

Supplementary Information

Recyclable Optical Bioplastics Platform for Solid State Red Light Harvesting via Triplet-triplet Annihilation Photon Upconversion

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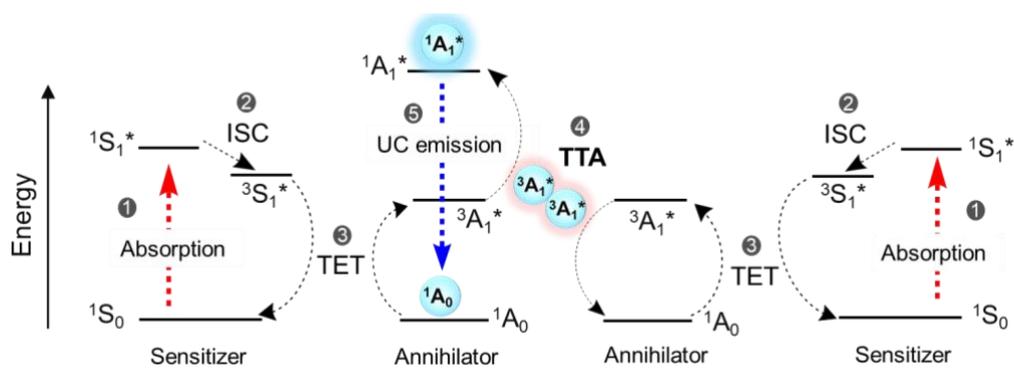


Fig. S1. Energy diagram of triple-triplet annihilation photon upconversion (TTA-UC).

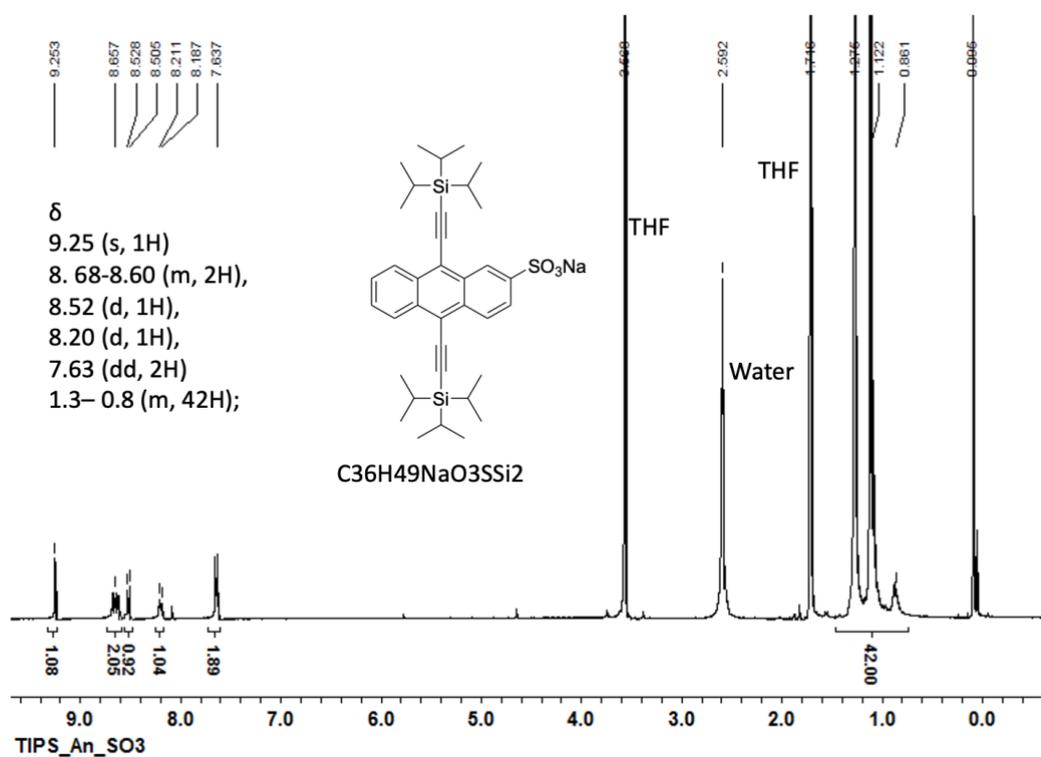
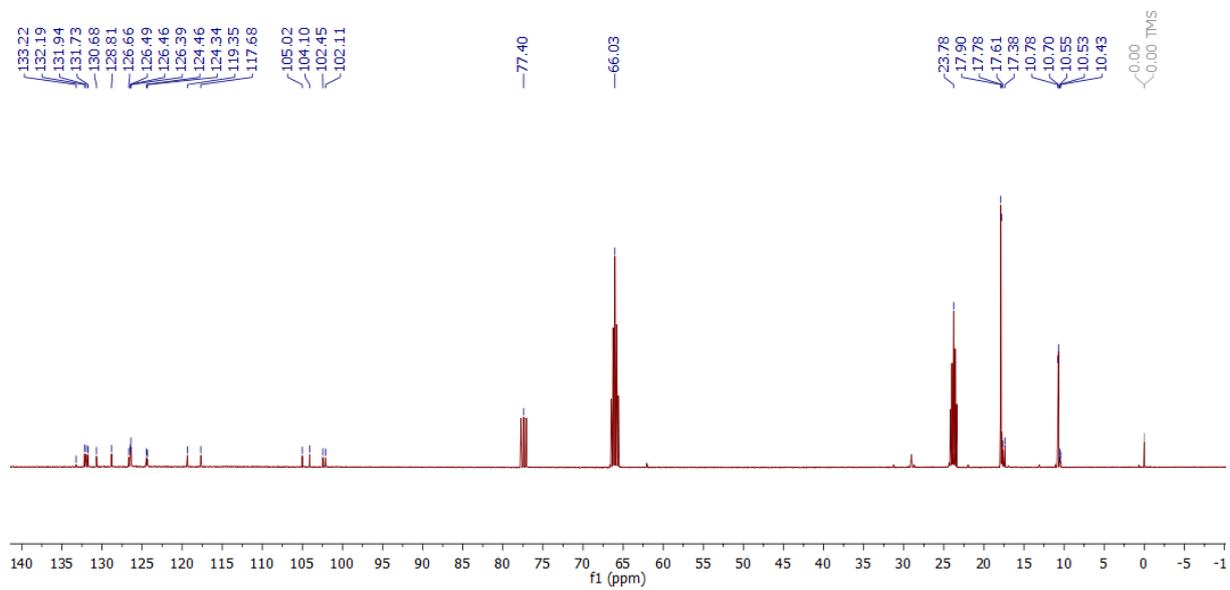
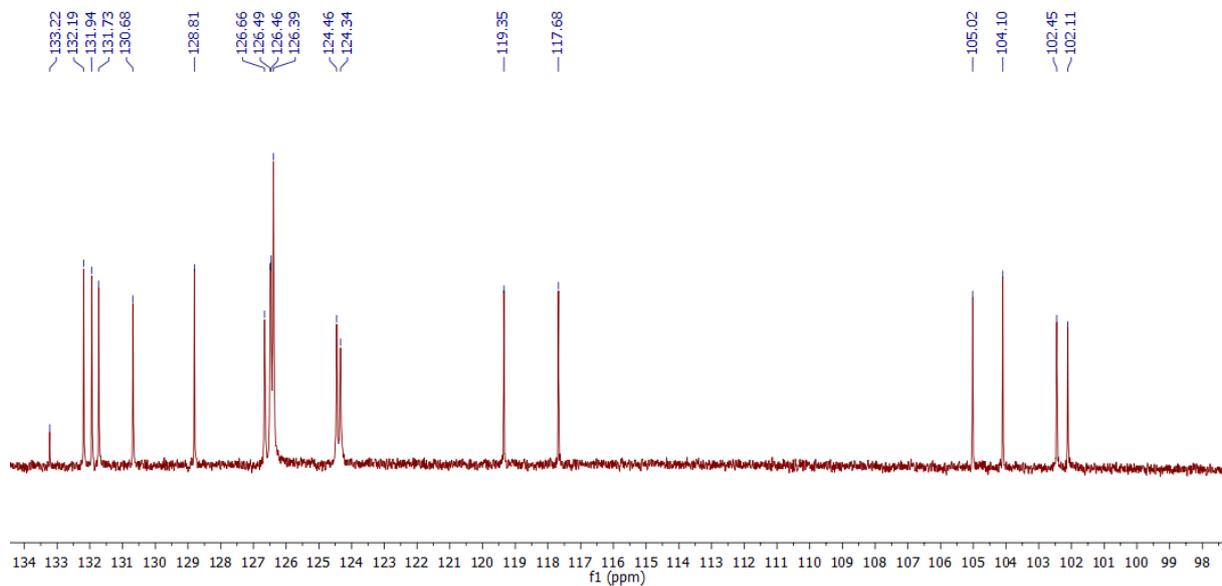


