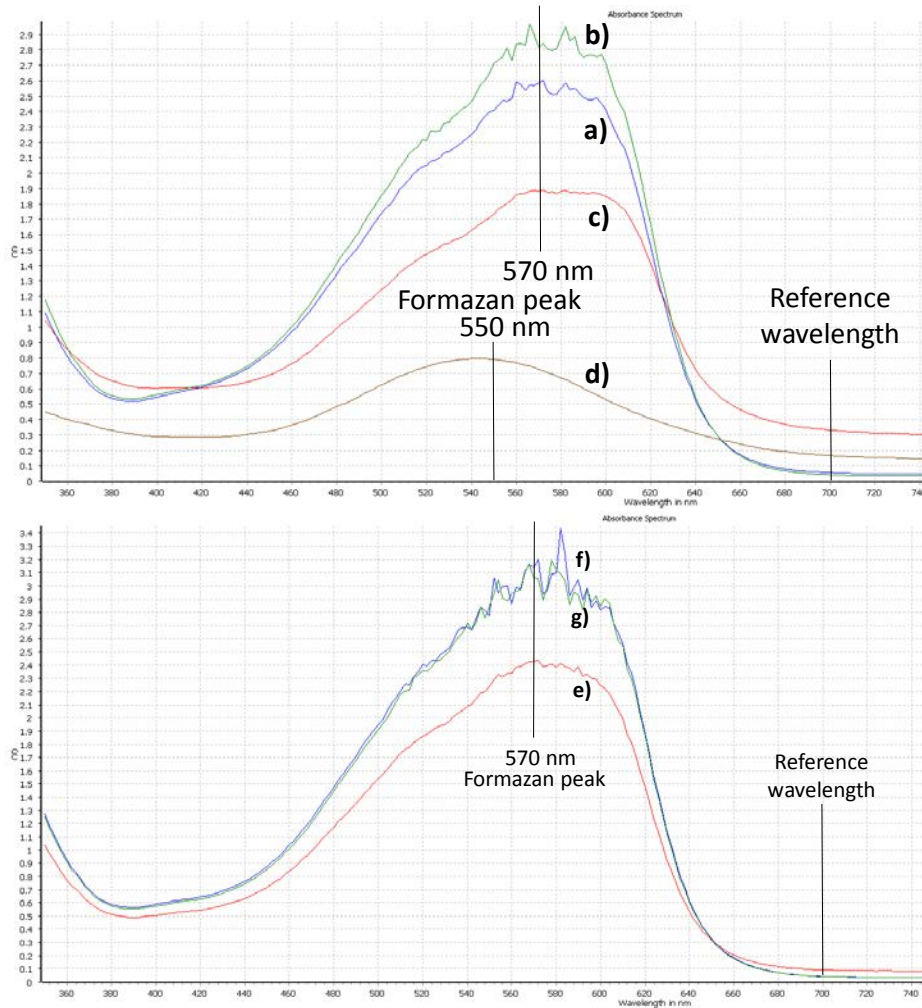


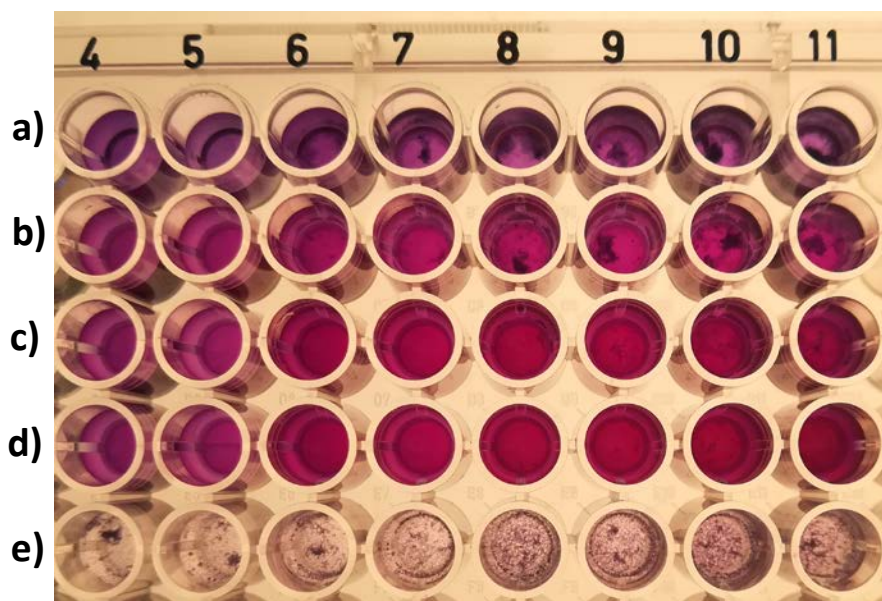
Supplementary material



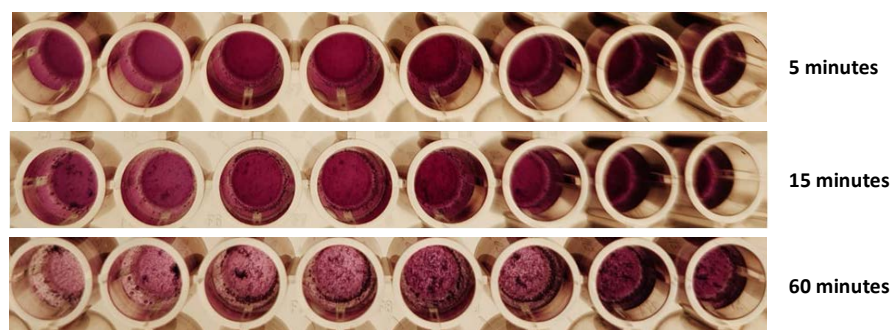
Suppl. Fig. 1. Absorption spectra of pure formazan dissolved in different solvents

Pure formazan crystals were weighed and dispersed in phosphate-buffered saline. Aliquots were serially diluted under constant stirring, and 100 μ l portions were added