Supplementary Information for:

Cell membrane coating integrity affects the internalization mechanism of biomimetic nanoparticles

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Supplementary Methods

Materials

Tetraethyl orthosilicate (TEOS), hexadecyltrimethylammonium bromide (CTAB), Au NPs (size: 80 nm), fetal bovine serum (FBS), antibiotic antimycotic solution (100×), trypsin-EDTA, 4',6-diamidino-2-phenylindole (DAPI), 1,1'-dioctadecyl-3,3,3',3'-tetramethylindocarbocyanine perchlorate (Dil) and bovine serum albumin (BSA) were purchased from Sigma-Aldrich Corporation (St Louis, MO, USA). Cyanine 5 (Cy5) was purchased from Lumiprobe Corporation (Hallandale Beach, FL, USA). 1,2dioleoyl-sn-glycero-3-phosphocholine (DOPC) and 1,2-distearoyl-sn-glycero-3phosphoethanolamine-N-(7-nitro-2-1,3-benzoxadiazol-4-yl) (NBD-DSPE) were purchased from Avanti Polar Lipids Inc. (Alabaster, AL, USA). Bicinchoninic acid (BCA) assay kit and 6-(7-Nitrobenzofurazan-4-ylamino)dodecanoic acid NHS ester (NBD-X) were purchased from Thermo Fisher Scientific (Waltham, MA). Dulbecco's modified Eagle's medium (DMEM) and RPMI 1640 medium were purchased from Biowest Corporation, France. Phosphate buffer saline (PBS) and Hank's balanced salt solution (HBSS) were obtained from HyClone. CellTiter-Glo assay was purchased from Promega Corporation, USA. ZIF-8 NPs¹, Fe₃O₄ NPs², poly(lactic-co-glycolic acid) (PLGA) NPs³ and porous silicon (PSi) NPs⁴ were synthesized according to the literatures. All chemical reagents were used directly without further purification unless specifically mentioned.

Fluorescence labeling

NBD labeled core materials (mesoporous SiO₂ NPs, nonporous SiO₂ NPs, Fe₃O₄ NPs, and PSi NPs) were obtained through sequential chemical surface modification. Briefly, 50 mg core NPs were reacted with 20 uL APTES in 10 mL ethanol at room temperature for 2 h. The as-synthesized amine-modified NPs were washed with ethanol twice and then reacted with NBD in ethanol at room temperature overnight. To make a negative surface charge of NPs on the cell membrane coating, the excess amine groups were reacted with succinic anhydride in ethanol. After rinsing with ethanol twice, the NBD labeled were successfully harvested. In the preparation of fluorescent PLGA NPs, NBD was covalently attached to the polymers by a carbodiimide method⁵. In order to obtain NBD labeled ZIF-8 NPs, NBD was encapsulated in the ZIF-8 NPs during the biomimetic growth process. The NBD labeled Au NPs were prepared by mixing the Au NPs with dye under stirring for 2 h and then washed three times with deionized water.

Preparation of GUVs, LB-SiO₂ NPs, E-SiO₂ NPs, Ex-SiO₂ NPs and BSA adsorbed NPs

Giant unilamellar vesicles (GUVs) comprising DOPC/NBD-DSPE (molar ratio: 150/1) were prepared by a natural swelling method as follows⁶. First, the above lipid mixture in chloroform was dried under nitrogen gas to obtain a thin, homogeneous lipid film. Subsequently, the residual chloroform in the film was completely removed by placing the bottle in a vacuum desiccator for 12 h. Then, 100 μ L of buffer containing 20 mM HEPES (pH 7.0), 150 mM NaCl and 100 mM sucrose were added to the resulting lipid film and pre-hydrated at 45 °C for 10 min. After that, a 900 μ L aliquot of the same buffer was added, and the glass bottle was resealed and incubated in an incubator at 37 °C for 2 h to produce the GUV suspension. The morphology of GUVs was observed under a confocal laser scanning microscopy (CLSM; Zeiss LSM 700, Carl Zeiss, Jena, Germany). The control membrane, DOPC lipid bilayer coated SiO₂ (LB-SiO₂) NPs, were prepared according to a procedure already described⁷.

