

Chemistry–A European Journal

Supporting Information

Flexible Alkylene Bridges as a Tool To Engineer Crystal Distyrylbenzene Structures Enabling Highly Fluorescent Monomeric Emission

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1. General procedure

^1H NMR and ^{13}C NMR spectra were recorded on BRUKER 500 (125 MHz) spectrometers or JEOL 400 (100 MHz) spectrometers for CDCl_3 solutions using tetramethylsilane (TMS) as an internal standard. Data for ^1H -NMR are reported as follows: chemical shift (δ ppm), multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, m = multiplet), integration, coupling constants (Hz). ^{13}C NMR spectra were reported as chemical shifts in ppm and multiplicity where appropriate. Melting points were recorded on a METTLER TOLEDO FP82HT Hot stage controlled with the METTLER TOLEDO Central Processor of the FP900 System. Recycling HPLC was performed at room temperature using a JAI LC-9204 Recycling Preparative HPLC with a refractive index (RI) detector: RI-50s and a UV detector: S-310 A model II controlled with TECHNO ALPHA Smart Chrom system.

UV-Vis spectra were recorded on a JASCO V-670 UV-vis spectrophotometer, and fluorescence spectra were recorded on a JASCO FP-6500 spectrofluorometer. The wavelengths obtained by the fluorescence spectrometer were converted to wavenumber using the equation $I(\tilde{\nu}) = \lambda^2 I(\lambda)^4$. Absolute quantum yields were measured by a Hamamatsu Photonics Quantaaurus QY apparatus.

All photophysical measurements were performed using dilute solutions with optical densities (ODs) around 0.1 at the maximum absorption wavelength in 1 cm path length quartz cells at room temperature (298 K). In addition, all sample solutions were deaerated by bubbling with argon gas for 15 min prior to the quantum yield.

Emission lifetimes were obtained using a Horiba FluoroCube time-correlated single-photon counting system. The excitation light sources were LED pulse lamps (NanoLED, 269 nm and 379 nm). The solution samples were dissolved in THF and purged with argon for 20 minutes before lifetime measurements. In all samples, the time-to-amplitude converter ranges were 50 ns, and the amounts of counts were 10000. Reptation rates were 1 MHz in solution samples and 100 kHz in solid samples.

Diffuse-reflectance spectra were recorded on a JASCO FP-6500 spectrofluorometer equipped with an integration sphere detector. Thus, the experimental error arising from the fluorescence, typically encountered with diffuse-reflectance spectrometers producing polychromatic outgoing light, was avoided. Samples and references were charged in a JASCO powder sample cell to obtain a sufficiently thick powder layer. Just before each measurement, the synchronous spectrum of the NaBr powder was measured as a reflectance spectrum of a standard reference $r_{\text{standard}}(\lambda)$. Then, a reflectance spectrum of each sample $r_{\text{sample}}(\lambda)$ was obtained following the same procedure. In addition to neat powder samples, the spectra of samples adsorbed on the NaBr powder were measured at a concentration of 1.0×10^{-3} M. The obtained reflectance spectra $r_{\text{sample}}(\lambda)$ and $r_{\text{standard}}(\lambda)$

were converted to Kubelka–Munk functions $f(r_\infty)$ using the following equation:

$$f(r_\infty) = \frac{(1 - r_\infty(\lambda))^2}{2r_\infty(\lambda)}, \text{ where } r_\infty(\lambda) = \frac{r_{\text{sample}}(\lambda)}{r_{\text{standard}}(\lambda)}$$

All diffuse-reflectance spectra are displayed at plots of the Kubelka-Munk functions, *i.e.*, $f(r_\infty)$ as a function of the wavelength, λ .

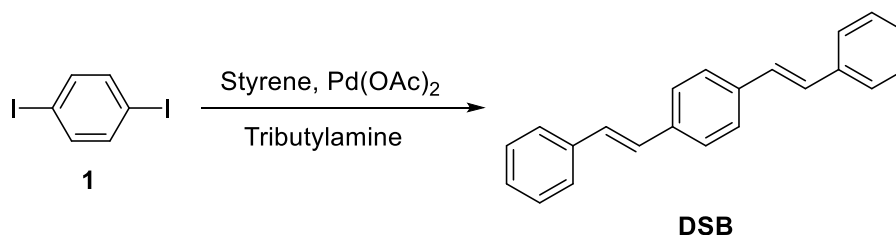
Single crystal X-ray structural analysis were obtained using a Rigaku AFC10 with Saturn 70 CCD detector (for **DBDMDB[7]**, **DB α MDB[7]-cross/parallel**, and **5DB α MDB[7]**) and a Rigaku FR-E+ with Hypix-6000 area detector (for **DSDMB**, **DBDB[6]**, **DBDB[7]** and **'BuDB α MDB[7]**). The single crystals of **DSDMB**, **DBDB[6]**, **DBDB[7]** were made from a mixed solvent of dichloromethane and methanol, and that of **DBDMDB[7]**, **DB α MDB[7]-cross/parallel**, **5DB α MDB[7]**, and **'BuDB α MDB[7]** were made from a mixed solvent of dichloromethane and hexane evaporated gradually under atmospheric conditions.

2. Materials

All solvents and chemicals were commercially available and used without further purification unless otherwise noted. Column chromatography was performed on silica gel (Silica Gel 60N, 63-210 μm , Kanto Chemical Co., Inc.). 1,4-Diiodobenzene, styrene, tributylamine, palladium(II) acetate, 1,4-diiodo-2,5-dimethylbenzene, potassium *tert*-butoxide, α -methylstyrene, tri(*o*-tolyl)phosphine, triethylamine, β -tetralone, *N*-phenylbis(trifluoromethanesulfonimide), 1,4-benzendiboronic acid bis(pinacol)ester, tetrakis(triphenylphosphine)palladium(0), 2,5-dibromo-*p*-xylene, bis(pinacolato)diboron, [1,1-bis(diphenylphosphino)ferrocene]palladium(II) dichloride dichloromethane adduct, methyltriphenylphosphonium bromide, α -tetralone, ethyltriphenylphosphonium bromide, [hydroxy(tosyloxy)iodo]benzene, 6-bromo-1-tetralone, ethyltriphenylphosphonium iodide, 1-pentene, ethylene glycol, *p*-toluensulphonic acid monohydrate, 2-chloro-2-methylpropane, and ammonium chloride were purchased from TCI (Tokyo, Japan). Tripotassium phosphate (K_3PO_4), *N,N*-dimethylformamide, ethanol, toluene, chloroform, methanol, 1,4-dioxane, ethyl acetate, hexane, tetrahydrofuran, and diethyl ether were purchased from Kanto Chem (Tokyo, Japan). 9-Borabicyclo 3.3.1 nonane 0.5M tetrahydrofuran, 1,3-dicyclohexylimidazolium tetrafluoroborate salt, and magnesium sulfate were purchased from Sigma-Aldrich Japan (Tokyo, Japan). Sodium hydroxide, magnesium, celite, and sodium hydrogen carbonate were purchased from Nacalai Tesque (Kyoto, Japan). Potassium acetate and nickel(II) chloride hexahydrate were purchased from Wako Pure Chem (Tokyo, Japan). Methylene chloride was purchased from AGC Chemicals (Tokyo, Japan). Spectrograde solvent (tetrahydrofuran) was purchased from Nacalai Tesque (Kyoto, Japan) and Wako Pure Chem (Tokyo, Japan).

3. Synthesis and characterization

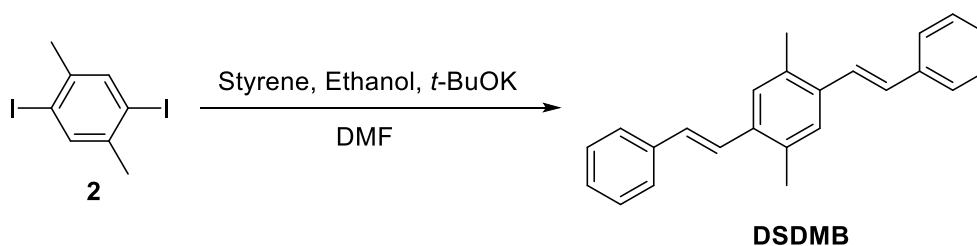
(1) Synthesis of 1,4-di(*E*-styryl)benzene (**DSB**)¹



Scheme S1. Synthesis of **DSB**.

A mixture of 1,4-diodobenzene (**1**) (0.66 g, 2.0 mmol), styrene (0.57 mL, 5.0 mmol), and Pd(OAc)₂ (0.01 g, 0.04 mmol) was dissolved in tributylamine (0.95 mL, 4.0 mmol). The reaction mixture was refluxed at 100°C for 1 day. The mixture solidified during the heating. The solid reaction mixture was broken up with a spatula and stirred with water. The solid was filtered, washed several times with fresh water, and filtered through Celite by hot DMF. Recrystallization from hot DMF and adding water until crystals began to appear in the hot filtrate gave **DSB** as a pale green solid. Yield: 48%; Mp: 257.0°C; ¹H-NMR (500 MHz, CDCl₃): δ 7.53 (d, *J* = 9.0 Hz, 8H), 7.37 (t, *J* = 7.5 Hz, 4H), 7.28 (s, 2H), 7.13 (t, *J* = 17.5 Hz, 4H) (**Figure S40**).

(2) Synthesis of ((1*E*,1'*E*)-(2,5-dimethyl-1,4-phenylene)bis(ethene-2,1-diyl))dibenzene (**DSDMB**)²

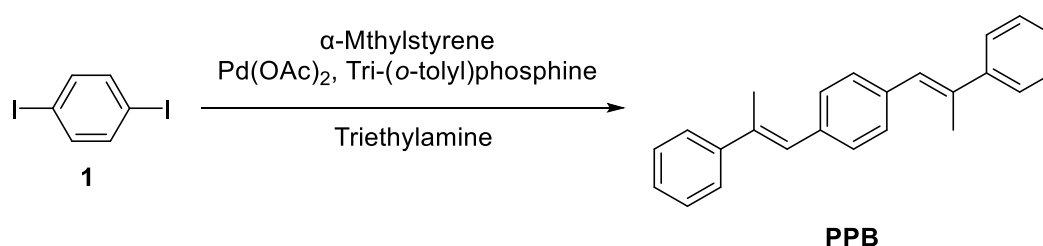


Scheme S2. Synthesis of **DSDMB**.

A mixture of 1,4-diiido-2,5-dimethylbenzene (**2**) (0.71 g, 2.0 mmol), styrene (2.29 mL, 20.0 mmol), potassium *tert*-butoxide (1.35 g, 12.0 mmol) and ethanol (0.05 mL, 0.9 mmol) was dissolved in DMF (10 mL) under argon atmosphere. The mixture was refluxed at 80°C for 6 hours, then cooled to room temperature. Water was added to the solution, and the mixture was extracted with diethyl ether. The organic layer was washed with water three times, dried over MgSO₄, filtered, and evaporated under reduced pressure to give a residue. The residue was chromatographed over silica gel, eluted with toluene, and

recrystallized from 3/1 (v/v) methanol/chloroform to give **DSDMB** as a pale-yellow solid. Yield: 6%; Mp: 196.7°C; $^1\text{H-NMR}$ (500 MHz, CDCl_3) δ 7.53 (d, $J = 7.3$ Hz, 4H), 7.44 (s, 2H), 7.37 (t, $J = 7.3$ Hz, 4H), 7.32 (d, $J = 16.2$ Hz, 2H), 7.27 (d, $J = 10.1$ Hz, 2H), 7.03 (d, $J = 15.9$ Hz, 2H), 2.44 (s, 6H) (δ 2.17 is the peak of acetone, mixed with CDCl_3) (**Figure S41**); $^{13}\text{C-NMR}$ (100 MHz, CDCl_3) δ 137.9, 135.7, 133.7, 129.6, 128.8, 127.7, 127.3, 126.7, 126.3, 19.7 (**Figure S42**).

(3) Synthesis of 1,4-bis((*E*)-2-phenylprop-1-en-1-yl)benzene (**PPB**)³

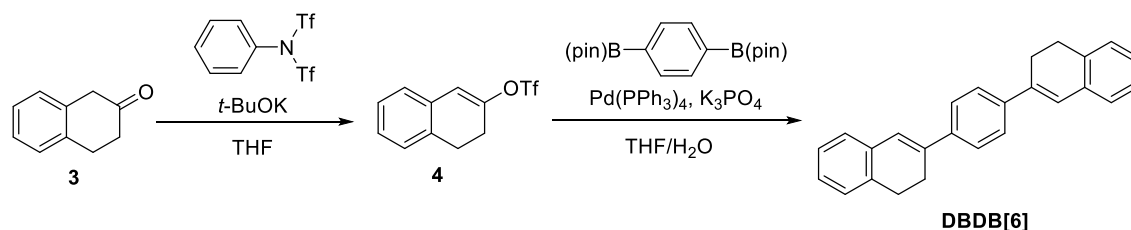


Scheme S3. Synthesis of **PPB**.

α -Methylstyrene (1.95 mL, 15.0 mmol), 1,4-diodobenzene (**1**) (1.65 g, 5.0 mmol), $\text{Pd}(\text{OAc})_2$ (0.02 g, 0.09 mmol) and tri-(*o*-tolyl)phosphine (0.06 g, 0.2 mmol) was added in triethylamine (4.16 mL) in thick glass tube sealed with screw cap and stirred at 130°C for 1 day. Water was added to the solution, and the mixture was extracted with dichloromethane. The organic layer was washed with water three times, dried over MgSO_4 , filtered, and evaporated under reduced pressure to give a residue. The residue was chromatographed over silica gel, eluting with 3/1 (v/v) hexane/dichloromethane, and recrystallized from dichloromethane to give **PPB** as a pale green solid. Yield: 25%; Mp: 172.9°C; $^1\text{H-NMR}$ (399 MHz, CDCl_3) δ 7.55-7.53 (m, 4H), 7.40-7.37 (m, 8H), 7.30 (t, $J = 7.3$ Hz, 2H), 6.85 (d, $J = 0.9$ Hz, 2H), 2.34 (s, 6H) (**Figure S43**); $^{13}\text{C-NMR}$ (100 MHz, CDCl_3) δ 144.2, 137.6, 136.7, 129.2, 128.5, 127.7, 127.3, 126.2, 17.8 (**Figure S44**).

PPB included small amounts of contamination (see in **Figure S43**). Thus, we measured absorption and fluorescence spectra of **PPB** in 80% water suspensions (**Figure S10**) to compare with the reported result⁴. As a result, λ_{abs} was almost the same and fluorescence spectra had a similar trend to the reported result, and we determined that using this **PPB** was no problem.

(4) Synthesis of 1,4-bis(3,4-dihydronaphthalen-2-yl)benzene (**DBDB[6]**)



Scheme S4. Synthesis of **DBDB[6]**.

3,4-Dihydronaphthalen-2-yl trifluoromethanesulfonate (**4**)⁵

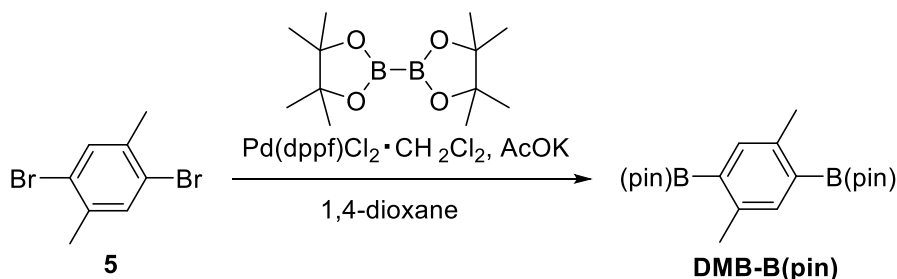
β-Tetralone (**3**, 1.5 g, 10 mmol) was dissolved in THF (30 mL) under an argon atmosphere. The reaction mixture was cooled to -20 °C, and potassium *tert*-butoxide (1.5 g, 13 mmol) was added to the mixture and stirred at 0 °C for 1 hour. The mixture was cooled to -20 °C, and *N*-phenylbis(trifluoromethanesulfonimide) (4.6 g, 13 mmol) was added to the mixture and stirred at -20 °C for 1 hour, and then stirred at 0 °C for 4 hours. Water was added, and the mixture was extracted with ethyl acetate. The organic layer was washed with water and saturated with aqueous NaHCO₃, dried over MgSO₄, filtered, and evaporated under reduced pressure to give a residue. The residue was chromatographed over silica gel, eluting with 4/1 (v/v) hexane/ethyl acetate to give crude **4** as a colorless oil. Yield: 95%; ¹H-NMR (399 MHz, CDCl₃) δ 7.21-7.19 (m, 2H), 7.16-7.13 (m, 1H), 7.10-7.06 (m, 1H), 6.48 (s, 1H), 3.06 (t, *J* = 8.2 Hz, 2H), 2.69 (t, *J* = 8.5 Hz, 2H) (**Figure S45**).

1,4-Bis(3,4-dihydronaphthalen-2-yl)benzene (**DBDB[6]**)⁶

A mixture of **4** (1.2 g, 4.0 mmol), 1,4-benzenediboronic acid bis(pinacolate) ester (0.60 g, 1.8 mmol), K₃PO₄ (2.6 g, 12 mmol), and Pd(PPh₃)₄ (0.23 g, 0.20 mmol) was dissolved in 5/1 (v/v) THF/water (20 mL) under argon atmosphere. The reaction mixture was refluxed at 50 °C for 2 hours and then cooled to room temperature. Water was added, and the mixture was extracted with dichloromethane. The organic layer was washed with water three times, dried over MgSO₄, filtered, and evaporated under reduced pressure to give a residue. The residue was recrystallized from 1/4 (v/v) hexane/dichloromethane to give **DBDB[6]** as a yellowish solid. Yield: 68%; Mp: 243.2°C; ¹H-NMR (399 MHz, CDCl₃) δ 7.57 (s, 4H), 7.22-7.13 (m, 8H), 6.91 (s, 2H), 2.97 (t, *J* = 8.2 Hz, 4H), 2.78 (t, *J* = 8.2 Hz, 4H) (**Figure S46**); ¹³C-NMR (100 MHz, CDCl₃) δ 140.1, 138.3, 135.0, 134.9, 127.4, 127.1, 126.8, 125.3, 124.3, 28.3, 26.3 (**Figure S47**); HRMS (EI) Calcd for C₂₆H₂₂: 334.1722, Found 344.1723 (**Figure S72**).

(5) Synthesis of 2,2'-(2,5-dimethyl-1,4-phenylene)bis(4,4,5,5-tetramethyl-1,3,2-

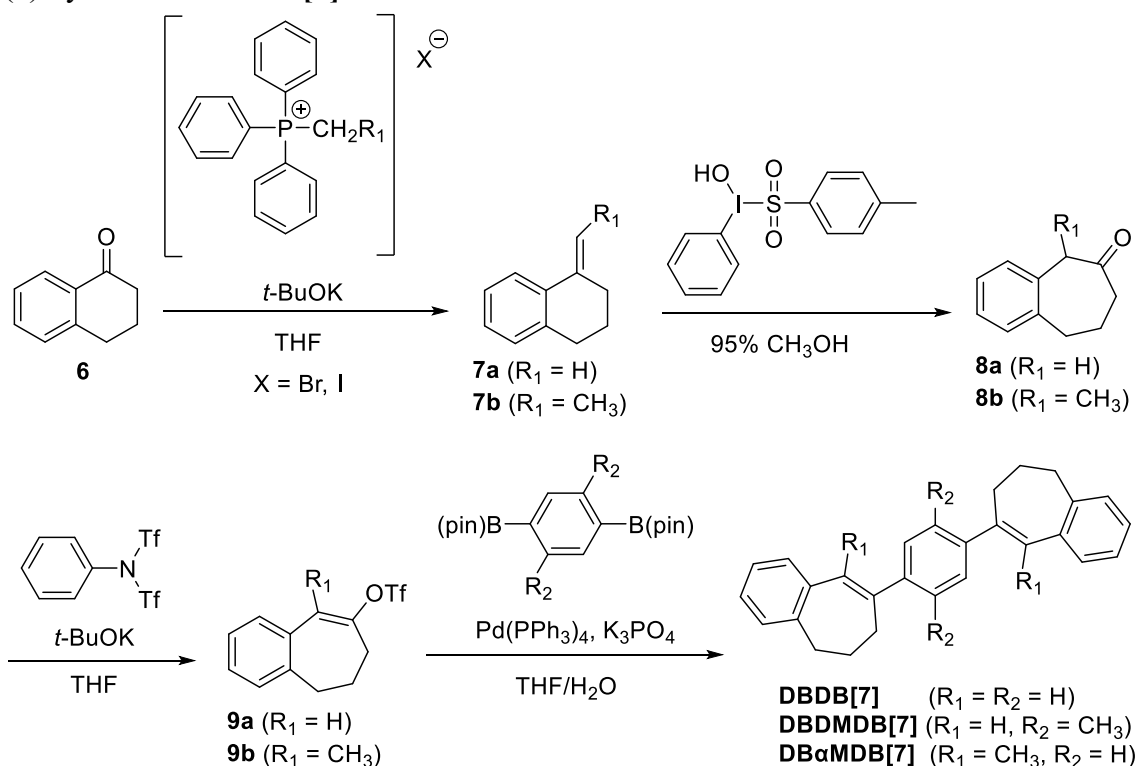
dioxaborolane) (**DMB-B(pin)**)⁷



Scheme S5. Synthesis of **DMB-B(pin)**.

A mixture of 2,5-dibromo-*p*-xylene (**5**) (1.32 g, 5.0 mmol), bis(pinacolate)diboron (3.26 g, 10.0 mmol), potassium acetate (1.47 g, 15.0 mmol) and $\text{Pd(dppf)Cl}_2 \cdot \text{CH}_2\text{Cl}_2$ (0.12 g, 0.15 mmol) was dissolved in 1,4-dioxane (35 mL) under argon atmosphere. The reaction mixture was refluxed at 80 °C overnight. Water was added, and the mixture was extracted with dichloromethane. The organic layer was washed with water three times, dried over MgSO_4 , filtered, and evaporated under reduced pressure to give a residue. The residue was chromatographed over silica gel, eluting with 1/2 (v/v) hexane/dichloromethane, and recrystallized from 3/1 (v/v) hexane/dichloromethane to give **DMB-B(pin)** as a colorless solid. Yield: 45%; $^1\text{H NMR}$ (399 MHz, CDCl_3) δ 7.53 (s, 2H), 2.48 (s, 6H), 1.34 (s, 24H) (**Figure S48**).

(6) Synthesis of **DBDB[7]s**



Scheme S6. Synthesis of **DBDB[7]**, **DBDMDB[7]**, and **DB α MDB[7]**.

1-Methylene-1,2,3,4-tetrahydronaphthalene (**7a**)⁸

Methyltriphenylphosphonium bromide (8.0 g, 22.3 mmol) was dissolved in THF (50 mL) under an argon atmosphere and stirred at 0 °C. Potassium *tert*-butoxide (1.7 g, 22.1 mmol), and α -tetralone (**6**, 1.99 mL, 15.0 mmol) were added to the mixture and stirred at room temperature for 1 hour. Saturated aqueous NH₄Cl was added, and the mixture was extracted with ethyl acetate. The organic layer was washed with water three times, dried over MgSO₄, filtered, and evaporated under reduced pressure to give a residue. The residue was chromatographed over silica gel, eluting with 3/1 (v/v) hexane/dichloromethane to give crude **7a** as colorless oil. Yield: 85%; ¹H-NMR (500 MHz, CDCl₃) δ 7.65 (d, J = 7.3 Hz, 1H), 7.18-7.13 (m, 2H), 7.10 (d, J = 7.3 Hz, 1H), 5.48 (s, 1H), 4.95 (s, 1H), 2.85 (t, J = 6.1 Hz, 2H), 2.55 (t, J = 5.8 Hz, 2H), 1.91-1.86 (m, 2H) (**Figure S49**).

1-Ethylidene-1,2,3,4-tetrahydronaphthalene (**7b**)

Following a similar procedure used for the synthesis of **7b** from ethyltriphenylphosphonium Iodide (6.16g, 14.7 mmol) and α -tetralone (**6**, 1.33 mL, 10.0 mmol) stirred at room temperature for 4 hours, chromatography over silica gel, eluting with 4/1 (v/v) hexane/dichloromethane gave crude **7b** as a colorless oil. Yield: 97%; ¹H-NMR (399 MHz, CDCl₃) δ 7.57-7.55 (m, 1H), 7.42 (dt, J = 8.5, 3.4 Hz, 0H), 7.18-7.07 (m, 3H), 6.14-6.08 (m, 1H), 5.57 (q, J = 7.3 Hz, 0H), 2.83 (t, J = 6.6 Hz, 1H), 2.76 (t, J = 6.2 Hz, 1H), 2.50 (t, J = 6.4 Hz, 1H), 2.43-2.39 (m, 1H), 1.93-1.79 (m, 5H) (**Figure S50**) (**7b** was a mixture of *E* and *Z* configuration).

5,7,8,9-Tetrahydro-6*H*-benzo[7]annulen-6-one (**8a**)⁹

7a (1.83 g, 12.7 mmol) was dissolved in 95% methanol (60 mL), and [hydroxy(tosyloxy)iodo]benzene (5.01 g, 12.8 mmol) was added. The solid dissolved evolution of heat rapidly to a yellow solution. The solution was stirred at room temperature for 20 min, and the solvent was removed in *vacuo*. Dichloromethane and water were added, and the mixture was extracted with dichloromethane. The organic layer was washed with water and brine three times, dried over MgSO₄, filtered, and evaporated under reduced pressure to give a residue. The residue was chromatographed over silica gel, eluting with 3/1 (v/v) hexane/ethyl acetate to give crude **8a** as a colorless oil. Yield: 84%; ¹H-NMR (500 MHz, CDCl₃) δ 7.22-7.14 (m, 4H), 3.73 (s, 2H), 2.95 (t, J = 6.3 Hz, 2H), 2.57 (t, J = 6.9 Hz, 2H), 2.02-1.97 (m, 2H) (**Figure S51**).

5-Methyl-5,7,8,9-tetrahydro-6*H*-benzo[7]annulen-6-one (**8b**)

Following a similar procedure used for the synthesis of **8a** from **7b** (1.36 g, 8.6 mmol), chromatography over silica gel, eluting with 3/1 (v/v) hexane/ethyl acetate gave crude **8b** as a colorless oil. Yield: 69%; ¹H-NMR (399 MHz, CDCl₃) δ 7.24-7.18 (m, 3H), 7.16-7.10 (m, 1H), 3.88 (q, *J* = 7.2 Hz, 1H), 2.97 (qd, *J* = 7.3, 4.6 Hz, 1H), 2.87-2.80 (m, 1H), 2.68-2.63 (m, 1H), 2.48 (qd, *J* = 5.9, 4.2 Hz, 1H), 2.09-2.04 (m, 1H), 1.98-1.87 (m, 1H), 1.45 (d, *J* = 6.9 Hz, 3H) (**Figure S52**) (The compound was a mixture of ketone and acetal (11:5))

6,7-Dihydro-5*H*-benzo[7]annulen-8-yl trifluoromethanesulfonate (**9a**)

Following a similar procedure used for the synthesis of **4** from **8a** (1.70 g, 10.6 mmol), THF (30 mL), potassium *tert*-butoxide (1.58g, 14.1mmol), and *N*-phenylbis(trifluoromethanesulfonimide) (4.95 g, 14.1 mmol), chromatography over silica gel, eluting with 6/1 (v/v) hexane/ethyl acetate, gave crude **9a** as a colorless oil. Yield: 93%; ¹H-NMR (500 MHz, CDCl₃) δ 7.21-7.16 (m, 3H), 7.11 (t, *J* = 4.1 Hz, 1H), 6.59 (s, 1H), 2.89 (t, *J* = 4.9 Hz, 2H), 2.79 (t, *J* = 6.6 Hz, 2H), 2.02-1.97 (m, 2H) (**Figure S53**) (The compound was a mixture of **7a** and *N*-phenylbis(trifluoromethanesulfonimide) (5:1))

9-Methyl-6,7-dihydro-5*H*-benzo[7]annulen-8-yl trifluoromethanesulfonate (**9b**)

Following a similar procedure used for the synthesis of **4** from **8b** (1.16 g, the amount of ketone = 4.4 mmol), chromatography over silica gel, eluting with 6/1 (v/v) hexane/ethyl acetate, gave crude **9b** as a colorless oil. Yield: 41%; ¹H-NMR (399 MHz, CDCl₃) δ 7.32-7.28 (m, 2H), 7.24-7.21 (m, 2H), 2.69 (t, *J* = 6.6 Hz, 2H), 2.32-2.22 (m, 4H), 2.17 (s, 3H) (**Figure S54**).

1,4-Bis(6,7-dihydro-5*H*-benzo[7]annulen-8-yl)benzene (**DBDB[7]**)

Following a similar procedure used for the synthesis of **DBDB[6]** from **9a** (0.90 g, 3.1 mmol) refluxed at 60 °C overnight, chromatography over silica gel, eluting with 9/1 (v/v) hexane/dichloromethane, and recrystallization from 3/1 (v/v) hexane/dichloromethane gave **DBDB[7]** as a yellowish solid. Yield: 17%; Mp: 181.2 °C; ¹H-NMR (399 MHz, CDCl₃) δ 7.52 (s, 4H), 7.26-7.21 (m, 4H), 7.19 (d, *J* = 6.4 Hz, 2H), 7.17-7.13 (m, 2H), 6.86 (s, 2H), 2.83 (t, *J* = 6.2 Hz, 4H), 2.67 (t, *J* = 6.9 Hz, 4H), 2.27-2.21 (m, 4H) (**Figure S55**); ¹³C-NMR (100 MHz, CDCl₃) δ 142.9, 142.5, 141.4, 137.6, 130.5, 129.1, 128.5, 126.7, 126.2, 126.1, 34.4, 32.4, 30.9 (**Figure S56**); HRMS (EI) Calcd for C₂₈H₂₆: 362.2035, Found 362.2028 (**Figure S73**).

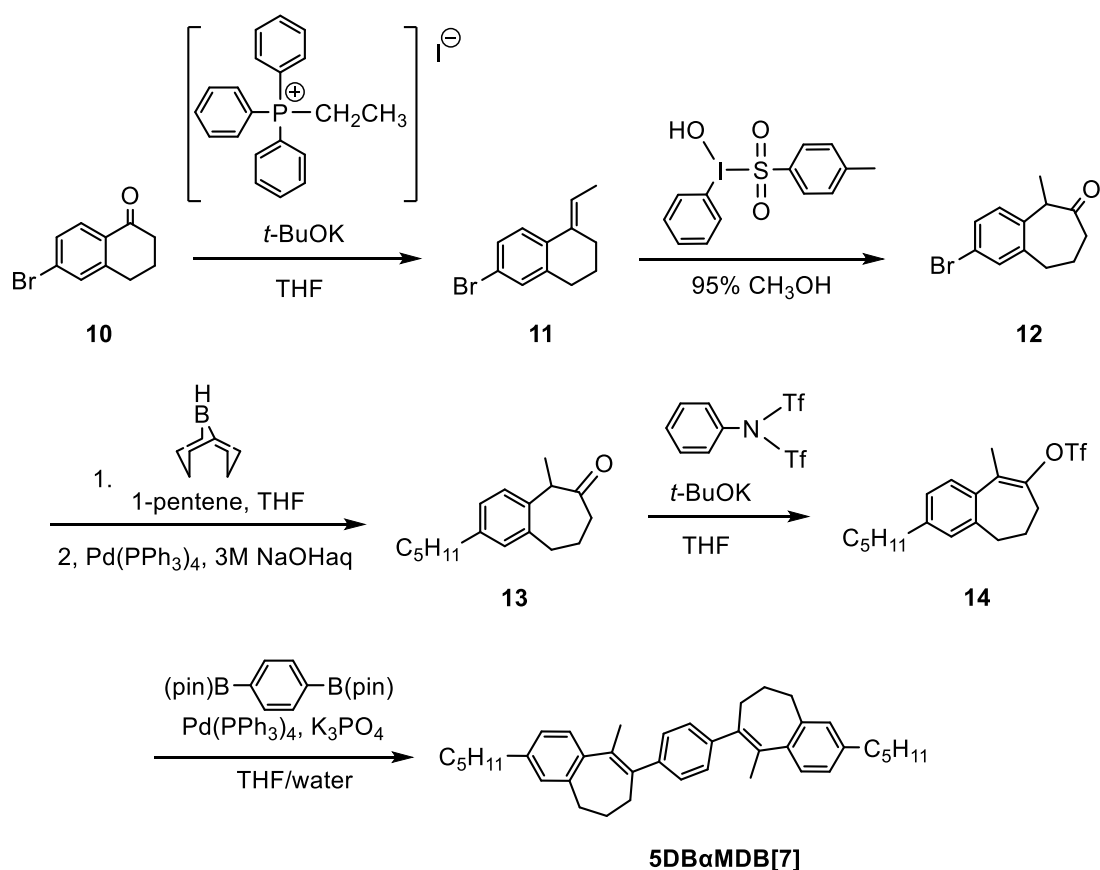
8,8'-(2,5-Dimethyl-1,4-phenylene)bis(6,7-dihydro-5*H*-benzo[7]annulene)

(DBDMDB[7])

Following a similar procedure used for the synthesis of **DBDB[6]** from **9a** (0.90 g, 3.1 mmol) and **DMB-B(pin)** (0.50 g, 1.4 mmol), refluxed at 50°C overnight, chromatography over silica gel, eluting with 9/1 (v/v) hexane/dichloromethane, and recrystallization from 2/1 (v/v) hexane/dichloromethane gave **DBDMDB[7]** as a colorless solid. Yield: 55%; Mp: 156.9 °C; ¹H-NMR (399 MHz, CDCl₃) δ 7.21-7.11 (m, 8H), 7.06 (s, 2H), 6.41 (s, 2H), 2.94-2.91 (m, 4H), 2.55 (t, *J* = 6.4 Hz, 4H), 2.34 (s, 6H), 2.20-2.13 (m, 4H) (**Figure S57**); ¹³C-NMR (100 MHz, CDCl₃) δ 144.4, 144.3, 141.4, 136.9, 131.9, 130.7, 130.3, 129.9, 129.2, 126.6, 126.1, 36.2, 35.4, 29.3, 19.6 (**Figure S58**); HRMS (EI) Calcd for C₃₀H₃₀: 390.2348, Found 390.2351 (**Figure S74**).

1,4-Bis(9-methyl-6,7-dihydro-5*H*-benzo[7]annulen-8-yl)benzene (DBαMDB[7])

Following a similar procedure used for the synthesis of **DBDB[6]** from **9b** (0.31 g, 1.0 mmol) refluxed at 50 °C for 1.5 hour, chromatography over silica gel, eluting with 9/1 (v/v) hexane/dichloromethane, and recrystallization from 3/1 (v/v) methanol/dichloromethane gave **DBαMDB[7]** as a colorless solid. Yield: 49%; Mp: 163.3 °C; ¹H-NMR (399 MHz, CDCl₃) δ 7.37 (dd, *J* = 7.8, 0.9 Hz, 2H), 7.31-7.27 (m, 6H), 7.23 (dd, *J* = 7.3, 1.8 Hz, 2H), 7.19 (td, *J* = 7.1, 1.4 Hz, 2H), 2.71 (t, *J* = 6.9 Hz, 4H), 2.20 (m, 8H), 2.10 (s, 6H) (**Figure S59**); ¹³C-NMR (100 MHz, CDCl₃) δ 143.6, 142.0, 140.3, 137.9, 131.8, 128.6, 128.3, 126.6, 126.2, 34.7, 32.9, 32.2, 20.1 (**Figure S60**); HRMS (EI) Calcd for C₃₀H₃₀: 390.2348, Found 390.2350 (**Figure S75**)



Scheme S7. Synthesis of **5DBaMDB[7]**.

