Supporting Information

Halogenation of Tyrosine perturbs large-scale protein self-organization

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Supplementary Methods

1. Lipophilicity analysis

The lipophilicity of the amino acid residues was measured against 150 mM buffers as described¹. The lipophilicity of the amino acid residues was measured against 150 mM buffers: MES buffer (pH 6.0), phosphate buffer (ph 7.0), Tris buffer (pH 8.0) and borate buffer (pH 9.0). All reactions and manipulations were performed at the room temperature: 21-23°C. The starting amino acids and the reagents were of commercial grade. The solvents were of standard grade ("ACS certified"), and they were used without additional purification. NMR spectra were measured at a spectrometer machine operating at 500 MHz 1H frequency at 298 K. First, amino acids were derivatized chemically to yield methyl esters of N-acetylamino acids. An amino acid (1 mmol) was mixed with dichloromethane (10 ml) and N, N-diisopropylethylamine (0.7 ml, 4 equiv.). Trimethylsilylchloride (0.4 ml, 3.2 equiv.) was added, and the mixture was stirred for about 15 min until solution was almost or completely clear. Acetic anhydride was added

in one equivalent (0.1 ml) and the reaction mixture was stirred for about 23 h, and then the reaction vial was left open under a fume hood overnight (20 h) allowing the solvent to evaporate. The residue was dissolved in methanol (2-3 ml), excess of trimethylsilylchloride (0.5 ml) was added, and the mixture was stirred for 60 h. Methanol was removed under reduced pressure, and the target substance was purified by two consecutive short silica gel (about 20 g) columns using ethyl acetate - methanol (40:1 to 20:1) mixture as an eluent. The obtained substance was subjected to distribution experiments as follows. An amino acid derivative (5-10 mg) was mixed with buffer (1 ml) and octan-1-ol (2 ml) in a glass vial, and the vial was vigorously shaken for the next 2-3 hours on a reciprocal laboratory shaker. The vial was centrifuged on a low speed for 1 min to allow full separation of phases. Samples of each phase (0.35 ml) were taken gently by standard 1.00 ml syringes and diluted with DMSO-d6 (0.20 ml) in identical standard 5 mm NMR tubes. 1H NMR spectra were measured in one scan and multiscan in datasets with identical settings for samples of both phases. The datasets were copied and applied for measurements without readjustment of acquisition or processing parameters (except zero-order phase) and without readjustment of the receiver gain. One-scan and multi-scan (8*n) measurement were performed on the samples. A single 90-degree pulse experiment with no dummy scans was used for single scan experiments, and a 30-degree pulse experiment with 8 dummy scans was used for collection of multiscan spectra. The spectra were processed in a conventional manner using Fourier transform with 1-2 Hz line broadening and a 5-degree polynomial baseline correction. The comparison of the relative analyte concentration in the phases was made by comparison of the absolute integral values and/or by overlaying resonances from two spectra on each other, whichever was more accurate. The intensities of the resonances were compared by overlay and integration of the spectra. A ratio between resonance intensity in octan-1-ol and buffer phase was considered a distribution coefficient. For each pH/substance combination the distribution experiments were performed in triplicate. DpH values were read out for several separate resonances for each sample. Averaging these values produced the final DpH value and

the standard deviation for a compound/pH combination. N-Acetyl-tyrosine methyl ester and N-acetyl-3-iodo-tyrosine methyl ester were reported earlier, and other compounds were synthesized according to the protocol. The final distribution coefficients logD are provided in Supplementary Table 7. A dedicated article on the used method and reference datasets is available in reference 1.

N-Acetyl-3-chloro-tyrosine methyl ester: ¹H NMR (DMSO-d₆, 500 MHz), δ : 9.98 (s, 1H, OH), 8.29 (d, *J* = 7.7 Hz, 1H, NH), 7.18 (d, *J* = 1.6 Hz, 1H, ArH), 6.97 (dd, *J* = 8.3 1.6 Hz, 1H, ArH), 6.86 (d, *J* = 8.3 Hz, 1H, ArH), 4.38 (ddd, *J* = 9.2, 7.8, 5.6 Hz, 1H, α -CH), 3.60 (s, 3H, CH₃O), 2.90 (dd, *J* = 13.8, 5.5 Hz, 1H, C<u>H</u>H), 2.75 (dd, *J* = 13.8, 9.3 Hz, 1H, CH<u>H</u>), 1.80 (s, 3H, CH3CO). 13C {1H} NMR (DMSO-d₆, 126 MHz), δ : 172.6 (ester CO), 169.8 (amide CO), 152.1 (C), 130.7 (CH), 129.4 (C), 129.0 (CH), 119.7 (C), 116.9 (CH), 54.1 (α -CH), 52.2 (CH₃O), 36.0 (CH₂), 22.7 (CH₃).

