

Physico chemical studies of resveratrol, methyl-jasmonate and cyclodextrins interactions: an approach to resveratrol bioproduction optimization

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Supplementary Information

Figure S1: ESI (+) MS Spectra corresponding to RSV with maltoheptaose (a) and RSV with β -CD (b) in the same experimental conditions.

Figure S2: dedicated Job plots of (a) β -CD (up)/resveratrol (down) and (b) DIMEB (up)/resveratrol (down)

Figure S3: Plot of the relative intensity of each complex between CD and MeJA versus cone voltage.

Figure S4: Fit between experimental RSV solubility data (β -CD ■, and RAMEB ×) and theoretical values predicted by formation constants obtained by ITC (red line and blue line, respectively).

Figure S5: Fit between experimental MEJA solubility data (β -CD ■, and RAMEB ×) and theoretical values predicted by formation constants obtained by ITC (red line and blue line, respectively).

Table SI : Association constant values, K (M^{-1}) calculated from solubility data

Figure S6: Enlarged contour plot of a T-ROESY experiment (spin lock: 300 ms, 22 dB) performed at 600 MHz on a β -CD/RSV/MeJA mixture (4.6 mM of β -CD with RSV and MeJA at saturation in D_2O). Rectangles represent the absence of a correlation between RSV and MeJA and ovals the presence of correlations between β -CD/RSV on one hand and β -CD/MeJA on the other hand

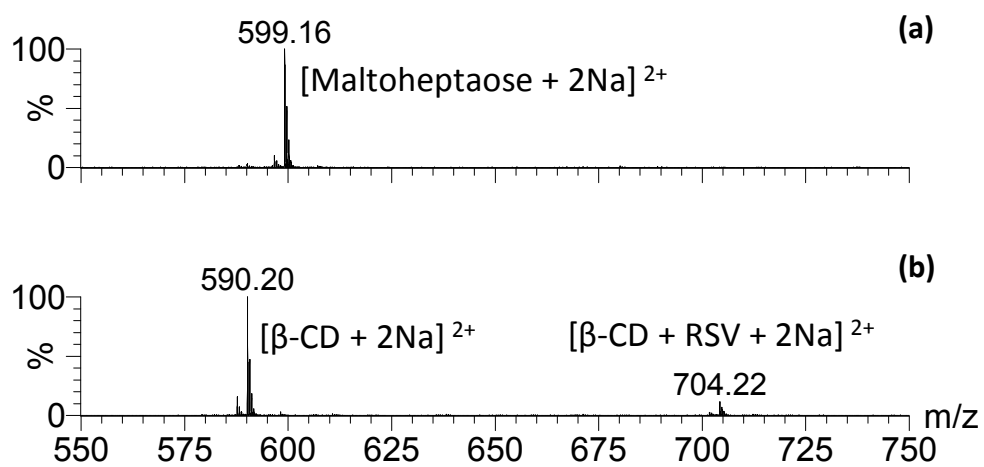
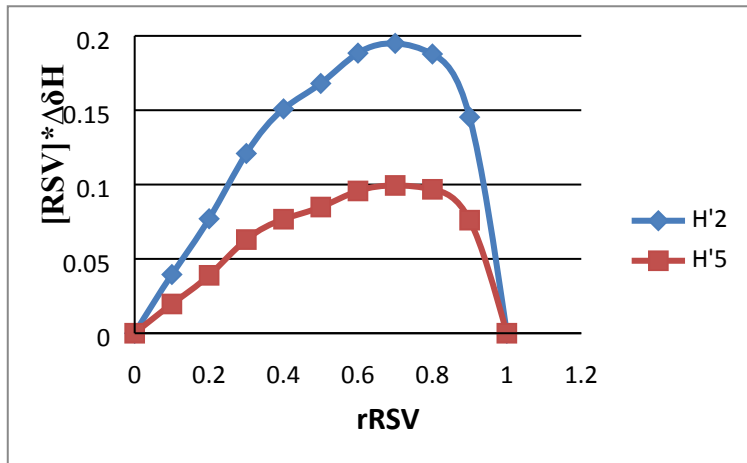
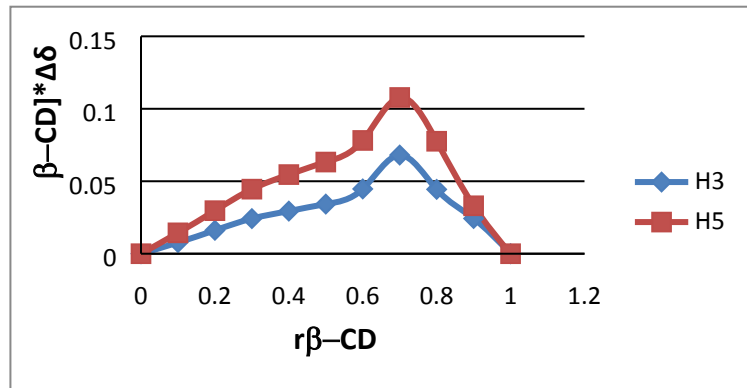


