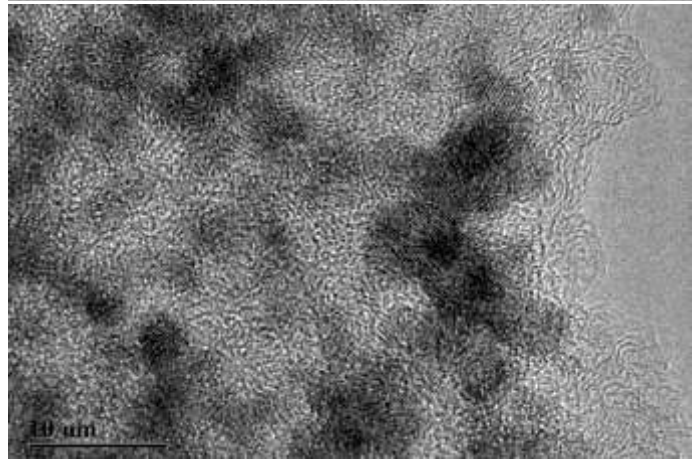
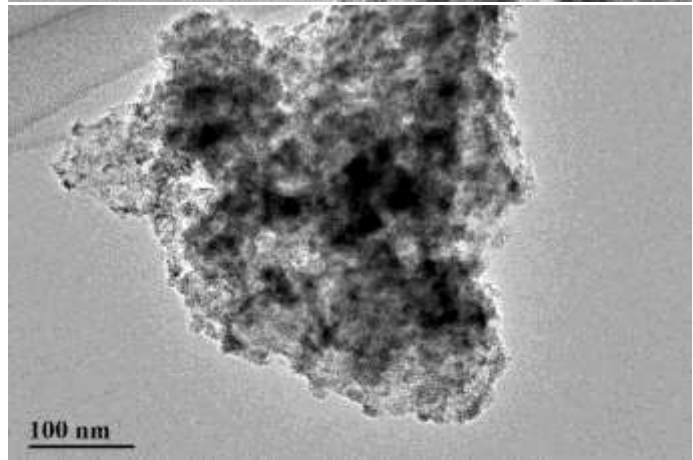
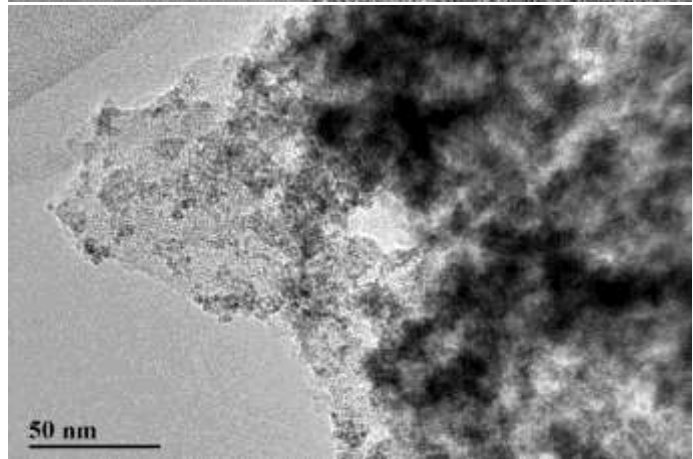
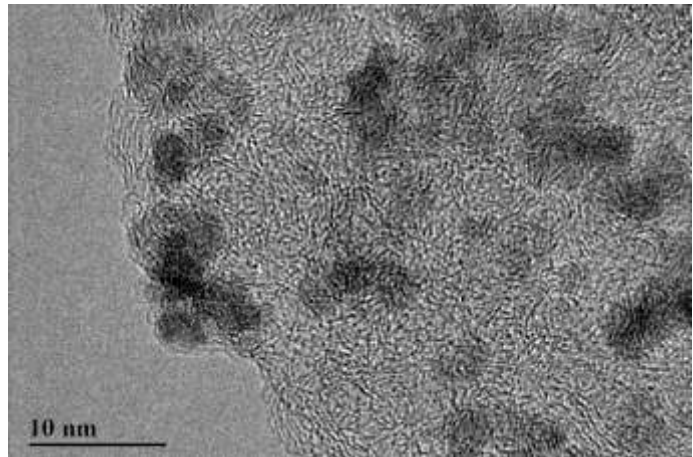