N-Acetyl-3-bromo-tyrosine methyl ester: ¹H NMR (DMSO-d₆, 500 MHz), δ : 10.06 (s, 1H, OH), 8.29 (d, *J* = 7.8 Hz, 1H, NH), 7.32 (d, *J* = 2.1 Hz, 1H, ArH), 7.01 (dd, *J* = 8.3, 2.1 Hz, 1H, ArH), 6.85 (d, *J* = 8.3 Hz, 1H, ArH), 4.38 (ddd, *J* = 9.2, 7.8, 5.6 Hz, 1H, α -CH), 3.60 (s, 3H, CH₃O), 2.90 (dd, *J* = 13.9, 5.6 Hz, 1H, C<u>H</u>H), 2.75 (dd, *J* = 13.8, 9.3 Hz, 1H, CH<u>H</u>), 1.80 (s, 3H, CH₃CO). ¹³C{¹H} NMR (DMSO-d₆, 126 MHz), δ : 172.6 (ester CO), 169.8 (amide CO), 153.1 (C), 133.7 (CH), 129.8 (C), 129.7 (CH), 116.6 (CH), 109.3 (C), 54.1 (α -CH), 52.2 (CH₃O), 35.9 (CH₂), 22.7 (CH₃).

N-Acetyl-3-iodo-tyrosine methyl ester: ¹H NMR (DMSO-d₆, 500 MHz), δ : 10.14 (s, 1H, OH), 8.28 (d, *J* = 7.7 Hz, 1H, NH), 7.51 (d, *J* = 2.1 Hz, 1H, ArH), 7.03 (dd, *J* = 8.2, 2.1 Hz, 1H, ArH), 6.78 (d, *J* = 8.2 Hz, 1H, ArH), 4.36 (ddd, *J* = 9.0, 8.0, 5.6 Hz, 1H, α -CH), 3.60 (s, 3H, CH₃O), 2.88 (dd, *J* = 13.9, 5.6 Hz, 1H, C<u>H</u>H), 2.73 (dd, *J* = 13.8, 9.2 Hz, 1H, CH<u>H</u>), 1.80 (s, 3H, CH₃CO). ¹³C{¹H} NMR (DMSO-d₆, 126 MHz), δ : 172.6 (ester CO), 169.8 (amide CO), 155.7 (C), 139.6 (CH), 130.6 (CH), 130.2 (C), 115.2 (CH), 84.7 (C), 54.2 (α -CH), 52.2 (CH₃O), 35.8 (CH₂), 22.7 (CH₃).

N-Acetyl-3,5-dichloro-tyrosine methyl ester: ¹H NMR (DMSO-d₆, 500 MHz), δ: 9.96

(s, 1H, OH), 8.31 (d, J = 7.9 Hz, 1H, NH), 7.23 (s, 2H, ArH), 4.42 (ddd, J = 9.5, 7.9, 5.4 Hz, 1H, α -CH), 3.61 (s, 3H, CH₃O), 2.94 (dd, J = 13.9, 5.4 Hz, 1H, C<u>H</u>H), 2.76 (dd, J = 13.8, 9.7 Hz, 1H, CH<u>H</u>), 1.80 (s, 3H, CH₃CO). ¹³C{¹H} NMR (DMSO-d₆, 126 MHz), δ : 172.4 (ester CO), 169.8 (amide CO), 148.0 (C), 130.8 (C), 129.6 (CH), 122.3 (C), 53.7 (α -CH), 52.3 (CH₃O), 35.6 (CH₂), 22.7 (CH₃).

N-Acetyl-3,5-dibromo-tyrosine methyl ester: ¹H NMR (DMSO-d₆, 500 MHz), δ : 9.76 (s, 1H, OH), 8.32 (d, *J* = 8.0 Hz, 1H, NH), 7.40 (s, 2H, ArH), 4.42 (ddd, *J* = 9.6, 8.0, 5.6 Hz, 1H, α -CH), 3.61 (s, 3H, CH₃O), 2.94 (dd, *J* = 13.8, 5.4 Hz, 1H, C<u>H</u>H), 2.76 (dd, *J* = 13.8, 9.6 Hz, 1H, CH<u>H</u>), 1.80 (s, 3H, CH₃CO). ¹³C{¹H}NMR (DMSO-d₆, 126 MHz), δ : 172.4 (ester CO), 169.8 (amide CO), 149.7 (C), 133.3 (CH), 132.2 (C), 112.1 (C), 53.8 (α -CH), 52.3 (CH₃O), 35.4 (CH₂), 22.7 (CH₃).

N-Acetyl-3,5-diiodo-tyrosine methyl ester: ¹H NMR (DMSO-d₆, 500 MHz), δ : 9.39 (s, 1H, OH), 8.30 (d, *J* = 8.0 Hz, 1H, NH), 7.59 (s, 2H, ArH), 4.37 (ddd, *J* = 9.4, 8.0, 5.6 Hz, 1H, α -CH), 3.60 (s, 3H, CH₃O), 2.89 (dd, *J* = 13.8, 5.4 Hz, 1H, C<u>H</u>H), 2.72 (dd, *J* = 13.8, 9.5 Hz, 1H, CH<u>H</u>), 1.80 (s, 3H, CH₃CO). ¹³C {¹H} NMR (DMSO-d₆, 126 MHz), δ : 172.4 (ester CO), 169.7 (amide CO), 154.4 (C), 140.1 (CH), 133.9 (C), 87.3 (C), 53.9 (α -CH), 52.3 (CH₃O), 35.0 (CH₂), 22.7 (CH₃).



NMR spectra for *N*-acetyl-3-halo-tyrosine methyl esters:

¹³C{¹H} NMR (DMSO-d₆, 126 MHz):



NMR spectra for *N*-acetyl-3,5-dihalo-tyrosine methyl esters:

¹H NMR (DMSO-d₆, 500 MHz):



$^{13}C\{^{1}H\}$ NMR (DMSO-d₆, 126 MHz):



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2. FtsZ structure simulation

Homology modelling

A homology model was created using the I-TASSER suite²(templates from *M. Jannaschii*, PDB code: 1W5A)³. Secondary structure refinements were performed using Modeller software⁴.

