Development of Force Field Parameters for the Simulation of Single- and Double-Stranded DNA Molecules and DNA-Protein Complexes

Supporting Information

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Table S1. Simulations of short DNA oligomers and comparison to NMR J-couplings. Jcouplings were calculated for each system and each temperature using the Karplus relation from Hasnoot et al.^{11,12} and compared to the experimentally determined J-couplings.^{1-3,5-10} DES-Amber T is the variant of the DES-Amber force field where all of the negative charge of the 3' terminal residues is localized on this residue, rather than split between the 5'- and 3'-terminal residues as is customary in the Amber force fields. For DES-Amber 3.20 RMS errors are reported before and after (italics) the refit of backbone torsion parameters.

Table S2. Simulations of dsDNA reversible duplex formation. For each sequence and each force field, the number of melting events and the calculated thermodynamic stability (kcal mol-1) at 310 K and 1 M oligomer concentration are reported.

Table S3. Simulations of dsDNA reversible duplex formation. For each sequence and each force field the fraction of structures where the two DNA strands are non-interacting %Diss (defined as frames where there are no contacts between the strands) and the average number of contacts $\langle \#C \rangle$ (defined as two heavy atoms with a distance $\langle 5 \text{ Å} \rangle$ between DNA strands in non B-DNA structures is reported.

Table S4. Long DNA oligomers simulations. The average heavy atoms RMSD (Å) with respect to the starting PDB structure was calculated for the last 49 µs of each 50 µs simulation. For the Drew-Dickerson dodecamer, three simulations were performed starting from the X-ray (1bna) or NMR (1d89, 1fxz) structures. RMSD data are also reported for the Amber-bsc1 simulations performed by other authors and deposited in the BIGNASim database.¹³ In the DES-Amber SF1.0 simulation of 2lwg the double helix is unstable and the two strands dissociate, as indicated by the high value of the RMSD of 18.32 Å. This data point was not used to compute the RMSD average over all simulations for the DES-Amber SF1.0 force field. For the DES-

Amber 3.20 force field 10 µs of simulation was performed and the last 9.9 µs was used for data analysis.

Table S5. Deviations from experimental J-couplings for simulations of ssRNA

tetranucleotides. RMS error from the experimental data (Hz) for J-couplings calculated for each system using the Karplus relation from refs. 12, 14, and 15. Experimental data was taken from refs. 16, 17, and 18. For each value of σ_{OW} tested, deviations from experiment are reported before and after the refit. For $\sigma_{\text{OW}} = 3.20$ we also report scalar coupling deviations for a torsion optimization where NOEs and nNOEs data was also used (3.20 NOE).

$\sigma_{OW} =$	3.18	3.20	3.23	3.25
$\pmb{\alpha}$	0.39	0.40	0.65	0.97
β	0.09	0.10	0.36	0.29
γ	0.11	0.19	0.32	0.34
δ	0.04	0.08	0.10	0.24
$\pmb{\varepsilon}$	0.62	0.56	0.73	0.66
ζ	0.34	0.38	0.56	0.49

Table S6. Magnitude of torsion corrections from refitting based on experimental NMR Jcoupling data. Sum of coefficients of torsion corrections (kcal mol⁻¹) for each dihedral angle obtained from refitting based on NMR data.

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Table S7. Modified force field parameters for DES-Amber 3.20. List of parameters that were modified with respect to standard DES-Amber during the development of DES-Amber 3.20. Torsion corrections for the α , β , γ , δ , ε , and ζ torsion angles were obtained from refitting based on experimental NMR J-coupling data as $V(\varphi) = \sum_{n=1}^{3} a_n sin(\varphi) + b_n cos(\varphi)$ and transformed into the cosine expansion $V(\varphi) = \sum_{n=1}^{3} k_n \cos(n\varphi - \varphi_n)$ that is commonly used in the Amber force field (and those transformed values are reported here).

		Exp ^{19,20}	DES-Amber 3.20			
Sequence	[NaCl]	ΔG_{310}	$t(\mu s)$	#	ΔG_{310}	
CACAG	0 _M	-3.6	500	18	$-4.2(3)$	
CACAG	1 M	-4.7	500	12	$-4.1(3)$	
CGCGG	0 Ml	-6.2	500	21	$-6.8(3)$	
CGCGG	$1 \mathrm{M}$	-7.3	500	16	$-6.6(5)$	
UAAGGUA	0 _M	-5.4	180	5	$-5.2(7)$	
UAAGGUA	1 M	-6.95	150	7	$-5.7(5)$	

Table S8. Simulations of dsRNA reversible duplex formation. For each sequence, the total simulation time (μs) , number of melting events, and the calculated thermodynamic stability (kcal mol−1) at 310 K are reported and compared against polymer model estimates at two different salt concentrations. The free energies in the presence of counterions only (0 M) were estimated from the polymer model at 0.1 M salt concentration.¹⁹

