Supporting Information

Constructing Molecular Bridge for High-Efficiency and Stable Perovskite Solar Cells based on P3HT

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Supplementary Methods



Supplementary Fig. 1. Synthetic routes of MDN, RDN, PDN, and TDN.

10-(2-(2-ethoxyethoxy)ethyl)-10H-phenoxazine (2):

The KOH (1.36 g, 24 mmol) was added slowly to a solution of 10Hphenoxazine (2.2 g, 12 mmol) and 1-bromo-2-(2-ethoxyethoxy)ethane (2.4 g, 12 mmol) in 30 mL tetrahydrofuran at room temperature, then the mixture was stirred for 16 hours at room temperature. The mixture was diluted with dichloromethane and washed with water. The organic layer was dried over Na₂SO₄ and evaporated under a vacuum. The crude product was purified by column chromatography (SiO₂, petroleum ether 40–60 °C / ethyl acetate = 20/1 v/v) and to obtain compound **2** (3.1 g, 86.4 % yield) as a red oil.

¹H NMR (600 MHz, DMSO-*d*6): 6.83–6.80 (m, 2H), 6.75–6.74 (m, 2H), 6.67–6.63 (m, 4H), 3.76 (t, *J* = 6.0 Hz, 2H), 3.65 (t, *J* = 6.0 Hz, 2H), 3.57–3.55 (m, 2H), 3.47–3.46 (m, 2H), 3.42–3.39 (m, 2H), 1.07 (t, *J* = 7.2 Hz, 2H); ¹³C NMR (150 MHz, DMSO-*d*6):144.54, 133.58, 124.37, 121.43, 115.41, 113.00, 70.66, 69.73, 67.13, 66.10, 44.46, 15.57;

FTMS (APCI) calcd for C₁₈H₂₂NO₃ (M+H⁺): 300.1600; found: 300.1591.

10-(2-(2-ethoxyethoxy)ethyl)-10H-phenoxazine-3-carbaldehyde (3):

The POCI₃ (3.07 g, 20.1 mmol) was added slowly to a solution of compound **2** (2 g, 6.7 mmol) and anhydrous N,N-dimethylformamide (4 mL) in 20 mL anhydrous chloroform at 0~5 °C under argon atmosphere, the mixture was stirred for 1 hour at 0~5 °C, and then the mixture was stirred for 16 hours at 65 °C. After the reaction completed, the mixture was quenched with sodium hydroxide solution. Then the mixture was extracted with dichloromethane for three times, and then the combined organic phase was dried over Na₂SO₄ and evaporated under vacuum. The crude product was purified by column chromatography (SiO₂, petroleum ether 40–60 °C / ethyl acetate = 5/1 v/v) and to obtain compound **3** (2 g, 91.3 % yield) as a yellow oil.

¹H NMR (600 MHz, DMSO-*d*6): 9.65 (s, 1H), 7.39 (dd, *J*₁ = 8.4 Hz, *J*₂ = 1.8 Hz, 1H), 7.00 (d, *J* = 1.8 Hz, 1H), 6.92 (d, *J* = 8.4 Hz, 2H), 6.86–6.85 (m, 2H), 6.77–6.74 (m, 1H), 6.70–6.68 (m, 1H), 3.86 (t, *J* = 6.0 Hz, 2H), 3.68 (t, *J* = 6.0 Hz, 2H), 3.57–3.55 (m, 2H), 3.46–3.44 (m, 2H), 3.41–3.37 (m, 2H), 1.05 (t, *J* = 7.2 Hz, 2H);

¹³C NMR (150 MHz, DMSO-*d*6): 190.44, 144.61, 144.21, 139.55, 131.77, 130.07, 129.02, 124.61, 123.01, 115.69, 113.97, 113.71, 112.78, 70.74, 69.71, 67.20, 66.11, 44.53, 15.56;

FTMS (APCI) calcd for C19H22NO4 (M+H⁺): 328.1549; found: 328.1528.

