

## SUPPORTING INFORMATION

### Formation and evolution of nanoscale calcium phosphate precursors under biomimetic conditions

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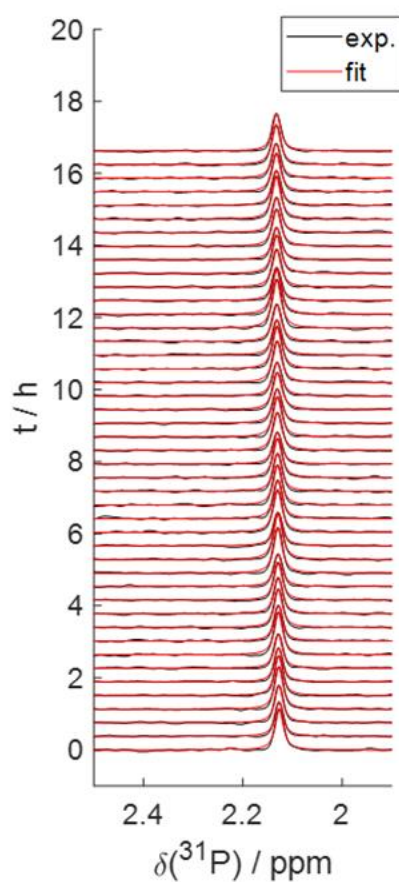
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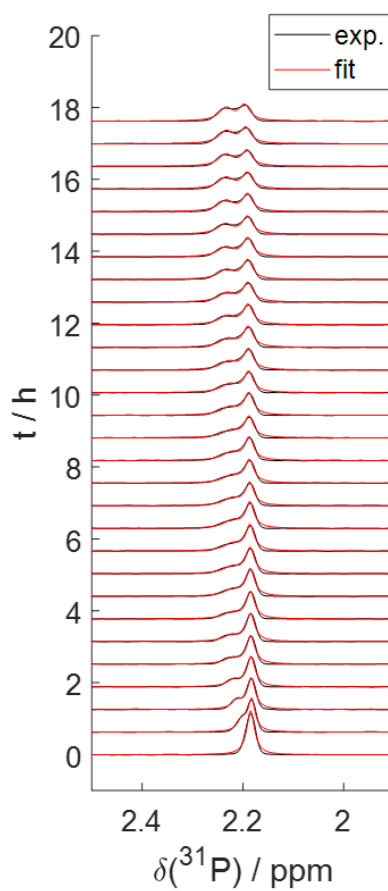
**Figure S1**



**Figure S1.** Negative control experiment of 1 mM Pi solution in which all salts (except Pi) from mSBF have been replaced by NaCl at equivalent concentrations to keep ionic strength constant (in HEPES buffer at pH 7.4). The  $^{31}\text{P}$  NMR time traces are constant over time and exhibit one single resonance of similar position and line width to mSBF at  $t = 0$ . This control allows us to assign unambiguously the 2.18 ppm  $^{31}\text{P}$  resonance of mSBF to free Pi.

