Quantification of structure-property relationships for plant polyesters reveals suberin and cutin idiosyncrasies

Artur Bento,^a Carlos J.S. Moreira,^{a,±} Vanessa G. Correia,^{a,±} Rita Escórcio,^a Rúben Rodrigues,^a Ana S. Tomé,^a Nathalie Geneix,^b Johann Petit,^c Bénédicte Bakan,^b Christophe Rothan,^c Oleksandr O. Mykhaylyk,^d Cristina Silva Pereira^{a,*}

^aInstituto de Tecnologia Química e Biológica António Xavier, Universidade Nova de Lisboa (ITQB NOVA), Av. da República, 2780-

157, Oeiras, Portugal

^bINRAE, UR BIA 1268, 44316, Nantes, France

^cUMR 1332 BFP, INRAE, Univ. Bordeaux, F-33140 Villenave d'Ornon, France

^d Soft Matter Analytical Laboratory, Dainton Building, Department of Chemistry, The University of Sheffield, S3 7HF, UK [±] equally contributing authors; ^{*}corresponding author: Cristina Silva Pereira (spereira@itqb.unl.pt)

Supporting Information contains 26 pages including 28 Figures and 4 Tables.

<u>Index</u>

- **Supplemental Tables and Figures** (by order of appearance in the manuscript)
 - **Figure S1** Schematics of the extraction process.
- **Figure S2** –The capacity of the cholinium hexanoate to mediate the hydrolysis of distinct ester-type bonds, up to 24h at 100 °C with stirring, was analyzed by GC-MS. Standard compounds representative of five ester-types were used: a primary aliphatic ester (PAE, octyl octanoate), a secondary aliphatic ester (SAE, octan-4-yl octanoate), an aromatic ester (ArE, ethyl 4-hydroxy-3-methoxy-cinnamate), a carbohydrate ester (CarbE, sucrose monolaurate) and a glycerol ester (GlyE, glyceryl trioctanoate). PAE and GlyE kinetic curves were adapted from a dataset published before by us¹.
- Figure S3 Representative GC-MS chromatograms of the model compound sucrose monolaurate after reaction with cholinium hexanoate at increasing time points. Over time the peak assigned to dodecanoic acid hydrolysis product of sucrose monolaurate, increased. IS internal standard (hexadecane); 1 dodecanoic acid, methyl ester; 2 dodecanoic acid, trimethylsilyl ester.
- **Figure S4** The capacity of the cholinium hexanoate to mediate the hydrolysis of primary aliphatic esters bonds in a synthetic polymer, poly(1,4-butyl adipate), up to 24h at 100 °C with stirring, was

analyzed by NMR: a) ¹H NMR in DMSO- d_6 in different time points, b) representative of ¹H-¹³C HMBC NMR after 24 h of reaction. The dashed red circle mark the region of the spectrum where the signal assigned to the acid resulting from the hydrolysis would be noticed.

