

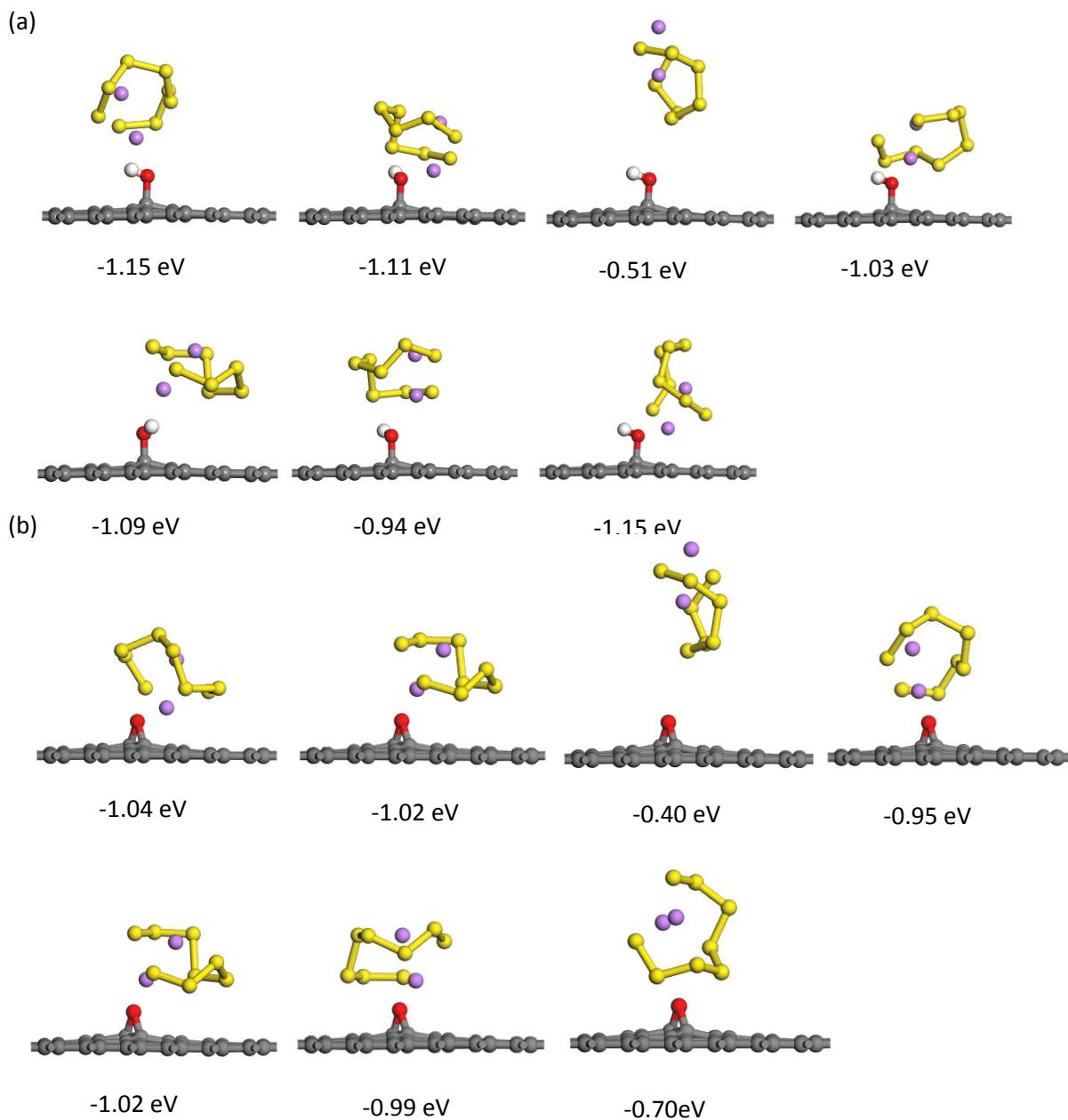
Supporting Information for

Interaction between functionalized graphene and sulfur compounds in lithium-sulfur battery – a density functional theory investigation

