

Supporting Information

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Handheld device for selective benzene sensing over toluene
and xylene

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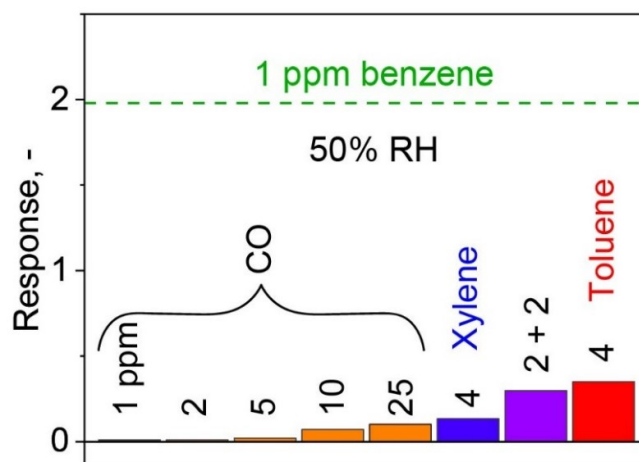


Fig. S1: Benzene selectivity at higher confounder concentrations. Detector responses to 1 – 25 ppm CO (orange), 4 ppm *m*-xylene (blue) and toluene (red), and a mixture of 2 ppm (each) *m*-xylene and toluene (purple) at 50% RH. For comparison, the response to 1 ppm benzene (dashed line) is shown.

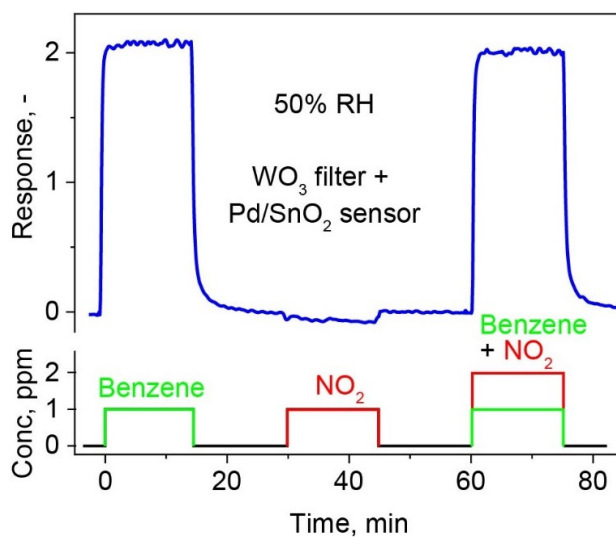


Fig. S2: Benzene selectivity over NO₂. Detector response when exposed subsequently to 1 ppm benzene ($0 \leq t \leq 15$ min), NO₂ ($30 \leq t \leq 45$ min) and their mixture ($60 \leq t \leq 75$ min) at 50% RH.