Fig. S2. 1H -NMR Spectra of Sodium TIPS-Anthracene Sulfonate in deuterated THF. 1H NMR (400 MHz, d^8 -THF)

a



b



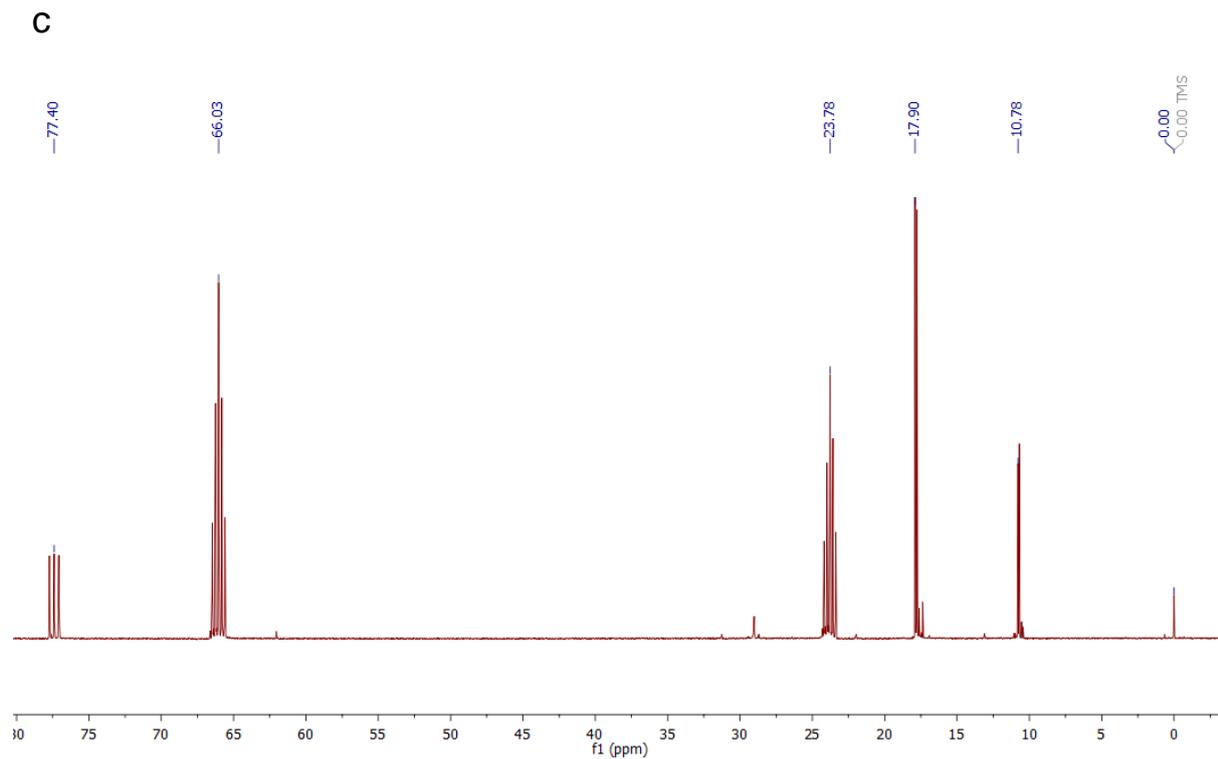


Fig. S3. ^{13}C carbon NMR of TIPS-AnS in CDCl_3 and d-THF mixture. **a)** Full spectrum **b)** sp 2 and sp carbons, **c)** aliphatic region (77.40 comes from chloroform and 66.03 & 23.78 are from THF).

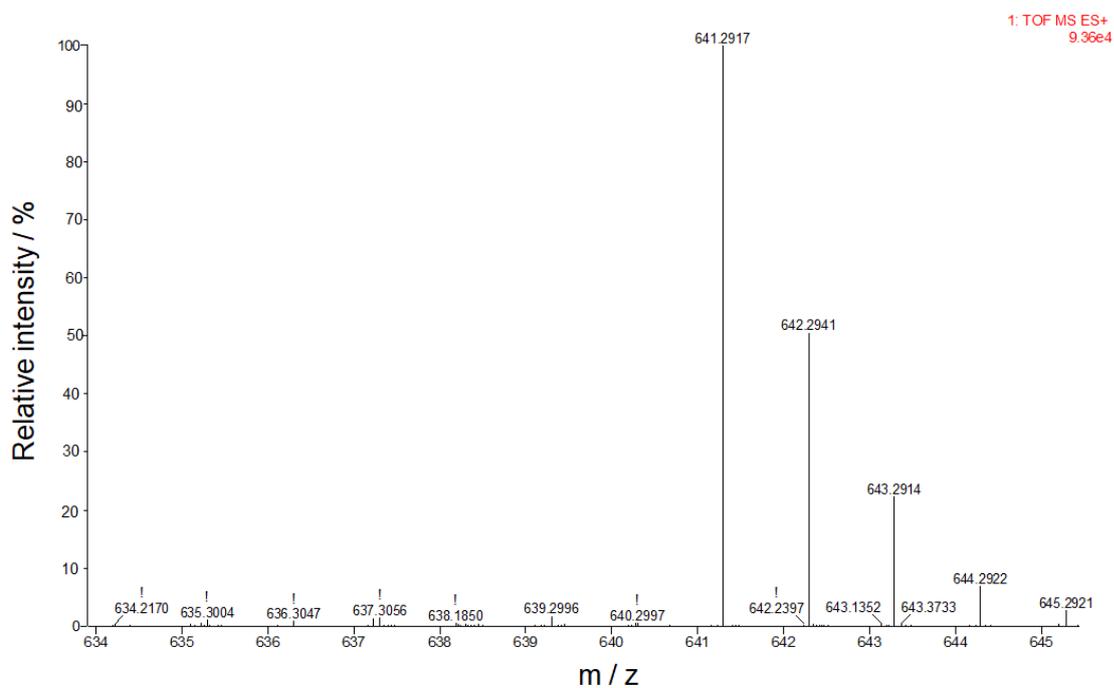


Fig. S4. HR-MS spectra of TIPS-Anthracene sodium-2-sulfonate

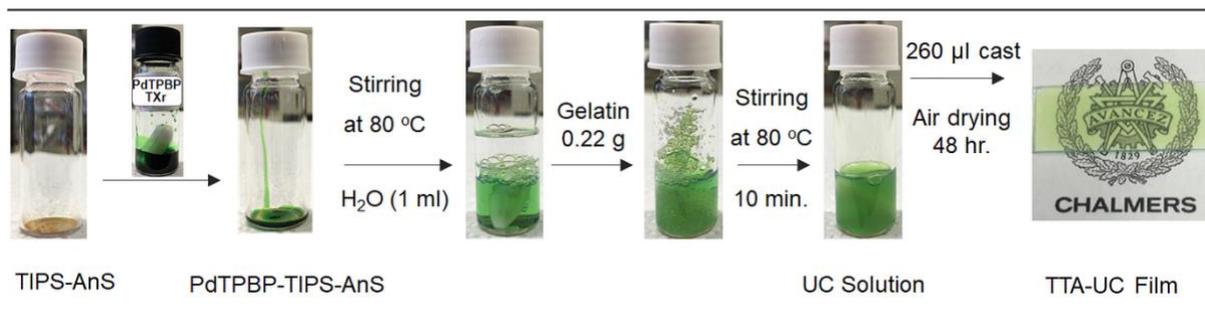


Fig. S5. Schematic of the preparation of G-TXr-TIPS-AnS-PdTPBP film (Sample-I).

