Metallomimetic Chemistry of a Cationic, Geometrically Constrained Phosphine in the Catalytic Hydrodefluorination and Amination of Ar–F Bonds

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1. General experimental considerations:

All preparations were carried out under an anhydrous N₂ atmosphere using standard Schlenk and glove box techniques. All glassware was oven dried and cooled under vacuum before use. Commercial reagents were purchased from Sigma Aldrich, Strem or Apollo Scientific and used without further purification unless indicated otherwise. **2** was prepared following the reported procedure.^[1] NMR spectra were recorded at room temperature using a Bruker AvanceIII-400 MHz spectrometer. Data for ¹H NMR are reported as follows: chemical shift (δ ppm), integration, multiplicity (s = singlet, d = doublet, t = triplet, q = quartet, quin = quintet, sep = septet, m = multiplet, br = broad), coupling constant (Hz), assignment.

2. X-ray Crystallography:

Single crystal X-ray diffraction data were collected on a Bruker KAPPA APEX II diffractometer equipped with an APEX II CCD detector using a TRIUMPH monochromator with a MoK α X-ray source (α = 0.710 73 Å). The crystals were mounted on a cryoloop with Paratone oil, and all data were collected at 100(2) K. Unit cell determination and refinement and data collection were done using the Bruker APEX-II suite,^[2] data reduction and integration were performed using SAINT v8.34A (Bruker, 2013)^[3] and absorption corrections and scaling were done using SADABS-2014/5 (Bruker,2014/5)^[4]. All the crystal structures were solved through OLEX2^[5] package using SHELXT^[6] and the structures were refined using SHELXL^[6]. All non-hydrogen atoms were refined anisotropically. All the figures were generated using Mercury 3.10.2.

3. Synthetic Procedures:

Synthesis of [1⁺][Cl]: Dilithiated hexaphenyl-carbodiphosphorane (2-Li₂) ligand was prepared in Et₂O (40 mL) solution from 2 (610 mg, 1.14 mmol) using a reported procedure.^[7] PCl₃ (1 mL, 1.14 mmol) was added at -78 °C. The reaction mixture was allowed to warm to room temperature and stirred for overnight. A pale-white powder precipitated from the colorless solution. The Et₂O solution was decanted and the remaining solid was dried under vacuum

and washed with hexane (50 mL). The solid obtained was dissolved in DCM and filtered to remove LiCI. All the volatiles were evaporated under vacuum to afford a yellow-creamy powder. Yield: 520 mg (76.11%). ¹**H-NMR** (500 MHz, CDCl₃), δ : 6.91 (m, 4H), 7.13 (t, *J* = 8 Hz, 4H), 7.41

(t, J = 8 Hz, 2H), 7.64-7.65 (br-m, 8H), 7.68 (m, 2H), 7.74 (t, J = 8 Hz, 2H), 7.97 (m, J = 8 Hz, 4H), 8.14 (t, 8 Hz, 2H). ¹³**C-NMR** (125.8 MHz, CDCl₃), δ : 120.57 (d, J = 72 Hz), 128.72, 129.94, 130.22, 130.76, 131.75, 132.75, 133.50, 133.77, 134.23, 151.27 (br) ppm. ³¹**P-NMR** (202.5 MHz, CDCl₃): $\delta = 29.80$ (s, CP_2), 54.24 (s, CP) ppm. HRMS (ES+): Calculated for C₃₇H₂₈P₃: 565.1404 [M-Cl]⁺; Obs: 565.1417.



Figure S1. ¹H NMR spectrum (500 MHz, in CDCl₃) of [1⁺][Cl].



Figure S2. ^{31}P NMR spectrum (202.5 MHz, in CDCI₃) of [1⁺][CI].



Figure S3. ¹³C NMR spectrum (125.8 MHz, in CDCI₃) of [1⁺][CI].

Synthesis of [1+][PF6]: KPF6 (175.2 mg, 0.952 mmol) was added to a 10 mL DCM solution of



[1⁺][CI] (520 mg, 0.865 mmol) and the mixture was stirred overnight at room temperature. The solution was filtered to separate KCI salt. All the volatiles were evaporated under vacuum to produce a yellow-beige powder. Dissolving this powder in 1:10 DCM/hexane mixture and slow evaporation of this solution at room temperature after 2 days afforded

yellow crystals. Yield: 605 mg (98.41%). ¹**H-NMR** (400 MHz, CDCl₃), δ : 6.82 (m, 4H), 7.00 (t, J = 8 Hz, 4H), 7.30 (t, J = 8 Hz, 2H), 7.51-7.63 (m, 12H), 7.91 (m, 4H), 8.05 (t, J = 8 Hz, 2H). ¹³**C-NMR** (100 MHz, C₆D₆), δ : 120.55 (d, J = 90 Hz), 129.85, 130.24, 130.66, 131.74, 132.66, 133.41, 133.66, 134.16, 151.37. ³¹**P-NMR** (162 MHz, CDCl₃), δ : -144.73 (sep, J = 711 Hz, PF₆), 31.41 (s, CP₂), 55.70 (s, CP) ppm. ¹⁹**F-NMR** (376.5 MHz, CDCl₃), δ : -73.85 (d, 711 Hz) ppm. **HRMS** (ES+): Calculated for C₃₇H₂₈P₃: 565.1404 [M-PF₆]⁺; Obs: 565.1417.



Figure S4. ¹H NMR spectrum (400 MHz, in CDCl₃) of [1⁺][PF₆].



Figure S5. ³¹P NMR spectrum (162 MHz, in CDCl₃) of $[1^+][PF_6]$.



Figure S6. ¹³C NMR spectrum (100 MHz, in CDCl₃) of [1⁺][PF₆].







Synthesis of [4+][PF₆]: 10 equivalents of pentafluoropyridine (232 μ L, 2.11 mmol) were added to a J-Young NMR tube containing 0.5 mL *o*-DFB solution of [1+][PF₆] (150 mg, 0.211 mmol). The reaction progress was monitored by NMR spectroscopy and left for heating at 80 °C for 3 hours until its full conversion to **4+**. All the volatiles were evaporated under vacuum that yielded product [**4+**][PF₆] (90% NMR yield).

*(4) is contaminated with a minute amount of oxidized (P=O) product of [1+][PF₆]. ¹H-NMR (400 MHz, CDCl₃), δ: 6.73 (m, 4H), 7.00 (t, J = 8 Hz, 4H), 7.27 (t, J = 8 Hz, 2H), 7.47 (m, 6H), 7.62 (m, 2H), 7.77-7.91 (m, 8H), 8.74 (m, 2H) ppm. ¹³C-NMR (100 MHz, CDCl₃), δ: 128.46, 129.79, 131.94, 132.59, 133.38, 134.01, 139.16 ppm. ³¹P-NMR (162 MHz, CDCl₃), δ: -144.57 (sep, J = 711 Hz, PF₆), -48.87 (dt, J = 666 Hz, 52 Hz, P-F), 13.91 (d, J = 52, CP₂) ppm. ¹⁹F-NMR (376.5 MHz, CDCl₃), δ: -133.71 (s, Ar-*F*), -89.32 (s, Ar-*F*), -73.50 (d, J = 715 Hz, PF₆), 1.92 (d, J = 666 Hz, P-F) ppm. HRMS (ES+): Calculated for C₄₂H₂₈NF₅P₃: 734.1355 [M-PF₆]⁺; Obs: 734.1357.



Figure S8. ¹H NMR spectrum (400 MHz, in CDCl₃) of [4⁺][PF₆].



Figure S9. ³¹P NMR spectrum (162 MHz, in CDCl₃) of [4⁺][PF₆].



Figure S10. ¹³C NMR spectrum (100 MHz, in CDCI₃) of $[4^+][PF_6]$.



Figure S11. ¹⁹F NMR spectrum (376.5 MHz, in CDCl₃) of [4⁺][PF₆].

Synthesis of [1+-F2][PF6] and [1+-O][PF6]: A J-Young NMR tube was charged with oDFB (0.5



mL) solution of $[4^+][PF_6]$ (152 mg, 0.173 mmol). The solution was heated at 110 °C and monitored by NMR spectroscopy. After 10 hours the reaction solution was filtered, evaporated, and fully dried. Crystals of $[1^+-F_2][PF_6]$ (Yield: 32 mg (25.07%)) were obtained from by slow evapoartion from DCM/hexane (1:10) solution at room temperature.

NMR data of [1+-*F*₂][*PF*₆]: ¹H-NMR (400 MHz, in DCM, DMSO-d₆ cap.), δ: 7.72-7.81 (m, 16H), 7.97-8.04 (m, 8H), 8.27 (m, 2H), 8.59 (m, 2H). ¹³C-

NMR (100 MHz, in DCM, DMSO-d₆ cap.), δ : 119.80 (t, J = 44 Hz), 127.39 (d, J = 17 Hz), 128.81 (t, J = 6 Hz), 129.94 (br), 131.56 (t, J = 5 Hz), 133.02 (d, J = 12 Hz), 133.28 ppm. ³¹**P-NMR** (162 MHz, in DCM, DMSO-d₆ cap.), $\delta = -145.11$ (sep, J = 711 Hz, PF_6), -2.08 (tt, J = 1067 Hz, 76 Hz, PF_2), 13.74 (d, J = 76 Hz, CP_2) ppm. ¹⁹**F-NMR** (376.5 MHz, in DCM, DMSO-d₆ cap.), δ : -74.03 (d, J = 711 Hz, PF_6), -37.41 (d, J = 1069 Hz, PF_2) ppm. **HRMS** (ES+): Calculated for C₃₇H₂₈F₂P₃: 603.1372 [M-PF₆]⁺; Obs: 603.1386.



Figure S12. ¹H NMR spectrum (400 MHz, in DCM, DMSO-d₆ cap.) of [1⁺-F₂][PF₆].



Figure S13. ³¹P NMR spectrum (162 MHz, in DCM, DMSO-d₆ cap.) of [1+-F₂][PF₆].



Figure S14. ¹³C NMR spectrum (100 MHz, in DCM, DMSO-d₆ cap.) of [1+-F₂][PF₆].



Figure S15. ¹⁹F NMR spectrum (376.5 MHz, in DCM, DMSO-d₆ cap.) of [1+-F₂][PF₆].

