

Synthesis, characterization, antitumor potential, and investigation of mechanism of
action of copper(II) complexes with acylpyruvates as ligands: Interactions with
biomolecules and kinetic study

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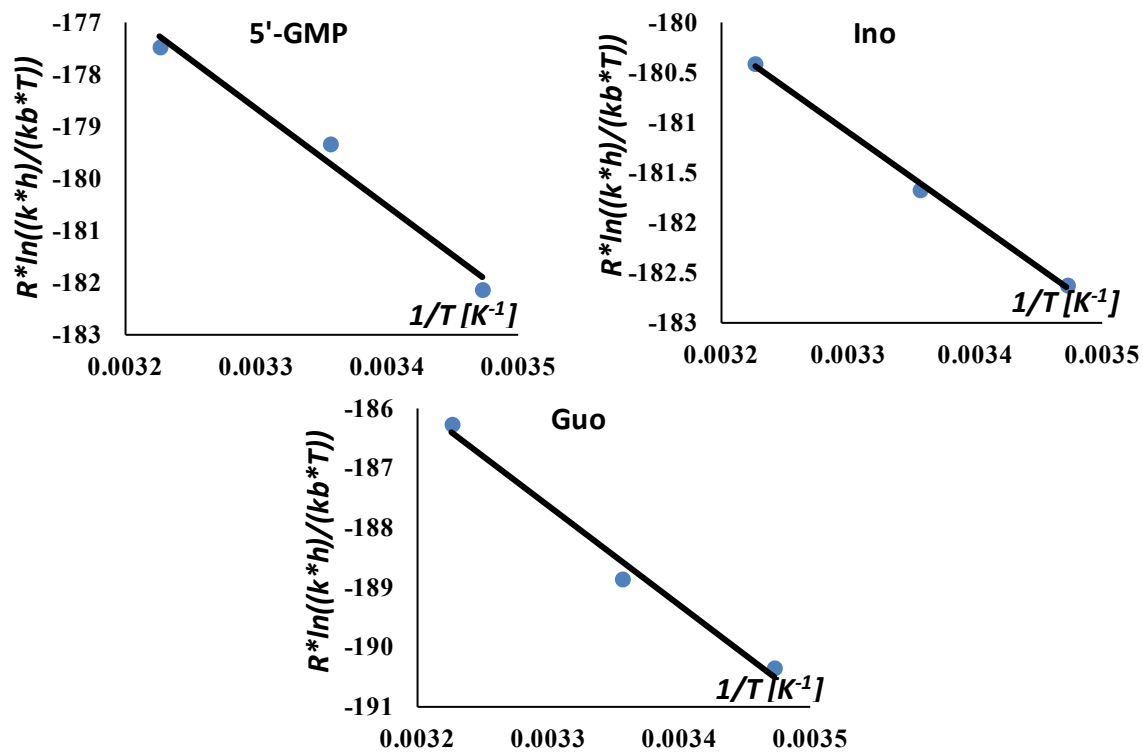


Fig. S1. Eyring plots for the reactions of copper(II) complex **F** with different nucleophiles in Hepes buffer (pH=7.2, 25mM Hepes).