7-bromo-10-(2-(2-ethoxyethoxy)ethyl)-10H-phenoxazine-3-carbaldehyde (4):

The N-Bromosuccinimide (547 mg, 3.22 mmol) in tetrahydrofuran (5 mL) was added slowly to a solution of compound **3** (960 mg, 2.94 mmol) in tetrahydrofuran (5 mL) at $0\sim5$ °C, the mixture was stirred for 1 hours at $0\sim5$ °C,

and then the mixture was stirred for 16 hours at room temperature. After the reaction completed, the mixture was quenched with saturated sodium bicarbonate solution. Then the mixture was extracted with dichloromethane for three times, and then the combined organic phase was dried over Na₂SO₄ and evaporated under vacuum. The crude product was used in next step without further purification.

¹H NMR (600 MHz, DMSO-*d*6): 9.65 (s, 1H), 7.40 (dd, $J_1 = 8.4$ Hz, $J_2 = 1.8$ Hz, 1H), 7.00–6.98 (m, 2H), 6.92 (d, J = 8.4 Hz, 1H), 6.86 (d, J = 2.4 Hz, 1H), 6.79 (d, J = 9.0 Hz, 1H), 3.83 (t, J = 6.0 Hz, 2H), 3.66 (t, J = 6.0 Hz, 2H), 3.55–3.54 (m, 2H), 3.45–3.43 (m, 2H), 3.39–3.36 (m, 2H), 1.04 (t, J = 7.2 Hz, 2H);

¹³C NMR (150 MHz, DMSO-*d*6): 190.48, 145.07, 144.21, 138.92, 131.54, 130.26, 129.13, 127.01, 118.27, 115.60, 113.96, 113.44, 113.08, 70.76, 69.71, 67.18, 66.11, 44.69, 15.54;

FTMS (APCI) calcd for C₂₆H₃₁BNO₂S₂ (M+H⁺): 464.1889; found: 464.1890.

7-(4-(bis(4-methoxyphenyl)amino)phenyl)-10-(2-(2-ethoxyethoxy)ethyl)-10H-phenoxazine-3-carbaldehyde (5):

A mixture of compound **4** (406 mg, 1.0 mmol), compound **9** (431 mg, 1.0 mmol), $Pd(PPh_3)_4$ (11 mg, 0.01 mmol), potassium carbonate (280 mg, 2.0 mmol) in 1,4dioxane (5 mL) and water (1 mL) was stirred at 100 °C under argon atmosphere for 16 hours. After the mixture was cooled to room temperature, 50 mL of water were added and water phase was extracted with dichloromethane for three times, and then the combined organic phase was dried over Na₂SO₄. After the solvent was evaporated under vacuum, the crude product was purified by column chromatography (SiO₂, petroleum ether 40–60 °C / ethyl acetate = 5/1 v/v) to give compound **5** as a yellow solid (530 mg, 84.1 % yield).

¹H NMR (600 MHz, DMSO-*d*6): 9.65 (s, 1H), 7.46 (d, J = 9.0 Hz, 2H), 7.39 (dd, $J_1 = 8.4$ Hz, $J_2 = 1.8$ Hz, 1H), 7.09 (dd, $J_1 = 8.4$ Hz, $J_2 = 1.8$ Hz, 1H), 7.05–7.02 (m, 5H), 6.94–6.91 (m, 6H), 6.88 (d, J = 8.4 Hz, 1H), 6.78 (d, J = 9.0

Hz, 1H), 3.90–3.87 (m, 2H), 3.75 (s, 6H),3.70 (t, *J* = 6.0 Hz, 2H), 3.57–3.56 (m, 2H), 3.46–3.45 (m, 2H), 3.41–3.37 (m, 2H), 1.05 (t, *J* = 7.2 Hz, 2H);

¹³C NMR (150 MHz, DMSO-*d*6): 190.43, 156.27, 148.15, 144.54, 144.45, 140.47, 139.29, 134.74, 130.54, 130.24, 130.04, 129.08, 127.21, 126.90, 121.56, 119.92, 115.45, 114.33, 113.73, 112.82, 112.78, 70.77, 69.71, 67.26, 66.12, 55.71, 44.53, 40.53, 15.56;

FTMS (APCI) calcd for C₁₉H₂₁BrNO₄ (M+H⁺): 406.0654; found: 406.0645.