Wild-type structure preparation

The initial homology model was first relaxed by a 10 ns MD simulation run to ensure a thermally equilibrated molecule was obtained. Titrations were performed using Karlsberg²⁺ on CUDA platforms, utilising a cavity algorithm with constant $0.7^{5,6}$. GTP and bound Mg²⁺ positions were taken by aligning 1W5A to the homology model with the MultiSeq tool in VMD^{4,7}. A preliminary minimisation procedure of 2000 steps in a protein dielectric constant $\varepsilon = 4$ was used to optimise side chains of residues within 8 Å of GTP; all other atoms were constrained. This model was used as a monomer reference structure in future calculations.

Mutant structure preparation

Crystal structures containing the mutated residues were found as the basis structures for the modified tyrosine (PDB codes: 4NX2 for Cl and 3GFD for I mutant respectively)^{8,9}. A methyl group replaced CA and the backbone for the purposes of charge calculation. Geometry optimisation of the hydrogen atoms was performed whilst heavy atoms were constrained, and the electrostatic potential (ESP) was calculated using the Gaussian 16 program¹⁰. Different basis sets were used according to the number of orbitals required for each halogen: 6-31G* for chlorine¹¹, 3-21G for iodine¹². Partial charges were calculated using RESP tools in the Amber antechamber¹³. The optimised phenyl ring was aligned with tyrosine ring of the reference structure and merged with the backbone. Charges for the backbone were taken from the tyrosine residue in the Amber force field. The charge of the methyl substituent of the ring was redistributed in the residue to give an overall neutral charge; final charges are shown in Supplementary Table 5. Bond, angle, dihedral and improper dihedral parameters were calculated based on the methyl ring with optimized hydrogens, merged with the backbone of the reference structure; the backbone was necessary to ensure all dihedrals could be calculated and to enable Amber to recognize the main chain atoms that connect the residue to the rest of the protein. This was done using the antechamber package of Amber¹³. All necessary parameters were taken from the standard parm10 database.

Dimer structure preparation

The 2004 *M. Jannaschii* dimer structure (1W5A) was used as a template, on which the monomer input structures were aligned using the MultiSeq plugin of VMD¹⁴. As a result of this, the N-terminal tail of the upper monomer pointed inwards and clashed with the dimer interface, which would lead to the calculation of large repulsive forces for the tail and a system crash during the minimization procedure. Therefore, the monomer PDB was aligned to the first 12 residues of each chain in the template structure. The coordinates were then saved and merged with the monomer PDB from the backbone nitrogen atom at the 11 position. The preliminary dimer structure was minimized using a dielectric constant of 4, in order to relax the protein structure. This model was used as a dimer reference structure in later analysis.

Molecular dynamics simulation

In this study, all molecular dynamics (MD) modelling was performed using the Amber software suite⁷, using the recommended ff14SB, TIP3P and gaff2 force fields for protein, water, and generalised molecules respectively¹⁵ and in-house developed GTP parameters¹⁶. In each simulation, the protein was solvated in a TIP3P¹⁷ periodic box, measuring 12 Å from the edges of the protein. The system was neutralized with K⁺ ions, which were distributed based on a Coulombic potential grid. No further ions were added, apart from the Mg²⁺ ions associated with the active site. The energy of the system was minimized for 12 000 steps, the first 10 000 steps using the steepest descent method. The system was then heated from 0 K to 300 K over the course of 500 ps using

an NVT ensemble, during which backbone atoms were constrained. Langevin dynamics were used to control the temperature with a collision frequency of 2.0 ps^{-1 18}. The system was equilibrated for 200 ps without constraints on backbone atoms preceding production. A 2 fs integration time-step was used for the simulation and the SHAKE algorithm¹⁹ was employed to constrain hydrogen bonds during the simulation. Non-bonded terms were cut off at 12 Å; particle mesh Ewald (PME) was used to treat long-range electrostatics²⁰. Simulations were performed on a CUDA- accelerated platform with 12 cores at 3.0 GHz and 128 GB RAM.

Supplementary Figures



Supplementary Figure 1. One round positive selection method to identify specific aaRS variants for orthogonal activating a halogenated Tyr analogue. a.) E. coli strain DH10b co-transformed with *Mj*TyrRS gene library plasmid (pBU18'1GK MjTyrRS library) and selection plasmid containing chloramphenicol acetyltransferase (cat) and small ubiquitin-like modifier tagged sfGFP bearing stop codon is grown in the presence of a specific HY and proper concentration of chloramphenicol (Cm). The cells containing the specific MiTyrRS variants could only survive and have green fluorescence in the presence of HYs in the medium with Cm. The promising clones are sequenced and MS is used to confirm the fidelity of HYs incorporation by screened MjTyrRSs. b-c.) MjTyrRS gene library plasmid (pBU18'1GK_MjTyrRS_library) and the dual reporter selection plasmid (pPAB26 cat (Q98TAG, D181TAG) MjtRNATyCUA his SUMO-sfGFPR2TAGstrep). pPAB26 has a p15A ori, a Kan resistance for plasmid propagation, and one copy of the suppressor MitRNATyr CUA, controlled by proK promoter. It contains a CAT encoding gene bearing amber codons at position Gln98 and Asp181 and SUMO fused sfGFP bearing TAG at Arg-2 position driven by T5 promoter.



