

Supplementary Information

Hybrid Assembly of Polymeric Nanofiber Network for Robust and Electronically Conductive Hydrogels

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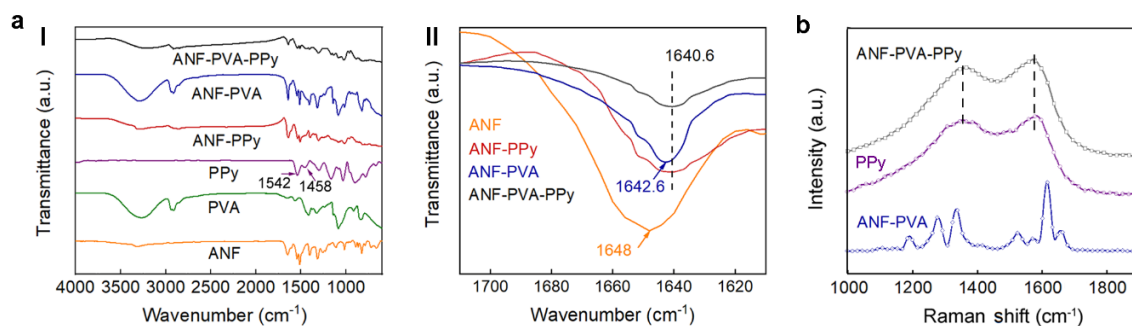
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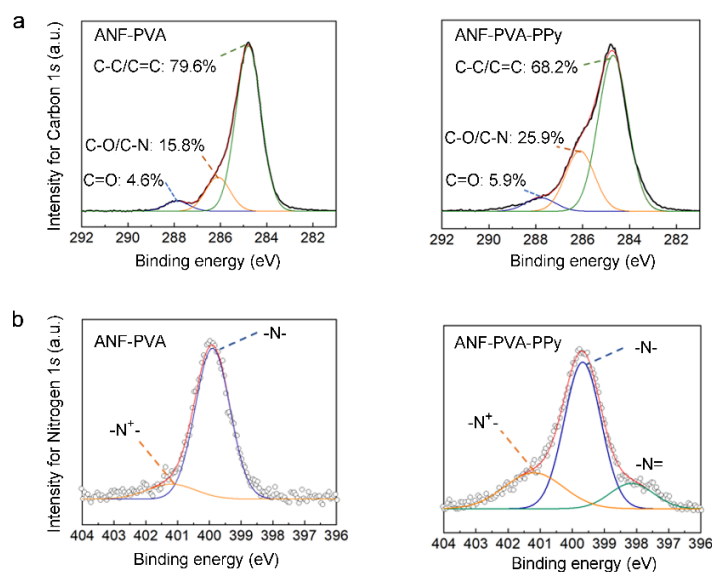
Supplementary Figures 1-17