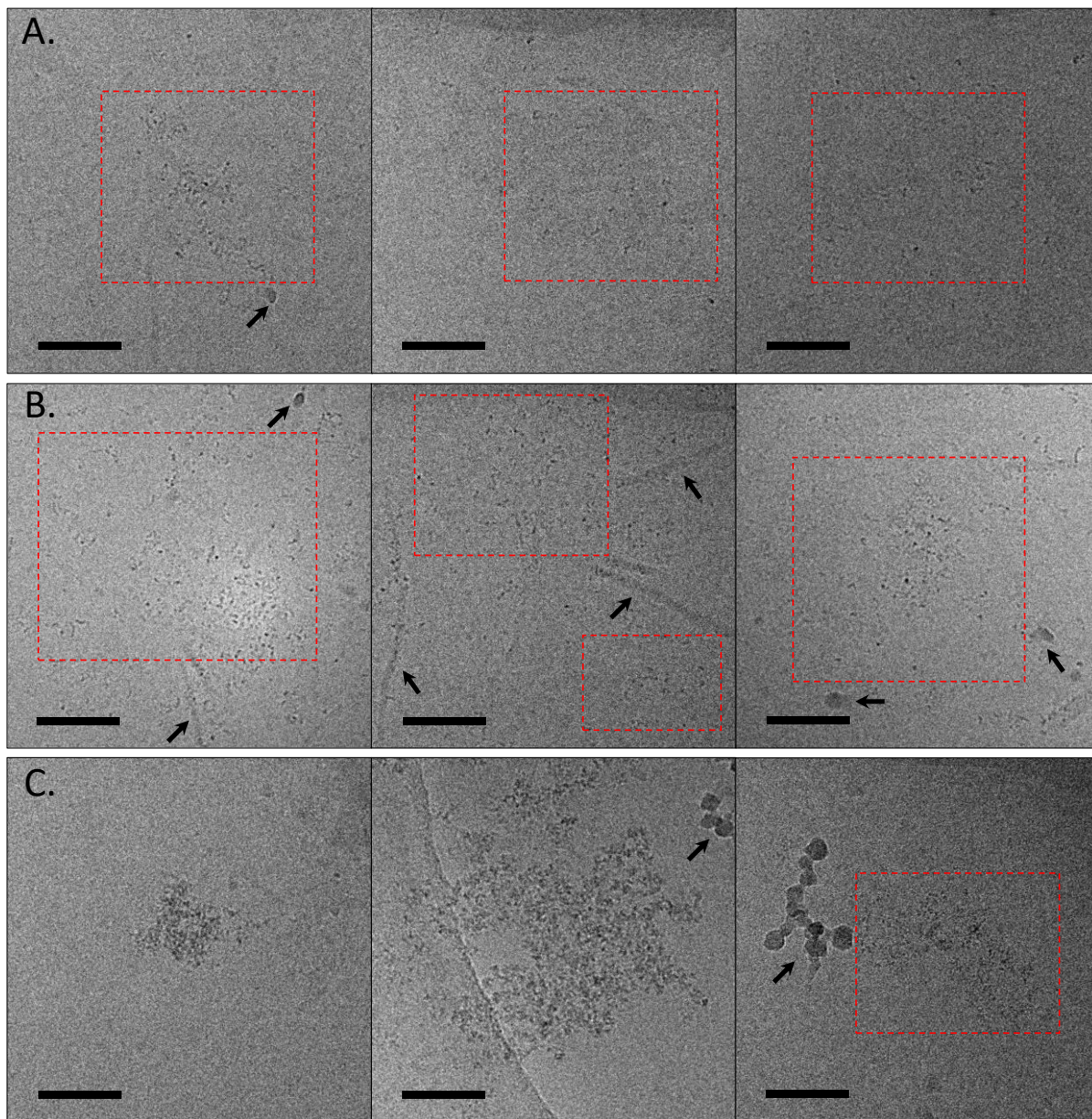
Please note that mSBF solution without Pi cannot be used as a control because of calcite precipitation in these conditions (see Oyane *et al. J. Biomed. Res. A* **2003**, 64A, 339-348).