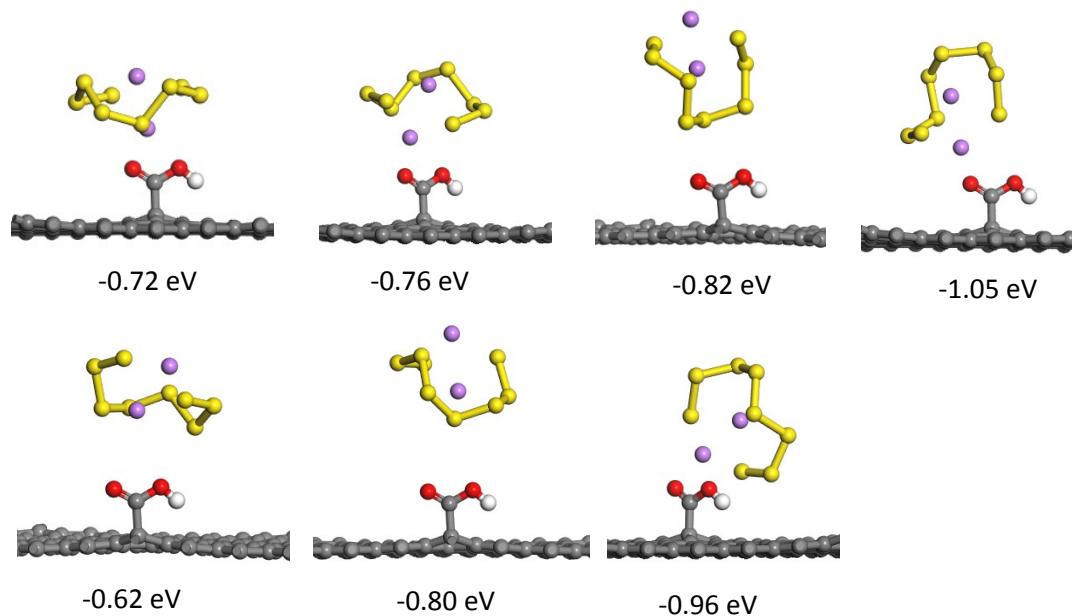
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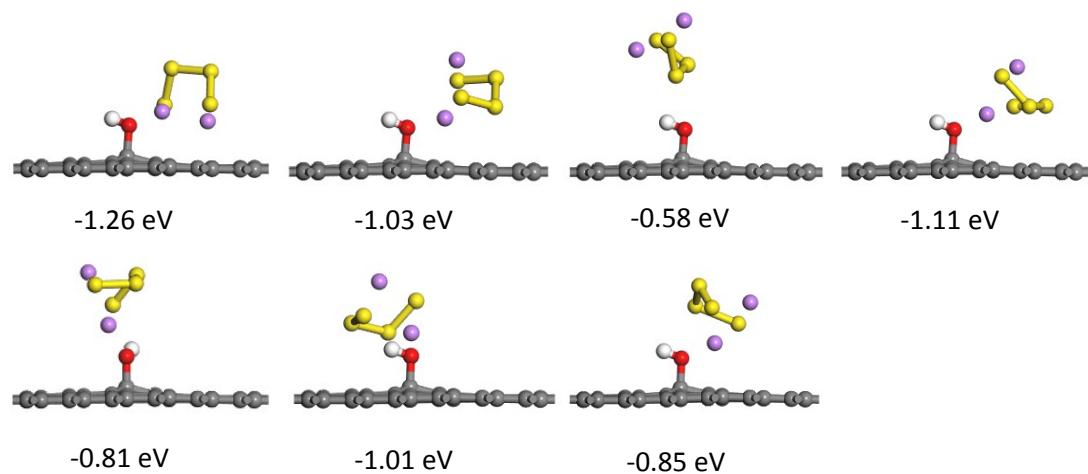
*Email: c2.yan@qut.edu.au



