

Supplementary Information for

Deactivation blocks proton pathways in the mitochondrial complex I

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SI Methods

FEP calculations

Alchemical free energy perturbation (FEP) calculations were used to study the hydration free energy of the gating region in both *active* and *deactive* states, based on the last snapshots of MD simulations S19 and S23 (Table S1), respectively. To enhance the sampling, the simulation system comprised subunits ND1, ND2 (residues 1-150), ND3, ND4L and ND6, and all water molecules within 3 Å of these subunits. Water molecules close to the cavity between Asp66^{ND3} and Glu34^{ND4L} were removed to prevent interference with the created/annihilated water wire, comprising five water molecules between Asp66 NDS and Glu34 $NDAL$, as observed in the unbiased MD simulations. Asp66 N^{D3} was modeled protonated, mimicking a state in which the water wire would enable pT to Glu34ND4L. The system was embedded in a POPC/POPE/cardiolipin lipid membrane, re-solvated in TIP3P water and neutralized with NaCl to a concentration of 150 mM by adding 135 Na⁺ and 65 Cl- ions. The FEP-model comprised *ca*. 120,000 atoms and was relaxed for 20 ns with initial harmonic constraints with a force constant of 3 kcal mol⁻¹ \AA ² acting on the position of the all protein atoms and water molecules. The harmonic constraints on the protein were thereafter removed, followed by equilibration and FEP simulations. External forces were applied to prevent new water molecules from entering into the protein interior. All simulation parameters were analogous to those used in the complete complex I system setup (see Main text methods), unless otherwise stated.

The FEP calculations were conducted by applying flat-bottom distance constraints between the introduced water molecules as well as $\overrightarrow{Asp66}^{N\overrightarrow{D}3}$ and Glu34^{ND4L} using the Colvars (1) module of NAMD. Bulk water molecules were prevented from entering into the protein interior by using a harmonic restraint introduced on the water oxygen positions with a force constant of 0.02 kcal mol- 1 Å-2 . No constraints were applied on the protein atoms during production simulations to allow for the structural relaxation of the cavity upon introduction of the water chain between Asp66 N_{D3} and Glu34ND4L. Sampling was done in 20 equidistant windows from λ=0 (wire completely decoupled) to λ=1 (wire fully coupled) in forward and backward directions with 1 ns equilibration and 10 ns production simulations for each window, with a total sampling for 220 ns for each FEP simulation. The convergence was further tested using two 44 ns replicas with 0.2 ns equilibration and 2 ns production per window. Benchmarking simulations (0.5 ns per window) were conducted to probe the effect of splitting vdW and electrostatics and the number of windows used. The FEP simulations included in total 2.5 μs MD sampling. All FEP simulations yielded similar results when compared to the full-scale simulations indicating robust results, with a standard deviation of *ca*. 1 kcal mol-1 for the statistical error. Estimation of the accuracy of the non-polarizable CHARMM36 force field in prediction of hydration free energies is outside the scope of the present work. To probe the transfer free energy from bulk to the protein interior, five water molecules were removed/created in a 50 x 50 x 50 Å water box, with a 150 mM NaCl concentration (11 Na⁺/Cl⁻) using the same FEP-protocol as for the complex I models, but without addition of constraints. All FEP simulations were conducted using the NAMD-FEP implementation and analyzed with the ParseFEP plugin (2) in VMD employing the Bennet acceptance ratio (BAR) method (3).

MD simulations with a hydronium species in the water chain

Simulations with an explicit H_3O^+ species were started from different snapshots of the deactive state simulation S23 with Asp66^{ND3} in a deprotonated state (see Table S1) and the H₃O⁺ modeled next to the Asp66^{ND3} carboxylate in the cavity connecting to Glu34^{ND4L}. These states were propagated for 110 ns and the water content of the cavity was subsequently evaluated in a cumulative manner. Classical force field parameters for the hydronium ion were derived at the B3LYP-D3/def2-TZVP level using Turbomole 7.5 (15), converting the molecular Hessian into force constants using Hess2FF (26), and applying the RESP procedure (27) for calculation of atomic point charges. Lennard-Jones parameters were taken from the TIP3P water parameters implemented in the CHARMM36 (28) force field. The full set of parameters is reported in SI Appendix, Table S9.