Figure S1: ESI (+) MS Spectra corresponding to RSV with maltoheptaose (a) and RSV with β -CD (b) in the same experimental conditions.

a



b

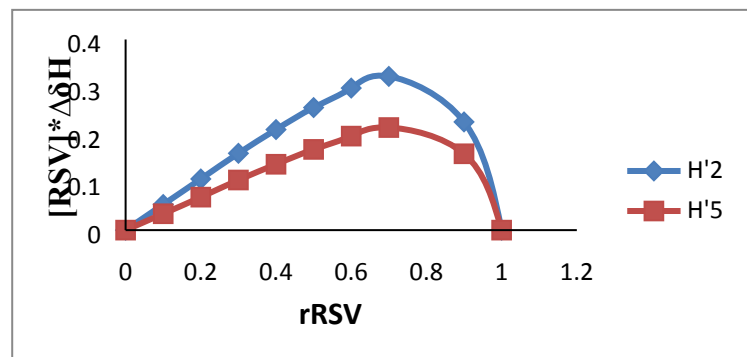
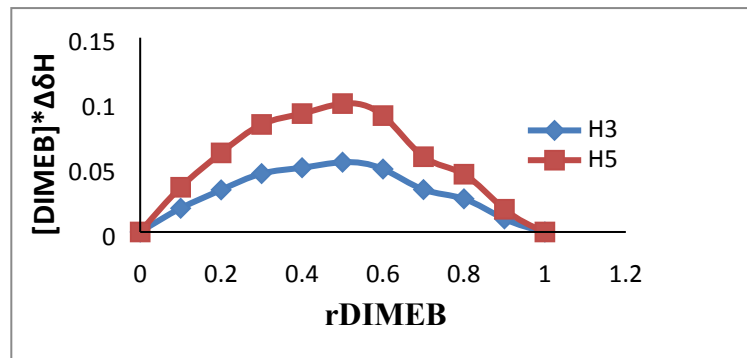


Figure S2: dedicated Job plots of (a) β -CD (up)/resveratrol (down) and (b) DIMEB (up)/resveratrol (down)

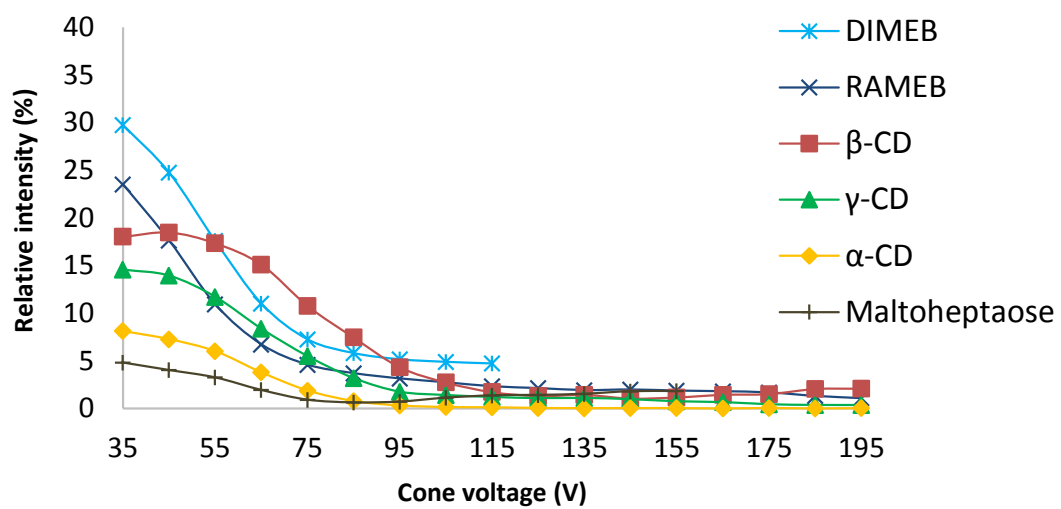


Figure S3: Plot of the relative intensity of each complex between CD and MeJA versus cone voltage.