Supplementary Figure 2. Selective incorporation of halogenated tyrosine analogues by one-round positive selection. a.) Growth assay of *E. coli* (DH10b) cell for evaluating the one-round positive selected *Mj*TyrRS towards Br₂Y (BRS) and Cl₂Y (CRS). Restreak of 100 colonies containing dual reporter selection plasmid and the potential BRSs encoding gene plasmid possible on new minimum medium²¹ agar plates supplied with 100μ g/mL Amp, 50μ g/mL Kan, 0.5 mM IPTG and 70μ g/mL Cm in presence or absence of 1mM Br₂Y or Cl₂Y. The represent fluorescence photos were taken under UV light. b.) Validation and characterization of HYs incorporation into SUMO-sfGFP

bearing an amber codon at Arg-2 by 9 unique synthetase mutants selected in this study (B4RS, B12RS, B37RS, B39RS, B48RS, C5RS, C45RS, C61RS, and C64RS). sfGFP fluorescence assay of DH10b containing selected MjTyrRS variants in the absence of HY or presence of ClY, Cl₂Y, BrY, Br₂Y, IY and I₂Y, respectively. Fluorescence values (excitation: 481 nm, emission: 511nm) are normalized to the lowest fluorescence value in DH10b and to the OD₆₀₀ of the bacterial culture. The data shown was an average of triplicate measurements (Mean \pm s.d.). (c) The amino acid sequences of MiTyrRS variants for recognizing Br₂Y and Cl₂Y are represented by WebLogo²². The amino acid type and randomized positions are shown on the x-axis. The overall height of the amino acid stacks plotted on the y-axis, indicates the sequence conservation at a given position, while the height of individual symbols within a stack indicates the relative frequency of an amino acid at that position. The colors of the amino acids correspond to their chemical properties; polar amino acids are shown in green, hydrophobic amino acids are shown in black, acidic amino acids are shown in red, basic amino acids are shown in blue, neutral amino acids are shown in purple. Source data of b. are provided as a Source Data file.



Supplementary Figure 3. The deconvoluted and non-deconvoluted ESI-MS spectra of SUMO-sfGFPR2TAG incorporated with HYs. The observed and expected molecular masses are as follows: a.) CIY-SUMO-sfGFP: observed:40250.83Da, expected: 40249.57Da. b.) Cl₂Y-SUMO-sfGFP: observed:40286.29Da, expected: 40284.01Da. c.) BrY-SUMO-sfGFP: observed: 40295.52Da, expected: 40294.02Da. d.) Br₂Y-SUMO-sfGFP: observed: 40372.92Da. e.) IY-SUMO-sfGFP: observed: 40343.12Da, expected: 40341.03Da. f.) I₂Y-SUMO-sfGFP: observed: 40468.99Da, expected: 40466.92Da. Sodium adduct species are indicated in the figure. Source data of a-f. are provided as a Source Data file.



Supplementary Figure 4. The deconvoluted and non-deconvoluted ESI-MS spectra of FtsZ-YFP-mts incorporated with HYs. The observed and expected molecular masses are as follows: a.) wild type FtsZ-YFP-mts (Y): main peak (observed: 68060.71 Da, expected: 68061.29 Da). b.) FtsZ(Y222ClY)-YFP-mts: main peak (observed: 68094.02 Da, expected: 68095.73 Da); c.) FtsZ(Y222Cl₂Y)-YFP-mts: main peak: observed:

68130.04 Da, expected: 68130.29 Da. d.) FtsZ(Y222BrY)-YFP-mts: main peak (observed: 68139.11 Da, expected: 68140.18 Da). e.) FtsZ(Y222 Br₂Y)-YFP-mts: main peak (observed: 68219.41 Da, expected: 68219.08 Da). f.) FtsZ(Y222IY)-YFP-mts: main peak (observed: 68186.04 Da, expected: 68187.19 Da). g.) FtsZ(Y222I₂Y)-YFPmts: main peak (observed: 68312.59 Da, expected: 68313.08 Da). All the satellite peaks are listed below: Peak1 (67685.53 Da), Peak2 (67719.94 Da), Peak5 (67765.03 Da), Peak6 (67844.63 Da), Peak7 (67808.00 Da) and Peak8 (67937.56 Da) represent the proteins that are around 375 Da smaller compared to the main peak, which probably resulted from the loss the of C-terminal α-helix of FtsZ-YFP-mts during ESI-MS measurement. The mts from MinD is a positively charged domain that is unstable under the acidic conditions applied in the ESI-MS measurement protocol due to a low ionic strength²³. It has been reported that the N-terminal or C-terminal α -helices can be prone to a loss under low ionic strength²⁴. Peak3 (observed: 68027.10 Da, expected: 68,027.52 Da) indicates FtsZ(Y222ClY)-YFP-mts without N-terminal Met but with 3 Na⁺ adduct. Peak4 (observed: 67652.90 Da, expected: 67,652.52 Da) represents FtsZ(Y222ClY)-YFP-mts without N-terminal Met and C-terminal mts (during ESI-MS measurement) but with 3 Na⁺ adduct. Peak 9 (observed: 68186.01 Da, expected: 68186.69 Da) is deiodination of FtsZ(Y222 I₂Y)-YFP-mts caused by formic acid during MS measurement²⁵. Source data of a-g. are provided as a Source Data file.