Supplementary Table 1

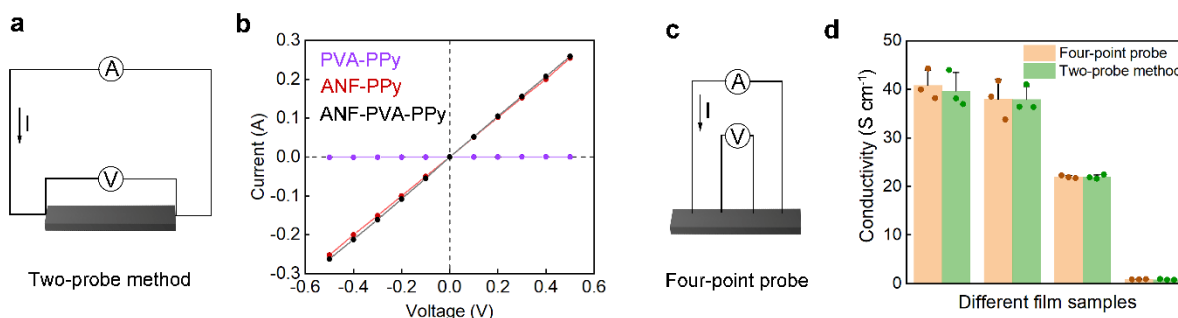
Supplementary Videos 1-2



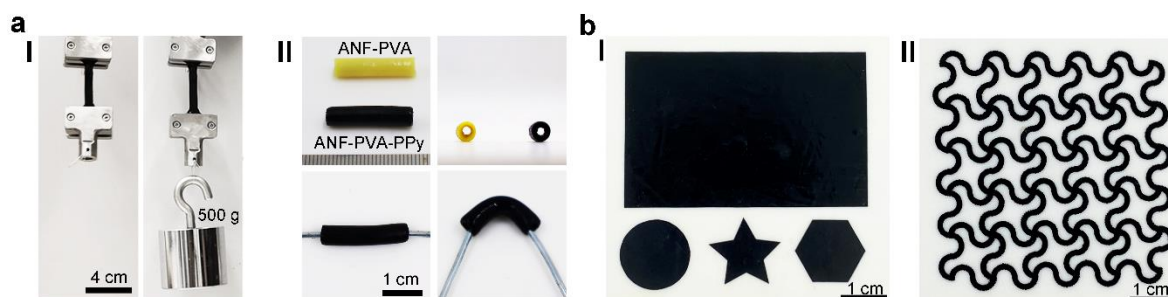
Supplementary Fig. 1 | Chemical structures characterized by FTIR and Raman spectroscopy. **a**, FTIR spectra (I) of ANF, PVA, PPy, ANF-PPy, ANF-PVA, and ANF-PVA-PPy. Magnified plot (II) showing spectra of aramid C=O stretching vibration in ANF, ANF-PVA, ANF-PPy and ANF-PVA-PPy. For ANF-PPy and ANF-PVA-PPy, characteristic peaks for C=C and =C–N symmetric and asymmetric ring stretching were measured at 1542 and 1458 cm^{-1} respectively, demonstrating the successful integration of PPy. For pure ANF, the aramid C=O stretching band was peaked at 1648 cm^{-1} . ANF-PVA has a red-shift of this peak to 1642.6 cm^{-1} , reflecting the effects of hydrogen bonding between ANF and PVA. A larger scale of red-shift to 1640.6 cm^{-1} was found in ANF-PPy composites, which indicates hydrogen bonding between ANF and PPy as well. **b**, Raman spectra of ANF-PVA, PPy, and ANF-PVA-PPy indicate the incorporation of PPy in the composites. Source data are provided as a Source Data file.



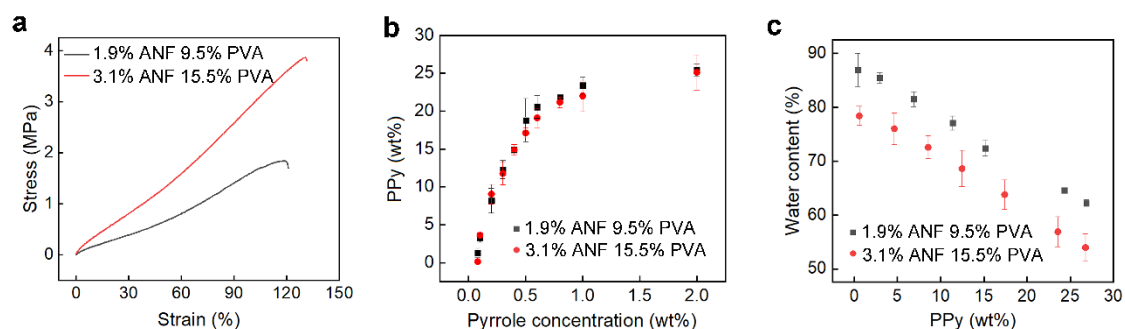
Supplementary Fig. 2 | XPS spectra of ANF-PVA and ANF-PVA-PPy. **a**, C 1s spectra of ANF-PVA and ANF-PVA-PPy. **b**, N 1s spectra of ANF-PVA and ANF-PVA-PPy. The C 1s spectra have 3 peaks at binding energies of 284.8, 286.1, and 287.9 eV, corresponding to C–C/C=C, C–O/C–N, and C=O, respectively. The presence of PPy is indicated by the increase of peak intensity at 286.1 eV. For N 1s spectra, 3 peaks at 398.1, 399.6, and 401.2 eV represent –N=, –N–, and –NH⁺– respectively. The emerging peak at 398.1 eV corresponding to –N= further confirms the incorporation of PPy into ANF-PVA-PPy. The significant increase of the peak intensity at 401.2 eV also demonstrates peroxidation of PPy in the presence of excess oxidant of FeCl_3 . Source data are provided as a Source Data file.



