Expanded View Figures

Figure EV1. Pore-forming assay using BakATM and BclxLATM bound to Ni²⁺-caged lipids in liposomes.

- **Figure EV1.** Pore-forming assay using Bak \triangle TM and BclxL \triangle TM bound to Ni²⁺-caged lipids in liposomes.
A, B The data are averaged for the time frame of 9,000–10,000 s and the standard deviation calculated from three caged lipids on Bak Δ TM activity. Liposomes were formed with OMM-like lipids and subjected to 50 nM Bak Δ TM \pm 20 nM CBid. (B) Influence of the indicated lipid compositions on Bak Δ TM autoactivity
- C Influence of cBid (300 nM) and BclxL Δ TM (300 nM) on Bak Δ TM (600 nM), when autoactive.
D Comparison of the effect of BclxL Δ TM on 600 nM Bak Δ TM pore formation with and without
- Comparison of the effect of BclxL Δ TM on 600 nM Bak Δ TM pore formation with and without a His-tag. The inhibitory activity of BclxL is highly dependent on its stable membrane association.
- E Far-UV-CD spectra of Bak $\Delta TM \pm E$. coli polar liposomes and the Puma-BH3 peptide as indicated.
- CD-detected thermal melting experiments of Bak Δ TM (black) with liposomes (red) and Puma-BH3 (blue).
- G Percentile dye release is shown after 420 s for 600 nM Bak $\Delta TM \pm 300$ nM of an activator. The data are averaged for the time frame of 9,000-10,000 s and the standard deviation calculated from three technical replicates, respectively. Both cBid and Puma-BH3 lead to enhanced liposome permeabilization.

Data information: (C–G) Liposomes were formed using E. coli polar lipids.

Figure EV1.

- **Figure EV2. NMR structural characterization of the Bak soluble domain and the TMH.**
A NMR secondary chemical shifts of Bak∆TM are in agreement with the α-helical secondary structure content as present in a crystal structu et al, 2006)) indicated at the top.
- B Correlation between the experimental and back-calculated RDCs for the NMR-refined Bak ΔTM structure.
- C Comparison of the refined BakATM structure containing the full N-terminus with the published crystal structure (2IMS.pdb (Moldoveanu et al, 2006)), which was used as an initial structure for the refinement.
- D Comparison of the sequence and structure of the Bak- and BclxL-TMH in nanodiscs. The high similarity is shown by color coding the amino acids according to their polarity (red: apolar, green: polar, blue: basic). Identical amino acids are highlighted in the sequence in gray.
- E SEC profiles of Bak-TMH in nanodiscs after cleaving off the GB1 tag color-coded according to the table on the right. The area under the curve was used to calculate the ratio of TMH/nanodisc. The table shows the applied assembly conditions as well as the expected and calculated amount of TMH/nanodisc. For a dimeric protein like the glycophorin A (GlyA) TMH, the expected and calculated ratios vary, which is not the case for the Bak-TMH.

Figure EV2.

- **Figure EV3. Biophysical characterization of soluble Bak** Δ **TM and BclxL** Δ **TM.
A 2D-[¹⁵N, ¹H]-HSQC spectra of Bak** Δ **TM (left) and BclxL** Δ **TM (right) using a protonated (black) or deuterated (red) sample.**
- B DLS experiments of Bak ΔTM (left) and BclxL ΔTM (right) upon incubation at different temperatures and varying times. The tables on the right state the hydrodynamic diameter in [nm], which is calculated from the measured hydrodynamic radii.
- C Tryptophan fluorescence emission spectra of Bak ΔTM at different GuHCl concentrations upon excitation at 280 nm.
- D Fluorescence emission spectra of SYPRO Orange binding to Bak ΔTM at different GuHCl concentrations.
- E 2D-[¹⁵N, ¹H]-HSQC spectra of Bak Δ TM in the absence (black) and presence of 1.5 M GuHCl (red) and 2.5 M GuHCl (magenta).

Figure EV3.

- **Figure EV4. Characterization of Bak∆TM stability and pore-forming activity in nanodiscs and liposomes.**
A Far-UV-CD thermal melting data of Bak∆TM bound to lipid nanodiscs. The melting point at 68.4°C corresponds to Bak∆T shows the melting of the MSP forming the nanodisc.
- B SDS-PAGE of Cu/phen crosslinking of Bak Δ TM bound to Δ H5 nanodiscs \pm Bid-BH3 peptide.
- C Size exclusion chromatogram of BakDTM bound to nanodiscs after activation by Bid-BH3. Inset: SDS–PAGE shows that activated BakDTM elutes before the main nanodisc peak. The MSP band is very faint due to increased destaining which is sometimes seen for MSP.
- D Pore-forming assay with 600 nM Bak Δ TM in liposomes of 30 nm size.
- E Comparison of the deuterium uptake of the active Bak species in liposomes (black) and DH5 nanodiscs (blue). The HDX-MS detected deuterium uptake is compared after 10 s at the top and 2 h at the bottom.

Figure EV4.

- Figure EV5. HDX-MS data of Bak∆TM.
A HDX peptide coverage maps of the different Bak∆TM samples. Bak∆TM: Total coverage was 100% with 126 peptides and a redundancy of 6.96. Bak∆TM + ∆H5 nanodiscs (inactive): Total coverage was 99% with 83 peptides and a redundancy of 4.47. BakΔTM + ΔH5 nanodiscs + Bid-BH3 (active): Total coverage was 97% with 76 peptides and a redundancy of 4.25.
- B HDX graphs of three peptides, located in different helices as indicated, displaying the relative deuterium uptake during the reaction time. Two biological replicates were measured for each state. (ND = Δ H5 nanodiscs).
- C HDX-MS experiments of unbound BakDTM and BakDTM bound to Ni-nanodiscs in the inactive state and after activation with the Bid-derived BH3 peptide. HDX was mapped to the BakDTM structure and colored according to the measured exchange rates after the incubation times indicated. Blue colors correspond to a lower (less accessible) and red colors to a higher (more accessible) fractional uptake of deuterium.

Figure EV5.