



Supporting Information

for *Adv. Sci.*, DOI: 10.1002/advs.202100498

High-performance Ammonium Cobalt Phosphate Nanosheet Electrocatalyst for Alkaline Saline Water Oxidation

Zhongxin Song^a, Kaixi (Cathy) Wang^{bc}, Qian Sun^b, Lei Zhang^b, Junjie Li^b, Dingjiu Li^a, Pok-Wai Sze^a, Yue Liang^a, Xueliang Sun^b, Xian-Zhu Fu^a, Jing-Li Luo^{a,*}

Supporting Information

High-performance Ammonium Cobalt Phosphate Nanosheet Electrocatalyst for Alkaline Saline Water Oxidation

Zhongxin Song ^a, Kaixi (Cathy) Wang ^{bc}, Qian Sun ^b, Lei Zhang ^b, Junjie Li ^b, Dingjiu Li ^a, Pok-Wai Sze ^a, Yue Liang ^a, Xueliang Sun ^b, Xian-Zhu Fu ^a, Jing-Li Luo ^{a,*}

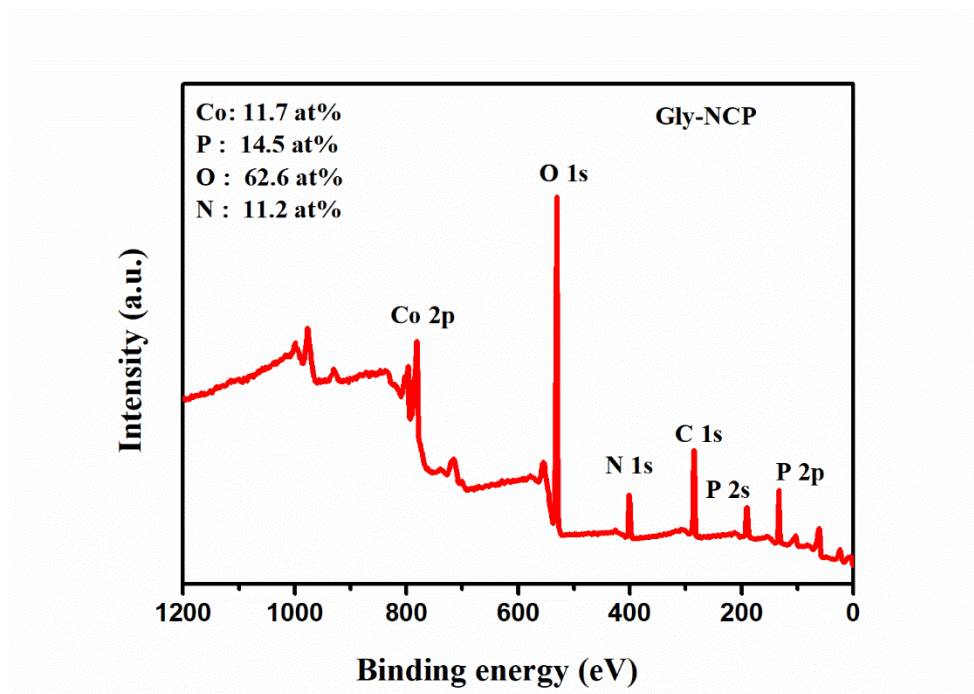


Figure S1. The XPS spectrum and element composition of as-prepared Gly-NCP.