NMR data of [1⁺-*O*][*PF*₆]: ¹H-NMR (400 MHz, CDCl₃), δ: 6.94 (m, 4H), 7.13 (t, J = 8 Hz, 4H), 7.43 (t, J = 8 Hz, 2H), 7.66-771 (m, 8H), 7.76 (t, J = 8 Hz, 2H), 7.90 (m, 2H), 8.08 (m, 4H), 8.39 (t, J = 8 Hz) ppm. ¹³C-NMR (100 MHz, CDCl₃), δ: 117.59 (d, J = 103 Hz), 125.84 (d, J = 103 Hz), 130.32, 132.59, 133.75, 134.50, 134.98, 140.95 (d, J = 110 Hz) ppm. ³¹P-NMR (162 MHz, CDCl₃): $\delta = -144.45$ (sep, J = 711 Hz, *P*F₆), 26.18 (d, J = 52 Hz, *CP*₂), 59.68 (t, J = 52 Hz, *P*=O) ppm. ¹⁹F-NMR (376.5 MHz, CDCl₃), δ : -73.55 (d, J = 711 Hz, *PF*₆) ppm. HRMS Calculated for C₃₇H₂₈OP₃: 581.1353 [M-PF₆]⁺; Obs: 581.1362.



Figure S16. Single crystal X-ray structure of [**1-O**⁺][PF₆]. Thermal ellipsoids at 50% probability, hydrogen atoms were omitted for clarity.



Figure S17. ¹H NMR spectrum (400 MHz, CDCl₃) of [1+-O][PF₆].



Figure S18. ³¹P NMR spectrum (162 MHz, CDCl₃) of [1+-O][PF₆].





Figure S20. ¹⁹F NMR spectrum (376.5 MHz, CDCl₃) of [1+-O][PF₆].

Perfluoro-4,4'-bipyridine (5): ¹⁹**F-NMR** (376.5 MHz, CDCl₃), δ: -137.87 (s, 2F), -87.12 (s, 2F) ppm.^[8]



Figure S21. ¹⁹F NMR spectrum (376.5 MHz, in CDCl₃) of 5.

General procedure:

0.1 equiv. of [1+][PF₆] was dissolved in *o*DFB. Next, 1.0 equivalent of Ar-F and 1.0 equivalent of PhSiH₃ (it is important to note that only PhSiF₃ was observed in the ¹⁹F NMR spectra after reaction completion, meaning that 0.3 equivalent of PhSiH₃ is consumed in these reactions) were added to the solution and transferred to a J-Young NMR tube. The tube was heated to monitor the progress of reaction (conditions specified in Table 1). After completion, the solution was evaporated and fully dried. The residue was dissolved in n-hexane and the resulting mixture was passed through celite column. Hexane was then evaporated and the product thoroughly dried. If the hydrodefluorination product is volatile, the reaction mixture was analyzed by quantitative ¹⁹F NMR with a sealed glass capillary containing DMSO-d₆.



2,3,5,6-tetrafluoropyridine (7): ¹⁹**F-NMR** (376.5 MHz, in *o*-DFB, DMSO cap.), *δ*: -141.00 (s, 2F), -92.57 (s, 2F) ppm. NMR data is consistent with the reported literature.^[9]



2,3,5,6-tetrafluorobenzonitrile (8): ¹⁹**F-NMR** (376.5 MHz, in *o*-DFB, DMSO cap.), δ : -136.78 (s, 2F), -134.57 (s, 2F) ppm. NMR data is consistent with the reported literature.^[9]



Methyl 4-(diethylamino)-2,3,5,6-tetrafluorobenzoate (9): ¹⁹**F-NMR** (376.5 MHz, CDCl₃), δ : -139.44 (s, 2F), -137.36 (s, 2F) ppm. NMR data is consistent with the reported literature.^[9]



1,2,4,5-tetrafluoro-3-(trifluoromethyl)benzene (10): ¹⁹**F-NMR** (376.5 MHz, in *o*-DFB, DMSO cap.), δ : -141.93 (s, 2F), -137.68 (s, 2F), -57.49 (t, *J* = 22.7 Hz, 3F) ppm. NMR data is consistent with the reported literature.^[9]



2,2',3,3',4,5,5',6,6'-nonafluoro-1,1'-biphenyl (11): ¹⁹**F-NMR** (376.5 MHz, -H CDCl₃), *δ*: -138.36 (s, 2F), -137.82 (s, 2F) ppm. NMR data is consistent with the reported literature.^[9]



Tris(2,3,5,6-tetrafluorophenyl)phosphane (12): ¹⁹**F-NMR** (376.5 MHz, CDCI₃), δ : -137.07 (m, 2F), -129.94 (m, 2F) ppm. NMR data is consistent with the reported literature.^[9]



Figure S22. ¹⁹F NMR spectrum (376.5 MHz, o-DFB) of 7.



Figure S23. ¹⁹F NMR spectrum (376.5 MHz, o-DFB) of 8.



Figure S24. ¹⁹F NMR spectrum (376.5 MHz, CDCl₃) of 9.



Figure S25. ¹⁹F NMR spectrum (376.5 MHz, *o*-DFB) of **10**.



Figure S26. ¹⁹F NMR spectrum (376.5 MHz, CDCl₃) of 11.



Figure S27. ¹⁹F NMR spectrum (376.5 MHz, CDCl₃) of **12**.



Figure S28. Stacked ³¹P (162 MHz) and ¹⁹F (376.5 MHz) NMR spectra (in *o*DFB) recorded at variable temperatures (VT) to show the resting state during the catalytic hydrodefluorination of C_5F_5N using [1+][PF₆].



Figure S29. Stacked ³¹P (162 MHz, in *o*DFB) NMR spectra recorded at variable temperatures (VT) to show the formation of [1⁺][PF₆] from the reaction of [4⁺][PF₆] with DIBAL-H.

Et₃Si-NEt₂:

Et₃Si-NEt₂ was synthesized according to the procedure described earlier.^[10] **H-NMR** (400 MHz, CDCl₃), δ : 0.53 (q, *J* = 7.7 Hz, 6H), 0.94 (dt, *J* = 15.6, 7.3 Hz, 15H), 2.79 (q, *J* = 6.8 Hz, 4H), ppm. ¹³C-NMR (100 MHz, CDCl₃), δ : 4.83, 7.64, 16.10, 40.13 ppm. ¹H-²⁹Si HMBC NMR (79.5 MHz, CDCl₃), δ : 8.8 ppm.

Et₃Si-N(H)Bn:

Et₃Si-N(H)Bn (Bn = PhCH₂) was synthesized according to the procedure described earlier.^[10] ¹H-NMR (500 MHz, CDCl₃), δ : 0.63 (q, J = 7.9 Hz, 6H), 0.73 (br-s, 1H), 1.01 (t, J = 7.9 Hz, 9H), 4.00 (d, J = 7.9 Hz, 2H), 7.25-7.39 (m, 5H) ppm; ¹³C-NMR (100 MHz, CDCl₃), δ : 4.78, 7.22, 46.16, 126.38, 126.95, 128.25, 144.58 ppm; ¹H-²⁹Si HMBC NMR (79.5 MHz, CDCl₃), δ : 9.3 ppm.



Figure S30. ¹H NMR spectrum (400 MHz, CDCl₃) of Et₃Si-NEt₂.



Figure S31. ¹³C NMR spectrum (100 MHz, CDCl₃) of Et₃Si-NEt₂.



Figure S32. ¹H-²⁹Si HMBC NMR spectrum (CDCl₃) of Et₃Si-NEt₂.





Figure S34. ¹³C NMR spectrum (100 MHz, CDCl₃) of Et₃Si-N(H)Bn.



Figure S35. ¹H-²⁹Si HMBC NMR spectrum (CDCl₃) of Et₃Si-N(H)Bn.

General procedure:

0.1 equiv. of [1+][PF₆] was dissolved in *o*DFB. Next, 1.0 equivalent of Ar-F and 1.0 equivalent of Et₃Si-NEt₂ or Et₃Si-N(H)Bn were added to the solution and transferred to a J-Young NMR tube. The tube was heated to monitor the progress of reaction (conditions specified in Table 2). After completion, the solution was evaporated and fully dried. The residue was dissolved in n-hexane and the resulting mixture was passed through celite column. Hexane was then evaporated and the product thoroughly dried.

N,N-diethyl-2,3,5,6-tetrafluoropyridin-4-amine (14):



¹**H-NMR** (400 MHz, CDCl₃), δ: 1.21 (t, J = 7.1 Hz, 6H), 3.42 (qt, J = 7.0 Hz, 1.5 Hz, 4H) ppm. ¹³**C-NMR** (100 MHz, CDCl₃), δ: 13.79, 46.57 (t, J = 4 Hz), 134.30 (dm, J = 248.9 Hz), 139.37, 145.17 (dt, J = 238.4, 16.6 Hz) ppm. ¹⁹**F-NMR** (376.5

MHz, CDCl₃), δ : -94.67 (s, 2F), -156.29 (s, 2F) ppm. NMR data is consistent with the literature.^[11]

N-benzyl-2,3,5,6-tetrafluoropyridin-4-amine (14a):



¹**H-NMR** (400 MHz, CDCl₃), δ: 4.71 (d, J = 5.7 Hz, 2H), 4.80 (bs, 1H), 7.32-7.42 (m, 5H) ppm. ¹³**C-NMR** (100 MHz, CDCl₃), δ: 48.85, 127.52, 128.29, 129.10, 131.32 (dm, J = 245.1 Hz), 137.52, 144.23 (dm, J = 240.8 Hz). ¹⁹**F-NMR** (376.50 MHz, CDCl₃), δ: -163.30 (s, 2F), -93.75 (s, 2F) ppm.