DFT benchmarking and reaction pathway calculations

The quantum chemical calculations were benchmarked at the DFT level using B3LYP-D3 (4,5,6), CAM-B3LYP-D3 (6,7), CAMh-B3LYP-D3 (6,8), ω B97XD (6,9), TPSSh-D3 (6,10), and by using the correlated *ab initio* random phase approximation (RPA) (11) and domain-based local pair natural orbital coupled cluster with single-double and perturbative triples (DLPNO-CCSD(T)) (12). def2- TZVP basis sets (13) were used for the DFT calculations and aug-cc-pVTZ (14) basis sets at the correlated levels. Single point energy calculations were performed for model systems comprising 27, 131, and 381 atoms, constructed from DFT models 1-3 (see Table S4), and involving proton transfer reactions between carboxylates (Asp66^{ND3}, Glu34^{ND4L}, Glu70^{ND4L}) and water molecules. The DFT calculations of the large model systems were performed using an implicit polarizable medium with ε=4, whereas the calculations for small and intermediate models were done in gas phase. The benchmarking calculations were performed using TURBOMOLE v7.4-7.5 (15) for the DFT and RPA calculations, and ORCA v4.2 (16) for the DLPNO-CCSD(T) calculations. See Table S7 for the quantum chemical benchmarking results.

Reaction pathways of the proton transfer reactions in the DFT models were optimized using a minimum energy reaction pathway optimization approach (29) that resembles the zero-temperature string method. To this end, the initial guess of the reaction pathway was constructed by optimization of the structures with the proton on the donor and acceptor states, followed by interpolating 19 intermediate structures between the two states using the linear synchronous transit method. The reaction pathway was refined by constraining equally spaced intermediate structures with a quadratic potential (29) until the root-mean-square difference of the gradient varied less than 10⁻⁴ au in subsequent optimization steps. Numerical estimation of the Hessian showed one imaginary frequency ($>$ 500 cm⁻¹) for the approximate transition state structure. See Ref. (29) for further methodological details of the reaction pathway optimization approach. The overall reaction process predicted by this approximate reaction pathway method resembles that obtained from our QM/MM umbrella sampling calculations, along the geometric reaction coordinate *R* (see methods).

Fig. S1. Molecular models. A, MD simulation setup of the mitochondrial complex I. **B**, QM and QM/MM models for probing proton transfer at the ND3/ND4L/ND6/ND2 interface. **C**, QM region used in QM/MM MD simulations of binding site 2, with residues shown in licorice, modeled at the DFT level (see *Methods*). **D**, models used in PB/MC-electrostatic calculations. **E**, models used for exploration of the Q10 dynamics with atomistic- and coarse-grained MD.

Fig. S2. Conformational changes linked to complex I deactivation. A, Comparison of dynamics inferred from MD simulations and B-factors obtained from cryoEM data. **B,** Conformational changes around the NDUFA10/5 subunit linked to the A/D transition in MD simulations. **C**, **D**, ion pairs at the NDUFA10/5 interface undergo conformational rearrangement during the MD simulations of the respective states.

Fig. S3. Structural characterization of the deactive model. A, Cross-correlation between calculated density during the MDFF simulations and the cryoEM density map of the *active* (EMDB: 4345) and *deactive* state (EMDB: 4356). **B,** The *root-mean-square-deviation* (RMSD) of the *deactive* state model during MDFF relative to experimentally refined structures of the D (PDB ID: 6g72) and A (PDB ID: 6g2j) states. The RMSD was calculated on a backbone level for the resolved regions. **C,** Subunit-wise RMSD relative of the D state model relative to cryoEM-refined models of the D (PDB ID: 6g72) and A (PDB ID: 6g2j) states. **D** and **E,** RMSD difference between the D state model to the cryoEM-refined models of the D (PDB ID: 6g72) and A (PDB ID: 6g2j) states. Unobserved regions in either cryoEM models are colored green. **F**, Bending angles (α) and dihedral angles (ϕ) between the hydrophilic and membrane domains in the *active* and *deactive* states. The angles were determined using the principal axes of a subset of specific subunits for the respective domains (hydrophilic: NDUFV1, NDUFV2, ND1; membrane: ND5, ND4, ND2).

Fig. S4. Comparison of proton pathways in the *active* **and** *deactive* **state MD simulations. A**, Water dynamics (red spheres) shown as an ensemble average of the 2 μ s MD simulations of the *active* state (*top*, simulations S1+S2) and *deactive* state (*bottom*, simulations S6+S7). **B**, Snapshot of individual buried water molecules (red spheres) after 1 µs MD simulation (*top: active* state simulations S1+S2; *bottom*: *deactive* state, simulations S6+S7).

