

Supplementary Information for: Quantum  
Anharmonic Calculations of Vibrational  
Spectra for Water Adsorbed on Titania  
Anatase (101) Surface: Dissociative versus  
Molecular Adsorption

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**Table S1: Effects of the U parameter and the dispersion terms on the relevant geometrical coordinates and binding energies for molecular H<sub>2</sub>O adsorption on the TiO<sub>2</sub> anatase (101) surface. The value chosen for the Hubbard U is 3.3 eV.**

	Layers	DFT $V_{xc}$		$r(\text{Ti}_{5c}-\text{O}_w)$ [Å]	$r(\text{O}_w-\text{H})$ [Å]	$d(\text{O}_{2c}-\text{H})$ [Å]	$\alpha(\text{H}-\text{O}_w-\text{H})$ [°]	BE <sup>a</sup> [eV]
Q-E present	4L	PBE	$\theta=0.25$	2.289	0.984	2.284	103.89	0.76
Q-E present	4L	PBE+U	$\theta=0.25$	2.289	0.986	2.240/2.237	103.54	0.84
Q-E present	4L	PBE-D3	$\theta=0.25$	2.284	0.986	2.184	103.45	1.00

<sup>a</sup>Experimental value 0.5-0.7 eV.<sup>1</sup>

**Table S2: Effects of the U parameter and the dispersion terms on the relevant geometrical coordinates and binding energies for dissociative H<sub>2</sub>O adsorption on the TiO<sub>2</sub> anatase (101) surface. The value chosen for the Hubbard U is 3.3 eV.**

	Layers	DFT $V_{xc}$	$\theta$	$r(\text{Ti}_{5c}-\text{O}_w)$ [Å]	$r(\text{O}_w-\text{H})$ [Å]	$d(\text{O}_{2c}-\text{H})$ [Å]	$d(\text{O}_w-\text{H}_{diss})$ [Å]	$\alpha(\text{O}_w-\text{H}_{diss}-\text{O}_{2c})$ [°]	BE [eV]
<b>Intrapair Dissociative Adsorption</b>									
Q-E present	4L	PBE	0.25	1.824	0.976	0.973	2.680	79.578	0.41
Q-E present	4L	PBE+U	0.25	1.848	0.977	0.972	2.685	80.048	0.55
Q-E present	4L	PBE-D3	0.25	1.822	0.977	0.972	2.735	78.015	0.57
<b>Interpair Dissociative Adsorption</b>									
Q-E present	4L	PBE	0.25	1.854	0.975	0.982	2.487	138.282	0.29
Q-E present	4L	PBE+U	0.25	1.885	0.976	0.983	2.410	139.841	0.43
Q-E present	4L	PBE-D3	0.25	1.863	0.976	0.984	2.331	142.441	0.51

**Table S3:** Comparison of harmonic frequencies for H<sub>2</sub>O and D<sub>2</sub>O adsorbed on the TiO<sub>2</sub> anatase (101) surface for different computational set-up. Finite Difference calculations have been performed with an atomic displacements of 0.01 Å and the frequencies are in cm<sup>-1</sup>.

	Layers	DFT	$\theta$	Method	H <sub>2</sub> O		D <sub>2</sub> O				
		$V_{xc}$				<b>Molecular Adsorption</b>					
Q-E present	4L	PBE	0.25	DFPT	1586	3580	3647	1163	2581	2673	
Q-E present	4L	PBE	0.25	FD	1578	3580	3670	1158	2583	2685	
Q-E present	4L	PBE+U	0.25	FD	1582	3549	3635	1161	2562	2659	
Q-E present	4L	PBE-D3	0.25	FD	1587	3560	3621	1164	2565	2653	
						<b>Intrapair Dissociative Adsorption</b>					
					(O <sub>w</sub> -H)	(O <sub>2c</sub> -H)		(O <sub>w</sub> - D)	(O <sub>2c</sub> - D)		
Q-E present	4L	PBE	0.25	DFPT	3760	3797		2738	2765		
Q-E present	4L	PBE	0.25	FD	3745	3760		2731	2738		
Q-E present	4L	PBE+U	0.25	FD	3752	3753		2731	2737		
Q-E present	4L	PBE-D3	0.25	FD	3748	3753		2731	2734		
						<b>Interpair Dissociative Adsorption</b>					
					(O <sub>w</sub> -H)	(O <sub>2c</sub> -H)		(O <sub>w</sub> - D)	(O <sub>2c</sub> - D)		
Q-E present	4L	PBE	0.25	DFPT	3775	3644		2748	2654		
Q-E present	4L	PBE	0.25	FD	3736	3614		2723	2634		
Q-E present	4L	PBE+U	0.25	FD	3718	3581		2710	2610		
Q-E present	4L	PBE-D3	0.25	FD	3717	3554		2709	2591		