4-(diethylamino)-2,3,5,6-tetrafluorobenzonitrile (15):



¹**H-NMR** (400 MHz, CDCl₃), δ: 1.18 (t, J = 7.1 Hz, 6H), 3.38 (q, J = 7.0 Hz, 4H) ppm. ¹³**C-NMR** (100 MHz, CDCl₃), δ: 13.84 (s), 46.94 (t, J = 4.3 Hz), 108.91 (s), 135.46 (s), 140.89 (dm, J = 242.1 Hz), 146.11 (s), 148.36 (dm, J = 259.2 Hz) ppm. ¹⁹**F-NMR** (376.5 MHz, CDCl₃), δ: -135.21 (s, 2F), -149.92 (s, 2F) ppm. NMR data is consistent with the literature.^[12]

4-(benzylamino)-2,3,5,6-tetrafluorobenzonitrile (15a):



¹H-NMR (400 MHz, CDCl₃), δ: 4.68 (d, J = 6.1, 2H), 4.72 (bs, 1H), 7.31-7.41 (m, 5H) ppm. ¹³C-NMR (100 MHz, CDCl₃), δ: 49.32, 108.82, 127.53, 128.34, 129.13, 133.00, 136.03 (dm, J = 247.5 Hz), 137.47, 148.04 (dm, J = 258.39 Hz). ¹⁹F-NMR (376.5 MHz, CDCl₃), δ: -158.95 (s, 2F), -135.04 (s, 2F) ppm.

Methyl 2-(diethylamino)-3,4,5,6-tetrafluorobenzoate (16):



¹**H-NMR** (400 MHz, CDCl₃), δ: 0.98 (td, J = 7.2, 0.5 Hz, 6H), 3.05 (qd, J = 7.2, 1.3 Hz, 4H), 3.90 (s, 3H) ppm. ¹⁹**F-NMR** (376.5 MHz, CDCl₃), δ: -159.32 (t, J = 20.2 Hz, 1F), -154.37 (t, J = 18.6 Hz, 1F), -144.84 (s, 1F), -142.54 (s, 1F) ppm.

Methyl 4-(diethylamino)-2,3,5,6-tetrafluorobenzoate (17):



¹H-NMR (400 MHz, CDCl₃), δ: 1.12 (t, J = 7.1 Hz, 6H), 3.31 (qt, J = 7.1, 1.1 Hz, 4H), 3.92 (s, 3H) ppm. ¹³C-NMR (100 MHz, CDCl₃), δ: 13.54, 46.78 (p), 48.08 (o), 52.76 (p), 52.99 (o), 120.16, 132.75, 136.94-148.41 (m), 163. 33 ppm. ¹⁹F-NMR (376.5 MHz, CDCl₃), δ: -150.67 (d, J = 12.8 Hz, 2F), -140.94 (d, J = 12.7

Hz, 2F) ppm.



N,N-diethyl-2,3,5,6-tetrafluoro-4-(trifluoromethyl)aniline (18):

¹**H-NMR** (400 MHz, CDCl₃), δ: 1.14 (t, J = 7.1 Hz, 6H), 3.33 (q, J = 7.1 Hz, 4H) ppm. ¹³**C-NMR** (100 MHz, CDCl₃), δ: 13.63, 46.73, 120.24, 122.93, 133.26, 142.29 (d, J = 241.4 Hz), 145.11 (d, J = 269.2 Hz) ppm; ¹⁹**F-NMR** (376.5 MHz, CDCl₃), δ : -55.42 (t, *J* = 19.5 Hz, 3F), -143.26 (t, *J* = 17.3 Hz, 2F), -149.96 (s, 2F) ppm. NMR data is consistent with the literature.^[13]

N,N-diethyl-2,2',3,3',4',5,5',6,6'-nonafluoro-[1,1'-biphenyl]-4-amine (19):



¹H-NMR (400 MHz, CDCl₃), δ : 1.15 (t, J = 7.1 Hz, 6H), 3.34 (q, J= 7.1 Hz, 4H) ppm. ¹³C-NMR (100 MHz, CDCl₃), δ : 13.15 (s), 47.30 (s), 98.31 (m), 103.52 (m), 132.16 (m), 138.71 (dm, J = 248.6 Hz), 142.90 (dm, J = 257.9 Hz), 144.00 (dm, J = 246.0 Hz),

145.45 (d, *J* = 248.2 Hz) ppm. ¹⁹**F-NMR** (376.5 MHz, CDCl₃), δ: -164.23 (s, 2F), -154.53 (s, 1F), -151.23 (s, 2F), -143.16 (s, 2F), -140.46 (s, 2F) ppm.

4-(bis(perfluorophenyl)phosphanyl)-N,N-diethyl-2,3,5,6-tetrafluoroaniline (20):



¹**H-NMR** (400 MHz, CDCl₃), δ : 1.14 (t, J = 7.1 Hz, 6H), 3.31 (q, J = 7.1, 4H) ppm. ¹³**C-NMR** (100 MHz, CDCl₃), δ : 13.50, 46.53, 97.69 (m), 106.11 (m), 132.96 (m), 137.63 (dm, J = 255.7 Hz), 142.43 (dd, J = 247.9, 15.2 Hz), 147.59 (d, J = 250.37 Hz), 148.47 (dm, J = 245.0 Hz).

¹⁹**F-NMR** (376.5 MHz, CDCl₃), δ: -160.08 (broad t, 4F), -149.53 (d, J = 14.0 Hz, 2F), -149.06 (broad t, 2F), -132. 89 (m, 2F), -130.49 (broad t, 4F). ³¹**P-NMR** (162 MHz, CDCl₃), δ: -75.13 (quin, J = 35.2, 1P). **HRMS** (AP+): Calculated for C₂₂H₁₁NF₁₄P: 586.0406 [MH]⁺; Obs: 586.0408.



Figure S36. ¹H NMR spectrum (400 MHz, CDCl₃) of 14.



Figure S37. ¹³C NMR spectrum (100 MHz, CDCl₃) of 14.



Figure S38. ¹⁹F NMR spectrum (376.5 MHz, CDCl₃) of 14.



Figure S39. ¹H NMR spectrum (400 MHz, CDCl₃) of 14a.



Figure S40. ¹³C NMR spectrum (100 MHz, CDCl₃) of 14a.



Figure S41. ¹⁹F NMR spectrum (376.5 MHz, CDCl₃) of 14a.



Figure S42. ¹H NMR spectrum (400 MHz, CDCI₃) of 15.



Figure S43. ¹³C NMR spectrum (100 MHz, CDCl₃) of 15.



Figure S44. $^{19}\mathsf{F}$ NMR spectrum (376.5 MHz, CDCl₃) of 15.



Figure S45. ¹H NMR spectrum (400 MHz, CDCl₃) of 15a.



Figure S46. ¹³C NMR spectrum (100 MHz, CDCl₃) of 15a.



Figure S47. ¹⁹F NMR spectrum (376.5 MHz, CDCl₃) of 15a.



Figure S48. ¹H NMR spectrum (400 MHz, CDCl₃) of ortho and para isomers of 16 and 17.



Figure S49. ¹³C NMR spectrum (100 MHz, CDCl₃) of ortho and para isomers of 16 and 17.



Figure S50. ¹⁹F NMR spectrum (376.5 MHz, CDCl₃) of ortho and para isomers of 16 and 17.



Figure S51. ¹H NMR spectrum (400 MHz, CDCl₃) of 18.





Figure S53. ¹⁹F NMR spectrum (376.5 MHz, CDCl₃) of 18.



Figure S54. ¹H NMR spectrum (400 MHz, CDCl₃) of 19.



Figure S55. ¹³C NMR spectrum (100 MHz, CDCl₃) of **19**.



Figure S56. ¹⁹F NMR spectrum (375.6 MHz, CDCl₃) of 19.



Figure S57. ¹H NMR spectrum (400 MHz, CDCI₃) of 20.



Figure S58. ¹³C NMR spectrum (100 MHz, CDCl₃) of 20.



Figure S59. ¹⁹F NMR spectrum (376.5 MHz, CDCl₃) of 20.



4. DFT calculations

DFT calculations were performed using Gaussian 09.2 Geometry optimization of all the molecules, intermediates, and the transition state were carried out using the BP86(D3)/def2-SVP or BP86(D3)/def2-TZVP basis sets implemented in the Gaussian 09 software.^[14-16] Thermal energy corrections were extracted from the results of frequency analysis performed at the same level of theory. Frequency analysis of all the molecules and intermediates contained no imaginary frequency showing that these are energy minima. The transition states geometries gave one imaginary frequency at expected reaction coordinates confirming that it is a first-order saddle point.



(BPa	86(D3)/def2-T.	ZVP):	
С	0.05436	0.48748	-0.75896
Р	-1.53610	0.00565	-0.30736
Р	1.56558	-0.03682	-0.11750
С	-2.32844	1.62531	-0.15768
С	-3.66061	1.83177	0.22623
С	-1.47547	2.71303	-0.43306
С	-4.13462	3.13650	0.36600
Н	-4.31825	0.98372	0.42668
С	-1.97484	4.01842	-0.32089
С	-3.29018	4.22429	0.10135
Н	-5.16649	3.30969	0.67469
Н	-1.34148	4.87387	-0.56540
Н	-3.66856	5.24207	0.21101
С	2.14050	1.50471	0.64706
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С	3.20675	1.64240	1.54540
С	1.76133	3.89953	0.68236
С	3.52027	2.90937	2.04087
Н	3.77610	0.76870	1.86970
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Н	3.04645	5.01803	2.00485

С	-2.39359	-0.98916	-1.55835
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С	-1.81795	-1.12092	-2.83048
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Н	-4.09653	-1.48006	-0.29901
С	-2.49563	-1.81859	-3.83311
Н	-0.83842	-0.67783	-3.02126
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Н	-5.30040	-2.69564	-2.09442
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Н	-4.27752	-2.92454	-4.35394
С	-1.75358	-0.87178	1.26664
С	-1.88606	-0.12592	2.45074
С	-1.74919	-2.27541	1.31750
С	-2.04027	-0.78287	3.67205
Н	-1.88187	0.96492	2.41329
С	-1.88822	-2.92436	2.54532
Н	-1.64248	-2.85790	0.40124
С	-2.04436	-2.18087	3.71940
Н	-2.16251	-0.20224	4.58784
Н	-1.87730	-4.01448	2.58389
Н	-2.16707	-2.69320	4.67516
С	2.76228	-0.50038	-1.40014
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Н	1.45405	0.06577	-3.02364
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Н	4.28415	-1.13789	0.01426
С	4.64675	-1.09279	-3.37643
Н	3.13509	-0.46618	-4.79029
Н	5.95405	-1.65706	-1.74588
Н	5.38267	-1.32601	-4.14775
С	1.54600	-1.39438	1.07130
С	1.37800	-1.12954	2.44033
С	1.68165	-2.71798	0.61791
С	1.36387	-2.18621	3.35044
Н	1.25969	-0.10256	2.78889
С	1.65816	-3.76925	1.53577
Н	1.82173	-2.92252	-0.44503
С	1.50653	-3.50314	2.90011
Н	1.23565	-1.98104	4.41391
Н	1.76792	-4.79673	1.18548
Н	1.49857	-4.32603	3.61692

