

## Electronic Supporting Information (ESI)

### Response surface methodology for optimization of methylene blue adsorption onto carboxymethyl cellulose-based hydrogel beads: Adsorption kinetic, isotherm, thermodynamic and reusability studies

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#### Preparation of CMC-Alg/GO hydrogel beads

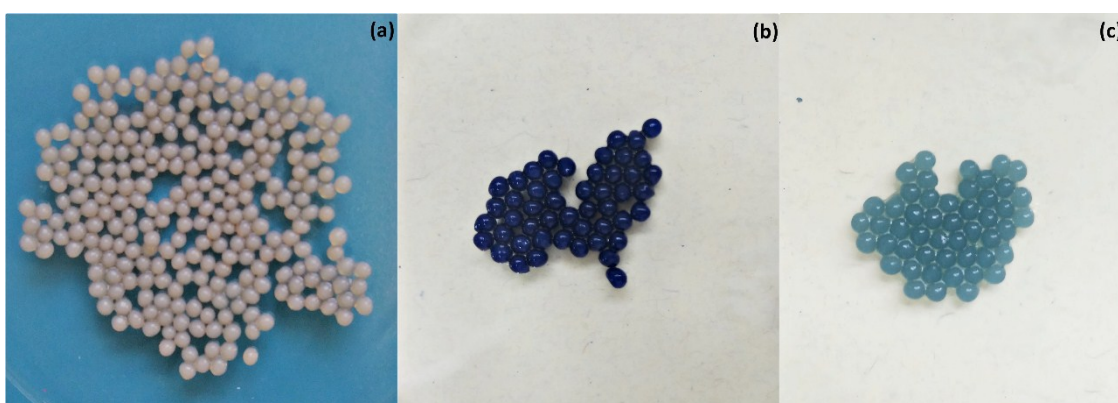


Fig.S1 Digital Images of wet CMC-Alg/GO beads (a) before MB adsorption, (b) after MB adsorption and (c) after MB desorption.

#### Characterization of CMC-Alg/GO beads

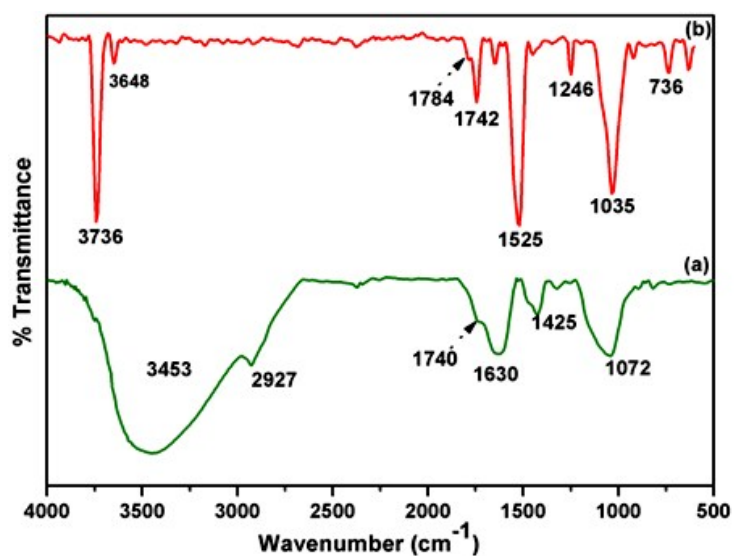


Fig.S2 ATR-FTIR spectra of CMC-Alg/GO (a) before and (b) after MB removal.

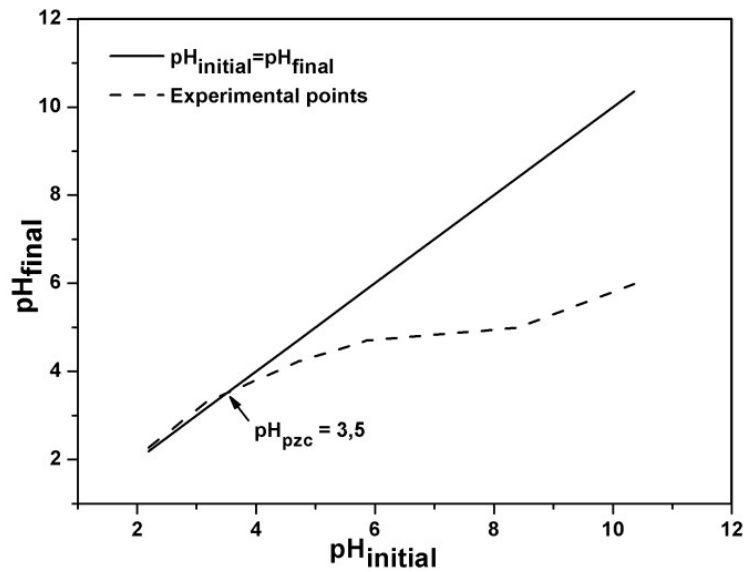


Fig.S3  $pH_{pzc}$  of wet CMC-Alg/GO beads.