4-methoxy-N-(4-methoxyphenyl)-N-phenylaniline (7):

A mixture of bis(4-methoxyphenyl)amine (2.29 g, 10 mmol), bromobenzene (1.73 g, 11 mmol), Pd₂dba₃ (91 mg, 0.1 mmol), t-Bu₃PHBF₄ (116 mg, 0.4 mmol), sodium tert-butoxide (1.44 g, 15 mmol) in anhydrous toluene (20 mL) was stirred at 110 °C under argon atmosphere for 16 hours. After evaporating the solvent under reduced pressure, the remaining crude product was purified by column chromatography (SiO₂, petroleum ether 40–60 °C /ethyl acetate = 25/1 v/v) to give compound **7** as a yellow oil (2.73 g, 89.5% yield).

¹H NMR (600 MHz, CDCl₃): 7.19 (t, *J* = 7.2 Hz, 2H), 7.08–7.05 (m, 4H), 6.96 (d, *J* = 7.8 Hz, 2H) , 6.89 (t, *J* = 7.2 Hz, 1H), 6.86–6.84 (m, 4H), 3.82 (s, 6H).

4-bromo-N,N-bis(4-methoxyphenyl)aniline (8):

The N-Bromosuccinimide (1.59 g, 8.95 mmol) in tetrahydrofuran (15 mL) was added slowly to a solution of compound **7** (2.73 mg, 8.95 mmol) in tetrahydrofuran (15 mL) at 0~5 °C, the mixture was stirred for 1 hours at 0~5 °C, and then the mixture was stirred for 16 hours at room temperature. After the reaction completed, the mixture was quenched with saturated sodium bicarbonate solution. Then the mixture was extracted with dichloromethane for three times, and then the combined organic phase was dried over Na₂SO₄ and evaporated under vacuum. The crude product was used in next step without further purification.

4-methoxy-N-(4-methoxyphenyl)-N-(4-(4,4,5,5-tetramethyl-1,3,2dioxaborolan-2-yl)phenyl)aniline (9):

A mixture of compound **8** (3.43 g, 8.95 mmol), bis(pinacolato)diboron (2.72 g, 10.74 mmol), Pd(dppf)Cl₂ (66 mg, 0.09 mmol), Potassium Acetate (1.8 g, 17.9 mmol) in anhydrous 1,4-dioxane (30 mL) was stirred at 100 °C under argon atmosphere for 72 hours. After evaporating the solvent under reduced pressure, the remaining crude product was purified by column chromatography (SiO₂, petroleum ether 40–60 °C /ethyl acetate = 25/1 v/v) to give compound **9** as a yellow solid (3.63 g, 94 % yield).

¹H NMR (600 MHz, CDCl₃): 7.62 (t, *J* = 9.0 Hz, 2H), 7.10–7.07 (m, 4H), 6.89 (d, *J* = 9.0 Hz, 2H) , 6.87–6.83 (m, 4H), 3.82 (s, 6H), 1.34 (s, 12H).

2-((7-(4-(bis(4-methoxyphenyl)amino)phenyl)-10-(2-(2-ethoxyethoxy)ethyl) -10H-phenoxazin-3-yl)methylene)malononitrile (MDN):

A mixture of compound **5** (520 mg, 0.83 mmol), malononitrile (110 mg, 1.66 mmol), pyridine (200 g, 2.46 mmol) in acetonitrile (10 mL) was stirred at 90 °C under argon atmosphere for 72 hours. After evaporating the solvent under reduced pressure, the remaining crude product was purified by column chromatography (SiO₂, petroleum ether 40–60 °C /ethyl acetate = 2/1 v/v) to give **MDN** as a red solid (510 mg, 90.6 % yield) (Tg = 64 °C, Td = 392 °C).