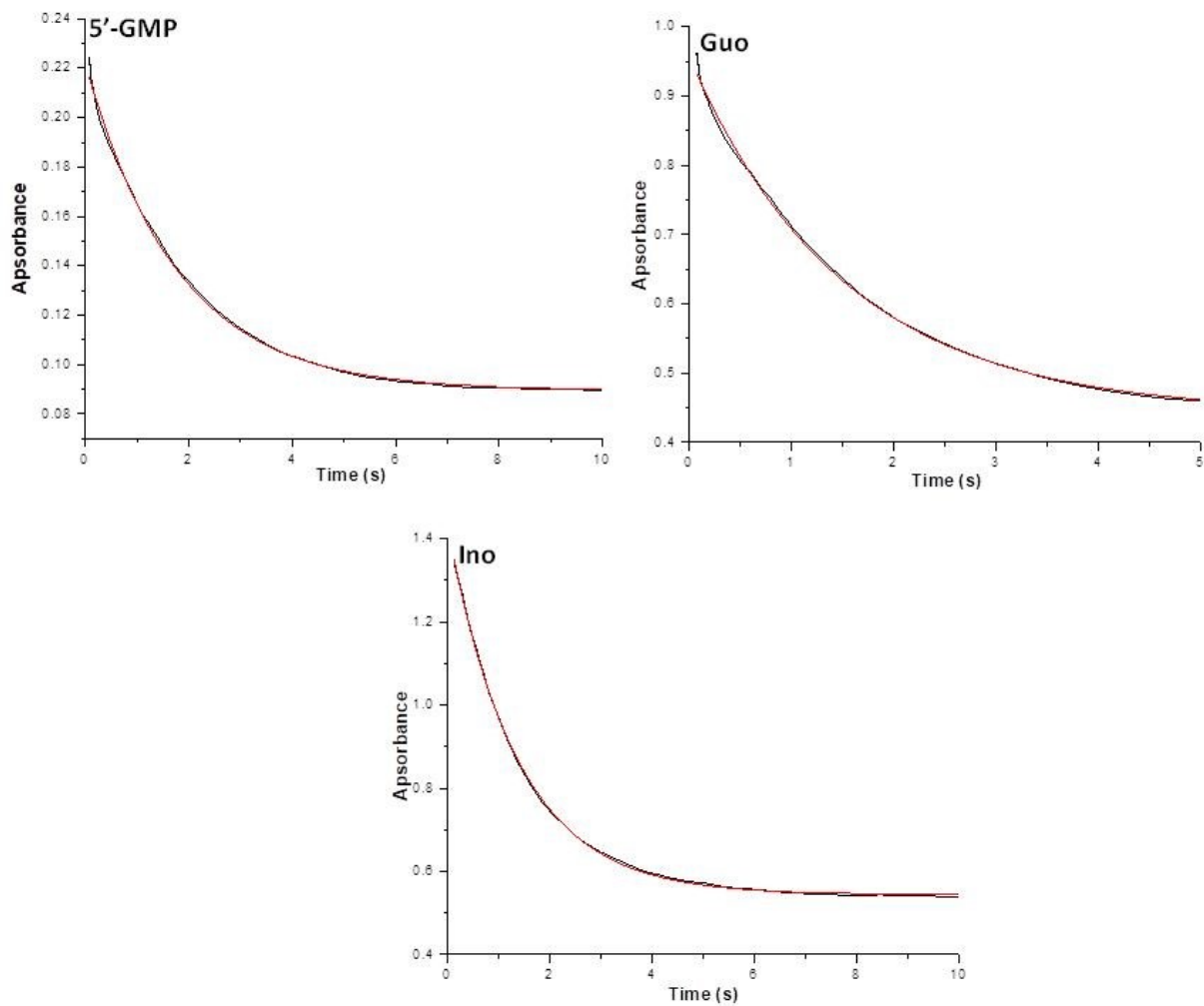


Fig. S2. Kinetic traces for the substitution reactions of copper(II) complexes and the nucleophiles.