TA/GC -0.02 0.49 -0.07 1.13			$\vert 0.08 \vert 0.23 \vert 0.25 \vert 3.94 \vert 2.79 \vert 5.59 \vert -1.47 \vert 6.95$			
TA/TA \vert -0.06 \vert 0.38 \vert -0.22 \vert 0.51 \vert 0.13 \vert 0.20 \vert -0.11 \vert 2.41 \vert 0.53 \vert 3.78 \vert -1.14 \vert 2.98						

Table S9. DES-Amber simulations of DNA duplexes. Base-pair and base-step parameters deviations from the reference X-ray structure. For each simulation, the average base-step and base-pair parameters were calculated. The RMSD deviation and mean signed deviation of the average MD values from the reference X-ray structure was computed for all base-airs and all base steps of each sequence. Here we report the average RMSD and MSD aggregated across all simulations by base-pair type and base-step type.

Table S10. DES-Amber 1.0 simulations of DNA duplexes. Base-pair and base-step

parameters deviations from the reference X-ray structure. For each simulation, the average base-step and base-pair parameters were calculated. The RMSD deviation and mean signed deviation of the average MD values from the reference X-ray structure was computed for all base-airs and all base steps of each sequence. Here we report the average RMSD and MSD aggregated across all simulations by base-pair type and base-step type.

Table S11. DES-Amber 3.20 simulations of DNA duplexes. Base-pair and base-step

parameters deviations from the reference X-ray structure. For each simulation, the average base-step and base-pair parameters were calculated. The RMSD deviation and mean signed deviation of the average MD values from the reference X-ray structure was computed for all base-airs and all base steps of each sequence. Here we report the average RMSD and MSD aggregated across all simulations by base-pair type and base-step type.

Table S12. Amber-bsc1 simulations of DNA duplexes. Base-pair and base-step parameters deviations from the reference X-ray structure. For each simulation, the average base-step and base-pair parameters were calculated. The RMSD deviation and mean signed deviation of the average MD values from the reference X-ray structure was computed for all base-airs and all base steps of each sequence. Here we report the average RMSD and MSD aggregated across all simulations by base-pair type and base-step type.

Table S13. Amber-OL15 simulations in TIP3P water of DNA duplexes. Base-pair and base-step parameters deviations from the reference X-ray structure. For each simulation, the average base-step and base-pair parameters were calculated. The RMSD deviation and mean signed deviation of the average MD values from the reference X-ray structure was computed for all base-airs and all base steps of each sequence. Here we report the average RMSD and MSD aggregated across all simulations by base-pair type and base-step type.

Table S14. Amber-OL15 simulations in OPC water of DNA duplexes. Base-pair and basestep parameters deviations from the reference X-ray structure. For each simulation, the average base-step and base-pair parameters were calculated. The RMSD deviation and mean signed deviation of the average MD values from the reference X-ray structure was computed for all base-airs and all base steps of each sequence. Here we report the average RMSD and MSD aggregated across all simulations by base-pair type and base-step type.

Table S15. Average helical Rise and Twist in simulations of DNA duplexes. For each simulation, the average Rise (Å) and Twist (degrees) parameters were calculated using DSSR. The X-ray values are reported for reference, with the corresponding PDB ID. Simulations performed with Amber-OL15 and the OPC water model gave very similar results as Amber-OL15, and simulations performed with DES-Amber 1.0 and DES-Amber 3.20 gave very similar results as DES-Amber, and are not reported in this table.

Figure S1. dT40 simulations. Last frame from the 100 µs simulations of dT40 in 0.4 M NaCl performed with the Amber-bsc1 (A) and CHARMM36 (B) force fields. (C) Number of basepairs formed as a function of time in the 1M NaCl simulations as calculated using DSSR.

Figure S2. Simulated tempering simulations of DNA duplex formation performed with the Amber-bsc1 force field. RMSD traces with respect to the ideal B-DNA structure as a function of simulation time for the 7 systems investigated. The oligonucleotide sequences are reported for each system.

Figure S3. Simulated tempering simulations of DNA duplex formation performed with the DES-Amber SF1.0 force field. RMSD traces with respect to the ideal B-DNA structure as a function of simulation time for the 7 systems investigated. The oligonucleotide sequences are reported for each system.

Figure S4. Simulated tempering simulations of DNA duplex formation performed with the DES-Amber force field. RMSD traces with respect to the ideal B-DNA structure as a function of simulation time for the 7 systems investigated. The oligonucleotide sequences are reported for each system.

Figure S5. Simulated tempering simulations of DNA duplex formation performed with the DES-Amber 3.20 force field. RMSD traces with respect to the ideal B-DNA structure as a function of simulation time for the 7 systems investigated. The oligonucleotide sequences are reported for each system.