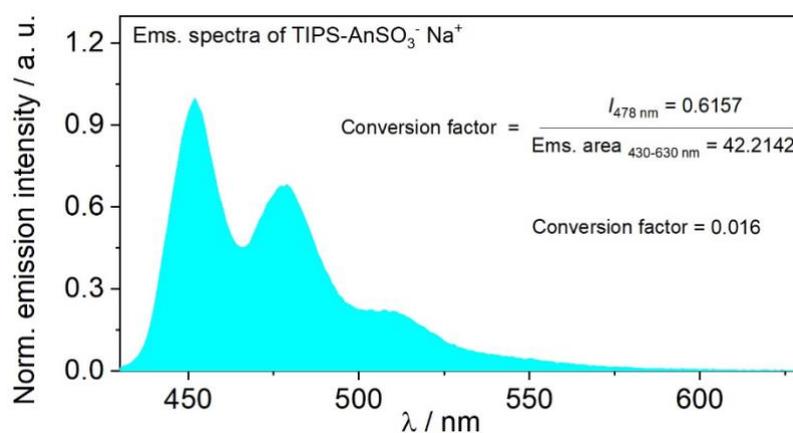


Fig. S6. Fluorescence emission spectra of TIPS-AnS (40 µM) in TXr liquid in air (Inset shows correction factor used for calculation of internal UC quantum yield of G-TXr-TIPS-AnS-PdTPBP films using UC emission intensities at 478 nm).

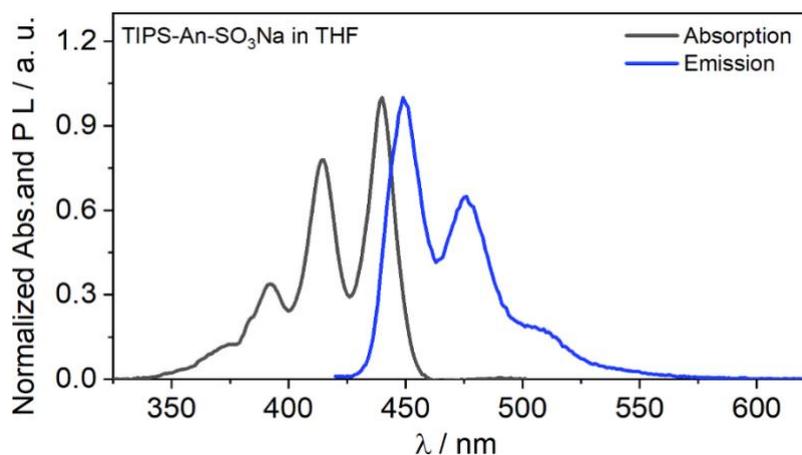


Fig. S7. Normalized absorption (black line) and photoluminescence (PL, blue lines) spectra of the TIPS-AnS (blue line, $\lambda_{\text{ex}} = 415 \text{ nm}$) in THF. Chromophore concentrations (TIPS-AnS = 40 µM).

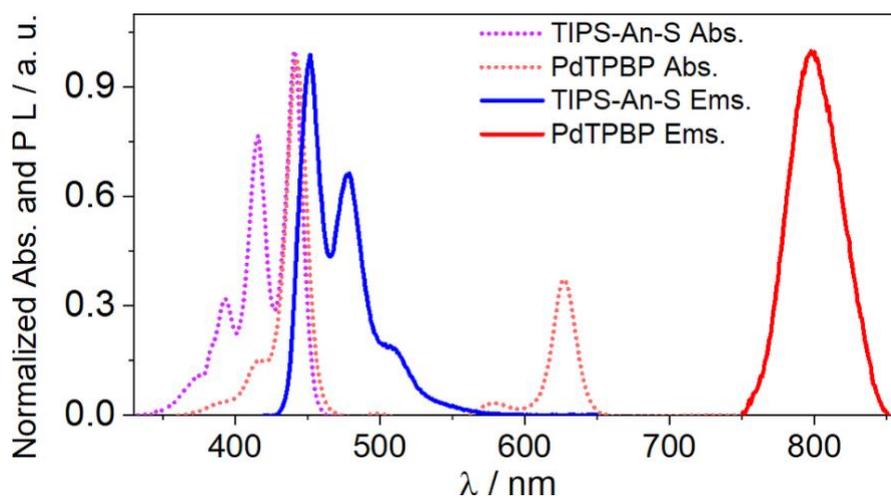


Fig. S8. Normalized absorption (dotted lines) and photoluminescence (PL, solid lines) spectra of the TXr-TIPS-AnS (blue line, $\lambda_e = 415$ nm) and TXr-PdTPBP (red line, $\lambda_{ex} = 633$ nm) in air. Chromophore concentrations TIPS-AnS = 40 μ M and PdTPBP = 5 μ M.

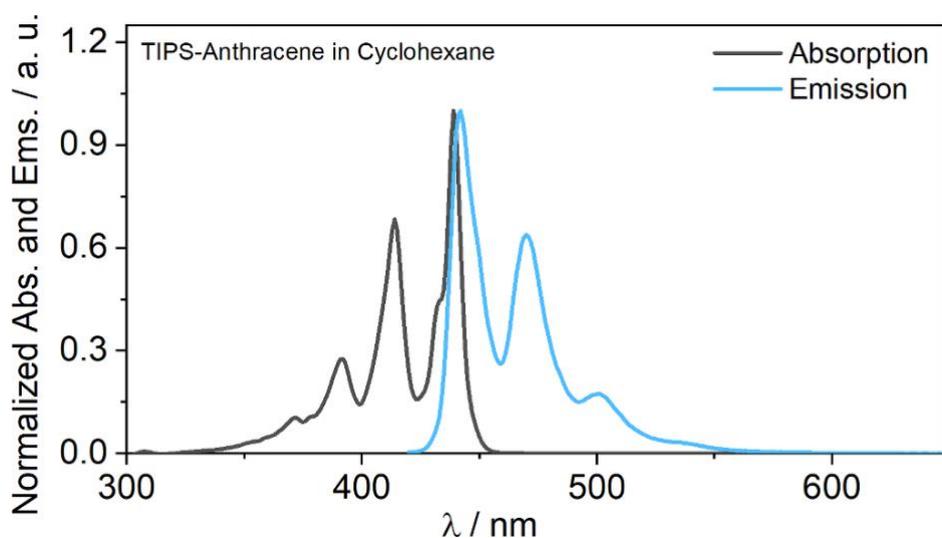


Fig. S9. Normalized absorption (black line) and photoluminescence (PL, blue lines) spectra of the TIPS-Anthracene (blue line, $\lambda_{ex} = 414$ nm) in cyclohexane. Chromophore concentrations (TIPS-AnS = 40 μ M). The peaks were observed at $\lambda_{abs} = 392$ nm, 414nm and 439 nm. $\lambda_{ems} = 442$ nm, 476nm and 501 nm.