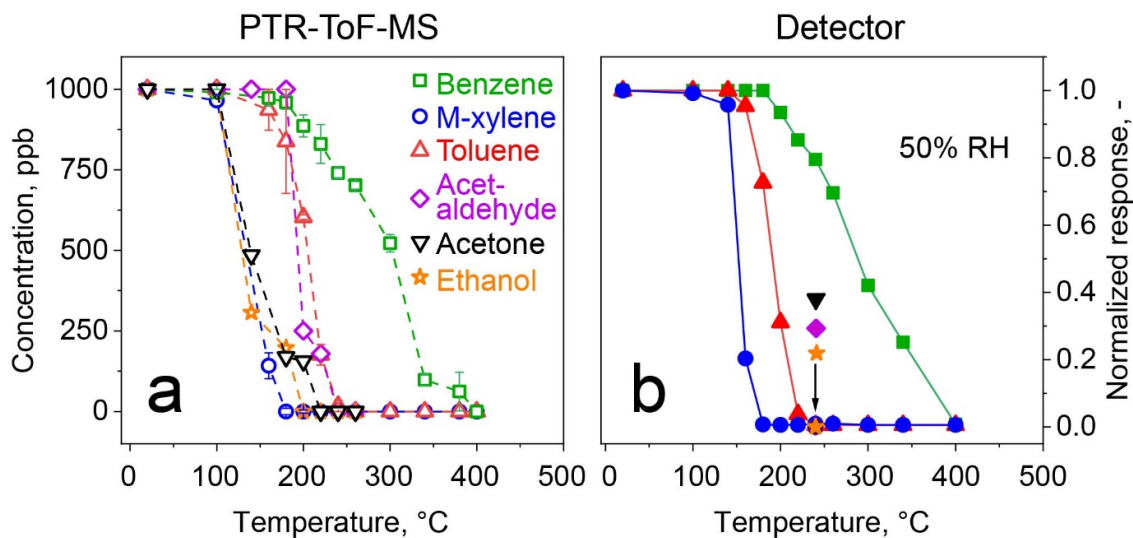


Fig. S3: Conversion on the WO_3 catalyst. Catalytic conversion of 1 ppm benzene (squares), *m*-xylene (circles), toluene (triangles), acetaldehyde (diamonds), acetone (inverse triangles) and ethanol (stars) over WO_3 , as quantified by **a)** PTR-ToF-MS and **b)** the sensor as a function of the WO_3 temperature. Error bars for benzene, toluene and *m*-xylene in (a) correspond to $n = 3$ identically prepared WO_3 particle filters. Note that the sensor response in (b) was normalized to its value at 25 °C for better comparison to PTR-ToF-MS.

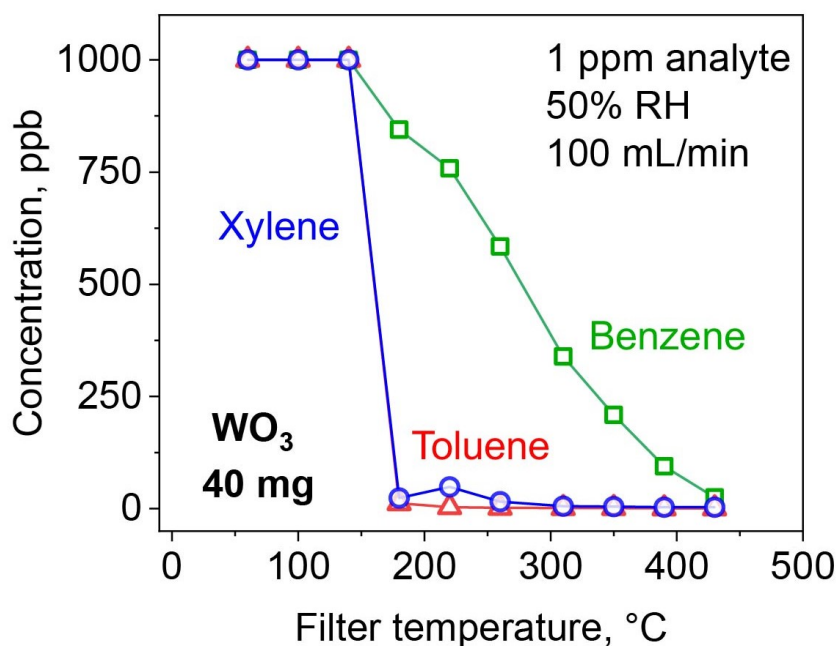


Fig. S4: Catalytic performance of the miniaturized filter. Concentration of 1 ppm *m*-xylene (circles), toluene (triangles) and benzene (squares) after a 40 mg WO_3 filter at 50% RH as a function of its temperature, as measured by PTR-ToF-MS.