To obtain endocytosed SiO₂ (E-SiO₂) NPs, CT26 cells were treated with 200 μ g/mL positively charged SiO₂ NPs for 12 h in 15 cm dishes. Then, the cells were detached with Trypsin-EDTA solution and washed by HBSS for three times. The as-prepared E-SiO₂ NPs were collected by centrifuging at 1200 × g. With respect to exocytosed SiO₂ (Ex-SiO₂) NPs, after the CT26 cells had been incubated with 200 μ g/mL positively charged SiO₂ NPs for 24 h in 15 cm dishes, the cell media were discarded and replaced with fresh media. After a further 12 h incubation, the cells were

discarded at 1200 g for 5 min and then the supernatants were further centrifuged at 10,000 g for 15 min to obtain $Ex-SiO_2$ NPs. To prepare BSA adsorbed NPs, both the positively charged NPs and negatively charged NPs (1 mg/mL) were incubated with BSA solution (1 mg/mL) at 37 °C for 4 h, and then the superfluous BSA was discarded by centrifugation.

DPD formulation

The dissipative particle dynamics (DPD) technique was used to study the cellular pathways of the CM-SiO₂ NPs, which has been widely applied in the studies of the NP-membrane interactions⁸⁻¹⁰. In the DPD simulations, a cluster of atoms can be represented by a single bead and each bead is governed by the Newton's equation of motion,

$$\frac{dv_i}{dt} = \frac{F_i}{m} \tag{1}$$

where the mass *m* of per bead is set to 1. The total force between beads *i* and *j* consists of three types of forces: conservative force $F_{ij}^{\ C}$, dissipative force $F_{ij}^{\ D}$, and random force $F_{ij}^{\ R}$. The total force on the bead *i* can be expressed as,

$$F_{i} = \sum_{i \neq j} F_{ij}^{C} + F_{ij}^{D} + F_{ij}^{R}$$
(2)

where the sum runs over all beads j within a certain cutoff radius r_c . The conservative force is a soft-repulsive force taken as,

$$F_{ij}^{C} = a_{ij}(1 - r_{ij})\mathbf{r}_{ij}$$
(3)

where a_{ij} is the maximum repulsion between the bead *i* and *j*, $r_{ij} = |\mathbf{r}_i - \mathbf{r}_j| / r_c$, and $\mathbf{r}_{ij} = \mathbf{r}_{ij} / |\mathbf{r}_{ij}|$. The dissipative force is given by,

$$F_{ij}^{D} = -\gamma \omega_{D}(r_{ij})(\mathbf{r}_{ij} \cdot \mathbf{v}_{ij})\mathbf{r}_{ij}$$
(4)

Where γ is the strength of friction, $\omega_D(r_{ij}) = (1 - r_{ij})^2$ and $\mathbf{v}_{ij} = \mathbf{v}_i - \mathbf{v}_j$. The random force is,

$$F_{ij}^{R} = \sigma \omega_{R}(r_{ij}) \xi_{ij} \mathbf{r}_{ij}$$
⁽⁵⁾

where ξ_{ij} is a zero-mean Gaussian random variable of unit variance, $\omega_R(r_{ij}) = 1 - r_{ij}$, and σ is a noise amplitude that equals to $\sqrt{2k_BT\gamma}$ ¹¹. We take r_c as the characteristic length scale and k_BT as the characteristic energy scale. The characteristic time scale is defined as $\tau = \sqrt{mr_c^2 / k_BT}$. Equation (1) is integrated in time with a velocity-Verlet algorithm at $\sigma = 3$ and $\Delta t = 0.01 \tau^{12}$. Two simulation systems were used here: the smaller system ($40 \times 100 \times 100 r_c^3$) was used for studying the 1, 2, 4 NPs aggregates and performed at $8 \times 10^4 \tau$; the larger system ($40 \times 150 \times 150 r_c^3$) was used for studying the 9,16 NPs aggregates and performed for $1.6 \times 10^5 \tau$. The periodic boundaries are applied in all the x, y, z directions of the simulation box. The bead density of each simulation box is a constant of 3. All simulations were performed in the NVT ensembles using the LAMMPS package¹³.

Supplementary Figures



Supplementary Fig. 1 TEM images of cell membrane vesicles at low-magnification (**a**) and high-magnification (**b**). Scale bars, 200 nm in (a) and 50 nm in (b).