to triplicate wells in a 96-well plate, followed by the addition of 100 μ l of the tested solvent. Plates were kept on a shaker at 200 rpm. Spectra were recorded 60 minutes after adding the solvents.

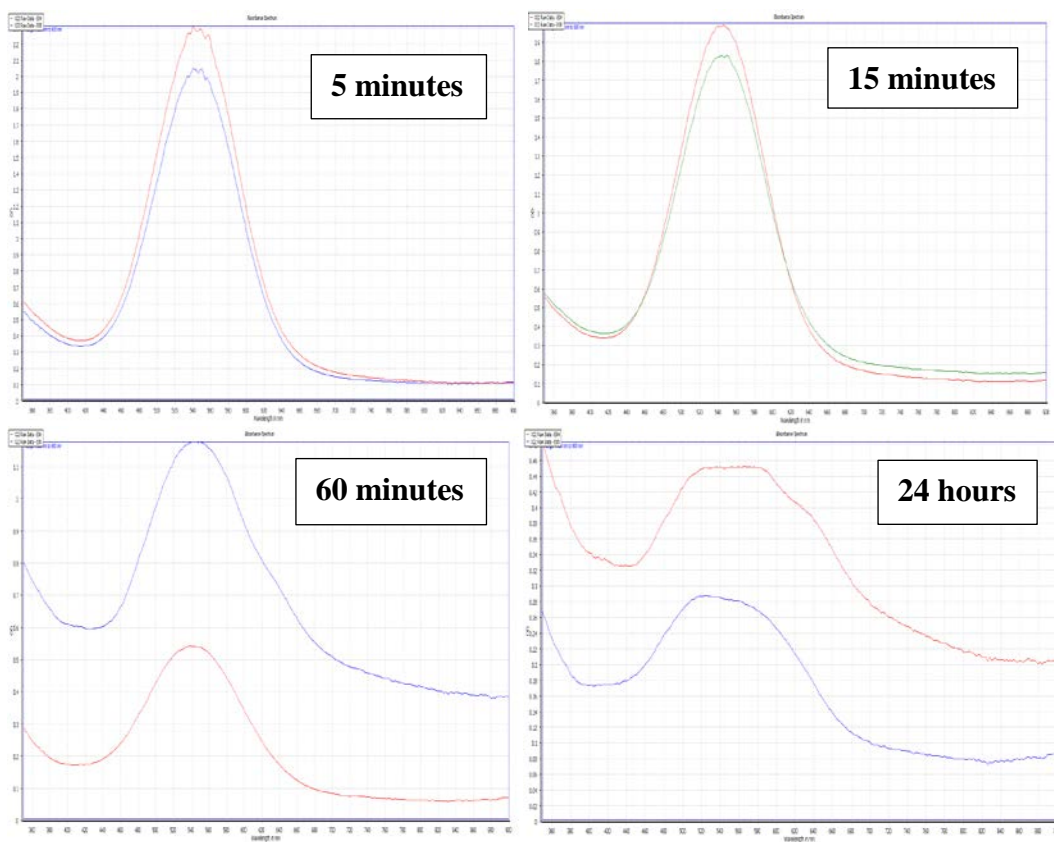
a) 10% SDS-buffered DMF; b) 5% SDS-buffered DMF; c) SDS-EDTA; d) ammonia-DMSO; e) 5% SDS-buffered DMSO; f) 10% SDS-buffered DMSO; g) 15% SDS-buffered DMSO



