

Supporting Information

Palladium and Nickel Catalyzed Suzuki Cross-Coupling with Alkyl Fluorides

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1. General information

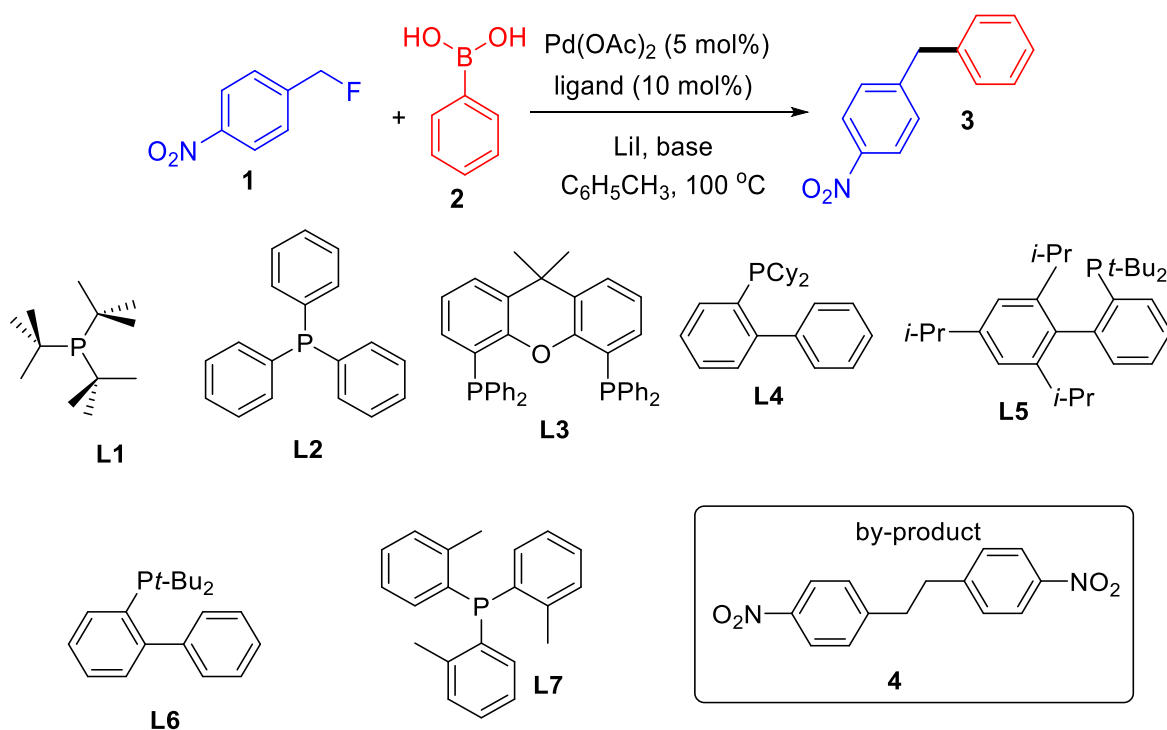
Commercially available organofluorines, ligands, catalysts, metal salts, boronic acids, reagents and solvents were used as purchased without further purification. The solvents were stored over 4Å molecular sieves prior to use.

NMR spectra were obtained at 400 MHz (^1H NMR), 376 MHz (^{19}F NMR) and 100 MHz (^{13}C NMR) in deuterated chloroform. Chemical shifts are reported in ppm relative to TMS. All reaction products were purified by column chromatography on silica gel (particle size 40-63 μm) as described below.

Single crystal X-ray analysis was performed at 100 K using a Siemens platform diffractometer with graphite monochromated Mo-K α radiation ($\lambda = 0.71073 \text{ \AA}$). Data were integrated with the Bruker SAINT program. Structure solution and refinement were performed using the SHELXT/PC suite and ShelXle. Intensities were corrected for Lorentz and polarization effects and an empirical absorption correction was applied using Blessing's method as incorporated into the program SADABS. Non-hydrogen atoms were refined with anisotropic thermal parameters and hydrogen atoms were included in idealized positions.

2. Optimization of the Pd-catalyzed Suzuki cross-coupling

Screening of phosphine ligands, base, temperature and reaction time

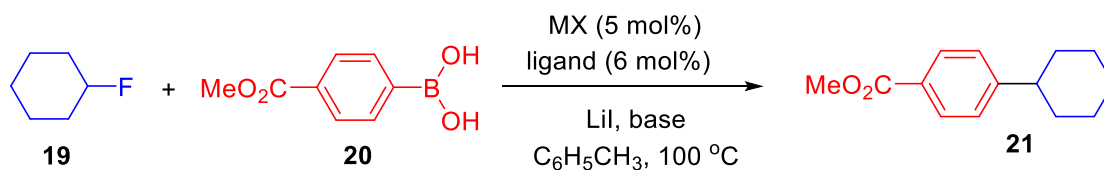


Entry	Ligand	Base	Time (h)	Temp. (°C)	Yield (%)
1	L2	Cs ₂ CO ₃	36	100	68
2	L2	K ₂ CO ₃	18	100	28
3	L2	K ₃ PO ₄	18	100	42
4	L1	K ₂ CO ₃	24	25	0
5	L1	K ₂ CO ₃	18	80	33
6	L1	K ₂ CO ₃	2	90	29 ^a
7	L1	K ₃ PO ₄	18	80	33
8	L1	K ₃ PO ₄	18	100	40
9	L1	Cs ₂ CO ₃	18	70	43
10	L1	Cs ₂ CO ₃	18	100	60 ^b
11	L1	Cs ₂ CO ₃	18	100	54 ^c
12	L3	Cs ₂ CO ₃	18	100	54
13	L4	Cs ₂ CO ₃	18	100	57
14	L5	Cs ₂ CO ₃	18	100	29
15	L6	Cs ₂ CO ₃	18	100	51
16	L7	Cs ₂ CO ₃	18	100	92
17	L7	Cs ₂ CO ₃	18	100	95 ^d
18	L7	Cs ₂ CO ₃	18	100	91 ^e

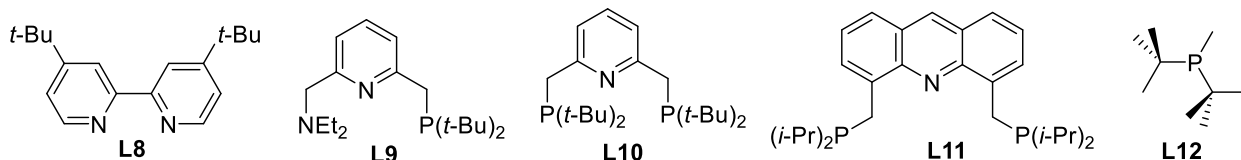
Conditions: **1** (0.2 mmol), **2** (0.6 mmol), LiI (0.3 mmol), Pd(OAc)₂ (5 mol%), ligand (10 mol%) in 2.0 mL solvent. ^aMicrowave reaction, ^b28% of **4** isolated, ^c*m*-xylene used as a solvent, ^d0.4 mmol of **2** used, ^e0.3 mmol of **2** used.

3. Optimization of the Ni-catalyzed Suzuki cross-coupling

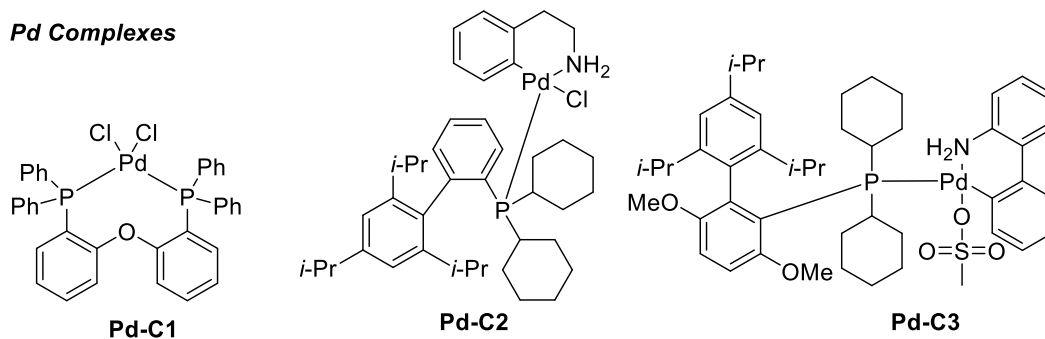
Screening of Ni complexes, base, solvent, temperature and reaction time



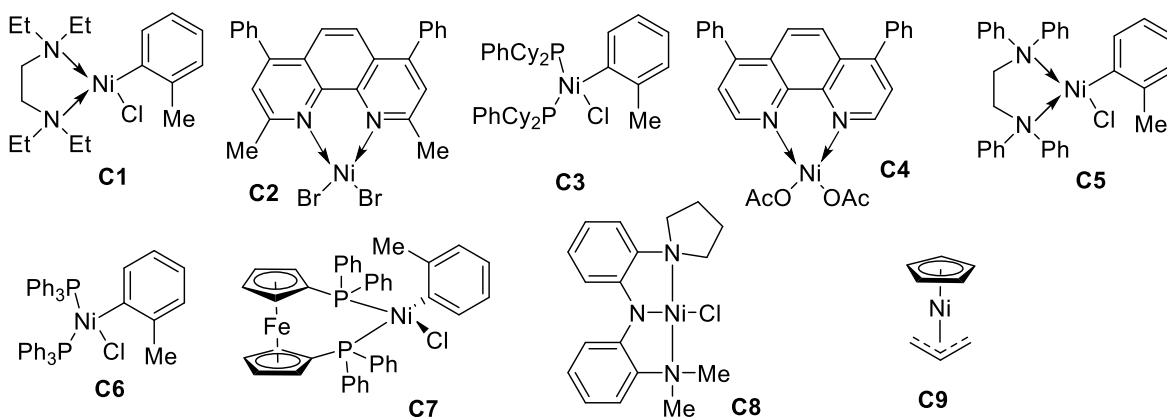
Ligands



Pd Complexes



Ni Complexes



Entry	Catalyst	Base	Solvent	Time (h)	Yield (%)
1	L7 Pd(OAc) ₂	Cs ₂ CO ₃	Toluene	24	0
2	- Pd(PPh ₃) ₄	Cs ₂ CO ₃	Toluene	24	<5
3	- Pd-C1	Cs ₂ CO ₃	Toluene	24	0
4	- Pd-C2	Cs ₂ CO ₃	Toluene	24	0
5	- Pd-C3	Cs ₂ CO ₃	Toluene	24	0
6	- Ni(PPh ₃) ₄	Cs ₂ CO ₃	Toluene	24	12
7	L8 Ni(NO ₃) ₂ ·6H ₂ O	Cs ₂ CO ₃	Toluene	24	24

8	L8	NiCl ₂	Cs ₂ CO ₃	Toluene	24	12
9	-	NiCl ₂ (PCy ₃) ₂	Cs ₂ CO ₃	Toluene	24	<5
10	-	NiCl ₂ (PPh ₃) ₂	Cs ₂ CO ₃	Toluene	24	<5
11	-	Ni(dppe)Cl ₂	Cs ₂ CO ₃	Toluene	24	<5
12	-	Ni(dppf)Cl ₂	Cs ₂ CO ₃	Toluene	24	<5
13	L8	NiBr ₂	Cs ₂ CO ₃	Toluene	24	15
14	L8	NiBr ₂ ·DME	Cs ₂ CO ₃	Toluene	24	12
15	L8	NiBr ₂ ·Diglyme	Cs ₂ CO ₃	Toluene	24	8
16	L8	NiI ₂	Cs ₂ CO ₃	Toluene	24	36
17	L8	Ni(COD) ₂	Cs ₂ CO ₃	Toluene	24	57
18	L8	Ni(ClO ₄) ₂	Cs ₂ CO ₃	Toluene	24	30
19	L8	NiO	Cs ₂ CO ₃	Toluene	24	0
20	L9	NiI ₂	Cs ₂ CO ₃	Toluene	24	30
21	L9	Ni(COD) ₂	Cs ₂ CO ₃	Toluene	24	48
22	L8	Ni(COD) ₂	Cs ₂ CO ₃	Toluene	48	71
23	L8	Ni(COD) ₂	K ₂ CO ₃	Toluene	48	56
24	L8	Ni(COD) ₂	K ₃ PO ₄	Toluene	48	45
25	L8	Ni(COD) ₂	KO <i>t</i> -Bu	Toluene	48	54
26	L8	Ni(COD) ₂	Cs ₂ CO ₃	IPA	48	0
27	L8	Ni(COD) ₂	Cs ₂ CO ₃	Dioxane	48	0
28	L8	Ni(COD) ₂	Cs ₂ CO ₃	DMF	48	0
29	L8	Ni(COD) ₂	Cs ₂ CO ₃	<i>t</i> -BuOH	48	0
30	L9	Ni(COD) ₂	Cs ₂ CO ₃	Toluene	48	15
31	L10	Ni(COD) ₂	Cs ₂ CO ₃	Toluene	48	0
32	L11	Ni(COD) ₂	Cs ₂ CO ₃	Toluene	48	13
33	L12	Pd(OAc) ₂	KO <i>t</i> -Bu	Toluene	24	0
34	-	C1	Cs ₂ CO ₃	Toluene	48	36
35	-	C2	Cs ₂ CO ₃	Toluene	48	28
36	-	C3	Cs ₂ CO ₃	Toluene	48	38
37	-	C4	Cs ₂ CO ₃	Toluene	48	0
38	-	C5	Cs ₂ CO ₃	Toluene	48	30
39	-	C6	Cs ₂ CO ₃	Toluene	48	35
40	-	C7	Cs ₂ CO ₃	Toluene	48	25
41	-	C8	Cs ₂ CO ₃	Toluene	48	12
42	-	C9	Cs ₂ CO ₃	Toluene	48	<5

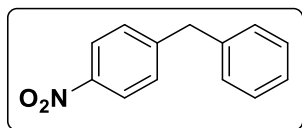
Conditions: **19** (0.2 mmol), **20** (0.4 mmol), LiI (0.3 mmol), **MX** (5 mol%) and ligand (6 mol%) or metal complex (5 mol%) in 2.0 mL solvent at 100 °C.

4. General procedure for the Pd-catalyzed Suzuki cross-coupling

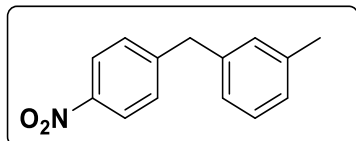
Method A: Pd(OAc)₂ (5 mol%), P(*o*-tolyl)₃ (10 mol%), alkyl fluoride (0.2 mmol, 1.0 eq), boronic acid (0.4 mmol, 2.0 eq), LiI (0.3 mmol, 1.5 eq) and Cs₂CO₃ (0.4 mmol, 2.0 eq) were added into 2 mL of dry toluene under nitrogen atmosphere. The reaction mixture was stirred at 100 °C in an oil bath and stirred at this temperature for 18 hours in the dark. The solvent was removed and the crude products were purified by silica flash column chromatography as described below.

Method B: Pd(OAc)₂ (5 mol%), P(*o*-tolyl)₃ (10 mol%), alkyl fluoride (0.4 mmol, 2.0 eq), boronic acid (0.2 mmol, 1.0 eq), LiI (0.6 mmol, 3.0 eq) and Cs₂CO₃ (0.4 mmol, 2 eq) were added into 2 mL of dry toluene under nitrogen atmosphere. The reaction mixture was stirred at 25 °C for 3 hours and then heated to 100 °C in an oil bath and stirred at this temperature for another 18 hours in the dark. The solvent was removed and the crude products were purified by silica flash column chromatography as described below.

4.1. Product purification and characterization

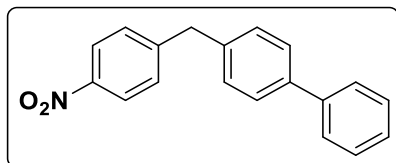


1-Benzyl-4-nitrobenzene (3). Compound **3** was obtained after column purification using hexanes/ethyl acetate (19:1) as mobile phase as a colorless oil in 95% yield (40 mg, 0.19 mmol) from 1-(fluoromethyl)-4-nitrobenzene (31 mg, 0.2 mmol) and phenylboronic acid (49 mg, 0.4 mmol) by following the general procedure described above (Method A). *R*_f = 0.5 (hexanes/EtOAc, 4:1); ¹H NMR (400 MHz, chloroform-*d*) δ = 8.13 (d, *J* = 8.4 Hz, 2H), 7.36 – 7.28 (m, 4H), 7.24 (m, 1H), 7.17 (d, *J* = 6.8 Hz, 2H), 4.07 (s, 2H); ¹³C NMR (100 MHz, chloroform-*d*) δ = 149.0, 146.7, 139.3, 129.8, 129.1, 128.9, 126.9, 123.9, 41.9; HRMS (ESI-TOF) *m/z*: [M+H]⁺ calcd for C₁₃H₁₂NO₂ 214.0863, found 214.0866.

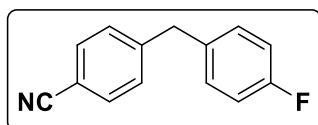


1-Methyl-3-(4-nitrobenzyl)benzene (5). Compound **5** was obtained after column purification using hexanes/ethyl acetate (49:1) as mobile phase as a colorless solid in 96% yield (43 mg, 0.192 mmol) from 1-(fluoromethyl)-4-nitrobenzene (31 mg, 0.2 mmol) and *m*-tolylboronic acid (54 mg,

0.4 mmol) by following the general procedure described above (Method A). $R_f = 0.6$ (hexanes/EtOAc, 19:1); $^1\text{H NMR}$ (400 MHz, chloroform-*d*) $\delta = 8.14$ (d, $J = 7.8$ Hz, 2H), 7.41 – 7.30 (m, 2H), 7.21 (dd, $J = 7.8, 7.6$ Hz, 1H), 7.07 (d, $J = 7.6$ Hz, 1H), 6.98 (d, $J = 7.8$ Hz, 2H), 4.04 (s, 2H), 2.33 (s, 3H); $^{13}\text{C NMR}$ (100 MHz, chloroform-*d*) $\delta = 149.1, 146.6, 139.2, 138.6, 129.8, 129.8, 128.8, 127.6, 126.1, 123.9, 41.8, 21.5$; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{14}\text{H}_{14}\text{NO}_2$ 228.1019, found 228.1024.

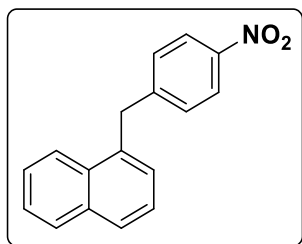


4-(4-Nitrobenzyl)-1,1'-biphenyl (6). Compound **6** was obtained after column purification using hexanes/ethyl acetate (49:1) as mobile phase as a colorless solid in 92% yield (53 mg, 0.184 mmol) from 1-(fluoromethyl)-4-nitrobenzene (31 mg, 0.2 mmol) and [1,1'-biphenyl]-4-ylboronic acid (80 mg, 0.4 mmol) by following the general procedure described above (Method A). $R_f = 0.6$ (hexanes/EtOAc, 19:1); $^1\text{H NMR}$ (400 MHz, chloroform-*d*) $\delta = 8.14$ (d, $J = 7.7$ Hz, 2H), 7.57 – 7.53 (m, 4H), 7.42 (dd, $J = 7.6, 7.5$ Hz, 2H), 7.37 – 7.31 (m, 3H), 7.23 (d, $J = 7.6$ Hz, 2H), 4.10 (s, 2H); $^{13}\text{C NMR}$ (100 MHz, chloroform-*d*) $\delta = 148.9, 146.7, 140.7, 139.9, 138.4, 129.8, 129.5, 128.9, 127.6, 127.4, 127.1, 123.9, 41.5$; HRMS (ESI-TOF) m/z : $[\text{M}-\text{H}]^-$ calcd for $\text{C}_{19}\text{H}_{14}\text{NO}_2$ 288.1030, found 288.1025.

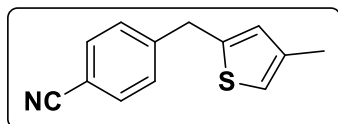


4-(4-Fluorobenzyl)benzonitrile (7). Compound **7** was obtained after column purification using hexanes/ethyl acetate (19:1) as mobile phase as a colorless solid in 86% yield (36 mg, 0.172 mmol) from 4-(fluoromethyl)benzonitrile (27 mg, 0.2 mmol) and (4-fluorophenyl)boronic acid (56 mg, 0.4 mmol) by following the general procedure described above (Method A). $R_f = 0.3$ (hexanes/EtOAc, 4:1); $^1\text{H NMR}$ (400 MHz, chloroform-*d*) $\delta = 7.57$ (d, $J = 8.2$ Hz, 2H), 7.26 (d, $J = 8.2$ Hz, 2H), 7.11 (dd, $J = 8.5, 5.4$ Hz, 2H), 7.03 – 6.94 (m, 2H), 4.00 (s, 2H); $^{13}\text{C NMR}$ (100 MHz, chloroform-*d*) $\delta = 161.8$ (d, $J_{\text{C-F}} = 245.0$ Hz), 146.6 (d, $J_{\text{C-F}} = 1.0$ Hz), 135.1 (d, $J_{\text{C-F}} = 3.2$ Hz), 132.5, 130.5 (d, $J_{\text{C-F}} = 7.9$ Hz), 129.7, 119.0, 115.7 (d, $J_{\text{C-F}} = 21.4$ Hz), 110.4, 41.2; $^{19}\text{F NMR}$

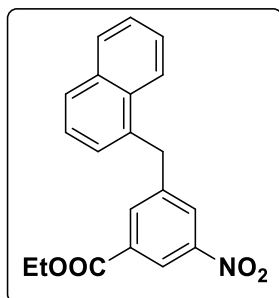
(376 MHz, chloroform-*d*) $\delta = -116.3$ (m, 1F); HRMS (ESI-TOF) m/z : $[M+H]^+$ calcd for $C_{14}H_{11}FN$ 212.0870, found 212.0873.