Supplementary Fig. 2 a, NBD labeled NPs were achieved by surface chemical modification of the NPs with 3-aminopropyltriethocysilane (APTES), NBD dodecanoic acid N-succinimidyl ester (NBD-X), and succinic anhydride, respectively. **b-e,** Fourier transform infrared spectroscopy (FTIR, Thermo Nicolet iS 50) spectra (**b**), thermogravimetric analysis (TGA; **c**), mean diameter (**d**) and zeta potential (**e**) of SiO₂, SiO₂-NH₂, SiO₂-NBD, and SiO₂-NBD-COOH NPs. The characteristic peak for amine

modification was observed at 1550 cm⁻¹ owing to $-NH_2$ bending vibration. The characteristic absorption at 1720 cm⁻¹ indicated that there are carboxylic groups on the final product (SiO₂-NBD-COOH). Due to only 0.42% of NBD was grafted onto SiO₂ NPs calculated from thermogravimetric analysis (TGA) results (c), the absorptions of NBD were difficult to be detected from FTIR. Alternatively, the fluorescence images of NBD labelled SiO₂ NPs (Fig. 1k) supported that the NBD was successfully modified onto the SiO₂ NPs. These modifications were further confirmed by the measurement of size and zeta potential, in which the zeta potential changed from -32.5 mV to 22.9 mV after amine modification. Data represents mean ± SD (n = 3).



Supplementary Fig. 3 TEM images of mesoporous SiO_2 NPs at low-magnification (**a**) and high-magnification (**b**). Scale bars, 200 nm in (a) and 50 nm in (b).



Supplementary Fig. 4 Physicochemical characterization of SiO₂ NPs, cell membrane vesicles and CM-SiO₂ NPs. **a,b**, Mean diameter (**a**) and zeta potential (**b**) of SiO₂ NPs, cell membrane vesicles and CM-SiO₂ NPs. Data represents mean \pm SD (n = 3). **c**, SDS-PAGE protein analysis of SiO₂ NPs, cell membrane vesicles and CM-SiO₂ NPs. **d**, Stability of CM-SiO₂ NPs in 1X PBS, determined by monitoring the particle size (diameter, nm), over a span of 6 day. Data represents mean \pm SD (n = 3).



Supplementary Fig. 5 Confocal laser scanning microscopy (CLSM) images showed the colocalization of the SiO₂ NPs (labeled with Cy5; red) and cell membranes (labeled with Dil; green) after being internalized by CT26 cells. The CM-SiO₂ NPs were incubated with CT26 cells for 4 h. The cell nuclei were stained with DAPI (blue). Scale bar, 5 μ m.



Supplementary Fig. 6 a,b, Mean diameter (**a**) and zeta potential (**b**) of positive charge and negative charge functionalized SiO₂ NPs before and after adsorption of BSA. **c**, Relative fluorescence intensity of positively charged and negatively charged NPs after addition of quencher. Data represents mean \pm SD (n = 3). One-way ANOVA followed by post hoc Tukey test was used to determine the significance of data. ns: not significant.



Supplementary Fig. 7 TEM images of cell membrane fully coated mesoporous SiO_2 NPs. Scale bars, 100 nm. According to these TEM images, the full cell membrane coating can be classified to two different types: tight and loose, depending on the size of encountered cell membrane vesicles during the coating process. Specifically, when the size of cell membrane vesicles is comparable to that of SiO_2 NPs, the SiO_2 NPs can be wrapped tightly by the cell membrane. By contrast, when the size of cell membrane vesicles is significantly bigger than that of SiO_2 NPs, the SiO_2 NPs are coated loosely.



Supplementary Fig. 8 a, Schematic illustration of the preparation of Ex-SiO₂ NPs. Positively charged mesoporous SiO₂ NPs are endocytosed into CT26 cancer cells and then exocytosed into the extracellular space after incubation. **b,c**, Mean diameter (**b**) and zeta potential (**c**) of positively charged mesoporous SiO₂ NPs and Ex-SiO₂ NPs. **d-f**, TEM images of bare SiO₂ NPs (**d**) and Ex-SiO₂ NPs at low-magnification (**e**) and high-magnification (**f**). Scale bars, 100 nm in d and f and 200 nm in e. **g**, Quantification of the ratio of full cell membrane coating for Ex-SiO₂ NPs. Data represents mean \pm SD (n=3). One-way ANOVA followed by post hoc Tukey test was used to determine the significance of data. p = 5.9E-5. **p < 0.01.



Supplementary Fig. 9 TEM images of RBC membrane vesicles at low-magnification (**a**) and high-magnification (**b**). Scale bars, 200 nm.