Figure S6. Comparison of calculated and experimental^{21,22} osmotic coefficients for DMP, Purine, Cytidine and Uridine. Osmotic coefficients were calculated for solutions of Na⁺-DMP (circles), K⁺-DMP (squares), and Purine (crosses) using the Amber (black), DES-Amber (red), and DES-Amber 3.20 (blue) force fields. For solutions of Cytidine (diamonds) and Uridine (triangles), calculations were performed using the Amber (black) and DES-Amber (red) force fields only as, for these two systems, DES-Amber 3.20 is identical to DES-Amber.

Figure S7. Comparison of the disordered states of rU40 (A) and dT40 (B) simulated with different force fields with FRET data.²³ (A) The average FRET end-to-end distance ($\leq R$ >_{FRET}) was calculated at 0.05 M (black) and 0.1 M (red) NaCl concentration using simulations with values of the sigma parameter for the LJ interaction between water and the phosphate oxygen ranging from 3.18 to 3.25. Results from simulations performed using Amber-bsc1 in OPC water and DES-Amber are also shown. The experimental value is shown as a solid line. (B) FRET intensity for dT40 was calculated from simulations at NaCl concentrations ranging from 0.05 M to 1.0 M using the DES-Amber and the DES-Amber 3.20 force fields, compared to the experimental data. The Förster radius was assumed to be 55 Å for rU40 and 56.4 Å for dT40.²³

Figure S8. MD simulations of the d(CGCGCGCGCGCG) Z-DNA dodecamer at KCl concentrations ranging between 0.1 and 1.5M. For each concentration 200 µs of simulation were performed. Backbone RMSD traces with respect to the starting X-ray structure (PDB entry 4ocb) are reported for simulations performed using the DES-Amber force field. Simulations performed using DES-Amber SF1.0 gave similar results. For the calculation of the number of Z-DNA base-pairs, a dual cutoff methods was used where a base pair was considered to be in a Z-DNA conformation when the RMSD with respect to the starting structure, calculated on the

heavy atoms of the nucleotide pair, was < 1 Å, while the base pair was considered to be broken when the RMSD was $>$ 3 Å.

Figure S9. MD simulations of a d(CGCGCGCGCGCG) Z-DNA dodecamer performed with the DES-Amber 3.20 force field. Number of Z-DNA base pairs as a function of time during 50 µs MD simulations performed at KCl concentrations between 0.1 and 1.5 M.

Figure S10. MD simulations of three complex RNA folds. Backbone RMSD deviations from the starting X-ray structure as function of time for three 50 µs simulations of three complex RNA folds performed with the DES-Amber 3.20 force field: (A) the ribosomal L1 stalk from Thermus
Thermophilus in complex with the ribosomal protein tthl1 (PDB entry 3u4m;²⁴ complex RMSD is in black, protein RMSD in red and RNA RMSD in blue), (B) the Sarcin/Ricin domain from ecoli 23S rRNA (PDB entry 3dw4²⁵) and (C) an all-RNA "G8" hairpin ribozyme (PDB entry 2 oue 26).

Figure S11. MD simulations of six proteins containing structural zinc ions. For each $5 \mu s$ simulation, the protein RMSD (red) and the RMSD of the heavy atoms of the residues coordinating zinc (blue, violet and indigo) are reported with respect to the experimentally determined structures (from top to bottom: $1AAY$,²⁷ 1F57,²⁸ 1PZW,²⁹ 2AP1, 2EXF,³⁰ and 5W9S³¹). In cases where the protein structure has been determined in complex with DNA, the DNA RMSD (green) and the complex RMSD (black) are also reported. The coordination of zinc is CCCC for 1PZW and 5W9S, CCCH for 2AP1 and 2EXF, CHHE for 1F57, and CCHH for 1AAY.

Figure S12. MD simulations of tRNA. The position of the Mg2+ ions was equilibrated by performing 36 µs of simulated tempering with position restraints where the tempered parameter (the Mg2+ LJ σ), allowed to vary between 2.61 and 3.045 Å. The last structure of the equilibration was used to perform 20 µs of unrestrained simulation using the DES-Amber and DES-Amber 3.20 force fields. The structure was found to be remarkably stable during the simulation, as indicated by the low backbone RMSD. A comparison of the initial structure

(blue) to the average structure of the last microsecond of simulation (red) is also shown. The position of the Mg^{2+} ions in 50 snapshots of the simulation is shown as cyan balls.

