

Supporting Information for
“Stability and strength of monolayer polymeric C₆₀”

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A. Computational details

Density functional theory calculations are performed using the Vienna *ab-initio* simulation package (VASP)^{1,2}. The projector augmented wave (PAW) potential is used with C $2s^22p^2$ valence states^{3,4}, under the generalized gradient approximation (GGA) with the Perdew-Burke-Ernzerhof parameterization revised for solids (PBEsol) as the exchange-correlation functional⁵. For structural relaxation, a plane-wave cutoff of 800 eV is used with a \mathbf{k} -mesh of 5×5 and 3×5 for qTP and qHP C_{60} respectively, until the energy difference between successive steps is below 10^{-6} eV and the forces on the atoms are below 10^{-2} eV/Å. A vacuum spacing larger than 17 Å is used to eliminate interactions between neighboring unit cells along the c direction.

The elastic tensor coefficients (including ionic relaxations) are calculated using the finite differences method^{6,7}, and then renormalized by the c lattice constant ($C_{ij}^{2D} = c \times C_{ij}^{3D}$)^{8,9}.

The phonon spectra are calculated under the harmonic approximation based on density functional perturbation theory¹⁰⁻¹² using the PHONOPY code^{13,14}. The supercell size is 2×2 for qTP C_{60} with an electronic \mathbf{k} -point grid of (Γ -centered) 2×2 , which is well converged, as shown in Fig. 1. The supercell size is 1×2 for qHP C_{60} because the lattice constant a is nearly two times the b . The thermodynamic properties are calculated using a phonon \mathbf{q} -mesh of 91×91 for both qTP and qHP C_{60} monolayers.

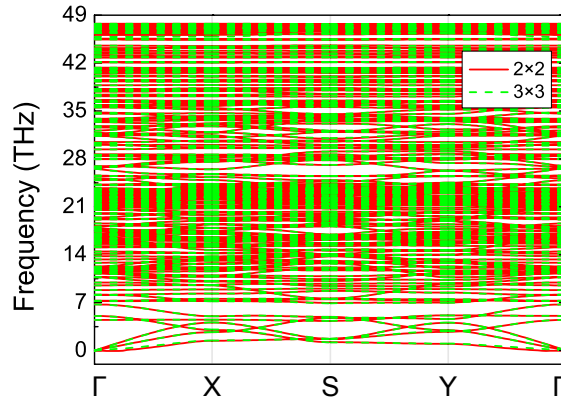


FIG. 1. Phonon spectra of qTP2 C_{60} calculated from 2×2 and 3×3 supercells.

The role of thermal expansion is investigated under the quasi-harmonic approximation. The Helmholtz free energy is calculated for anisotropically contracted and expanded lattice with a strain step of 0.4%, and at least 25 configurations are included for each phase when

fitting the Gibbs free energy for a temperature step of 10 K between 0–900 K.

B. Phonon group velocity

Figure 2 shows the calculated phonon group velocity v_g in 2D qTP2 and qHP C_{60} . The elastic constants can be estimated from the phonon speed of sound, i.e. phonon group velocity v_g near Γ ¹⁵

$$C_{11} = \rho^{2D} \times (v_x^{LA})^2, \quad (1)$$

$$C_{22} = \rho^{2D} \times (v_y^{LA})^2, \quad (2)$$

$$C_{66} = \rho^{2D} \times (v_x^{TA})^2, \quad (3)$$

where ρ^{2D} is the mass density in 2D and $v_{x/y}^{LA/TA}$ is the speed of sound for the longitudinal or transverse acoustic mode (LA/TA) along x/y . The computed elastic constants are listed in Table 2 in the main text.

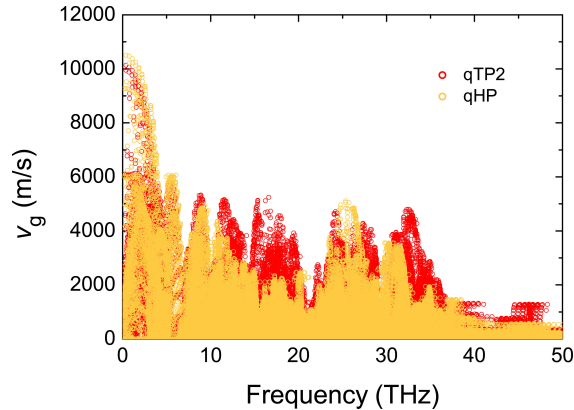


FIG. 2. Phonon group velocity in 2D qTP2 and qHP C_{60} .