Fig. S5. MD simulations exploring proton transfer along the ND3/ND4L/ND6 region. The simulations show the effect of modeling different protonation states in Asp66 N^{D3} , Glu34ND4L/Tyr59ND6, and Glu70ND4L, mimicking proton transfer via the region in the *active* (left, A-D) and *deactive* (right, F-I) state simulations. **E**, **J**, Hydration average of the tunnels from MD simulations in the *active* (red) and *deactive* (blue) states with Asp66^{ND3} protonated. The shown diameter is proportional to the water content (see *Methods*). **K**, Hydration fraction along the tunnels connecting the acidic residues along the ND3/ND4L/ND6 gating region when $Glu34^{ND4L}$ is protonated. The tunnel distances correspond to the beads shown in panel K. Functional residues along the tunnel are indicated with differently shaped markers (diamond: D66^{ND3}, circle: E34^{ND4L}, hexagon: E70^{ND4L}, pentagon: E34^{ND2}, (deactive only) square: M63^{ND6}). The mean position is indicated with a larger size marker. **L-O**, Hydration dynamics during 110 ns trajectories with a classically modelled H_3O^+ species, placed next to Asp66^{ND3} (panels L, M) or Glu34^{ND4L} (panels N, O) in the deactive state. The cavity, marked with a black frame, remains dry on the simulation timescale in the studied states, suggesting that the thermodynamic cost of hydrating this region could be high in the deactive state.

Fig. S6. Water cluster analysis and hydration dynamics. A, B, Top/side view of water cluster analysis for the ND1/ND3/ND4L/ND6 region obtained based on 1000 ns MD simulations of the *active* and *deactive* states (simulations S1 and S6, Table S1). The water shown are high probability water molecules identified in the cluster analysis (25). **C,** Hydration dynamics in the ND1/ND3/ND4L/ND6 region during *active* and *deactive* state MD simulations (simulations S1 and S6, Table S1). **D,** Hydration dynamics in the ND1/ND3/ND4L/ND6 region induced by shifting the proton during consecutive 100 ns steps along the Asp66^{ND3}, Glu34^{ND4L}, Glu70^{ND4L} (simulations S18-S21, Table S1) The simulations were initiated from 1000 ns frame shown in panel C.

Fig. S7. Free energy perturbation (FEP) calculations. A, Free energy for introduction of a water wire (*exnihilation*) in the ND3/ND4L/ND6 region in the *active* and *deactive* states, and water removal (*annihilation*) from the bulk, using alchemical FEP. **B**, Free energy profiles from λ=0 (fully decoupled) to $\lambda = 1$ (fully coupled) for all replicas. Note that intermediate λ values represent unphysical states. **C**, Decomposition of the free energy into vdW and electrostatic contributions. **D**, Benchmarking simulations to probe the effect of splitting the vdW and electrostatic coupling/decoupling into separate simulations and the dependence on number of sampling windows. The benchmarking calculations were performed on the *deactive* state model. **E**, Backbone RMSD of TM3-ND6 for the *active* and *deactive* FEP simulations (λ=1). The RMSD of each simulation is relative to the first frame and calculated by aligning the whole ND6 subunit, indicating small local rearrangement within the TM3^{ND6} region. **F**, Comparison of the last snapshot of the *active* and *deactive* state FEP simulations (pale colors) with the last snapshots from simulations S19 and S23 (dark colors), respectively. TM3^{ND6} shows small conformational changes, whereas larger conformational changes could be limited by the FEP simulations timescale.

Fig. S8. Convergence of hybrid QM/MM free energy calculations and sampling of reaction coordinates. A, definition of reaction coordinates R_1 and R_2 used in the QM/MM umbrella sampling (US) simulations. **B, C, D,** convergence of QM/MM US calculations and overlap of reaction coordinates in the individual simulations. **E, F, G,** Sampled proton transfer distances in the QM/MM US simulations during the respective simulations shown in **B-D,** suggesting that the proton is transferred in a Grotthuss-type transfer mechanism (sampling along the diagonal). Results from DFT reaction pathway optimizations are shown as black dots. **H,** effect of modeling ND2 ion-pair opening in the DFT cluster models by conformational and/or protonation changes.

Fig. S9. Disease related mutations in the mitochondrial complex I. A, location of mitochondrial disease related mutations (residues marked in red) (22). **B,** Closeup of disease related mutations near the proton pathway at the ND3/ND4L/ND6 interface, and **C**, *in silico* models of the mutations. Introduced substitutions form contacts with water molecules or proton donor/acceptor groups within the wire. F41C^{ND4L} could lead to a formation of a disulfide bridge with Cys40^{ND6}, the backbone of which is located <10 Å in the wild type structure. See also Table S6. **D-G**, introduced single point mutations: **D**, E34DND4L, **E**, E34AND4L, **F**, E34QND4L, **G**, E70AND4L perturb the proton pathways at the ND6/ND4L interface. **H, I**, effect of single point mutations of key residues on complex I activity. The experimental data was obtained from Refs. (23,24).