6-Bromo-1-ethylidene-1,2,3,4-tetrahydronaphthalene (**11**)

Following a similar procedure used for the synthesis of **7a** from ethyltriphenylphosphonium iodide (6.15 g, 14.7 mmol) and 6-bromo-1-tetralone (**10**, 2.27 g, 10.1 mmol) stirred at room temperature for 4 hours, chromatography over silica gel, eluting with 3/1 (v/v) hexane/dichloromethane, gave crude **11** as colorless oil. Yield: 90%; ¹H-NMR (399 MHz, CDCl₃) δ 7.40 (d, *J* = 7.8 Hz, 1H), 7.28-7.24 (m, 1H), 7.22 (s, 1H), 6.09 (q, *J* = 7.0 Hz, 1H), 2.71 (t, *J* = 6.2 Hz, 2H), 2.47 (t, *J* = 6.4 Hz, 2H), 1.90-1.74 (m, 5H) (**Figure S61**) (**11** was a mixture of *E* and *Z* configuration).

2-Bromo-5-methyl-5,7,8,9-tetrahydro-6H-benzo[7]annulen-6-one (**12**)

Following a similar procedure used for the synthesis of **8a** from **11** (2.15 g, 9.1 mmol), chromatography over silica gel, eluting with 3/1 (v/v) hexane/ethyl acetate, gave crude **12** as colorless solid: Yield: 74%; ¹H-NMR (399 MHz, CDCl₃) δ 7.36 (dd, *J* = 8.2, 2.3 Hz, 1H), 7.30 (d, *J* = 1.8 Hz, 1H), 7.08 (d, *J* = 8.2 Hz, 1H), 3.82 (q, *J* = 7.0 Hz, 1H), 2.97-2.90 (m, 1H), 2.82-2.76 (m, 1H), 2.65 (td, *J* = 11.1, 7.2 Hz, 1H), 2.51-2.45 (m, 1H), 2.13-2.02 (m, 1H), 1.97-1.86 (m, 1H), 1.43 (d, *J* = 7.3 Hz, 3H) (**Figure S62**).

5-Methyl-2-pentyl-5,7,8,9-tetrahydro-6H-benzo[7]annulen-6-one (**13**)¹⁰

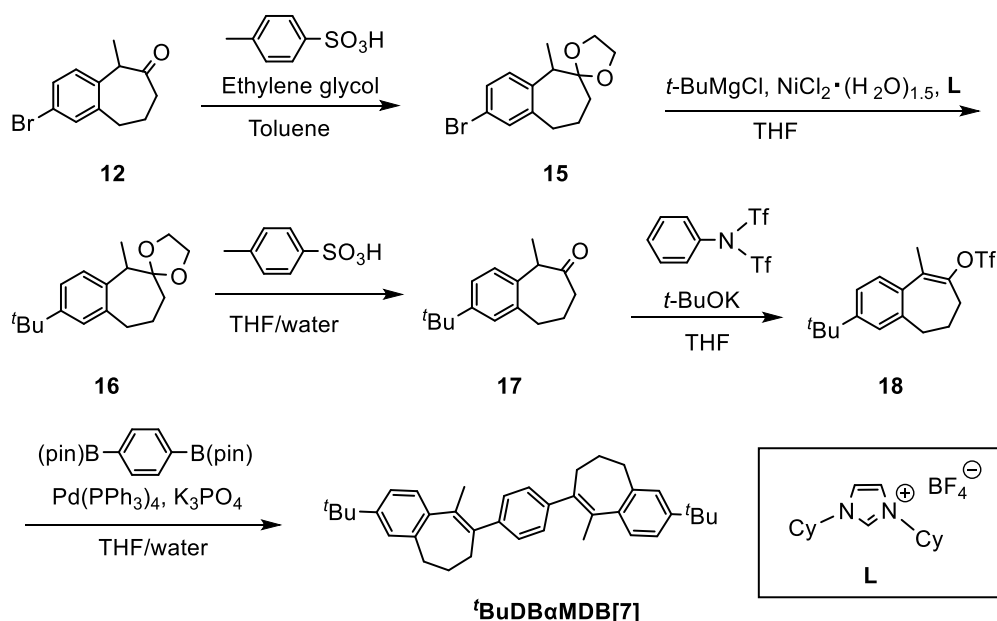
0.5 M 9-BBN in THF (9.8 mL) was stirred at 0°C under argon atmosphere. 1-Pentene (0.54 mL, 4.9 mmol) was added in the mixture and stirred room temperature for 4 hours. Then, 3M aqueous NaOH (3.0 mL), **12** (0.81 g, 3.2 mmol), and Pd(PPh₃)₄ (0.13 g, 0.1 mmol) was added in the mixture and refluxed overnight at 70°C. Water was added, and the mixture was extracted with dichloromethane. The organic layer was washed with water three times, dried over MgSO₄, filtered, and evaporated to give residue. The residue was chromatographed over silica gel, eluting with 6/1 (v/v) hexane/ethyl acetate to give crude **13** as yellow oil. Yield: 77%; ¹H-NMR (399 MHz, CDCl₃) δ 7.11 (d, *J* = 7.8 Hz, 1H), 7.06-7.04 (m, 1H), 6.95 (s, 1H), 3.86 (q, *J* = 7.2 Hz, 1H), 2.98-2.91 (m, 1H), 2.84-2.77 (m, 1H), 2.66 (td, *J* = 11.1, 7.2 Hz, 1H), 2.56 (t, *J* = 7.8 Hz, 2H), 2.50-2.45 (m, 1H), 2.13-2.02 (m, 1H), 1.96-1.85 (m, 1H), 1.64-1.57 (m, 2H), 1.44 (d, *J* = 7.3 Hz, 3H), 1.37-1.29 (m, 4H), 0.90 (t, *J* = 6.9 Hz, 3H) (**Figure S63**).

9-Methyl-3-pentyl-6,7-dihydro-5H-benzo[7]annulen-8-yl trifluoromethanesulfonate (**14**)

Following a similar procedure used for the synthesis of **4** from **13** (0.60 g, 2.5 mmol), chromatographed over silica gel, eluting with 6/1 (v/v) hexane/ethyl acetate, gave crude **14** as colorless oil. Yield: 63%; ¹H-NMR (399 MHz, CDCl₃) δ 7.21 (d, *J* = 7.8 Hz, 1H), 7.10 (d, *J* = 7.8 Hz, 1H), 7.03 (s, 1H), 2.66 (t, *J* = 6.6 Hz, 2H), 2.58 (t, *J* = 8.0 Hz, 2H), 2.28-2.24 (m, 4H), 2.14 (s, 3H), 1.66-1.58 (m, 2H), 1.35-1.32 (m, 4H), 0.90 (t, *J* = 6.9 Hz, 3H) (**Figure S64**).

1,4-Bis(9-methyl-3-pentyl-6,7-dihydro-5H-benzo[7]annulen-8-yl)benzene
(**5DBαMDB[7]**)

Following a similar procedure used for the synthesis of **DBDB[6]** from **14** (0.59 g, 1.6 mmol) refluxed at 50°C for 3 hours, chromatography over silica gel, eluting with 9/1 (v/v) hexane/dichloromethane, and recrystallization from 3/1 (v/v) methanol/dichloromethane gave **5DBαMDB[7]** as colorless solid. Yield: 67%; ¹H-NMR (399 MHz, CDCl₃) δ 7.25 (d, *J* = 9.6 Hz, 6H), 7.08 (dd, *J* = 8.0, 1.6 Hz, 2H), 7.02 (s, 2H), 2.65 (t, *J* = 6.6 Hz, 4H), 2.58 (t, *J* = 7.8 Hz, 4H), 2.21-2.13 (m, 8H), 2.05 (s, 6H), 1.67-1.60 (m, 4H), 1.37-1.32 (m, 8H), 0.89 (t, *J* = 6.6 Hz, 6H) (**Figure S65**); ¹³C-NMR (100 MHz, CDCl₃) δ 142.1, 141.4, 140.9, 140.2, 137.6, 131.9, 128.7, 128.3, 126.5, 126.1, 35.9, 34.8, 33.0, 32.3, 31.9, 31.4, 22.7, 20.2, 14.2 (**Figure 66**); HRMS (FAB) Calcd for C₄₀H₅₀: 530.3913, Found 530.3923 (**Figure S76**).



Scheme S8. Synthesis of **'Bu DBaMDB[7]**.

2-Bromo-5-methyl-5,7,8,9-tetrahydrospiro[benzo[7]annulene-6,2'-[1,3]dioxolane] (**15**)

12 (1.70 g, 6.7 mmol), *p*-toluenesulfonic acid monohydrate (0.46 g, 2.4 mmol) and ethylene glycol (15 ml) was dissolved in Toluene (100 ml) and refluxed at 130°C for 1 day. Saturated aqueous NaHCO_3 was added, and the mixture was extracted with ethyl acetate. The organic layer was washed with water three times, dried over MgSO_4 , filtered, and evaporated under reduced pressure to give residue. The residue was chromatographed over silica gel, eluting with 3/1 (v/v) hexane/ethyl acetate to give crude **15** as colorless solid. Yield: 60%; $^1\text{H-NMR}$ (399 MHz, CDCl_3) δ 7.26-7.24 (m, 1H), 7.21 (s, 1H), 7.00 (d, $J = 7.8$ Hz, 1H), 3.96-3.89 (m, 4H), 3.05 (q, $J = 7.3$ Hz, 1H), 2.94-2.87 (m, 1H), 2.75-2.70 (m, 1H), 2.07-2.00 (m, 1H), 1.84-1.66 (m, 3H), 1.35 (d, $J = 6.9$ Hz, 3H) (**Figure S67**)

$\text{NiCl}_2 \cdot (\text{H}_2\text{O})_{1.5}$ ¹¹

$\text{NiCl}_2 \cdot (\text{H}_2\text{O})_6$ (2.0 g, 8.4 mmol) was finely ground using a mortar and pestle and transferred to a 50 ml round bottom flask containing a stir chip. The flask was placed under high vacuum, and stirred at 100°C for 20 minutes, then 120°C for 10 minutes to give $\text{NiCl}_2 \cdot (\text{H}_2\text{O})_{1.5}$. For more information, see Ref[11].

2-(*tert*-Butyl)-5-methyl-5,7,8,9-tetrahydrospiro[benzo[7]annulene-6,2'-[1,3]dioxolane] (**16**)¹¹

15 (1.50 g, 5.0 mmol), $\text{NiCl}_2 \cdot (\text{H}_2\text{O})_{1.5}$ (0.08 g, 0.5 mmol) and 1,3-

dicyclohexylimidazolium tetrafluoroborate salt (0.18 g, 5.6 mmol) was added into 30 ml 2neck flask under argon atmosphere and stirred at -10°C. 1M *tert*-butylmagnesium chloride in THF (10 ml) was added in the mixture, and stirred at -10°C for 1.5 hours. Ice and saturated aqueous NH₄Cl was added, and the mixture was extracted with ethyl acetate. The organic layer was washed with water three times, dried over MgSO₄, filtered, and evaporated under reduced pressure to give residue. The residue was chromatographed over silica gel, eluting with 4/1 (v/v) hexane/ethyl acetate, but the resulting compound remained much unreacted **15**, so the same reaction was performed again. The resulting compound (1.25 g, 10/7 (mol/mol) Br/^tBu), **15** (0.30 g, 1.0 mmol), NiCl₂ · (H₂O)_{1.5} (0.05 g, 0.3 mmol), 1,3-dicyclohexylimidazolium tetrafluoroborate salt (0.10 g, 0.3 mmol) and 1M *tert*-butylmagnesium chloride in THF (6 ml) were used. Crude **16** as a colorless solid was obtained. Yield: 40%; ¹H-NMR (500 MHz, CDCl₃) δ 7.14 (dd, *J* = 7.9, 1.8 Hz, 1H), 7.05 (dd, *J* = 5.0, 2.9 Hz, 2H), 4.00-3.89 (m, 4H), 3.07 (q, *J* = 7.3 Hz, 1H), 2.96-2.88 (m, 1H), 2.80-2.75 (m, 1H), 2.07-2.01 (m, 1H), 1.83-1.68 (m, 3H), 1.36 (d, *J* = 7.6 Hz, 3H), 1.29 (s, 9H) (some amounts of **15** and isomerization product were remained) (**Figure S68**)

2-(*tert*-Butyl)-5-methyl-5,7,8,9-tetrahydrospiro[benzo[7]annulene-6,2'-[1,3]dioxolane] (**17**)

16 (0.66 g, 2.4 mmol) and *p*-toluenesulfonic acid monohydrate (0.2 g, 1.0 mmol) were dissolved in 3/1 (v/v) THF/water (50 ml) and refluxed at 50°C overnight. Saturated aqueous NaHCO₃ was added, and the mixture was extracted with dichloromethane. The organic layer was washed with water three times, dried over MgSO₄, filtered, and evaporated under reduced pressure to give crude **17** as a colorless solid. We continued to the next reaction step.

3-(*tert*-Butyl)-9-methyl-6,7-dihydro-5H-benzo[7]annulen-8-yl trifluoromethanesulfonate (**18**)

Following a similar procedure used for the synthesis of **4** from **17** (0.57 g), chromatography over silica gel, eluting with 6/1 (v/v) hexane/ethyl acetate, gave crude **18** as colorless oil. Yield: 78%; ¹H-NMR (500 MHz, CDCl₃) δ 7.30 (dd, *J* = 8.1, 2.0 Hz, 1H), 7.23 (d, *J* = 8.2 Hz, 1H), 7.22 (d, *J* = 2.1 Hz, 1H), 2.68 (t, *J* = 6.7 Hz, 2H), 2.29-2.24 (m, 4H), 2.15 (s, 3H), 1.33 (s, 9H) (**Figure S69**)

1,4-Bis(3-(*tert*-butyl)-9-methyl-6,7-dihydro-5H-benzo[7]annulen-8-yl)benzene (**BuDBaMDB[7]**)

Following a similar procedure used for the synthesis of **DBDB[6]** from **18** (0.68 g, 1.9 mmol) refluxed at 60°C for 2 hours, chromatography over silica gel, eluting with 6/1 (v/v)

hexane/dichloromethane, recrystallization from 3/1 (v/v) hexane/dichloromethane, and purifying by recycling HPLC gave **'BuDB α MDB[7]** as a colorless solid. Yield: 5.7%; $^1\text{H-NMR}$ (500 MHz, CDCl_3) δ 7.30 (s, 4H), 7.28 (s, 4H), 7.22 (s, 2H), 2.70 (t, $J = 6.7$ Hz, 4H), 2.20 (s, 8H), 2.08 (s, 6H), 1.36 (s, 18H) (**Figure S70**); $^{13}\text{C-NMR}$ (100 MHz, CDCl_3) δ 149.4, 142.1, 140.6, 139.9, 137.6, 131.8, 128.3, 126.2, 125.5, 123.0, 34.8, 34.6, 32.9, 32.6, 31.5, 20.0 (**Figure S71**); HRMS (FAB) Calcd for $\text{C}_{38}\text{H}_{46}$: 502.3600, Found 502.3603 (**Figure S77**).

4. Photophysical properties

(1) Absorption and fluorescence spectra in THF and in the solid states

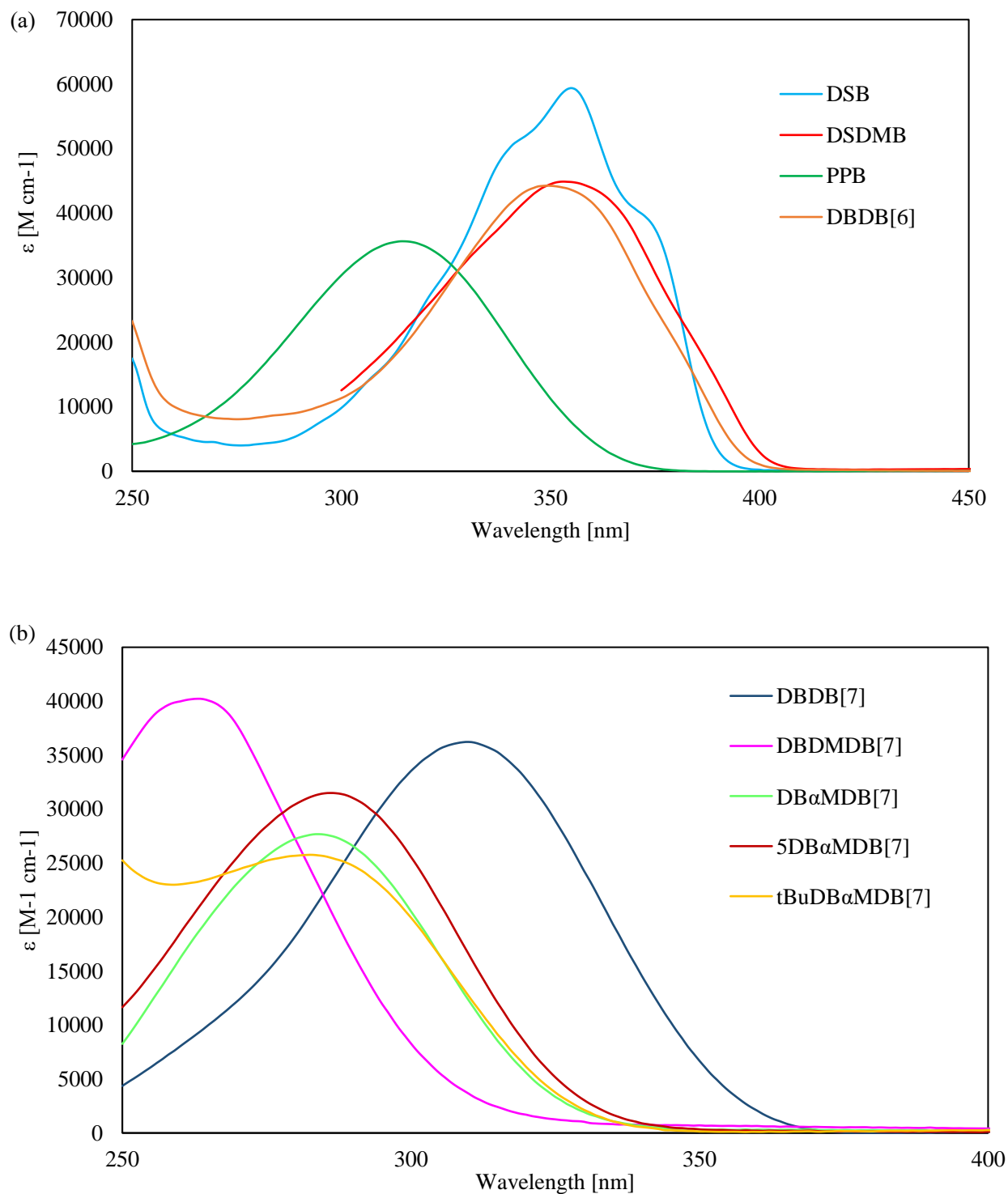


Figure S1. Absorption spectra of (a) DSBs (b) DBDB[7]s in THF.

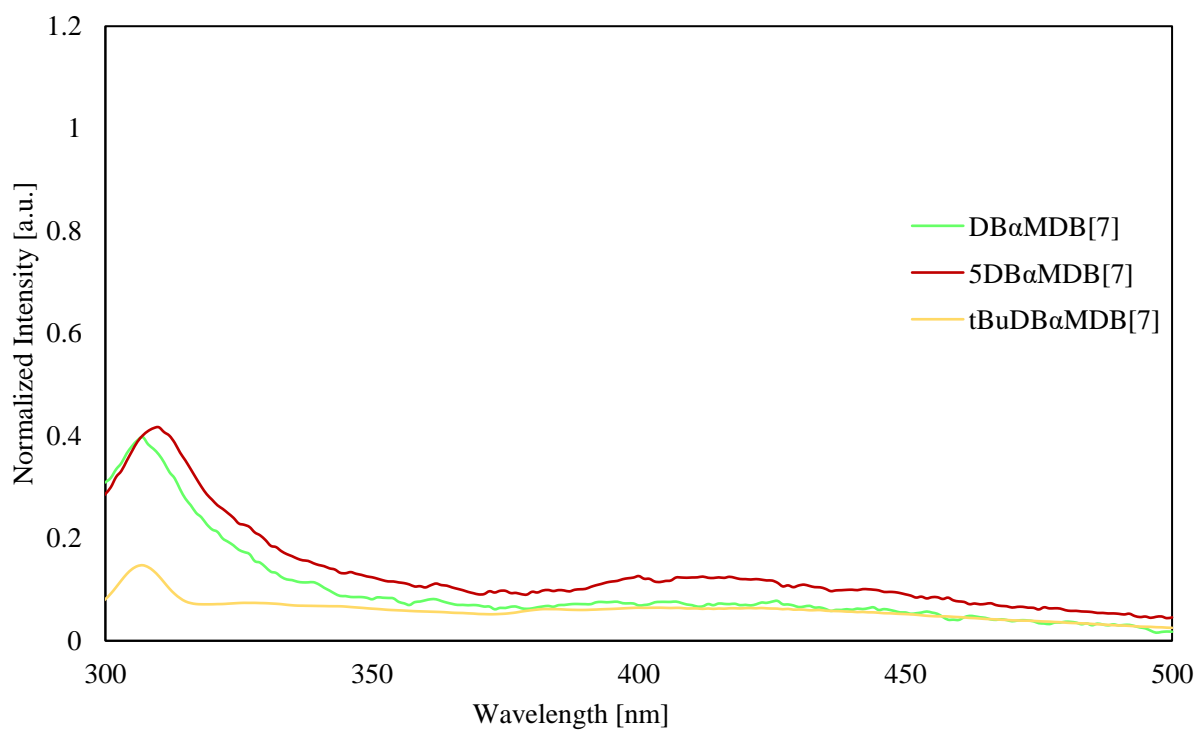
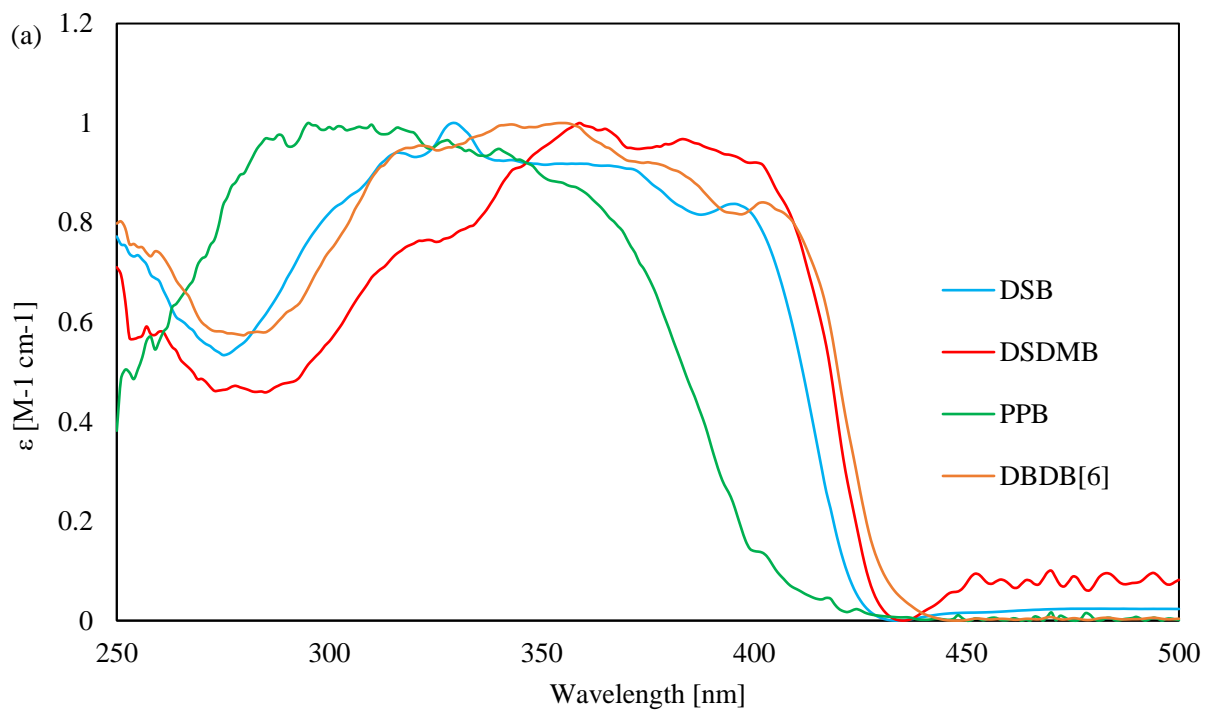


Figure S2. Fluorescence spectra of **DB α MDB[7]**s in THF (excitation wavelength of each compound is λ_{abs}). Small peaks around 310 nm were assigned to Raman scattering.



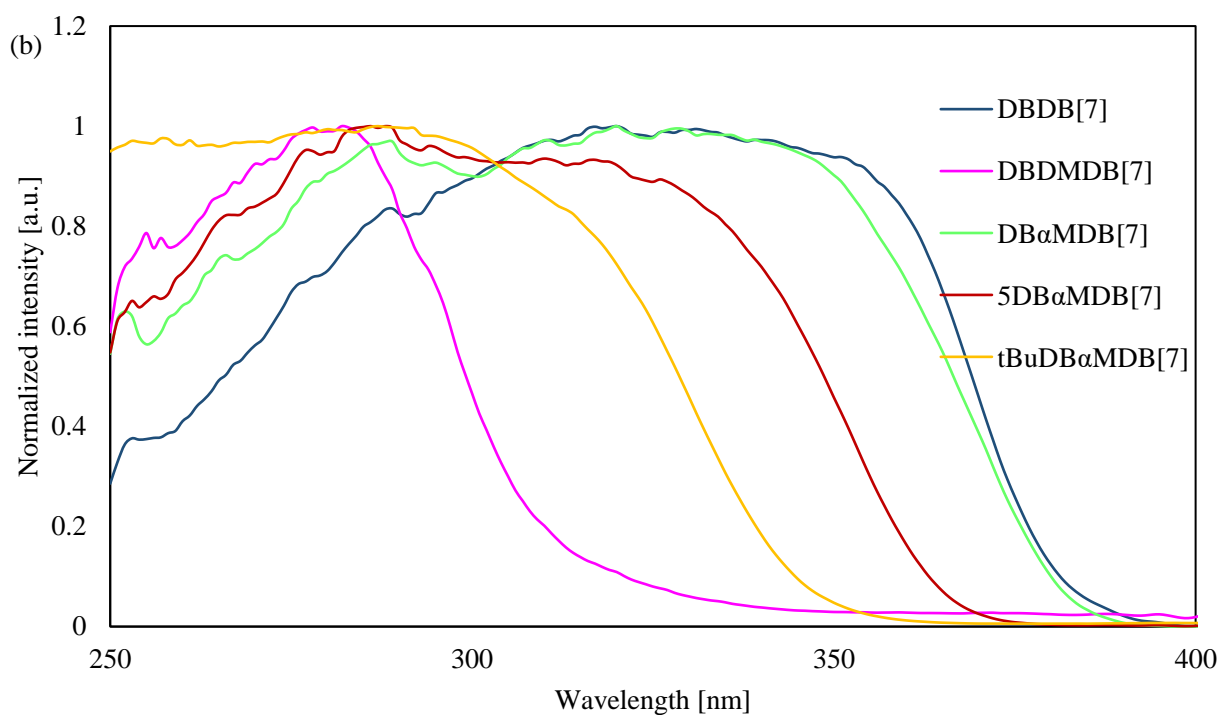


Figure S3. Diffuse-reflectance spectra of (a) **DSBs** (b) **DBDB[7]**s dispersed on NaBr powder.

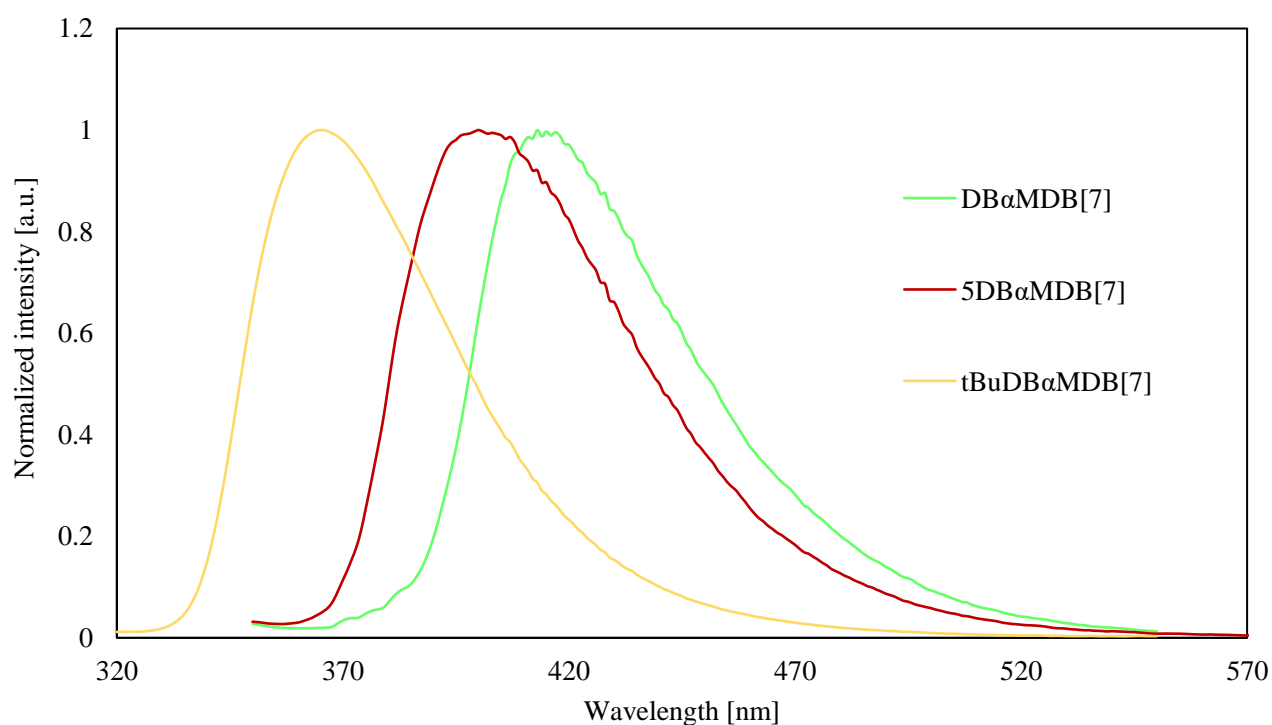


Figure S4. Fluorescence spectra of **DBαMDB[7]**s in the solid states (excitation wavelength of each compounds is λ_{abs}).

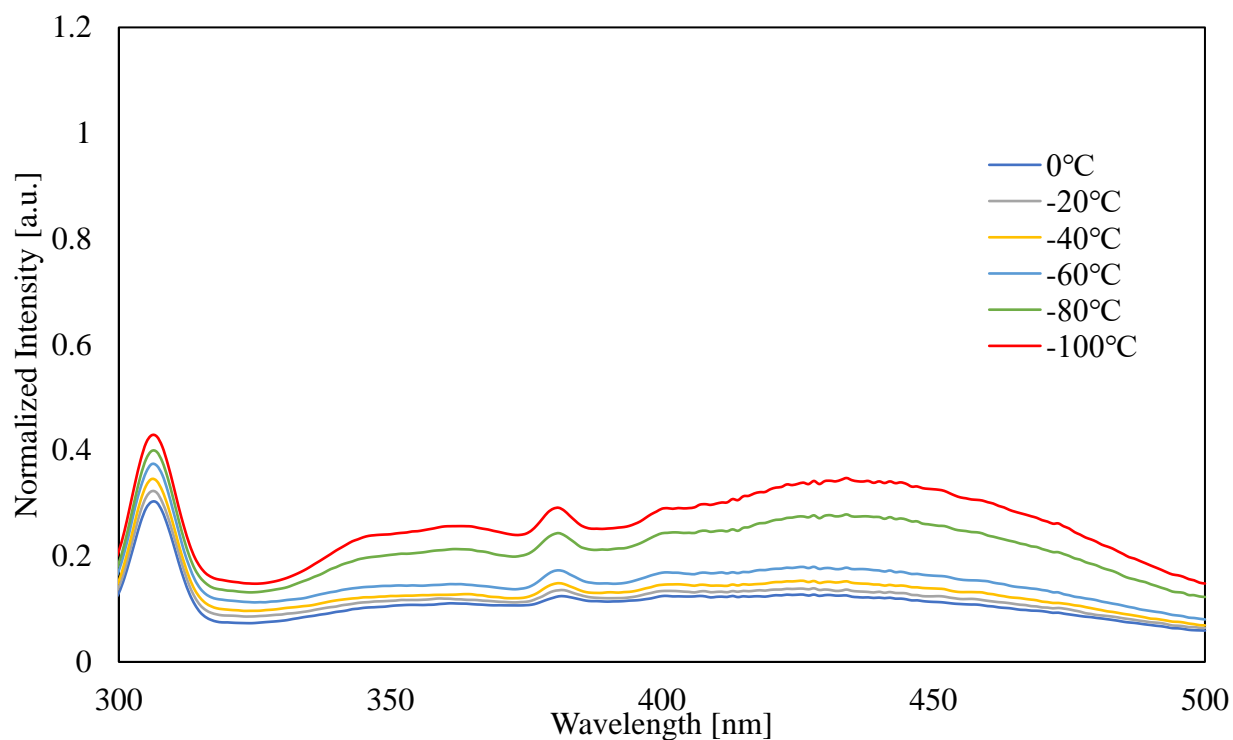


Figure S5. Fluorescence spectra of **DBaMDB[7]** in THF from 0°C to -100°C at 1.0×10^{-5} M (excitation wavelength is 284 nm). Small peaks around 310 nm were assigned to Raman scattering.

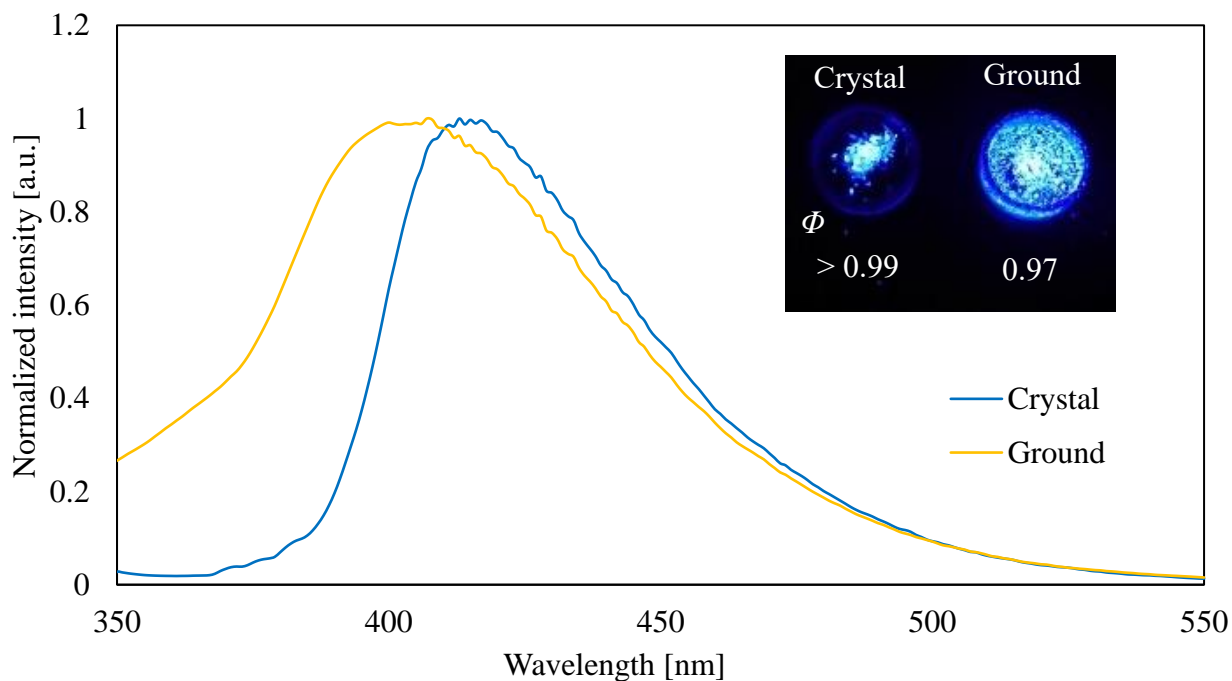


Figure S6. Fluorescence spectra (excitation at 330 nm) and photographs of crystal-**DBaMDB[7]** and ground-**DBaMDB[7]** (irradiation with 365 nm UV light).