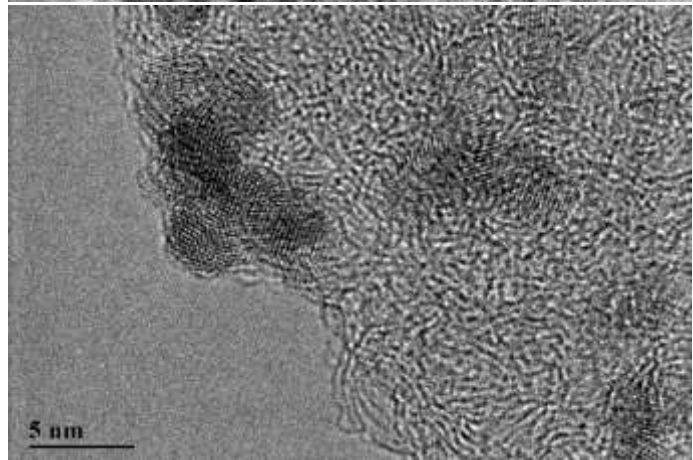
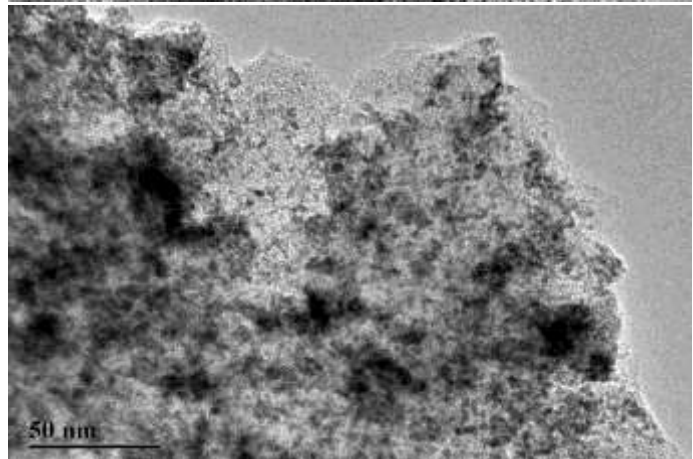
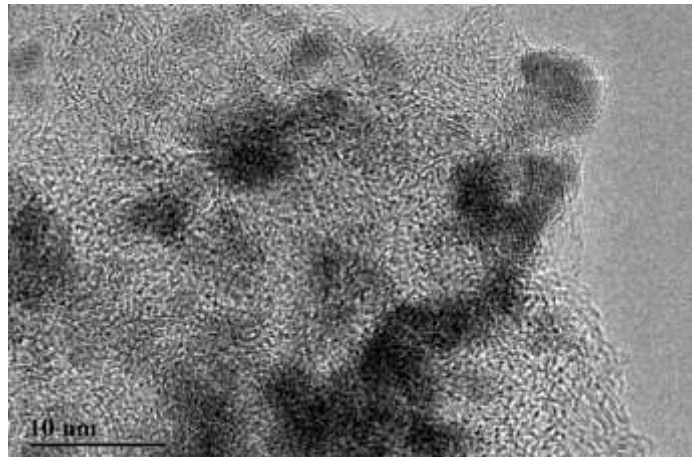
TEM of 20% Pd[OH]₂/C Sigma-Aldrich (3#).











Palladium: PDF 05-0681

Pattern: PDF 05-0681 Radiation: 1.54060 Quality: Star (*)

Formula		Pd		d	2θ	I fix	h	k	l
Name		Palladium		2.24600	40.115	101	1	1	1
Name (mineral)		Palladium, syn		1.94500	46.652	43	2	0	0
Name (common)				1.37600	68.085	26	2	2	0
				1.17300	82.096	25	3	1	1
				1.12320	86.598	9	2	2	2
				0.97230	104.792	4	4	0	0
				0.89240	119.351	14	3	3	1
				0.86970	124.677	12	4	2	0
Lattice:		Cubic		Mol. weight = 106.4					
S.G.:		Fm-3m (225)		Volume [CD] = 58.85					
				Dx =					
				Dm =					
				I/Icor = -1.000					
a =	3.88980	alpha =							
b =		beta =							
c =		gamma =							
a/b =	1.00000	Z =	4						
c/b =	1.00000								
<p>Color: Black Sample Source Or Locality: Sample from Johnson Matthey Company, Ltd Analysis: Spectroscopic analysis shows <0.1% Ag, Si; <0.01% Ca, Cu, Mg, Pt; 0.0001% Pb Temperature Of Data Collection: Pattern taken at 26 C Deleted By or Rejected By: Deleted by 46-1043; F#N is higher, Mayo 11/94</p>									
<p>Primary Reference Publication: Natl. Bur. Stand. (U.S.), Circ. 539 Detail: volume I, page 21 (1953) Authors: Swanson, Tetge.</p>									
Radiation:		CuKα1		Filter:		F			
Wavelength:		1.54060		d-spacing:					
SS/FOM:		55.9 (0.0179,8)							