Figure S13. MD simulations of a leucine zipper/DNA complex. (A) Backbone RMSD from the X-ray structure (PDB ID 2DGC)³² (B) The number of H-bonds and the number of saltbridges between the protein and DNA is reported for simulations performed with DES-Amber (blue and red) and DES-Amber 3.20 (cyan and magenta). (C) Superimposition of the average structure of the last 2.5 μ s of simulation (red -DNA, blue – protein) with the X-ray structure (gray)

Figure S14. MD simulations of the Trp-repressor/operator complex. (A) Backbone RMSD from the X-ray structure (PDB ID 1TRO).³³ (B) The number of H-bonds and the number of saltbridges between the protein and DNA is reported for simulations performed with DES-Amber (blue and red) and DES-Amber 3.20 (cyan and magenta). (C) Superimposition of the average structure of the last 2.5 μ s of simulation (red -DNA, blue – protein) with the X-ray structure (gray)

Figure S15. MD simulations of the P22 c2 repressor protein operator complex. (A) Backbone RMSD from the X-ray structure (PDB ID 3 JXC).³⁴ (B) The number of H-bonds and the number of salt-bridges between the protein and DNA is reported for simulations performed with DES-Amber (blue and red) and DES-Amber 3.20 (cyan and magenta). (C) Superimposition of the average structure of the last 2.5 μ s of simulation (red -DNA, blue – protein) with the X-ray structure (gray)

Figure S16. MD simulations of CI repressor in complex with the OL2 operator half-site. (A) Backbone RMSD from the X-ray structure (PDB ID 3ZHM).³⁵ The protein flexible tail residues were omitted from the RMSD calculation. (B) The number of H-bonds and the number of salt-bridges between the protein and DNA is reported for simulations performed with DES-Amber (blue and red) and DES-Amber 3.20 (cyan and magenta). (C) Superimposition of the average structure of the last 2.5 μ s of simulation (red -DNA, blue – protein) with the X-ray structure (gray)

Figure S17. MD simulations of ETV6 bound to DNA. (A) Backbone RMSD from the X-ray structure (PDB ID 4MHG).³⁶ (B) The number of H-bonds and the number of salt-bridges between the protein and DNA is reported for simulations performed with DES-Amber (blue and red) and DES-Amber 3.20 (cyan and magenta). (C) Superimposition of the average structure of the last 2.5 μ s of simulation (red -DNA, blue – protein) with the X-ray structure (gray)

Figure S18. MD simulations of the λ -repressor. (A) Backbone RMSD from the X-ray structure (PDB ID 3BDN).³⁷ The large dimerization domain was omitted from the RMSD calculation as it is loosely attached to the DNA binding domain and displays substantial hinge motion. (B) The number of H-bonds and the number of salt-bridges between the protein and DNA is reported for simulations performed with DES-Amber (blue and red) and DES-Amber 3.20 (cyan and magenta). (C) Superimposition of the average structure of the last $2.5 \mu s$ of simulation (red -DNA, blue – protein) with the X-ray structure (gray)

Figure S19. MD simulations of the zinc finger of human CXXC5 in complex with CpG DNA. (A) Backbone RMSD from the X-ray structure (PDB ID 5W9S).³⁸ The protein flexible tail residues were omitted from the RMSD calculation. (B) The number of H-bonds and the number of salt-bridges between the protein and DNA is reported for simulations performed with DES-Amber (blue and red) and DES-Amber 3.20 (cyan and magenta). (C) Superimposition of the average structure of the last 2.5 μ s of simulation (red -DNA, blue – protein) with the X-ray structure (gray).

Figure S20. MD simulations of the ZIF268 zinc finger-DNA complex. (A) Backbone RMSD from the X-ray structure (PDB ID $1AAY$).³⁹ The protein flexible tail residues were omitted from the RMSD calculation. (B) The number of H-bonds and the number of salt-bridges between the protein and DNA is reported for simulations performed with DES-Amber (blue and red) and DES-Amber 3.20 (cyan and magenta). (C) Superimposition of the average structure of the last 2.5 μ s of simulation (red -DNA, blue – protein) with the X-ray structure (gray)

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Figure S21. MD simulations of Z-DNA in 1 M NaCl. (A) Probability distribution for the backbone torsion angles of cytosine (C) and guanosine (G). The position of the backbone conformers (ZI, ZI', ZII, ZII'')⁴⁰ is indicated.

Figure S22. MD simulations of complex RNA folds. 50 μ s of simulation of a Sarcin/Ricin domain, of a Hairpin ribozyme and of the L1 stalk were performed with the DES-Amber 3.20 force-field. The ϵ RMSD score⁴¹ was calculated with respect to the X-ray structure. In the bottom panel, the average structure of the last $2.5 \mu s$ of simulation is superimposed to the starting X-ray structures (PDB IDs $3\text{DW4},^{25}\text{2OUE}^{26}$ and 3U4M^{24}).

Development of a generalized force field for the description of structural zinc ions compatible with DES-Amber

Structural zinc ions are a rather common feature of proteins that bind nucleic acids. To simulate these proteins and their interactions with DNA and RNA, we have created a set of parameters for structural zinc ions by adapting and applying to DES-Amber the EZAFF protocol developed by the group of Kenneth Merz⁴² to cover the very common cases of tetracoordinated zinc ions. In the EZAFF force field, a preliminary QM calculation is performed to compute the partial charges on the system as well as the corresponding bond angles and bond lengths. Such an approach, while highly accurate, is computationally expensive and impractical in the context of the DES-Amber workflow and makes calculation harder to reproduce across different platforms.