- **Table S1** Elemental analysis of tomato cutin, pepper cutin, cork suberin and potato suberin. Relative percentage of Nitrogen (N), Carbon (C), Hydrogen (H) and Oxygen (O) in the different samples.
- **Table S2** Total carbohydrate quantification of cutin and suberin samples.
- Figure S5 2D-¹H-¹H COSY-NMR spectrum (COrrelated SpectroscopY) of tomato cutin extracted using cholinium hexanoate (2 h) in DMSO-*d*₆ at 60 °C.
- Figure S6 2D-¹H-¹³C HSQC-NMR spectrum (Heteronuclear Single Quantum Coherence) of tomato cutin extracted using cholinium hexanoate (2 h) in DMSO-*d*₆ at 60 °C. The red rectangles mark signals used for the integration of specific functional groups, all of which showed no overlap of resonances with other chemical groups as shown by the bi-dimensional NMR correlations done.
- Figure S7 2D-¹H-¹³C HMBC-NMR spectrum (Heteronuclear Multiple Bond Correlation) of tomato cutin extracted using cholinium hexanoate (2 h) in DMSO- *d*₆ at 60 °C.
- Figure S8 2D-¹H-¹H COSY-NMR spectrum (COrrelated SpectroscopY) of pepper cutin isolated with cholinium hexanoate (2 h) in DMSO- *d*₆ at 60 °C.
- Figure S9 2D-¹H-¹³C HSQC-NMR spectrum (Heteronuclear Single Quantum Coherence) of pepper cutin isolated with cholinium hexanoate (2 h) in DMSO- *d*₆ at 60 °C. The red rectangles mark signals used for the integration of specific functional groups, all of which showed no overlap of resonances with other chemical groups as shown by the bi-dimensional NMR correlations done.
- Figure S10 2D-¹H-¹³C HMBC-NMR spectrum (Heteronuclear Multiple Bond Correlation) of pepper cutin isolated with cholinium hexanoate (2 h) in DMSO- *d*₆ at 60 °C.
- Figure S11 2D-¹H-¹H COSY-NMR spectrum (COrrelated SpectroscopY) of cork suberin in DMSOd₆ at 60 °C.
- Figure S12 2D-¹H-¹³C HSQC-NMR spectrum (Heteronuclear Single Quantum Coherence) of cork suberin in DMSO- *d*₆ at 60 °C.
- Figure S13 2D-¹H-¹³C HMBC-NMR spectrum (Heteronuclear Multiple Bond Correlation) of cork suberin isolated with cholinium hexanoate (2 h) in DMSO- *d*₆ at 60 °C.
- Figure S14 2D-¹H-¹H COSY-NMR spectrum (COrrelated SpectroscopY) of potato suberin in DMSOd₆ at 60 °C.
- Figure S15 2D-¹H-¹³C HSQC-NMR spectrum (Heteronuclear Single Quantum Coherence) of potato suberin in DMSO- *d*₆ at 60 °C.
- Figure S16 2D-¹H-¹³C HMBC-NMR spectrum (Heteronuclear Multiple Bond Correlation) of potato suberin isolated with cholinium hexanoate (2 h) in DMSO- *d*₆ at 60 °C.
- Table S3 Monomeric composition of cutin purified from tomato or pepper using cholinium hexanoate
 (2 h), identified by NMR. Chemical structures and ¹H and ¹³C chemical shifts (ppm) of the cutin

monomers. Signals not resolved in the NMR spectra due to overlapping signals or not yet assigned are not shown in this table. Most of the signals have been assigned before by us¹.

- Table S4 Monomeric composition of suberin extracted from cork or potato using cholinium hexanoate (2 h), identified by NMR. Chemical structures and ¹H and ¹³C chemical shifts (ppm) of the cutin monomers. Signals not resolved in the NMR spectra due to overlapping signals or not yet assigned are not shown in this table. Most of the signals have been assigned before by us².
- Figure S17 Quantification of crystalline cellulose concentration with ImageJ software, of cutin from tomato, pepper, Micro-tom wild type and mutants purified with cholinium hexanoate. The result was expressed in mg of crystalline cellulose per mg of cutin. The error bars represent the standard deviations of three independent measurements.
- **Figure S18** SEM micrographs of cutin samples extracted from Micro-Tom tomatoes, wild type (a) and the mutants *cus1* (b) and *gpat6* (c), using cholinium hexanoate extraction for 2h.
- Figure S19 2D-¹H-¹H COSY-NMR spectrum (COrrelated SpectroscopY) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) in DMSO- *d*₆ at 60 °C.
- Figure S20 2D-¹H-¹³C HSQC-NMR spectrum (Heteronuclear Single Quantum Coherence) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) in DMSO- *d*₆ at 60 °C. The red rectangles mark signals used for the integration of specific functional groups, all of which showed no overlap of resonances with other chemical groups as shown by the bi-dimensional NMR correlations done.
- Figure S21 2D-¹H-¹³C HMBC-NMR spectrum (Heteronuclear Multiple Bond Correlation) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) in DMSO- *d*₆ at 60 °C.
- Figure S22 2D-¹H-¹H COSY-NMR spectrum (COrrelated SpectroscopY) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) from the *cus1* mutant in DMSO- *d*₆ at 60 °C.
- Figure S23 2D-¹H-¹³C HSQC-NMR spectrum (Heteronuclear Single Quantum Coherence) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) from the *cus1* mutant in DMSO- *d*₆ at 60 °C. The red rectangles mark signals used for the integration of specific functional groups, all of which showed no overlap of resonances with other chemical groups as shown by the bi-dimensional NMR correlations done.
- Figure S24 2D-¹H-¹³C HMBC-NMR spectrum (Heteronuclear Multiple Bond Correlation) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) from the *cus1* mutant in DMSO- d_6 at 60 °C.
- Figure S25 2D-¹H-¹H COSY-NMR spectrum (COrrelated SpectroscopY) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) from the *gpat6* mutant in DMSO- *d*₆ at 60 °C.
- Figure S26 2D-¹H-¹³C HSQC-NMR spectrum (Heteronuclear Single Quantum Coherence) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) from the *gpat6* mutant in DMSO- d_6 at 60 °C. The red rectangles mark signals used for the integration of specific functional groups, all of which showed no overlap of resonances with other chemical groups as shown by the bi-dimensional NMR correlations done.