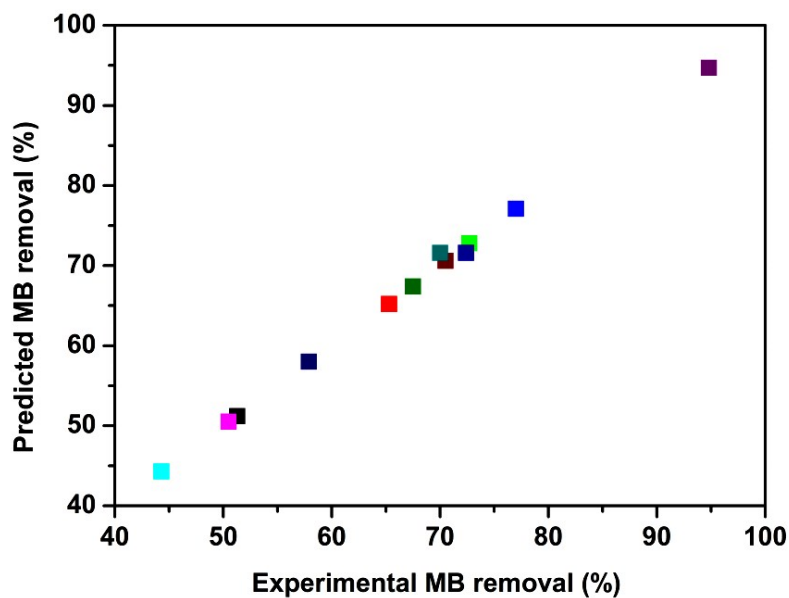


Fig.S4 Plot of the experimental and model-predicted values of the MB removal rate (%).

### Optimization of process parameters using response surface methodology

The ANOVA analysis expresses the equation and relationship between the response and significant variables. Thus, p-values inferior than 5% announce that the model has statistically significance and values superior than 10 % indicate the variables of the model are statistically not significant. In order to verify the validity of both regression models (% Removal and adsorption capacity mg/g), two experiments (run 7 and 8) were deactivated because of their high distance between the experimental and predicted values by the models. The test of student was statistically significant (p-value (%) <5%). The analysis of variance (ANOVA) (**Table S2**) revealed for both responses that the models were valid which was explained by the lower p-value (%) than 5%. There is an excellent correlation between the experimental and predicted values as proved by nearness between  $R^2$  and adjusted  $R^2$  value.

**Table S1 Analysis of variance (ANOVA) for quadratic model for Methylene Blue adsorption**

Response	Source	Sum of squares	D <sub>f</sub>	Mean square	F-value	P-value (%)	Comment
MB removal (%)	Regression	2099.47	9	233.275	187.53	0.0622	SD = 1.14 %
	Residuals	3.92	3	1.31	-	-	CV = 1.6%
	Lack of fit	0.08	1	0.08	0.0417	85.7	R <sup>2</sup> = 0.998
	Pure error	3.84	2	1.92	-	-	R <sup>2</sup> <sub>adjusted</sub> = 0.993
MB uptake (mg/g)	Regression	3629.04	9	403.27	39.34	0.587	SD = 3.20 mg/g
	Residuals	30.75	3	10.25	-	-	CV = 8.14%
	Lack of fit	22.11	1	22.11	5.19	15.2	R <sup>2</sup> = 0.992
	Pure error	8.64	2	4.32	-	-	R <sup>2</sup> <sub>adjusted</sub> = 0.966

**Table S2. Estimated regression coefficients for MB removal efficiency (%) in coded units**

Terms	Coefficients	SE coefficient	t-value	P-value (%)
Constant	71.600	0.659	108.490	0.01
X <sub>1</sub>	4.575	0.571	8.000	0.407
X <sub>2</sub>	8.375	0.404	20.720	0.0246
X <sub>3</sub>	-3.675	0.571	-6.430	0.763
X <sub>1</sub> <sup>2</sup>	-16.987	0.719	-23.620	0.0166
X <sub>2</sub> <sup>2</sup>	11.962	0.719	16.630	0.0473
X <sub>3</sub> <sup>2</sup>	-10.887	0.719	-15.140	0.0626
X <sub>1</sub> X <sub>2</sub>	-2.425	0.571	-4.240	2.40
X <sub>1</sub> X <sub>3</sub>	1.475	0.989	1.490	23.3
X <sub>2</sub> X <sub>3</sub>	-9.975	0.571	-17.450	0.0410