**Figure S2**



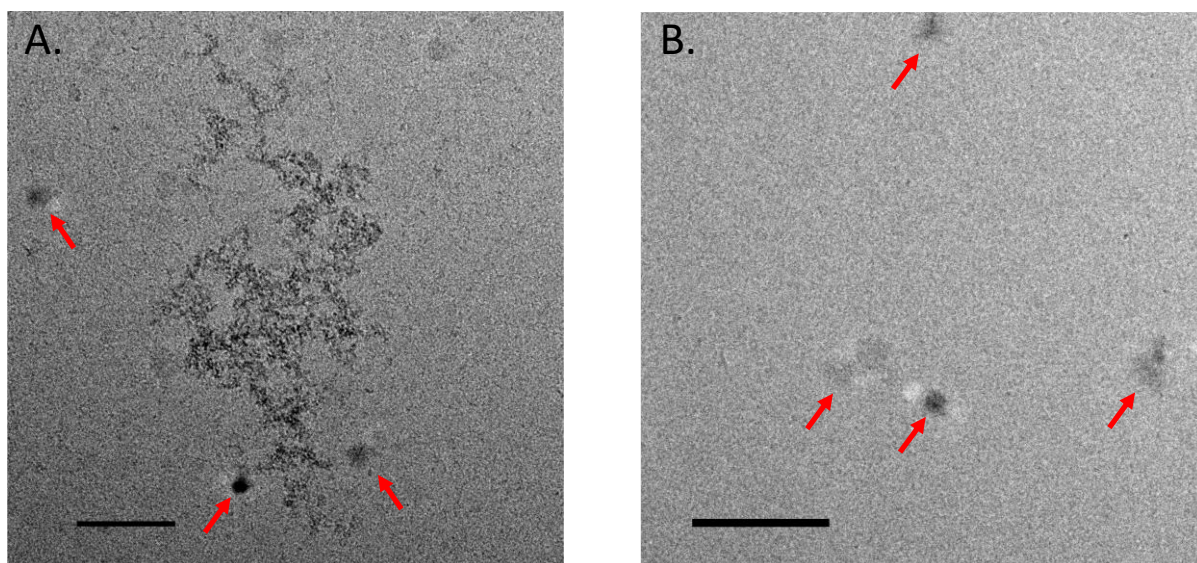
**Figure S2.** <sup>31</sup>P NMR time traces of 5 mM P<sub>i</sub> in mSBF as used in the DOSY experiments. The evolution is qualitatively the same as in the case of neat mSBF. However, the evolution is slowed down due to the excess in phosphate, which enables the detection of longer DOSY experiments in the slow exchange regime.

Figure S3



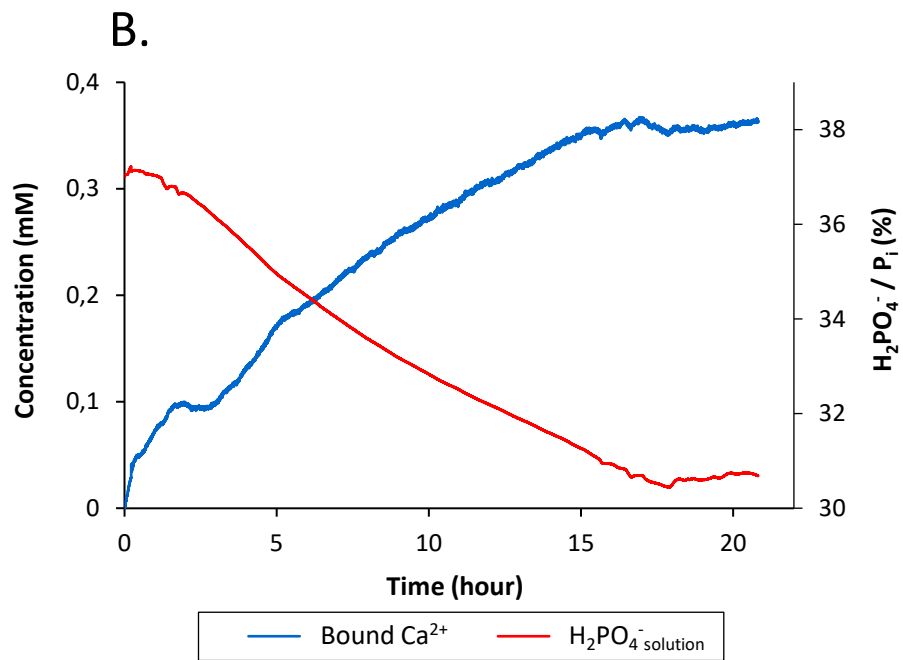
**Figure S3.** Evolution of PNS formation in mSBF over time revealed through cryo-TEM. (a) 1h, (b) 5.5h and (c) 23h after preparation. Scale bars = 50 nm. Red rectangles surround PNS units, black arrows point pollutions due to the freezing process, the plasma treatment of the grid or the solution container

**Figure S4**



**Figure S4.** Cryo-TEM micrographs corresponding to the zones analyzed through EDX. (A) zone containing PNS. (B) zone empty of PNS (B.). Scalebar = 100 nm. Red arrows exhibit pollutions due to the freezing process and the plasma treatment of the grid.

Figure S5



**Figure S5.** Calculation of bound- $\text{Ca}^{2+}$  involved in PNS (blue curve) and  $\text{H}_2\text{PO}_4^-$  concentration (red curve) in mSBF over time.