¹H NMR (600 MHz, DMSO-*d*6): 9.65 (s, 1H), 7.46 (d, J = 9.0 Hz, 2H), 7.39 (dd, $J_1 = 8.4$ Hz, $J_2 = 1.8$ Hz, 1H), 7.09 (dd, $J_1 = 8.4$ Hz, $J_2 = 1.8$ Hz, 1H), 7.05–7.02 (m, 5H), 6.94–6.91 (m, 6H), 6.88 (d, J = 8.4 Hz, 1H), 6.78 (d, J = 9.0 Hz, 1H), 3.90–3.87 (m, 2H), 3.75 (s, 6H),3.70 (t, J = 6.0 Hz, 2H), 3.57–3.56 (m, 2H), 3.46–3.45 (m, 2H), 3.41–3.37 (m, 2H), 1.05 (t, J = 7.2 Hz, 3H);

¹³C NMR (150 MHz, DMSO-*d*6): 190.43, 156.27, 148.15, 144.54, 144.45, 140.47, 139.29, 134.74, 130.54, 130.24, 130.04, 129.08, 127.21, 126.90, 121.56, 119.92, 115.45, 114.33, 113.73, 112.82, 112.78, 70.77, 69.71, 67.26, 66.12, 55.71, 44.53, 40.53, 15.56;

(Z)-5-((7-(4-(bis(4-methoxyphenyl)amino)phenyl)-10-(2-(2-ethoxyethox y)ethyl)-10H-phenoxazin-3-yl)methylene)-3-ethyl-2-thioxothiazolidin-4one (RDN):

A mixture of compound **5** (518 mg, 0.82 mmol), 3-ethylrhodanine (264 mg, 1.64 mmol), pyridine (200 g, 2.46 mmol) in acetonitrile (10 mL) was stirred at 90 °C under argon atmosphere for 48 hours. After evaporating the solvent under reduced pressure, the remaining crude product was purified by column chromatography (SiO₂, petroleum ether 40–60 °C /ethyl acetate = 2/1 v/v) to give **RDN** as a red solid (400 mg, 63.1% yield) (*T*g = 68 °C, *T*d = 382 °C).

¹H NMR (600 MHz, DMSO-*d*6): 7.56 (s, 1H), 7.41 (d, J = 8.4 Hz, 2H), 7.11–7.02 (m, 7H), 6.93–6.88 (m, 7H), 6.76–6.83 (m, 4H), 4.04–4.01 (m, 2H), 3.89–3.84 (m, 2H), 3.75 (s, 6H), 3.70–3.67 (m, 2H), 3.57–3.56 (m, 2H), 3.47–3.45 (m, 2H), 3.41–3.37 (m, 2H), 1.17 (t, J = 7.2 Hz, 3H), 1.04 (t, J = 7.2Hz, 3H);

¹³C NMR (150 MHz, DMSO-*d*6): 171.01, 156.29, 147.67, 144.87, 144.42, 140.46, 135.45, 134.89, 132.82, 131.27, 127.24, 126.86, 121.56, 119.87, 115.46, 112.81, 70.77, 69.72, 67.32, 67.25, 66.13, 55.72, 40.52, 31.43, 22.54, 15.59, 14.44, 12.40;

FTMS (APCI) calcd for C₄₄H₄₄N₃O₆S₂ (M+H⁺): 774.2672; found: 774.2676.