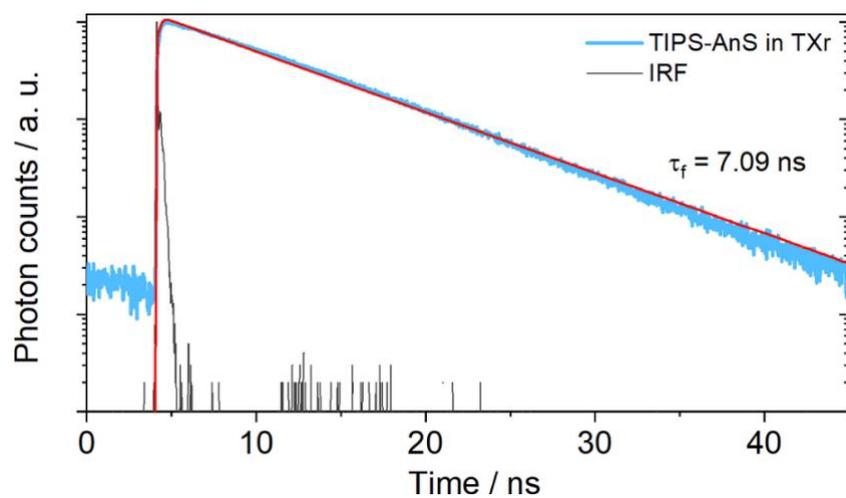


Fig. S10. Fluorescence decay profile of TIPS-AnS in TXr. Chromophore concentration TIPS-AnS=40 μM .

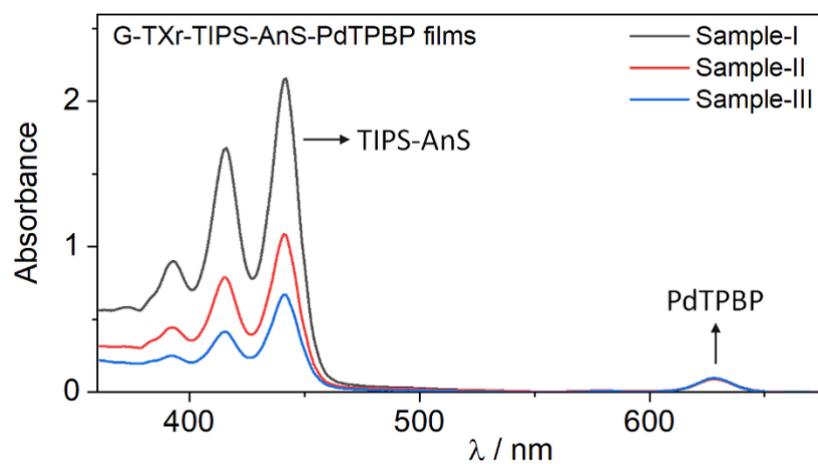


Fig. S11. Full absorption spectra of G-TXr-TIPS-AnS-PdTPBP films. TIPS-AnS concentrations (Sample-I = 5 $\text{mmol} \cdot \text{kg}^{-1}$, Sample-II = 2.5 $\text{mmol} \cdot \text{kg}^{-1}$ and Sample-III = 1 $\text{mmol} \cdot \text{kg}^{-1}$).

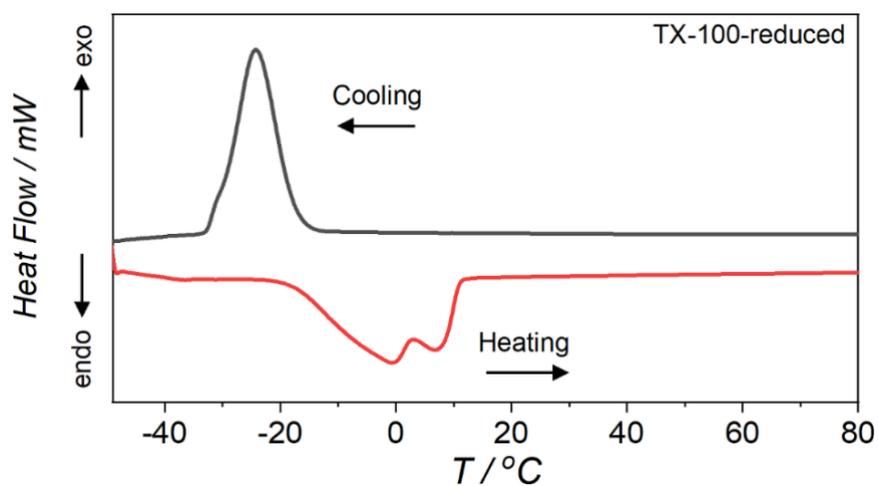


Fig. S12. DSC thermogram of vacuum dried TX-100-reduced. Scanning rate 10 °C min⁻¹.

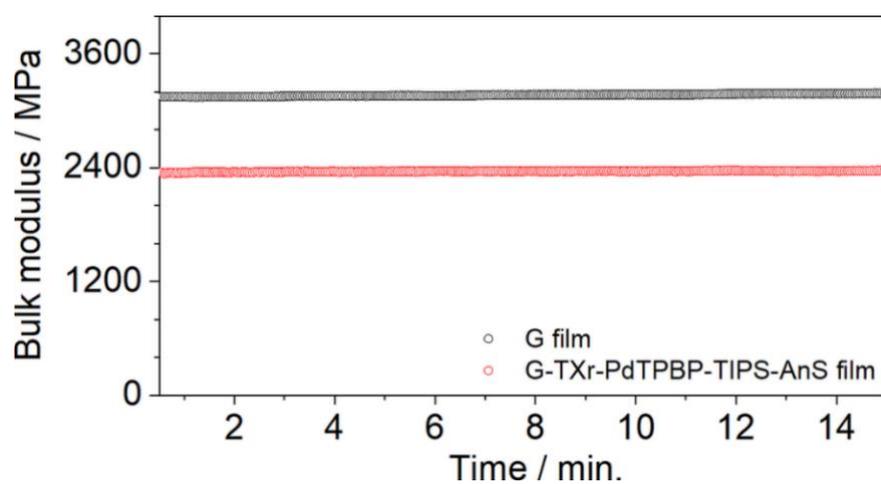


Fig. S13. Time sweeps dynamic mechanical analysis plots of the G film and the G-TXr-TIPS-AnS-PdTPBP film at 30 °C, a strain of 0.05 % and frequency of 1 Hz.

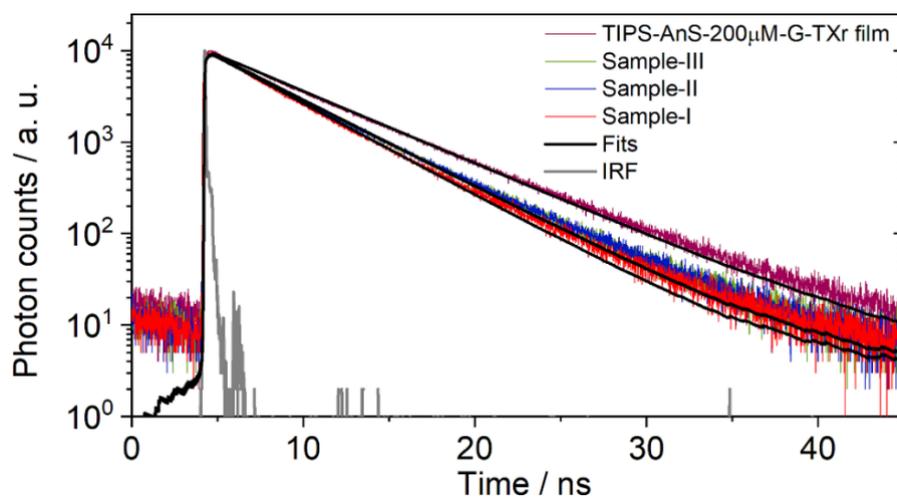


Fig. S14. Fluorescence decay profile of TIPS-AnS in the G-TXr-TIPS-AnS film or G-TXr-TXr-TIPS-AnS-PdTPBP films. TIPS-AnS concentrations (Sample-I = 5 mmol kg^{-1} , Sample-II = 2.5 mmol kg^{-1} and Sample-III = 1 mmol kg^{-1}) and G-TXr-TIPS-AnS film (TIPS-AnS = $200\text{ }\mu\text{mol kg}^{-1}$).