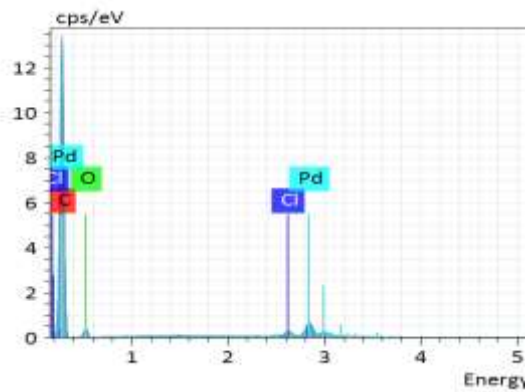
Palladium oxide: PDF 88-2434

Pattern: PDF 88-2434 Radiation: 1.54060 Quality: Calculated

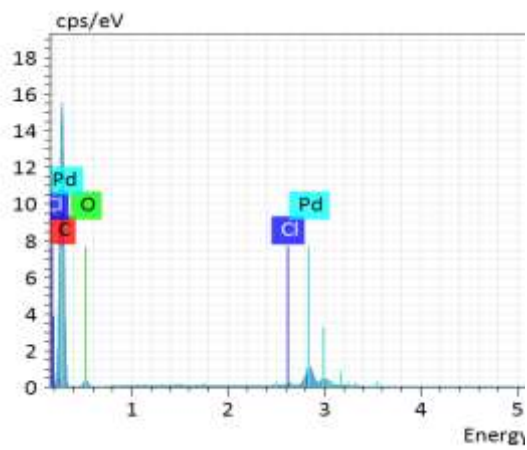
Formula Pd O Name Palladium Oxide Name (mineral) Name (common) Palladium oxide - HP		<table border="1"> <thead> <tr> <th>d</th> <th>2θ</th> <th>I fix</th> <th>h</th> <th>k</th> <th>l</th> </tr> </thead> <tbody> <tr><td>2.69150</td><td>33.261</td><td>515</td><td>0</td><td>0</td><td>2</td></tr> <tr><td>2.60849</td><td>34.352</td><td>999</td><td>1</td><td>0</td><td>1</td></tr> <tr><td>2.10859</td><td>42.854</td><td>438</td><td>1</td><td>1</td><td>0</td></tr> <tr><td>1.65987</td><td>55.300</td><td>342</td><td>1</td><td>1</td><td>2</td></tr> <tr><td>1.53746</td><td>60.135</td><td>148</td><td>1</td><td>0</td><td>3</td></tr> <tr><td>1.49100</td><td>62.213</td><td>108</td><td>2</td><td>0</td><td>0</td></tr> <tr><td>1.34575</td><td>69.835</td><td>34</td><td>0</td><td>0</td><td>4</td></tr> <tr><td>1.30425</td><td>72.400</td><td>117</td><td>2</td><td>0</td><td>2</td></tr> <tr><td>1.29446</td><td>73.036</td><td>144</td><td>2</td><td>1</td><td>1</td></tr> <tr><td>1.13440</td><td>85.537</td><td>62</td><td>1</td><td>1</td><td>4</td></tr> </tbody> </table>						d	2θ	I fix	h	k	l	2.69150	33.261	515	0	0	2	2.60849	34.352	999	1	0	1	2.10859	42.854	438	1	1	0	1.65987	55.300	342	1	1	2	1.53746	60.135	148	1	0	3	1.49100	62.213	108	2	0	0	1.34575	69.835	34	0	0	4	1.30425	72.400	117	2	0	2	1.29446	73.036	144	2	1	1	1.13440	85.537	62	1	1	4
d	2θ	I fix	h	k	l																																																																				
2.69150	33.261	515	0	0	2																																																																				
2.60849	34.352	999	1	0	1																																																																				
2.10859	42.854	438	1	1	0																																																																				
1.65987	55.300	342	1	1	2																																																																				
1.53746	60.135	148	1	0	3																																																																				
1.49100	62.213	108	2	0	0																																																																				
1.34575	69.835	34	0	0	4																																																																				
1.30425	72.400	117	2	0	2																																																																				
1.29446	73.036	144	2	1	1																																																																				
1.13440	85.537	62	1	1	4																																																																				
Lattice: Tetragonal S.G.: I4/mmm (139)		Mol. weight = 122.4 Volume [CD] = 47.87 Dx = Dm = I/floor = 10.080																																																																							
a = 2.98200 b = c = 5.38300 a/b = 1.00000 c/b = 1.80516	alpha = beta = gamma = Z = 2																																																																								
<p>ICSD Collection Code: 041617 Remark From ICSD/CSD: REM M Stable above 12 GPa, elongated rocksalt-type Remark From ICSD/CSD: REM M Cell at ambient pressure 3.042, 5.351, cp. 24692 Remark From ICSD/CSD: REM M PDF 43-1024 Test From ICSD: No R value given Test From ICSD: At least one TF missing Remark From ICSD/CSD: REM PRE Mentioned Article Title: Structural behavior of palladium(II) oxide and a palladium suboxide at high pressure: an energy-dispersive X-raydiffraction study Wyckoff Sequence: b a (I4/MMM) ANX: AX</p>																																																																									
<p>Structure Publication: Phys. Rev. B: Condens. Matter Detail: volume 52, page 9259 (1995) Authors: Christy, A.G., Clark, S.M. Primary Reference Publication: Calculated from ICSD using POWD-12++</p>																																																																									
Radiation: CuKα1 Wavelength: 1.54060 SS/FOM: 999.9 (0,10)		Filter: Not specified d-spacing:																																																																							

SEM-EDS

1# Sample, Pd/C STREM Chemicals: Cl=1.7%_{wt}



2# Sample, Pd/C Sigma-Aldrich: Cl=0.8%_{wt}



3# Sample, Pd[OH]₂ /C Sigma-Aldrich: negligible