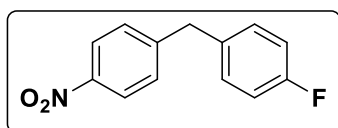
1-(4-Nitrobenzyl)naphthalene (8). Compound **8** was obtained after column purification using hexanes/ethyl acetate (32:1) as mobile phase as a colorless solid in 87% yield (46 mg, 0.174 mmol) from 1-(fluoromethyl)naphthalene (32 mg, 0.2 mmol) and (4-nitrophenyl)boronic acid (67 mg, 0.4 mmol) by following the general procedure described above (Method A). $R_f = 0.6$ (hexanes/EtOAc, 4:1); 1H NMR (400 MHz, chloroform-*d*) $\delta = 8.12$ (d, $J = 8.7$ Hz, 2H), 7.90 (dd, $J = 7.5, 1.9$ Hz, 1H), 7.84 (dd, $J = 8.7, 8.7$ Hz, 2H), 7.53 – 7.42 (m, 3H), 7.37 – 7.30 (m, 3H), 4.54 (s, 2H); ^{13}C NMR (100 MHz, chloroform-*d*) $\delta = 148.7, 146.7, 134.8, 134.2, 131.9, 129.5, 129.0, 128.1, 127.8, 126.5, 126.0, 125.7, 124.0, 123.9, 39.1$; HRMS (ESI-TOF) m/z : $[M+H]^+$ calcd for $C_{17}H_{14}NO_2$ 264.1019, found 264.1019.



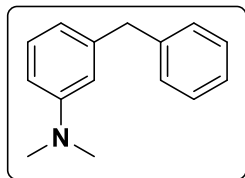
4-((4-Methylthiophen-2-yl)methyl)benzonitrile (9). Compound **9** was obtained after column purification using hexanes/ethyl acetate (19:1) as mobile phase as a colorless solid in 96% yield (41 mg, 0.192 mmol) from 4-(fluoromethyl)benzonitrile (27 mg, 0.2 mmol) and (4-methylthiophen-2-yl)boronic acid (57 mg, 0.4 mmol) by following the general procedure described above (Method A). $R_f = 0.4$ (hexanes/EtOAc, 4:1); 1H NMR (400 MHz, chloroform-*d*) $\delta = 7.59$ (d, $J = 8.2$ Hz, 2H), 7.34 (d, $J = 8.2$ Hz, 2H), 6.75 (s, 1H), 6.61 (s, 1H), 4.14 (s, 2H), 2.21 (s, 3H); ^{13}C NMR (100 MHz, chloroform-*d*) $\delta = 146.0, 141.6, 137.8, 132.5, 129.4, 128.4, 119.8, 119.0, 110.6, 36.3, 15.8$; HRMS (ESI-TOF) m/z : $[M+H]^+$ calcd for $C_{13}H_{12}NS$ 214.0685, found 214.0685.



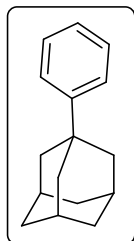
Ethyl 3-(naphthalen-1-ylmethyl)-5-nitrobenzoate (10). Compound **10** was obtained after column purification using hexanes/ethyl acetate (9:1) as mobile phase as a colorless oil in 89% yield (59 mg, 0.178 mmol) from 1-(fluoromethyl)naphthalene (32 mg, 0.2 mmol) and (3-(ethoxycarbonyl)-5-nitrophenyl)boronic acid (96 mg, 0.4 mmol) by following the general procedure described above (Method A). $R_f = 0.3$ (hexanes/EtOAc, 4:1); $^1\text{H NMR}$ (400 MHz, chloroform-*d*) $\delta = 8.68$ (dd, $J = 1.7, 1.7$ Hz, 1H), 8.27 (dd, $J = 1.7, 1.7$ Hz, 1H), 8.19 (dd, $J = 1.7, 1.7$ Hz, 1H), 7.93 – 7.79 (m, 3H), 7.53 – 7.43 (m, 3H), 7.33 (dd, $J = 7.0, 1.4$ Hz, 1H), 4.58 (s, 2H), 4.41 (q, $J = 7.1$ Hz, 2H), 1.40 (t, $J = 7.1$ Hz, 3H); $^{13}\text{C NMR}$ (100 MHz, chloroform-*d*) $\delta = 164.7, 148.7, 143.6, 135.6, 134.4, 134.3, 132.4, 131.7, 129.2, 128.3, 127.9, 127.4, 126.6, 126.1, 125.8, 123.7, 122.6, 62.1, 38.8, 14.4$; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{20}\text{H}_{18}\text{NO}_4$ 336.1230, found 336.1225.



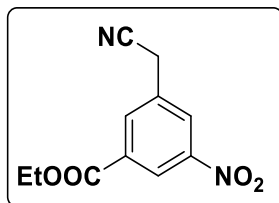
1-Fluoro-4-(4-nitrobenzyl)benzene (11). Compound **11** was obtained after column purification using hexanes/ethyl acetate (19:1) as mobile phase as a colorless solid in 97% yield (45 mg, 0.194 mmol) from 1-(fluoromethyl)-4-nitrobenzene (31 mg, 0.2 mmol) and (4-fluorophenyl)boronic acid (56 mg, 0.4 mmol) by following the general procedure described above (Method A). $R_f = 0.7$ (hexanes/EtOAc, 4:1); $^1\text{H NMR}$ (400 MHz, chloroform-*d*) $\delta = 8.15$ (d, $J = 8.7$ Hz, 2H), 7.32 (d, $J = 8.7$ Hz, 2H), 7.17 – 7.10 (m, 2H), 7.06 – 6.95 (m, 2H), 4.05 (s, 2H); $^{13}\text{C NMR}$ (100 MHz, chloroform-*d*) $\delta = 161.9$ (d, $J_{\text{C-F}} = 245.4$ Hz), 148.7 (d, $J_{\text{C-F}} = 1.0$ Hz), 146.7, 135.0 (d, $J_{\text{C-F}} = 3.3$ Hz), 130.5 (d, $J_{\text{C-F}} = 8.0$ Hz), 129.7, 123.9, 115.8 (d, $J_{\text{C-F}} = 21.4$ Hz), 41.0; $^{19}\text{F NMR}$ (376 MHz, chloroform-*d*) $\delta = -116.1$ (m, 1F); HRMS (ESI-TOF) m/z : $[\text{M}-\text{H}]^-$ calcd for $\text{C}_{13}\text{H}_9\text{FNO}_2$ 230.0623, found 230.0625.



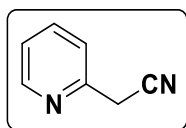
3-Benzyl-*N,N*-dimethylaniline (12). Compound **12** was obtained after column purification using hexanes/ethyl acetate (9:1) as mobile phase as a colorless oil in 89% yield (38 mg, 0.178 mmol) from 3-(fluoromethyl)-*N,N*-dimethylaniline (31 mg, 0.2 mmol) and phenylboronic acid (49 mg, 0.4 mmol) by following the general procedure described above (Method A). $R_f = 0.6$ (hexanes/EtOAc, 4:1); $^1\text{H NMR}$ (400 MHz, chloroform-*d*) $\delta = 7.25 - 7.19$ (m, 3H), 7.18 - 7.11 (m, 3H), 7.01 (dd, $J = 7.8, 1.2$ Hz, 1H), 6.55 - 6.45 (d, $J = 8.1$ Hz, 2H), 4.07 (s, 2H), 2.86 (s, 6H); $^{13}\text{C NMR}$ (100 MHz, chloroform-*d*) $\delta = 149.9, 142.4, 141.0, 133.5, 129.3, 129.2, 128.3, 125.9, 114.8, 111.8, 40.7, 40.2$. The spectroscopic data of 3-benzyl-*N,N*-dimethylaniline (**12**) are in accordance with the literature.¹



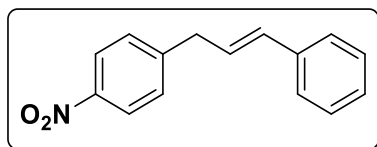
1-Phenyladamantane (13). Compound **13** was obtained after preparative TLC purification using hexanes as mobile phase as a colorless solid in 41% yield (17 mg, 0.082 mmol) from 1-fluoroadamantane (31 mg, 0.2 mmol) and phenylboronic acid (49 mg, 0.4 mmol) by following the general procedure described above (Method A). $R_f = 0.9$ (hexanes); $^1\text{H NMR}$ (400 MHz, chloroform-*d*) $\delta = 7.41 - 7.37$ (m, 2H), 7.36 - 7.29 (m, 2H), 7.19 (dd, $J = 7.6, 7.6$ Hz, 1H), 2.11 (bs, 3H), 1.93 (bs, 6H), 1.85 - 1.71 (m, 6H); $^{13}\text{C NMR}$ (100 MHz, chloroform-*d*) $\delta = 151.5, 128.2, 125.6, 125.0, 43.3, 37.0, 36.3, 29.1$. The spectroscopic data of 1-phenyladamantane (**13**) are in accordance with the literature.²



Ethyl 3-(cyanomethyl)-5-nitrobenzoate (14). Compound **14** was obtained after column purification using hexanes/ethyl acetate (3:1) as mobile phase as a colorless solid in 91% yield (42 mg, 0.182 mmol) from 2-fluoroacetonitrile (24 mg, 0.4 mmol) and (3-(ethoxycarbonyl)-5-nitrophenyl)boronic acid (48 mg, 0.2 mmol) by following the general procedure described above (Method B). $R_f = 0.3$ (hexanes/EtOAc, 1:1); $^1\text{H NMR}$ (400 MHz, chloroform-*d*) $\delta = 8.81$ (dd, $J = 1.7, 1.7$ Hz, 1H), 8.41 (dd, $J = 1.7, 1.7$ Hz, 1H), 8.35 (dd, $J = 1.7, 1.7$ Hz, 1H), 4.45 (q, $J = 7.1$ Hz, 2H), 3.95 (s, 2H), 1.43 (t, $J = 7.1$ Hz, 3H); $^{13}\text{C NMR}$ (100 MHz, chloroform-*d*) $\delta = 163.9, 148.9, 134.7, 133.5, 132.7, 126.8, 124.5, 116.2, 62.5, 23.6, 14.4$; HRMS (ESI-TOF) m/z : $[\text{M-H}]^-$ calcd for $\text{C}_{11}\text{H}_9\text{N}_2\text{O}_4$ 233.0568, found 233.0566.

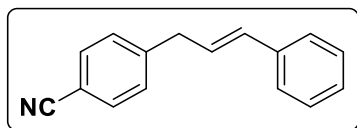


2-(Pyridin-2-yl)acetonitrile (15). Compound **15** was obtained after column purification using hexanes/ethyl acetate (4:1) as mobile phase as a colorless oil in 83% yield (20 mg, 0.166 mmol) from 2-fluoroacetonitrile (24 mg, 0.4 mmol) and pyridin-2-ylboronic acid (25 mg, 0.2 mmol) by following the general procedure described above (Method B). $R_f = 0.4$ (hexanes/EtOAc, 1:1); $^1\text{H NMR}$ (400 MHz, chloroform-*d*) $\delta = 8.59$ (d, $J = 7.7$ Hz, 1H), 7.74 (dd, $J = 7.8, 7.7$ Hz, 1H), 7.43 (d, $J = 7.8$ Hz, 1H), 7.27 (m, 1H), 3.95 (s, 2H); $^{13}\text{C NMR}$ (100 MHz, chloroform-*d*) $\delta = 150.6, 150.0, 137.5, 123.1, 122.3, 117.1, 26.7$; HRMS (ESI-TOF) m/z : $[\text{M+H}]^+$ calcd for $\text{C}_7\text{H}_7\text{N}_2$ 119.0604, found 119.0601.

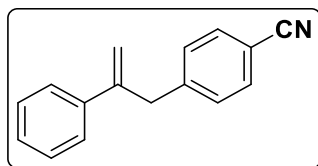


(E)-3-(4-Nitrophenyl)-1-phenylpropene (16). Compound **16** was obtained after column purification using hexanes/ethyl acetate (19:1) as mobile phase as a colorless solid in 81% yield (39 mg, 0.162 mmol) from 1-(fluoromethyl)-4-nitrobenzene (31 mg, 0.2 mmol) and (*E*)-phenylethenylboronic acid (59 mg, 0.4 mmol) by following the general procedure described above (Method A). $R_f = 0.5$ (hexanes/EtOAc, 4:1); $^1\text{H NMR}$ (400 MHz, chloroform-*d*) $\delta = 8.17$ (d, $J = 8.7$ Hz, 2H), 7.42 – 7.34 (m, 4H), 7.31 (dd, $J = 8.0, 7.3$ Hz, 2H), 7.24 (d, $J = 7.3$ Hz, 1H), 6.49 (d, $J = 15.9$ Hz, 1H), 6.31 (dt, $J = 15.9, 6.8$ Hz, 1H), 3.65 (d, $J = 6.8$ Hz, 2H); $^{13}\text{C NMR}$ (100 MHz,

chloroform-*d*) δ = 148.1, 146.8, 137.0, 132.7, 129.6, 128.8, 127.7, 127.1, 126.3, 123.9, 39.2; HRMS (ESI-TOF) m/z : $[M+H]^+$ calcd for C₁₅H₁₄NO₂ 240.1019, found 240.1021.



(*E*)-3-(4-Cyanophenyl)-1-phenylpropene (17). Compound **17** was obtained after column purification using hexanes/ethyl acetate (19:1) as mobile phase as a colorless solid in 89% yield (39 mg, 0.178 mmol) from 4-(fluoromethyl)benzonitrile (27 mg, 0.2 mmol) and (*E*)-phenylethenylboronic acid (59 mg, 0.4 mmol) by following the general procedure described above (Method A). R_f = 0.4 (hexanes/EtOAc, 4:1); ¹H NMR (400 MHz, chloroform-*d*) δ = 7.57 (d, J = 7.8 Hz, 2H), 7.40 – 7.24 (m, 6H), 7.22 (m, 1H), 6.45 (d, J = 15.7 Hz, 1H), 6.26 (dt, J = 15.7, 6.9 Hz, 1H), 3.57 (d, J = 6.9 Hz, 2H); ¹³C NMR (100 MHz, chloroform-*d*) δ = 145.9, 137.0, 132.6, 132.4, 129.6, 128.7, 127.7, 127.3, 126.3, 119.1, 110.3, 39.4; HRMS (ESI-TOF) m/z : $[M+H]^+$ calcd for C₁₆H₁₄N 220.1121, found 220.1123.

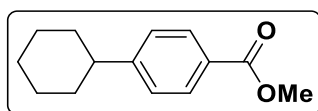


4-(2-Phenylallyl)benzonitrile (18). Compound **18** was obtained after column purification using hexanes/ethyl acetate (19:1) as mobile phase as a colorless oil in 88% yield (38 mg, 0.176 mmol) from 4-(fluoromethyl)benzonitrile (27 mg, 0.2 mmol) and (1-phenylvinyl)boronic acid (59 mg, 0.4 mmol) by following the general procedure described above (Method A). R_f = 0.4 (hexanes/EtOAc, 8:2); ¹H NMR (400 MHz, chloroform-*d*) δ = 7.56 – 7.49 (m, 2H), 7.40 – 7.34 (m, 2H), 7.34 – 7.20 (m, 5H), 5.53 (d, J = 1.2 Hz, 1H), 5.05 (d, J = 1.2 Hz, 1H), 3.88 (s, 2H); ¹³C NMR (100 MHz, chloroform-*d*) δ = 145.8, 145.4, 140.1, 132.3, 129.7, 128.5, 127.9, 126.2, 119.1, 115.6, 110.2, 41.9; HRMS (ESI-TOF) m/z : $[M+H]^+$ calcd for C₁₆H₁₄N 220.1126, found 220.1125.

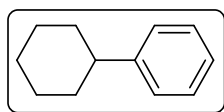
5. General procedure for the Ni-catalyzed Suzuki cross-coupling

Ni(cod)₂ (5 mol%), **L8** (6 mol%), alkyl fluoride (0.2 mmol, 1.0 eq), boronic acid (0.4 mmol, 2.0 eq), LiI (0.3 mmol, 1.5 eq) and Cs₂CO₃ (0.4 mmol, 2.0 eq) were added into 2 mL of dry toluene under nitrogen atmosphere. The reaction mixture was stirred at 25 °C for 3 hours and then heated to 100 °C in an oil bath and stirred at this temperature for 48 hours in the dark. The solvent was removed and the crude products were purified by silica flash column chromatography as described below.

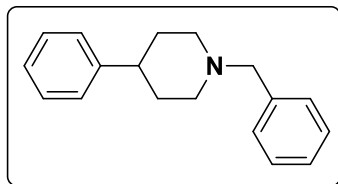
5.1. Product purification and characterization



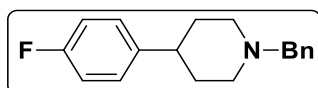
Methyl 4-cyclohexylbenzoate (21). Compound **21** was obtained after preparative TLC purification using hexanes/ethyl acetate (49:1) as mobile phase as a colorless oil in 71% yield (31 mg, 0.142 mmol) from fluorocyclohexane (17 mg, 0.2 mmol) and (4-(methoxycarbonyl)phenyl)boronic acid (72 mg, 0.4 mmol) by following the general procedure described above. $R_f = 0.7$ (hexanes/EtOAc, 19:1); ¹H NMR (400 MHz, chloroform-*d*) $\delta = 7.98 - 7.92$ (m, 2H), 7.30 – 7.23 (m, 2H), 3.89 (s, 3H), 2.56 (m, 1H), 1.93 – 1.80 (m, 4H), 1.76 (m, 1H), 1.52 – 1.33 (m, 4H), 1.27 (m, 1H); ¹³C NMR (100 MHz, chloroform-*d*) $\delta = 167.3, 153.6, 129.8, 127.9, 127.0, 52.1, 44.9, 34.3, 26.9, 26.2$; HRMS (ESI-TOF) m/z : [M+H]⁺ calcd for C₁₄H₁₉O₂ 219.1380, found 219.1379.



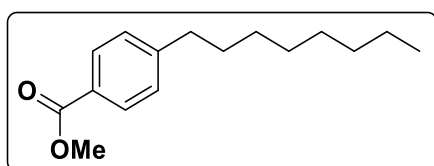
Cyclohexylbenzene (22). Compound **22** was obtained after preparative TLC purification using hexanes as mobile phase as a colorless oil in 63% yield (20 mg, 0.126 mmol) from fluorocyclohexane (17 mg, 0.2 mmol) and phenylboronic acid (49 mg, 0.4 mmol) by following the general procedure described above. $R_f = 0.8$ (hexanes); ¹H NMR (400 MHz, chloroform-*d*) $\delta = 7.31 - 7.25$ (m, 2H), 7.23 – 7.13 (m, 3H), 2.49 (m, 1H), 1.92 – 1.82 (m, 4H), 1.74 (m, 1H), 1.51 – 1.34 (m, 4H), 1.25 (m, 1H); ¹³C NMR (100 MHz, chloroform-*d*) $\delta = 148.2, 128.4, 127.0, 125.9, 44.8, 34.6, 27.1, 26.3$. The spectroscopic data of cyclohexylbenzene (**22**) are in accordance with the literature.³



1-Benzyl-4-phenylpiperidine (23). Compound **23** was obtained after column purification using hexanes/ethyl acetate (9:1) as mobile phase as a colorless solid in 69% yield (35 mg, 0.138 mmol) from 1-benzyl-4-fluoropiperidine (38 mg, 0.2 mmol) and phenylboronic acid (49 mg, 0.4 mmol) by following the general procedure described above. $R_f = 0.3$ (hexanes/EtOAc, 4:1); $^1\text{H NMR}$ (400 MHz, chloroform-*d*) $\delta = 7.38 - 7.26$ (m, 6H), $7.26 - 7.11$ (m, 4H), 3.55 (s, 2H), 3.03 – 2.98 (m, 2H), 2.49 (m, 1H), 2.15 – 2.01 (m, 2H), 1.86 – 1.74 (m, 4H); $^{13}\text{C NMR}$ (100 MHz, chloroform-*d*) $\delta = 146.7, 138.6, 129.4, 128.5, 128.3, 127.1, 127.0, 126.2, 63.7, 54.4, 42.9, 33.7$; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{18}\text{H}_{22}\text{N}$ 252.1747, found 252.1751.

