Supplementary information

Modulation of toll-like receptor 1 intracellular domain structure and activity by Zn2+ ions.

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Supplementary Figure 1. ¹H,¹⁵N-HSQC NMR spectrum of TLR1-TIR recorded at 35 °C, pH 6.3. The assignment of amide groups is indicated.



Supplementary Figure 2. A fragment of ¹H,¹³C-CT-HSQC NMR spectrum of TLR1-TIR recorded at 35 °C, pH 6.3. The assignment of methyl groups is indicated.



Supplementary Figure 3. NMR relaxation parameters of ¹⁵N nuclei (rates of longitudinal (R1) and transverse (R2) relaxation, heteronuclear equilibrium NOE (¹H,¹⁵N NOE)) and internal mobility parameters - generalized order parameter $S_0^2S_1^2$ and exchange contribution to the transverse relaxation R_{ex} , measured at two concentrations (blue corresponds to 3.3 mg/ml and orange - to 10.3 mg/ml).



Supplementary Figure 4. Analysis of TLR1-TIR ¹⁵N NMR relaxation. **a)** The R₁R₂ product. Orange line represents 15 s⁻², theoretical maximal for this parameter at 600 MHz in the absence of slow motions. Residues with R₁R₂ above the line are experiencing the motions in the us-ms timescale. **b)** Correlation plot of R₂ and η_{xy} . Region between two orange lines represents the theoretically possible values in the absence of slow motions. These residues were selected to calculate the rotational correlation time value. Points outside the specified region correspond to the residues, experiencing the motions in the us-ms timescale.



Supplementary Figure 5. Crystals of TLR1-TIR of P6₂22 space group.



Supplementary Figure 6. ¹H,¹⁵N-HSQC NMR spectra of 100 μ M TLR1-TIR recorded at 35 °C, pH 7.4 in the presence of 0/50/100/200 μ M of Zn²⁺ ions.



Supplementary Figure 7. Hydrodynamic properties of the TLR1-TIR in the presence and in the absence of Zn²⁺ ions. **a)** Intensity of NMR signal I as a function of gradient power Gz in the PGSTE-watergate experiment recorded for the 100 μ M sample of TLR1-TIR in the absence of ZnCl₂ (pH 7.4, 30 °C). The decay is approximated by a theoretical dependence, shown by the solid blue line. Hydrodynamic radius, corresponding to the measured diffusion coefficient (124·10⁻¹² m²s⁻¹) is indicated. **b)** Mass-weighted distributions of hydrodynamic radii measured by DLS for the 100 μ M sample of TLR1-TIR in the absence of Zn²⁺ ions (gray bars) and in the presence of 300 μ M ZnCl₂ (black bars). **c)** Fragment of the reference TROSY-HSQC spectrum recorded for the 200 μ M TLR1-TIR/Zn 1:1 sample that was used to measure the cross-correlated relaxation rates and estimate the rotational diffusion. Spectrum was recorded at pH 7.4 and 30 °C overnight. Correlation times of rotational diffusion obtained for each cross-peak of the Zn-bound TLR1-TIR are indicated.



Supplementary Figure 8. Overlay of ¹H,¹⁵N-HSQC NMR spectra of 100 μ M TLR1-TIR recorded at 30 °C, pH 7.4 in the presence of 100 μ M Co²⁺ ions (red) and without them (blue).



Supplementary Figure 9. Overlay of ¹H,¹⁵N-HSQC NMR spectra of 100 μ M TLR1-TIR recorded at 30 °C, pH 7.4 in the presence of 100 μ M Co²⁺ ions (blue) or 100 μ M of Cu²⁺ ions (red).



Supplementary Figure 10. The influence of Zn^{2+} and Co^{2+} ions on TLR1-TIR disulfide bond formation. **a)** Overlay of ¹H,¹⁵N-HSQC NMR spectra of 100 µM TLR1-TIR recorded at 30 °C, pH 7.4 after the treatment with GSH/GSSG overnight (shown in blue, cysteines oxidized) and after the addition of 50 µM ZnCl₂ (shown in red). Spectra are different, the zinc binding does not cause the disulfide formation. **b)** Overlay of ¹H,¹⁵N-HSQC NMR spectra of 100 µM TLR1-TIR recorded at 30 °C, pH 7.4 after the treatment with GSH/GSSG overnight (shown in blue, cysteines oxidized) and after the disulfide formation. **b)** Overlay of ¹H,¹⁵N-HSQC NMR spectra of 100 µM TLR1-TIR recorded at 30 °C, pH 7.4 after the treatment with GSH/GSSG overnight (shown in blue, cysteines oxidized) and after the addition of 50 µM CoCl₂ (shown in red). Co²⁺ catalyzes the disulfide formation in TLR1-TIR.



mutants, recorded for the 100 μ M TLR1-TIR at 30 °C, pH 7.4 in the presence of 50 μ M ZnCl₂. Signals from the Trp side chains, which were used for the analysis are highlighted by a dashed circle and additionally provided at a larger size at the left upper corner of each spectrum.



three cysteine mutants (red), recorded for the 100 μ M TLR1-TIR at 30 °C, pH 7.4 in the absence of ZnCl₂.



Supplementary Figure 13. Normalized frequencies of conformations with distances between possible Zn2+ coordinators less than 5.0Å. Total 2000 conformations (100 loop conformations for every NMR structure) were sampled with Rosetta Loopmodel protocol. Only atoms with frequencies >15 are shown.



Supplementary Figure 14. The properties of **Zn1** state. **a)** Root Mean Square Fluctuations (RMSF) for systems with C667-H669-C686 coordination mode. The most significant differences (up to 3Å) between two types of systems are concentrated around the region I666 - E670. Only heavy atoms (except hydrogens) are considered. **b)** Root Mean Square Displacement (RMSD) of the BB-loop over the course of simulation. Only heavy atoms (except hydrogens) are considered.



Supplementary Figure 15. The properties of **Zn2** state. **a)** Distance between I666_O and Zn in the unrestrained test simulation. The coordination is stable even without specialized parameters for the Zn coordination by the backbone oxygen. **b)** Root Mean Square Fluctuations (RMSF) for systems with the C6667-I666 backbone coordination mode in restrained simulations. The most significant differences (up to 3Å) between two types of systems are in the areas of R671_O which also joins the coordination sphere and F695 buried in the hydrophobic core. Only heavy atoms (except hydrogens) are considered. **c)** Root Mean Square Displacement (RMSD) of the BB-loop over the course of simulation. Only heavy atoms (except hydrogens) are considered.





residue is shown in yellow.