Ρ	0.17141	2.28438	-1.12584	
Sur	m of electronic	and zero-poi	nt Energies=	-2450.857814
Sur	m of electronic	and thermal	Energies=	-2450.823779
Sur	m of electronic	and thermal	Enthalpies=	-2450.822835
Sur	m of electronic	and thermal	Free Energies=	-2450.924188
(BP	86(D3)/def2-S	VP):		
Ρ	1.56208	-0.05822	0.09775	
Ρ	0.23324	2.19157	1.38888	
Р	-1.52787	-0.01238	0.37707	
С	0.08123	0.41638	0.87834	
С	2.09880	1.55197	-0.58320	
С	3.11522	1.76205	-1.53524	
Н	3.67139	0.91099	-1.95963	
С	3.40010	3.07288	-1.95170	
Н	4.17832	3.25071	-2.70965	
С	2.69874	4.16021	-1.39314	
Н	2.92785	5.18519	-1.72431	
С	1.71761	3.95228	-0.40766	
Н	1.20093	4.81589	0.04147	
С	1.39042	2.64156	-0.01038	
С	-1.45656	2.68025	0.81231	
С	-1.95550	3.99821	0.84894	
Н	-1.30229	4.83146	1.15436	
С	-3.29266	4.24855	0.49772	
Н	-3.67320	5.28181	0.50587	
С	-4.15548	3.18834	0.15362	
Н	-5.20613	3.39435	-0.10191	
С	-3.68037	1.86765	0.14216	
Н	-4.35433	1.03904	-0.12568	
С	-2.33061	1.62166	0.46160	
С	1.45476	-1.33735	-1.18604	
С	1.31064	-0.98721	-2.54591	
Н	1.25501	0.06996	-2.84616	
С	1.23219	-2.00011	-3.51375	
Н	1.12330	-1.73020	-4.57497	
С	1.28129	-3.35236	-3.12943	
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Н	1.43996	-4.75828	-1.47204	
С	1.49661	-2.69510	-0.79602	
Н	1.61854	-2.96340	0.26539	
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Н	4.25526	-1.21456	-0.27041		
С	5.06450	-1.48107	1.72979		
Н	6.02912	-1.86821	1.36669		
С	4.82695	-1.36042	3.11269		
Н	5.60979	-1.65184	3.83015		
С	3.59410	-0.87223	3.57993		
Н	3.40989	-0.77963	4.66141		
С	2.59199	-0.50056	2.66723		
Н	1.62204	-0.11492	3.02032		
С	-1.76925	-0.63231	-1.32646		
С	-1.80331	0.30888	-2.38056		
Н	-1.71038	1.38472	-2.16231		
С	-1.96980	-0.13220	-3.70210		
Н	-2.01404	0.60148	-4.52188		
С	-2.08751	-1.50736	-3.97607		
Н	-2.22280	-1.85108	-5.01341		
С	-2.02828	-2.44438	-2.92938		
Н	-2.10451	-3.52086	-3.14575		
С	-1.87138	-2.01258	-1.60235		
Н	-1.83001	-2.74834	-0.78481		
С	-2.34346	-1.21175	1.48400		
С	-1.67809	-1.59851	2.66520		
Н	-0.67144	-1.19815	2.86723		
С	-2.30213	-2.48663	3.55725		
Н	-1.78240	-2.79309	4.47822		
С	-3.58782	-2.98065	3.27453		
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Н	-5.25759	-2.98603	1.87364		
С	-3.63407	-1.71287	1.19497		
Н	-4.14942	-1.43054	0.26304		
Sum	of electronic	and zero-poir	nt Energies=	-2448.982637	
Sum	of electronic	and thermal I	=neraies-	-2448 948673	

Sum of electronic and thermal Energies= Sum of electronic and thermal Enthalpies= Sum of electronic and thermal Free Energies= -2448.982637 -2448.948673 -2448.947729 -2449.048981

F	F (3)		
(BP86	6(D3)/def2-T2	ZVP):	
С	-0.00000	-1.12830	0.00000
С	1.20938	-0.42279	-0.00001
С	-1.20938	-0.42280	0.00000

N

∕F

F,

С	1.13522	0.97271	-0.00002	
С	-1.13522	0.97271	0.00002	
Ν	-0.00000	1.64147	-0.00000	
F	2.37823	-1.07821	0.00000	
F	2.27302	1.68036	0.00001	
F	-2.27302	1.68036	-0.00000	
F	-2.37824	-1.07820	-0.00000	
F	0.00001	-2.46202	0.00000	
Sur	n of electronic	and zero-poi	nt Energies=	-744.736117
Sur	n of electronic	and thermal	Energies=	-744.727514
Sur	n of electronic	and thermal	Enthalpies=	-744.726570
Sur	n of electronic	and thermal	Free Energies=	-744.770121
(BPa	86(D3)/def2-S	<i>VP</i>):		
С	-0.00000	1.13775	-0.00000	
С	1.21731	0.42677	0.00001	
С	-1.21731	0.42679	-0.00001	
С	1.14207	-0.97983	0.00003	
С	-1.14206	-0.97982	-0.00002	
Ν	-0.00001	-1.64385	-0.00000	
F	2.38333	1.07741	-0.00000	
F	2.27753	-1.68235	-0.00001	
F	-2 27755	-1 68231	0 00001	
	2.21100	1.00201	0.00001	

-743.858087 -743.849437 -743.848493 -743.892139



		(TS1)	
(BP8	6(D3)/def2-T2	ZVP):	
С	-0.34637	-3.51076	2.04439
С	-0.83690	-3.76581	3.32705
С	-1.33341	-2.72018	4.11863
С	-1.34884	-1.40891	3.63997
С	-0.82213	-1.13984	2.37134
С	-0.33205	-2.19518	1.56733
Ρ	0.25326	-1.67425	-0.07305
С	-0.86038	-2.28017	-1.35743

С	0.19384	0.04950	0.13719
С	1.87441	-2.40083	-0.39229
С	-1.03354	-3.66337	-1.55334
С	-1.91790	-4.11391	-2.53179
С	-2.62662	-3.19380	-3.31570
С	-2.45173	-1.82164	-3.12439
С	-1.56915	-1.36225	-2.14404
С	2.79047	-2.52477	0.66696
С	4.04560	-3.08211	0.43111
С	4.39111	-3.51444	-0.85486
С	3.48454	-3.38348	-1.91041
С	2.22425	-2.82557	-1.68487
Р	-0.88072	0.43982	1.49295
С	0.02777	1.82532	2.21196
С	-2.93923	0.92275	0.96518
F	-3.18398	1.33903	2.31937
Р	1.39427	1.28116	-0.08735
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С	-3.49678	1.75050	-1.26342
Ν	-4.13905	0.65257	-1.62918
С	-4.21089	-0.30897	-0.71935
С	-3.66588	-0.23805	0.55092
С	-0.33332	2.55017	3.35509
С	0.37286	3.71153	3.67517
С	1.39665	4.18269	2.84039
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С	1.06220	2.29854	1.37465
F	-2.23167	3.09414	0.26451
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F	-3.76128	-1.27688	1.41945
F	-4.85923	-1.43329	-1.07761
С	1.19564	2.29387	-1.56879
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С	3.81397	0.73299	1.18137
С	5.14317	0.30871	1.22083
С	5.77252	-0.13131	0.05231
С	5.07250	-0.15443	-1.15863
С	3.74855	0.28203	-1.21098
С	-0.07378	2.37279	-2.16114
С	-0.28919	3.23381	-3.23714
С	0.76202	4.01914	-3.72144
С	2.02829	3.94873	-3.12829
С	2.25087	3.09105	-2.05009
Н	0.03897	-4.32793	1.43202

Н	-0.83941	-4.78661	3.71159	
Н	-1.72039	-2.93376	5.11612	
Н	-1.77375	-0.60510	4.24292	
Н	-0.47694	-4.38518	-0.95246	
Н	-2.06031	-5.18524	-2.68114	
Н	-3.32538	-3.55189	-4.07331	
Н	-3.02353	-1.10107	-3.70930	
Н	-1.44168	-0.29309	-1.97383	
Н	2.51751	-2.19109	1.66970	
Н	4.75706	-3.17953	1.25200	
Н	5.37297	-3.95628	-1.03297	
Н	3.75478	-3.72172	-2.91188	
Н	1.51103	-2.73287	-2.50540	
Н	-1.16969	2.22251	3.97385	
Н	0.10836	4.27017	4.57431	
Н	1.92123	5.10559	3.09112	
Н	2.51883	3.86224	1.01648	
Н	3.32331	1.07694	2.09350	
Н	5.68922	0.32703	2.16528	
Н	6.81264	-0.45989	0.08482	
Н	5.55961	-0.51043	-2.06727	
Н	3.20959	0.27882	-2.15985	
Н	-0.89195	1.76934	-1.76725	
Н	-1.28173	3.29379	-3.68428	
Н	0.59429	4.69212	-4.56398	
Н	2.84569	4.56409	-3.50739	
Н	3.24057	3.03678	-1.59281	
Cur	m of alactropia	and zoro pai	nt Enorgian-	2105 59/162

-3195.540383 -3195.539439 -3195.662331

(*BP86(D3)/def2-SVP*):

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С	0.22636	0.06199	0.17903
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Ρ	-0.00692	-1.64940	-0.09259
С	1.37133	2.16840	1.46535
С	2.19201	3.25887	1.80254
С	0.24031	1.84529	2.26284
С	1.89287	4.00859	2.95379
Н	3.05636	3.52508	1.17333
С	-0.08419	2.63184	3.38372
С	0.76296	3.69536	3.73682
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Н	-1.00139	2.42674	3.95590
Н	0.52793	4.30444	4.62327