Suppl. Fig. S2. Image of a 96-well plate taken 60 minutes after the addition of the solvents
 Pure formazan was added at the quantities indicated below to a duplicate wells containing 100 μ l of PBS. Hundred μ l of selected solvents were added and the plate was shaken at 200 rpm.
Solvents: a) SDS-EDTA; b) 5% SDS- buffered DMF; c) 7.5% SDS-buffered DMF; d) 10% SDS-buffered DMF; e) ammonia-DMSO
Quantity of formazan (μ g per well): wells 4 & 5, 12.5 μ g; wells 6 & 7, 25.0 μ g; wells 8 & 9, 37.5 μ g; wells 10 & 11, 50.0 μ g

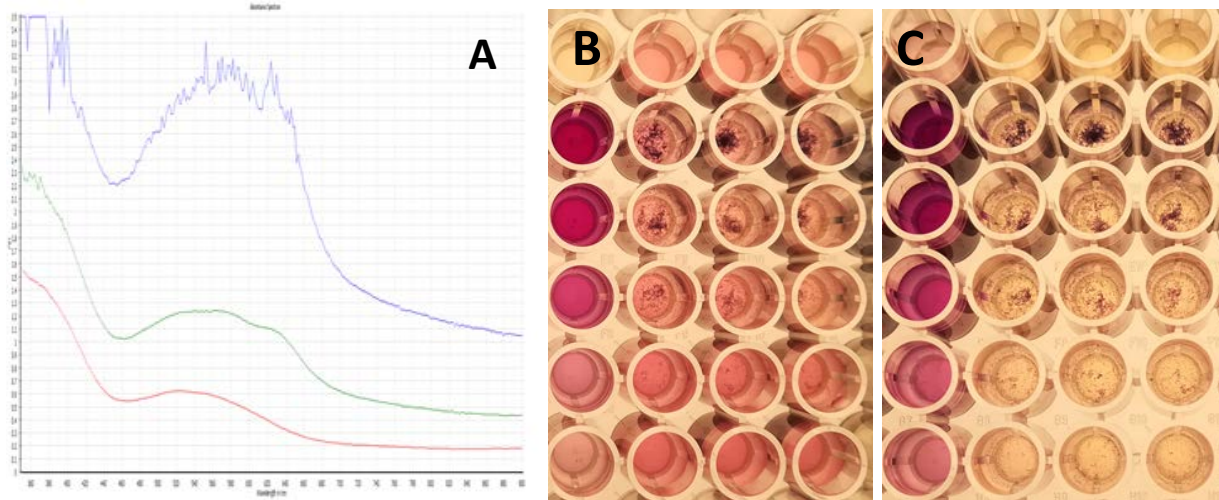


Suppl. Fig. S3. Time-dependent aggregation of formazan dissolved in ammonia-DMSO
 Pure formazan was added at the quantities indicated below to duplicate wells containing 100 μ l of PBS. Hundred μ l of solvents were added and the 96-well plates were shaken at 200 rpm. Images were taken 5, 15, and 60 minutes after adding the solvent.
