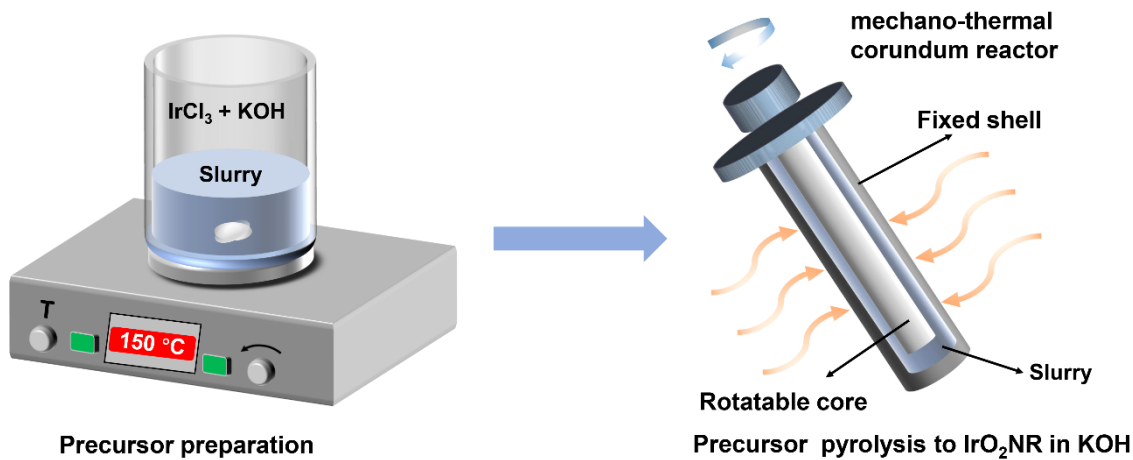
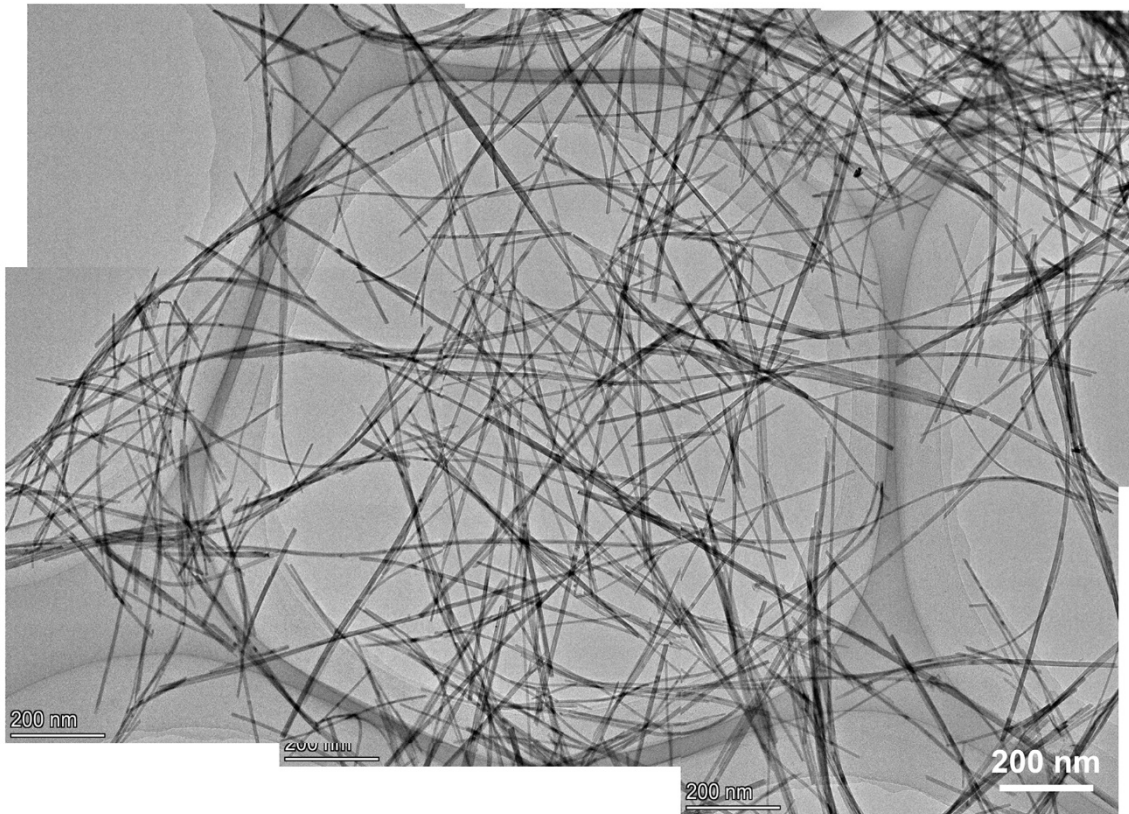


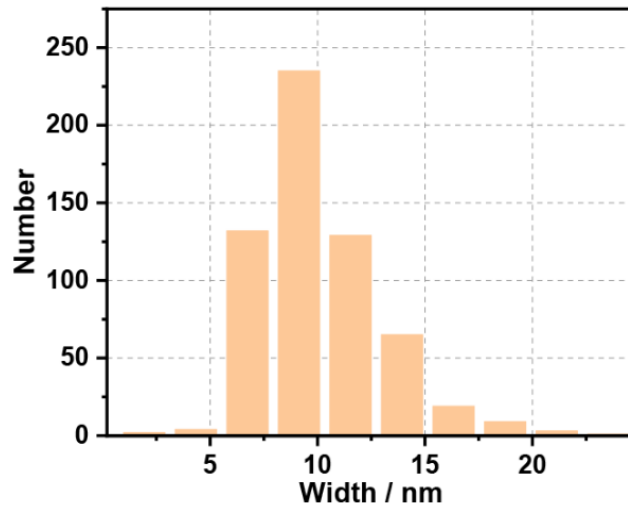
1 Supplementary Figures



3 **Supplementary Figure 1. The schematic molten-alkali mechanochemical reaction for**
4 **synthesizing IrO_2NR .** The slurry of IrCl_3 and KOH aqueous solution was transformed into a homemade
5 mechano-thermal corundum reactor that was fixed in a muffle furnace. The reactor was heated to different
6 temperatures to obtain samples.

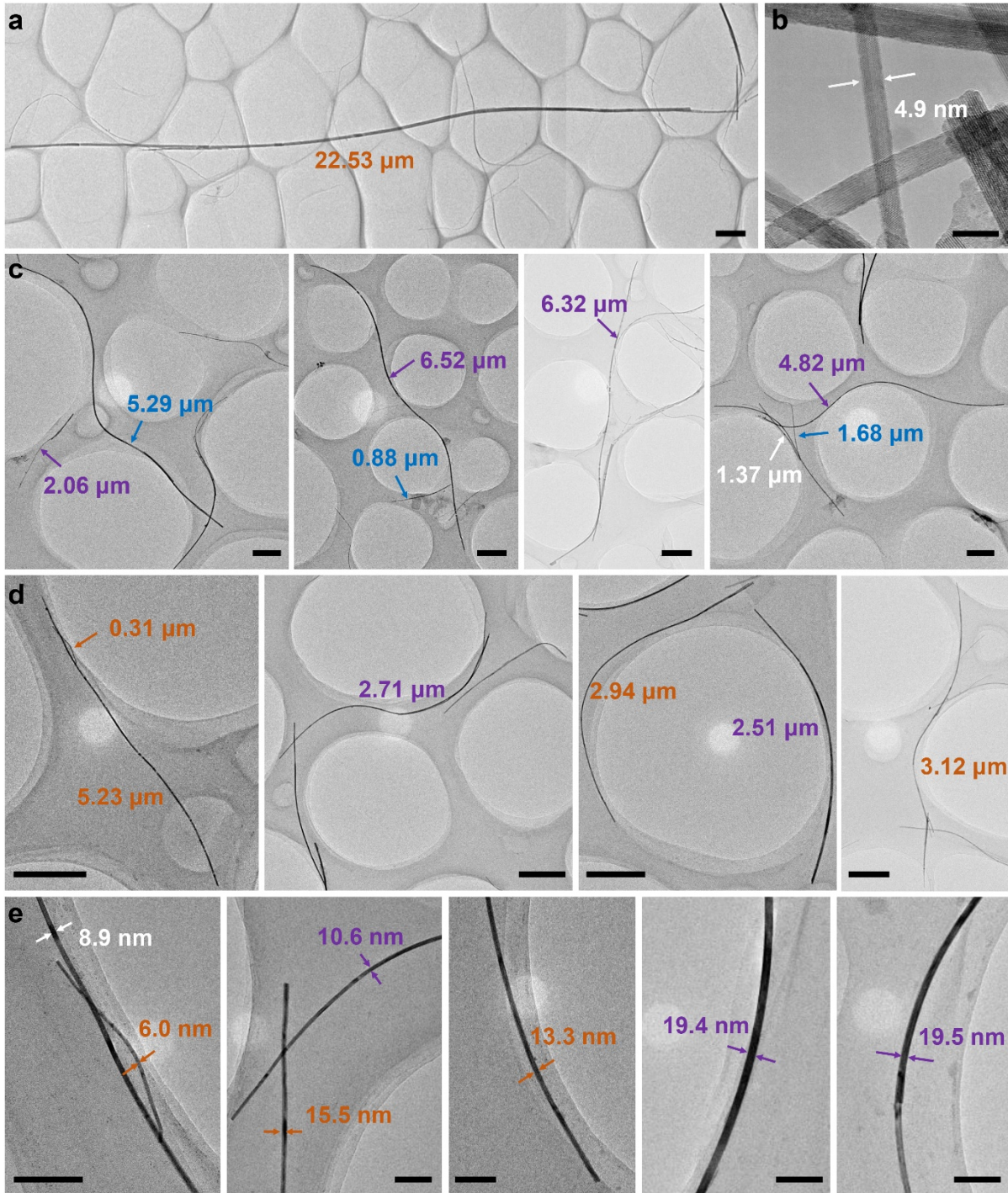


1
2 **Supplementary Figure 2. TEM image of IrO₂NR.** TEM images show nanoribbon morphology at a
3 high magnification.



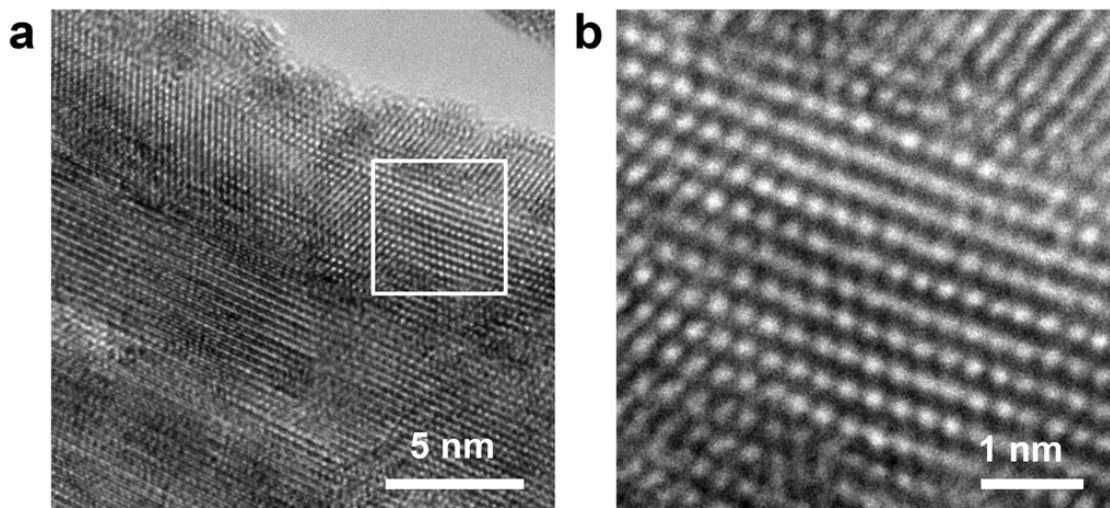
1

2 **Supplementary Figure 3. Statistics histogram showing the width of IrO₂NR.** The width values
3 are obtained by 500 nanoribbons.

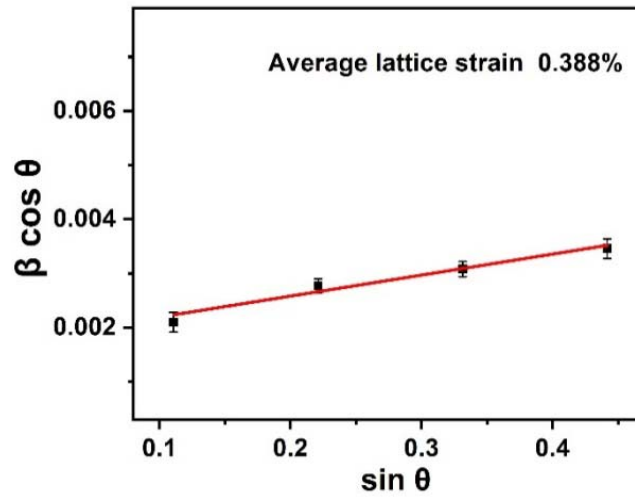


1

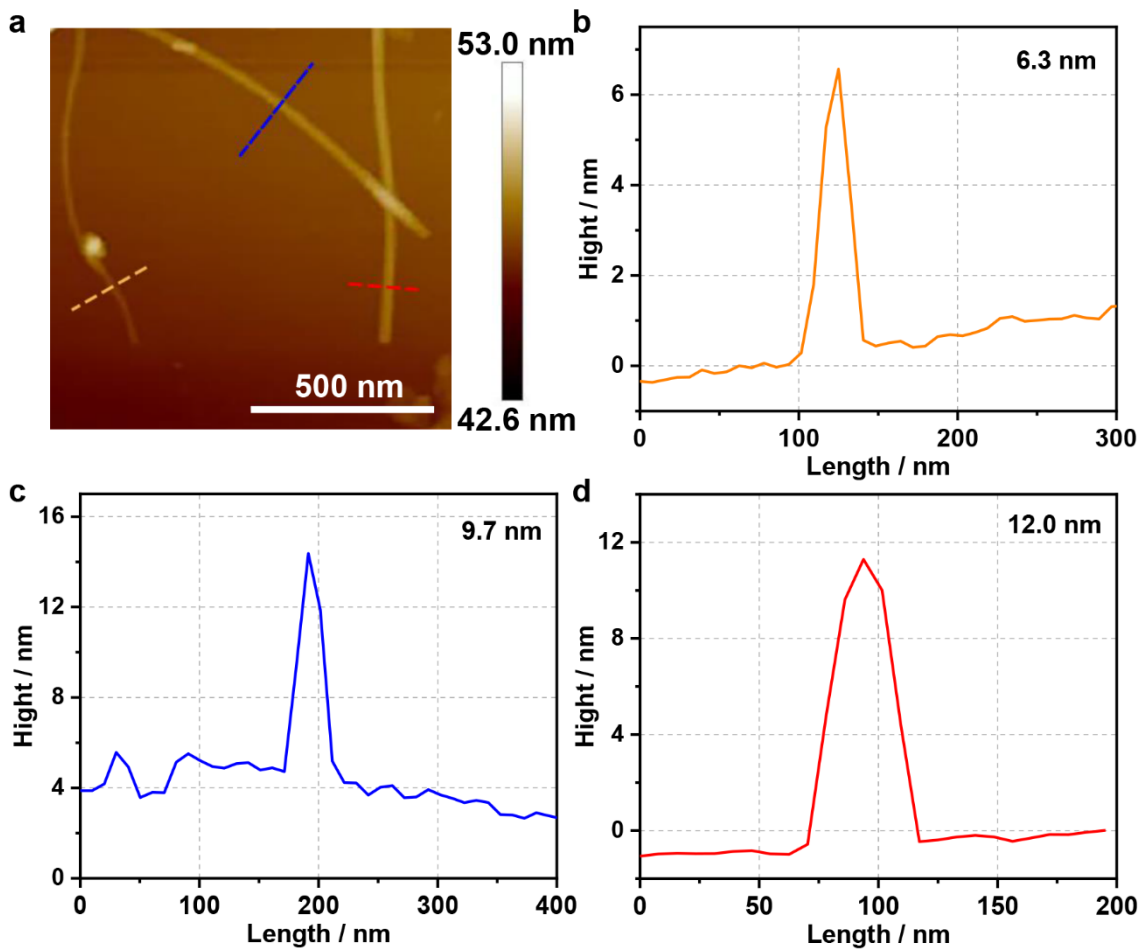
2 **Supplementary Figure 4. Gallery of IrO₂NR.** (a) Typical image of a long IrO₂NR. Figure bar, 1
 3 μm; insert bar, 100 nm. (b) Typical image of a thin IrO₂NR. Figure bar, 10 nm. (c) and (d) IrO₂NR of
 4 different lengths. Figure bar, 500 nm. (e) Micrographs of IrO₂NR with different widths. Figure bar,
 5 100 nm.