C -0.96292 -1.03133 2.3	J . U !
	9367
C -0.78724 -3.45083 2.0	2532
C -1.51866 -1.24840 3.6	6643
C -1.30094 -3.66043 3.3	1722
H -0.49885 -4.30735 1.3	9546
C -1.66643 -2.56673 4.1	2824
H -1.83833 -0.39715 4.2	8610
H -1.42228 -4.68719 3.6	9480
H -2.07810 -2.74691 5.1	3323
C 3.22471 0.32584 0.0	1035
C 376269 -020433 -11	8434
C 3 91668 0 18014 1 2	3269
C = 4.99981 - 0.86485 - 1.1	5227
H = 3.22018 - 0.09590 - 2.1	3637
C = 5.15978 - 0.03330 - 2.1	5057
H = 3.49060 = 0.58729 = 2.10	6322
C = 5.70150 -0.00008 = 0.00008	6065
$\Box = 5.10130 - 0.99090 0.0$	0000
\square 5.41040 -1.20255 -2.0	0031
$\Box 5.71001 -0.57201 2.1$	3003
$\Box 0.07779 -1.49979 0.0 \\ C 4.50759 0.49507 4.4 \\ C 4.50759 0.49507 0.4 \\ C 4.50759 0.49507 $	1002
C = 1.52750 = 2.10527 = 1.4	9152
C 2.08440 2.83217 -1.9	9110
	0800
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H 3.67275 2.63359 -1.5	4713
C 0.15449 3.34592 -3.1	3950
H -0.62985 1.94799 -1.6	6837
C 1.30453 3.98331 -3.6	3904
H 3.46272 4.22921 -3.4	5992
H -0.83818 3.54696 -3.5	6814
H 1.21928 4.68833 -4.4	8064
C 1.48343 -2.60400 -0.4	9205
C 2.42365 -2.85702 0.5	3251
C 1.69704 -3.08344 -1.8	0195
C 3.57468 -3.60067 0.2	4030
H 2.25164 -2.47273 1.5	5004
C 2.85886 -3.82179 -2.0	8379
H 0.95256 -2.89285 -2.5	9110
C 3.79123 -4.08327 -1.0	6491
H 4.31012 -3.80238 1.0	3356
H 3.02895 -4.20408 -3.1	0200
H 4.69560 -4.67074 -1.2	8736
C -1.26559 -1.96344 -1.3	6148
C -2.13924 -3.06967 -1.2	7936
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н -2.108// -3./4961 -0.4	
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Н	-3.78004	-4.12440	-2.22837
Н	-2.36875	-0.57354	-4.29596
Н	-3.91149	-2.54572	-4.16950
Ρ	-0.81161	0.56356	1.52924
С	-2.73025	1.15363	0.97911
С	-2.61651	2.17383	-0.03265
С	-3.54401	0.04031	0.55435
С	-3.16276	1.98007	-1.29925
С	-4.06880	-0.00507	-0.73731
Ν	-3.89500	0.93447	-1.65797
F	-1.78671	3.21689	0.23379
F	-2.92239	2.91350	-2.24632
F	-4.80411	-1.07484	-1.08406
F	-3.72483	-0.98854	1.41745
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(BP86(D3)/def2-TZVP):				
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Ρ	-0.00813	-1.56546	-0.26116	
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С	0.42801	-2.37208	1.29571	
С	0.40813	-3.74819	1.53599	
С	0.78473	-1.45584	2.30590	
С	0.71019	-4.22007	2.81780	
Н	0.13030	-4.44360	0.74125	
С	1.11278	-1.94323	3.57533	
С	1.05109	-3.31877	3.83018	
Н	0.67465	-5.29061	3.02561	
Н	1.41772	-1.25337	4.35967	
Н	1.28157	-3.68803	4.83084	
С	-1.30503	2.12814	1.53705	
С	-0.23705	1.75404	2.37527	
С	-2.16830	3.17304	1.87889	
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С	-1.96038	3.87215	3.07115	
Н	-3.00623	3.42892	1.22721	

С	-0.90002	3.51577	3.90802
Н	0.78308	2.20203	4.22715
Н	-2.62985	4.68807	3.34767
Н	-0.73786	4.05677	4.84168
С	-1.43602	-2.35239	-1.03016
С	-1.52391	-2.45021	-2.42867
С	-2.49822	-2.79035	-0.22142
С	-2.68404	-2.96405	-3.01345
Н	-0.68537	-2.13711	-3.05427
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Н	-2.42367	-2.71439	0.86391
С	-3.74507	-3.38830	-2.20773
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Н	-4.47631	-3.64014	-0.18554
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С	1.39429	-1.67514	-1.40631
C	2.27219	-2.76962	-1.38580
C	1.62435	-0.60101	-2.28351
Č	3.37765	-2.78551	-2.23863
Ĥ	2.11639	-3.59006	-0.68372
C	2.72488	-0.63026	-3.14088
Ĥ	0.96454	0.26902	-2.26498
C	3.60408	-1.71845	-3.11377
Ĥ	4.07455	-3.62407	-2.20639
н	2,91540	0.21194	-3.80747
н	4 48013	-1 72508	-3 76388
C	-3.20032	0.41872	0.15212
č	-3.57421	-0.17319	1.37342
č	-4 10819	0 44942	-0.91522
č	-4 84292	-0 72922	1 52041
Ĥ	-2 86643	-0 19658	2 20568
C	-5 37270	-0 13013	-0 76764
н	-3.83930	0.92660	-1.85771
C	-5.74174	-0.71437	0.44600
Ĥ	-5.13334	-1.17532	2.47308
н	-6.07348	-0.11404	-1.60373
н	-6.73414	-1.15399	0.56007
C	-1 40711	2 08048	-1 46097
č	-1.22918	3.47150	-1.45861
Ĉ	-1.42610	1.38007	-2.68218
č	-1.07487	4,15451	-2.66811
Ĥ	-1.18398	4.02097	-0.51784
C	-1 27579	2 06910	-3 88575
н	-1 55734	0 29467	-2 69312
C	-1 09949	3 45784	-3 87927
н	-0.92790	5 23557	-2 66126
н	-1 29295	1 52192	-4 82978
н	-0.97638	3,99573	-4.82060
P	0.85178	0.34727	1.82778
C	2,31539	0.77392	0.77039
Č	3.41449	-0.08359	0.64079
-			

С	2.34528	1.93246	-0.01196
С	4.43204	0.23586	-0.26585
С	3.41686	2.15960	-0.87667
Ν	4.43053	1.32726	-1.00144
F	3.49830	-1.22857	1.34466
F	5.46377	-0.60669	-0.40842
F	3.42375	3.26625	-1.63519
F	1.33799	2.83150	0.04925
F	1.86131	0.62702	3.17888

-3195.646734 -3195.603179 -3195.602235 -3195.722850

(BP86(D3)/def2-SVP):

С	0.27883	0.02481	0.36613
Р	-0.00487	1.56255	-0.34536
Р	1.55357	-1.09072	0.13860
С	-0.46049	2.45787	1.17197
С	-0.46011	3.85306	1.33499
С	-0.82160	1.59024	2.23169
С	-0.79405	4.39185	2.59154
Н	-0.17815	4.51352	0.49943
С	-1.18590	2.14224	3.47179
С	-1.14755	3.53837	3.65115
Н	-0.77497	5.48219	2.74276
Н	-1.49890	1.47930	4.28885
Н	-1.40678	3.96374	4.63311
С	1.31014	-2.06497	1.65063
С	0.22790	-1.65545	2.46582
С	2.16207	-3.11369	2.03558
С	-0.00352	-2.33688	3.67651
С	1.92274	-3.78203	3.24952
Н	3.01308	-3.40118	1.39778
С	0.84444	-3.39236	4.06173
Н	-0.84604	-2.03776	4.31223
Н	2.58381	-4.60493	3.56222
Н	0.65758	-3.91366	5.01335
С	1.42324	2.30994	-1.17751
С	1.50770	2.31062	-2.58724
С	2.50046	2.78090	-0.39484
С	2.68551	2.76005	-3.20854
Н	0.65492	1.96747	-3.19457
С	3.66959	3.23003	-1.02515
Н	2.42458	2.78501	0.70255
С	3.76535	3.21224	-2.42928
Н	2.75615	2.76370	-4.30717
Н	4.51316	3.58729	-0.41572
Н	4.68658	3.56254	-2.92018
С	-1.43226	1.56446	-1.48386
С	-2.38205	2.60775	-1.47825

	1 61266	0 42565	2 21520
C	-1.01300	0.43505	-2.31530
	2 26267	2.51010	-2.30043
	-2.20307	0.26202	-0.00204
	-2.74011	0.30292	-3.14072
	-0.69544	-0.39947	-2.27579
	-3.09070	1.40029	-3.14000
	-4.20990	3.31072	-2.20029
	-2.09003	-0.52370	-3.76200
	-4.58974	1.32331	-3.77084
	3.23637	-0.37305	0.23684
	3.53744	0.33520	1.42370
	4.20020	-0.48380	-0.78485
C	4.79657	0.92864	1.58406
Н	2.77512	0.42436	2.21521
С	5.45383	0.13470	-0.62561
Н	3.98321	-1.05059	-1.70232
С	5.75331	0.83594	0.55409
Н	5.03371	1.46953	2.51336
Н	6.20338	0.05751	-1.42819
Н	6.74038	1.30843	0.67719
С	1.47109	-2.11794	-1.36763
С	1.22852	-3.50620	-1.32662
С	1.55326	-1.44909	-2.61223
С	1.07753	-4.22096	-2.52730
Н	1.13358	-4.02772	-0.36262
С	1.40406	-2.17169	-3.80596
Н	1.73283	-0.36150	-2.64914
С	1.16571	-3.55843	-3.76385
Н	0.88230	-5.30387	-2.49390
Н	1.47154	-1.65018	-4.77335
Н	1.04429	-4.12390	-4.70071
Р	-0.85528	-0.25282	1.85369
С	-2.30753	-0.73972	0.80290
С	-3.42005	0.10580	0.63865
С	-2.30003	-1.92883	0.05315
С	-4.42217	-0.25929	-0.28438
С	-3.36555	-2.20045	-0.82297
Ν	-4.38943	-1.38070	-0.97837
F	-3.51516	1.27808	1.28257
F	-5.44784	0.57196	-0.48204
F	-3.34099	-3.32168	-1.54922
F	-1 26960	-2 79187	0 13824
F	-1 87528	-0 46497	3 23702
Sum	of electronic	and zero-poir	nt Energies=
Sum	of electronic	and thermal F	=neraies=
Sum of electronic and thermal Energies=			