In zinc complexes, a substantial fraction of the positive charge of zinc is usually delocalized on the coordination sphere rather than localized on the zinc ion and in the original EZAFF approach the preliminary QM calculation is used to accurately quantify this redistribution of charge. To approximately reproduce this characteristic feature without resorting to QM calculations, we defined special residue types for the amino acids coordinating the zinc ions and assigned 0.3 units of charge of the zinc ion on each of these residues. To preserve the total charge of the system, we assigned a charge of 0.6 to tetracoordinated zinc ion (note that in DES-Amber charges are scaled by 0.9, so the total charge on zinc is $+1.8$ rather than $+2$). We also created parameters for an exacoordinated zinc ion of charge 0, but these were not extensively tested. Bond lengths were set to the average values observed for each atom type in high resolution structures taken from the Protein Data Bank⁴³ and bond angles were set to 109.5 degrees for tetracoordinated zing and 90 degrees for exacoordinated zinc, corresponding to a tetrahedral and octahedral geometry, respectively. Force constants for bonds and angles were taken from EZAFF; force-constants for dihedral angles were set to 0, as in EZAFF. Lennard-Jones

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parameters for zinc were also those of EZAFF. Files containing the full set of parameters are distributed with the Supporting Information.

We validated the force field by performing simulations of 5 μ s each for six proteins containing a total of 11 tetra coordinated structural zinc ions in different coordination spheres (namely CCCC, CCCH, CCHH and CHHE). In all simulations the zinc binding site is stable and its geometry is well maintained throughout the simulation, as evidenced by the RMSD of the zinc and surrounding residues heavy atoms being ≤ 1 Å for all simulations investigated (Figure S11). This result is comparable to the level of accuracy obtained for the EZAFF force field 42 and suggests that this approximate parametrization is adequate to perform stable simulations of proteins containing tetracoordinated structural zinc ions.

Parameterization strategy for non-standard residues

In some nucleic acid structures, nucleotides with a chemical structure slightly different the canonical one can be present due to changes in pH (protonation) or post-transcriptional modifications (methylation, oxidation, etc.). To develop parameters for these non-standard nucleotides, we started from the parameters for the most similar nucleotide in DES-Amber (the "parent" nucleotide) and followed the principle of introducing the smallest possible perturbation to these parameters. To this aim, we retain the bonded and non-bonded parameters for the atoms that are unchanged in the two nucleotides (the parent and the non-standard) and then assign GAFF bonded and non-bonded parameters^{44,45} for the non-standard components and fit torsion parameters to torsion scans at the MP2 level. Charges for the non-standard components and the atoms bound to them are assigned according to the GAFF protocol with the constraints of preserving the total charge of the nucleobase.

Optimization of nucleic acid backbone non-bonded parameters against experimental data

The study described in the main text is mostly focused on the optimization of *nucleobase* nonbonded parameters and torsion potentials. Other parameters, however, may also influence the accuracy of nucleic acid simulations. Backbone non-bonded parameters are arguably the most important parameters not considered so far that may influence the outcome of simulations. Here we describe our attempts at tuning some of the backbone non-bonded parameters, namely the phosphate-water interactions and the 2' hydroxyl-group charge distribution in RNA. We found that tuning of these parameters resulted in a force field that improved the description of some of the systems investigated; in particular, notable improvements were observed in the description of protein-nucleic acid complexes.

It is worth pointing out that we are here addressing rather small quantitative deviations from experimental measurements in the context of an otherwise fairly accurate force field. Although somewhat minor, it is possible that some of the discrepancies observed in the simpler test systems (for which it is possible to get quantitative results) may be significant enough to limit our ability to describe some of the more complex nucleic acid systems. This type of fine-tuning of backbone non-bonded parameters would be difficult to perform in the context of force fields like Amber-bsc1, as the parameters of that force field provide a much less accurate description of many of the simpler systems considered in this study, like Na-DMP, K-DMP, Uracyl, Thymine, and rU40.

Reparametrization of the 2' alcohol group charges.