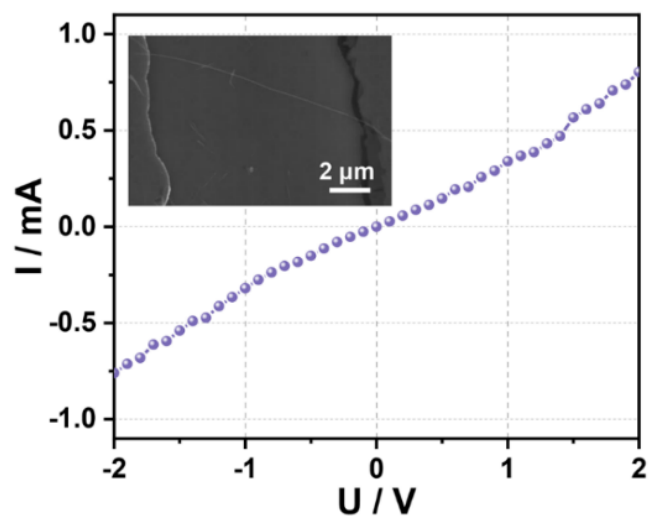
1
2 **Supplementary Figure 5. Different magnification HETEM images. (a)** Low magnification and **(b)**
3 high magnification HRTEM images of IrO₂NR showing the lattice tension/compression.
4



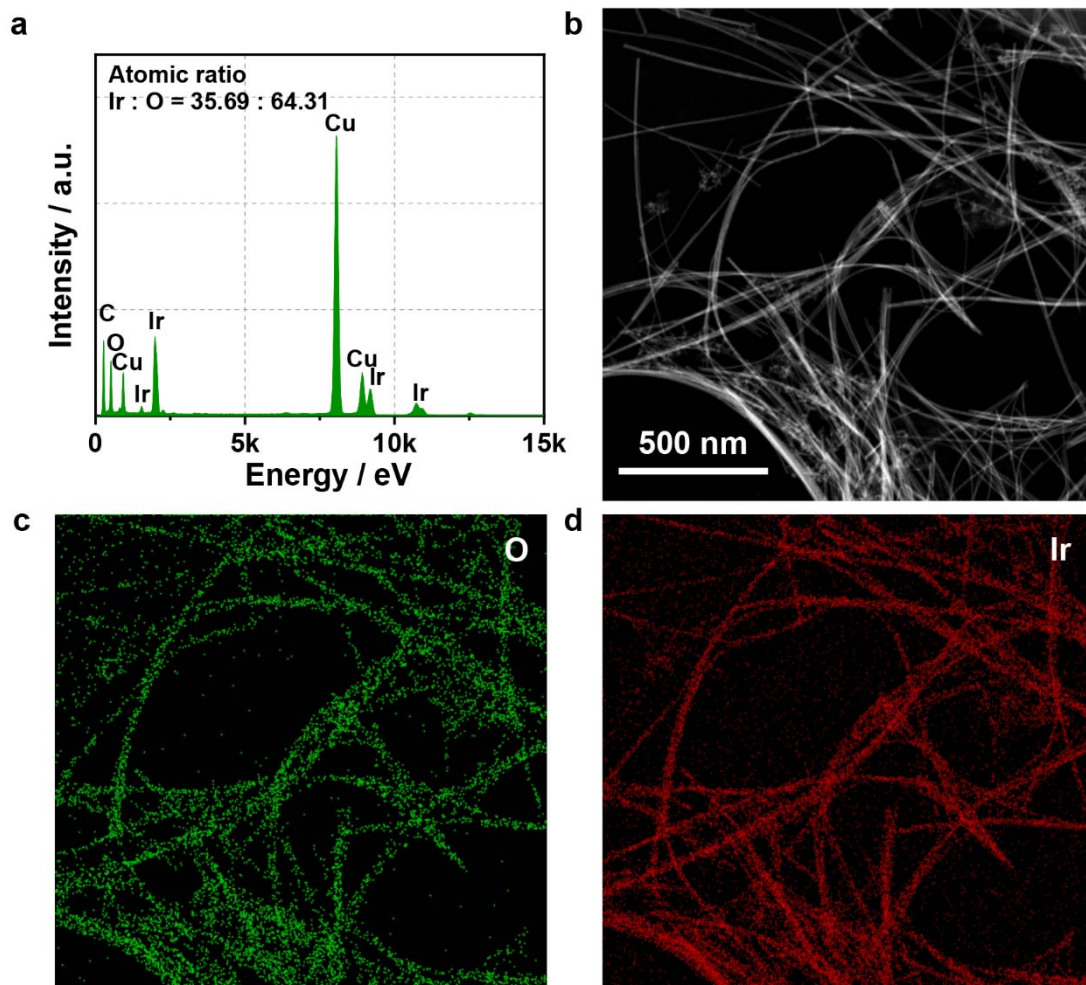
1
 2 **Supplementary Figure 6. Plot for Williamson-Hall analysis to calculate the lattice strain of**
 3 **IrO₂NR.** According to the Williamson-Hall equation: $\beta \times \cos\theta = K\lambda / D + 4\varepsilon \times \sin\theta$, where β is the
 4 full width at half-maximum of the peak, θ is the Bragg angle, K is the shape factor, D is the
 5 crystallite size, and λ is the wavelength of X-ray, the lattice strain of IrO₂NR is calculated to be
 6 0.388%. Error bars represent standard deviation, n = 3 independent replicates.
 7



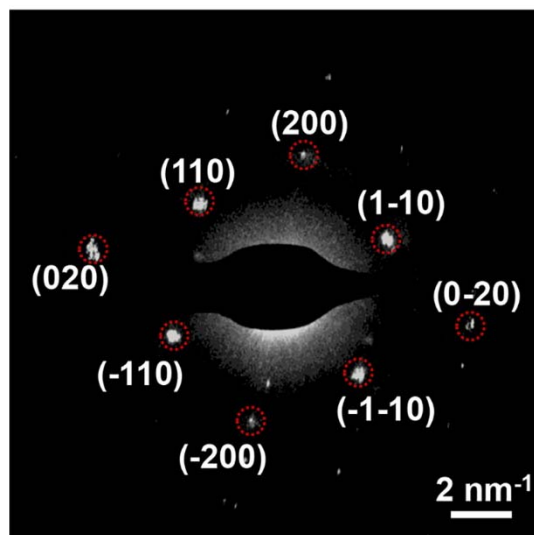
1
 2 **Supplementary Figure 7. Thickness of IrO₂NR.** (a) AFM image and (b-d) the corresponding
 3 height profiles showing the thickness of the IrO₂NR.
 4



1
2 **Supplementary Figure 8. I – V curve.** A representative bottom-contact device was fabricated from
3 a single IrO₂NR on an ITO substrate. Insert is the optical photograph of the device.
4



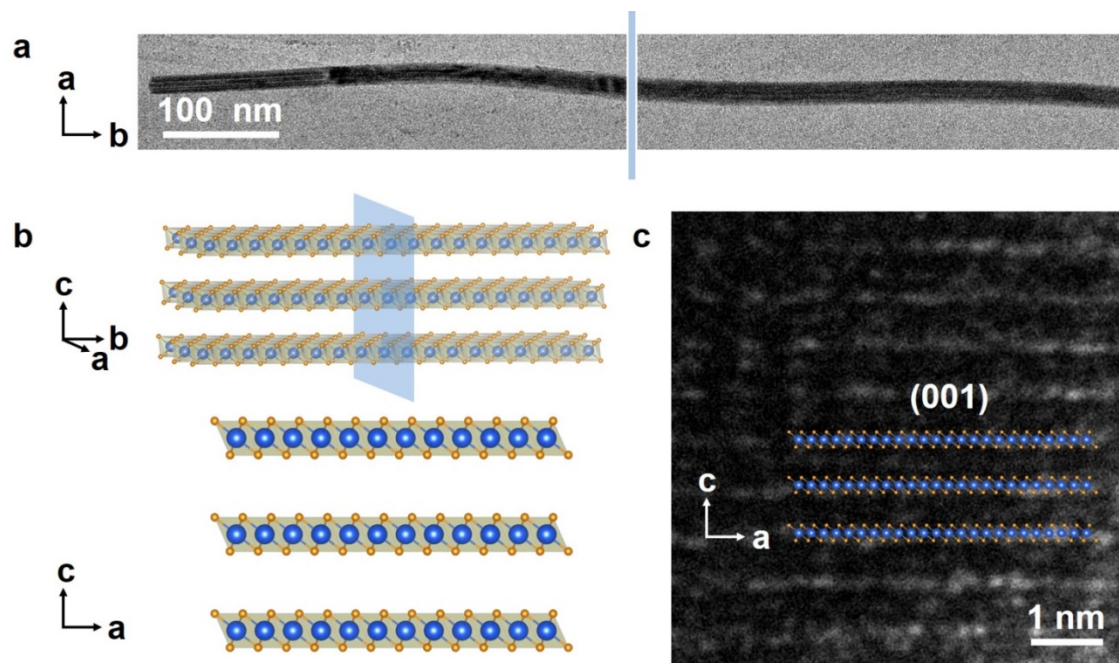
1
 2 **Supplementary Figure 9. Chemical composition analysis of IrO₂NR.** (a) TEM-EDS spectrum of
 3 IrO₂NR holding on a Cu grid. (b) HAADF-STEM image and corresponding elemental EDS mapping
 4 image showing the distributions of (c) O and (d) Ir elements.



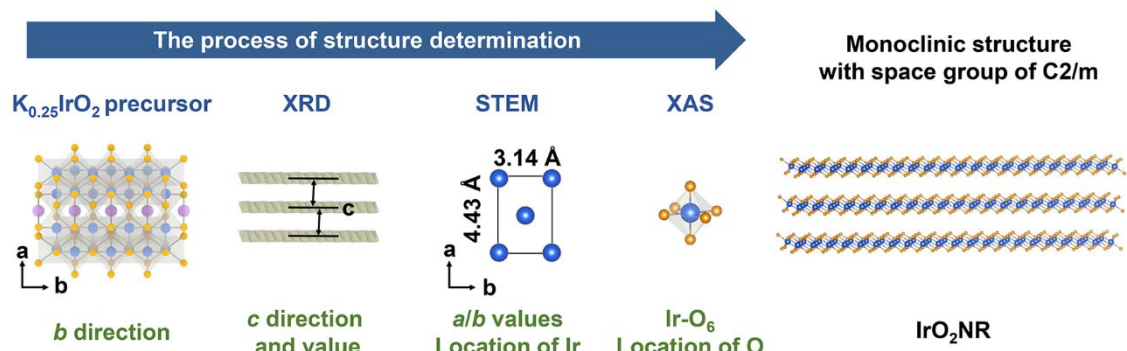
1

2 **Supplementary Figure 10. SAED pattern.** SAED pattern of IrO₂NR from the [001] direction.

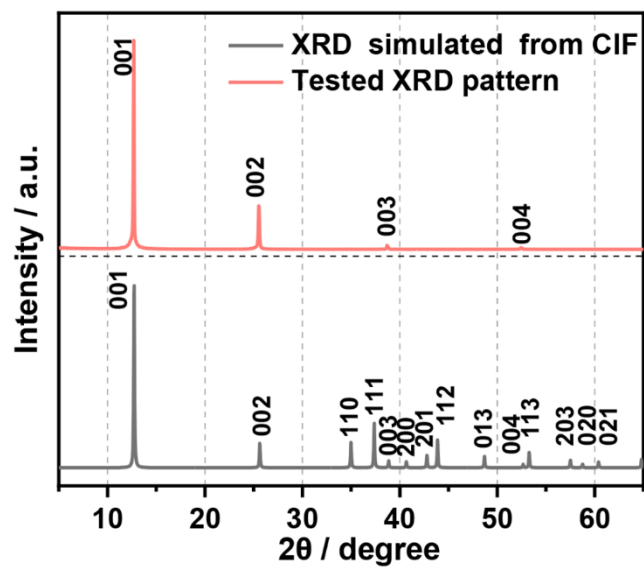
3



1
 2 **Supplementary Figure 11. Cross section of IrO₂NR.** (a) and (b) Schematic showing that the
 3 IrO₂NR was embedded in epoxy resin and sliced into pieces to observe the cross section. (c) HRTEM
 4 image of the cross section for IrO₂NR showing the (001) plane with distance of 0.69 nm.
 5

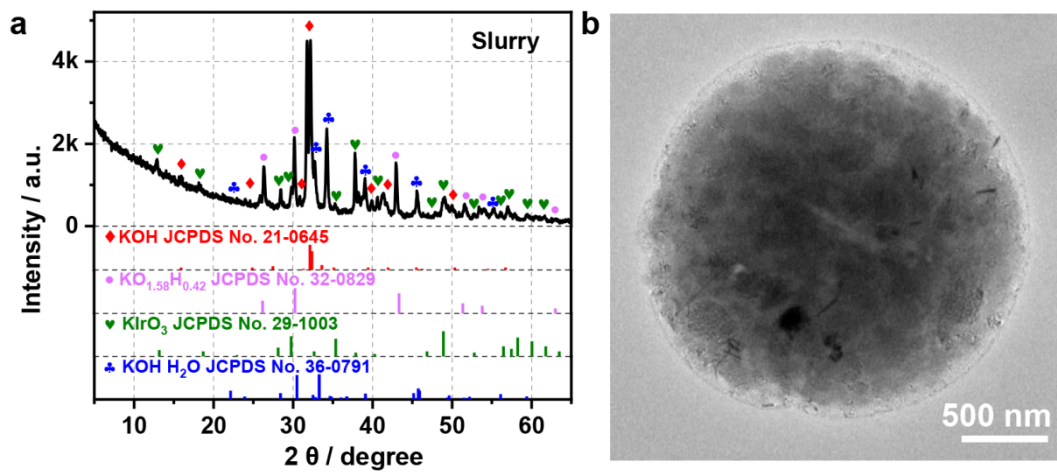


1
2 **Supplementary Figure 12. Structure determination process.** The schematic diagram showing the
3 process of the structure determination of IrO₂NR.

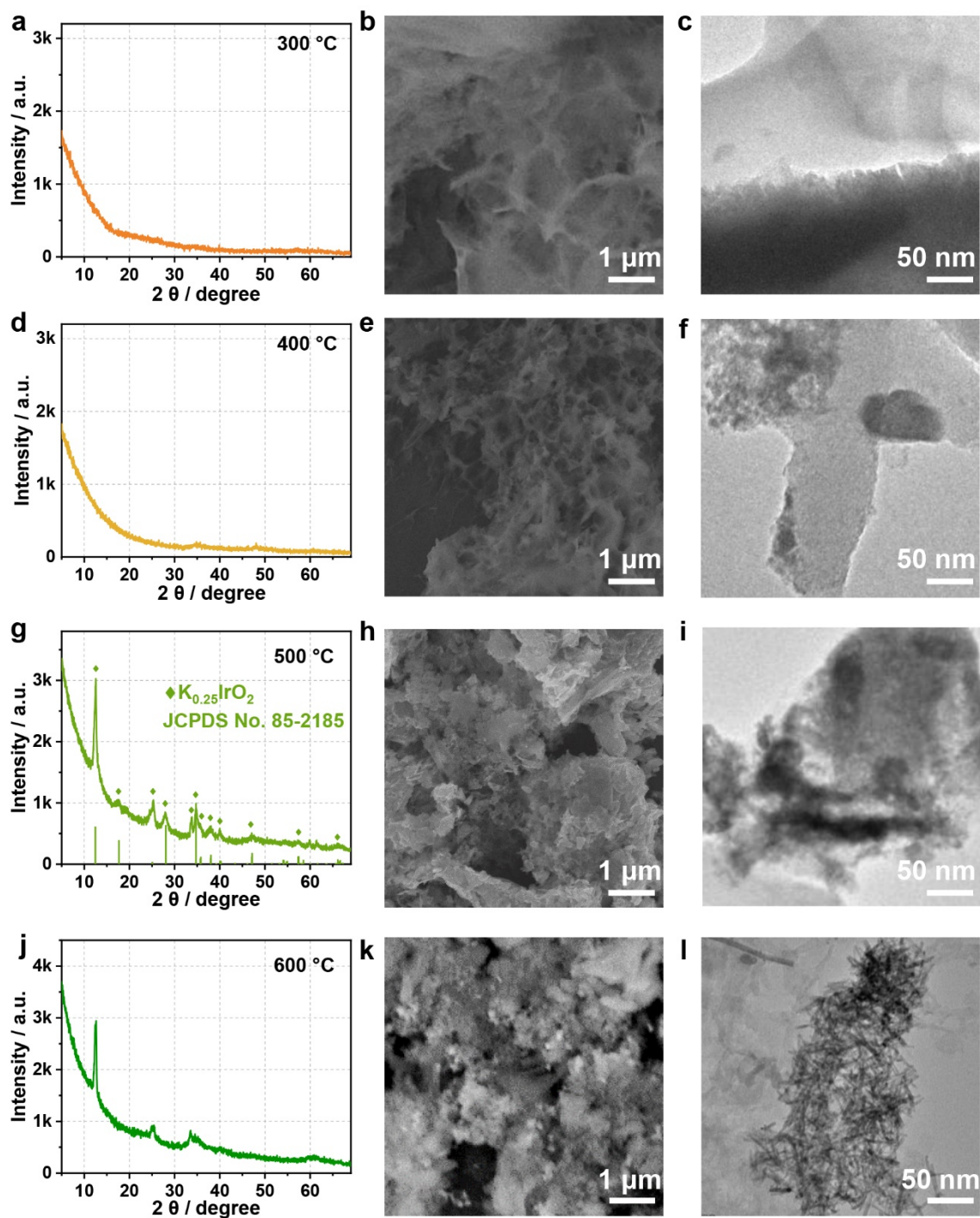


1

2 **Supplementary Figure 13. XRD patterns.** The simulated and tested XRD patterns of IrO₂NR.

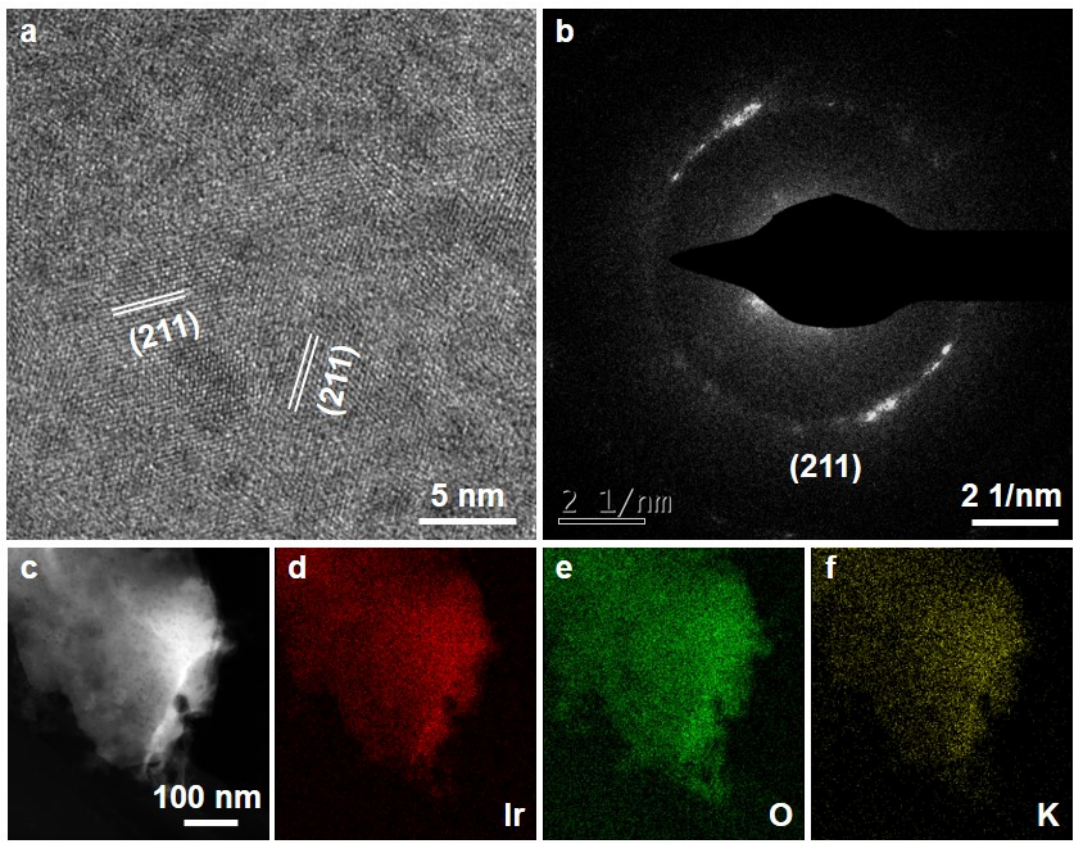


1
2 **Supplementary Figure 14. Characterization of the initial slurry before washing by water. (a)**
3 **XRD pattern and (b) TEM image.**

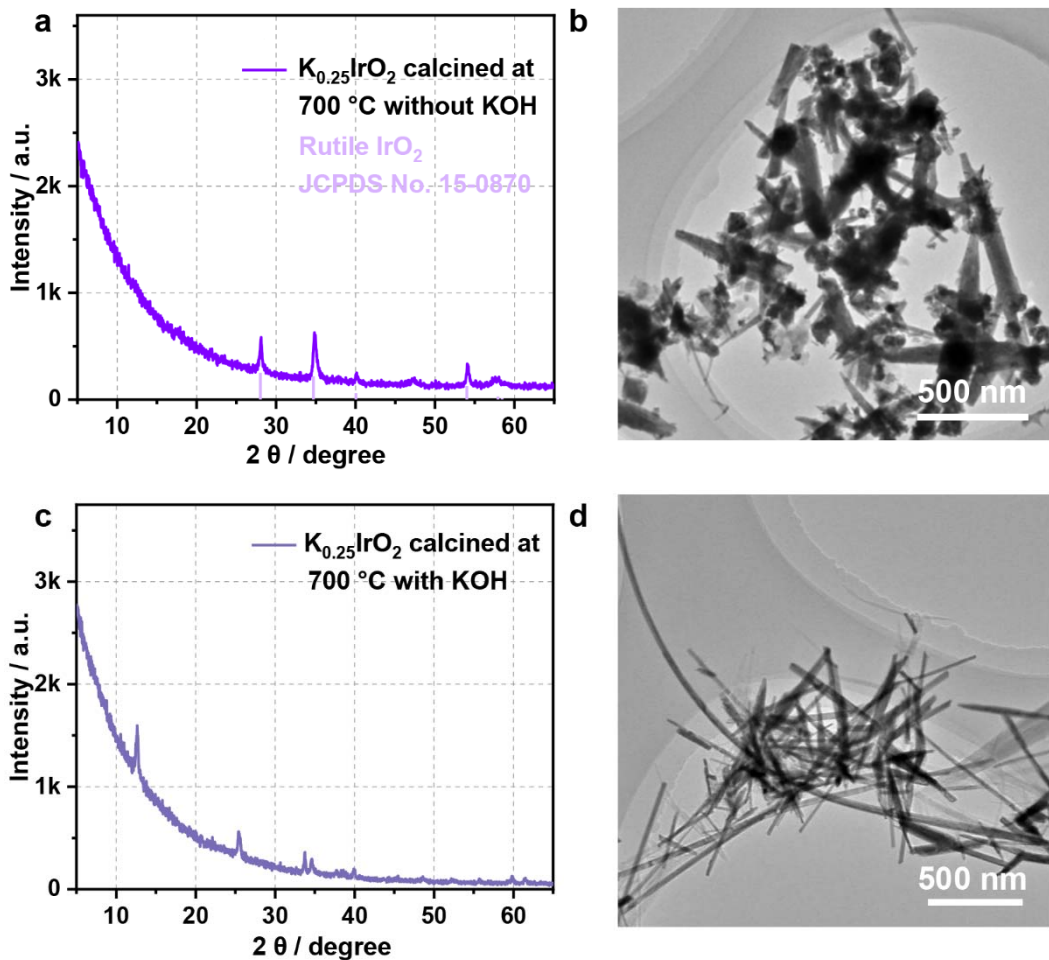


1

2 **Supplementary Figure 15. The phase and morphology of samples at different reaction**
 3 **temperatures. (a)** XRD pattern, **(b)** SEM image and **(c)** TEM image of sample annealed at 300 °C.
 4 **(d)** XRD pattern, **(e)** SEM image and **(f)** TEM image of sample annealed at 400 °C. **(g)** XRD pattern,
 5 indexed as $K_{0.25}IrO_2$; **(h)** SEM image and **(i)** TEM image of sample annealed at 500 °C. **(j)** XRD
 6 pattern, **(k)** SEM image and **(l)** TEM image of sample annealed at 600 °C.