(2) Lifetimes

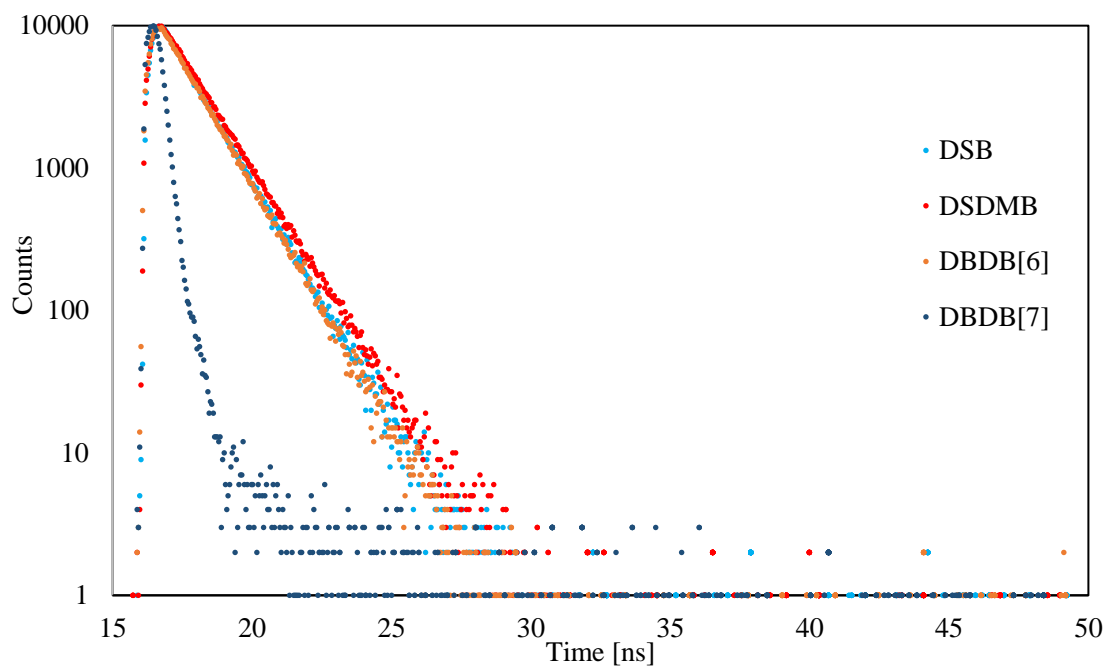


Figure S7. Time resolved fluorescence decay of **DSB**, **DSDMB**, **DBDB[6]**, and **DBDB[7]** in THF obtained on excitation at 379 nm and probing at the maxima of the fluorescence bands of each compounds, cutting wavelengths below 400 nm.

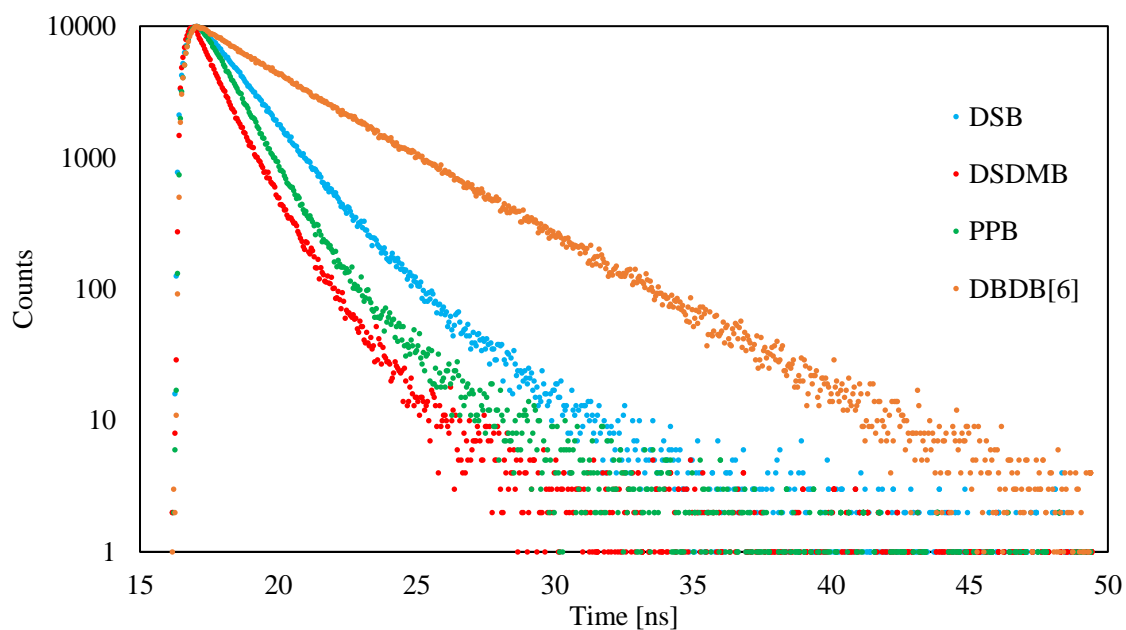


Figure S8. Time resolved fluorescence decay of **DSBs** in the solid states obtained on excitation at 379 nm and probing at the maxima of the fluorescence bands of each compounds, cutting wavelengths below 425 nm.

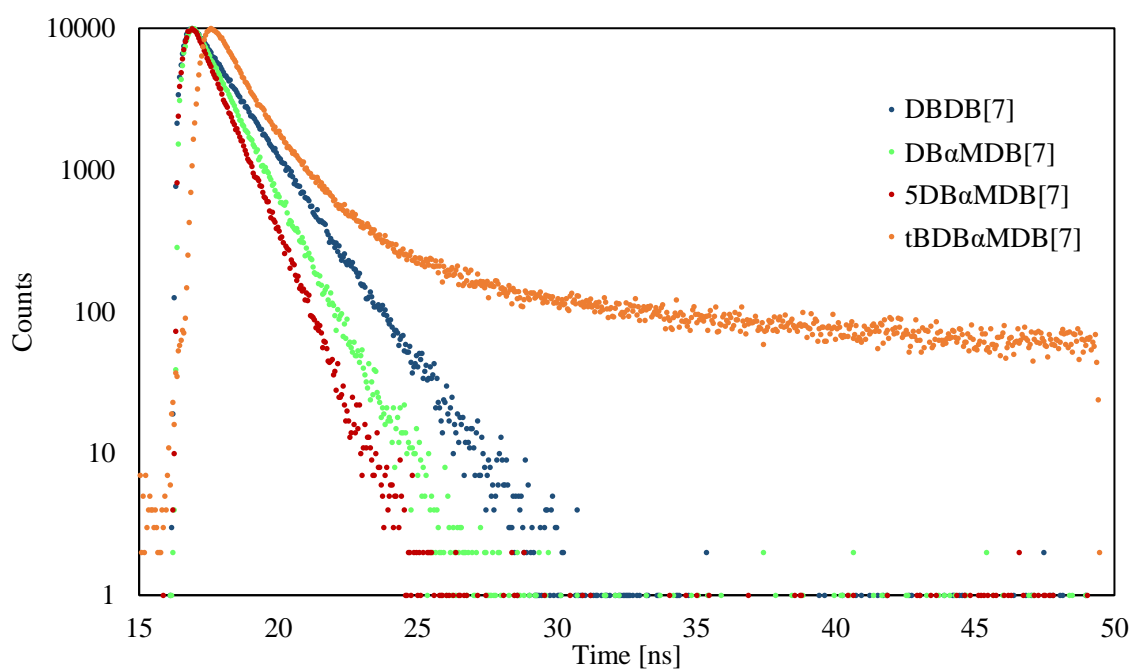


Figure S9. Time-resolved fluorescence decay of **DBDB[7]** and **DB α MDB[7]** in the solid states obtained on excitation at **DBDB[7]**, **DB α MDB[7]**, and **5DB α MDB[7]**: 379 nm and **tBuDB α MDB[7]**: 269 nm, and probing at **DBDB[7]**, **5DB α MDB[7]**, and **tBuDB α MDB[7]**: 410 nm and **DB α MDB[7]**: 413 nm, cutting wavelengths below 400 nm.

(3) Photophysical properties of aggregated states

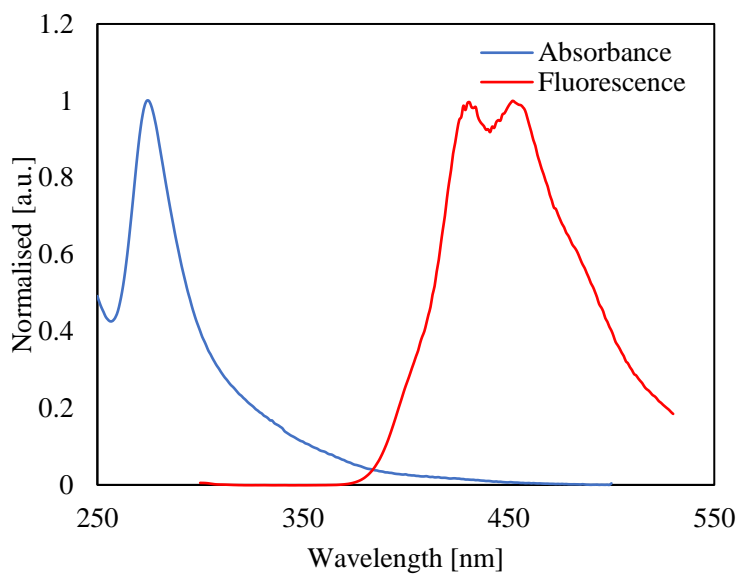


Figure S10. Absorption and fluorescence spectra of **PPB** in 2/8 (v/v) THF/water solution at 1.0×10^{-5} M.

Table S1. Photophysical properties of **DBDB[7]** in THF/water mixed solvent at 1.0×10^{-5} M.

Water Fraction (vol%)	0	10	20	30	40	50	60	70	80	90
λ_{abs} [nm]	321	320	320	320	321	321	321	322	307	320
ϵ [M cm^{-1}]	42000	40000	42000	41000	42000	42000	42000	12000	24000	38000
λ_{fl} [nm] ^a	411	413	413	413	413	413	413	392	392	420
Φ [-]	0.108	0.106	0.092	0.103	0.104	0.103	0.099	0.572	0.689	0.446

a) Excitation wavelengths correspond to the maxima of the absorption bands.

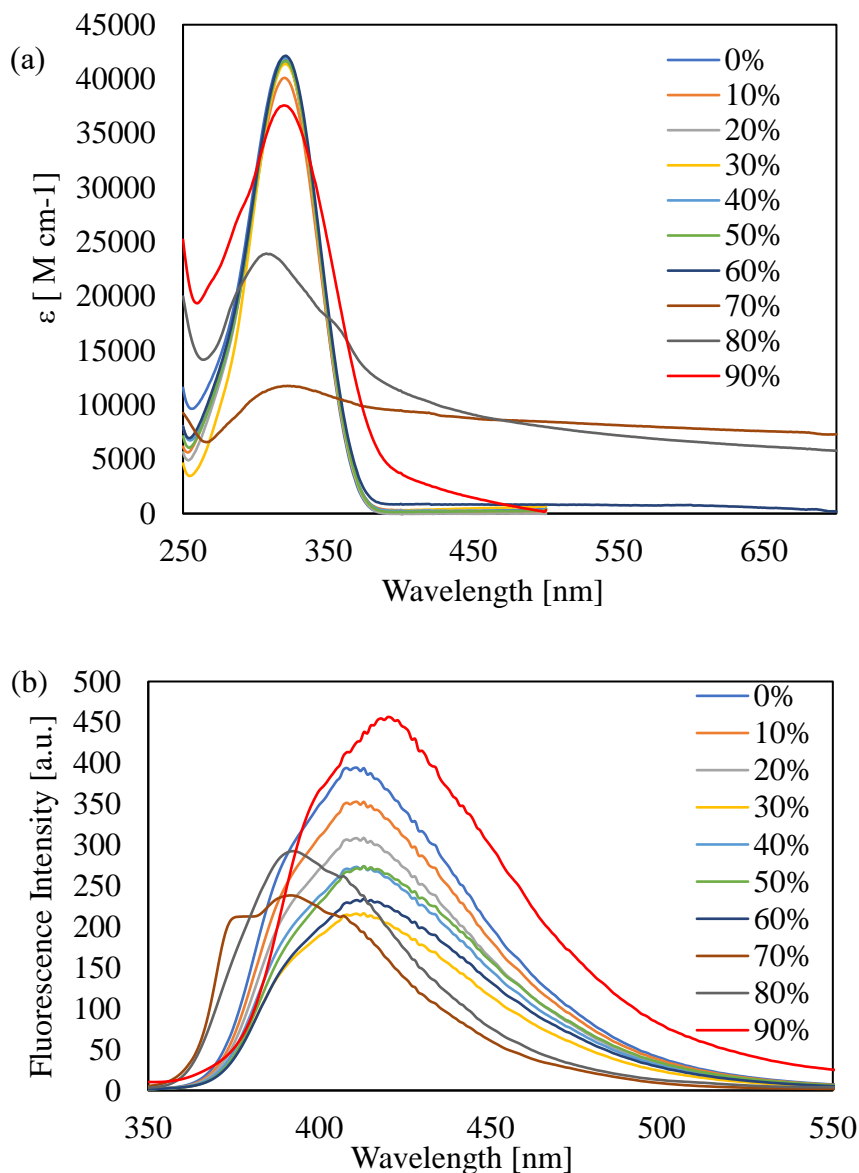


Figure S11. (a) Absorption and (b) fluorescence spectra of **DBDB[7]** in THF/water mixed solvent at 1.0×10^{-5} M.

Table S2. Photophysical properties of **DBDMDB[7]** in THF/water mixed solvent at 1.0×10^{-5} M.

Water Fraction (vol%)	0	10	20	30	40	50	60	70	80	90
λ_{abs} [nm]	274	275	275	275	275	274	275	275	282	278
ϵ [M cm ⁻¹]	39000	40000	55000	38000	39000	43000	37000	32000	31000	35000
λ_{fl} [nm] ^a	394	397	394	396	396	397	394	385	398	394
Φ [-]	0.014	0.009	0.007	0.013	0.009	0.015	0.014	0.021	0.023	0.062

a) Excitation wavelengths correspond to the maxima of the absorption bands.

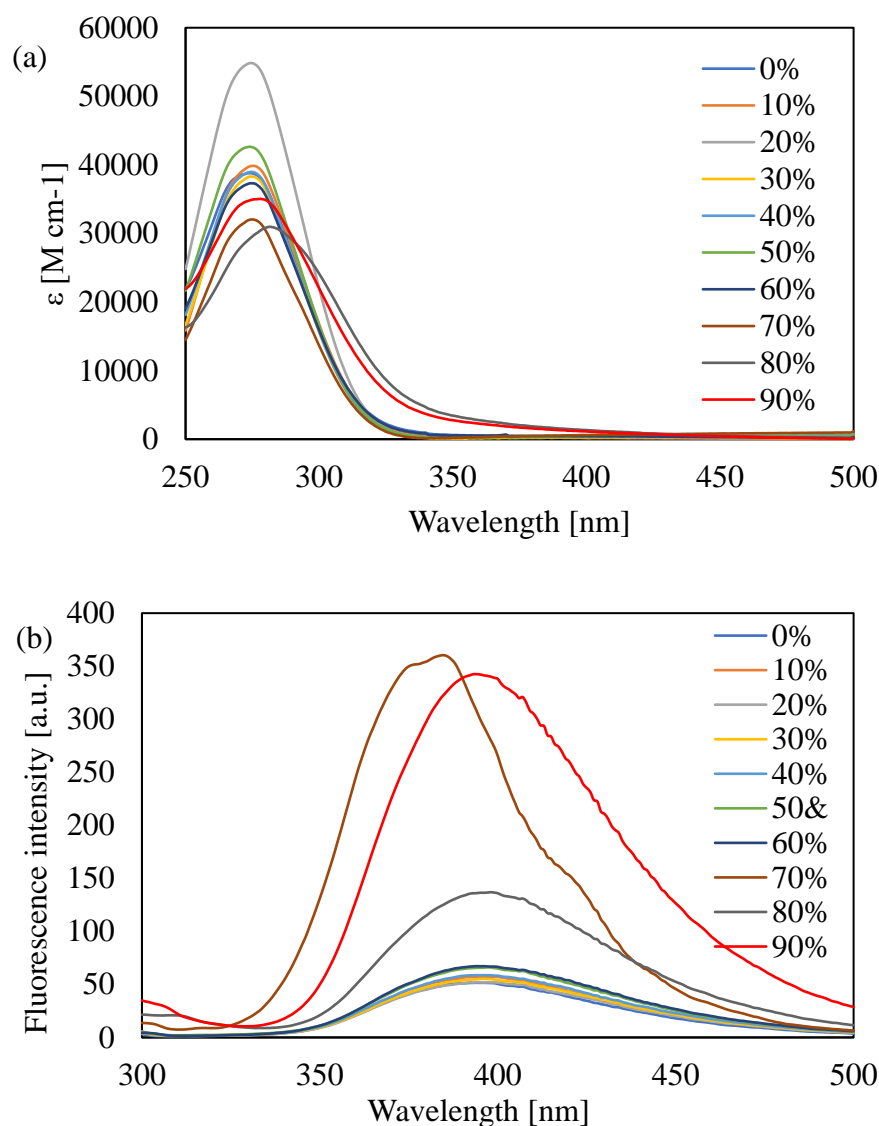


Figure S12. (a) Absorption and (b) fluorescence spectra of **DBDMDB[7]** in THF/water mixed solvent at 1.0×10^{-5} M.

Table S3. Photophysical properties of **DBaMDB[7]** in THF/water mixed solvent at 1.0×10^{-5} M.

Water Fraction (vol%)	0	10	20	30	40	50	60	70	80	90
λ_{abs} [nm]	284	283	284	285	283	284	284	286	295	286
ϵ [M cm^{-1}]	30000	30000	29000	27000	30000	30000	30000	25000	24000	26000
λ_{fl} [nm] ^a	-	-	-	-	-	-	-	359	428	423
Φ [-] ^a	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	<0.01	0.05	0.04	0.12

a) Excitation wavelengths correspond to the maxima of the absorption bands.

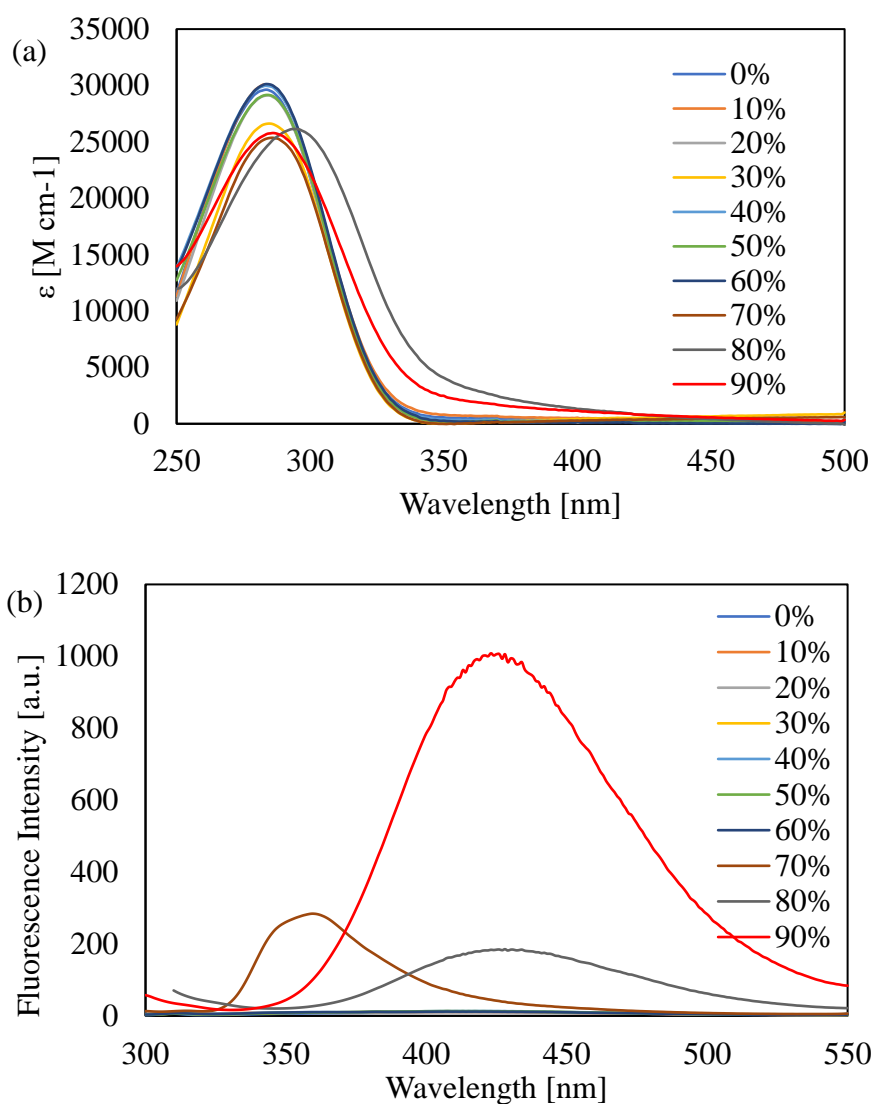


Figure S13. (a) Absorption and (b) fluorescence spectra of **DBaMDB[7]** in THF/water mixed solvent at 1.0×10^{-5} M.

(4) Solvatochromism of **DBDB[7]**

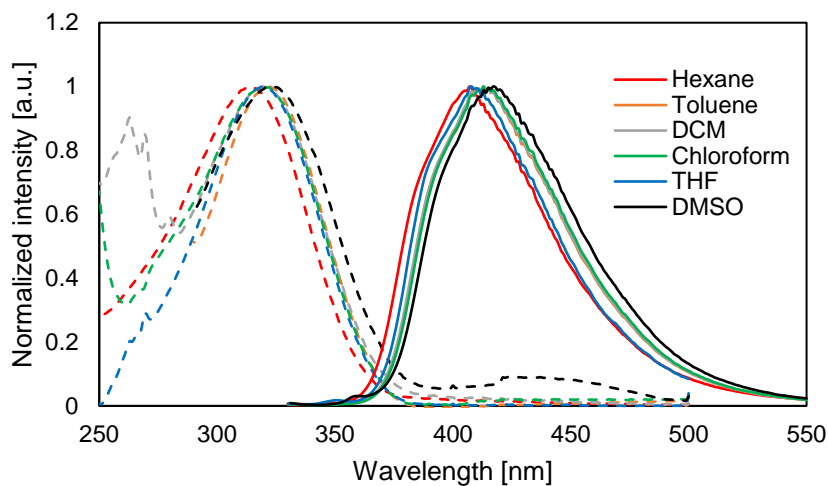


Figure S14. Absorption (dashed lines) and fluorescence (solid lines) spectra of **DBDB[7]** in hexane, toluene, dichloromethane (DCM), chloroform, THF, and dimethyl sulfoxide (DMSO) with ODs of approximately 0.1.

Table S4. Photophysical properties of **DBDB[7]** in various solvents.

Solvent	Hexane	Toluene	DCM	Chloroform	THF	DMSO
λ_{abs} [nm]	315	321	320	320	320	324
λ_{fl} [nm] ^a	407	413	413	413	408	417
Φ [-] ^a	0.08	0.11	0.09	0.11	0.11	0.14

a) Excitation wavelengths correspond to the maxima of the absorption bands.

(5) Photophysical properties of PMMA cast films of **DBDB[7]s**.

Methods of making cast films.

PMMA ($M_w = 100,000$) and **DBDB[7]s** (about 0.1 wt%) was dissolved in THF to reach a concentration of about 0.025 g/ml. The mixture was stirred at room temperature for 3 hours, then ultrasonicated for 20 seconds. The mixture was dropped on a quartz plate and dried at room temperature for 2 days to give a PMMA cast film with 0.1 wt% **DBDB[7]s**.

Table S5. Photophysical properties of the PMMA cast films doped with **DBDB[7]s**.

Dopant	λ_{abs} [nm]	λ_{fl} [nm] ^a	Φ [-]
DBDB[7]	309	407	0.55 ^b
DBDMDB[7]	272	384	0.13 ^c
DBαMDB[7]	292	407	0.24 ^d
5DBαMDB[7]	293	407	0.42 ^d
'BuDBαMDB[7]	278	408	0.48 ^d

Excitation wavelengths: a) corresponds to the maxima of the absorption bands.; b) 330 nm; c) 290 nm; d) 300 nm.

λ_{fl} of the cast film of **'BuDB α MDB[7]** is about 40 nm red-shifted compared to that in the solid states. Unlike in the solid state, **'BuDB α MDB[7]** in PMMA cast film may adopt almost the same conformation to **DB α MDB[7]-cross**, because λ_{fl} of PMMA cast films of **DB α MDB[7]s** are very similar.

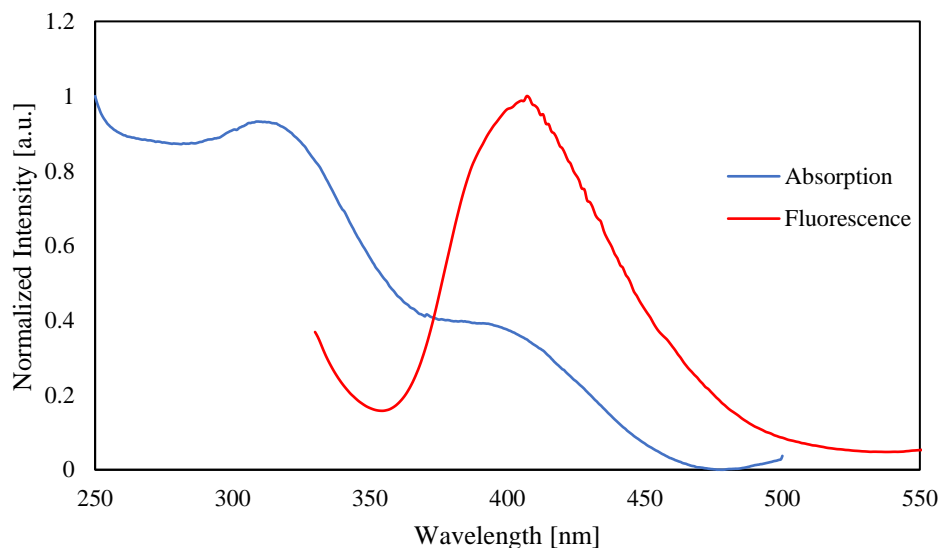


Figure S15. Absorption and fluorescence spectra of the **DBDB[7]**-doped cast film.

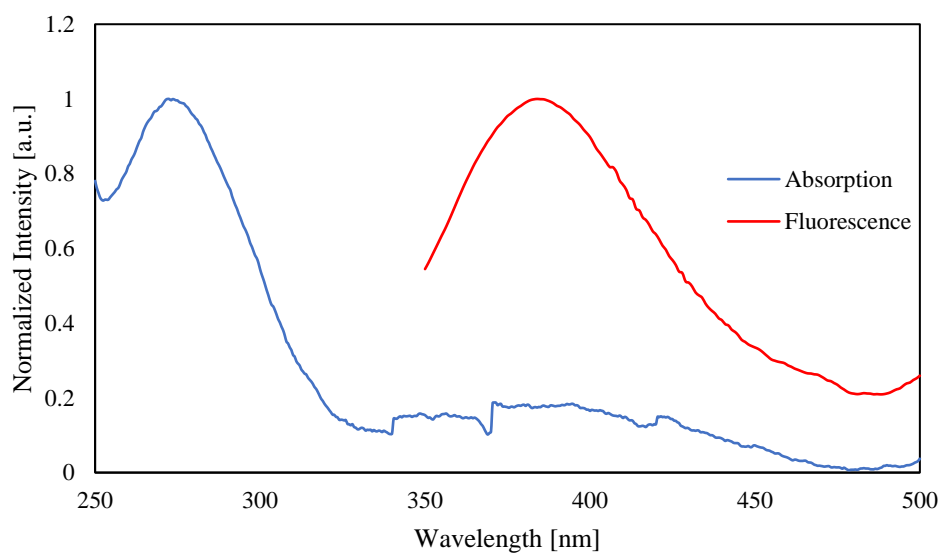


Figure S16. Absorption and fluorescence spectra of the **DBDMDB[7]**-doped cast film.

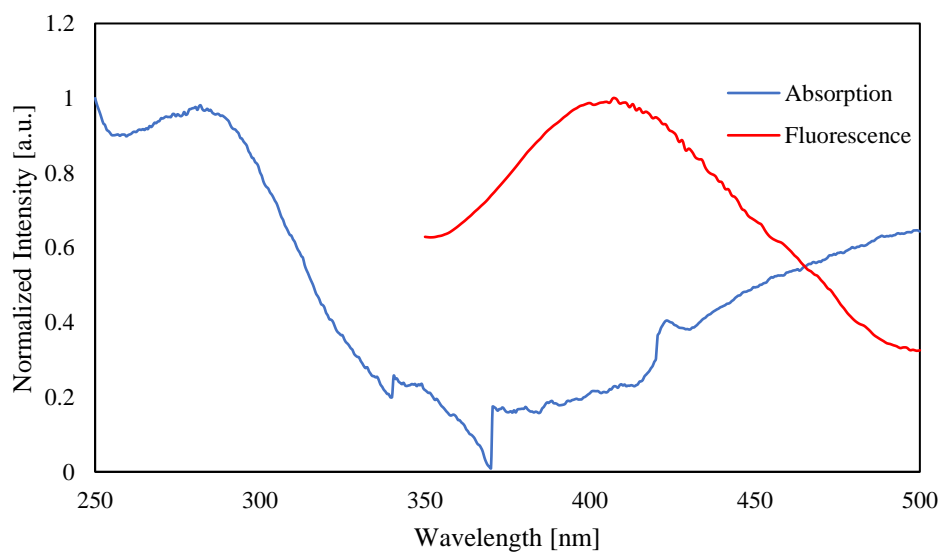


Figure S17. Absorption and fluorescence spectra of the **DBaMDB[7]**-doped cast film.

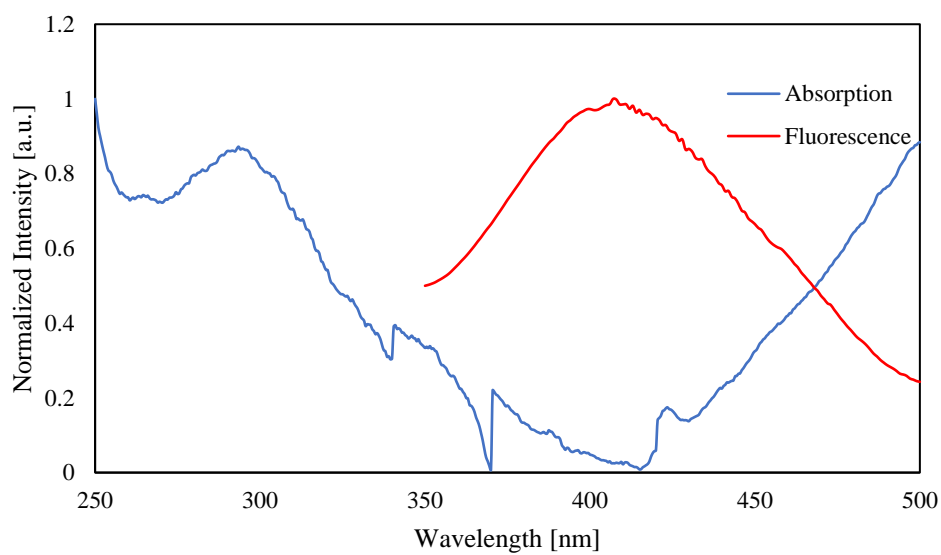


Figure S18. Absorption and fluorescence spectra of the **5DBaMDB[7]**-doped cast film.

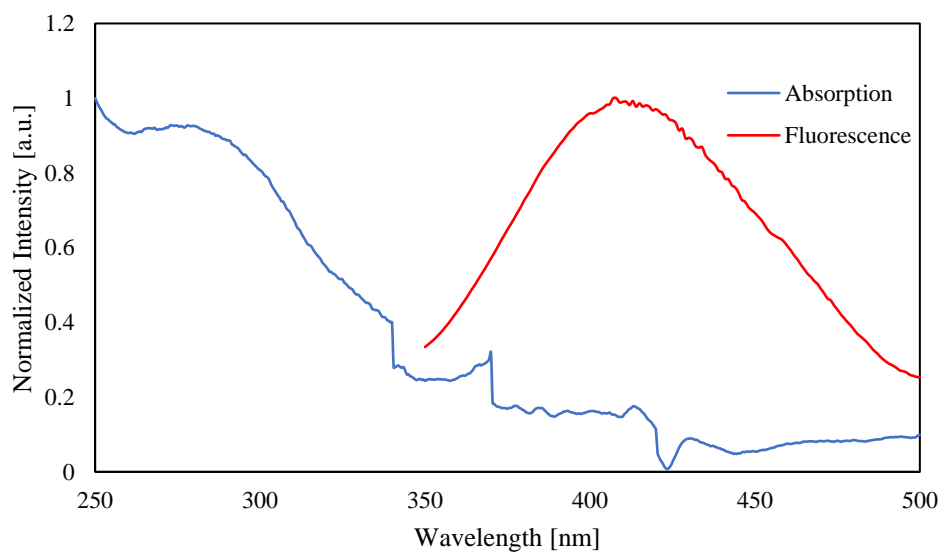


Figure S19. Absorption and fluorescence spectra of the **BuDBaMDB[7]**-doped cast film.

5. Physical properties

Table S6. Density and occupancy of **DSB**¹², **DSDMB**, **PPB**⁴, **DBDB[7]**, **DBDMDB[7]**, **DB α MDB[7]-cross**, and **DB α MDB[7]-parallel** in the single crystal.

Entry	Density (g/cm³)	Occupancy (%)
DSB	1.229	83.9
DSDMB	1.204	83.2
PPB	1.178	81.1
DBDB[7]	1.271	84.4
DBDB[7]	1.213	81.5
DBαMDB[7]-cross	1.243	83.1
DBαMDB[7]-parallel	1.210	81.0

Table S7. Melting points (T_m) of all compounds.

Entry	T_m (°C)
DSB	257.0
DSDMB	196.7
PPB	172.9
DBDB[6]	243.2
DBDB[7]	158.1
DBDMDB[7]	156.9
DBαMDB[7]	160.7
5DBαMDB[7]	125.6
^tBuDBαMDB[7]	311.3

6. The single-crystal x-ray structure

Although, the quality of single crystal X-ray crystallographic analysis of **DBDB[6]** is not high, we added **DBDB[6]** to the comparison of packing structures for more meaningful discussion of fluorescent property of crystallin distyrylbenzenes.

