

Supplementary Information

Flower-like ZnO-Ag₂O nanocomposite for label and mediator free direct sensing of dinitrotoluene

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Contents:

1. Optimization of amount of coating for fabrication of ZnO-Ag₂O composite nanoflowers modified gold electrode.
2. Estimation of the electrode surface area

1. Optimization of amount of coating for fabrication of ZnO-Ag₂O composite nanoflowers modified gold electrode

To optimize the amount of coating used for the fabrication of the modified gold electrode, DPV measurements in PBS with 20 μ M 2,4 DNT were performed using the electrode coated with different amount (1-7 μ L) of the nanoparticle slurry as shown in **Fig. 1S**. It was observed that when the coating amount was 5 μ L, maximum current response was obtained. Therefore, 5 μ L amount was fixed for fabrication of the modified electrode for further studies.

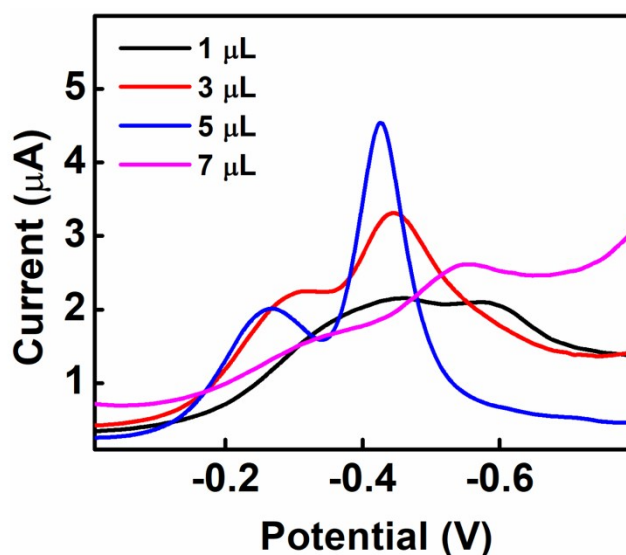


Figure 1S. DPV for 2,4-DNT (20 μ M in pH 7, 0.1 mM PBS) with different amount of coating (1 μ L-7 μ L).

2. Estimation of the electrode surface area

For estimation of the electrode surface area, scan rate studies were performed with solution of 0.1 mM potassium ferricyanide in 0.1 M KCl, and cyclic voltammograms were recorded for different scan rates (20-150 mVs⁻¹) using the modified electrode. A plot of peak current (I_p ; in Amperes) vs. square root of scan rate ($v^{1/2}$; in (V s⁻¹)^{1/2}) was plotted for the anodic peak current values as shown in **Fig. 2S** (a-b), and the value of the slope obtained was used in Randles-Sevcik equation at 25°C (S.1). The electrode surface area calculated was 0.033 cm².

$$I_p = 2.69 \cdot 10^5 n^{3/2} A D^{1/2} C v^{1/2} \quad (\text{S.1})$$

where, I_p = peak current value in Amperes, n = number of electrons involved in the electrochemical process (1 for electrochemical oxidation/reduction of pot. ferricyanide), A = electrode area in cm², D = diffusion coefficient in cm² s⁻¹, C = concentration in mol cm⁻³, v = scan rate in V s⁻¹.

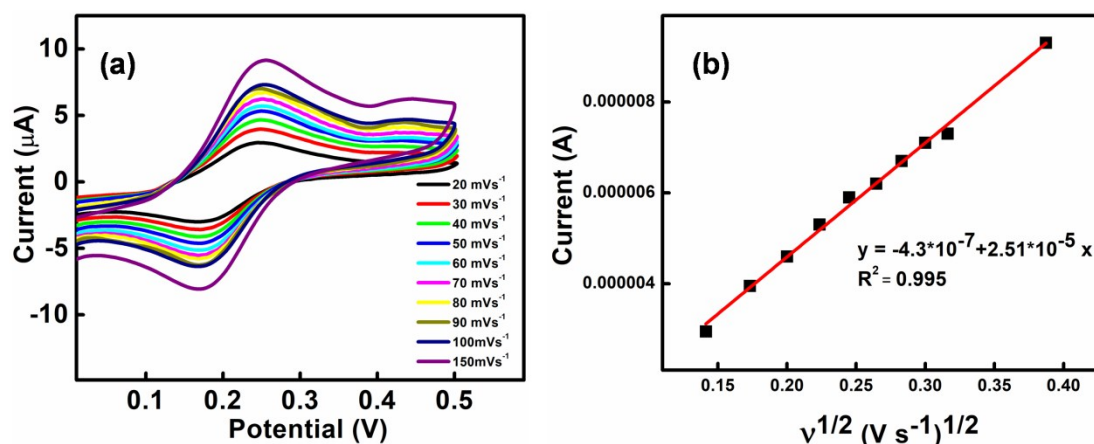


Figure 2S (a) CV for different scan rates (20-150 mVs⁻¹) using the modified electrode in solution of 0.1 mM potassium ferricyanide in 0.1 M KCl (b) plot of peak current (I_p ; in Amperes) vs. square root of scan rate ($v^{1/2}$; in (V s⁻¹)^{1/2})