- Figure S27 2D-¹H-¹³C HMBC-NMR spectrum (Heteronuclear Multiple Bond Correlation) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) from the *gpat6* mutant in DMSO-*d*₆ at 60 °C.
- **Figure S28** Wide-ranging NMR spectral characterisation of Micro-tom cutins: wild-type (a), *cus1* (b) and *gpat6* (c) mutants; upon isolation with cholinium hexanoate (2 h). The ¹H NMR spectra of all samples (I); HSQC regions corresponding to aliphatics (II) and CH/CH2-X aliphatics (III). Some correlations (unlabeled) are uncertain or unidentified.



Figure S1 – Schematics of the extraction process



Figure S2 –The capacity of the cholinium hexanoate to mediate the hydrolysis of distinct ester-type bonds, up to 24h at 100 °C with stirring, was analyzed by GC-MS. Standard compounds representative of five ester-types were used: a primary aliphatic ester (PAE, octyl octanoate), a secondary aliphatic ester (SAE, octan-4-yl octanoate), an aromatic ester (ArE, ethyl 4-hydroxy-3-methoxy-cinnamate), a carbohydrate ester (CarbE, sucrose monolaurate) and a glycerol ester (GlyE, glyceryl trioctanoate). PAE and GlyE kinetic curves were adapted from a dataset published before by us¹.



Figure S3 - Representative GC-MS chromatograms of the model compound sucrose monolaurate after reaction with cholinium hexanoate at increasing time points. Over time the peak assigned to dodecanoic acid - hydrolysis product of sucrose monolaurate, increased.

IS – internal standard (hexadecane); 1 – dodecanoic acid, methyl ester; 2 – dodecanoic acid, trimethylsilyl ester.



Figure S4 - The capacity of the cholinium hexanoate to mediate the hydrolysis of primary aliphatic esters bonds in a synthetic polymer, poly(1,4-butyl adipate), up to 24h at 100 °C with stirring, was analyzed by NMR: a) ¹H NMR in DMSO-*d*₆ in different time points, b) representative of ¹H-¹³C HMBC NMR after 24 h of reaction. The dashed red circle marks the region of the spectrum where the signal assigned to the acid resulting from the hydrolysis would be noticed.

Table S1 – Elemental analysis of tomato cutin, pepper cutin, cork suberin and potato suberin. Relative percentage of Nitrogen (N), Carbon (C), Hydrogen (H) and Oxygen (O) in the different samples

	Sample	Tomato Cutin	Pepper Cutin	Cork Suberin	Potato suberin
Elemental Analysis	Nitrogen (N) ^a (%)	< 1	< 1	1.60 ± 0.08	2.33 ± 0.07
	Carbon (C) (%)	63.98 ± 0.16	67.63 ± 0.02	64.07 ± 0.02	60.65 ± 0.11
	Hydrogen (H) (%)	9.08 ± 0.12	9.79 ± 0.03	8.86 ± 0.06	8.54 ± 0.07
	Oxygen (O) (%)	26.96 ± 0.27	22.60 ± 0.04	26.08 ± 0.04	28.50 ± 0.11

a) corresponding peak areas were outside the calibration curve, the suberin values may be subject to error yet shown since the monomer 4-(2-aminoethyl)phenol contains N. Deviation from theoretical percentage (should be <0.4%);

Table S2 – Total carbohydrate quantification of cutin and suberin samples.