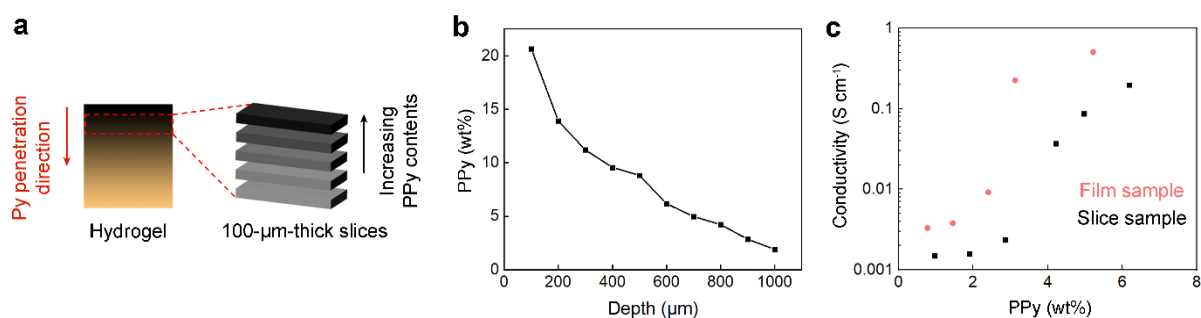
Supplementary Fig. 3 | Measurement of electrical conductivity. **a**, A schematic of conductivity measured by two-probe method. **b**, Current measured with bias voltage from -0.5 to 0.5 V by two-probe method, showing linear responses of CNHs. **c**, A schematic of conductivity measured by four-point probe. **d**, Conductivities measured by the two different methods are consistent, based on various ANF-PVA-PPy samples with different levels of conductivity. Data in **d** represent their means \pm SDs from $n=3$ independent samples. Source data are provided as a Source Data file.



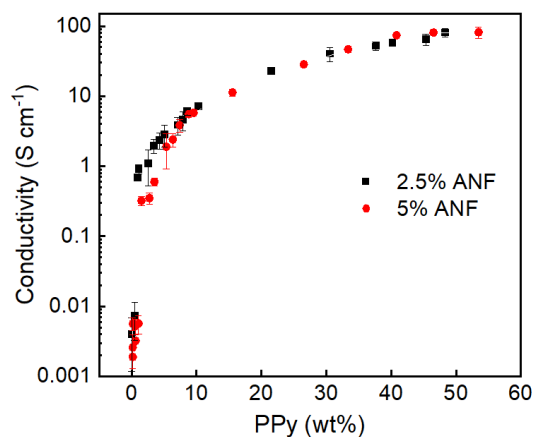
Supplementary Fig. 4 | Photographs of bulk and film ANF-PVA-PPy. **a**, Photographs of a strip sample under tension (I) and hollow tubes which can withstand bending (II). **b**, Photographs of films with various shapes (I) and patterned by infrared laser machining (II).



Supplementary Fig. 5 | Control of PPy content for ANF-PVA-PPy. **a**, Tensile stress-strain curves for 1.9% ANF 9.5% PVA and 3.1% ANF 15.5% PVA hydrogels. **b**, The PPy content in ANF-PVA-PPy hydrogels is tunable with various pyrrole concentrations during polymerization. The PPy content raised with the increase of pyrrole concentration until 1 wt% and then reached a plateau. **c**, The water content of conductive hydrogels. Both the CNHs prepared from 1.9% ANF 9.5% PVA and 3.1% ANF 15.5% PVA matrices demonstrate linear decrease in water content with increasing PPy content. Data in **b** and **c** represent their means \pm SDs from $n=3$ independent samples. Source data are provided as a Source Data file.

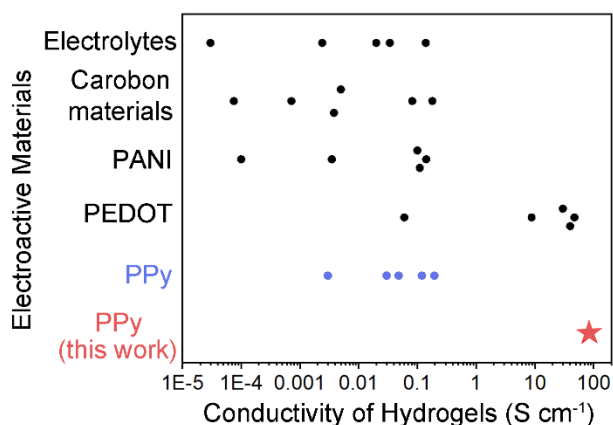


Supplementary Fig. 6 | PPy distribution within bulk hydrogels. **a**, Schematics of slicing a bulk ANF-PVA-PPy hydrogel with 100- μ m-thick layers. **b**, PPy content of each layer as a function of depth from the surface. The sample was obtained from Py (0.4 M), $FeCl_3$ (0.48 M), and 3.1% ANF 15.5% PVA matrix (18 mm in diameter and 8 mm in height) during the synthesis. **c**, Electrical conductivity as a function of PPy content for CNHs with 3.1% ANF 15.5% PVA matrix, obtained from both slice samples and film samples. The results show consistency of measured threshold for electrical percolation. The lower absolute values of conductivity for slice samples may be attributed to the structural defects introduced during the mechanical cryo-sectioning process. Source data are provided as a Source Data file.



