

Supporting Information: Capturing free-radical polymerization by synergistic *ab-initio* calculations and topological reactive molecular dynamics

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Extra details on the TraPPE-UA force field; Templates used in the matching topology algorithm.

TraPPE-UA Force Field

The TraPPE-UA¹⁻⁶ force fields parameters were obtained by site <http://trappe.oit.umn.edu>. In the tables are listed all force field parameters.

Table S1: Non bonded interaction parameters $v_{NB}(r_{ij}) = 4\epsilon_{ij}[(\frac{r_{ij}}{\sigma_{ij}})^{12} - (\frac{r_{ij}}{\sigma_{ij}})^6] + \frac{q_i q_j}{4\pi\epsilon_0 r_{ij}}$; $\sigma_{ij} = \frac{1}{2}(\sigma_{ii} + \sigma_{jj})$; $\epsilon_{ij} = (\epsilon_{ii}\epsilon_{jj})^{1/2}$

ID	Atom	$\epsilon/k_B(K)$	$\sigma(\text{\AA})$	q(e)
1	[CH ₂]=CH _x	85.0	3.675	0.0
2	CH _x =[C](sp ²)-CH _y (sp ²)	22.0	3.850	0.0
3	CH _x -O-[C]=O	40.0	3.820	0.4
4	CH _x -[O]-C=O	55.0	2.800	-0.25
5	CH _x -[CH ₂]-O-CH _y	46.0	3.95	0.25
6	CH _x -[CH ₂]-CH _x	46.0	3.95	0.0
7	CH _x -O-C=[O]	79.0	3.050	-0.4
8	[CH ₃]-CH _x	98.0	3.75	0.0

Table S2: Bonds interaction parameters: $U_B(r_{ij}) = K(r_{ij} - r_0)^2$

ID	Atom	$r_0(\text{\AA})$	k(kcal/mol)
1	CH _x =CH _y	1.33	300
2	CH _x (sp ²)-(CH _y =O)	1.52	300
3	O-(C=O)	1.344	300
4	CH _x -O	1.41	300
5	CH _x -CH _y	1.54	300
6	OC=O	1.2	300

Table S3: Bending interaction parameters: $U_{bend}(\theta) = \frac{k_\theta}{2}(\theta - \theta_0)^2$

ID	Atom	$\theta_0(\text{deg})$	k_θ (K/rad ²)
1	CH _x =(C)-CH _y	119.7	70420
2	O-(C)[=O]-CH _x ,	110.0	70600
3	O=(C)[-OCH _x]-CH _x	125.0	62500
4	C[=O]-O-CH _x	115.0	62500
5	CH _x -(CH _y)-O	112.0	50300
6	CH _x -(CH ₂)-CH _y	114.0	62500

Table S4: Dihedral parameters for $U_{tor}^1(\phi) = c_0 + c_1 \cos(\phi) + c_2 \cos(2\phi) + c_3 \cos(3\phi) + c_4 \cos(4\phi)$

ID	Atom	c_0/k_B	c_1/k_B	c_2/k_B	c_3/k_B	c_4/k_B (K)
1	<chem>CH2=(C)-(C)[=O]-O</chem>	823.03	47.91	-773.13	1.99	0.00
2	<chem>CH2=(C)-(C)[-O]=O</chem>	823.03	-47.91	-773.13	-1.99	0.00
3	<chem>CHx-(O)-(C)(sp2)-C(sp2)</chem>	1820.74	417.41	-1373.14	30.19	0.00
4	<chem>CHx-(CH2)-(O)-C(sp2)</chem>	2029.99	-751.83	-538.95	-22.10	-51.27
6	<chem>O-(C)(sp2)-(C)(sp2)-CH3</chem>	195.19	149.30	164.38	-24.12	28.45
8	<chem>CHx-(O)-(C)=O</chem>	1820.74	-417.41	-1373.14	-30.19	0.00
9	<chem>O=(C)-(C)(sp2)-CH3</chem>	195.19	-149.30	164.38	24.12	28.45

Table S5: Top table: dihedral parameters for $U_{tor}^2(\phi) = c_0 + c_1(1 + \cos(\phi)) + c_2(1 - \cos(2\phi)) + c_3(1 + \cos(3\phi))$. Bottom table: dihedral parameters for $U_{tor}^3(\phi) = \sum_{n=1}^4 A_n \cos^{n-1}(\phi)$

Id	Atom	c_0/k_B	c_1/k_B	c_2/k_B	c_3/k_B (K)
5	<chem>CHx-(CH2)-(CH2)-O</chem>	0.00	176.62	-53.34	769.93
7	<chem>CHx-(CH2)-(CH2)-CHy</chem>	0.00	355.03	-68.19	791.32
10	<chem>CHx-(CH2)-(C)=O</chem>	2035.58	-736.90	57.84	-293.23