Supplementary Figure 5. Self-assembly of halogenated FtsZ in live *E. coli* cells. Localization and patterns of a.) Wild type FtsZ-YFP-mts (Y), b.) BrY, c.) Cl₂Y, d.) ClY, e.) Br₂Y, f.) IY, and g.) I₂Y incorporated FtsZ-YFP-mts in living *E. coli* cells. Scale bar, 2μm. The white lines indicate the position for the line plot. h-i) FRAP analysis of dynamics of FtsZ in *E. coli*. h.) Representative snapshots of *E. coli* cell overexpressed with wild type FtsZ-YFP-mts and FtsZ(Y222IY)-YFP-mts after photo-bleaching. Scale

bar: 1µm. i.) Fluorescence recovery curves of wild type FtsZ-YFP-mts and halogenated FtsZ(Y222X)-YFP-mts (X = ClY, BrY, Br₂Y, I₂Y, Cl₂Y, and IY) after photo-bleaching. Data points in i. are shown as means with standard deviations. The replicate numbers are indicated in the figure. Source data of a-g. and i. are provided as a Source Data file.



Supplementary Figure 6. Dimer (left) and monomer (right) of FtsZ structures. The monomer is rotated 90 degrees, looking down the main axis of the dimer and looking into the GTP binding pocket. The main protein body is shown in cartoon representation, GTP is displayed in Van der Waals representation and tyrosine 222 is displayed in licorice representation. The NTD is colored blue, CTD is red and the central H7 helix is shown in cyan. The C-terminal tail is displayed in transparent cyan.



Supplementary Figure 7. EPS surface of monomer conformations at 500 ns from. a.) "front", facing H7 helix, b.) "back", 180° rotation around y axis and facing CTT, c.) "above", 90° rotation around x axis and facing into GTP binding pocket and NTD interface in dimer, d.) "below", 180° around x axis and facing T7 loop and CTD interface in dimer (depicted for Wild type FtsZ in cartoon representation). Calculated using Pymol APBS tools using the dielectric constants $\varepsilon = 80$ for protein and $\varepsilon = 4$ for water. Surface shown for range ±5 eV.



Supplementary Figure 8. Depiction of solvation binding energy scheme. Solvent dielectric constant, ε , used for each system is shown in bold on left hand side, solute (protein/GTP moiety) dielectric constant $\varepsilon = 4$ for all calculations. Evaluation of the solvation binding energies shown in Fig. 4c were calculated according to this scheme. ΔG_1 signifies the coulombic energy of complex formation from its individual components, where a homogeneous dielectric constant ($\varepsilon = 4$) is used to represent the solvent and the solute. ΔG_2 and ΔG_4 represent the free energy difference gained when the solute is moved to a heterogeneous dielectric environment (solvent, $\varepsilon = 80$; solute, $\varepsilon = 4$) from a homogeneous dielectric environment ($\varepsilon = 4$). Here we do not assume that FtsZ is in the membrane, but rather in solution and the value of 4 refers to the protein dielectric constant. The protein dielectric value of 4 was chosen in order to allow proper treatment of polarization and small backbone fluctuations, the effects which would lower the dielectric constant significantly if treated explicitly²⁶⁻³¹. The overall binding energy, $-\Delta G_3$, was thus calculated by $-\Delta G_3 = \Delta G_1 + \Delta G_2 - \Delta G_4$. The components were calculated as following; ΔG_2 and ΔG_4 were computed using Adaptive Possion-Boltzmann Solver (APBS) to solve the linearized Poisson-Boltzmann equation^{32,33}. ΔG_1 was calculated using CHARMM (v46b1) software³⁴.

Supplementary tables

AA Position	<i>Mj</i> TyrRS	Library	Degenerate Codons
32	Y	All AAs except C, F, W, Y	VNK
65	L	All AAs	NNS
67	А	A, G, L, V	GBC / CTC
70	Н	A, G, L, V	GBC / CTC
108	F	All AAs	NNS
109	Q	All AAs except W	NDT / VHG
158	D	A, G, L, V	GBC / CTC
162	L	All AAs except W	NDT / VHG

Supplementary Table 1. *Mj*TyrRS gene library used in this study to select *Mj*TyrRS towards HYs

Supplementary Table 2. Sequences of aaRS mutant variants with specificity towards Br_2Y and Cl_2Y . BX (X: number) represents the clones from the Figure S2a plate containing Br_2Y while CX (X: number) shows the clones from the Figure S2a plate containing Cl_2Y . BXRS (X: number) and CXRS (X: number) represent the corresponding selected aaRS towards Br_2Y and Cl_2Y .

Clana	Synthetase	Position							
Cione		Y32	L65	A67	H70	F108	Q109	D158	L162
WT	<i>Mj</i> TyrRS	Y	L	А	Н	F	Q	D	L
B4	B4RS	V	Е	А	G	Q	Q	А	С
B12	B12RS	L	Е	А	G	Т	Q	А	Р
B37	B37RS	G	Е	А	G	Ν	Q	А	Q
B39	B39RS	Н	Е	А	G	L	Q	А	А
B48	B48RS	G	Е	А	G	Q	Q	А	Ν
C5	C5RS	G	Е	А	G	Ν	S	А	Е
C45	C45RS	А	Е	А	G	Е	Q	А	С
C61	C61RS	V	Е	А	G	А	Q	А	Q
C64	C64RS	V	Е	А	G	Q	Q	А	С

Supplementary Table 3. Expected and observed molecular weights of FtsZ-YFP-mts variants measured with ESI-MS*.