Quantity of formazan added per well: 12.5 μ g - wells 1 & 2; 25.0 μ g - wells 3 & 4; 37.5 μ g - wells 5 & 6; 50.0 μ g - wells 7 & 8.

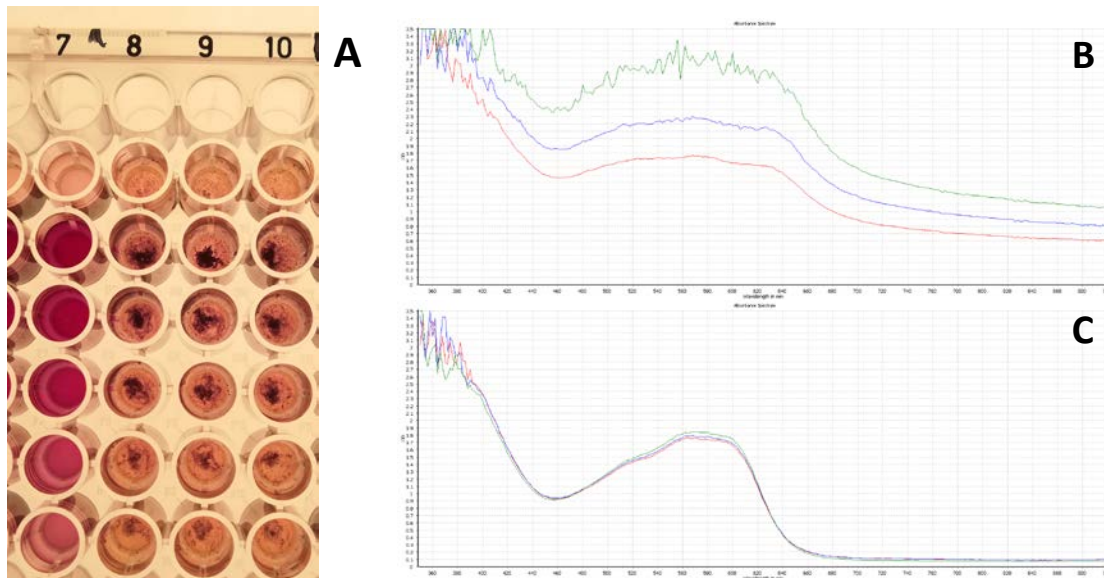


Suppl. Fig. S4. Time-dependent changes of the absorption spectrum of formazan dissolved in ammonia-DMSO

Spectra of duplicate wells were recorded 5, 15, 60 minutes, and 24 hours after adding ammonia-DMSO to wells containing 12.5 μg formazan in 100 μl PBS.



Suppl. Fig S5. Absorbance spectra of triplicate wells recorded 1 hour after the addition of ammonia-DMSO to wells containing 18×10^8 *E. coli* cells/ml subjected to the MTT assay (A). Images of a 96-well plate taken 1 hour (B) and 24 hours (C) after adding ammonia-DMSO. Wells contained medium (top row) or *E. coli* in the range 0.8×10^8 – 18×10^8 cells/ml. Ammonia-DMSO was added to solubilize formazan crystals after 30 min. of incubation with MTT solution.