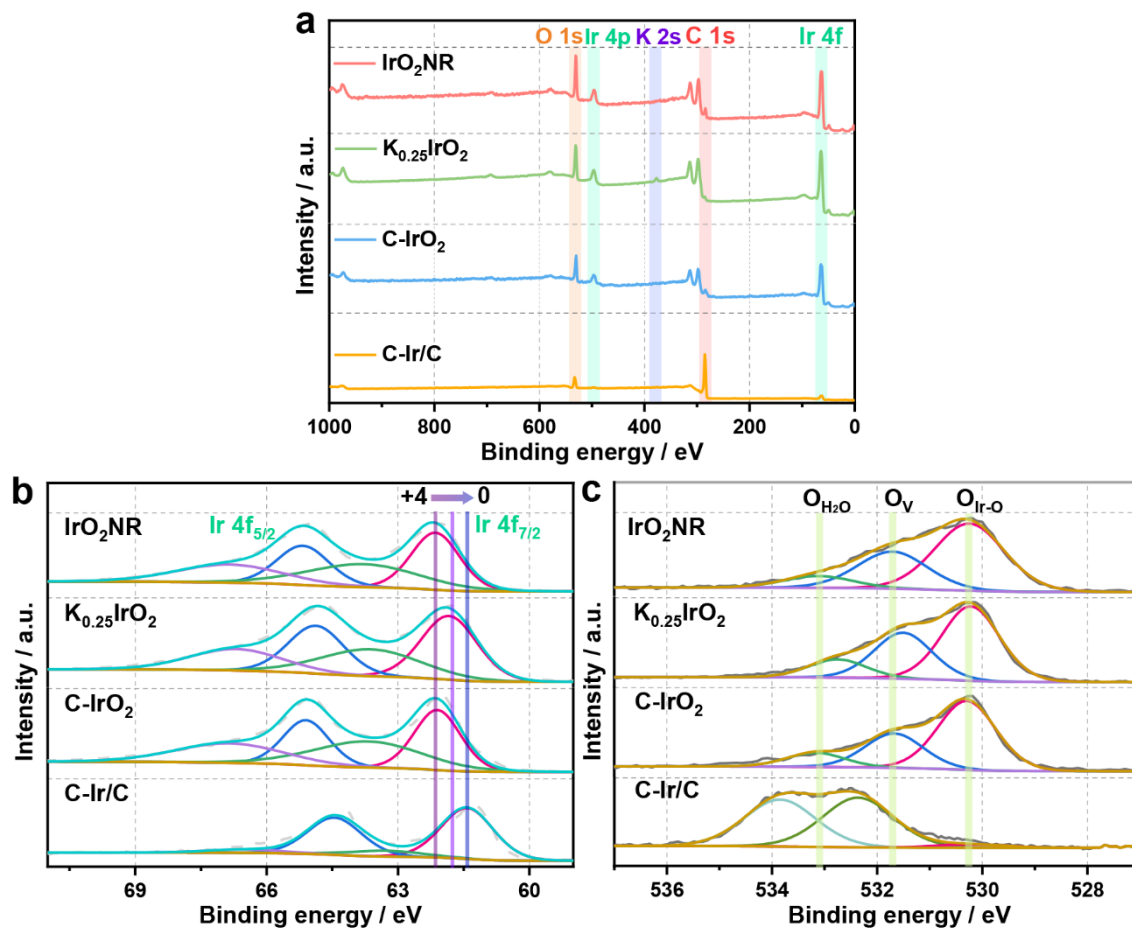


1
2 **Supplementary Figure 16. Chemical composition analysis of $K_{0.25}IrO_2$ obtained at reaction**
3 **temperature of 500 °C. (a) HRTEM image, (b) SAED, (c) HAADF-STEM image and**
4 **corresponding elemental EDS mapping image showing the distributions of (d) Ir, (e) O, and (f) K**
5 **elements.**



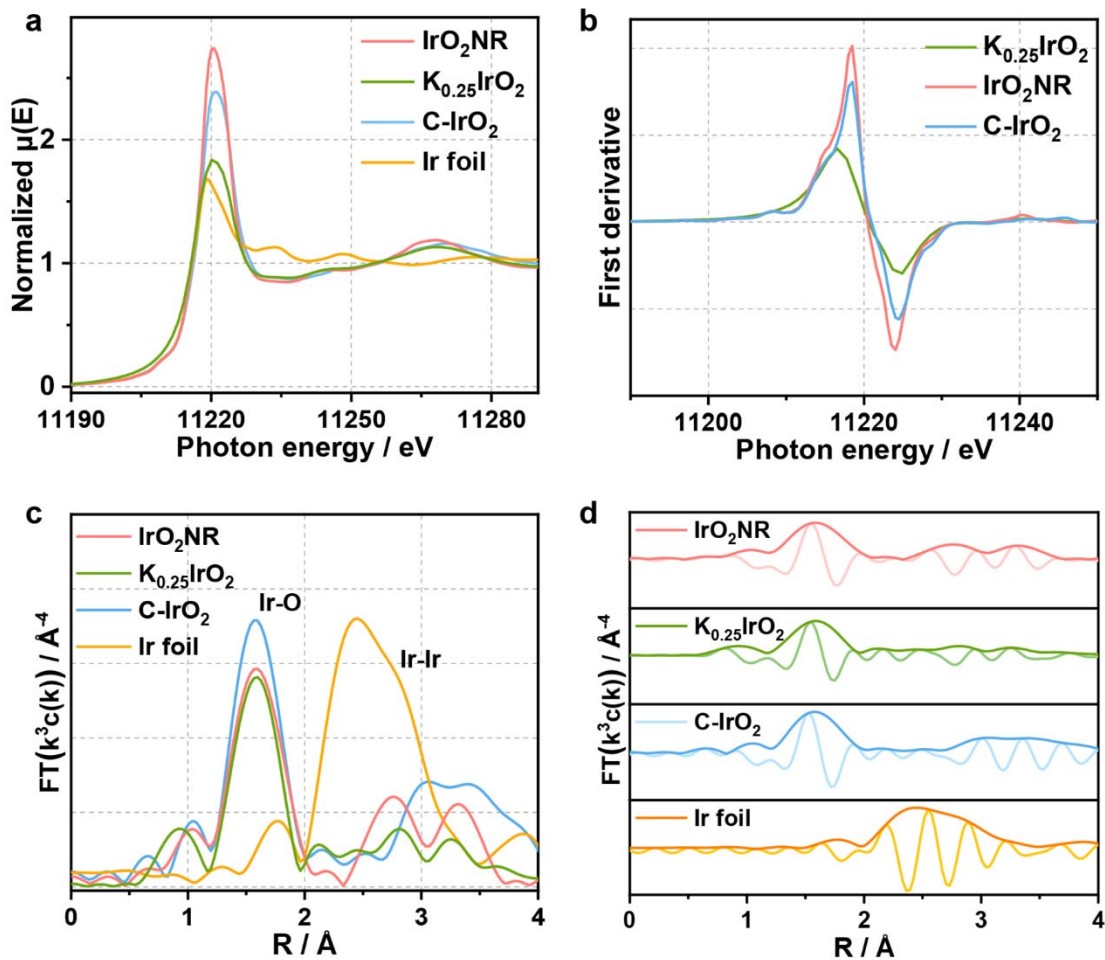
1

2 **Supplementary Figure 17. The comparison experiments showing the important role of the**
 3 **alkaline condition.** (a) XRD pattern of sample by calcining $K_{0.25}IrO_2$ at 700 °C for 2 h without KOH,
 4 which is indexed as Rutile IrO_2 . (b) TEM image of the above mentioned Rutile IrO_2 . (c) XRD pattern
 5 of the sample that calcined the mixture of $K_{0.25}IrO_2$ and KOH at 700 °C for 2 h, where the layered
 6 structure appears. (d) TEM image of the above mentioned sample.



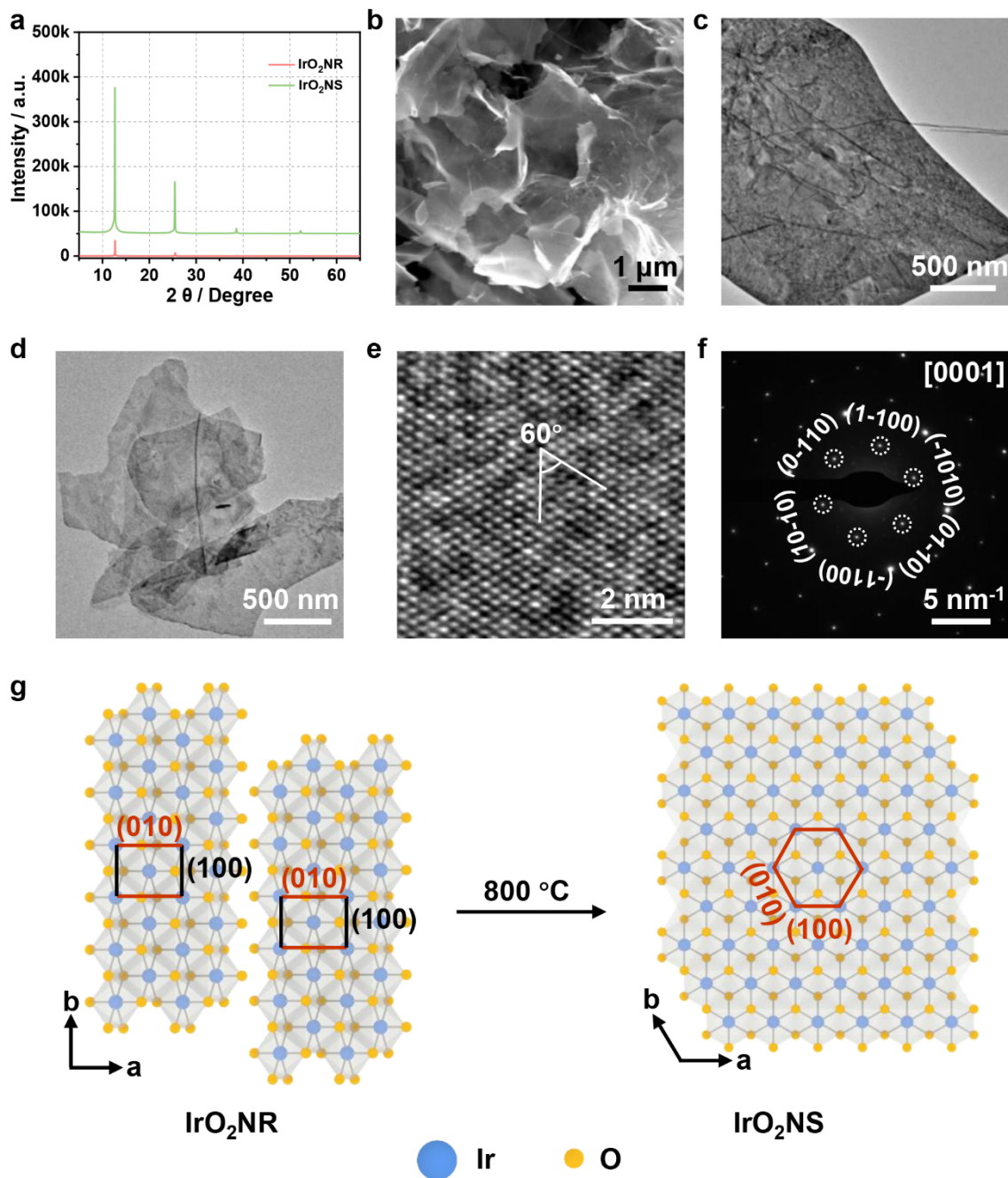
1

2 **Supplementary Figure 18. XPS spectra.** (a) Full spectra; (b) Ir 4f peaks and (c) O 1s peaks for
 3 IrO₂NR, K_{0.25}IrO₂, C-IrO₂ and C-Ir/C. The O 1s for C-Ir/C may come from the adsorbed oxygen
 4 species with carbon materials.

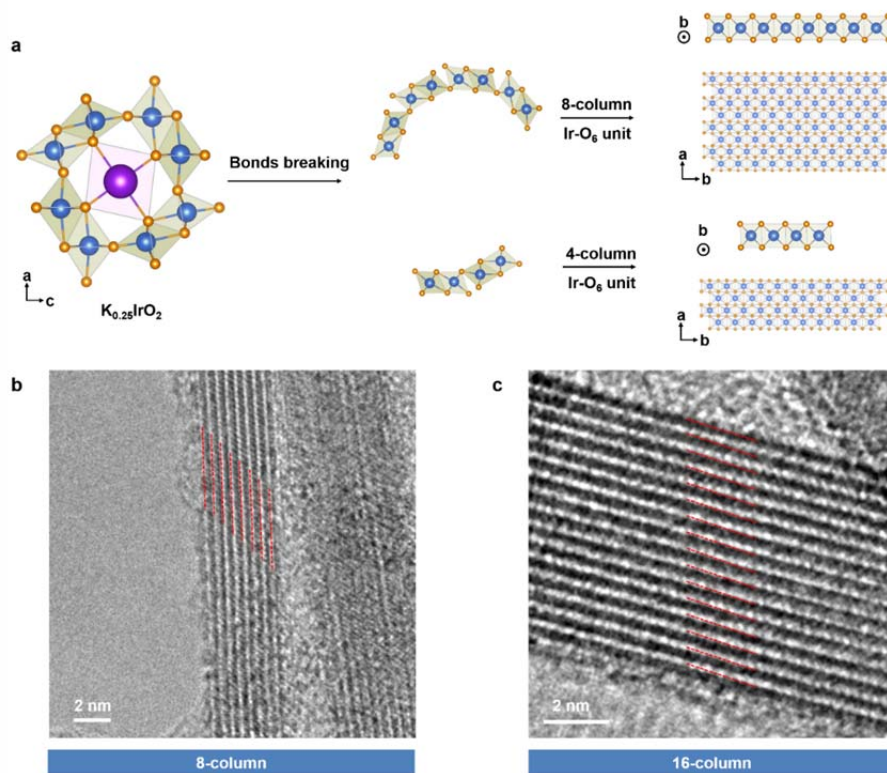


1
2
3
4
5
6

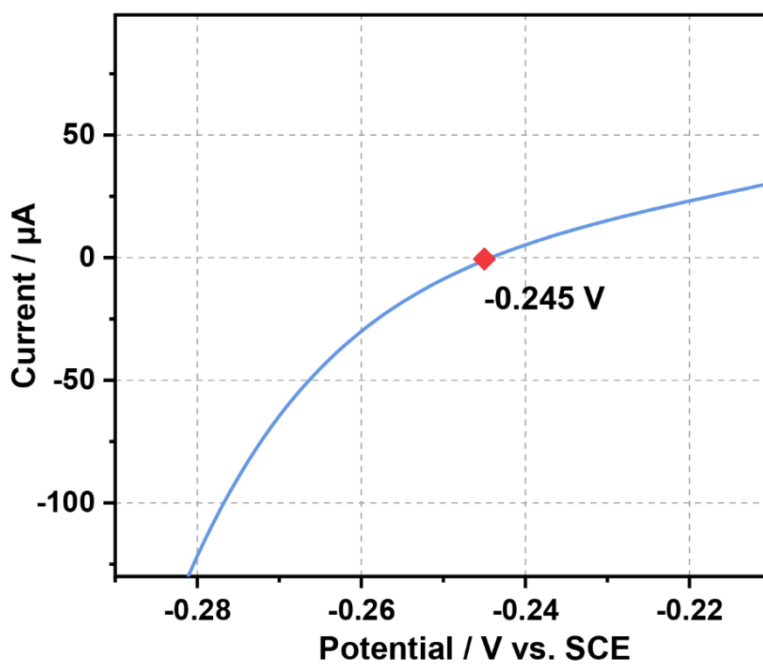
Supplementary Figure 19. XAS analysis of IrO₂NR, K_{0.25}IrO₂, C-IrO₂, and Ir foil. (a) XANES spectra for IrO₂NR, K_{0.25}IrO₂, C-IrO₂ and Ir foil. **(b)** The first derivative of normalized intensity for IrO₂NR, K_{0.25}IrO₂ and C-IrO₂ in **(a)**. **(c)** FT-EXAFS spectra for IrO₂NR, K_{0.25}IrO₂, C-IrO₂ and Ir foil. **(d)** The stacked magnitude and imaginary parts of IrO₂NR, K_{0.25}IrO₂, C-IrO₂, and Ir foil.