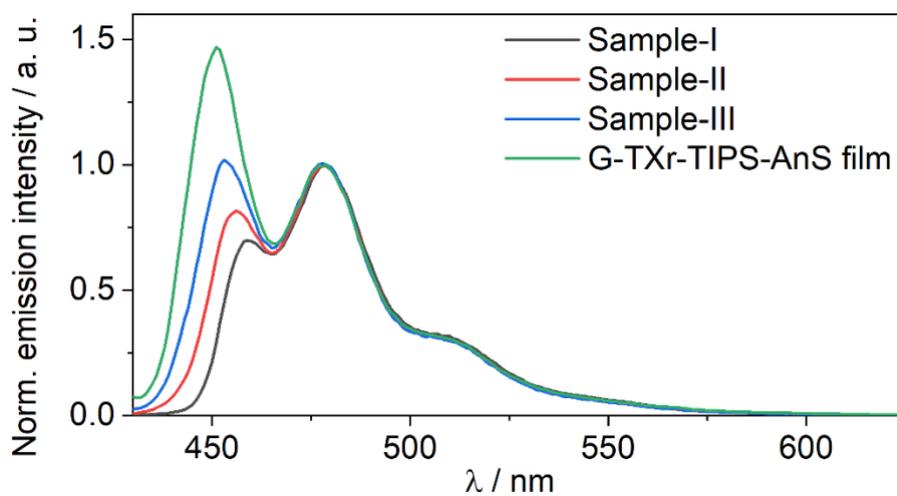


Fig. S15. Normalized emission spectra of TIPS-AnS in G-TXr-TIPS-AnS-PdTPBP films. TIPS-AnS concentrations (Sample-I = 5 mmol kg^{-1} , Sample-II = 2.5 mmol kg^{-1} and Sample-III = 1 mmol kg^{-1}) and G-TXr-TIPS-AnS film (TIPS-AnS = $200\text{ }\mu\text{mol kg}^{-1}$). The spectra are normalized corresponding to the wavelength at 478 nm .

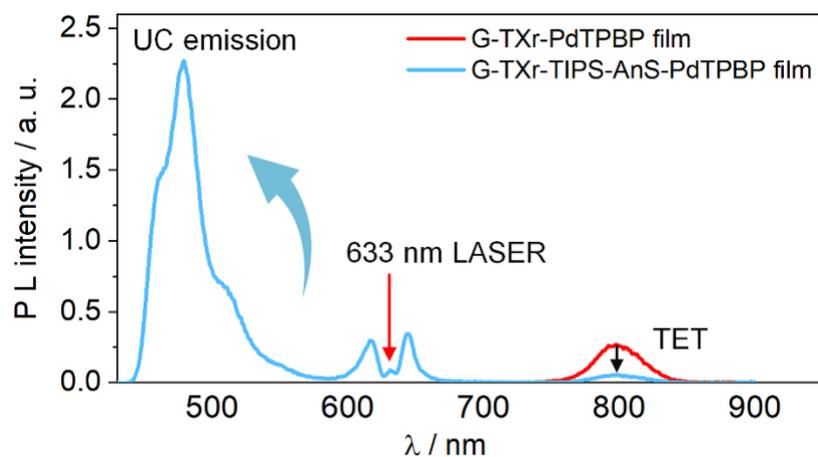


Fig. S16. Photoluminescence spectra of PdTPBP in the absence and presence of TIPS-AnS in sample-III. $\lambda_{\text{ex}} = 633$ nm, excitation intensity = 33.8 mW cm^{-2} .

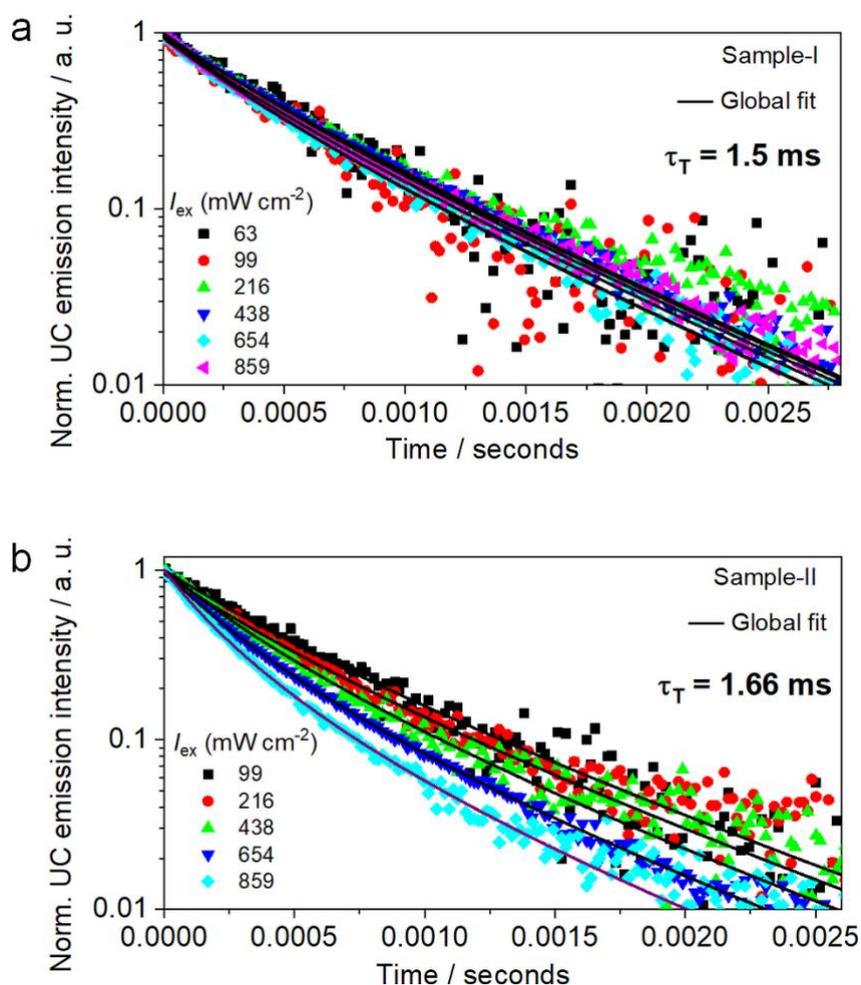


Fig. S17. Time resolved upconversion emission vs excitation intensity plot of G-TXr-TIPS-AnS-PdTPBP films in air. **a)** sample-I and **b)** sample-II.

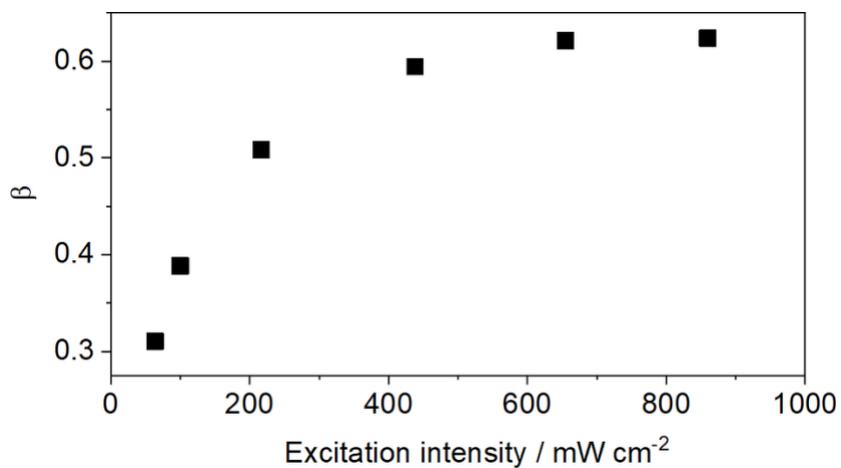


Fig. S18. Fitted value of β for the G-TXr-TIPS-AnS-PdTPBP Sample-III at various excitation intensities.