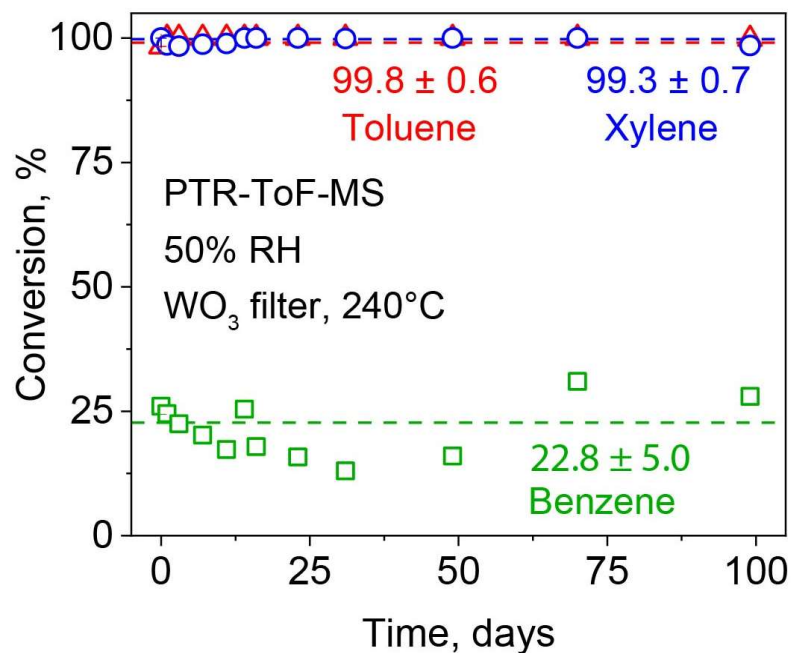


Fig. S5: Catalyst stability. Conversion of a mixture of 1 ppm benzene (squares), toluene (triangles) and *m*-xylene (circles) at 50% RH over a WO_3 filter after storage at ambient condition for 100 days, as measured by PTR-ToF-MS.

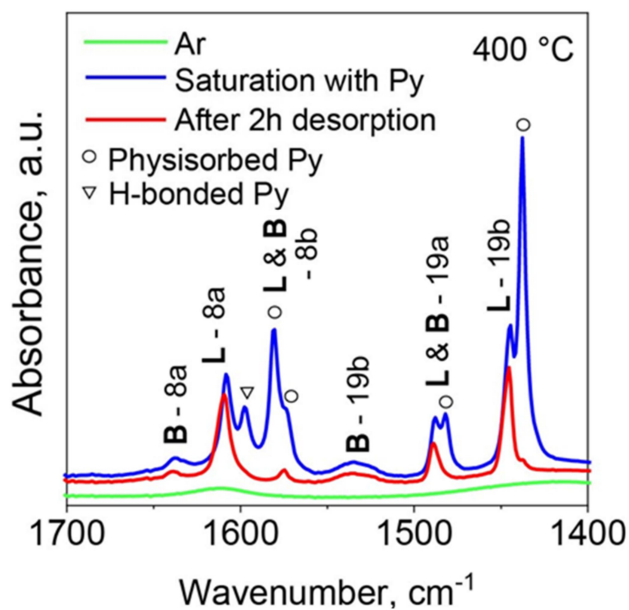


Fig. S6: IR measurements of WO_3 nanoparticles after pyridine adsorption. IR profiles when pretreated at 400°C under argon (green), saturated with pyridine (blue) and after desorption for 2 h (red). H-bonded pyridine (triangles), physisorbed pyridine (circles) and characteristic vibration modes of Brønsted (**B**) and Lewis (**L**) acid sites are indicated.