Sum of electronic and thermal Energies= Sum of electronic and thermal Enthalpies= Sum of electronic and thermal Free Energies=

-3192.892956 -3192.849531 -3192.848586 -3192.968671

PhSiH₃

(BP8	86(D3)/def2-T2	ZVP):	
С	-2.35380	-0.00017	

0.01082

С	-1.65231	-1.21000	0.00301
С	-0.25447	-1.20780	-0.01032
С	0.46927	0.00022	-0.01404
С	-0.25469	1.20789	-0.00992
С	-1.65276	1.20968	0.00342
Si	2.34829	0.00012	0.00597
Н	2.87555	-0.05597	1.40543
Н	2.86376	-1.19346	-0.73161
Н	2.85871	1.24985	-0.63559
Н	-3.44571	-0.00038	0.01845
Н	-2.19545	-2.15746	0.00353
Н	0.27793	-2.16296	-0.02334
Н	0.27783	2.16290	-0.02248
Н	-2.19611	2.15702	0.00424

-522.949514 -522.942249 -522.941305 -522.982128



(TS2)	
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(BP86(D3)/def2-TZVP):				
С	-0.62250	-0.36921	-0.03764	
Ρ	-1.01818	0.21544	-1.59946	
Ρ	-1.55378	-1.22711	1.11560	
С	0.63874	0.18786	-2.29684	
С	0.90811	0.19542	-3.67063	
С	1.66738	0.14748	-1.33136	
С	2.23466	0.14410	-4.10312	
Н	0.08704	0.22360	-4.39041	
С	2.98938	0.12877	-1.78684	
С	3.26486	0.11229	-3.15963	
Н	2.46180	0.12881	-5.17005	
Н	3.81794	0.13345	-1.08514	
Н	4.30516	0.07695	-3.48729	
С	-0.53502	-0.92481	2.56807	
С	0.67432	-0.26083	2.29864	

С	-0.90708	-1.26579	3.87463
С	1.48400	0.12376	3.37047
С	-0.08090	-0.88952	4.93496
Н	-1.84120	-1.80010	4.06169
С	1.09986	-0.18239	4.68003
Н	2.41823	0.65201	3.18386
Н	-0.36147	-1.13832	5.95933
Н	1.73724	0.12589	5.51057
С	-2.19896	-0.82060	-2.48628
С	-3.36632	-0.28797	-3.05546
С	-1.96001	-2.20603	-2.52776
С	-4.30249	-1.14569	-3.63901
Н	-3.55019	0.78663	-3.02961
С	-2.90172	-3.05494	-3.10689
Н	-1.04860	-2.61579	-2.09208
С	-4.07694	-2.52557	-3.65560
Н	-5.21188	-0.73342	-4.07892
Н	-2.71975	-4.13039	-3.12748
Н	-4.81549	-3.19210	-4.10436
С	-1.61381	1.93004	-1.60082
С	-0.83275	2.95219	-2.16544
С	-2.78833	2.25209	-0.89564
С	-1.22576	4.28536	-2.02452
Н	0.09202	2.71076	-2.69171
С	-3.17514	3.58499	-0.76540
Н	-3.38658	1.46842	-0.42853
С	-2.39377	4.60239	-1.32541
Н	-0.60875	5.07854	-2.44890
Н	-4.07869	3.83148	-0.20615
Н	-2.68757	5.64566	-1.20178
С	-1.65686	-3.02036	0.84654
С	-0.55138	-3.66231	0.26318
С	-2.75248	-3.77812	1.29066
С	-0.55105	-5.04759	0.10675
Н	0.30728	-3.07382	-0.06676
С	-2.74742	-5.16629	1.12756
Н	-3.61051	-3.29027	1.75592
С	-1.65175	-5.80027	0.53397
Н	0.31181	-5.54027	-0.34420
Н	-3.60267	-5.75308	1.46652
Н	-1.65214	-6.88446	0.40929
С	-3.24880	-0.61963	1.28619
С	-3.58272	0.33971	2.25495
С	-4.19418	-1.00214	0.31434

С	-4.85295	0.91995	2.24432	
Н	-2.84903	0.64383	3.00150	
С	-5.45757	-0.40969	0.30736	
Н	-3.94101	-1.74912	-0.44052	
С	-5.78628	0.55266	1.26903	
Н	-5.11177	1.66502	2.99806	
Н	-6.18503	-0.70055	-0.45188	
Н	-6.77547	1.01357	1.26151	
С	1.01687	1.98017	0.69446	
С	2.02744	2.85151	0.25894	
С	-0.10265	2.62520	1.24727	
С	1.84882	4.23944	0.34052	
С	-0.17815	4.01807	1.29043	
N	0.77221	4.80856	0.83892	
F	3.19378	2.40614	-0.24127	
F	2.82120	5.04545	-0.10840	
F	-1.27001	4.59824	1.81541	
F	-1.14815	1.93024	1.74816	
F	3.02989	-0.00231	1.05350	
Ρ	1.04524	0.07666	0.47477	
Si	3.17602	-1.91473	0.93462	
Н	1.62178	-1.49133	0.43306	
Н	2.75143	-3.23601	0.30961	
Н	3.04369	-2.09296	2.40188	
С	4.90625	-1.76191	0.23048	
С	5.34635	-2.72555	-0.69634	
С	5.80047	-0.73745	0.60643	
С	6.63573	-2.67074	-1.23510	
Н	4.67068	-3.52801	-1.00375	
С	7.08977	-0.68430	0.07477	
Н	5.48041	0.03039	1.31288	
С	7.50848	-1.64959	-0.84939	
Н	6.95897	-3.42514	-1.95504	
Н	7.77241	0.11117	0.38001	
Н	8.51594	-1.60415	-1.26754	
Sum	n of electronic	and zero-poi	nt Eneraies=	-3718.565465

Sum of electronic and thermal Energies= Sum of electronic and thermal Enthalpies= Sum of electronic and thermal Free Energies=

-3718.514373 -3718.513429 -3718.651499

PhSi(F)H₂:

С	2.69890	-0.00030	0.18163
С	2.00654	-1.21171	0.08297
С	0.62287	-1.20974	-0.11200

С	-0.09270	0.00025	-0.20964
С	0.62325	1.20995	-0.11136
С	2.00693	1.21138	0.08359
Si	-1.94072	0.00059	-0.44847
Н	-2.37207	-1.23451	-1.16446
F	-2.70483	-0.00104	0.98368
Н	-2.37188	1.23734	-1.16171
Н	3.78048	-0.00051	0.33191
Н	2.54649	-2.15796	0.15587
Н	0.09397	-2.16344	-0.19273
Н	0.09464	2.16385	-0.19158
Н	2.54718	2.15742	0.15697

-622.309742 -622.301941 -622.300997 -622.343697



(6+)

(BP86(D3)/def2-TZVP):

С	-0.45038	-3.65521	1.75958
С	-0.83539	-4.05445	3.04393
С	-1.25265	-3.10392	3.98211
С	-1.29700	-1.74851	3.63967
С	-0.88682	-1.34104	2.36859
С	-0.46156	-2.29563	1.43470
Ρ	0.04413	-1.57856	-0.15381
С	-1.22451	-2.02471	-1.37869
С	0.11819	0.06827	0.28914
С	1.57082	-2.38777	-0.68922
С	-1.42951	-3.37428	-1.71937
С	-2.45306	-3.72225	-2.60033
С	-3.27779	-2.73091	-3.14711
С	-3.07168	-1.38872	-2.81874
С	-2.04670	-1.03722	-1.93700
С	2.52232	-2.77014	0.27180
С	3.71235	-3.37167	-0.13447
С	3.96139	-3.58814	-1.49458
С	3.02235	-3.19642	-2.45282

С	1.82576	-2.59565	-2.05473	
Ρ	-0.96363	0.45091	1.87523	
С	0.12791	1.88651	2.30140	
С	-2.45069	0.88678	0.84677	
Ρ	1.38932	1.16538	0.00558	
С	-2.51446	2.06090	0.08558	
С	-3.59689	2.27177	-0.77135	
Ν	-4.59559	1.41933	-0.89744	
С	-4.57583	0.32649	-0.16332	
С	-3.54194	0.01921	0.72918	
С	-0.12424	2.64442	3.44764	
С	0.67407	3.75603	3.73597	
С	1.71646	4.11523	2.87510	
С	1.96989	3.36302	1.72314	
С	1.17602	2.24698	1.44530	
F	-1.52820	2.97991	0.14278	
F	-3.63091	3.38195	-1.52242	
F	-3.59298	-1.13583	1.42152	
F	-5.58924	-0.53546	-0.30779	
С	1.31049	2.16926	-1.50541	
С	3.08891	0.52233	0.07763	
С	3.68818	0.31142	1.33122	
С	4.97734	-0.21759	1.40763	
С	5.66693	-0.55241	0.23745	
С	5.06401	-0.36476	-1.00992	
С	3.77914	0.17422	-1.09446	
С	0.14867	2.14407	-2.28812	
С	0.04344	2.95931	-3.41628	
С	1.09925	3.80680	-3.76453	
С	2.26422	3.83420	-2.98852	
С	2.37497	3.01705	-1.86316	
Н	-0.12129	-4.39864	1.03142	
Н	-0.80911	-5.11167	3.31280	
Н	-1.55103	-3.42073	4.98273	
Н	-1.65398	-1.00607	4.35769	
Н	-0.78528	-4.15267	-1.30607	
Н	-2.61111	-4.77008	-2.86037	
Н	-4.08233	-3.00807	-3.83011	
Н	-3.71470	-0.61258	-3.23685	
Н	-1.89026	0.00639	-1.66570	
Н	2.33097	-2.59731	1.33204	
Н	4.45035	-3.66815	0.61188	
Н	4.89271	-4.06322	-1.80743	
Н	3.21695	-3.36404	-3.51329	

Н	1.08663	-2.30268	-2.80243
Н	-0.94758	2.36581	4.11004
Н	0.48013	4.34592	4.63319
Н	2.33510	4.98559	3.09970
Н	2.78287	3.64783	1.05307
Н	3.15103	0.57306	2.24513
Н	5.44673	-0.36285	2.38203
Н	6.67625	-0.96302	0.29807
Н	5.59470	-0.63978	-1.92249
Н	3.31616	0.32494	-2.07066
Н	-0.67069	1.48502	-2.00258
Н	-0.86470	2.93579	-4.02042
Н	1.01725	4.44633	-4.64495
Н	3.09023	4.49178	-3.26412
Н	3.29507	3.02678	-1.27580
Н	-1.73916	0.71284	3.09522