FtsZ-YFP-mts variants	Expected mass (Da)	Observed mass (Da)
Y (black line)	68061.29	68060.71
ClY (blue line)	68095.73	68094.02
Cl ₂ Y (yellow line)	68130.29	68130.04
BrY (green line)	68140.18	68139.11
Br ₂ Y (purple line)	68219.08	68219.41
IY (grey line)	68187.19	68186.04
I ₂ Y (red line)	68313.08	68312.59

*The accuracy of ESI-MS for intact protein molecules when using modern quadrupole time-of-flight (QTOF) MS has been documented to be 20 p.p.m when compared with the theoretical value³⁵.

Supplementary Table 4. Average ring diameter and treadmilling dynamics of Wild type FtsZ-YFP-mts and halogenated FtsZ-YFP-mts. n: ring numbers used for determining velocities.

Protein	Average ring diameter (μm)	Treadmilling velocity (nm/s)	
Wild type FtsZ-YFP-mts	0.80±0.18, n=149	21.63±5.11, n=115	
FtsZ(Y222ClY)-YFP-mts	0.62 ± 0.15 , n=150	15.10±4.78, n=93	
FtsZ(Y222Cl ₂ Y)-YFP-mts	No ring	No ring	
FtsZ(Y222BrY)-YFP-mts	0.89±0.19, n=150	15.61±6.15, n=101	
FtsZ(Y222 Br ₂ Y)-YFP-mts	0.92±0.18, n=150	14.52±4.49, n=93	
FtsZ(Y222 IY)-YFP-mts	No ring	No ring	
FtsZ(Y222 I2Y)-YFP-mts	1.03±0.21, n=149	11.36±4.19, n=104	

Supplementary Table 5. Atomic partial charges calculated for the tyrosine mutants shown in Fig. 4b using Gaussian 16 and RESP tools integrated in Amber software. The backbone CA was replaced by a methyl group; one hydrogen was removed from the methyl group to merge with the peptide backbone and the charge was redistributed to the CB and HB atoms in order to give an overall neutral charge. Backbone charges were taken directly from the tyrosine parameters from Amber force field. Basis sets used for the calculation of the electrostatic potential (ESP) are listed in the table.

	Cl ₂ Y, deprot		
Basis Set	6-31G*		
Charges	CB	-0.088522	
	CG	-0.017128	
	CD1	-0.258483	
	CD2	-0.258483	
	CE1	-0.047197	
	CE2	-0.047197	
	CZ	0.497556	
	OH	-0.777270	
	C11	-0.157343	
	C12	-0.157343	
	HB2	0.028949	
	HB3	0.028949	
	HD1	0.140856	
	HD2	0.140856	
	-		
	Ν	-0.4157	
	Н	0.2719	
	CA	-0.0014	
	HA	0.0876	
	С	0.5973	

O -0.5679

Supplementary Table 6. XYZ coordinates of 2,6-dichloro-4-methylphenolate after geometry optimization

CB	1	-0.24215	3.351338	-0.004059	
CG	2	-0.13521	1.833026	-0.006476	
CD1	3	-1.318698	1.077129	0.003353	
CD2	4	1.142828	1.24217	0.000326	
CE1	5	-1.219918	-0.339489	-0.003036	
CE2	6	1.246207	-0.151132	0.010944	
CZ	7	0.071039	-0.895503	0.011775	
OM	8	0.249437	-2.204104	-0.023587	
C11	9	-2.468533	-1.161441	0.004776	
C12	10	2.575478	-0.827044	0.003721	
HB1	11	0.43412	3.817144	-0.736895	
HB2	12	0.005252	3.802511	0.972084	
HD3	13	-1.259964	3.678975	-0.252759	
HD1	14	-2.291942	1.564658	-0.003648	
HD2	15	2.03438	1.868555	-0.011499	

	logD						
structure	pH 6	pH 7	pH 8	pH 9			
OH CO ₂ CH ₃ NHAc	+0.28±0.04	+0.43±0.08	+0.29±0.03	+0.32±0.03			
CI CI CO ₂ CH ₃ NHAc	+0.94±0.04	+1.00±0.06	+0.76±0.07	+0.34±0.04			
Br CO ₂ CH ₃ NHAc	+1.10±0.05	+1.17±0.04	+0.96±0.07	+0.49±0.06			
	+1.49±0.06	+1.58±0.04	+1.29±0.02	+0.79±0.03			
CI CI CO ₂ CH ₃ NHAc	+1.50±0.06	+1.16±0.04	+0.15±0.02	-0.72±0.04			
Br CO ₂ CH ₃ NHAc	+1.72±0.05	+1.33±0.08	+0.34±0.03	-0.49±0.08			
	+2.18±0.05	+1.95±0.05	+0.85±0.04	0.00±0.04			

Supplementary Table 7 Experimental distribution coefficients (octan-1-ol/buffer) for amino acid derivatives.