C. Phonon density of states and entropy

With increasing temperature, the free energy F of qTP1 C_{60} drops faster than that of qTP2 C_{60} due to their smaller vibrational frequencies. As shown in Fig. 3(a), the cumulative phonon density of states (phDOS) of qTP1 fullerene is much larger than that of qTP2 and qHP fullerene at frequencies below 7 THz. The low phonon frequencies in qTP1 C_{60} give rise to higher entropy S in Fig. 3(b), especially at temperatures below 50 K. Consequently,

the Helmholtz free energy of qTP1 C_{60} drops faster because of its higher entropy S , which gives the curve of the free energy F as a function of temperature T ^{16,17}

$$S = -\partial F/\partial T. \quad (4)$$

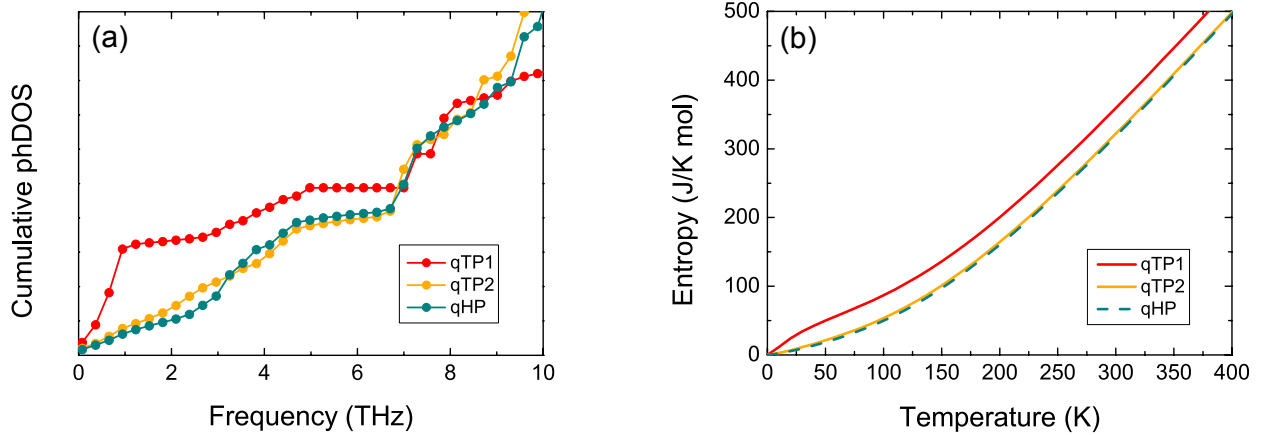


FIG. 3. (a) Cumulative phDOS and (b) phonon entropy of 2D qTP1, qTP2 and qHP C_{60} .

D. 1D qTP C_{60} chains

For structural relaxation of 1D qTP C_{60} chains, a plane-wave cutoff of 800 eV is used with a \mathbf{k} -mesh of 5 along the chain direction, until the energy difference between successive steps is below 10^{-6} eV and the forces on the atoms are below 10^{-2} eV/Å. A vacuum spacing larger than 20 Å is used to eliminate interactions between neighboring unit cells along the interchain directions. The top view of the crystal structure of 1D qTP C_{60} chain is shown in Fig. 4(a). The strong covalent [2+2] bonds along a leads to high electron localization function (ELF) values. The band structures in Fig. 4(b) are calculated using screened hybrid functional HSE06^{18–20}, with a direct band gap of 1.81 eV at the X high-symmetry point.

The interatomic force constants are calculated under the harmonic approximation based on density functional perturbation theory^{10–12} using the PHONOPY code^{13,14}, with a supercell of four qTP C_{60} molecules along the a direction. No imaginary mode is observed in Fig. 4(c), showing its dynamic stability. The extra zero-frequency mode at Γ other than the three acoustic modes is the typical torsional acoustic mode in 1D materials^{21,22}. For thermal

expansion calculations, the Helmholtz free energy is calculated for 8 uniaxially contracted and expanded configurations with a strain step of 0.2% to fit the Gibbs free energy.