Supplementary Fig. 10 a,b, Mean diameter (**a**) and zeta potential (**b**) of negative charge and positive charge functionalized mesoporous SiO₂ NPs before and after coating with cell membranes. **c,d,** TEM images of negative charge (**c**) and positive charge (**d**) functionalized mesoporous SiO₂ NPs after coating with cell membranes. Scale bars, 100 nm. **e**, Quantification of the ratio of full cell membrane coating for negative charge and positive charge functionalized mesoporous SiO₂ NPs. **f**, Quantification of the ratio of full cell membrane coating with different weight ratios of cell membranes to core materials (SiO₂ NPs). Data represents mean ± SD (*n* = 3). One-way ANOVA followed by post hoc Tukey test was used to determine the significance of data in **e** and **f**. *p* = 6.7E-4 (**e**). ***p* < 0.01.



Supplementary Fig. 11 a–e, TEM images of nonporous Stöber SiO₂ NPs of varied sizes: 30 nm (**a**), 62 nm (**b**), 100 nm (**c**), 149 nm (**d**), and 190 nm (**e**). Scale bars, 100 nm. **f**, Particle size distributions of five sizes of nonporous Stöber SiO₂ NPs (n = 120). The data were calculated from TEM images. **g**,**h**, Mean diameter (**g**) and zeta potential (**h**) of nonporous Stöber SiO₂ NPs before and after coating with cell membranes. Data represents mean ± SD (n = 3).



Supplementary Fig. 12 Mean diameter (**a**) and zeta potential (**b**) of Fe₃O₄ NPs, ZIF-8 NPs, Au NPs, PLGA NPs, and PSi NPs before and after coating with cell membranes. Data represents mean \pm SD (*n* = 3).



Supplementary Fig. 13 Cell viability of CT26 cells after incubation with SiO₂ NPs, LB-SiO₂ NPs and CM-SiO₂ NPs for 24 h at different concentrations. Data represents mean \pm SD (n = 8). One-way ANOVA followed by post hoc Tukey test was used to determine the significance of data. p = 1.7E-8 (SiO₂ NPs vs. LB-SiO₂ NPs), p = 6.3E-8 (SiO₂ NPs vs. CM-SiO₂ NPs). **p < 0.01.



Supplementary Fig. 14 TEM images of CT26 cells (**a**), HeLa cells (**b**) and MCF-7 cells (**c**) after 4 h incubation with LB-SiO₂ NPs. Insets below are magnified of each image in the area highlighted with the respective yellow dashed box. Scale bars, 2 μ m (top); 200 nm (bottom).



Supplementary Fig. 15 a, Representative CLSM images of RAW264.7 cells after 4 h incubation with SiO₂ NPs, LB-SiO₂ NPs and CM-SiO₂ NPs. The SiO₂ cores were labeled with Cy5 (red), and the cell nuclei were stained with DAPI (blue). Scale bars, 20 μ m. **b**, Flow cytometric analysis of RAW264.7 cells incubated with blank solution, SiO₂ NPs, LB-SiO₂ NPs and CM-SiO₂ NPs. **c**, Quantification of cellular uptake in RAW264.7 cells. Data represents mean ± SD (*n* = 3). One-way ANOVA followed by post hoc Tukey test was used to determine the significance of data. ****p* < 0.001.



Supplementary Fig. 16 Influence of pharmacological inhibitors on CM-SiO₂ NPs uptake. Fluorescence intensity histograms (**a**) and fluorescence quantification (**b**) of cellular uptake by CM-SiO₂ NPs (50 µg/mL) in CT26 cells treated with low temperature (4 °C), chlorpromazine (CPZ), genistein (GEN), and cytochalasin D (CytD), as determined by flow cytometry. Data were normalized to cells treated only with CM-SiO₂ NPs at 37 °C in **b**. Data represents mean ± SD (*n* = 3). One-way ANOVA followed by post hoc Tukey test was used to determine the significance. *****p* < 0.0001.



Supplementary Fig. 17 Schematic illustration of the models in the simulations. The coated part of the CM-SiO₂ NP is shown in orange and the uncoated part is shown in blue. For cell membranes, the heads and the tails of the membrane's phospholipids are shown in cyan and white, respectively. The receptor of the cellular membrane is shown in ice blue.