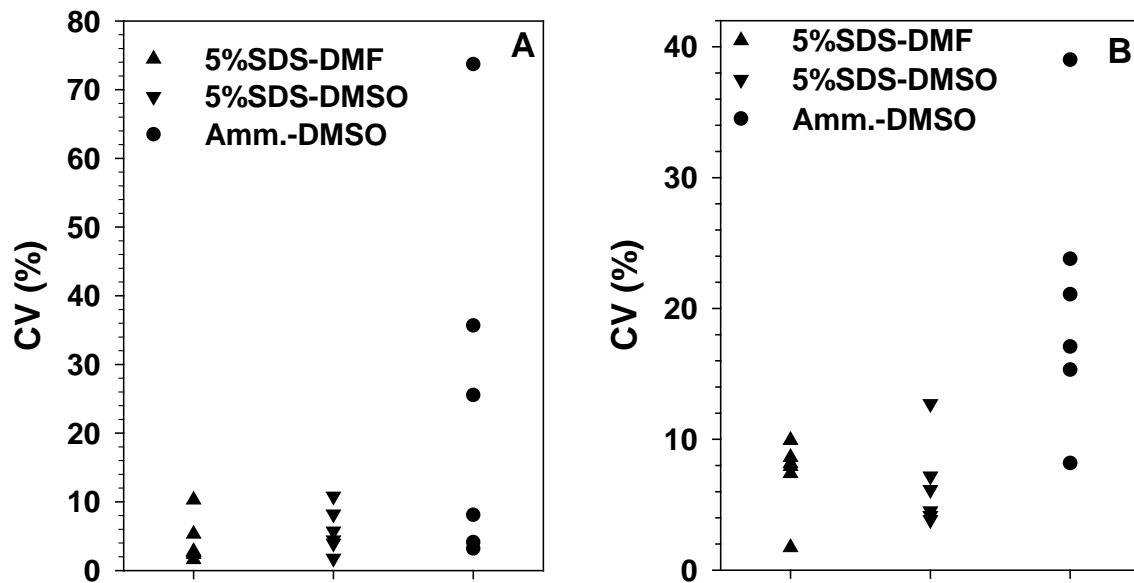


Suppl. Fig S6. Image of a 96-well plate (A) and absorbance spectra of triplicate wells B & C).

A) Wells containing 100 μ l of BHI medium (top row) or 100 μ l of *S. aureus* culture ($1.5 - 57 \times 10^8$ cells/ml) were subjected to the MTT assay. Formazan was solubilized with ammonia-DMSO. The image was captured 2 hours after addition of the solvent.

B) Absorption spectra of triplicate wells containing 8×10^8 *S. aureus* cells/ml subjected to the MTT assay and solubilized with ammonia-DMSO. Spectra were recorded 2 hours after the addition of the solvent.

C) Absorption spectra of triplicate wells containing 8×10^8 cells/ml, 2 hours after addition of 5% SDS-buffered DMSO. All other conditions were as in Panel B.



Suppl. Fig S7. Inter assay coefficient of variation (CV).
Panel A: *E. coli*; Panel B: *S. aureus*

<i>E. coli</i>	R ²	p
5% SDS-DMF	0.992	<0.001
5% SDS-DMSO	0.991	<0.001
Ammonia-DMSO	0.788	0.018
<i>S. aureus</i> (up to 5.7x10⁹ cells/ml)		
5% SDS-DMF	0.769	0.022
5% SDS-DMSO	0.841	0.010
Ammonia-DMSO	0.288	0.272
<i>S. aureus</i> (up to 5.7x10⁹ cells/ml)		
5% SDS-DMF	0.979	0.010
5% SDS-DMSO	0.990	0.005
Ammonia-DMSO	0.921	0.040

Suppl. Table 1. Coefficient of determination (R²) and p values obtained by linear regression analysis for *E. coli* and *S. aureus* cultures subjected to the MTT assay and solubilized with the listed solvents.

Data were derived from Figs. 3 D and 4 D. For *S. aureus*, R² and p values were calculated for the entire range of cell densities (0 - 5.7x10⁹ cells/ml) and for the linear part of the graph (0 – 8x10⁸ cells/ml)