2-((10-(2-(2-ethoxyethoxy)ethyl)-10H-phenoxazin-3-yl)methylene)malononitrile (PDN):

A mixture of compound **3** (560 mg, 1.71 mmol), malononitrile (226 mg, 3.42 mmol), pyridine (406 g, 5.14 mmol) in acetonitrile (10 mL) was stirred at 90 °C under argon atmosphere for 48 hours. After evaporating the solvent under reduced pressure, the remaining crude product was purified by column chromatography (SiO₂, petroleum ether 40–60 °C /ethyl acetate = 3/1 v/v) to give **PDN** as a blood red solid (600 mg, 93.4 % yield).

¹H NMR (600 MHz, CDCl₃-*d*): 7.36 (s, 1H), 7.27–7.26 (d, J = 6.0 Hz, 1H), 6.85–6.82 (t, J = 9.0 Hz, 1H), 6.80–6.77 (t, J = 9.0 Hz, 1H), 6.69–6.66 (m, 3H), 3.84–3.82 (t, J = 6.0 Hz, 2H), 3.77–3.75 (t, J = 6.0 Hz, 2H), 3.66–3.65 (m, 2H), 3.58–3.56 (m, 2H), 3.53–3.50 (m, 2H), 1.22–1.19 (t, J = 8.6 Hz, 3H);

¹³C NMR (150 MHz, CDCl₃-*d*): 157.04, 144.76, 144.48, 140.06, 130.85, 130.64, 124.09, 139.65, 115.96, 115.07, 113.86, 112.97, 111.75, 71.20, 69.89, 67.16, 66.81, 44.85, 29.71, 15.17;

FTMS (APCI) calcd for C₂₂H₂₁N₃O₃ (M+H⁺): 375.1583; found: 375.1687.

2-([2,2':5',2"-terthiophen]-5-ylmethylene)malononitrile (TDN):

A mixture of [2,2':5',2"-terthiophene]-5-carbaldehyde (276 mg, 1 mmol), malononitrile (132 mg, 2 mmol), pyridine (237 g, 3 mmol) in acetonitrile (15 mL) was stirred at 90 °C under argon atmosphere for 48 hours. After evaporating the solvent under reduced pressure, the remaining crude product was purified by column chromatography (SiO₂, petroleum ether 40–60 °C /dichloromethane = 1/1 v/v) to give **TDN** as a dark red solid (280 mg, 86.4 % yield).

¹H NMR (600 MHz, CDCl₃-*d*): 7.76 (s, 1H), 7.64–7.63 (d, J = 6.8 Hz, 1H), 7.36–7.35 (d, J = 6.0 Hz, 1H), 7.631–7.30 (d, J = 6.0 Hz, 1H), 7.26–7.25 (d, J = 6.2 Hz, 2H), 7.17–7.16 (d, J = 5.8 Hz, 1H), 7.07–7.06 (d, J = 6.9 Hz, 1H);

¹³C NMR (150 MHz, CDCl₃-*d*): 150.05, 149.08, 140.65, 140.19, 136.16, 133.53, 133.42, 128.25, 128.21, 125.95, 124.99, 124.95, 124.37, 114.29, 113.49, 76.04;

FTMS (APCI) calcd for C₄₂H₃₉N₄O₅ (M+H⁺): 323.985; found: 323.9945.



Supplementary Fig. 2. (**a**) UV–vis absorption spectra and (**b**) photoluminescence spectra of materials in THF solution and as thin films.



Supplementary Fig. 3. CV curves of MDN and RDN.



Supplementary Fig. 4. 2D GIWAXS patterns based on the structures of (**a**) Si-wafers/perovskite/P3HT, (**b**) Si-wafers/perovskite/M-P3HT, and (**c**) Si-wafers/perovskite/R-P3HT.



Supplementary Fig. 5. (a) UV–vis absorption spectra and (b) photoluminescence spectra of P3HT, M-P3HT, and R-P3HT films.



Supplementary Fig. 6. (a) HOMO region and (b) cut-off energy region of UPS.