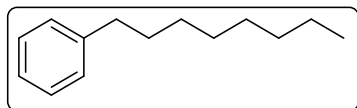


1-Benzyl-4-(4-fluorophenyl)piperidine (24). Compound **24** was obtained after column purification using hexanes/ethyl acetate (9:1) as mobile phase as a colorless solid in 65% yield (35 mg, 0.13 mmol) from 1-benzyl-4-fluoropiperidine (38 mg, 0.2 mmol) and (4-fluorophenyl)boronic acid (56 mg, 0.4 mmol) by following the general procedure described above. $R_f = 0.3$ (hexanes/EtOAc, 4:1); $^1\text{H NMR}$ (400 MHz, chloroform-*d*) $\delta = 7.37 - 7.29$ (m, 4H), 7.25 (m, 1H), 7.20 – 7.13 (m, 2H), 6.99 – 6.93 (m, 2H), 3.54 (s, 2H), 3.05 – 2.94 (m, 2H), 2.47 (tt, $J = 10.5, 5.5$ Hz, 1H), 2.07 (td, $J = 11.1, 3.9$ Hz, 2H), 1.83 – 1.67 (m, 4H); $^{13}\text{C NMR}$ (100 MHz, chloroform-*d*) $\delta = 161.4$ (d, $J_{\text{C-F}} = 243.5$ Hz), 142.3 (d, $J_{\text{C-F}} = 3.1$ Hz), 138.5, 129.4, 128.3, 128.2, 127.1, 115.2 (d, $J_{\text{C-F}} = 21.0$ Hz), 63.6, 54.3, 42.1, 33.8; $^{19}\text{F NMR}$ (376 MHz, chloroform-*d*) $\delta = -117.5$ (m, 1F); HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{18}\text{H}_{21}\text{FN}$ 270.1653, found 270.1653.

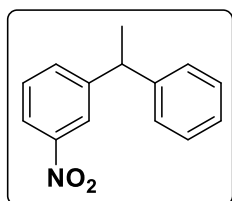


Methyl 4-octylbenzoate (25). Compound **25** was obtained after column purification using hexanes/ethyl acetate (49:1) as mobile phase as a colorless oil in 72% yield (36 mg, 0.144 mmol) from 1-fluorooctane (26 mg, 0.2 mmol) and (4-(methoxycarbonyl)phenyl)boronic acid (72 mg, 0.4

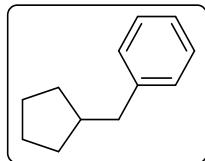
mmol) by following the general procedure described above. $R_f = 0.8$ (hexanes/EtOAc, 19:1); ^1H NMR (400 MHz, chloroform-*d*) $\delta = 7.94$ (d, $J = 8.4$ Hz, 2H), 7.23 (d, $J = 8.4$ Hz, 2H), 3.90 (s, 3H), 2.65 (t, $J = 7.2$ Hz, 2H), 1.69 – 1.56 (m, 2H), 1.38 – 1.20 (m, 10H), 0.88 (t, $J = 7.2$ Hz, 3H); ^{13}C NMR (100 MHz, chloroform-*d*) $\delta = 167.3, 148.6, 129.7, 128.5, 127.8, 52.0, 36.1, 32.0, 31.2, 29.5, 29.4, 29.3, 22.8, 14.2$; HRMS (ESI-TOF) m/z : $[\text{M}+\text{H}]^+$ calcd for $\text{C}_{16}\text{H}_{25}\text{O}_2$ 249.1849, found 249.1852.



Octylbenzene (26). Compound **26** was obtained after preparative TLC purification using hexanes as mobile phase as a colorless oil in 74% yield (28 mg, 0.148 mmol) from 1-fluorooctane (26 mg, 0.2 mmol) and phenylboronic acid (49 mg, 0.4 mmol) by following the general procedure described above. $R_f = 0.8$ (hexanes); ^1H NMR (400 MHz, chloroform-*d*) $\delta = 7.30 - 7.23$ (m, 2H), 7.18 – 7.14 (m, 3H), 2.59 (t, $J = 7.2$ Hz, 2H), 1.60 (p, $J = 7.2$ Hz, 2H), 1.39 – 1.18 (m, 10H), 0.87 (t, $J = 7.2$ Hz, 3H); ^{13}C NMR (100 MHz, chloroform-*d*) $\delta = 143.1, 128.5, 128.4, 125.7, 36.2, 32.0, 31.7, 29.6, 29.5, 29.4, 22.8, 14.3$. The spectroscopic data of octylbenzene (**26**) are in accordance with the literature.³



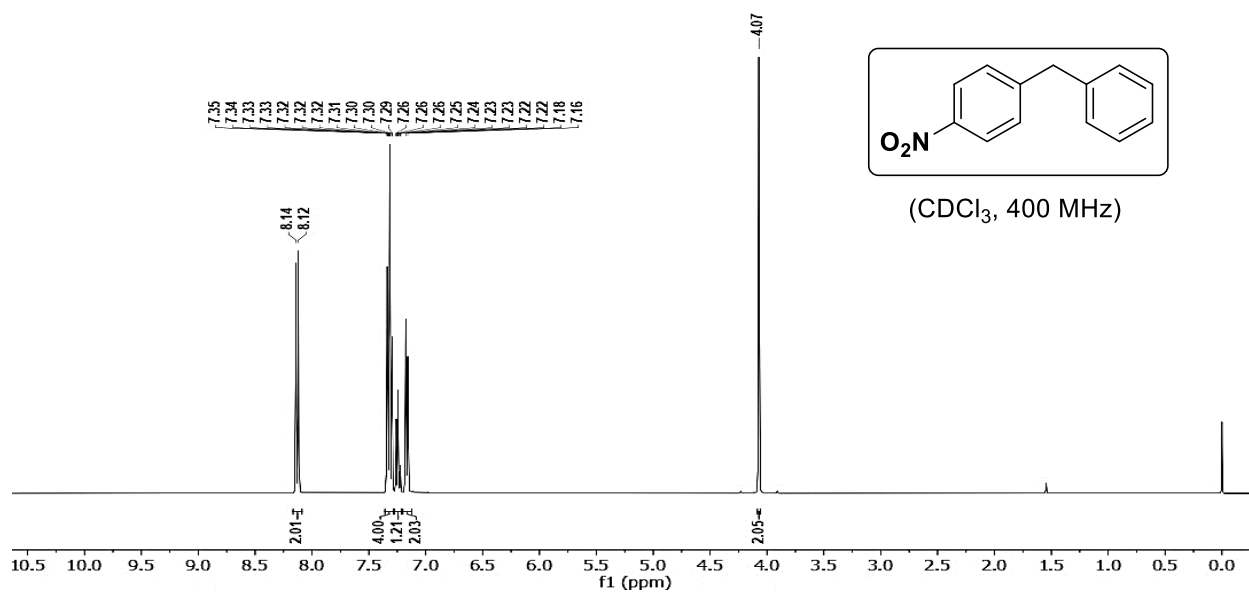
1-Nitro-3-(1-phenylethyl)benzene (27). Compound **27** was obtained after column purification using hexanes/ethyl acetate (9:1) as mobile phase as a colorless solid in 24% yield (11 mg, 0.048 mmol) from 1-(1-fluoroethyl)-3-nitrobenzene (34 mg, 0.2 mmol) and phenylboronic acid (49 mg, 0.4 mmol) by following the general procedure described above. $R_f = 0.4$ (hexanes/EtOAc, 9:1); ^1H NMR (400 MHz, chloroform-*d*) $\delta = 8.10$ (dd, $J = 1.8, 1.6$ Hz, 1H), 8.04 (dd, $J = 7.8, 1.6$ Hz, 1H), 7.53 (ddd, $J = 7.8, 1.6, 0.8$ Hz, 1H), 7.42 (dd, $J = 7.8, 7.8$ Hz, 1H), 7.35 – 7.27 (m, 2H), 7.25 – 7.16 (m, 3H), 4.25 (q, $J = 7.2$ Hz, 1H), 1.68 (d, $J = 7.2$ Hz, 3H); ^{13}C NMR (100 MHz, chloroform-*d*) $\delta = 148.6, 148.6, 144.9, 134.1, 129.4, 128.9, 127.7, 126.8, 122.5, 121.4, 44.7, 21.7$. The spectroscopic data of 1-nitro-3-(1-phenylethyl)benzene (**27**) are in accordance with the literature.⁴



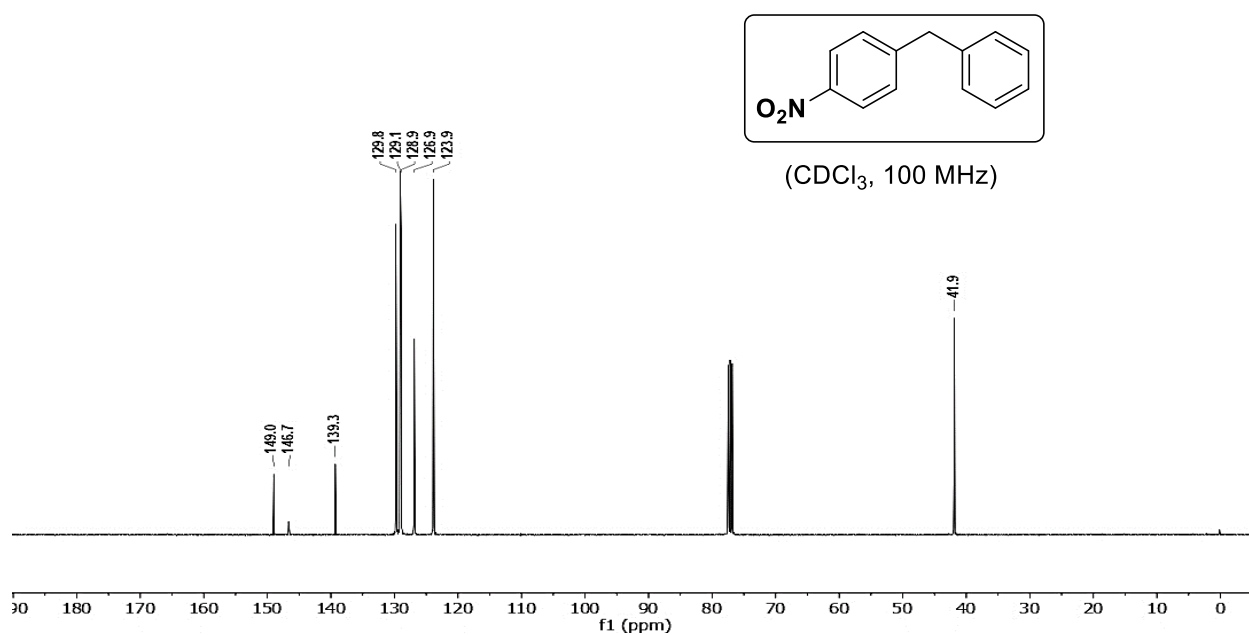
(Cyclopentylmethyl)benzene (29). Compound **29** was obtained after column purification using hexanes as mobile phase as a colorless oil in 68% yield (22 mg, 0.136 mmol) from 6-fluorohex-1-ene (20 mg, 0.2 mmol) and phenylboronic acid (49 mg, 0.4 mmol) by following the general procedure described above. $R_f = 0.9$ (hexanes); $^1\text{H NMR}$ (400 MHz, chloroform-*d*) $\delta = 7.32 - 7.21$ (m, 2H), 7.19 – 7.11 (m, 3H), 2.59 (d, $J = 7.4$ Hz, 2H), 2.07 (m, 1H), 1.75 – 1.46 (m, 6H), 1.19 (m, 1H), 0.87 (m, 1H); $^{13}\text{C NMR}$ (100 MHz, chloroform-*d*) $\delta = 142.5, 128.9, 128.3, 125.7, 42.3, 42.2, 32.6, 25.1$. The spectroscopic data of (cyclopentylmethyl)benzene (**29**) are in accordance with the literature.³

6. ^1H , ^{13}C and ^{19}F NMR spectra

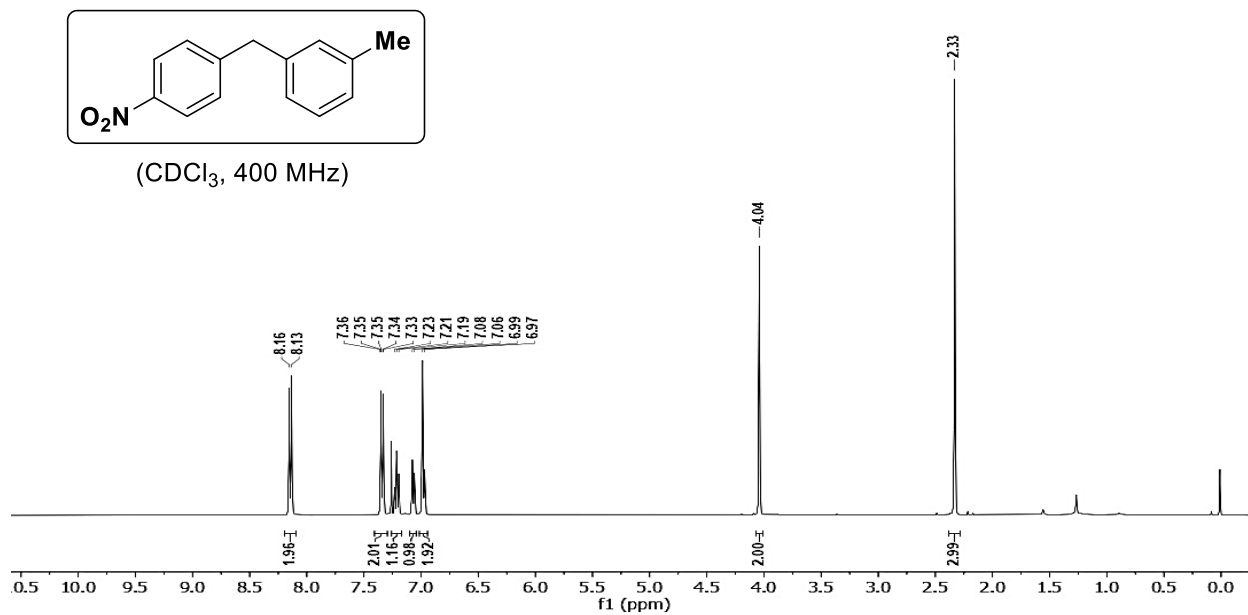
^1H NMR spectrum of 1-benzyl-4-nitrobenzene (3).



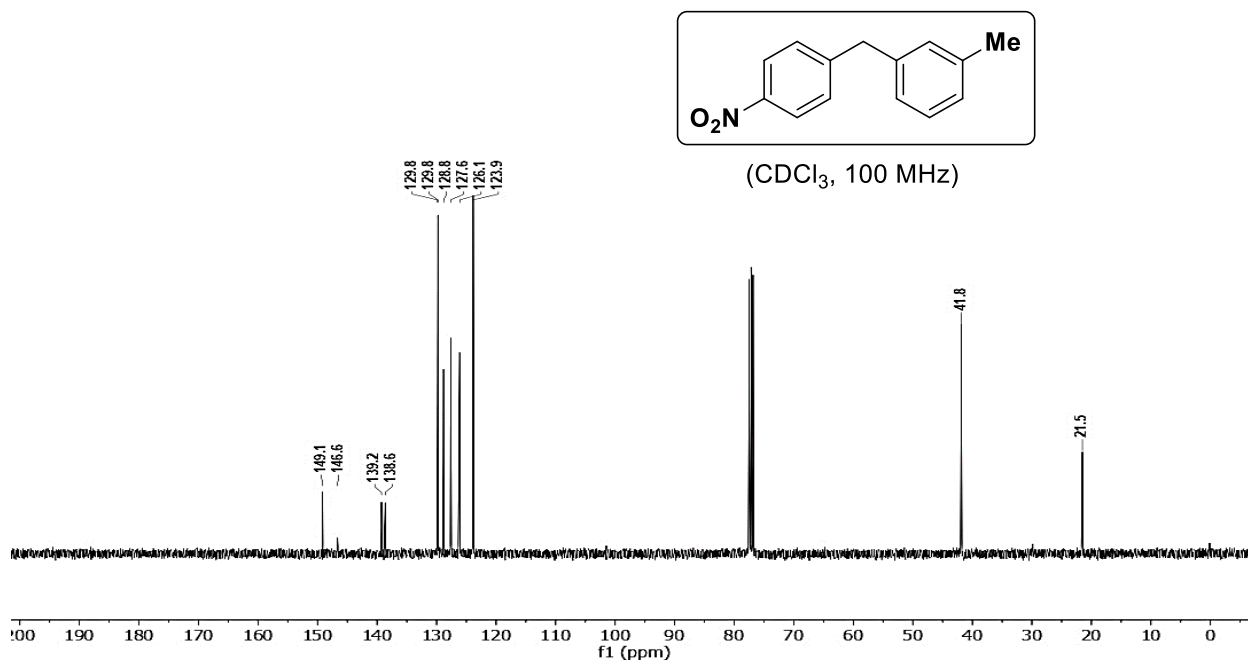
^{13}C NMR spectrum of 1-benzyl-4-nitrobenzene (3).



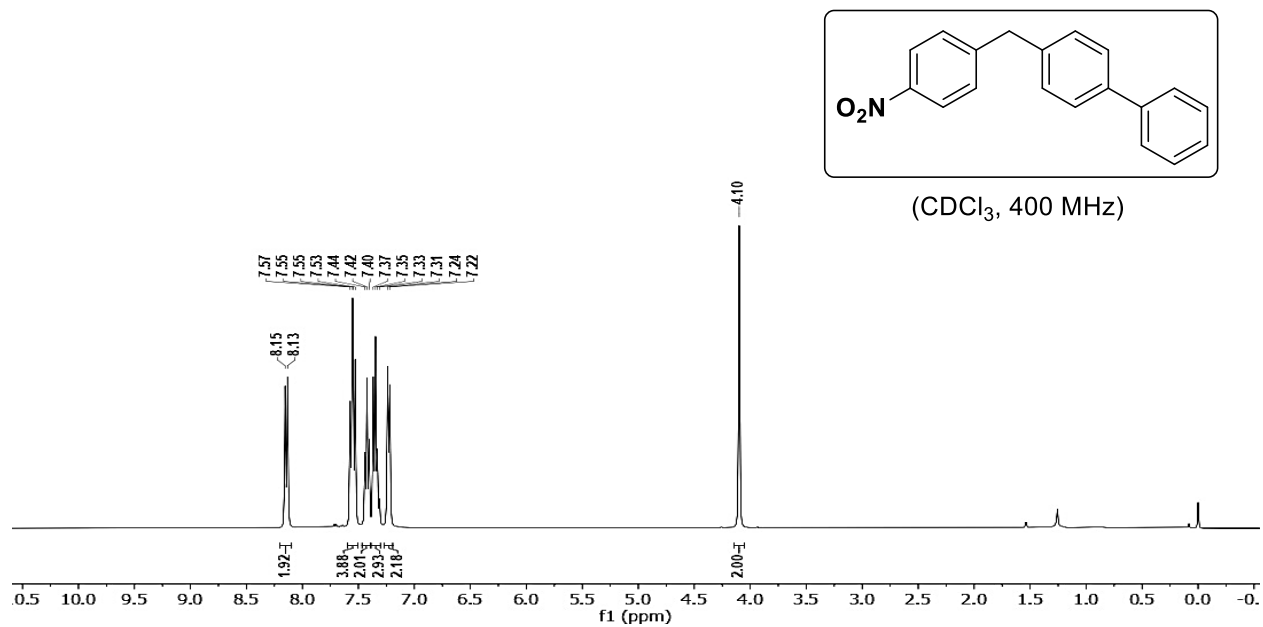
¹H NMR spectrum of 1-methyl-3-(4-nitrobenzyl)benzene (5).



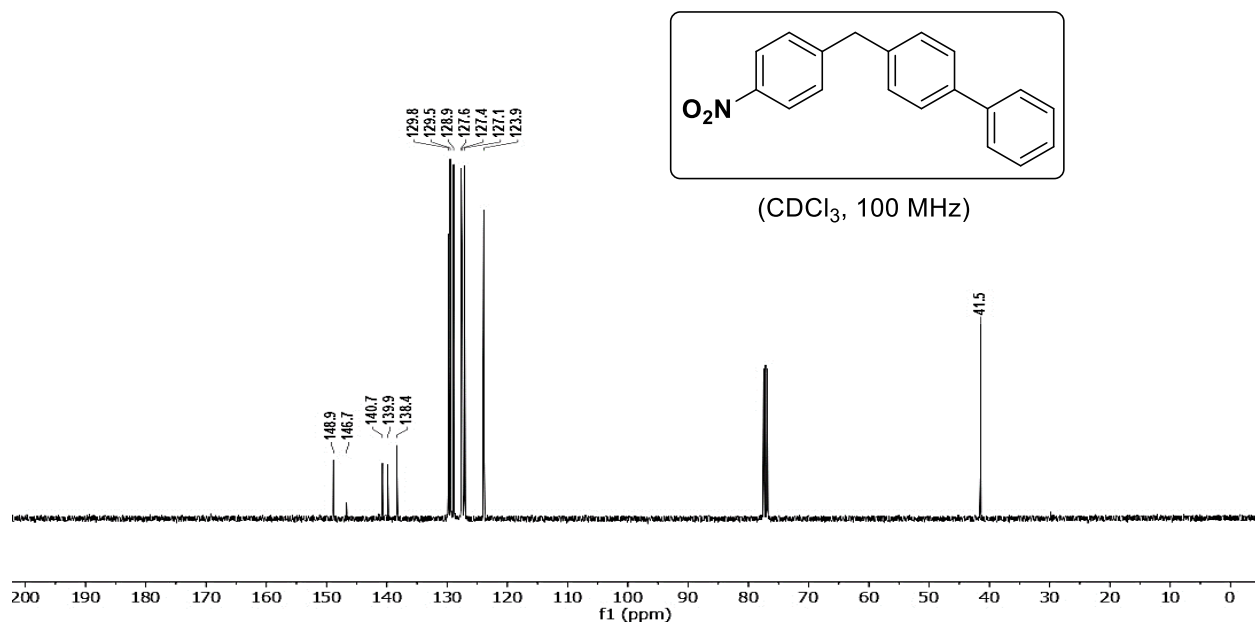
¹³C NMR spectrum of 1-methyl-3-(4-nitrobenzyl)benzene (5).