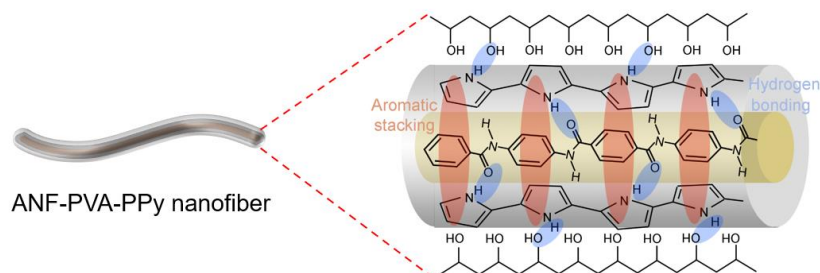
Supplementary Fig. 7 | Effects of PPy contents on the conductivity of ANF-PPy hydrogels.

The electrical conductivity of ANF-PPy with sparser nanofiber template (2.5% ANF) is higher than that of 5% ANF before the PPy content reaches 10 wt%. Then samples showed similar electrical conductivity under same PPy contents. Data represent their means \pm SDs from n=3 independent samples. Source data are provided as a Source Data file.

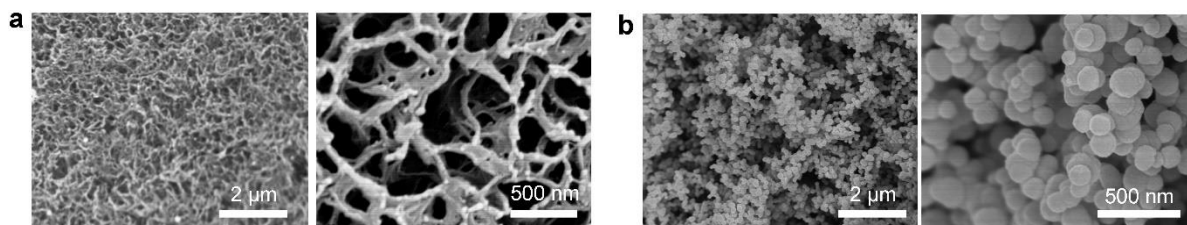


Supplementary Fig. 8 | Comparison of conductivities between CNHs and other reported conductive hydrogels.

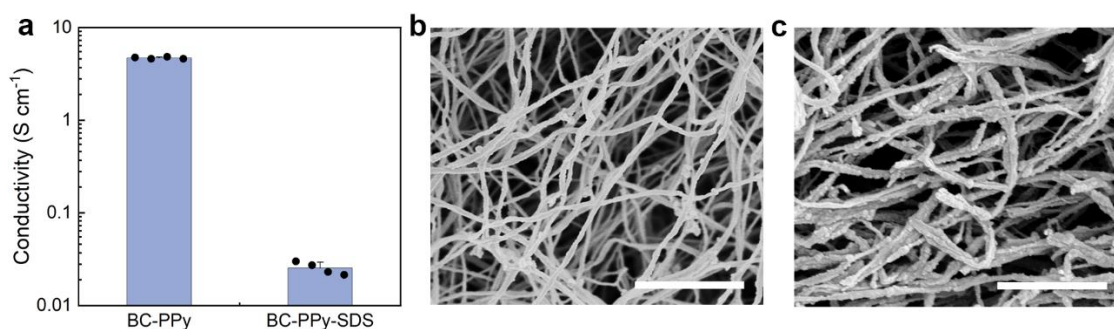
Data points were categorized based on the electroactive components in the hydrogels. More details are shown in Supplementary Table 1. Source data are provided as a Source Data file.



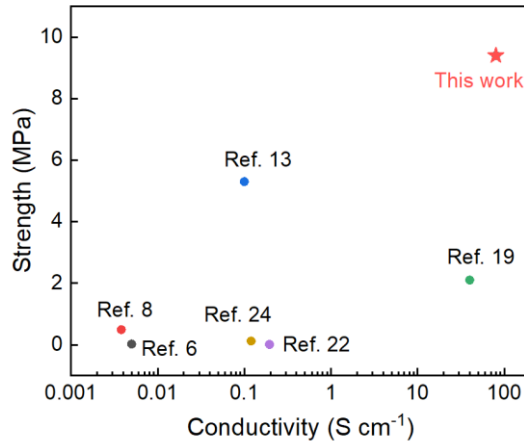
Supplementary Fig. 9 | Schematic illustration of possible intermolecular interactions in ANF-PVA-PPy. The aromatic rings and amide groups on ANFs afford attractions to PPy via π - π stacking and hydrogen bonding. Hydrogen bonding is also possible between PPy and PVA, as well as PVA and ANF.