1
 2 **Supplementary Figure 20. The phase, morphology and formation mechanism of samples**
 3 **calcined at temperature of 800 °C. (a) XRD pattern of IrO₂NS and IrO₂NR; (b) SEM image of**
 4 **IrO₂NS; (c) and (d) TEM image of IrO₂NS, most of the sample showing nanosheet structure with**
 5 **several nanoribbons in it. (e) HRTEM image, and (f) SAED pattern of IrO₂NS; (g) Schematic**
 6 **diagram illustrating the structure evolution of IrO₂NR to IrO₂NS.**

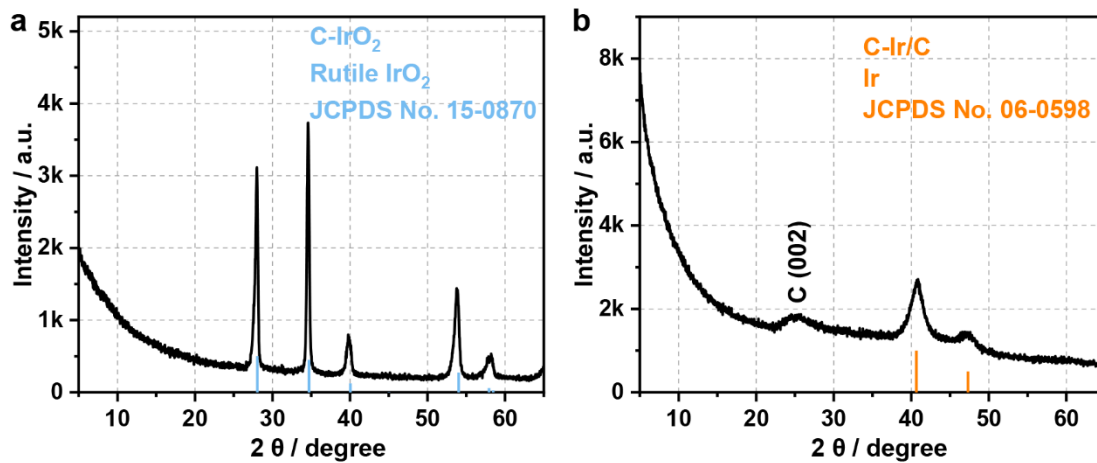


1
 2 **Supplementary Figure 21. The conversion of $K_{0.25}IrO_2$ to IrO_2NR .** (a) Schematic showing that
 3 the columns of $[IrO_6]$ subunits that perpendicular to the b direction are even numbers, taking 4 and 8
 4 columns as examples. (b) and (c) Corresponding HRTEM images showing the even number layers of
 5 $[IrO_6]$ subunits.



1

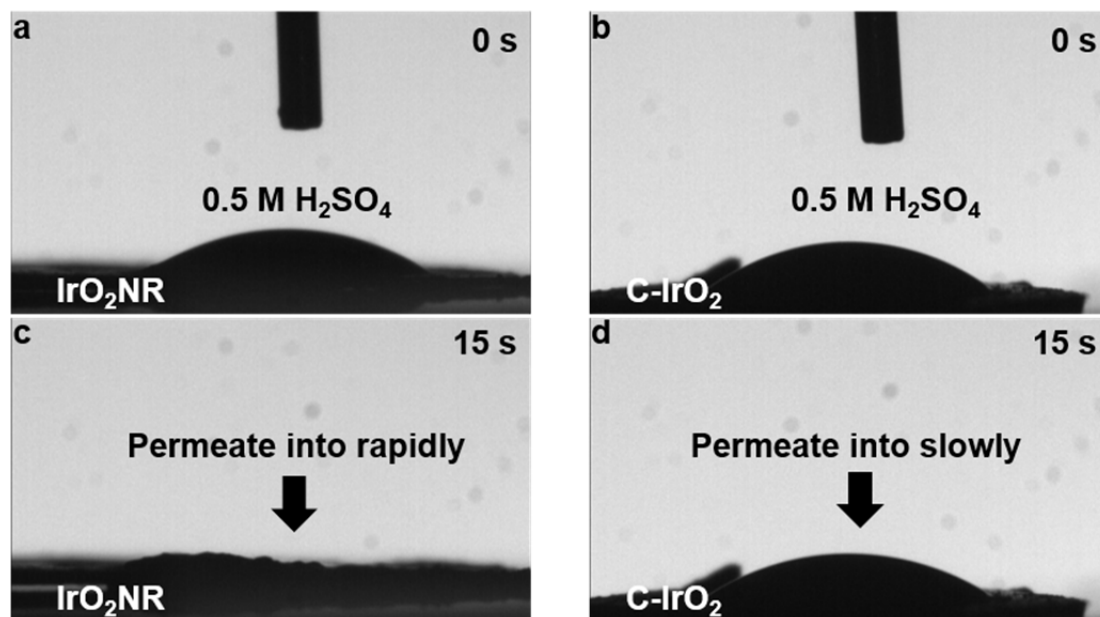
2 **Supplementary Figure 22. Calibration of the saturated calomel electrode (SCE).** Calibration of
3 the SCE electrode with respect to reversible hydrogen electrode (RHE) in 0.5 M H₂SO₄ electrolytes
4 bubbled with pure hydrogen gas at room temperature. Scan rate: 1 mV s⁻¹.



1

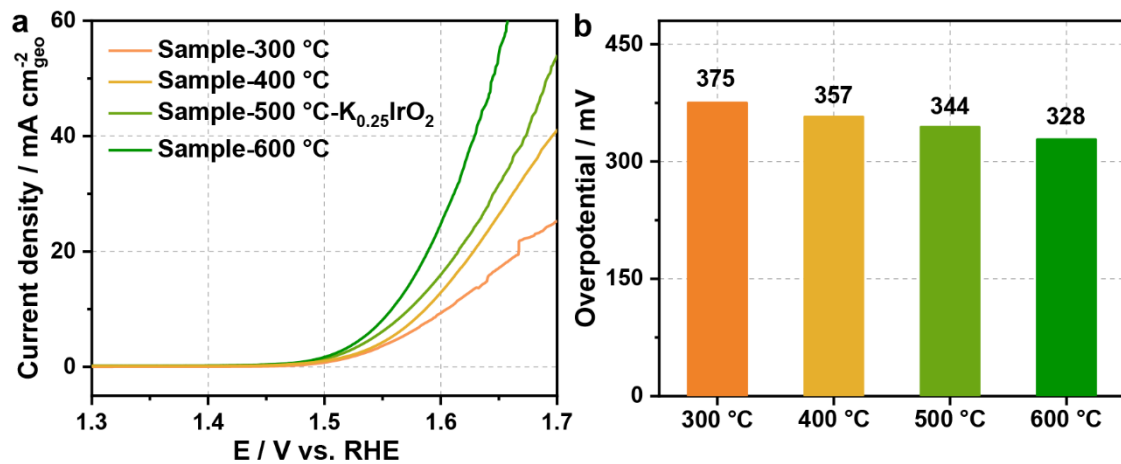
2 **Supplementary Figure 23. XRD patterns of commercial samples. (a) C-IrO₂ and (b) C-Ir/C.**

3

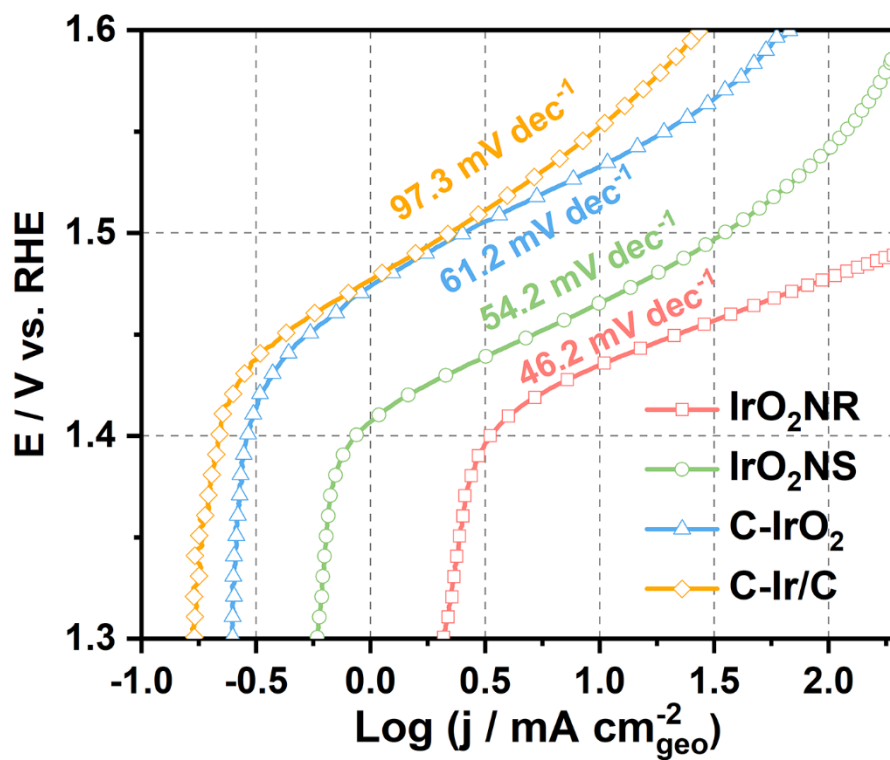


1

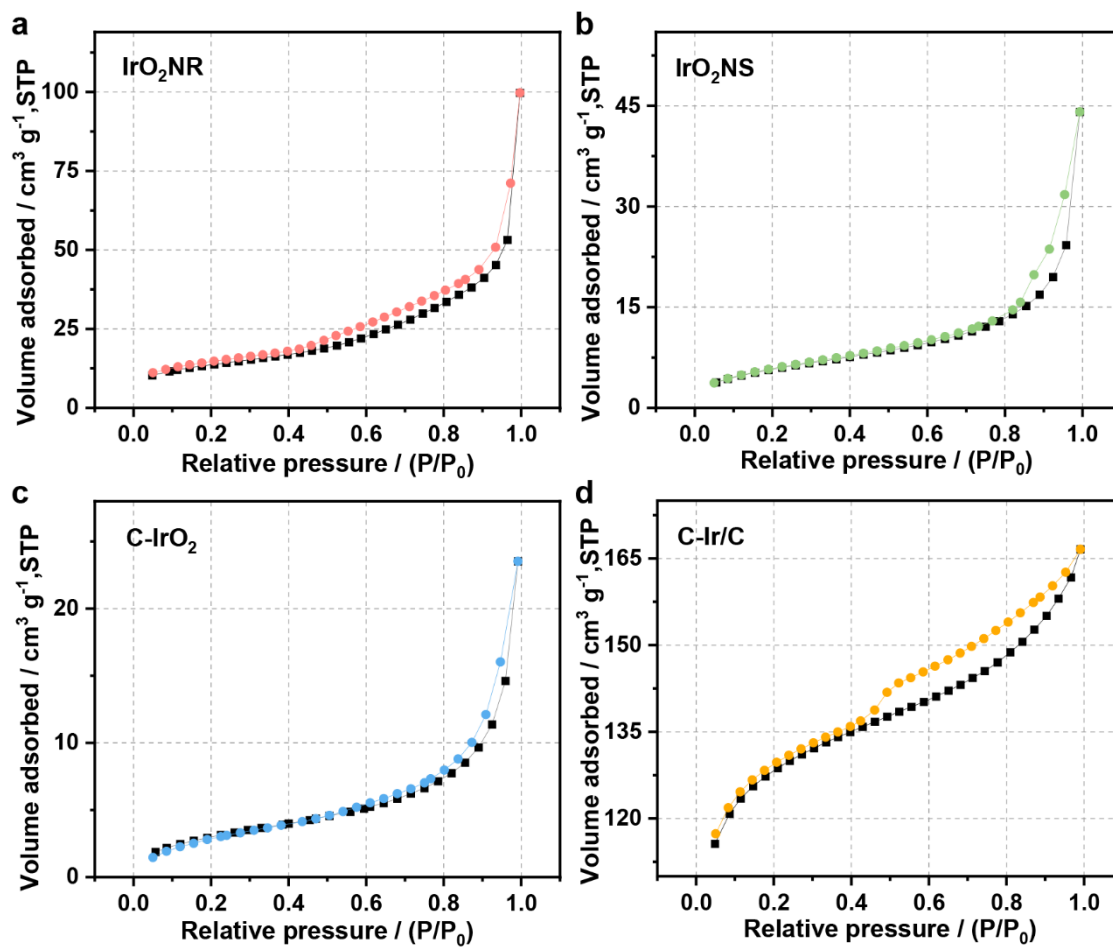
2 **Supplementary Figure 24. The contact angles at different time.** H₂SO₄ (0.5 M) droplet on
3 IrO₂NR tablet at (a) 0 s and (c) 15 s. H₂SO₄ (0.5 M) droplet on C-IrO₂ tablet at (b) 0 s and (d) 15 s.
4 The H₂SO₄ permeate into the IrO₂NR more rapidly than that into C-IrO₂.



1
2 **Supplementary Figure 25. OER performance of samples obtained at different reaction steps. (a)**
3 **The LSV curves, and (b) histograms showing the overpotentials of corresponding samples in 0.5 M**
4 **H₂SO₄.**

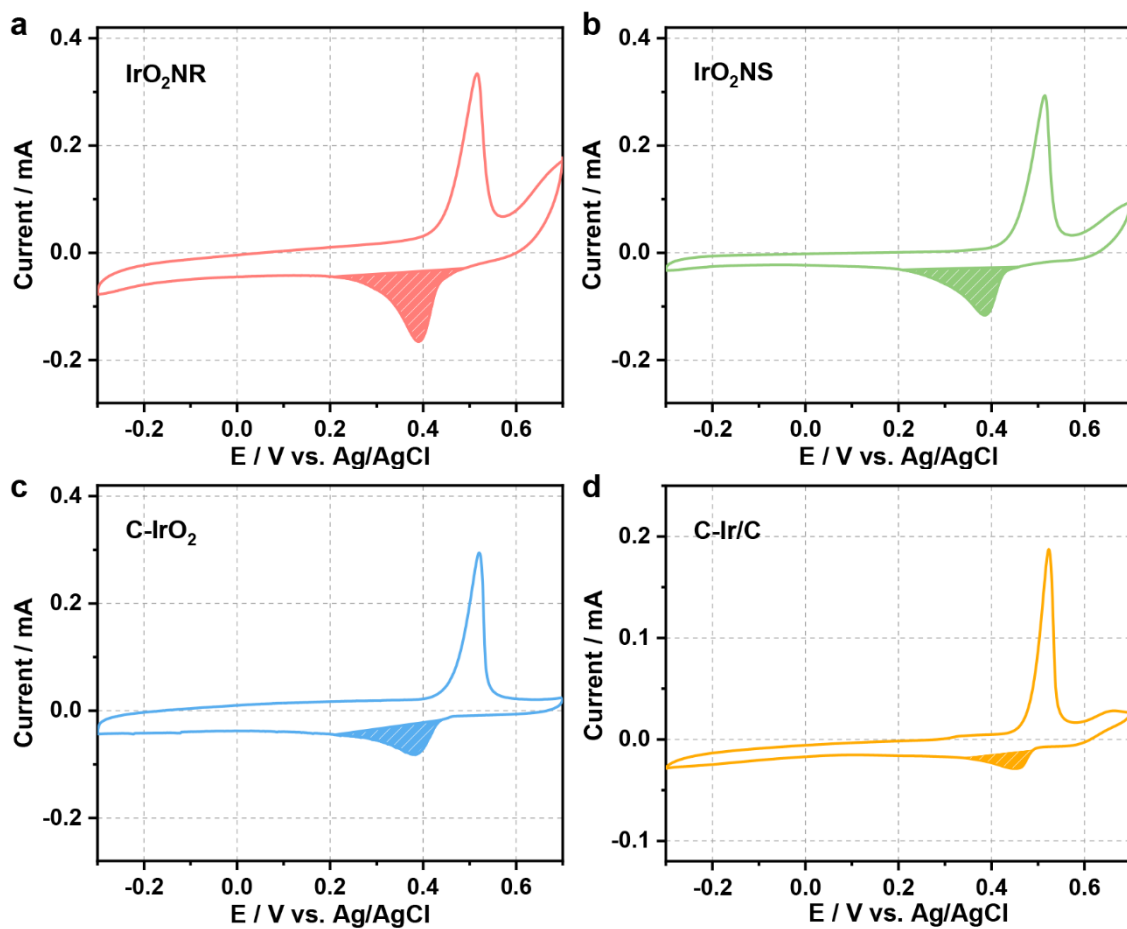


1
 2 **Supplementary Figure 26. Tafel plots.** Tafel plots and corresponding Tafel slopes of IrO₂NR,
 3 IrO₂NS, C-IrO₂, and C-Ir/C.



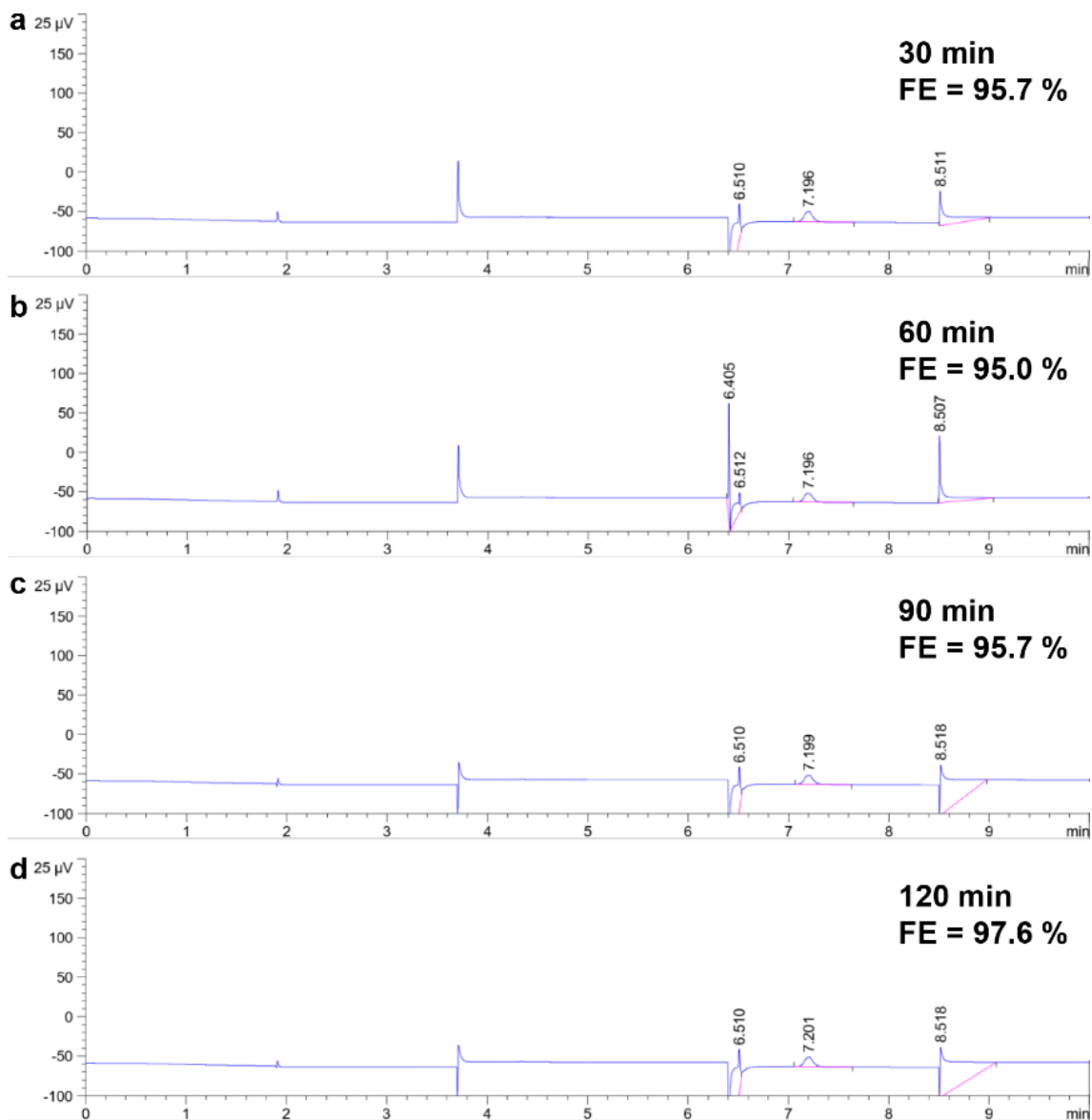
1

2 **Supplementary Figure 27. BET surface area measurements.** N₂ adsorption-desorption isotherms
 3 obtained for (a) IrO₂NR; (b) IrO₂NS; (c) C-IrO₂; and (d) C-Ir/C.