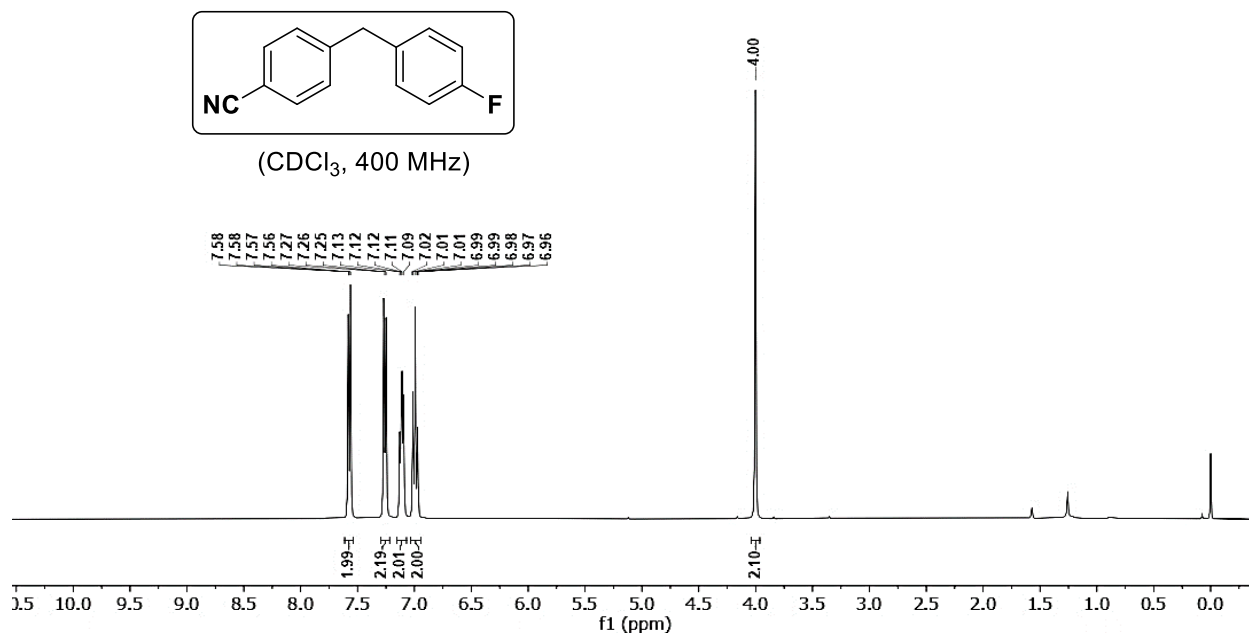
¹H NMR spectrum of 4-(4-nitrobenzyl)-1,1'-biphenyl (6).



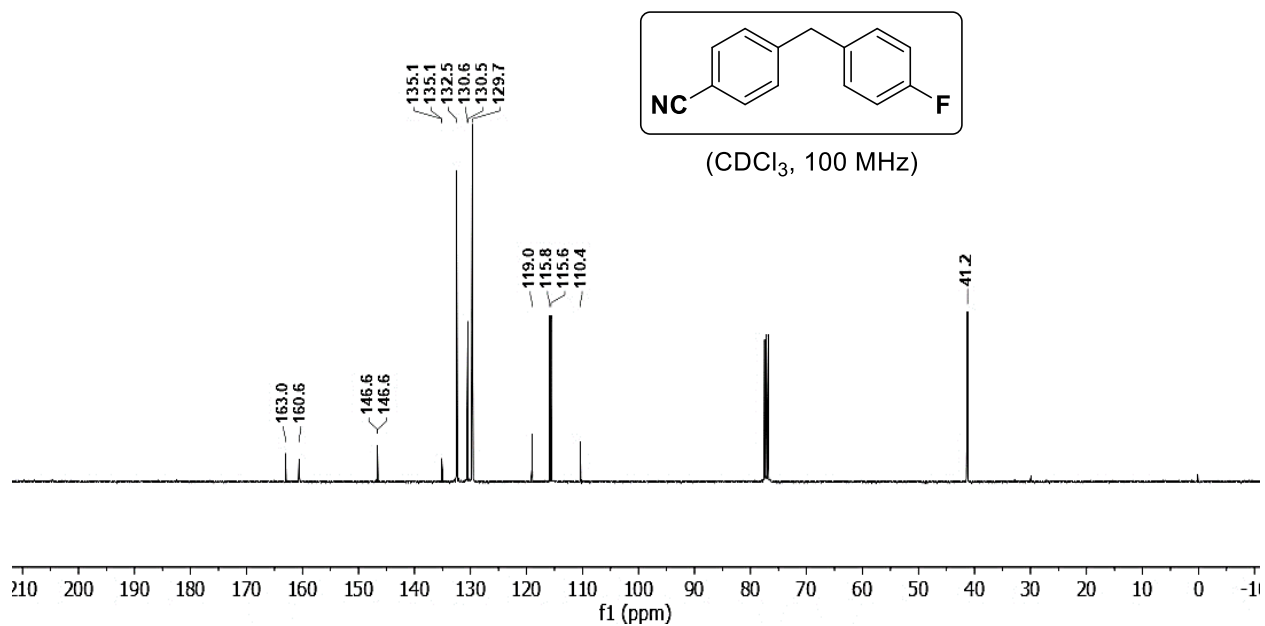
¹³C NMR spectrum of 4-(4-nitrobenzyl)-1,1'-biphenyl (6).



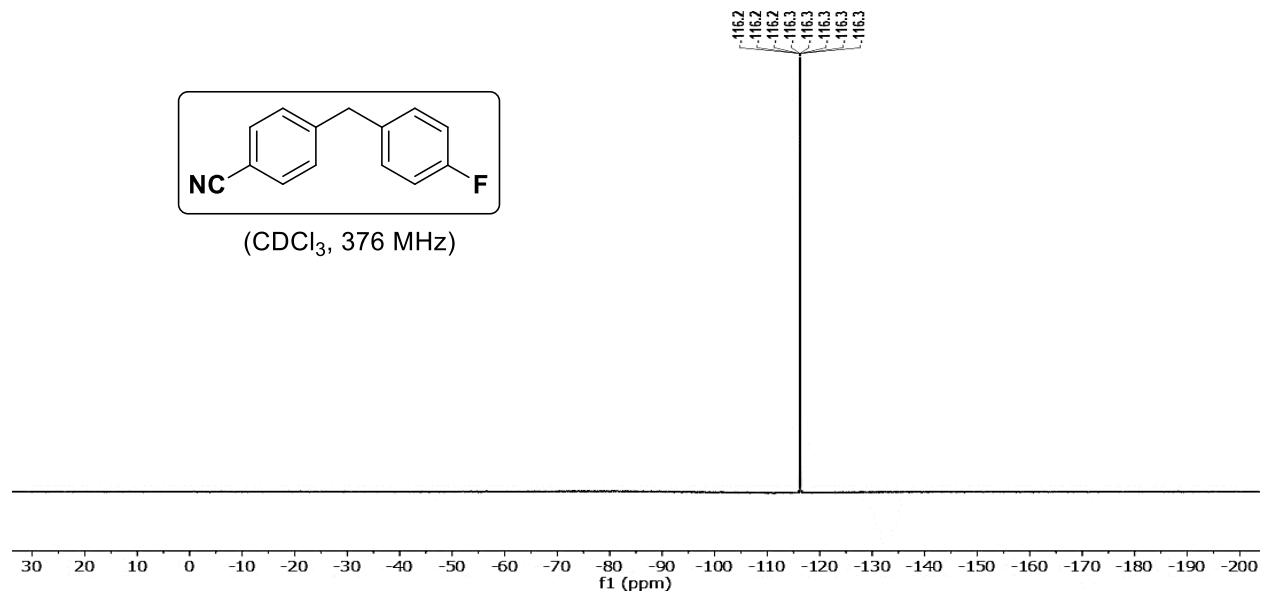
¹H NMR spectrum of 4-(4-fluorobenzyl)benzonitrile (7).



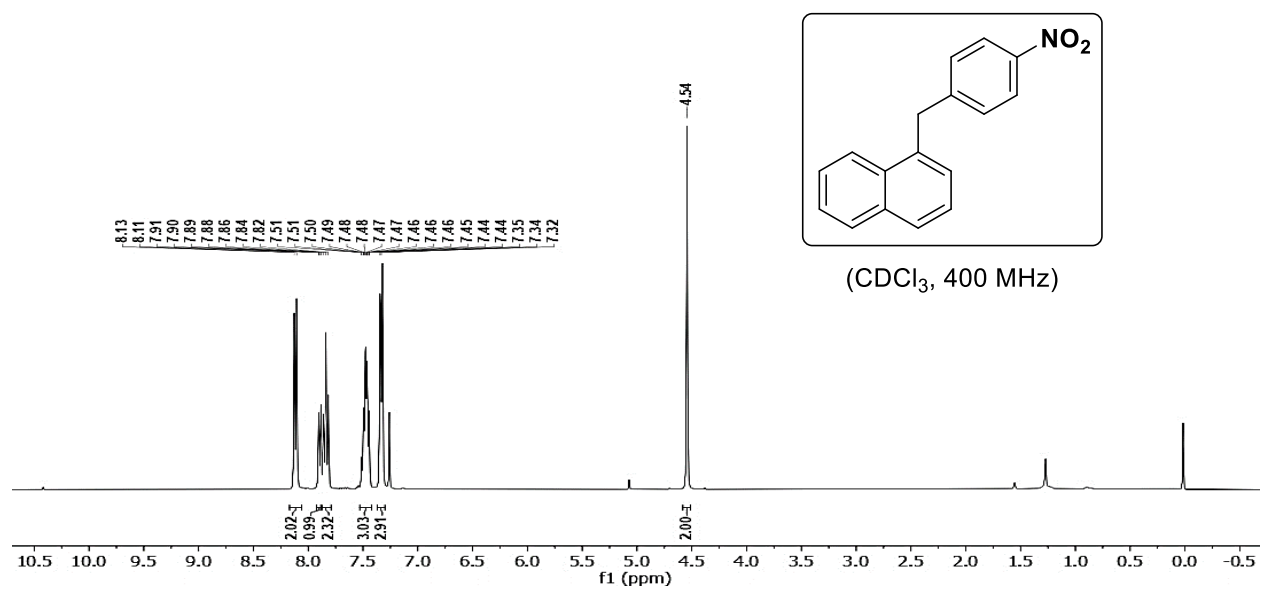
¹³C NMR spectrum of 4-(4-fluorobenzyl)benzonitrile (7).



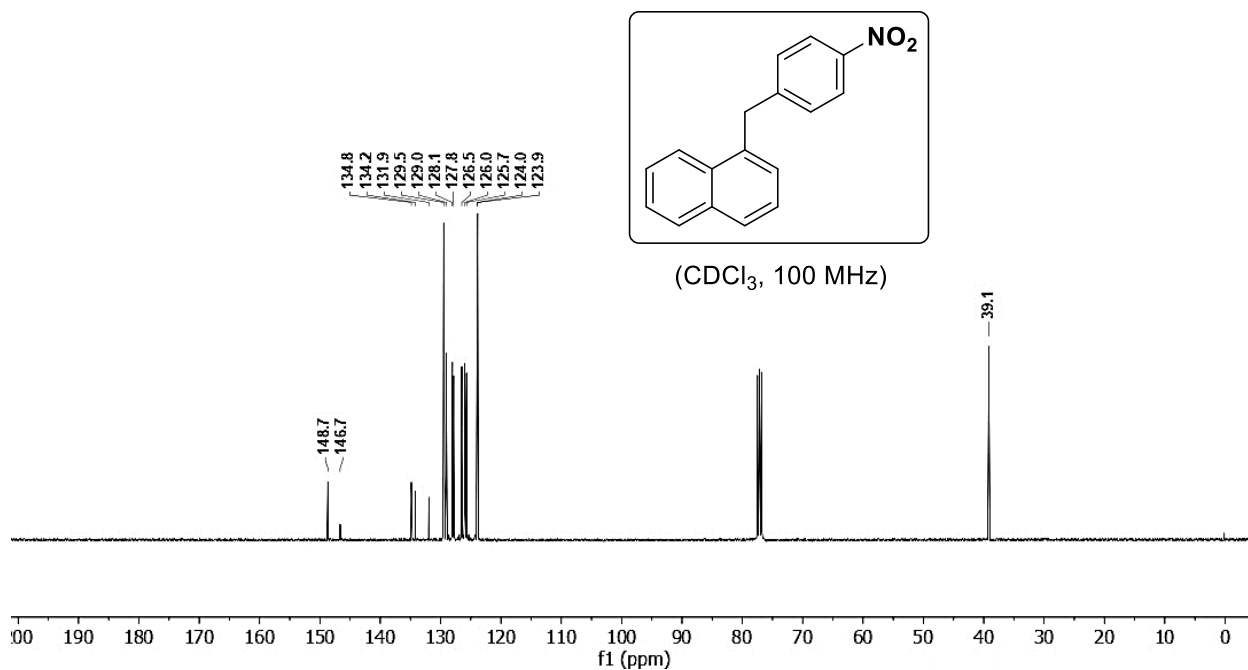
^{19}F NMR spectrum of 4-(4-fluorobenzyl)benzonitrile (7).



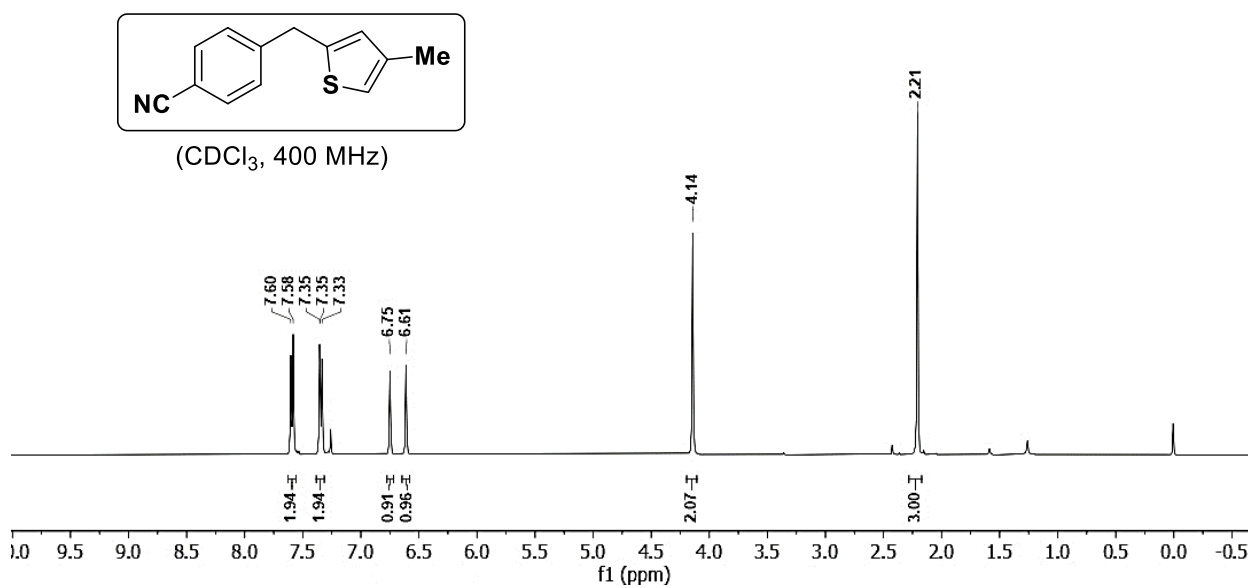
^1H NMR spectrum of 1-(4-nitrobenzyl)naphthalene (8).



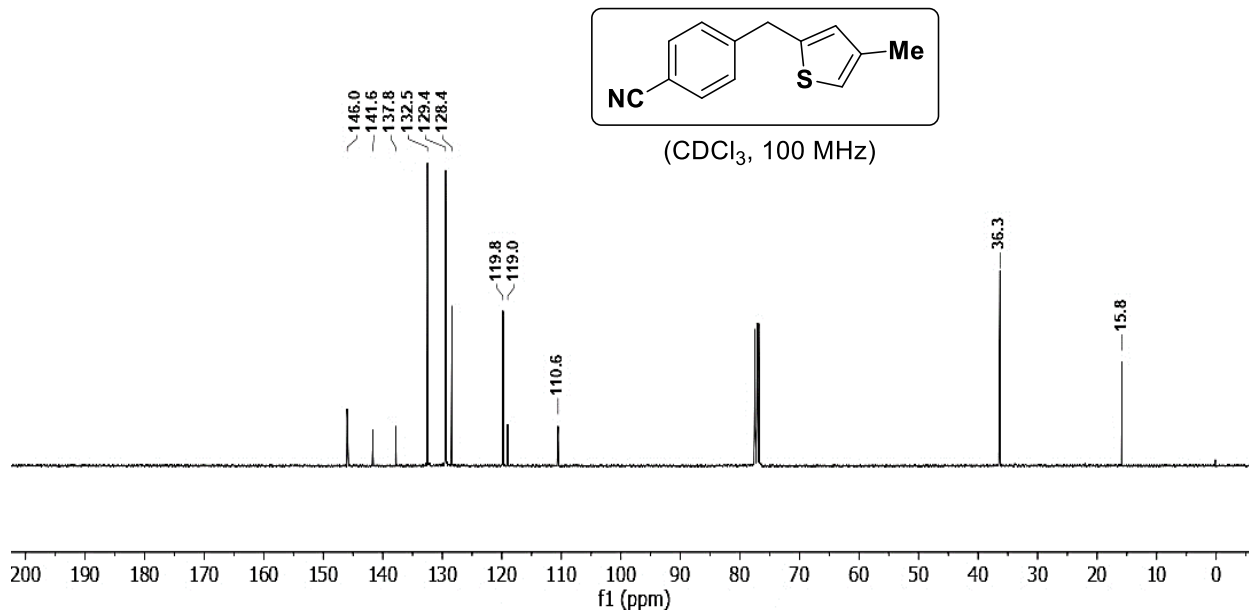
¹³C NMR spectrum of 1-(4-nitrobenzyl)naphthalene (8).



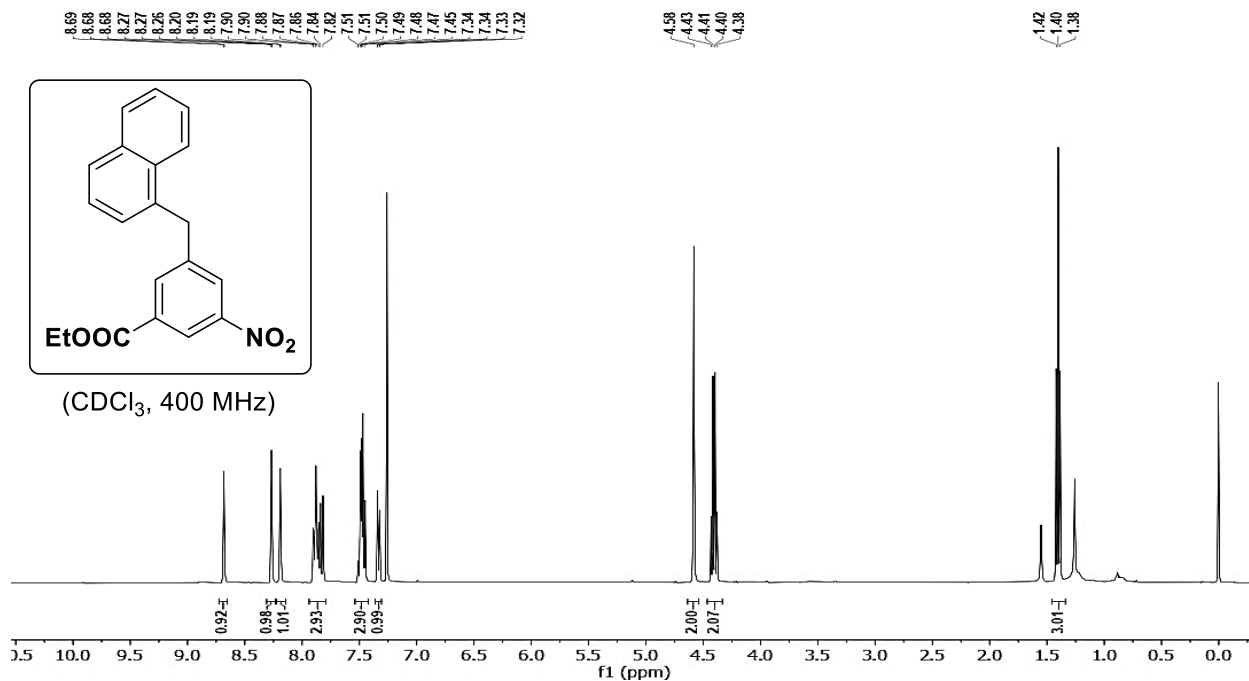
¹H NMR spectrum of 4-((4-methylthiophen-2-yl)methyl)benzonitrile (9).