**Table S4:** Convergence test of the harmonic frequencies *vs* the k-point sampling of the Brillouin zone. All the frequencies are in cm<sup>-1</sup>.

	Molecular H <sub>2</sub> O			Intrapair H <sub>2</sub> O		Interpair H <sub>2</sub> O	
Γ-point only	1586	3580	3647	3760	3797	3775	3644
2×2×1	1589	3582	3647	3746	3793	3784	3663
4×4×1	1589	3581	3647	3746	3793	3784	3663

**Table S5:** Convergence test of the harmonic frequencies *vs* the slab thickness. All the frequencies are in cm<sup>-1</sup>.

	Molecular H <sub>2</sub> O			Intrapair H <sub>2</sub> O		Interpair H <sub>2</sub> O	
4 Layers	1586	3580	3647	3760	3797	3775	3644
6 Layers	1586	3579	3645	3761	3795	3772	3653
8 Layers	1588	3577	3643	3761	3795	3769	3650

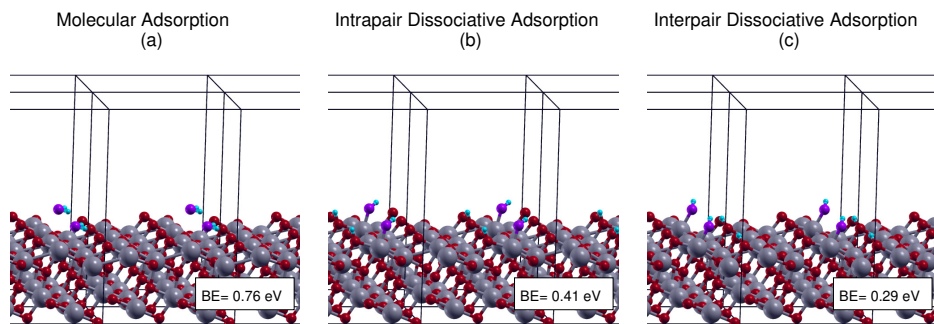


Figure S1: Supercell optimized geometries of molecular and dissociative H<sub>2</sub>O adsorption on the TiO<sub>2</sub> anatase (101) surface. Titanium atoms are in gray, surface oxygen in red, molecular oxygen in violet and hydrogen in cyan.

**Table S6: Comparison of vibrational frequencies of H<sub>2</sub>O in gas phase. All the frequencies are in cm<sup>-1</sup>.**

		Gas Phase Molecular H <sub>2</sub> O					
		Bending		Symmetric Stretching		Asymmetric Stretching	
		Frequency	$\Delta$	Frequency	$\Delta$	Frequency	$\Delta$
Experiment <sup>2-4</sup>		1595		3657		3756	
Q-E PBE	Harmonic	1580		3731		3843	
	Quasi-classical	1545	-35	3602	-129	3683	-160
	DC SCIVR	1546	-34	3602	-129	3682	-161
NWChem PBE	Harmonic	1605		3707		3809	
	Quasi-classical	1577	-28	3587	-120	3657	-152
	DC SCIVR	1578	-27	3588	-119	3658	-151
NWChem PBE0	Harmonic	1641		3874		3975	
	Quasi-classical	1612	-29	3758	-116	3831	-144
	DC SCIVR	1612	-29	3758	-116	3832	-143
NWChem B3LYP	Harmonic	1638		3807		3905	
	Quasi-classical	1609	-29	3691	-116	3756	-149
	DC SCIVR	1608	-30	3690	-117	3756	-149