Samples	ng (carbohydrate)/mg (plant polyester)	carbohydrate content / %
Tomato cutin	0.489 ± 0.127	$4.89E^{-5} \pm 1.27E^{-5}$
Pepper cutin	0.124 ± 0.016	$1.25\mathrm{E}^{-5}\pm1.79\mathrm{E}^{-6}$
Cork suberin	0.21 ± 0.033	$2.10E^{-5} \pm 3.34E^{-6}$
Potato suberin	1.16 ± 0.31	$9.70E^{-5} \pm 5.51E^{-5}$
Micro-Tom wt	0.387 ± 0.072	$3.87E^{-5} \pm 7.19E^{-6}$
Micro-Tom cus1	1.022 ± 0.496	$1.02E^{-4} \pm 4.96E^{-5}$
Micro-Tom gpat6	0.541 ± 0.175	$5.53E^{-5} \pm 3.00E^{-5}$



Figure S5 – 2D-¹H-¹H COSY-NMR spectrum (COrrelated SpectroscopY) of tomato cutin extracted using cholinium hexanoate (2 h) in DMSO- d_6 at 60 °C.



Figure S6 – $2D^{-1}H^{-13}C$ HSQC-NMR spectrum (Heteronuclear Single Quantum Coherence) of tomato cutin extracted using cholinium hexanoate (2 h) in DMSO- d_6 at 60 °C. The red rectangles mark signals used for the integration of specific functional groups, all of which showed no overlap of resonances with other chemical groups as shown by the bi-dimensional NMR correlations done.



Figure S7 – 2D-¹H-¹³C HMBC-NMR spectrum (Heteronuclear Multiple Bond Correlation) of tomato cutin extracted using cholinium hexanoate (2 h) in DMSO- d_6 at 60 °C.



Figure S8 – 2D-¹H-¹H COSY-NMR spectrum (COrrelated SpectroscopY) of pepper cutin isolated with cholinium hexanoate (2 h) in DMSO- d_6 at 60 °C.



Figure S9 – $2D^{-1}H^{-13}C$ HSQC-NMR spectrum (Heteronuclear Single Quantum Coherence) of pepper cutin isolated with cholinium hexanoate (2 h) in DMSO- d_6 at 60 °C. The red rectangles mark signals used for the integration of specific functional groups, all of which showed no overlap of resonances with other chemical groups as shown by the bi-dimensional NMR correlations done.



Figure S10 – 2D-¹H-¹³C HMBC-NMR spectrum (Heteronuclear Multiple Bond Correlation) of pepper cutin isolated with cholinium hexanoate (2 h) in DMSO- d_6 at 60 °C.



Figure S11 – 2D-¹H-¹H COSY-NMR spectrum (COrrelated SpectroscopY) of cork suberin in DMSO- d_6 at 60 °C.



Figure S12 – 2D-¹H-¹³C HSQC-NMR spectrum (Heteronuclear Single Quantum Coherence) of cork suberin in DMSO- d_6 at 60 °C.



Figure S13 – 2D-¹H-¹³C HMBC-NMR spectrum (Heteronuclear Multiple Bond Correlation) of cork suberin isolated with cholinium hexanoate (2 h) in DMSO- d_6 at 60 °C.



Figure S14 – 2D-¹H-¹H COSY-NMR spectrum (COrrelated SpectroscopY) of potato suberin in DMSO- d_6 at 60 °C.



Figure S15 – 2D-¹H-¹³C HSQC-NMR spectrum (Heteronuclear Single Quantum Coherence) of potato suberin in DMSO- d_6 at 60 °C.



Figure S16 – 2D-¹H-¹³C HMBC-NMR spectrum (Heteronuclear Multiple Bond Correlation) of potato suberin isolated with cholinium hexanoate (2 h) in DMSO- d_6 at 60 °C.

Table S3 – Monomeric composition of cutin purified from tomato or pepper using cholinium hexanoate (2 h), identified by NMR. Chemical structures and ¹H and ¹³C chemical shifts (ppm) of the cutin monomers. Signals not resolved in the NMR spectra due to overlapping signals or not yet assigned are not shown in this table. Most of the signals have been assigned before by us¹.