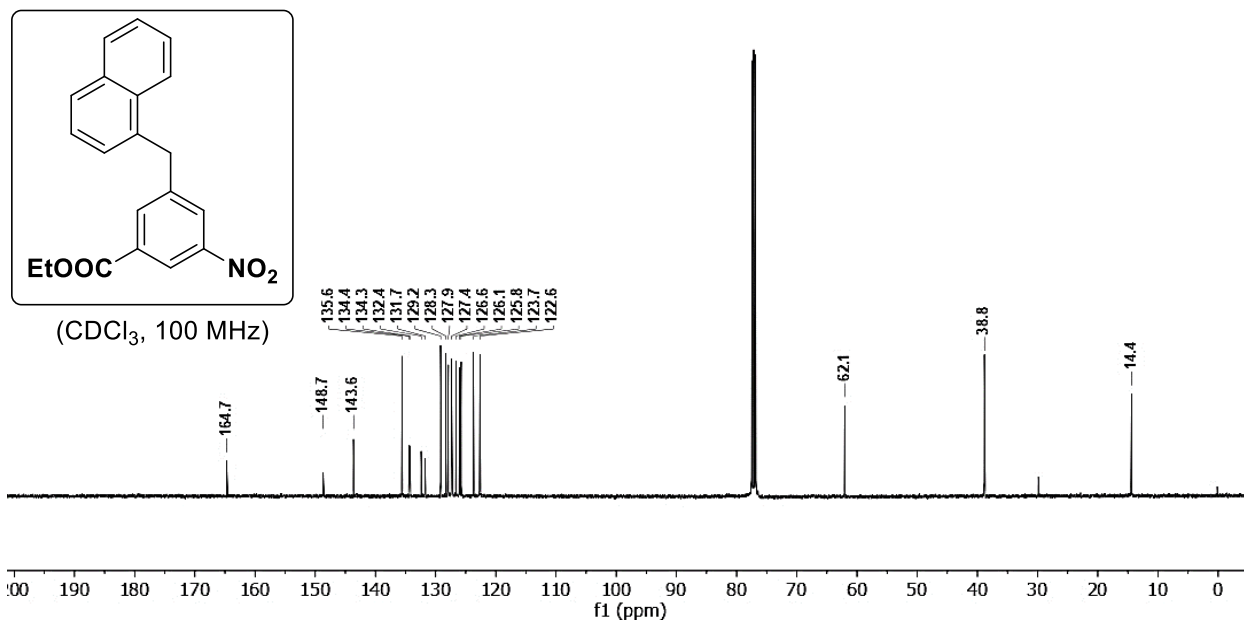
^{13}C NMR spectrum of 4-((4-methylthiophen-2-yl)methyl)benzonitrile (9).



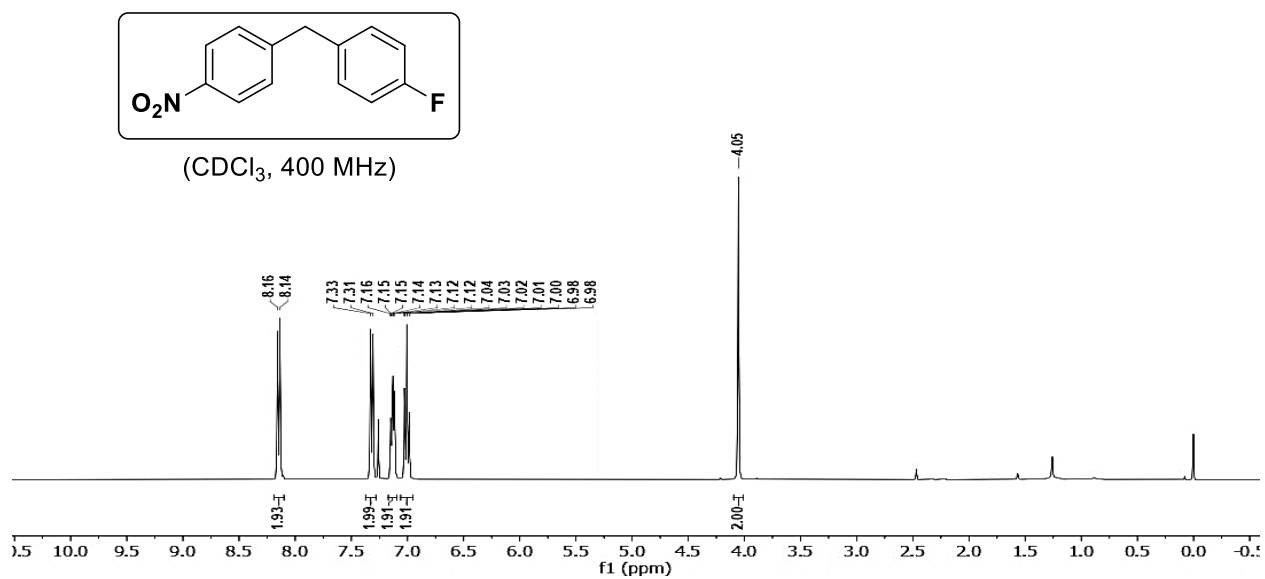
^1H NMR spectrum of ethyl 3-(naphthalen-1-ylmethyl)-5-nitrobenzoate (10).



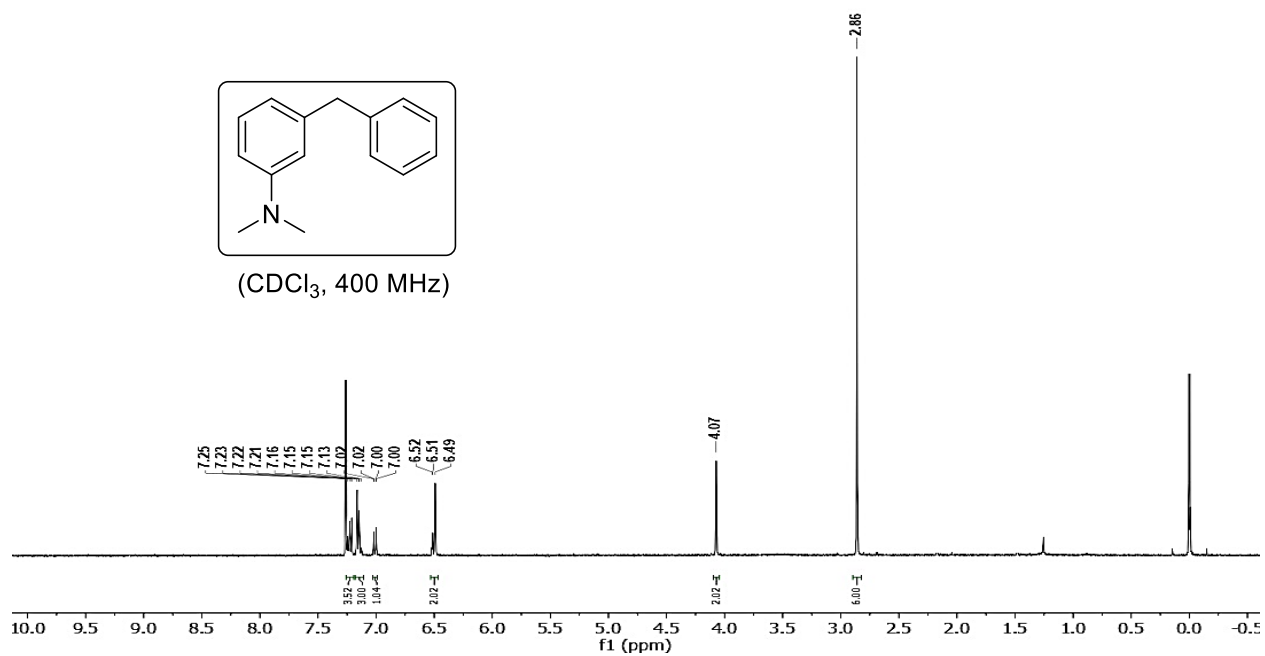
¹³C NMR spectrum of ethyl 3-(naphthalen-1-ylmethyl)-5-nitrobenzoate (10).



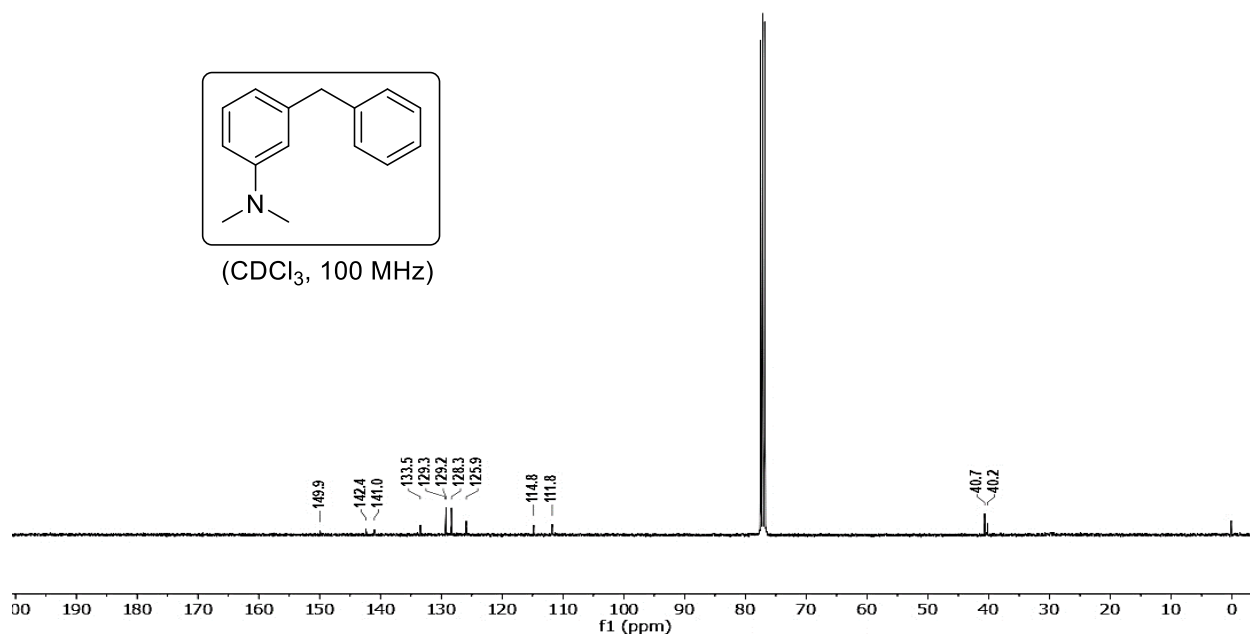
¹H NMR spectrum of 1-fluoro-4-(4-nitrobenzyl)benzene (11).



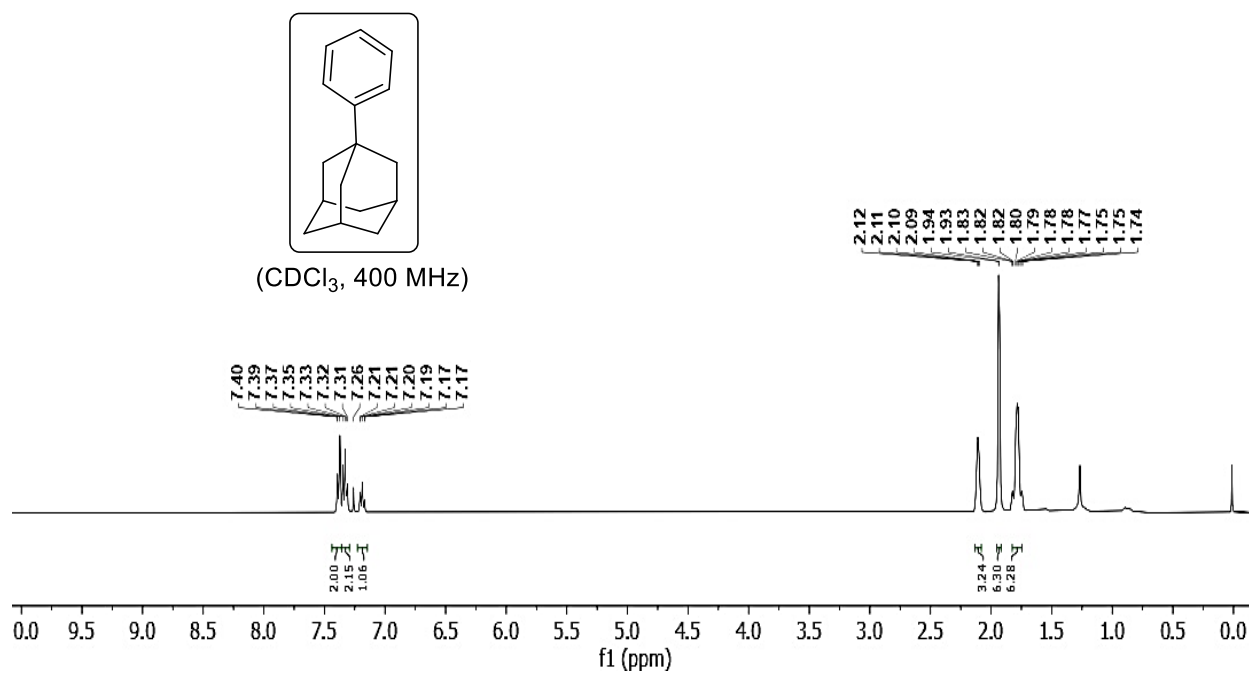
¹H NMR spectrum of 3-benzyl-*N,N*-dimethylaniline (12).



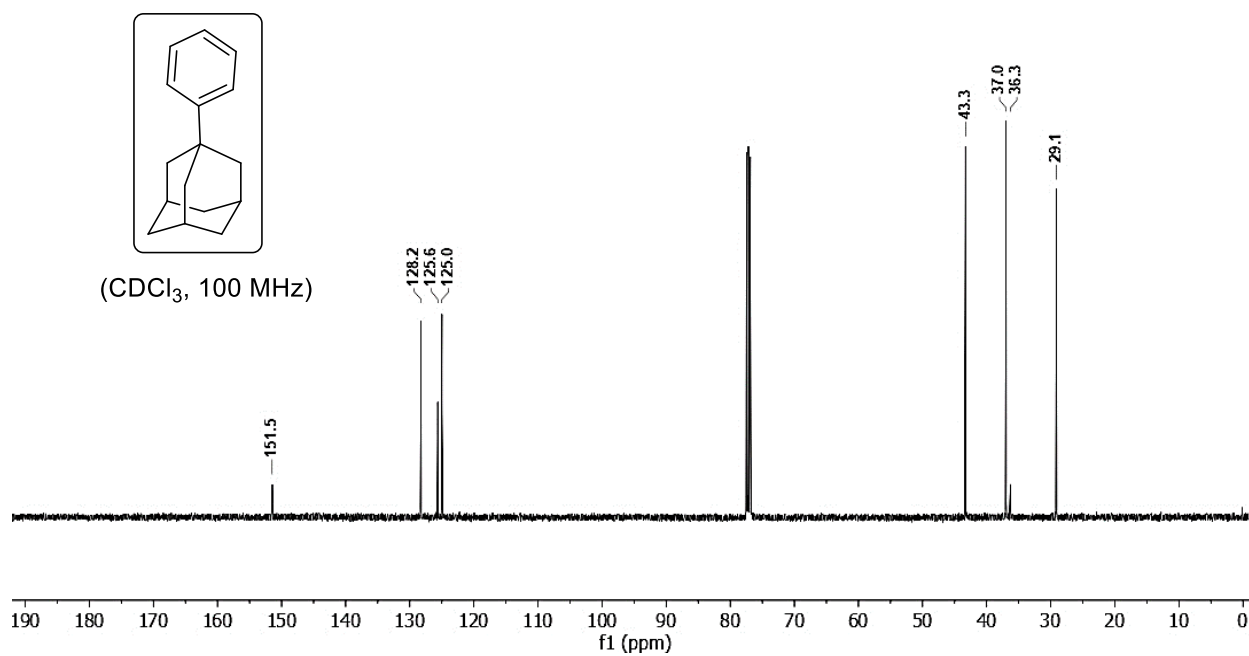
¹³C NMR spectrum of 3-benzyl-*N,N*-dimethylaniline (12).



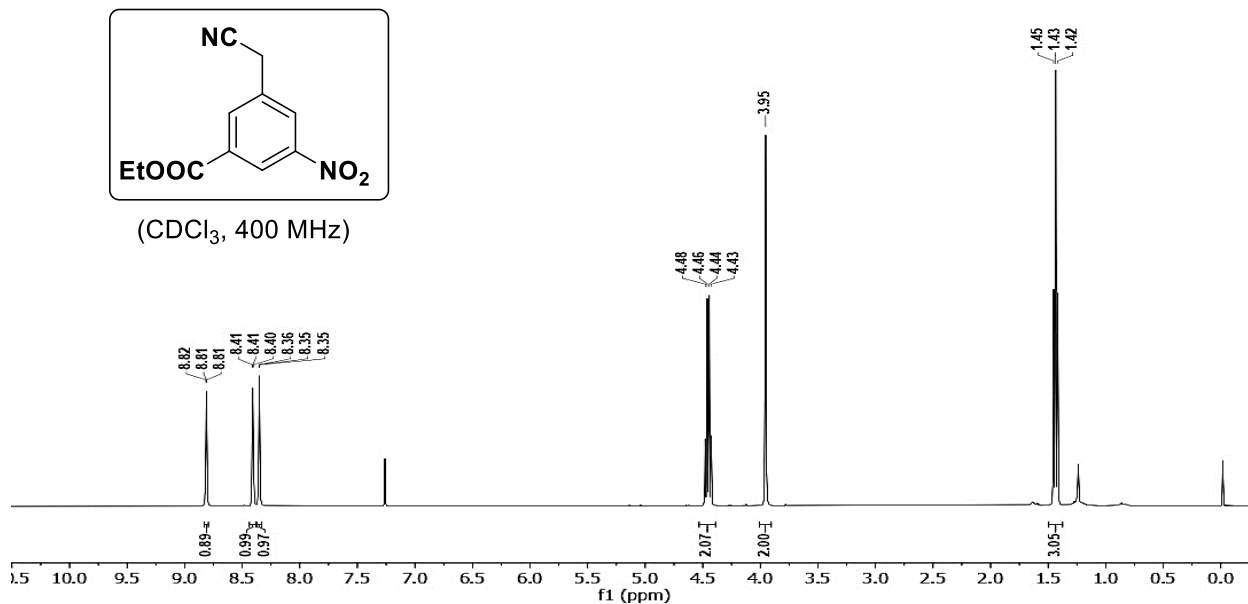
¹H NMR spectrum of 1-phenyladamantane (13).



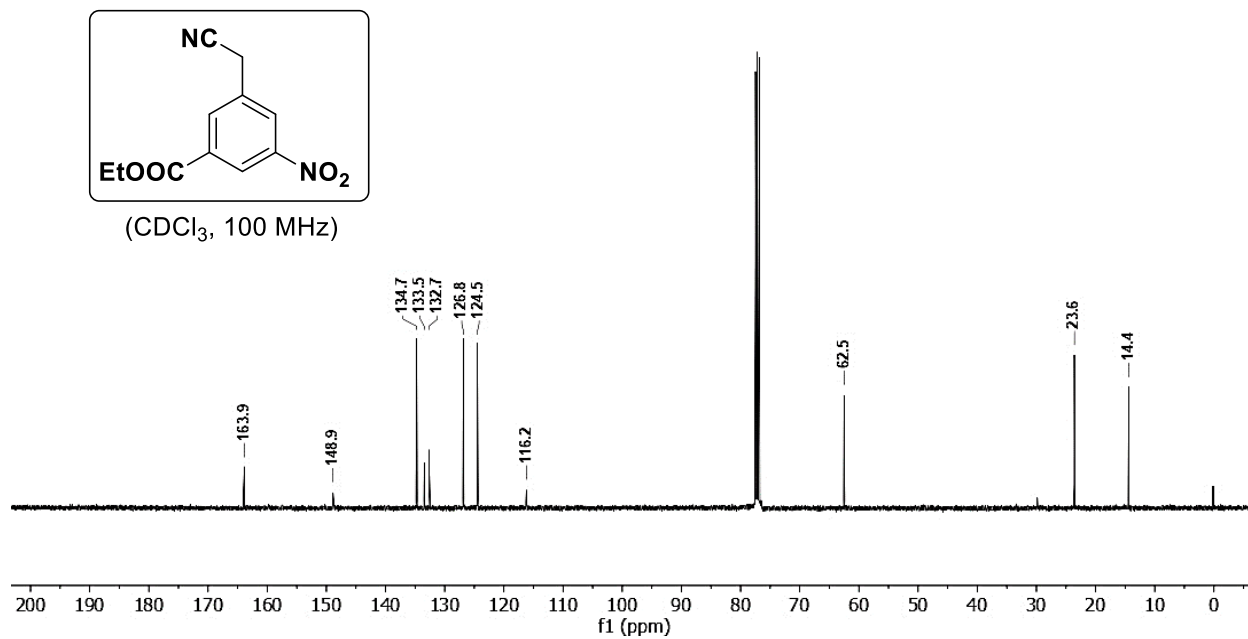
¹³C NMR spectrum of 1-phenyladamantane (13).