**Table S7: Comparison of vibrational frequencies of D<sub>2</sub>O in gas phase. All the frequencies are in cm<sup>-1</sup>.**

		Gas Phase Molecular D <sub>2</sub> O					
		Bending		Symmetric Stretching		Asymmetric Stretching	
		Frequency	$\Delta$	Frequency	$\Delta$	Frequency	$\Delta$
Experiment <sup>2,3,5</sup>		1178		2671		2788	
Q-E PBE	Harmonic	1156		2689		2815	
	Quasi-classical	1138	-18	2617	-72	2733	-82
	DC SCIVR	1138	-18	2618	-71	2732	-83
NWChem PBE	Harmonic	1174		2673		2789	
	Quasi-classical	1159	-15	2607	-66	2713	-76
	DC SCIVR	1160	-14	2608	-65	2712	-77
NWChem PBE0	Harmonic	1200		2793		2912	
	Quasi-classical	1185	-15	2730	-63	2839	-73
	DC SCIVR	1186	-14	2730	-63	2838	-74
NWChem B3LYP	Harmonic	1198		2745		2860	
	Quasi-classical	1183	-15	2680	-65	2786	-74
	DC SCIVR	1182	-16	2680	-65	2786	-74

(a) H<sub>2</sub>O Molecular Adsorption

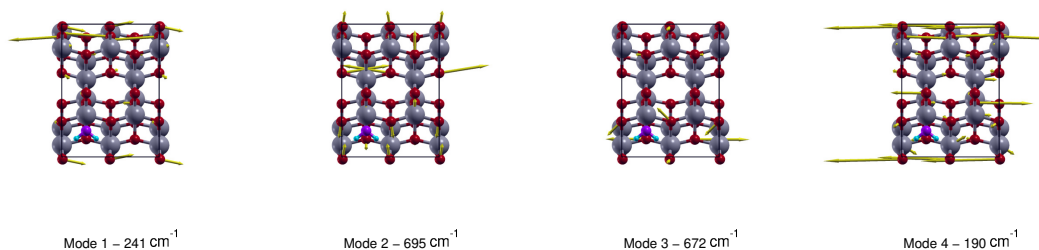


Figure S2: Plot of normalized eigen-displacements for the first 4 modes entering in the subspace for molecular adsorbed H<sub>2</sub>O.

(b) D<sub>2</sub>O Molecular Adsorption

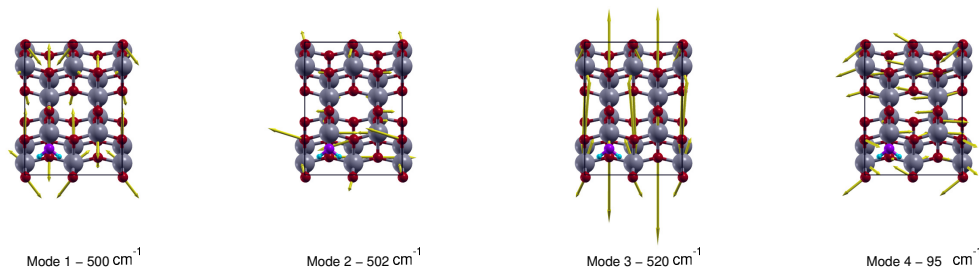
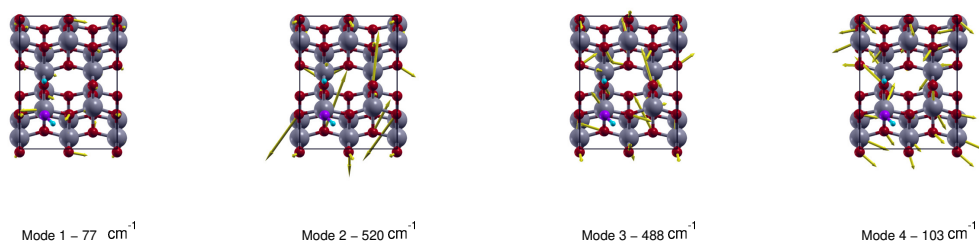


Figure S3: Plot of normalized eigen-displacements for the first 4 modes entering in the subspace for molecular adsorbed D<sub>2</sub>O.