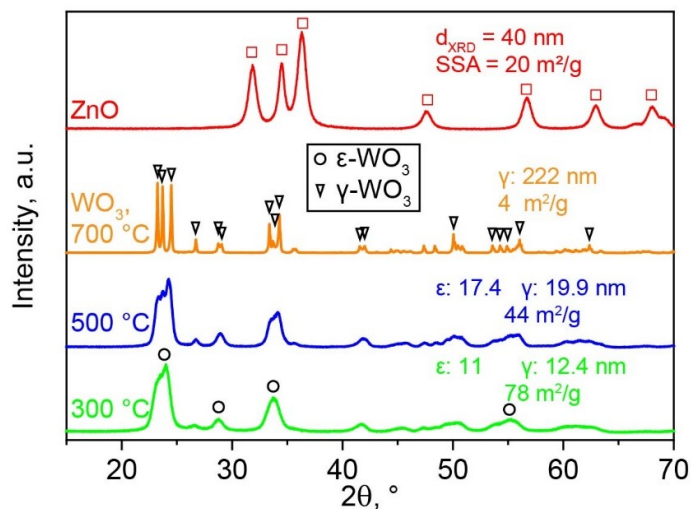


Fig. S7: XRD analysis of WO₃ at different annealing temperatures and ZnO. XRD patterns and reference peaks of the flame-made WO₃ annealed at 300 °C (green), 500 °C (blue), 700 °C (orange) and ZnO (red). Peak locations for monoclinic γ- (triangles), ε-WO₃ (circles) and wurtzite ZnO (squares) are indicated. Crystal sizes (d_{XRD}) as well as N₂ adsorption-measured specific surface areas (SSA) are indicated.

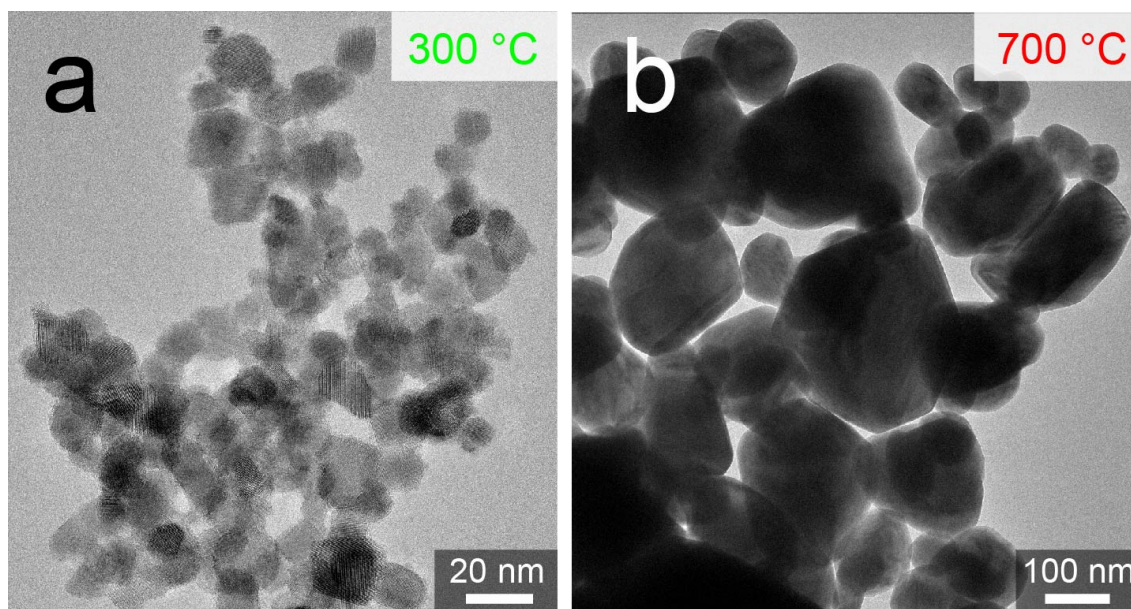


Fig. S8: Morphology characterization of WO₃ at different annealing temperatures. TEM images of WO₃ annealed at **a)** 300 °C and **b)** 700 °C.

