

## Supporting Information

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Unravelling Alkali-Metal-Assisted Domain Distribution of Quasi-2D Perovskites for Cascade Energy Transfer toward Efficient Blue Light-Emitting Diodes

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## **Supporting Information**

**Table S1**. Summary of the tri-exponential fitting results for PL lifetime curves of Q-2D CsPbBr<sub>3</sub> films with various NaBr concentrations. Time-resolved PL lifetime curves were fitted by the tri-exponential function:

$$I = A_1 \exp\left(-\frac{t}{\tau_1}\right) + A_2 \exp(-\frac{t}{\tau_2}) + A_3 \exp(-\frac{t}{\tau_3})$$
(S1)

Where, I is the normalized photoluminescence intensity;  $A_1$ ,  $A_2$  and  $A_3$  are the fractions of the three decay components and  $A_1 + A_2 + A_3 = 1$ ;  $\tau_1$ ,  $\tau_2$  and  $\tau_3$  stand for the lifetime constants of a fast

component, a middle component and a slow component, respectively. The average lifetime ( $\tau_{ave}$ ) was calculated with the  $A_i$  and  $\tau_i$  (i= 1, 2, 3) values according to the following equation:

Perovskite Emitter	Emission [nm]	τ1 [ns]	A1 [%]	τ2 [ns]	A2 [%]	τ3 [ns]	A3 [%]	τ <sub>ave</sub> [ns]	χ²
0% NaBr	478	2.20	13.21	10.87	38.82	40.51	47.97	34.82	1.07
15% NaBr	485	3.27	30.21	12.01	46.02	55.01	23.77	40.27	1.10
30% NaBr	485	3.64	34.12	12.88	45.40	68.30	20.48	49.11	1.12
45% NaBr	487	4.16	26.82	16.49	46.94	75.97	26.24	57.20	1.15

$$\tau_{ave} = \frac{A_1\tau_1 + A_2\tau_2 + A_3\tau_3}{A_1 + A_2 + A_3}$$

(S2)



**Figure S1.** Trap density extraction by dark current-voltage measurement of the hole-only device, with the device structure of ITO/PEDOT/perovskite/MoO<sub>3</sub>/Ag. The light-blue lines represent the ohmic regime of each case, and the dark-blue lines indicate the trap-filled limit (TFL) regime with the onset voltage ( $V_{TFL}$ ). The trap density ( $N_t$ ) of Q-2D perovskite films with various ratios of NaBr can be extracted by the equation  $N_t = \frac{2\varepsilon_0\varepsilon_r VTFL}{qL2}$ , where  $\varepsilon_0$ , q and L represent vacuum permittivity, elementary charge and thickness of the perovskite film, respectively;  $\varepsilon_r$  is the average relative dielectric constant of Q-2D perovskite films with various ratios of NaBr.



**Figure S2.** Scanning electron microscope (SEM) images of Q-2D perovskite films with various ratios of NaBr.



**Figure S3**. Photoluminescence image of quasi-2D perovskite films with various ratios of NaBr under ultraviolet lamp excitation (365 nm).



**Figure S4**. XPS spectra of perovskite films without and with 30% NaBr: (a) Entire spectra, (b) Na 1s spectra, (c) Cs 3d spectra and (d) Pb 4f spectra.



**Figure S5.** (a) High-binding energy secondary-electron cut-off regions of Q-2D perovskites without and with 30% NaBr. (b) VB-edge regions. The VB is calculated using the equation of  $VB = 21.2 - (E_{cutoff} - E_{onset})$ . The calculated VBs were 6.0 and 5.8 for 0% NaBr and 30% NaBr-based perovskite films, respectively.



Figure S6. Electroluminescence (EL) spectra of perovskite LEDs.



**Figure S7.** Hole mobility measurement of Q-2D perovskites without and with 30% NaBr by SCLC method, with a device structure of ITO/PEDOT:PSS/perovskite/MoO<sub>3</sub>/Ag.

**Table S2.** Fitting parameters for the kinetics shown in **Figure 5e**. The kinetics are fitted by the multiple-exponential function,  $\Delta A(t) = a_1 \exp(-t/\tau_1) + a_2 \exp(-t/\tau_2) + a_3 \exp(-t/\tau_3) - c_1 \exp(-t/\tau_{et})$ , where  $a_1$ ,  $a_2$ ,  $a_3$  and  $c_1$  are the amplitudes;  $\tau_1$ ,  $\tau_2$  and  $\tau_3$  are the decay time constants and  $\tau_{et}$  is the electron transfer time constant.

	GSB [nm]	τ <sub>et</sub> [ps]	τ <sub>1</sub> [ps]	τ2 [ps]	τ3 [ps]	$\chi^2$
n=2	414		5.1	45.6	8190.7	0.99
n=3	440		20.3	387.1	100000	0.97
n≥4	475	30.4	290.5	20715.7		0.99

**Table S3.** Fitting parameters for the kinetics shown in **Figure 5f**. The kinetics are fitted by the multiple-exponential function,  $\Delta A(t) = a_1 \exp(-t/\tau_1) + a_2 \exp(-t/\tau_2) + a_3 \exp(-t/\tau_3) - c_1 \exp(-t/\tau_{et})$ , where  $a_1$ ,  $a_2$ ,  $a_3$  and  $c_1$  are the amplitudes;  $\tau_1$ ,  $\tau_2$  and  $\tau_3$  are the decay time constants and  $\tau_{et}$  is the electron transfer time constant.

	GSB [nm]	τ <sub>et</sub> [ps]	τ <sub>1</sub> [ps]	τ2 [ps]	τ <sub>3</sub> [ps]	$\chi^2$
n=2	418		4.8	43.0	8163.3	0.98
n=3	445		2.5	55.5	12707.5	0.99
n≥4	488	22.2	229.4	16319.5		0.99



**Figure S8**. Bleach recovery dynamics for pristine and 30% NaBr-incorporated perovskite films. Solid lines are the fits of the kinetics by exponential function.



Figure S9. Stability measurement of the PeLEDs with an initial luminance of 100 cd/m<sup>2</sup>.