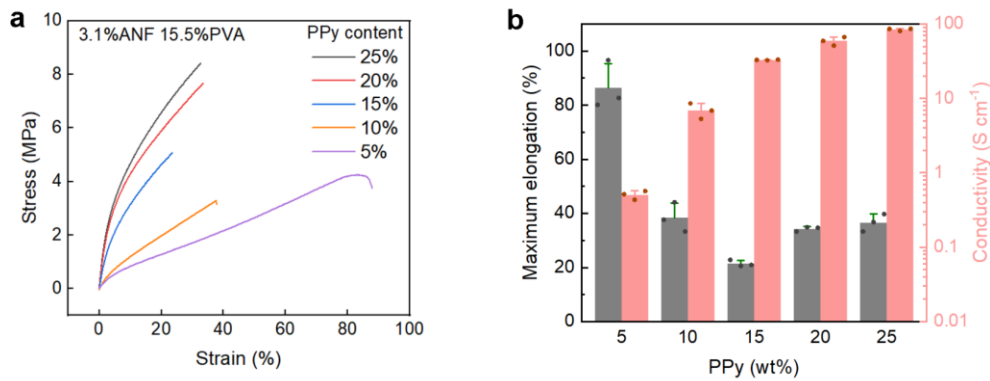
Supplementary Fig. 10 | SEM images of PPy distributed in various matrices. ANF-PPy (a) exhibited fibrillar networks while PPy formed spherical particles randomly distributed in the PVA matrix (b). The SEM images are based on cross-sections of hydrogel samples.



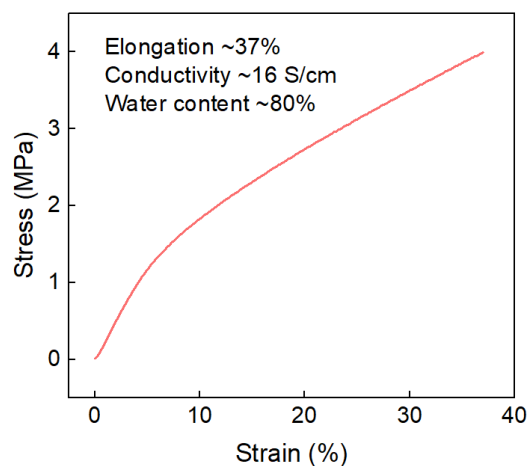
Supplementary Fig. 11 | Templating effect of bacterial cellulose (BC) hydrogels. a, Comparison of electrical conductivities between BC-PPy and BC-PPy-SDS hydrogels prepared under similar polymerization conditions. b-c, SEM images of BC (b) and BC-PPy (c). The PPy polymerized along the nanofibers indicates the templating effect of BCs. Scale bars: 1 μ m. Data in a represent their means \pm SDs from n=4 independent samples. Source data are provided as a Source Data file.



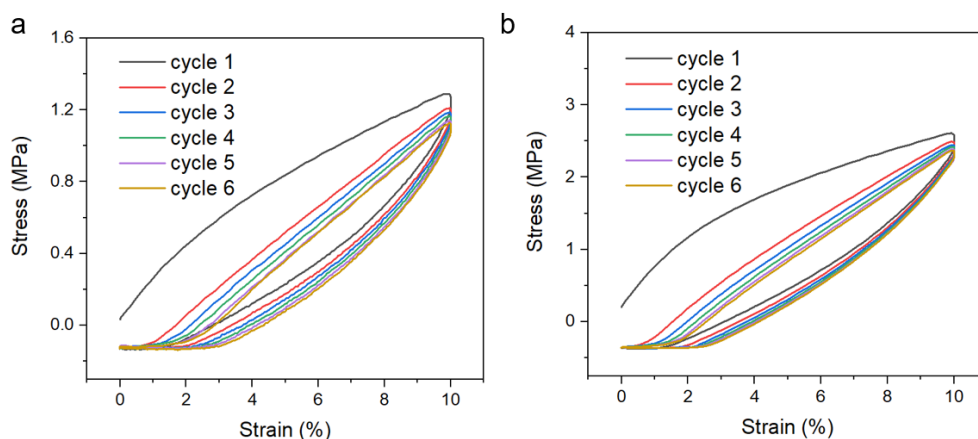
Supplementary Fig. 12 | Comparison of conductivities and mechanical strengths between CNHs and other reported conductive hydrogels. CNH exhibits outstanding electronic conductivity and mechanical strength. Source data are provided as a Source Data file.