Supplementary Fig. 18 a, Schematic illustration of cell membrane coating degree calculation. **b,** TEM image of partially coated NPs. Scale bar, 50 nm. The calculation process is as follows:

The coating area of cell membrane:

$$S_{coating} = \int_0^{\frac{\pi}{2}} 2\pi r R \cdot d\varphi = \int_0^{\frac{\pi}{2}} 2\pi R^2 \cos\varphi \cdot d\varphi = 2\pi R^2 \int_0^{\frac{\pi}{2}} \cos\varphi \cdot d\varphi = 2\pi R^2 (1 - \sin\varphi),$$

Surface area of NPs: $S_{NPs} = 4\pi R^2$,

Coating degree(%) =
$$\frac{S_{coating}}{S_{NPs}} \times 100\% = \frac{2\pi R^2 (1 - \sin\varphi)}{4\pi R^2} \times 100\% = \frac{1 - \sin\varphi}{2} \times 100\%$$

= $\frac{1 - \sin\left(\frac{\pi - \theta}{2}\right)}{2} \times 100\%$,

Where θ can be measured directly from the TEM image (Supplementary Fig. 18b).



Supplementary Fig. 19 Effect of cell membrane coating degree on position (**a**) and wrapping ratio (**b**) of CM-SiO₂ NPs at the end of wrapping process. Data represents mean \pm SD (n = 20).



Supplementary Fig. 20 a-d, TEM images for time evolution of CM-SiO₂ NPs incubated with CT26 cells for 0.5 h (**a**), 1 h (**b**), 2 h (**c**), and 4 h (**d**). Scale bars, 2 µm; insets, 200 nm. **e**, Schematic of intracellular trafficking of CM-SiO₂ NPs. Partially coated NPs enter cells mainly in the multiple form and then those NPs are gradually trapped in early endosomes (2 h), late endosomes (2~4 h) and lysosomes (≥ 4 h) during the vesicle transport and maturation, respectively. **f**, Histogram distribution of the number of NPs trapped in a single early endosome. 158 early endosomes were analyzed from TEM images (2 h).



Supplementary Fig. 21 a, TEM images of CM-SiO₂ NPs (1 μ g/mL) incubated with CT26 cells for 2 h. Scale bars, 2 μ m; insets, 200 nm. **b,** Histogram distribution of the number of NPs trapped in a single early endosome. 120 early endosomes were analyzed from TEM images.



Supplementary Fig. 22 a-d, Schematic illustration of three different locations of two NPs when they interacted with the cell membrane: separated (a), contacted (b), and rotated with a high coating degree (30%; c) and a low coating degree (15%; d). **e**, Rotating angle as a function of coating degree. Data represents mean \pm SD (n = 20).



Supplementary Fig. 23 The two-dimension model applied to predict the required coating degree for half wrapping the NPs aggregates with n=2 (a) and n=3 (b). The orange region shows the coated region to drive the membrane bending around the CM-SiO₂ NPs. The purple region shows the coated region to drive the additional membrane deformation (including bending and stretching), which connects with the neighboring curved membrane. R, r₁ and r₂ stand for the radius of the SiO₂ NP, the thickness of the coating membrane and the thickness of the wrapping membrane, respectively.



Supplementary Fig. 24 Final snapshots (t=160000 τ) of the interactions between different aggregated numbers (2, 4, and 9) of NPs and coating degrees (16%, 25%, 33% and 40%).



Supplementary Fig. 25 a,b, Representative DPD simulation snapshots of 16 aggregated CM-SiO₂ NPs show the setup of the simulation system (t=0 τ ; **a**) and the final equilibrated NP-membrane structure at the top view and the profile view (t=160000 τ ; **b**). The coating degree of each NP is 33%. **c**, Quantification of the final position of NPs. Data represents mean ± SD (*n* = 20).



Supplementary Fig. 26 a,b, Representative DPD simulation snapshots of 9 aggregated CM-SiO₂ NPs with inconsistent coating degrees show the setup of the simulation system (t=0 τ ; **a**) and the final equilibrated NP-membrane structure at the top view and the profile view (t=160000 τ ; **b**). The distribution of the coating degree for each NP is shown in the inset. **c**, Quantification of the final position of NPs. Data represents mean ± SD (*n* = 20).



Supplementary Fig. 27 TEM images showing the process of a single CM-SiO₂ NP enter into CT26 cells. Scale bars, 100 nm.