-3096.295683 -3096.252895 -3096.251951 -3096.372070



(TS3) (BP86(D3)/def2-TZVP): С 0.09677 -3.58641 1.95311 С -0.45621 -3.94533 3.18821 С -1.18698 -3.01476 3.93415 С -1.37140-1.71040 3.46203 С -0.81042 -1.34368 2.23682 С -0.07014 -2.28236 1.48528 Ρ 0.49776 -1.65505 -0.12594 С -0.59234 -2.34508 -1.39222 С 0.26943 0.15689 0.02836 С 2.18395 -2.19511 -0.45059С -0.55029 -3.71930 -1.69289С -1.48995 -4.26248 -2.56806 С -2.47596 -3.44510 -3.13596

С	-2.51581	-2.08015	-2.84120
С	-1.57295	-1.52700	-1.97236
С	3.10389	-2.24656	0.61157
С	4.41967	-2.63332	0.36416
С	4.82175	-2.96442	-0.93530
С	3.91080	-2.90112	-1.99367
С	2.58984	-2.51471	-1.75705
Р	-0.98259	0.26625	1.42001
С	-0.36116	1.82391	2.10014
С	-2.68292	0.47094	0.87782
Р	1.26794	1.40919	-0.05648
С	-3.07346	1.68088	0.19782
С	-4.19515	1.71903	-0.61095
Ν	-4.96442	0.68060	-0.89636
С	-4.58779	-0.47171	-0.35769
С	-3.49615	-0.64808	0.46823
С	-0.90012	2.51498	3.18922
С	-0.36016	3.75326	3.54934
С	0.70802	4.30158	2.83027
С	1.24660	3.61738	1.73565
С	0.72050	2.37281	1.38132
F	-2.32117	2.80918	0.37163
F	-4.51277	2.89092	-1.19900
F	-3.18163	-1.89309	0.92295
F	-5.31194	-1.55943	-0.69182
С	0.94531	2.38358	-1.54381
С	3.04798	1.11740	0.05328
С	3.68846	1.16787	1.30300
С	5.05284	0.88699	1.39372
С	5.77445	0.53962	0.24740
С	5.13268	0.46636	-0.99343
С	3.77239	0.75716	-1.09656
С	-0.33349	2.30447	-2.11887
С	-0.66505	3.12717	-3.19500
С	0.27652	4.03108	-3.69852
С	1.54906	4.11832	-3.12254
С	1.88743	3.30181	-2.04203
Н	0.66007	-4.31640	1.36886
Н	-0.31318	-4.95738	3.56985
Н	-1.61521	-3.30465	4.89473
Н	-1.95127	-0.98300	4.03184
Н	0.21241	-4.36212	-1.24795
Н	-1.46065	-5.32781	-2.80154
Н	-3.21849	-3.87791	-3.80820

Н	-3.29057	-1.44481	-3.27234
Н	-1.61546	-0.46657	-1.72270
Н	2.78818	-1.98601	1.62339
Н	5.13474	-2.67508	1.18662
Н	5.85173	-3.27276	-1.12228
Н	4.22594	-3.15869	-3.00594
Н	1.87420	-2.47327	-2.57996
Н	-1.74374	2.08823	3.73473
Н	-0.77868	4.29552	4.39842
Н	1.12212	5.26830	3.12055
Н	2.06760	4.05274	1.16228
Н	3.12464	1.43654	2.19815
Н	5.55396	0.94381	2.36131
Н	6.84131	0.32228	0.32111
Н	5.69291	0.18151	-1.88487
Н	3.27605	0.70775	-2.06724
Н	-1.07188	1.61210	-1.71120
Н	-1.66214	3.06799	-3.63336
Н	0.01687	4.67390	-4.54125
Н	2.28053	4.82632	-3.51528
Н	2.88015	3.37624	-1.59420
Н	-2.63308	0.56767	2.40874

-3096.250410 -3096.207521 -3096.206577 -3096.327609



(BP86(D3)/def2-TZVP):

		-		
Ν	0.00000	0.00000	1.37068	
С	0.00000	1.13515	0.70193	
С	0.00000	1.19307	-0.69704	
С	0.00000	0.00000	-1.41406	
С	0.00000	-1.19307	-0.69704	
С	0.00000	-1.13515	0.70193	
F	0.00000	-2.37737	-1.33673	
F	0.00000	-2.27434	1.41088	
F	0.00000	2.27434	1.41088	
F	0.00000	2.37737	-1.33673	
Н	0.00000	0.00000	-2.50360	
Sum of electronic and zero-point Energies=				

-645.457844

Sum of electronic and thermal Energies=	-645.450196
Sum of electronic and thermal Enthalpies=	-645.449252
Sum of electronic and thermal Free Energies=	-645.489935

Me₃SiNMe₂

(BP86(D3)/def2-SVP):

Si	-0.61666	0.00001	-0.00456
С	-1.33066	1.54702	-0.82916
Н	-2.43974	1.48580	-0.83568
Н	-1.05679	2.47969	-0.29308
Н	-0.98453	1.63538	-1.88013
С	-1.10082	-0.00016	1.83261
Н	-2.20390	-0.00018	1.96845
Н	-0.69582	-0.89713	2.34804
Н	-0.69583	0.89670	2.34822
С	-1.33070	-1.54683	-0.82944
Н	-0.98458	-1.63501	-1.88043
Н	-1.05686	-2.47961	-0.29353
Н	-2.43979	-1.48559	-0.83595
Ν	1.13775	0.00000	-0.23648
С	1.91360	-1.20980	-0.01589
Н	1.31698	-2.11531	-0.24865
Н	2.81258	-1.23210	-0.67821
Н	2.28856	-1.31510	1.03675
С	1.91365	1.20976	-0.01587
Н	2.81264	1.23204	-0.67819
Н	1.31707	2.11530	-0.24861
Н	2.28861	1.31503	1.03678

Sum of electronic and zero-point Energies=-543.Sum of electronic and thermal Energies=-543.4Sum of electronic and thermal Enthalpies=-543.4Sum of electronic and thermal Enthalpies=-543.4Sum of electronic and thermal Free Energies=-543.4

-543.419276 -543.407108 -543.406164 -543.457072



(*BP86(D3)/def2-SVP*): C 0.29110 0.12284 -0.17980

Р	0.19254	-1.51734	-0.64634
Р	1.45666	0.95990	0.73251
С	-0.23916	-1.24340	-2.40473
С	-0.00945	-2.16308	-3.44088
С	-0.65094	0.08511	-2.68829
С	-0.13806	-1.74264	-4.77894
Н	0.31788	-3.18953	-3.21082
С	-0.75528	0.49889	-4.02701
С	-0.48426	-0.41230	-5.06705
Н	0.06074	-2.45200	-5.59705
Н	-1.04160	1.53630	-4.25512
Н	-0.54669	-0.07470	-6.11312
С	1.22425	2.60987	-0.00365
С	0.26057	2.67364	-1.03631
С	2.12685	3.66527	0.21993
С	0.27257	3.78818	-1.89936
С	2.09476	4.79240	-0.61849
Н	2.88406	3.58602	1.01625
С	1.18984	4.83450	-1.69341
Н	-0.45124	3.82943	-2.72537
Н	2.79951	5.62164	-0.45261
Н	1.19274	5.69358	-2.38215
С	1.74716	-2.45120	-0.50446
С	1.95223	-3.29851	0.60732
С	2.79377	-2.21449	-1.42250
С	3.20944	-3.89448	0.80315
Н	1.12852	-3.49990	1.31073
С	4.04053	-2.82445	-1.22434
Н	2.63452	-1.54822	-2.28364
С	4.25167	-3.65664	-0.10969
Н	3.37166	-4.55477	1.66904
Н	4.85745	-2.63682	-1.93713
Н	5.23526	-4.12655	0.04520
С	-1.09363	-2.48884	0.21349
С	-1.79430	-3.52777	-0.43269
С	-1.39476	-2.14615	1.54981
С	-2.80321	-4.21581	0.26008
Н	-1.58026	-3.77742	-1.48361
С	-2.39484	-2.84897	2.23981
Н	-0.88098	-1.29572	2.02592
С	-3.10226	-3.87804	1.59266
Н	-3.37018	-5.01124	-0.24741
Н	-2.64746	-2.56789	3.27339
Н	-3.90663	-4.40860	2.12482

С	3.20994	0.54909	0.36926
С	3.66140	0.82842	-0.94248
С	4.08841	-0.02620	1.30858
С	4.98658	0.54926	-1.30166
Н	2.96923	1.26868	-1.67866
С	5.41002	-0.32838	0.93135
Н	3.75519	-0.23136	2.33655
С	5.86169	-0.03437	-0.36539
Н	5.33957	0.78368	-2.31800
Н	6.09228	-0.78521	1.66467
Н	6.90172	-0.25828	-0.64967
С	1.24702	0.93433	2.54494
С	0.95255	2.09784	3.28429
С	1.28759	-0.32471	3.18959
С	0.70346	2.00004	4.66399
Н	0.89209	3.07524	2.78207
С	1.04437	-0.41245	4.56893
Н	1.50251	-1.23915	2.61064
С	0.75000	0.75028	5.30590
Н	0.46366	2.90736	5.23950
Н	1.07806	-1.39216	5.07033
Н	0.55044	0.67964	6.38636
Р	-0.94644	1.22869	-1.23850
С	-2.28476	0.67506	-0.03280
С	-3.24399	-0.30600	-0.32965
С	-2.38329	1.29210	1.22391
С	-4.20764	-0.63738	0.64292
С	-3.38117	0.88248	2.12321
Ν	-4.26385	-0.05810	1.82771
F	-3.21449	-0.98145	-1.49788
F	-5.09453	-1.59977	0.37322
F	-3.44928	1.45019	3.33105
F	-1.49776	2.25541	1.56915
Ν	-2.18310	2.31037	-2.06932
С	-3.33029	1.71683	-2.76360
Н	-4.17451	1.47750	-2.06797
Н	-3.72377	2.44063	-3.51152
Н	-3.07281	0.78328	-3.29283
С	-2.67785	3.42838	-1.25172
Н	-3.02382	4.25685	-1.90855
Н	-3.54529	3.12810	-0.61162
Н	-1.90187	3.83624	-0.57746
Sum	of algotrapia	and zoro poir	ot Enorgiao-