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Sample	λ _{max}	(nm)	λ _{em} ((nm)	Eg (eV)	<i>Е</i> номо (eV)	<i>Е</i> ∟имо (eV)		
MDN	509 ^s	532 ^f	661 ^s	694 ^f	2.03	-5.31	-3.28		
RDN	514 ^s	532 ^f	667 ^s	694 ^f	2.03	-5.26	-3.24		
P3HT	53	38	66	60	1.93	-4.58	-2.65		
M-P3HT	53	38	66	60	1.93	-4.65	-2.72		
R-P3HT	53	38	66	61	1.93	-4.61	-2.68		

Supplementary Table 1. Optical and electrochemical properties.

^sResults from absorption in THF solution. ^fResults from absorption of films.



Supplementary Fig. 7. Band-edges of the stacked films in the device.



Supplementary Fig. 8. Conductivity of P3HT, M-P3HT and R-P3HT films.



Supplementary Fig. 9. Cross-sectional SEM images of PSCs with (**a**) P3HT, and (**b**) R-P3HT.



Supplementary Fig. 10. The top view SEM images with fresh devices of (**a**) perovskite, (**b**) perovskite/P3HT, (**c**) perovskite/M-P3HT, and (**d**) perovskite/R-P3HT.

Device	Voc (V)	J _{sc} (mA⋅cm⁻²)	FF (%)	PCE (%)
P3HT-REV ^{a)}	0.92	23.21	58.73	12.48
P3HT-FOR ^{b)}	0.82	23.04	36.75	6.98
M-P3HT-REV	1.16	24.58	80.17	22.87
M-P3HT-FOR	1.16	24.44	78.61	22.28
R-P3HT-REV	1.05	23.90	66.54	16.66
R-P3HT-FOR	1.00	23.90	58.19	13.87

Supplementary Table 2. Photovoltaic parameters of champion devices based on different HTMs measured under AM 1.5G Illumination.

^{a)} Reverse scan; ^{b)} Forward scan.



Supplementary Fig. 11. The statistical distributions of 20 devices with different concentrations of MDN and RDN of (**a**) *PCE*, (**b**) J_{SC} , (**c**) *FF*, and (**d**) V_{OC} .

Supplementary Table 3. Comparison of the device performance obtained with different HTM. The average values and standard deviations are calculated from 20 devices fabricated at the same conditions.

Device	Voc (V)	J _{SC} (mA⋅cm⁻²)	FF (%)	PCE (%)
РЗНТ	0.88 ± 0.03	23.01 ± 0.67	52.91 ± 5.14	10.73 ± 1.03
2-M-P3HT	1.09 ± 0.02	24.12 ± 0.51	71.94 ± 3.85	18.96 ± 1.05
4-M-P3HT	1.15 ± 0.02	24.55 ± 0.32	75.02 ± 3.09	21.19 ± 0.83
6-M-P3HT	1.06 ± 0.02	24.29 ± 0.48	70.96 ± 1.57	18.23 ± 0.52
R-P3HT	1.03 ± 0.03	23.87 ± 0.49	62.30 ± 2.60	15.33 ± 0.64



Supplementary Fig. 12. *J*–*V* characteristics of the best-performing based on the different HTMs (P3HT, 2-P3HT, 4-P3HT, 6-P3HT, and R-P3HT, respectively).



Supplementary Fig. 13. High-resolution XPS spectra of (**a**) Pb 4*f* spectra, and (**b**) I 3*d* spectra.



Supplementary Fig. 14. High-resolution XPS spectra of (**a**) I 3*d* spectra, (**b**) S 2*p* spectra, and (**c**) O 1*s* spectra.

Device	A 1	τ1	A 2	$ au_2$	Tave
perovskite	2.09	32.54	0.49	1192.64	1071.71
perovskite/P3HT	0.47	62.28	0.11	633.75	464.75
perovskite/R-P3HT	0.49	29.81	0.08	525.17	397.38
perovskite/M-P3HT	0.65	12.62	0.05	416.51	302.38

Supplementary Table 4. Time-resolved PL decay fitting parameters.