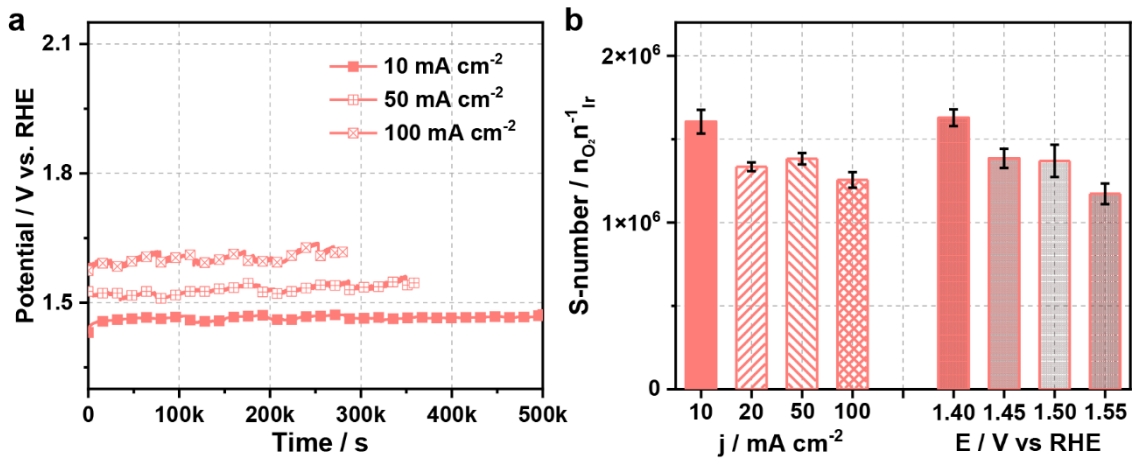


1

2 **Supplementary Figure 28. ECSA measurements.** The CV curves of (a) IrO₂NR; (b) IrO₂NS; (c)
 3 C-IrO₂; and (d) C-Ir/C in 0.1 M HClO₄ electrolyte containing 1 mM mercury nitrate at the scan rate
 4 of 100 mV s⁻¹.

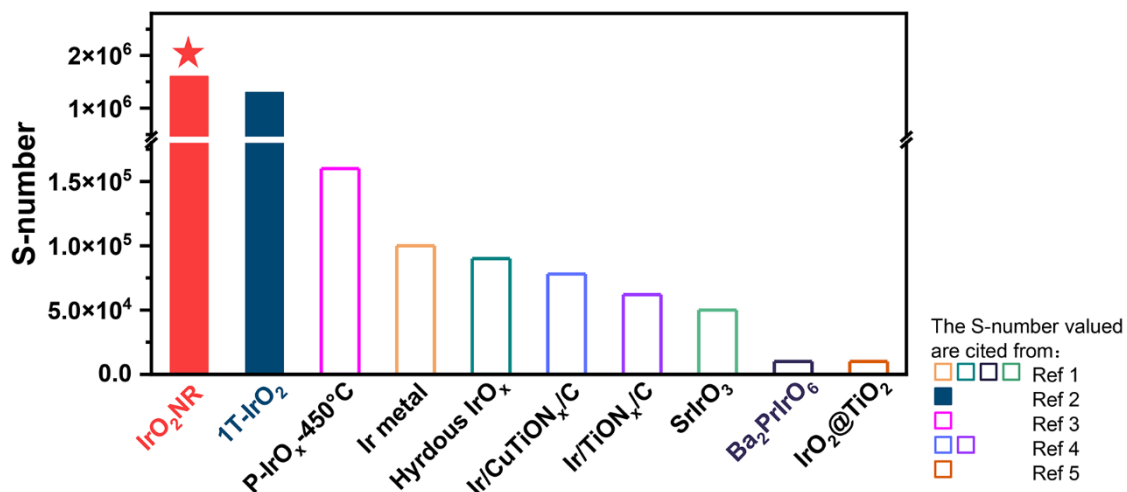


1
2 **Supplementary Figure 29. Gas chromatograph data of the produced oxygen by IrO₂NR at**
3 **different reaction time at a current density of 20 mA cm_{geo}⁻². (a) 30 min; (b) 60 min; (c) 90 min;**
4 **and (d) 120 min.**



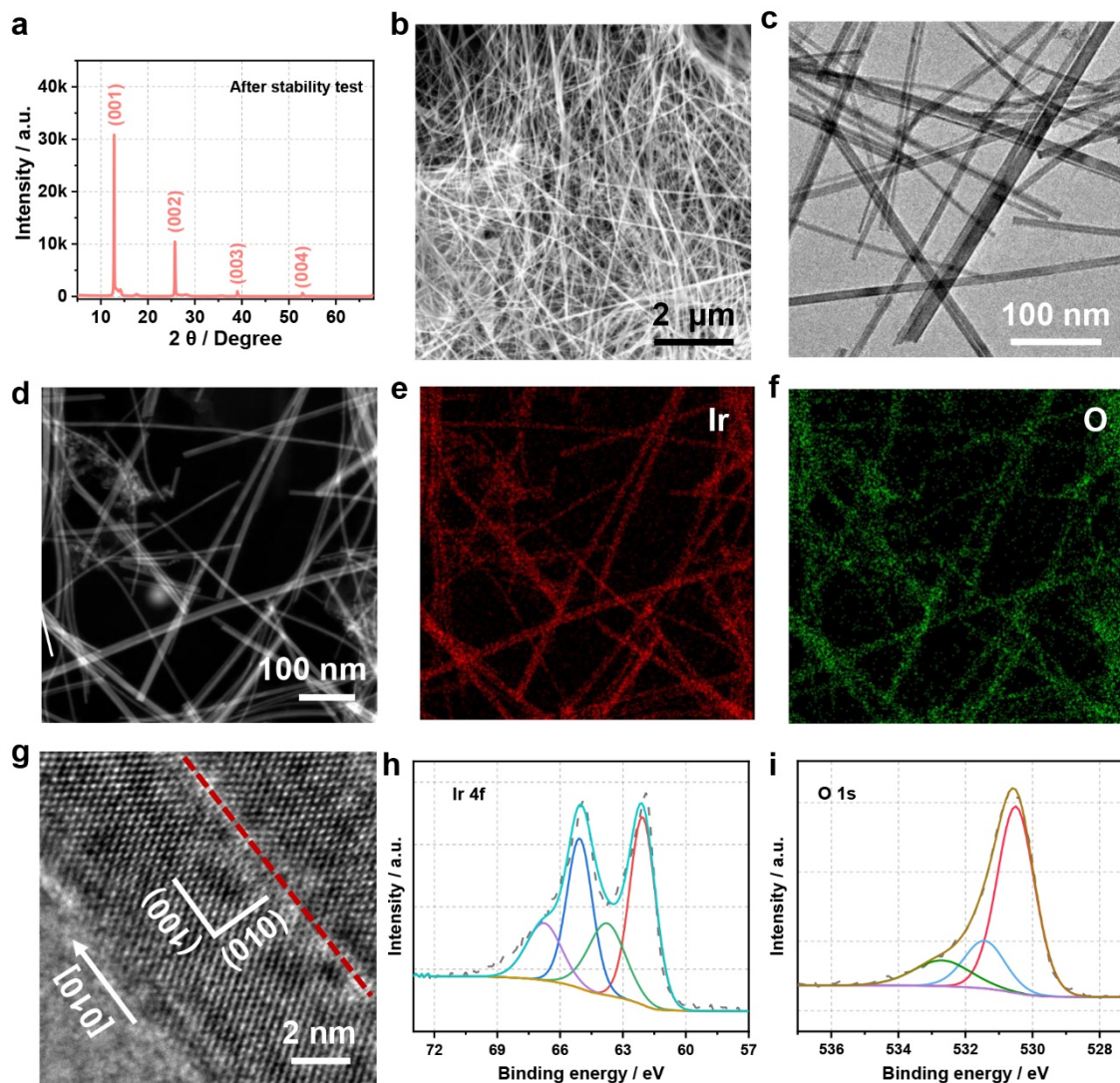
1

2 **Supplementary Figure 30. Stability tests for IrO₂NR.** (a) Chronopotentiometric curves of IrO₂NR
 3 at a constant current density of 10, 50 and 100 mA cm⁻²; and (b) S-numbers of IrO₂NR at different
 4 current densities and different applied potentials (Error bars represent standard deviation, n = 3
 5 independent replicates).

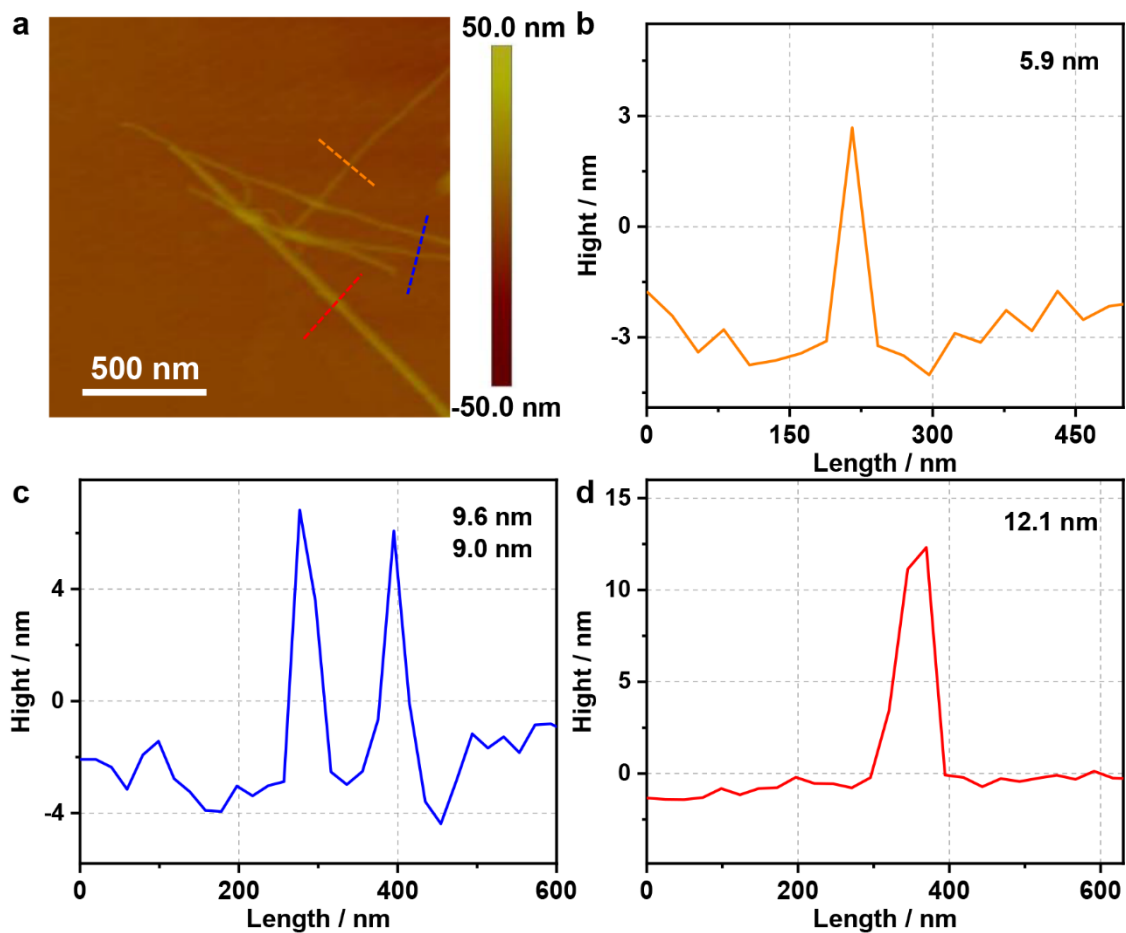


1

2 **Supplementary Figure 31. Comparison of S-number values.** S-number values of IrO₂NR
 3 compared with other reported iridium-based OER catalysts¹⁻⁵.

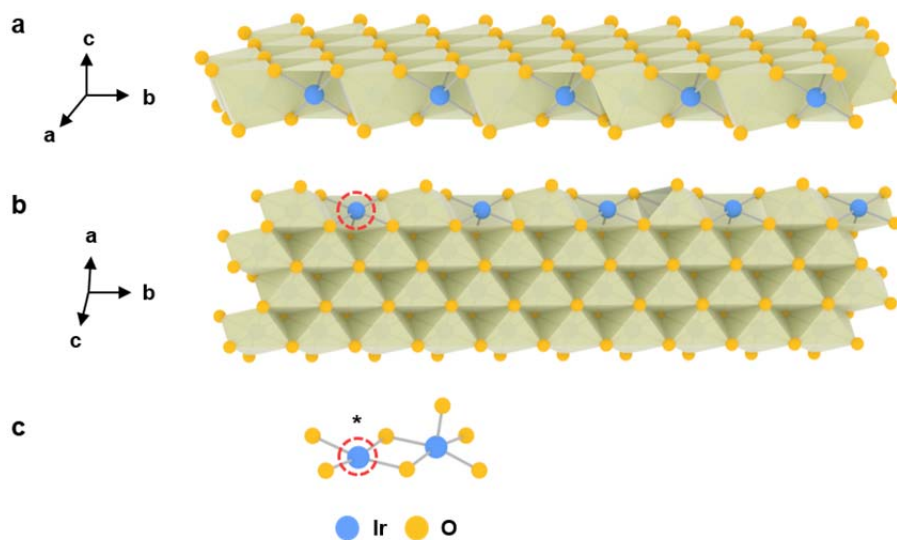


1
 2 **Supplementary Figure 32. Structure of IrO₂NR after long stability test with the constant**
 3 **current density of 10 mA cm⁻² in 0.5 M H₂SO₄ electrolyte. (a) XRD pattern showing the**
 4 **maintained layered structure of IrO₂NR. (b) SEM image, (c) TEM image, (d) HAADF-STEM image**
 5 **and corresponding elemental EDS mapping image showing the distribution of (e) Ir and (f) O**
 6 **elements. The IrO₂NR are uniformly dispersed without aggregation or morphology change after**
 7 **undergoing the long-term OER. (g) HRTEM image showing clearly the regular atomic arrangement.**
 8 **The XPS spectra of (h) Ir 4f peak and (i) O 1s peak. The valence state of Ir keeps almost constant**
 9 **with the original IrO₂NR.**



1
2
3
4

Supplementary Figure 33. Thickness of IrO₂NR after stability test. (a) AFM image and (b-d) the corresponding height profiles showing the thickness of the IrO₂NR.



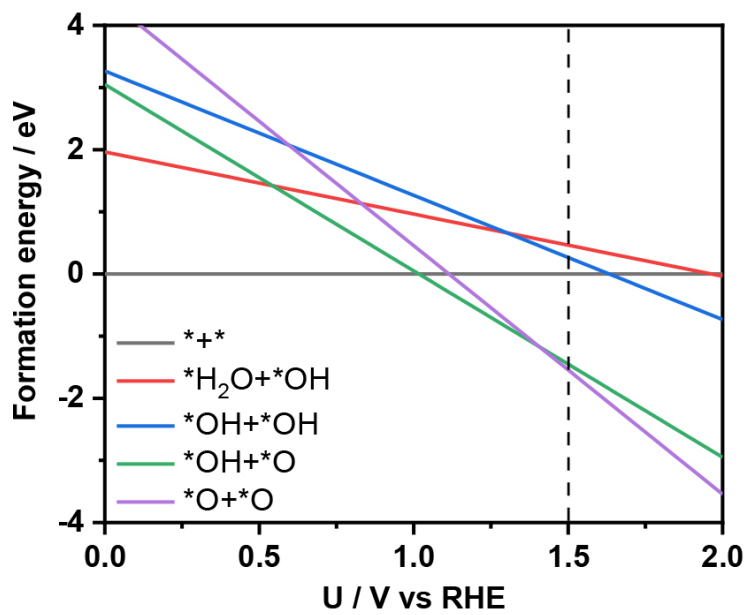
1

2 **Supplementary Figure 34. Structure of IrO₂NR. (a) and (b)** Crystal structures of the IrO₂NR with

3 an exposed edge for the (001) plane during theoretical calculation with different viewing angles. (c)

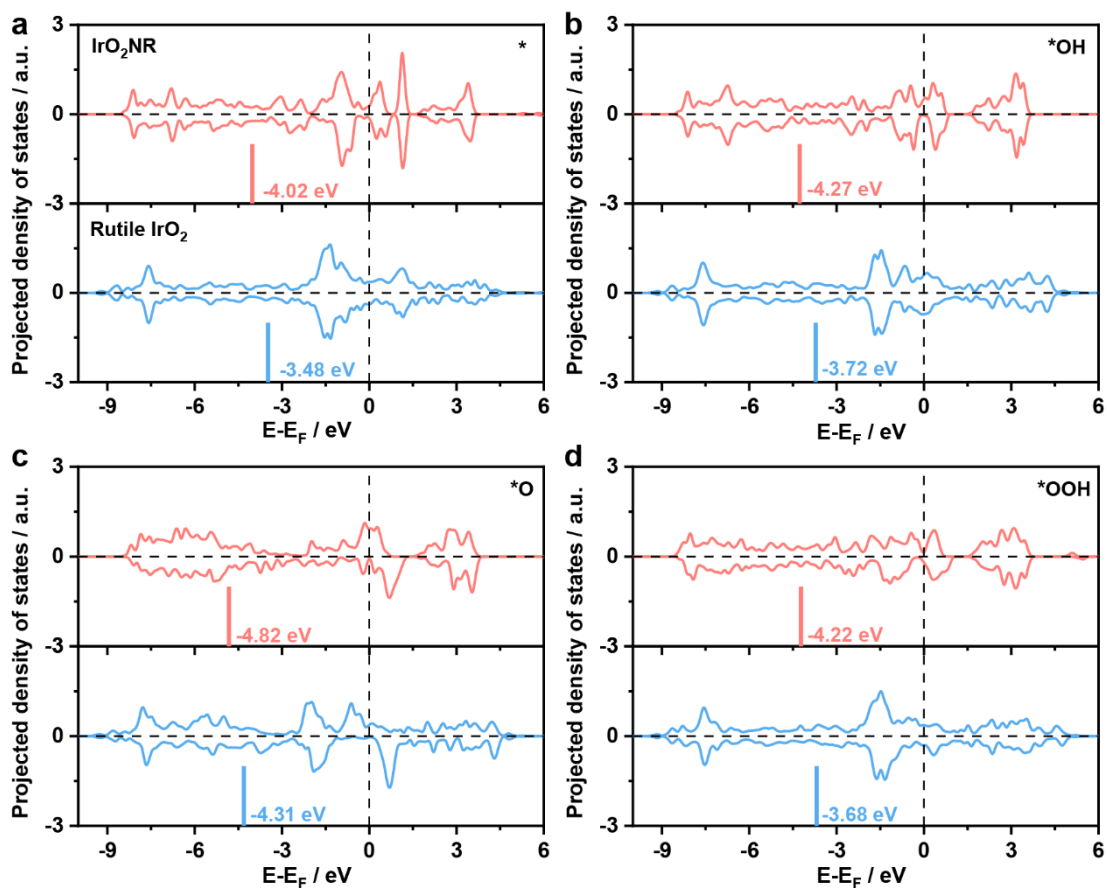
4 The ball-and-stick model extracted from (b) to show the surface location of active Ir in Figure 4c.

