Cobalt Ferrite Supported on Reduced Graphene Oxide as a T₂ Contrast Agent for Magnetic

Resonance Imaging

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Figure SI1. EDS analysis of a representative CoFe₂O₄-rGO composite (the corresponding SEM image and probe marker are inset).

Sample	Fe (wt%)
CoFe ₂ O ₄	36.5
rHGO- 5 %	2
rHGO – 10 %	3
rHGO- 16 %	5
rHGO -30 %	9
rIGO- 5 %	3
rIGO – 10 %	4
rlGO- 16 %	5
rIGO -30 %	15

Table SI1. Concentration of Fe in the CoFe₂O₄ nanoparticles and composites, as measured by ICP-OES.

Table SI2. Average particle size (Dav) and respective standard deviation (σ) for the CoFe₂O₄ NPs, CoFe₂O₄-rHGO composites and CoFe₂O₄-rIGO composites (as estimated from TEM images).

Sample	Particle size			
	Dav (nm)ª	σ (nm) ^ь		
CoFe ₂ O ₄	12.3	2.6		
rHGO-5%	3	1		
rHGO -10%	3	0.8		
rHGO -16%	4	1		
rHGO -30%	7	2		
rIGO -5%	5	1		
rIGO -10%	7	1		
rIGO -16%	9	3		
rIGO -30%	12	2		



Figure SI2. (a) High-resolution O1s XPS spectra of the two 30% rGO composites; (b) Concentration of the surface oxygen atomic percentage as a function of the $CoFe_2O_4$ loading (data extracted from the corresponding XPS spectra).



Figure SI3. Low-field regions of the hysteresis loops at room temperature for: (a) CoFe₂O₄-rHGO composites and (b) CoFe₂O₄-rIGO composites.



Figure SI4. Room temperature field-dependent magnetic measurements for: (a) rHGO and (b) rIGO.

Table SI3. Magnetization at 30000 Oe ($M_{H=30000 \text{ Oe}}$), saturation magnetization (M_r) and magnetic coercivity (H_c) for the CoFe₂O₄ nanoparticles, CoFe₂O₄-rHGO composites and CoFe₂O₄-rIGO composites (extracted from the M vs. H loops).

CoFe ₂ O ₄	CoEe O 5%		CoEo O 10%		CoEo O 16%		CoFe O 30%		CoFo O 100%
Loading wt%	Core	204 376	COFe ₂ O ₄ 10%		COFE20410%		C0Fe20430%		COFE204100%
Composite	rHGO	rlGO	rHGO	rlGO	rHGO	rlGO	rHGO	rIGO	-
М _{н=30000 Ое} (emu/g)	0.08	0.50	0.84	4.70	2.50	10.00	12.70	24.71	55.51
M _r (emu/g)	0.00	0.00	0.00	0.05	0.02	0.78	0.29	2.20	10.00
H _c (Oe)	18.00	8.00	20.00	23.00	28.00	87.00	30.00	97.00	199



Figure SI5. Plot correlating the average particle size of $CoFe_2O_4$ with magnetization at 30,000 Oe for the two sets of rGO composites studied.

Table SI4. Particle size, with corresponding r_2 values, of two commercial contrast agents, two reported research samples and the present cobalt ferrite composites/nanoparticles.

Material	Particle size	r ₂ value	Reference
	(nm)	$(\mathbf{m}\mathbf{M}^{-1}\mathbf{sec}^{-1})$	
Feridex	120-180	98.3	YX. J. Wang, Quantitative imaging in
(Ferumoxides)			medicine and surgery, 2013, 3 , 1.
Resovist	45-60	151.0	YX. J. Wang, Quantitative imaging in
(Ferucarbotran)			medicine and surgery, 2013, 3 , 1.
Au–Fe ₃ O ₄ dumbbell	28	114.0	C. Xu, J. Xie, Angewandte Chemie
			International Edition, 2008, 47, 173-176.
CoFe ₂ O ₄ -GO	5-13	72.7	G. Wang, Chemical Engineering Journal,
			2016, 289 , 150-160.
CoFe ₂ O ₄ -rHGO 30%	7 (±2)	55.6	This work
CoFe ₂ O ₄ -rIGO 30%	12 (±2)	102.1	This work
CoFe ₂ O ₄	12 (±3)	112.4	This work



Figure SI6. Plot of the T_1 relaxation rate $(1/T_1)$ for: (a) CoFe₂O₄-rHGO composites and pure CoFe₂O₄, (b) CoFe₂O₄-rIGO composites and pure CoFe₂O₄; all samples were suspended in aqueous solution at different Fe concentrations.

Table SI5. Chronological evaluation of the ζ -potentials for the 30 wt% CoFe₂O₄-rHGO and 30 wt% CoFe₂O₄-rIGO composites dispersed in aqueous PBS (samples measured at 70 µg mL⁻¹). Mean values and standard deviations obtained from triplicate measurements.

Samples	ζ-potential in PBS (mV)			
	0 time	10 hours	24 hours	
30 wt% CoFe ₂ O ₄ -rHGO	-31.5 ±6.99	-31.8 ± 5.38	-35.0 ± 5.06	
30 wt% CoFe ₂ O ₄ -rIGO	-36.4 ± 6.09	-35.4 ±5.23	-39.2 ± 5.60	



Figure SI7. Photographs of 30 wt% CoFe₂O₄-rHGO and 30 wt% CoFe₂O₄-rIGO composites, dispersed at 70 μ g mL⁻¹ in PBS, after various time points.



Figure SI8. (a-b) TEM images of the 30 wt% $CoFe_2O_4$ -rHGO in PBS after 0 h and 24 h, respectively; (c-d) TEM images of the 30 wt% $CoFe_2O_4$ -rIGO in PBS after 0 h and 24 h, respectively.