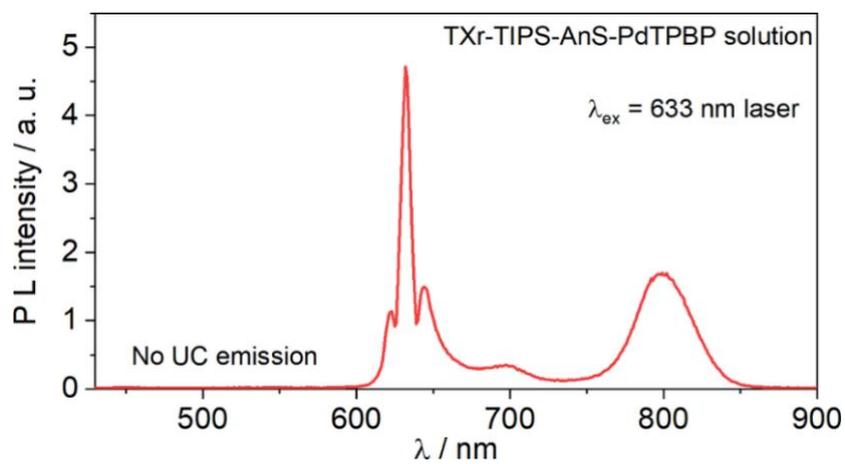


Fig. S19. Photoluminescence spectrum of the TXr-TIPS-AnS-PdTPBP solution at 633 nm laser (excitation intensity of 15 W cm⁻²) in air.

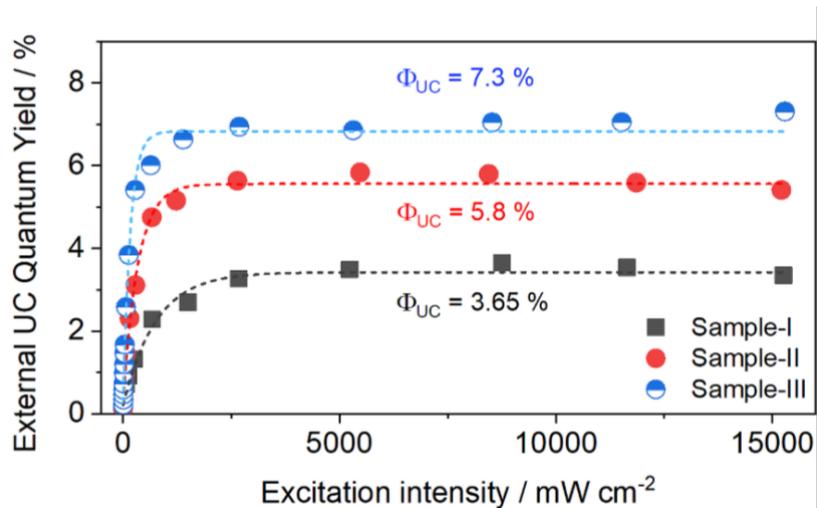


Fig. S20. Plot showing external UC quantum yields of G-TXr-TIPS-AnS-PdTPBP films calculated without conversion factor.

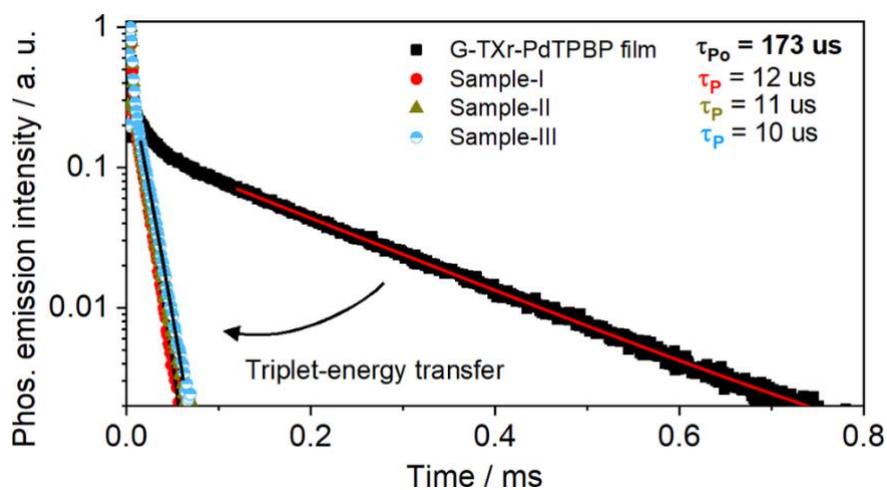


Fig. S21. Comparative phosphorescence decay profiles of G-TX-PdTPBP and TTA-UC film (Sample-I, Sample-II, and Sample-III).

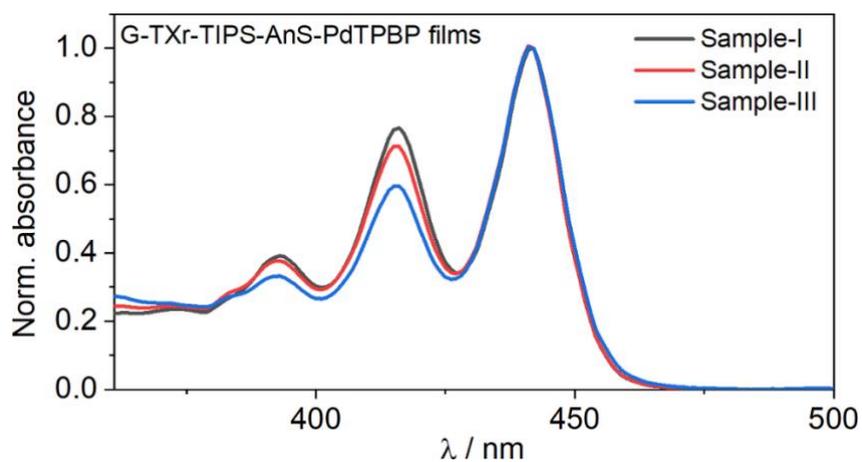


Fig. S22. Normalized absorption spectra of G-TXr-TIPS-AnS-PdTPBP films. TIPS-AnS concentrations (Sample-I = 5 mmol kg⁻¹, Sample-II = 2.5 mmol.kg⁻¹ and Sample-III = 1 mmol kg⁻¹).

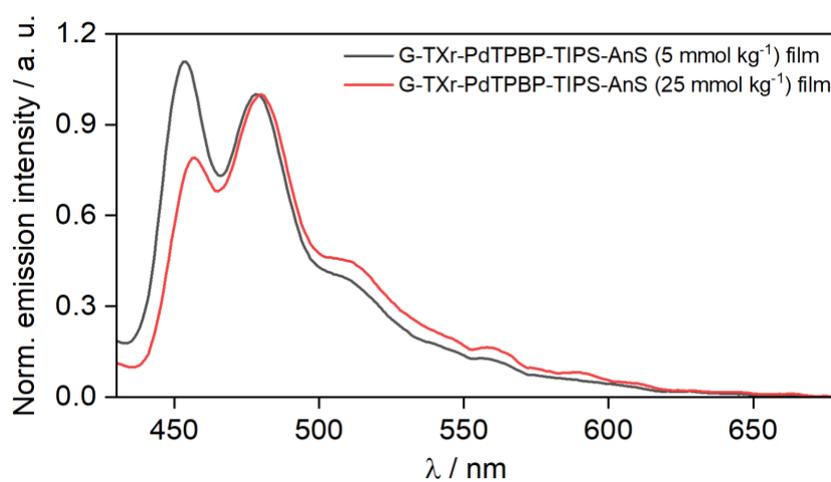


Fig. S23. Normalized emission spectra of G-TXr-TIPS-AnS-PdTPBP films. TIPS-AnS concentrations (TIPS-AnS = 5 mmol kg⁻¹ and TIPS-AnS = 25 mmol.kg⁻¹).

