

## Supporting Information

for Adv. Sci., DOI 10.1002/advs.202201586

Harvesting Water-Evaporation-Induced Electricity Based on Liquid–Solid Triboelectric Nanogenerator

Jingu Chi, Chaoran Liu\*, Lufeng Che, Dujuan Li, Kai Fan, Qing Li, Weihuang Yang, Linxi Dong\*, Gaofeng Wang\* and Zhong Lin Wang\*

## Supporting Information

# Harvesting Water-Evaporation Induced Electricity Based on Liquid-Solid Triboelectric Nanogenerator

Jingu Chi, Chaoran Liu<sup>\*</sup>, Lufeng Che, Dujuan Li, Kai Fan, Qing Li, Weihuang Yang, Linxi Dong<sup>\*</sup>, Gaofeng Wang<sup>\*</sup>, Zhong Lin Wang<sup>\*</sup>

## 1. The streaming current and potential mechanism

To further illuminate the energy harvesting mechanism, we assume the porous  $Al_2O_3$  sheet consisting of multiple vertical parallel nanochannels (capillary tubes). First, the water flow climbs upwards to a certain height inside the nanochannel driven by the capillary force. Then, the height of the water flow declines, owing to the evaporation of water on the sheet surface. Consequently, a pressure difference  $\Delta P$  occurs between the two ends of fluid flow inside the nanochannel. The pressure difference will drive the flow climb upwards again. Therefore, sustained evaporation will lead to continuous upward water flow. In another word, the water flow rate depends on the evaporation rate. The water evaporates mainly on the surface of LS-TENG sheet, which supplied by the water flow in the surficial vicinity nanochannels. Thus, we focus on the surficial nanochannels flow. The surficial nanochannels water flow forms an EDL at the liquid-solid interface. For a single nanochannel filled with water, the net positive charges transport in upward water flow forms a streaming current  $I_{ss}$ . Meanwhile, the induced electric field created by the resulting polarization of charge distribution along the flowing axis leads to a streaming potential  $V_{ss}$ .<sup>[1,2]</sup> The  $I_{ss}$  and  $V_{ss}$  can be given as<sup>[3,4]</sup>

$$I_{\rm ss} = \frac{\varepsilon_0 \varepsilon_{\rm r} \zeta A}{\eta l} \Delta P$$

$$V_{\rm ss} = \frac{\varepsilon_0 \varepsilon_{\rm r} \zeta}{\sigma \eta} \Delta P$$
(S1)

where  $\zeta$  is the zeta potential,  $\Delta P$  is the water pressure difference between two ends of a nanochannel.  $\varepsilon_{\rm r}$ ,  $\eta$ ,  $\sigma$ , l and A are water relative permittivity, viscosity, conductivity, nanochannel length and cross sectional area, respectively. According to Hagen–Poiseuille equation, the volume flow rate inside the single nanochannel can be expressed as  $Q_{\rm ss} = A^2 \Delta P / 8\pi \eta l$ .<sup>[4]</sup> The total streaming current  $I_{\rm st}$ , potential  $V_{\rm st}$  and flow rate  $Q_{\rm st}$  of the LS-TENG can be considered as parallel connection of all the surficial vicinity nanochannels.

## WILEY-VCH

Therefore,  $I_{st} = nI_{ss}$ ,  $V_{st} = V_{ss}$  and  $Q_{st} = nQ_{ss}$ , where *n* is the amount of surficial nanochannels. Then, we can express  $I_{st}$  and  $V_{st}$  as follow:

$$I_{\rm st} = \frac{8\pi\varepsilon_0\varepsilon_{\rm r}\zeta}{A}Q_{\rm st}$$

$$V_{\rm st} = \frac{8\pi\varepsilon_0\varepsilon_{\rm r}\zeta l}{\sigma A^2}\frac{Q_{\rm st}}{n}$$
(S2)

## 2. Measuring the LS-TENG sealed in a sink

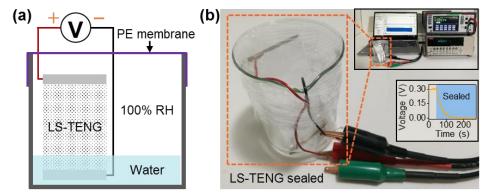


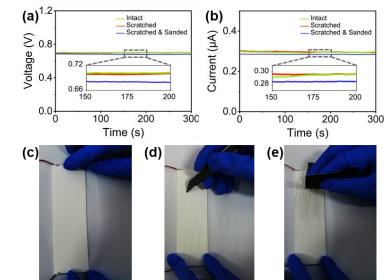
Figure S1. Measuring the LS-TENG sealed in a sink. (a) Schematic and (b) experimental photograph of measuring the LS-TENG sealed.