(c) H<sub>2</sub>O Intrapair Dissociative Adsorption – O<sub>w</sub>-H Stretching



(d) H<sub>2</sub>O Intrapair Dissociative Adsorption – O<sub>2c</sub>-H Stretching

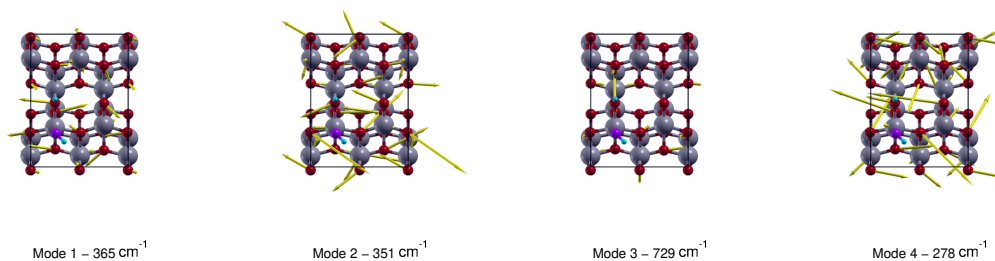
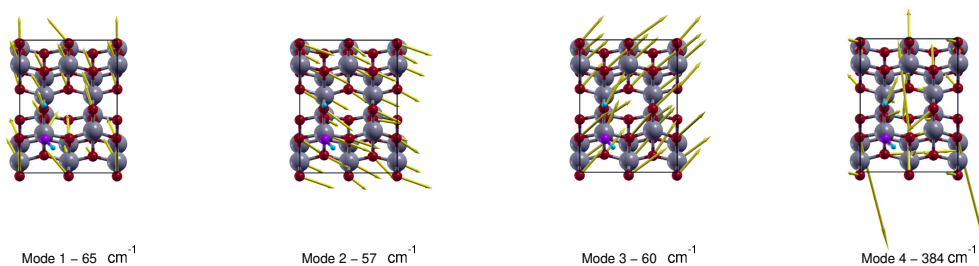


Figure S4: Plot of normalized eigen-displacements for the first 4 modes entering in the subspace for intrapair dissociative adsorbed H<sub>2</sub>O.

(e) D<sub>2</sub>O Intrapair Dissociative Adsorption – O<sub>w</sub>-H Stretching



(f) D<sub>2</sub>O Intrapair Dissociative Adsorption – O<sub>2c</sub>-H Stretching

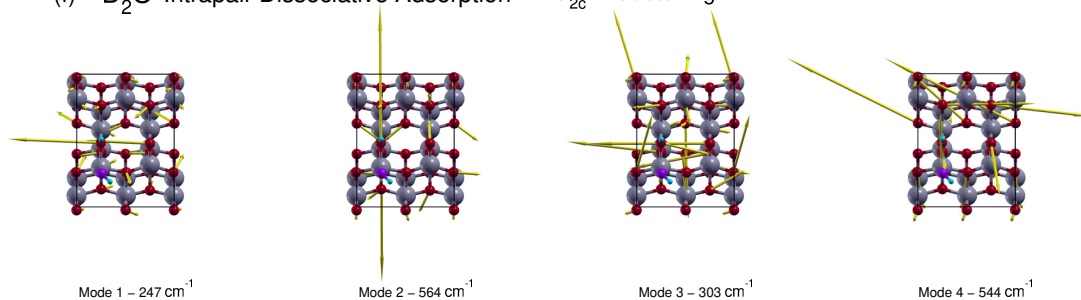
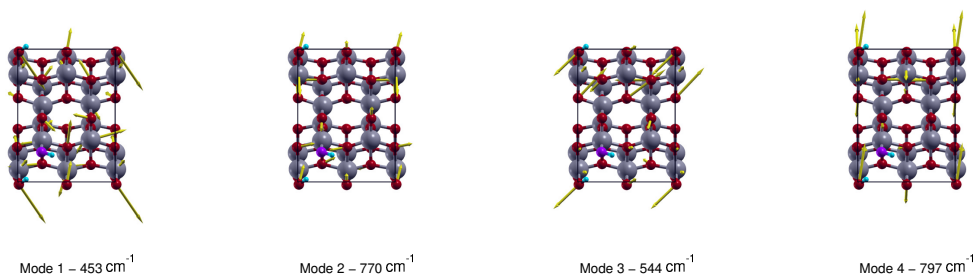