The solubility of a guest G in the presence of CD (leading to a CD-G complex with a formation constant K), can be calculated as follows:

$$[G]_T = [G] + [CD-G] \quad (1)$$

$$[CD]_T = [CD] + [CD-G] \quad (2)$$

$$K = [CD-G] / ([CD] * [G]) \quad (3)$$

Combining equation (3) with equation (2):

$$[CD-G] = K * ([CD]_T - [CD-G]) * [G] \quad (4)$$

In phase solubility experiments, [G] is equal to the intrinsic G solubility S_0 , thus:

$$[CD-G] = (K * [CD]_T * S_0) / (1 + K * S_0) \quad (5)$$

As a result, the total guest concentration (ie. its solubility in the presence of CD) is equal to:

$$[G]_T = [G] + [CD-G] = S_0 + (K * [CD]_T * S_0) / (1 + K * S_0) \quad (6)$$

Accordingly, formation constants obtained by ITC were used to simulate the RSV and MEJA solubility in the presence of β -CD and RAMEB. Least square fitting between experimental and theoretical data were then realized (Figure S4 and S5, for RSV and MEJA, respectively), with a floating value for the intrinsic guest solubility.

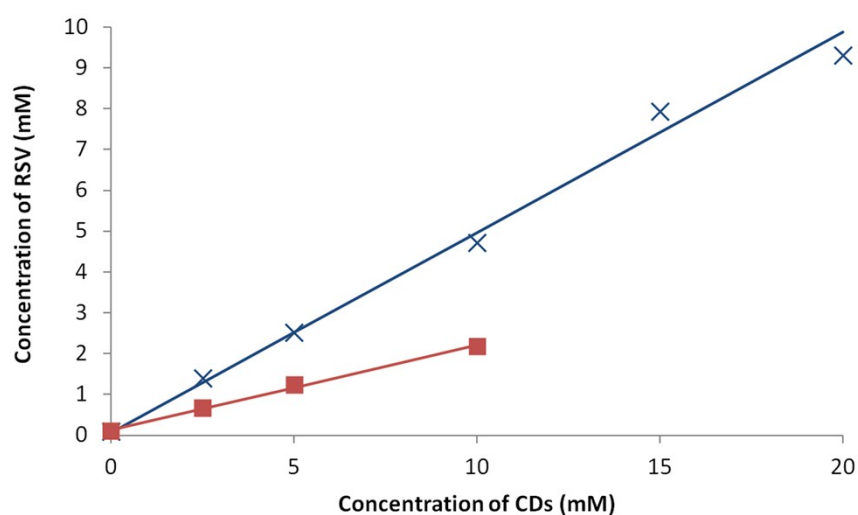


Figure S4: Fit between experimental RSV solubility data (β -CD ■ and RAMEB ×) and theoretical values predicted by formation constants obtained by ITC (red line and blue line, respectively).

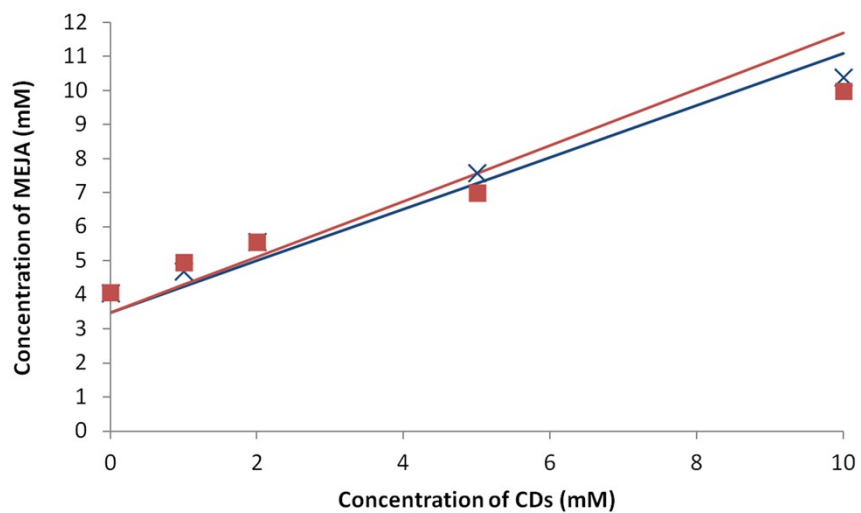


Figure S5: Fit between experimental MeJA solubility data (β -CD ■ and RAMEB ×) and theoretical values predicted by formation constants obtained by ITC (red line and blue line, respectively).