The 2' hydroxyl group in RNA plays an important role in determining the local conformational preference of the RNA backbone.⁴⁶ In a previous publication,⁴⁷ we have shown that the charge distribution used for the hydroxyl groups in the Amber force fields results in a poor description of the properties of water-alcohol mixtures (as assessed by KBIs) and the wrong order of stability for donor and acceptor hydrogen bonds to the hydroxyl group compared to QM calculations. These deficiencies can be remediated by rebalancing the charges of the alcohol group. As we did previously for the Ser and Thr side chains in the context of the DES-Amber protein force field, we therefore changed the charge distribution on the 2' atoms by decreasing the charge of O2' by 0.07 (from -0.6139 to -0.6839) and of HO2' by 0.03 (from 0.4186 to 0.3886) and increasing the charge of C2' by 0.1 (from 0.0670 to 0.1670). These changes allow us to quantitatively reproduce QM hydrogen bonding energies and KBIs of alcohol/water mixtures.⁴⁷

Assessment of nucleic acid non-bonded parameters based on osmotic coefficients and aggregation propensities

The osmotic coefficient of Purine calculated using standard Amber and TIP3P water is in excellent agreement with experiment (RMS error 0.05) but it is slightly overestimated in DES-Amber and its variants (RMS error 0.14). We address this imbalance by optimization of the LJ interaction between the aromatic carbon atoms and water. We therefore changed the σ of this LJ interaction from 3.2325 Å to 3.2825 Å and ε from 0.12 kcal mol⁻¹ to 0.1 kcal mol⁻¹. Using these parameters, the experimentally measured osmotic coefficients could be reproduced within experimental error (RMS error of 0.01). Note that this change only affects Adenine.

Osmotic coefficient calculations for sodium dimethyl-phosphate (Na-DMP) and potassium dimethyl-phosphate (K-DMP) solutions obtained from calculations performed with the DES-Amber force field are in very good agreement with the experimental measurements²² (Figure S6) suggesting that phosphate LJ parameters developed by Steinbrecher and coworkers⁴⁸ and adopted in this force field are already close to optimal for these systems. Osmotic coefficient calculations for solutions of Uridine (rU) and Cytidine (rC) ribonucleotides result in osmotic coefficients that are slightly smaller than experiment²¹ (Figure S6). This finding could be explained by assuming that the sugar ring or the Uridine and Cytidine nucleosides may be too hydrophobic in DES-Amber. Based on this observation, if the force field were fully transferable from small to larger molecules, we would have expected that a simulation of poly-U or poly-C would result in a structural ensemble that is too compact. Simulations of rU40 and dT40, two polynucleotides that are mostly disordered in solution, however, result in disordered ensembles that are slightly more expanded that the FRET data suggests (Figure S7), a result that is at odd with the expectations from the osmotic coefficient of the individual components. These results suggest that parameters obtained from small molecule calculations, while providing a good starting point, may not be fully transferable and some degree of additional fine-tuning may be required when applying these parameters to larger, more complex systems.

In an attempt to fine-tune the hydration properties of the phosphate backbone in the context of a large RNA or DNA macromolecule, we added to the Steinbrecher-Case parameters⁴⁸ that describe the phosphate group an additional set of LJ parameters to specifically modulate the interaction between the phosphate O2 oxygen atoms and the water oxygen atoms. Similar NBfix corrections have been recently used by several authors to fine-tune the properties of charged groups and hydrogen bonds in the context of non-polarizable force-fields.^{49–55} In particular, we tested a number of increases in the sigma for the interaction (σ_{OW}) between the original value of 3.14 and 3.25 Å. This change introduces the smallest perturbation that would influence the

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hydration properties of this group without changing the backbone torsion potentials or other LJ interactions, which would further complicate the interpretation of the results. This set of changes results in osmotic coefficients for Na-DMP and K-DMP that are still in good agreement with experiment (Figure S6). Simulations of rU40 and dT40 are only weakly affected by these changes, yielding disordered ensembles that are slightly more compact than DES-Amber and in good agreement with the experimental data (Figure S7). These results indicated that small changes in the LJ interaction of the phosphate group still result in force fields that exhibit reasonable global hydration properties and we decided to perform further testing of this force field to further quantify the effect of these parameter changes on the structure or stability of small ssRNA molecules.

Influence of phosphate non-bonded parameters on the structural properties of small ssRNA oligomers

We performed simulation of the five tetranucleotides described in the main text, namely AAAA, UUUU, CAAG, CCCC, and CGAA using values for σ_{OW} of 3.18, 3.20, 3.23, and 3.25 Å and measured the deviation from NMR scalar couplings and NOEs experimentally measured for these systems. We find that even small changes of σ_{OW} result in deviations from the experiment due to the formation of substantial amounts of intercalated structures (previously observed in simulations performed with a number of Amber force fields^{17,56}) that are inconsistent with the experimental data, as shown by average error on the J-couplings that grows from 0.98 to > 1.3 Hz (Table S5). The amount of intercalated structures appears to increase with the increase of the value of σ_{OW} and the intercalated structures become dominant for values larger than 3.18 Å, a result consistent with previous observation of the importance of hydration for determining the structural properties of these tetranucleoties.⁵⁷