Figure S5: Plot of normalized eigen-displacements for the first 4 modes entering in the subspace for intrapair dissociative adsorbed D<sub>2</sub>O.

(g) H<sub>2</sub>O Interpair Dissociative Adsorption – O<sub>2c</sub>-H Stretching



(h) H<sub>2</sub>O Interpair Dissociative Adsorption – O<sub>w</sub>-H Stretching

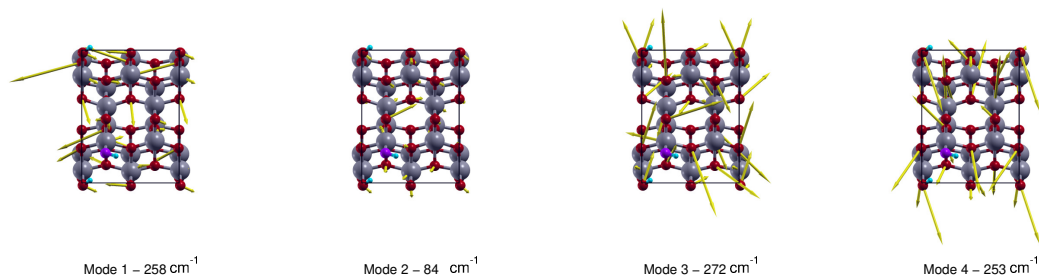
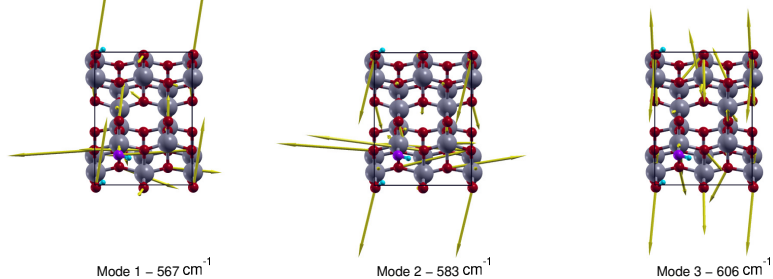


Figure S6: Plot of normalized eigen-displacements for the first 4 modes entering in the subspace for interpair dissociative adsorbed H<sub>2</sub>O.

(i) D<sub>2</sub>O Interpair Dissociative Adsorption – O<sub>2c</sub>-H Stretching



(j) D<sub>2</sub>O Interpair Dissociative Adsorption – O<sub>w</sub>-H Stretching

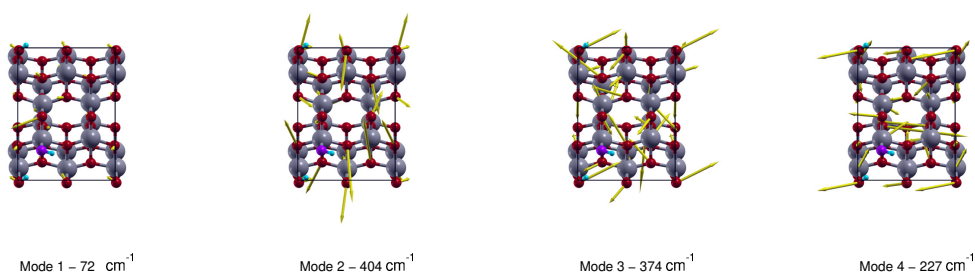


Figure S7: Plot of normalized eigen-displacements for the first 4 modes entering in the subspace for the interpair dissociative adsorbed D<sub>2</sub>O.

## References

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