Plasmid	Replicatio n Origin	Resista nce	Description
pBU18'1GK_ <i>Mj</i> TyrRS_ Library	pUC	AmpR	<i>Mj</i> TyrRS Library (Supplementary Figure 1-2)
pPAB26_cat (Q98TAG, D181TAG) <i>Mj</i> tRNATyrCUA- his-SUMO-sfGFPR2TAG-strep	p15A	KanR	One-round Selection plasmid (Fig.S1, Fig.S2); SUMO-sfGFP production (Supplementary Figure 3)
pBU18'1GK_B4RS(B4RS)	pUC	AmpR	
pBU18'1GK_B12RS(B12RS)	pUC	AmpR	tRNA Synthetase
pBU18'1GK_B37RS(B372RS)	pUC	AmpR	(Supplementary Figure
pBU18'1GK_B39RS(B39RS)	pUC	AmpR	1-2); B48RS and
pBU18'1GK_C5RS(C12RS)	pUC	AmpR	SUMO afGED
pBU18'1GK_C45RS(C45RS)	pUC	AmpR	production
pBU18'1GK_C61RS(C61RS)	pUC	AmpR	(Supplementary Figure
pBU18'1GK_B48RS(B48RS)	pUC	AmpR	(Suppremental) 1 igure 3)
pBU18'1GK_C64RS(C64RS)	pUC	AmpR	,
pULTRA_B48RS_tRNATyrCU A	CloDF13	SpecR	Halogenated FtsZ
pULTRA_C64RS_tRNATyrCU A	CloDF13	SpecR	(Fig.1.d.e.f, Fig.2, Fig.3
pET28a_FtsZ (Y222TAG)-	ColE1	KanR	4)
YFP-mts			,
pET11b_wt_FtsZ-YFP-mts*	ColE1	AmpR	Wild type FtsZ protein expression (Fig.1.d.e.f, Fig.2, Fig.3, Supplementary Figure 4)

Supplementary Table 8. List of plasmids used in this study

AmpR: Ampicillin resistance; KanR: Kanamycin resistance; SpecR: Spectinomycin resistance. *Ramirez-Diaz DA, García-Soriano DA, Raso A, Mücksch J, Feingold M, Rivas G, Schwille P: **Treadmilling analysis reveals new insights into dynamic FtsZ** ring architecture. *PLoS biology* 2018, **16**:e2004845.

Gene constructs

Nucleotide sequence of MjTyrRS

Mutated amino acids are highlighted with red color

ATGGACGAATTTGAAATGATAAAGAGAAACACATCTGAAATTATCAGCGAGGAAGAGT TAAGAGAGGTTTTAAAAAAGATGAAAAATCTGCT<mark>TAC</mark>ATAGGTTTTGAACCAAGTGG ATTTGATATAATTATA<mark>TTG</mark>TTG<mark>GCT</mark>GATTTA<mark>CAT</mark>GCCTATTTAAACCAGAAAGGAGAGTT GGATGAGATTAGAAAAATAGGAGATTATAACAAAAAAGTTTTTGAAGCAATGGGGTTA AAGGCAAAATATGTTTATGGAAGTGAATTCCAGCTTGATAAGGATTATACACTGAATGT CTATAGATTGGCTTTAAAAAACTACCTTAAAAAGAGCAAGAAGGAGTATGGAACTTATAG GATTCTCATTATTAGGCGTTGATGTTGCAGTTGGAGGGATGGAGCAGAGAAAAATACA CGGGTTTGGATGGAGAAGGAAAGATGAGTTCTTCAAAAGGGAATTTTATAGCTGTTGA TGACTCTCCAGAAGAGATTAGGGCTAAGATAAAGAAAGCATACTGCCCAGCTGGAGTT GTTGAAGGAAATCCAATAATGGAGATAGCTAAATACTTCCTTGAATATCCTTTAACCATA AAAAGGCCAGAAAAATTTGGTGGAGATTTGACAGTTAATAGCTATGAGGAGTTAGAGA GTTTATTTAAAAATAAGGAATTGCATCCAATGtATTTAAAAAATGCTGTAGCTGAAGAAC TTATAAAGATTTTAGAGCCAATTAGAAAGAGATTATAA

Nucleotide sequence of Chloramphenicol Acetyltransferase (CAT(Q98TAG, D181TAG))

Nucleotide sequence CAT(Q98TAG, D181TAG) with 2 amber stop - codons highlighted with red colour at position 98 and 181.

Nucleotide sequence of SUMO-His-sfGFP(R2TAG)

In-frame TAG stop codon is highlighted in red.