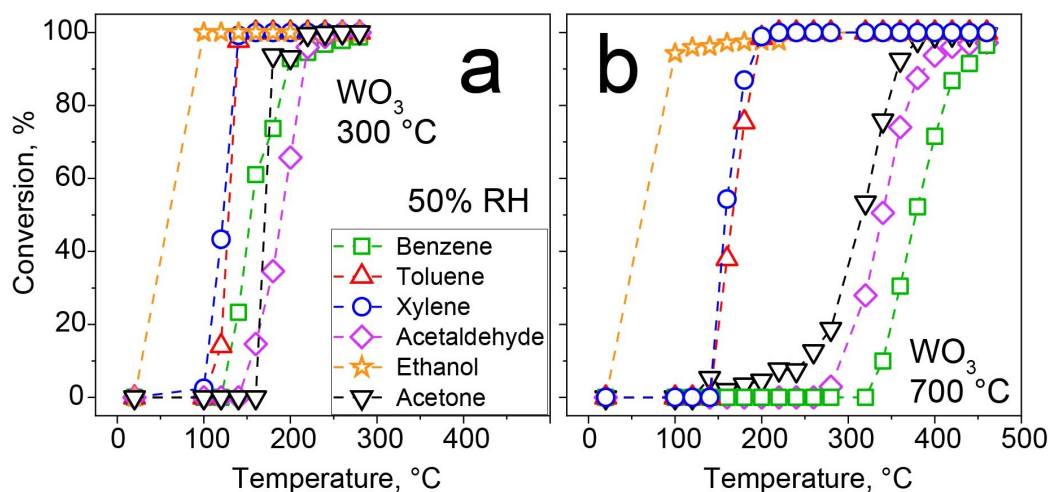


Fig. S9: Catalytic performance of differently annealed WO_3 . Catalytic conversion of 1 ppm benzene (squares), *m*-xylene (circles), toluene (triangles), acetaldehyde (diamonds), acetone (inverted triangles) and ethanol (stars) over WO_3 annealed at **a)** 300 °C and **b)** 700 °C at 50% RH, as measured by PTR-ToF-MS.

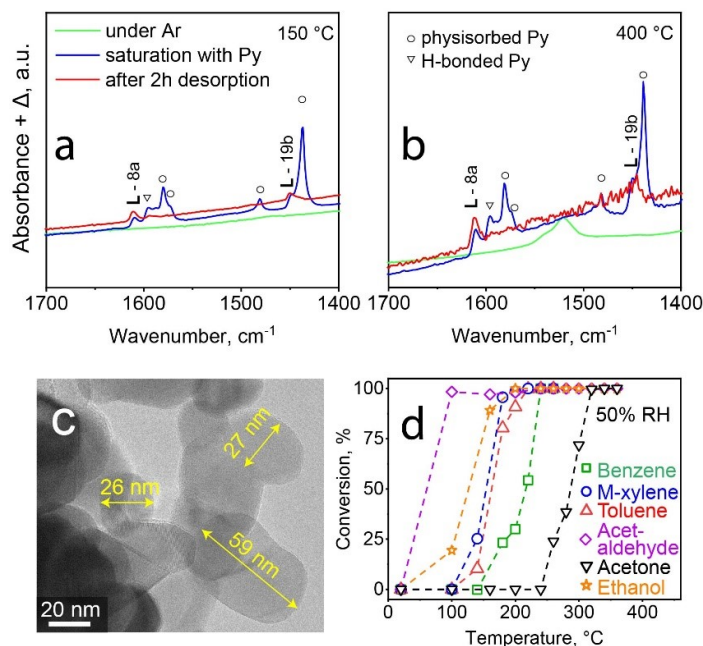


Fig. S10: ZnO material and catalytic characterization. IR profiles when pretreated at 150 °C **a)** and 400 °C **b)** under argon (green), saturated with pyridine (blue) and after desorption for 2 h (red). H-bonded pyridine (triangles), physisorbed pyridine (circles) and characteristic vibration modes of Lewis (L) acid sites are indicated. **c)** HRTEM bright field image of the flame-made ZnO particles. Diameters of some particles are indicated. **d)** Catalytic conversion of 1 ppm benzene (squares), *m*-xylene (circles), toluene (triangles), acetaldehyde (diamonds), acetone (inverted triangles) and ethanol (stars) over ZnO at 50% RH, as measured by PTR-ToF-MS. The ZnO catalyst was tested at a catalyst surface area of 3.6 m² that is identical to the WO_3 .