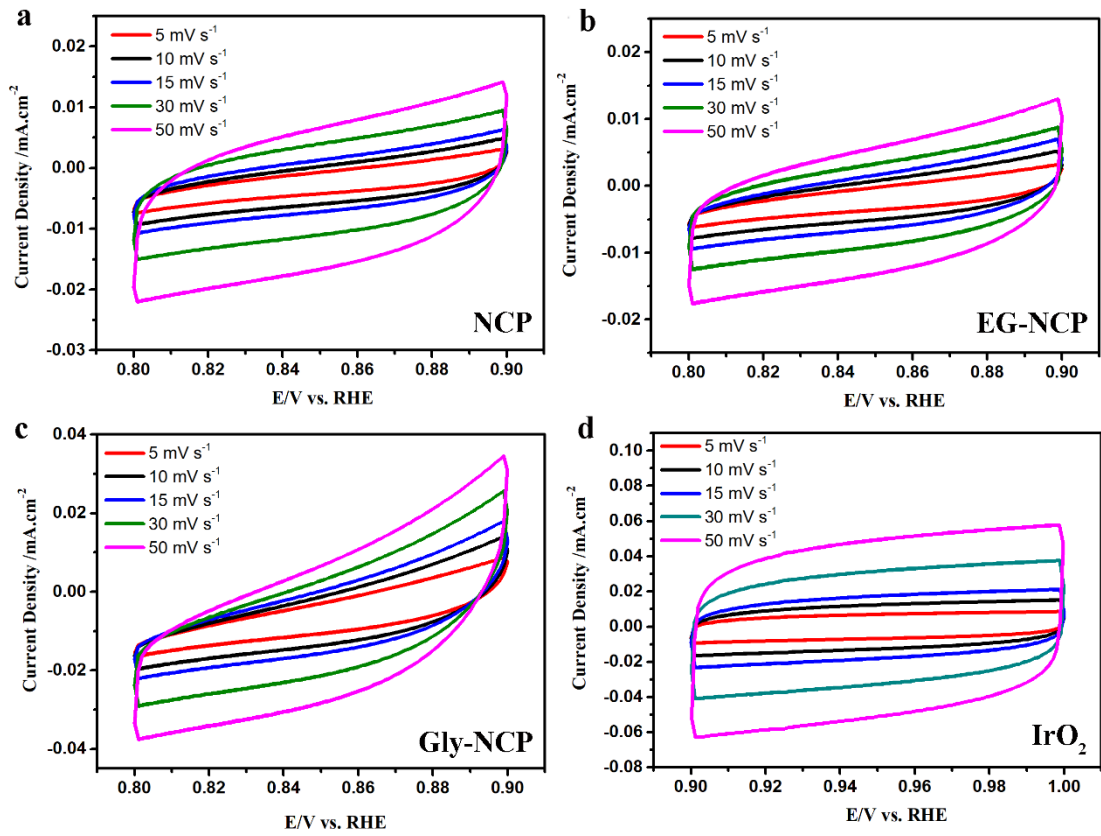


Figure S2. The CV curves at different scan rates for NCP, EG-NCP, Gly-NCP nanosheets and the reference of IrO_2 .