It has been shown that reweighting with experimental data can be an effective approach to refining protein and nucleic acid force fields.18,58,59,60 Here we used a curated dataset of Jcouplings18,60 to refine the RNA backbone torsion potentials by reweighting of the simulation ensemble.⁶⁰ Adding NOEs and not-observed NOEs (nNOEs) data to the fit only slightly affected the results (Table S5), as torsion potentials optimized only to J-couplings also reproduced well the NOE data. We find that, for each value of σ_{OW} between 3.18 and 3.25 Å, it is possible to obtain a force field that is in good agreement with the experimental data for the tetranucleotides (Average error on J-coupling < 0.9 Hz, Table S5); and, for some of these force fields, we verified that they can fold the UUCG and GGAC tetraloops to within 1 Å of the experimentally determined structure and also fold the CGCGG and CAGAG RNA duplexes to the correct Aform. In general, the resulting torsion correction appears to favor formation of the A-form of RNA with the largest changes observed in the α torsion term. In other words, there is a substantial degree of correlation in this dataset and care should be taken not to overfit to it. To help preventing overfitting, here we use an L2 regularization term restraining the coefficients of the fit to 0 with a force constant of 20 kcal mol⁻¹ \AA^{-1} . We find that the strength of the correction terms generally increases as the value of σ_{OW} increases (Table S6) and some torsion corrections can become as large as 1 kcal mol⁻¹ for $\sigma_{\text{OW}} = 3.25$ Å.

This result indicates that the conformation of small ssRNA molecules is strongly influenced by both torsions and non-bonded potentials and so force fields that satisfactorily describe such systems can be obtained using a range of different parameters. We decided to pick the force field obtained using a value for σ_{OW} of 3.20 Å for further, more extensive testing on RNA and DNA systems as this force field provides a description of the solution properties of DMP that is still in good agreement with the experimental data while improving the description of rU40 and dT40 and requiring rather minor corrections to the torsions potentials to accurately describe the structural properties of the RNA tetranucleotides. To make this variant compatible with Mg^{2+}

we used osmotic coefficient calculations of $MgCl₂$, ⁶¹ MgAce₂, ⁶² and MgDMP₂ to calibrate the $Mg²⁺-OW LJ σ, changing it from 3.004 to 2.90 Å. We name this DES-Amber variant DES-$ Amber 3.20.

Assessment of the accuracy of DES-Amber 3.20 for DNA simulations

Refinement of backbone torsions based on structural properties of small DNA oligomers

As a first step, we initially performed a refinement of the backbone torsion parameters based on a total of 713 experimental NMR scalar couplings that probe the structural properties of 18 DNA oligomers ranging from 2 to 6 nucleotides (Table S1). In stark contrast to the RNA results, the change of the σ_{OW} for the water-phosphate LJ interaction does not affect the scalar couplings of the oligonucleotides. The RMS error from the experimental J-coupling is 0.88 Hz for DES-Amber and 0.87 Hz for DES-Amber 3.20. As the agreement with the experiment is already good in the starting force field, torsion refitting only reduces the error to 0.76 Hz; only the δ torsion corrections are larger than 0.1 kcal mol⁻¹, while corrections of less than 0.06 kcal mol⁻¹ are applied to the other torsions (Table S7).

Thermodynamic stability of DNA double helix

We determined the thermodynamic stability of the DNA double helix for 7 sequences ranging from 4 to 7 nucleotides by performing ST simulations of 500 μ s for each system (Figure S5) and computing the fraction of helix at each temperature. The helix stability at 310 K for each system

is reported in Table S2. The RMS error in the free energy of helix formation at 310 K is 2.0 kcal mol⁻¹ for Amber-bsc1, 2.2 kcal mol⁻¹ for DES-Amber and 1.1 kcal mol⁻¹ for DES-Amber 3.20.

Thermodynamic stability of a DNA hairpin

We determined the thermodynamic stability of a DNA hairpin of sequence d(GCGAAGC) by running 295 µs of ST simulation and computing the fraction of helix at each temperature. The hairpin folds reversibly hundreds of times in simulation to within less than 1 Å RMSD from the experimentally-determined solution structure (PDB entry 1kr8).⁶³ The fraction of helix formed at 310 K is 13.5%, corresponding to a ΔG_{fold} at 310K of 1.1 kcal mol⁻¹ for DES-Amber 3.20, which is comparable to the values of 1.3 and 1.5 kcal mol⁻¹ obtained for DES-Amber and DES-Amber SF1.0, respectively.