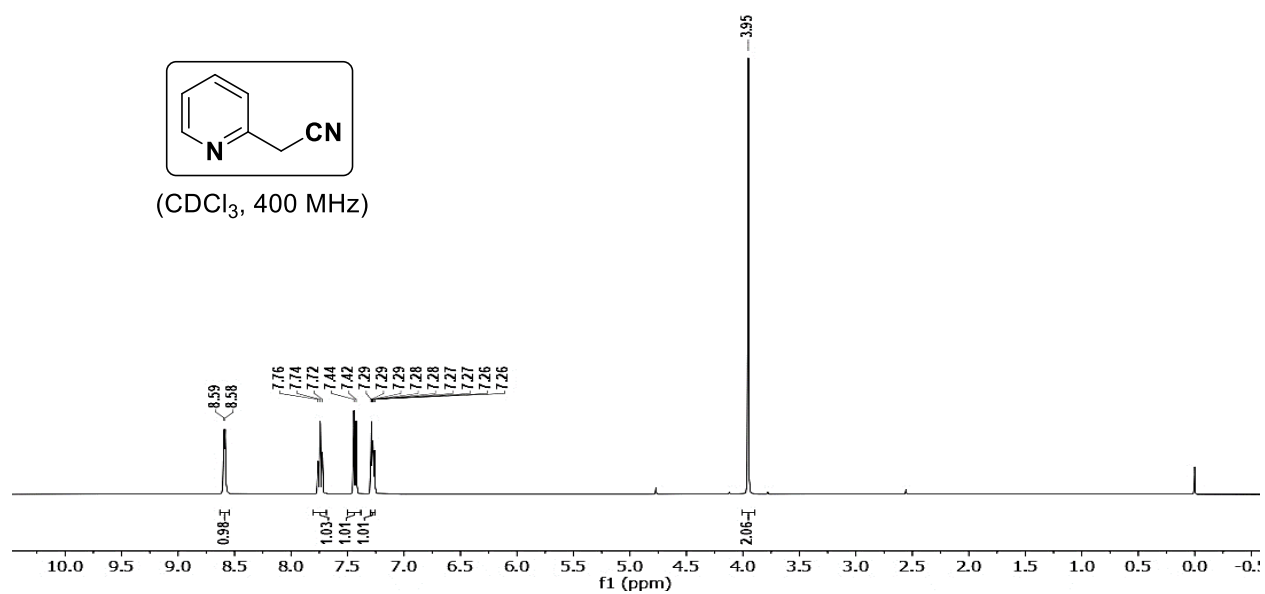
¹H NMR spectrum of ethyl 3-(cyanomethyl)-5-nitrobenzoate (14).



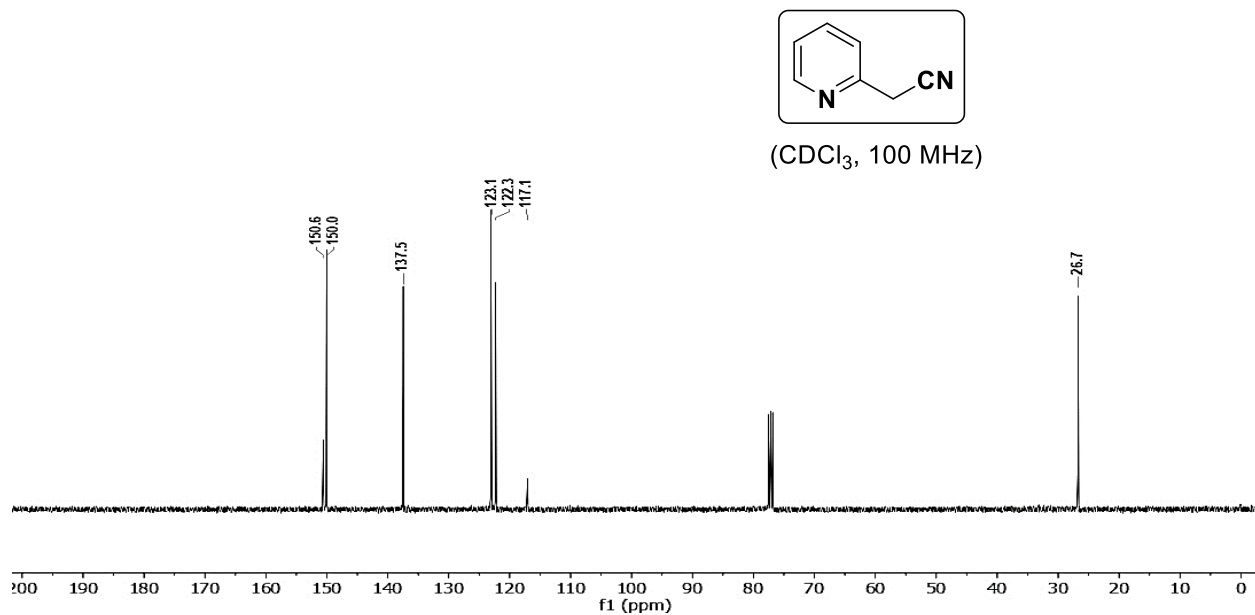
¹³C NMR spectrum of ethyl 3-(cyanomethyl)-5-nitrobenzoate (14).



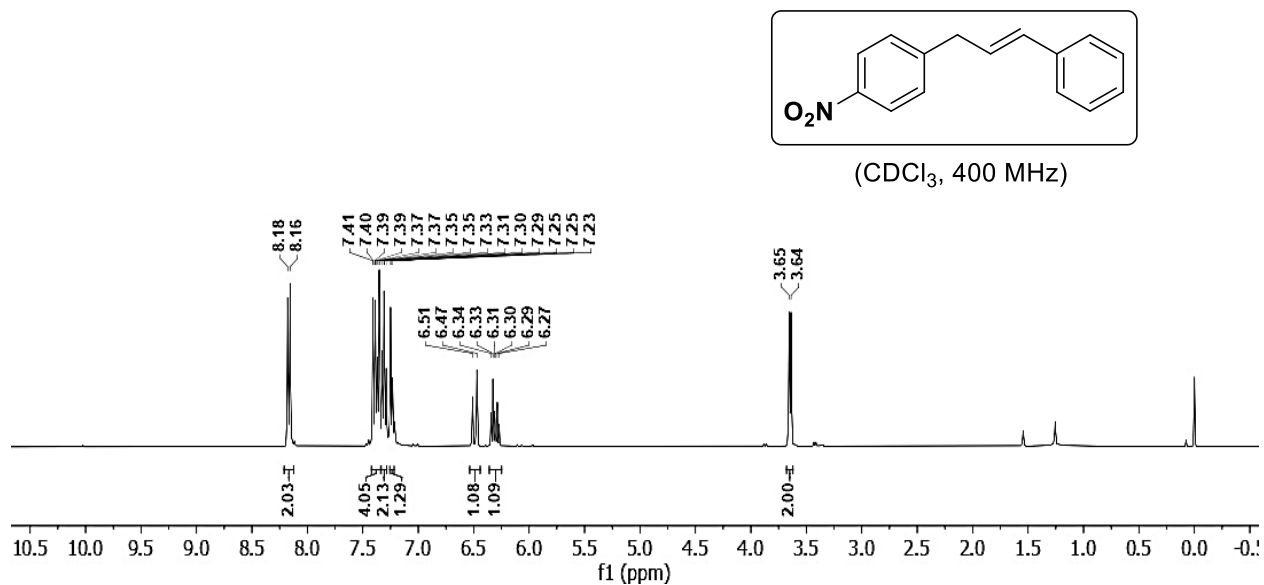
¹H NMR spectrum of 2-(pyridin-2-yl)acetonitrile (15).



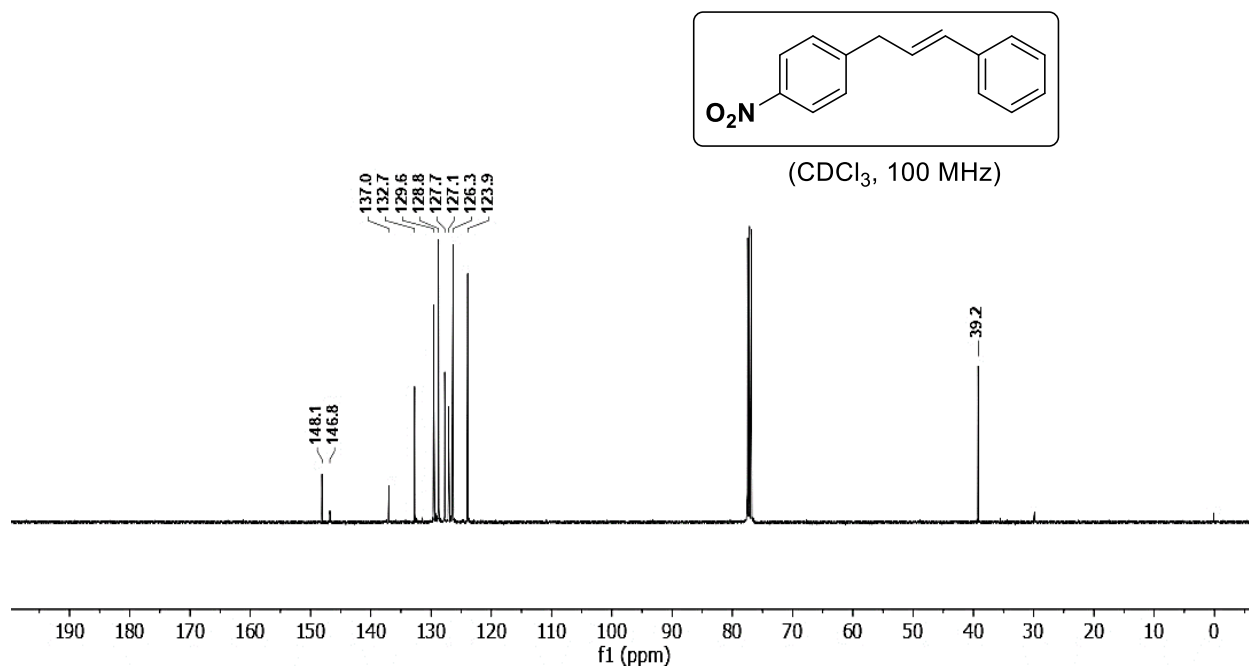
¹³C NMR spectrum of 2-(pyridin-2-yl)acetonitrile (15).



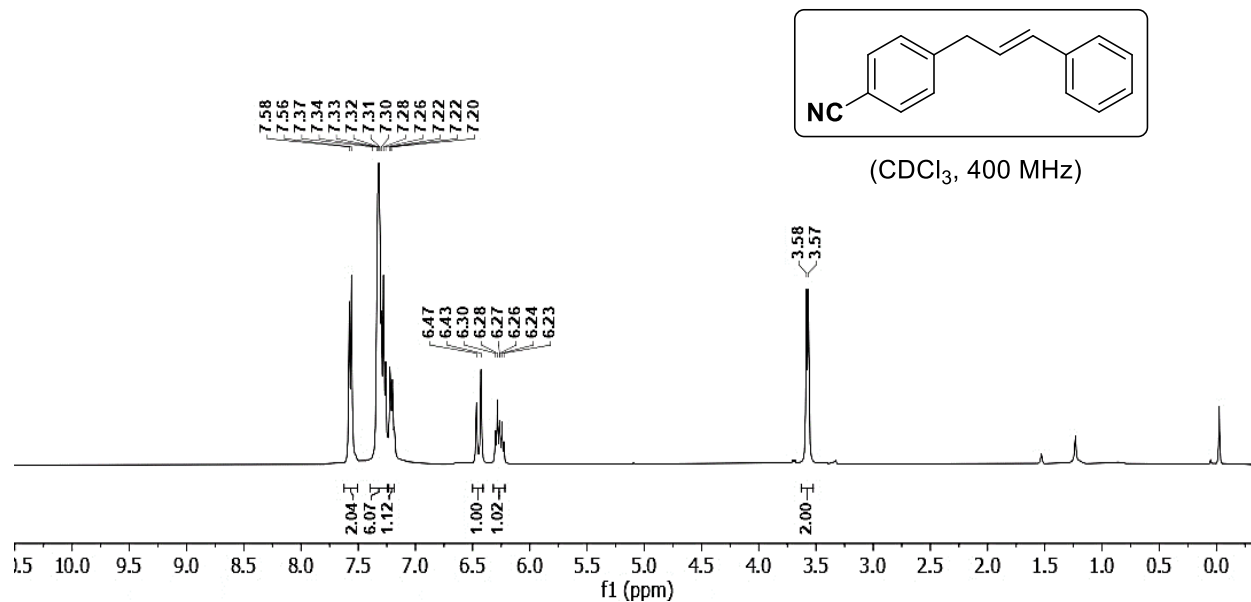
¹H NMR spectrum of (*E*)-3-(4-nitrophenyl)-1-phenylpropene (16).



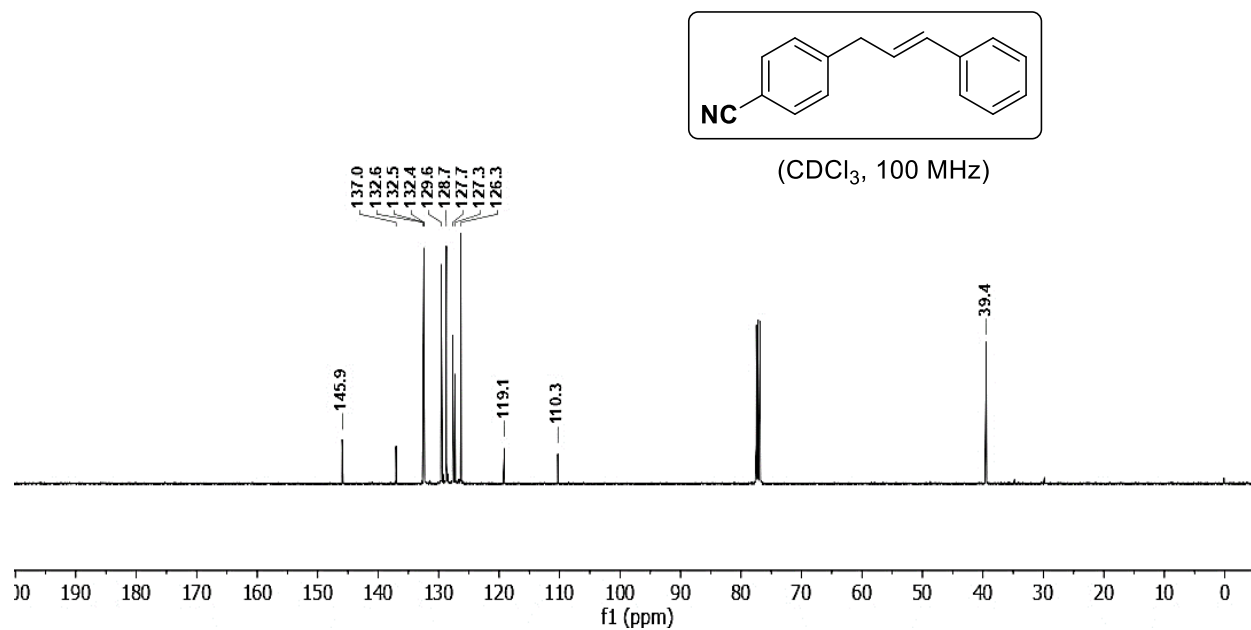
¹³C NMR spectrum of (*E*)-3-(4-nitrophenyl)-1-phenylpropene (16).



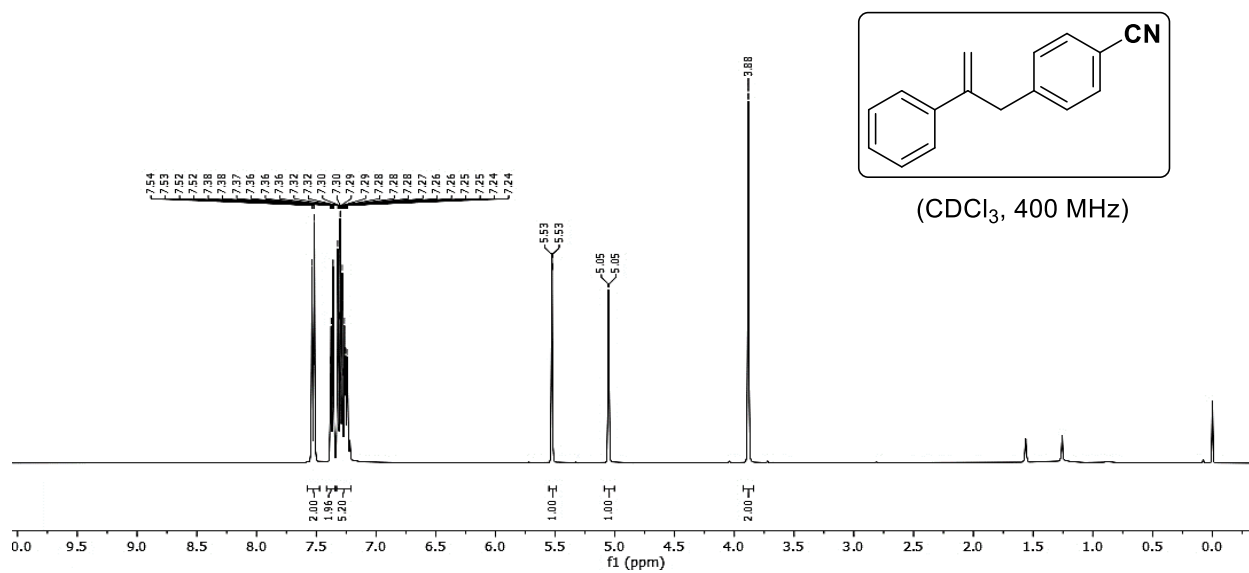
¹H NMR spectrum of (*E*)-3-(4-cyanophenyl)-1-phenylpropene (17).



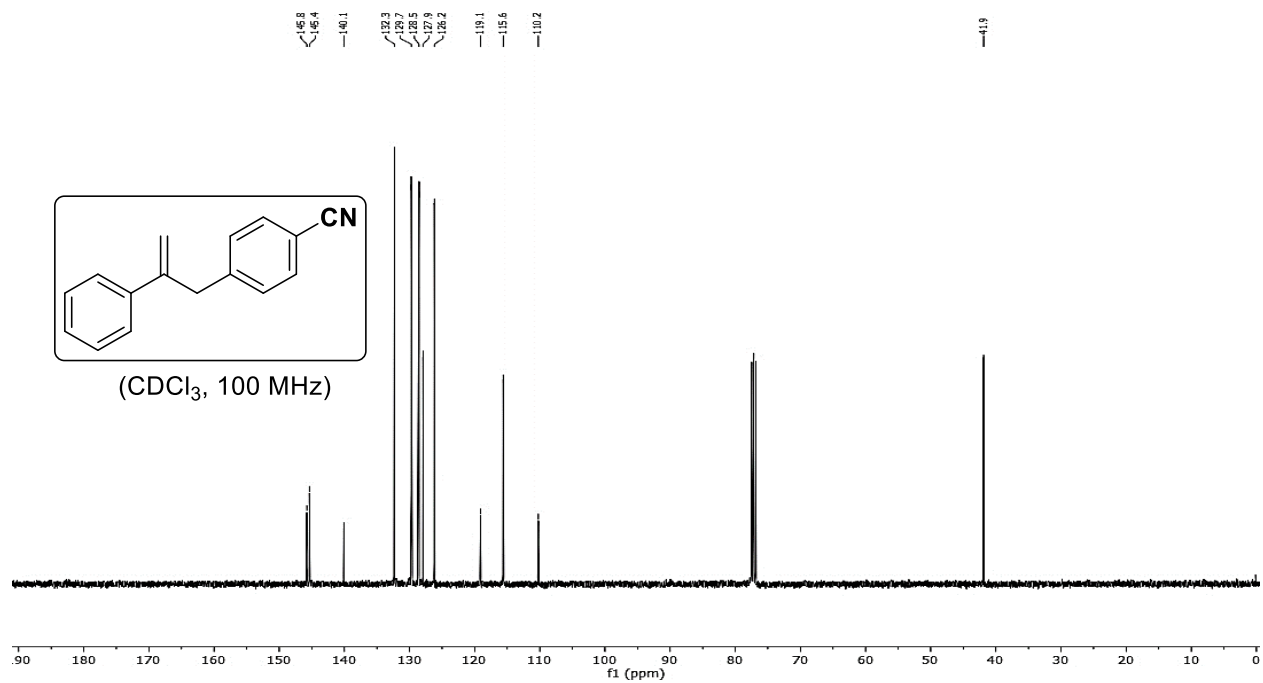
¹³C NMR spectrum of (*E*)-3-(4-cyanophenyl)-1-phenylpropene (17).



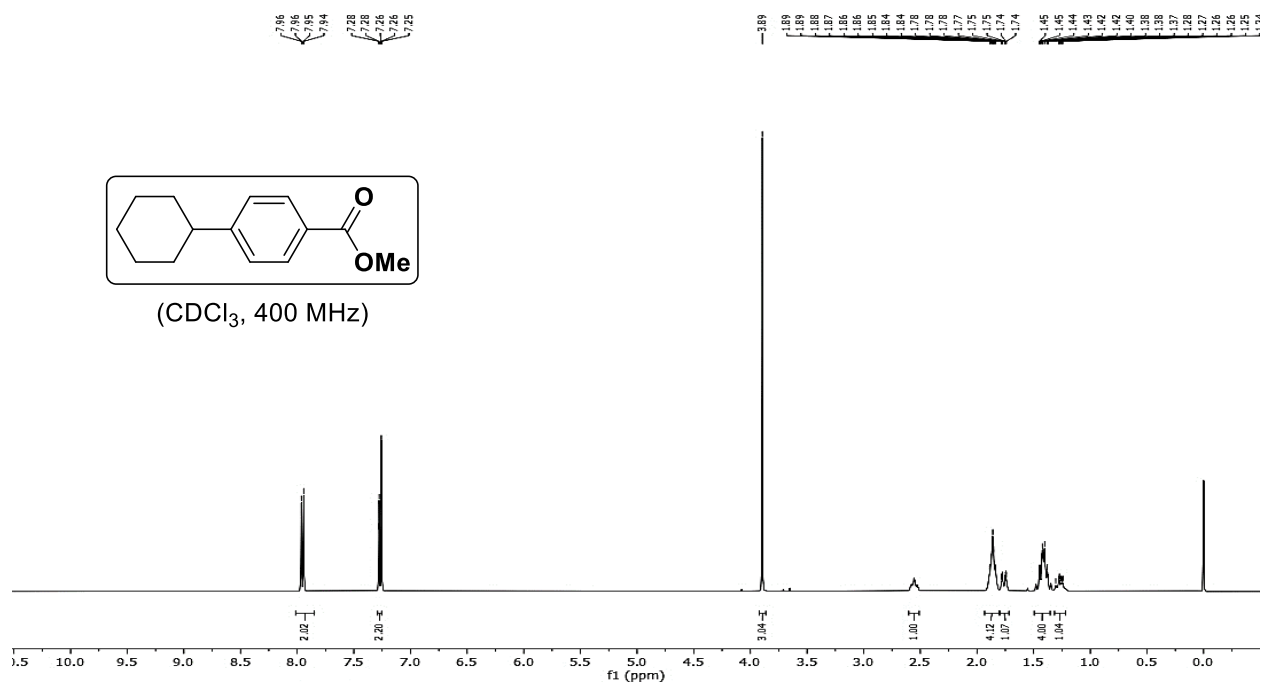
¹H NMR spectrum of 4-(2-phenylallyl)benzonitrile (18).