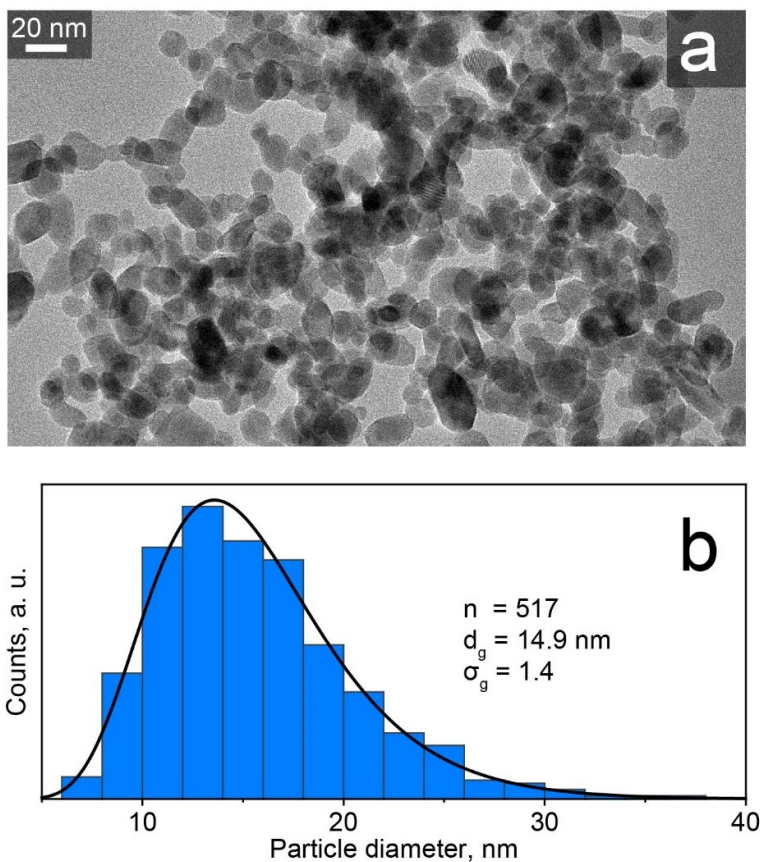


Fig. S11: WO₃ nanoparticle images and size analysis. a) TEM bright field image. **b)** Particle size distribution with a mean geometric diameter (d_g) and respective geometric deviation (σ_g) of $n = 517$ particles.

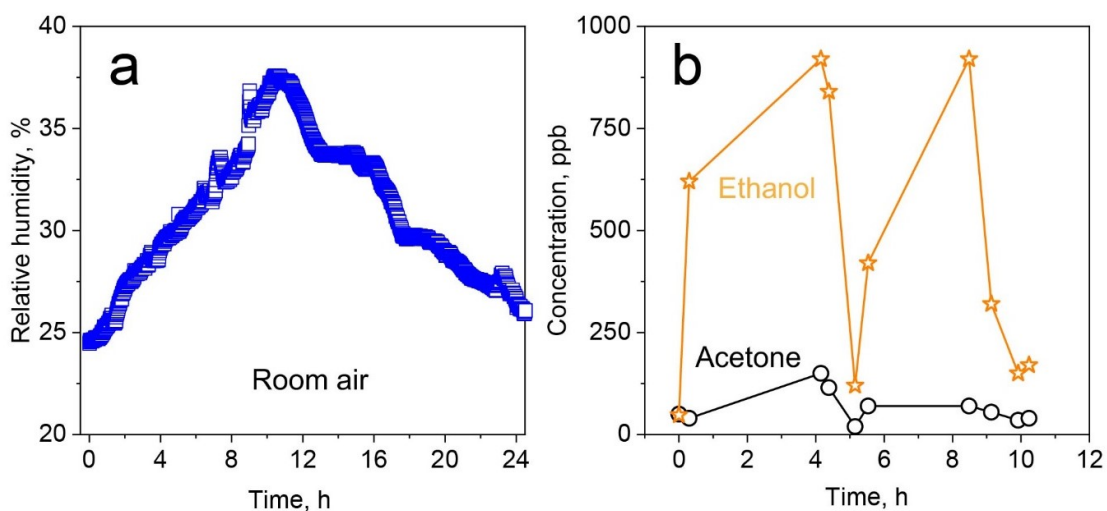


Fig. S12: Humidity and analyte variations during indoor air measurement. a) Indoor air RH as measured with a commercial sensor (Sensirion SHT2x) over 24 h. **b)** Corresponding ethanol (stars) and acetone (circles) concentrations determined by PTR-ToF-MS.