Supplementary Fig. 15. The statistical distributions of 20 devices w/ or w/o PS interface of (a) *PCE*, (b) J_{SC} , (c) V_{OC} , and (d) *FF*.

Supplementary Table 5. Comparison of the device performance obtained w/ or w/o PS. The average values and standard deviations are calculated from 20 devices fabricated at the same conditions.

Device	Voc (V)	J _{SC} (mA⋅cm⁻²)	FF (%)	PCE (%)
РЗНТ	0.88 ± 0.03	23.01 ± 0.67	52.91 ± 5.14	10.73 ± 1.03
PS/P3HT	0.96 ± 0.05	23.44 ± 0.35	58.02 ± 6.19	13.07 ± 1.36
M-P3HT	1.15 ± 0.02	24.55 ± 0.32	75.02 ± 3.09	21.19 ± 0.83
PS/M-P3HT	1.10 ± 0.01	24.34 ± 0.37	68.60 ± 2.65	18.35 ± 0.69
R-P3HT	1.03 ± 0.03	23.87 ± 0.49	62.30 ± 2.60	15.33 ± 0.64
PS/R-P3HT	1.01 ± 0.03	23.79 ± 0.47	64.73 ± 2.98	15.47 ± 0.83



Supplementary Fig. 16. The J-V curves of the champion devices based on the w/ or w/o PS interface.

Supplementary Table 6. Performances of the champion devices based on the w/ or w/o PS interface.

Device	Voc (V)	Jsc (mA⋅cm⁻²)	FF (%)	PCE (%)
РЗНТ	0.92	23.21	58.73	12.48
PS/P3HT	0.98	23.57	65.03	15.01
M-P3HT	1.16	24.58	80.17	22.87
PS/M-P3HT	1.10	24.79	71.43	19.53
R-P3HT	1.05	23.90	66.54	16.66
PS/R-P3HT	1.01	23.42	71.31	16.82



Supplementary Fig. 17. The statistical distributions of 20 devices with different concentrations of MDN and RDN interface of (**a**) PCE, (**b**) J_{SC} , (**c**) V_{OC} , and (**d**) *FF*.

Supplementary Table 7. Comparison of the device performance obtained with different concentrations of MDN and RDN interface. The average values and standard deviations are calculated from 20 devices fabricated at the same conditions.

Device	Voc (V)	J _{sc} (mA⋅cm⁻²)	FF (%)	PCE (%)
РЗНТ	0.88 ± 0.03	23.01 ± 0.67	52.91 ± 5.14	10.73 ± 1.03
2mg/ml MDN	1.05 ± 0.02	24.01 ± 0.49	58.05 ± 2.48	14.62 ± 0.74
4mg/ml MDN	1.07 ± 0.02	24.06 ± 0.50	63.56 ± 3.60	16.30 ± 1.06
6mg/ml RDN	0.99 ± 0.02	23.84 ± 0.60	54.62 ± 3.43	12.85 ± 0.65
4mg/ml RDN	0.98 ± 0.05	23.61 ± 0.51	58.03 ± 3.33	13.45 ± 1.04



Supplementary Fig. 18. The J-V curves of the champion devices based on the different concentrations of MDN and RDN interface.

Supplementary Table 8. Performances of the champion devices based on th	е
different concentration of MDN and RDN interface.	

Device	Voc (V)	Jsc (mA≀cm⁻²)	FF (%)	PCE (%)
РЗНТ	0.92	23.21	58.73	12.48
2mg/ml MDN	1.07	24.46	61.74	16.12
4mg/ml MDN	1.09	24.49	70.27	18.68
6mg/ml MDN	0.97	24.02	61.51	14.39
4mg/ml RDN	1.08	23.77	61.71	15.81



Supplementary Fig. 19. 2D GIWAXS patterns based on the structures of (**a**) Si-wafer/MDN, (**b**) Si-wafer/perovskite/MDN, and (**c**) Si-wafer/perovskite/MDN/P3HT.