¹³C NMR (201.42 MHz, DMSO-*d*6) δ 28.34 (C-2), 34.67 (C-1).

¹H NMR (800.33 MHz, DMSO-*d*6) δ 1.25 (m, 2H, H-2), 1.30 (m, 2H, H-3), 2.02 (m, 2H, H-1), 3.38 (m, 2H, H-4). ¹³C NMR (201.42 MHz, DMSO-*d*6) δ 28.34

(C-2), 34.67 (C-1), 36.64 (C-3), 74.31 (C-4).

ω-hydroxyacids



¹H NMR (800.33 MHz, DMSO-*d*6) δ 1.25 (m, 2H, H-2), 1.41 (m, 2H, H-3), 2.02 (m, 2H, H-1), 3.41 (m, 2H, H-4).

¹³C NMR (201.42 MHz, DMSO-*d*6) δ 28.34 (C-2), 32.03 (C-3), 34.67 (C-1), 67.64 (C-4).

¹H NMR (800.33 MHz, DMSO-*d*6) δ 1.25 (m, 2H, H-2), 1.42 (m, 2H, H-5), 1.41 (m, 2H, H-5), 1.43 (m, 2H, H-3), 2.02 (m, 2H, H-1), 2.82 (m, 2H, H-4), 3.41 (m, 2H, H-6) ¹³C NMR (201.42 MHz, DMSO-*d*6) δ 26.68 (C-3), 28.34 (C-2), 32.03 (C-5) 34.67 (C-1), 55.43 (C-4), 67.64 (C-6)

Polyhydroxyacids



¹H NMR (800.33 MHz, DMSO-*d*6) δ 1.25 (m, 2H, H-2), 1.30 (m, 2H, H-3), 1.41 (m, 2H, H-5), 2.02 (m, 2H, H-1), 3.38 (m, 2H, H-4), 3.41 (m, 2H, H-6). ¹³C NMR (201.42 MHz, DMSO-*d*6) δ 28.34 (C-2), 32.03 (C-5), 36.64 (C-3), 34.67 (C-1), 67.64 (C-6), 74.31 (C-4).

SECONDARY ALIPHATIC ESTERS



¹H NMR (800.33 MHz, DMSO-*d*6) δ 1.47 (m, 2H, H-4), 1.54 (m, 2H, H-2), 2.25 (m, 2H, H-1), 4.77 (m, 2H, H-3).

 ^{13}C NMR (201.42 MHz, DMSO-*d*6) δ 27.60 (C-2), 33.05 (C-4), 33.27 (C-1), 72.60 (C-3).

PRIMARY ALIPHATIC ESTERS



¹H NMR (800.33 MHz, DMSO-*d*6) δ 1.49 (m, 2H, H-4), 1.54 (m, 2H, H-2), 2.25 (m, 2H, H-1), 3.98 (m, 2H, H-3).

¹³C NMR (201.42 MHz, DMSO-*a*6) δ 24.02 (C-4), 27.60 (C-2), 33.27 (C-1), 63.05 (C-3). **Table S4** – Monomeric composition of suberin extracted from cork or potato using cholinium hexanoate (2 h), identified by NMR. Chemical structures and ¹H and ¹³C chemical shifts (ppm) of the cutin monomers. Signals not resolved in the NMR spectra due to overlapping signals or not yet assigned are not shown in this table. Most of the signals have been assigned before by us².





¹³C NMR (201.42 MHz, DMSO-*d*6) δ 26.13 (C-3), 27.07 (C-6), 29.86 (C-5), 29.35 (C-4), 36.40 (C-2), 129.98 (C-7), 172.20 (C-1) ¹H NMR (800.33 MHz, DMSO-*d*6) δ 1.24 (m, 2H, H-4), 1.43 (m, 2H, H-5), 1.46 (m, 2H, H-3), 2.03 (m, 2H, H-2), 2.80 (m, 2H, H-6) ¹³C NMR (201.42 MHz, DMSO-*d*6) δ 26.13 (C-3), 29.35 (C-4), 27.83 (C-5), 36.40 (C-2), 56.38 (C-6), 172.20 (C-1) ¹H NMR (800.33 MHz, DMSO-*d*6) δ 1.24 (m, 2H, H-4), 1.46 (m, 2H, H-3), 2.03 (m, 2H, H-2) ¹³C NMR (201.42 MHz, DMSO-*d*6) δ 26.13 (C-3), 29.35 (C-4), 36.40 (C-2), 172.20 (C-1)