	RSV	MeJA
β -CD	3 053	481
RAMEB	10 700	570

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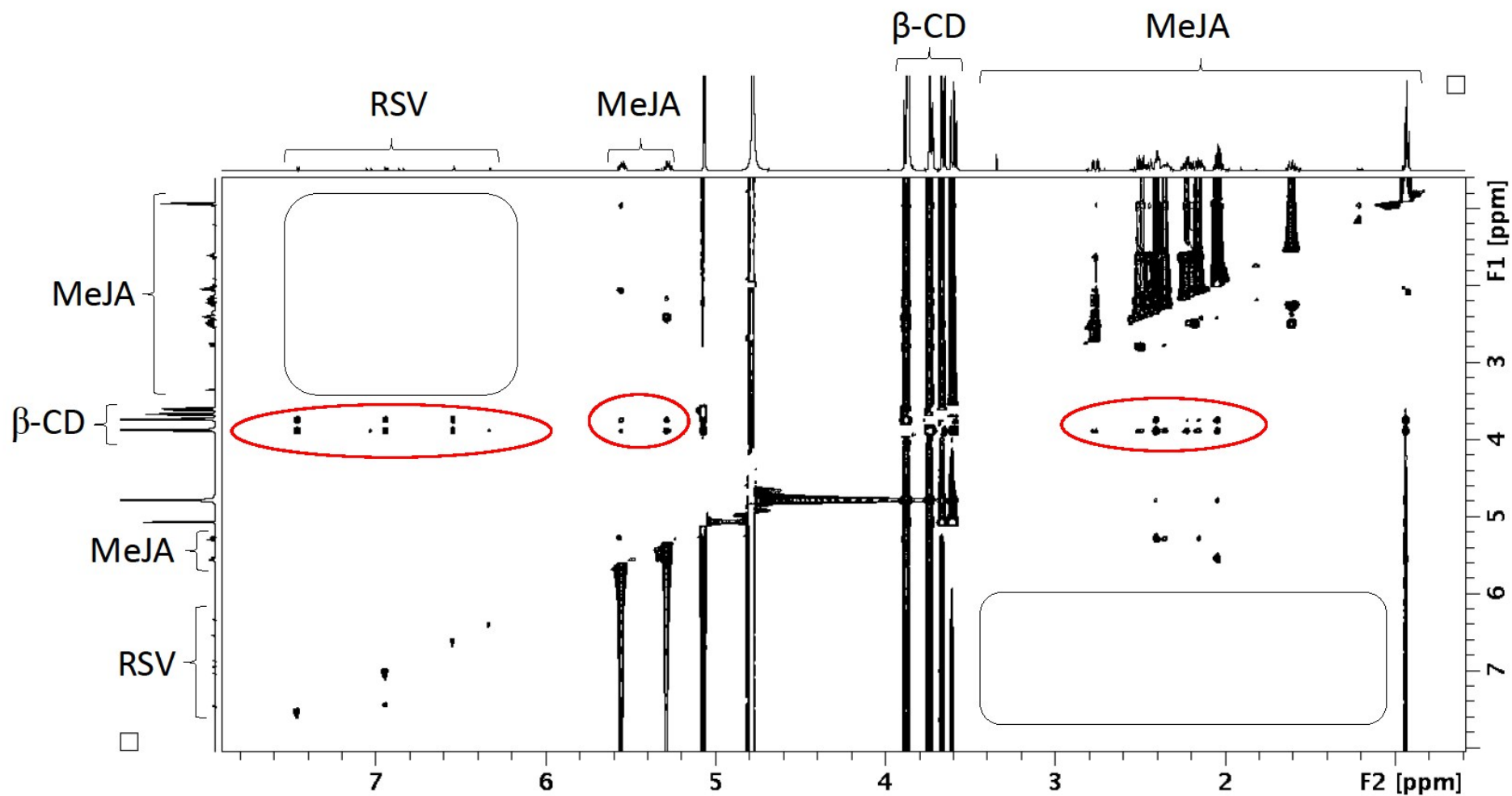


Figure S6: Enlarged contour plot of a T-ROESY experiment (spin lock: 300 ms, 22 dB) performed at 600 MHz on a β -CD/RSV/MeJA mixture (4.6 mM of β -CD with RSV and MeJA at saturation in D_2O). Rectangles represent the absence of a correlation between RSV and MeJA and ovals the presence of correlations between β -CD/ RSV on one hand and β -CD/MeJA on the other hand