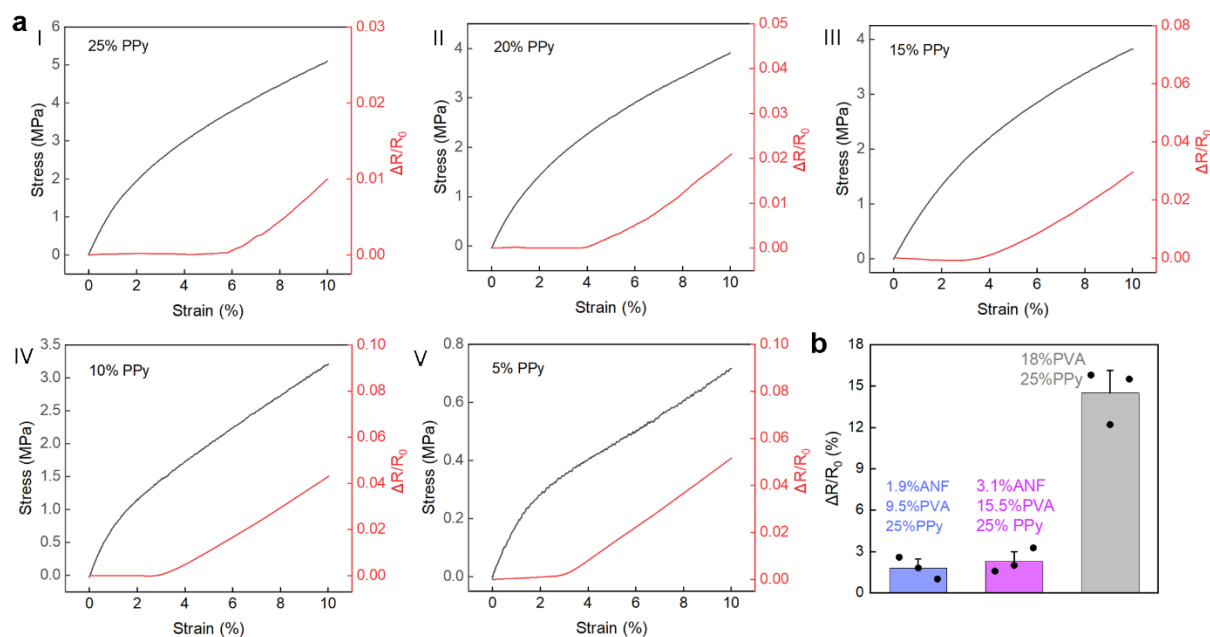
Supplementary Fig. 13 | Tensile behaviors of CNHs with various PPy contents based on matrices with 3.1% ANF 15.5% PVA. **a**, Stress-strain curves. **b**, Maximum elongation and conductivity of various samples. Data in **b** represent their means \pm SDs from $n=3$ independent samples. Source data are provided as a Source Data file.



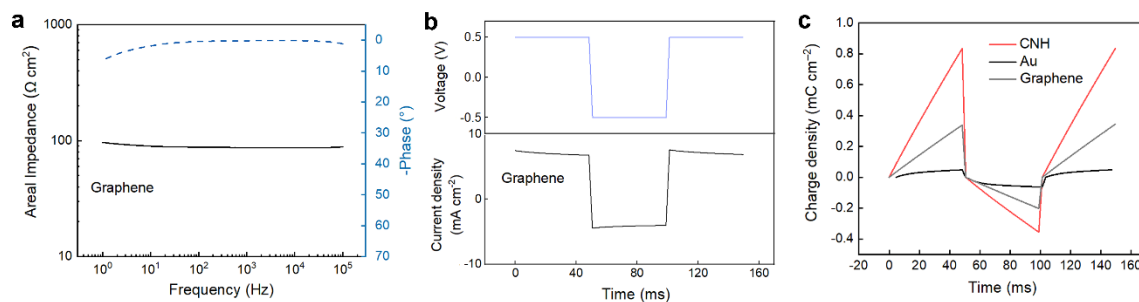
Supplementary Fig. 14 | Tensile stress-strain curve for a representative CNH. In an optimized configuration, the hydrogel shows a stretchability of 37%, conductivity of 16 S/cm, and water content of 80%. The hydrogel was synthesized based on 1.9% ANF 9.5% PVA matrix and 0.3 wt% pyrrole, polymerized under pH 7 and 0°C for 2 h. Source data are provided as a Source Data file.