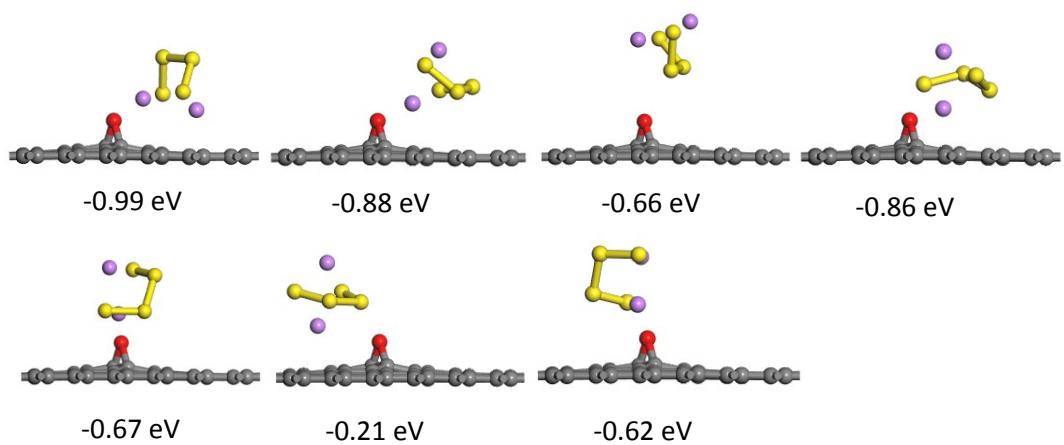
(c)



(d)



(e)



(f)

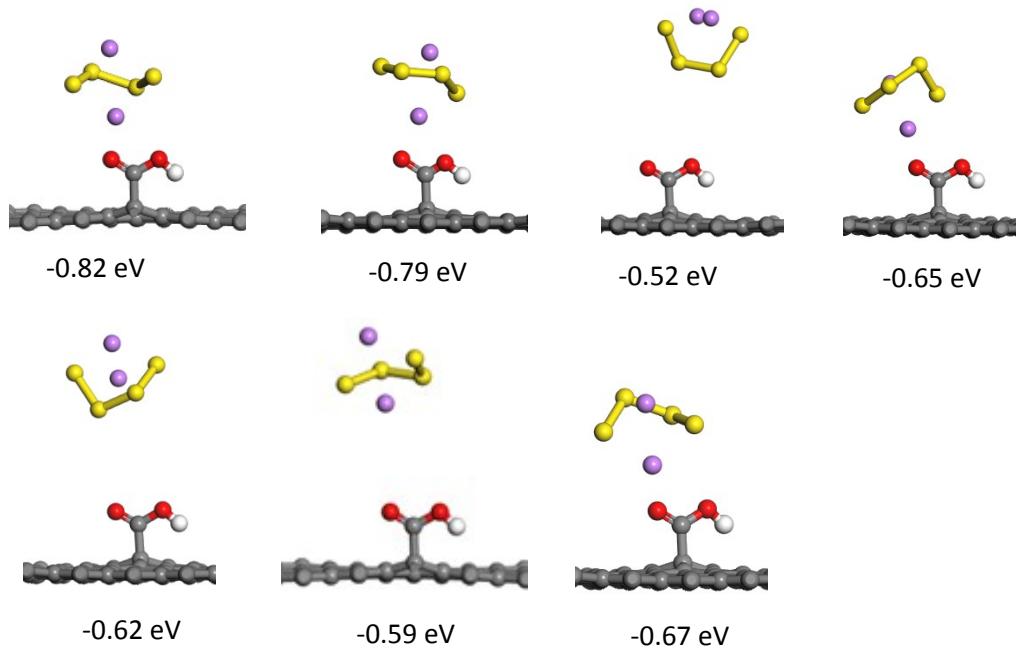


Fig. S1 Different configurations and corresponding adsorption energies of Li_2S_8 and Li_2S_4 on hG (a,d), eG (b,e) and cG (c,f)