#### (a)<sub>0.6</sub> 0.3 40mm 10mm 30mm 20mm 40mm 30mm 20mm 10mm 0.2 Voltage (V) Current (µA) 0.4 0.1 0.2 0.0 -0.1 0.0 Ó 100 200 300 100 200 300 Ò Time (s) Time (s) (b)<sub>0.6</sub> 0.3 85% 75% 65% 85% 75% 65% 55% 45% 55% 45% 0.2 Voltage (V) Current (µA) 0.4 0.1 0.2 0.0 -0.1 0.0 ō 100 200 300 100 200 300 ō Time (s) Time (s) (c)<sub>0.6</sub> 0.3 pH 7 — pH 6 pH 4 — pH 3 pH 5 pH 7 \_\_\_\_ pH 4 \_\_\_ рН 6 рН 3 pH 5 0.2 Current (µA) Voltage (V) 0.4 0.1 0.2 0.0 0.0 -0.1 100 200 300 100 200 300 Ò Ó Time (s) Time (s)

#### 3. The real-time output voltage and current

Figure S2. The real-time voltage  $V_{oc}$  and current  $I_{sc}$  under different conditions such as (a) width, (b) relative humidity and (c) pH.

## WILEY-VCH



## 4. The robustness and mechanical strength of the LS-TENG

Figure S3. The robustness and mechanical strength test experiments. (a) Output voltage. (b) Output current. (c)

Intact LS-TENG. (d) Scratched LS- TENG. (e) Scratched and Sanded LS- TENG.

#### 5. Rising process of output voltage and current of LS-TENG

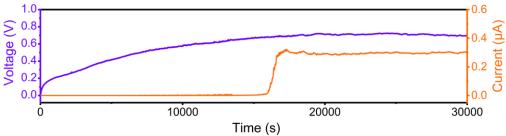


Figure S4. The rising process of output voltage and current of LS-TENG.

## 6. Absorption water rate of LS-TENG

We have supplemented the experiment to characterize the water delivery efficiency of commercial porous  $Al_2O_3$  ceramic sheet by the weighing method. The absorption water process of the  $Al_2O_3$  sheet is shown in Figure S5. Firstly, we weight the dry  $Al_2O_3$  sheet with the size of  $130 \times 40 \times 1$  mm3. The bottom electrode is immersed into water in an unsealed sink. Then, the immersed  $Al_2O_3$  sheet will be weighted every ~ 30 minutes. The experiment was carried out at the room temperature ~ 11.4 °C and relative humidity ~ 55.8 %. Also, we can obtain the average absorption rate as 6.57 mg/min.

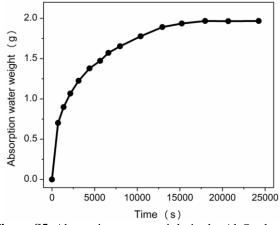


Figure S5. Absorption water weight in the Al<sub>2</sub>O<sub>3</sub> sheet.

## WILEY-VCH

## Video legends

Video S1. Measuring the output voltage of a LS-TENG

Video S2. Charging a capacitor with ten LS-TENGs connected in series and parallel

Video S3. Powering a calculator with ten LS-TENGs connected in series and parallel

## References

- [1] R. Zhang, S. Wang, M.H. Yeh, C. Pan, L. Lin, R. Yu, Y. Zhang, L. Zheng, Z. Jiao, Z.L.
   Wang, Adv. Mater. 2015, 27, 6482.
- [2] J. Yang, F.Z. Lu, L.W. Kostiuk, D.Y. Kwok, J. Micromech. Microeng. 2003, 13, 963.
- [3] W. Olthuis, B. Schippers, J. Eijkel, A. van den Berg, Sens. Actuators, B 2005, 111, 385.
- [4] X.B. Zhou, W.L. Zhang, C.L. Zhang, Y. Tan, J.C. Guo, Z.N. Sun, X. Deng, ACS Appl. Mater. Interfaces 2020, 12, 11232.