5



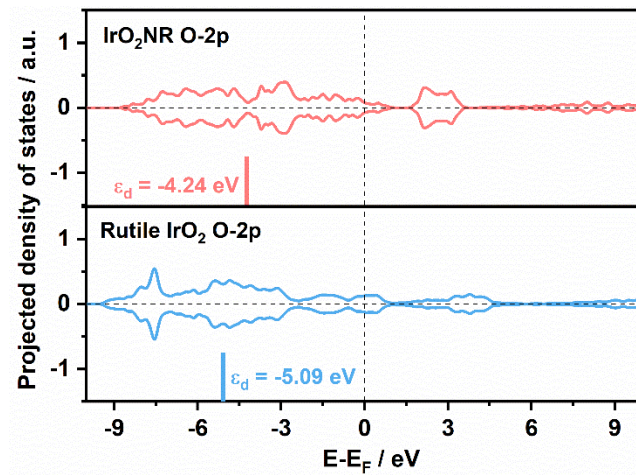
1

2 **Supplementary Figure 35. Surface termination of the IrO₂NR under acidic conditions.**
 3 Theoretical Pourbaix diagram of IrO₂NR (100) surface.



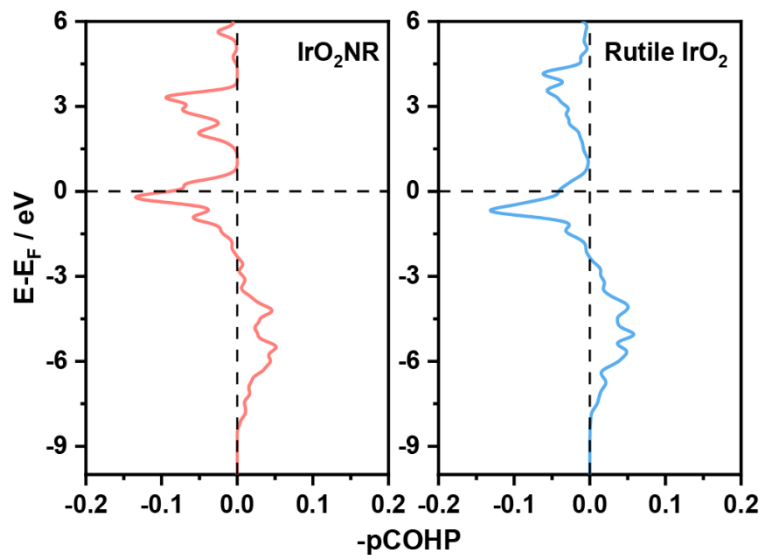
1
2
3
4
5

Supplementary Figure 36. The electronic properties of the Ir atoms. Comparison of *d*-orbital distribution of Ir atoms for the (a) *, (b) *OH, (c) *O and (d) *OOH in IrO₂NR (red curve) and Rutile IrO₂ (blue curve).



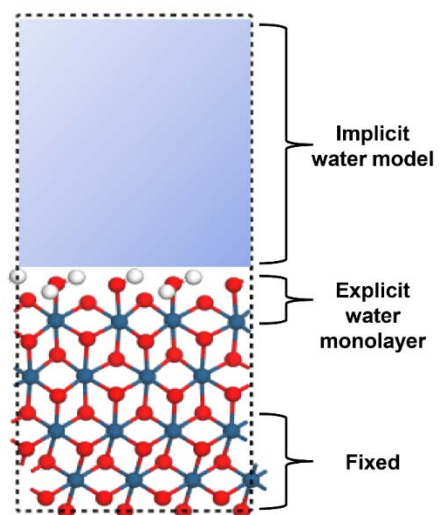
1

2 **Supplementary Figure 37. The electronic properties of the O atoms.** Comparison of the
3 $2p$ -orbital distribution of the O atoms in the Rutile IrO₂ and IrO₂NR.
4



1
2
3
4

Supplementary Figure 38. Comparisons of projected crystal orbital Hamilton population. The Ir-O bonds in IrO₂NR (left) and the Ir-O bonds in Rutile IrO₂ (right).



1
2 **Supplementary Figure 39. Implicit solvation model.** Schematic implicit solvation model for the
3 calculation of Pourbaix diagram of IrO₂NR.

1 **Supplementary Tables**2 **Supplementary Table 1. Crystallographic information for IrO₂NR and C-IrO₂.**

Material	Crystal system	Bravais lattice	Unit-cell dimension	Space group
IrO₂NR	Monoclinic	C-centered	$a = 4.43 \pm 0.01 \text{ \AA}, b = 3.14 \pm 0.02 \text{ \AA}, c = 6.95 \pm 0.03 \text{ \AA}; \alpha = \beta = \gamma = 90^\circ$	C2/m (12)
C-IrO₂	Tetragonal	Primitive	$a = b = 4.498 \text{ \AA}, c = 3.154 \text{ \AA}; \alpha = \beta = \gamma = 90^\circ$	P4₂/mm (136)
IrO₂NS	Trigonal	Primitive	$a = b = 3.11 \text{ \AA}, c = 6.91 \text{ \AA}; \alpha = \beta = 90^\circ, \gamma = 120^\circ$	P-3m1 (164)

3

1 **Supplementary Table 2. The BET areas and ECSAs of IrO₂NR, C-IrO₂ and C-Ir/C.**

Material	IrO₂NR	IrO₂NS	C-IrO₂	C-Ir/C
BET area / m² g⁻¹	47.3	22.0	11.8	397.6
ECSA / m² g⁻¹	31.3	21.6	15.5	16.3

2

1 **Supplementary Table 3. OER activities of IrO₂NR, C-IrO₂ and C-Ir/C tested at 1.5 V vs RHE.**

Material	IrO₂NR	IrO₂NS	C-IrO₂	C-Ir/C
Overpotential @ 10 mA cm_{geo}⁻² / mV	205	235	303	322
Tafel slope / mV dec⁻¹	46.2	54.2	61.2	97.3
Geometric activity / mA cm_{geo}⁻²	456.8	57.4	2.6	2.2
Mass activity / mA mg_{Ir}⁻¹	2354.5	295.9	13.4	48.6
Specific activity / mA cm_{BET}⁻²	4.27	1.15	0.1	0.01
Specific activity / mA cm_{ECSA}⁻²	6.45	1.17	0.07	0.3
TOF_{BET} / s_{BET}⁻¹	9.27	3.10	0.15	0.01
TOF_{ECSA} / s_{ECSA}⁻¹	14.01	3.07	0.12	0.3

2

1 **Supplementary Table 4. Comparison of overpotential, Tafel slope and TOF of IrO₂NR with**
 2 **previously reported Ir-based catalysts.**

Catalyst	Electrolyte	$\eta@10$ mA $\text{cm}_{\text{geo}}^{-2} / \text{mV}$	Tafel slope / mV dec^{-1}	Potential at 1 mA $\text{cm}_{\text{BET}}^{-2} /$ V	Mass activity / mA $\text{mg}_{\text{Ir}}^{-1}$ @ Potential vs RHE	TOF/ s^{-1} @ Potential vs RHE	Ref
IrO₂NR	0.5 M H ₂ SO ₄	205	46.2	1.48	2354.5@1.50 V	9.27@1.50 V ^a 14.01@1.50 V ^b	This work
3%IrO ₂ @BCNT		291	52	/	/	/	6
H-Ti@IrO _x		277	29	1.58 V@ 0.04 mA $\text{cm}_{\text{ECSA}}^{-2}$	1500@1.58 V	/	7
IrRu@Te		220	35	/	590@1.50 V	/	8
RuIrO _x		233	42	/	/	/	9
Rh ₂₂ Ir ₇₈ /VXC		292	101	/	1174@1.53 V	5.095@1.53 V ^b	10
Li-IrO _x		300	39	1.55	1000@1.575 V	0.31@1.53 V ^a	11
IrO _x on IrCo		247	49	/	/	/	12
IrO ₂ /GCN		278	57	/	1280@1.60 V	0.175@1.60 V ^d	13
Ru@IrO _x		282	69.1	/	644.8@1.56 V	/	14
Ir nanosheets		240	49	/	260@1.50 V	/	15
6H-SrIrO ₃		248	/	1.55	75@1.525 V	/	16
Ir/GF		290	46	/	200@1.62 V	/	17
IrO ₂ /CNT		293	67	/	11.2@1.53 V	/	18
IrCoNi/CFP		303	54	/	750@1.53 V	0.26@1.53 V ^a	19
IrO ₂ -RuO ₂ @Ru		281	53.1	/	/	0.039@1.55 V ^d	20
IrO _x -Ir		290	~44	/	100@1.53 V	~1.1@1.53 V ^b	21
IrO _x /SrIrO ₃		270	/	1.47	/	/	22
IrNiO _x		/	/	/	676@1.53 V	/	23
La ₂ LiIrO ₆		/	/	/	~13@1.50 V	/	24
1T-IrO ₂		235	49	/	152@1.50 V	2.0@1.50 V ^b	2
ultrathin iridium		328	45.4	/	209@1.53 V	/	25

nanosheets							
S-doped M-SrIrO ₃ nanosheets		228	58.4	/	41.5@1.55 V	/	26
Ir nanosheets		254	72.5	/	1200@1.55 V	0.25@1.56 V ^c	27
Mesoporous IrOx nanosheets		~250	47	/	~1300@1.55 V	/	28
IrO ₂ nanoneedles	1 M H ₂ SO ₄	313	57	1.55 V@0.03 mA cm _{ECSA} ⁻²	51.6@1.55 V	/	29
Ir nanodendrites	0.05 M H ₂ SO ₄	400	56	1.57	100@1.52V	/	30
3R-IrO ₂		188	52	/	690.4@1.50 V	5.7@1.50 V ^b	31
SrCo _{0.9} Ir _{0.1} O _{3-δ}		~290	/	1.56 V@10 mA cm _{BET} ⁻²	/	2.56@1.50 V ^a	32
IrNiCu double-layered nanoframe	0.1 M HClO ₄	~303	48	/	120@1.50 V	/	33
Y ₂ Ir ₂ O ₇		/	50	/	100@1.54 V	/	34
H _{3.6} IrO ₄ ·3.7H ₂ O		/	/	1.51	/	/	35
npIrx-NS		~300	~54	/	/	/	36
Amorphous Ir nanosheets		255	40	/	221.8 @1.53 V	0.16@1.53 V ^c	37

1 ^abased on the surface number sites calculated on BET.

2 ^bbased on the moles of surface metal atoms on ECSA.

3 ^cbased on the total number sites of mass.

4 ^dbased on the moles of surface metal atoms in mol.

5

1 **Supplementary Table 5. comparison of BET, ECSA and specific activity of IrO₂NR with**
 2 **previously reported Ir-based catalysts.**

Catalyst	BET / m ² g ⁻¹	Specific activity based on BET / mA cm _{BET} ⁻²	ECSA / m ² g ⁻¹	Specific activity based on ECSA / mA cm _{ECSA} ⁻²	Ref
IrO₂NR	47.3	1@1.48 V vs RHE 4.3@1.50 V vs RHE	31.3	1@1.47 V vs RHE 6.4@1.50 V vs RHE	This work
H-Ti@IrO _x	/	/	/	0.04@1.58 V vs RHE	7
Li-IrO _x	27	1@1.55 V vs RHE	/	~1@1.53 V vs RHE	11
6H-SrIrO ₃	0.3	~0.03@1.48 V vs RHE	/	/	16
1T-IrO ₂	21.9	~1.3@1.55 V vs RHE	15.4	~1.8@1.55 V vs RHE	2
IrO ₂ nanoneedles	196.5	0.031@1.55 V vs RHE	205.1	0.03@1.55 V vs RHE	29
Ir nanodendrites	39.2	1@1.57 V vs RHE	/	/	30
3R-IrO ₂	14.8	~4.2@1.5 V vs RHE	14.0	~4.21@1.5 V vs RHE	31
SrCo _{0.9} Ir _{0.1} O _{3-δ}	0.175	~1@1.48 V vs RHE	/	/	32
H _{3.6} IrO ₄ ·3.7H ₂ O	4.01	1@1.51 V vs RHE	/	/	35

3

1 **Supplementary Table 6. ICP-MS results of IrO₂NR under different current densities and**
 2 **potentials for 3600 s.**

	Current density / mA cm⁻²				Potential / V vs. RHE			
	10	20	50	100	1.40	1.45	1.50	1.55
Concentration of Ir (ppb)	0.107	0.255	0.616	1.378	0.009	0.067	0.362	1.128
nO ₂ (μmol)	94.2	186.6	467.0	948.8	7.1	47.9	258.4	689.1
S-number	160501 6	133467 1	138292 1	125587 4	162867 2	138545 4	136983 7	117258 2

3

1 **Supplementary Table 7. ZPVEs and entropic correction at 300 K.**

	H₂O	H₂	*O	*OH	*OOH
ZPVE (eV)	0.60	0.30	0.07	0.37	0.46
ΔS (eV)	0.59	0.41	0.09	0.12	0.16

2

1 **Supplementary Notes**

2 **Supplementary Note 1. The simulated coordinates for IrO₂NR.**

3 01-sub

4 O Ir

5	1.0000000000000000					
6	15.0000000000000000	0.0000000000000000	0.0000000000000000			
7	0.0000000000000000	5.4324110000000001	0.0000000000000000			
8	0.0000000000000000	0.0000000000000000	25.9223649999999992			

9 O Ir

10 19 8

11 Selective dynamics

12 Direct

13	0.41906500000000034	0.25794799999999990	0.02596400000000019	F	F	F
14	0.41906500000000034	0.92384899999999970	0.12480999999999965	F	F	F
15	0.4471198203822812	0.6545491573385520	0.2340191417309872	T	T	T
16	0.4733495942004957	0.4099674867967978	0.3457988609085443	T	T	T
17	0.4510552718603226	0.1375283621595616	0.4332698356881374	T	T	T
18	0.41906500000000034	0.75794799999999990	0.02596400000000019	F	F	F
19	0.41906500000000034	0.42384899999999970	0.12480999999999965	F	F	F
20	0.4422174904000905	0.1574790054584634	0.2325569340693466	T	T	T
21	0.4620263186325192	0.9051790564068785	0.3406259311725576	T	T	T
22	0.58093499999999966	0.08303399999999978	0.00000000000000000	F	F	F
23	0.58093499999999966	0.74893600000000005	0.09884600000000018	F	F	F
24	0.58093499999999966	0.41483800000000031	0.19769300000000010	F	F	F
25	0.5995550574035357	0.1514906889058718	0.3109581747931270	T	T	T
26	0.6251726828718004	0.9164016834594557	0.4147991366877084	T	T	T
27	0.58093499999999966	0.58303399999999978	0.00000000000000000	F	F	F
28	0.58093499999999966	0.24893600000000005	0.09884600000000018	F	F	F
29	0.58093499999999966	0.91483800000000031	0.19769200000000035	F	F	F
30	0.5992910936771227	0.6625385755164241	0.3085092669953869	T	T	T
31	0.6136816569837211	0.4032003522254856	0.4074041802867656	T	T	T
32	0.50000000000000000	0.50344199999999997	0.06240499999999983	F	F	F
33	0.50000000000000000	0.16934299999999978	0.16125100000000000	F	F	F
34	0.5220876905389951	0.9165325830657767	0.2704034770555377	T	T	T
35	0.5465663529252881	0.6799750677583399	0.3792005652002331	T	T	T
36	0.50000000000000000	0.00344199999999997	0.06240499999999983	F	F	F
37	0.50000000000000000	0.66934299999999978	0.16125100000000000	F	F	F
38	0.5240494336193062	0.4007618178081752	0.2718000858633810	T	T	T
39	0.5386840228307030	0.1397281512224765	0.3865812420588636	T	T	T

40

41 02-OH

42 H O Ir

43 1.0000000000000000

1	15.000000000000000	0.000000000000000	0.000000000000000			
2	0.000000000000000	5.432411000000001	0.000000000000000			
3	0.000000000000000	0.000000000000000	25.922364999999992			
4	H	O	Ir			
5	1	20	8			
6	Selective dynamics					
7	Direct					
8	0.4869046722266670	0.6796695669154855	0.4734471476813779	T	T	T
9	0.4190650000000034	0.2579479999999990	0.0259640000000019	F	F	F
10	0.4190650000000034	0.9238489999999970	0.1248099999999965	F	F	F
11	0.4493551104986169	0.6538115026585298	0.2352068924827673	T	T	T
12	0.4685265947716387	0.4032705463194092	0.3444880126933831	T	T	T
13	0.4460404200917793	0.1281450585898191	0.4336860409467891	T	T	T
14	0.4190650000000034	0.7579479999999990	0.0259640000000019	F	F	F
15	0.4190650000000034	0.4238489999999970	0.1248099999999965	F	F	F
16	0.4391142014764139	0.1564624534972288	0.2315223094660780	T	T	T
17	0.4559209689777723	0.9089744956201233	0.3396461914434053	T	T	T
18	0.5809349999999966	0.0830339999999978	0.0000000000000000	F	F	F
19	0.5809349999999966	0.7489360000000005	0.0988460000000018	F	F	F
20	0.5809349999999966	0.4148380000000031	0.1976930000000010	F	F	F
21	0.5991559407180789	0.1494965021496423	0.3113624108423823	T	T	T
22	0.6159069397987182	0.9099919555610447	0.4173677061722748	T	T	T
23	0.5809349999999966	0.5830339999999978	0.0000000000000000	F	F	F
24	0.5809349999999966	0.2489360000000005	0.0988460000000018	F	F	F
25	0.5809349999999966	0.9148380000000031	0.1976920000000035	F	F	F
26	0.5930621796413594	0.6606447617978193	0.3061233218646185	T	T	T
27	0.6056735869717875	0.4017952874591937	0.4077403919680206	T	T	T
28	0.4537724124361938	0.6764993969329485	0.4406407748655715	T	T	T
29	0.5000000000000000	0.5034419999999997	0.0624049999999983	F	F	F
30	0.5000000000000000	0.1693429999999978	0.1612510000000000	F	F	F
31	0.5201963175305339	0.9235384113552986	0.2702674644800035	T	T	T
32	0.5345665558677728	0.6780359439945117	0.3807995373832371	T	T	T
33	0.5000000000000000	0.0034419999999997	0.0624049999999983	F	F	F
34	0.5000000000000000	0.6693429999999978	0.1612510000000000	F	F	F
35	0.5228683837382202	0.3909902872523834	0.2716706941679512	T	T	T
36	0.5331267307575491	0.1336485933205033	0.3863388366432710	T	T	T
37						
38	03-O					
39	O Ir					
40	1.000000000000000					
41	15.000000000000000	0.000000000000000	0.000000000000000			
42	0.000000000000000	5.432411000000001	0.000000000000000			
43	0.000000000000000	0.000000000000000	25.922364999999992			
44	O	Ir				