ID	Atom	A_1/k_B	A_2/k_B	A_3/k_B	A_4/k_B (K)
11	<chem>CHx-CH2-C(=O)-O</chem>	373.05	919.04	268.15	1737.21

Templates used in the matching topology algorithm

In the templates files a certain number of atoms can represent the "edges" of the reaction whose close atoms topology does not need to be updated. In particular, in each HDDMA monomer, two half monomers are available to react since each half monomer contains a reactive ($-\text{CH}=\text{CH}_2$) vinyl group, while the edge points are represented by the most peripherals atoms with respect to the vinyl units. In the following, three templates are needed to allow the reaction to proceed. The details of the three templates are reported in Figures S1 and S2.

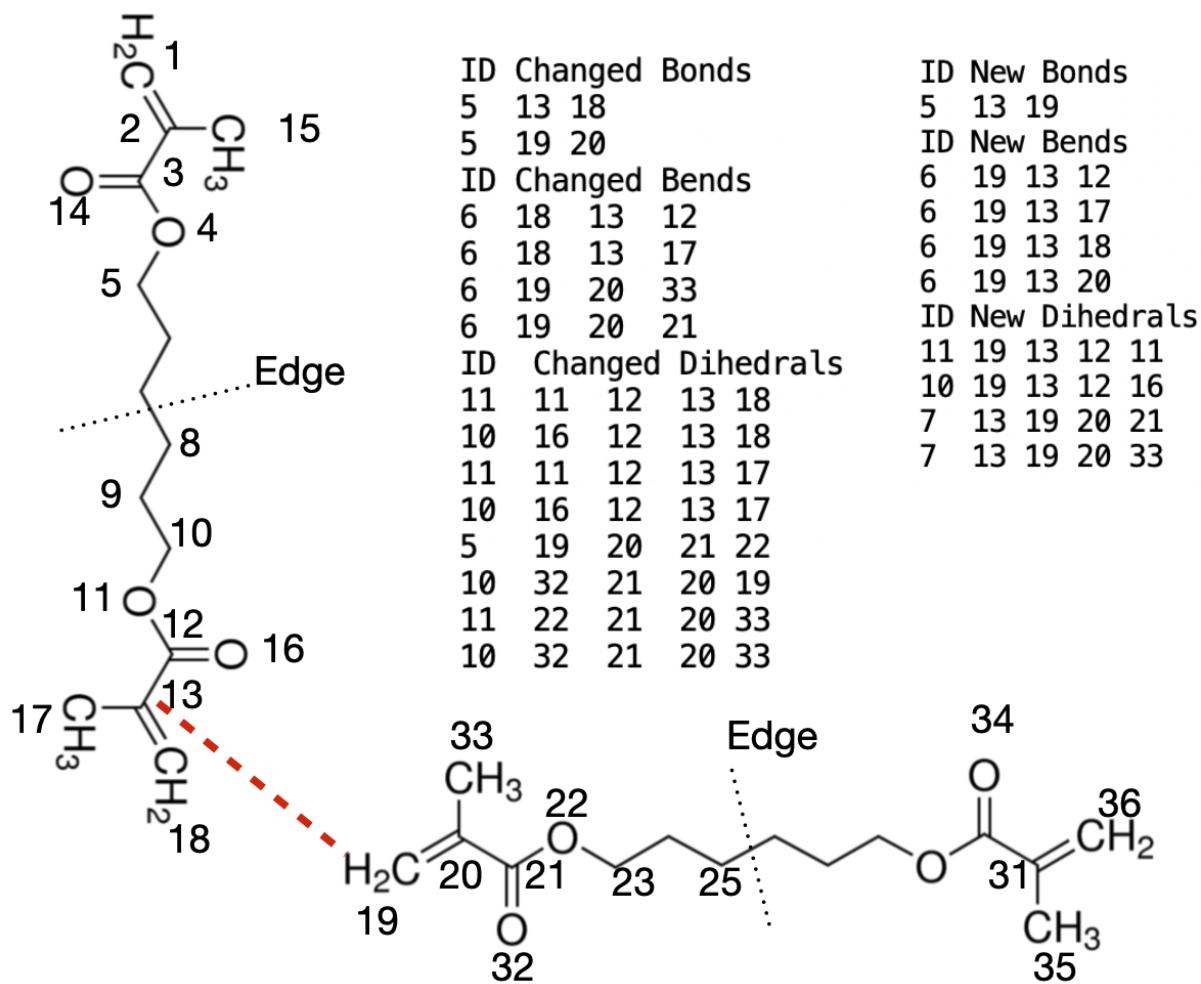


Figure S1: Edges are represented by black dotted line and between the two reacting carbons a red dotted line is present. Topology changes done in the reaction of two monomers.