Supplementary Fig. 20. Schematic diagram of the molecular stacking with (**a**) PS interface and (**b**) MDN interface.



Supplementary Fig. 21. Snapshot of PVK/MDN.



Supplementary Fig. 22. Snapshot of PVK-VI/MDN.



Supplementary Fig. 23. Snapshot of PVK/RDN.



Supplementary Fig. 24. Snapshot of PVK-VI/RDN.



Supplementary Fig. 25. LOL- π maps of MDN and RDN at different isosurface values. The π orbitals are displayed in green.



Supplementary Fig. 26. Molecular structure of (a) MDN, (b) PDN, (c) and TDN.



Supplementary Fig. 27. The J-V curves of the champion devices based on the different HTMs.

Supplementary Table 9. Photovoltaic parameters of champion	devices based
on different HTMs measured under AM 1.5G Illumination.	

Device	Voc (V)	J _{sc} (mA≀cm⁻²)	FF (%)	PCE (%)
РЗНТ	1.02	23.07	56.41	13.29
2-P-P3HT	1.11	23.39	66.04	17.22
4-P-P3HT	1.13	23.84	76.10	20.47
6-P-P3HT	1.12	22.49	63.39	16.01
1-T-P3HT	1.11	23.92	70.82	18.73
2-T-P3HT	1.11	24.47	71.68	19.53
3-T-P3HT	1.11	23.08	62.62	16.09



Supplementary Fig. 28. The statistical distributions of 20 devices with different HTMs of (a) PCE, (b) V_{OC} , (c) J_{SC} , (d) FF.

Supp	olementar	'y Table	10.	Comparise	on of the	e dev	rice perforr	mance obtai	ned
with	different	HTMs.	The	average	values	and	standard	deviations	are
calculated from 20 devices fabricated at the same conditions.									

Device	Voc (V)	J _{sc} (mA ⋅ cm ⁻²)	FF (%)	PCE (%)
РЗНТ	1.01 ± 0.04	22.24 ± 1.04	53.10 ± 2.90	11.93 ± 1.22
2-P-P3HT	1.10 ± 0.03	23.02 ± 0.52	61.33 ± 2.66	15.48 ± 0.96
4-P-P3HT	1.12 ± 0.02	23.48 ± 0.68	67.93 ± 4.51	17.89 ± 1.23
6-P-P3HT	1.07 ± 0.05	22.93 ± 0.58	59.46 ± 3.70	14.55 ± 0.81
1-T-P3HT	1.07 ± 0.04	23.49 ± 0.45	68.24 ± 2.74	17.23 ± 1.05
2-T-P3HT	1.08 ± 0.03	23.65 ± 0.59	68.65 ± 3.27	17.57 ± 1.28
3-T-P3HT	1.07 ± 0.04	23.13 ± 0.51	58.54 ± 2.36	14.51 ± 0.79



Supplementary Fig. 29. Hole mobility measurement characteristics by the space charge limited current (SCLC) method for the devices of (**a**) ITO/PEDOT:PSS/perovskite/P3HT/Ag, (**b**) ITO/PEDOT:PSS/perovskite/M-/P3HT/Ag, (**c**) ITO/PEDOT:PSS/perovskite/R-P3HT/Ag.



Supplementary Fig. 30. The top view SEM images with aging devices of (**a**) perovskite/P3HT, (**b**) perovskite/M-P3HT, and (**c**) perovskite/R-P3HT.



Supplementary Fig. 31. The water contact angle of perovskite with or without different HTMs.



Supplementary Fig. 32. Primitive cell of FA_{0.875}MA_{0.125}PbI₃. Only one Pb–I octahedron is displayed for clarity. The arrows denote the orientations of C–H bond in FA molecule.



Supplementary Fig. 33. Slab model with a vacuum of 20 Å of FA_{0.875}MA_{0.125}Pbl₃ after structural optimization.



¹H NMR, ¹³C NMR and MS spectra