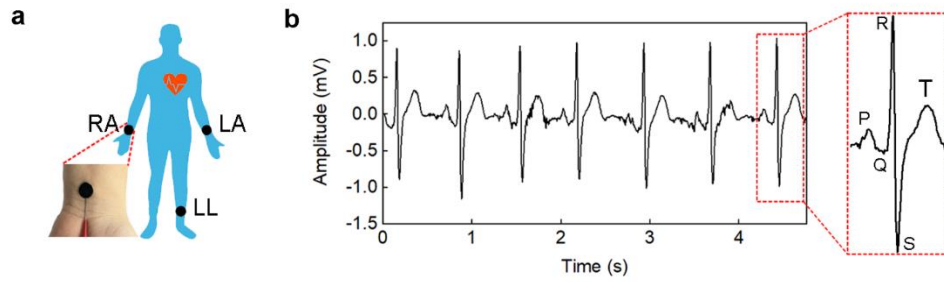
Supplementary Fig. 15 | CNHs with considerable stability under cyclic deformation. Cyclic stress-strain curves for CNHs based on 1.9% ANF, 9.5% PVA, 10% PPy (**a**) and 3.1% ANF, 15.5% PVA, 15% PPy (**b**). Source data are provided as a Source Data file.



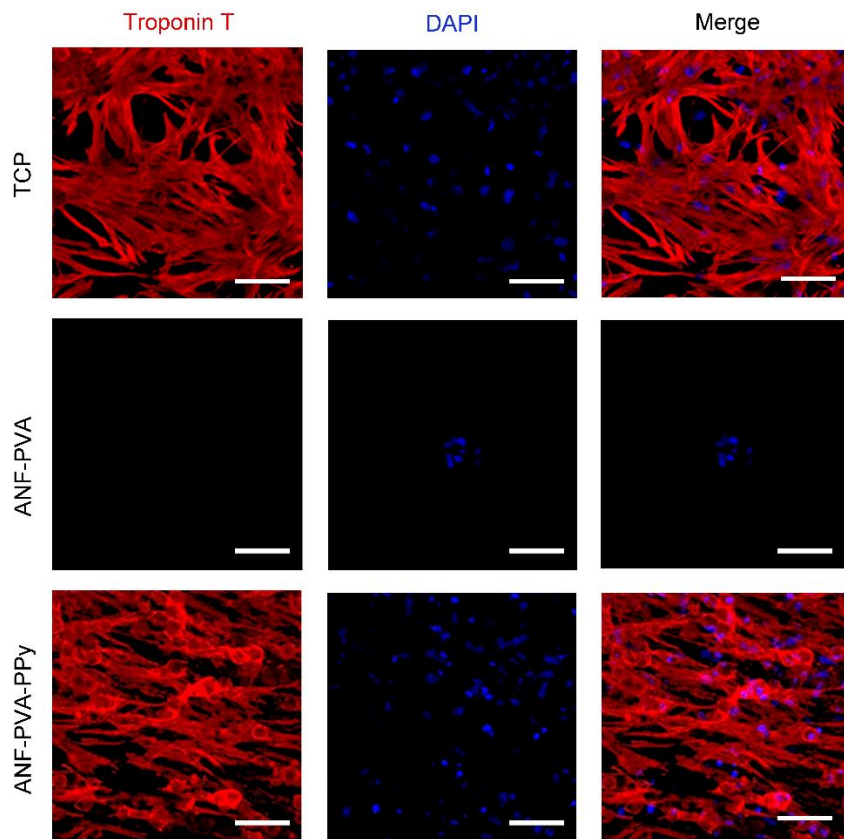
Supplementary Fig. 16 | Mechanical and electrical behaviors of CNHs under tensile deformation. **a**, The samples are based on 1.9% ANF 9.5% PVA matrices and incorporated with a PPy content of 25 wt% (**I**), 20 wt% (**II**), 15 wt% (**III**), 10 wt% (**IV**), and 5 wt% (**V**). The stress and change of resistance ($\Delta R/R_0$) are recorded simultaneously under imposed tensile strain, showing strain-invariant conductance within the initial range of deformation. **b**, Comparison of electrical responses of CNHs and PVA-PPy on tensile deformation of 10%. Data in **b** represent their means \pm SDs from $n=3$ independent samples. Source data are provided as a Source Data file.