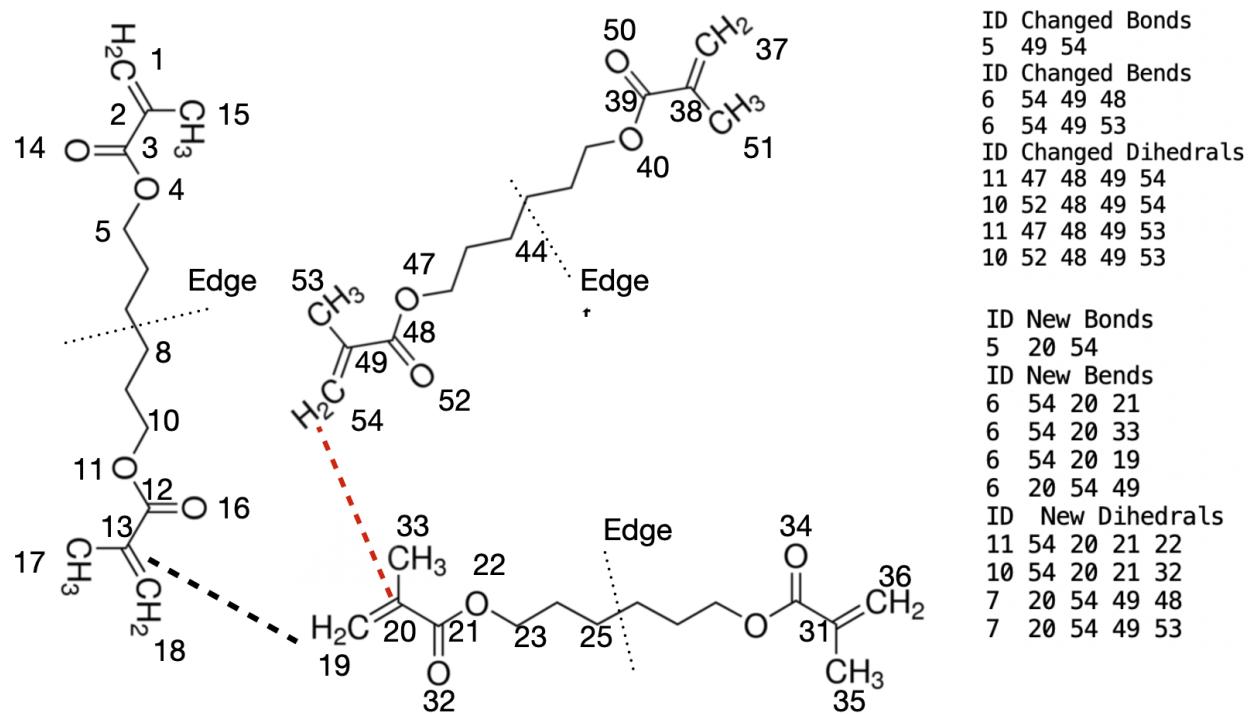


Figure S2: Edges are represented by black dotted line and between the two reacting carbons a red dotted line is present. Topology changes done in the reaction of one monomer with a just reacted monomer.

References

- (1) Martin, M. G.; Siepmann, J. I. Transferable potentials for phase equilibria. 1. United-atom description of n-alkanes. *J. Phys. Chem. B* **1998**, *102*, 2569–2577.
- (2) Wick, C. D.; Martin, M. G.; Siepmann, J. I. Transferable potentials for phase equilibria. 4. United-atom description of linear and branched alkenes and of alkylbenzenes. *J. Phys. Chem. B* **2000**, *104*, 8008–8016.
- (3) Chen, B.; Poto, J. J.; Siepmann, J. I. Monte Carlo calculations for alcohols and their mixtures with alkanes. Transferable potentials for phase equilibria. 5. United-atom description of primary, secondary and tertiary alcohols. *J. Phys. Chem. B* **2001**, *105*, 3093–3104.
- (4) Stubbs, J. M.; Poto, J. J.; Siepmann, J. I. Transferable potentials for phase equilibria. 6. United-atom description for ethers, glycols, ketones and aldehydes. *J. Phys. Chem. B* **2004**, *108*, 17596–17605.
- (5) Kamath, G.; Robinson, J.; Poto, J. J. Application of TraPPE-UA force field for determination of vapor-liquid equilibria of carboxylate esters. *Fluid Phase Equilibria* **2006**, *240*, 46–55.
- (6) Maerzke, K. A.; Schultz, N. E.; Ross, R. B.; Siepmann, J. I. TraPPE-UA force field for acrylates and Monte Carlo simulations for their mixtures with alkanes and alcohols. *J. Phys. Chem. B* **2009**, *113*, 6415–6425.