(1) Molecular structures

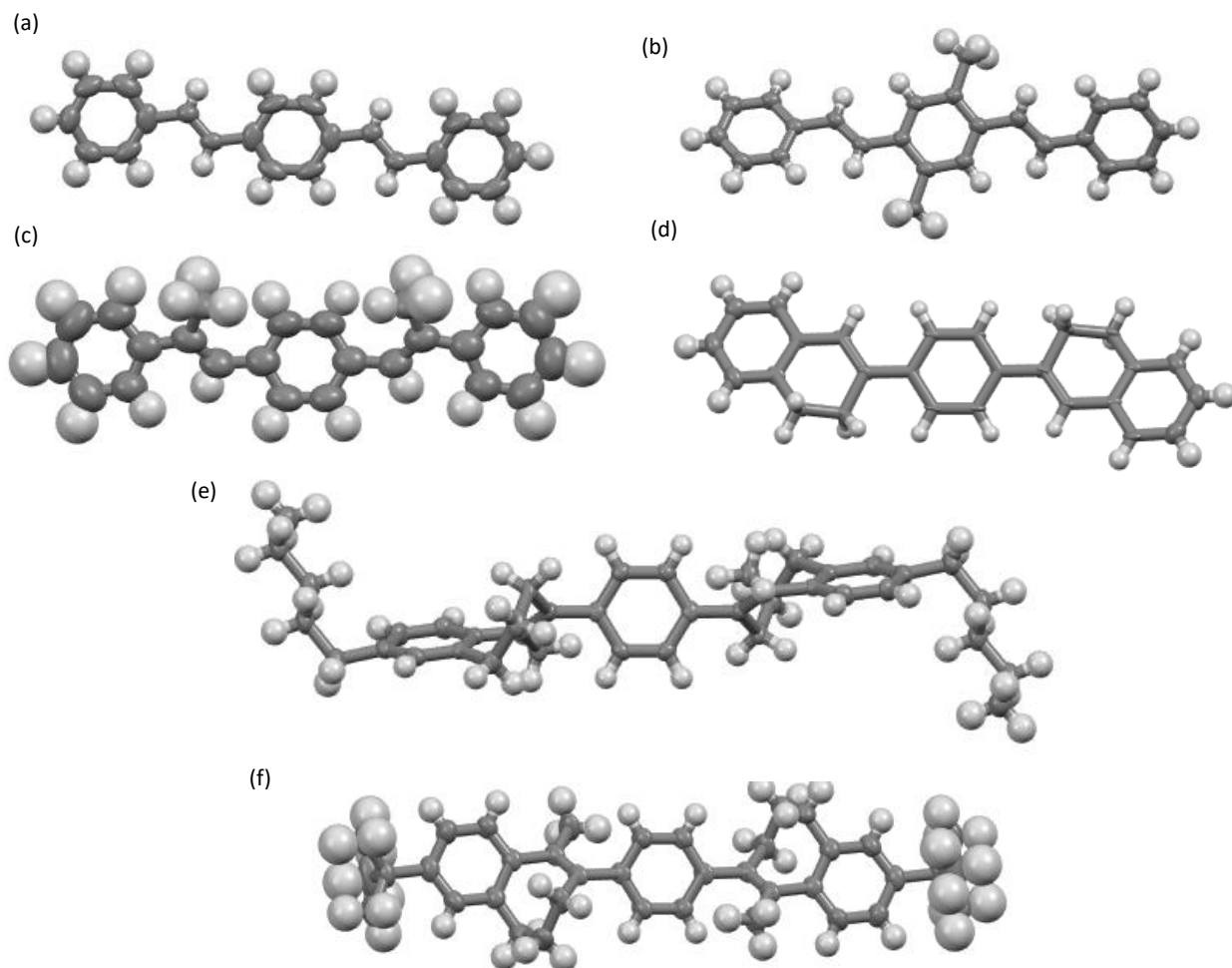


Figure S20. The front view of (a) **DSB**¹², (b) **DSDMB**, (c) **PPB**¹⁴, (d) **DBDB[6]**, (e) **5DBaMDB[7]**, and (f) **'BuDBaMDB[7]** molecule in single crystal.

(2) Single crystal X-ray

Table S8. Crystal data for **DSB**¹², **DSDMB**, **PPB**⁴, **DBDB[6]**, **DBDB[7]**, and **DBDMDB[7]**.

Compound Name	DSB	DSDMB	PPB	DBDB[6]	DBDB[7]	DBDMDB[7]
Space Group	P-1	P2 ₁ /n	P2 ₁ 2 ₁ 2 ₁	P2 ₁ /c	P-1	P2 ₁ /c
Cell Lengths (Å)	a 5.9165 b 7.4503 c 51.924	a 11.3650 b 7.0916 c 21.4912	a 36.813 b 6.2947 c 7.5555	a 6.2540 b 7.5229 c 36.785	a 7.8987 b 9.0080 c 13.8356	a 16.046 b 8.373 c 8.128
Cell Angles (°)	α 88.804 β 89.643 γ 89.945	α 90 β 98.708 γ 90	α 90 β 90 γ 90	α 90 β 92.181 γ 90	α 99.206 β 102.460 γ 90.999	α 90 β 101.706 γ 90
Cell Volume (Å ³)	2288.25	1712.14	1750.81	1729.41	947.477	1069.31
Z	6	4	4	4	2	2

Table S9. Crystal data for **DBαMDB[7]-cross**, **DBαMDB[7]-parallel**, **5DBαMDB[7]**, and **^tBuDBαMDB[7]**.

Compound Name	DBαMDB[7]-cross	DBαMDB[7]-parallel	5DBαMDB[7]	^tBuDBαMDB[7]
Space Group	P2 ₁ /c	P2 ₁ /c	P-1	P-1
Cell Lengths (Å)	a 10.2319 b 11.0510 c 9.6513	a 8.3896 b 8.4692 c 15.4856	a 8.149 b 9.767 c 10.899	a 5.8767 b 6.1160 c 20.3049
Cell Angles (°)	α 90 β 107.110 γ 90	α 90 β 103.274 γ 90	α 107.924 β 105.686 γ 102.185	α 93.452 β 91.590 γ 93.560
Cell Volume (Å ³)	1043	1070.9	752.521	726.704
Z	5	5	1	2

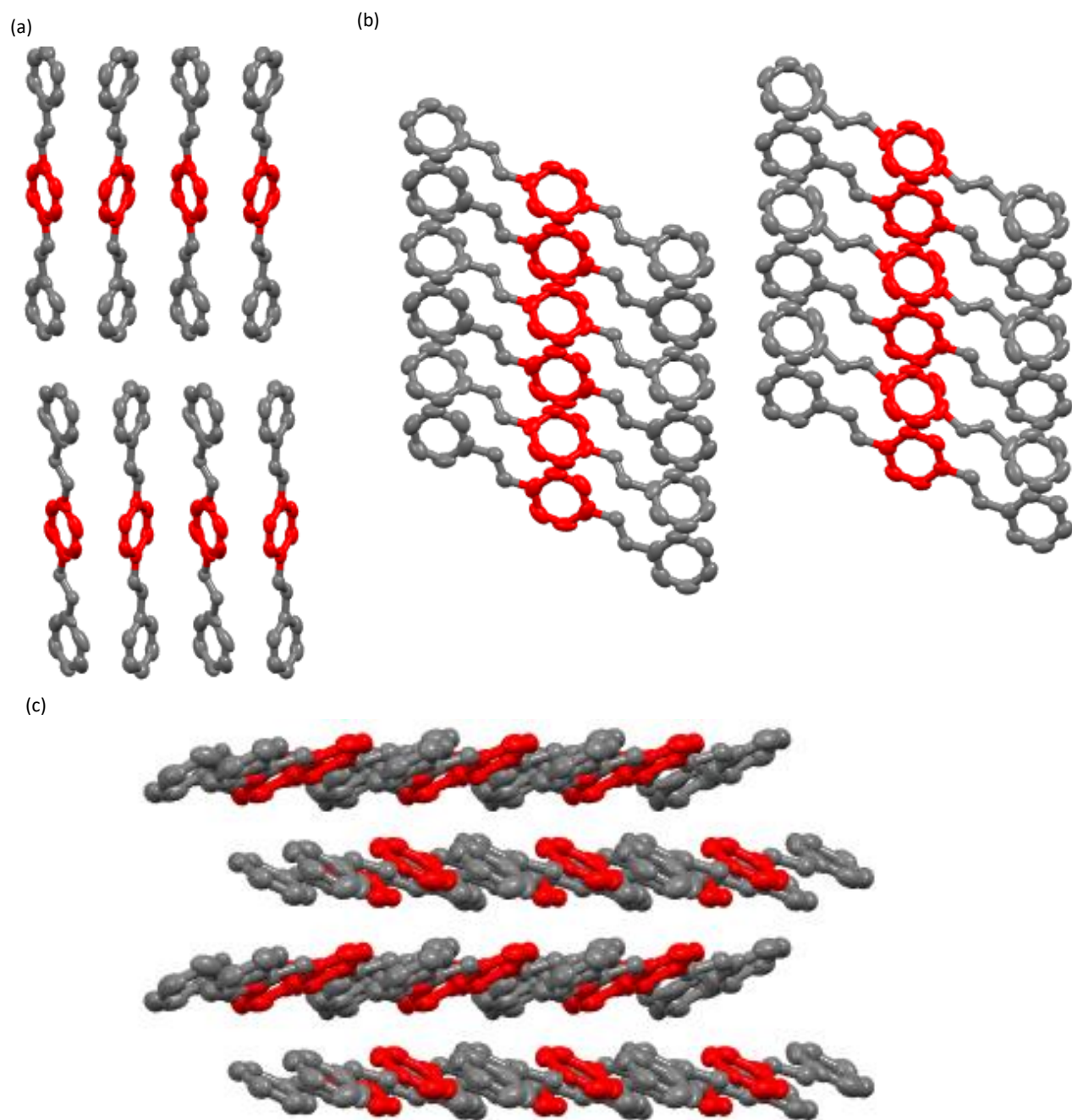


Figure S21. Projection of the structure of DSB^{12} viewed along (a) a -axis, (b) b -axis, and (c) c -axis directions. Hydrogen atoms are omitted for clarity, and the central benzene rings are shown in red.

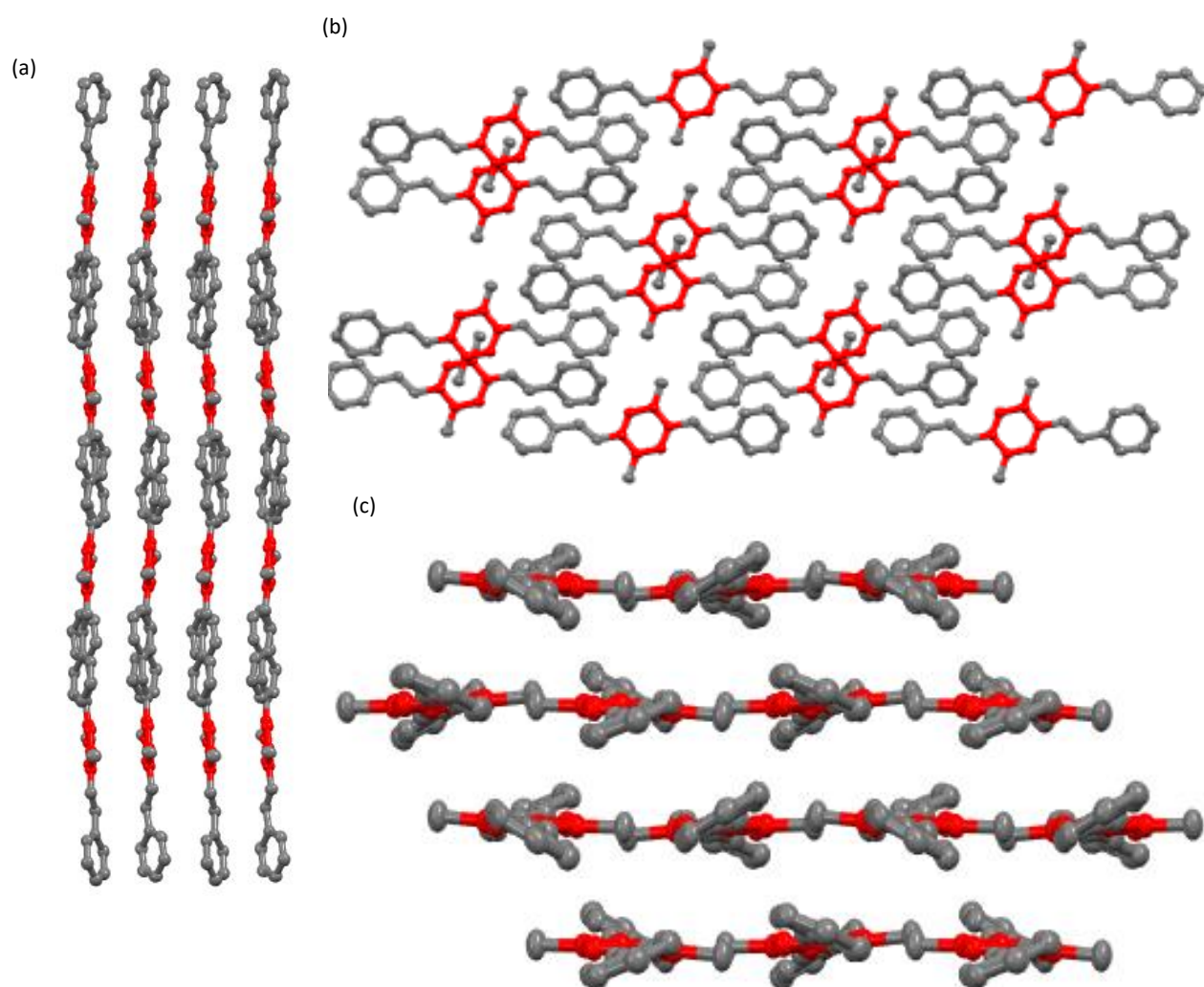


Figure S22. Projection of the structure of **DSDMB** viewed along (a) *a*-axis, (b) *b*-axis, and (c) *c*-axis directions. Hydrogen atoms are omitted for clarity, and the central benzene rings are shown in red.

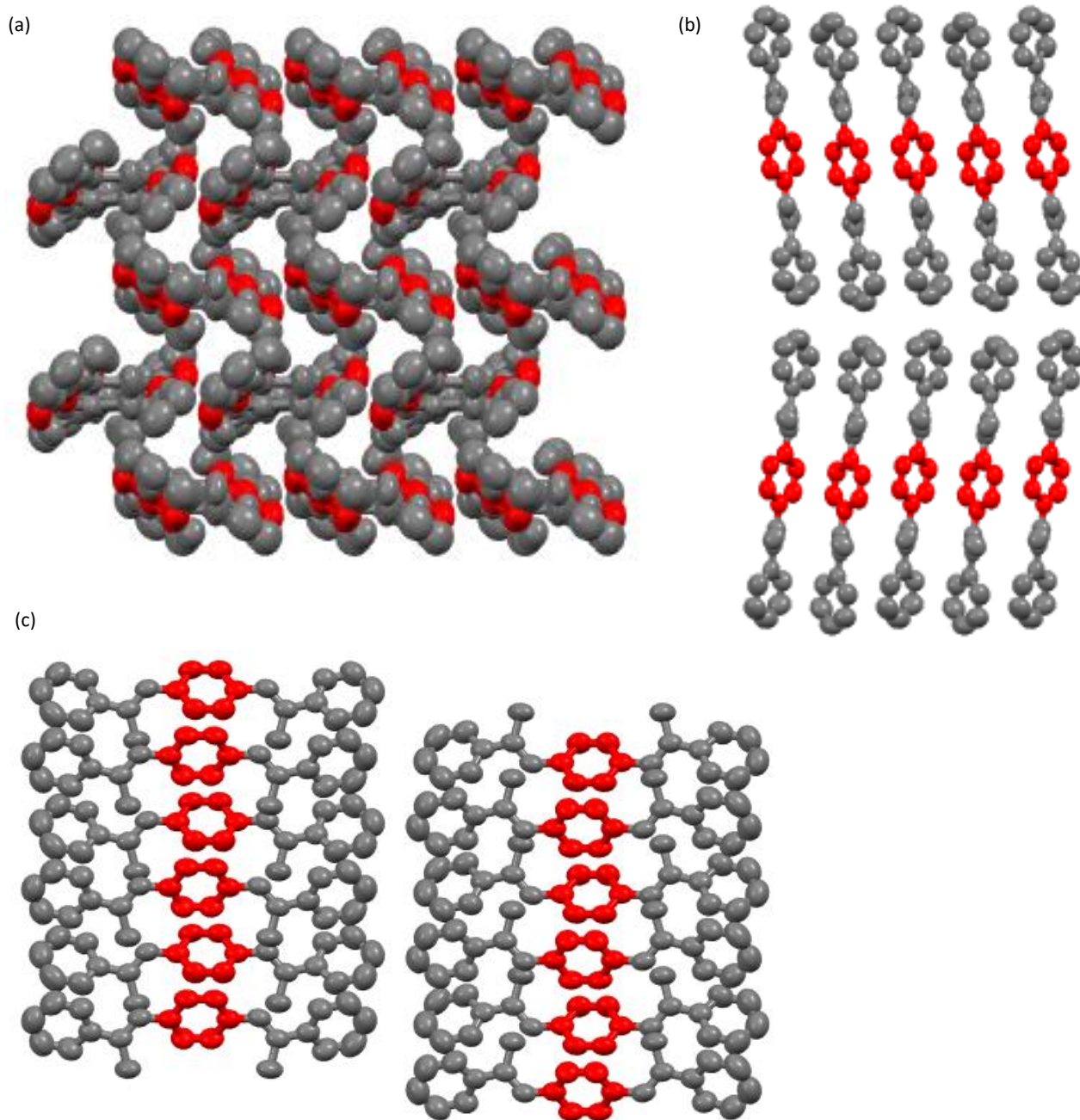


Figure S23. Projection of the structure of PPB⁴ viewed along (a) *a*-axis, (b) *b*-axis, and (c) *c*-axis directions. Hydrogen atoms are omitted for clarity, and the central benzene rings are shown in red.

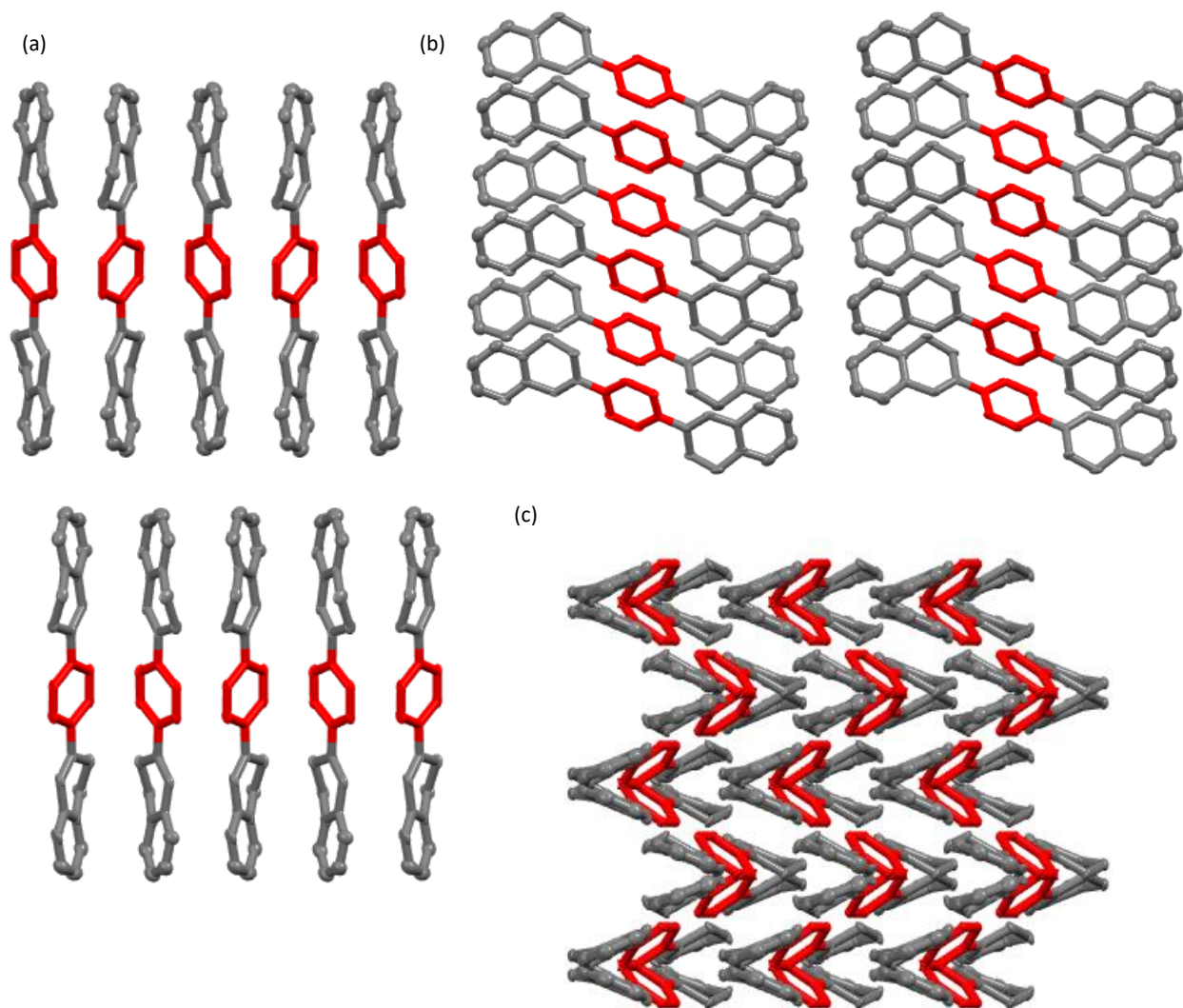


Figure S24. Projection of the structure of DBDB[6] viewed along (a) a -axis, (b) b -axis, and (c) c -axis directions. Hydrogen atoms are omitted for clarity, and the central benzene rings are shown in red.

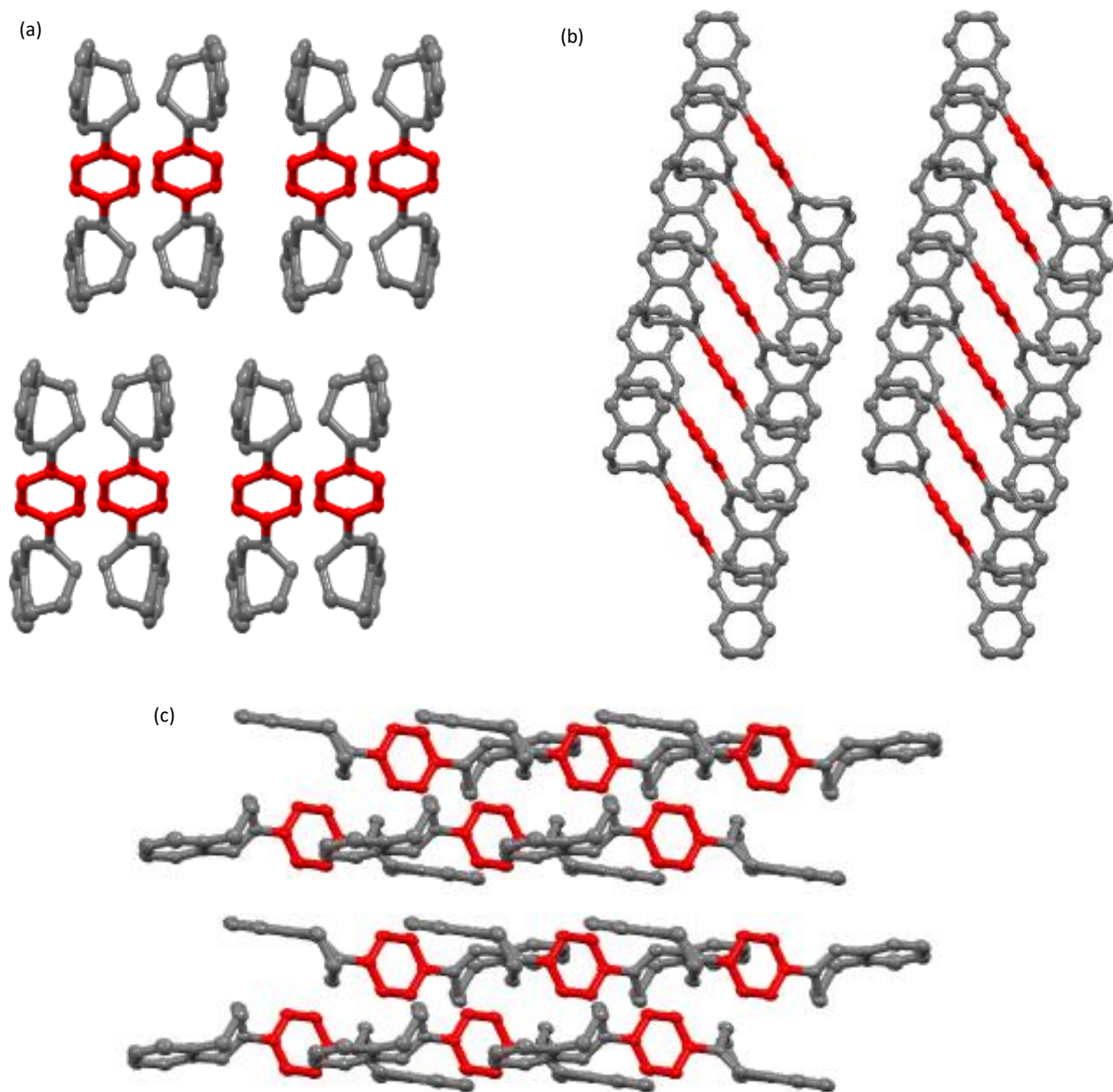


Figure S25. Projection of the structure of DBDB[7] viewed along (a) a -axis, (b) b -axis, and (c) c -axis directions. Hydrogen atoms are omitted for clarity, and the central benzene rings are shown in red.

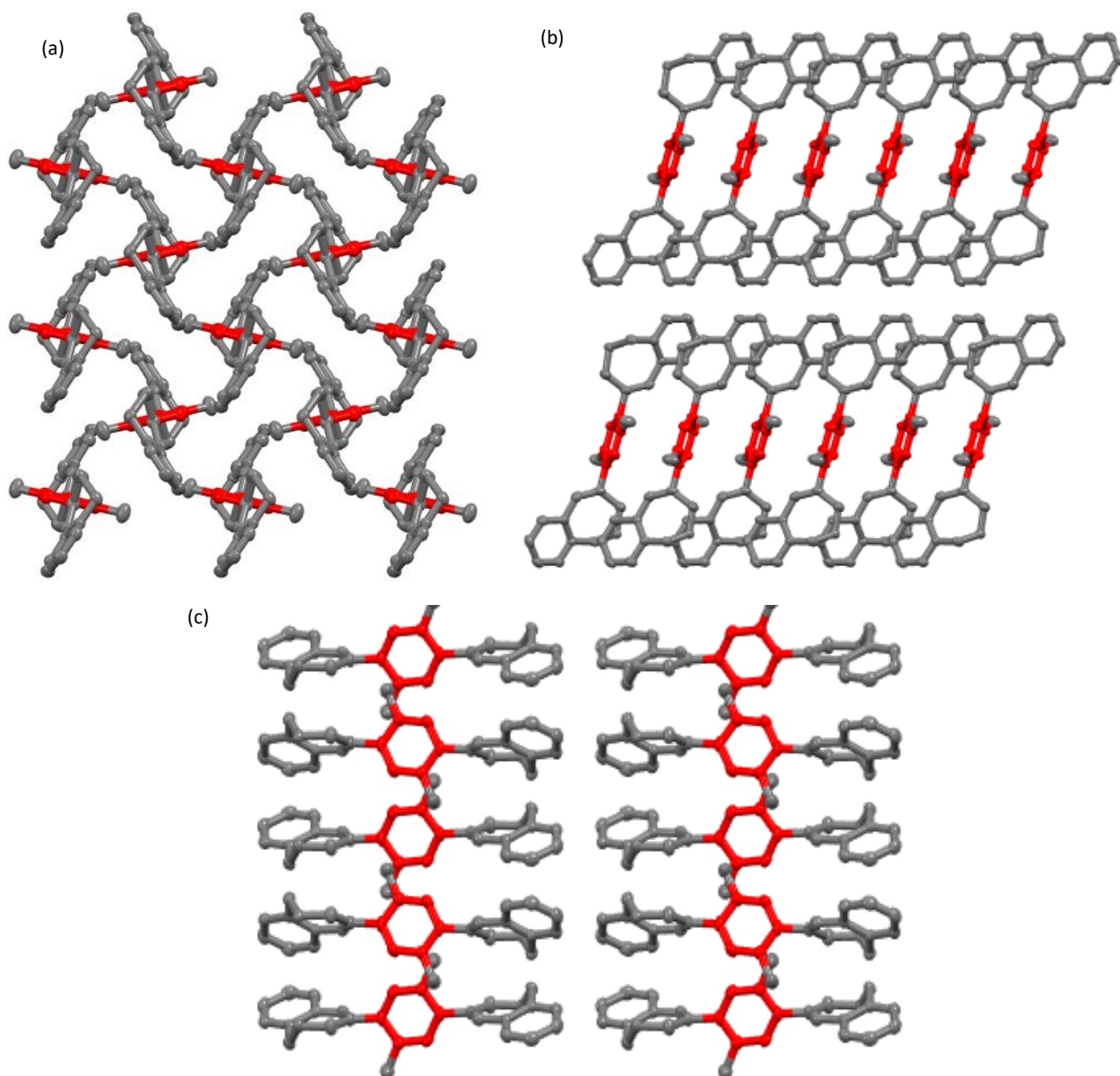


Figure S26. Projection of the structure of DBDMDB[7] viewed along (a) *a*-axis, (b) *b*-axis, and (c) *c*-axis directions. Hydrogen atoms are omitted for clarity, and the central benzene rings are shown in red.

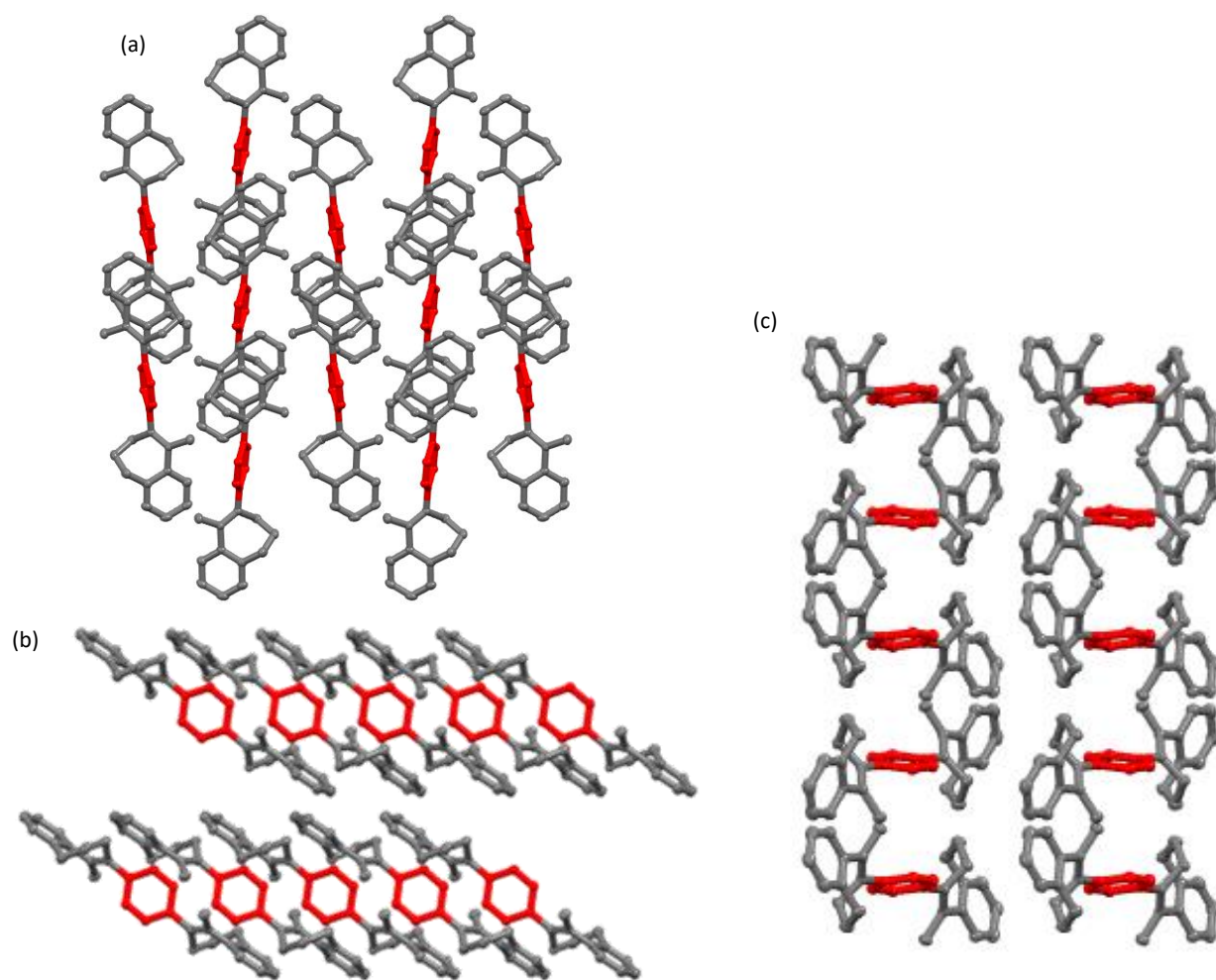


Figure S27. Projection of the structure of **DBaMDB[7]-cross** viewed along (a) *a*-axis, (b) *b*-axis, and (c) *c*-axis directions. Hydrogen atoms are omitted for clarity, and the central benzene rings are shown in red.

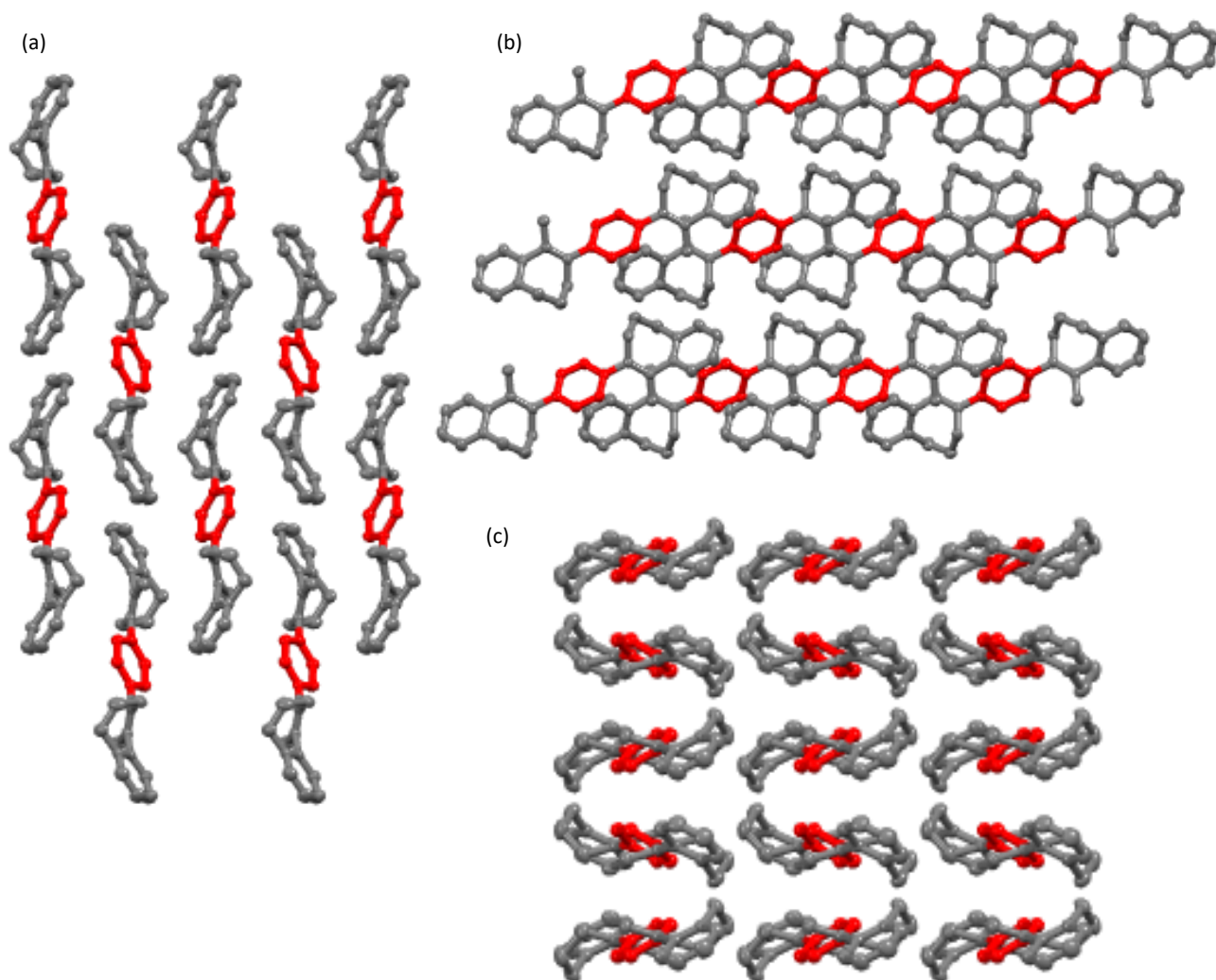


Figure S28. Projection of the structure of DB α MDB[7]-parallel viewed along (a) *a*-axis, (b) *b*-axis, and (c) *c*-axis directions. Hydrogen atoms are omitted for clarity, and the central benzene rings are shown in red.