Sum of electronic and zero-point Energies= Sum of electronic and thermal Energies= -3227.480441 -3227.433654 Sum of electronic and thermal Enthalpies= Sum of electronic and thermal Free Energies=

-3227.432710 -3227.559330



(TS3-N)

(BP86(D3)/def2-SVP)
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Ċ	0.40697	-0.05986	0.23320
P	0.81489	1.61260	0.26445
Р	1.12534	-1.47751	-0.43008
С	0.27908	1.95178	1.98736
С	0.66140	3.06358	2.75245
С	-0.52393	0.92657	2.55906
С	0.26911	3.13838	4.10350
Н	1.28932	3.85500	2.31368
С	-0.91048	1.00720	3.90501
С	-0.49893	2.10989	4.67745
Н	0.57852	4.00057	4.71395
Н	-1.52486	0.20850	4.34553
Н	-0.78412	2.16577	5.73924
С	0.53568	-2.67099	0.82400
С	-0.36295	-2.13040	1.78060
С	1.02263	-3.98096	0.96204
С	-0.73155	-2.89160	2.90049
С	0.63563	-4.74420	2.08000
Н	1.72741	-4.39427	0.22321
С	-0.21767	-4.19263	3.05294
Н	-1.41830	-2.46429	3.64543
Н	1.01906	-5.76895	2.20039
Н	-0.49252	-4.78418	3.93977
С	2.58397	1.95782	0.05158
С	3.06098	2.46343	-1.17796
С	3.48591	1.59487	1.07571
С	4.44413	2.60580	-1.37692
Н	2.35311	2.75082	-1.97121
С	4.86375	1.74934	0.86894
Н	3.10671	1.18722	2.02596
С	5.34261	2.25056	-0.35600
Н	4.82032	3.00080	-2.33315
Н	5.57004	1.46777	1.66451
Н	6.42614	2.36569	-0.51416
С	-0.09684	2.64344	-0.92397
С	-0.01238	4.05297	-0.87087

С	-0.89379	2.00835	-1.89924
С	-0.73715	4.82198	-1.79313
Н	0.61449	4.55129	-0.11389
С	-1.61413	2.78801	-2.81987
Н	-0.97562	0.91053	-1.91459
С	-1.53846	4.18924	-2.76411
Н	-0.68420	5.92076	-1.75108
Н	-2.26226	2.29049	-3.55667
Н	-2.11647	4.79760	-3.47683
С	2.94940	-1.50488	-0.38420
С	3.57324	-1.69236	0.87153
С	3.73087	-1.30029	-1.54033
С	4.97198	-1.69447	0.96099
Н	2.96250	-1.84461	1.77547
С	5.13241	-1.27677	-1.43496
Ĥ	3.25369	-1.17505	-2.52396
С	5.75221	-1.48083	-0.19059
Ĥ	5.45702	-1.85919	1.93561
H	5.74198	-1.11127	-2.33642
н	6.85082	-1.47703	-0.11703
C	0.53552	-1.88879	-2.10141
č	-0.14848	-3.09098	-2.37473
č	0.63332	-0.88275	-3 09198
č	-0 74488	-3 27667	-3 63270
н	-0 25803	-3 86016	-1 59657
C	0.04890	-1 08532	-4 35055
й	1 13766	0.07296	-2 87155
C	-0 64897	-2 27849	-4 61666
н	-1 30511	-4 20148	-3 83705
н	0 11031	-0 30047	-5 11026
н	-1 13108	-2 12/00	-5 595/0
D	-0.04307	-0 13213	-0.090 + 0
Ċ	-0.94397	-0.45215	0.46058
C C	-2.02000	-0.20 4 07	0.40000
Ĉ	-2 77560	-1 22627	-0 58238
C	-2.77300	1 22027	-1.04674
C	-3.77200	-0 80023	-1.04074
	-3.32043	-0.09923	1 0/727
	-4.04040	2 11820	-1.947.57
	-2.02313	2.11039	-1 30886
	-4.21055	-1 8/556	-2.65152
	-3.70233	2 47562	-2.05152
	-2.22402	-2.47505	-0.40401
	-2.90170	-0.00173	2.23031
	-3.02212	0.22212	2.00000
	-4.70732	0.30440	2.27344
	-4.10390	-0.00423	3.09010
	-3.30/99	1.19930	∠.ŏŏUŏ/
	-3.0/062	-2.084/5	2.1//8/
	-3.95439	-2.43/45	3.19821
	-4.01//8	-2.00257	1.58340
н	-3.05275	-2.86105	1.69802

Sum of electronic and zero-point Energies= -3227.454835 Sum of electronic and thermal Energies= -3227.408297 Sum of electronic and thermal Enthalpies= -3227.407353 Sum of electronic and thermal Free Energies= -3227.534104

F	N F
Ĭ	
F	F

NMe₂ (**7-N**)

(BP86	(D3)/def2·	-SVP):
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С	0.61464	0.00003	0.00010
С	-0.15922	-1.19896	-0.05933
С	-0.15942	1.19901	0.05906
С	-1.55913	-1.13036	-0.03887
С	-1.55923	1.13032	0.03878
Ν	-2.24525	-0.00010	0.00002
F	0.44914	-2.39994	-0.17324
F	-2.25375	-2.27595	-0.09556
F	-2.25408	2.27575	0.09579
F	0.44903	2.39996	0.17285
Ν	1.99124	0.00028	0.00044
С	2.77188	1.08619	-0.58380
Н	3.19693	1.76772	0.18761
Н	3.61435	0.65168	-1.16531
Н	2.15713	1.68668	-1.27717
С	2.77185	-1.08613	0.58386
Н	3.19610	-1.76761	-0.18805
Н	3.61487	-0.65207	1.16488
Н	2.15733	-1.68656	1.27748

Sum of electronic and zero-point Energies= Sum of electronic and thermal Energies= -778.497431 Sum of electronic and thermal Enthalpies= Sum of electronic and thermal Free Energies=

Me₃SiF

(BP86(D3)/def2-SVP):

Si	0.00001	0.00033	0.02383
С	1.44911	1.06192	-0.53161
Н	1.50352	1.10548	-1.64045
Н	1.35043	2.10116	-0.15396
Н	2.41073	0.65104	-0.15895
С	-1.64395	0.72305	-0.53336
Н	-1.70537	0.75469	-1.64224
Н	-2.49380	0.11283	-0.16215
Н	-1.77392	1.75861	-0.15501
С	0.19612	-1.78523	-0.53226
Н	1.14750	-2.21752	-0.15750
Н	-0.63809	-2.41411	-0.15676

-778.509758

-778.496487

-778.548349



Sum of electronic and thermal Enthalpies= Sum of electronic and thermal Free Energies=





Figure S61. DFT calculated (BP86-D3/def2SVP)^[15,16] potential energy surface of the proposed mechanism of 1⁺ catalyzed C-N bond cross-coupling reaction. Free Gibbs energies (enthalpies) are given relative to the starting materials.



Figure S62. Frontier orbitals of 1+.

5. References

- (1) S. C. Böttger, C. Poggel C. and J. Sundermeyer, Organometallics 39, 3789-3793 (2020).
- (2) Bruker (2012). Apex-II. Bruker AXS Inc., Madison, Wisconsin, USA.
- (3) Bruker (2013). SAINT v8.34A. Bruker AXS Inc., Madison, Wisconsin, USA.
- (4) Bruker (2014/5). Sadabs, 2014/5. Bruker AXS Inc., Madison, Wisconsin, USA.
- (5) O. V. Dolomanov, L. J. Bourhis, R. J. Gildea, J. A. K. Howard, H. J. Puschmann. Appl. Crystallogr. 2009, 42, 339-341.
- (6) G. M. Sheldrick, Acta Cryst. A, 2015, 71, 3-8.
- (7) S. C. Böttger, C. Poggel C. and J. Sundermeyer. Organometallics 39, 3789-3793 (2020).
- (8) A. V. Gutov, E. B. Rusanov, A. B. Ryabitskii & A. N. Chernega. Octafluoro-4,4'-bipyridine and its derivatives: Synthesis, molecular and crystal structure. *J. Fluor. Chem.* **131**, 278–281 (2010).
- (9) Y. Pang, M. Leutzsch, N. Nöthling, F. Katzenburg & Josep Cornella. Catalytic Hydrodefluorination via Oxidative Addition, Ligand Metathesis, and Reductive Elimination at Bi(I)/Bi(III) Centers. J. Am. Chem. Soc. 2021, 143, 12487–12493
- (10) H. Yoshida, T. Minabe, J. Ohshita & A. Kunai. Aminosilylation of arynes with aminosilanes: synthesis of 2silylaniline derivatives. *Chem. Commun.* 3454 (2005).
- (11) R. D. Chambers, G. Sandford & J. Trmcic. Continuous flow glassware reactors for the laboratory. *J. Fluor. Chem.* **128**, 1439–1443 (2007).
- (12) N. Matsuda, K. Hirano, T. Satoh, & M. Miura. Copper-Catalyzed Direct Amination of Electron-Deficient Arenes with Hydroxylamines. *Org. Lett.* **13**, 2860–2863 (2011).

- (13) L. J. Bole, L. Davin, A. R. Kennedy, R. McLellan & E. Hevia. Magnesium-mediated arylation of amines via C–F bond activation of fluoroarenes. *Chem. Commun.* **55**, 4339–4342 (2019).
- (14) Gaussian 09, Revision D.01, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnengerg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery Jr, J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, T. Keith, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, N. J. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, Ö. Farkas, J. B. Foresman, J. V. Ortiz, J. Cioslowski and D. J. Fox, Gaussian, Inc., Wallingford CT, 2010.
- (15) a) A. D. Becke, Phys. Rev. A: At., Mol. Opt. Phys. 1988, 38, 3098-3100. b) J. P. Perdew, Phys. Rev. B: Condens. Matter Mater. Phys. 1986, 34, 7406.
- (16) F. Weigend, R. Ahlrichs, Phys. Chem. Chem. Phys. 2005, 7, 3297-3305.