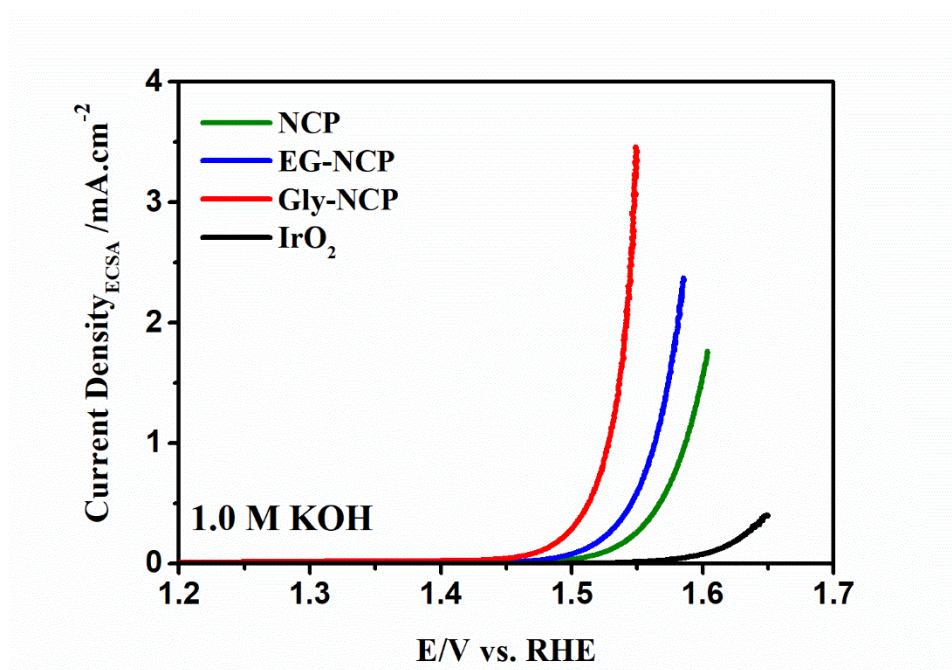


Figure S3. The ECSA normalized LSV curves for NCP, EG-NCP, Gly-NCP nanosheets (scan rate: 10 mVs⁻¹)

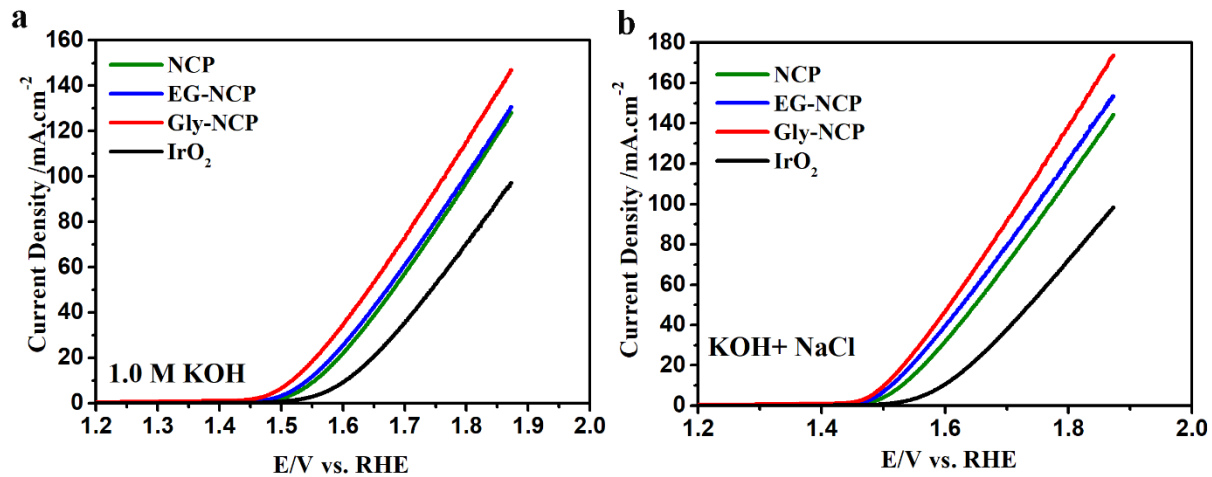


Figure S4. The LSV curves for NCP, EG-NCP, Gly-NCP nanosheets in (a) 1.0 M KOH and (b) 1.0 M KOH+0.5 M NaCl before jR correction (scan rate: 10 mV s^{-1}).