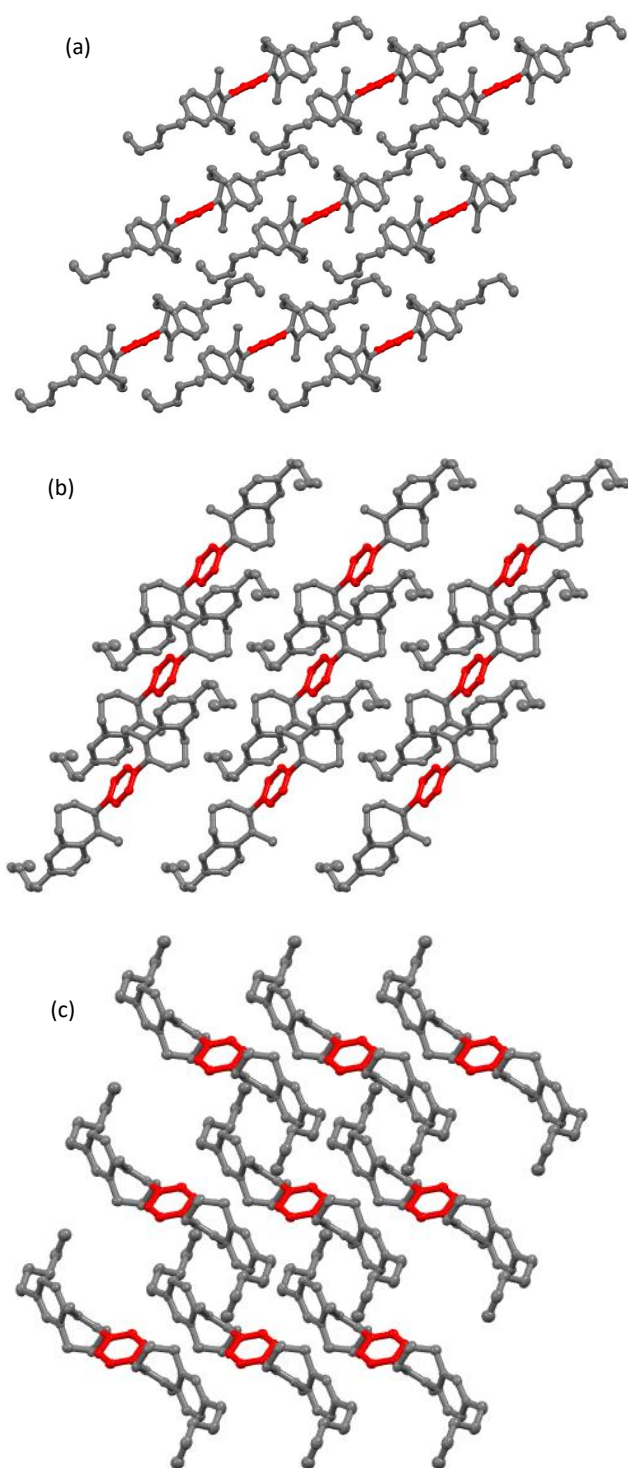


Figure S29. Projection of the structure of **5DBaMDB[7]** viewed along (a) *a*-axis, (b) *b*-axis, and (c) *c*-axis directions. Hydrogen atoms are omitted for clarity, and the central benzene rings are shown in red.

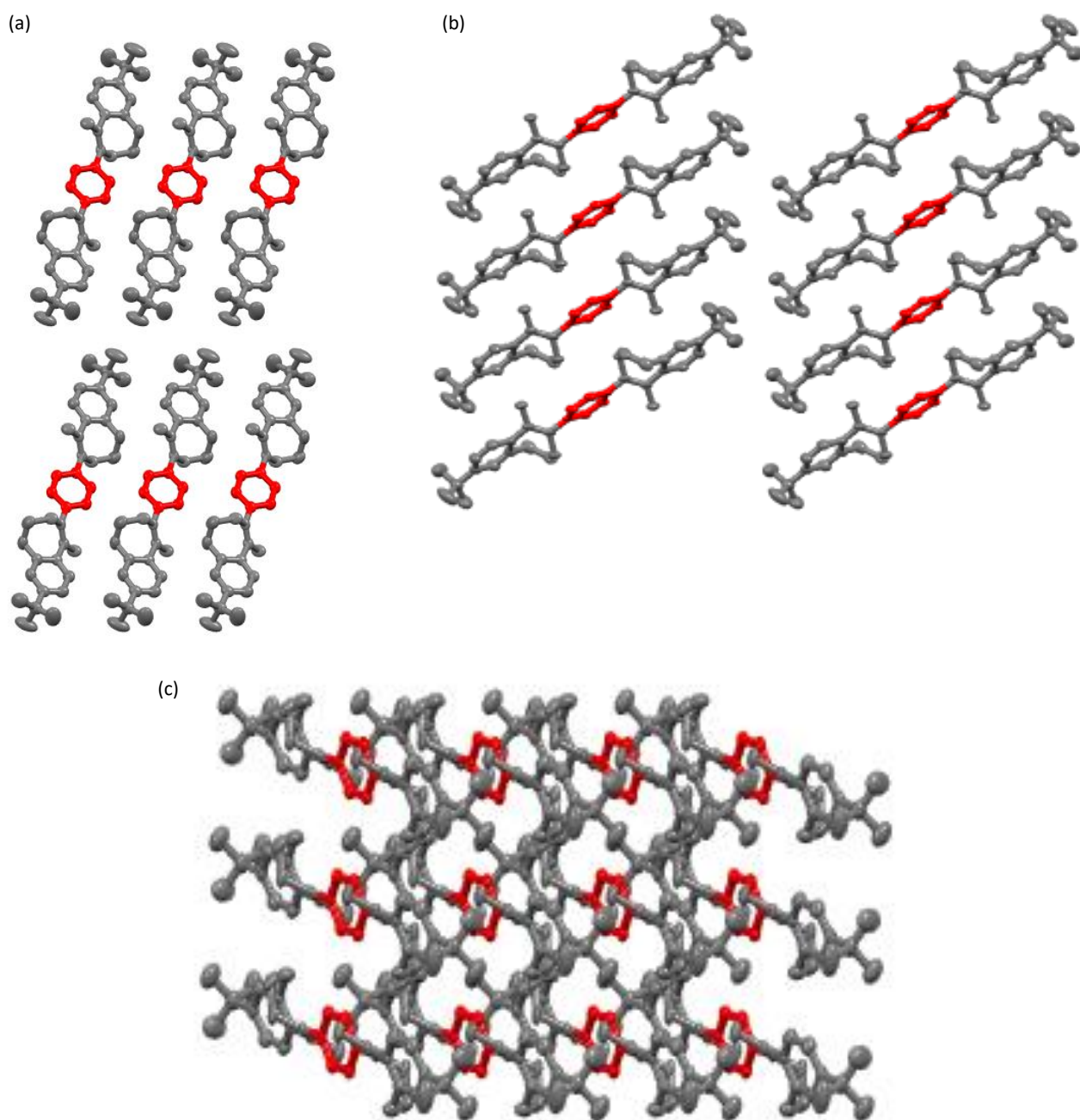


Figure S30. Projection of the structure of 'BuDBaMDB[7] viewed along (a) a -axis, (b) b -axis, and (c) c -axis directions. Hydrogen atoms are omitted for clarity, and the central benzene rings are shown in red.

7. Theoretical study

Molecular and crystal structures used in the computational calculations were from single-crystal X-ray analysis. Calculations of excitation transition of single molecule were performed using Gaussian 09 program, and calculations of NBO analysis were using Gaussian 16 program.

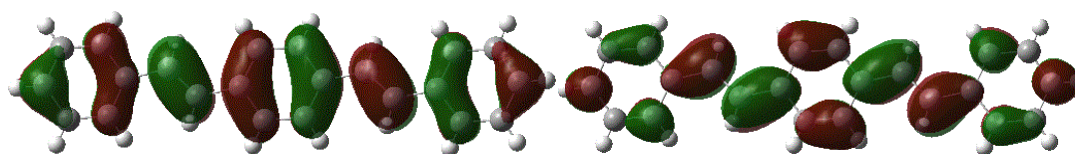
(1) Excitation transition of single molecule

In **5DB α MDB[7]**, methyl groups were used instead of pentyl groups because the calculation did not converge.

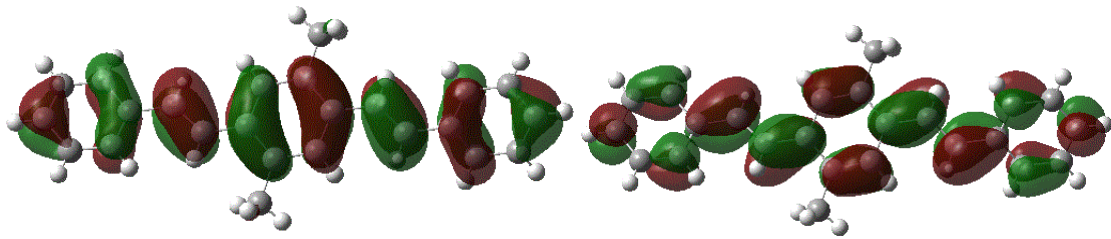
Table S10. Excitation transition properties of each compound.

Entry	Transition property	Oscillator strengths	ΔE [eV]	λ_{abs} [nm]
DSB	0.672 (HOMO→LUMO)	1.8823	3.8217	324.42
	0.184 (HOMO-1→LUMO+1)			
DSDMB	0.673 (HOMO→LUMO)	1.7476	3.6935	335.69
	0.181 (HOMO-1→LUMO+1)			
PPB	0.669 (HOMO→LUMO)	1.7209	4.2083	294.62
	0.193 (HOMO-1→LUMO+1)			
DBDB[6]	0.668 (HOMO→LUMO)	1.5320	3.6154	342.93
	0.205 (HOMO-1→LUMO+1)			
DBDB[7]	0.664 (HOMO→LUMO)	1.4343	4.2444	292.11
	-0.206 (HOMO-1→LUMO+1)			
	0.477 (HOMO→LUMO)			
DBDMDB[7]	-0.460 (HOMO-1→LUMO+1)	1.2504	5.1075	242.75
	-0.125 (HOMO-2→LUMO+2)			
	-0.102 (HOMO-3→LUMO)			
DBαMDB[7]-cross	0.671 (HOMO→LUMO)	1.3063	4.2887	289.10
	0.238 (HOMO-1→LUMO+1)			
DBαMDB[7]-parallel	0.652 (HOMO→LUMO)	1,2507	4.6772	265.08
	0.220 (HOMO-1→LUMO+1)			
5DBαMDB[7]	0.629 (HOMO→LUMO)	1.4077	4.6106	268.91
	0.278 (HOMO-1→LUMO+1)			
^tBuDBαMDB[7]	0.628 (HOMO→LUMO)	1.5794	4.7347	261.86
	0.276 (HOMO-1→LUMO+1)			
	-0.102 (HOMO-3→LUMO+5)			

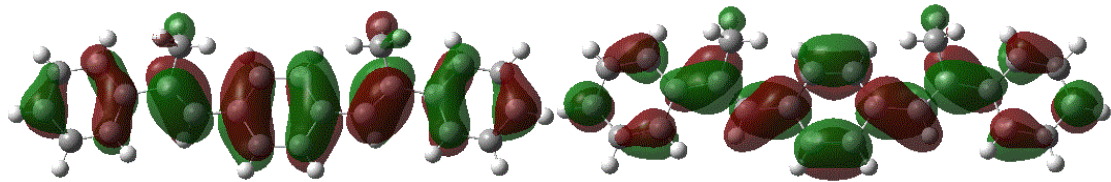
DSB



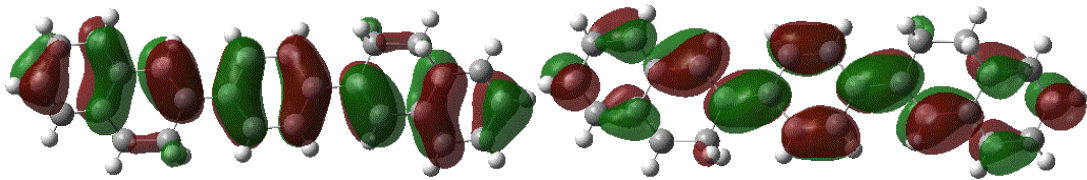
DSDMB



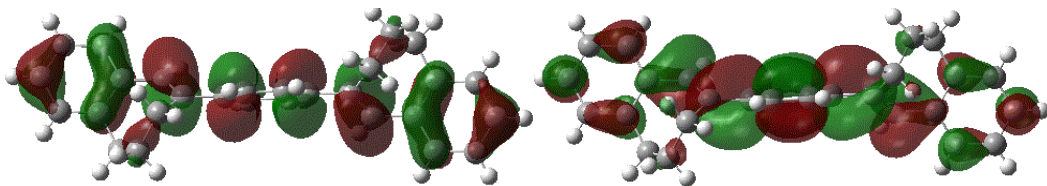
PPB



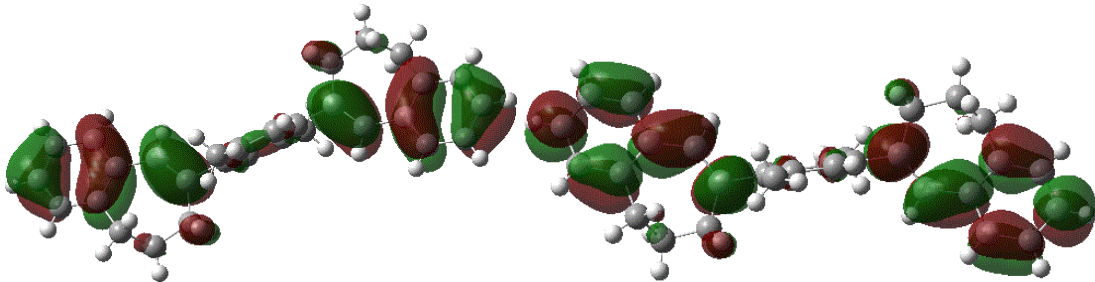
DBDB[6]



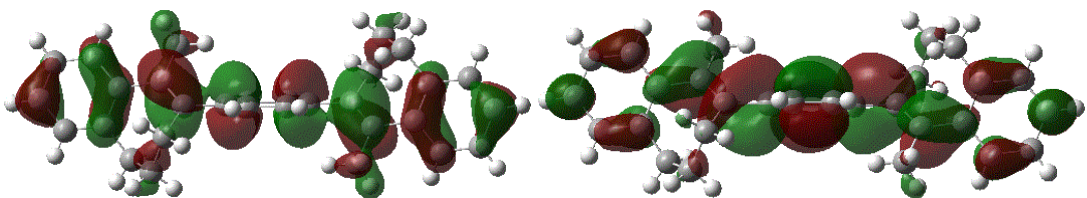
DBDB[7]



DBDMDB[7]



DBaMDB[7]-cross



DBaMDB[7]-parallel

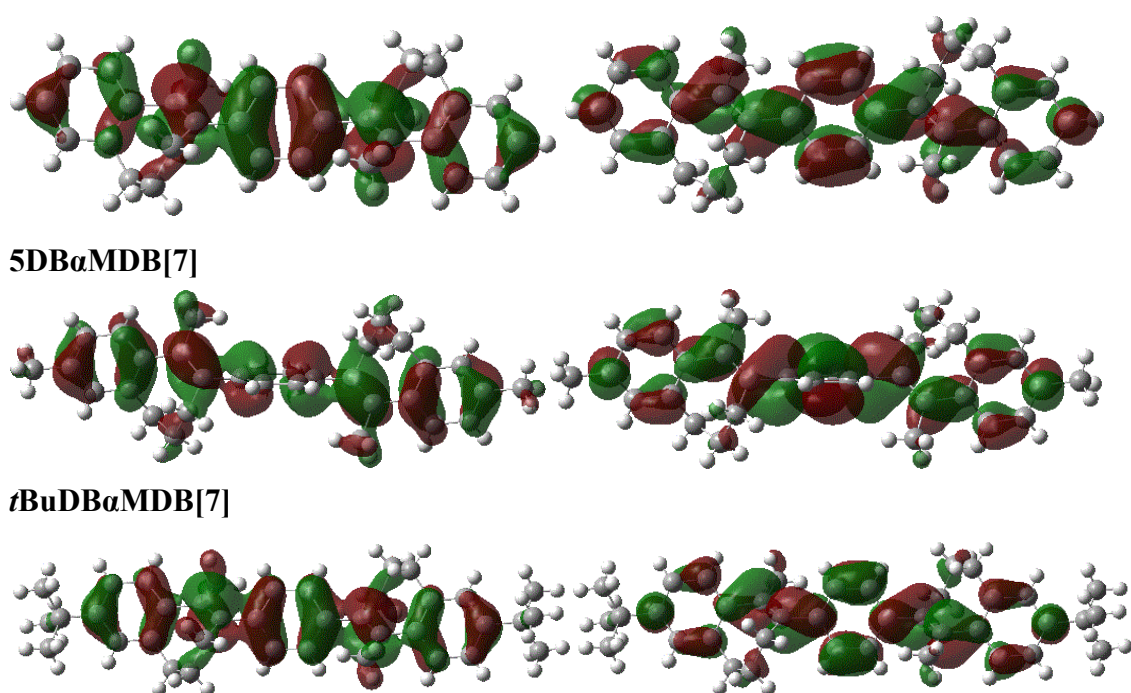


Figure S31. HOMO (left) and LUMO (right) of each compound at single crystal structure.

Table S11. Energy levels of HOMO and LUMO of each compound.

Compound	HOMO [eV]	LUMO [eV]
DSB	-7.18435	-0.15238
DSDMB	-7.07170	-0.16191
PPB	-7.33565	0.24245
DBDB[6]	-6.94897	-0.09497
DBDB[7]	-7.11959	0.53035
DBDMDB[7]	-7.58110	1.31948
DB α MDB[7]-cross	-7.31551	0.53171
DB α MDB[7]-parallel	-7.31279	1.03920
5DB α MDB[7]	-7.33674	1.03784
<i>t</i> BuDB α MDB[7]	-7.41647	0.95185

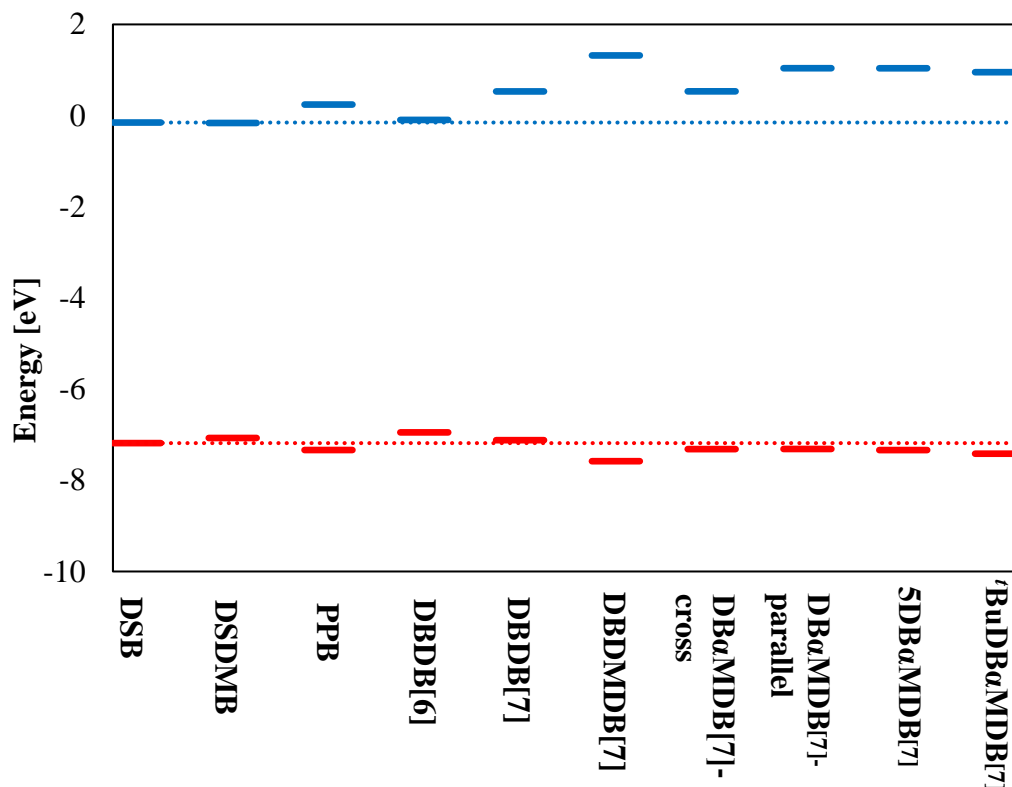


Figure S32. Energy levels of HOMO (red) and LUMO (blue) of each compound.

(2) NBO analysis

The center molecule is shown in black. “Unit color A to color B” means the interaction between occupied orbital of C=C bond of color A and unoccupied orbital of C=C bond of color B.

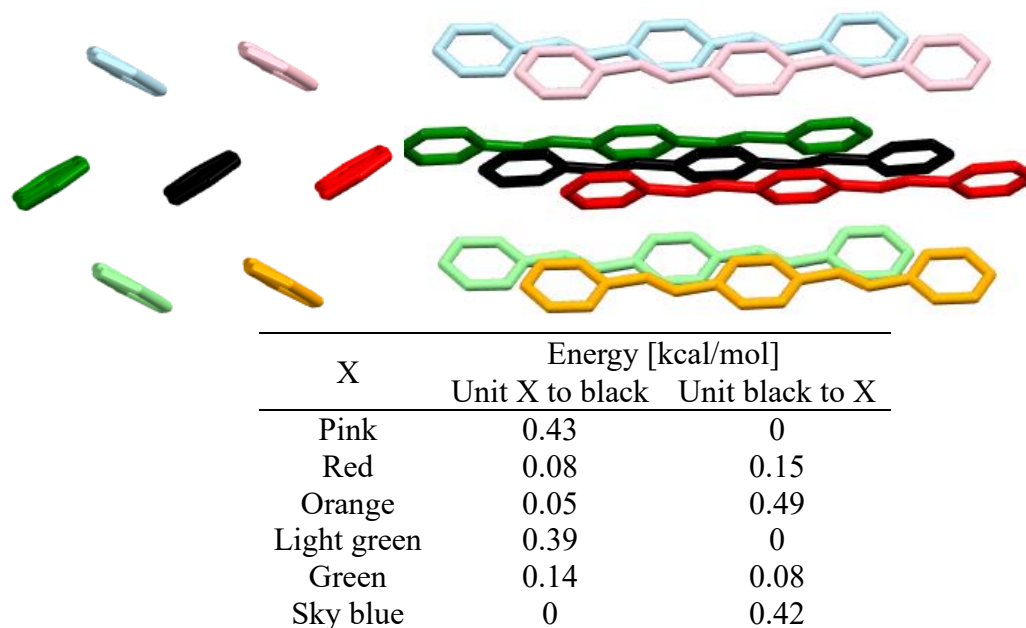
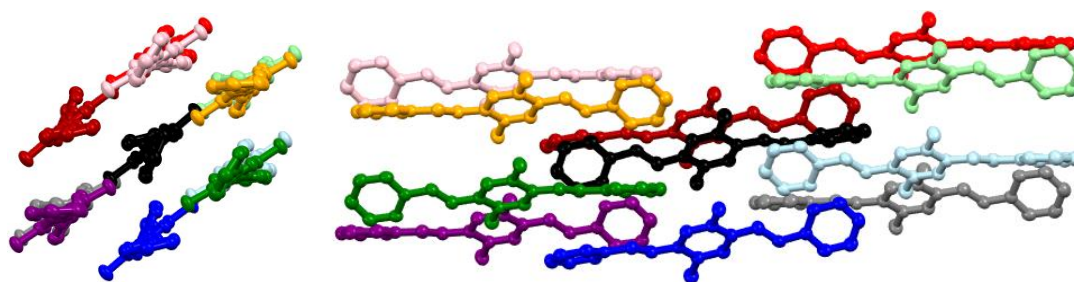


Figure S33. Intermolecular interaction of NBO of DSB.



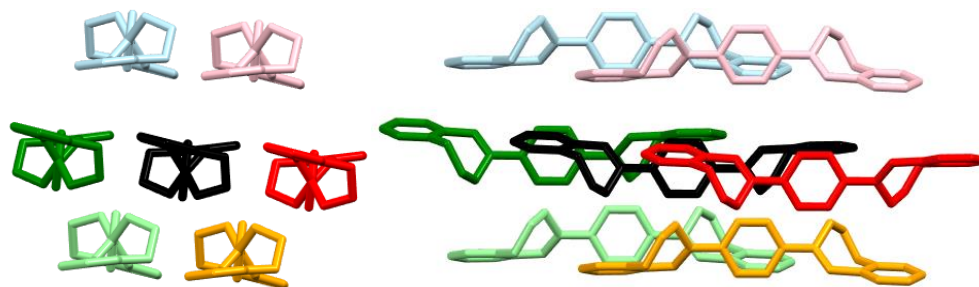
X	Energy [kcal/mol]	
	Unit X to black	Unit black to X
Pink	0.30	0
Red	0.20	0.47
Orange	0	0
Light green	0	0
Green	0	0.30
Sky blue	0.47	0.19
Blue	0.36	0.36
Purple	0	0
Gray	0	0
Brown	0.63	0.61

Figure S34. Intermolecular interaction of NBO of **DSDMB**.



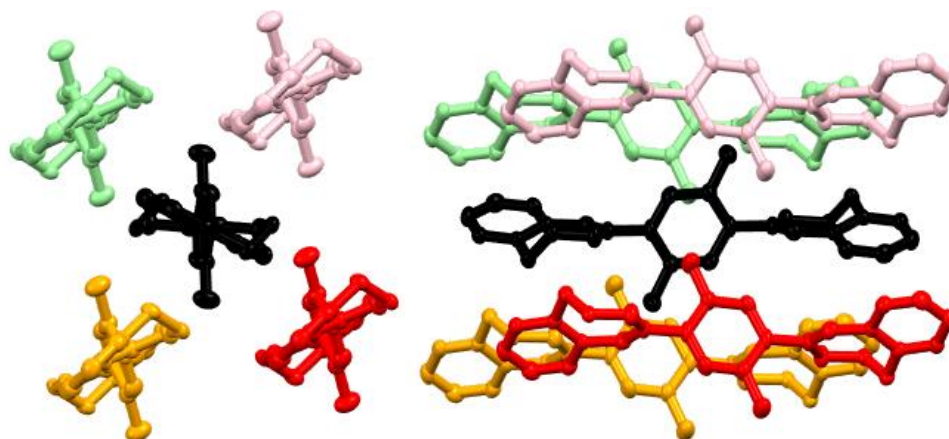
X	Energy [kcal/mol]	
	Unit X to black	Unit black to X
Pink	0.14	0
Red	0	0
Orange	0	0.31
Light green	0.20	0.11
Green	0	0
Sky blue	0	0.15

Figure S35. Intermolecular interaction of NBO of **PPB**.



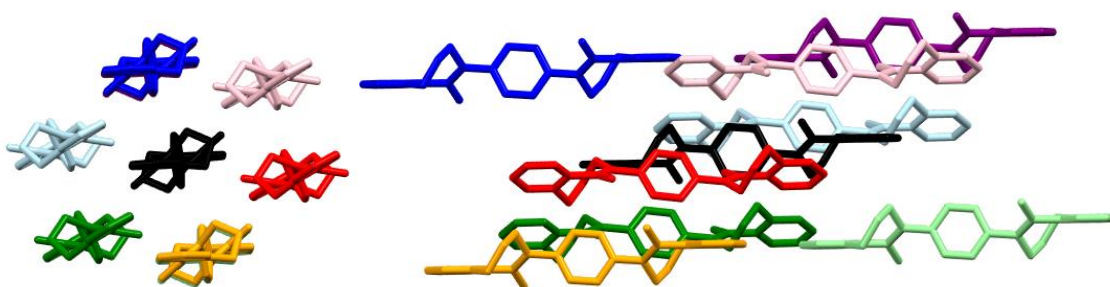
X	Energy [kcal/mol]	
	Unit X to black	Unit black to X
Pink	0.41	0.41
Red	0	0
Orange	0.42	0.42
Light green	0.52	0.52
Green	0	0
Sky blue	0.16	0.16

Figure S36. Intermolecular interaction of NBO of **DBDB**[7].



X	Energy [kcal/mol]	
	Unit X to black	Unit black to X
Pink	0	0
Red	0	0
Orange	0	0
Light green	0	0

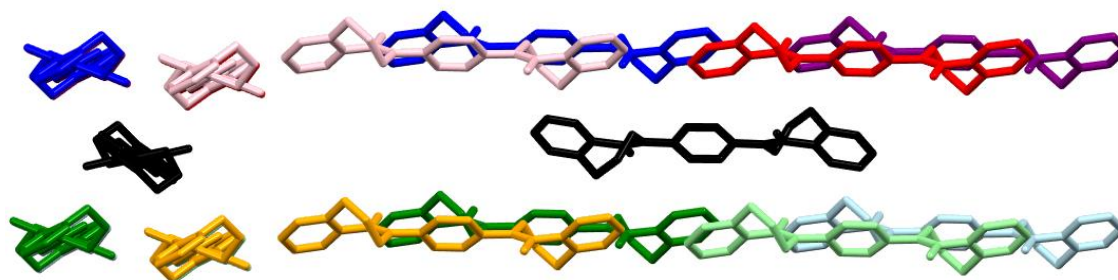
Figure S37. Intermolecular interaction of NBO of **DBDMDB**[7].



X	Energy [kcal/mol]	
	Unit X to black	Unit black to X
Pink	0.28	0.15
Red	0.15	0.19
Orange	0	0
Light green	0	0
Green	0.19	0.15
Sky blue	0.15	0.28

Blue	0	0
Purple	0	0

Figure S38. Intermolecular interaction of NBO of DB α MDB[7]-cross.



X	Energy [kcal/mol]	
	Unit X to black	Unit black to X
Pink	0	0
Red	0	0.08
Orange	0	0
Light green	0.07	0
Green	0	0.14
Sky blue	0	0
Blue	0.13	0
Purple	0	0

Figure S39. Intermolecular interaction of NBO of DB α MDB[7]-parallel.

8. References

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9. NMR chart

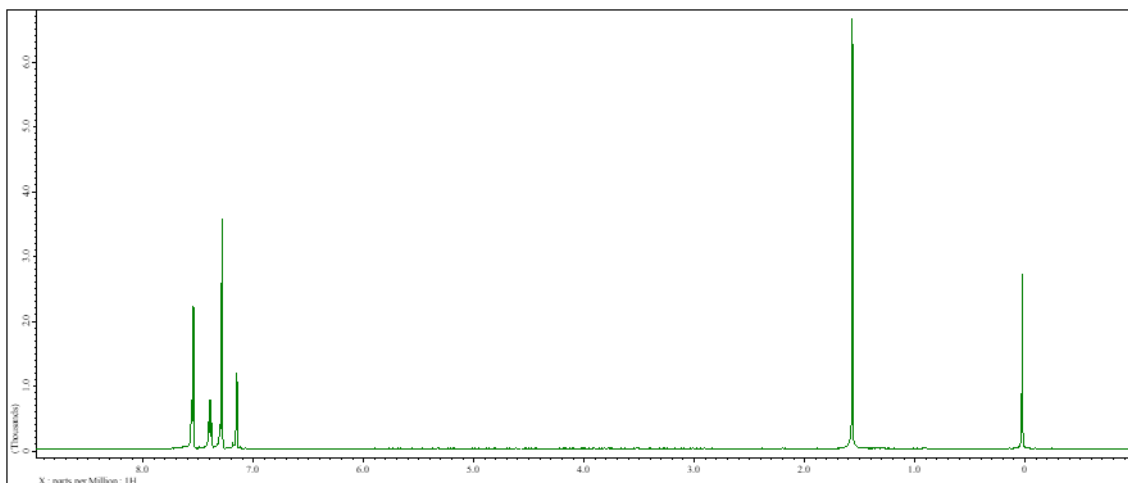


Figure S40. ^1H -NMR spectra of **DSB** (500 MHz, CDCl_3).

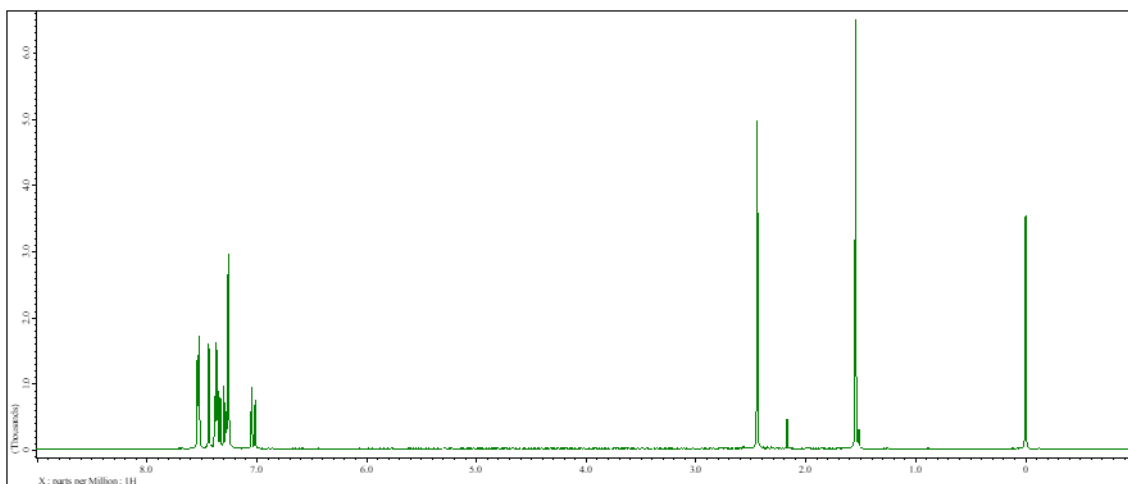


Figure S41. ^1H -NMR spectra of **DSDMB** (500 MHz, CDCl_3).

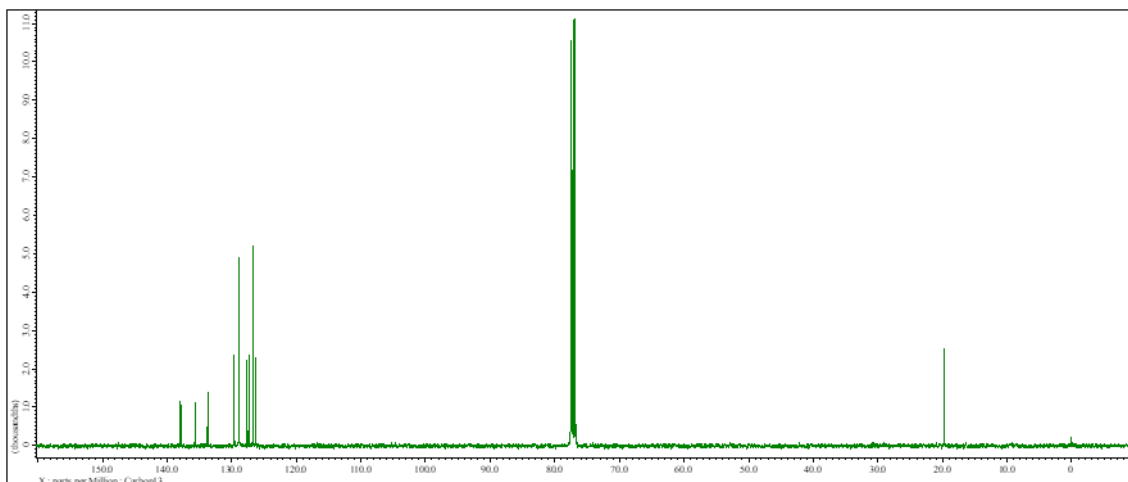


Figure S42. ^{13}C -NMR spectra of **DSDMB** (100 MHz, CDCl_3).