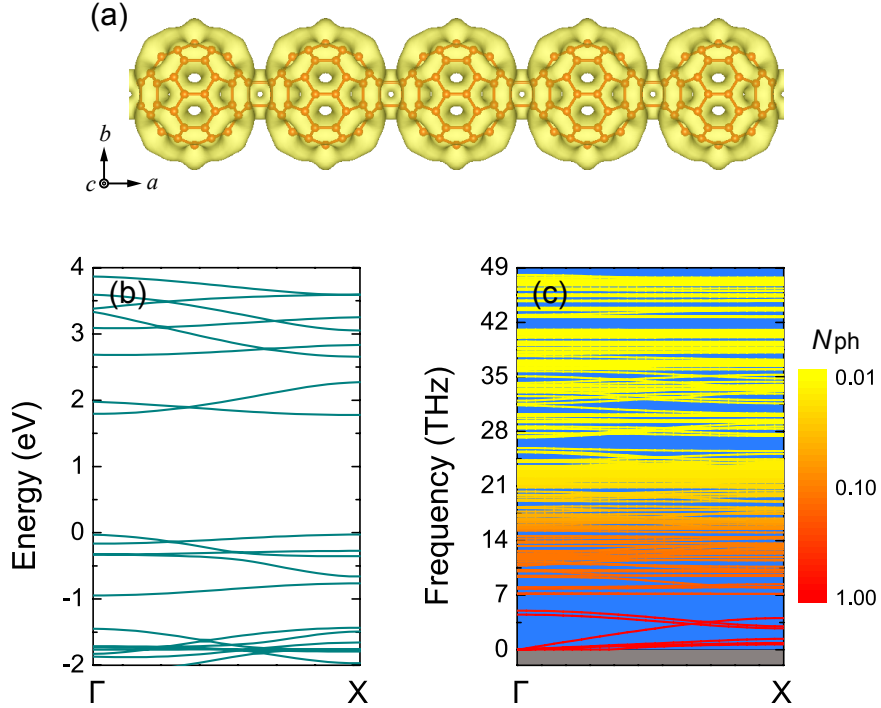


FIG. 4. (a) ELF of 1D qTP C₆₀ chain. The default isosurface level in VESTA²³ is used. (b) HSE06 band structures of 1D qTP C₆₀. (c) Phonon dispersion of 1D qTP C₆₀ with the phonon occupation number N_{ph} determined from the Bose-Einstein distribution function at 300 K.

E. Relative thermodynamic stability between 3D, 2D, 1D and 0D C₆₀

For structural relaxation of bulk polymeric C₆₀, a plane-wave cutoff of 800 eV is used with a \mathbf{k} -mesh of $5 \times 5 \times 3$ and $3 \times 5 \times 3$ for qTP and qHP C₆₀ respectively, until the energy difference between successive steps is below 10^{-6} eV and the forces on the atoms are below 10^{-2} eV/Å. The cohesive energy E_c for bulk single crystal polymeric C₆₀ is calculated and compared with their low-dimensional counterparts. As shown in Fig. 5, both 3D qTP1 and qHP C₆₀ are thermodynamically more stable than their 2D counterparts, whereas 3D qTP2 C₆₀ is energetically less favored than 2D qTP2 C₆₀. Among all the bulk single crystal polymeric C₆₀, 3D qTP1 C₆₀ has the lowest E_c while 3D qHP C₆₀ has the highest E_c .

To include the contribution of phonons, the Helmholtz free energy of 3D, 2D, 1D and 0D C₆₀ is calculated using the static lattice constants. The supercell size is $2 \times 2 \times 1$ and

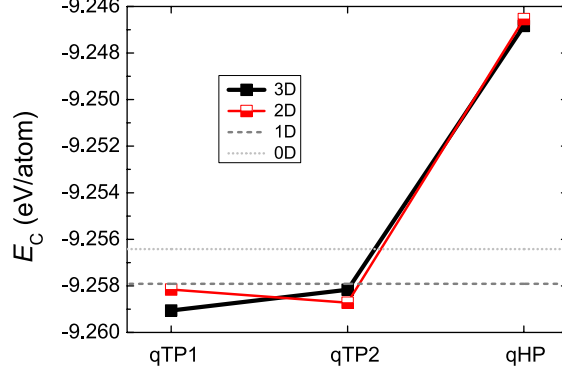


FIG. 5. Cohesive energy of bulk single crystal polymeric C_{60} and their low-dimensional counterparts.

$1 \times 2 \times 1$ for 3D qTP and qHP C_{60} respectively. Figure 6 shows the Helmholtz free energy of 3D, 2D and 1D qTP C_{60} as a function of temperature, with the free energy of monolayer qTP2 C_{60} set to zero to compare the relative stability. Interestingly, bulk qTP1 C_{60} has the lowest free energy in the entire temperature range (0 – 1000 K), whereas bulk qTP2 C_{60} has the highest free energy above 50 K. It should be noted that, different from the Gibbs free energy in the main text that includes the contribution of thermal expansion, for all the structural phases in Fig. 6 thermal expansion is not included because the computational cost for thermal expansion in anisotropic bulk materials is much higher than that in their 2D counterparts.

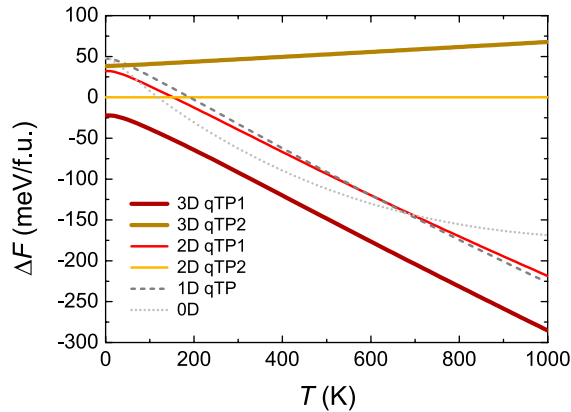


FIG. 6. Relative thermodynamic stability of bulk polymeric C_{60} , monolayer fullerene networks, one-dimensional fullerene chain and zero-dimensional fullerene molecule.

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