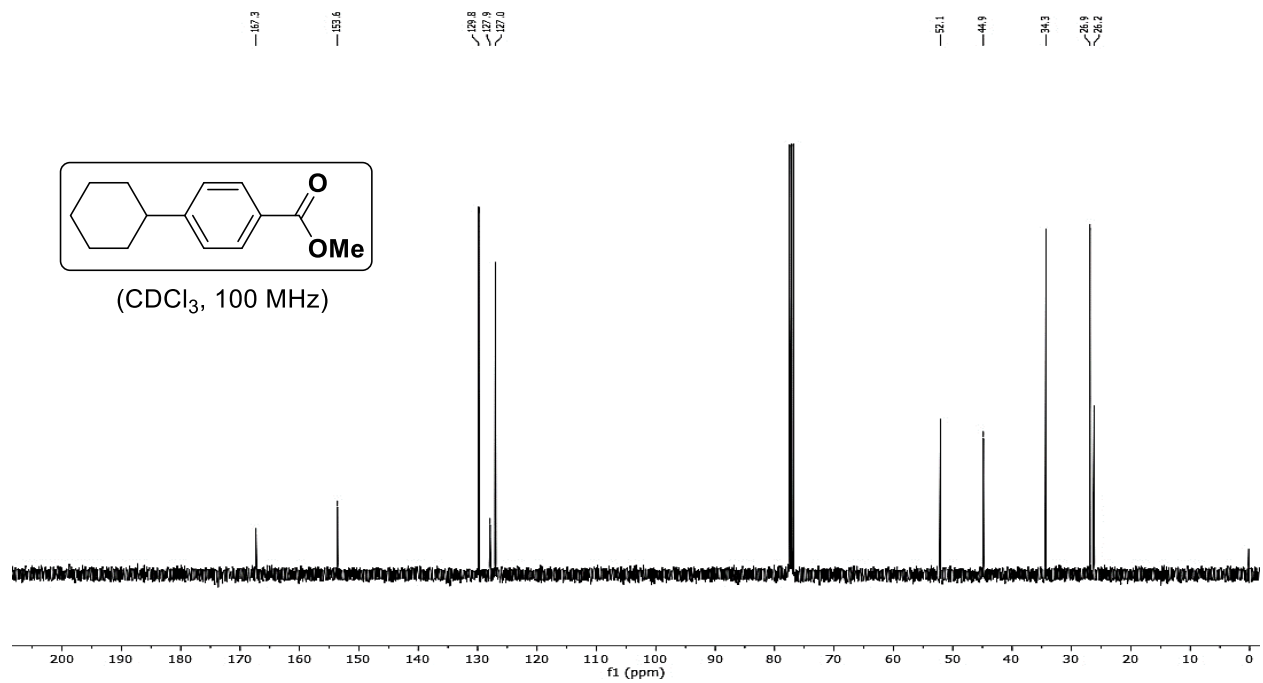
¹³C NMR spectrum of 4-(2-phenylallyl)benzonitrile (18).



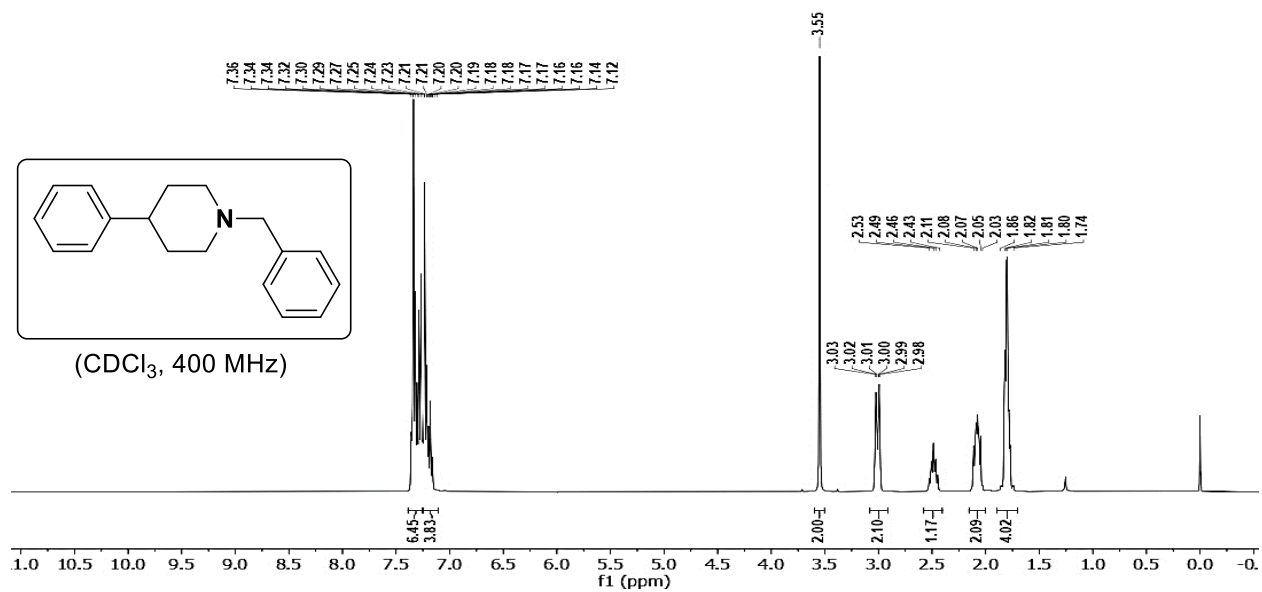
¹H NMR spectrum of methyl 4-cyclohexylbenzoate (21).



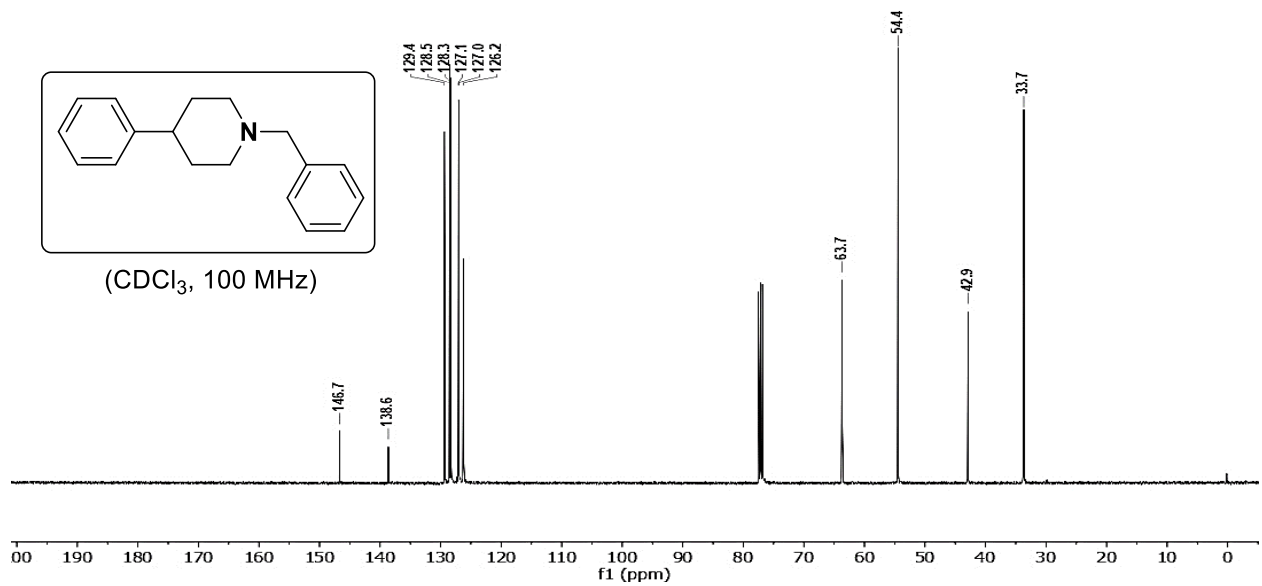
¹³C NMR spectrum of methyl 4-cyclohexylbenzoate (21).



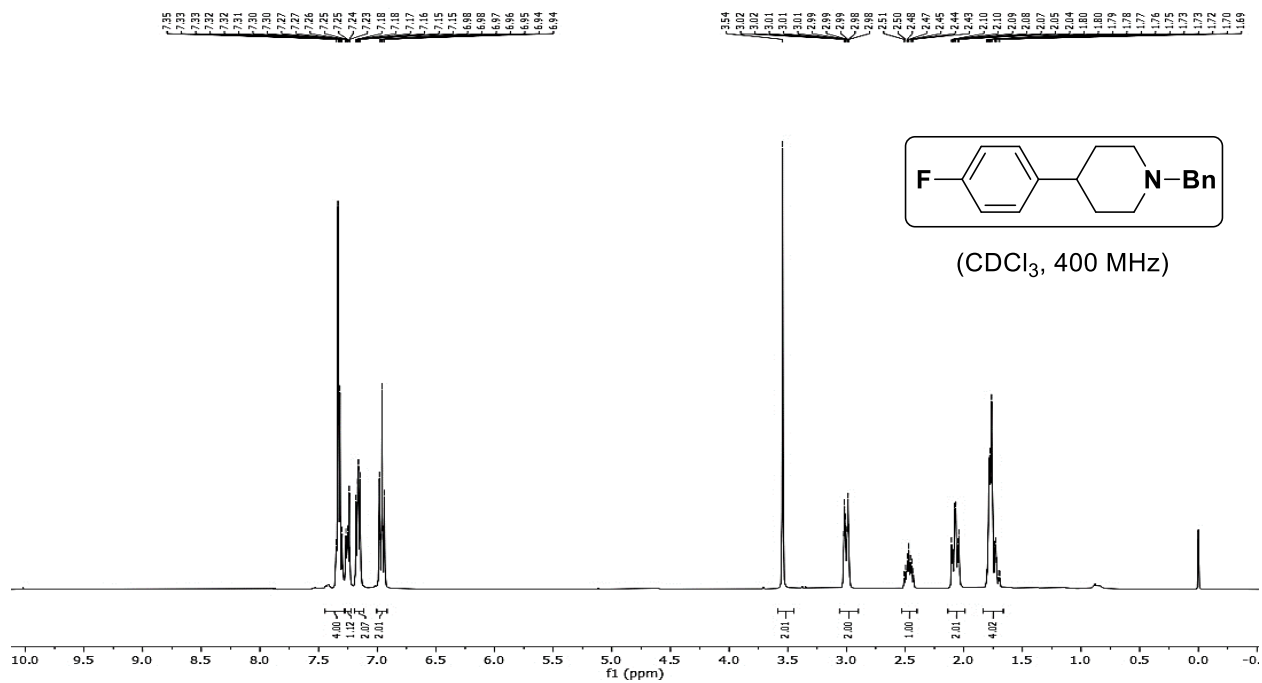
¹H NMR spectrum of 1-benzyl-4-phenylpiperidine (23).



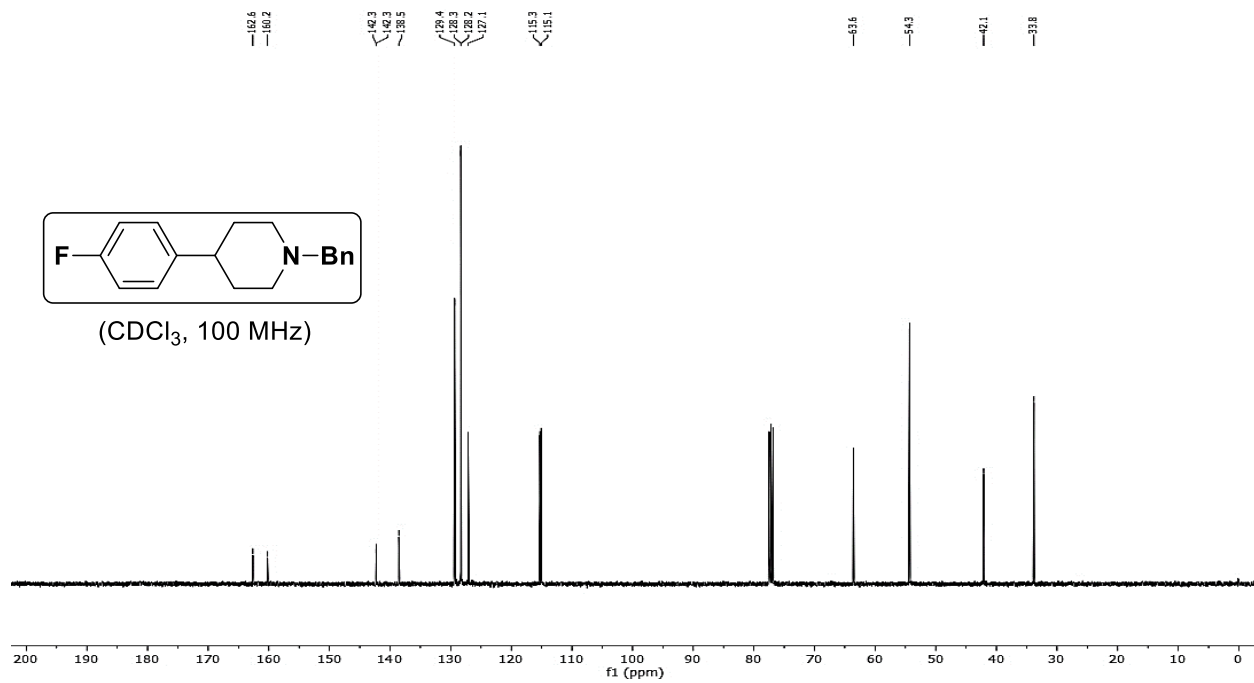
¹³C NMR spectrum of 1-benzyl-4-phenylpiperidine (23).



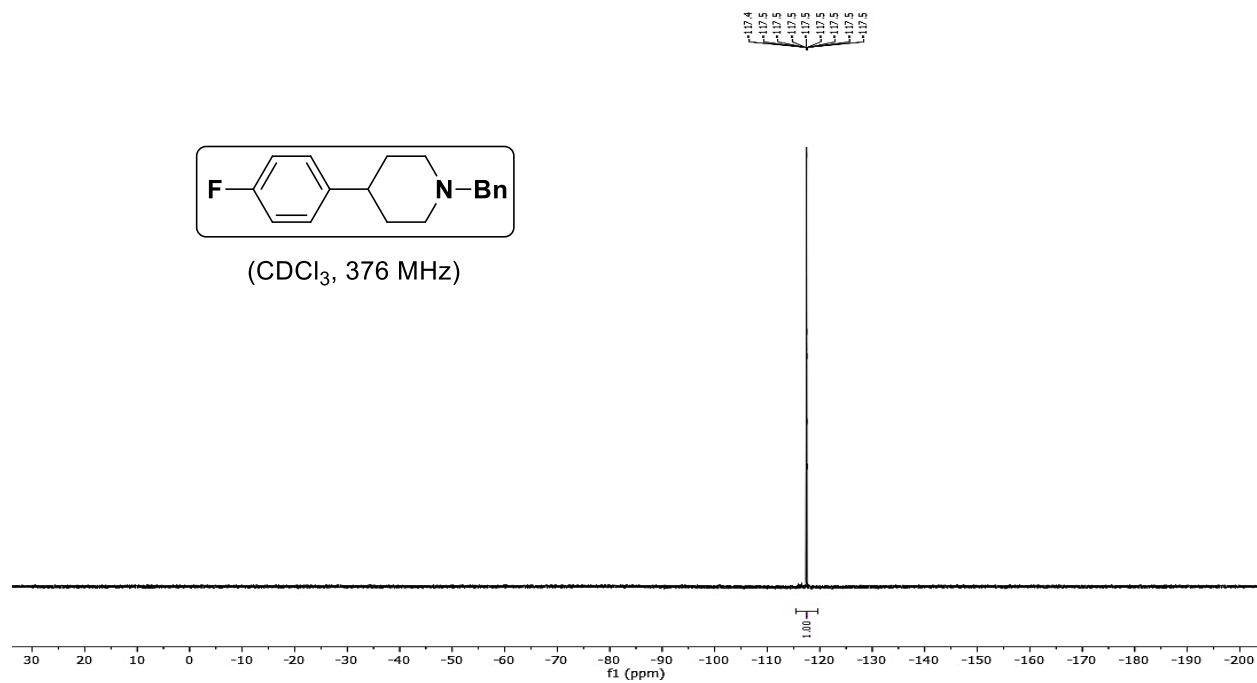
¹H NMR spectrum of 1-benzyl-4-(4-fluorophenyl)piperidine (24).



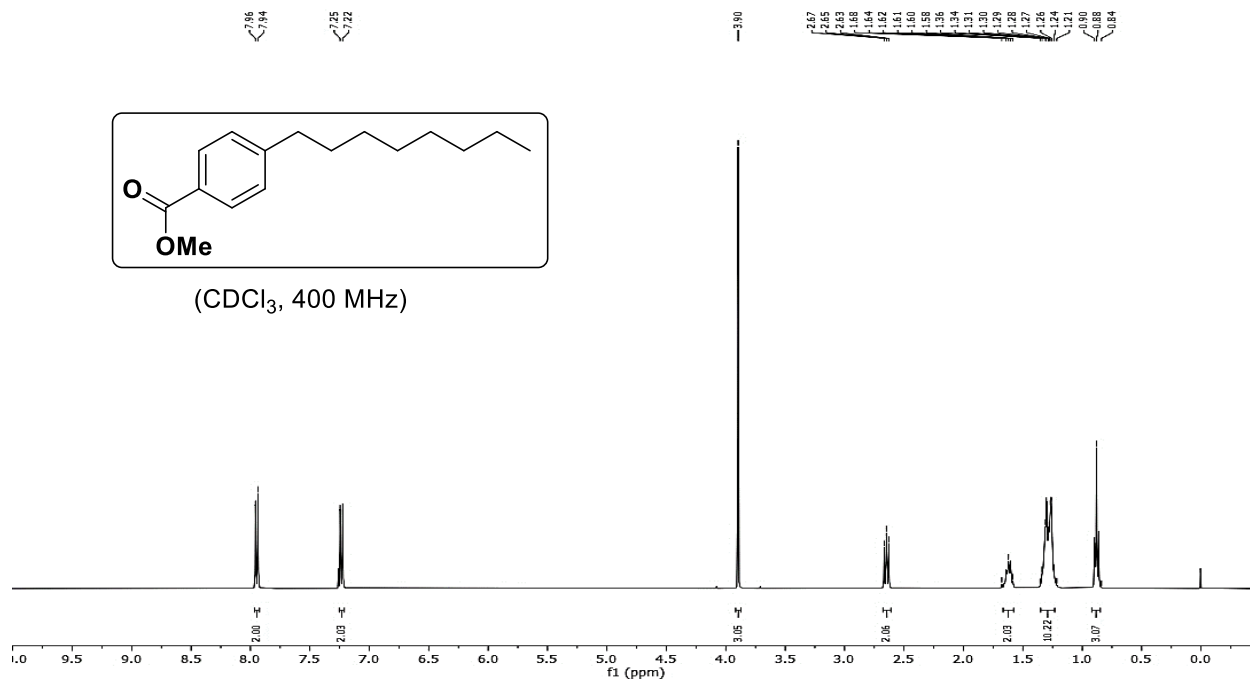
¹³C NMR spectrum of 1-benzyl-4-(4-fluorophenyl)piperidine (24).