Fig. S10. Q10 dynamics explored by atomistic and coarse-grained MD. A, hydrogen-bonding binding mode, with the Q headgroup forming contacts with Tyr108NDUFS2 and His59NDUFS2. **B**, interaction energy decomposition of the binding energy contributions (in kcal mol-1). **C**, Distribution of Q₁₀-Tyr108^{NDUFS2} distances explored from classical (simulations S62-S77) and coarse-grained MD simulations (simulations S78-S87). The Q starting positions in the simulations are indicated with an asterisk. **D**, Dynamics of Q₁₀ explored in independent MD (in blue, simulations S62, S53, S70, S71) and CGMD (simulations S78, S79) trajectories.

Fig. S11. Dynamics of Q₁₀ in the membrane-bound binding region. A, Selection of key residues that form contact with Q10 during the *active*/*deactive* state simulations **B,** Distance plot between Q/QH2 and the selected residues. The distances were measured from Glu(CD)/Asp(CG)/Arg(CZ)/Tyr(OH) to the closest carbonyl oxygen of Q10.

Fig. S12. Conformational changes around the Q tunnel linked to the A/D transition. A, unresolved loop regions/structural differences in cryoEM structures around ND1, ND3, NDUFS2, and NDUFS7 of the mammalian complex I (marked with arrows). **B, C,** Conformational dynamics around ND1, ND3, NDUFS2, and NDUFS7 in the *active* (subunit specific color) and the *deactive* (light brown) states MD simulations, with **B**, showing the Q-bound simulations (simulation S1/S6, Table S1), and **C,** showing the *apo*-state simulations (simulation S12, S14, Table S1).

Fig. S13. Conformational changes around TM3ND3/ND6 and the ND3/ND4L/ND6 gating region in **A**, *Thermus thermophilus* (purple, Nqo11/Nqo10, PDB ID: 4HEA(17)), **B**, *Thermosynechococcus elongatus* (green, NdhE/NdhG, PDB ID: 6TJV (18)), **C**, *Yarrowia lipolytica* (blue, ND4L/ND6, PDB ID: 6RFR (19)), and **D**, mouse complex I *active* (red) and *deactive* (pale) models derived in this study.

Table S2. Non-standard protonation states determined based on PBE/MC calculations (see main text *Methods*). HSE – ε-protonated (neutral) His, HSD - δ-protonated (neutral) His, HSP – ε/δprotonated (charged) His.

Table S3. Secondary structure restraints used for modeling the D state during MDFF.

Additional restraints centered on the MDFF setup, for modeling the D state.

Table S4. Overview of QM/MM MD and QM/MM free energy simulations. QM/MM model 1 – QM region shown in Fig. S1C; QM/MM model 2 – QM region in Fig. S1B. QM models 3-5 were constructed from QM models 1-2 for the DFT benchmarking (see SI Methods).

Table S5. Multiple sequence alignment (20) of ND6 (top) and ND4L (bottom) from different species.

Rickettsia prowa Arabidopsis tha Paracoccus denitrif Rhodobacter capsu Sus se Bos to Homo sa Petromyzon mai Mycobacterium tubercu Streptomyces coelic Yarrowia lipoly Neurospora ci Drosophila melanogo Aquifex aeo Caenorhabditis ele Rhizobium leguminos Escherichia Escherichia Shigella fle. Pseudomonas aerug Buchnera aphid Buchnera aphid Buchnera aphid

Mus muse Escherichia Thermus thermonl

Table S6. Disease related mutations of subunits ND1/ND3/ND4L/ND6, encoded by mitochondrial DNA as contained in MITOMAP (22).

Table S7. Benchmarking the quantum chemical calculations using different density functionals against *ab initio* random-phase approximation (RPA) and domain-based local pair natural orbital coupled cluster theory with single-double and perturbative triples (DLPNO-CCSD(T)). **Top**: 381 atom DFT-model system (see Table S4) at the def2-TZVP/e=4 level. **Middle**: 131 atom DFT-model system (see Table S4) at the def2-TZVP level and RPA/aug-cc-pVTZ level. **Bottom:** 27 atom DFTmodel system, comprising an array of three water molecules, between two carboxylates (side chains of D66^{ND3} and E34^{ND4L}), at the def2-TZVP level for the DFT calculations and using aug-ccpVTZ basis set for the DLPNO-CCSD(T) and RPA calculations. Energies are given in kcal mol⁻¹.

Table S9. Force field parameters for H₃O⁺ used in the classical simulations in the deactive state.

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