Supplementary Fig. 28 Gating strategy to quantify nanoparticle binding and uptake *in vitro* (Fig. 3bd, Supplementary Fig. 15b and Supplementary Fig. 16a). Three times of this experiment was repeated independently with similar results.

Supplementary Tables

Supplementary Table 1 Synthesis conditions of nonporous Stöber SiO $_2$ NPs with a variety of sizes

Size (nm)	95% Ethanol (mL)	28-30% NH₄OH (mL)	TEOS (mL)	Temperature (K)
30 ± 2.9	40	0.6	0.5	298
62 ± 8.1	40	0.65	0.5	298
100 ± 7.0	40	1.1	0.5	298
149 ± 7.8	40	1.2	0.5	298
190 ± 8.6	40	1.5	0.5	298

Supplementary Table 2 Interaction parameters a_{ij} between the beads *i* and *j*. H, R, T and W represent the lipid head, receptor, lipid tail and water beads, respectively. P₁ and P₂ represent the coated and uncoated beads of the CM-SiO₂ NP, where the coated part is considered as the ligand and the uncoated part represents the SiO₂. The increase of a_{ij} represents the increase of the repulsion between the beads. Interaction parameters between the same beads a_{ii} are set as 25 except for P₂, which slightly increases to reflect the repulsion within the SiO₂ NPs by the surface charge coatings. The a_{ij} between the receptor and the ligand is set as 1 to mimic their strong interactions. Other parameters are described in the CG model section.

a _{ij}	Η	R	Т	W	P ₁	P ₂
Н	25	25	100	25	25	25
R	25	25	100	25	1	25
Т	100	100	25	100	100	100
W	25	25	100	25	25	40
P ₁	25	1	100	25	25	25
P ₂	25	25	100	40	25	30

Supplementary Movies

Supplementary Movies 1-3. Interaction between the cell membrane and the CM-SiO2 NPs with increasing the aggregation numbers: Supplementary Movie 1 (n=1), Supplementary Movie 2 (n=2), and Supplementary Movie 3 (n=9). The coating degree of each NP is 33%. It demonstrated that the higher number of NP aggregates were more likely to rotate spontaneously after entering into the membranes.

Supplementary Notes

Supplementary Note 1. Estimation of the coating degree by geometric constraints

With the geometric constraints and the symmetricity of the two-dimension model (Supplementary Fig. 23), the required coating degree (c) can be estimated by:

$$c_{\min} = \frac{A_{orange}}{A_{total}} = \frac{1 - \sin\theta}{2}$$
(6)

$$c_{\max} = \frac{A_{orange} + A_{purple-max}}{A_{total}} = \frac{1 - \sin\left(\theta - \frac{\alpha}{2}\right)}{2}$$
(7)

Here, the A_{orange} stands for the area of the orange region and $A_{purple-max}$ stands for the maximum of the area of the purple region. The θ and α can be derived by:

$$\theta = \frac{2\pi - \varphi}{4} \text{ (for 2 NPs)}$$
(8)

$$\theta = \frac{2\pi - \varphi}{3} \text{ (for 3 NPs)}$$
(9)

$$\alpha = \arctan\left(\frac{r_2 \sin\left(\frac{\varphi}{2}\right)}{R + r_1 + r_2 - r \cos\left(\frac{\varphi}{2}\right)}\right)$$
(10)

$$\varphi = 2 \arcsin\left(\frac{R}{R + r_1 + r_2}\right) \tag{11}$$

According to the experimental conditions, the required coating degree for 2 NPs is: $c_{\min} = 29\%$ and $c_{\max} = 32\%$; the required coating degree for 3 NPs is: $c_{\min} = 23\%$ and $c_{\max} = 26\%$. According to the simulation conditions, the required coating degree for 2 NPs is: $c_{\min} = 29\%$ and $c_{\max} = 36\%$; the required coating degree for 3 NPs is: $c_{\min} = 23\%$ and $c_{\max} = 30\%$. When compared with the real three-dimension simulation results for 4 NPs (~33%) and 9 NPs (~30%), the estimated values are slightly larger. There are two possible reasons: 1. In the real three-dimension situation, the required coating degree for two dimensions needs to be satisfied, whereas only one dimension needs to be satisfied in a two-dimension model. 2. The actual additional deformation connecting the two neighboring curved membrane is much more complicated than the hypothetical one in the two-dimension model, which may need more driving energies provided by the coating membranes.

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