1	20	8				
2	Selective dynamics					
3	Direct					
4	0.4190650000000034	0.2579479999999990	0.0259640000000019	F	F	F
5	0.4190650000000034	0.9238489999999970	0.1248099999999965	F	F	F
6	0.4506213435184632	0.6568840874230705	0.2358826528955412	T	T	T
7	0.4674781563964113	0.4017191746660626	0.3431075694857109	T	T	T
8	0.4573671385236630	0.1513213230981630	0.4375853242497879	T	T	T
9	0.4190650000000034	0.7579479999999990	0.0259640000000019	F	F	F
10	0.4190650000000034	0.4238489999999970	0.1248099999999965	F	F	F
11	0.4395338541121853	0.1561349346689531	0.2316806931647079	T	T	T
12	0.4624684101476215	0.9142992000138565	0.3413876735687739	T	T	T
13	0.4583247969381578	0.6599673983021526	0.4360430519211883	T	T	T
14	0.5809349999999966	0.0830339999999978	0.0000000000000000	F	F	F
15	0.5809349999999966	0.7489360000000005	0.0988460000000018	F	F	F
16	0.5809349999999966	0.4148380000000031	0.1976930000000010	F	F	F
17	0.6016265308471409	0.1554368935451549	0.3102450175213794	T	T	T
18	0.6223036866206140	0.9052844807512053	0.4086612247496957	T	T	T
19	0.5809349999999966	0.5830339999999978	0.0000000000000000	F	F	F
20	0.5809349999999966	0.2489360000000005	0.0988460000000018	F	F	F
21	0.5809349999999966	0.9148380000000031	0.1976920000000035	F	F	F
22	0.5909188039156626	0.6596605627479961	0.3051418404574474	T	T	T
23	0.6194287127516339	0.4086377767262041	0.4051088904614694	T	T	T
24	0.5000000000000000	0.5034419999999997	0.0624049999999983	F	F	F
25	0.5000000000000000	0.1693429999999978	0.1612510000000000	F	F	F
26	0.5217444420812252	0.9276692841804395	0.2709750305103393	T	T	T
27	0.5371201908335426	0.6632725168640359	0.3840989215630520	T	T	T
28	0.5000000000000000	0.0034419999999997	0.0624049999999983	F	F	F
29	0.5000000000000000	0.6693429999999978	0.1612510000000000	F	F	F
30	0.5224513120508993	0.3900244917680420	0.2712175215135299	T	T	T
31	0.5376182709382881	0.1512397974224304	0.3867738537579117	T	T	T

32						
33	04-OOH					
34	H O Ir					
35	1.0000000000000000					
36	15.0000000000000000	0.0000000000000000	0.0000000000000000			
37	0.0000000000000000	5.4324110000000001	0.0000000000000000			
38	0.0000000000000000	0.0000000000000000	25.9223649999999992			

39	H	O	Ir			
40	1	21	8			
41	Selective dynamics					
42	Direct					
43	0.4720329614738420	0.6464561274848877	0.5173192365208771	T	T	T
44	0.4190650000000034	0.2579479999999990	0.0259640000000019	F	F	F

1	0.4190650000000034	0.9238489999999970	0.1248099999999965	F	F	F
2	0.4480437220497948	0.6550524967693305	0.2344693105191512	T	T	T
3	0.4715672711257574	0.4091725904050069	0.3448848023231698	T	T	T
4	0.4494696992809179	0.1443702587225965	0.4328245565288801	T	T	T
5	0.4190650000000034	0.7579479999999990	0.0259640000000019	F	F	F
6	0.4190650000000034	0.4238489999999970	0.1248099999999965	F	F	F
7	0.4407871240435655	0.1582154797030442	0.2319743664960739	T	T	T
8	0.4602890615263362	0.9094940953218242	0.3402952235302220	T	T	T
9	0.5809349999999966	0.0830339999999978	0.0000000000000000	F	F	F
10	0.5809349999999966	0.7489360000000005	0.0988460000000018	F	F	F
11	0.5809349999999966	0.4148380000000031	0.1976930000000010	F	F	F
12	0.5997887931942437	0.1520144686175755	0.3108272481400659	T	T	T
13	0.6231902310863675	0.9174138219553138	0.4141971464961759	T	T	T
14	0.5809349999999966	0.5830339999999978	0.0000000000000000	F	F	F
15	0.5809349999999966	0.2489360000000005	0.0988460000000018	F	F	F
16	0.5809349999999966	0.9148380000000031	0.1976920000000035	F	F	F
17	0.5976084005880233	0.6642132219336677	0.3075249493112006	T	T	T
18	0.6135707822523041	0.4033430571898000	0.4065116086074406	T	T	T
19	0.5185462945748707	0.6465715389818154	0.4898291958197232	T	T	T
20	0.4678132550128926	0.6707265026463550	0.4454923420111937	T	T	T
21	0.5000000000000000	0.5034419999999997	0.0624049999999983	F	F	F
22	0.5000000000000000	0.1693429999999978	0.1612510000000000	F	F	F
23	0.5215581759085860	0.9199915717490112	0.2704639004679797	T	T	T
24	0.5410569529877362	0.6781303171646037	0.3811651948603809	T	T	T
25	0.5000000000000000	0.0034419999999997	0.0624049999999983	F	F	F
26	0.5000000000000000	0.6693429999999978	0.1612510000000000	F	F	F
27	0.5240942114161827	0.3988690320131885	0.2717791338671527	T	T	T
28	0.5375125698806702	0.1415826832235977	0.3860381793981448	T	T	T
29						
30						

1 **Supplementary Note 2. The simulated coordinates for Rutile IrO₂.**

2 01-sub

3 O Ir

4 1.0000000000000000

5 6.3790139999999997 0.0000000000000000 0.0000000000000000

6 0.0000000000000000 6.4275140000000004 0.0000000000000000

7 0.0000000000000000 0.0000000000000000 27.1048849999999995

8 O Ir

9 33 16

10 Selective dynamics

11 Direct

12 0.0000000000000000 0.5000000000000000 0.1127090000000024 F F F

13 0.0000000000000000 0.0000000000000000 0.2312760000000011 F F F

14 0.9998445680351926 0.5010017369335630 0.3521987881139923 T T T

15 0.9994965058634052 0.0016695960319556 0.4694566958742364 T T T

16 0.0000000000000000 0.5000000000000000 0.0218169999999986 F F F

17 0.0000000000000000 0.0000000000000000 0.1403839999999974 F F F

18 0.9998819793757620 0.5008972018637352 0.2594521381489161 T T T

19 0.9996259223036761 0.0019192165512308 0.3788791698653289 T T T

20 0.2500000000000000 0.8083540000000013 0.0672629999999970 F F F

21 0.2500000000000000 0.3083540000000013 0.1858300000000028 F F F

22 0.2494157646963543 0.8090788817774327 0.3055969149996448 T T T

23 0.2495907823033064 0.3106286536138581 0.4253387398246818 T T T

24 0.2500000000000000 0.1916459999999987 0.0672629999999970 F F F

25 0.2500000000000000 0.6916459999999987 0.1858300000000028 F F F

26 0.2493828154747827 0.1929000641092782 0.3054636508096125 T T T

27 0.2495815292875117 0.6930558945980061 0.4250517030320726 T T T

28 0.5000000000000000 0.5000000000000000 0.1127090000000024 F F F

29 0.5000000000000000 0.0000000000000000 0.2312760000000011 F F F

30 0.4998198059594037 0.5007391366584167 0.3505882334525952 T T T

31 0.4995544737545041 0.0014153080660952 0.4696174333714332 T T T

32 0.5000000000000000 0.5000000000000000 0.0218169999999986 F F F

33 0.5000000000000000 0.0000000000000000 0.1403839999999974 F F F

34 0.4998843037734757 0.5008953976136499 0.2596797790019997 T T T

35 0.4995913062981768 0.0018333004320948 0.3790363607518746 T T T

36 0.7500000000000000 0.8083540000000013 0.0672629999999970 F F F

37 0.7500000000000000 0.3083540000000013 0.1858300000000028 F F F

38 0.7502442101018467 0.8091002149315418 0.3056004516223246 T T T

39 0.7494529891548373 0.3106332557036335 0.4253109183090793 T T T

40 0.7500000000000000 0.1916459999999987 0.0672629999999970 F F F

41 0.7500000000000000 0.6916459999999987 0.1858300000000028 F F F

42 0.7502852598433740 0.1929070926387134 0.3054594489419337 T T T

43 0.7494504559824992 0.6930585580597578 0.4250282178903759 T T T

44 0.9992729854811668 0.5034711991814431 0.4964397147839072 T T T

1	0.2500000000000000	0.5000000000000000	0.0672629999999970	F	F	F
2	0.2500000000000000	0.0000000000000000	0.1858300000000028	F	F	F
3	0.2475069015541578	0.5009915241560096	0.3054876608785878	T	T	T
4	0.2492901012064446	0.0017640358393325	0.4252172614315575	T	T	T
5	0.0000000000000000	0.0000000000000000	0.0672629999999970	F	F	F
6	0.0000000000000000	0.5000000000000000	0.1858300000000028	F	F	F
7	0.9998254840168888	0.0009951528151384	0.3055600425645030	T	T	T
8	0.9995095101943213	0.5019398334215505	0.4296801104279055	T	T	T
9	0.7500000000000000	0.5000000000000000	0.0672629999999970	F	F	F
10	0.7500000000000000	0.0000000000000000	0.1858300000000028	F	F	F
11	0.7522189572992751	0.5009991211868521	0.3054923822539625	T	T	T
12	0.7498078830423235	0.0017579660655349	0.4252075578576048	T	T	T
13	0.5000000000000000	0.0000000000000000	0.0672629999999970	F	F	F
14	0.5000000000000000	0.5000000000000000	0.1858300000000028	F	F	F
15	0.4998190680808338	0.0009988354452719	0.3056387029642698	T	T	T
16	0.4995109677516775	0.5018519584487785	0.4217974338177495	T	T	T

17

18 02-OH

19 H O Ir

20	1.0000000000000000					
21	6.3790139999999997	0.0000000000000000	0.0000000000000000			
22	0.0000000000000000	6.4275140000000004	0.0000000000000000			
23	0.0000000000000000	0.0000000000000000	27.1048849999999995			

24 H O Ir
25 1 34 16

26 Selective dynamics

27 Direct

28	0.6338218103958381	0.5067305354378833	0.5109118083149641	T	T	T
29	0.0000000000000000	0.5000000000000000	0.1127090000000024	F	F	F
30	0.0000000000000000	0.0000000000000000	0.2312760000000011	F	F	F
31	0.0002115946427766	0.5006692981053429	0.3518085521966868	T	T	T
32	-0.0006099061285491	0.0012056575053149	0.4693501256153819	T	T	T
33	0.0000000000000000	0.5000000000000000	0.0218169999999986	F	F	F
34	0.0000000000000000	0.0000000000000000	0.1403839999999974	F	F	F
35	-0.0001616623193010	0.5005377364994646	0.2597714134179485	T	T	T
36	-0.0001995536195784	0.0014122165233374	0.3789067445012633	T	T	T
37	0.2500000000000000	0.8083540000000013	0.0672629999999970	F	F	F
38	0.2500000000000000	0.3083540000000013	0.1858300000000028	F	F	F
39	0.2498668038117612	0.8087345875183251	0.3053588169655004	T	T	T
40	0.2491681853642929	0.3099691856292135	0.4247271754293133	T	T	T
41	0.2500000000000000	0.1916459999999987	0.0672629999999970	F	F	F
42	0.2500000000000000	0.6916459999999987	0.1858300000000028	F	F	F
43	0.2498793038273987	0.1926416531991275	0.3052695071543272	T	T	T
44	0.2492589770498049	0.6928085795571646	0.4244340201172746	T	T	T

1	0.5000000000000000	0.5000000000000000	0.1127090000000024	F	F	F
2	0.5000000000000000	0.0000000000000000	0.2312760000000011	F	F	F
3	0.4996246563607662	0.5007058506002751	0.3512755880197697	T	T	T
4	0.5000965351963679	0.0011627109843875	0.4692013539045547	T	T	T
5	0.5000000000000000	0.5000000000000000	0.0218169999999986	F	F	F
6	0.5000000000000000	0.0000000000000000	0.1403839999999974	F	F	F
7	0.5001122421505795	0.5006116887766602	0.2596781999178207	T	T	T
8	0.4995845123596138	0.0014303246404307	0.3789413676719597	T	T	T
9	0.7500000000000000	0.8083540000000013	0.0672629999999970	F	F	F
10	0.7500000000000000	0.3083540000000013	0.1858300000000028	F	F	F
11	0.7500473298671265	0.8087843372403861	0.3053729732162634	T	T	T
12	0.7503705996796457	0.3097062221571728	0.4240462945554899	T	T	T
13	0.7500000000000000	0.1916459999999987	0.0672629999999970	F	F	F
14	0.7500000000000000	0.6916459999999987	0.1858300000000028	F	F	F
15	0.7500577850612676	0.1926522401496229	0.3052822621686517	T	T	T
16	0.7504789256230621	0.6930821964265689	0.4238444840446888	T	T	T
17	0.9919481473566180	0.5024887867277223	0.4953921166567681	T	T	T
18	0.4916258834264045	0.5026283776001284	0.4972301991872556	T	T	T
19	0.2500000000000000	0.5000000000000000	0.0672629999999970	F	F	F
20	0.2500000000000000	0.0000000000000000	0.1858300000000028	F	F	F
21	0.2496574493718979	0.5006550766711961	0.3055484969244238	T	T	T
22	0.2493632091686847	0.0013633269078379	0.4252728188657499	T	T	T
23	0.0000000000000000	0.0000000000000000	0.0672629999999970	F	F	F
24	0.0000000000000000	0.5000000000000000	0.1858300000000028	F	F	F
25	-0.0000243257944562	0.0007067859798631	0.3055622792645232	T	T	T
26	0.0003751045719662	0.5014558878886503	0.4286777017602615	T	T	T
27	0.7500000000000000	0.5000000000000000	0.0672629999999970	F	F	F
28	0.7500000000000000	0.0000000000000000	0.1858300000000028	F	F	F
29	0.7503421554100849	0.5007302908895575	0.3056135304513852	T	T	T
30	0.7499413235432304	0.0013226072190987	0.4251436777818714	T	T	T
31	0.5000000000000000	0.0000000000000000	0.0672629999999970	F	F	F
32	0.5000000000000000	0.5000000000000000	0.1858300000000028	F	F	F
33	0.4999342224940794	0.0007045927376406	0.3055960218412249	T	T	T
34	0.5004319618172495	0.5013785617706408	0.4252213354183028	T	T	T

35

36 03-O

37 O Ir

38	1.0000000000000000		
39	6.3790139999999997	0.0000000000000000	0.0000000000000000
40	0.0000000000000000	6.4275140000000004	0.0000000000000000
41	0.0000000000000000	0.0000000000000000	27.1048849999999995

42 O Ir

43 34 16

44 Selective dynamics

1	Direct					
2	0.0000000000000000	0.5000000000000000	0.1127090000000024	F	F	F
3	0.0000000000000000	0.0000000000000000	0.2312760000000011	F	F	F
4	0.9998689644201714	0.5009002032862475	0.3519682429236804	T	T	T
5	0.9993344891123284	0.0018118839891236	0.4692436650365079	T	T	T
6	0.0000000000000000	0.5000000000000000	0.0218169999999986	F	F	F
7	0.0000000000000000	0.0000000000000000	0.1403839999999974	F	F	F
8	0.9999333076180618	0.5008703815970384	0.2598442601605263	T	T	T
9	0.9995413867453676	0.0020125748078738	0.3788676535902920	T	T	T
10	0.2500000000000000	0.8083540000000013	0.0672629999999970	F	F	F
11	0.2500000000000000	0.3083540000000013	0.1858300000000028	F	F	F
12	0.2499127295844903	0.8088026942427462	0.3054991807877735	T	T	T
13	0.2494234004703997	0.3105887187729434	0.4243089340770135	T	T	T
14	0.2500000000000000	0.1916459999999987	0.0672629999999970	F	F	F
15	0.2500000000000000	0.6916459999999987	0.1858300000000028	F	F	F
16	0.2499135989388178	0.1932366051173780	0.3053566572894766	T	T	T
17	0.2494245778724460	0.6934107911073747	0.4239175605647564	T	T	T
18	0.5000000000000000	0.5000000000000000	0.1127090000000024	F	F	F
19	0.5000000000000000	0.0000000000000000	0.2312760000000011	F	F	F
20	0.4998618233891925	0.5009105452674546	0.3519674233019505	T	T	T
21	0.4993441147968616	0.0018272829415091	0.4692435260638592	T	T	T
22	0.5000000000000000	0.5000000000000000	0.0218169999999986	F	F	F
23	0.5000000000000000	0.0000000000000000	0.1403839999999974	F	F	F
24	0.4999336059489501	0.5008698334069562	0.2598441626926167	T	T	T
25	0.4995363575278933	0.0020289141258509	0.3788676927922908	T	T	T
26	0.7500000000000000	0.8083540000000013	0.0672629999999970	F	F	F
27	0.7500000000000000	0.3083540000000013	0.1858300000000028	F	F	F
28	0.7499129015292626	0.8088032403638902	0.3054996207517020	T	T	T
29	0.7494394270523926	0.3105891976385393	0.4243026078236614	T	T	T
30	0.7500000000000000	0.1916459999999987	0.0672629999999970	F	F	F
31	0.7500000000000000	0.6916459999999987	0.1858300000000028	F	F	F
32	0.7499134914126016	0.1932366406207862	0.3053565385032874	T	T	T
33	0.7494132118942283	0.6934129428582144	0.4239109108329677	T	T	T
34	0.9993185444826654	0.5037874790445082	0.4953276145215443	T	T	T
35	0.4993734011744317	0.5036879808826554	0.4953242154944987	T	T	T
36	0.2500000000000000	0.5000000000000000	0.0672629999999970	F	F	F
37	0.2500000000000000	0.0000000000000000	0.1858300000000028	F	F	F
38	0.2499169030561023	0.5009770799872626	0.3057677369430566	T	T	T
39	0.2494291759223654	0.0019432604995545	0.4251638409715845	T	T	T
40	0.0000000000000000	0.0000000000000000	0.0672629999999970	F	F	F
41	0.0000000000000000	0.5000000000000000	0.1858300000000028	F	F	F
42	0.9998892883281378	0.0010193371458187	0.3056226939312163	T	T	T
43	0.9994073756976700	0.5021101886134252	0.4286552763636916	T	T	T
44	0.7500000000000000	0.5000000000000000	0.0672629999999970	F	F	F