Structural properties of DNA double helices

To verify that he force field is able to recapitulate the structure of double-stranded DNA, we performed simulations of 10 µs at 300 K for 17 DNA duplexes between 10 and 17 nucleotides long and compared the heavy-atom RMSD to the structures determined experimentally by X-ray or NMR. The heavy-atom RMSD deviations observed for DES-Amber 3.20 were not only comparable to those of DES-Amber but also to those of Amber-bsc1 (Table S4), a force field that is able to very accurately reproduce the structural features of double-stranded B-DNA.^{64,65}

Structural properties of non-canonical DNA structures

We performed simulations for a number of non-canonical DNA structures to verify that the force field can correctly recapitulate the stability of non-Watson-Crick base pairs. In particular, we performed 20 μ s simulations of two G-quadruplex structures^{66,67} and a triplex structure⁶⁸ as well as 50 μ s simulations of a Z-DNA helix⁶⁹ in salt concentrations ranging between 0.1 M and 1.5 M. The triplex and quadruplex structures were stable for the 20 µs of simulation (Figure S8). As observed for the DES-Amber force field, in simulations of the G-quadruplexes, the core quadruplex structure remained very close to the experimentally determined structure for the duration of the simulations, while the loops were more flexible. The Z-DNA structure is expected to be unstable in monovalent salt solution, although its stability is expected to increase at higher salt concentration. Within the first 10-20 µs, we observed fraying of the terminal base pairs in all salt conditions, but at the lowest salt concentration of 0.1 M [NaCl] the Z-DNA duplex melted almost completely, while core residues remained stable in the simulations run at higher salt concentration (Figure S9).

Assessment of the accuracy of DES-Amber 3.20 for RNA simulations.

We described previously the performance of this force field for rU40 and small ssRNA tetranucleotides. Not surprisingly, simulations of small ssRNA tetranucleotides are in excellent agreement with the NMR experimental data (Table S5) as the backbone torsion parameters of the force field were fitted to reproduce this data.

Thermodynamic stability of the RNA double-helix

We verified that this force field can describe the stability of A-RNA double-helical structures by performing ST simulations of three RNA A-helices of sequences CACAG, CGCGG, and UAAGGUA. The duplexes were simulated in 0 M (counterions only) and 1 M NaCl concentration. In each simulation, we observed between 5 and 21 helix formation events (defined as transitions between structures with greater than 10 Å RMSD and structures within 2 Å RMSD from the canonical A-helix structure), allowing us to estimate the stability of the duplexes with a statistical error of ~ 0.3 -0.7 kcal mol⁻¹ (Table S8). The RMS error in the free energies at 310 K with respect to the experimental data is 0.7 kcal mol⁻¹, indicating that the force field can reproduce the helix stability reasonably well, although the ionic strength dependence on stability might be weaker than what is observed in experiment.

Thermodynamic stability of RNA tetraloops

We verified that the force field is also able to recapitulate the structure of two RNA tetraloops, namely the ggcGCAAgcc and gccUUCGggc tetraloops. To this aim, we performed 500 µs of ST simulation for each of the tetraloops. In the two simulations, we observed multiple independent reversible folding events to structures that were within 1 Å RMSD from the experimentally determined structure. The calculated thermodynamic stability of the two tetraloops at 310 K is $0.0(3)$, $-0.5(3)$ kcal mol⁻¹ if only structures with the correct tetraloop configuration are considered as folded, or $-0.8(5)$ and $-1.1(4)$ kcal mol⁻¹ if all hairpin structures are considered folded. These values can be compared to the experimentally determined stabilities⁷⁰ of -3.4 and -5.44 kcal mol-¹ suggesting that, although the correct structure is recapitulated by the force field, the thermodynamic stability of the tetraloops is underestimated by $3-5$ kcal mol⁻¹; this

underestimation is a problem that is commonly observed for the Amber RNA force fields^{71,72} and may be related to the limited ability of point-charge force fields to properly describe hydrogen bonding.⁷¹

Structure and stability of complex RNA folds

Finally, we performed 50 µs simulations of three complex RNA folds, namely the ribosomial L1 stalk from Thermus Thermophilus in complex with the ribosomal protein tthl1 (PDB entry $3u4m²⁴$), the Sarcin/Ricin domain from Ecoli 23S rRNA (PDB entry $3dw4²⁵$), and an all-RNA "G8" hairpin ribozyme (PDB entry 2oue²⁶). The structural Mg^{2+} and Na⁺ ions and the nucleobase modifications present in the structures were retained during the simulations as they were found to be important for structural stability.

We find that in all three cases, although some fluctuations are observed, overall the RNA fold is stable on the simulation timescale (Figure S10), suggesting that the force field provides a reasonable representation of the overall structure of non-canonical RNA structures. We also calculated the eRMSD score for each structure (Figure S22) which has values of \sim 1, indicating that some rearrangement does occur at the level of the base-pairs. In both the hairpin ribozyme and the Sarcin/Ricin domain simulations fraying occurs at the terminal residues; furthermore, in the ribozyme simulation the relative orientation of the two duplexes can fluctuate, further contributing to the overall RMSD from the X-ray structure. The Sarcin/Ricin domain features a peculiar S-loop where a bulged Guanine forms a characteristic GpU dinucleotide platform. In the 50 µs of simulation, the bulged Guanine residue is found to flip away from the GpU platform and becomes solvent exposed. This motion induces only minor perturbations in the remainder of the structure that closely maintains the X-ray conformation (Figure S22).

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