SATURATED CHAIN





¹H NMR (800.33 MHz, DMSO-*d*6) δ 1.24 (m, 2H, H-4), 1.41 (m, 2H, H-5), 1.46 (m, 2H, H-3), 2.03 (m, 2H, H-2), 3.39 (m, 2H, H-6) ¹³C NMR (201.42 MHz, DMSO-*d*6) δ 26.13 (C-3), 29.35 (C-4), 32.95 (C-5), 36.40 (C-2), 61.32 (C-6), 172.20 (C-1)

 α,ω -diacids (n= 1-5)



¹H NMR (800.33 MHz, DMSO-*d*6) δ 1.24 (m, 2H, H-4), 1.46 (m, 2H, H-3), 2.03 (m, 2H, H-2) ¹³C NMR (201.42 MHz, DMSO-*d*6) δ 26.13 (C-3), 29.35 (C-4), 36.40 (C-2), 172.20 (C-1)

Fatty alcohols (n= 1-5)



¹H NMR (800.33 MHz, DMSO-*d*6) δ 0.86 (m, 3H, H-4), 1.24 (m, 2H, H-3), 1.41 (m, 2H, H-1), 3.39 (m, 2H, H-2) ¹³C NMR (201.42 MHz, DMSO-*d*6) δ 14.13 (C-4), 29.35 (C-3), 32.95 (C-2), 61.32 (C-1)

Fatty acids (n= 1-5)



¹H NMR (800.33 MHz, DMSO-*d*6) δ 0.86 (m, 3H, H-5), 1.24 (m, 2H, H-4), 1.46 (m, 2H, H-3), 2.03 (m, 2H, H-2) ¹³C NMR (201.42 MHz, DMSO-*d*6) δ 14.13 (C-5), 26.13 (C-3), 29.35 (C-4), 36.40 (C-2), 172.20 (C-1)

ACYLGLYCEROL ESTERS

1-monoacylglycerol (1-MAG)



¹H NMR (800.33 MHz, DMSO-*d*6) δ 3.37 (m, 2H, H-3), 3.65 (m, 1H, H-2), 3.93 (m, 1H, H-1), 4.03 (m, 1H, H-1') ¹³C NMR (201.42 MHz, DMSO-*d*6) δ 63.35 (C-3), 65.88 (C1,1'), 69.96 (C-2)

2-monoacylglycerol (2-MAG)



¹H NMR (800.33 MHz, DMSO-*d*6) δ 3.50 (m, 2H, H-1,3), 4.71 (m, 1H, H-2) ¹³C NMR (201.42 MHz, DMSO-*d*6) δ 60.17(C-1,3), 76.13 (C-2)

1,2-diacylglycerol (1,2-DAG)



¹H NMR (800.33 MHz, DMSO-*d*6) δ 3.50 (m, 2H, H-3), 4.08 (m, 2H, H-1), 4.27 (m, 2H, H-1'), 4.95 (m, 1H, H-2) ¹³C NMR (201.42 MHz, DMSO-*d*6) δ 60.17 (C-3), 62.76 (C-1'), 62.81 (C-1), 72.46 (C-2)

1,3-diacylglycerol (1,3-DAG)



¹H NMR (800.33 MHz, DMSO-*d*6) δ 3.87 (m, 1H, H-2), 3.99 (m, 4H, H-1,3) ¹³C NMR (201.42 MHz, DMSO-*d*6) δ 65.25 (C-1,3), 66.72 (C-2)

1,2,3-triacylglycerol (1,2,3-TAG)



¹H NMR (800.33 MHz, DMSO-*d*6) δ 4.12 (m, 2H, H-1,3),4.24 (m, 2H, H-1',3'), 5.17 (m, 1H, H-2) ¹³C NMR (201.42 MHz, DMSO-*d*6) δ 62.33 (C1'-3'), 62.40 (C1-3), 69.32 (C2)



Figure S17 – Quantification of crystalline cellulose concentration with ImageJ software of cutin from tomato, pepper and Micro-tom (wild type and mutants) purified with cholinium hexanoate. The result was expressed in mg of crystalline cellulose per mg of cutin in sample (a). The error bars represent the standard deviations of three independent measurements. Standard calibration curve for quantitative detection of crystalline cellulose based on direct dot-blot immunoassay using Image J software (b). The error bars represent the standard deviations of three independent measurements.