1	0.7500000000000000	0.0000000000000000	0.1858300000000028	F	F	F
2	0.7499182810285098	0.5009796978011276	0.3057687879354140	T	T	T
3	0.7494295141586830	0.0019447972812111	0.4251619733868104	T	T	T
4	0.5000000000000000	0.0000000000000000	0.0672629999999970	F	F	F
5	0.5000000000000000	0.5000000000000000	0.1858300000000028	F	F	F
6	0.4998887467097261	0.0010207903689170	0.3056225398537408	T	T	T
7	0.4994103929723537	0.5021007607696568	0.4286544999212406	T	T	T
8						
9	04-OOH					
10	H O Ir					
11	1.0000000000000000					
12	6.3790139999999997	0.0000000000000000	0.0000000000000000			
13	0.0000000000000000	6.4275140000000004	0.0000000000000000			
14	0.0000000000000000	0.0000000000000000	27.1048849999999995			
15	H O Ir					
16	1	35	16			
17	Selective dynamics					
18	Direct					
19	0.5297212590431493	0.4177615828041949	0.5609517968202882	T	T	T
20	0.0000000000000000	0.5000000000000000	0.1127090000000024	F	F	F
21	0.0000000000000000	0.0000000000000000	0.2312760000000011	F	F	F
22	-0.0003451184592728	0.4944432468660376	0.3519644386127907	T	T	T
23	0.0001882990651607	-0.0054455271463742	0.4691720724593572	T	T	T
24	0.0000000000000000	0.5000000000000000	0.0218169999999986	F	F	F
25	0.0000000000000000	0.0000000000000000	0.1403839999999974	F	F	F
26	0.0001276516495146	0.4987379466838284	0.2597312933866688	T	T	T
27	-0.0007298965815942	-0.0027922360552167	0.3787117324938758	T	T	T
28	0.2500000000000000	0.8083540000000013	0.0672629999999970	F	F	F
29	0.2500000000000000	0.3083540000000013	0.1858300000000028	F	F	F
30	0.2498096361876740	0.8055174447751586	0.3054163506269030	T	T	T
31	0.2510967212646291	0.3038137292842731	0.4248298644130823	T	T	T
32	0.2500000000000000	0.1916459999999987	0.0672629999999970	F	F	F
33	0.2500000000000000	0.6916459999999987	0.1858300000000028	F	F	F
34	0.2498905817199911	0.1899157653294576	0.3053085905288905	T	T	T
35	0.2472976696229873	0.6880587838577156	0.4228715500751000	T	T	T
36	0.5000000000000000	0.5000000000000000	0.1127090000000024	F	F	F
37	0.5000000000000000	0.0000000000000000	0.2312760000000011	F	F	F
38	0.5005286551200049	0.4945350070325660	0.3515087287865385	T	T	T
39	0.4985998470488681	-0.0108192852744165	0.4689134234697506	T	T	T
40	0.5000000000000000	0.5000000000000000	0.0218169999999986	F	F	F
41	0.5000000000000000	0.0000000000000000	0.1403839999999974	F	F	F
42	0.4999508058355370	0.4988785258575315	0.2597686749591206	T	T	T
43	0.5000766116819096	-0.0033615787454754	0.3787958157552416	T	T	T
44	0.7500000000000000	0.8083540000000013	0.0672629999999970	F	F	F

1	0.7500000000000000	0.3083540000000013	0.1858300000000028	F	F	F
2	0.7500462242857603	0.8049686590591850	0.3053428337911113	T	T	T
3	0.7505304900937947	0.3037502268207589	0.4262380358970056	T	T	T
4	0.7500000000000000	0.1916459999999987	0.0672629999999970	F	F	F
5	0.7500000000000000	0.6916459999999987	0.1858300000000028	F	F	F
6	0.7501131133095940	0.1893299063095829	0.3052318979352714	T	T	T
7	0.7494399131834137	0.6866586398525100	0.4238359169425996	T	T	T
8	0.0070337350979972	0.5026060251618678	0.4955031256750325	T	T	T
9	0.5658966608529270	0.3662788821623024	0.5277136118529102	T	T	T
10	0.4762856737701269	0.5268733806416662	0.4981488362517004	T	T	T
11	0.2500000000000000	0.5000000000000000	0.0672629999999970	F	F	F
12	0.2500000000000000	0.0000000000000000	0.1858300000000028	F	F	F
13	0.2495569599619835	0.4978797297798127	0.3056818313388325	T	T	T
14	0.2494960075735005	-0.0046518753176292	0.4247810136447266	T	T	T
15	0.0000000000000000	0.0000000000000000	0.0672629999999970	F	F	F
16	0.0000000000000000	0.5000000000000000	0.1858300000000028	F	F	F
17	-0.0000968684984893	-0.0024551339087700	0.3054452958349328	T	T	T
18	-0.0002363537065004	0.4958766678608142	0.4288260701893173	T	T	T
19	0.7500000000000000	0.5000000000000000	0.0672629999999970	F	F	F
20	0.7500000000000000	0.0000000000000000	0.1858300000000028	F	F	F
21	0.7505418782974969	0.4967459283923896	0.3056142407824793	T	T	T
22	0.7497601130295259	-0.0042913429674015	0.4251003239125429	T	T	T
23	0.5000000000000000	0.0000000000000000	0.0672629999999970	F	F	F
24	0.5000000000000000	0.5000000000000000	0.1858300000000028	F	F	F
25	0.4999895325035672	-0.0024098543549113	0.3055359256907156	T	T	T
26	0.4988734466362944	0.4958633544827046	0.4266112243991225	T	T	T
27						

1 **Supplementary Note 3. The code input file.**

2

3 SYSTEM = Opt

4

5 ISYM = 0

6 ISPIN = 2

7

8 PREC = Normal

9 ALGO = FAST

10

11 ENCUT = 520

12 EDIFF = 1E-04

13 EDIFFG = -0.03

14

15 ISMEAR = 0

16 SIGMA = 0.05

17

18 LREAL = Auto

19 LWAVE = .FALSE.

20 LCHARG = .FALSE.

21

22 NSW = 999

23 IBRION = 2

24

25 GGA = RP

26

27

1 **Supplementary References**

- 2 1. Geiger, S., Kasian, O., Ledendecker, M., Pizzutilo, E., Mingers, A. M., Fu, W. T., Diaz-Morales,
3 O., Li, Z. Z., Oellers, T., Fruchter, L., Ludwig, A., Mayrhofer, K. J. J., Koper, M. T. M. & Cherevko,
4 S. The stability number as a metric for electrocatalyst stability benchmarking. *Nat. Catal.* **1**, 508-515
5 (2018).
- 6 2. Dang, Q., Lin, H. P., Fan, Z. L., Ma, L., Shao, Q., Ji, Y. J., Zheng, F. F., Geng, S. Z., Yang, S. Z.,
7 Kong, N. N., Zhu, W. X., Li, Y. Y., Liao, F., Huang, X. Q. & Shao, M. W. Iridium metallene oxide for
8 acidic oxygen evolution catalysis. *Nat. Commun.* **12**, 6007 (2021).
- 9 3. da Silva, C. D. F., Claudel, F., Martin, V., Chattot, R., Abbou, S., Kumar, K., Jiménez-Morales, I.,
10 Cavaliere, S., Jones, D., Rozière, J., Solà-Hernandez, L., Beauger, C., Faustini, M., Peron, J., Gilles,
11 B., Encinas, T., Piccolo, L., Barros de Lima, F. H., Dubau, L. & Maillard, F. Oxygen evolution
12 reaction activity and stability benchmarks for supported and unsupported IrO_x electrocatalysts. *ACS*
13 *Catal.* **11**, 4107-4116 (2021).
- 14 4. Lončar, A., Escalera-López, D., Ruiz-Zepeda, F., Hrnjić, A., Šala, M., Jovanovič, P., Bele, M.,
15 Cherevko, S. & Hodnik, N. Sacrificial Cu layer mediated the formation of an active and stable
16 supported iridium oxygen evolution reaction electrocatalyst. *ACS Catal.* **11**, 12510-12519 (2021).
- 17 5. Pham, C. V., Bühler, M., Knöppel, J., Bierling, M., Seeberger, D., Escalera-López, D., Mayrhofer,
18 K. J. J., Cherevko, S. & Thiele, S. IrO₂ coated TiO₂ core-shell microparticles advance performance of
19 low loading proton exchange membrane water electrolyzers. *Appl. Catal. B: Environ.* **269**, 118762
20 (2020).
- 21 6. Liu, N., Duan, Z. Y., Zhang, Q. Q. & Guan, J. Q. Insights into active species of ultrafine iridium
22 oxide nanoparticle electrocatalysts in hydrogen/oxygen evolution reactions. *Chem. Eng. J.* **419**,
23 129567 (2021).
- 24 7. Yu, Z. P., Xu, J. Y., Li, Y. F., Wei, B., Zhang, N., Li, Y., Bondarchuk, O., Miao, H. W., Araujo, A.,
25 Wang, Z. C., Faria, J. L., Liu, Y. Y. & Liu, L. F. Ultrafine oxygen-defective iridium oxide
26 nanoclusters for efficient and durable water oxidation at high current densities in acidic media. *J.*
27 *Mater. Chem. A* **8**, 24743-24751 (2020).
- 28 8. Xu, J. Y., Lian, Z., Wei, B., Li, Y., Bondarchuk, O., Zhang, N., Yu, Z. P., Araujo, A., Amorim, I.,
29 Wang, Z. C., Li, B. & Liu, L. F. Strong electronic coupling between ultrafine iridium-ruthenium

- 1 nanoclusters and conductive, acid-stable tellurium nanoparticle support for efficient and durable
2 oxygen evolution in acidic and neutral media. *ACS Catal.* **10**, 3571-3579 (2020).
- 3 9. Zhuang, Z. W., Wang, Y., Xu, C. Q., Liu, S. J., Chen, C., Peng, Q., Zhuang, Z. B., Xiao, H., Pan, Y.,
4 Lu, S. Q. Yu, R., Cheong, W. C., Cao, X., Wu, K. L., Sun, K. A., Wang, Y., Wang, D. S., Li, J. & Li,
5 Y. D. Three-dimensional open nano-netcage electrocatalysts for efficient pH-universal overall water
6 splitting. *Nat. Commun.* **10**, 4875 (2019).
- 7 10. Guo, H. Y., Fang, Z. W., Li, H., Fernandez, D., Henkelman, G., Humphrey, S. M. & Yu, G. H.
8 Rational design of rhodium-iridium alloy nanoparticles as highly active catalysts for acidic oxygen
9 evolution. *ACS Nano* **13**, 13225-13234 (2019).
- 10 11. Gao, J. J., Xu, C. Q., Hung, S. F., Liu, W., Cai, W. Z., Zeng, Z. P., Jia, C. M., Chen, H. M., Xiao,
11 H., Li, J., Huang, Y. Q. & Liu, B. Breaking long-range order in iridium oxide by alkali ion for
12 efficient water oxidation. *J. Am. Chem. Soc.* **141**, 3014-3023 (2019).
- 13 12. Meng, G., Sun, W. M., Mon, A. A., Wu, X., Xia, L. Y., Han, A. J., Wang, Y., Zhuang, Z. B., Liu, J.
14 F., Wang, D. S. & Li, Y. D. Strain regulation to optimize the acidic water oxidation performance of
15 atomic-layer IrO_x. *Adv. Mater.* **31**, 1903616 (2019).
- 16 13. Chen, J. Y., Cui, P. X., Zhao, G. Q., Rui, K., Lao, M. M., Chen, Y. P., Zheng, X. S., Jiang, Y. Z.,
17 Pan, H. G., Dou, S. X. & Sun, W. P. Low-coordinate iridium oxide confined on graphitic carbon
18 nitride for highly efficient oxygen evolution. *Angew. Chem. Int. Ed.* **58**, 12540-12544 (2019).
- 19 14. Shan, J. Q., Guo, C. X., Zhu, Y. H., Chen, S. M., Song, L., Jaroniec, M., Zheng, Y. & Qiao, S. Z.
20 Charge-redistribution-enhanced nanocrystalline Ru@IrO_x electrocatalysts for oxygen evolution in
21 acidic media. *Chem* **5**, 445-459 (2019).
- 22 15. Jiang, B., Guo, Y. N., Kim, J., Whitten, A. E., Wook, K., Kani, K., Rowan, A. E., Henzie, J. &
23 Yamauchi, Y. Mesoporous metallic iridium nanosheets. *J. Am. Chem. Soc.* **140**, 12434-12441 (2018).
- 24 16. Yang, L., Yu, G. T., Ai, X., Yan, W. S., Duan, H. L., Chen, W., Li, X. T., Wang, T., Zhang, C. H.,
25 Huang, X. R., Chen, J. S. & Zou, X. X. Efficient oxygen evolution electrocatalysis in acid by a
26 perovskite with face-sharing IrO₆ octahedral dimers. *Nat. Commun.* **9**, 5236 (2018).
- 27 17. Zhang, J., Wang, G., Liao, Z. Q., Zhang, P. P., Wang, F. X., Zhuang, X. D., Zschech, E. & Feng,
28 X. L. Iridium nanoparticles anchored on 3D graphite foam as a bifunctional electrocatalyst for
29 excellent overall water splitting in acidic solution. *Nano Energy* **40**, 27-33 (2017).
- 30 18. Guan, J. Q., Li, D., Si, R., Miao, S., Zhang, F. X. & Li, C. Synthesis and demonstration of

- 1 subnanometric iridium oxide as highly efficient and robust water oxidation catalyst. *ACS Catal.* **7**,
2 5983-5986 (2017).
- 3 19. Feng, J. R., Lv, F., Zhuang, W. Y., Li, P. H., Wang, K., Yang, C., Wang, B., Yang, Y., Zhou, J. H.
4 & Lin, F. Iridium-based multimetallic porous hollow nanocrystals for efficient overall-water-splitting
5 catalysis. *Adv. Mater.* **29**, 1703798 (2017).
- 6 20. Li, G. Q., Li, S. T., Ge, J. J., Liu, C. P. & Xing, W. Discontinuously covered IrO₂-RuO₂@Ru
7 electrocatalysts for the oxygen evolution reaction: how high activity and long-term durability can be
8 simultaneously realized in the synergistic and hybrid nano-structure. *J. Mater. Chem. A* **5**,
9 17221-17229 (2017).
- 10 21. Lettenmeier, P., Wang, L., Golla-Schindler, U., Gazdzicki, P., Canas, N. A., Handl, M., Hiesgen,
11 R., Hosseiny, S. S., Gago, A. S. & Friedrich, K. A. Nanosized IrO_x-Ir catalyst with relevant activity
12 for anodes of proton exchange membrane electrolysis produced by a cost-effective procedure. *Angew.
13 Chem. Int. Ed.* **55**, 742-746 (2016).
- 14 22. Seitz, L. C., Dickens, C. F., Nishio, K., Hikita, Y., Montoya, J., Doyle, A., Kirk, C., Vojvodic, A.,
15 Hwang, H. Y. & Nørskov, J. k. A highly active and stable IrO_x/SrIrO₃ catalyst for the oxygen
16 evolution reaction. *Science* **353**, 1011-1014 (2016).
- 17 23. Nong, H. N., Reier, T., Oh, H. S., Gliech, M., Paciok, P., Vu, T. H. T., Teschner, D., Heggen, M.,
18 Petkov, V., Schlogl, R., Jones, T. & Strasser, P. A unique oxygen ligand environment facilitates water
19 oxidation in hole-doped IrNiO_x core-shell electrocatalysts. *Nat. Catal.* **1**, 841-851 (2018).
- 20 24. Grimaud, A., Demortiere, A., Saubanere, M., Dachraoui, W., Duchamp, M., Doublet, M. L. &
21 Tarascon, J. M. Activation of surface oxygen sites on an iridium-based model catalyst for the oxygen
22 evolution reaction. *Nat. Energy* **2**, 16189 (2017).
- 23 25. Cheng, Z. F., Huang, B. L., Pi, Y. C., Li, L. G., Shao, Q. & Huang, X. Q. Partially hydroxylated
24 ultrathin iridium nanosheets as efficient electrocatalysts for water splitting. *Natl. Sci. Rev.* **7**,
25 1340-1348 (2020).
- 26 26. You, M. S., Gui, L. Q., Ma, X., Wang, Z. B., Xu, Y., Zhang, J., Sun, J., He, B. B. & Zhao, L.
27 Electronic tuning of SrIrO₃ perovskite nanosheets by sulfur incorporation to induce highly efficient
28 and long-lasting oxygen evolution in acidic media. *Appl. Catal. B: Environ.* **298**, 120562 (2021).
- 29 27. Xie, Y. H., Long, X., Li, X. W., Chang, C. F., Qu, K. G. & Yang, Z. H. The template synthesis of
30 ultrathin metallic Ir nanosheets as a robust electrocatalyst for acidic water splitting. *Chem. Commun.*

- 1 **57**, 8620-8623 (2021).
- 2 28. Jiang, B., Kim, J., Guo, Y. N., Wu, K. C. W., Alshehri, S. M., Ahamad, T., Alhokbany, N., Henzie,
3 J. & Yamachi, Y. Efficient oxygen evolution on mesoporous IrO_x nanosheets. *Catal. Sci. Technol.* **9**,
4 3697-3702 (2019).
- 5 29. Lim, J., Park, D., Jeon, S. S., Roh, C. W., Choi, J., Yoon, D., Park, M., Jung, H. & Lee, H.
6 Ultrathin IrO₂ nanoneedles for electrochemical water oxidation. *Adv. Funct. Mater.* **28**, 1704796
7 (2018).
- 8 30. Oh, H. S., Nong, H. N., Reier, T., Gliech, M. & Strasser, P. Oxide-supported Ir nanodendrites
9 with high activity and durability for the oxygen evolution reaction in acid PEM water electrolyzers.
10 *Chem.Sci.* **6**, 3321-3328 (2015).
- 11 31. Fan, Z. L., Ji, Y. J., Shao, Q., Geng, S. Z., Zhu, W. X., Liu, Y., Liao, F., Hu, Z. W., Chang, Y. C.,
12 Pao, C. W., Li, Y. Y., Kang, Z. H. & Shao, M. W. Extraordinary acidic oxygen evolution on new
13 phase 3R-iridium oxide. *Joule* **5**, 3221-3234 (2021).
- 14 32. Chen, Y. B., Li, H. Y., Wang, J. X., Du, Y. H., Xi, S. B., Sun, Y. M., Sherburne, M., Ager, J. M.,
15 Fisher, A. C. & Xu, Z. C. J. Exceptionally active iridium evolved from a pseudocubic perovskite for
16 oxygen evolution in acid. *Nat. Commun.* **10**, 572 (2019).
- 17 33. Park, J., Sa, Y. J., Baik, H., Kwon, T., Joo, S. H. & Lee, K. Iridium-based multimetallic
18 nanoframe@nanoframe structure: an efficient and robust electrocatalyst toward oxygen evolution
19 reaction. *ACS Nano* **11**, 5500-5509 (2017).
- 20 34. Lebedev, D., Povia, M., Waltar, K., Abdala, P. M., Castelli, I. E., Fabbri, E., Blanco, M. V.,
21 Fedorov, A., Coperet, C., Marzari, N. & Schmidt, T. J. Highly active and stable iridium pyrochlores
22 for oxygen evolution reaction. *Chem. Mater.* **29**, 5182-5191 (2017).
- 23 35. Zhang, R. H., Pearce, P. E., Pimenta, V., Cabana, J., Li, H. F., Dalla Corte, D. A., Abakumov, A.
24 M., Rouse, G., Giaume, D., Deschamps, M. & Grimaud, A. First example of protonation of
25 ruddlesden–popper Sr₂IrO₄: a route to enhanced water oxidation catalysts. *Chem. Mater.* **32**,
26 3499-3509 (2020).
- 27 36. Chatterjee, S., Peng, X., Intikhab, S., Zeng, G. S., Kariuki, N. N., Myers, D. J., Danilovic, N. &
28 Snyder, J. Nanoporous iridium nanosheets for polymer electrolyte membrane electrolysis. *Adv.*
29 *Energy Mater.* **11**, 2101438 (2021).
- 30 37. Wu, G., Zheng, X. S., Cui, P. X., Jiang, H. Y., Wang, X. Q., Qu, Y. T., Chen, W. X., Lin, Y., Li, H.,

- 1 Han, X., Hu, Y. M., Liu, P. G., Zhang, Q. H., Ge, J. J., Yao, Y. C., Sun, R. B., Wu, Y. E., Gu, L., Hong,
- 2 X. & Li, Y. D. A general synthesis approach for amorphous noble metal nanosheets. *Nat. Commun.*
- 3 **10**, 4855 (2019).