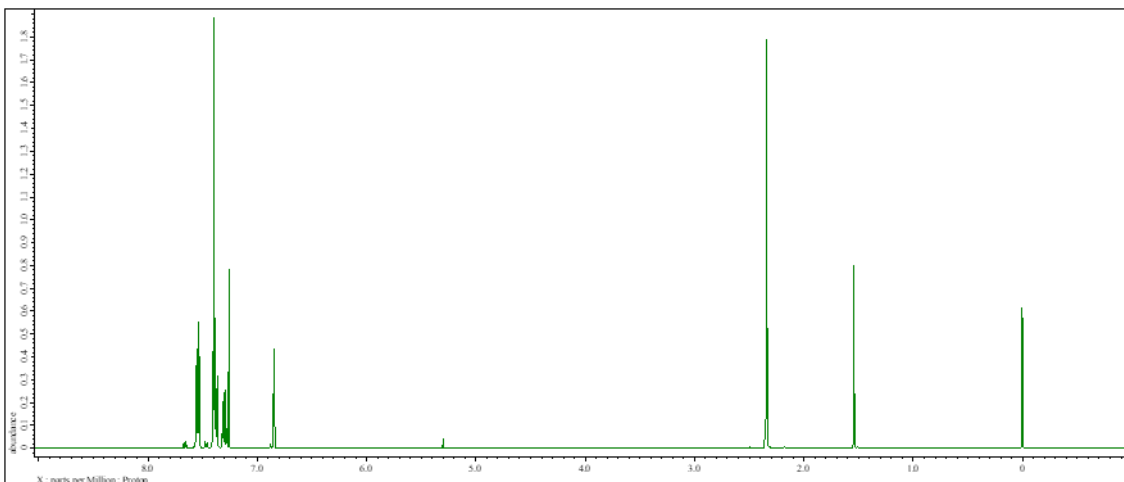


Figure S43. ^1H -NMR spectra of **PPB** (399 MHz, CDCl_3).

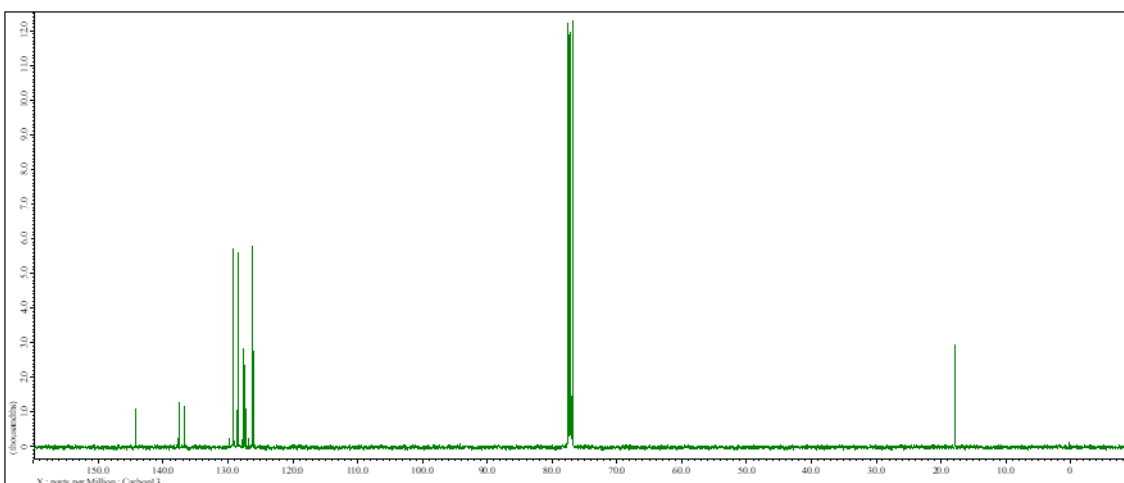


Figure S44. ^{13}C -NMR spectra of **PPB** (100 MHz, CDCl_3).

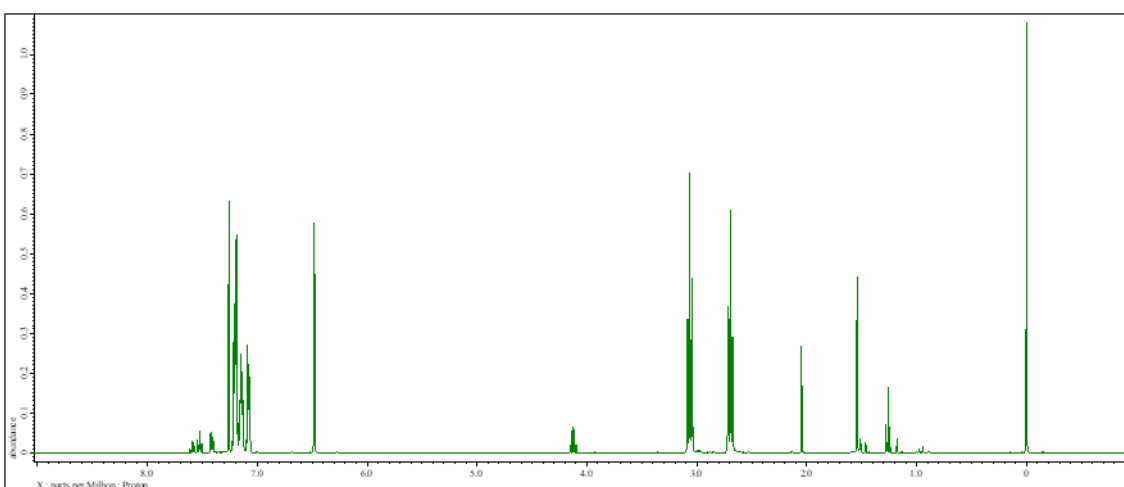


Figure S45. ^1H -NMR spectra of **4** (399 MHz, CDCl_3).

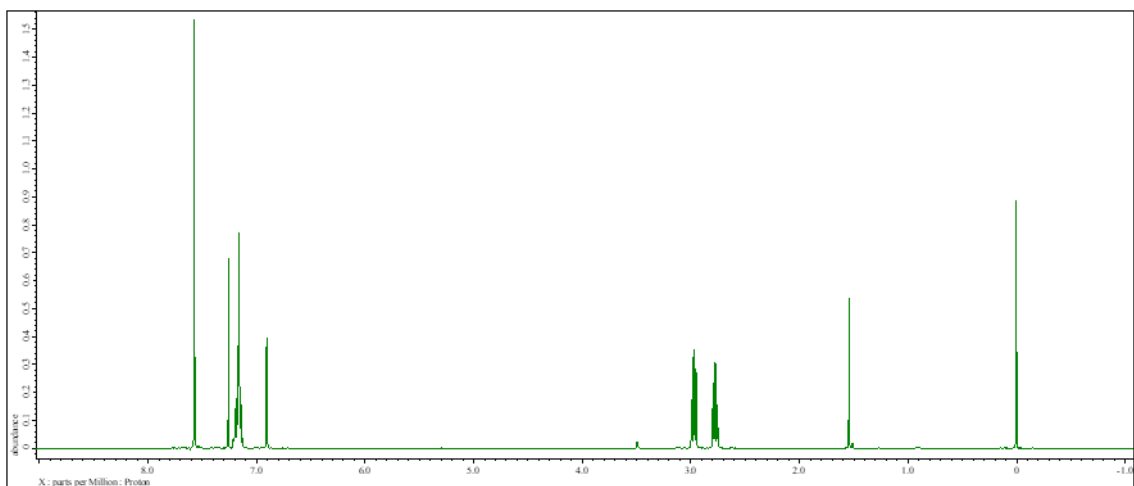


Figure S46. ^1H -NMR spectra of **DBDB[6]** (399 MHz, CDCl_3).

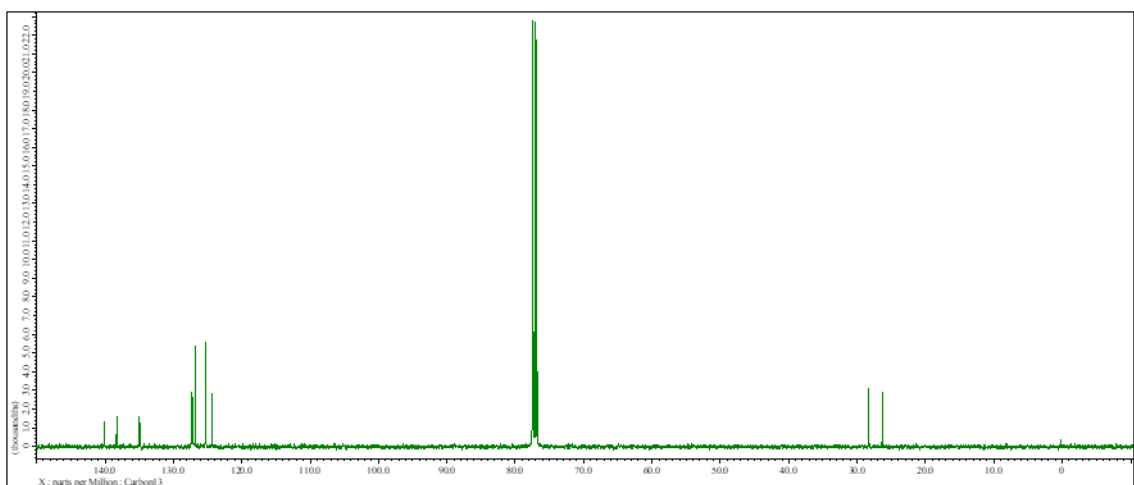


Figure S47. ^{13}C -NMR spectra of **DBDB[6]** (100 MHz, CDCl_3).

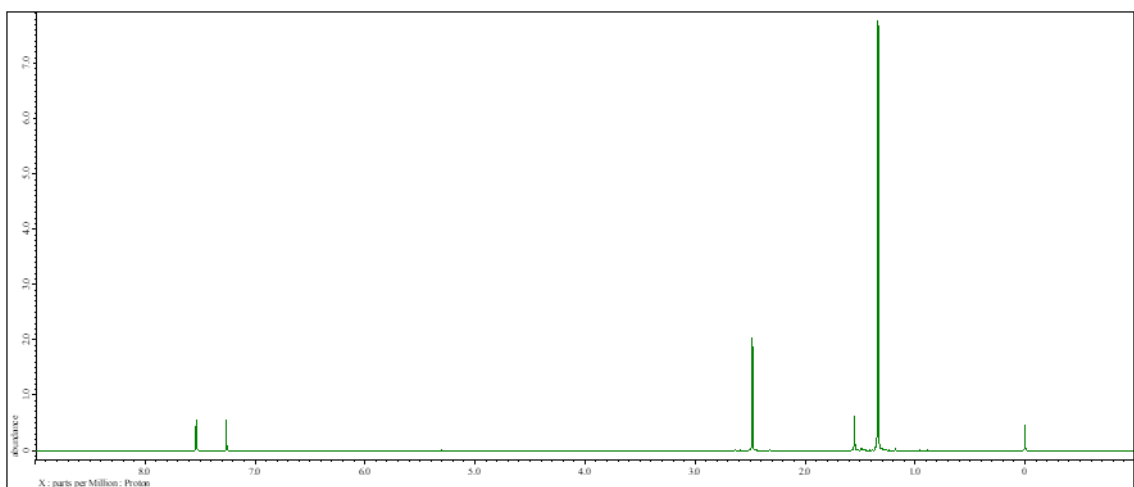


Figure S48. ^1H -NMR spectra of **DMB-B(pin)** (399 MHz, CDCl_3),

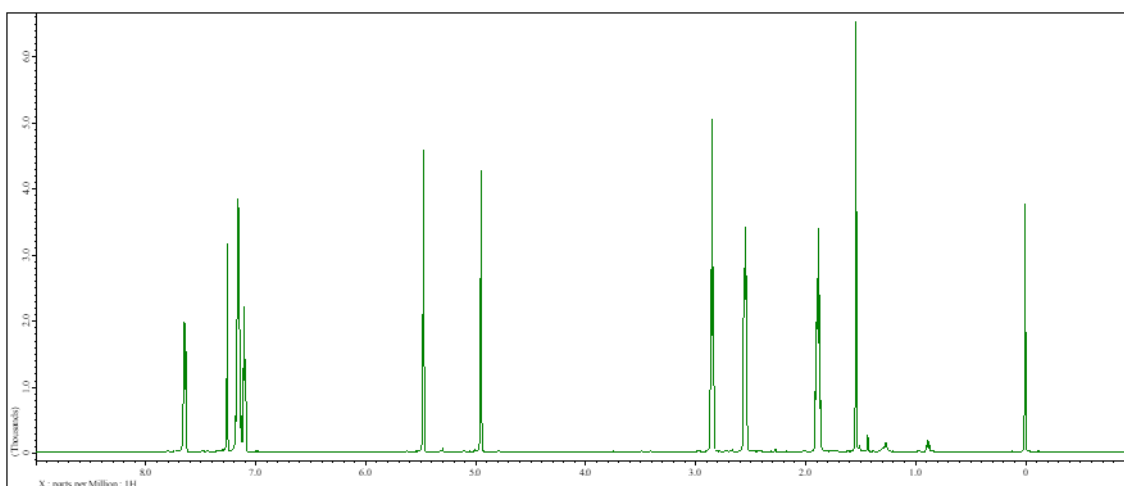


Figure S49. ^1H -NMR spectra of **7a** (500 MHz, CDCl_3).

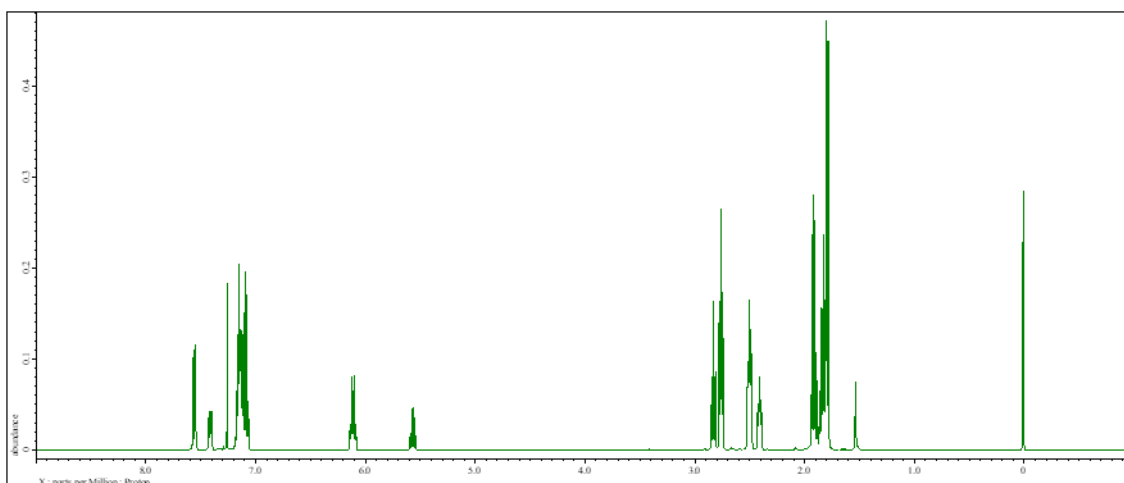


Figure S50. ^1H -NMR spectra of **7b** (399 MHz, CDCl_3).

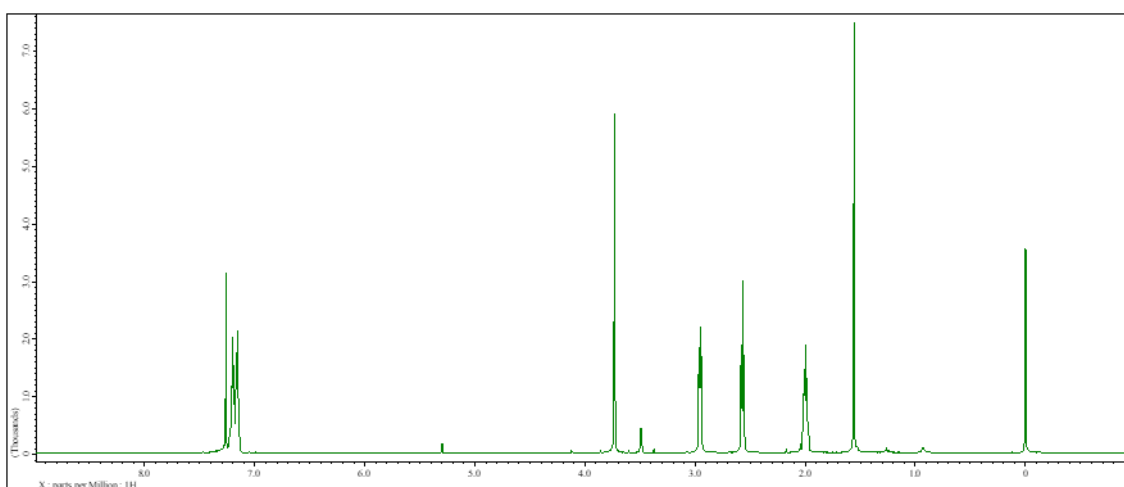


Figure S51. ^1H -NMR spectra of **8a** (500 MHz, CDCl_3).

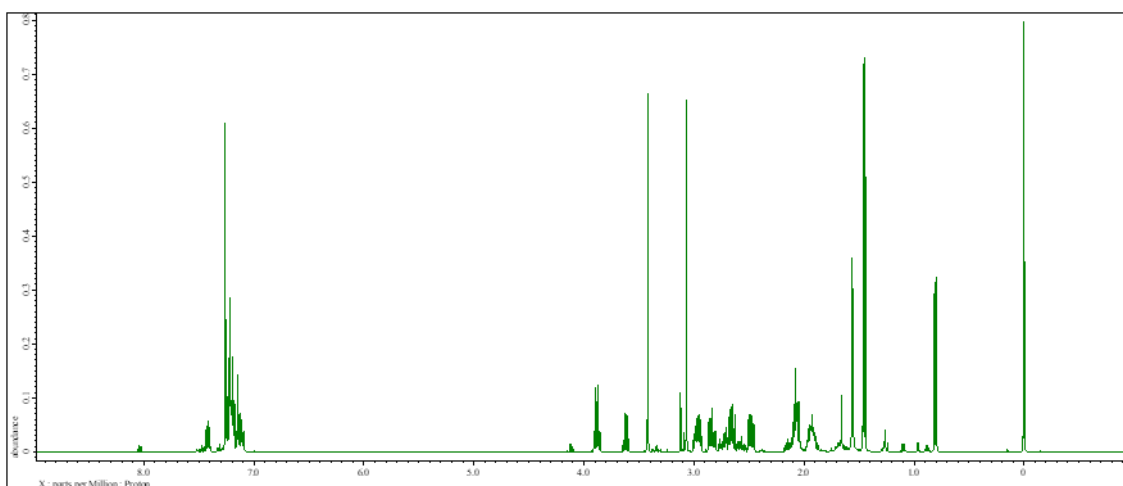


Figure S52. $^1\text{H-NMR}$ spectra of **8b** (399 MHz, CDCl_3).

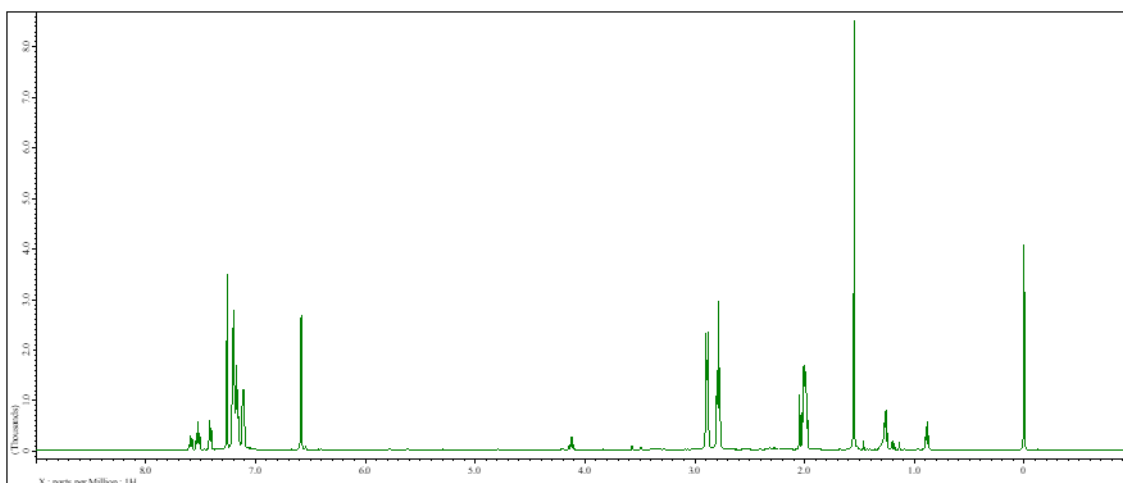


Figure S53. $^1\text{H-NMR}$ spectra of **9a** (500 MHz, CDCl_3).

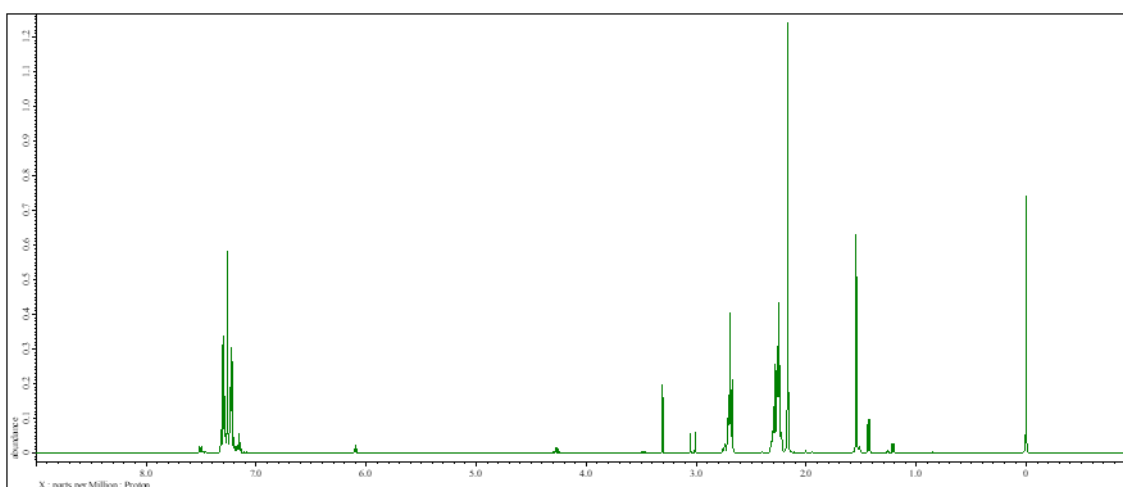


Figure S54. $^1\text{H-NMR}$ spectra of **9b** (399 MHz, CDCl_3).

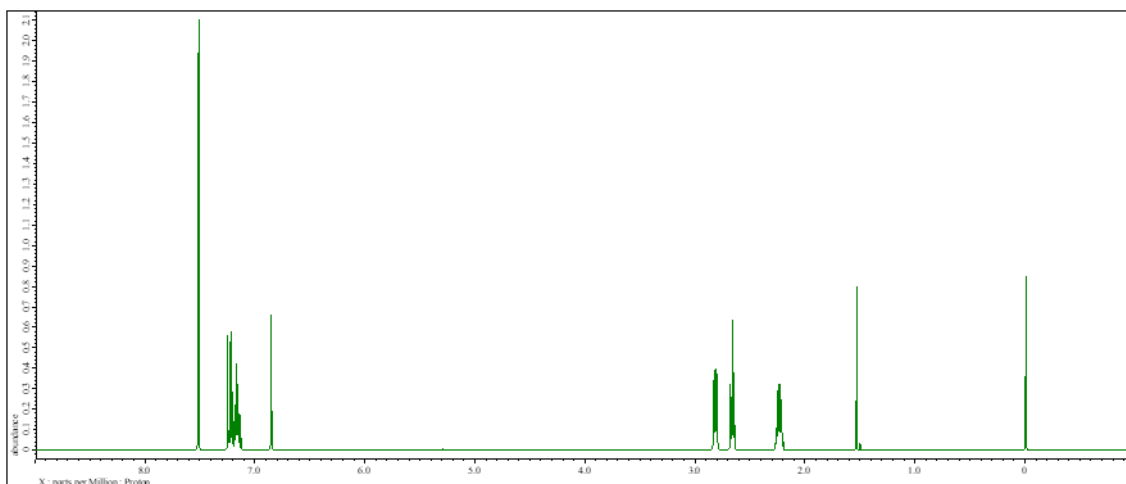


Figure S55. ¹H-NMR spectra of DBDB[7] (399 MHz, CDCl₃).

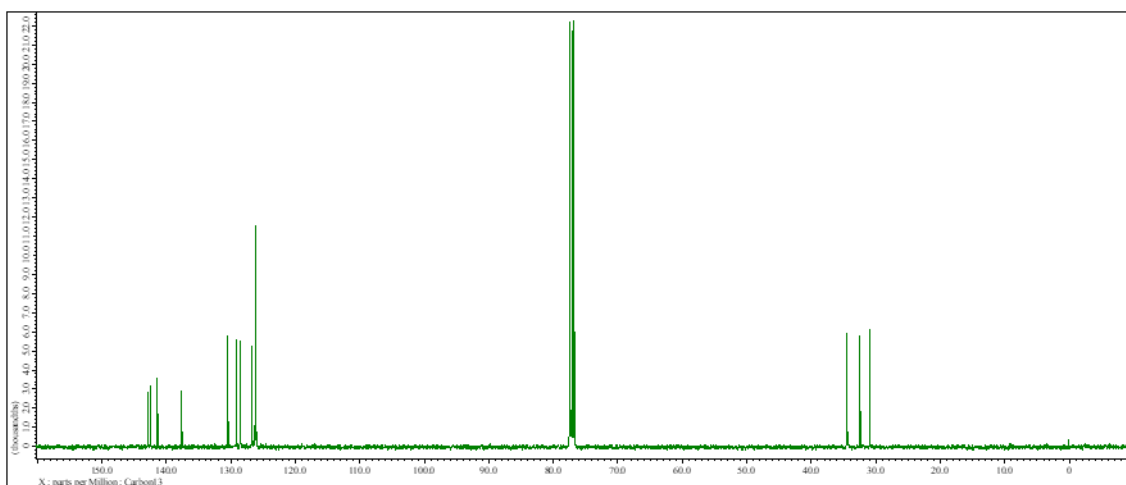


Figure S56. ¹³C-NMR spectra of DBDB[7] (100 MHz, CDCl₃)

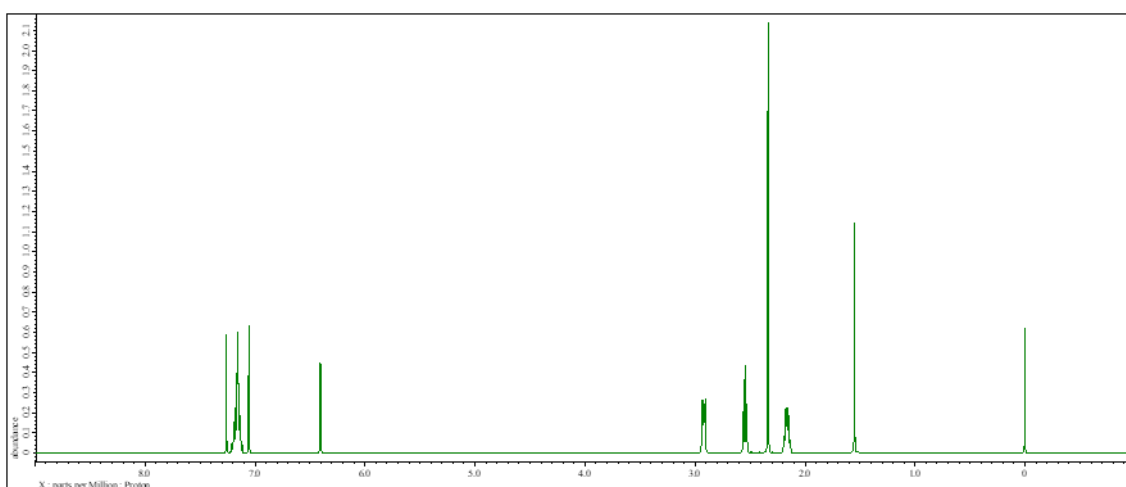


Figure S57. ¹H-NMR spectra of DBDMDB[7] (399 MHz, CDCl₃).

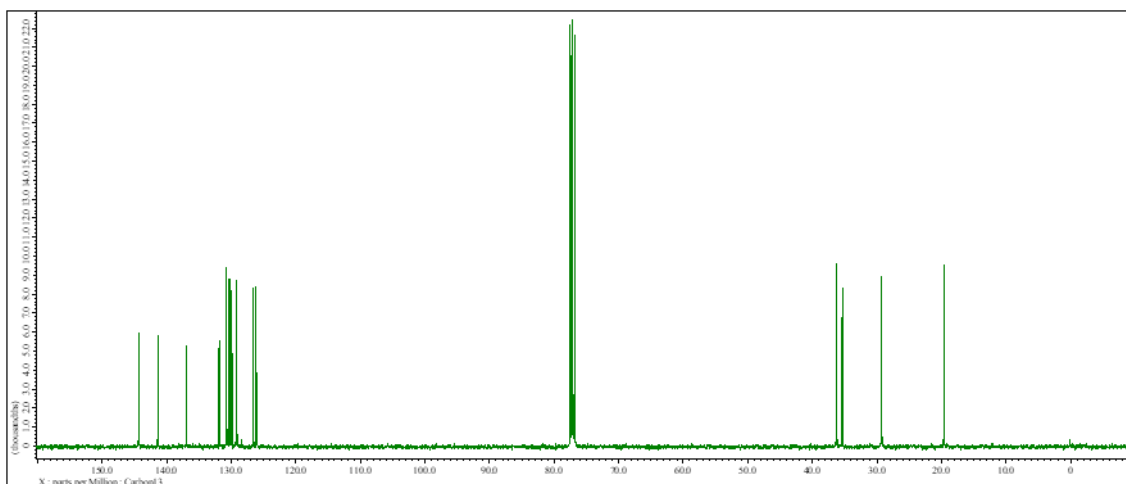


Figure S58. ¹³C-NMR spectra of **DBDMDB[7]** (100 MHz, CDCl₃).

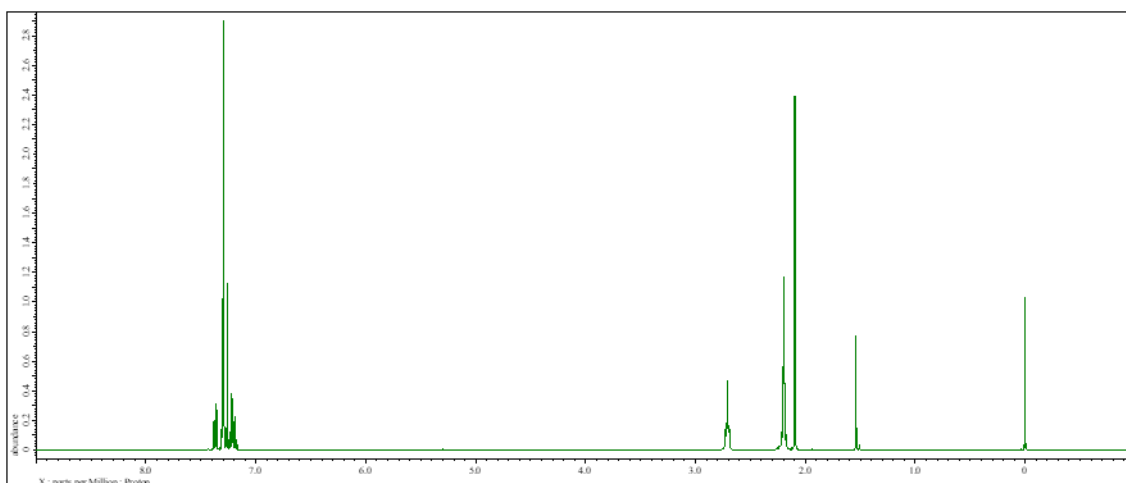


Figure S59. ¹H-NMR spectra of **DB α MDB[7]** (399 MHz, CDCl₃).

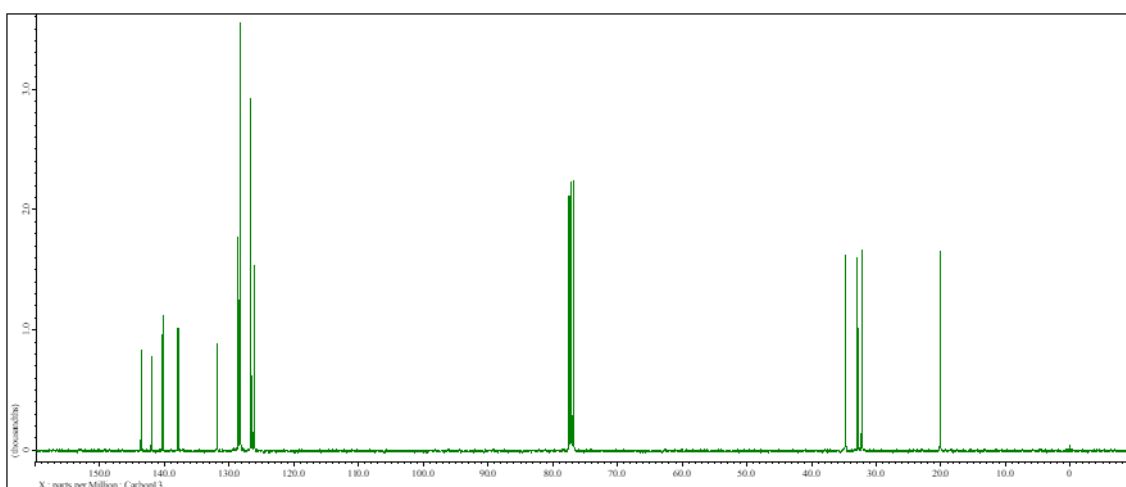


Figure S60. ¹³C-NMR spectra of **DB α MDB[7]** (100MHz, CDCl₃).

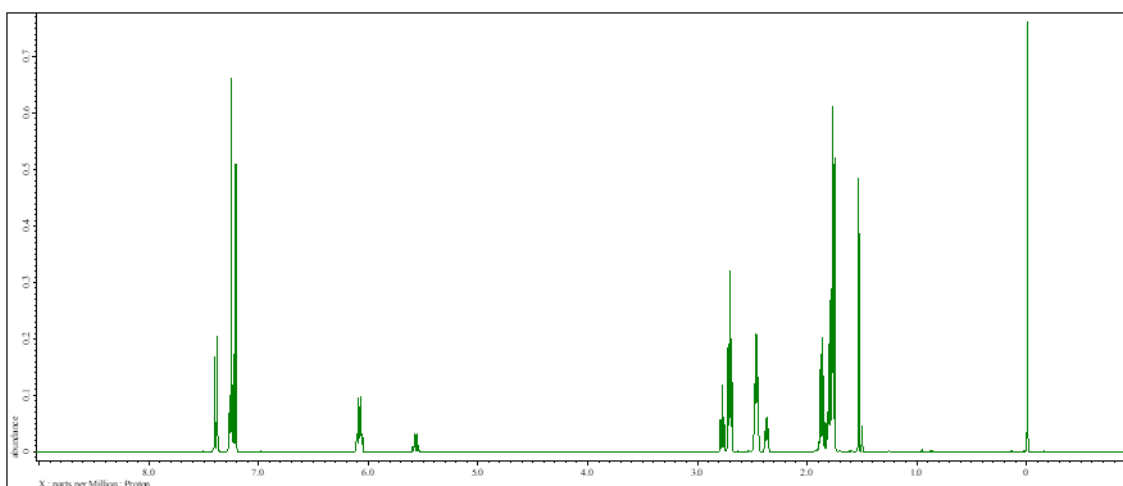


Figure S61. $^1\text{H-NMR}$ spectra of **11** (399 MHz, CDCl_3).