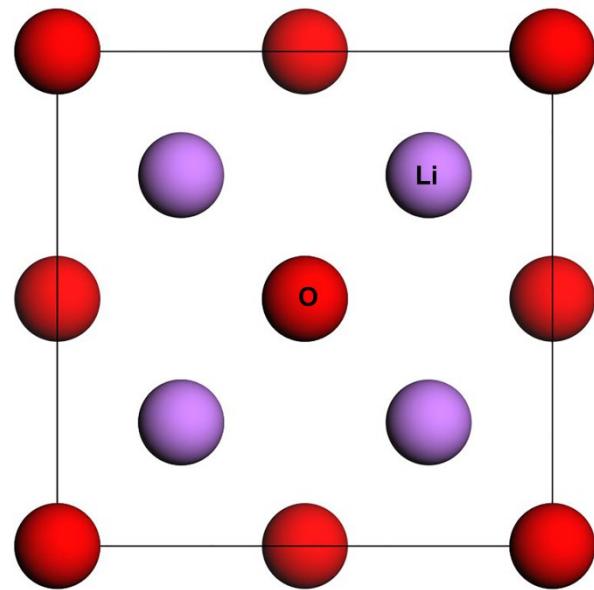


Fig. S2 Optimized structure of Li_2O

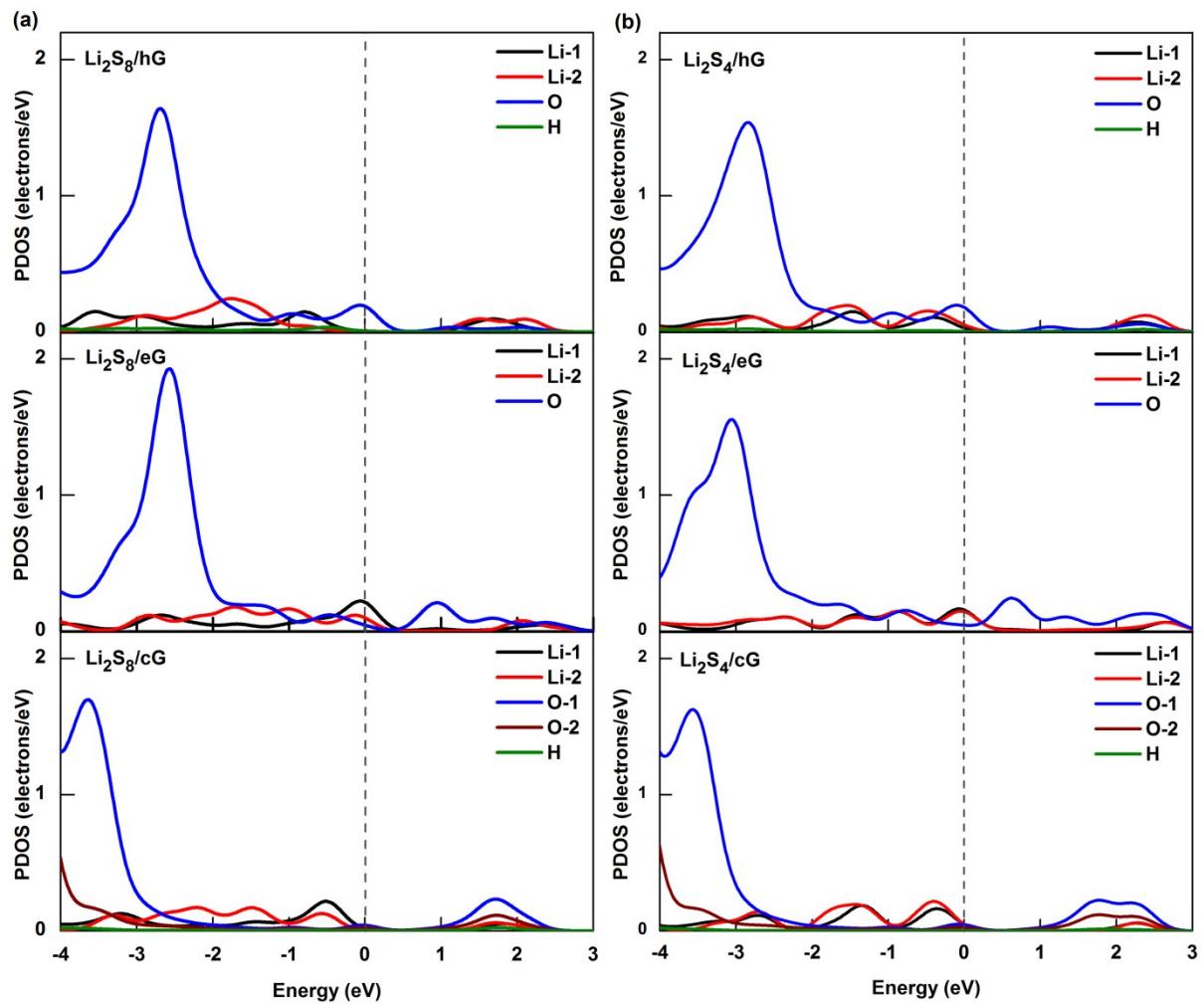


Fig. S3 Atomic partial density of states near the Fermi energy region for (a) Li₂S₈ and (b) Li₂S₄ on hG, eG and cG

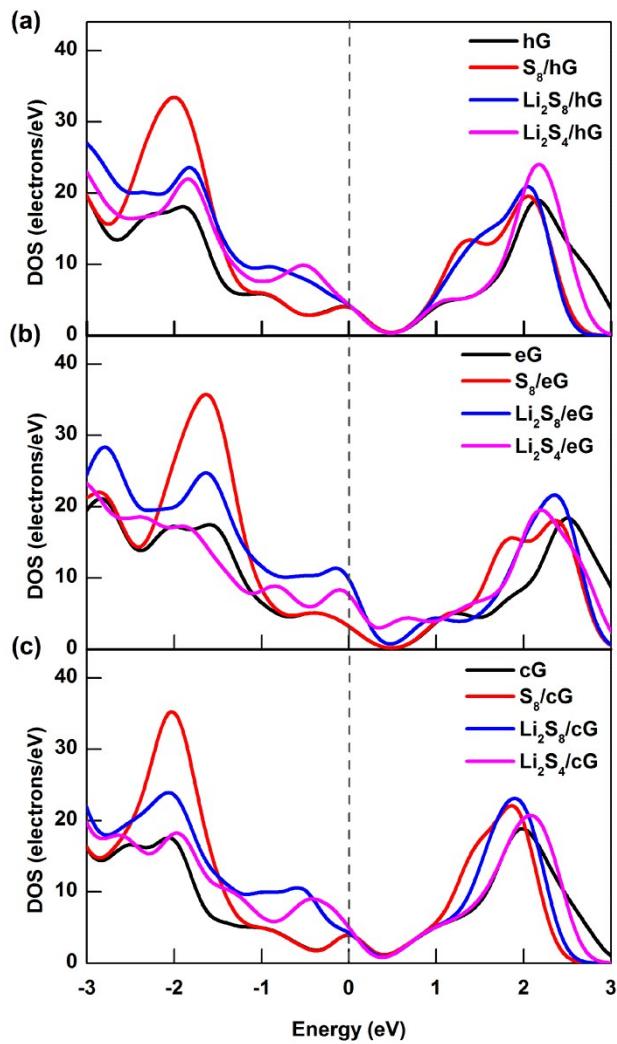


Fig. S4 Total density of states near the Fermi energy region for S8, Li₂S₈ and Li₂S₄ on a) hG, b) eG and c) cG