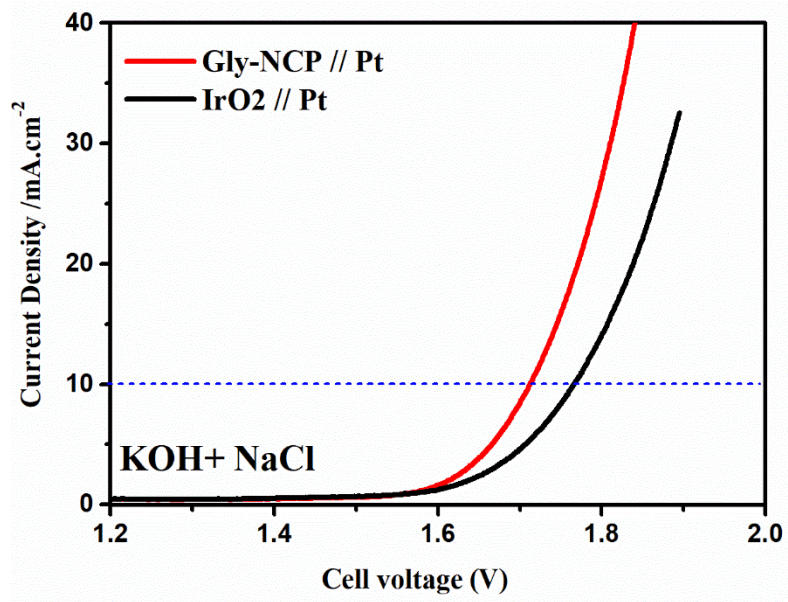


Figure S5. The LSV curve for overall alkaline saline water electrolysis of Gly-NCP//Pt and IrO₂//Pt under two electrode system (scan rate: 10 mVs⁻¹).

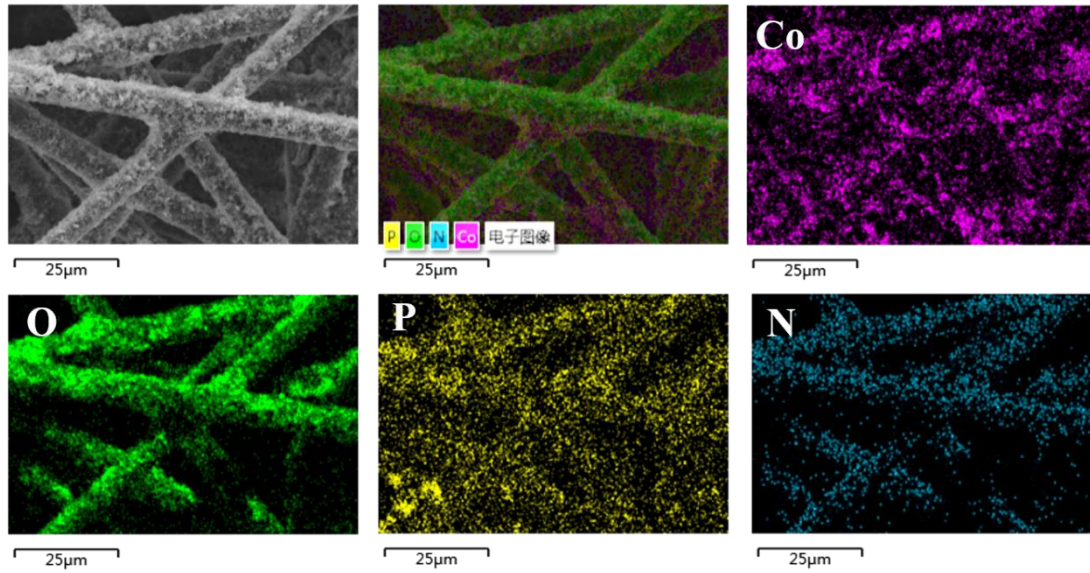


Figure S6. The SEM image and EDS maps of Gly-NCP electrode after 20 hours of stability test in 1.0 M KOH+0.5 M NaCl.