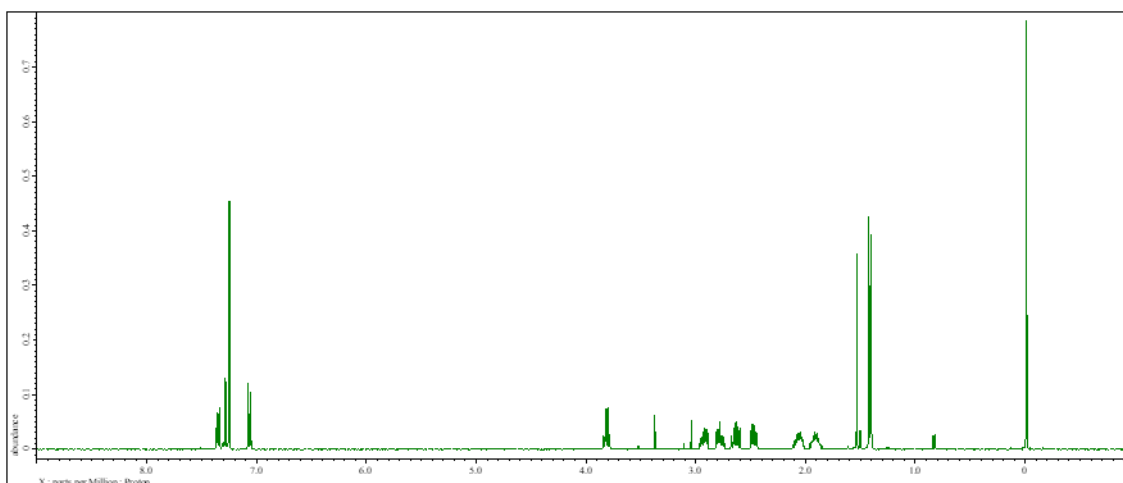


Figure S62. $^1\text{H-NMR}$ spectra of **12** (399 MHz, CDCl_3).

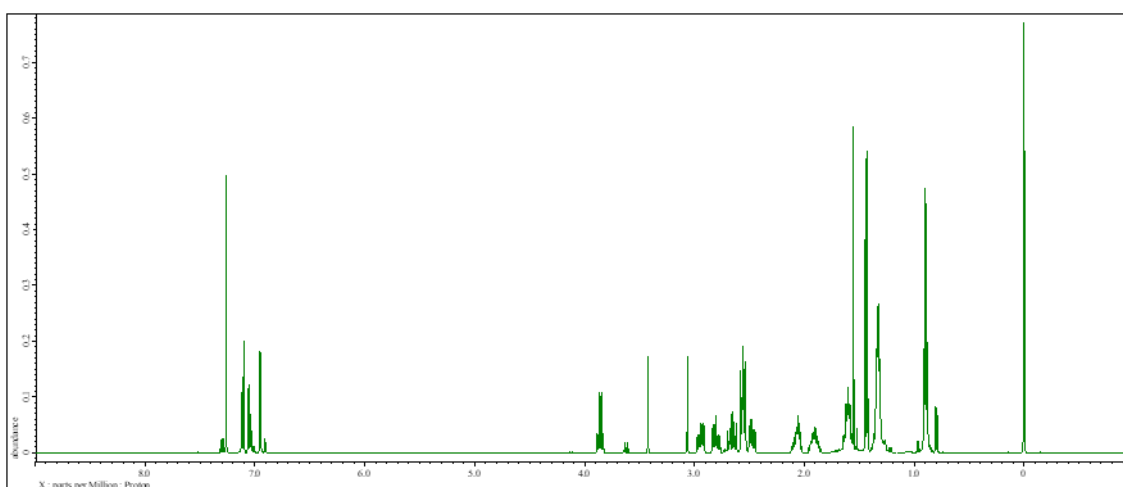


Figure S63. $^1\text{H-NMR}$ spectra of **13** (399 MHz, CDCl_3).

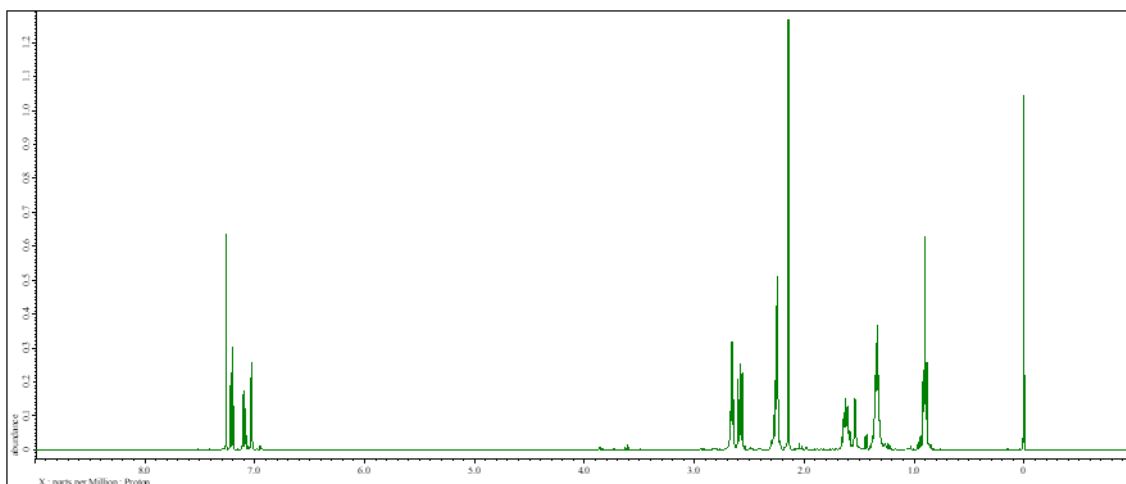


Figure S64. ^1H -NMR spectra of **14** (399 MHz, CDCl_3).

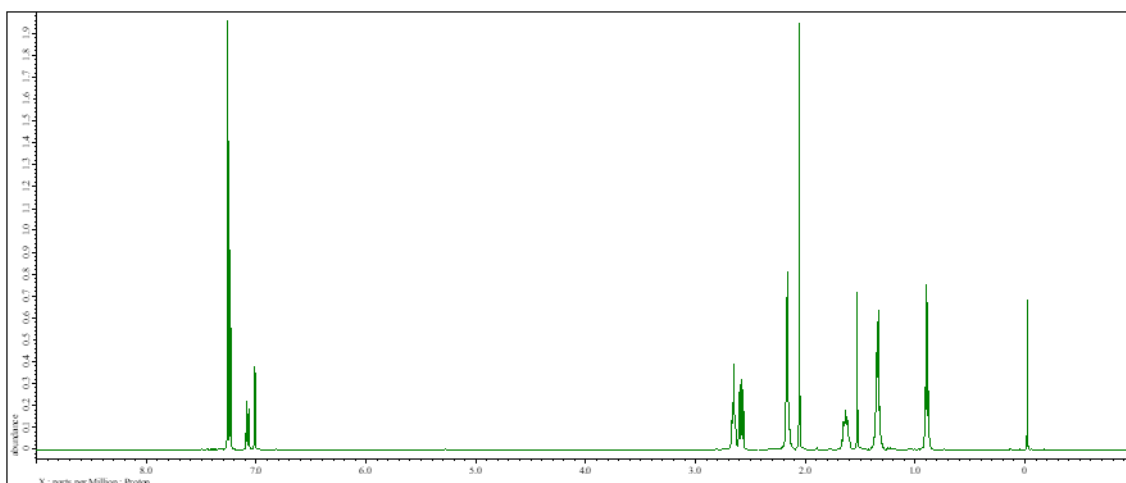


Figure S65. ^1H -NMR spectra of **5DBaMDB[7]** (399 MHz, CDCl_3).

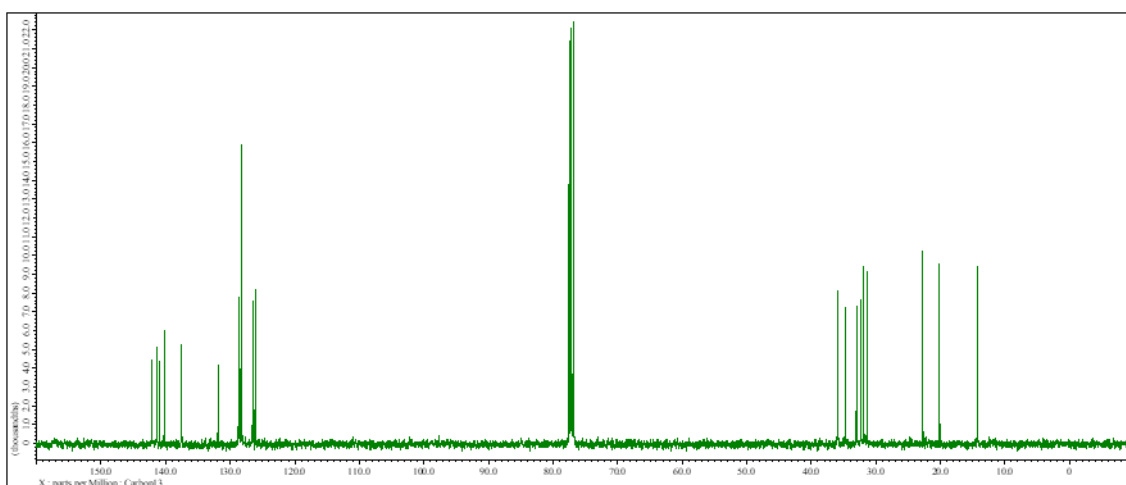


Figure S66. ^{13}C -NMR spectra of **5DBaMDB[7]** (100 MHz, CDCl_3).

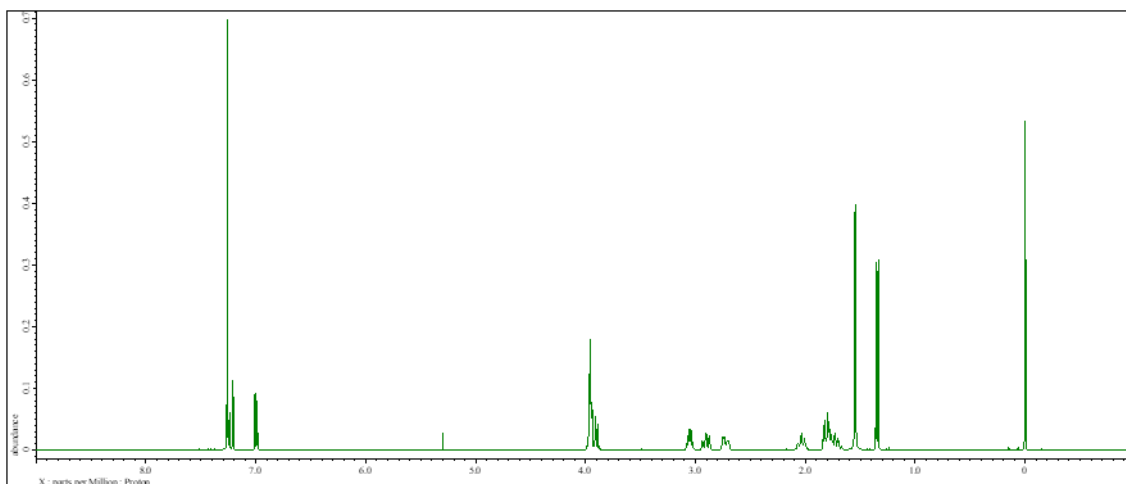


Figure S67. ^1H -NMR spectra of **15** (399 MHz, CDCl_3).

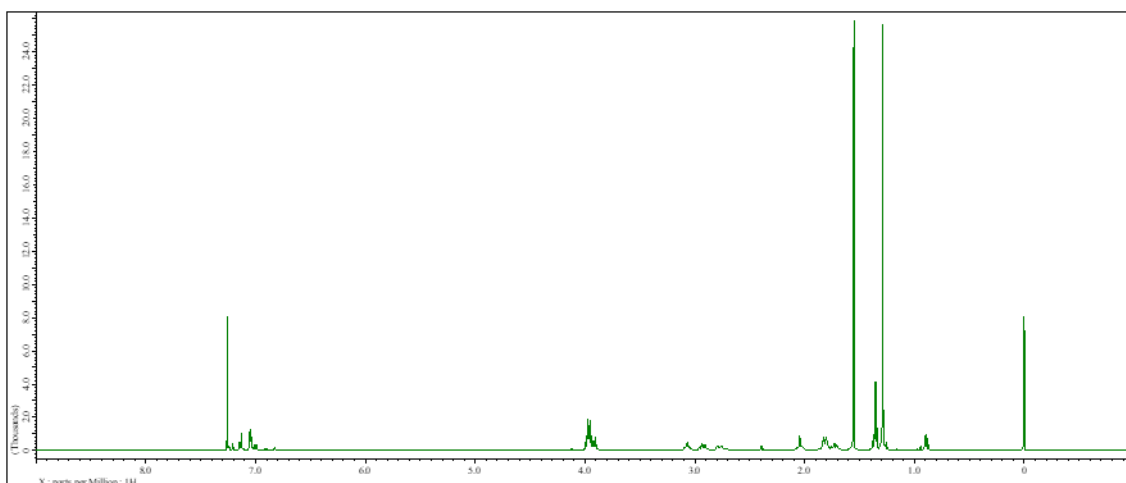


Figure S68. ^1H -NMR spectra of **16** (500 MHz, CDCl_3).

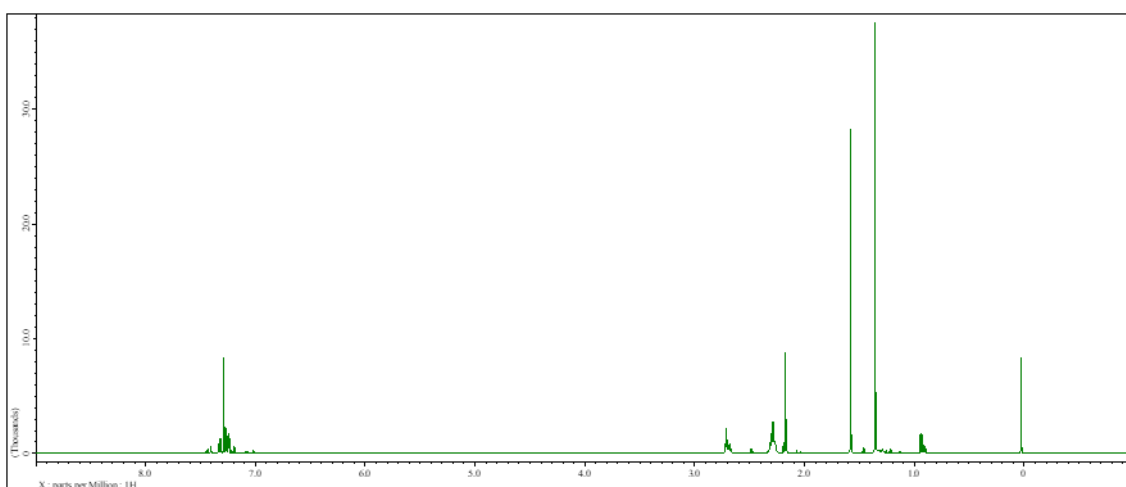


Figure S69. ^1H -NMR spectra of **18** (500 MHz, CDCl_3).

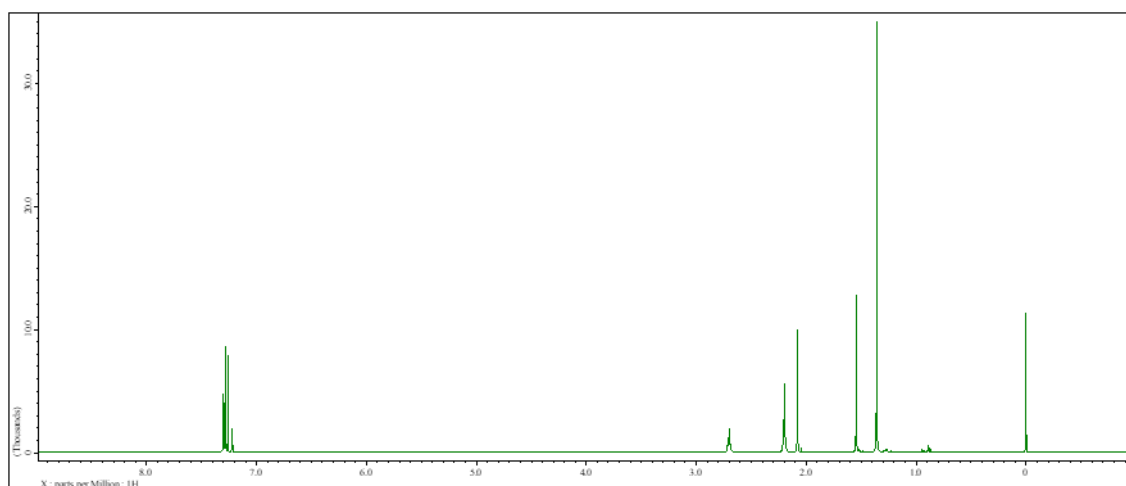


Figure S70. ^1H -NMR spectra of **'BuDB α MDB[7]** (500 MHz, CDCl_3).

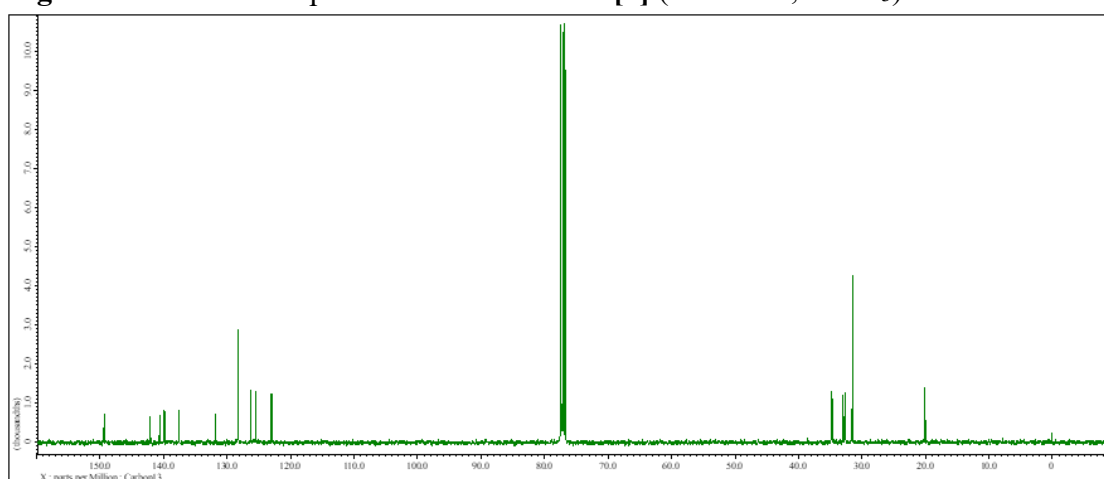


Figure S71. ^{13}C -NMR spectra of **'BuDB α MDB[7]** (100 MHz, CDCl_3).

10. High resolution mass spectrometry (HRMS) chart

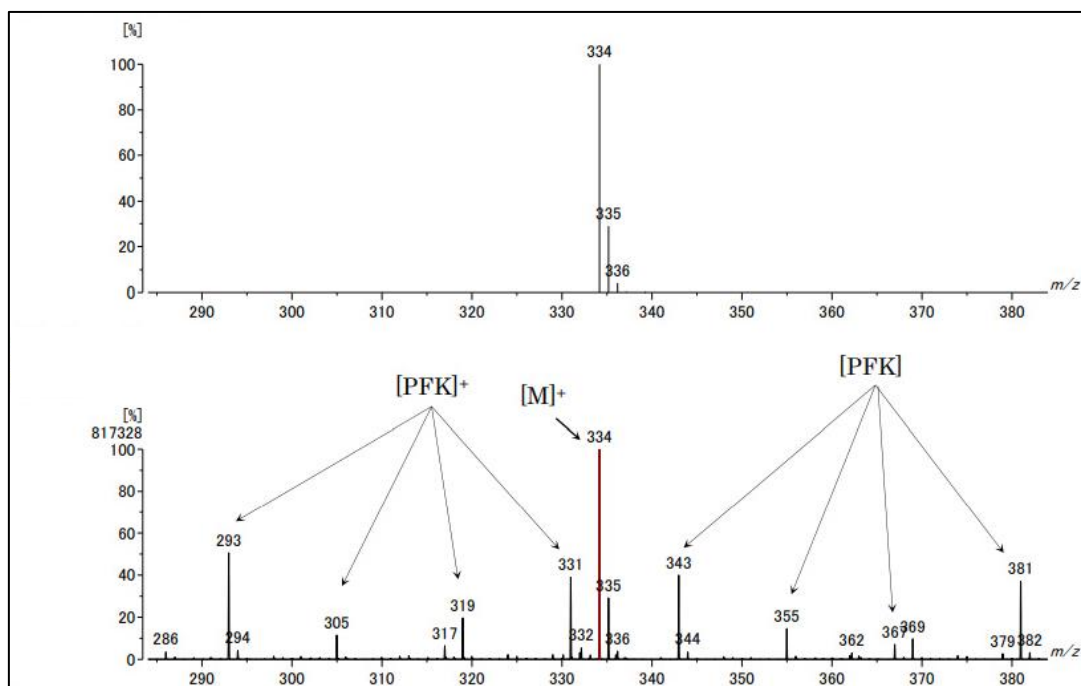


Figure S72. HR-MS spectra of DBDB[6] (above: calculated data; below: experimental data).

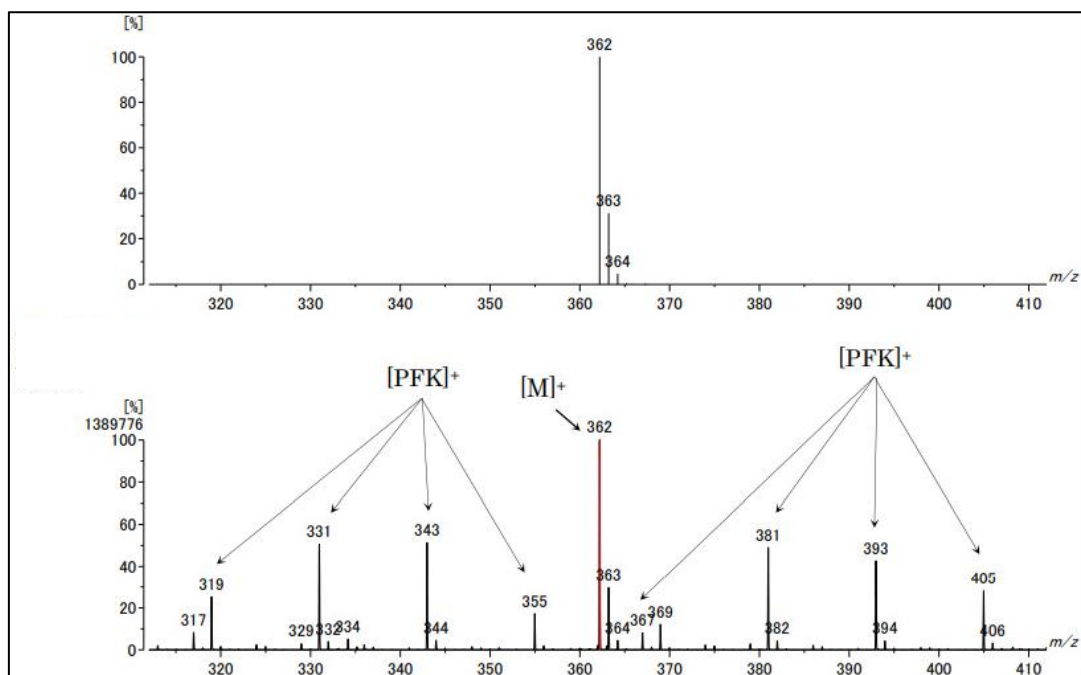


Figure S73. HR-MS spectra of DBDB[7] (above: calculated data; below: experimental data).

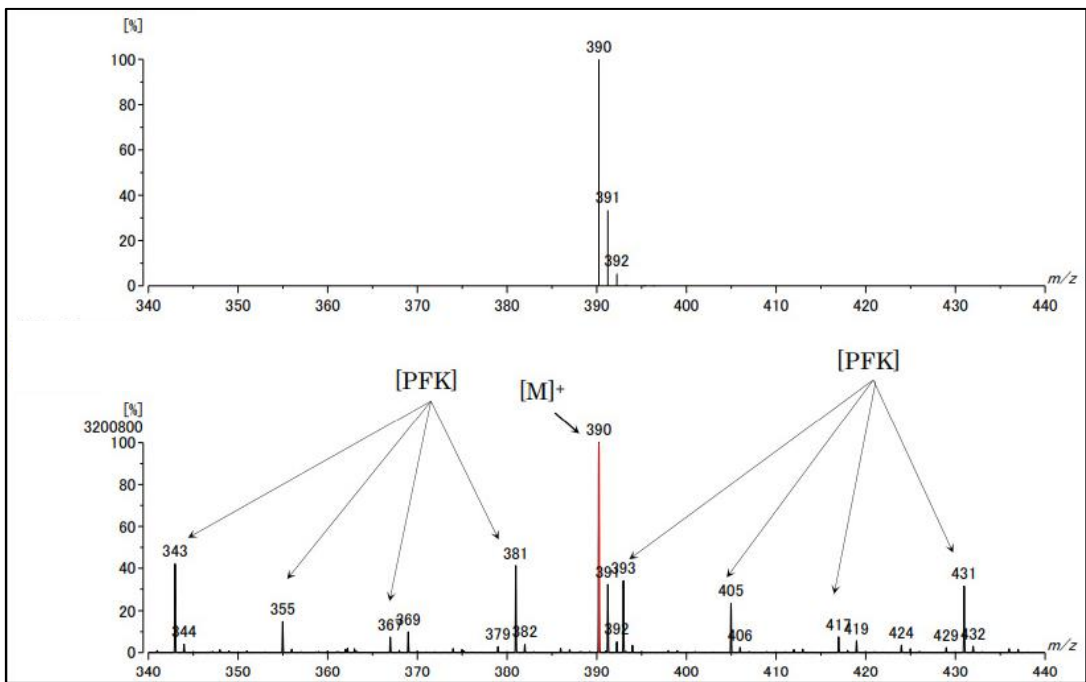


Figure S74. HR-MS spectra of **DBDMDB[7]** (above: calculated data; below: experimental data).

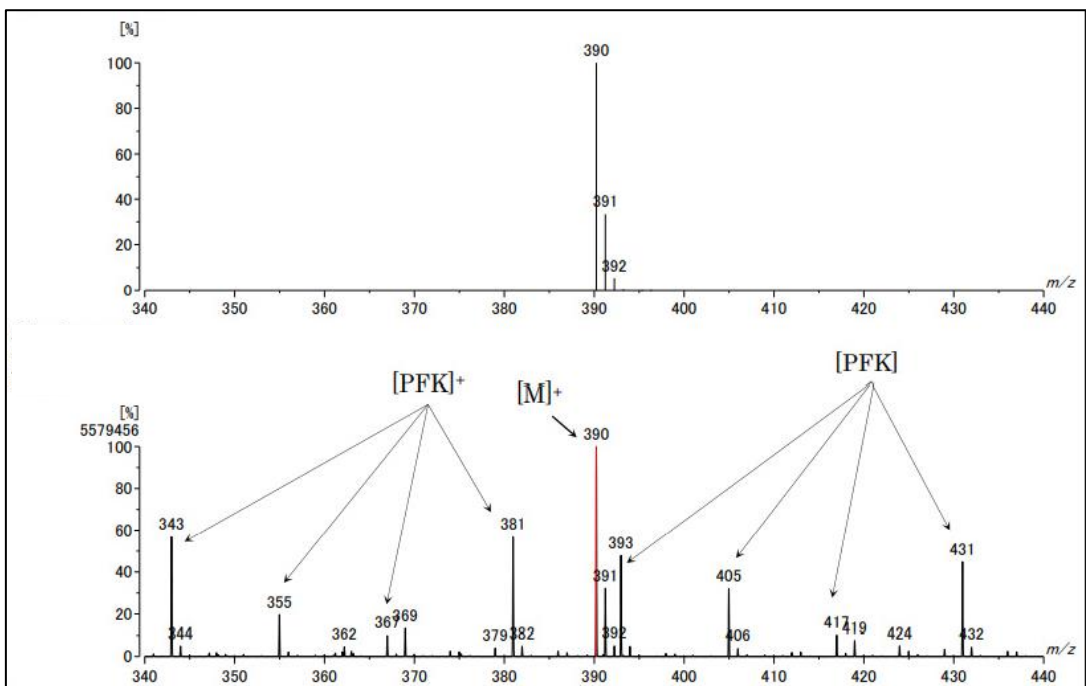


Figure S75. HR-MS spectra of **DBaMDB[7]** (above: calculated data; below: experimental data).

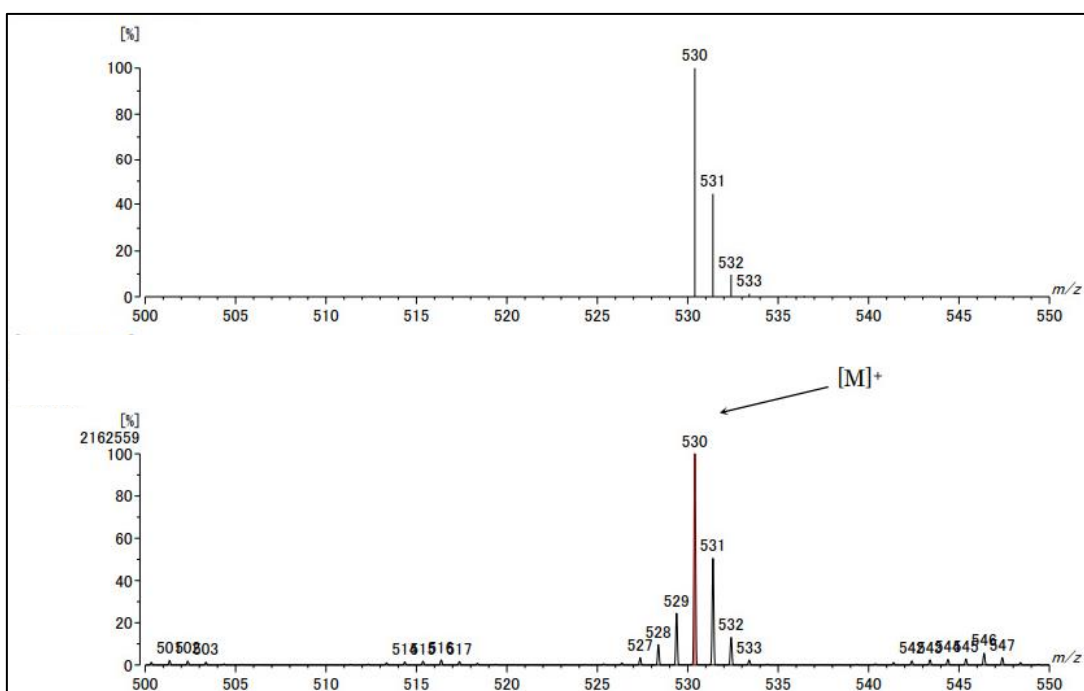


Figure S76. HR-MS spectra of **5DBaMDB[7]** (above: calculated data; below: experimental data).

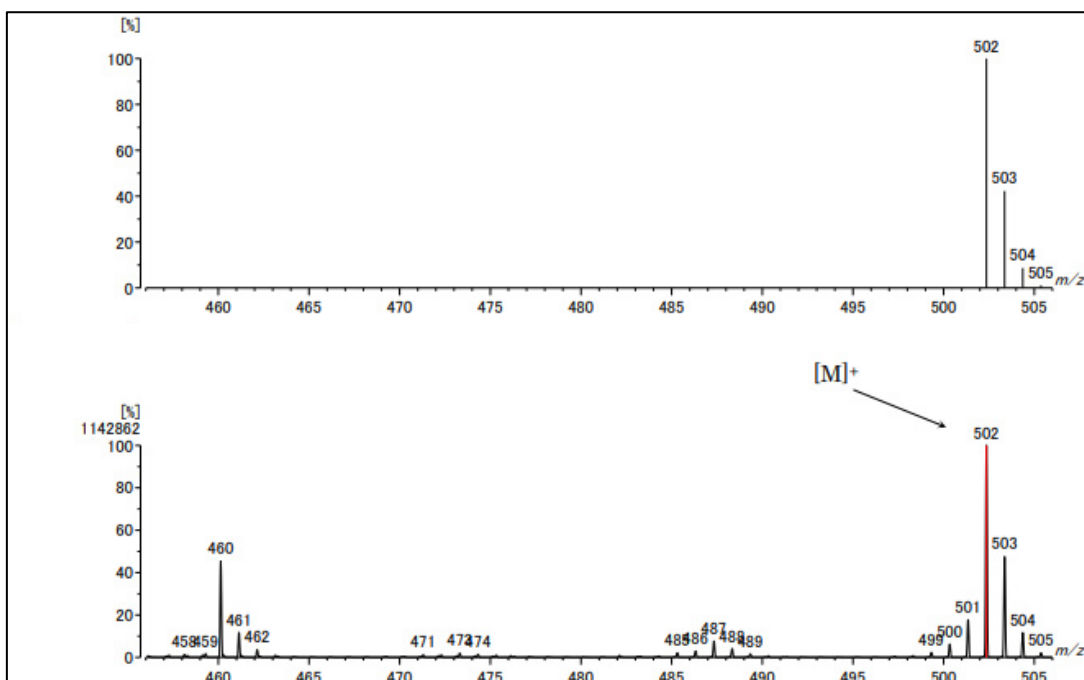


Figure S77. HR-MS spectra of **'BuDBaMDB[7]** (above: calculated data; below: experimental data).

Crystal Structures (Cambridge Crystallographic Data Centre)

CCDC 2168494-2168501

checkCIF/PLATON report

You have not supplied any structure factors. As a result the full set of tests cannot be run.

THIS REPORT IS FOR GUIDANCE ONLY. IF USED AS PART OF A REVIEW PROCEDURE FOR PUBLICATION, IT SHOULD NOT REPLACE THE EXPERTISE OF AN EXPERIENCED CRYSTALLOGRAPHIC REFEREE.