Table S1 Energy components related to adsorption energies of S₈, Li₂S₈ and Li₂S₄ to microporous graphene

| Pore Size (Å) | E_{graphene} (eV) | E_S where S = S ₈ (eV) | E_{Total} (eV) | E_{ads} (eV) |
|---------------|----------------------------|---|-------------------------|-----------------------|
| 7.5 | -74621.756 | -86654.006 | -161276.987 | -1.22 |
| 10 | -74621.601 | -86654.006 | -161276.332 | -0.71 |
| 12.5 | -74621.595 | -86654.006 | -161276.244 | -0.64 |
| 15 | -74621.584 | -86654.006 | -161276.210 | -0.62 |
| 17.5 | -74621.525 | -86654.006 | -161276.126 | -0.60 |
| 20 | -74621.516 | -86654.006 | -161276.118 | -0.60 |
| Pore Size (Å) | E_{graphene} (eV) | E_S where S = Li ₂ S ₈ (eV) | E_{Total} (eV) | E_{ads} (eV) |
| 7.5 | -74621.756 | -87066.246 | -161689.556 | -1.55 |
| 10 | -74621.601 | -87066.246 | -161688.871 | -1.02 |
| 12.5 | -74621.595 | -87066.246 | -161688.825 | -0.98 |
| 15 | -74621.584 | -87066.246 | -161688.726 | -0.90 |
| 17.5 | -74621.525 | -87066.246 | -161688.667 | -0.90 |
| 20 | -74621.516 | -87066.246 | -161688.496 | -0.73 |

| Pore Size (Å) | E_{graphene} (eV) | E_S where S = Li_2S_8 (eV) | E_{Total} (eV) | E_{ads} (eV) |
|---------------|----------------------------|--|-------------------------|-----------------------|
| 7.5 | -74621.756 | -87066.246 | -161689.556 | -1.55 |
| 10 | -74621.601 | -87066.246 | -161688.871 | -1.02 |
| 12.5 | -74621.595 | -87066.246 | -161688.825 | -0.98 |
| 15 | -74621.584 | -87066.246 | -161688.726 | -0.90 |
| 17.5 | -74621.525 | -87066.246 | -161688.667 | -0.90 |
| 20 | -74621.516 | -87066.246 | -161688.496 | -0.73 |

Table S2 Summary of adsorption energies between different substrates and S species

| Substrate | Adsorption Energy (eV) | | |
|-----------|------------------------|-------------------------|-------------------------|
| | S_8 | Li_2S_8 | Li_2S_4 |
| Graphene | -0.64 | -0.93 | -0.74 |
| hG | -0.63 | -1.23 | -1.43 |
| eG | -0.63 | -1.08 | -1.07 |
| cG | -0.65 | -1.29 | -0.95 |

Table S3 Total DOS value at E_f of Sulfur species in different substrates

| Species | Total DOS value at E_f (electrons/eV) |
|---|--|
| $S_8/\text{Graphene}$ | 4.32665 |
| $\text{Li}_2\text{S}_8/\text{Graphene}$ | 9.33993 |
| $\text{Li}_2\text{S}_4/\text{Graphene}$ | 6.48636 |
| hG | 3.98315 |
| S_8/hG | 3.99742 |
| $\text{Li}_2\text{S}_8/\text{hG}$ | 4.23578 |
| $\text{Li}_2\text{S}_4/\text{hG}$ | 4.30917 |
| eG | 3.15841 |
| S_8/eG | 3.14733 |
| $\text{Li}_2\text{S}_8/\text{eG}$ | 9.82511 |
| $\text{Li}_2\text{S}_4/\text{eG}$ | 7.54303 |
| cG | 3.91877 |
| S_8/cG | 3.91533 |
| $\text{Li}_2\text{S}_8/\text{cG}$ | 4.24859 |
| $\text{Li}_2\text{S}_4/\text{cG}$ | 4.93481 |

Table S4 Bond lengths after adsorption

| | Li2S8-hG | Li2S8-eG | Li2S8-cG | Li2S4-hG | Li2S4-eG | Li2S4-cG |
|-------------------|----------|----------|----------|----------|----------|----------|
| Li-O distance (Å) | 1.869 | 1.958 | 1.988 | 1.993 | 2.010 | 1.979 |
| Li-S distance | 2.476 | 2.477 | 2.476 | 2.463 | 2.449 | 2.435 |