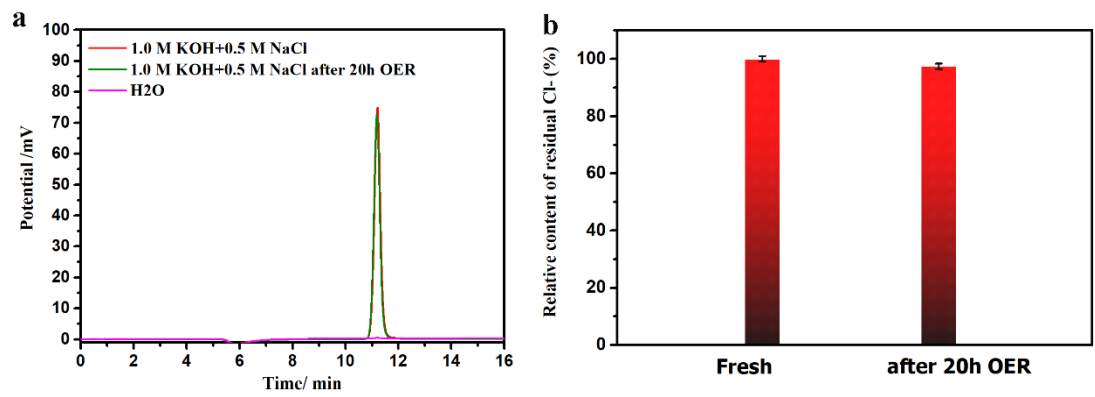


Figure S7. The Ion-Chromatography traces of diluted electrolyte (a) and relative content of residual Cl⁻ in KOH+NaCl electrolyte (b) before and after 20 h OER test catalyzed by Gly-NCP at 10 mA cm⁻².

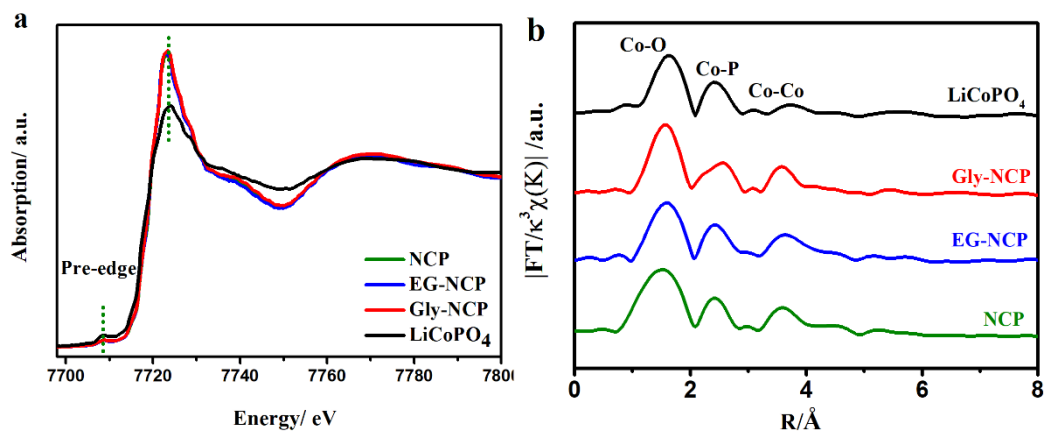


Figure S8. (a) XANES spectra and (b) FT-EXAFS spectra of as-prepared NCP, EG-NCP, Gly-NCP nanosheets and the reference of LiCoPO₄.

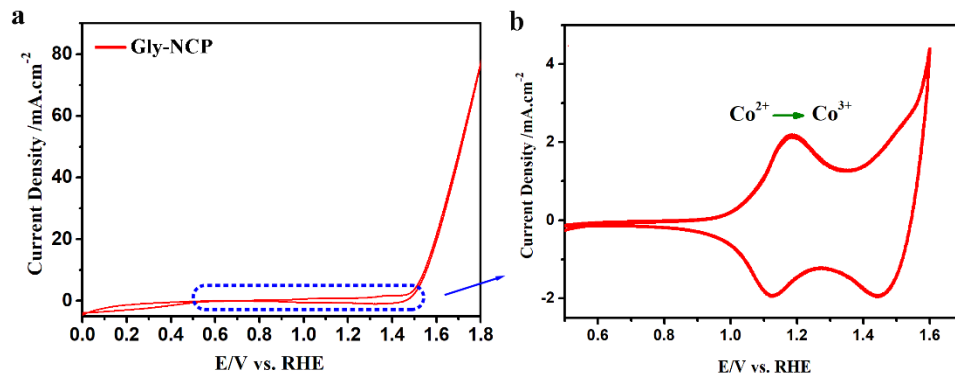


Figure S9. The CV curve of Gly-NCP electrode in 1.0 M KOH+0.5 M NaCl (scan rate: 50 mVs⁻¹).