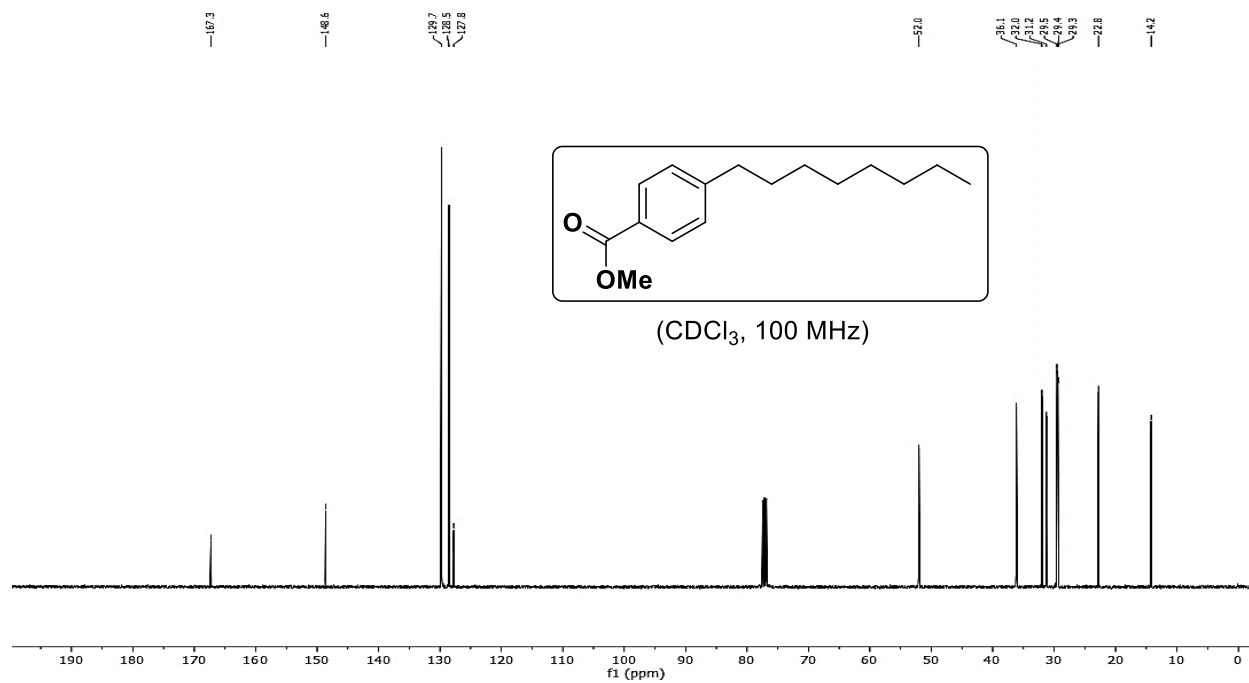
¹⁹F NMR spectrum of 1-benzyl-4-(4-fluorophenyl)piperidine (24).



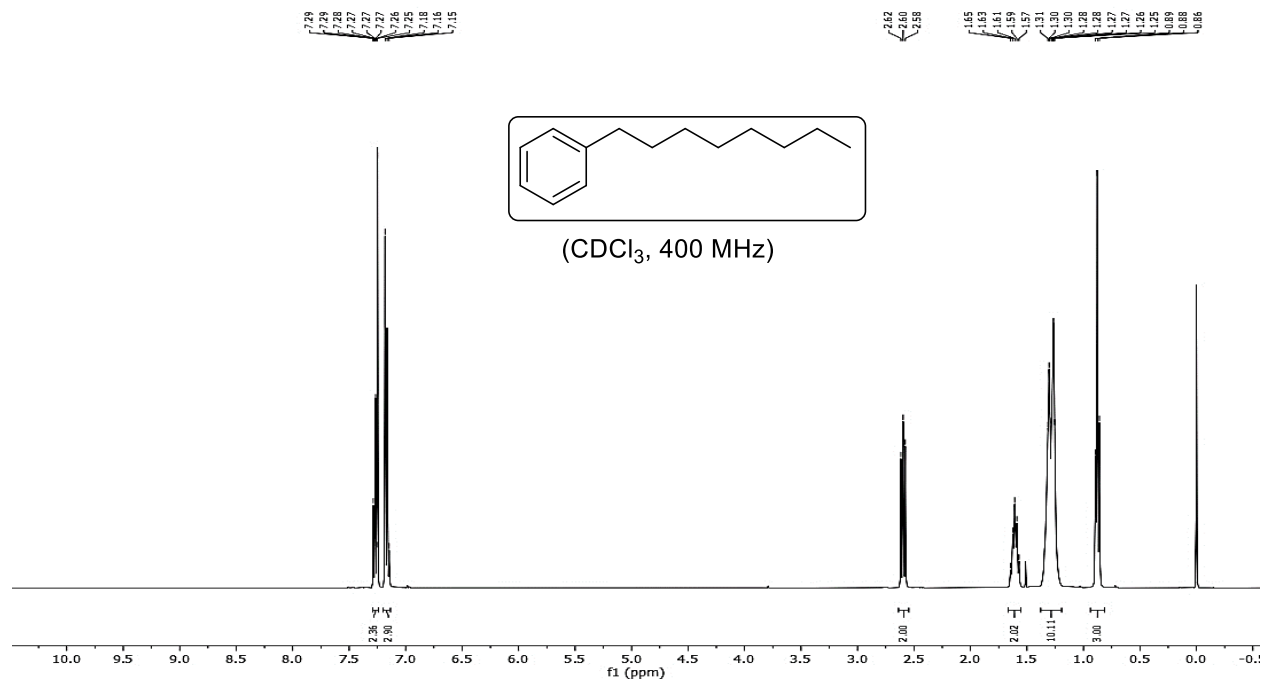
¹H NMR spectrum of methyl 4-octylbenzoate (25).



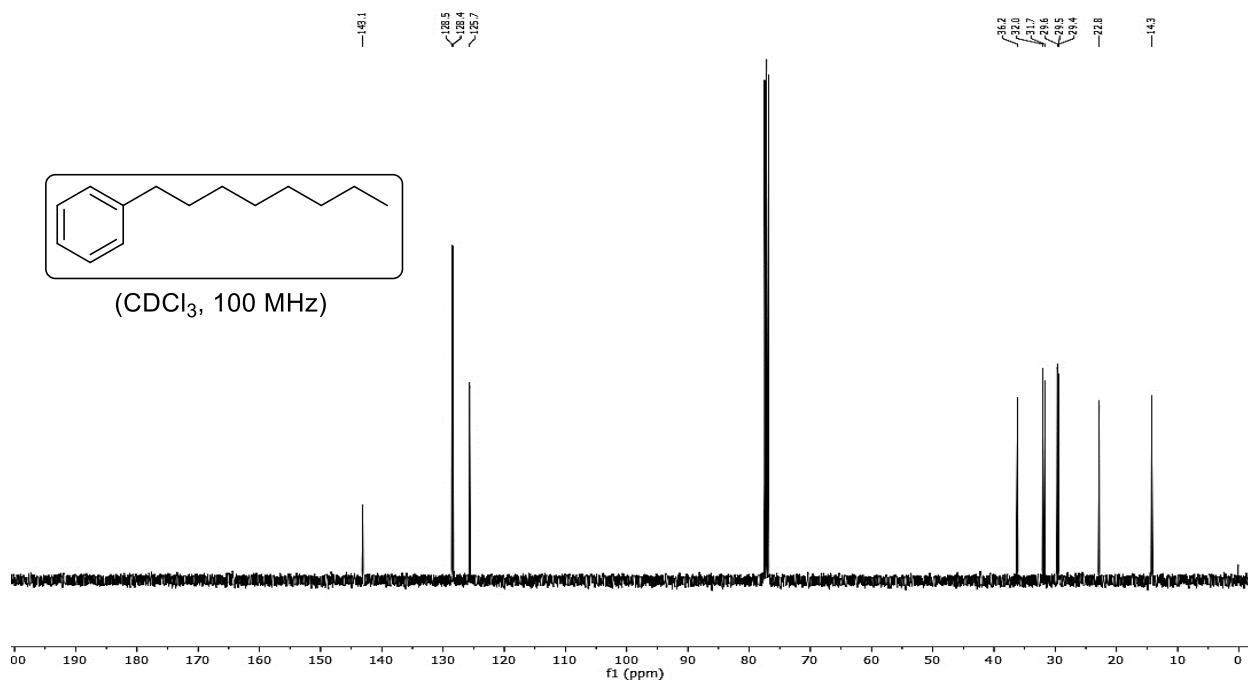
^{13}C NMR spectrum of methyl 4-octylbenzoate (25).



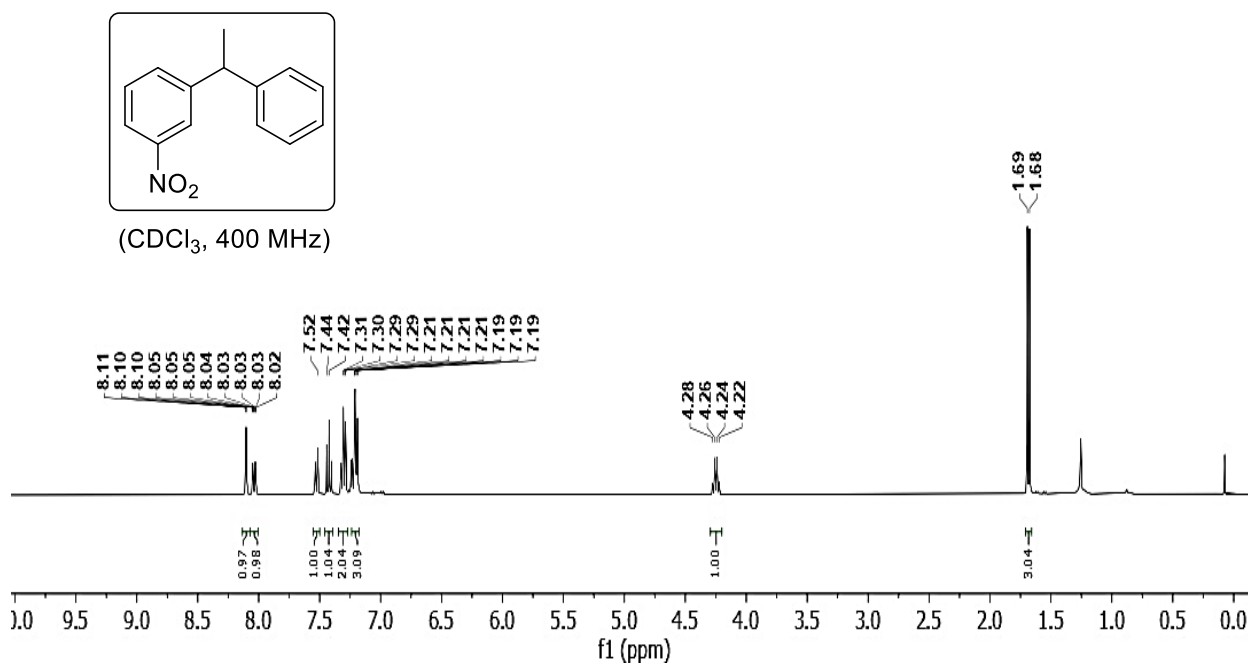
^1H NMR spectrum of octylbenzene (26).



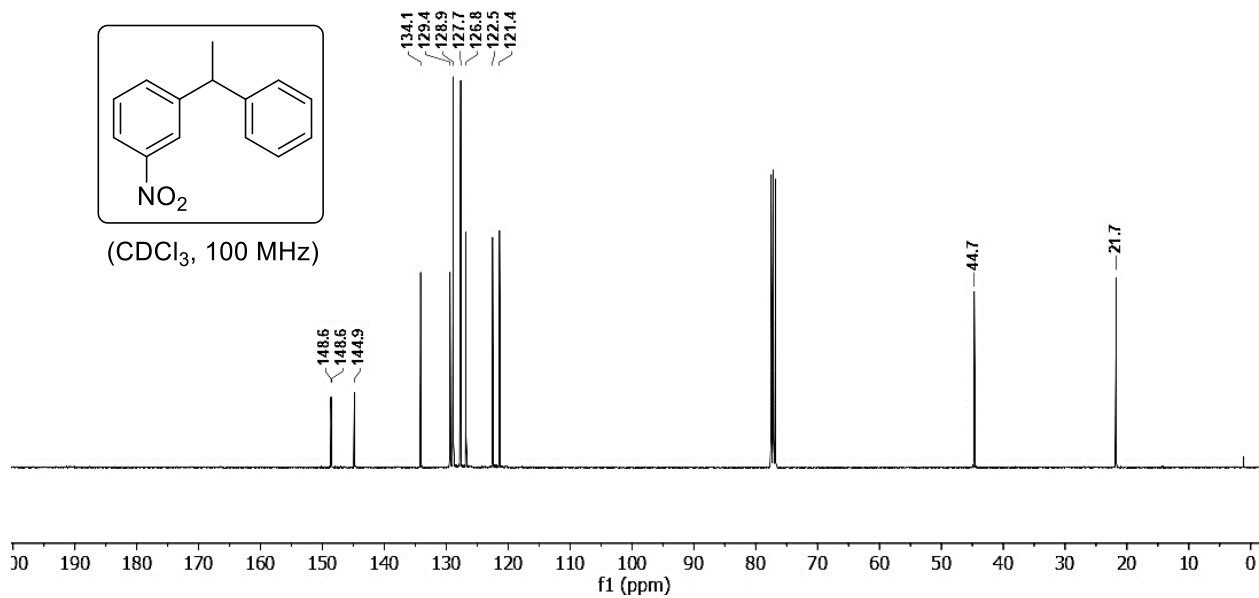
^{13}C NMR spectrum of octylbenzene (26).



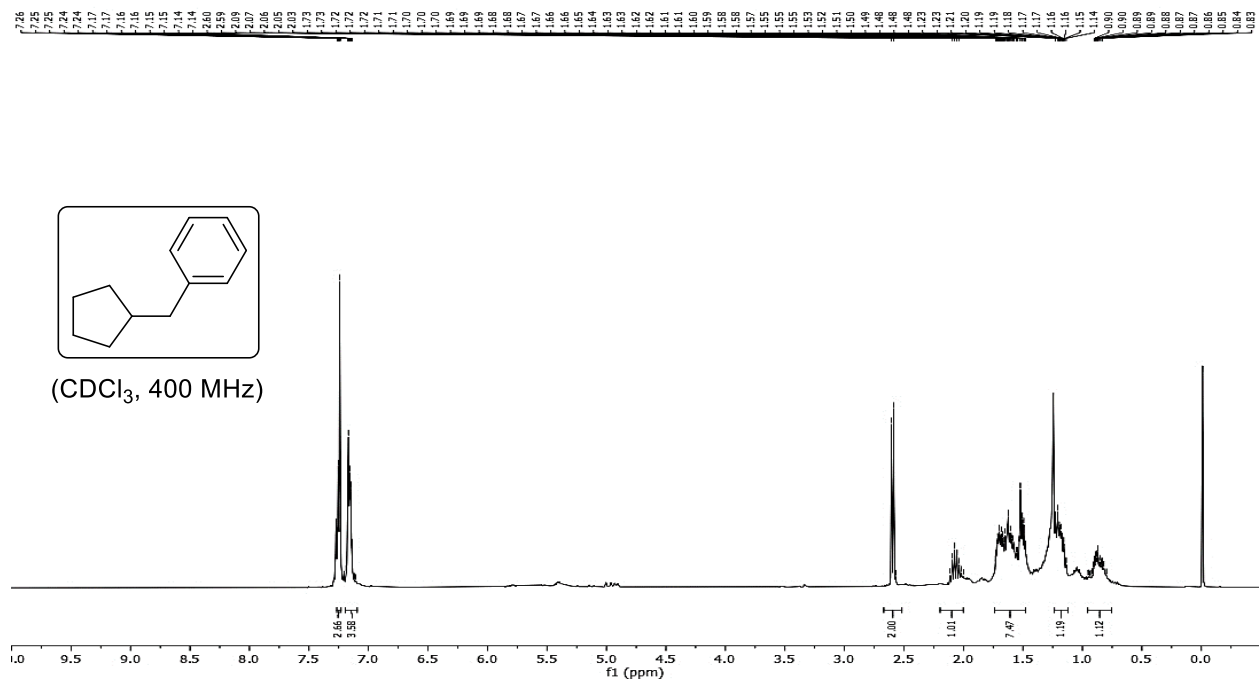
^1H NMR spectrum of 1-nitro-3-(1-phenylethyl)benzene (27).



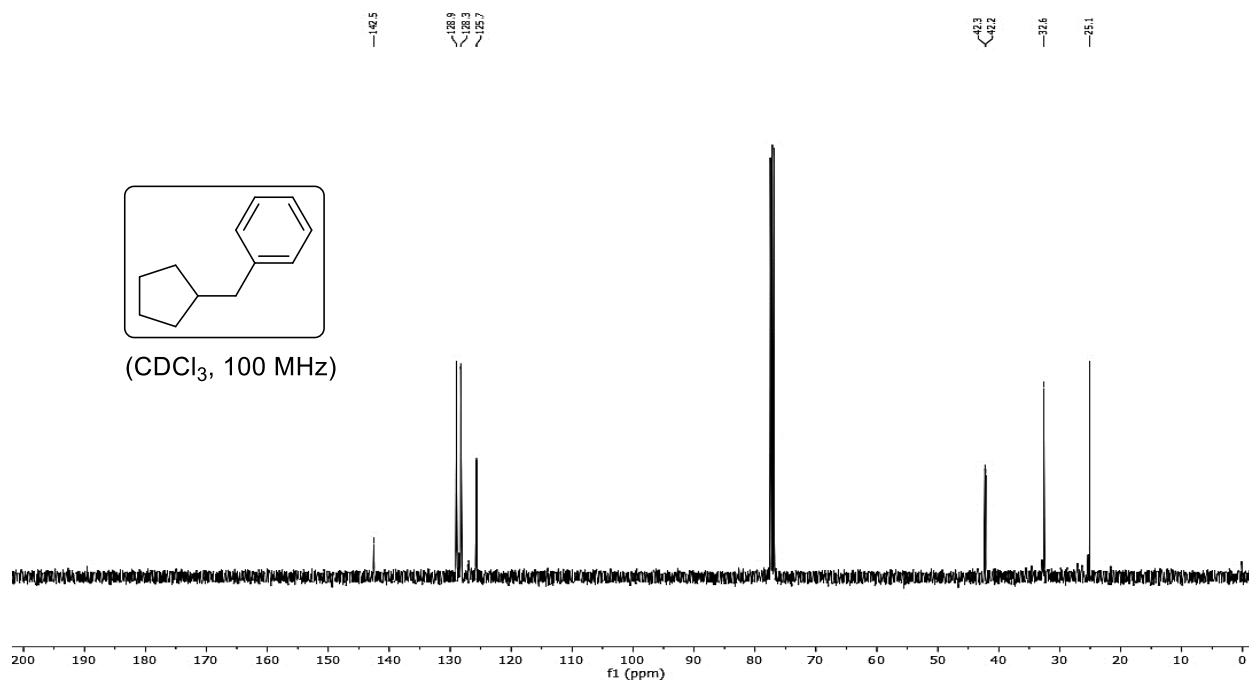
¹³C NMR spectrum of 1-nitro-3-(1-phenylethyl)benzene (27).



¹H NMR spectrum of (cyclopentylmethyl)benzene (29).



^{13}C NMR spectrum of (cyclopentylmethyl)benzene (29).



7. Crystallographic analysis⁵

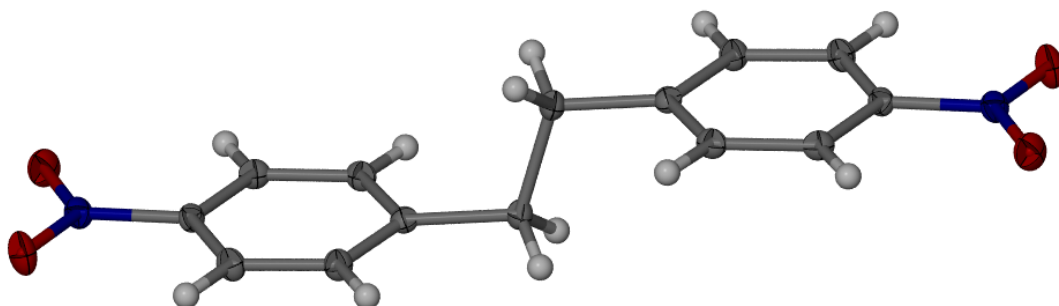


Figure S1. X-ray structure (ellipsoids are drawn at the 50% probability level) of 1,2-bis(4-nitrophenyl)ethane (**4**)

A single crystal was obtained by slow evaporation of a solution containing the homo-coupling product in a mixture of hexanes and ethyl acetate (4:1). Single crystal X-ray analysis was performed at 100 K using a Siemens platform diffractometer with graphite monochromated Mo- $K\alpha$ radiation ($\lambda = 0.71073 \text{ \AA}$). Data were integrated with the Bruker SAINT program. Structure solution and refinement were performed using the SHELXT/PC suite and ShelXL. Intensities were corrected for Lorentz and polarization effects and an empirical absorption correction was applied using Blessing's method as incorporated into the program SADABS. Non-hydrogen atoms were refined with anisotropic thermal parameters and hydrogen atoms were included in idealized positions. Crystal data: $C_{14}H_{12}N_2O_4$, $M = 272.26$, pale yellow needle, $0.214 \times 0.115 \times 0.068 \text{ mm}^3$, monoclinic, space group $P2_1/c$, $a = 4.4892(6)$, $b = 13.2061(19)$, $c = 10.6589(15) \text{ \AA}$, $\alpha = 90$, $\beta = 98.845(2)$, $\gamma = 90$, $V = 624.40(15) \text{ \AA}^3$, $Z = 2$.

8. References

1. W-C. Dai, B. Yang, S-H. Xu, Z-X. Wang. *J. Org. Chem.* **2021**, *86*, 2235-2243.
2. Buden, M. E.; Bardagi, J. I.; Puiatti, M.; Rossi, R. A. *J. Org. Chem.* **2017**, *82*, 8325-8333.
3. R.B. Bedford, P.B. Brenner, E. Carter, T.W. Carvell, P.M. Cogswell, T. Gallagher, J.N. Harvey, D.M. Murphy, E.C. Neeve, J. Nunn, D.R. Pye. *Chem. Eur. J.* **2014**, *20*, 7935-7938.
4. V. Saini, L. Liao, Q. Wang, R. Jana, M.S. Sigman. *Org. Lett.* **2013**, *15*, 5008-5011.
5. CCDC 1992313 (compound **4**) contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre.