Supporting Information

Dissecting the mechanism of the non-heme iron endoperoxidase FtmOx1 using substrate analogs

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Experimental Procedures

General

NMR spectra were acquired on an Agilent DD2 600 MHz spectrometer (600 MHz for ¹H-NMR and 150 MHz for ¹³C-NMR) (Santa Clara, CA, USA). NMR spectra were recorded in DMSO-d₆ (99.9 atom%) enriched, AICON) and CDCl₃ (99.5 atom% enriched, Kanto). ¹H-NMR chemical shifts were reported in δ value based on residual DMSO- d_6 (2.50 ppm) and CDCl₃ (7.26 ppm) as reference. ¹³C-NMR chemical shifts were reported in δ value based on DMSO- d_6 (39.52 ppm) and CDCl₃ (77.16 ppm) as a reference. Data are reported as follows: chemical shift, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet), coupling constant (Hz), and integration. LC-HR-MS(ESI) measurements were obtained on Thermo Q Exactive Orbitrap mass spectrometer coupling with a Shimadzu LC-20AD UPLC equipped with a Waters ACQUITY BEH C18 column (2.1 × 100 mm, 1.7 μm particles). The LC-MS analysis was run using an Agilent 1100 Series separation module with an Agilent G1956B MS detector. RP-HPLC was performed on an Agilent 1260 Series separation module with a diode array detector. HPLC analysis was carried out using a C18 column (ACE Excel 5 C18, 4.6 × 250 mm, 5 μm, 1 ml/min). Semipreparative HPLC was carried out using a C18 column (Titank, 10 × 250 mm, 5 µm, 4 ml/min). Protein-substrate complexes were first screened for crystallization using a Crystal Gryphon auto dispenser (Art Robbins Instruments, LLC), optimized crystals were sent to the Shanghai Synchrotron Radiation Facility (SSRF) to collect X-ray diffraction data under beamline BL02U1 and BL19U1 with a Pilatus3 6M image plate detector. Biological reagents, chemicals, and media were purchased from standard commercial sources unless stated otherwise.

Fermentation of *Aspergillus fumigatus* Af293 and purification of fumitremorgin B (1), verruculogen (2), and 12,13-dihydroxy-fumitremorgin C (4)

Strain *A. fumigatus* Af293 cells were cultured on PDA medium at 25°C for 3 days, and agar plugs (5-mm-diameter) were placed into 10 Erlenmeyer flasks (250 ml), each containing 100 ml of potato dextrose broth (PDB). The flasks were incubated at 28°C on a rotary shaker at 220 rpm for 3 days to generate the seed culture, from which 5 ml of culture was separately inoculated into 200 aseptic-bags, each containing 80 g of autoclaved rice and 120 ml distilled H₂O. These aseptic-bags were transferred to an incubator and fermented at 28°C for 30 days. Then, 16 kg of whole cultures were extracted with EtOAc (3×30 L) and concentrated under reduced pressure to yield a dark brown gum (61 g).

The crude extracts were fractionated by silica gel vacuum liquid chromatography (85×300 mm column packing with 300 g 300~400 mesh silica gel) using a CH₂Cl₂-MeOH gradient (100:0, 99:1, 98:2, 97:3, 96:4, and 0:100; each 2800 ml) to afford 6 fractions (G1–G6). Fraction G3 (10.5 g), dissolved in CH₂Cl₂ and MeOH (1:1), was applied to a Sephadex LH-20 column (45×1200 mm packing with 1600 ml resin) using MeOH as the mobile phase to yield 11 sub-fractions (G3N1–G3N11). LC-MS analysis of these fractions revealed the existence of the target compounds (fumitremorgin B and verruculogen) in fraction G3N3 based on characteristic UV absorption and

HR-ESI-MS spectra. Then, G3N3 (65.6 mg) was further purified by semi-preparative RP-HPLC using a C18 column (Titank, 10×250 mm, 5μ m) with a flow rate of 4.0 ml/min and a gradient elution (During 0–16 min, ACN was increased from 10% to 99% in 0.1% HCOOH-containing H₂O and then the column was eluted for 5 mins with 99% ACN in 0.1% HCOOH-containing H₂O) to obtain fumitremorgin B (**1**, 35.0 mg, t_R = 16.8 min), verruculogen (**2**, 7.0 mg, t_R = 15.5 min), and 12,13-dihydroxy-fumitremorgin C (**4**, 15 mg, t_R = 9.7 min).

Fumitremorgin B (1): white powder; ¹H- and ¹³C-NMR data, see Table S1; HR-MS (ESI+) detected at m/z 462.2369, *cald*. for $[M - H_2O + H]^+ C_{27}H_{32}N_3O_5$: m/z 462.2387.

Verruculogen (**2**): white powder; ¹H- and ¹³C-NMR data, see Table S2; HR-MS (ESI+) detected at m/z 494.2266, *cald*. for [M – H₂O + H]⁺ C₂₇H₃₃N₃O₇: m/z 494.2285.

12,13-dihydroxy-fumitremorgin C (**4**): pale yellow powder; ¹H- and ¹³C-NMR data, see Table S4; HR-MS (ESI+) detected at m/z 394.1757, *cald*. for [M – H₂O + H]⁺ C₂₂H₂₄N₃O₄: m/z 394.1761.

Amino acid sequences of FtmOx1 with the N-terminal strep-tag

MGDRGPEFWSHPQFEKTVDSKPQLQRLAADADVDRMCRLLEEDGAFILKGLLPFDVVESFNRELDVQM AIPPPKGERLLADKYPPHFKYVPNVATTCPTFRNTVLINPVIHAICEAYFQRTGDYWLSAAFLREIESGMP AQPFHRDDATHPLMHYQPLEAPPVSLSVIFPLTEFTEENGATEVILGSHRWTEVGTPERDQAVLATMDP GDVLIVRQRVVHAGGGNRTTAGKPRRVVLAYFNSVQLTPFETYRTMPREMVESMTVLGQRMLGWRTM KPSDPNIVGINLIDDKRLENVLQLKAADSPALEVDLQGDHGLSA*

Structure determination and analysis

The FtmOx1•Co^{II} crystal shows identical space group and unit cell with previous reported FtmOx1•Fe^{II} structure (pdb entry 4Y5T), the FtmOx1•Co^{II}•αKG•7 ternary complex crystal has the same space group type (P21) compared with FtmOx1•Fe^{II} structure, while having an expanded unit cell. A Cell Content Analysis suggested that in the FtmOx1•Co^{II}•αKG•7 crystals, there are three dimers within one asymmetric unit, instead of having only one dimer as in the FtmOx1•Fe^{II} crystals. Such differences might be due to the small re-arrangement of the crystal lattice upon compound binding. The structure of the FtmOx1•Co^{II}•αKG•7 complex was solved by molecular replacement using Phaser as implemented in the Phenix/CCP4 suite¹, using a model of PDB Entry 4Y5T (FtmOx1•Fe^{II} structure). The initial solution was refined using phenix refine, and the resulting Fo-Fc omit map showed clear electron density for the substrate 7 located at the hydrophobic subpocket; the shape of the omit electron density fits quite well with compound 7. The omit map located at the polar sub-pocket suggested the existence of α KG. The model was further improved using iterative cycles of manual model building in COOT², followed by computational refinement using phenix.refine³⁻⁴. After ligands were placed, potential sites of solvent molecules were identified by the automatic water-picking algorithm in COOT and phenix.refine. The positions of these automatically picked waters were checked manually during model building. The final model suggested a Real Space Correlation Coefficient (RSCC) ranging from 0.84 to 0.93 in each chain, indicating a good fit between compound 7 and electron density. Assuming the similar thermal factors for the ligand as surrounding residues, and the ligand occupancy is estimated to be in the

range of 0.81 to 0.86. Figures were generated with PyMOL (Schrödinger, LLC), and surface electrostatics were calculated with APBS⁵.

Single-turnover enzymatic reaction of FtmOx1 and Y224F using fumitremorgin B (1) as substrate

These enzymatic reactions were performed in an anaerobic Coy chamber. For reactions of FtmOx1, The anaerobic reaction mixture (200 μ l, in 100 mM Tris-HCl, pH 7.5) contained 140 μ M fumitremorgin B (**1**), 140 μ M FtmOx1 containing 126 μ M Fe^{II}, 140 μ M α KG, and 420 μ M ascorbate (or w/o ascorbate). The reaction mixture was sealed and initiated by adding 200 μ l of oxygen-saturated buffer (1.2 mM of oxygen) and incubated for 10 min at room temperature. The enzymatic reactions was quenched by adding 1 ml dichloromethane, the precipitated protein was removed by centrifugation at 13,000 g for 10 min, and the dichloromethane layer was carefully separated for further analysis. The reaction mixture was extracted once more using a second 1 ml volume of dichloromethane. The combined dichloromethane layers were concentrated by rotatory evaporation in vacuo, and the residue was re-dissolved in 1 ml of acetonitrile and subjected to LC-HR-MS(ESI) analysis on a Waters ACQUITY BEH C18 column (2.1 × 100 mm, 1.7 μ m particles) with a flow rate of 0.35 ml/min (During 0–12 min, ACN was increased from 5% to 99% in 0.1% HCOOH-containing H₂O). The reaction setup procejures of Y224F and LCMS analysis of enzymatic products were same to those of FtmOx1.

Single-turnover enzymatic conversion of 13-oxo-fumitremorgin B (7) by FtmOx1

The enzymatic reactions were performed in an anaerobic Coy chamber. For the single-turnover condition, the anaerobic reaction mixture (200 μ l, in 100 mM Tris-HCl, pH 7.5) contained 140 μ M 13-oxo-fumitremorgin B (7), 140 μ M FtmOx1 containing 168 μ M Fe^{II}, and 336 μ M α KG. The reaction mixture was sealed and initiated by adding 200 μ l of oxygen-saturated buffer (1.2 mM of oxygen) and incubated for 10 min at room temperature. The enzymatic reactions were quenched by adding 1 ml dichloromethane, the precipitated protein was removed by centrifugation at 13,000 *g* for 10 min, and the dichloromethane layer was carefully separated for further analysis. The reaction mixture was extracted once more using a second 1 ml volume of dichloromethane. The combined dichloromethane layers were concentrated by rotatory evaporation *in vacuo*, and the residue was re-dissolved in 1 ml of acetonitrile and subjected to LC-MS analysis on a C18 column (ACE Excel 5 C18, 4.6 × 250 mm, 5 μ m) with a flow rate of 1.0 ml/min (During 0–15 min, ACN was increased from 10% to 99% in 0.1% HCOOH-containing H₂O and then the column was eluted for 5 mins with 99% ACN in 0.1% HCOOH-containing H₂O).

Single-turnover enzymatic conversion of verruculogen (2) by FtmOx1 and characterization of 13-oxo-verruculogen (3)

The enzymatic reactions were performed in an anaerobic Coy chamber. The anaerobic reaction mixture (400 μ l, in 100 mM Tris-HCl, pH 7.5) contained 140 μ M vertuculogen (2), 140 μ M FtmOx1 containing 168 μ M Fe^{II}, and 168 μ M α KG. The reaction mixture was sealed and initiated by adding

200 µl of oxygen-saturated buffer (1.2 mM of oxygen) and incubated for 30 min at room temperature. The resulting reaction mixtures contained a final concentration of 70 µM **2**, 70 µM FtmOx1 containing 84 µM Fe^{II}, and 84 µM α KG. The enzymatic reaction was quenched by adding 1 ml dichloromethane, the precipitated protein was removed by centrifugation at 13,000 *g* for 10 min, and the dichloromethane layer was carefully separated. The reaction mixture was extracted once more using a second 1 ml volume of dichloromethane. The combined dichloromethane layers were concentrated by rotatory evaporation, and the residue was re-dissolved in 100 µl of acetonitrile and subjected to LC-HR-MS(ESI) analysis on a Waters ACQUITY BEH C18 column (2.1 × 100 mm, 1.7 µm particles) with a flow rate of 0.35 ml/min (During 0–12 min, ACN was increased from 5% to 99% in 0.1% HCOOH-containing H₂O and then the column was eluted for 3 min with 99% ACN in 0.1% HCOOH-containing H₂O).

We synthesized the 13-oxo-verruculogen (**3**) and used it as standard in the LC-MS analysis mentioned above. To a solution of verruculogen (**2**, 5 mg) in acetone (2.5 ml) was added 2,3-Dichloro-5,6-dicyano-1,4-benzoquinone (DDQ; 98%; 120 mg) under an ice-water bath. After stirring for 8 hours, the reaction mixture was filtered by 0.22 μ m membrane and analyzed by LC-MS. The reaction mixture was concentrated and redissolved in 1 ml MeOH. The sample was further purified by a semi-prep HPLC running with ACN-H₂O (During 0–15 min, ACN was increased from 10% to 99% in 0.1% HCOOH-containing H₂O and then the column was eluted for 5 mins with 99% ACN in 0.1% HCOOH-containing H₂O) at 4 ml/min using a C18 column (Titank, 10 × 250 mm) to obtain 13-oxo-verruculogen (**3**, 2.8 mg, *t*_R = 13.2 min).

13-oxo-verruculogen (3): pale-yellow powder; ¹H- and ¹³C- NMR data, see Table S3; HR-MS (ESI+) detected at m/z 510.2210, *cald*. for [M + H]⁺ C₂₇H₃₁N₃O₇: m/z 510.2234.

Decomposion of verruculogen (3) into verruculogen TR-2 (6) using excess ascorbate and Fe(II)

Verruculogen (**3**, 5 mg, 9.78 mmol) was dissolved in 300 µL DMSO and diluted with 3 mL Tris-HCl buffer (100 mM, pH 7.5), which was further added with sodium ascorbate (96.9 mg, 50 equiv.) and FeSO₄·7H₂O (13.6 mg, 5 equiv.) at room temperature. After stirring for 2 h, the reaction mixture was extracted with 5 ml of dichloromethane for three times, the combined dichloromethane layers were then concentrated by rotatory evaporation, and the residue was re-dissolved in 200 µl of acetonitrile. The sample was subjected to the semi-preparative RP-HPLC for purification, running with a C18 column (Titank, 10 × 250 mm, 5 µm) and eluted with ACN-H₂O (During 0–15 min, ACN was increased from 10% to 99% in 0.1% HCOOH-containing H₂O and then the column was eluted for 5 mins with 99% ACN in 0.1% HCOOH-containing H₂O) at 4 ml/min, to obtain verruculogen TR-2 (**5**, 0.6 mg, t_R = 8.2 min, 14% yeild).

verruculogen TR-2 (**6**): pale-yellow powder; ¹H- and ¹³C-NMR data, see Table S6; HR-MS (ESI+) detected at m/z 412.1861, *cald*. for [M – H₂O + H]⁺ C₂₂H₂₆N₃O₅: m/z 412.1867.

Supplementary schemes



Scheme S1. Tyrosyl radical-based catalytic cycle of the proposed mechanism of prostaglandin H synthase (PGHS or COX).

COX catalyzed the conversion of arachidonic acid (AA) and two molecules of O_2 to prostaglandin endoperoxide G_2 (PGG₂), based on a branched-chain model⁶⁻⁷.



Scheme S2. Proposed mechanism of Nvfl-catalyzed endoperoxidation.

a COX-like mechanism



b CarC-like mechanism



Scheme S3. The complete mechanistic models of COX-like and CarC-like proposed for FtmOx1-catalysis.



Scheme S4. The proposed ferryl-regeneration model from an undisclosed *Nature* reader.

Supplementary figures



Figure S1a. HR-ESI-MS spectrum of fumitremorgin B (1).



Figure S1b. ¹H NMR spectrum of fumitremorgin B (**1**) (600 MHz, CDCl₃-*d*₆).



Figure S1c. ¹³C NMR spectrum of fumitremorgin B (1) (150 MHz, $CDCl_3-d_6$).



Figure S2a. HR-ESI-MS spectrum of verruculogen (2).



Figure S2b. ¹H NMR spectrum of verruculogen (2) (600 MHz, DMSO-*d*₆).



Figure S2c. ¹³C NMR spectrum of verruculogen (2) (150 MHz, DMSO-*d*₆).



Figure S3b. ¹H NMR spectrum of 13-oxo-verruculogen (3) (600 MHz, DMSO-*d*₆).



Figure S3c. ¹³C NMR spectrum of 13-oxo-verruculogen (3) (150 MHz, DMSO-*d*₆).



Figure S4a. HR-ESI-MS spectrum of 12, 13-dihydroxy-fumitremorgin C (4).



Figure S4b. ¹H NMR spectrum of 12,13-dihydroxy-fumitremorgin C (4). (600 MHz, CDCl₃).



Figure S4c. ¹³C NMR spectrum of 12,13-dihydroxy-fumitremorgin C (4). (150 MHz, CDCl₃).



Figure S5a. HR-ESI-MS spectrum of 12-hydroxy-13-oxo-fumitremorgin C (5).



Figure S5b. ¹H NMR spectrum of 12-hydroxy-13-oxo-fumitremorgin C (5) (600 MHz, DMSO-*d*₆).



Figure S5c. ¹³C NMR spectrum of 12-hydroxy-13-oxo-fumitremorgin C (5) (150 MHz, DMSO-*d*₆).



Figure S6a. HR-ESI-MS spectrum of verruculogen TR-2 (6).



Figure S6c. ¹³C NMR spectrum of verruculogen TR-2 (6) (150 MHz, DMSO-*d*₆).



Figure S7a. HR-ESI-MS spectrum of 13-oxo-fumitremorgin B (7).



Figure S7b. ¹H NMR spectrum of 13-oxo-fumitremorgin B (7) (600 MHz, DMSO-*d*₆).



Figure S7c. ¹³C NMR spectrum of 13-oxo-fumitremorgin B (7) (150 MHz, DMSO-d₆).



Figure S8a. HR-ESI-MS spectrum of 13-oxo-21-hydroxy-fumitremorgin B (8).



Figure S9. Overall architecture of the FtmOx1•Co^{II}•αKG•7 ternary complex.

Monomers of FtmOx1 are colored in green and light blue, respectively. The cobalt is shown as a pink sphere, and the substrates are shown as orange (7) and yellow (α KG) sticks.



Figure S10. Composite omit map (*2mFo-DFc*) showing substrate 13-oxo-fumitremorgin B (**7**) and α KG at the FtmOx1 active site.

The map was generated by phenix using simple refinement after omitting each region. The electron density around each compound is shown in grey mesh, with the contour level at 1.0 rmsd. The residues coordinating with the metallo-center are shown as stick.



Figure S11. Superimposition of the FtmOx1 ternary structure with FtmOx1•Fe^{II}• α KG structure (pdb entry 4Y5S).

The FtmOx1•Fe^{II}• α KG complex structure is shown in white and yellow with 50% transparency. The regions of FtmOx1•Fe^{II}• α KG complex structure that had been re-arranged upon substrate 7 binding were colored in red with arrows to show the moving directions.



Figure S12. Interaction diagram between compound **7** and FtmOx1. The figure was prepared using Ligplot.



Figure S13. Superimposition of the FtmOx1•Co^{II}• α KG•7 complex structure with the FtmOx1•Fe^{II}• α KG•1 complex structure.

The FtmOx1•Co^{II}• α KG•7 complex structure is shown in green and blue, while the FtmOx1•Fe^{II}• α KG•1 complex structure is shown in white and wheat.



Figure S14. Active site comparison between the $FtmOx1 \cdot Co^{II} \cdot \alpha KG \cdot 7$ complex and the $FtmOx1 \cdot Fe^{II} \cdot \alpha KG \cdot 1$ complex (pdb entry 7ETK).

(The opposite view complementary to Figure 4a in the main text)

Supplementary Tables

		2 (_).
Pos.	δ _H ^a , mult (/ in Hz)	δc ^b , type
2		131.3, C
3	5.97, d (10.0)	49.2, CH
5		166.4, C
6	4.45, dd (10.1, 6.9)	58.9, CH
7a	1.94, m, overlapped	20.1 CH ₂
7b	2.47, m	2 9.1, CH2
8	2.10, m	22.8, CH ₂
9	3.64, dd (9.2, 4.5)	45.4, CH ₂
11		170.6, C
12		83.1, C
12-OH	4.01, brs	
13	5.77, brs	69.1, CH
13-OH	4.70, m	
14		104.5, C
15		120.4, C
16	7.85, d (8.7)	121.5, CH
17	6.80, dd (8.7, 2.2)	109.5, CH
18		156.4, C
19	6.69, d (2.2)	94.0, CH
20		138.0, C
21	4.54, m	41.9, CH ₂
22	5.04, m	120.7, CH
23		134.8, C
24	1.85, s	18.4, CH ₃
25	1.70, s	25.9, CH ₃
26	4.70, m, overlapped	123.1, CH
27		135.4, C
28	1.99, s	18.5, CH₃
29	1.63, s	25.7, CH₃
30	3.84, s	55.9, CH ₃
$^a Recorded$ at 600 MHz in CDCl3. $^b Recorded$ at 150 MHz in CDCl3.		

Table S1.¹H and ¹³C-NMR data of fumitremorgin B (1).

Pos.	$\delta_{\mathrm{H}^{\mathrm{a}}}$ mult (J in Hz)	δ _c ^b , type
2		131.7, C
3	6.06, d (10.2)	49.0, CH
5		170.9, C
6	4.48, dd (10.0, 7.2)	58.8, CH
7a	2.50, m	
7b	2.11, m, overlapped	29.2, CH ₂
8a	2.12, m, overlapped	
8b	1.98, m	22.7, CH2
9	3.64, m	45.5, CH ₂
11		166.3, C
12		82.7, C
13	5.66, s	68.8, CH
14		105.7, C
15		121.1, C
16	7.90, d (8.6)	121.8, CH
17	6.83, dd (8.6, 2.2)	109.5, CH
18		156.5, C
19	6.60, d (2.2)	94.0, CH
20		136.3, C
21	6.64, d (8.2)	85.9, CH
22	5.05, m	118.6, CH
23		143.3, C
24	2.00, brs	18.9, CH ₃
25	1.74, brs	25.8, CH ₃
26a	2.02, d (13.4)	51.3, CH ₂
26b	1.68, dd (13.5, 10.2)	
27		82.2, C
28	1.01, s	27.2, CH ₃
29	1.72, s	24.3, CH ₃
30	3.84, s	55.9, CH ₃
13-OH	4.78, brs	
^a Recorded at 6	00 MHz in CDCl ₃ . ^b Recorded at	150 MHz in $CDCl_3$.

Table S2. ¹H- and ¹³C-NMR data of verruculogen (2).

SUPPORTING INFORMATION Table S3. ¹H- and ¹³C-NMR data of 13-oxo-verruculogen (3).

Pos.	$\delta_{ ext{H}^{ ext{a}}}$ mult (J in Hz)	$\delta_{ m C^b}$, type
2		147.2, C
3	6.14, d (9.8)	47.4, CH
5		172.7, C
6	4.69, t (8.3)	60.0, CH
7a	2.28, m	28.4, CH ₂
7b	2.07, m	
8	1.76, m	22.8, CH ₂
9a	3.60, m, overlapped	45.3, CH ₂
9b	3.54, m, overlapped	
11		164.2, C
12		81.3, C
13		180.4, C
14		108.6, C
15		118.2, C
16	8.02, d (8.6)	121.6, CH
17	6.97, dd (8.6, 2.2)	112.3, CH
18		157.2, C
19	6.99, d (2.2)	95.5, CH
20		137.0, C
21	7.08, d (8.4)	86.2, CH
22	4.97, d (8.4)	117.1, CH
23		145.1, C
24	2.03, s	18.7, CH ₃
25	1.71, s	25.3, CH ₃
26a	1.87, m, overlapped	
26b	1.71, m, overlapped	33.3, СП2
27		81.6, C
28	1.00, s	26.2, CH ₃
29	1.57, s	23.6, CH ₃
30	3.81, s	55.4, CH ₃
^a Recorded at 6	600 MHz in DMSO- <i>d</i> 6. ^b Recorded at	150 MHz in DMSO- d_6 .

Table S4. ¹H- and ¹³C-NMR data of 12,13-dihydroxy-fumitremorgin C (4). $\delta_{\rm H^a}$ mult (*J* in Hz) $\delta_{\rm C^b}$, type Pos. 1-NH 7.68, s 2 130.2, C 3 5.87, dd (9.6, 0.9) 50.3, CH 5 171.1, C 6 4.43, dd (10.2, 6.8) 58.9, CH 7a 2.48, m 29.3, CH₂ 7b 2.00, m 8 2.09, m 22.7, CH₂ 9 3.65, m 45.6, CH₂ 166.3, C 11 12 83.1, C 13 5.75, s 68.8, CH 14 105.5, C 15 120.8, C 7.80, d (8.7) 121.4, CH 16 17 6.81, d (8.7, 2.1) 110.0, CH 18 156.8, C 19 6.84, d (2.1) 95.1, CH 20 137.7, C 21 4.80, d (9.6) 124.1, CH 22 134.8, C 23 1.67, s 25.8, CH3 24 2.00, s 18.4, CH₃ 25 3.83, s 55.9, CH₃ ^aRecorded at 600 MHz in CDCl₃. ^bRecorded at 150 MHz in CDCl₃.

Pos.	$\delta_{ ext{H}^{ ext{a}}}$ mult (J in Hz)	$\delta_{ m C^b}$, type
2		147.1, 0
3	5.86, d (9.3)	48.8, CH
5		173.1, 0
6	4.54, t (7.8)	60.0, CH
7a	2.28, m	20.0 CU
7b	1.82, m, overlapped	28.9, UI
8	1.82, m, overlapped	22.6, CH
9a	3.42, m, overlapped	1E 2 CU
9b	3.36, m, overlapped	45.3, Сп
11		164.4, 0
12		82.5, C
L2-0H	7.83, brs	
13		180.2, 0
14		108.0, 0
15		117.6, 0
16	7.91, d (8.4)	121.2, C
17	6.89, dd (8.4, 2.3)	111.4, C
18		157.0, 0
19	6.99, d (2.3)	96.0, CH
20		138.3, 0
21	4.82, d (9.1)	122.9, C
22		136.2, 0
23	1.67, d (0.9)	25.6, CH
24	1.88, d (0.9)	18.4, CH
25	3.80, s	55.3, CH

Pos.	$\delta_{ m H^a}$ mult (J in Hz)	$\delta_{ ext{C}}^{ ext{b}}$, type
1-NH	10.64, s	
2		137.2, C
3	5.36, dd (8.2, 3.9)	49.8, CH
5		170.9, C
6	4.40, t (7.8)	59.1, CH
7a	1.93, m	
7b	1.87, m	29.7, CH ₂
8	2.73, m	22.8, CH ₂
9	3.51, m (overlapped)	45.4, CH ₂
11		166.7, C
12		83.8, C
12-0H	8.36, brs	
13		69.0, C
13-0H	5.50, s	
14		107.0, C
15		121.0, C
16	7.62, d (8.7)	121.2, C
17	6.61, dd (8.7, 2.3)	108.9, CH
18		155.5, C
19	6.88, d (2.3)	95.2, CH
20		131.7, C
21a	2.29, m	40.2 CH
21b	1.93, m	48.3, CH2
22		68.3, C
22-0H	6.19, brs	
23	0.97, s	29.2, CH ₃
24	1.07, s	30.7, CH ₃
25	3.75, s	55.6, CH ₃
^a Recorded at 600	MHz in DMSO-d ₆ . ^b Recorded at 1	50 MHz in DMSO-d6.

Pos.	$\delta_{ m H^a}$ mult (/ in Hz)	$\delta_{ m C^b}$, type
2		147.2, C
3	6.10, d (9.8)	48.6, CH
5		165.1, C
6	4.73, m	60.5, CH
7	2.26, m	29.0, CH ₂
8a	1.85, m	22.1 CU
8b	1.87, m, overlapped	23.1, CH2
9	3.45, m	45.7, CH ₂
11		173.4, C
12		82.3, C
13		180.9, C
14		108.3, C
15		118.1, C
16	8.00, d (8.6)	122.9, CH
17	6.95, dd (8.6, 2.2)	112.3, CH
18		157.6, C
19	6.93, d (2.2)	95.5, CH
20		139.0, C
21a	4.62, m	427 CH
21b	4.66, m	42.7, CH2
22	4.96, m	119.3, CH
23		136.1, C
24	1.85, s	18.6, CH ₃
25	1.69, s	25.8, CH ₃
26	4.73, d (9.6)	122.0, CH
27		137.5, C
28	1.87, s	19.0, CH ₃
29	1.60, s	25.7, CH ₃
30	3.81, s	55.9, CH₃

Table S8. Crystallographic data collection and refinement statistics for the FtmOx1•Co^{II}• α KG•7 ternary complex.

Data Collection		
Wavelength (Å)	0.97852	
Resolution range (Å)	30.00 - 2.87 (2.92 - 2.87)	
Space group	P 1 2 ₁ 1	
Unit cell	104.05.85.44.160.77	
a, b, c (Å)	108.84	
β (°)	100.04	
Unique reflections	60482 (2949)	
Multiplicity	6.4 (6.3)	
Completeness (%)	98.5 (97.6)	
Mean I/sigma(I)	13.39 (2.44)	
CC _{1/2}	0.980 (0.838)	
R _{sym}	0.140 (0.608)	
Refinement		
Rwork	0.1608 (0.2161)	
R _{free}	0.2110 (0.2601)	
Number of non-hydrogen	14026	
atoms	14020	
macromolecules	13587	
ligands	302	
water	137	
Protein residues	1706	
RMS bonds (Å)	0.005	
RMS angles (°)	0.794	
Ramachandran favored	98.06	
(%)	70.00	
Ramachandran outliers	0	
(%)	0	
Average B-factor (Å ²)	52.77	
macromolecules	52.76	
ligands	58.19	
solvent	42.23	
¹ Statistics for the highest-resolution shell are shown in		
parentheses.		
$^2R_{free}$ is calculated with 5% of the data randomly omitted		
from refinement.		

References

(1) McCoy, A. J.; Grosse-Kunstleve, R. W.; Adams, P. D.; Winn, M. D.; Storoni, L. C.; Read, R. J., Phaser crystallographic software. *J. Appl. Crystallogr.* **2007**, *40*, 658–674.

(2) Emsley, P.; Lohkamp, B.; Scott, W. G.; Cowtan, K., Features and development of Coot. *Acta. Crystallogr. D. Biol. Crystallogr.* **2010**, *66*, 486–501.

(3) Afonine, P. V.; Grosse-Kunstleve, R. W.; Echols, N.; Headd, J. J.; Moriarty, N. W.; Mustyakimov, M.; Terwilliger, T. C.; Urzhumtsev, A.; Zwart, P. H.; Adams, P. D., Towards automated crystallographic structure refinement with phenix.refine. *Acta. Crystallogr. D. Biol. Crystallogr.* **2012**, *68*, 352–67.

(4) Liebschner, D.; Afonine, P. V.; Baker, M. L.; Bunkoczi, G.; Chen, V. B.; Croll, T. I.; Hintze, B.; Hung, L.-W.; Jain, S.; McCoy, A. J.; Moriarty, N. W.; Oeffner, R. D.; Poon, B. K.; Prisant, M. G.; Read, R. J.; Richardson, J. S.; Richardson, D. C.; Sammito, M. D.; Sobolev, O. V.; Stockwell, D. H.; Terwilliger, T. C.; Urzhumtsev, A. G.; Videau, L. L.; Williams, C. J.; Adams, P. D., Macromolecular structure determination using X-rays, neutrons and electrons: recent developments in Phenix. *Acta. Crystallogr. D.* **2019**, *75* (10), 861–877.

(5) Baker, N. A.; Sept, D.; Joseph, S.; Holst, M. J.; McCammon, J. A., Electrostatics of nanosystems: application to microtubules and the ribosome. *Proc. Natl. Acad. Sci. U. S. A.* **2001**, *98* (18), 10037–41.

(6) Tsai, A. L.; Kulmacz, R. J., Prostaglandin H synthase: resolved and unresolved mechanistic issues. *Arch. Biochem. Biophys.* **2010**, *493* (1), 103–24.

(7) Dietz, R.; Nastainczyk, W.; Ruf, H. H., Higher oxidation states of prostaglandin H synthase. Rapid electronic spectroscopy detected two spectral intermediates during the peroxidase reaction with prostaglandin G2. *Eur. J. Biochem.* **1988**, *171* (1–2), 321–8.