CTAAGCCAGAGGTCAAGCCAGAAGTCAAGCCTGAGACTCACATCAATTTAAAGGTGTC CGATGGATCTTCAGAGATCTTCTTCAAGATCAAAAAGACCACTCCTTTAAGAAGGCTG ATGGAAGCGTTCGCTAAAAGACAGGGGTAAGGAAATGGACTCCTTAAGATTCTTGTACG ACGGTATTAGAATCCAAGCTGATCAGACCCCTGAAGATTTGGACATGGAGGATAACGAT ATTATTGAGGCTCACCGCGAACAGATTGGTGGCATG<mark>TAG</mark>AAAGGCGAAGAGCTGTTCA CTGGTGTCGTCCCTATTCTGGTGGAACTGGATGGTGATGTCAACGGTCATAAGTTTTCC GTGCGTGGCGAGGGTGAAGGTGACGCAACTAATGGTAAACTGACGCTGAAGTTCATCT GTACTACTGGTAAACTGCCGGTACCTTGGCCGACTCTGGTAACGACGCTGACTTATGGT GTTCAGTGCTTTGCTCGTTATCCGGACCATATGAAGCAGCATGACTTCTTCAAGTCCGC CATGCCGGAAGGCTATGTGCAGGAACGCACGATTTCCTTTAAGGATGACGGCACGTAC AAAACGCGTGCGGAAGTGAAATTTGAAGGCGATACCCTGGTAAACCGCATTGAGCTG AAAGGCATTGACTTTAAAGAAGACGGCAATATCCTGGGCCATAAGCTGGAATACAATTT TAACAGCCACAATGTTTACATCACCGCCGATAAACAAAAAATGGCATTAAAGCGAAT AAAACACTCCAATCGGTGATGGTCCTGTTCTGCTGCCAGACAATCACTATCTGAGCAC GCAAAGCGTTCTGTCTAAAGATCCGAACGAGAAACGCGATCATATGGTTCTGCTGGAG

TTCGTAACCGCAGCGGGCATCACGCATGGTATGGATGAACTGTACAAAAGCGCTTGGA GCCACCCGCAGTTCGAAAAATAA

Nucleotide sequence of FtsZ(Y222TAG)-YFP-mts

Y222 is highlighted in red.

ATGTTTGAACCAATGGAACTTACCAATGACGCGGTGATTAAAGTCATCGGC GTCGGCGGCGGCGGCGGTAATGCTGTTGAACACATGGTGCGCGAGCGCAT TGAAGGTGTTGAATTCTTCGCGGTAAATACCGATGCACAAGCGCTGCGTAA AACAGCGGTTGGACAGACGATTCAAATCGGTAGCGGTATCACCAAAGGAC TGGGCGCTGGCGCTAATCCAGAAGTTGGCCGCAATGCGGCTGATGAGGATC GCGATGCATTGCGTGCGGCGCTGGAAGGTGCAGACATGGTCTTTATTGCTG CGGGTATGGGTGGTGGTACCGGTACAGGTGCAGCACCAGTCGTCGCTGAA GTGGCAAAAGATTTGGGTATCCTGACCGTTGCTGTCGTCACTAAGCCTTTC AACTTTGAAGGCAAGAAGCGTATGGCATTCGCGGAGCAGGGGGATCACTGA ACTGTCCAAGCATGTGGACTCTCTGATCACTATCCCGAACGACAAACTGCT GAAAGTTCTGGGCCGCGGTATCTCCCTGCTGGATGCGTTTGGCGCAGCGAA CGATGTACTGAAAGGCGCTGTGCAAGGTATCGCTGAACTGATTACTCGTCC GGGTTTGATGAACGTGGACTTTGCAGACGTACGCACCGTAATGTCTGAGAT GGGCTACGCAATGATGGGTTCTGGCGTGGCGAGCGGTGAAGACCGTGCGG AAGAAGCTGCTGAAATGGCTATCTCTTCTCCGCTGCTGGAAGATATCGACC TGTCTGGCGCGCGCGCGCGTGCTGGTTAACATCACGGCGGGCTTCGACCTGC GTCTGGATGAGTTCGAAACGGTAGGTAACACCATCCGTGCATTTGCTTCCG AGCTGCGCGTAACCGTTGTTGCGACAGGTATCGGCATGGACAAACGTCCTG AAATCACTCTGGTGACCAATAAGCAGGTTCAGCAGCCAGTGATGGATCGCT ACCAGCAGCATGGGATGGCTCCGCTGACCCAGGAGCAGAAGCCGGTTGCT AAAGTCGTGAATGACAATGCGCCGCAAACTGCGCCCCCTCGACCTGCAGG CGGCCGCATGGTGAGCAAGGGCGAGGAGCTGTTCACCGGGGTGGTGCCCA TCCTGGTCGAGCTGGACGGCGACGTAAACGGCCACAAGTTCAGCGTGTCC

GGCGAGGGCGAGGCGATGCCACCTACGGCAAGCTGACCCTGAAGCTGAT CTGCACCACCGGCAAGCTGCCCGTGCCCTGGCCCACCCTCGTGACCACCC TGGGCTACGGCCTGCAGTGCTTCGCCCGCTACCCCGACCACATGAAGCAGC ACGACTTCTTCAAGTCCGCCATGCCCGAAGGCTACGTCCAGGAGCGCACC ATCTTCTTCAAGGACGACGGCAACTACAAGACCCGCGCCGAGGTGAAGTT CGAGGGCGACACCCTGGTGAACCGCATCGAGCTGAAGGGCATCGACTTCA AGGAGGACGGCAACATCCTGGGGCACAAGCTGGAGTACAACTACAACAGC CACAACGTCTATATCACCGCCGACAAGCAGAAGAACGGCATCAAGGCCAA CTTCAAGATCCGCCACAACATCGAGGACGGCGGCGTGCAGCTCGCCGAC ACTACCAGCAGAACACCCCCATCGGCGACGGCGGCGTGCAGCTCGCCGAC AACCACTACCTGAGCTACCAGTCCGCCCTGAGCAAAGACCCCAACGAGAA GCGCGATCACATGGTCCTGCTGGCGGCGCGCCCGTGCTGCCGGACAAGAAGAAGAA AAGGCTTCCTCAAACGCTTGTTCGGAGGAC

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