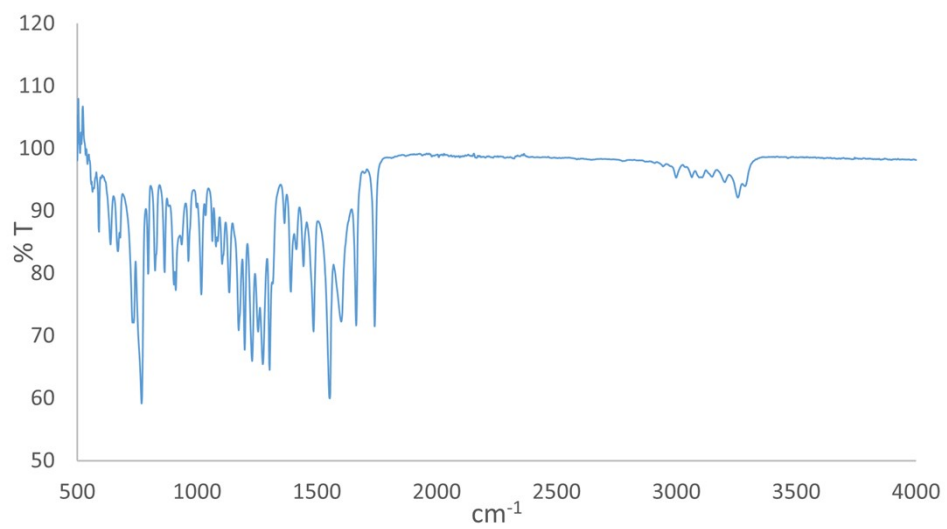


Fig S3. IR spectra of compound **A**

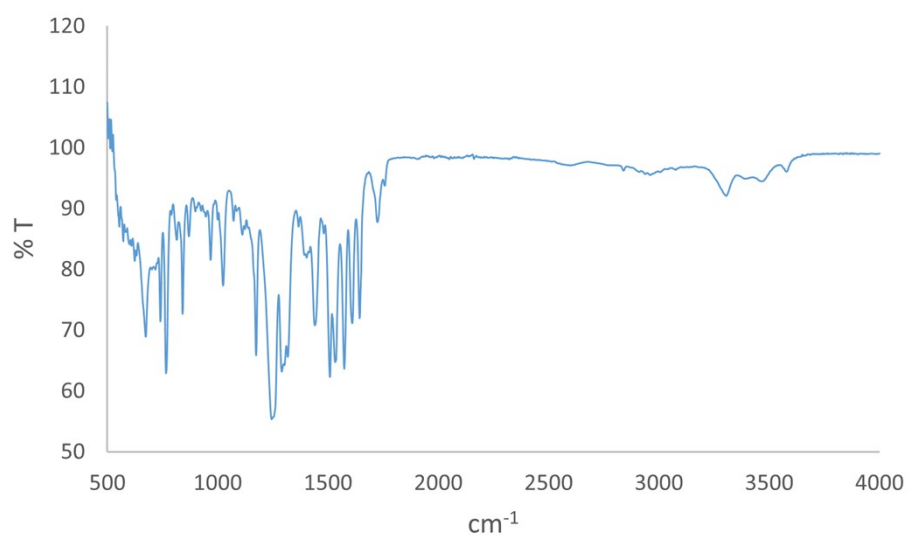


Fig S4. IR spectra of compound **B**

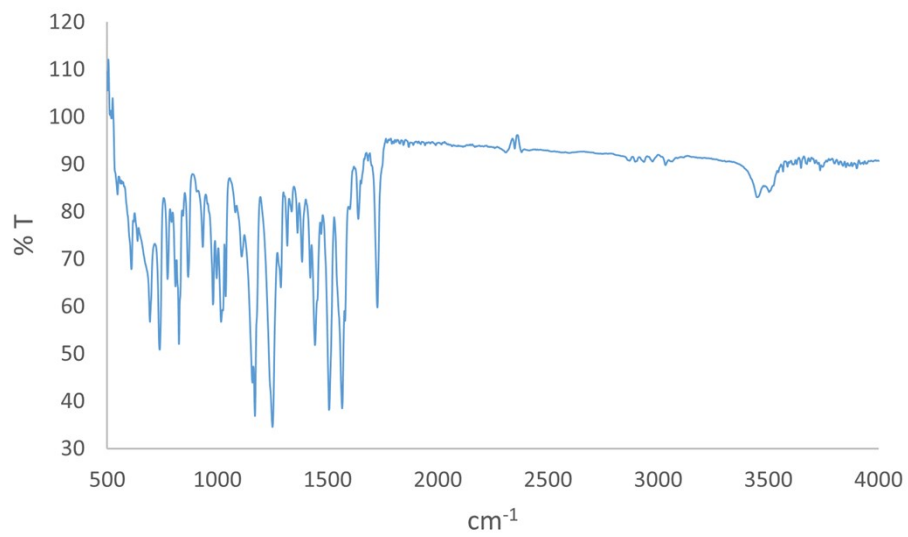


Fig S5. IR spectra of compound C

Table S1. Observed *pseudo*-first order rate constant as a function of the nucleophile concentration and temperature for the substitution reactions in 25mM Hepes buffer (pH = 7,2).

T/K	10 ⁻⁵ C _L /M	k _{obs} /s ⁻¹		
		5'-GMP	Guo	Ino
288	5	0.680(7)	0.512(6)	0.553(7)
	7.5	1.080(6)	0.633(7)	0.810(7)
	10	1.300(6)	0.840(6)	1.093(6)
	12.5	1.698(7)	1.071(7)	1.398(6)
	17.5	1.945(6)	1.261(7)	1.618(6)
	20	2.168(6)		
298	5	1.040(7)	0.681(6)	0.600(6)
	7.5	1.546(7)	0.793(6)	0.984(6)
	10	1.963(6)	1.050(7)	1.322(6)
	12.5	2.486(7)	1.285(7)	1.564(7)
	17.5	2.860(6)	1.630(6)	1.896(7)
	20	3.080(6)		
308	5	1.480(6)	0.944(7)	0.984(6)
	7.5	2.060(6)	1.171(7)	1.368(7)
	10	2.598(6)	1.490(6)	1.794(6)
	12.5	3.197(6)	1.873(7)	2.136(6)
	17.5	3.689(7)	2.285(6)	2.496(6)
	20	4.176(7)		