No syntax errors found. CIF dictionary Interpreting this report

Datablock: DBDB7

Bond precision: C-C = 0.0027 A

Wavelength=0.71073

Cell: a=7.8987(6) b=9.0080(6) c=13.8356(14)
 alpha=99.206(7) beta=102.460(8) gamma=90.999(6)
Temperature: 100 K

	Calculated	Reported
Volume	947.48(14)	947.48(14)
Space group	P -1	P -1
Hall group	-P 1	-P 1
Moiety formula	C28 H26	C28 H26
Sum formula	C28 H26	C28 H26
Mr	362.49	362.49
Dx, g cm ⁻³	1.271	1.271
Z	2	2
Mu (mm ⁻¹)	0.071	0.071
F000	388.0	388.0
F000'	388.14	
h, k, lmax	11, 13, 20	11, 13, 19
Nref	6095	5227
Tmin, Tmax	0.992, 0.999	0.784, 1.000
Tmin'	0.979	

Correction method= # Reported T Limits: Tmin=0.784 Tmax=1.000
AbsCorr = MULTI-SCAN

Data completeness= 0.858

Theta(max)= 31.103

R(reflections)= 0.0758(3984)

wR2(reflections)=
0.1698(5227)

S = 1.092

Npar= 253

The following ALERTS were generated. Each ALERT has the format

test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

● Alert level G

PLAT012_ALERT_1_G	No	_shelx_res_checksum Found in CIF	Please Check
PLAT720_ALERT_4_G	Number of Unusual/Non-Standard Labels		3 Note
PLAT767_ALERT_4_G	INS Embedded LIST 6 Instruction Should be LIST 4		Please Check
PLAT941_ALERT_3_G	Average HKL Measurement Multiplicity		3.2 Low

- 0 **ALERT level A** = Most likely a serious problem - resolve or explain
- 0 **ALERT level B** = A potentially serious problem, consider carefully
- 0 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight
- 4 **ALERT level G** = General information/check it is not something unexpected

- 1 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
 - 0 ALERT type 2 Indicator that the structure model may be wrong or deficient
 - 1 ALERT type 3 Indicator that the structure quality may be low
 - 2 ALERT type 4 Improvement, methodology, query or suggestion
 - 0 ALERT type 5 Informative message, check
-

It is advisable to attempt to resolve as many as possible of the alerts in all categories. Often the minor alerts point to easily fixed oversights, errors and omissions in your CIF or refinement strategy, so attention to these fine details can be worthwhile. In order to resolve some of the more serious problems it may be necessary to carry out additional measurements or structure refinements. However, the purpose of your study may justify the reported deviations and the more serious of these should normally be commented upon in the discussion or experimental section of a paper or in the "special_details" fields of the CIF. checkCIF was carefully designed to identify outliers and unusual parameters, but every test has its limitations and alerts that are not important in a particular case may appear. Conversely, the absence of alerts does not guarantee there are no aspects of the results needing attention. It is up to the individual to critically assess their own results and, if necessary, seek expert advice.

Publication of your CIF in IUCr journals

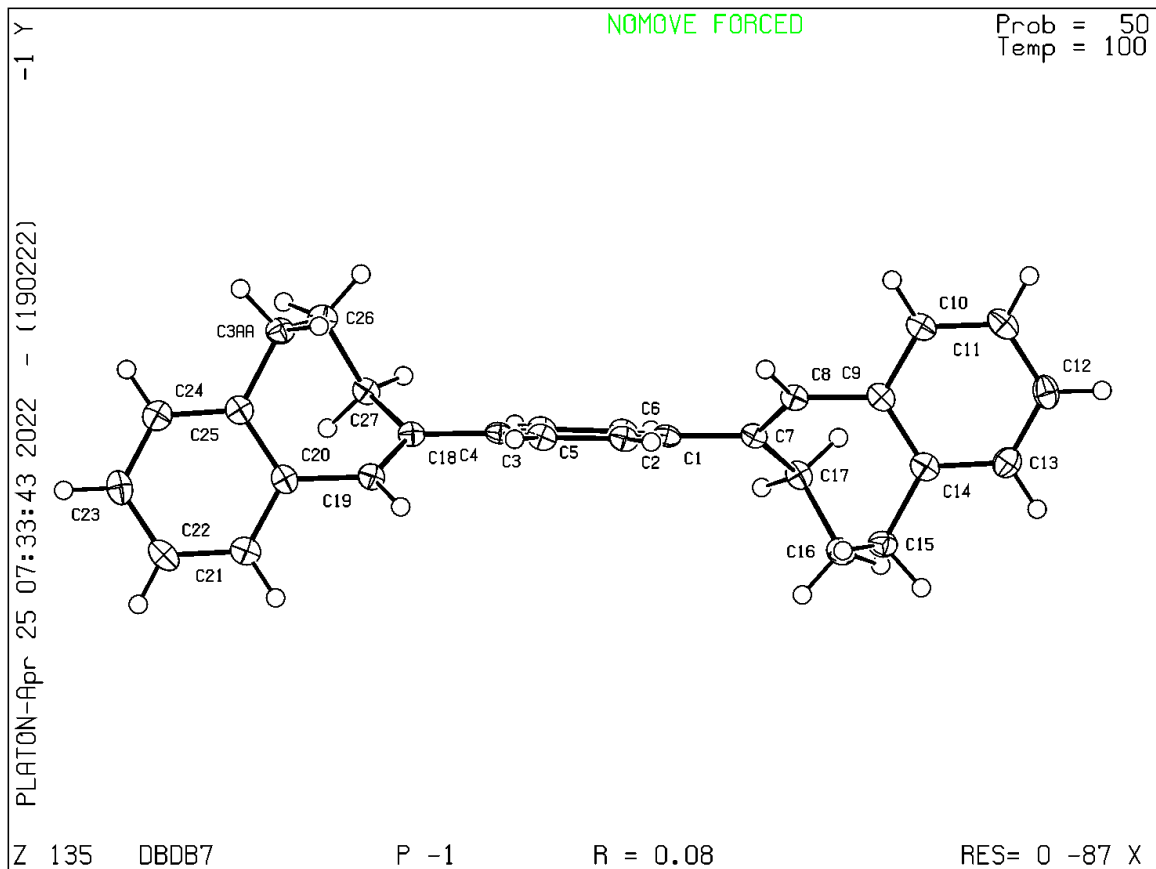
A basic structural check has been run on your CIF. These basic checks will be run on all CIFs submitted for publication in IUCr journals (*Acta Crystallographica*, *Journal of Applied Crystallography*, *Journal of Synchrotron Radiation*); however, if you intend to submit to *Acta Crystallographica Section C* or *E* or *IUCrData*, you should make sure that **full publication checks** are run on the final version of your CIF prior to submission.

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Please refer to the *Notes for Authors* of the relevant journal for any special instructions relating to CIF submission.

PLATON version of 19/02/2022; check.def file version of 19/02/2022

Datablock DBDB7 - ellipsoid plot



The following ALERTS were generated. Each ALERT has the format

test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

● **Alert level C**

DIFMX02_ALERT_1_C The maximum difference density is > 0.1*ZMAX*0.75

The relevant atom site should be identified.

PLAT082_ALERT_2_C High R1 Value	0.11	Report
PLAT084_ALERT_3_C High wR2 Value (i.e. > 0.25)	0.32	Report
PLAT097_ALERT_2_C Large Reported Max. (Positive) Residual Density	0.58	eA-3
PLAT340_ALERT_3_C Low Bond Precision on C-C Bonds	0.00688	Ang.

● **Alert level G**

CHEMS02_ALERT_1_G Please check that you have entered the correct

`_publ_requested_category` classification of your compound;

FI or CI or EI for inorganic; FM or CM or EM for metal-organic;

FO or CO or EO for organic.

From the CIF: `_publ_requested_category` CHOOSE FI FM FO CI CM CO or A

From the CIF: `_chemical_formula_sum` :C30 H30

PLAT012_ALERT_1_G N.O.K. <code>_shelx_res_checksum</code> Found in CIF	Please	Check
PLAT072_ALERT_2_G SHELXL First Parameter in WGHT Unusually Large	0.10	Report
PLAT083_ALERT_2_G SHELXL Second Parameter in WGHT Unusually Large	5.52	Why ?
PLAT941_ALERT_3_G Average HKL Measurement Multiplicity	3.0	Low

- 0 **ALERT level A** = Most likely a serious problem - resolve or explain
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- 5 **ALERT level G** = General information/check it is not something unexpected

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 - 3 ALERT type 3 Indicator that the structure quality may be low
 - 0 ALERT type 4 Improvement, methodology, query or suggestion
 - 0 ALERT type 5 Informative message, check
-
-

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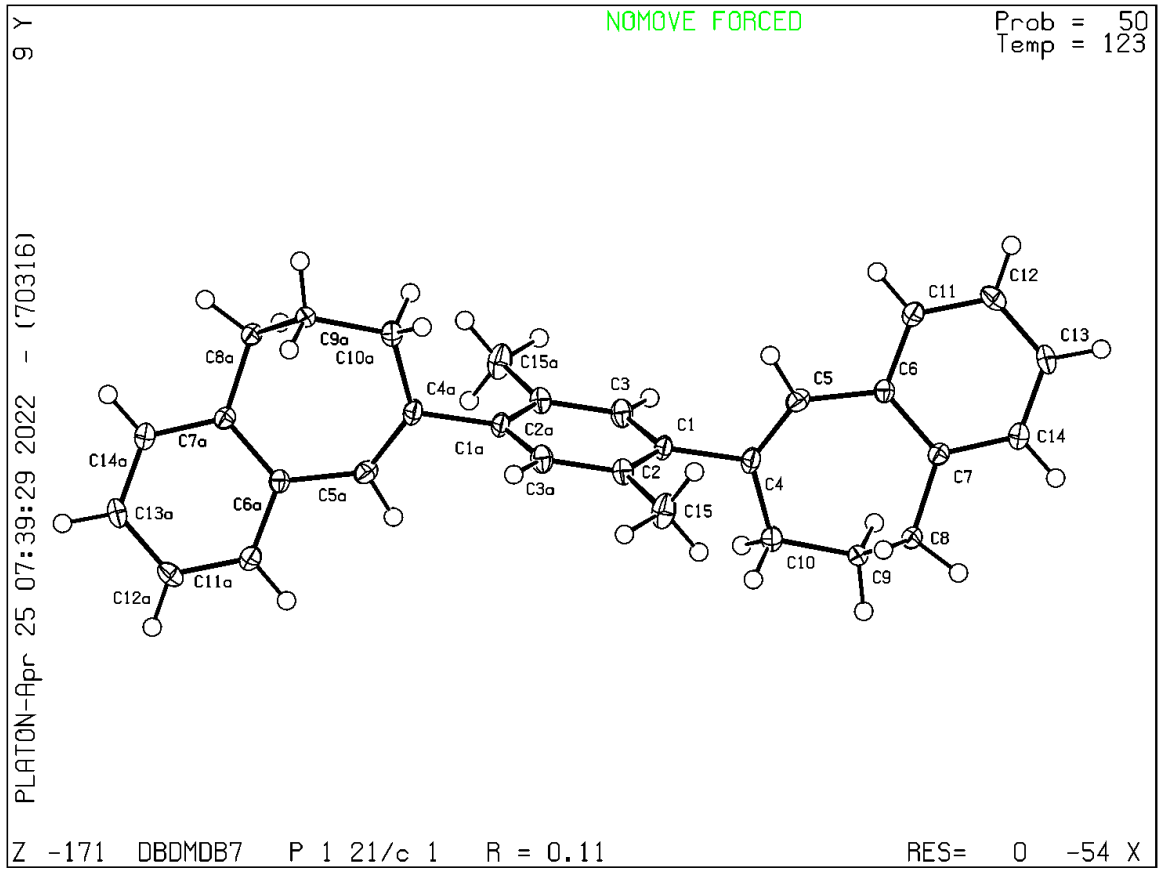
Publication of your CIF in IUCr journals

A basic structural check has been run on your CIF. These basic checks will be run on all CIFs submitted for publication in IUCr journals (*Acta Crystallographica*, *Journal of Applied Crystallography*, *Journal of Synchrotron Radiation*); however, if you intend to submit to *Acta Crystallographica Section C* or *E* or *IUCrData*, you should make sure that full publication checks are run on the final version of your CIF prior to submission.

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PLATON version of 19/02/2022; check.def file version of 19/02/2022



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test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

Alert level G

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PLAT013_ALERT_1_G	N.O.K.	_shelx_hkl_checksum Found in CIF	Please Check
PLAT042_ALERT_1_G	Calc. and Reported Moiety Formula Strings Differ		Please Check
PLAT045_ALERT_1_G	Calculated and Reported Z Differ by a Factor ...	0.400	Check
PLAT941_ALERT_3_G	Average HKL Measurement Multiplicity	2.7	Low

0 **ALERT level A** = Most likely a serious problem - resolve or explain
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Publication of your CIF in IUCr journals

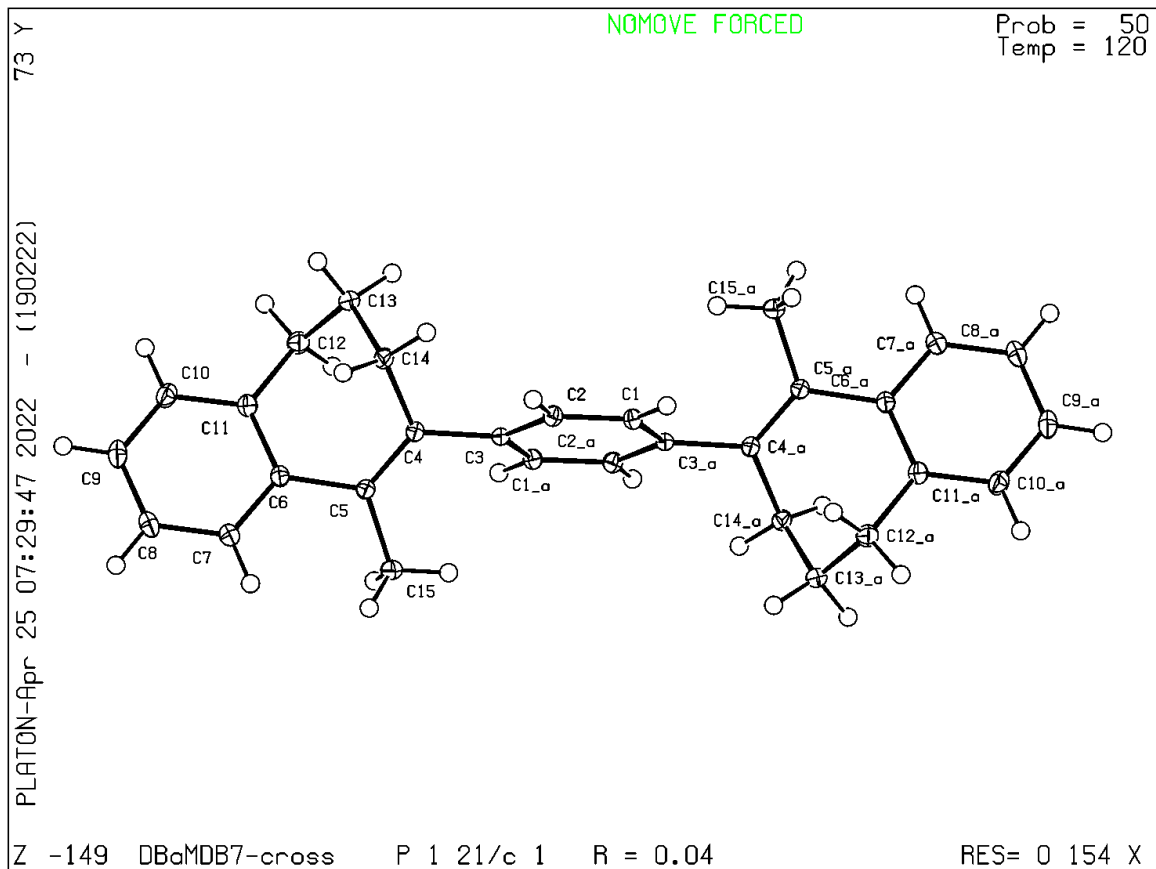
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Datablock DBaMDB7-cross - ellipsoid plot



The following ALERTS were generated. Each ALERT has the format

test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

Alert level G

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PLAT042_ALERT_1_G	Calc.	and Reported Moiety Formula Strings Differ	Please Check
PLAT045_ALERT_1_G	Calculated	and Reported Z Differ by a Factor ...	0.400 Check
PLAT941_ALERT_3_G	Average HKL	Measurement Multiplicity	2.7 Low

- 0 **ALERT level A** = Most likely a serious problem - resolve or explain
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- 4 **ALERT level G** = General information/check it is not something unexpected

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 - 1 ALERT type 3 Indicator that the structure quality may be low
 - 0 ALERT type 4 Improvement, methodology, query or suggestion
 - 0 ALERT type 5 Informative message, check
-

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Publication of your CIF in IUCr journals

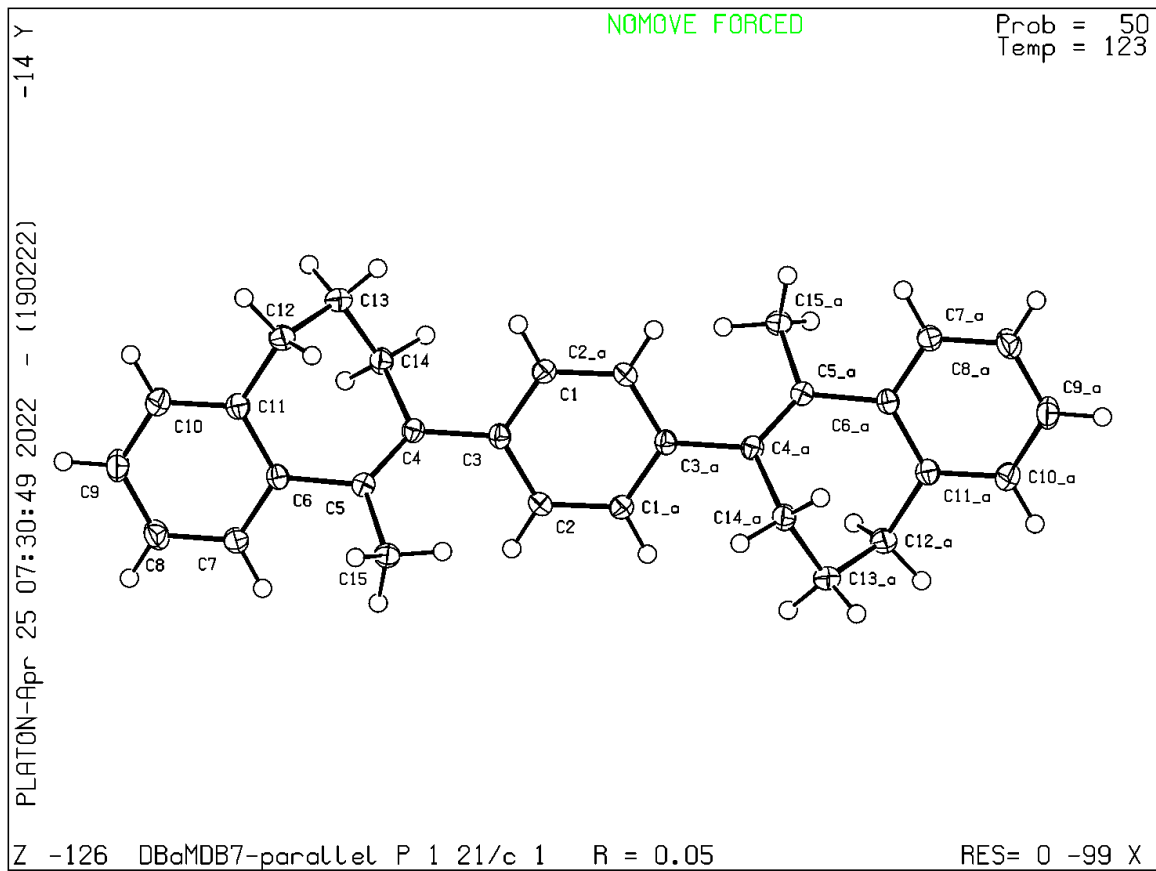
A basic structural check has been run on your CIF. These basic checks will be run on all CIFs submitted for publication in IUCr journals (*Acta Crystallographica*, *Journal of Applied Crystallography*, *Journal of Synchrotron Radiation*); however, if you intend to submit to *Acta Crystallographica Section C* or *E* or *IUCrData*, you should make sure that full publication checks are run on the final version of your CIF prior to submission.

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Datablock DBaMDB7-parallel - ellipsoid plot



The following ALERTS were generated. Each ALERT has the format

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Click on the hyperlinks for more details of the test.

● **Alert level G**

CHEMS02_ALERT_1_G Please check that you have entered the correct
_publ_requested_category classification of your compound;
FI or CI or EI for inorganic; FM or CM or EM for metal-organic;
FO or CO or EO for organic.
From the CIF: _publ_requested_category CHOOSE FI FM FO CI CM CO or A
From the CIF: _chemical_formula_sum :C40 H50
PLAT012_ALERT_1_G N.O.K. _shelx_res_checksum Found in CIF Please Check
PLAT941_ALERT_3_G Average HKL Measurement Multiplicity 3.1 Low

- 0 **ALERT level A** = Most likely a serious problem - resolve or explain
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-
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 - 1 ALERT type 3 Indicator that the structure quality may be low
 - 0 ALERT type 4 Improvement, methodology, query or suggestion
 - 0 ALERT type 5 Informative message, check
-

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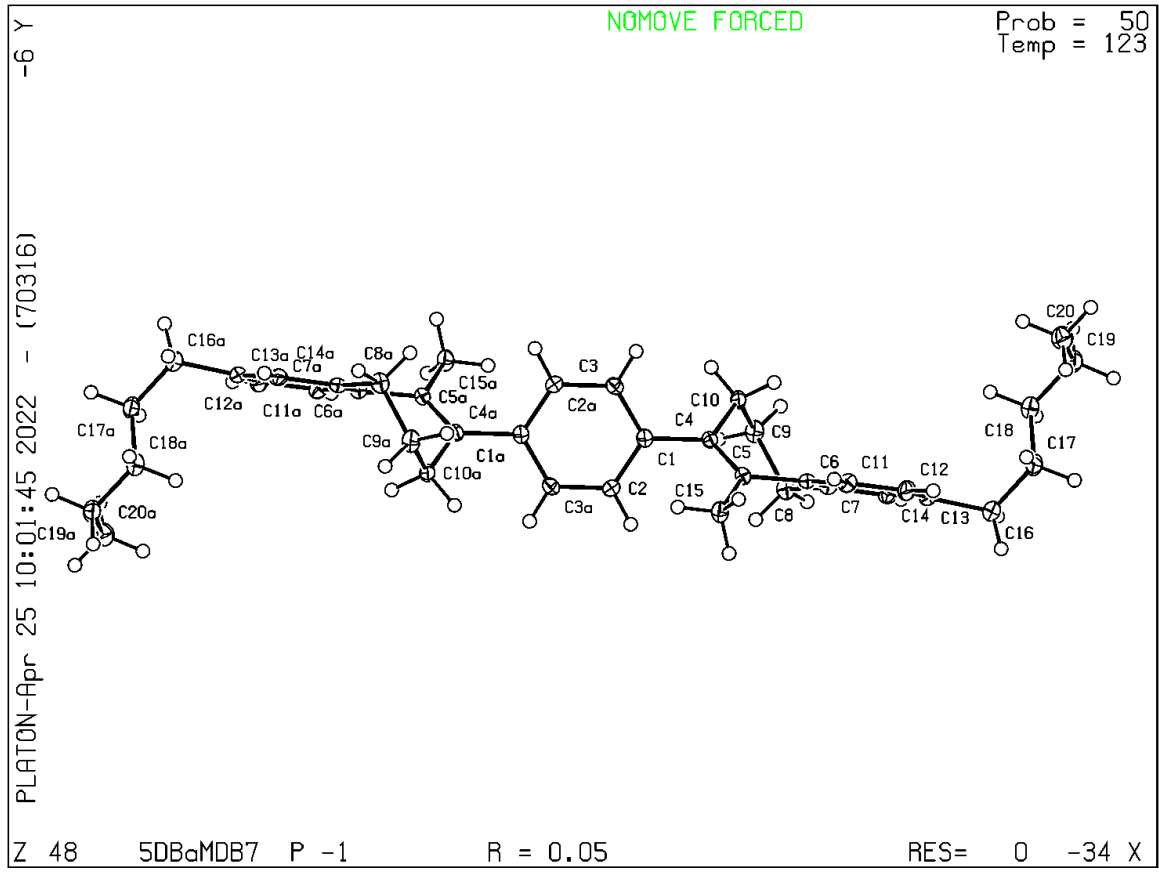
Publication of your CIF in IUCr journals

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PLATON version of 19/02/2022; check.def file version of 19/02/2022



checkCIF/PLATON report

You have not supplied any structure factors. As a result the full set of tests cannot be run.

THIS REPORT IS FOR GUIDANCE ONLY. IF USED AS PART OF A REVIEW PROCEDURE FOR PUBLICATION, IT SHOULD NOT REPLACE THE EXPERTISE OF AN EXPERIENCED CRYSTALLOGRAPHIC REFEREE.

No syntax errors found. CIF dictionary Interpreting this report

Datablock: tBuDBaMDB7

Bond precision: C-C = 0.0031 A

Wavelength=0.71073

Cell: a=5.8767 (4) b=6.1160 (6) c=20.3049 (14)
 alpha=93.452 (7) beta=91.590 (6) gamma=93.560 (7)
Temperature: 100 K

	Calculated	Reported
Volume	726.70 (10)	726.70 (10)
Space group	P -1	P -1
Hall group	-P 1	-P 1
Moiety formula	C38 H46	C19 H23
Sum formula	C38 H46	C19 H23
Mr	502.75	251.37
Dx, g cm ⁻³	1.149	1.149
Z	1	2
Mu (mm ⁻¹)	0.064	0.064
F000	274.0	274.0
F000'	274.09	
h, k, lmax	8, 8, 29	8, 8, 28
Nref	4664	3951
Tmin, Tmax	0.989, 0.998	0.654, 1.000
Tmin'	0.987	

Correction method= # Reported T Limits: Tmin=0.654 Tmax=1.000
AbsCorr = MULTI-SCAN

Data completeness= 0.847

Theta (max)= 31.066

R(reflections)= 0.0816 (3172)

wR2(reflections)=
0.2331 (3951)

S = 1.088

Npar= 176

The following ALERTS were generated. Each ALERT has the format
test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

● **Alert level C**

DIFMX02_ALERT_1_C The maximum difference density is > 0.1*ZMAX*0.75

The relevant atom site should be identified.

PLAT097_ALERT_2_C Large Reported Max. (Positive) Residual Density 0.58 eA-3

● **Alert level G**

PLAT012_ALERT_1_G	N.O.K. _shelx_res_checksum Found in CIF	Please Check
PLAT013_ALERT_1_G	N.O.K. _shelx_hkl_checksum Found in CIF	Please Check
PLAT042_ALERT_1_G	Calc. and Reported Moiety Formula Strings Differ	Please Check
PLAT045_ALERT_1_G	Calculated and Reported Z Differ by a Factor ...	0.500 Check
PLAT072_ALERT_2_G	SHELXL First Parameter in WGHT Unusually Large	0.11 Report
PLAT180_ALERT_4_G	Check Cell Rounding: # of Values Ending with 0 =	3 Note
PLAT767_ALERT_4_G	INS Embedded LIST 6 Instruction Should be LIST 4	Please Check
PLAT941_ALERT_3_G	Average HKL Measurement Multiplicity	2.6 Low

- 0 **ALERT level A** = Most likely a serious problem - resolve or explain
- 0 **ALERT level B** = A potentially serious problem, consider carefully
- 2 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight
- 8 **ALERT level G** = General information/check it is not something unexpected

- 5 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
 - 2 ALERT type 2 Indicator that the structure model may be wrong or deficient
 - 1 ALERT type 3 Indicator that the structure quality may be low
 - 2 ALERT type 4 Improvement, methodology, query or suggestion
 - 0 ALERT type 5 Informative message, check
-
-

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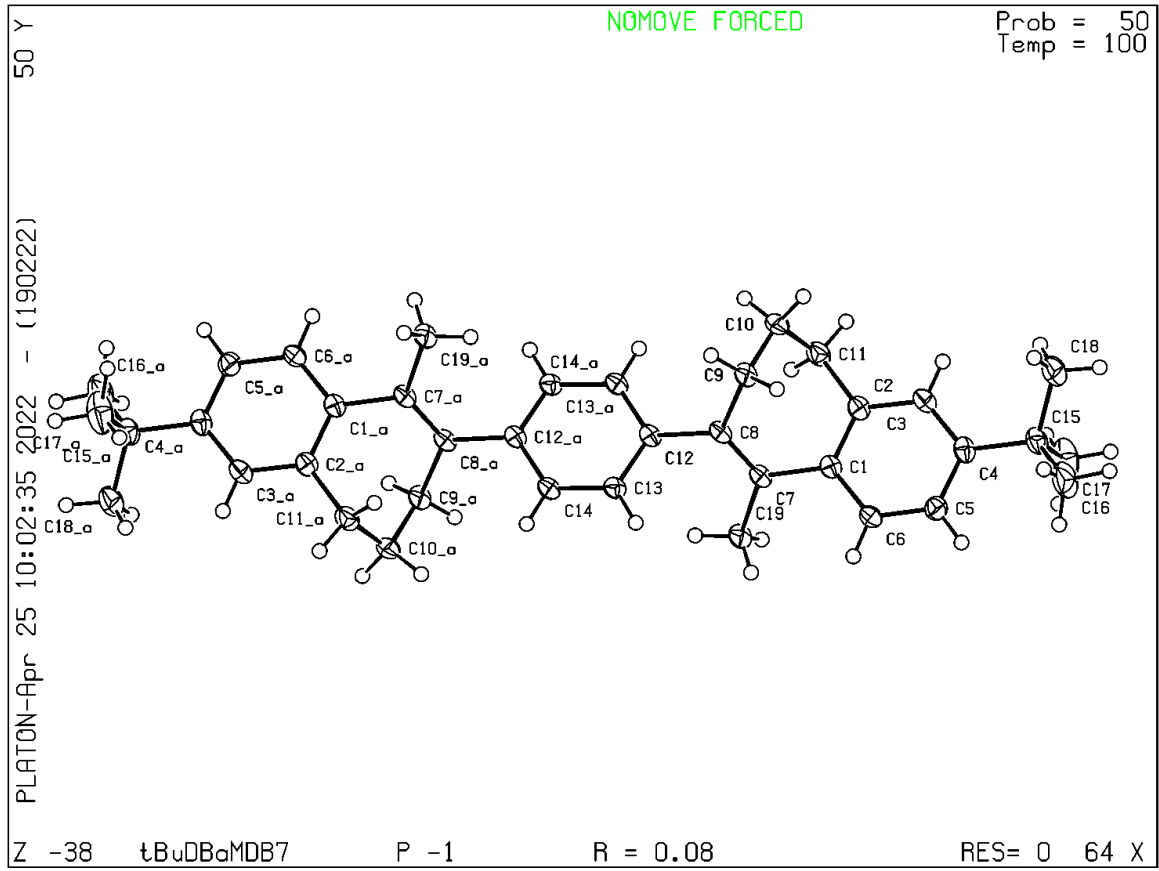
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test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

Alert level C

DIFMX02_ALERT_1_C The maximum difference density is $> 0.1 * ZMAX * 0.75$

The relevant atom site should be identified.

PLAT094_ALERT_2_C Ratio of Maximum / Minimum Residual Density 2.02 Report

PLAT097_ALERT_2_C Large Reported Max. (Positive) Residual Density 0.51 eA-3

Alert level G

PLAT012_ALERT_1_G No _shelx_res_checksum Found in CIF Please Check

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 - 3 **ALERT level C** = Check. Ensure it is not caused by an omission or oversight
 - 1 **ALERT level G** = General information/check it is not something unexpected
-
- 2 ALERT type 1 CIF construction/syntax error, inconsistent or missing data
 - 2 ALERT type 2 Indicator that the structure model may be wrong or deficient
 - 0 ALERT type 3 Indicator that the structure quality may be low
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 - 0 ALERT type 5 Informative message, check
-
-

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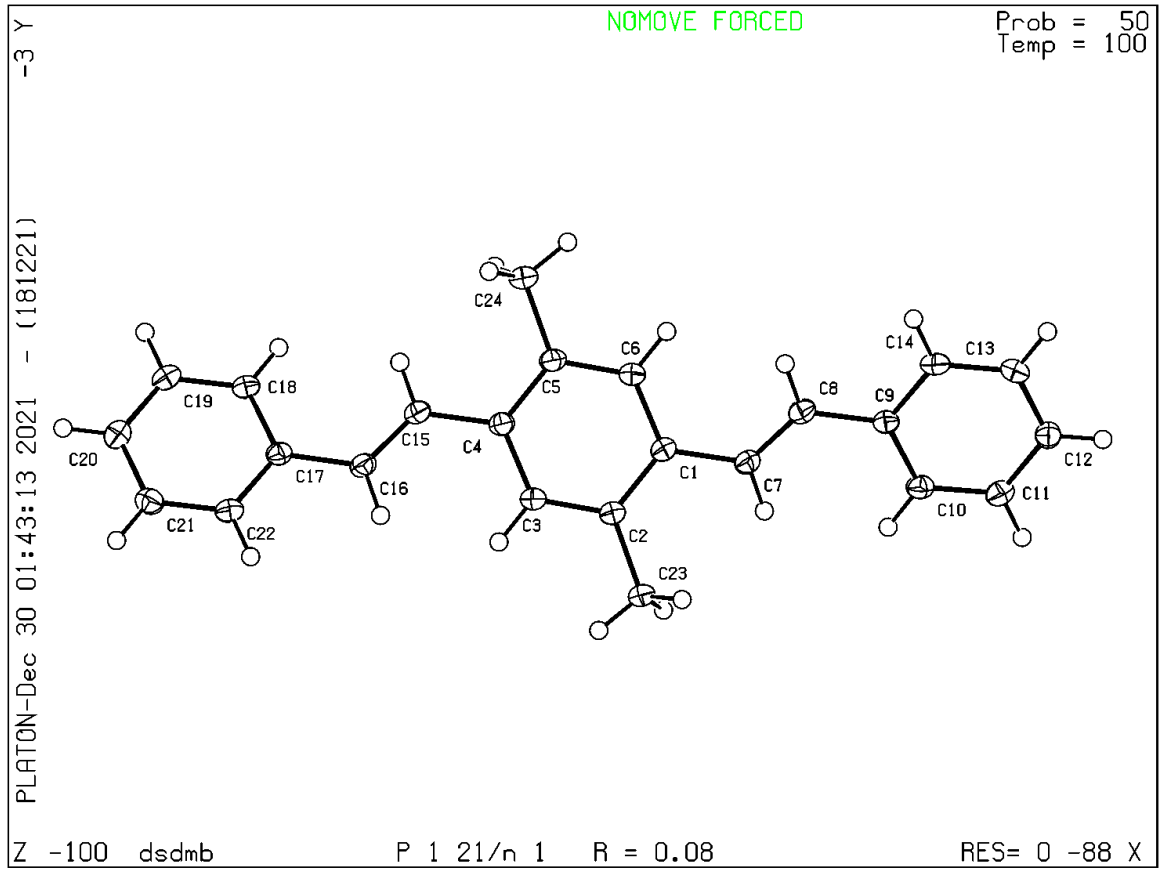
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PLATON version of 18/12/2021; check.def file version of 18/12/2021



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test-name_ALERT_alert-type_alert-level.

Click on the hyperlinks for more details of the test.

 **Alert level A**

PLAT211_ALERT_2_A ADP of Atom C17 is N.P.D. or (nearly) 2D . Please Check

 **Alert level B**

PLAT097_ALERT_2_B Large Reported Max. (Positive) Residual Density 0.63 eA-3

 **Alert level C**

DIFMN02_ALERT_2_C The minimum difference density is < -0.1*ZMAX*0.75

_refine_diff_density_min given = -0.494

Test value = -0.450

DIFMN03_ALERT_1_C The minimum difference density is < -0.1*ZMAX*0.75

The relevant atom site should be identified.

DIFMX02_ALERT_1_C The maximum difference density is > 0.1*ZMAX*0.75

The relevant atom site should be identified.

PLAT082_ALERT_2_C High R1 Value 0.13 Report

PLAT084_ALERT_3_C High wR2 Value (i.e. > 0.25) 0.34 Report

PLAT098_ALERT_2_C Large Reported Min. (Negative) Residual Density -0.49 eA-3

PLAT213_ALERT_2_C Atom C6 has ADP max/min Ratio 3.5 oblate

PLAT340_ALERT_3_C Low Bond Precision on C-C Bonds 0.0061 Ang.

 **Alert level G**

PLAT012_ALERT_1_G No _shelx_res_checksum Found in CIF Please Check

PLAT083_ALERT_2_G SHELXL Second Parameter in WGHT Unusually Large 13.22 Why ?

PLAT767_ALERT_4_G INS Embedded LIST 6 Instruction Should be LIST 4 Please Check

PLAT941_ALERT_3_G Average HKL Measurement Multiplicity 2.7 Low

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