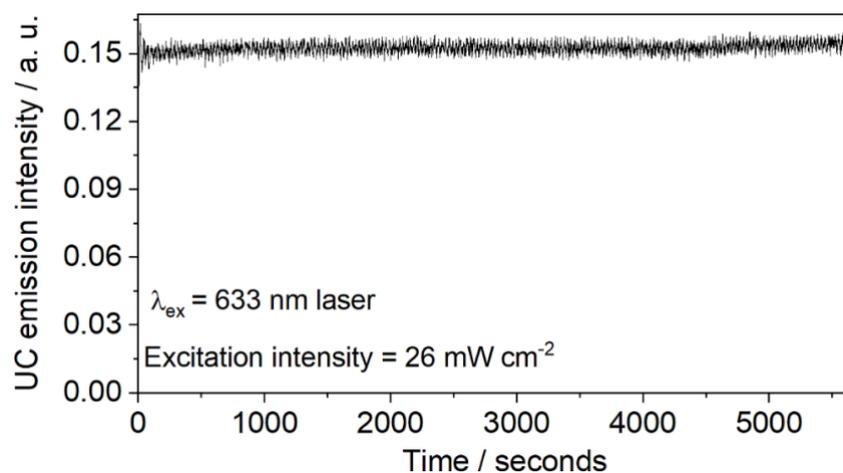


Fig. S24. UC emission intensity changes of the G-TXr-TIPS-AnS-PdTPBP film-I at $\lambda_{em} = 478$ nm upon continuous excitation with 633 nm laser for 5600 seconds (excitation intensity of 26 mW cm^{-2}) in air.

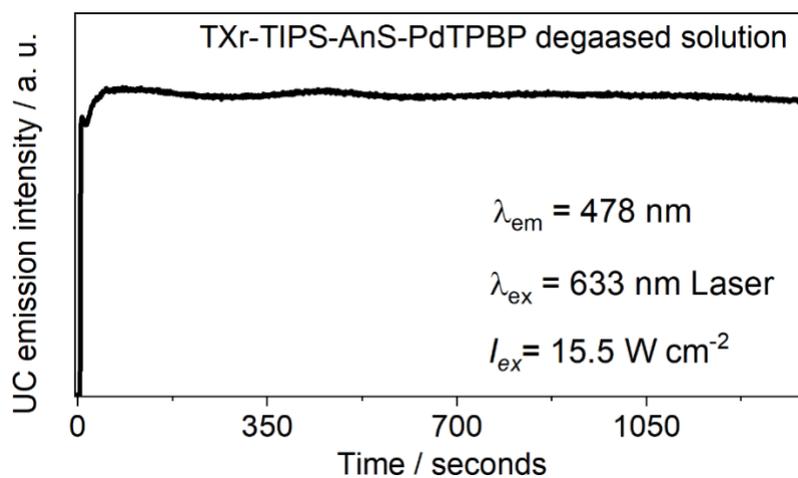


Fig. S25. Changes in upconversion emission intensity ($\lambda_{em} = 478$) of degassed TXr-TIPS-AnS-PdTPBP solution upon 633 nm laser excitation (excitation intensity of 15 W cm^{-2}).

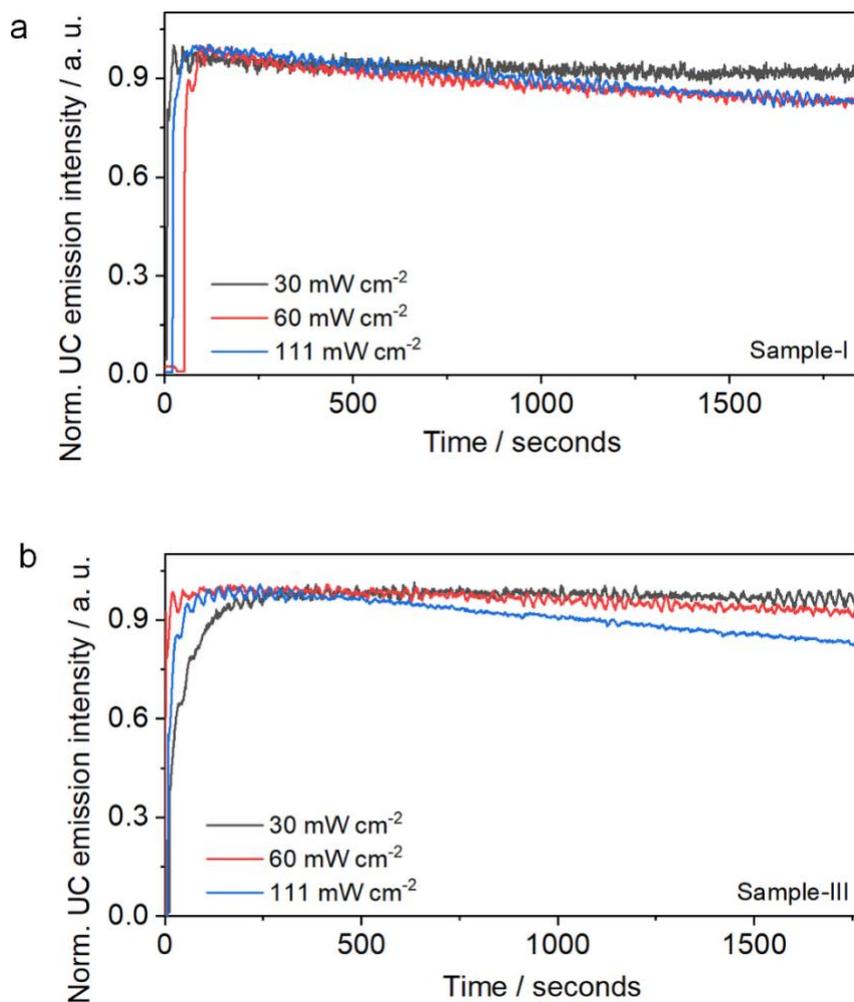


Fig. S26. UC emission intensity changes of the G-TXr-TIPS-AnS-PdTPBP film-I and film-III at $\lambda_{em} = 478$ nm upon continuous excitation with 633 nm laser for 1800 seconds at varying excitation intensities in air.

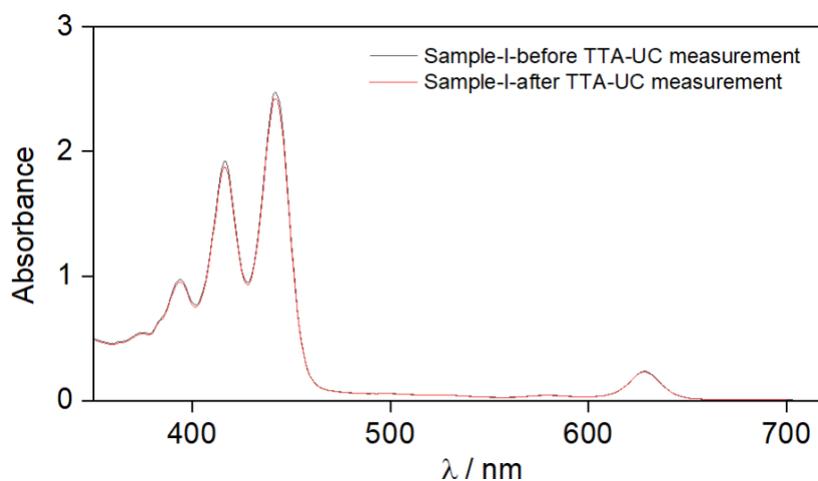


Figure S27. Absorption spectra of G-TXr-TIPS-AnS-PdTPBP film (Sample-I) before and after TTA-UC measurement.