**Table S3. Estimated regression coefficients for MB removal capacity (mg/g) in coded units**

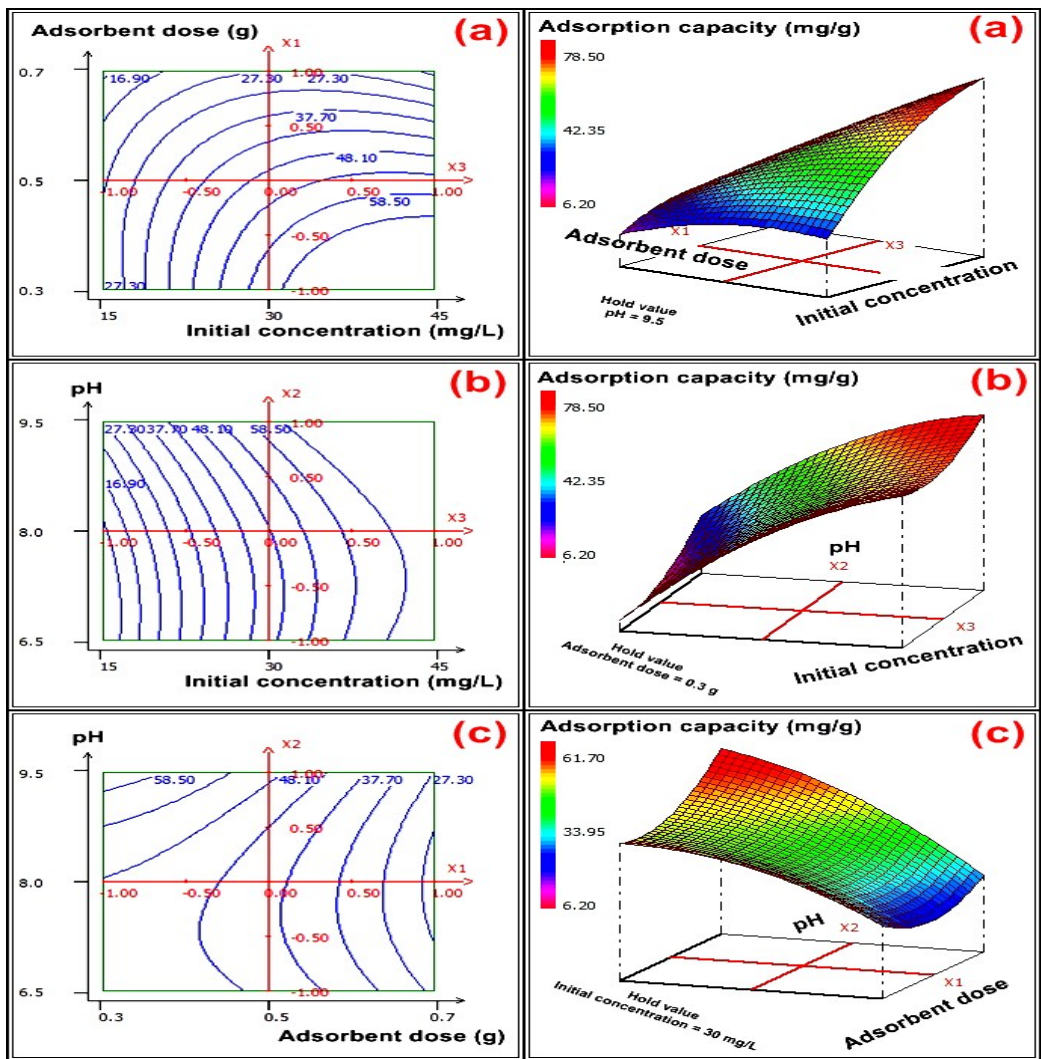
Terms	Coefficients	SE coefficient	t-value	P-value (%)
Constant	39.30	1.84	21.26	0.0228
X <sub>1</sub>	-13.700	1.600	-8.56	0.335
X <sub>2</sub>	4.088	1.131	3.61	3.65
X <sub>3</sub>	16.525	1.600	10.32	1.194
X <sub>1</sub> <sup>2</sup>	-5.200	2.014	-2.58	8.2
X <sub>2</sub> <sup>2</sup>	6.300	2.014	3.13	5.2
X <sub>3</sub> <sup>2</sup>	-9.025	2.014	-4.48	2.07
X <sub>1</sub> X <sub>2</sub>	-3.550	1.600	-2.22	11.3
X <sub>1</sub> X <sub>3</sub>	-11.650	2.772	-4.20	2.46
X <sub>2</sub> X <sub>3</sub>	-2.425	1.600	-1.51	22.7

**MB adsorption capacity response:**

As exposed in **Table S4**, the linear model terms (adsorbent dose (X<sub>1</sub>), pH (X<sub>2</sub>) and initial dye concentration (X<sub>3</sub>)), quadratic model terms (X<sub>1</sub><sup>2</sup>, X<sub>2</sub><sup>2</sup> and X<sub>3</sub><sup>2</sup>) and interactive model term (X<sub>1</sub>X<sub>3</sub>) are statistically significant. According to the BBD results, the final model is given by equation below:

$$\hat{Y}_2 = 39.30 - 13.7 X_1 + 4.088 X_2 + 16.525 X_3 - 5.2 X_1^2 + 6.3 X_2^2 - 9.025 X_3^2 - 11.65 X_1 X_3$$

It is obvious that the capacity of MB adsorbed gets increased with increase in the initial MB concentration which enhances the interaction between MB and CMC-Alg/GO beads (**Fig. S5.a and S5.b**).



**Fig. S5** 2D-3D response surface plots of the effect (a) the adsorbent dose and initial concentration, (b) the pH and initial concentration and (c) the pH and adsorbent dose on MB adsorption capacity (mg/g) by CMC-Alg/GO beads.