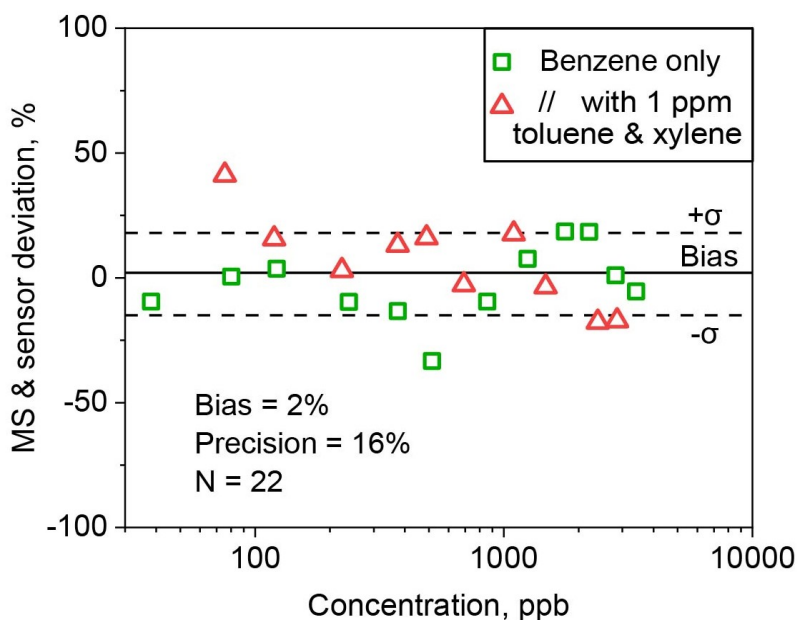


Fig. S13: Detector bias and precision. Relative deviation between the detector and PTR-ToF-MS as a function of the concentration for room air spiked with benzene only (squares) and with additional 1 ppm toluene & *m*-xylene (triangles) for $n = 22$ samples. Bias and precision (σ) are indicated.

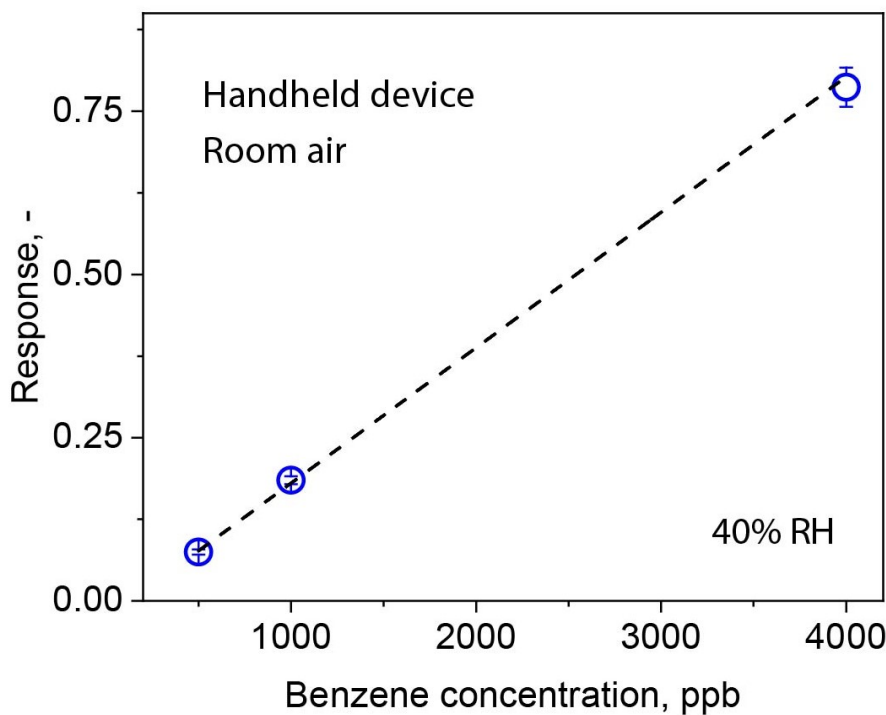


Fig. S14: Hand-held detector calibration. Detector response when exposed to calibrated gas standards at 40% RH. Error bars correspond to $n = 3$ subsequent measurements.