Table S1. Comparison of electrocatalytic activity towards seawater oxidation for recently reported electrocatalysts

| Electrocatalysts | Electrolyte | Overpotential | Tafel slope mVdec ⁻¹ | Stability | Reference |
|---|----------------------------|-----------------------------------|------------------------------------|--|-----------|
| 3D core-shell NiMoN@NiFeN | 1.0 M KOH + 0.5 M NaCl | 286 mV @ 100 mA cm ⁻² | - | Current increase <10% after 100 h | [1] |
| Na ₂ Co _{1-x} Fe _x P ₂ O ₇ | 0.5 M NaCl + 0.1 M KOH | 285 mV @ 100 mA cm ⁻² | 56 | 40 mV increase after 100 h | [2] |
| Ni ₂ P-Fe ₂ P | 1 M KOH seawater | 305 mV @ 100 mA cm ⁻² | - | 36 h stable at 100 mA cm ⁻² | [3] |
| NiFe/NiS _x -Ni foam | 1 M KOH+ 0.5 M NaCl | 380 mV @ 1500 mA cm ⁻² | - | 500 h stable at 400-1000 mA cm ⁻² | [4] |
| S-(Ni,Fe)OOH | 1 M KOH+ 0.5 M NaCl | 278 mV @ 100 mA cm ⁻² | 48.9 | 100 h stable at 100 mA cm ⁻² | [5] |
| NiFe-LDH | 0.1 M KOH+0.5m NaCl | 359 mV @ 10 mA cm ⁻² | 50 | 480 mV increase after 2h operation | [6] |
| NiCo-DEA | seawater | Onset potential of 1.31 V | 51 | Keep 96% current after 8h at 1.5 V | [7] |
| NiNS | Overall seawater splitting | 48.3 mA cm ⁻² at 1.8 V | 112 | Stable during 12 h operation | [8] |
| Gly-NCP | 1.0 M KOH + 0.5 M NaCl | 268 mV @ 100 mA cm ⁻² | 39 | 13 mV increase after 20h operation | This work |

References

- [1] L. Yu, Q. Zhu, S. W. Song, B. McElhenny, D. Z. Wang, C. Z. Wu, Z. J. Qin, J. M. Bao, Y. Yu, S. Chen, Z. F. Ren, *Nat. Commun.* **2019**, 10, 10.
- [2] H. J. Song, H. Yoon, B. Ju, D. Y. Lee, D. W. Kim, *Acs Catalysis* **2020**, 10, 702.
- [3] L. Wu, L. Yu, F. Zhang, B. McElhenny, D. Luo, A. Karim, S. Chen, Z. Ren, *Adv. Funct. Mater.* **2021**, 31, 2006484.
- [4] Y. Kuang, M. J. Kenney, Y. T. Meng, W. H. Hung, Y. J. Liu, J. E. Huang, R. Prasanna, P. S. Li, Y. P. Li, L. Wang, M. C. Lin, M. D. McGehee, X. M. Sun, H. J. Dai, *Proc. Natl. Acad. Sci. U. S. A.* **2019**, 116, 6624.
- [5] L. Yu, L. Wu, B. McElhenny, S. Song, D. Luo, F. Zhang, Y. Yu, S. Chen, Z. Ren, *Energy Environ. Sci.* **2020**, 13, 3439.
- [6] F. Dionigi, T. Reier, Z. Pawolek, M. Gliech, P. Strasser, *ChemSusChem* **2016**, 9, 962.
- [7] J. J. Zheng, *Electrochim. Acta* **2017**, 247, 381.
- [8] Y. Q. Zhao, B. Jin, A. Vasileff, Y. Jiao, S. Z. Qiao, *Journal of Materials Chemistry A* **2019**, 7, 8117.