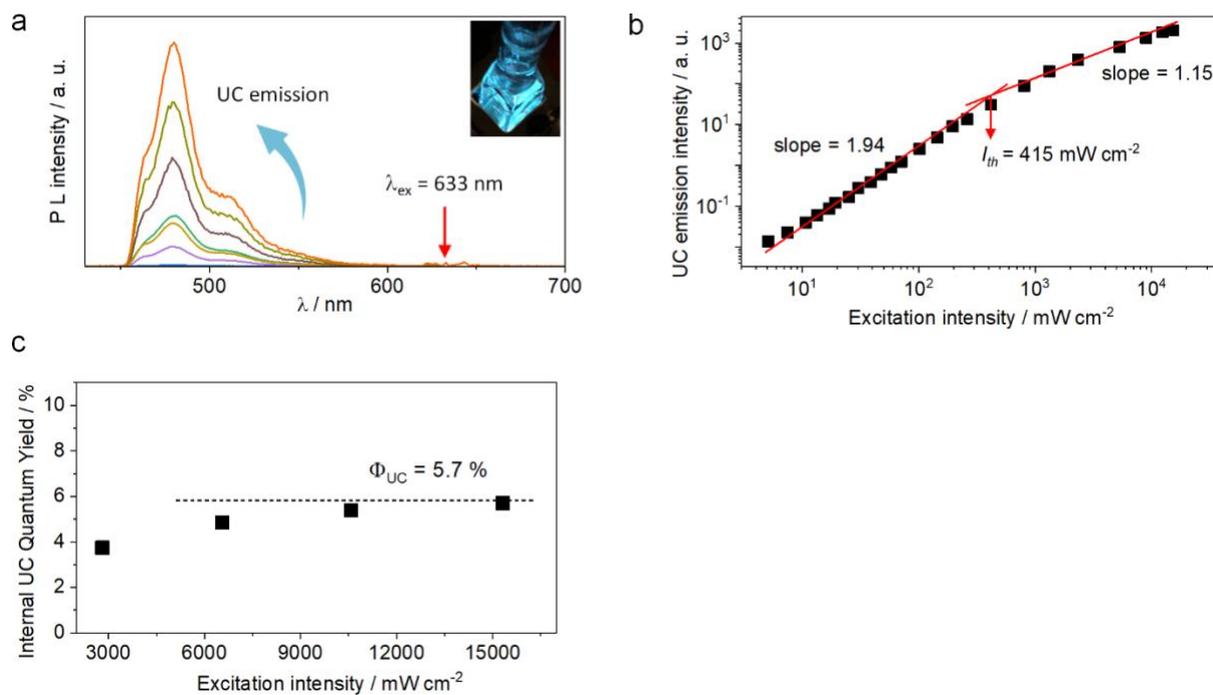


Fig. S28. Upconversion properties of deaerated TXr-TIPS-AnS-PdTPBP solution. a) PL spectra upon 633 nm laser excitation, inset show photograph of UC emission in the cuvette (no short pass filter used). **b)** Log–log plot of excitation intensity dependence of UC emission. **c)** Internal UC quantum yield vs. excitation intensity plot.

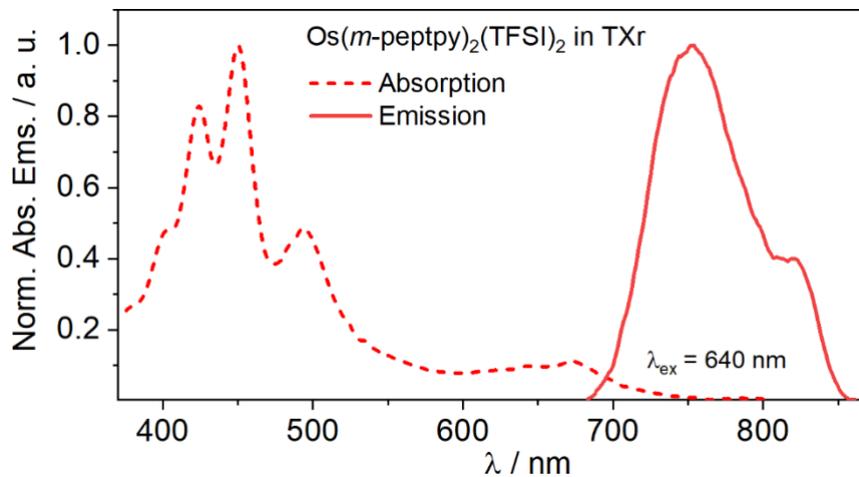


Fig. S29. Normalized absorption and emission spectra of $\text{Os}(m\text{-peptpy})_2(\text{TFSI})_2$ in TXr in air ($\lambda_{\text{ex}} = 640 \text{ nm}$).

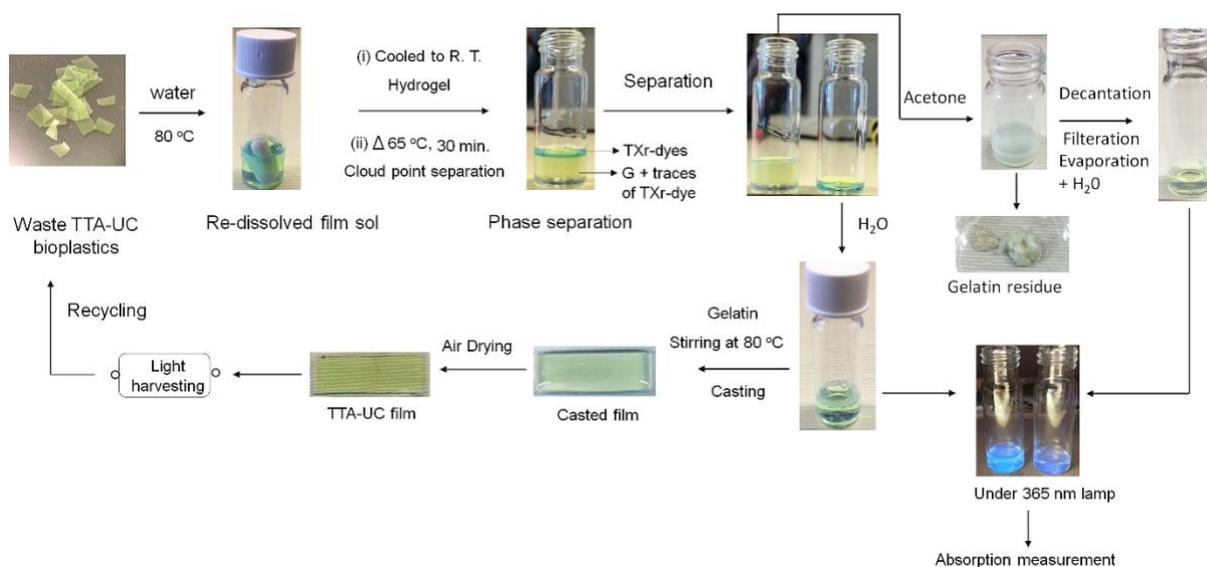


Fig. S30. Schematic of the recycling of G-TXr-TIPS-AnS-PdTPBP film.

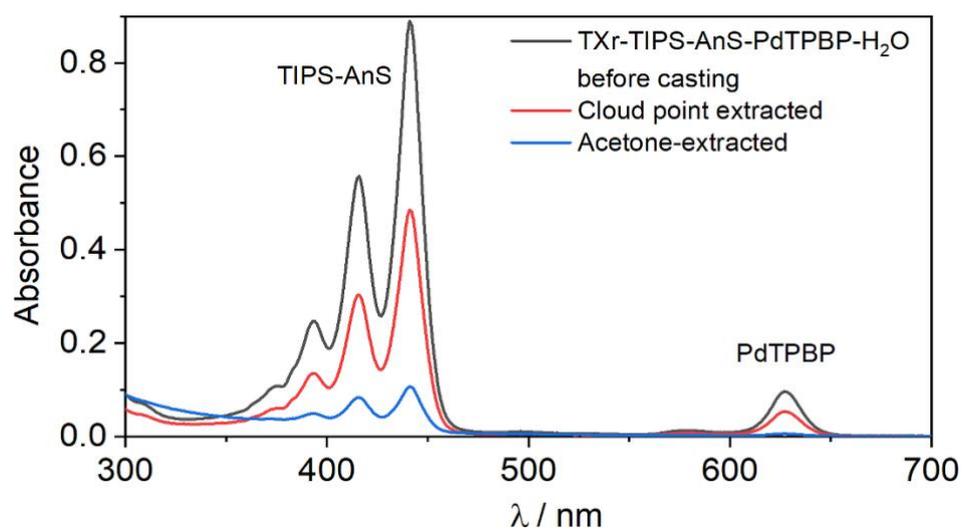


Fig. S31. Absorption spectra of TXr-TIPS-AnS-PdTPBP aqueous solution before casting into film and after cloud point and acetone extraction. The absorbance measurements carried out using 1 mm pathlength quartz cuvette.