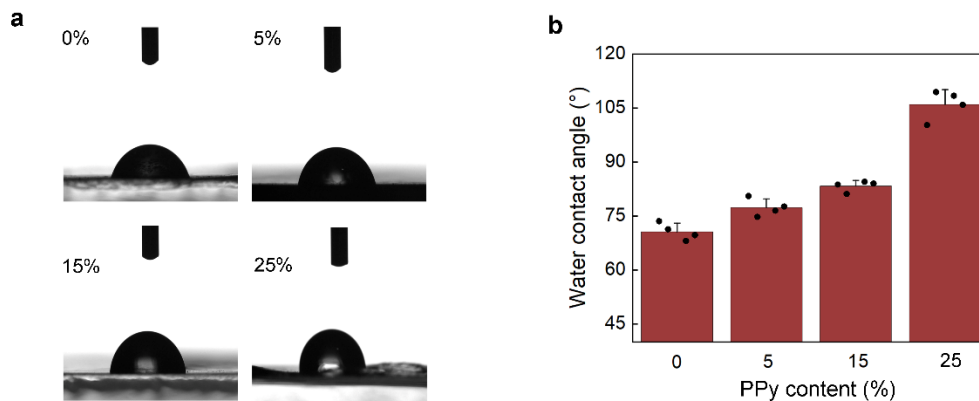
Supplementary Fig. 17 | Comparison of electrochemical properties. **a**, **b**, Electrochemical impedance spectroscopy (EIS) (**a**) and pulsed current injection curves for a graphene electrode with the voltage switching between -0.5 and 0.5 V vs. Ag/AgCl (**b**). **c**, Cyclic electrochemical current pulse injection curves of electrodes based on CNH, Au and graphene, with the voltage switching between -0.5 V and 0.5 V vs. Ag/AgCl. Source data are provided as a Source Data file.



Supplementary Fig. 18 | Measurement of ECG using CNH electrodes. **a**, The setup for the measurement. **b**, ECG signals captured by the CNH electrodes, showing details of QRS, T and P waves. Source data are provided as a Source Data file.



Supplementary Fig. 19 | Immunostaining of troponin (red) and nuclei (blue) of neonatal rat cardiomyocytes cultured on TCPs, ANF-PVA and ANF-PVA-PPy respectively. Cells are well attached to ANF-PVA-PPy substrates, showing good alignment and expression of cardiac-specific protein troponin T. Scale bar: 50 μm .



Supplementary Fig. 20 | Wettability of ANF-PVA-PPy with different PPy contents. a, photos of water droplets (3 μ l) on the surfaces of ANF-PVA-PPy (3.1% ANF, 15.5% PVA) with various PPy contents after 10s contact. **b,** Water contact angles as a function of PPy content. Data in **b** represent their means \pm SDs from n=4 independent samples. Source data are provided as a Source Data file.

Supplementary Table 1 | Summary of conductivities of other reported electroconductive hydrogels.

Electroactive Materials	Hydrogel Matrix	Conductivity (S cm ⁻¹)	Ref.	
Electrolytes	NaCl	Hydroxypropyl cellulose/PVA	0.034	1
	LiCl	PAAm	0.139	2
	sodium citrate	Aramid nanofiber/PVA	0.02	3
	Na ₂ SO ₄	PEG	0.0024	4
	choline chloride	Poly α-lipoic acid	0.00003	5
Carbon materials	GO	GO	0.005	6
	graphene	PAAm	0.000075	7
	graphene	Collagen	0.0038	8
	rGO	Polydopamine/PAAm	0.18	9
	CNTs	P(AAm-AAc)	0.082	10
	CNTs	Chitosan/Gelatin	0.00072	11
PANI	PANI	Chitosan/PEG/PANI	0.0035	12
	PANI	PVA/PANI	0.1	13
	PANI	PSS/PANI	0.142	14
	PANI	Phytic acid/PANI	0.11	15
	PANI	N-fluorenylmethoxycarbonyl diphenylalanine/PANI	0.0001	16
PEDOT	PEDOT	PEDOT/PSS	8.8	17
	PEDOT	PEDOT/PSS	47.4	18
	PEDOT	PEDOT/PSS	40	19
	PEDOT	PEDOT/sulfonated lignin/PAAc	0.06	20
	PEDOT	PU/ PEDOT/PSS	30	21
PPy	PPy	Agarose	0.195	22
	PPy	Silk fibroin	0.03	23
	PPy	PAAm	0.12	24
	PPy	Cellulose nanofibers/PVA	0.048	25
	PPy	PAAm/Chitosan	0.003	26

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