Figure S18 - SEM micrographs of cutin samples extracted from Micro-Tom tomatoes, wild type (a) and the mutants *cus1* (b) and *gpat6* (c), using cholinium hexanoate extraction for 2h.



Figure S19 – 2D-¹H-¹H COSY-NMR spectrum (COrrelated SpectroscopY) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) in DMSO- d_6 at 60 °C.



Figure S20 – 2D-¹H-¹³C HSQC-NMR spectrum (Heteronuclear Single Quantum Coherence) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) in DMSO- d_6 at 60 °C. The red rectangles mark signals used for the integration of specific functional groups, all of which showed no overlap of resonances with other chemical groups as shown by the bi-dimensional NMR correlations done.



Figure S21 – 2D-¹H-¹³C HMBC-NMR spectrum (Heteronuclear Multiple Bond Correlation) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) in DMSO- d_6 at 60 °C.



Figure S22 – 2D-¹H-¹H COSY-NMR spectrum (COrrelated SpectroscopY) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) from the *cus1* mutant in DMSO- d_6 at 60 °C.



Figure S23 – 2D-¹H-¹³C HSQC-NMR spectrum (Heteronuclear Single Quantum Coherence) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) from the *cus1* mutant in DMSO- d_6 at 60 °C. The red rectangles mark signals used for the integration of specific functional groups, all of which showed no overlap of resonances with other chemical groups as shown by the bi-dimensional NMR correlations done.



Figure S24 – 2D-¹H-¹³C HMBC-NMR spectrum (Heteronuclear Multiple Bond Correlation) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) from the *cus1* mutant in DMSO- d_6 at 60 °C.



Figure S25 – 2D-¹H-¹H COSY-NMR spectrum (COrrelated SpectroscopY) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) from the *gpat6* mutant in DMSO- d_6 at 60 °C.



Figure S26 – 2D-¹H-¹³C HSQC-NMR spectrum (Heteronuclear Single Quantum Coherence) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) from the *gpat6* mutant in DMSO- d_6 at 60 °C. The red rectangles mark signals used for the integration of specific functional groups, all of which showed no overlap of resonances with other chemical groups as shown by the bi-dimensional NMR correlations done.



Figure S27 – 2D-¹H-¹³C HMBC-NMR spectrum (Heteronuclear Multiple Bond Correlation) of Micro-Tom cutin isolated with cholinium hexanoate (2 h) from the *gpat6* mutant in DMSO- d_6 at 60 °C.



Figure S28 - Wide-ranging NMR spectral characterisation of Micro-tom cutins: wild-type (a), *cus1* (b) and *gpat6* (c) mutants; upon isolation with cholinium hexanoate (2 h). The ¹H NMR spectra of all samples (I); HSQC regions corresponding to aliphatics (II) and CH/CH2-X aliphatics (III). Some correlations (unlabelled) are uncertain or unidentified.

References

- Moreira, C. J. S.; Bento, A.; Pais, J.; Petit, J.; Escórcio, R.; Correia, V. G.; Pinheiro, Â.; Haliński, Ł. P.; Mykhaylyk, O. O.; Rothan, C.; et al. An Ionic Liquid Extraction That Preserves the Molecular Structure of Cutin Shown by Nuclear Magnetic Resonance. *Plant Physiol.* 2020, *184* (2), 592–606. https://doi.org/10.1104/pp.20.01049.
- (2) Correia, V. G.; Bento, A.; Pais, J.; Rodrigues, R.; Haliński, P.; Frydrych, M.; Greenhalgh, A.;
 Stepnowski, P.; Vollrath, F.; King, A. W. T.; et al. The Molecular Structure and Multifunctionality of the Cryptic Plant Polymer Suberin. *Mater. Today Bio* 2020, *5*, 100039. https://doi.org/10.1016/j.mtbio.2019.100039.